Rules of Stereoselectivity in Tandem Oxidative Polycyclization Reaction with Rhenium(VII) Oxides

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The tandem oxidative polycyclization reaction with rhenium-(VII) reagents, first reported in 1995,1 represents a powerful methodology by which polyene alcohols can be converted into poly-THF products with very high diastereoselectivity in a single step.

Kennedy's pioneering work on the monocyclization reaction with simple bis-homoallylic alcohols has suggested that the stereochemical outcome of this reaction leads consistently to trans-THF products.² We have confirmed this general rule in our early studies with mono- and also with a few bis-cyclization reactions, ^{1,3} and similar results were also reported by McDonald.⁴ Therefore. we were surprised to observe the exclusive formation of 2a, rather than its expected isomer 3a, in the triple oxidative cyclization reaction with 1a and trifluoroacetylperrhenate (CF₃CO₂ReO₃) (Scheme 1).5 What we discovered was inconsistent with the findings of an independent study by McDonald who reported that the reaction of very similar trienol substrates, 1b and 1c, with the same oxidant, CF₃CO₂ReO₃, afforded the all-trans products, 3b and 3c, respectively.6

Considering the immense synthetic importance of the tandem oxidative polycyclization reaction, we felt that the discrepancy between ours and McDonald's results had to be resolved. More importantly, to allow the use of this reaction in a stereochemically predictable way, one must understand the stereochemical relationship between the polyenol substrate and the poly-THF product. Here, on the basis of a systematic study, we confirm that the polycyclization of 1 with CF₃CO₂ReO₃ produces 2 and not 3. Moreover, we propose a set of rules for predicting the stereochemistry of the poly-THF products obtained by tandem oxidative cyclization reaction with CF₃CO₂ReO₃.⁷

The relative and absolute stereochemistries of the triple cyclization product 2a were elucidated by carrying out the reaction in a stepwise manner (Scheme 2). Partial oxidative cyclization of 1a with CF₃CO₂ReO₃ afforded a mixture of the mono- and bis-THF products, 4a and 5a, along with recovered starting material. The reaction of the mixture of 4a and 5a with an additional amount of CF₃CO₂ReO₃ afforded 2a. The stereochem-

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Scheme 1. Oxidative Cyclization of 1 with CF₃CO₂ReO₃

Scheme 2. Stepwise Oxidative Polycyclization of 1a with CF₃CO₂ReO₃a

1a
$$\stackrel{a}{\underset{R}{\longrightarrow}}$$
 $\stackrel{PO}{\underset{A}{\longrightarrow}}$ 4 $\stackrel{A}{\underset{R}{\longrightarrow}}$ 2a $\stackrel{B}{\underset{R}{\longrightarrow}}$ 2 $\stackrel{A}{\underset{R}{\longrightarrow}}$ \stackrel

^a Prepared in situ from 1 equiv of Re₂O₇ and 1.2 equiv of TFAA. Key: (a) CF₃CO₂ReO₃ (2 equiv), TFAA (2.6 equiv), CH₂Cl₂, 2 h, (Yield: 4a, 11%; 5a, 28%, recovered 1a, 37%); (b) CF₃CO₂ReO₃ (1.5 equiv), TFAA (2 equiv), CH₂Cl₂, 6 h; (c) (S)-PhC(OMe)(CF₃)COCl, DMAP, CH₂Cl₂; (d) (R)-PhC(OMe)(CF₃)COCl, DMAP, CH₂Cl₂.

