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Original article

In vitro and in vivo anticancer activity evaluation of ursolic acid derivatives

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ABSTRACT

Twenty-three ursolic acid (1) derivatives 2-24 (ten novel compounds 8-10, 14-17 and 22-24) modified at the C-3 and the C-28 positions were synthesized, and their structures were confirmed by IR, ¹H NMR, MS, and elemental analysis. The single crystals of compounds 15 and 17 were obtained. The cytotoxic activity of the derivatives was evaluated against HepG2, BGC-823, SH-SY5Y, HeLa and HELF cells by the MTT assay. The induction of apoptosis and affects on the cell cycle distribution with compound 14 were assessed by fluorescence microscopy, flow cytometry and the activity of caspase-3 in HepG2 cells. Compounds 14-17 had more significant antiproliferative ability against the four cancer cell lines and low cytotoxicity to human embryonic lung fibroblast cells (HELF). Compounds 11, 14-16, 21 and 23 were particularly active against HepG2 cell growth. Compound 14 was selected to investigate cell apoptosis and cell cycle distribution. Flow cytometric analysis and morphologic changes of the cell exhibited that treatment of HepG2 cells with compound 14 led to cell apoptosis accompanied by cell cycle arrest at the S phase in a dose-dependent manner. Furthermore, the activity of the caspase-3 enzyme was increased in the treated cells. In vivo studies using H22 xenografts in Kunming mice were conducted with compound 14 at doses of 50, 100 and 150 mg/kg body weight. The results revealed that the medium dosage group (100 mg/kg) showed significant anticancer activity (45.6 \pm 4.3%) compared to the control group.

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1. Introduction

Ursolic acid (UA, 3β -hydroxy-urs-12-en-28-oic acid 1), a pentacyclic triterpene acid, exists abundantly in the plant kingdom. Ursolic acid and its derivatives have been reported to have interesting bioactivity, including anti-HIV [1], antihepatodamage [2], antimalarial [3], antimicrobial and anti-inflammatory activity [4]. Recent studies have shown that ursolic acid has marked anti-tumor effects and cytotoxic activity toward various types of cancer cell lines *in vitro* [5–7].

As an effective natural anticancer drug, considerable structural modification has been performed on ursolic acid to obtain potential anticancer compounds. Ursolic acid has been treated with different anhydrides in pyridine or tetrahydrofuran to obtain a series of 3-O-acyl derivatives [8,9]. Amino acid was connected to ursolic acid after its 17-COOH was chlorinated with oxalyl chloride [10]. Some derivatives have shown significant cytotoxicity against some tumor cell lines [10,11]. The structure—activity relationship demonstrated that a hydrogen donor group at either position 3 and/or 28 was

essential for cytotoxic activity, and a significant improvement in

However, derivatives have not been thoroughly explored for their anti-tumor activity. To search for potentially important anti-tumor drug candidates, study their structure—activity relationships and explore their mechanisms, a series of UA derivatives were synthesized, and their cytotoxic activities were evaluated against human hepatoma cells HepG2, human gastric cancer cells BGC-823, human neuroblastoma cells SH-SY5Y, human cervical carcinoma cells HeLa, and human embryonic lung fibroblast HELF. Investigations of the mechanism and anti-tumor activity *in vivo* were performed on compound **14** and UA.

2. Results and discussion

2.1. Chemistry

Chemical transformations of the active compound were carried out to study the structure—activity relationships.

Ursolic acid (1) was acetylated to give 3-O-acetylursolic acid. Treatment of Ursolic acid or 3-O-acetylursolic acid with the bromodiolefine afforded fatty esters **2–4** (Scheme 1) or **8–10** (Scheme 2),

cell growth inhibition was observed when an acetyl group was introduced on the 3-OH, and when amino alkyl groups were introduced at the 17-COOH position.

However, derivatives have not been thoroughly explored for

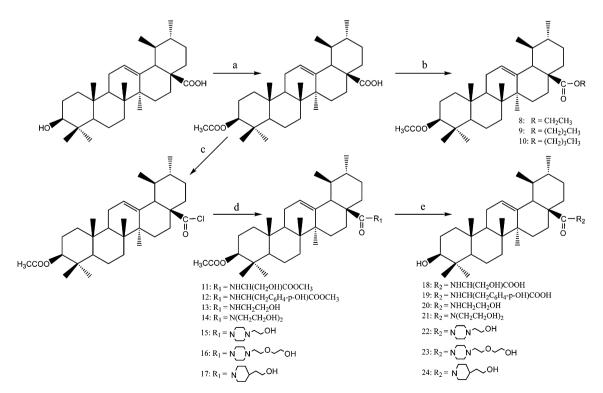
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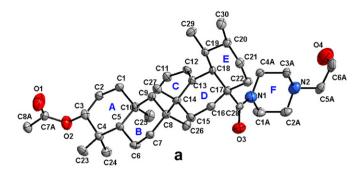
Scheme 1. Synthesis of ester derivatives of ursolic acid. Reagents and conditions: (a) CH₃(CH₂)_nBr, K₂CO₃, DMF, r.t.; (b) Br(CH₂)_nBr, K₂CO₃, DMF, r.t.; (c) AgNO₃/CH₃CN, COO(CH₂)_nBr.

respectively. Ursolic acid (1) was treated with dibromo-diolefine to obtain intermediate products, which were further reacted with AgNO₃ to yield nitrate esters **5–7** (Scheme 1). 3-O-acetylursolic acid was treated with oxalyl chloride to give the 28-acylchloride, which was then condensed with the appropriate amino compound (methyl ester of serine, tyrosine, 2-aminoethanol, 2-aminodiethanol, *N*- (2-Hydroxyethyl) piperazine, 1-hydroxyethylethoxypiperazine or

4-piperidineethanol) in the presence of triethylamine to provide compounds **11–17** (Scheme 2). Saponification of compounds **11–17** gave the corresponding compounds **18–24** (Scheme 2). All the new compounds were fully characterized by various spectroscopic methods, including Infrared (IR), ¹H NMR, mass spectra (MS) and element analysis. In addition, single crystal of compound **15** and **17** were obtained by layering hexane on a mixture of dichloromethane



Scheme 2. Synthesis of amides and other ester derivatives of ursolic acid. Reagents and conditions: (a) anhydride/Pyr/DMAP, r.t.; (b) $CH_3(CH_2)_nBr$, K_2CO_3 , DMF, r.t.; (c) $(CO)_2CI$, CH_2CI_2 , r.t.; (d) CH_2CI_2 , Et_3N , HR, r.t.; (e) NaOH, CH_3OH /THF, r.t.



