Organic & Biomolecular Chemistry

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Cite this: Org. Biomol. Chem., 2011, 9, 8301

www.rsc.org/obc PAPER

Synthesis and biological evaluation of potential new inhibitors of the bacterial transferase MraY with a β-ketophosphonate structure†

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Received 8th July 2011, Accepted 14th September 2011 DOI: 10.1039/c1ob06124k

Stable analogs of bacterial transferase MraY substrate or product with a pyrophosphate surrogate in their structure are described. β-ketophosphonates were designed as pyrophosphate bioisosteres and were investigated as UDP-GlcNAc mimics. The developed strategy allows introduction of structural diversity at a late stage of the synthesis. The biological activity of the synthesized compounds was evaluated on the MraY enzyme.

Introduction

Enzymes involved in peptidoglycan biosynthesis are key targets for the development of new antibiotics because there are no parallels of this pathway in eukaryotes. Peptidoglycan is a heteropolymer of bacterial cell walls consisting of long glycan chains made of β -1 \rightarrow 4 alternating units of N-acetylmuramoyl-peptides (MurNAcpeptides) and N-acetylglucosamine (GlcNAc) which are crosslinked together through the short peptide chains. This macromolecule protects the cell from internal osmotic pressure effects and contributes to the cell shape maintenance. Peptidoglycan biosynthesis is a complex process which takes place in various cell compartments: cytoplasm, membrane and periplasm² (Fig. 1).

UDP-*N*-acetylglucosamine (UDP-Glc*N*Ac) is first transformed into UDP-*N*-acetylmuramoyl-pentapeptide (UDP-Mur*N*Acpentapeptide) by the Mur enzymes (MurA, MurB, MurC, MurD, MurE and MurF).³ Then, at the plasma membrane, two distinctive proteins (MraY and MurG) synthesize polyprenyl-linked precursors (lipid I and lipid II) that carry one complete cell-wall subunit (Fig. 2).⁴

Thus, MraY⁵ transfers the phospho-MurNAc-pentapeptide moiety from the cytoplasmic precursor to the membrane ac-

ceptor undecaprenyl phosphate (C₅₅-P) yielding undecaprenyl-pyrophosphoryl-MurNAc-pentapeptide (lipid I) and releasing uridine monophosphate (UMP). Thereafter, MurG⁶ catalyzes the transfer of the GlcNAc moiety from UDP-GlcNAc to lipid I yielding undecaprenyl-pyrophosphoryl-MurNAc-(pentapeptide)-GlcNAc (lipid II). Lipid II is then translocated through the membrane by a specific flippase. This translocation activity was shown for FtsW⁷ from *E. coli*. In the outside of the membrane the lipid II is then used as a substrate for the polymerisation reactions, transglycosylation and transpeptidation, catalyzed by the penicillin-binding proteins (PBPs) and the monofunctional transglycosylases (MGTs).⁸

The enzymes involved in this biosynthetic pathway have been demonstrated as ubiquitous and essential to bacteria^{2,3,4a,8,9} and any anomaly in this biosynthesis leads to cell lysis. Therefore, enzymes implicated in this process are ideal targets in the search for new antibacterials.10 Indeed, several families of widely used antibiotics, such as β-lactams and lipoglycopeptides, are already known to interfere with certain enzymes implicated in peptidoglycan biosynthesis. However, because of the ever-increasing resistance¹¹ to these antibacterial agents and in the context of a program¹² directed to the synthesis and biological evaluation of new antibacterials, we are focusing on the development of MraY inhibitors. Indeed, this enzyme has been little exploited so far as a consequence of its trans-membrane localisation. 13 Therefore, since there is no drug in clinical use targeting this essential enzyme, this can be expected to delay the emergence of resistance. Our goal is to target compounds which are stable analogs of MraY substrate or product and displaying a pyrophosphate surrogate in their structure (Fig. 3). Furthermore, some of these compounds can also be considered as potential inhibitors of the MurG enzyme due to their structural analogy with the substrate of this enzyme. Results concerning the synthesis of a first compound have been published in a preliminary form¹⁴ and we are now aiming at reporting our full results dealing with a series of related compounds.

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[†] Electronic supplementary information (ESI) available: 1. Numbering system, 2. Comprehensive experimental section, 3. ¹H, ¹³C and ³¹P NMR spectra. See DOI: 10.1039/c1ob06124k

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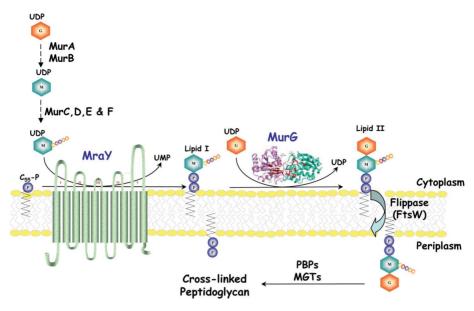


Fig. 1 Peptidoglycan biosynthesis pathway.

Fig. 2 Enzymatic reactions catalyzed by MraY and MurG enzymes.

Fig. 3 General structure of MraY and MurG substrates and products and targeted β -ketophosphonate isosteres.

The pyrophosphate present in substrate and product of the reaction catalyzed by MraY is on one hand linked to a GlcNAc derivative and on the other hand to either uridine or undecaprenol. This diphosphate moiety is supposed to interact with the divalent cation Mg^{2+} which is a co-factor of the enzymatic reaction, suggested to proceed in either one or two steps. Furthermore, it is prone to hydrolysis due to its positioning in the anomeric position of the glycosyl derivative. That is why our strategy has been to replace this pyrophosphate, site of the enzymatic reaction, by a bioisostere displaying a β -ketophosphonate structure. In the resulting compounds, the replacement of the sugar anomeric oxygen atom is expected to enhance the stability of the inhibitors with

regards to hydrolysis. Furthermore, the carbonyl group is proposed as an alternative electrophilic center to the one involved in the enzymatic reaction. The β -ketophosphonate moiety displaying both a phosphonate anion and a polarized carbonyl group should be able to chelate the Mg^{2+} co-factor. Consequently, the targeted β -ketophosphonate pyrophosphate bioisosteres should be pertinent mimics of the substrate or the product, since they have the required electronic properties to coordinate the metallic co-factor of the reaction as well as being stable to hydrolysis.

Results and discussion

The design and synthesis of pyrophosphate analogs of glycosyl nucleotides have already been described (Fig. 4). Thus, methylene diphosphonate¹⁷ and 1-*C*-phosphonophosphate¹⁸ analogs of UDP-Glc or UDP-Glc*N*Ac, respectively, have been reported. Different ways of diphosphate-type connections between Glc*N*Ac and uridine moieties have also been investigated.¹⁹ In some cases, the pyrophosphate has been replaced by a propenyl phosphonate,²⁰ a hydroxypropyl phosphonate²¹ or a propyl phosphate.²²

Fig. 4 Structures of pyrophosphate analogs of glycosyl nucleotides previously described.

Other mimics have no phosphorus atoms in their structure, the pyrophosphate moiety being changed to a malonic linkage, 23 α , β -ethylenic ester or amide with or without dihydroxylation of the double bond. 24 In addition, an oxycarbonyl aminosulfonyl linker 25 and a carbonyl linked to O- or N-sulfamoyl uridine 26 have also been reported as well as allophanates and biuret derivatives. 27 Galactose-linked uridine derivatives without charge or dipole contributions in the linker have been synthesized via cross metathesis. 28 Finally, replacement of the pyrophosphate by a triazole moiety has also been recently described. 29 However, none of these pyrophosphate analogs display a glycosyl β -ketophosphonate skeleton. Nevertheless, it should be mentioned that a uridine-containing phosphonate ester analog has previously been synthesized and tested as inhibitor of solubilised E. coli transferase MraY (IC $_{50}$ 3.7 mM). 30

Chemical synthesis

Two complementary strategies to obtain the targeted compounds have been investigated (Fig. 5). On one hand, the retrosynthesis relies on the condensation of variously substituted lithiomethylenephosphonates on the ester function of a conveniently protected methyl N-acetylglucosaminyl acetate (path a) and on the other hand it involves the esterification of a β -ketophosphonate derivative using different alcohol derivatives (path b). The residue \mathbb{R}^3 to be introduced on the phosphonate

Path a Ph O O O TBSO AcHN
$$CO_2Me$$
 OP $P - OR^3$ P

Fig. 5 Retrosynthetic analysis.

has been chosen to mimic either the substrate or the product of the reaction catalyzed by the transferase MraY. Accordingly, the uridine moiety present in the substrate has been introduced as it is or as a C_3 -alkyl uracil analog and the undecaprenyl chain, part of lipid I, has been replaced by a citronellyl or a simple hexadecanyl moiety.

The ester building block 3 (Scheme 1) was synthesized from orthogonally protected α -1-C-allyl-N-acetylglucosamine 1 readily obtained in four steps (50% overall yield as an α/β mixture of anomers) from commercial GlcNAc. ³¹ Protection of the secondary alcohol function as its *tert*-butyldimethylsilyl ether derivative 2 allowed isolation of pure α anomer from the 24/1 α/β mixture. Finally, ozonolysis of the alkene under basic conditions in the presence of methanol gave the ester 3. ³²

Scheme 1 Synthesis of C-glycosyl ester 3. Reagents and conditions: a) TBSCl, imidazole, DMF, 97%; b) O₃, CH₂Cl₂, MeOH, NaOH, 65%.

According to path a, we next turned to the synthesis of unsymmetrical methylphosphonates 7 and 8 substituted by an uridinyl or a citronellyl moiety. Their preparation (Scheme 2) involved NaH treatment, in the presence of methyl iodide, of commercially available dibenzyl *H*-phosphonate, leading to dibenzyl methylphosphonate 4. It was followed by benzyl monodeprotection in the presence of DABCO in refluxing toluene³³ to afford monobenzyl methylphosphonate 5. Esterification with isopropylidene *N*-Boc uridine³⁴ or citronellol was performed under Mitsunobu conditions in the presence of triphenyl phosphine and diisopropyl azodicarboxylate affording the phosphonate 7 or 8 as a diastereomeric mixture.

