



JOURNAL OF CRYSTAL GROWTH

Journal of Crystal Growth 288 (2006) 70-74

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# Preparation and characterization of nanocrystalline SnO<sub>2</sub> thin films by PECVD

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Available online 18 January 2006

#### **Abstract**

Nanocrystalline  $SnO_2$  thin films were deposited by RF inductively coupled plasma enhanced chemical vapor deposition (PECVD) using dibutyltin diacetate as precursor. The effects of substrate–nozzle distance  $(D_{sn})$ , RF power and deposition time on the phase and morphologies of the  $SnO_2$  thin films have been studied. The as-deposited  $SnO_2$  thin films were well crystallized even without additional substrate heating and post-annealing. The grain size and film thickness decreased continuously with the increase of  $D_{sn}$  and RF power while increased with increasing deposition time. The deposition parameters showed notable effects on the microstructure of the  $SnO_2$  thin films, thus it was possible to optimize the microstructure of the  $SnO_2$  thin films by adjusting deposition parameters to achieve high-performance  $SnO_2$ -based thin film gas sensors.

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PACS: 81.15.Gh; 81.07.Bc

Keywords: A3. Chemical vapor deposition processes; B1. Tin oxide

#### 1. Introduction

SnO<sub>2</sub> is an important functional material and has been extensively investigated for gas sensors [1], electrocatalytic anodes [2] and optical-conductive coatings for solar cells [3]. Various processing routes, both physical and chemical deposition methods, have been utilized to prepare SnO<sub>2</sub> thin films, including pulsed laser deposition [4], radio frequency sputtering [5], sol–gel process [6], spray pyrolysis [7], electron beam evaporation [8] and PECVD [9]. Among them, PECVD was chosen as an ideal fabrication method for preparing nanostructured SnO<sub>2</sub> thin films basically due to its low deposition temperature.

For gas sensing application, the sensing mechanism of SnO<sub>2</sub> gas sensors is largely based on the surface reactions, where the concentration change of the surface adsorbed oxygen leading to conductivity change is associated with

\*Corresponding author. Tel.: +6567906387; fax: +6567902631. *E-mail address:* hhuang@ntu.edu.sg (H. Huang). the detection of a specific gas in the environment. Recently, due to large surface-to-volume ratio and small grain size comparable to the depth of the surface space-charge layer, nanostructured SnO2 thin films have stimulated great interests. Tan et al. [10] found that the nanocrystalline  $x\text{SnO}_2$ - $(1-x)\alpha\text{Fe}_2\text{O}_3$  with grain size down to 10 nm showed good sensitivity and selectivity to ethanol. Study on grain size effect in gas sensing by Yamazoe et al. [11,12] showed that the CO and H<sub>2</sub> gas sensitivity was greatly enhanced when the reduction of grain size to the range comparable to or less than two times depth of the surface space-charge layer. A linear relationship between SnO<sub>2</sub> sensor surface area and its sensitivity to H<sub>2</sub> was reported by Li and Kawi [13]. It was proven that microstructure optimization of sensor materials to achieve small grain size and large surface area was an effective way to obtain high gas sensitivity and selectivity for SnO<sub>2</sub>-based gas sensors. Therefore, it is necessary to know how the preparation parameters affect on the microstructure of SnO<sub>2</sub> thin films before microstructure modifications are carried out.

In this paper, nanostructured  $SnO_2$  thin films were prepared by PECVD and the effects of deposition parameters on the microstructure of  $SnO_2$  thin films were studied. These preliminary results are helpful to optimize the microstructure of  $SnO_2$  thin films to achieve high-performance  $SnO_2$ -based gas sensor.

## 2. Experimental procedure

The deposition system was a custom-designed PECVD system, and details of the system are shown in Ref. [14]. Inductively coupled plasma (ICP) was used as the plasma source and it was generated by a 13.56 MHz, 0.4–1.5 kW RF generator coupled on a copper tape circled around the cylindrical quartz deposition chamber. Dibutyltin diacetate (Aldrich, 98% purity), (C<sub>4</sub>H<sub>9</sub>)<sub>2</sub>Sn(OOCCH<sub>3</sub>)<sub>2</sub>, was used as precursor. The precursor was soaked at 90 °C. Ar was used as the carrier gas at a flow rate of 100 and 100 sccm O<sub>2</sub> was used as the reaction gas. A 4" SiO<sub>2</sub>/Si wafer was used as the substrate and it was put in the downstream of the plasma for deposition. Neither additional substrate heating during deposition nor post-annealing was performed.

The phase of the films was studied by a Rigaku X-ray diffractometer (XRD) using Cu  $K_{\alpha}$  radiation. The mean grain size was estimated by Scherrer's equation. A Leo 1550 field emission-type scanning electron microscopy (SEM) was used to observe the morphologies of the SnO<sub>2</sub> thin films. Thickness of the films was measured by a Filmetric F20 Thin-Film Measurement System and a J.A. Woollam VB-250 ellipsometer.