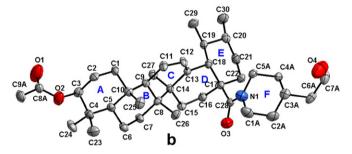


Fig. 1. Molecular structure of compounds. (a) compound 15; (b) compound 17. H atoms have been omitted for clarity.

and methanol. These compounds crystallize in the orthorhombic space group P21 (Fig. 1).

2.2. Cytotoxic activity

The *in vitro* cytotoxic activity was evaluated for all of the synthesized compounds against the HepG2, BGC-823, SH-SY5Y, HeLa and HELF cell lines. Ursolic acid, paclitaxel and compounds **2–24** were dissolved in DMSO, dissolved in ethanol or diluted with

culture medium containing 0.1% DMSO, respectively. The control cells were treated with culture medium containing 0.1% DMSO. Paclitaxel was used as a positive control. If 50% inhibition could not be reached at the highest concentration, then $>\!100~\mu\text{M}$ was given. The results are shown in Table 1.

It can be seen from the IC_{50} values that ursolic acid suppressed proliferation of the above four cancer cell lines in different extents (IC_{50} values of 33.12–68.82 μ M), and some modified compounds from it exhibited more effective, such as **11**, **13**–**16**, **21** and **23**. These derivatives showed similar inhibition activity against the HepG2, BGC-823 and SH-SY5Y cell lines, while the proliferation inhibition of HeLa cell lines was superior to other kinds of cancer cell lines.

The 3-OH of the modified ursolic acid derivatives was acetylated, resulting in 3-O-acetyl ursolic acids. When the 17-COOH was substituted by with a fatty alkyl group, the effect was significantly reduced as the length of the carbon chain was increased (2 > 3 > 4). Furthermore, the activity or even disappeared when the 3-OH and 17-COOH were both esterified (such as **8–10**), which suggested that keeping a polar group at either the 3-OH and/or 17-COOH position is essential for cytotoxic activity. On one hand, it may affect the hydrogen bonding, but on the other hand, the long chains may prevent cells from connecting with the mother nucleus.

NO-donating compounds **5**–**7** did not show significant antitumor activity as reported in the literature [12]. In our study, compound **5** had similar activity to UA, while compounds **6**–**7** showed decreased activity, which may be due to the same reasons as the long fatty alkyl group.

The previous study [10] showed that significant further improvement of cell growth inhibition was observed when an acetyl group was introduced at the 3-OH position and amino alkyl groups were introduced at the 17-COOH position. Similar results were seen in our study, where compounds 11 and 13–17 had more potent anti-tumor effects than compounds 18 and 20–24, which had a free 3-OH group. Interestingly, compounds 12 and 19 were exceptions, exhibiting little antiproliferative effects, suggesting that

Table 1 The *in vitro* activity of 1-24 (expressed as IC₅₀ (μ M)) against HepG2, BGC-823, SH-SY5Y and HeLa human cancer cell lines, as well as the human embryonic lung fibroblast cell line HELF.^a

Compoud	$IC_{50} (\mu M)^a$				
	HepG2	BGC-823	SH-SY5Y	HeLa	HELF
1	68.82 ± 3.71	66.38 ± 0.66	54.62 ± 3.87	33.12 ± 0.63	21.28 ± 1.76
2	70.77 ± 3.88	69.32 ± 2.03	66.28 ± 0.45	29.18 ± 4.72	$81.92 \pm 3.19^{**}$
3	89.11 ± 4.02	79.68 ± 1.92	$92.52 \pm 3.22^*$	$48.03 \pm 2.98^*$	$> 100^{\rm b}$
4	>100	>100	>100	$63.24 \pm 3.03^{**}$	>100
5	79.58 ± 0.45	84.24 ± 1.25	71.07 ± 2.35	36.21 ± 1.07	$68.06 \pm 2.11^{**}$
6	87.44 ± 2.10	$98.28 \pm 1.23^*$	$89.57 \pm 3.12^*$	$69.39 \pm 4.13^{**}$	$82.56 \pm 0.85^{**}$
7	>100	>100	>100	>100	>100
8	>100	>100	>100	>100	>100
9	>100	>100	>100	>100	>100
10	>100	>100	>100	>100	>100
11	$35.26 \pm 1.26^{**}$	$29.01 \pm 4.13^{**}$	$37.18 \pm 3.55^{**}$	$20.61 \pm 1.89^*$	22.34 ± 2.08
12	>100	82.95 ± 2.65	>100	$66.97 \pm 2.45^{**}$	$35.81 \pm 1.56^*$
13	$46.62\pm2.42^*$	48.47 ± 4.22	49.41 ± 1.34	16.27 ± 2.33	27.82 ± 1.56
14	$20.25 \pm 1.52^{**}$	$15.52 \pm 0.56^{**}$	$13.24 \pm 0.89^{**}$	$10.87 \pm 3.21^{**}$	38.06 ± 3.99
15	$20.46 \pm 2.53^{**}$	$13.86 \pm 2.45^{**}$	$11.7\pm1.08^{**}$	$7.24 \pm 3.11^{**}$	$31.04 \pm 2.13^*$
16	$15.26 \pm 4.10^{**}$	$12.83 \pm 2.30^{**}$	$9.51\pm0.45^{**}$	$6.28\pm0.88^{**}$	$38.92 \pm 2.43^*$
17	59.38 ± 0.42	51.34 ± 3.25	40.12 ± 2.13	12.43 ± 1.42	30.02 ± 2.03
18	83.03 ± 2.45	79.3 ± 2.14	72.21 ± 1.88	$41.65 \pm 1.45^*$	$60.33 \pm 1.22^{**}$
19	>100	>100	>100	>100	>100
20	$82.36 \pm 3.23^*$	78.26 ± 2.06	65.01 ± 2.96	$32.33 \pm 0.08^*$	$52.75 \pm 1.08**$
21	$35.66 \pm 0.33^*$	$39.27 \pm 2.11^*$	$21.38 \pm 3.06^*$	$14.36 \pm 4.26^*$	$36.92\pm2.09^*$
22	>100	98.26 ± 2.06	>100	$52.33 \pm 0.08^*$	$32.75\pm1.08^*$
23	$36.71 \pm 1.36^{**}$	$35.04 \pm 2.03^{**}$	$30.28 \pm 2.56^{**}$	23.78 ± 3.19	28.06 ± 3.58
24	>100	>100	>100	67.29 ± 2.01	>100
Paclitaxol	$10.32 \pm 0.02^{**}$	$9.24 \pm 0.13^{**}$	$8.56\pm0.07^{**}$	$6.62\pm0.21^{**}$	$7.17 \pm 0.08^*$

Data are presented as mean \pm SD (n=6). *P<0.05; **P<0.01. compared with control.

