

## Beyond the Wade–Mingos Rules in Bare 10- and 12-Vertex Germanium Clusters: Transition States for Symmetry Breaking Processes

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**Abstract:** The lowest energy structures of bare  $\text{Ge}_n^z$  clusters ( $n = 10, 12$ ;  $z = -6, 0, +2, +4$ ) obtained using density functional theory (DFT) at the hybrid B3LYP level often are relatively low-symmetry polyhedra not readily recognizable by the Wade–Mingos rules. However, such optimized structures may arise from higher symmetry transition states through symmetry breaking processes. Thus the lowest energy structures for the  $\text{Ge}_{10}^{6-}$  and  $\text{Ge}_{12}^{6-}$  clusters with hyperelectronic *arachno*  $2n + 6$  skeletal electron counts are derived from pentagonal and hexagonal prism transition states, respectively, and retain the pentagonal and hexagonal faces of the prisms upon symmetry-breaking optimization. In addition, a variety of capped cube, prism, and antiprism transition states are found for the hypoelectronic  $\text{Ge}_{10}^{4+}$ ,  $\text{Ge}_{12}$ , and  $\text{Ge}_{12}^{4+}$  clusters, which go to low-energy low-symmetry optimized structures, typically  $C_s$  or  $C_i$ , upon following the normal modes of the imaginary vibrational frequencies.

### 1. Introduction

The Wade–Mingos rules<sup>1–4</sup> historically were derived in order to relate the structures of polyhedral boranes and isoelectronic compounds to the number of skeletal electrons.<sup>5</sup> However, they subsequently have been used to explain the shapes of other cluster structures isoelectronic and isolobal with boranes. According to the Wade–Mingos rules the polyhedra in the so-called *closo* boranes  $\text{B}_n\text{H}_n^{2-}$  and isoelectronic compounds with  $2n + 2$  skeletal electrons are the most spherical deltahedra, namely polyhedra in which all faces are triangles and the vertices are as similar as possible. These deltahedral boranes can be considered to be three-dimensional aromatic systems<sup>6,7</sup> with  $2n$  of the  $2n + 2$  skeletal electrons being used for surface bonding analogous to the  $\sigma$ -bonding in benzene. The remaining two skeletal electrons are used for an  $n$ -center two-electron core bond involving overlap of inward pointing radial orbitals from each of the  $n$  vertex atoms at the center of the deltahedron. This

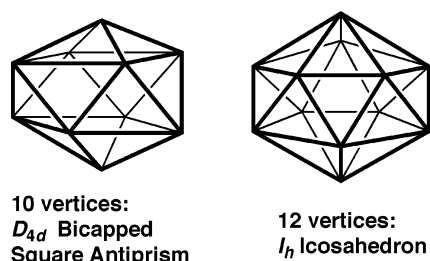
latter bond in the deltahedral boranes plays an analogous role to the  $\pi$ -bonding in benzene. For the 10- and 12-vertex structures of interest in this paper the most spherical deltahedra found in the *closo* boranes  $\text{B}_{10}\text{H}_{10}^{2-}$  and  $\text{B}_{12}\text{H}_{12}^{2-}$  and related compounds are the  $D_{4d}$  bicapped square antiprism and the  $I_h$  regular icosahedron, respectively (Figure 1). Note that in counting skeletal electrons in the clusters of interest in this paper either a BH or bare Ge vertex is a donor of two skeletal electrons.

Now consider hyperelectronic (electron-rich) polyhedral boranes having more than  $2n + 2$  skeletal electrons. The so-called *nido* boranes with  $n$  vertices have  $2n + 4$  skeletal electrons and polyhedral structures with one nontriangular face. Frequently such *nido* borane structures can be derived from a *closo* borane structure with  $n + 1$  vertices by removing one vertex and its associated edges. Thus the 10-vertex *nido* borane polyhedron, found in the long-known relatively stable<sup>8</sup>  $\text{B}_{10}\text{H}_{14}$ , can be obtained by removal of the unique degree 6 vertex from the 11-vertex *closo* deltahedron (Figure 2). Similarly the 12-vertex *nido* borane polyhedron, found in the ligand<sup>9</sup>  $\text{C}_2\text{B}_{10}\text{H}_{12}^{2-}$  obtained by reduction of the carborane  $\text{C}_2\text{B}_{10}\text{H}_{12}$ , can be formally obtained by removal

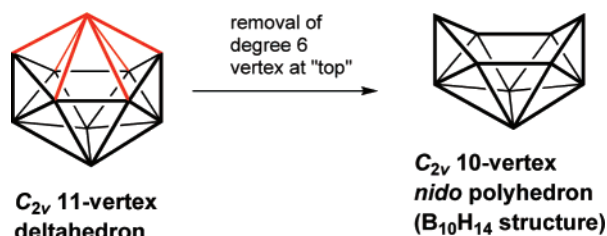
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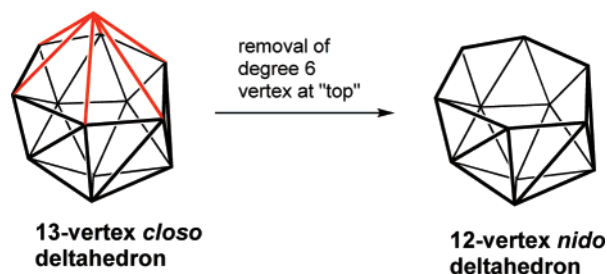
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**Figure 1.** The most spherical (*closo*) deltahedra with 10 and 12 vertices.



**Figure 2.** Conversion of the 11-vertex most spherical deltahedron to the 10-vertex *nido* polyhedron (found in  $B_{10}H_{14}$ ) by removal of the unique degree 6 vertex (the "top" vertex) and the associated edges (colored red).



