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The Synthesis and Biological Evaluation of a Novel Series of Antimicrobials of the Oxazolidinone Class

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Abstract—A novel series of antimicrobials of the oxazolidinone class is disclosed. These compounds are characterized relative to previously described analogues by a 'halostilbene-derived' pharmacophore and demonstrate enhanced antimicrobial activity against key Gram-positive pathogens when compared to Linezolid. © 2002 Elsevier Science Ltd. All rights reserved.

The oxazolidinones represent a novel class of antimicrobial agent that possess excellent activity against antibiotic-sensitive and MIC90 resistant Gram-positive bacteria.1 The prototype member of this class, Linezolid, has recently been approved for a number of clinical applications in man including the treatment of nosocomial and community-acquired pneumonia and skin infections caused by Staphylococcus aureus/Methicillin resistant S. aureus, Vancomycin resistant Enterococci, and Streptococcus pneumoniae (Pen-S). We were interested in identifying second generation compounds that had greater antimicrobial potency and less potential for side effects such as myelosuppression, including anaemia, and thrombocytopenia.2 To this end, we pursued a series of stilbene/'heterostilbene'-based analogues of general structure A whose conception was based on the structure and activity of a previously disclosed α, β unsaturated ester³ and a series of amide based phenyloxazolidinone derivatives.4 In this context, it should be remembered that vinyl fluorides are a well-known class of amide bond mimetics.⁵

The compounds wherein the halogen group is fluoro were constructed as outlined in Scheme 1. Synthesis of the corresponding bromo analogues is shown in Scheme 2. The sequence to access the target vinyl fluorides featured as key components the palladium(0) catalyzed oxazolidinone/aryl halide coupling reaction recently disclosed by us^{6,7} and the known but critical aldehyde to α -fluoro vinyl stannane conversion introduced by McCarthy and co-workers.^{8–10} The α -fluoro vinyl stannane could be subsequently coupled under Pd(0) catalyzed conditions with the requisite aryl halide. This reaction proved to be especially capricious with only 11 out of 45 reactions providing product in modest yield (range 2–40%).

Scheme 1. Synthesis of fluorostilbenes: (a) *p*-TsCl, pyridine, -5°C, 2 h, 95%; (b) potassium phthalimide, DMF, 80°C, 5 h, 75%; (c) 4-bromo-2-fluorobenzaldehyde, Pd₂dba₃, Cs₂CO₃, BINAP, toluene, 100°C, 16 h, 77%; (d) fluoromethyl phenyl sulfone, (EtO)₂P(O)Cl, LiN(TMS)₂, THF, 93%; (e) (i) N₂H₂-H₂O, EtOH; (ii) A₂O, pyridine/CH₂Cl₂Cl₃(f) *n*-Bu₃SnH (4 equiv), AIBN, benzene, 80°C, 81%; (g) R-X (1.01 equiv), CuI (0.95 equiv), Pd₂dba₃ (0.05 equiv), *Pt*-Bu₃ (0.2 equiv), dioxane (0.2 M), 2–8 h, 75°C.

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Scheme 2. Synthesis of bromostilbenes: (a) polystyrene-PPh₃ (6 equiv), CBr₄ (3 equiv), CH₂Cl₂, -5°C, 1 h, 93%; (b) N₂H₄–H₂O (3 equiv), THF/EtOH, 70°C, 3 h; (c) Ac₂O, pyridine, rt, 5 min, 77% total for (b)+(c); (d) R-B(OH)₂, Pd(PPh₃)₄ (0.1 equiv), 10% Na₂CO₃/H₂O, DME, 70°C, 16 h.

Scheme 2 featured a new and useful version of the Corey–Fuchs¹¹ aldehyde to dibromoolefin conversion using polymer bound triphenylphosphine. This modification avoids the tedious removal of triphenyphosphine that is associated with the parent procedure. The 1,1-dibromoolefins coupled effectively with a variety of aryl halides to provide target analogues in poor to good yields (range 12–82%).

The compounds were evaluated for antimicrobial activity against a panel of representative Gram-positive pathogens including clinically relevant strains. Many compounds demonstrated significantly better activity than Linezolid against several of these organisms. In