^a The concentration of the drug that inhibited cells growth by 50%

 $[^]b\,$ When 50% inhibition could not reached at the highest concentration, then ${>}100~\mu\text{M}$ was given.

the introduction of a more bulky group (i.e., benzene ring) could increase the steric hindrance and decrease binding to the target.

The introduction of a diethanolamine structure at the 17-COOH position was done to determine whether a disubstituted amide structure was beneficial to the activity. Compounds **14** and **21** demonstrated better anticancer activities than ethanolamines **13** and **20**, in which the 3-OH group was free or acetylated, respectively.

The piperidine derivatives were relatively less active than the piperazidine, as the IC_{50} values showed, suggesting that the piperazine ring had better pharmacodynamic effects than the piperidine ring.

Considering the cytotoxicity of UA and its derivatives, the proliferation of HELF cells was tested. The activity of the UA ester compound decreased compared with UA. Among of the UA amide conjugates, those which possessed strong inhibition shared similar toxicity to UA, and certain modifications showed lower toxicity than UA (compounds **13, 14,** etc.).

Compound **14** showed significant activity against cancer cell lines (HepG2, C823, SH-SY5Y and HeLa) and lower cytotoxicity against HELF compared with ursolic acid (Fig. 2).

2.3. Investigation of apoptosis

It has been reported that UA induces apoptosis and growth inhibition at the G1 phase of the cell cycle in certain cancer cell systems [13]. Compound **14** was selected to analyze the mechanism of growth inhibition of HepG2 and BGC-823 cells by the following methods.

2.3.1. Fluorescence staining detection of cell apoptosis

The morphologic changes in the cell after treatment with compound **14** were assessed by fluorescence microscopy after staining with acridine orange (AO). Because AO can penetrate the normal cell membrane, the cells become stained uniformly to yield yellow-green fluorescence, while in apoptotic cells, apoptotic bodies are formed as a result of nuclear shrinkage, blebbing and broken fragments of different sizes. Therefore, staining showed intense green particles under a microscope (Fig. 3). With an increase in the concentration of the drug, this phenomenon was evident and the number of the apoptotic cells was increased, which suggested that compound **14** could induce cell apoptosis.

2.3.2. Annexin V-EGFP/PI apoptosis detection

Annexin V-EGFP/PI double staining kit was used to identify the cells in different states. HepG2 cells were double-labeled by

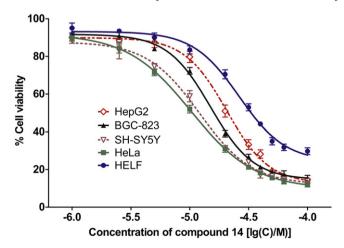


Fig. 2. Dose-response curves of the antiproliferative effect of compound 14 on HepG2, BGC-823, SH-SY5Y, HeLa and HELF cells.

Annexin V-EGFP/PI after they were treated with different concentrations of compound **14**; the control cells were treated with medium. The results are representative of three independent experiments. Four quadrant images were observed by flow cytometric analysis: the B1 area represented damaged cells appearing in the process of cell collection, the B2 region showed necrotic cells and later period apoptotic cells, the B3 area showed normal cells, and the early apoptotic cells were located in the B4 area. Our results (Fig. 4) showed that the early apoptotic cells were significantly more populous than in the untreated group and in a dose-dependent manner. In addition, necrosis and advanced apoptotic cells showed the same characteristics as the early apoptotic cells. The results illustrated that compound **14** suppressed cell proliferation by inducing apoptosis.

2.3.3. Measurement of caspase-3 activity

The occurrence of the apoptotic process was confirmed by the presence of caspase-3 (which plays a crucial role and is a key element in the execution phase of apoptosis). The enzyme activity of caspase-3 was determined by spectrophotometry, as shown in Fig. 5. After incubation with compound **14** for 24 h, the activity of caspase-3 increased compared to the control. The results clearly showed that an induction of cell apoptosis took place when the cells were exposed to the drugs, and the derivative had better effect than UA based on the increased absorbance.

2.4. Investigation of cell cycle distribution

To determine the possible role of cell cycle arrest in UA derivative-induced growth inhibition, HepG2 cells were treated with different concentrations of compound **14**. Cell cycle distribution was observed by flow cytometric analysis after staining of the DNA with propidium iodide (PI). With an increase of concentration of compound **14**, the cells accumulated in the G0/G1 phase, and G2/M phase cells were gradually reduced, while S period cells were

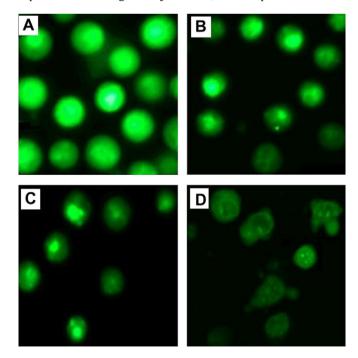


Fig. 3. Fluorescence images of acridine orange-stained HepG2 cells. Apoptotic cells were observed based on morphological changes in HepG2 cells after treatment with compound 14 at different concentrations 0 (control, A), 10 (B), 20 (C), 30 μ M (D) for 24 h.(For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