**Figure 3.** Conversion of the 13-vertex most spherical deltahedron to the 12-vertex *nido* polyhedron (found in the  $C_{2B_{10}H_{12}^{2-}}$  ligand in  $(\eta^5-C_5H_5)CoC_2B_{10}H_{12}$ ) by removal of the unique degree 6 vertex (the "top" vertex) and the associated edges (colored red).

of a degree 6 vertex from the 13-vertex *closo* polyhedron found in metallaboranes such as  $(\eta^5-C_5H_5)CoC_2B_{10}H_{12}$  (Figure 3).

The Wade–Mingos rules in borane chemistry have been extended to systems even more hyperelectronic than the *nido* compounds such as the *arachno* compounds with  $2n + 6$  skeletal electrons and two nontriangular faces or one large nontriangular face and the *hypho* compounds with  $2n + 8$  skeletal electrons and an even more open structure. In principle, the *arachno* and *hypho* structures with  $n$  vertices can be derived from *closo* structures with  $n + 2$  or  $n + 3$  vertices, respectively, by removal of two or three vertices, respectively. However, as the structures become electron-rich the increasingly open polyhedra become increasingly less recognizable.

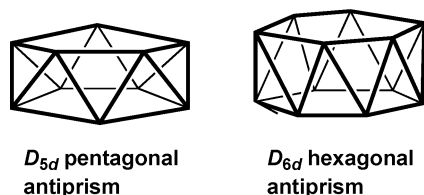
The Wade–Mingos rules<sup>1–4</sup> are more difficult to apply to hypoelectronic (electron-poor) clusters containing fewer than the  $2n + 2$  skeletal electrons of *closo* structures. Such systems are not found in borane and carborane derivatives containing exclusively boron and carbon vertices so that they were not considered in Wade's original work.<sup>1,2</sup> However,

hypoelectronic structures are found in isoelectronic metal carbonyl clusters and bare post-transition element clusters. Hypoelectronic structure types found in systems with  $n$  vertices and less than  $2n + 2$  skeletal electrons include the following: (1) a capped deltahedron with  $m < n$  vertices, typically for a system with  $m + 2$  skeletal electrons such as the capped octahedral osmium carbonyl cluster<sup>10</sup>  $Os_7(CO)_{21}$  ( $v = 7$  but  $m = 6 \Rightarrow 14$  skeletal electrons) and (2) "flattened" deltahedra with  $f$  "flattened" vertices pushed toward the center of the deltahedron,<sup>11</sup> typically for a system with  $v - f + 2$  skeletal electrons such as the  $In_{11}^{7-}$  cluster found in  $K_8In_{11}$  ( $v = 11$ ,  $f = 3 \Rightarrow 18$  skeletal electrons).<sup>12</sup>

In recent years the chemistry of bare post-transition-metal clusters has expanded greatly from the original work of Zintl and co-workers.<sup>13–16</sup> Such clusters can be considered to be formally isoelectronic with boranes and carboranes.<sup>17</sup> Thus a bare group 14 element vertex (Si, Ge, Sn, Pb) is a donor of two skeletal electrons like a B–H vertex in boranes. Similarly a bare group 15 element vertex (P, As, Sb, Bi) is a donor of three skeletal electrons like a C–H vertex in carboranes. However, in many cases the polyhedra found in bare post-transition-metal clusters are different from those found in boranes and related compounds. Furthermore, they do not relate obviously to polyhedra suggested by the Wade–Mingos rules,<sup>1–4</sup> particularly in the cases of electron-poor clusters containing bare group 13 elements (Al, Ga, In, Tl), which are donors of only one skeletal electron. In order to understand such unusual polyhedra and the chemical bonding in such structures we have performed density functional theory (DFT) studies of germanium clusters containing from 5 to 12 germanium atoms.<sup>18–23</sup> Germanium was chosen as a model vertex atom to minimize the charges on clusters isoelectronic with the known molecules of interest.

Our studies as well as the work of others have indicated major differences between isoelectronic boranes and carboranes, on the one hand, and bare germanium and other post-transition-metal clusters, on the other hand. Examples are the following: (1) The antiaromaticity of the icosahedral  $E_{12}^{2-}$  ( $E = Si, Ge$ ) as compared with the strong aromaticity of the isoelectronic  $B_{12}H_{12}^{2-}$  as noted above.<sup>24,25</sup> (2) The lowest energy structure for  $Ge_{11}^{2-}$  is not the most spherical 11-vertex deltahedron<sup>22</sup> found in the stable borane  $B_{11}H_{11}^{2-}$ . (3) The lowest energy structure for  $Ge_8^{2-}$  is the spherically aromatic  $T_d$  tetracapped tetrahedron rather than the bisdisphenoid found in  $B_8H_8^{2-}$  and related compounds.<sup>19</sup> These differences between isoelectronic bare germanium and borane clusters appear to be a consequence of the fact that the external germanium lone pair electrons can participate in the skeletal bonding, whereas no comparable electrons are available from the B–H vertices of boranes.

We have also observed many low-energy low-symmetry (not readily recognizable) germanium clusters  $Ge_n^z$ , particularly those with fewer than  $2n + 2$  skeletal electrons. Such polyhedra often arise during the optimization of more obvious symmetrical polyhedra by following imaginary vibrational frequencies. The more symmetrical and thus more recognizable polyhedra can then be considered as transition states linking isomeric polyhedra of low symmetry. Thus a useful way of characterizing unsymmetrical low-energy



**Figure 4.** The pentagonal and hexagonal antiprisms as possible *arachno* polyhedra with two nontriangular faces (namely pentagons and hexagons, respectively).

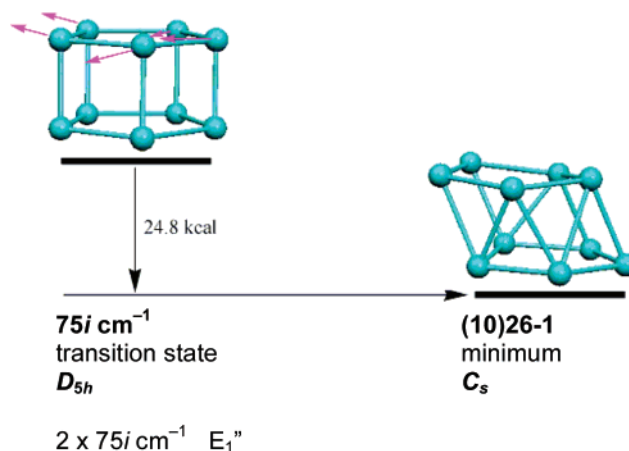
polyhedra is by the more symmetrical transition states from which they arise.