Access to the targeted β -ketophosphonates 9 and 10 was then tentatively carried out, by condensation on the ester 3 of lithio

OBn
$$H o OBn$$
 $A o P o OP$
 $O o$

Scheme 2 Synthesis of unsymmetrical methyl phosphonates 7 and 8. Reagents and conditions: a) i. NaH, THF, -15 °C to r.t.; ii. MeI, -40 °C to r.t., 80%; b) i. DABCO, toluene, reflux; ii. H+, H2O, 82%; c) DIAD, PPh3, THF. 83% for 7 and 95% for 8.

derivatives of phosphonates 7 and 8, generated in the presence of *n*-butyllithium (Scheme 3). However, even if these conditions proved successful for the synthesis of the β -ketophosphonate 10, they did not allow that of the uridinyl related compound 9. It has to be noted that at least 3 equivalents of the nucleophilic entity are required to achieve the reaction due to both the acidity of the acetamide and that of the methylene within the resulting β-ketophosphonate. However, this excess of reagent renders its separation from the formed compound 10 difficult and precludes its use in higher excess in order to improve the yield.

Scheme 3 Towards targeted β-ketophosphonates 9 and 10 according to path a. Reagents and conditions: a) i. nBuLi, THF, -78 °C, 1 h; ii. 3, THF, -78 °C to r.t., 57% for 10.

In an alternative manner, the synthesis of the β ketophosphonate 9 was envisaged according to the path b (Scheme 4). It first involved the condensation of the dibenzyl methylenephosphonate anion, generated from 4 in the presence of *n*-butyllithium, on the ester 3 giving the dibenzyl β ketophosphonate 11. Then, mono-benzyl deprotection in the presence of DABCO led to 12 and was followed by the esterification with isopropylidene N-Boc uridine under Mitsunobu conditions in the presence of triphenylphosphine and diisopropyl azodicarboxylate affording the targeted compound 9. It has to be mentioned that N-Boc protection of uridine is necessary to achieve this reaction since the esterification of 12 with isopropylidene uridine under the same conditions was unsuccessful. In order to compare biological activity, compound 11 was also submitted to simultaneous acidic hydrolysis of benzylidene acetal and silyl ether to afford 13 which was then mono-deprotected with DABCO followed by DOWEX H+ treatment to provide 14. The complete deprotection of the phosphonate 9 was then undertaken. Acidic hydrolysis was first performed and allowed the simultaneous removal of benzylidene, isopropylidene, tertbutyldimethylsilyl and Boc groups. Finally, cleavage of the benzyl ester was achieved by hydrogenolysis in the presence of palladium on charcoal in methanol leading to 16 as a first analog of the MraY substrate.

In a complementary manner, access to other substrate or product analogs of the reaction catalyzed by MraY were targeted (Scheme 5). Therefore, on the one hand, a simplified structure of the uridine moiety has been chosen in which the ribose of uridine has been replaced by a C_5 acyclic alkyl chain, 35 retaining the same five bond linkage between the alcohol function and the uracil residue as in uridine. The imide function present in this 1-(5'hydroxypentyl)uracil was protected as its tert-butyl carbamate 17 to avoid side reactions. On the other hand, the undecaprenyl chain present within the product of the enzymatic reaction was replaced by either citronellyl or hexadecanyl. The citronellyl moiety has been preferred to a terpenic alcohol due to the better stability of homo-allylic phosphonate in acidic conditions compared to allylic phosphonate. Accordingly, phosphonate 12 has been reacted with 1-(5'-hydroxypentyl)-N-Boc uracil 17, citronellol and hexadecanol, under Mitsunobu conditions, affording the protected βketophosphonates 18, 10 and 19, respectively, as diastereomeric mixtures. Acidic hydrolysis of acid labile protective groups was then carried out in the presence of 1/1 TFA/H2O which led to 20, 21 and 23, respectively. Compound 21 resulted from the addition of water on the citronellyl double bond, catalyzed by the acidic conditions that were used. To limit this side reaction, the acidic hydrolysis of citronellyl β-ketophosphonate 10 was also performed in smoother conditions in the presence of 1/1/1 TFA/H₂O/THF and afforded 22. Final benzyl ester cleavage of compounds 20, 21 and 23 was achieved by hydrogenolysis in the presence of palladium on charcoal 10% leading to the final βketophosphonates 24, 25 and 26, respectively.

Biological evaluation

The *in vitro* biological evaluation (Table 1) of compounds **13–16** and 20-26, on MraY purified from Bacillus subtilis, was carried out as described in the experimental section. The residual activity of the enzyme was measured in the presence of 1 mM of the tested compounds. For the most active inhibitors, the IC50 value was determined.

Commercially available tunicamycin from Streptomyces sp. was used as a positive control in the tests and resulted in an IC50 value

Scheme 4 Synthesis of substrate analog 16 according to path b. Reagents and conditions: a) i. nBuLi, -78 °C, 1 h; ii. 3, -78 °C to r.t., 83%; b) i. DABCO, toluene, reflux; ii. Dowex H+, MeOH, 90% for 12 and 88% for 14; c) TFA/H₂O: 1/1, 97% for 13 from 11 and 94% for 15; d) 6, DIAD, PPh₃, THF, 46% or PyBOP, DIEA, DMF, 20%; e) H₂, Pd/C 10%, MeOH, 85%.

Scheme 5 Generalisation to the synthesis of β-ketophosphonates 24, 25 and 26. Reagents and conditions: a) DIAD, PPh₃, THF, 27% for 18, 58% for 10, 42% for 19; b) TFA/H₂O: 1/1, 93% for 20, 94% for 21, 90% for 23; c) TFA/H₂O/THF: 1/1/1, 84% for 22; d) H₂, Pd/C 10%, MeOH, 94% for 24, 100% for **25**, 87% for **26**.

Table 1 Inhibitory activity of synthesized compounds on the MraY enzyme

Compound		13	14	15	20	21	22	23
HO O O OBn OBn P-OR1	R ¹ =	Bn	Н	N NH HO OH	NN NH	,×́\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	*****	÷C ₁₆ H ₃₃
IC ₅₀ (mM)		NI	NI	NI	0.94	0.19	NI	0.82
Compound		16	24	25	26			
HO O OH OH OH O'S OR 1	R ¹ =	HO OH	NN NH	, он	÷C ₁₆ H ₃₃			
IC_{50} (mM)		NI	NI	1.3	0.85			
NI: no inhibition at 1 mM co	oncentra	ation.						

equal to 0.012 mM. Furthermore, the inhibitors were also tested on the MurG enzyme from E. coli which catalyzes the second membrane step of peptidoglycan biosynthesis (Fig. 1), leading to lipid II formation.

The detected inhibitory activities were proven to be rather weak concerning MraY inhibition since compounds 20, 23, 25 and 26 resulted in IC₅₀ values in the mM range while compounds 13, 14, 15, 16, 22 and 24 proved not to be inhibitors at 1 mM concentration. Interestingly, compound 21 exhibited significant inhibitory activity with an IC₅₀ value equal to 0.19 mM. Finally, no inhibition of MurG activity was observed for the tested compounds at 1 mM concentration. Therefore, the replacement of the phosphorus atom within the pyrophosphate moiety by a carbonyl group leads to low inhibitory activity which could indicate the requirement of a tetrahedral geometry instead of a trigonal one, in the related position, to ensure accurate interaction of the inhibitor with the enzyme. Furthermore, by comparison with compound 22, the activity observed for compound 21 is encouraging and suggests a possible interaction with the enzyme through a hydrogen bond involving the hydroxyl group present in the inhibitor.

Conclusion

In summary, we have developed a simple and efficient method for the synthesis of new pyrophosphate analogs of glycosyl nucleotides from N-acetylglucosamine. The strategy offers the advantage of allowing introduction of structural diversity at a late stage of the synthesis rendering easy access to a series of compounds. The method has been illustrated by the synthesis of several analogs of MraY substrate or product. The activity of the synthesized compounds has been evaluated on purified MraY and MurG. Unexpectedly, the inhibitory activities were low, except for one compound and the reason for this weak activity could lie in the change in the geometry of the phosphorus atom, site of the enzymatic reaction, probably preventing efficient recognition of the compounds by the enzyme. This study led to reaching some insight into the requirements for inhibition of the trans-membrane protein MraY. Taking into account these results, our current work is now focusing on new pyrophosphate mimics with a tetrahedral geometry in the β position of the pyrophosphate. Special attention will also be paid to the introduction of the pentapeptide side chain in C₃ position of GlcNAc moiety since it can be important for binding to the enzyme.

Experimental

Chemical synthesis

¹H NMR (500 MHz), ¹³C NMR (125 MHz) and ³¹P NMR (202 MHz) spectra were recorded in CDCl₃ unless indicated on a Bruker Avance or Avance II. Chemical shifts (δ) are reported in ppm and coupling constants are given in Hz. Optical rotations were measured on a Perkin-Elmer 341 polarimeter with a sodium (589 nm) lamp at 20 °C. Mass spectra, electrospray (ESI) and high resolution (HRMS) were recorded by the Service de Spectrométrie de Masse, ICSN Gif sur Yvette. Tunicamycin from Streptomyces sp. was purchased from Sigma-Aldrich®. All reactions were carried out under a nitrogen atmosphere, and were

monitored by thin-layer chromatography with Merck 60F-254 precoated silica (0.2 mm) on glass. Flash chromatography was performed with Merck Kieselgel 60 (200-500 µm); the solvent systems were given v/v. Spectroscopic ¹H, ¹³C and ³¹P NMR, MS and/or analytical data were obtained using chromatographically homogeneous samples.

