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Original article

Discovery of azaisoerianin derivatives as potential antitumors agents



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ABSTRACT

A series of N-methyl-diarylamines ${\bf 2}$ was designed and synthesized as a novel class of CA-4 and isoCA-4 analogues. Compounds ${\bf 2b}$ and ${\bf 2m}$ showed excellent antiproliferative activity with mean GI_{50} values at a nanomolar level in a diverse set of human cancer cells. These compounds also inhibited tubulin assembly at a micromolar range, arrested the cellular cycle in the G2/M phase and induced apoptosis at very low concentrations. Preliminary $in\ vitro$ results revealed that ${\bf 2b}$ and ${\bf 2m}$ displayed substantial efficacy as potent antivascular agents. Docking studies indicates that these lead compounds showed a binding mode similar to those observed with isoCA-4 at the colchicine binding site of tubulin.

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1. Introduction

Combretastatin A-4 (CA-4) a natural stilbene isolated from the African willow tree, *Combretum caffrum* in 1989 by Pettit [1] has been found to be a potent anticancer agent which strongly inhibits tubulin assembly by binding to the colchicine site. [2] CA-4 also shows potent cytotoxicity against a wide range of human cancer cells including multi drugs resistant (MDR) cancer cells [3]. Moreover, CA-4 has been demonstrated to selectively target the vascular network of tumors, inducing irreversible shutdown of blood flow to neoplastic cells [4]. Currently, two water-soluble prodrug derivatives are in clinical trials: disodium phosphate CA-4P (fosbretabulin, ZybrestatTM) [5] and AVE8062 (ombrabulin) [6]. To date, CA-4P [7,8] either as a single agent or in combination therapy is undergoing several advanced clinical trials worldwide for the treatment of age-related macular degeneration or anaplastic thyroid cancer.

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Despite their remarkable anticancer properties, the main problem associated with CA-4 and its prodrugs is the ready isomerization of the Z-double bond into the inactive E-form during storage, administration and metabolism [9,10]. In an ongoing project aimed at developing stable CA-4 analogues [11-19], we recently solved the stability problem of CA-4 by the discovery of the non-isomerizable isocombretastatin A-4 (isoCA-4) which holds biological activities comparable to that of CA-4 [20,21]. This structural isomer of the natural product, having a 1,1diarylethylene scaffold, is chemically and metabolically stable [22] and easy to synthesize [23–25] at a multigrams scale without the need to control the olefin geometry. By chemical modifications on the B-ring of isoCA-4, we have also identified other stable and promising antiproliferative agents [21,26] such as isoNH₂CA-4 and isoFCA-4. We next have demonstrated that the reduction of the 1,1-ethylene bond of isoCA-4 led to (\pm) -isoerianin, in which racemic and both enantiomers displayed excellent anti-cancer activities comparable to that of the natural isomer erianin [27]. It seems that the carbon (Csp³) linker in isoerianin is not involved in the interaction in the colchicine binding site. It serves only to position correctly the two aromatic rings of the molecule [27]. Herein, we planned to prepare a new class of isoerianin derivatives, named azaisoerianins (Fig. 1), in which the two aromatic

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Fig. 1. Structures of CA-4 and synthetic prodrugs, isoCA-4 and derivatives, erianin, isoerianin and targeted azaisoerianin.

rings are connected through a nitrogen atom in place of the carbon linker in isoerianin compounds. Indeed, the diphenylamine skeleton is a common privileged structure found in biologically active compounds, and compounds derived from this scaffold manifest diverse activities, including anticancer activity [28,29]. This design has several advantages as follow: (i) the ease in synthesis where various side-arms can be introduced on nitrogen atom in a convergent access, (ii) the absence of a chiral center, (iii) the rapid access to weakly basic and chargeable derivatives in a solubilization strategy, (iv) the discovery of a third generation of candidates having promising antiproliferative and antitubulin activities and a better understanding of structure-activity relationships (SAR). In this article are described the synthesis and the biological evaluation of novel azaisoerianin derivatives 2 and **3**. Preliminary in vitro efficacy of these compounds in terms of cytotoxicity, inhibition of tubulin assembly, cell cycle, apoptosis and in vitro antivascular activity is described. The possible binding mode of the most bioactive substrates 2b and 2m on tubulin is also reported.

2. Results and discussion

2.1. Chemistry

The synthesis of azaisoerianin derivatives 2a-q and 3a-c, is summarized in Scheme 1. Diarylanilines 1a-1 were prepared from commercially available methoxylated anilines which were coupled in a sealed tube with a variety of (hetero)aryl halides under our previously reported conditions (Pd(OAc)₂ as the catalyst, Xantphos as the ligand, Cs₂CO₃ as the base in hot dioxane) [30-33]. Treatment of the resulting diarylamines 1 with sodium hydride followed by reaction with MeI gave the best results for the N-alkylation process furnishing the desired trisubstituted amines 2. By analogy with highly cytotoxic N-propylbenzylanilines, [34] N-propyl derivatives 3 were prepared using a large excess of nPrBr as the alkylating agent and NaH as the base. Reduction of the nitro group of **2c** using HCl and iron powder yielded azaisoerianin derivative **2d** with 3-amino-4-methoxy groups on the B-ring identical to the Bring of AVE8062 drug. Debenzylation of the phenol function of 2f was carried out by hydrogenation to afford azaisoerianin 2g having the greatest resemblance with isoerianin and isoCA-4. Compound **2j** was synthesized from 4-chloro derivative **2i** using dimethylamine in the presence of Pd₂(dba)₃.CHCl₃ (5 mol%), and Johnphos (10 mol%) as the ligand and *t*BuONa as the base in hot toluene [35]. Finally, the removal of the *N*-protected tosyl group of **2p** was accomplished by the use of TBAF in DMF at 100 °C to afford **2q** (49%).

2.2. Biological results

New synthesized azaisoerianin compounds were evaluated for their cytotoxic effects against HCT116 human carcinoma cell line and for tubulin polymerization inhibitory activity. *Iso*erianin [27], *iso*CA-4 [21] and CA-4 [36] were included as reference compounds. The cytotoxicity and inhibition of tubulin polymerization (ITP) results are presented in Table 1. Without any exception, all secondary diarylamines 1 were not cytotoxic (data not shown).

In the series of compounds 2 having a benzene substituted Bring, our findings showed that compound 2b, having a fluorine atom at the C3'-position, displayed the best cytotoxicity with a GI₅₀ values of 20 nM comparable to that of the reference isoerianin. Additionally, the water-soluble hydrochloride salt 2b-HCl displayed a high level of cytotoxicity against HCT116 cells ($GI_{50} = 28 \text{ nM}$) comparable to that of the parent molecule 2b (GI₅₀ = 22 nM). Replacement of the F atom with H or a NH2 on the C3'-position generated compounds 2a and 2d (GI₅₀ = 64 nM and 66 nM, respectively) which were three fold less active than 2b. Contrary to our expectations, a dramatic loss of antiproliferative activity was observed with azaisoerianin 2g, having a OH group on C3'-position. This compound and its corresponding acetate 2h displayed only modest cytotoxicity against HCT116 cells (GI₅₀ = 650-700 nM). We next showed that the introduction of a pyridine or a quinoline nucleus as B-ring resulted in products 2k ($GI_{50} = 650$ nM) and **2l** ($GI_{50} = 2100$ nM), respectively, with a significant loss of potency. We next investigated the role on cytotoxicity of derivatives having an indole nucleus which was found useful as B-ring in the CA-4 series [37]. It is significant to note that in the azaisoerianin series, N-methyl-indol-5-yl compound 2m was consistently the more potent compound. This lead compound (2m) was about four times more cytotoxic than the reference isoerianin with a low GI₅₀ value of 7 nM. The absence of the trimethoxyphenyl-ring in the 5indoyl series reduced the cytotoxicity as it was observed with

$$\begin{array}{c} R^1 \\ R^2 \\ R^3 \end{array} \qquad \begin{array}{c} \text{3a: } R^1 = R^2 = R^3 = \text{OMe, } R^4 = \text{OBn} \\ \text{3b: } R^1 = R^2 = R^3 = \text{OMe, } R^4 = \text{OH} \\ \end{array} \qquad \begin{array}{c} R^1 \\ R^2 \\ R^1 \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\ \end{array} \qquad \begin{array}{c} \text{3c: } R^1 = R^2 = \text{OMe} \\$$

"Reagents and conditions: (a) Pd(OAc)₂, Xantphos, ArX, Cs₂CO₃, dioxane, 100-130 °C, sealed tube. (b) H₂, Pd/C, AcOEt. (c) NaH, MeI, DMF, RT. (d) HCl, MeOH, Et₂O, RT. (e) Fe, HCl, EtOH/H₂O. (f) Pyridine, DMAP, (Ac)₂O, CH₂Cl₂. (g) Pd₂(dba)₃CHCl₃, t-Bu₂P(o-biphenyl), NHMe₂, tBuONa, toluene, 100 °C. (h) TBAF, DMF, 100 °C. (i) NaH, nPrBr, DMF, RT.