A previous paper from our group<sup>26</sup> considers low-energy low-symmetry structures of hypoelectronic 11-vertex bare germanium clusters  $\text{Ge}_{11}^z$ . In that case, consideration of transition states leading to such clusters is less useful since relatively few chemically relevant 11-vertex polyhedra are readily recognizable because of their generally low symmetries. The present paper discusses the transition states leading to the low-energy low-symmetry structures of 10- and 12-vertex bare germanium clusters  $\text{Ge}_{10}^z$   $z = +4$  and  $\text{Ge}_{12}^z$   $z = 0, +4$  found in our previous work.<sup>21,23</sup> In this case, the low-symmetry structures often arise from readily recognizable symmetrical transition states. This paper characterizes such transition states, both for hypoelectronic  $\text{Ge}_{10}^z$  and  $\text{Ge}_{12}^z$  clusters as well as for hyperelectronic  $\text{Ge}_{10}^{6-}$  and  $\text{Ge}_{12}^{6-}$  clusters. The  $\text{Ge}_{10}^{6-}$  and  $\text{Ge}_{12}^{6-}$  clusters with the  $2n + 6$  *arachno* skeletal electron counts are of interest since the readily recognizable pentagonal and hexagonal antiprism structures satisfying the *arachno* requirement of  $2n + 6$  skeletal electrons and two obvious nontriangular faces (Figure 4) are not the lowest energy structures.

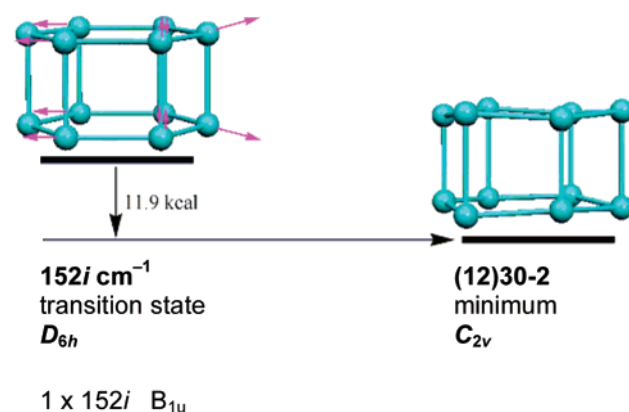
## 2. Theoretical Methods

The density functional theory (DFT) methods used in this paper are described in our previous papers on  $\text{Ge}_{10}^z$  and  $\text{Ge}_{12}^z$  clusters.<sup>21,23</sup> Thus the geometry optimizations were carried out at the hybrid DFT B3LYP level<sup>27–30</sup> with the 6-31G(d) (valence) double- $\zeta$  quality basis functions extended by adding one set of polarization (d) functions. The Gaussian 98 package of programs<sup>31</sup> was used in which the fine grid (75, 302) is the default for numerically evaluating the integrals and the tight ( $10^{-8}$ ) hartree stands as default for the self-consistent field convergence. The symmetries were maintained during the initial geometry optimization processes. In the systems of interest in this paper the transition states of interest were optimized structures with significant imaginary vibrational frequencies, typically above  $100i \text{ cm}^{-1}$ . Symmetry breaking using the normal modes of these transition states defined by these imaginary vibrational frequencies was then used to determine optimized structures with minimum energies. Both the transition states and the final optimized structures are discussed in this paper with their relationships being depicted in Figures 5–11 (and Figures 1S–6S in the Supporting Information) with the energy differences between the transition state and the final structure approximately according to scale.

One might raise the legitimate question if this method is suitable for describing subtle electronic effects governing the



**Figure 5.** The distortion of the pentagonal prism transition state in  $\text{Ge}_{10}^{6-}$  along the  $\text{E}_1$  normal mode of the  $75i \text{ cm}^{-1}$  vibrational frequency to give the global minimum (10)26-1.



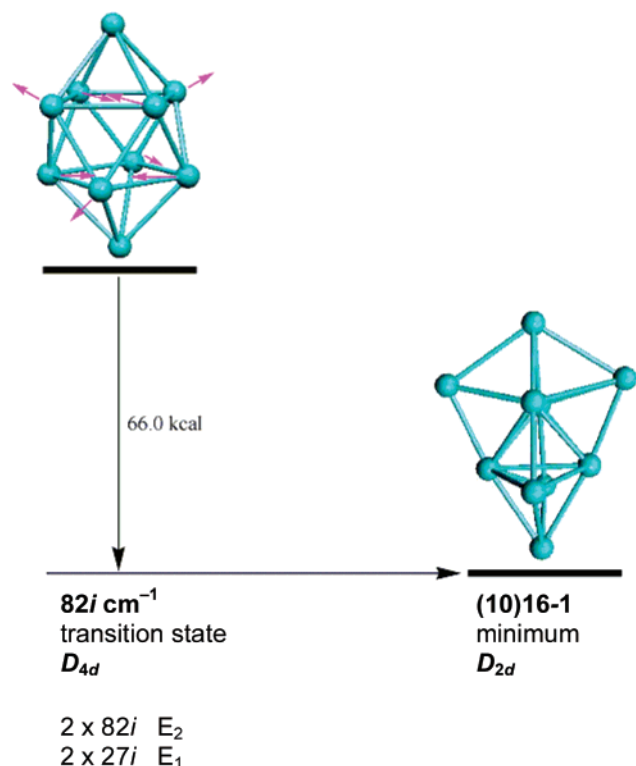
**Figure 6.** The distortion of the hexagonal prism transition state in  $\text{Ge}_{12}^{6-}$  along the  $\text{B}_{1u}$  normal mode of the  $152i \text{ cm}^{-1}$  vibrational frequency to give the lowest energy polyhedral structure (12)30-2.

structure of germanium clusters. In this connection Archibong and St. Amant<sup>32</sup> have shown that CCSD(T) calculations on  $\text{Ge}_6^z$  ( $z = 0, -1$ ) give similar results to those obtained at the B3LYP DFT level of theory.