Benzyl (3-N-tert-butyloxycarbonyl-2',3'-O-isopropylidene)-uridin-5'-yl 3-(2-acetamido-4,6-O-(R)-benzylidene-3-O-tert-butyldimethylsilyl-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 9

At 0 °C, DIAD (95 µL, 0.47 mmol, 1.5 eq.) was added dropwise to a solution of 12 (200 mg, 0.32 mmol, 1 eq.), 6 (121 mg, 0.32 mmol, 1 eq.) and PPh₃ (124 mg, 0.47 mmol, 1.5 eq.) in THF (5 mL). The reaction mixture was stirred at r.t. for 16 h then concentrated. Purification by flash chromatography (CH₂Cl₂/acetone, 8:2 to 7:3) afforded 9 (146 mg, 46%, white solid) as a mixture of epimers (d.r. = 2/1): R_f 0.46 (CH₂Cl₂/acetone = 7:3); ¹H NMR δ 7.50–7.34 (m, 10H, H_{ar}), 7.28, 7.27 (2d, 1H, $J_{H6u-H5u} = 8.1$ Hz, H_{6u}° , H_{6u}^{*}), 6.28 (d, 1H, $J_{NH-H2'} = 8.2$ Hz, NH), 5.71, 5.66 (2d, 1H, $J_{H5u-H6u} = 8.1$ Hz, H_{50}° , H_{50}^{*}), 5.66, 5.58 (2d, 1H, $J_{H1''-H2''} = 2.1$ Hz, $H_{1''}^{*}$, $H_{1''}^{\circ}$), 5.50 (s, 1H, H₇), 5.18-5.08 (m, 2H, CH₂Ph), 4.98, 4.96 (2dd, 1H, $J_{\text{H2''-H3''}} = 6.4 \text{ Hz}, J_{\text{H2''-H1''}} = 2.1 \text{ Hz}, H_{2''}^*, H_{2''}^\circ), 4.81 \text{ (dd, 0.7H,}$ $J_{\text{H3"*-H2"*}} = 6.4 \text{ Hz}, J_{\text{H3"*-H4"*}} = 3.7 \text{ Hz}, H_{\text{3"}}^*), 4.79-4.73 \text{ (m, 1.3H,}$ $H_{1'}$, $H_{3''}$ °), 4.37–4.33, 4.33–4.29 (2 m, 1H, $H_{4''}$ °, $H_{4''}$ *), 4.29–4.19 $H_{3'}$), 3.67–3.61 (m, 1H, $H_{6'ax}$), 3.54–3.46 (m, 2H, $H_{4'}$, $H_{5'}$), 3.14, 3.08 (2dd, 1H, $J_{\text{H1a-P}} = 22.3$ Hz, $J_{\text{H1a-H1b}} = 13.9$ Hz, H_{1a}° , H_{1a}^{*}), 3.11, 2.99 (2dd, 1H, $J_{\text{H3a-H3b}} = 16.6$ Hz, $J_{\text{H3a-H1'}} = 7.0$ Hz, H_{3a}° , H_{3a}^{*}), 3.07, 3.04 (2dd, 1H, $J_{\text{HIb-P}} = 22.9$ Hz, $J_{\text{HIb-HIa}} = 13.9$ Hz, H_{Ib}° , H_{1b} *), 2.75, 2.70 (2dd, 1H, $J_{H3b-H3a} = 16.6$ Hz, $J_{H3b-H1'} = 6.6$ Hz, H_{3b}°, H_{3b}*), 1.93, 1.91 (2 s, 3H, CH₃CO*, CH₃CO°), 1.59, 1.58 (2 s, 9H, CO₂tBu*, CO₂tBu°), 1.55, 1.36, 1.34 (3 s, 6H, CMe₂), 0.83 (s, 9H, SitBu), 0.08, -0.01 (2 s, 4H, SiMe*), 0.07, -0.01 (2 s, 2H, SiMe°); ¹³C NMR δ 197.8 (d, J_{C2-P} = 6.5 Hz, C_2), 170.7, 170.7 (CH_3CO° , CH_3CO^{*}), 160.4 (C_{4u}), 148.3 (C_{2u}), 147.6, 147.6 $(CO_{Boc}^{\circ}, CO_{Boc}^{*}), 141.7, 141.3 (C_{6u}^{\circ}, C_{6u}^{*}), 137.3 (Cq_{ar}), 135.3 (d,$ $J_{\text{Cq-P}} = 5.0 \text{ Hz}, \text{Cq}_{\text{ar}}$, 129.3, 129.1, 129.0, 128.5, 128.3, 128.2, 126.4, 126.4 (CH_{ar}), 114.7, 114.7 ($\underline{C}Me_2^*$, $\underline{C}Me_2^\circ$), 102.1, 102.1, 102.0, $101.9\,(C_{5u},C_{7}),95.7,95.0\,({C_{1''}}^{\circ},{C_{1''}}^{*}),87.4,87.3\,(\underline{C}M{e_{3}}^{\circ},\underline{C}M{e_{3}}^{*}),$ 86.0, 85.8 (2d, $J_{C4''-P} = 7.0 \text{ Hz}$, $C_{4''}^{\circ}$, $C_{4''}^{*}$), 84.5, 84.5 ($C_{2''}^{*}$, $C_{2''}^{\circ}$), 83.2, 83.2 (C_4° , $C_{4'}$ *), 80.5 ($C_{3''}$), 70.7, 70.6 ($C_{1'}^{\circ}$, $C_{1'}$ *), 70.1, 70.1 $(C_{3'}^*, C_{3'}^\circ)$, 69.2 $(C_{6'})$, 69.0, 68.9 (2d, $J_{CH2Ph-P} = 6.5 \text{ Hz}$, CH_2Ph^* , CH_2Ph°), 65.7, 65.6 (2d, $J_{C5''-P} = 6.5$ Hz, $C_{5''}^{\circ}$, $C_{5''}^{*}$), 65.3, 65.3 $(C_{5'}^*, C_{5'}^\circ)$, 54.1, 54.1 $(C_{2'}^\circ, C_{2'}^*)$, 43.8, 43.7 (C_3^*, C_3°) , 42.0, 41.8 $(2d, J_{C1-P} = 127.0 \text{ Hz}, C_1^*, C_1^\circ), 27.6 (CMe_3), 27.2, 25.4 (CMe_2^\circ),$ 27.1, 25.3 (CMe₂*), 25.8 (SitBu), 23.2, 23.1 (CH₃CO*, CH₃CO°), 18.2 (SitBu), -3.9, -4.7 (SiMe); 31 P NMR δ 21.9 (s, 0.66P, P*), 21.7 (s, 0.33P, P°); MS (ESI): $m/z = 1022 [M + Na]^{+} 100\%$; HRMS calcd for C₄₈H₆₆N₃NaO₁₆PSi⁺ 1022.3848, found 1022.3858.

Benzyl (R)-citronellyl 3-(2-acetamido-4,6-O-(R)-benzylidene-3-*O-tert*-butyldimethylsilyl-2-deoxy-α-D-glucopyranosyl)-2oxopropylphosphonate 10

Condensation of 8 on 3. The reaction was performed as described for 11 with a solution of 8 (446 mg, 1.37 mmol, 3 eq.) in THF (10 mL), nBuLi (630 μ L, 1.42 mmol, 3.1 eq.) and a solution of 3 (220 mg, 0.46 mmol, 1 eq.) in THF (5 mL). Two successive purifications by flash chromatography (CH_2Cl_2 /acetone, 9:1 and cyclohexane/acetone, 2:1) afforded 10 (205 mg, 57%, pale yellow oil) as a mixture of epimers (d.r. = 52/48).