#### 3. Results and discussions

## 3.1. Effect of substrate-nozzle distance

Substrate-nozzle distance  $(D_{\rm sn})$  was denoted as the distance between the substrate and the injection nozzle of vapor outlet. It should be indicated that the substrate

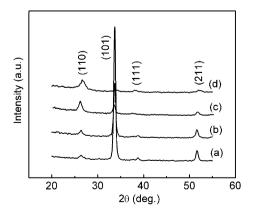


Fig. 1. XRD patterns of the  $SnO_2$  thin films deposited at  $D_{\rm sn}=$  (a) 12.5 cm, (b) 16.5 cm, (c) 20.5 cm and (d) 24.5 cm.

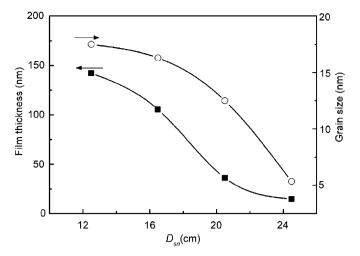


Fig. 2. Effect of  $D_{\rm sn}$  on the thickness and grain size of the SnO<sub>2</sub> thin films.

was exactly located in the middle of the plasma for  $D_{\rm sn}=12.5\,{\rm cm}$  and the substrates position shifted to the downstream of the plasma for  $D_{\rm sn}$  larger than 12.5 cm. Fig. 1 shows the XRD patterns of the SnO<sub>2</sub> thin films deposited at various  $D_{\rm sn}$ . It indicated that the as-deposited SnO<sub>2</sub> thin films were well crystallized even without additional substrate heating. All diffraction peaks could be indexed by Cassiterite SnO<sub>2</sub> (PDF 41-1445). A shift of preferred orientation from the (101) plane to (110) plane was noted with the increase of  $D_{\rm sn}$  from 12.5 to 24.5 cm. Fig. 2 shows the effect of  $D_{\rm sn}$  on the thickness and mean grain size calculated by Scherrer's equation. As the  $D_{\rm sn}$  increased from 12.5 to 24.5 cm, the grain size decreased from 17 to 5 nm while the thickness of the films also decreased dramatically.

Fig. 3 shows the SEM micrographs of the  $\mathrm{SnO}_2$  thin films deposited at  $D_{\mathrm{sn}} = (a)$  12.5 cm, (b) 16.5 cm and (c) 20.5 cm. The  $\mathrm{SnO}_2$  thin film deposited in the middle of the plasma ( $D_{\mathrm{sn}} = 12.5$  cm) was porous and showed flower-like microstructure (Fig. 3a), which was possibly caused by the competition effect between deposition and plasma sputtering. The grains in the  $\mathrm{SnO}_2$  thin film deposited at 16.5 cm was regular in uniform size of 40 nm, and some tetragonal grains could be observed (Fig. 3b), which indicated that the films were well crystallized and developed. The SEM micrographs confirmed that the grain size of the films decreased with the increase of  $D_{\mathrm{sn}}$ .

## 3.2. Effect of the RF power

Fig. 4 shows the XRD patterns of the  $SnO_2$  thin films deposited at  $D_{\rm sn}=20.5\,\rm cm$  with RF power of (a) 400 W, (b) 800 W, (c) 1200 W and (d) 1500 W. Similar shift of preferred orientation from (101) plane to (110) plane as shown in Fig. 1 was observed with the increase of RF power from 400 to 1500 W. Fig. 5 shows the effect of RF power on the thickness and grain size of the  $SnO_2$ 

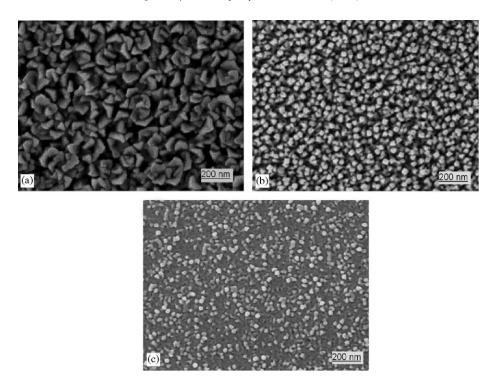


Fig. 3. SEM micrographs of the  $SnO_2$  thin films deposited at  $D_{sn} = (a) 12.5 \, \text{cm}$ , (b) 16.5 cm and (c) 20.5 cm.

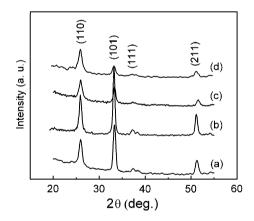


Fig. 4. XRD patterns of the SnO<sub>2</sub> thin films deposited at  $D_{\rm sn}=20.5\,{\rm cm}$  with RF power of (a) 400 W, (b) 800 W (c) 1200 W and (d) 1500 W.