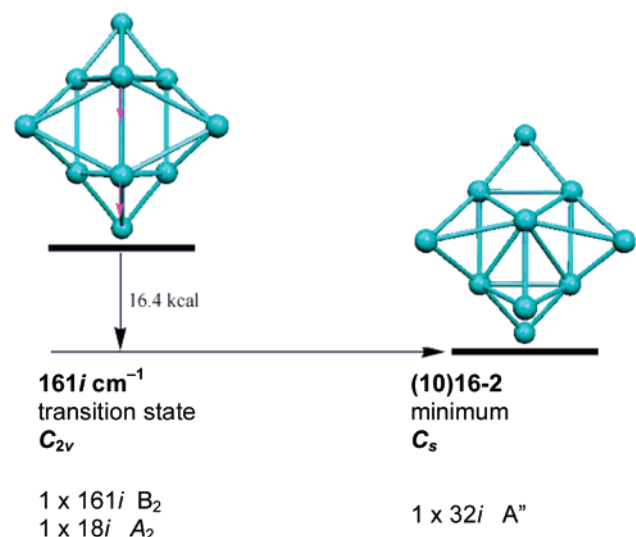
The individual structures are labeled according to the number of skeletal electrons and relative energies with the specific relative energy designations matching those in the previous papers.<sup>21,23</sup> In addition, structures with 10 and 12 vertices are distinguished by the designations (10) and (12), respectively, in front of their structure labels. Thus the lowest energy structure of the 10-vertex 26 skeletal electron system  $\text{Ge}_{10}^{6-}$ , designated as 26-1 in the previous paper,<sup>21</sup> is now designated as (10)26-1 (see Figure 5) in order to differentiate it from the lowest energy 12-vertex structure with 26 skeletal electrons, which would now be designated as (12)26-1.

## 3. Results

**3.1. The Systems  $\text{Ge}_{10}^{6-}$  and  $\text{Ge}_{12}^{6-}$  with the *arachno*  $2n + 6$  Skeletal Electron Counts.** The Wade–Mingos rules<sup>1–4</sup> suggest that the  $D_{5d}$  pentagonal antiprism (Figure 4) should be the global minimum for the 26 ( $=2n + 6$  for  $n = 10$ ) skeletal electron 10-vertex system  $\text{Ge}_{10}^{6-}$  with an *arachno* skeletal electron count. Indeed a pentagonal antiprismatic

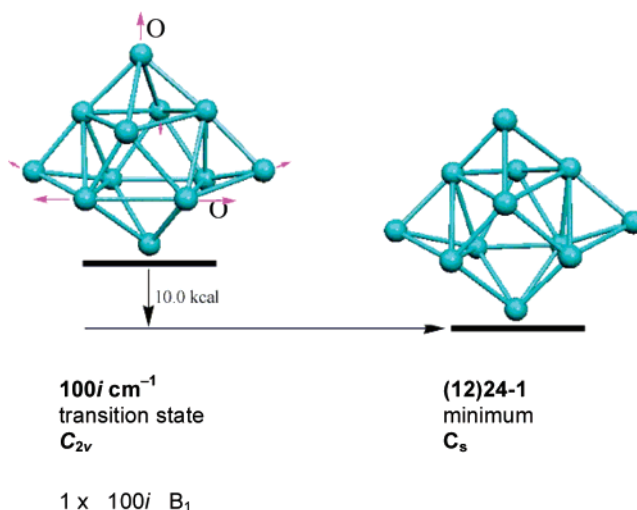


**Figure 7.** The distortion of the  $D_{4d}$  bicapped square antiprism transition state in  $\text{Ge}_{10}^{4+}$  along the  $E_2$  normal mode of the  $82i \text{ cm}^{-1}$  vibrational frequency to give the lowest energy polyhedral structure (10)16-1.

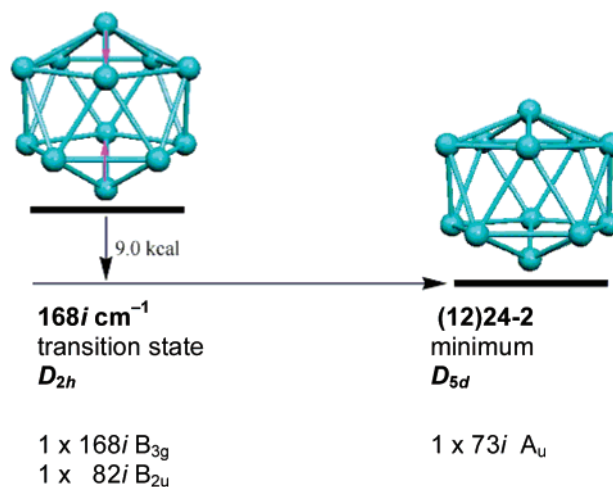


**Figure 8.** The distortion of the  $C_{2v}$  tetracapped trigonal prism transition state in  $\text{Ge}_{10}^{4+}$  along the  $B_2$  normal mode of the  $161i$  vibrational frequency to give the  $C_s$  polyhedral structure (10)16-2.

structure (10)26-3 is found for  $\text{Ge}_{10}^{6-}$  but at  $17.1 \text{ kcal/mol}$  above the global minimum<sup>21</sup> (10)26-3. Optimization of a  $D_{5h}$  pentagonal prism starting structure leads to a pentagonal prismatic transition state with a  $75i \text{ cm}^{-1}$  imaginary vibrational mode. Following the corresponding  $E_1''$  normal mode lowers the energy of the structure by  $24.8 \text{ kcal/mol}$  leading to the global minimum (10)26-1 observed<sup>21</sup> for  $\text{Ge}_{10}^{6-}$  (Figure 5). The polyhedron in (10)26-1 is derived from a  $D_{5h}$  pentagonal prism by distortion of the top pentagonal face



**Figure 9.** The distortion of the  $C_{2v}$  tetracapped square antiprism transition state in  $\text{Ge}_{12}$  along the normal mode of the  $100i \text{ B}_1$  vibrational frequency to give the  $C_s$  polyhedral structure (12)24-1.