Mitsunobu with 12 and (R)-citronellol. At 0 °C, DIAD (70 μL, 0.36 mmol, 1.5 eq.) was added dropwise to a solution of 12 (150 mg, 0.24 mmol, 1 eq.), (R)-citronellol (44 µL, 0.24 mmol, 1 eq.) and PPh₃ (93 mg, 0.36 mmol, 1.5 eq.) in THF (4 mL). The reaction mixture was stirred at r.t. for 16 h then concentrated. Two successive purifications by flash chromatography (cyclohexane/acetone, 2:1 and CH₂Cl₂/acetone, 9:1) afforded 10 (107 mg, 58%, pale yellow oil) as a mixture of epimers (d.r. = 1/1): R_f 0.39 (cyclohexane/acetone = 2:1); IR (cm⁻¹) 1719 (CO), 1250 (PO); ¹H NMR δ 7.51–7.33 (m, 10H, H_{ar}), 6.68, 6.54 (2d, 1H, $J_{NH-H2'}$ = 8.6 Hz, NH*, NH°), 5.51, 5.50 (2 s, 1H, H_{7} *, H_{7} °), 5.19–5.02 (m, 3H, CH₂Ph, H_{6"}), 4.81 (ddd, 0.5H, $J_{\text{HI}}^{\circ} \circ_{-\text{H3a}}^{\circ} = 7.9 \text{ Hz}$, $J_{\text{HI}}^{\circ} \circ_{-\text{H2}}^{\circ} \circ =$ 5.9 Hz, $J_{\rm HI'^{\circ}-H3b^{\circ}} = 4.9$ Hz, $H_{\rm I'^{\circ}}$), 4.77 (ddd, 0.5H, $J_{\rm HI'^{*}-H3a^{*}} = 8.3$ Hz, $J_{\text{H}^{1/*}-\text{H}^{2/*}} = 5.9$ Hz, $J_{\text{H}^{1/*}-\text{H}3b^*} = 4.4$ Hz, $H_{1'}^*$), 4.36-4.28 (m, 1H, $H_{2'}$), 4.20–4.13 (m, 1H, $H_{6'eq}$), 4.11–3.97 (m, 2H, $H_{1''}$), 3.87–3.80 $(m, 1H, H_{3'}), 3.69-3.62 (m, 1H, H_{6'ax}), 3.57-3.43 (m, 2H, H_{4'}, H_{5'}),$ 3.30 (dd, 0.5H, $J_{\text{H3a}^{\circ}-\text{H3b}^{\circ}} = 17.0 \text{ Hz}$, $J_{\text{H3a}^{\circ}-\text{H1}^{\circ}} = 7.9 \text{ Hz}$, $H_{\text{3a}^{\circ}}$), 3.22 (dd, 0.5H, $J_{\text{H3a*-H3b*}} = 17.2 \text{ Hz}$, $J_{\text{H3a*-H1'*}} = 8.3 \text{ Hz}$, H_{3a}^*), 3.18 (dd, 0.5H, $J_{\text{H1a}^{\circ}-\text{P}^{\circ}} = 23.1 \text{ Hz}$, $J_{\text{H1a}^{\circ}-\text{H1b}^{\circ}} = 13.5 \text{ Hz}$, $H_{\text{1a}^{\circ}}$), 3.06, 3.06 (AB from ABX, 1H, $J_{AB} = 13.3$ Hz, $J_{AX} = 23.3$ Hz, $J_{BX} = 21.6$ Hz, H_{1a}^* , H_{1b}^*), 3.01 (dd, 0.5H, $J_{H1b}\circ_{-P}\circ = 21.8$ Hz, $J_{H1b}\circ_{-H1a}\circ =$ 13.5 Hz, H_{1b}°), 2.74 (dd, 0.5H, $J_{H3b}^{\circ}_{-H3a}^{\circ} = 17.0$ Hz, $J_{H3b}^{\circ}_{-H1}^{\circ} =$ 4.9 Hz, H_{3b}°), 2.59 (dd, 0.5H, $J_{H3b^*-H3a^*} = 17.2$ Hz, $J_{H3b^*-H1'^*} = 4.4$ Hz, H_{3b}^*), 2.03–1.85 (m, 2H, $H_{5''}$), 1.89, 1.87 (2 s, 3H, CH_3CO^* , CH_3CO°), 1.74–1.62 (m, 1H, $H_{2''a}$), 1.67 (s, 3H, $H_{8''}$), 1.59 (s, 3H, $H_{9''}$), 1.59–1.38 (m, 2H, $H_{3''}$, $H_{2''b}$), 1.35–1.25, 1.21–1.11 (2 m, 2H, $H_{4"}$), 0.89, 0.87 (2d, 3H, $J_{H10"-H3"} = 6.5$ Hz, $H_{10"}^{\circ}$, $H_{10"}^{*}$), 0.83, 0.82 (2 s, 9H, SitBu°, SitBu*), 0.09, 0.01 (2 s, 3H, SiMe*), 0.06, -0.01 (2 s, 3H, SiMe°); ¹³C NMR δ 197.6, 197.4 (2d, J_{C2-P} = 5.0 Hz, C₂°, C₂*), 170.7, 170.6 (CH₃CO*, CH₃CO°), 137.5, 137.4 (Cq_{ar}*, Cq_{ar}°), 135.6, 135.5 (2d, $J_{Cq-P} = 5.5 \text{ Hz}$, Cq_{ar}^{*} , Cq_{ar}°), 131.6 ($C_{7''}$), 129.2, 129.1, 129.1, 129.0, 128.9, 128.5, 128.3, 128.2, 126.4 (CH_{ar}), 124.5 ($C_{6''}$), 102.0 ($C_{7'}$), 83.4, 83.4 ($C_{4'}$ *, $C_{4'}$ °), 70.6, 70.5 ($C_{1'}$ *, $C_{1'}^{\circ}$), 70.2, 70.2 $(C_{3'}^{*}, C_{3'}^{\circ})$, 69.4 $(C_{6'})$, 68.6, 68.6 (2d, $J_{CH2Ph-P}$ = 6.0 Hz, CH₂Ph), 65.8, 65.8 ($C_{5'}^{\circ}$, $C_{5'}^{*}$), 65.7, 65.5 (2d, $J_{C1''-P} = 7.0$ Hz, $C_{1''}^*$, $C_{1''}^\circ$), 54.1, 54.0 ($C_{2'}^\circ$, $C_{2'}^*$), 44.2, 44.0 (C_3° , C_3^*), 42.8, 42.5 (2d, J_{C1-P} = 124.0 Hz, C_1), 37.4, 37.3 (2d, $J_{C2''-P}$ = 6.0 Hz, $C_{2''}$ °, $C_{2''}^*$), 37.1, 37.0 ($C_{4''}^*$, $C_{4''}^\circ$), 29.2, 29.1 ($C_{3''}^\circ$, $C_{3''}^*$), 25.9 (SitBu, $C_{8''}$), 25.5 ($C_{5''}$), 23.2, 23.1 (CH_3CO° , CH_3CO^{*}), 19.4, 19.3 ($C_{10''}$ *, $C_{10''}$ °), 18.3 (SitBu), 17.8 ($C_{9''}$), -3.9, -4.7, -4.7 (SiMe); ³¹P NMR δ 20.9 (s, 0.52P, P*), 20.6 (s, 0.48P, P*); MS (ESI): m/z = 772 [M + H]+ 100%; HRMS calcd for C₄₁H₆₂NNaO₉PSi+ 794.3829, found 794.3831.

Dibenzyl 3-(2-acetamido-4,6-O-(R)-benzylidene-3-O-tert-butyldimethylsilyl-2-deoxy- α -D-glucopyranosyl)-2-oxopropylphosphonate 11

To a solution of 17 (890 mg, 3.2 mmol, 3.25 eq.) in THF (15 mL) was added dropwise nBuLi (1.5 mL, 3.43 mmol, 3.5 eq.) at -78 °C. After 1 h stirring at -78 °C, the mixture was added to a cold solution of 3 (475 mg, 0.99 mmol, 1 eq.) in THF (3 mL). The reaction mixture was slowly warmed to r.t. overnight and quenched with saturated aqueous NH₄Cl solution. The aqueous phase was extracted with EtOAc and the combined

organic layers were dried (MgSO₄) and concentrated. Purification by flash chromatography (CH₂Cl₂/acetone, 8:2) afforded 11 (600 mg, 83%, white solid): R_f 0.47 (CH₂Cl₂/acetone = 8:2); $[\alpha]_D^{20}$ +26 $(c 1.0, \text{CH}_2\text{Cl}_2); \text{IR (cm}^{-1}) 1718 (\text{CO}), 1250 (\text{PO}); {}^{1}\text{H NMR } \delta 7.51-$ 7.47 (m, 2H, H_{ar}), 7.42–7.30 (m, 13H, H_{ar}), 6.62 (d, 1H, J_{NH-H2}) 8.7 Hz, NH), 5.50 (s, 1H, H_7), 5.12, 5.02 (AB from ABX, 2H, J_{AB} = 11.6 Hz, $J_{A-P} = 9.5$ Hz, $J_{B-P} = 10.0$ Hz, CH_2Ph), 5.01 (d, 2H, $J_{H-P} =$ 8.4 Hz, CH₂Ph), 4.78 (ddd, 1H, $J_{H1'-H3a} = 8.0$ Hz, $J_{H1'-H2'} = 5.7$ Hz, $J_{\rm HI'-H3b} = 5.1$ Hz, $H_{\rm I'}$), 4.31 (ddd, 1H, $J_{\rm H2'-H3'} = 9.8$ Hz, $J_{\rm H2'-NH} =$ 8.7 Hz, $J_{\text{H2'-H1'}} = 5.7$ Hz, $H_{\text{2'}}$), 4.15 (dd, 1H, $J_{\text{H6'eq-H6'ax}} = 10.3$ Hz, $J_{\text{H6'eq-H5'}} = 4.3 \text{ Hz}, H_{\text{6'eq}}), 3.83 \text{ (dd, 1H, } J_{\text{H3'-H2'}} = 9.8 \text{ Hz}, J_{\text{H3'-H4'}} = 8.3$ Hz, H₃, 3.65 (dd, 1H, $J_{\text{H6'ax-H6'eq}} = 10.3$ Hz, $J_{\text{H6'ax-H5'}} = 9.2$ Hz, $H_{\text{6'ax}}$), 3.51 (dd, 1H, $J_{\text{H4'-H5'}} = 9.6 \text{ Hz}$, $J_{\text{H4'-H3'}} = 8.3 \text{ Hz}$, $J_{\text{H4'}} = 8.3 \text{ Hz}$ $J_{\text{H5'-H4'}} = 9.6 \text{ Hz}, J_{\text{H5'-H6'ax}} = 9.2 \text{ Hz}, J_{\text{H5'-H6'eq}} = 4.3 \text{ Hz}, H_{\text{5'}}), 3.19 \text{ (dd,}$ 1H, $J_{\text{H3a-H3b}} = 17.1$ Hz, $J_{\text{H3a-H1}'} = 8.0$ Hz, H_{3a}), 3.10, 3.05 (AB from ABX, 2H, $J_{AB} = 13.5 \text{ Hz}$, $J_{A-P} = 23.2 \text{ Hz}$, $J_{B-P} = 21.9 \text{ Hz}$, H_{1a} , H_{1b}), 2.63 (dd, 1H, $J_{H3b-H3a}$ = 17.1 Hz, $J_{H3b-H1'}$ = 5.1 Hz, H_{3b}), 1.85 (s, 3H, CH₃CO), 0.83 (s, 9H, SitBu), 0.10, 0.01 (2 s, 6H, SiMe); ¹³C NMR δ 197.4 (d, J_{C2-P} = 4.7 Hz, C_2), 170.7 (CH₃CO), 137.4 (Cq_{ar}), 135.4, 135.3 (2d, $J_{Cq-P} = 6.0 \text{ Hz}$, Cq_{ar}), 129.2, 129.1, 129.0, 129.0, 128.9, 128.5, 128.2, 128.2, 126.4 (CH_{ar}), 102.0 (C₇), 83.4 (C₄), 70.5 (C₁), 70.1 (C₃), 69.3 (C₆), 68.6, 68.5 (2d, $J_{\text{CH2Pb-P}} = 6.5 \text{ Hz}$, CH₂Ph), 65.7 (C_5), 54.0 (C_2), 43.9 (C_3), 42.9 (d, $J_{C1-P} = 124.8 \text{ Hz}, C_1$), 25.8 (SitBu), 23.0 (<u>C</u>H₃CO), 18.2 (SitBu), -3.9, -4.7 (SiMe); ³¹P NMR δ 21.4 (s); MS (ESI): $m/z = 1469 [2M + Na]^+ 100\%$; HRMS calcd for C₃₈H₅₀NNaO₉PSi⁺ 746.2890, found 746.2898.