Scheme 1. Synthesis of compounds 1–3.

compounds $\mathbf{2n}$ (GI₅₀ = 1100 nM) and $\mathbf{2o}$ (GI₅₀ = 70 nM). Contrary to other observations, [37] the N-methyl substitution of the indole moiety in **2m** plays an important role for the antiproliferative activity. Indeed, its des-methyl analogue 2q showed a 25-fold reduction in cytotoxicity. Finally, a marked loss in cytotoxicity was observed with compounds 3b (GI₅₀ = 1400 nM) and 3c $(GI_{50} = 200 \text{ nM})$ having a N-propyl group in comparison with their N-methyl analogues (e.g. 3c vs 2m; 3b vs 2g). To further characterize the cytotoxicity profiles of these compounds, we next investigated the effect of the most active substances 2a, 2b, 2d and **2m** on the proliferation of three tumor cell lines (hormone-independent breast cancer (MDA-MB231)), human glioblastoma (U87-MG), myelogenous leukemia (K562) and on the normal primary human umbilical vein endothelial (HUVEC). As shown in Table 2, all examined compounds displayed very potent activity against all cell lines tested, in the nanomolar range. Again, compound 2b and indole 2m displayed the highest level of cytotoxicity with GI₅₀ values ranging from 2 to 39 nM.

To investigate whether compounds 2 and 3 exert their activities by interacting with microtubules, their effect on the in vitro polymerization of purified tubulin was next examined (see Table 1). It is of interest to note that highly cytotoxic derivatives 2b and Nmethylindole 2m were also very potent as tubulin inhibitors and were comparable to isoerianin with IC₅₀ μM values (2.7 μM and 1.3 μM for **2b** and **2m**, respectively). Interestingly, derivative **2g**, having the greatest resemblance to isoCA-4 and isoerianin which was poorly cytotoxicity against HCT116 cells was one of the more potent agent as tubulin inhibitor with an IC₅₀ value of 1.5 μ M. Similarly, N-propyl derivatives 3b and 3c which displayed a modest level of cytotoxicity were very potent in the tubulin assay with IC₅₀ values ranging from 2.8 to 3.6 μM . These results showed an important discrepancy between the potency of antitubulin activity and cytotoxicity of N-propyl derivatives 3, indicating that with hindered substituents on the nitrogen atom of derivatives 3, it might be possible to design news series of antivascular agents with low cytotoxicity.

Table 1Cytotoxicity against HCT 116 cells^a and ITP of selected compounds **2** and **3**.

	MeO OMe	MeO N F OMe	MeO Ne OMe	MeO NO ₂ OMe
Compound Cytotoxicity GI ₅₀ ^b [nM]	2a 64 ± 4	2b 20 ± 1.5	2b.HCl 28 ± 2	2c 190 ± 15
ITP IC ₅₀ ^c [μΜ)	5.5 ± 0.5	2.7 ± 0.2	3 ± 0.2	28 ± 2
	MeO NH ₂ OMe	MeO N N N N N N N N N N N N N N N N N N N	MeO N OH OMe	MeO N OAc OMe
Compound Cytotoxicity GI_{50}^{b} [nM]	$ 2\mathbf{d} \\ 66 \pm 5 $	2e 170 ± 12	2g 650 ± 55	2h 700 ± 55
ITP IC_{50}^{c} [µM)	3.0 ± 0.2	170 ± 12 14 ± 1.8	1.5 ± 0.2	8.0 ± 0.5
	MeO NMe ₂	MeO N OMe	MeO N N N O O Me	MeO N
Compound Cytotoxicity GI ₅₀ ^b [nM]	2j 250 ± 25	2k 650 ± 62	2l 2100 ± 180	2m 7 ± 0.6
ITP IC ₅₀ ^c [μM]	18 ± 1.2	NA ^d	NA ^d	1.3 ± 0.1
	MeO N	MeO N N N N N N N N N N N N N N N N N N N	MeO N N N N N N N N N N N N N N N N N N N	MeO N OH OMe
Compound Cytotoxicity GI ₅₀ ^b [nM]	2n 1100 ± 120	2o 70 ± 5	2q 180 ± 14	$egin{array}{ccc} {f 3b} & {}^{f OMe} \\ 1400 \pm 145 \end{array}$
ITP IC ₅₀ ^c [μM]	NA^d	3.4 ± 0.5	NA ^d	2.8 ± 0.3
	MeO N N N N O O Me	MeO OMe	MeO OMe	MeO OMe OH
Compound Cytotoxicity IC ₅₀ ^b [nM]	3c 200 ± 16	iso CA-4 2 ± 0.2^{e}	isoerianin 28 ± 2^{e}	CA-4 2 ± 0.2^{e}
ITP IC_{50}^{c} [μM]	3.6 ± 0.5	2 ± 0.2 2.0 ± 0.3	3.0 ± 0.2^{e}	2 ± 0.2 1.0 ± 0.2^{e}

- ^a HCT116 Human colon carcinoma.
- b Gl₅₀ is the concentration of compound needed to reduce cell growth by 50% following 72 h cell treatment with the tested drug (average of three experiments).
- c IC₅₀ is the concentration of compound required to inhibit 50% of the rate of microtubule assembly (average of three experiments).
- d NA not active.
- ^e The GI₅₀ values for isoCA-4, isoerianin and CA-4 were determined in this study.

To gain further insight the mechanism of action of azaisoerianin derivatives **2**, the most cytotoxic compounds **2b** and **2m** were assayed for their effects on cell cycle distribution. K562 and HCT116 cells were treated with **2b** and **2m** at different concentrations for

Table 2Cytotoxicity against human cancer cell lines, HCT116, MDA-MB231, U87-MG, K562 and HUVEC.

Compounds	Cytotoxicity (GI ₅₀ nM) ^a							
	HCT116 ^b	MDA-MB231 ^c	U87-MG ^d	K562 ^e	HUVECf			
2a	64 ± 4	55 ± 4	30 ± 2	39 ± 4	35 ± 3			
2b	20 ± 1.5	24 ± 2	7 ± 0.6	39 ± 5	22 ± 2			
2d	66 ± 5	80 ± 7	20 ± 1.7	62 ± 5	25 ± 3			
2m	7 ± 0.6	16 ± 1.8	2.2 ± 0.3	8 ± 0.7	7 ± 0.5			
<i>Iso</i> erianin ^g	28 ± 2	35 ± 3	30 ± 3	25 ± 2	45 ± 5			

- ^a GI₅₀ is the concentration of compound needed to reduce cell growth by 50% following 72 h cell treatment with the tested drug (average of three experiments).
 ^b HCT116 colon carcinoma
 - ^c MDA-MB-231 hormone-independent breast cancer.
- d U87-MG Glioblastoma.
- ^e K562, myelogenous leukaemia.
- f HUVEC, Human umbilical vein endothelial cells.
- ^g The Gl₅₀ values of *iso*erianin was determined in this study.

24 h. The results presented in Fig. 2, demonstrate that there was an accumulation of K562 and HCT116 cells in the G_2/M phase of the cellular cycle after treatment with ${\bf 2b}$ and ${\bf 2m}$ at very low concentrations. One note that there is a better perturbation of cycle cell progression in K562 cells after treatment with these drugs, and indolic derivative ${\bf 2m}$ seems to be slightly more effective than ${\bf 2b}$ in K562 and HCT116 cells.

The ability of **2b** and **2m** to induce apoptosis in HCT116 cells was further characterized using caspase-3 and caspase-7 assays. The enzymatic activity of these caspases was measured by monitoring the cleavage of the fluorogenic substrate Z-DEVD-R110 in cancer cells. The results presented in Table 3 show a significant dose-dependent increase in proteolytic activity of caspases in the cells treated for 24 h with diarylmethylamines **2b** and **2m**. A spectacular 4-fold increase in apoptosis was evidenced by **2m** in HCT116 cells at a low concentration of 7.5 nM. These results indicate that, in addition to their antiproliferative and antitubulin effects, the treatment of cancer cells with **2b** and **2m** activate caspases system leading to programmed cell death.

To identify compounds that disrupt the neovascularization formation, an assay using endothelial colony-forming cells (ECFCs) was performed with a panel of diarylamines **2** and **3**. In the

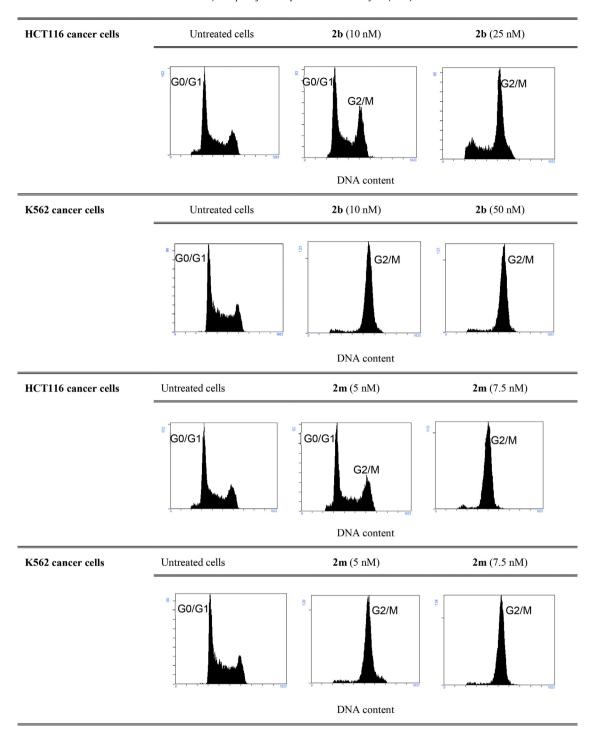


Fig. 2. Effects of 2b and 2m on cell cycle distribution in HCT116 and K562 cells determined by flow cytometry analysis. DNA content was assessed via propidium iodide staining.

presence of the vascular endothelial growth factor (VEGF), ECFCs form tube-like structures that express the endothelial marker (CD31) when co-cultured on an ADSC feeder layer [38]. The inhibition of endothelial tube formation (IC $_{50}$) was determined after 96 h of treatment using ECFCs, VEGF and compounds **2a-b**, **2d** and **2m** as angiogenesis inhibitors.