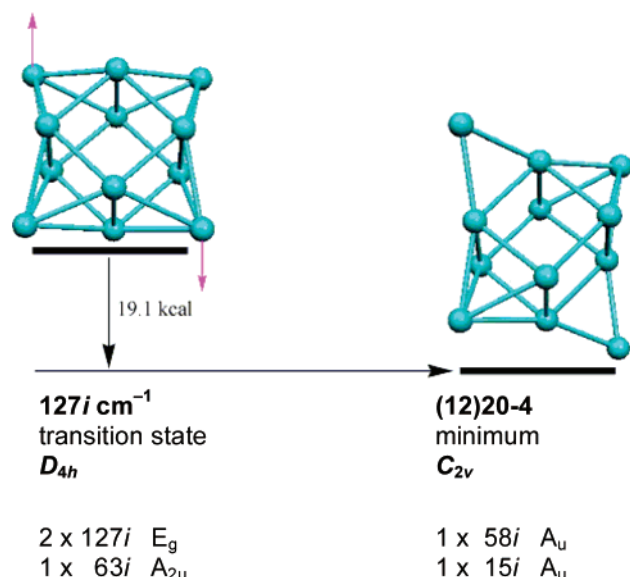


**Figure 10.** The distortion of the  $D_{2h}$  irregular icosahedron transition state in  $\text{Ge}_{12}$  along the  $B_{2g}$  normal mode of the  $168i$  vibrational frequency to give the  $D_{5d}$  bicapped pentagonal antiprismatic structure (12)24-2.

relative to the bottom face so that the five rectangular faces linking the two pentagonal faces in the original prism become two rectangular and six triangular faces by the addition of diagonals across three of the original rectangular faces. This distortion necessarily destroys the original  $C_5$  axis in the pentagonal prism. Also note that the (10)26-1 polyhedron is intermediate between the pentagonal prism with five rectangular faces between the two pentagonal faces and the pentagonal antiprism with ten triangular faces between the two pentagonal faces.

The Wade–Mingos rules<sup>1–4</sup> also suggest that the  $D_{6d}$  hexagonal antiprism (Figure 4) should be the global minimum for the 30 ( $=2n + 6$  for  $n = 12$ ) skeletal electron 12-vertex system  $\text{Ge}_{12}^{6-}$  with the *arachno* skeletal electron count. Indeed a hexagonal antiprismatic structure (12)30-5 is found for  $\text{Ge}_{12}^{6-}$  but at  $23.1 \text{ kcal/mol}$  above the lowest energy polyhedral structure (12)30-2.<sup>23</sup> Optimization of the  $D_{6h}$  hexagonal prism leads to a hexagonal prismatic transition





**Figure 11.** The distortion of the  $D_{4h}$  tetracapped cube transition state in  $\text{Ge}_{12}^{4+}$  along the  $E_g$  normal mode of the 127i vibrational frequency to give the  $C_{2v}$  structure **(12)20-4**.

state with a 152i  $\text{cm}^{-1}$  imaginary vibrational mode. Following the corresponding  $B_1$  normal mode lowers the energy of the structure by 11.9 kcal/mol leading to the lowest energy polyhedral structure **(12)30-2** calculated for  $\text{Ge}_{10}^{6-}$  (Figure 6). This structure retains the topology of the hexagonal prism (i.e., the two hexagonal faces and the six quadrilateral faces between the two hexagonal faces) but destroys the  $C_6$  axis.

**3.2. The Hypoelectronic Cluster  $\text{Ge}_{10}^{4+}$ .** The global minimum **(10)16-1** for  $\text{Ge}_{10}^{4+}$  with 16 skeletal electrons is derived from a relatively high energy  $D_{4d}$  bicapped square antiprism transition state (Figure 7). Following the  $E_2$  normal mode corresponding to the largest imaginary vibrational frequency (82i) reduces the energy of the system by a gigantic 66.0 kcal/mol. The symmetry is concurrently reduced from  $D_{4d}$  to  $D_{2d}$  leading to a polyhedron derived from two interlocking planar pentagons at right angles to each other, which resembles a “Siamese twin” of two pentagonal bipyramids (Figure 7). Note that the pentagonal bipyramid building block of structure **(10)16-1** is the most spherical deltahedron with seven vertices and thus requires  $2n + 2 = 16$  skeletal electrons for  $n = 7$  by the Wade–Mingos rules.<sup>1–4</sup>

The next higher energy structure **(10)16-2** for  $\text{Ge}_{10}^{4+}$ , at 14.5 kcal/mol above **(10)16-1** discussed above, is derived from a  $C_{2v}$  tetracapped trigonal prism transition state (Figure 8). Following the  $B_2$  normal mode of the largest imaginary frequency (161i) adds diagonals across two rectangular faces of the underlying trigonal prism with concurrent conversion of the corresponding two rectangular face caps to triangular face caps. The symmetry is thus reduced from  $C_{2v}$  to  $C_s$  and the energy by 16.4 kcal/mol (Figure 8). A relatively small imaginary vibrational frequency of 32i remains in **(10)16-2**, which might arise from a numerical integration error.<sup>33,34</sup>

A  $C_{2v}$  bicapped cube is the transition state to a  $C_s$  higher energy structure **(10)16-4** for  $\text{Ge}_{10}^{4+}$  (Figure 1S), which lies 20.6 kcal/mol above the global minimum **(10)16-1** (Figure 7). Following the  $B_1$  normal mode corresponding to the

relatively large 342i imaginary vibrational frequency in the bicapped cube reduces the symmetry from  $C_{2v}$  to  $C_s$ . However, the energy is reduced by only 7.9 kcal/mol with no significant change in the polyhedron topology.