Scheme 3. Synthesis of **2c** and **3c** from L-(+)-diethyl tartrate^a

^a Key: (a) i. TsOH, MeOH-H₂O, rt, 16 h. ii. MsCl, Et₃N, CH₂Cl₂, 0 °C, 2 h. (b) i. AD-mix- β , MeSO₂NH₂, tert-BuOH-H₂O, 18 h. ii. Pyridine, reflux, 2 h. (c) i. AD-mix-α, MeSO₂NH₂, tert-BuOH-H₂O, 18 h. ii. Pyridine, reflux, 2 h.

istry of the free hydroxyl group in 2a, 4a, and 5a was determined on the basis of 19 F NMR spectral data of their (R) and (S) Mosher's esters (see Supporting Information).8 The structure of 2a was further corroborated by 2D ¹H-¹H COSY, TOCSY, and ROESY experiments with the bis-benzoate ester of 2a. Finally, we prepared both compounds 2a9 and 3a10 by independent asymmetric synthesis. These experiments confirmed unequivocally that the tris-THF product obtained by the tandem oxidative cyclization of 1a is 2a and not 3a.

Compound 2b was easily prepared from 2a via a three-step sequence, desilation to produce corresponding diol, selective monotosylation of the primary alcohol, and LAH reductive cleavage of the tosylate function. Compounds 2c and 3c were prepared from the enantiomerically pure acetonide 6 (Scheme 3) which was synthesized from L-(+)-diethyl tartrate (see Supporting Information). Comparison of the spectral properties of our compounds, 2b, 2c, and 3c, with the original spectra kindly provided by McDonald revealed that the correct structures of compounds 49 and 48 described in ref 6 are also 2b and 2c, respectively, and not 3b and 3c as reported.

The above-described observation that trans, trans, trans-trienols 1a−c underwent stereoselective triple cyclization to give a trans,cis,cis-tris-THF product was quite intriguing considering our previously reported observations with a cis, cis-(4,8)-dienol substrate that afforded trans, trans-bis-THF product. Apparently, the relative configuration of the resultant THF rings strongly depends on the configuration of the vicinal oxygen functions formed in

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⁽⁶⁾ Towne, B. T.; McDonald, F. E. J. Am. Chem. Soc. 1997, 119, 6022. (7) In contrast to other rhenium(VII) oxidants, polycyclization with trifluoroacetylperrhenate was found to proceed with high stereoselectivity and high yields, particularly with acid-sensitive substrates (ref 6). Therefore, this study was carried out with trifluoroacetylperrhenate.

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Scheme 4. Tandem Oxidative Cyclization Reactions with CF₃CO₂ReO₃^a

^a Key: R=(CH₂)₂OBPS (a) CF₃CO₂ReO₃ (2.5 equiv) and TFAA (3 equiv), CH₂Cl₂, 6 h.

Scheme 5. Synthesis of 15^a

$$8 \xrightarrow{\frac{a}{50\%}} \xrightarrow{C_{12}H_{25}} \xrightarrow{OMS} \xrightarrow{OMS} \xrightarrow{OS} \xrightarrow$$

^a Key: (a) i. CF₃CO₂ReO₃, lutidine, CH₂Cl₂, 18 h. ii. MsCl, Et₃N, CH₂Cl₂, 0 °C, 2 h. (b) AD-mix-α, MeSO₂NH₂, tert-BuOH-H₂O, 18 h. (c) Pyridine, reflux, 2 h. (d) i. p-Nitrobenzoic acid, DEAD, PPh₃, C₆H₆, 3 h, ii. LiOH, THF-H₂O (1:1), 60 °C, 3 h.

previous cyclizations, which arises from the geometry of the double bonds in the polyenol substrate.

To further understand the stereochemical rules of this reaction, we have prepared four dienol substrates 8-11 (see Supporting Information) and subjected them to the tandem oxidative biscyclization reaction with trifluoroacetyl perrhenate (CF₃CO₂ReO₃), which yielded the only isolatable products 12–15 (Scheme 4) in 40-48% yields. We used both NMR spectroscopy and independent asymmetric synthesis to determine the stereochemistries of the products. For example, the structures of compounds 12-14 were determined on the basis of 2D ¹H-¹H COSY, TOCSY, and ROESY of their bis-nitrobenzoate derivatives. We used 2D ¹H−¹H COSY and TOCSY to locate the signals of H-7 and H-10 which show an nOe correlation in the case of the bis-benzoate esters of 12 and 13 and no such correlation in the case of 14. This analysis clearly indicated that the B-ring is cis in 12 and 13 and is trans in 14. The structure of 15 was determined by an independent synthesis starting with compound 8 (Scheme 5). Compound 14 was converted to 15 in two steps (Scheme 5), reinforcing our NMR structure determination of 14.