Compounds **2k**, **2l** having a pyridine and quinoline B-rings respectively, as well as **2n** were found to be inactive as angiogenesis inhibitors (Table 4). All other selected compounds inhibit the endothelial tube formation in a concentration-dependent manner

(data not shown) with remarkable low IC_{50} values as for examples with **2b** or **2m** ($CI_{50} = 25$ and 8 nM, respectively). One can note that N-propyl derivatives **3b** and **3c** which displayed a low level of cytotoxicity (Table 1) inhibit angiogenesis with nanomolar IC_{50} values of 73 and 50 nM, respectively.

These results were next confirmed by a second *in vitro* model of angiogenesis [39] in which human umbilical vein endothelial cells (HUVECs), when seeded on Matrigel™, aggregate to form a reticular vascular network of capillary-like vessels. The results presented in Table 2 (see above) revealed that after 72 h of

Table 3Apoptotic effects of **2b** and **2m** in HCT116 cells.

Rhodamine 110 relative fluorescence intensity ^a (% of control)								
2b 25 nM	2b 50 nM	2m 7.5 nM	2m 15 nM					
(300)	(452)	(393)	(471)					

^a The results are expressed in the percentage of apoptotic cells detected following 24 h of treatment with **2b** and **2m** at different concentrations.

incubation, ${\bf 2b}$ and ${\bf 2m}$ exhibited good growth inhibition activity against HUVEC cells with ${\rm GI}_{50}$ values of 22 and 7 nM, respectively. To evaluate whether ${\bf 2b}$ and ${\bf 2m}$ could affect models of newly formed blood vessels, the HUVEC tube disruption assay was performed. Addition of ${\bf 2b}$ at a concentration of 10 nM, and more significantly at 100 nM, rapidly disrupted the integrity of the network (Fig. 3). This effect was visible after only 3 h of treatment at doses of 10 and 100 nM which were not cytotoxic for such incubation time (data not shown). This antivascular effect is even more pronounced with indole derivative ${\bf 2m}$ which destroyed neovascularization at a very low dose of 5 nM. All together these results suggested that azaisoerianin derivatives ${\bf 2}$, and particularly ${\bf 2m}$ and ${\bf 2b}$, might be considered and evaluated as vascular disrupting agents for further $in\ vivo$ studies.

2.3. Docking study

The biological IPT and antivascular results demonstrated that tubulin is the target of these diarylamines **2** and **3**. Molecular docking calculations were performed with **2b** and **2m** in order to investigate their possible binding mode in the colchicine binding site of tubulin. For this purpose, the X-ray structure of tubulin DAMA-colchicine complex (accession code 1SA0) [40] was used. Fig. 4 illustrates the docking-derived superimposition of **2b** and **2m**, with *iso*CA-4 (green). As expected, these diarylamines showed a binding pose matching the one calculated under the same conditions for *iso*CA-4, used as a reference compound with its A-trimethoxyphenyl ring placed in the proximity of Cys241. Additionally, the 4′-OMe substituent of **2b** accepts a stabilizing H-bond donated by a polar hydrogen atom belonging to the sidechain amide moiety of Asn101.

3. Conclusions

We designed and synthesized a series of original N-methyl- and N-propyldiarylamines ${\bf 2}$ and ${\bf 3}$, respectively as aza-analogues of isoerianin and isoCA-4. High inhibition of cell growth was observed with compounds ${\bf 2b}$ and ${\bf 2m}$ in a range of human cancer cells. These potent drugs also inhibited tubulin assembly with micromolar IC $_{50}$ values. Cell cycle analysis of K562 and HCT116 cells treated with nanomolar concentrations of ${\bf 2b}$ and ${\bf 2m}$ indicated that cells were arrested at the G2/M phase. Moreover, ${\bf 2b}$ and ${\bf 2m}$ induced apoptosis in HCT116 cells and destroyed in vitro models of neovasculature at very low doses. Docking studies revealed that these compounds adopted an orientation similar to that of isoCA-4 in the colchicine binding-site of tubulin.

Table 4 Inhibition of endothelial tube formation using ECFCs.

Inhibition of endothelial colony tube formation													
Compounds	2a	2b	2c	2k	21	2m	2n	20	2q	3b	3c	isoerianin	isoCA-4
(IC ₅₀) nM	56	25	300	2000	NA ^a	8	NA ^a	67	250	73	50	70	7

^a Not active.

4. Experimental

4.1. General considerations

The compounds were all identified by usual physical methods, i.e. ¹H NMR, ¹³C NMR, IR, MS. ¹H and ¹³C NMR spectra were measured in CDCl₃ with a Bruker Avance 300. ¹H chemical shifts are reported in ppm from an internal standard TMS or of residual chloroform (7.27 ppm). The following abbreviations are used: m (multiplet), s (singlet), d (doublet), br s (broad singlet), t (triplet), dd (doublet of doublet), td (triplet of doublet). ¹³C chemical shifts are reported in ppm from the central peak of CDCl₃ (77.14). IR spectra were measured on a Bruker Vector 22 spectrophotometer (neat, cm⁻¹). Mass spectra were obtained with a LCT Micromass spectrometer. Analytical TLC was performed on Merck precoated silica gel 60F plates. Merck silica gel 60 (230–400 mesh) was used for column chromatography.

4.1.1. 3,4,5-Trimethoxy-N-(4-methoxyphenyl)-N-methylaniline **2a**

In a sealed tube and under an argon atmosphere were added successively, Pd(OAc)₂ (11 mg; 5 mol%), Xantphos (29 mg; 5 mol%), 1-bromo-4-methoxybenzene (187 mg; 1 mmol), trimethoxyaniline (275 mg; 1.5 mmol), Cs₂CO₃ (652 mg; 2 mmol) in dioxane (2 mL). After heating at 100 °C for 12 h, the resulting suspension was cooled to room temperature and filtered through a pad of Celite eluting with ethyl acetate, and the inorganic salts were removed. The filtrate was concentrated and a rapid purification by silica gel column chromatography of the residue to eliminate the excess of 3,4,5-trimethoxyaniline gave the desired product 1a which was immediately engaged in the alkylation step. 1a: (75 mg, 26%); ¹H NMR (300 MHz, CDCl₃) δ 7.04 (d, 2H, J = 9.0 Hz), 6.85 (d, 2H, J = 9.0 Hz), 6.15 (s, 2H), 5.52 (s, 1H), 3.79 (s, 6H), 3.76 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 155.1, 153.9 (2), 141.6, 136.1, 131.5, 121.8 (2), 114.7 (2), 93.7 (2), 61.1, 55.9 (2), 55.6. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 3278, 2993, 1600, 1501, 1224, 1124, 1007. MS (APCI) m/z: 290 [M + H]⁺. To a diarylamine 1a (50 mg; 0.17 mmol) solution in DMF (2 mL) was added at 0 °C, NaH (8 mg; 0.34 mmol), and iodomethane (48 mg; 28 μL, 0.34 mmol). The mixture was stirred at r.t. for a night and diluted in EtOAc (2 mL) and washed with aqueous NaHCO3. The organic layer was dried, evaporated, and the residue was subjected to flash chromatography to give **2a** as a yellow oil (35 mg, 67%). ¹H NMR (300 MHz, CDCl₃) δ 7.05 (d, 2H, J = 8.9 Hz), 6.88 (d, 2H, I = 8.9 Hz), 6.05 (s, 2H), 3.81 (s, 3H), 3.79 (s, 3H), 3.76 (s, 6H), 3.25 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 155.9, 153.6 (2), 146.5, 142.5, 131.6, 125.1 (2), 114.8 (2), 95.1 (2), 61.2, 56.1 (2), 55.6, 41.0. IR (neat) ν_{max} cm $^{-1}$: 1575, 1503, 1465, 1419, 1242, 1127. MS (APCI) m/z: 304 $[M + H]^+$. HRMS calcd for $C_{17}H_{22}NO_4$ $[M + H]^+$ 304.1549, obsd. 304.1548.