Benzyl 3-(2-acetamido-4,6-*O*-(*R*)-benzylidene-3-*O*-tert-butyldimethylsilyl-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 12

To a solution of 11 (600 mg, 0.83 mmol) in toluene (8 mL) was added DABCO (110 mg, 0.99 mmol, 1.2 eq.). The reaction mixture was then refluxed for 7 h and concentrated. The residue was taken up in methanol, acidified with DOWEX H⁺ (50WX8-100) ion exchange resin. Methanol was removed in vacuo to afford 12 (475 mg, 90%, white solid): $[\alpha]_D^{20}$ +18 (c 1.0, CH₂Cl₂); ¹H NMR (acetone-d₆) δ 7.54–7.32 (m, 11H, NH, H $_{\rm ar}$), 5.62 (s, 1H, H $_{7'}$), 5.18– 5.08 (m, 2H, CH₂Ph), 4.66 (ddd, 1H, $J_{H1'-H3a} = 6.8$ Hz, $J_{H1'-H3b} =$ 6.3 Hz, $J_{\text{H1'-H2'}} = 5.9$ Hz, $H_{\text{1'}}$), 4.28 (ddd, 1H, $J_{\text{H2'-H3'}} = 10.0$ Hz, $J_{\text{H2'-NH}} = 9.4 \text{ Hz}, J_{\text{H2'-H1'}} = 5.9 \text{ Hz}, H_{2'}, 4.09 \text{ (dd, 1H, } J_{\text{H6'eq-H6'ax}} =$ 9.7 Hz, $J_{\text{H6'eq-H5'}} = 4.5$ Hz, $H_{6'\text{eq}}$), 3.99 (dd, 1H, $J_{\text{H3'-H2'}} = 10.0$ Hz, $J_{\text{H3'-H4'}} = 8.9 \text{ Hz}, H_{\text{3'}}), 3.68 \text{ (dd, 1H, } J_{\text{H6'ax-H5'}} = J_{\text{H6'ax-H6'eq}} = 9.7 \text{ Hz},$ $H_{6'ax}$), 3.62 (ddd, 1H, $J_{H5'-H6'ax} = 9.7$ Hz, $J_{H5'-H4'} = 9.2$ Hz, $J_{H5'-H6'eq} =$ 4.5 Hz, H₅), 3.52 (dd, 1H, $J_{H4'-H5'} = 9.2$ Hz, $J_{H4'-H3'} = 8.9$ Hz, H₄), 3.37 (dd, 1H, $J_{\text{Hla-P}}$ = 22.8 Hz, $J_{\text{Hla-Hlb}}$ = 13.4 Hz, H_{la}), 3.35 (dd, 1H, $J_{H3a-H3b}$ = 17.1 Hz, $J_{H3a-H1'}$ = 6.8 Hz, H_{3a}), 3.24 (dd, 1H, J_{H1b-P} = 21.9 Hz, $J_{\text{H1b-H1a}} = 13.4$ Hz, H_{1b}), 3.00 (dd, 1H, $J_{\text{H3b-H3a}} = 17.1$ Hz, $J_{\text{H3b-H1'}} = 6.3 \text{ Hz}, H_{3b}$), 1.84 (s, 3H, CH₃CO), 0.82 (s, 9H, SitBu), $0.08, 0.00 (2 \text{ s}, 6\text{H}, \text{SiMe}); {}^{13}\text{C NMR (acetone-d}_6) \delta 199.9 (d, J_{C2-P} =$ $5.0 \text{ Hz}, C_2$, $170.5 \text{ (CH}_3\text{CO)}$, 139.0 (Cq_{ar}), $137.6 \text{ (d, } J_{Cq-P} = 6.0 \text{ Hz}$, Cq_{ar}), 129.6, 129.4, 129.3, 129.1, 128.7, 128.7, 127.3 (CH_{ar}), 102.5 (C_{7}) , 84.4 (C_{4}) , 72.0 (C_{1}) , 71.0 (C_{3}) , 69.7 (C_{6}) , 68.0 $(d, J_{CH2Ph-P} =$ 5.0 Hz, CH₂Ph), 65.9 (C₅), 54.6 (C₂), 43.8 (d, $J_{C1-P} = 124.0$ Hz, C₁), 43.5 (C₃), 26.3 (SitBu), 23.1 (CH₃CO), 18.8 (SitBu), -3.8, -4.5 (SiMe); ³¹P NMR (acetone-d₆) δ 18.8 (s); MS (ESI): m/z = 632 $[M - H]^-$ 100%; HRMS calcd for $C_{31}H_{43}NO_9PSi^-$ 632.2445, found 632.2460.

Dibenzyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2oxopropylphosphonate 13

To a suspension of 11 (100 mg, 0.14 mmol) in water (5 mL) was added trifluoroacetic acid (5 mL) at 0 °C. The reaction mixture was stirred at r.t. for 1 h and concentrated. Purification by flash chromatography (CH₂Cl₂/MeOH, 85:15) and lyophilization afforded 13 (70 mg, 97%, white solid): R_f 0.48 (CH₂Cl₂/MeOH = 85:15); $[\alpha]_D^{20}$ +38 (c 1.0, CH₂Cl₂); ¹H NMR (DMSO-d₆) δ 7.67 (d, 1H, $J_{NH-H2'}$ = 7.8 Hz, NH), 7.41–7.31 (m, 10H, H_{ar}), 5.07–4.98 (m, 5H, $OH_{4'}$, CH_2Ph), 4.88 (d, 1H, $J_{OH3'-H3'} = 5.3$ Hz, $OH_{3'}$), 4.44 (dd, 1H, $J_{OH6'-H6'b} = 6.0$ Hz, $J_{OH6'-H6'a} = 5.8$ Hz, $OH_{6'}$), 4.40 (ddd, 1H, $J_{\text{HI'-H3a}} = 9.1 \text{ Hz}, J_{\text{HI'-H2'}} = 5.6 \text{ Hz}, J_{\text{HI'-H3b}} = 4.6 \text{ Hz}, H_{\text{I'}}), 3.71 \text{ (ddd,}$ 1H, $J_{\text{H2'-H3'}} = 10.0$ Hz, $J_{\text{H2'-NH}} = 7.8$ Hz, $J_{\text{H2'-H1'}} = 5.6$ Hz, $H_{\text{2'}}$), 3.56 (ddd, 1H, $J_{\text{H6'a-H6'b}} = 11.6 \text{ Hz}$, $J_{\text{H6'a-OH6'}} = 5.8 \text{ Hz}$, $J_{\text{H6'a-H5'}} = 2.5 \text{ Hz}$, $H_{6'a}$), 3.50 (dd, 1H, $J_{H1a-P} = 21.8$ Hz, $J_{H1a-H1b} = 14.4$ Hz, H_{1a}), 3.47 (dd, 1H, (ddd, 1H, $J_{H6'b-H6'a}$ = 11.6 Hz, $J_{H6'b-OH6'}$ = 6.0 Hz, $J_{H6'b-H5'}$ = 5.7 Hz, H_{6'b}), 3.41 (dd, 1H, $J_{\text{H1b-P}} = 21.5$ Hz, $J_{\text{H1b-H1a}} = 14.4$ Hz, H_{1b}), 3.40 (ddd, 1H, $J_{H3'-H2'} = 10.0$ Hz, $J_{H3'-H4'} = 8.0$ Hz, $J_{H3'-OH3'} =$ 5.3 Hz, H_{3'}), 3.35 (ddd, 1H, $J_{H5'-H4'} = 8.8$ Hz, $J_{H5'-H6'b} = 5.7$ Hz, $J_{\text{H5'-H6'a}} = 2.5 \text{ Hz}, H_{\text{5'}}), 3.15 \text{ (ddd, 1H, } J_{\text{H4'-H5'}} = 8.8 \text{ Hz}, J_{\text{H4'-H3'}} =$ 8.0 Hz, $J_{\text{H4'-OH4'}} = 5.3$ Hz, $H_{\text{4'}}$), 2.90 (dd, 1H, $J_{\text{H3a-H3b}} = 16.1$ Hz, $J_{\text{H3a-HI'}} = 9.1 \text{ Hz}, H_{3a}$), 2.66 (dd, 1H, $J_{\text{H3b-H3a}} = 16.1 \text{ Hz}, J_{\text{H3b-HI'}} =$ 4.6 Hz, H_{3b}), 1.77 (s, 3H, CH₃CO); ¹³C NMR (DMSO-d₆) δ 200.5 (d, $J_{C2-P} = 6.0 \text{ Hz}$, C_2), 169.3 (CH₃CO), 136.2 (d, $J_{Cq-P} = 6.0 \text{ Hz}$, Cq_{ar}), 128.4, 128.2, 127.7 (CH_{ar}), 74.9 ($C_{5'}$), 70.6 ($C_{4'}$), 70.3 ($C_{3'}$), 69.3 ($C_{1'}$), 67.0, 66.9 (2d, $J_{CH2Ph-P} = 6.0$ Hz, CH_2Ph), 60.9 ($C_{6'}$), 52.6 (C_2), 41.8 (d, J_{C1-P} = 126.0 Hz, C_1), 41.4 (C_3), 22.6 (CH_3CO); ³¹P NMR (DMSO-d₆) δ 21.8 (s); MS (ESI): m/z = 544 [M + Na]+ 100%; HRMS calcd for C₂₅H₃₂NNaO₉P+ 544.1712, found 544.1707.

Benzyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2oxopropylphosphonate 14

To a solution of 13 (34 mg, 65 µmol) in toluene (1 mL) was added DABCO (9 mg, 78 µmol, 1.2 eq.). The reaction mixture was then refluxed for 4 h and concentrated. The residue was taken up in water, acidified with DOWEX H⁺ (50WX8-100) ion exchange resin. Water was removed in vacuo to afford 14 (25 mg, 88%, white solid): ¹H NMR (DMSO-d₆) δ 7.73 (d, 1H, $J_{NH-H2'}$ = 8.0 Hz, NH), 7.42–7.28 (m, 5H, H_{ar}), 4.94 (d, 2H, $J_{CH2Ph-P} = 7.5$ Hz, CH_2Ph), 4.66 (ddd, 1H, $J_{\text{HI'-H3a}} = 8.3$ Hz, $J_{\text{HI'-H3b}} = J_{\text{HI'-H2'}} = 5.2$ Hz, $H_{\text{I'}}$), 3.72 (ddd, 1H, $J_{\text{H2'-H3'}} = 10.3 \text{ Hz}$, $J_{\text{H2'-NH}} = 8.0 \text{ Hz}$, $J_{\text{H2'-H1'}} = 5.2 \text{ Hz}$, $H_{2'}$), 3.56 (dd, 1H, $J_{H6'a-H6'b}$ = 11.5 Hz, $J_{H6'a-H5'}$ = 2.2 Hz, $H_{6'a}$), 3.46 (dd, 1H, $J_{H6'b-H6'a}$ = 11.5 Hz, $J_{H6'b-H5'}$ = 5.7 Hz, $H_{6'b}$), 3.41 (dd, 1H, 8.8 Hz, $J_{\text{H5'-H6'b}} = 5.7$ Hz, $J_{\text{H5'-H6'a}} = 2.2$ Hz, $H_{\text{5'}}$), 3.17, 3.12 (AB from ABX, 2H, $J_{AB} = 13.3$ Hz, $J_{A-P} = 21.7$ Hz, $J_{B-P} = 22.0$ Hz, H_1), 3.12 (dd, 1H, $J_{\text{H4'-H5'}}$ = 8.8 Hz, $J_{\text{H4'-H3'}}$ = 8.1 Hz, $H_{\text{4'}}$), 2.88 (dd, 1H, $J_{\text{H3a-H3b}} = 16.2 \text{ Hz}, J_{\text{H3a-H1}'} = 8.3 \text{ Hz}, H_{\text{3a}}), 2.75 \text{ (dd, 1H, } J_{\text{H3b-H3a}} =$ 16.2 Hz, $J_{\text{H3b-H1}'}$ = 5.2 Hz, H_{3b}), 1.76 (s, 3H, CH₃CO); ¹³C NMR (DMSO-d₆) δ 201.5 (d, J_{C2-P} = 5.5 Hz, C₂), 169.3 (CH₃CO), 137.2 (d, $J_{Cq-P} = 7.0 \text{ Hz}$, Cq_{ar}), 128.3, 127.8, 127.4 (CH_{ar}), 74.8 (C₅), 70.8 ($C_{4'}$), 70.4 ($C_{3'}$), 69.5 ($C_{1'}$), 66.1 (d, $J_{CH2Ph-P} = 5.0$ Hz, CH_2Ph), $61.0 (C_{6})$, $52.7 (C_{2})$, $43.9 (d, J_{C1-P} = 120.0 Hz, C_{1})$, $41.1 (C_{3})$, 22.6(CH₃CO); ³¹P NMR (DMSO-d₆) δ 16.2 (s); MS (ESI): m/z = 430 $[M - H]^- 100\%$.

Benzyl uridin-5'-yl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 15

To a suspension of 9 (146 mg, 0.15 mmol) in water (5 mL) was added trifluoroacetic acid (5 mL) at 0 °C. The reaction mixture was stirred at r.t. for 2 h and concentrated. Purification by flash chromatography (CH₂Cl₂/MeOH, 3:1) and lyophilization afforded 15 (91 mg, 94%, white solid) as a mixture of epimers (d.r. = 2/1): $R_f 0.10 (CH_2Cl_2/MeOH = 3:1)$; ¹H NMR (DMSO d_6) δ 11.33 (s, 1H, H_{3u}), 7.70 (d, 1H, $J_{NH-H2'}$ = 7.7 Hz, NH), 7.62, 7.58 (2d, 1H, $J_{H6u-H5u}$ = 8.0 Hz, H_{6u}° , H_{6u}^{*}), 7.42–7.31 (m, 5H, H_{ar}), 5.78 (d, 1H, $J_{\text{H1''-H2''}} = 5.5 \text{ Hz}$, $H_{\text{1''}}$), 5.61, 5.57 (2d, 1H, $J_{\text{H5u-H6u}} =$ 8.0 Hz, H_{5u}° , H_{5u}^{*}), 5.48, 5.47 (2d, 1H, $J_{OH2''-H2''} = 5.5$ Hz, $OH_{2''}^{\circ}$, $OH_{2''}^*$), 5.29 (d, 1H, $J_{OH3''-H3''} = 4.8$ Hz, $OH_{3''}$), 5.10–5.01 (m, 3H, $CH_2Ph, OH_{4'}), 4.91 (d, 1H, J_{OH3'-H3'} = 5.1 Hz, OH_{3'}), 4.47-4.40 (m,$ 1H, OH₆), 4.40 (ddd, 1H, $J_{H1'-H3a} = 9.0$ Hz, $J_{H1'-H2'} = 5.5$ Hz, $J_{H1'-H3b} =$ 4.2 Hz, $H_{1'}$), 4.21–4.10 (m, 2H, $H_{5''}$), 4.04 (ddd, 1H, $J_{H2''-H1''} = 5.5$ Hz, $J_{\text{H2''-OH2''}} = 5.5$ Hz, $J_{\text{H2''-H3''}} = 5.5$ Hz, $H_{\text{2''}}$), 4.02–3.91 (m, 2H, $H_{3''}$, $H_{4''}$), 3.71 (ddd, 1H, $J_{H2'-H3'} = 9.9$ Hz, $J_{H2'-NH} = 7.7$ Hz, $J_{H2'-H1'} =$ 5.5 Hz, H₂), 3.60–3.28 (m, 6H, H₁, H₃, H₅, H₆), 3.14 (ddd, 1H, $J_{\text{H4'-H3'}} = 8.4 \text{ Hz}, J_{\text{H4'-H5'}} = 8.4 \text{ Hz}, J_{\text{H4'-OH4'}} = 5.4 \text{ Hz}, H_{\text{4'}}), 2.89 \text{ (dd,}$ 1H, $J_{\text{H3a-H3b}} = 16.0$ Hz, $J_{\text{H3a-H1'}} = 9.0$ Hz, H_{3a}), 2.65, 2.65 (2dd, 1H, $J_{\text{H3b-H3a}} = 16.0 \text{ Hz}, J_{\text{H3b-H1}'} = 4.2 \text{ Hz}, H_{3b}^{\circ}, H_{3b}^{*}), 1.78, 1.78 (2 \text{ s}, 3 \text{H},$ CH₃CO*, CH₃CO°); 13 C NMR (DMSO-d₆) δ 200.7, 200.7 (2d, $J_{C2-P} = 6.0 \text{ Hz}, C_2), 169.4 (CH_3\underline{C}O), 163.0 (C_{4u}), 150.6 (C_{2u}), 140.7,$ 140.6 (C_{6u}° , C_{6u}^{*}), 136.2 (d, $J_{Cq-P} = 6.5$ Hz, Cq_{ar}), 128.5, 128.3, 127.7 (CH_{ar}), 102.0 (C_{5u}), 88.2, 88.2 (C_{1"}°, C_{1"}*), 82.1 (d, $J_{C4"-P}$ = 6.0 Hz, $C_{4''}$), 74.9 ($C_{5'}$), 72.6 ($C_{2''}$), 70.6 ($C_{4'}$), 70.2 ($C_{3'}$), 69.5 ($C_{3''}$), $69.3 (C_{1'}), 67.1 (d, J_{CH2Ph-P} = 5.5 Hz, CH_2Ph), 65.5, 65.3 (2d, J_{CS''-P} =$ 5.5 Hz, $C_{5''}$, $C_{5''}$, 60.9 ($C_{6'}$), 52.7 ($C_{2'}$), 41.5 (C_{3}), 41.4, 41.4 (2d, $J_{\text{Cl-P}} = 127.0 \text{ Hz}, C_1^{\circ}, C_1^{*}, 22.6 (\underline{\text{CH}}_3\text{CO}); ^{31}\text{P NMR (DMSO-d}_6)$ δ 22.2 (s, 0.66P, P*), 22.0 (s, 0.33P, P°); MS (ESI): m/z = 1337 $[2M + Na]^+$ 100%; HRMS calcd for $C_{27}H_{36}N_3NaO_{14}P^+$ 680.1833, found 680.1832.

Uridin-5'-vl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2oxopropylphosphonate 16

A suspension of **15** (22 mg, 33 µmol) and Pd/C 10% (22 mg) in methanol (2 mL) was purged from air and filled with H₂. The reaction was stirred at r.t. for 3 h then diluted with methanol, filtered through a Celite pad and concentrated. Lyophilization afforded **16** (16 mg, 84%, white solid): $[\alpha]_D^{20}$ +34 (c 1.0, H₂O); ¹H NMR (DMSO-d₆) δ 11.27 (s, 1H, H_{3u}), 8.22 (d, 1H, $J_{NH-H2'} = 6.0$ Hz, NH), 7.85 (d, 1H, $J_{H6u-H5u}$ = 8.0 Hz, H_{6u}), 5.80 (d, 1H, $J_{H1''-H2''}$ = 5.5 Hz, $H_{1''}$), 5.62 (2d, 1H, $J_{H5u-H6u} = 8.0$ Hz, H_{5u}), 4.40–4.32 (m, 1H, $H_{1'}$), 4.11–4.05 (m, 1H, $H_{2''}$), 4.05–3.99 (m, 1H, $H_{3''}$), 3.97– $3.91 \text{ (m, 1H, H}_{4''}), 3.91-3.83 \text{ (m, 2H, H}_{5''}), 3.77-3.70 \text{ (m, 1H, H}_{2'}),$ 3.61-3.54 (m, 1H, $H_{6'a}$), 3.47-3.25 (m, 3H, $H_{3'}$, $H_{5'}$, $H_{6'b}$), 3.09-2.96(m, 2H, H₄, H_{3a}), 2.90 (dd, 1H, $J_{H1a-P} = 21.5$ Hz, $J_{H1a-H1b} = 11.5$ Hz, H_{1a}), 2.81–2.67 (m, 2H, H_{1b} , H_{3b}), 1.76 (s, 3H, CH_3CO); ¹³C NMR (DMSO-d₆) δ 203.7 (d, J_{C2-P} = 5.0 Hz, C_2), 169.6 (CH₃CO), 163.1 (C_{4u}), 150.8 (C_{2u}), 140.9 (C_{6u}), 101.9 (C_{5u}), 87.4 ($C_{1''}$), 83.5 $(d, J_{C4''-P} = 6.5 \text{ Hz}, C_{4''}), 74.8 (C_{5'}), 73.2 (C_{2''}), 71.3 (C_{4'}), 70.7 (C_{3'}),$ 70.3 ($C_{3''}$), 69.5 ($C_{1'}$), 63.5 (d, $J_{C5''-P} = 5.0 \text{ Hz}$, $C_{5''}$), 61.4 ($C_{6'}$), 52.7 (C_2) , 45.9 (d, $J_{C1-P} = 107.0 \text{ Hz}$, C_1), 41.3 (C_3), 22.7 (CH₃CO); ³¹P NMR (DMSO-d₆) δ 8.9 (s); MS (ESI): $m/z = 566 [M - H]^- 100\%$; HRMS calcd for $C_{20}H_{29}N_3O_{14}P^-$ 566.1387, found 566.1384.