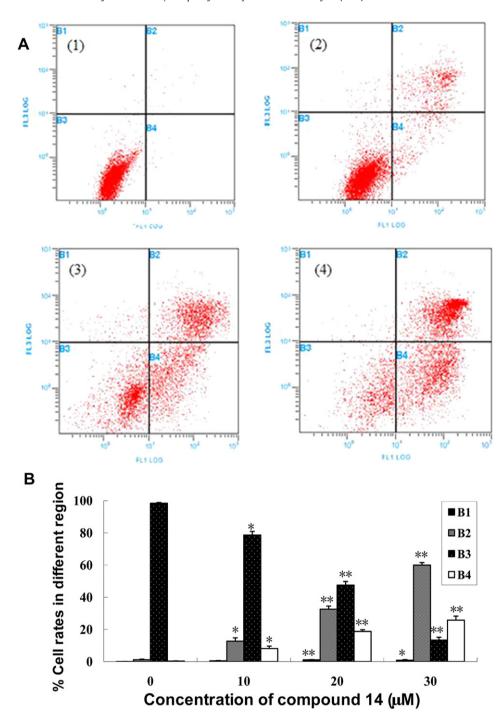


Fig. 4. Annexin V-EGFP/PI apoptosis detection assay. (A) Demonstration of apoptosis or necrosisby flow cytometric analysis. (1) Untreated HepG2 cells. (2)–(4) Cells were treated with increasing concentrations of compound 14 (10, 20 and 30 μM) for 24 h. (B) Dose-dependent induction of apoptosis or necrosisby compound 14. *P < 0.05; **P < 0.01, compared with control.

gradually increased (Fig. 6), which inferred that compound 14 arrested HepG2 cells in S phase.

2.5. Anti-tumorigenic activity of compound 14 against H22 xenograft tumor growth

Model Kunming mice bearing H22 hepatic carcinomas were established to observe the apoptosis of cells induced by UA and derivative compound **14** at different dosages. The medium dosage group (100 mg/kg) of compound **14** showed significant anticancer activity (45.6 \pm 4.3%) as compared to the UA (150 mg/kg)-treated

group (36.2 \pm 3.6%) and cyclophosphamide (CTX)-treated group (55.0 \pm 1.2%), even at the high dosage of 150 mg/kg (43.0 \pm 2.6%) and low dosage of 150 mg/kg (36.6 \pm 1.8%) (Fig. 7D). The bodyweight of this group was also visibly increased (Fig. 7B). These data proved that compound **14** also exerted a preferable proliferation inhibition effect *in vivo*.

3. Conclusion

More and more studies have shown that ursolic acid has significant anti-tumor effects and exhibits cytotoxic activity

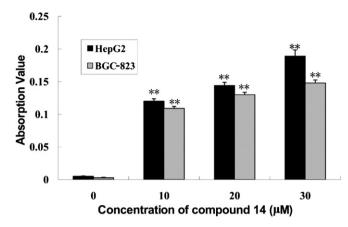


Fig. 5. Measurement of caspase-3 activity. Dose-dependent induction of caspase-3 in human HepG2 and BGC-823 cell lines. Data are mean \pm SD (n=6). *P<0.05; *P<0.01, compared with control.

against various kinds of cancer cell lines *in vitro*, which could be explained through the inhibition of DNA replication, activation of caspases, inhibition of protein tyrosine kinases and/or induction of Ca²⁺ release [14–16]. Another ursolic acid-induced apoptotic mechanism involves down-regulation of the cellular inhibitor of apoptosis gene [19] and the inhibition of NF-κB activity [20].

The above study results indicate that (1) ursolic acid can be modified to more potent bioactive derivatives. The 3-OH or 17-COOH esterified compounds showed decreasing inhibitory activity, and the effect was significantly reduced or even eliminated when they were esterified at the same time. Thus, keeping a polar group at either the 3-OH and/or 17-COOH position is essential for cytotoxic activity. (2) The inhibition effect of the ursolic acid ester compounds on the cancer cells decreases with increasing length of the carbon chain. We infer that long chains may prevent cells from connecting with the mother nucleus. (3) Significant improvement of the cell growth inhibition was achieved when an acetyl group was introduced at the 3-OH position and amino alkyl groups were introduced at the 17-COOH position. (4) The introduction of a more bulky group could increase the steric hindrance and decrease binding to the target. (5) A disubstituted amide structure is more likely to associate with the biomacromolecule through electrostatic interactions than monosubstituted amides. Moreover, a piperazine ring assumes a better position relative to a piperidine ring and improves activity. (6) Preliminary mechanistic studies demonstrated that compound 14 may inhibit cell growth by inducing apoptosis, arresting cell cycle progression at the S phase in HepG2 cells and increasing the activity of caspase-3. (7) The in vivo tests demonstrated that certain UA derivatives can provide better anti-tumor activity than UA. These findings can serve as the basis for further research on the chemical modifications, structure-activity relationships and bioactivity of UA and other triterpenoid acids.

4. Experimental

4.1. General

Ursolic acid was purchased from China Liuyang ATA Natural Product R&D Co., Ltd., in over 90% purity. Silica gel (200–300 mesh) used in column chromatography was provided by Tsingtao Marine Chemistry Co. Ltd. 3-O-acetylursolic acid was prepared according to a previous procedure [1]. Other reagents were obtained from commercial suppliers in analytically pure or chemically pure forms.

Melting points were determined on an electrically heated X-4 digital visual melting point apparatus and are uncorrected. IR spectra were recorded on Shimadzu FTIR-8400S. ¹H NMR spectra were recorded on a BRUKER AV-400 spectrometer with TMS as an internal standard in CDCl₃ or DMSO-d₆. Electrospray ionization (ESI) mass spectra were measured on an Agilent 1100 IC/MSD Trap XCT and are reported as *m*/*z*. Elemental analyses were recorded on Vario Micro elemental analysis apparatus.

4.2. Synthesis

4.2.1. General procedure for the preparation of compounds (8-10)

3-O-acetylursolic acid (1 mmol) and K_2CO_3 (2 mmol) were added to DMF (30 mL) and stirred at room temperature for 4 h. The bromoalkane (4 mmol) was then dripped slowly into the mixture. After being stirred for another 8 h, the reaction mixture was poured into the 100 mL distilled water and partitioned with ethyl acetate (3 \times 50 mL). The organic layer was washed with saturated sodium chloride, dried over Na₂SO₄, and purified via silica gel column chromatography with petroleum ether/ethyl acetate.