4.1.2. N-(3-Fluoro-4-methoxyphenyl)-3,4,5-trimethoxy-N-methylaniline **2b**

By following the general procedure described above for **1a**. Heating at 100 °C 3,4,5-trimethoxyaniline (275 mg; 1.5 mmol) and 4-bromo-2-fluoro-1-methoxybenzene (205 mg; 1 mmol) for 5 h gave **1b** (181 mg; 59%). 1 H NMR (300 MHz, CDCl₃) δ 6.92–6.84 (m, 2H), 6.76 (d, 1H, J = 8.7 Hz), 6.22 (s, 2H), 5.45 (s, 1H), 3.87 (s, 3H),

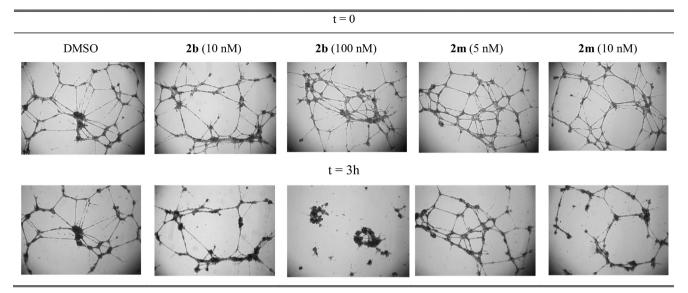


Fig. 3. Inhibition of HUVEC tube formation on a Matrigel by 2b (at 10 and 100 nM) and 2m (at 5 and 10 nM). Images were taken 3 h after addition of compound 2b and 2m.

3.80 (s, 9H). ¹³C NMR (75 MHz, CDCl₃) δ 154.0 (2), 153.0 (d, I = 244.5 Hz), 142.4 (d, I = 11.2 Hz), 140.1, 137.6 (d, I = 9.0 Hz), 132.7, 115.1 (d, I = 2.2 Hz), 114.5 (d, I = 2.2 Hz), 107.8 (d, I = 20.2 Hz), 95.4 (2), 61.1, 57.1, 56.1 (2). 19 F NMR (188 MHz, CDCl₃) δ –131.0. IR (neat) $\nu_{\rm max}/{\rm cm}^{-1}$: 1604, 1502, 1464, 1262, 1129, 1030. MS (APCI) m/z: 308 $[M + H]^+$. By following the general procedure described above for **2a.** Diarylamine **1b** (102 mg; 0.33 mmol), NaH (12 mg; 0.5 mmol) and CH₃I (71 mg; 0.5 mmol) gave **2b** as a yellow oil (80 mg; 75%). ¹H NMR (300 MHz, CDCl₃) δ 6.87 (dd, 1H, J = 9.3 Hz, J = 9.0 Hz), 6.75 (dd, 1H, J = 13.3 Hz, J = 2.7 Hz), 6.71-6.66 (m, 1H), 6.18 (s, 2H), 3.84(s, 3H), 3.80 (s, 3H), 3.76 (s, 6H), 3.22 (s, 3H). ¹³C NMR (75 MHz, $CDCl_3$) δ 153.6 (2), 152.7 (d, I = 243.7 Hz), 145.4, 143.4 (d, I = 8.4 Hz), 142.3 (d, I = 11.1 Hz), 133.2, 115.9 (d, I = 2.7 Hz), 114.6 (d, I = 2.3 Hz), 109.1 (d, J = 20.5 Hz), 98.2 (2), 61.0, 56.9, 56.1 (2), 40.8. ¹⁹F NMR (188 MHz, CDCl₃) δ –133.4. IR (neat) $\nu_{\rm max}/{\rm cm}^{-1}$: 2999, 1604, 1581, 1503, 1242, 1125. MS (APCI) m/z: 322 [M + H]⁺. HRMS calcd for $C_{17}H_{21}FNO_4 [M + H]^+$ 322.1455, obsd. 322.1450.

4.1.3. N-(3-fluoro-4-methoxyphenyl)-3,4,5-trimethoxy-N-methylbenzenaminium chloride **2b.HCl**

To a solution of **2b** (97 mg; 1 mmol) in MeOH $-Et_2O$ (1:1) (10 mL) was added HClg within 10 min and then the mixture was stirred for 1 h at RT. The precipitated salt **2b.HCl** (92 mg; 86%) was

isolated as green amorphous solid after filtration. 1 H NMR (300 MHz, DMSO) δ 7.06 (t, J = 9.4 Hz, 1H), 6.83 (dd, J = 13.8, 2.5 Hz, 1H), 6.73 (d, J = 8.9 Hz, 1H), 6.60 (s, 1H), 6.25 (s, 2H), 3.78 (s, 3H), 3.69 (s, 6H), 3.61 (s, 3H), 3.19 (s, 3H). 13 C NMR (75 MHz, DMSO) δ 153.86 (2C), 152.19 (d, J = 233.9 Hz), 145.37, 143.40 (d, J = 8.5 Hz), 141.91 (d, J = 11.1 Hz), 133.19, 116.20, 115.32, 108.60 (d, J = 20.2 Hz), 99.00 (2C), 60.56, 56.88, 56.28 (2C), 41.02. IR (neat) $\nu_{\rm max}/{\rm cm}^{-1}$: 2999, 1612, 1521, 1488, 1275, 1122, 1005. MS (APCI) m/z: 322 [M] $^+$. HRMS calcd for C₁₇H₂₁FNO₄ [M] $^+$ 322.1449, obsd. 322.1442.

4.1.4. 3,4,5-Trimethoxy-N-(4-methoxy-3-nitrophenyl)-N-methylaniline **2c**

By following the general procedure described above for **1a**. Heating at 100 °C 3,4,5-trimethoxyaniline (275 mg; 1.5 mmol) and 4-iodo-1-methoxy-2-nitrobenzene (279 mg; 1 mmol) in the presence of Pd(OAc)₂ (22 mg; 10 mol%), Xantphos (58 mg; 10 mol%) for 1 h 30 gave **1c** (307 mg; 92%). 1 H NMR (300 MHz, CDCl₃) δ 7.55 (d, 1H, J=2.8 Hz), 7.21 (dd, 1H, J=9.0 Hz, J=2.8 Hz), 7.01 (d, 1H, J=9.0 Hz), 6.25 (s, 2H), 5.59 (s, 1H), 3.92 (s, 3H), 3.81 (s, 3H), 3.80 (s, 6H). 13 C NMR (75 MHz, CDCl₃) δ 154.2 (2), 147.6, 140.2, 139.0, 137.3, 133.4, 124.2, 115.3, 114.9, 96.3 (2), 61.2, 57.2, 56.2 (2). IR (neat) $\nu_{\rm max}/$ cm $^{-1}$: 1600, 1535, 1347, 1302, 1230, 1122, 1021. MS (APCI) m/z: 335 [M + H] $^+$.

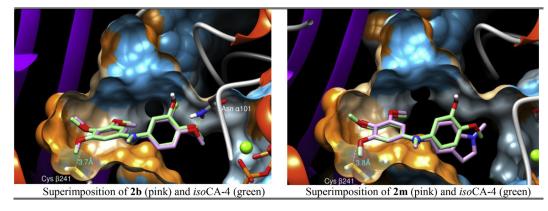


Fig. 4. Calculated binding mode for **2b** and **2m** in the colchicine binding site of tubulin and superimposition of the latter with that of *iso*CA-4. (Central nitrogen atom and (if applicable) fluorine atom are represented as spheres in order to increase their visibility.)

By following the general procedure described above for **2a**. Diarylamine **1c** (167 mg; 0.5 mmol), NaH (18 mg; 0.75 mmol) and CH₃I (106 mg; 0.75 mmol) gave **2c** as an orange oil (172 mg; 99%).

¹H NMR (300 MHz, CDCl₃) δ 7.41 (d, 1H, J = 2.9 Hz), 7.09 (dd, 1H, J = 9.1 Hz, J = 2.9 Hz), 6.97 (d, 1H, J = 9.1 Hz), 6.26 (s, 2H), 3.91 (s, 3H), 3.83 (s, 3H), 3.79 (s, 6H), 3.27 (s, 3H).

¹³C NMR (75 MHz, CDCl₃) δ 154.0 (2), 146.9, 144.6, 143.0, 140.1, 134.6, 124.6, 115.0, 114.8, 100.3 (2), 61.1, 57.2, 56.2 (2), 40.9. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1589, 1525, 1502, 1443, 1364, 1254, 1234, 1122. MS (APCI) m/z: 349 [M + H]⁺. HRMS calcd for C₁₇H₂₁N₂O₆ [M + H]⁺ 349.1400, obsd. 349.1392.