Benzyl 5-(3-N-tert-butyloxycarbonyl-uracil-1-yl)pentyl 3-(2-acetamido-4,6-O-(R)-benzylidene-3-O-tert-butyldimethylsilyl-2-deoxy- α -D-glucopyranosyl)-2-oxopropyl-phosphonate 18

At 0 °C, DIAD (47 μL, 0.24 mmol, 1.5 eq.) was added dropwise to a solution of 12 (100 mg, 0.16 mmol, 1 eq.), 17 (47 mg, 0.16 mmol, 1 eq.) and PPh₃ (62 mg, 0.24 mmol, 1.5 eq.) in THF (2.5 mL). The reaction mixture was stirred at r.t. for 16 h then concentrated. Purification by preparative chromatography (CHCl₃/MeOH, 95:5) afforded 18 (39 mg, 27%, white solid) as a mixture of epimers (d.r. = 65/35): R_f 0.35 (cyclohexane/acetone = 1:1); ¹H NMR δ 7.49–7.32 (m, 10H, H_{ar}), 7.15, 6.99 (2d, 1H, $J_{\text{H6u-H5u}} = 8.0 \text{ Hz}, H_{6u}^*, H_{6u}^\circ), 6.57 \text{ (2d, 1H, } J_{\text{NH-H2'}} = 8.5 \text{ Hz}, \text{NH*},$ NH°), 5.69, 5.64 (2d, 1H, $J_{H5u-H6u} = 8.0$ Hz, H_{5u}^* , H_{5u}°), 5.50, 5.49 (s, 1H, H₇), 5.16–5.02 (m, 2H, CH₂Ph), 4.79 (ddd, 0.35H, $J_{\text{HI}^{\circ}-\text{H3a}^{\circ}} = 7.6 \text{ Hz}, J_{\text{HI}^{\circ}-\text{H3b}^{\circ}} = 6.2 \text{ Hz}, J_{\text{HI}^{\circ}-\text{H2}^{\circ}} = 4.8 \text{ Hz}, H_{\text{I}^{\circ}}),$ 4.76 (ddd, 0.65H, $J_{\text{HI}'^*-\text{H3a}^*} = 7.3$ Hz, $J_{\text{HI}'^*-\text{H2}'^*} = 5.7$ Hz, $J_{\text{HI}'^*-\text{H3b}^*} =$ 5.4 Hz, $H_{1'}^*$), 4.32–4.24 (m, 1H, $H_{2'}$), 4.17–4.12 (m, 1H, $H_{6'eq}$), 4.08-3.90 (m, 2H, $H_{1''}$), 3.85-3.78 (m, 1H, $H_{3'}$), 3.73-3.61 (m, 3H, $H_{5''}$, $H_{6'ax}$), 3.54–3.45 (m, 2H, $H_{4'}$, $H_{5'}$), 3.17 (dd, 0.35H, $J_{H3a} \circ_{-H3b} \circ =$ 16.8 Hz, $J_{\text{H3a}}\circ_{-\text{HI'}}\circ = 7.6$ Hz, $H_{\text{3a}}\circ)$, 3.17 (dd, 0.35H, $J_{\text{H1a}}\circ_{-\text{P}}\circ = 22.6$ Hz, $J_{\text{H1a}} \circ_{\text{-H1b}} \circ = 14.6 \text{ Hz}$, $H_{\text{1a}} \circ$), 3.14 (dd, 0.65H, $J_{\text{H3a*-H3b*}} = 17.2$ Hz, $J_{\text{H3a*-H1'*}} = 7.3$ Hz, H_{3a}^*), 3.07 (dd, 0.65H, $J_{\text{H1b}}^{\circ}_{\text{-P}}^{\circ} = 21.7$ Hz, $J_{\text{H1b}} \circ_{-\text{H1a}} \circ = 14.6 \text{ Hz}, H_{\text{1b}} \circ), 3.06, 3.02 \text{ (AB from ABX, 1.3H, } J_{\text{AB}} =$ 13.5 Hz, $J_{AX} = 23.0$ Hz, $J_{BX} = 22.3$ Hz, H_{1a}^* , H_{1b}^*), 2.81 (dd, 0.35H, $J_{\text{H3b}}\circ_{-\text{H3a}}\circ = 16.8$ Hz, $J_{\text{H3b}}\circ_{-\text{H1}}\circ = 6.2$ Hz, $H_{\text{3b}}\circ$), 2.66 (dd, 0.65H, $J_{\text{H3b*-H3a*}} = 17.2 \text{ Hz}$, $J_{\text{H3b*-H1'*}} = 5.4 \text{ Hz}$, H_{3b} *), 1.89 (2 s, 3H, CH₃CO*, CH₃CO°), 1.72–1.58 (m, 4H, H_{2"}, H_{4"}), 1.59 (s, 9H, CO₂tBu), 1.41–1.30 (m, 2H, H_{3"}), 0.82 (s, 9H, SitBu), 0.08, 0.00 (2 s, 1.3H, SiMe*), 0.06, -0.01 (2 s, 0.7H, SiMe°); 13 C NMR δ 197.9, 197.8 (2d, $J_{C2-P} = 5.5 \text{ Hz}$, C_2° , C_2^{*}), 170.7, 170.6 (CH₃CO°, CH_3CO^*), 160.9, 160.9 (C_{4u}^* , C_{4u}°), 149.1 (C_{2u}), 148.0 (CO_{Boc}), 143.9, 143.7 (C_{6u}^* , C_{6u}°), 137.4 (Cq_{ar}), 135.5 (d, $J_{Cq-P} = 5.0$ Hz, Cq_{ar}), 129.2, 129.1, 129.0, 128.9, 128.5, 128.2, 126.4, 126.3 (CH_{ar}), 102.0, 101.9, 101.9 (C_{5u}, C₇), 86.9, 86.9 (CMe₃°, CMe₃*), 83.3 $(C_{4'})$, 70.7, 70.5 $(C_{1'}^{\circ}, C_{1'}^{*})$, 70.2 $(C_{3'})$, 69.3 $(C_{6'})$, 68.7 $(d, J_{CH2Pb-P} =$ 6.5 Hz, CH₂Ph), 66.4, 66.2 (2d, $J_{C1''-P} = 7.0$ Hz, $C_{1''}^*$, $C_{1''}^\circ$), 65.6, $65.5\,({C_{5'}}^*,{C_{5'}}^\circ),\,54.1,\,54.0\,({C_{2'}}^\circ,{C_{2'}}^*),\,49.2,\,49.1\,({C_{5''}}^*,{C_{5''}}^\circ),\,43.8,$ 43.3 (C_3^* , C_3°), 42.5, 42.4 (2d, $J_{Cl-P} = 125.0 \text{ Hz}$, C_1^* , C_1°), 29.7, 29.6 (2d, $J_{C2''-P} = 6.5 \text{ Hz}$, $C_{2''}^{\circ}$, $C_{2''}^{*}$), 28.4, 28.3 ($C_{4''}^{\circ}$, $C_{4''}^{*}$), 27.6 (CMe_3) , 25.8 (SitBu), 23.1 (CH_3CO) , 22.5, 22.5 $(C_{3''}^{\circ}, C_{3''}^{**})$, 18.2 (SitBu), -3.9, -4.7 (SiMe); ³¹P NMR δ 21.1 (s, 0.35P, P°), 20.9 (s, 0.65P, P*); MS (ESI): $m/z = 936 [M + Na]^+ 100\%$; HRMS calcd for C₄₅H₆₃N₃O₁₃PSi⁻ 912.3868, found 912.3856.