4.2.1.1. 3β -Acetoxy-urs-12-en-28-oic acid ethyl ester (**8**). According to the general procedure, 3-O-acetylursolic acid was treated with bromoethane, and then purified on silica gel column using petroleum ether/ethyl acetate (v/v 10:1) as eluent to give compound **8** ($R_{\rm f}=0.59$).

Yield: 87.29%; white powder; mp: 177–180 °C; IR (KBr) υ : 2927, 1735, 1718, 1455, 1374, 1250, 1028 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ : 5.23 (t, J = 3.6 Hz, 1 H, H-12), 4.49 (dd, J = 7.2, 10.0 Hz, 1 H, H-3), 4.04 (q, J = 6.4 Hz, 2 H, CH₂), 2.22 (d, J = 11.6 Hz, 1 H, H-18), 2.03 (s, 3 H, CH₃COO), 1.24 (t, J = 7.6 Hz, 3 H, CH₂CH₃), 1.07 (s, 3 H, CH₃), 0.98 (s, 3 H, CH₃), 0.94–0.91 (s, 6 H, 2 × CH₃), 0.84–0.78 (m, 9 H, 3 × CH₃), 0.76 (s, 3 H, CH₃); ESI-MS m/z: 527.6 (M + H)⁺; Anal. Calcd for C₃₄H₅₄O₄: C 77.52, H 10.33; Found: C 77.41, H 10.27.

4.2.1.2. 3β -Acetoxy-urs-12-en-28-oic acid propyl ester (**9**). According to general procedure, 3-O-acetylursolic acid was treated with 1-bromopropane, and then purified on silica gel column using petroleum ether/ethyl acetate (v/v 10:1) as eluent to afford compound **9** ($R_f = 0.57$).

Yield: 85.31%; amorphous powder; mp: 175–176 °C; IR (KBr) υ : 2948, 1736, 1722, 1456, 1374, 1257, 1028 cm $^{-1}$; 1 H NMR (400 MHz, CDCl $_{3}$) δ : 5.23 (t, J = 3.4 Hz, 1 H, H-12), 4.49 (t, J = 7.6 Hz, 1 H, H-3), 3.73–3.69 (m, 2 H, COOC $_{1}$), 2.29 (d, J = 10.6 Hz, 1 H, H-18), 2.16–2.07 (m, 2 H, CH $_{2}$ C $_{1}$), 1.43–1.24 (m, 6 H, 2 \times CH $_{3}$), 0.99–0.94 (m, 9 H, 3 \times CH $_{3}$), 0.83 (d, 3 H, CH $_{3}$), 0.77 (s, 3 H, CH $_{3}$), 0.75 (s, 3 H, CH $_{3}$); ESI-MS m/z: 541.6 (M + H) $_{1}$; Anal. Calcd for C $_{35}$ H $_{56}$ O $_{4}$: C 77.73, H 10.44; Found: C 77.44, H 10.37.

4.2.1.3. 3β -Acetoxy-urs-12-en-28-oic acid n-butyl ester (**10**). According to the general procedure, 3-O-acetylursolic acid was treated with n-butyl bromide, and then purified on silica gel column using petroleum ether/ethyl acetate (v/v 10:1) as eluent to obtain compound **10** ($R_{\rm f} = 0.56$).

Yield: 86.57%; amorphous powder; mp: 107–109 °C; IR (KBr) υ: 2941, 1735, 1716, 1455, 1372, 1244, 1027 cm $^{-1}$; 1 H NMR (400 MHz, CDCl₃) δ: 5.22 (t, J=3.4 Hz, 1 H, H-12), 4.48 (t, J=7.8 Hz, 1 H, H-3), 4.05–3.99 (m, 2 H, COOCH₂), 2.22 (d, J=11.2 Hz, 1 H, H-18), 2.03 (s, 3 H, CH₃COO), 2.00–1.94 (m, 2 H, CH₂CH₂), 1.60–1.55 (m, 2 H, CH₂CH₂), 1.43–1.19 (m, 12 H, 4 × CH₃), 1.08 (s, 3 H, CH₃), 0.98–0.82 (m, 6 H, 2 × CH₃), 0.76 (s, 3 H, CH₃); ESI-MS m/z: 555.6 (M + H) $^+$; Anal. Calcd for C₃₆H₅₈O₄: C 77.93, H 10.54; Found: C 77.67, H 10.49.

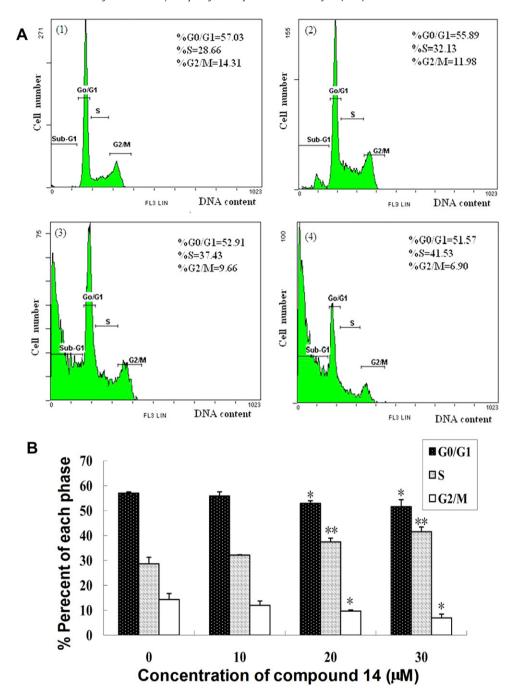


Fig. 6. Investigation of cell cycle distribution. (A) Demonstration of apoptosis by flow cytometric analysis. (1) Untreated HepG2 cells. (2)–(4) Cells was treated with increasing concentrations of compound 14 (10, 20 and 30 μ M) for 24 h. (B) Dose-dependent induction of apoptosis by compound 14. *P < 0.05; **P < 0.01, compared with control.