4.1.5. 4-Methoxy-N-methyl-N-(3,4,5-trimethoxyphenyl)benzene-1,3-diamine **2d**

To a (EtOH/H₂O: 4/1) solution (1.5 mL) of **2c** (140 mg; 0.4 mmol) were added iron (225 mg; 4 mmol) and a drop of HCl 12N. After refluxing for 2 h, the mixture was filtered through a pad of Celite, concentrated to give a residue which was subjected to flash chromatography to give **2d** as a brown oil (96 mg, 76%). $^1{\rm H}$ NMR (300 MHz, CDCl₃) δ 6.74 (d, 1H, J=8.3 Hz), 6.49–6.44 (m, 2H), 6.09 (s, 2H), 3.84 (s, 3H), 3.79 (s, 3H), 3.76 (s, 6H), 3.21 (s, 3H), NH₂ not seen. $^{13}{\rm C}$ NMR (75 MHz, CDCl₃) δ 153.6 (2), 146.6, 143.8, 143.2, 136.9, 131.7, 113.3, 111.3, 110.8, 95.7 (2), 61.1, 56.2 (2), 55.9, 41.0. IR (neat) $\nu_{\rm max}/{\rm cm}^{-1}$: 1612, 1581, 1502, 1148, 1248, 1126. MS (APCl) m/z: 319 [M + H]+. HRMS calcd for C₁₇H₂₃N₂O₄ [M + H]+ 319.1658, obsd. 319.1654.

4.1.6. N-Methyl-N-(3,4,5-trimethoxyphenyl)naphthalen-2-amine

By following the general procedure described above for 1a. Heating at 100 °C 3,4,5-trimethoxyaniline (275 mg; 1.5 mmol) and 2-iodonaphtalene (254 mg; 1 mmol) in the presence of Pd(OAc)₂ (22 mg; 10 mol%), Xantphos (58 mg; 10 mol%) for 3 h gave 1d (213 mg; 69%). ¹H NMR (300 MHz, CDCl₃) δ 7.77 (d, 2H, J = 8.5 Hz), 7.67 (d, 1H, J = 8.2 Hz), 7.46 - 7.42 (m, 2H), 7.36 - 7.22 (m, 2H), 6.45 (s, 2H), 5.86 (s, 1H), 3.89 (s, 3H), 3.84 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 153.9 (2), 141.4, 139.2, 134.7, 133.2, 129.2, 129.1, 127.7, 126.5, 126.5, 123.5, 119.9, 111.5, 96.4 (2), 61.1, 56.1 (2). IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1631, 1596, 1503, 1465, 1223, 1127, 1006. MS (APCI) m/z: 310 [M + H]⁺. By following the general procedure described above for 2a. Diarylamine 1d (155 mg; 0.5 mmol), NaH (18 mg; 0.75 mmol) and CH₃I (106 mg; 0.75 mmol) gave **2e** as a yellow oil (140 mg; 87%). ¹H NMR (300 MHz, CDCl₃) δ 7.72 (t, 2H, J = 7.8 Hz), 7.66 (d, 1H, J = 9.0 Hz), 7.45-7.39 (td, 1H, J = 8.6 Hz, J = 1.3 Hz), 7.33-7.28 (m, 1H), 7.25 (d, 1H, I = 2.5 Hz), 7.17 (dd, 1H, I = 8.9 Hz, I = 2.4 Hz), 6.39 (s, 2H), 3.87 (s, 3H), 3.79 (s, 6H), 3.41 (s, 3H). 13 C NMR (75 MHz, CDCl₃) δ 153.9 (2), 146.9, 145.4, 134.8, 134.4, 128.7, 128.5, 127.6, 126.7, 126.4, 123.5, 121.0, 112.5, 100.9 (2), 61.1, 56.2 (2), 41.0. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1629, 1589, 1502, 1240, 1128, 1005. MS (APCI) m/z: 324 [M + H]⁺. HRMS calcd for $C_{20}H_{22}NO_3 [M + H]^+$ 324.1600, obsd. 324.1602.

4.1.7. N-(3-(benzyloxy)-4-methoxyphenyl)-3,4,5-trimethoxy-N-methylaniline **2f**

Compound **2f** was prepared from diarylaniline **1e** (yield 61%) according to literature [41]. ¹H NMR (300 MHz, CDCl₃) δ 7.40–7.27 (m, 5H), 6.86 (d, 1H, J = 8.4 Hz), 6.67–6.62 (m, 2H), 6.06 (s, 2H), 5.08 (s, 2H), 3.88 (s, 3H), 3.81 (s, 3H), 3.72 (s, 6H), 3.20 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 153.6 (2), 148.7, 146.1, 145.8, 142.9, 137.1, 128.6 (2), 127.9, 127.4 (2), 115.1, 112.9, 110.1, 96.2 (2), 71.1, 61.1, 56.5, 56.1 (2), 41.0. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1579, 1502, 1452, 1411, 1230, 1125. MS (ESI) m/z: 432.3 [M + Na]⁺. HRMS calcd for C₂₄H₂₈NO₅ [M + H]⁺ 410.1967, obsd. 410.1964.

4.1.8. AzaisoCA-4 2g

Compound **2g** was prepared from diarylaniline **2f** using H₂ and Pd/C (10 mol%) in EtOAc (yield 98%) according to literature [1]. ¹H

NMR (300 MHz, CDCl₃) δ 6.79 (d, 1H, J=8.7 Hz), 6.67 (d, 1H, J=2.7 Hz), 6.53 (dd, 1H, J=8.7 Hz, J=2.7 Hz), 6.14 (s, 2H), 5.64 (s, 1H), 3.87 (s, 3H), 3.80 (s, 3H), 3.77 (s, 6H), 3.22 (s, 3H). 13 C NMR (75 MHz, CDCl₃) δ 153.6 (2), 146.3, 146.1, 143.7, 142.3, 132.4, 113.4, 111.5, 109.2, 97.0 (2), 61.1, 56.4, 56.2 (2), 41.0. IR (neat) $\nu_{\rm max}/{\rm cm}^{-1}$: 2959, 1581, 1501, 1447, 1249. MS (APCI) m/z: 342 [M + Na]⁺. HRMS calcd for $C_{17}H_{22}NO_5$ [M + H]⁺ 320.1498, obsd. 320.1494.

4.1.9. 2-Methoxy-5-(methyl(3,4,5-trimethoxyphenyl)amino)phenyl acetate **2h**

At 0 °C, to a solution of **2g** (101 mg; 0.316 mmol) in CH₂Cl₂ (1 mL) were added pyridine (54 μL), and DMAP (0.016 mmol) and Ac₂O (42 μL). After one hour of stirring, H₂O (3 mL) was added to the mixture. After extraction with EtOAc (3 × 3 mL), the organic layer was dried, evaporated, and the residue was subjected to flash chromatography to give **2h** as a yellow oil (76 mg, 67%). ¹H NMR (300 MHz, CDCl₃) δ 6.92–6.84 (m, 2H), 6.75 (d, J = 2.4 Hz, 1H), 6.17 (s, 2H), 3.80 (s, 6H), 3.77 (s, 6H), 3.24 (s, 3H), 2.28 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 169.0, 153.7 (2), 146.4, 145.7, 142.9, 140.2, 132.8, 119.3, 116.8, 113.3, 97.4 (2), 61.1, 56.4, 56.2 (2), 40.9, 20.7. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1721, 1574, 1452, 1375, 1204. MS (ESI) m/z: 362 [M + H]⁺. HRMS calcd for C₁₉H₂₃NO₆Na [M + Na]⁺ 384.1423, obsd. 384.1426.

4.1.10. N-(4-Chlorophenyl)-3,4,5-trimethoxy-N-methylaniline 2i

By following the general procedure described above for 1a. Heating at 100 °C 3.4.5-trimethoxyaniline (275 mg: 1.5 mmol) and 1-chloro-4-iodobenzene (59 mg; 0.25 mmol) in the presence of Pd(OAc)₂ (22 mg; 10 mol%), Xantphos (58 mg; 10 mol%) for 12 h gave **1f** as an orange solid (40 mg; 55%). m.p. = $109 \, ^{\circ}$ C. 1 H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 7.20 \text{ (d, 2H, } I = 8.9 \text{ Hz)}, 6.95 \text{ (d, 2H, } I = 8.9 \text{ Hz)},$ 6.30 (s, 2H), 3.82 (s, 3H), 3.80 (s, 6H), NH not seen. ¹³C NMR (75 MHz, CDCl₃) δ 154.0 (2), 142.5, 138.9, 133.3, 129.4 (2), 125.3, 118.6 (2), 96.8 (2), 61.1, 56.1 (2). IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1611, 1590, 1508, 1228, 1119. MS (APCI) m/z: 294, 296 [M + H]⁺. By following the general procedure described above for 2a. Diarylamine 1f (29 mg; 0.1 mmol), NaH (5 mg; 0.2 mmol) and CH₃I (28 mg; 0.2 mmol) gave 2i as a yellow-orange solid (27 mg; 88%). m.p. = 91 °C. ¹H NMR (300 MHz, CDCl₃) δ 7.17 (d, 2H, J = 8.5 Hz), 6.82 (d, 2H, J = 8.5 Hz), 6.31 (s, 2H), 3.84 (s, 3H), 3.79 (s, 6H), 3.26 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 153.9 (2), 147.9, 144.8, 134.6, 129.0 (2), 124.6, 118.9 (2), 101.0 (2), 61.0, 56.1 (2), 40.6. IR (neat) $\nu_{\text{max}}/$ cm⁻¹: 2999, 1609, 1579, 1510, 1451, 1237.