**3.3. The Neutral 12-Vertex Cluster  $\text{Ge}_{12}$ .** The lowest energy structures for the neutral  $\text{Ge}_{12}$  cluster with 24 skeletal electrons are relatively low-symmetry structures derived by distortion of higher energy transition states. The global minimum for  $\text{Ge}_{12}$ , namely **(12)24-1** (Figure 9), is derived from a  $C_{2v}$  tetracapped square antiprism transition state by distortion along the  $B_1$  normal mode corresponding to the 100i  $\text{cm}^{-1}$  vibrational frequency. The symmetry is reduced from  $C_{2v}$  to  $C_s$ , and two pairs of triangular faces in the original tetracapped square antiprism are opened (following the arrows marked by “O” in Figure 9) up into two quadrilateral faces. Removal of the two vertices capping triangular faces from the **(12)24-1** polyhedron leaves a 10-vertex polyhedron with 14 triangular faces and one quadrilateral face. This is a 10-vertex *nido* polyhedron and thus expected by the Wade–Mingos rules<sup>1–4</sup> to have the  $2n + 4 = 24$  for  $n = 10$  skeletal electrons found in  $\text{Ge}_{12}$ .

The next higher energy structure **(12)24-2** for  $\text{Ge}_{12}$ , at 21.9 kcal/mol above **(12)24-1** discussed above, is derived from an irregular  $D_{2h}$  icosahedron transition state by distortion along the  $B_{3g}$  normal mode corresponding to the largest imaginary frequency at 168i  $\text{cm}^{-1}$  (Figure 10). This normal mode involves compression of a pair of antipodal vertices and leads to a compressed (oblate) bicapped pentagonal antiprism still retaining the topology of the original icosahedron (i.e., no edges are broken). This process lowers the energy by 9.0 kcal/mol (Figure 10).

The next higher energy structure **(12)24-3** for  $\text{Ge}_{12}$ , at 23.5 kcal/mol above **(12)24-1** discussed above, is derived from a  $D_{6d}$  hexagonal antiprism transition state by distortion along the  $E_4$  normal mode corresponding to the largest imaginary frequency at 143i  $\text{cm}^{-1}$  (Figure 2S). This process lowers the energy by a very large 58.9 kcal/mol and the symmetry from  $D_{6d}$  to  $D_{2d}$ . Each of the hexagonal faces of the original hexagonal antiprism becomes a pair of trapezoidal faces by formation of new transannular edges.

A higher energy structure for  $\text{Ge}_{12}$ , namely **(12)24-6** at 28.2 kcal/mol above the global minimum **(12)24-1**, is obtained from a  $D_{4h}$  tetracapped cube by distortion along the  $E_g$  normal mode corresponding to the 114i  $\text{cm}^{-1}$  vibrational frequency (Figure 3S). The process lowers the energy by 12.8 kcal/mol and the symmetry from  $D_{4h}$  to  $C_{2v}$ . The polyhedron in the resulting structure **(12)24-6** can be described as a pair of  $C_{2v}$  bicapped trigonal prisms sharing their uncapped rectangular faces.

**3.4. The Lowest Energy Structure of  $\text{Ge}_{12}^{2+}$ .** The lowest energy structure for  $\text{Ge}_{12}^{2+}$ , namely the  $D_{3d}$  hexacapped octahedron **(12)22-1**, is derived from a  $D_{4h}$  tetracapped cube transition state (Figure 4S) by a distortion along the  $E_g$  normal mode corresponding to the 95i  $\text{cm}^{-1}$  vibrational frequency. This process replaces the original 4-fold axis with a 3-fold axis and lowers the energy by 42.8 kcal/mol. Since the  $D_{3d}$  point group of the hexacapped octahedron is obviously not a subgroup of the  $D_{4h}$  point group of the

**Table 1.** Seven Transition States within 25 kcal/mol of the Final Optimized Structures

figure	species	transition state	imaginary frequencies <sup>a</sup>	symmetry breaking <sup>c</sup>	energy gain <sup>b</sup> / binding energy/atom <sup>d</sup> (transition state/minimum)
Hyperelectronic ( <i>arachno</i> ) Species					
5	Ge <sub>10</sub> <sup>6-</sup>	pentagonal prism	75i(E <sub>1</sub> '')	D <sub>5h</sub> →C <sub>s</sub>	24.8/+8.1/5.6
6	Ge <sub>12</sub> <sup>6-</sup>	hexagonal prism	152i(B <sub>1u</sub> )	D <sub>6h</sub> →C <sub>2v</sub>	11.9/-19.9/-20.9
Hypoelectronic Species					
8	Ge <sub>10</sub> <sup>4+</sup>	tetracapped trigonal prism	161i(B <sub>2</sub> ), 18i(A <sub>2</sub> )	C <sub>2v</sub> →C <sub>s</sub>	16.4/-55.9/-57.5
S1	Ge <sub>10</sub> <sup>4+</sup>	bicapped cube	342i(B <sub>1</sub> ), 48i(B <sub>2</sub> )	C <sub>2v</sub> →C <sub>s</sub>	7.9/-56.1/-56.9
9	Ge <sub>12</sub>	tetracapped square antiprism	100i(B <sub>1</sub> )	C <sub>2v</sub> →C <sub>s</sub>	10.0/-112.2/-113.0
10	Ge <sub>12</sub>	irregular icosahedron (D <sub>2h</sub> )	168i(B <sub>3g</sub> ), 82i	D <sub>2h</sub> →D <sub>5d</sub> (73i) <sup>c</sup>	9.0/-110.4/-111.2
11	Ge <sub>12</sub> <sup>4+</sup>	tetracapped cube	127i(E <sub>g</sub> ), 62i(A <sub>2u</sub> )	D <sub>4h</sub> →C <sub>2v</sub> (58i)	19.1/-69.8/-71.4

<sup>a</sup> Residual imaginary vibrational frequencies are given in parentheses. <sup>b</sup> Difference between the energy of the transition state and that of the optimized structure (kcal/mol). <sup>c</sup> This represents a symmetry change rather than symmetry breaking. <sup>d</sup> Difference between the energy of the cluster and the sum of the energies of the components divided by the number of atoms (kcal/mol).

tetracapped cube transition state, this is not a simple symmetry breaking process.