On the basis of the stereochemical relationships between 8-11 and 12-15 as well as our previously reported results, we propose the following rules for the single step polycyclization of polydisubstituted alkenols with $CF_3CO_2ReO_3$:

- 1. With simple bishomoallylic alcohol (where the hydroxyl group is the only strong coordination site of rhenium), the first THF ring is always produced with *trans* configuration.
- 2. If the two vicinal oxygen functions formed in the first cyclization have a *threo* relationship, the next cyclization produces a *cis*-THF ring.
- 3. If the vicinal oxygen functions formed in the first cyclization have an *erythro* relationship, the next cyclization produces a *trans*-THF ring.

The above rules reflect a dramatic change in the stereochemical course of the tandem oxidative cyclization reaction when proceeding from the first cyclization to the subsequent ones. A plausible explanation for this phenomenon arises from the ability of the newly formed THF ring to chelate the Re atom during the next oxidative cyclization. As illustrated in Scheme 6 (top), in the first cyclization step, the noncoordinating alkyl group has a high preference to take a less sterically demanding pseudoequatorial position in the proposed 3+2 transition state (I), ¹¹ leading to a *trans*-THF ring. However, in cases where the group R possesses

Scheme 6. Plausible Transition States for Oxidative Cyclization with Re(VII) considering a 3 + 2 Addition Mechanism^a

$$R = alkyl$$

$$CF_3CO_2ReO_3$$

$$R = R^*$$

$$CF_3CO_2ReO_3$$

$$R = R^*$$

^a The other ligands on Re are omitted for clarity.

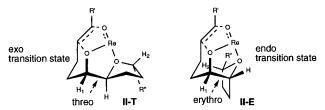


Figure 1.

a potential coordination site, the substrate may become a bidentate ligand to rhenium. In that case, a pseudoaxial position would be energetically preferred in the transition state (II), leading to a cis-THF ring. Nevertheless, the coordinating efficiency of this bidentate ligand depends on the relative configuration of the two oxygen functions. With a threo relative configuration the reaction proceeds via a sterically favored exo-type transition state, II-T. By contrast, the erythro configuration requires a sterically disfavored endo-type transition state, II-E, rendering the non-chelated structure, I, energetically more favorable.

Another observation of *cis*-THF rings in the tandem oxidative cyclization reaction with Re(VII) has been recently reported by Morimoto and Iwai who studied highly substituted systems (tertiary alcohols and trisubstituted double bonds). ¹² Their results are consistent with our model. (1.) As observed in our less substituted substrates, the first cyclization produces predominantly a *trans*-THF ring. (2.) The second cyclization produces predominantly a *cis*-THF ring. This may be expected for tertiary alcohols because, for substrates having an alkyl group instead of H_1 , there is no steric advantage of I over II. Yet, when coordination to Re becomes highly sterically demanding, e.g. with the *erythro* substrate 8 in ref 1, the transition state is II-E (having a methyl group instead of H_1 and H_2 and R'' = Pr) which results in rather low *cis* selectivity (2:1).

In conclusion, we propose here a set of stereochemical rules for the tandem oxidative cyclization reaction with CF₃CO₂ReO₃. Theoretical calculations that support these rules are underway.

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Supporting Information Available: Synthetic scheme of compounds **6** and **8–11**; ¹⁹F NMR spectral data of both Mosher's esters of **2a**, **4**, **5** and two model compounds **I** and **II**; ¹H and ¹³C NMR data of compounds **8–15**, ¹H and 2D ¹H–¹H NMR spectra of bis-nitrobenzoate derivatives of **12–14** (15 pages, print/PDF). See any current masthead page for ordering information and Web access instructions.

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