4.1.11. N,N,N'-Trimethyl-N'-(3,4,5-trimethoxyphenyl)benzene-1,4-diamine **2i**

In a sealed tube and under an argon atmosphere were added successively, Pd₂(dba)₃.CHCl₃ (52 mg; 5 mol%), Johnphos (30 mg; 10 mol%), dimethylamine (large excess), 2i (307 mg; 1 mmol), NaOtBu (134 mg; 1.4 mmol) in toluene (10 mL). After heating at 100 °C for 24 h, the resulting suspension was cooled to room temperature and filtered through a pad of Celite eluting with ethyl acetate, and the inorganic salts were removed. The filtrate was concentrated and a rapid purification by silica gel column chromatography of the residue to eliminate the excess of 3,4,5trimethoxyaniline gave the desired product 2j as a white solid (98 mg, 31%). m.p. = 75 °C. ¹H NMR (300 MHz, CDCl₃) δ 7.05 (d, 2H, J = 7.3 Hz), 6.75 (d, 2H, J = 7.3 Hz), 6.01 (s, 2H), 3.78 (s, 3H), 3.76 (s, 6H), 3.24 (s, 3H), 2.95 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 153.6 (2), 147.9, 147.8, 146.9, 139.1, 126.0 (2), 113.9 (2), 93.8 (2), 61.2, 56.1 (2), 41.1 (2), 41.0. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1611, 1583, 1504, 1128. MS (ESI) m/z: 339 [M + Na]⁺. HRMS calcd for C₁₈H₂₅N₂O₃ [M + H]⁺ 317.1865, obsd. 317.1860.

4.1.12. 6-Methoxy-N-methyl-N-(3,4,5-trimethoxyphenyl)pyridin-3-amine **2k**

By following the general procedure described above for 1a. Heating at 100 °C 3,4,5-trimethoxyaniline (275 mg; 1.5 mmol) and 5-bromo-2-methoxypyridine (187 mg; 1 mmol) in the presence of Pd(OAc)₂ (11 mg; 5 mol%), Xantphos (29 mg; 5 mol%) for 5 h gave **1g** as a vellow oil (160 mg; 54%), ¹H NMR (300 MHz, CDCl₃) δ 7.98 (d. 1H. I = 2.3 Hz), 7.42 (dd. 1H. I = 8.8 Hz, I = 2.8 Hz), 6.72 (d. 1H. I = 8.8 Hz), 6.10 (s, 2H), 5.36 (s, 1H), 3.92 (s, 3H), 3.79 (s, 3H), 3.78 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 160.1, 154.1 (2), 141.2, 139.0, 133.5, 133.1, 132.2, 111.2, 93.9 (2), 61.2, 56.2 (2), 53.8. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1599, 1486, 1367, 1230, 1125, 1028. MS (APCI) m/z: 291 [M + H]⁺. By following the general procedure described above for 2a. Diarylamine 1g (290 mg; 1 mmol), NaH (48 mg; 2 mmol) and CH₃I (284 mg; 2 mmol) gave **2k** as a yellow oil (267 mg; 88%). ¹H NMR (300 MHz, CDCl₃) δ 7.93 (d, 1H, J = 2.8 Hz), 7.34 (dd, 1H, J = 8.8 Hz, J = 2.8 Hz), 6.68 (d, 1H, J = 8.8 Hz), 6.03 (s, 2H), 3.90 (s, 3H), 3.76 (s, 3H), 3.74 (s, 6H), 3.22 (s, 3H). 13 C NMR (75 MHz, CDCl₃) δ 160.2, 153.7 (2), 145.8, 141.2, 139.7, 134.9, 132.2, 111.2, 95.6 (2), 61.0, 56.1 (2), 53.6, 41.0. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1587, 1491, 1275, 1235, 1127, 1028. MS (APCI) m/z: 305 [M + H]⁺. HRMS calcd for $C_{16}H_{21}N_2O_4$ $[M + H]^+$ 305.1501, obsd. 305.1501.

4.1.13. N-Methyl-N-(3,4,5-trimethoxyphenyl)quinolin-3-amine 21

By following the general procedure described above for 1a. Heating at 100 °C 3,4,5-trimethoxyaniline (275 mg; 1.5 mmol) and 3-bromoquinoline (207 mg; 1 mmol) in the presence of Pd(OAc)₂ (44 mg; 20 mol%), Xantphos (29 mg; 5 mol%) and Cs₂CO₃ (534 mg; 1.64 mmol) for 3 h gave **1h** as a yellow oil (167 mg; 54%). ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 8.69 \text{ (d, 1H, } I = 2.7 \text{ Hz}), 8.00 - 7.97 \text{ (m, 1H)}, 7.64$ (d, 1H, I = 2.7 Hz), 7.61-7.58 (m, 1H), 7.50-7.40 (m, 2H), 6.41 (s, 3H),3.84 (s, 3H), 3.77 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 154.0 (2), 144.9, 143.4, 138.2, 137.8, 133.6, 129.0, 129.0, 127.2, 126.4 (2), 116.7, 97.1 (2), 61.1, 56.1 (2). IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 3338, 1602, 1506, 1237, 1124. MS (APCI) m/z: 311 [M + H]⁺. By following the general procedure described above for 2a. Diarylamine 1h (155 mg; 0.5 mmol), NaH (18 mg; 0.75 mmol) and CH₃I (107 mg; 0.75 mmol) gave 21 as a yellow oil (110 mg; 68%). ¹H NMR (300 MHz, CDCl₃) δ 8.62 (d, 1H, J = 2.8 Hz, 8.00–7.94 (m, 1H), 7.68–7.63 (m, 1H), 7.51–7.42 (m, 2H), 7.39 (d, 1H, J = 2.8 Hz), 6.38 (s, 2H), 3.85 (s, 3H), 3.78 (s, 6H), 3.38 (s, 2H), 3.85 (s, 3H), 3.78 (s, 6H), 3.85 (s, 6H), 3.3H). ¹³C NMR (75 MHz, CDCl₃) δ 154.1 (2), 145.4, 144.2, 142.7, 142.6, 135.2, 129.0, 127.1, 126.5, 126.3, 117.3, 101.5 (2), 61.1, 56.2 (2), 40.82 (one C missing). IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1588, 1504, 1245, 1229, 1128. MS (APCI) m/z: 325 [M + H]⁺.

4.1.14. N,1-Dimethyl-N-(3,4,5-trimethoxyphenyl)-1H-indol-5-amine **2m**

To an Emrys Optimizer were added 3,4,5-trimethoxyaniline (260 mg; 1.42 mmol), 6-bromo-1-methyl-1*H*-indole (150 mg; 0.71 mmol), Pd(OAc)₂ (16 mg; 10 mol%), Xantphos (41 mg; 10 mol %), Cs₂CO₃ (463 mg; 1.42 mmol) in dioxane (5 mL). The reaction vessel was then placed in the Emrys Optimizer and exposed to microwave irradiation according to the following specifications: temperature: 130 °C, time: 3 h, fixed hold time: on, sample absorption: high, pre-stirring: 60 s. After cooling to room temperature, the mixture was filtered through a pad of Celite eluting with ethyl acetate, and the inorganic salts were removed. The filtrate was concentrated and a rapid purification by silica gel column chromatography of the residue to eliminate the excess of 3,4,5trimethoxyaniline gave the desired product 1i as a yellow oil which was immediately engaged in the alkylation step. 1i: (108 mg, 49%); ¹H NMR (300 MHz, CDCl₃) δ 7.35 (s, 1H), 7.23 (s, 1H), 7.05– 7.02 (m, 2H), 6.38 (d, 1H, J = 3.0 Hz), 6.15 (s, 2H), 5.50 (brs, 1H), 3.77 (s, 2H)(s, 3H), 3.76 (s, 3H), 3.74 (s, 6H). 13 C NMR (75 MHz, CDCl₃) δ 154.0 (2), 143.0, 135.1, 133.8, 131.3, 129.6, 129.2, 117.9, 113.4, 110.0, 100.6, 93.3 (2), 61.2, 56.1 (2), 33.1. IR (neat) ν_{max}/cm^{-1} : 3370, 1605, 1505, 1247, 1125, 1009. MS (APCI) m/z: 313 [M + H]⁺. By following the general procedure described above for **2a**. Diarylamine **1i** (31 mg; 0.1 mmol), NaH (12 mg; 0.5 mmol) and CH₃I (71 mg; 0.5 mmol) gave **2m** as a yellow oil (21 mg; 65%). ¹H NMR (300 MHz, CDCl₃) δ 7.43 (d, 1H, J = 2.0 Hz), 7.30 (d, 1H, J = 8.6 Hz), 7.08–7.05 (m, 2H), 6.45 (d, 1H, J = 3.0 Hz), 6.03 (s, 2H), 3.81 (s, 3H), 3.80 (s, 3H), 3.74 (s, 6H), 3.33 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 153.5 (2), 147.4, 141.7, 134.4, 130.7, 129.6, 129.3, 120.7, 117.4, 110.1, 101.0, 93.6 (2), 61.1, 56.1 (2), 41.5, 33.1. IR (neat) ν_{max}/cm^{-1} : 1602, 1581, 1505, 1490, 1418, 1234, 1123. MS (APCI) m/z: 327 [M + H]⁺. HRMS calcd for C₁₉H₂₃N₂O₃ [M + H]⁺ 327.1709, obsd.327.1698.