**3.5. Several Structures for Ge<sub>12</sub><sup>4+</sup>.** The global minimum for Ge<sub>12</sub><sup>4+</sup>, namely (12)20-1, is a C<sub>2v</sub> structure that does not resemble anything actually found in a chemical system.<sup>23</sup> The next highest energy structure for Ge<sub>12</sub><sup>4+</sup>, namely (12)-20-2 at 8.0 kcal/mol above the global minimum (12)20-1, is a D<sub>4h</sub> double cube, which can be derived from an O<sub>h</sub> cuboctahedron transition state by distortion along the E<sub>u</sub> normal mode of the highest imaginary vibration, namely the 417i cm<sup>-1</sup> frequency (Figure 5S). This process is a drastic rearrangement that lowers the energy of the system by a gigantic 164.4 kcal/mol.

The next higher energy structure for Ge<sub>12</sub><sup>4+</sup>, namely (12)-20-3 at 8.9 kcal/mol above (12)20-1, is derived from a D<sub>4h</sub> double square antiprism transition state. Following the E<sub>g</sub> normal mode corresponding to the highest imaginary vibrational frequency, namely the 123i cm<sup>-1</sup> frequency, lowers the energy by a large 62.0 kcal/mol and the symmetry from D<sub>4h</sub> to C<sub>i</sub> to give a much more open structure than the original double square antiprism (Figure 5S).

The next structure for Ge<sub>12</sub><sup>4+</sup>, namely the C<sub>2v</sub> structure (12)20-4 at 10.5 kcal/mol above the global minimum (12)-20-1, is derived from a D<sub>4h</sub> tetracapped cube transition state (Figure 11). Distortion along the E<sub>g</sub> normal mode of the largest imaginary frequency at 127i reduces the symmetry from D<sub>4h</sub> to C<sub>2v</sub> with two opposite face caps becoming edge caps. This process results in an energy gain of 19.1 kcal/mol.

## 4. Discussion

The figures in the text and the Supporting Information depict 12 transition states leading to previously obtained<sup>21,23</sup> optimized structures for Ge<sub>n</sub><sup>6-</sup> (*n* = 10, 12), Ge<sub>10</sub><sup>4+</sup>, Ge<sub>12</sub>, Ge<sub>12</sub><sup>2+</sup>, and Ge<sub>12</sub><sup>4+</sup> including global minima. The polyhedra in most of these optimized low-energy structures are not readily recognizable by the Wade–Mingos rules.<sup>1-4</sup> Of these 12 transition states, seven of them (Table 1) are within 25 kcal/mol of the optimized structure and thus are potentially chemically significant. The remaining five higher energy transition states, at energies ranging from 42.8 to 164.4 kcal/mol above the corresponding optimized structure, are significant mainly in indicating the route used for the DFT optimizations. These are the closest ones to the final

optimized structures of any of the initial structures investigated. This is particularly true for the D<sub>4h</sub> tetracapped cube transition state leading to the D<sub>3d</sub> hexacapped octahedron global minimum (12)22-1 of Ge<sub>12</sub><sup>4+</sup> (Figure S3), since D<sub>3d</sub> is not a subgroup of D<sub>4h</sub> so that the conversion of the tetracapped cube to a hexacapped octahedron is not a simple symmetry breaking process.

A true transition state exhibits exactly one significant imaginary vibrational frequency, the normal mode of which indicates the pathway to the corresponding minimum point. Additional small imaginary frequencies significantly below 100i cm<sup>-1</sup> can be attributed to errors arising from the numerical integration procedure.<sup>33,34</sup> The seven transition states listed in Table 1 all have exactly one imaginary vibrational frequency at or above 100i cm<sup>-1</sup> except for the pentagonal prism transition state for Ge<sub>10</sub><sup>6-</sup> (Figure 5), which has a single imaginary vibrational frequency at 75i cm<sup>-1</sup>, which was followed in the optimization to the global minimum (10)26-1 for Ge<sub>10</sub><sup>6-</sup>. Two of the optimized structures in Table 1, namely the D<sub>5d</sub> bicapped pentagonal antiprism structure (12)24-2 for Ge<sub>12</sub>, derived from a D<sub>2h</sub> irregular icosahedron transition state (Figure 11), and the C<sub>2v</sub> structure (12)20-4 for Ge<sub>12</sub><sup>4+</sup>, derived from a D<sub>4h</sub> tetracapped cube transition state (Figure 11), have residual imaginary vibrational frequencies below 100i cm<sup>-1</sup> after optimization. These structures are low-energy structures for Ge<sub>12</sub> and Ge<sub>12</sub><sup>4+</sup>, respectively, but are not their global minima.

Most of the high-energy transition states correspond to structures that are relatively far removed from the final structure and have more than one imaginary vibrational frequency large enough to be significant, thereby corresponding more accurately to higher order saddle points. The most extreme case of this type found in this work is the cuboctahedron for Ge<sub>12</sub><sup>4+</sup>, which has three imaginary vibrational frequencies above 100 cm<sup>-1</sup>, namely 417i cm<sup>-1</sup> (E<sub>u</sub>), 290i cm<sup>-1</sup> (T<sub>2u</sub>), and 115i (T<sub>1g</sub>) as well as smaller imaginary vibrational frequencies at 36i (A<sub>2g</sub>) and 29i (E<sub>g</sub>). Following the normal mode corresponding to the 417i cm<sup>-1</sup> vibrational frequency results in the gigantic energy lowering of 164.4 kcal/mol and a rather drastic rearrangement leading to a D<sub>4h</sub> double cube stationary point (12)20-2. This corresponds to the lowest energy chemically realistic structure<sup>23</sup> found for Ge<sub>12</sub><sup>4+</sup>.

