

## ADDITIONS AND CORRECTIONS

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Page 4115. Equation 5 for the molecular orientation analysis should be corrected in the following way:

$$\varphi = \tan^{-1} \left( \sqrt{2 \frac{I_{\text{IP}}}{I_{\text{OP}}}} \right)$$

The infrared absorption-intensity ratio in the equation represents the ratio of the absorption index ( $k$ ) in the in-plane and out-of-plane directions. The absorption index of a direction is proportional to the probability of vibrational transition per unit of time, which is also proportional to  $\langle \phi_f | \mu_j E_j | \phi_i \rangle^2$  where  $\mu_j$  and  $E_j$  are the dipole moment and electric field in the  $j$  direction, respectively, and  $\phi_f$  and  $\phi_i$  are the wave functions of the initial and final vibrational states via the interaction with the dipole moment, respectively. When we discuss the orientation of a dipole, therefore, the square root of the ratio of absorbance intensities must be calculated before applying the inverse tangent function. This is also mentioned in a different way by Chollet et al.<sup>1</sup>

### References and Notes

- (1) Chollet, P.-A.; Messier, J.; Rosillio, C. *J. Chem Phys.* **1976**, *64*, 1042.

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