4.1.15. N-(4-Methoxyphenyl)-N,1-dimethyl-1H-indol-5-amine 2n

By following the general procedure described above for 1i. Heating at 130 °C under microwave irradiation 4-methoxyaniline (123 mg; 1 mmol) and 6-bromo-1-methyl-1H-indole (105 mg; 0.5 mmol) for 3 h gave 1j as a yellow oil (50 mg; 40%). ¹H NMR (300 MHz, CDCl₃) δ 7.37 (d, 1H, J = 1.7 Hz), 7.30 (d, 1H, J = 8.7 Hz), 7.08-7.02 (m, 4H), 6.94-6.89 (m, 2H), 6.47 (d, 1H, J = 3.0 Hz), 5.31(br s, 1H), 3.86 (s, 3H), 3.79 (s, 3H). 13 C NMR (75 MHz, CDCl₃) δ 153.7, 139.5, 136.9, 133.2, 129.4, 129.2, 118.6 (2), 116.2, 114.7 (2), 110.6, 109.8, 100.3, 55.7, 32.9. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1618, 1571, 1509, 1422, 1240. MS (APCI) m/z: 253 [M + H]⁺. By following the general procedure described above for 2a. Diarylamine 1j (63 mg; 0.25 mmol), NaH (12 mg; 0.5 mmol) and CH₃I (71 mg; 0.5 mmol) gave **2n** as a white solid (34 mg; 51%). m.p. = 113 °C; 1 H NMR (300 MHz, CDCl₃) δ 7.25 (d, 1H, I = 1.9 Hz), 7.17–7.14 (m, 1H), 6.94 (d, 1H, I = 2.9 Hz), 6.90 (dd, 1H, I = 8.7 Hz, I = 2.0 Hz), 6.78–6.69 (m, 4H), 6.32 (d, 1H, I = 3.0 Hz), 3.68 (s, 6H), 3.20 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 153.4, 144.9, 143.0, 133.7, 129.5, 129.3, 119.2 (2), 119.1, 114.8, 114.6 (2), 110.0, 100.8, 55.8, 41.8, 33.0. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1620, 1508, 1487, 1422, 1234, 1031. MS (APCI) m/z: 267 [M + H]⁺. HRMS calcd for $C_{17}H_{19}N_2O [M + H]^+$ 267.1497, obsd. 267.1494.

4.1.16. N-(3,5-Dimethoxyphenyl)-N,1-dimethyl-1H-indol-5-amine

By following the general procedure described above for 1i. Heating at 130 °C under microwave irradiation 3,5dimethoxyaniline (153 mg; 1 mmol) and 6-bromo-1-methyl-1Hindole (105 mg; 0.5 mmol) for 3 h gave 1k as a yellow oil (35 mg; 25%). ¹H NMR (300 MHz, CDCl₃) δ 7.45 (d, 1H, J = 1.9 Hz), 7.30 (d, 1H, J = 9.4 Hz), 7.13–7.07 (m, 2H), 6.45 (d, 1H, J = 3.0 Hz), 6.14 (d, 2H, I = 2.1 Hz), 6.01 (t, 1H, I = 2.1 Hz), 5.68 (br s, 1H), 3.81 (s, 3H), 3.76 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ 161.2 (2), 148.7, 134.2, 134.1, 129.6, 129.1, 118.6, 114.5, 109.9, 100.7, 93.6 (2), 91.3, 55.3 (2), 33.0. IR (neat) $v_{\text{max}}/\text{cm}^{-1}$: 1592, 1518, 1448, 1154. MS (APCI) m/z: 305 [M + Na]⁺. By following the general procedure described above for 2a. Diarylamine 1k (28 mg; 0.1 mmol), NaH (12 mg; 0.5 mmol) and CH₃I (71 mg; 0.5 mmol) gave **20** as a yellow oil (27 mg; 93%). ¹H NMR (300 MHz, CDCl₃) δ 7.47 (d, 1H, J = 2.0 Hz), 7.32 (d, 1H, J = 8.6 Hz), 7.09 (d, 2H, I = 6.9 Hz, I = 2.5 Hz), 6.46 (d, 1H, I = 3.1 Hz), 5.94 (s, 3H), 3.81 (s, 3H), 3.70 (s, 6H), 3.32 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 161.4 (2), 152.5, 141.2, 134.8, 129.6, 129.3, 121.5, 118.9, 110.2, 101.1, 93.6 (2), 89.6, 55.3 (2), 41.2, 33.1. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 3433, 1608, 1578, 1419, 1236, 1138. MS (APCI) m/z: 297 [M + H]⁺. HRMS calcd for $C_{18}H_{21}N_2O_2 [M + H]^+$ 297.1603, obsd. 297.1601.

4.1.17. N-Methyl-1-(phenylsulfonyl)-N-(3,4,5-trimethoxyphenyl)-1H-indol-5-amine **2p**

By following the general procedure described above for **1i**. Heating at 130 °C under microwave irradiation 3,4,5-trimethoxyaniline (183 mg; 1 mmol) and 5-bromo-1-(phenylsulfonyl)-1H-indole (167 mg; 0.5 mmol) for 4 h gave **1l** as a yellow oil (162 mg; 74%). 1H NMR (300 MHz, CDCl₃) $^\delta$ 7.89–7.84 (m, 3H),

7.53-7.47 (m, 2H), 7.43-7.37 (m, 2H), 7.17 (d, 1H, I = 2.0 Hz), 7.02(dd, 1H, I = 2.1 Hz, I = 8.8 Hz), 6.53 (dd, 1H, I = 3.6 Hz, I = 0.5 Hz),6.26 (s, 2H), 5.74 (s, 1H), 3.80 (s, 3H), 3.74 (s, 6H). ¹³C NMR (75 MHz, $CDCl_3$) δ 153.9 (2), 140.3, 139.8, 138.2, 133.8, 132.5, 131.9, 130.2, 129.2 (2), 127.0, 126.7 (2), 117.4, 114.3, 109.9, 109.3, 95.6 (2), 61.0, 56.0 (2). IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 2927, 1600, 1504, 1450, 1336, 1227. MS (ESI) m/z: 461 $[M + Na]^+$. By following the general procedure described above for 2a. Diarylamine 11 (110 mg; 0.25 mmol), NaH (12 mg; 0.5 mmol) and CH₃I (71 mg; 0.5 mmol) gave 2p as a yellow oil (81 mg; 72%). ¹H NMR (300 MHz, CDCl₃) δ 7.89–7.86 (m, 3H), 7.55– 7.50 (m, 2H), 7.42 (dd, 2H, I = 10.4 Hz, I = 4.7 Hz), 7.13 (d, 1H, I = 2.2 Hz), 7.02 (dd, 1H, I = 8.9 Hz, I = 2.2 Hz), 6.57 (d, 1H, J = 3.6 Hz), 6.16 (s, 2H), 3.82 (s, 3H), 3.73 (s, 6H), 3.28 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 153.7 (2), 146.1, 145.7, 138.3, 133.8, 133.0, 131.9, 130.4, 129.3 (2), 127.0, 126.8 (2), 119.6, 114.2, 112.7, 109.4, 98.0 (2), 61.1, 56.1 (2), 41.2. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1586, 1504, 1446, 1368, 1123. MS (ESI) m/z: 453 [M + H]⁺, 475 [M + Na]⁺. HRMS calcd for $C_{25}H_{27}N_2O_5S [M + H]^+ 453.1484$, obsd.453.1487.

4.1.18. NN-Methyl-N-(3,4,5-trimethoxyphenyl)-1H-indol-5-amine **2q**

A THF (1 mL) solution of **2p** (47 mg; 0.1 mmol) was added 0.5 mL of a TBAF solution in THF (0.5 mmol) and stirred at 100 °C for 6 h. After cooling, the solution was filtered through a pad of Celite, eluting with ethyl acetate. The filtrate was concentrated and a purification by silica gel column chromatography gave the desired product **2q** as a white solid (15 mg, 49%). m.p. 166 °C. 1 H NMR (300 MHz, CDCl₃) δ 8.34 (s, 1H), 7.44 (s, 1H), 7.36 (d, 1H, J = 8.6 Hz), 7.22 (t, 1H, J = 2.7 Hz), 7.02 (dd, 1H, J = 8.6 Hz, J = 1.9 Hz), 6.52 – 6.51 (m, 1H), 6.04 (s, 2H), 3.80 (s, 3H), 3.74 (s, 6H), 3.33 (s, 3H). 13 C NMR (75 MHz, CDCl₃) δ 153.6 (2), 147.4, 142.1, 133.4, 130.8, 128.8, 125.1, 121.0, 117.1, 112.0, 102.7, 93.8 (2), 61.2, 56.1 (2), 41.5. IR (neat) $\nu_{\rm max}/$ cm $^{-1}$: 3433, 1608, 1578, 1505, 1450, 1419, 1236. MS (ESI) m/z: 335 [M + Na]+. HRMS calcd for C18H21N2O3 [M + H]+ 313.1552, obsd. 331.1548.