4.2.2. General procedure for the preparation of N-[3 β -acetoxy-urs-12-en-28-oyl]-amino acid methyl-esters, alkamines, piperazidines, piperidine (14–17)

Oxalyl chloride (0.6 mL) was added in three equal portions to a CH_2Cl_2 solution of (insert the related compound number) (1 mmol) and stirred at room temperature for 36 h. The mixture was concentrated to dryness under reduced pressure to obtain crude 3-O-acetylursolyl chloride. This intermediate was dissolved in a CH_2Cl_2 (20 mL) solution, and then treated with amino acid methylester, alkamines, piperazidines or piperidine. The mixture was stirred at room temperature for 4 h and treated with triethylamine until the pH was 8–9. Next, the mixture was concentrated to dryness. The residue was partitioned in 10 mL water and adjusted with 2 N HCl to

pH 3—4 and filtered. The crude material was purified on a silica gel column with the appropriate eluent to yield compounds **14**—**17**.

4.2.2.1. N-[3 β -acetoxy-urs-12-en-28-oyl]-2-aminodiethanol (14). According to the general procedure, 3-O-acetylursolic acid was treated with diethanolamine, and then purified on silica gel column using petroleum ether/ethyl acetate (v/v 1:3) as eluent to give compound 14 ($R_{\rm f}=0.53$).

Yield: 50.37%; white powder; mp: 226–228 °C; IR (KBr) υ : 3392, 2946, 1734, 1595, 1371, 1245 cm $^{-1}$; 1 H NMR (400 MHz, DMSO-d $_{6}$) δ : 5.08 (t, J = 3.6 Hz, 1 H, H-12), 4.42 (t, J = 6.8 Hz, 1 H, H-3), 3.42 (t, J = 9.2 Hz, 2 H, N (CH $_{2}$) $_{2}$), 2.33 (t, J = 6.8 Hz, 1 H, CH $_{2}$ CH $_{2}$), 2.00 (s, 3 H, CH $_{3}$ COO), 1.04 (s, 3 H, CH $_{3}$), 0.91–0.90 (m, 6 H, 2 \times CH $_{3}$),

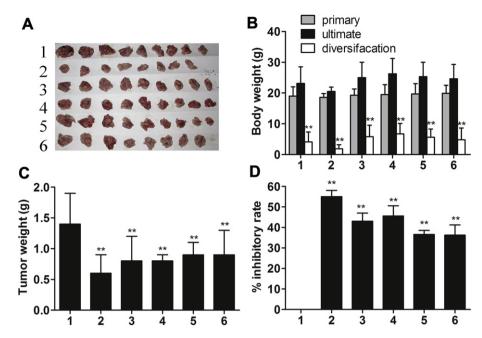


Fig. 7. Effect of anti-tumor in vivo. H22 tumor-bearing mice were administered i.p. 7 days with normal saline (1) CTX at 15 mg/kg (2) compound 14 at 50, 100 and 150 mg/kg (3–5) UA at 150 mg/kg (6) The figure shows the tumor mode (A) and body weight (B) of the control and therapeutic groups. (C) shows the tumor weight and (D) shows the inhibitory rates after administration. Data represent means \pm S.D. from three independent experiments performed in duplicate. *P < 0.05; **P < 0.01 vs. the normal saline (1).

0.83-0.81 (m, 9 H, 3 \times CH₃), 0.70 (s, 3 H, CH₃); ESI-MS m/z: 586.9 (M + H) $^+$; Anal. Calcd for C₃₆H₅₉NO₅: C 73.8, H 10.15, N 2.39; Found: C 73.59, H 10.11, N 1.98.

4.2.2.2. N-[3 β -acetoxy-urs-12-en-28-oyl]-amino-N- (2-hydroxyethyl) piperazine (**15**). According to the general procedure, 3-O-acety-lursolic acid was treated with N- (2-Hydroxyethyl) piperazine, and then purified on silica gel column using chloroform/acetone (v/v 3:1) to give compound **15** (R_f = 0.43).

Yield: 46.81%; white powder; mp: 196–198 °C; IR (KBr) υ : 3417, 2943, 1724, 1595, 1368, 1251 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ : 5.22 (s, 1 H, H-12), 4.48 (t, J=6.0 Hz, 1 H, H-3), 3.63 (t, J=5.2 Hz, 2 H, CH₂OH), 3.22 (t, J=3.2 Hz, 1 H, H-3), 2.55 (t, J=5.2 Hz, 2 H, -N (CH₂)₂), 2.47 (br, 4 H, -N (CH₂CH₂)₂), 2.04 (s, 3 H, CH₃CO), 1.07 (s, 3 H, CH₃), 0.95 (s, 3 H, CH₃), 0.93 (d, J=6.4 Hz, 3 H, CH₃), 0.91 (s, 3 H, CH₃), 0.86 (d, J=6.4 Hz, 3 H, CH₃), 0.85 (s, 3 H, CH₃), 0.75 (s, 3 H, CH₃); LC-MS m/z 611.6 (M + H)⁺, 633.6 (M + Na)⁺; Anal. Calcd for C₃₈H₆₂N₂O₄: C74.71, H 10.23, N 4.59; Found: C 74.66, H 10.16, N 4.08.

4.2.2.3. N-[3 β -acetoxy-urs-12-en-28-oyl]-amino-1-hydroxyethylethoxy piperazine (**16**). According to the general procedure, treatment of 3-O-acetylursolic acid with 1-hydroxyethylethoxy piperazine, and then purified on silica gel column using chloroform/acetone (v/v 3:1) afforded compound **16** ($R_f = 0.36$).

Yield: 37.87%; white powder; mp: 105–107 °C; IR (KBr) υ : 3446, 2945, 1734, 1633, 1370, 1246 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ: 5.21 (s, 1 H, H-12), 4.48 (t, J = 4.0 Hz, 1 H, H-3), 3.71–3.61 (m, 8 H, 4 × C $\underline{\text{H}}_2$), 2.58 (t, J = 5.2 Hz, 2 H, -N (C $\underline{\text{H}}_2$)₂), 2.48 (br, 4 H, -N (CH₂C $\underline{\text{H}}_2$)₂), 2.04 (s, 3 H, C $\underline{\text{H}}_3$ CO), 1.06 (s, 3 H, CH₃), 0.93 (d, J = 6.0 Hz, 3 H, CH₃), 0.86–0.84 (m, 9 H, 3 × CH₃), 0.84 (s, 3 H, CH₃), 0.73 (s, 3 H, CH₃); LC-MS m/z 656.1 (M + H)⁺, 677.6 (M + Na)⁺; Anal. Calcd for C₄0H₆6N₂O₅·H₂O: C 73.35 (71.39), H 10.16 (9.89), N 4.28 (4.17); Found: C 71.86, H 10.03, N 3.74.