In general the binding energies per atom are higher for the neutral clusters and decrease both in the negatively and positively charged species as expected. Notable is also the finding that the binding energies for the minima have only slightly lower values than those of the corresponding transition states.

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**Supporting Information Available:** Figures 1S–6S as cited in the text. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## References

- (1) Wade, K. *Chem. Commun. (Cambridge)* **1971**, 792.
- (2) Wade, K. *Adv. Inorg. Chem. Radiochem.* **1976**, 18, 1.
- (3) Mingos, D. M. P. *Nat. Phys. Sci.* **1972**, 99, 236.
- (4) Mingos, D. M. P. *Acc. Chem. Res.* **1984**, 17, 311.
- (5) Williams, R. E. *Chem. Rev.* **1992**, 92, 177.
- (6) King, R. B.; Rouvray, D. H. *J. Am. Chem. Soc.* **1977**, 99, 7834.
- (7) King, R. B. *Chem. Rev.* **1991**, 101, 1119.
- (8) Kasper, J. S.; Lucht, C. M.; Harker, D. *Acta Crystallogr.* **1950**, 3, 436.
- (9) Dustin, D. F.; Dunks, G. B.; Hawthorne, M. F. *J. Am. Chem. Soc.* **1973**, 95, 1109.
- (10) Eady, C. R.; Johnson, B. F. G.; Lewis, J.; Mason, R.; Hitchcock, P. B.; Thomas, K. M. *Chem. Commun. (Cambridge)* **1977**, 385.
- (11) King, R. B. *Rev. Roum. Chim.* **2002**, 47, 1005.
- (12) Sevov, S. C.; Corbett, J. D. *Inorg. Chem.* **1991**, 30, 4875.
- (13) Zintl, E.; Goubeau, J.; Dullenkopf, W. *Z. Phys. Chem., Abt. A* **1931**, 154, 1.
- (14) Zintl, E.; Harder, A. *Z. Phys. Chem., Abt. A* **1931**, 154, 47.
- (15) Zintl, E.; Dullenkopf, W. *Z. Phys. Chem., Abt. B* **1932**, 16, 183.
- (16) Zintl, E.; Kaiser, H. *Z. Anorg. Allgem. Chem.* **1933**, 211, 113.
- (17) King, R. B. *Inorg. Chim. Acta* **1982**, 57, 79.
- (18) Ge<sub>5</sub>, Ge<sub>6</sub>, and Ge<sub>7</sub>: King, R. B.; Silaghi-Dumitrescu, I.; Kun, A. *J. Chem. Soc., Dalton Trans.* **2002**, 3999.
- (19) Ge<sub>8</sub>: King, R. B.; Silaghi-Dumitrescu, I.; Lupan, A. *Dalton Trans.* **2005**, 1858.
- (20) Ge<sub>9</sub>: King, R. B.; Silaghi-Dumitrescu, I. *Inorg. Chem.* **2003**, 42, 6701.
- (21) Ge<sub>10</sub>: King, R. B.; Silaghi-Dumitrescu, I.; Uță, M. M. *Inorg. Chem.* **2006**, 45, 4974.
- (22) Ge<sub>11</sub>: King, R. B.; Silaghi-Dumitrescu, I.; Lupan, A. *Inorg. Chem.* **2005**, 44, 3579.
- (23) Ge<sub>12</sub>: King, R. B.; Silaghi-Dumitrescu, I.; Uță, M. M. *Dalton Trans.* **2007**, 364.
- (24) King, R. B.; Heine, T.; Corminboeuf, C.; Schleyer, P. v. R. *J. Am. Chem. Soc.* **2004**, 126, 430.
- (25) Chen, Z.; Neukemans, S.; Wang, X.; Janssens, E.; Zhou, Z.; Silverans, R. E.; King, R. B.; Schleyer, P. v. R.; Lievens, P. *J. Am. Chem. Soc.* **2006**, 128, 12829.
- (26) King, R. B.; Silaghi-Dumitrescu, I.; Lupan, A. *Chem. Phys.* **2006**, 327, 344.
- (27) Vosko, S. H.; Wilk, L.; Nusair, M. *Can. J. Phys.* **1980**, 58, 1200.
- (28) Lee, C.; Yang, W.; Parr, R. G. *Phys. Rev. B* **1988**, 37, 785.
- (29) Becke, A. D. *J. Chem. Phys.* **1993**, 98, 5648.
- (30) Stephens, P. J.; Devlin, F. J.; Chabalowski, C. F.; Frisch, M. J. *J. Phys. Chem.* **1994**, 98, 11623.
- (31) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Zakrzewski, V. G.; Montgomery, J. A., Jr.; Stratmann, R. E.; Burant, J. C.; Dapprich, S.; Millam, J. M.; Daniels, A. D.; Kudin, K. N.; Strain, M. C.; Farkas, O.; Tomasi, J.; Barone, V.; Cossi, M.; Cammi, R.; Mennucci, B.; Pomelli, C.; Adamo, C.; Clifford, S.; Ochterski, J.; Petersson, G. A.; Ayala, P. Y.; Cui, Q.; Morokuma, K.; Rega, N.; Salvador, P.; Dannenberg, J. J.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Cioslowski, J.; Ortiz, J. V.; Baboul, A. G.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Gomperts, R.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Andres, J. L.; Gonzalez, C.; Head-Gordon, M.; Replogle, E. S.; Pople, J. A. *Gaussian 98, Revision A.11.3*; Gaussian, Inc.: Pittsburgh, PA, 2002.
- (32) Archibong, E. F.; St-Amant, A. *J. Chem. Phys.* **1998**, 109, 962.
- (33) Xie, Y.; Schaefer, H. F.; King, R. B. *J. Am. Chem. Soc.* **2000**, 122, 8746.
- (34) Papas, B. N.; Schaefer, H. F. *J. Mol. Struct. THEOCHEM* **2006**, 768, 175.

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