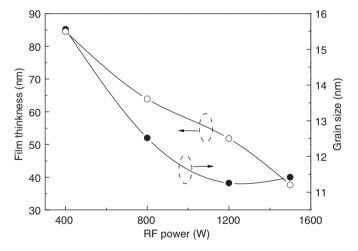


Fig. 5. Effect of RF power on the thickness and grain size of the  $SnO_2$  thin films.

thin films. It showed that the grain size as well as the films thickness decreased with the increasing RF power. Higher RF power increases both electron density and energy, accelerating reactions of precursor in the plasma phase, which are both prerequisites for the formation of SnO<sub>2</sub> and SnO [15]. Thus, the concentration of the SnO<sub>2</sub> components is higher in the plasma, and results in a higher nuclei rate of SnO<sub>2</sub> in the plasma. It is known that higher nuclei rate consumes most of the precursor in a shorter time and results in a smaller grain size. However, as the clusters in the plasma with higher RF power are

more active, they are likely to deposit on the chamber wall and cause precursor loss, which results in thinner films.

Fig. 6 shows the SEM micrographs of the  $SnO_2$  thin films deposited at  $D_{\rm sn} = 20.5\,\rm cm$  with RF power at (a)  $400\,\rm W$ , (b)  $800\,\rm W$ , (c)  $1200\,\rm W$  and (d)  $1500\,\rm W$ . The  $SnO_2$  thin films deposited at  $400\,\rm W$  were very smooth and dense, while the  $SnO_2$  thin films deposited at higher RF power showed rough surface morphologies. The grain size of the  $SnO_2$  thin films seemed to decrease with the increasing RF

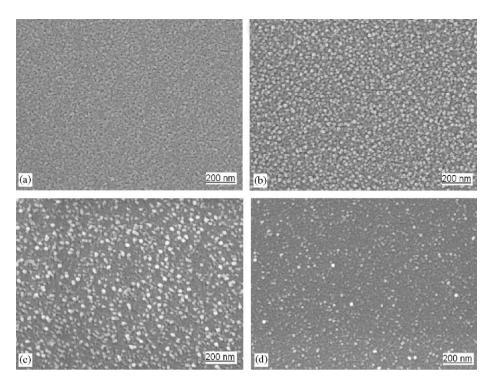


Fig. 6. SEM micrographs of the  $SnO_2$  thin films deposited at  $D_{sn} = 20.5$  cm with RF power of (a) 400 W, (b) 800 W, (c) 1200 W and (d) 1500 W.

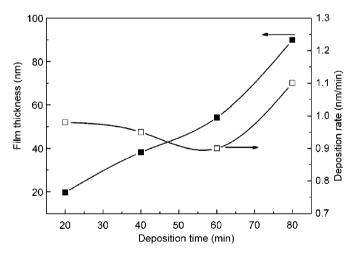


Fig. 7. Thickness and average deposition rate of the  ${\rm SnO_2}$  thin films deposited at increasing deposition time.

power, which was in accordance with the results obtained by Scherrer's equation (Fig. 5).

# 3.3. Effect of deposition time

Fig. 7 shows the thickness and average deposition rate of the  $SnO_2$  thin films deposited at increasing deposition time. The thickness of the  $SnO_2$  thin films increased with the

increase of deposition time and the average deposition rate ranged from 0.9 to 1.1 nm/min.

Fig. 8 shows the SEM micrographs of the  $SnO_2$  thin films deposited at 20.5 cm for (a) 20 min, (b) 40 min, (c) 60 min and (d) 80 min. The  $SnO_2$  thin film deposited for 20 min was very dense and the grain size was 18 nm. With the deposition time increased to 80 min, the surface of the as-deposited films was rough and the grain size increased to 32 nm.

## 4. Conclusions

Nanocrystalline SnO<sub>2</sub> thin films were deposited on SiO<sub>2</sub>/Si substrate by PECVD using dibutyltin diacetate as precursor. The as-deposited SnO<sub>2</sub> thin films were well crystallized even without additional substrate heating during deposition. The deposition parameters such as substrate distance, RF power and deposition time showed notable effects on the microstructure of the SnO<sub>2</sub> thin films, so the grain size and morphologies of the SnO<sub>2</sub> thin films can be optimized via the tuning of deposition parameters to achieve high-performance SnO<sub>2</sub>-based gas sensors.

## Acknowledgments

This work is sponsored by the A\*STAR of Singapore (No. 0221010022).

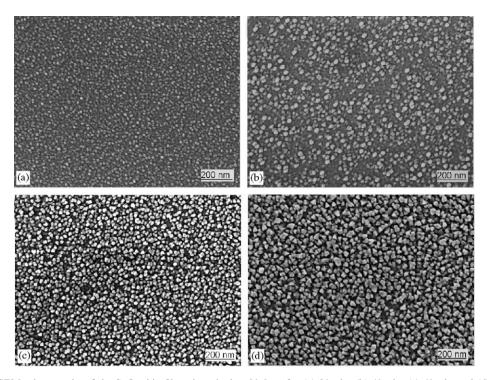


Fig. 8. SEM micrographs of the SnO<sub>2</sub> thin films deposited at 20.5 cm for (a) 20 min, (b) 40 min, (c) 60 min and (d) 80 min.

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