4.2.2.4. N-[3 β -acetoxy-urs-12-en-28-oyl]-amino-4-piperidineethanol (17). According to the general procedure, treatment of 3-O-

acetylursolic acid with 4-piperidineethanol, and then purified on silica gel column using chloroform/acetone (v/v 3:1) afforded compound 17 ($R_{\rm f}=0.61$).

Yield: 58.72%; white powder; mp: 138–140 °C; IR (KBr) υ:3409, 2943, 1728, 1585, 1367, 1253 cm $^{-1}$; 1 H NMR (400 MHz, CDCl $_{3}$) δ: 5.22 (s, 1 H, H-12), 4.48 (d, J = 4.0 Hz, 1 H, H-3), 4.37 (t, J = 6.4 Hz, 2 H, CH $_{2}$ OH), 3.72–3.70 (m, 2 H, CH $_{2}$ CH $_{2}$ OH), 2.73–2.70 (m, 4 H, -N-(CH $_{2}$) $_{2}$), 2.43 (s, 3 H, CH $_{3}$ COO), $\overline{2}$.04 (d, J = 3.6 Hz, 1 H, H-18), 1.51–1.49 (m, 1 H, (CH $_{2}$) $_{3}$ -CH $_{2}$ -), 1.30–1.27 (m, 4 H, -N (CH $_{2}$ CH $_{2}$) $_{2}$ CH $_{2}$ -), 1.07 (s, 3 H, CH $_{3}$), 0.94 (s, $\overline{3}$ H, CH $_{3}$), 0.86–0.85 (m, 12 H, 4 × CH $_{3}$), 0.76 (s, 3 H, CH $_{3}$); LC-MS m/z 610.6 (M + H) $_{2}$ +, 632.6 (M + Na) $_{2}$ +; Anal. Calcd for C $_{39}$ H $_{63}$ NO $_{4}$: C 76.8, H 10.41, N 4.28; Found: C 76.75, H 10.32, N 4.21.

4.2.3. General procedure for the synthesis of N-[3β -hydroxy-urs-12-en-28-oyl]- amino acids alkamines, piperazidines, piperidine (**22–24**)

A solution of **15–17** in aqueous NaOH (4 N) in CH₃OH:THF (1:1.5, v: v) was stirred for 4 h at room temperature and concentrated under reduced pressure. The residue was suspended in water, adjusted with 2 N HCl to pH 3–4 and then filtered. The filter was washed with water to pH 6–7, and the material was dried to obtain compounds **22–24**.

4.2.3.1. Synthesis of $N-[3\beta-hydroxy-12-en-28-oyl]-amino-N-$ (2-Hydroxyethyl)piperazine (22). According to the general procedure, compound 22 was obtained by hydrolysis of compound 15. R_f [chloroform/acetone (v/v 4:1)] = 0.29.

Yield: 81.38%; mp: 208–210 °C; IR (KBr) υ: 3380, 2926, 1635, 1368 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ: 5.22 (s, 1 H, H-12), 3.63 (t, J=6.0 Hz, 2 H, CH₂OH), 3.22 (t, J=3.2 Hz, 1 H, H-3), 2.55 (t, J=5.2 Hz, 2 H, -N (CH₂)₂), 2.47 (br, 4 H, -N (CH₂CH₂)₂), 2.17 (t, J=3.6 Hz, 1 H, H-18), 1.07 (s, 3 H, CH₃), 0.98 (s, 3 H, CH₃), 0.94 (d, J=6.0 Hz, 3 H, CH₃), 0.91 (s, 3 H, CH₃), 0.87 (d, J=6.4 Hz, 3 H, CH₃), 0.77 (s, 3 H, CH₃), 0.75 (s, 3 H, CH₃); LC-MS m/z 570.0 (M + H)⁺, 591.5 (M + Na)⁺; Anal. Calcd for C₃₆H₆₀N₂O₃: C 76.01, H 10.63, N 4.92; Found: C 75.59, H 10.59, N 4.81.

4.2.3.2. N-[3 β -acetoxy-urs-12-en-28-oyl]-amino-1-hydroxyethylethox-ypiperazine (**23**). According the general procedure, compound **23** was obtained by hydrolysis of compound **16**. R_f [chloroform/acetone (v/v 4:1)] = 0.27.

Yield: 87.26%; mp: 111–113 °C; IR (KBr) υ : 3414, 2925, 1616, 1376, 1227 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ : 5.21 (s, 1 H, H-12), 3.71 (t, J=4.4 Hz, 2 H, CH₂CH₂OH), 3.71–3.61 (m, 8 H, 4 \times CH₂), 2.58 (t, J=5.2 Hz, 4 H, -N (CH₂)₂), 2.49 (br, 4 H, -N (CH₂CH₂)₂), 1.05 (s, 3 H, CH₃), 0.94–0.93 (m, 6 H, 2 \times CH₃), 0.88–0.85 (m, 9 H, 3 \times CH₃), 0.76 (s, 3 H, CH₃); LC-MS m/z 613.6 (M + H)⁺, 635.6 (M + Na)⁺; Anal. Calcd for C₃₈H₆₄N₂O₄: C 74.46, H 10.52, N 4.57; Found: C 74.21, H 10.44, N 4.41.

4.2.3.3. N-[3 β -acetoxy-urs-12-en-28-oyl]-amino-4-piperidineeth-anol (**24**). According the general procedure, compound **24** was obtained by hydrolysis of compound **17**. R_f [chloroform/acetone (v/v 8:1)] = 0.34.