Table 1. Structure-activity relationships in halostilbene oxazolidinones

Compd	R^1	R ²	S. aureus MIC ^a (µg/mL)	S. epidermidis MIC ^a (µg/mL)	E. faecium MIC ^a (µg/mL)	M. cat. MIC ^a (μg/mL)	S. pneumo. MIC ^a (µg/mL)	MPS ^b IC ₅₀ (μM)	MAO B ^c IC ₅₀ (μM)
			(48/1112)	(\$8,1112)	(F8/1112)	(µ8/1112)	(F8/1112)		
Linezolid 1			4	1	2	4	1	12 (7)	17 (5)
2	5-CN-3-Pyridyl	Н	4	2	2	16	16	9.4 ± 1.4	> 300 (2)
3	5-CN-3-Pyridyl	F	i	0.5	1	4	< 0.125	< 0.1 (2)	> 300 (2)
4	5-CN-3-Pyridyl	Br	1	0.25	< 0.125	2	< 0.125	1.3 ± 0.2	> 300 (2)
5	3-Pyridyl	H	4	2	2	16	16	7.8 ± 1.0	59 (2)
6	3-Pyridyl	F	4	0.25	1	8	0.5	1.6 ± 0.1	98 (2)
7	3-Pyridyl	Br	2	1	2	8	1	12 (2)	121 (4)
8	F	3-Pyridyl	8	4	4	32	4	(-)	(.)
9	4-Pyridyl	Br	8	2	1	16	i	14(3)	186 (2)
10	4-MeO-Pyridin-3-yl	Br	8	4	2	32	2	- 1(0)	(-)
11	2-CN-Ph	Br	16	4	4	128	2	36 (2)	> 150 (2)
12	3-CN-Ph	F	> 128	128	> 128	> 128	2	> 300 (2)	> 300 (2)
13	3-CN-Ph	Br	0.25	0.125	0.125	1	0.25	24 (2)	> 150 (4)
14	4-CN-Ph	Br	4	1	2	> 128	0.5	76 (2)	> 150 (4)
15	3-Acetyl-Ph	F	0.5	1	2	> 128	1	69 ± 12	> 300 (2)
16	3-Acetyl-Ph	Br	2	ĺ	1	4	1	**	(-)
17	4-Acetyl-Ph	Br	64	64	32	128	i		
18	3-CHO-Ph	F	1	0.5	1	64	4	9.9 ± 0.8	> 300 (2)
19	3-CHO-Ph	Br	0.5	0.25	0.5	1	i	33 (2)	> 300 (2)
20	3-CONH ₂ -Ph	Br	4	4	2	16	4	(-)	(_)
21	3-CO ₂ Me-Ph	Br	8	8	4	> 128	32	150 (2)	> 300 (2)
22	3-CH ₂ OH-Ph	Br	2	1	0.5	4	1	8 (2)	> 150 (4)
23	3-NH ₂ -Ph	Br	2	1	2	8	2	- (-)	(.)
24	3-N-Acetyl-Ph	Br	4	2	2	64	2		
25	5- <i>i</i> Pr-2-MeO-Phenyl	Br	> 128	> 128	> 128	32	64		
26	Ph	Br	16	16	16	128	2	> 300 (2)	< 9.4 (2)
27	H	F	8	8	8	32	2	(_)	(_)
28	PhSO ₂ —	F	8	8	8	64	8		
29	2-Furyl	Br	1	0.25	0.5	1	0.5		
30	2-Thiazolyl	F	2	2	2	8	2		< 9.4 (1)
31	2-CN-Thiophen-3-yl	F	> 128	16	16	> 128	8	> 300 (2)	285 (2)
32	5-Ac-Thiophen-2-yl	Br	1	0.5	0.5	2	0.5	(-)	(-)
33	2-Me-2 <i>H</i> -Tetrazol-5-yl	F	8	1	4	32	2	4.9 ± 0.5	> 300 (2)
34	5-Pyrimidinyl	F	2	i	2	4	<u>-</u>		475 (1)
35	2-Amino-5-pyrimidinyl	F	16	4	8	32	8		(1)
36	3,5-di-Me-Oxazol-4-yl	Br	32	16	32	64	8	54 (2)	21 (2)

^aNational Committee for Clinical Laboratory Standards. 1997. Methods for Dilution Antimicrobial Susceptibility Tests for Bacteria that Grow Aerobically: Approved Standard M7-A5. National Committee for Clinical Laboratory Standards, Wayne, PA, USA.

 $^{^{}b}$ Mitochondrial Protein Synthesis (MPS) IC₅₀ expressed as a mean \pm STD or as a mean of (n) experiments. See ref 12 for assay protocol.

 $^{^{\}circ}$ Monoaminoxidase B IC₅₀ expressed as a mean of (n) experiments. Assay adapted from a Molecular Probes kit. The assay was carried out using rat primary hepatocytes because the rat was the primary model for toxicity used in assessing these compounds. We have reported the MAO B data because, in all cases, the compounds were significantly more inhibitory using substrate for this isozyme than for the MAO A isozyme, where little if any inhibition was noted.

addition, the compounds were evaluated in two key toxicity/selectivity assays (a) inhibition of mitochondrial protein synthesis (this effect may be, at least in part, responsible for the hematologic toxicities associated with Linezolid)¹² and (b) inhibition of monoamine oxidase. Although this is not a significant clinical problem with Linezolid, it is a class effect for oxazolidinones¹³ and needs to be avoided. The biological SAR data is highlighted in Table 1. Several compounds from either series, gratifyingly, showed increased potency, reduced MPS and MAO inhibitory activity compared to Linezolid. Although it was difficult to draw SAR conclusions on the series in a general sense (no clear cut patterns or direct comparison between the bromo and fluoro series being impossible due to the relative inaccessibility of target compounds) some general comments are instructive. Compounds containing halogen-substituted double bonds were as a rule more potent than their corresponding des-halo counterparts (2–4 and 5– 7). Within a halogenated series, the bromo analogues were generally more potent than the corresponding fluoro compounds (3-4, 6-7, 12-13, 15-16, and 18-19). Activity of compounds containing the (Z)-fluorostilbene unit was greater than that of the corresponding (E)-isomer (6 and 8). A variety of R¹ groups were well tolerated with a general trend that heterocycles bearing electron-withdrawing groups (higher sigma values) were the most active in the antimicrobial screens. There was a potency enhancement associated with substituents on the meta position of phenyl groups (13 and 14, 16 and 17) or with a 3-pyridyl moiety to form a 'heterostilbene' (7 and 9). This was not generally true for analogues that bore a phenyl group substituted with an electron-withdrawing group in the para position, for example 17.

Representative compounds with potency comparable to 1 were selected for pharmacokinetic evaluation in cassette dosing format in the rat. Compound 6 displayed characteristics that were considered to be appropriate for in vivo evaluation in animal infection models. For example, the PK profile of 6 following a 5 mpk iv dose was characterized by a low plasma clearance (CLp=0.25 L/h kg), with a volume of distribution of 0.42 L/kg and an apparent elimination half life of \sim 1.2 h. When administered orally at the same dose, the bioavailability of the compound was found to be 65%, n=6). The outcome of these in vivo studies will be reported upon in due course.

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