4.1.19. N-(3-(benzyloxy)-4-methoxyphenyl)-3,4,5-trimethoxy-N-propylaniline 3a

To a diarylamine **1e** (197 mg; 0.5 mmol) solution in DMF (5 mL) was added at 0 °C, NaH (24 mg; 1 mmol), and 1-bromopropane (122 mg; 1 mmol). The mixture was stirred at r.t. for 2 h, diluted in EtOAc (5 mL) and washed with aqueous NaHCO₃. The organic layer was dried, evaporated, and the residue was subjected to flash chromatography to give **3a** as a brown oil (208 mg, 95%). ¹H NMR (300 MHz, CDCl₃) δ 7.40–7.27 (m, 5H), 6.86–6.83 (m, 1H), 6.63–6.59 (m, 2H), 6.00 (s, 2H), 5.09 (s, 2H), 3.89 (s, 3H), 3.80 (s, 3H), 3.71 (s, 6H), 3.50–3.44 (m, 2H), 1.61–1.59 (m, 2H), 0.89 (t, 3H, J = 7.4 Hz). ¹³C NMR (75 MHz, CDCl₃) δ 153.7 (2), 148.7, 145.8, 145.4, 141.6, 137.1, 131.8, 128.6 (2), 127.9, 127.4 (2), 116.0, 112.8, 111.0, 96.3 (2), 71.1, 61.2, 56.5, 56.1 (2), 54.7, 21.0, 11.6. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 2925, 1610, 1502, 1464, 1225, 1127. MS (APCI) m/z: 438 [M + H]⁺. HRMS calcd for C₂₆H₃₂NO₅ [M + H]⁺ 438.2280, obsd. 438.2285.

4.1.20. 2-methoxy-5-(propyl(3,4,5-trimethoxyphenyl)amino) phenol **3b**

3b brown oil (140 mg; 99%) was prepared by following the general procedure described above for **2g**. ¹H NMR (300 MHz, CDCl₃) δ 6.78 (d, 1H, J = 8.7 Hz), 6.65 (d, 1H, J = 2.6 Hz), 6.51 (dd, 1H, J = 8.7 Hz, J = 2.6 Hz), 6.11 (s, 2H), 5.60 (s, 1H), 3.88 (s, 3H), 3.81 (s, 3H), 3.77 (s, 6H), 3.57–3.52 (m, 2H), 1.70–1.60 (m, 2H), 0.94 (t, 3H, J = 7.4 Hz). ¹³C NMR (75 MHz, CDCl₃) δ 153.7 (2), 146.3, 145.2, 142.6, 142.2, 132.3, 114.1, 111.6, 109.7, 97.5 (2), 61.2, 56.4, 56.2 (2), 54.7, 21.0, 11.6. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 2959, 1581, 1501, 1466, 1261, 1126, 1027. MS (ESI) m/z: 370 [M + Na]⁺. HRMS calcd for C₁₉H₂₆NO₅ [M + H]⁺ 348.1811, obsd. 348.1812.

4.1.21. 1-Methyl-N-propyl-N-(3,4,5-trimethoxyphenyl)-1H-indol-5-amine **3c**

3c yellow oil (120 mg; 34%) was prepared from **1i** by following the general procedure described above for **3a**. ¹H NMR (300 MHz, CDCl₃) δ 7.45 (d, 1H, J = 1.7 Hz), 7.32 (d, 1H, J = 8.6 Hz), 7.08–7.05 (m, 2H), 6.47 (d, 1H, J = 3.0 Hz), 5.99 (s, 2H), 3.81 (s, 6H), 3.73 (s, 6H), 3.66 (t, 2H, J = 6.5 Hz), 1.76–1.60 (m, 2H), 0.97 (t, 3H, J = 7.4 Hz). ¹³C NMR (75 MHz, CDCl₃) δ 153.5 (2), 146.7, 139.9, 134.5, 130.2, 129.4, 129.3, 121.8, 118.7, 110.1, 100.9, 93.3 (2), 61.1, 56.0 (2), 55.0, 33.0, 21.0, 11.6. IR (neat) $\nu_{\text{max}}/\text{cm}^{-1}$: 1606, 1580, 1507, 1487, 1237, 1125. MS (ESI) m/z: 377 [M + Na]⁺. HRMS calcd for C₂₁H₂₇N₂O₃ [M + H]⁺ 355.2022, obsd. 355.2018.

4.2. Biology

4.2.1. Cell culture and proliferation assay

Cancer cell lines were obtained from the American type Culture Collection (Rockville, MD) and were cultured according to the supplier's instructions. Briefly, A549 lung carcinoma, U87-MG human glioblastoma, MDA-MB231 cells were grown in Dulbecco minimal essential medium (DMEM) containing 4.5 g/L glucose supplemented with 10% FCS and 1% glutamine. Human K562 leukemia and HCT116 colorectal carcinoma cells were grown in RPMI 1640 containing 10% FCS and 1% glutamine. Human umbilical vein endothelial cells (HUVECs) were obtained from Clonetics (Lonza, Walkersville, MD, USA) and cultured according to the supplier's instructions. Briefly, HUVECs from three to six passages were subcultured to confluence onto 0.2% gelatin coated tissue culture flasks in endothelial cell growth medium (EGM2) containing growth factors and 2% FCS. All cell lines were maintained at 37 $^{\circ}\text{C}$ in a humidified atmosphere containing 5% CO2. Cell viability was assessed using Promega CellTiter-Blue TM reagent according to the manufacturer's instructions. Cells were seeded in 96-well plates $(5 \times 103 \text{ cells/well})$ containing 50 µL growth medium. After 24 h of culture, the cells were supplemented with 50 µL of the tested compound dissolved in DMSO (less than 0.1% in each preparation). After 72 h of incubation, 20 µL of resazurin was added for 2 h before recording fluorescence (λ ex = 560 nm, λ em = 590 nm) using a Victor microtiter plate fluorimeter (Perkin-Elmer, USA). The IC50 corresponds to the concentration of the tested compound that caused a decrease of 50% in fluorescence of drug treated cells compared with untreated cells. Experiments were performed in triplicate.

4.2.2. Tubulin binding assay

Sheep brain tubulin was purified according to the method of Shelanski [42] by two cycles of assembly-disassembly and then diluted in the assembly buffer containing 0.1 M MES, 0.5 mM MgCl₂, 1 mM EGTA, and 1 mM GTP, pH 6.6 to a final concentration around 2–3 mg/mL. Tubulin assembly was monitored by fluorescence according to reported procedure [43] using DAPI as fluorescent molecule. Assays were realized on 96-well plates prepared with Biomek NKMC and Biomek 3000 from Beckman Coulter and read at 37 °C on Wallac Victor fluorimeter from Perkin Elmer. The IC₅₀ value of each compound was determined as the concentration required to decrease the maximum assembly rate of tubulin by 50% compared to the rate in the absence of compound. The IC₅₀ values for all compounds were compared to the IC₅₀ of *iso*CA-4 and *iso*erianin measured the same day under the same conditions.

4.2.3. Cell cycle analysis

Exponentially growing cancer cells (K562, HCT116) were incubated with tested compound or DMSO for 24 h. Cell-cycle profiles

were determined by flow cytometry on a FC500 flow cytometer (Beckman–Coulter, France) as described previously [44].

4.2.4. Apoptosis assay

Apoptosis was measured by the Apo-one homogeneous caspase-3/7 assay (Promega Co, WI) according to the manufacturer's recommendations. Briefly, cells were subcultured on a 96-well plate with 5×10^4 cells/well in 100 μ L medium. After 24 h of incubation, the medium in the 96-well plate was discarded and replaced with medium containing different concentrations of **2b** and **2m** (7.5, 15, 25 and 50 nM). The treated cells were incubated for 24 h, each well then received 100 μ L of a mixture of caspase substrate and Apo-one caspase 3/7 buffer. After 1 h of incubation, the fluorescence of sample was measured using a Victor microtiter plate fluorimeter (Perkin–Elmer, USA) at 527 nm.

4.2.5. Inhibition of endothelial tube formation using ECFCs The protocol was achieved according to ref. [38].

4.2.6. Cord disruption assay

HUVECs (2×10^4 cells per well) were plated in 96-well plates on a thick layer of Matrigel (Becton Dickinson; 10 mg mL $^{-1}$, 60 μ L per well) and allowed to align for 24 h. **2b**, **2m** or vehicle (DMSO) were added to the formed cords and left for 3 h. Images were taken 3 h after the addition of compounds.

4.2.7. Molecular modelling

Coordinates for compounds **2b**, **2m** and *iso*CA-4 were generated using CORINA v3.44 software [45]. Molecules were then freely docked in the colchicine binding site between chains C and D from PDB structure 1SA0 using GOLD v5.1 software [46]. CHEMPLP with default parameters was used as an objective function [47]. Structures of the complexes were exported for further examination and depiction with Chimera v1.8.1 software [48], including hydrogen bonds detection, close contact analysis and representation of the solvent-accessible surface colored according to polarity.

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Appendix A. Supplementary data

Supplementary data related to this article can be found at http://dx.doi.org/10.1016/j.ejmech.2014.03.032.

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