Yield: 89.32%; mp:201–203 °C; IR (KBr) υ: 3426, 2926, 1610, 1369, 1270 cm $^{-1}$; 1 H NMR (400 MHz, CDCl $_{3}$) δ: 5.22 (s, 1 H, H-12), 4.37–4.35 (m, 2 H, CH $_{2}$ CH $_{2}$ OH), 3.71 (t, J = 6.0 Hz, 2 H, CH $_{2}$ OH), 3.21 (t, J = 3.2 Hz, 1 H, H-3), 2.76–2.71 (m, 4 H, -N- (CH $_{2}$) $_{2}$), 2.18 (t, J = 10.8 Hz, 1 H, H-18), 1.54–1.49 (m, 1 H, (CH $_{2}$) $_{3}$ -CH $_{2}$), 1.31–1.25 (m, 4 H, -N (CH $_{2}$ CH $_{2}$) $_{2}$ CH-), 1.07 (s, 3 H, CH $_{3}$), 0.98 (s, 3 H, CH $_{3}$), 0.94 (d, J = 6.0 Hz, 3 H, CH $_{3}$), 0.91 (s, 3 H, CH $_{3}$), 0.87 (d, J = 6.4 Hz, 3 H, CH $_{3}$), 0.78 (s, 3 H, CH $_{3}$), 0.76 (s, 3 H, CH $_{3}$); LC-MS m/z 568.6 (M + H) $^{+}$, 590.6 (M + Na) $^{+}$; Anal. Calcd for C $_{37}$ H $_{61}$ NO $_{3}$: C 78.25, H 10.83, N 2.47; Found: C 78.01, H 10.79, N 2.32.

4.3. Single crystal X-ray diffraction

Single crystal X-ray diffraction experiment was performed on Rigaku Saturn-724, equipped with graphite-monochromated Mo-K α radiation ($\lambda=0.71073$ Å) at room temperature. The structures were solved by direct methods and refined on F2 by full-matrix least-squares methods using the SHELXTL-97 program package. All non-hydrogen atom positions were located using difference Fourier methods as implemented in SHELXL-97. All H atoms were placed in calculated positions on their corresponding O and C atoms.

4.4. Cell lines and culture

HepG2, BGC-823, SH-SY5Y, HeLa and HELF cells were all obtained from our own laboratory. They were maintained in RPMI 1640 medium supplemented with 10% FBS, penicillin (100 U/mL), and streptomycin (100 μ g/mL) in a humidified atmosphere of 5.0% CO₂ at 37 °C.

4.5. MTT assay for cell viability/proliferation

The cytotoxicity of the derivatives was determined by the MTT assay. Cells (1 \times 10^4/well) were plated in 100 μL of medium/well in 96-well plates. After incubation overnight, the cells were treated with different concentrations of drugs in RPMI 1640 with 10% FBS for 24 h. Finally, MTT was added, and the cells were incubated for another 4 h. The MTT-formazan formed by metabolically viable cells was dissolved in 150 mL of DMSO and shaken for 10 min. The absorbance was then measured on an ELISA reader at a test wavelength of 490 nm. Each test was repeated at least three times. The concentration of the compound, which gives the 50% growth inhibition value, corresponds to the IC50.

4.6. Apoptosis tests

4.6.1. Morphologic evaluation by fluorescence microscopy

Cells incubated with different concentrations of compound 14 (10–30 μM) for 24 h were collected, washed with PBS and

suspended in solution at a cell density of $1\times10^6/mL$. Then, 95 μL of cell suspension was stained with 5 μL (200 $\mu g/mL$ in PBS) of AO for 15 min at room temperature in the dark. Stained cells were visualized with a Nikon fluorescence microscope using a 488 nm excitation while measuring fluorescence at 515 nm.

4.6.2. Annexin V-EGFP/PI double staining

Cells in different states were observed with an annexin V-EGFP/Pl double staining kit. HepG2 cells were treated with different concentrations of compound $14~(10-30~\mu\text{M})$ for 24 h. After incubation, a total of 3×10^5 cells were harvested from the treated and normal samples. The cells were washed twice with PBS, and then $5~\mu\text{L}$ of annexin V-EGFP and propidiumiodide (PI) were added in the dark for 15 min. The cells were then analyzed by analytical flow cytometry using a Cell Lab Quanta SC (Beckman, USA). All experiments were performed three times.

4.6.3. Caspase-3 enzyme assay

The activity of the caspase-3 protein was determined following the test kit instructions (R&D Systems, Inc.). Twenty-four hours after culturing, $5\times 10^5-10^6$ cells were collected, washed twice with PBS and centrifuged at 500 g for 5 min. Then, 25 μ L of lysis buffer was added to the cells on ice for 10 min, and then 50 μ L of 2 \times reaction buffer 3 and 5 μ L of caspase-3 substrate (DEVD-pNA) were added. After incubating at 37 °C for 1–2 h in darkness, the absorbance was measured at 405 nm, with the lysis buffer and reactionbuffer (25 μ L lysis buffer + 50 μ L 2 \times reaction buffer) as control.

4.7. Cell cycle assay

The cell cycle was analyzed by flow cytometry. Briefly, HepG2 cells were treated with different concentrations of compound $14\,$ (10–30 $\mu M)$ for 24 h. After incubation, a total of 1 \times 10 8 cells were harvested from the treated and normal samples. The cells were washed twice with PBS and fixed in 75% ice-cold ethanol for at least overnight. The sample was concentrated by removing the ethanol and staining the cellular DNA with fluorescent solution (1% (v/v) Triton X-100, 0.01% RNase, 0.05% PI) for 30 min at 4 $^\circ$ C in darkness. The cell cycle distribution was then detected by flow cytometry. All experiments were performed three times.

4.8. Evaluation of therapeutic effect in vivo

Kunming mice (16–19 g body weight, 6–8 weeks old) were maintained under identical laboratory conditions and given standard food pellets and water ad libitum. After inoculation with murine hepatoma H22 cells for 24 h, tumor-bearing mice were randomized into six experimental groups, each with 10 mice. All compounds were dissolved in 10% DMSO solution. Group I was the negative control (normal saline) and group II was the positive control (CTX; 1.5 mg/mL). Group III-VI was treated with 20, 40 or 60 μ M compound 14 and UA (60 μ M), respectively. All of the compounds were administered at a dose of 0.2 ml/10 g bodyweight for seven successive days. At the end of the experiment, the animals were sacrificed and the tumors were dissected and weighed. The tumor inhibitory rates were then calculated.

4.9. Statistical analysis

The data were presented as means \pm standard deviations of three determinations. Statistical analysis was performed using Student's t-test and one-way analysis of variance. Multiple comparisons of the means were done by the least significance difference (LSD) test. A probability value of <0.05 was considered

significant. All computations were made by employing SPSS statistical software (version 16.0).

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Appendix. Supplementary data

Supplementary data associated with this article can be found in online version at doi:10.1016/j.ejmech.2011.03.050.

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