Benzyl hexadecanyl 3-(2-acetamido-4,6-O-(R)-benzylidene-3-O-tert-butyldimethylsilyl-2-deoxy- α -D-glucopyranosyl)-2-oxopropylphosphonate 19

At 0 °C, DIAD (54 μ L, 0.27 mmol, 1.5 eq.) was added dropwise to a solution of **12** (114 mg, 0.18 mmol, 1 eq.), hexadecan-1-ol (44 mg, 0.18 mmol, 1 eq.) and PPh₃ (71 mg, 0.27 mmol, 1.5 eq.) in THF (3 mL). The reaction mixture was stirred at r.t. for 16 h then concentrated. Two successive purifications by flash chromatography (cyclohexane/acetone, 2:1 and CH₂Cl₂/acetone, 9:1) afforded **19** (66 mg, 42%, colorless oil) as a mixture of epimers (d.r. = 55/45): $R_{\rm f}$ 0.35, 0.25 (CH₂Cl₂/acetone = 9:1); ¹H NMR δ 7.52–7.31 (m, 10H, H_{ar}), 6.68, 6.53 (2d, 1H, $J_{\rm NH-H2'}$ = 8.6 Hz, NH*,

NH°), 5.51, 5.50 (2 s, 1H, H_7 *, H_7 °), 5.19–5.02 (m, 2H, CH₂Ph), $4.81 \text{ (ddd, } 0.45\text{H, } J_{\text{HI}'-\text{H3a}} = 7.6 \text{ Hz, } J_{\text{HI}'-\text{H2'}} = 5.7 \text{ Hz, } J_{\text{HI}'-\text{H3b}} = 5.0 \text{ Hz,}$ $H_{1'}^{\circ}$), 4.77 (ddd, 0.55H, $J_{H1'-H3a} = 8.3$ Hz, $J_{H1'-H2'} = 5.5$ Hz, $J_{H1'-H3b} =$ 4.4 Hz, $H_{1'}^*$), 4.36–4.28 (m, 1H, $H_{2'}$), 4.20–4.13 (m, 1H, $H_{6'eq}$), 4.09-3.94 (m, 2H, $H_{1''}$), 3.84 (dd, 1H, J = 9.6 Hz, J = 8.1 Hz, $H_{3'}$), $H_{4'}, H_{5'}), 3.29 \text{ (dd, } 0.45H, J_{H3a-H3b} = 17.0 \text{ Hz}, J_{H3a-H1'} = 7.6 \text{ Hz}, H_{3a}^{\circ}),$ 3.22 (dd, 0.55H, $J_{H_{3a-H_{3b}}} = 17.2$ Hz, $J_{H_{3a-H_{1}'}} = 8.3$ Hz, H_{3a}^*), 3.18 (dd, 0.45H, $J_{\text{Hla-P}} = 23.0 \text{ Hz}$, $J_{\text{Hla-Hlb}} = 13.4 \text{ Hz}$, H_{la}°), 3.06, 3.04 (AB from ABX, 1.1H, $J_{AB} = 12.7$ Hz, $J_{A-P} = 22.3$ Hz, $J_{B-P} = 22.7$ Hz, H_{1a}^* , H_{1b}^*), 3.01 (dd, 0.45H, J_{H1b-P} = 22.0 Hz, $J_{H1b-H1a}$ = 13.4 Hz, H_{1b}°), 2.75 (dd, 0.45H, $J_{H3b-H3a} = 17.0 \text{ Hz}$, $J_{H3b-H1'} = 5.0 \text{ Hz}$, H_{3b}°), 2.59 (dd, 0.55H, $J_{\text{H3b-H3a}} = 17.2$ Hz, $J_{\text{H3b-H1'}} = 4.4$ Hz, H_{3b}^*), 1.89, 1.87 (2 s, 3H, CH₃CO*, CH₃CO°), 1.69–1.57 (m, 2H, H_{2"}), 1.35– 1.22 (m, 26H, $H_{3''}$ to $H_{15''}$), 0.88 (t, 3H, $J_{H16''-H15''} = 6.7$ Hz, $H_{16''}$), 0.83, 0.82 (2 s, 9H, SitBu*, SitBu°), 0.09, 0.01 (2 s, 3.3H, SiMe*), 0.06, -0.01 (2 s, 2.7H, SiMe°); ¹³C NMR δ 197.7, 197.4 (2d, J_{C2-P} = 5.0 Hz, C₂°, C₂*), 170.7, 170.6 (CH₃CO*, CH₃CO°), 137.4, 137.4 $(Cq_{ar}^*, Cq_{ar}^\circ)$, 135.6, 135.5 (2d, $J_{Cq-P} = 6.0 \text{ Hz}$, $Cq_{ar}^*, Cq_{ar}^\circ)$, 129.1, 129.1, 129.0, 128.9, 128.9, 128.5, 128.2, 128.1, 126.4 (CH_{ar}), 102.0 (C_{7}) , 83.4, 83.4 $(C_{4}^{*}, C_{4}^{\circ})$, 70.6, 70.5 $(C_{1}^{\circ}, C_{1}^{*})$, 70.2 (C_{3}) , 69.4 (C_{6}) , 68.6 (d, $J_{CH2Ph-P} = 6.5$ Hz, $CH_{2}Ph$), 67.3, 67.1 (2d, $J_{C1''-P} =$ 6.5 Hz, $C_{1''}^*$, $C_{1''}^\circ$), 65.7, 65.7 ($C_{5'}^*$, $C_{5'}^\circ$), 54.0, 53.9 ($C_{2'}^\circ$, $C_{2'}^*$), 44.1, 43.9 (C_3^* , C_3°), 42.7, 42.5 (2d, $J_{Cl-P} = 124.0$ Hz, C_1^* , C_1°), 32.0, 29.8, 29.8, 29.7, 29.6, 29.5, 29.2, 25.6, 25.5, 23.1, 23.1 ($C_{3''}$ to $C_{15''}$), 30.5, 30.3 (2d, $J_{C2''-P} = 6.0 \text{ Hz}$, $C_{2''}^{\circ}$, $C_{2''}^{*}$), 25.8 (SitBu), 22.8 (<u>C</u>H₃CO), 18.2 (SitBu), 14.2 (C_{16"}), -3.9, -4.7, -4.7 (SiMe); ³¹P NMR δ 20.9 (s, 0.55P, P*), 20.6 (s, 0.45P, P°); MS (ESI): m/z =880 [M + Na] $^+$ 100%; HRMS calcd for $C_{47}H_{75}NO_9PSi^-$ 856.4949, found 856.4951.

Benzyl 5-uracil-1-yl-pentyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 20

To a suspension of 18 (37 mg, 40 µmol) in water (2 mL) was added trifluoroacetic acid (2 mL) at 0 °C. The reaction mixture was stirred at r.t. for 3 h and concentrated. Purification by flash chromatography (CH₂Cl₂/MeOH, 8:2) afforded **20** (23 mg, 93%, white solid) as a mixture of epimers: R_f 0.50 (CH₂Cl₂/MeOH = 8:2); ¹H NMR (DMSO-d₆) δ 11.19 (s, 1H, H_{3u}), 7.72–7.67 (m, 1H, NH), 7.65, 7.62 (2d, 1H, $J_{H6u-H5u} = 7.8$ Hz, H_{6u}° , H_{6u}^{*}), 7.43– 7.30 (m, 5H, H_{ar}), 5.54, 5.53 (2d, 1H, $J_{H6u-H5u} = 7.8$ Hz, H_{5u}° , H_{5u}^*), 5.10–4.98 (m, 2H, CH₂Ph), 4.39 (ddd, 1H, $J_{H1'-H3a} = 9.2$ Hz, $J_{\text{H1'-H2'}} = 5.3 \text{ Hz}, J_{\text{H1'-H3b}} = 4.5 \text{ Hz}, H_{\text{1'}}), 3.99 - 3.88 \text{ (m, 2H, H_{\text{1''}})}, 3.71$ (ddd, 1H, $J_{\text{H2'-H3'}} = 9.4$ Hz, $J_{\text{H2'-NH}} = 7.3$ Hz, $J_{\text{H2'-H1'}} = 5.3$ Hz, J_{H2}), 3.62 (t, 2H, $J_{\text{H5"-H4"}} = 7.2$ Hz, $H_{\text{5"}}$), 3.56 (ddd, 1H, $J_{\text{H6'a-H6'b}} = 11.6$ Hz, J = 5.8 Hz, J = 1.9 Hz, $H_{6'a}$), 3.47 (ddd, 1H, $J_{H6'b-H6'a} = 11.6$ Hz, J = 5.8 Hz, J = 5.5 Hz H_{6'b}), 3.43–3.26 (m, 4H, H₁, H_{3'}, H_{5'}), 3.18-3.10 (m, 1H, H₄), 2.90 (dd, 1H, $J_{H3a-H3b} = 16.2$ Hz, $J_{H3a-H1'} =$ 9.2 Hz, H_{3a}), 2.64 (dd, 1H, $J_{H3b-H3a} = 16.2$ Hz, $J_{H3b-H1'} = 4.7$ Hz, H_{3b}), 1.78, 1.76 (2 s, 3H, CH₃CO*, CH₃CO°), 1.62–1.50 (m, 4H, $H_{2''},\,H_{4''}),\,1.33-1.21$ (m, 2H, $H_{3''});\,^{13}C$ NMR (DMSO-d₆) δ 200.6 (C_2) , 169.3 (CH_3CO) , 163.7 (C_{4u}) , 150.9 (C_{2u}) , 145.6 (C_{6u}) , 136.3 (d, d) $J_{\text{Cq-P}} = 6.3 \text{ Hz}, \text{Cq}_{\text{ar}}$, 128.4, 128.2, 127.7 (CH_{ar}), 100.8 (C_{5u}), 74.9 $(C_{3'})$, 70.6 $(C_{4'})$, 70.3 $(C_{5'})$, 69.4 $(C_{1'})$, 66.9 $(2d, J_{CH2Ph-P} = 5.8 \text{ Hz},$ CH_2Ph^* , CH_2Ph°), 65.6 (2d, $J_{C_1"-P} = 5.5 \text{ Hz}$, $C_{1"}^*$, $C_{1"}^\circ$), 60.9 ($C_{6'}$), 52.6 ($C_{2'}$), 47.2 ($C_{5''}$), 42.3 (C_{1}), 41.3 (C_{3}), 29.3 (d, $J_{C2''-P} = 5.5$ Hz, $C_{2''}$), 27.9 ($C_{4''}$), 22.6 (<u>C</u>H₃CO), 21.8 ($C_{3''}$); ³¹P NMR (DMSO-d₆) δ 21.16 (m); MS (ESI): m/z = 634 [M + Na]⁺ 100%; HRMS calcd for $C_{27}H_{37}N_3O_{11}P^-$ 610.2166, found 610.2194.

Benzyl (*R*)-7-hydroxy-3,7-dimethyloctyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 21

To a suspension of 10 (205 mg, 0.27 mmol) in water (5 mL) was added trifluoroacetic acid (5 mL) at 0 °C. The reaction mixture was stirred at r.t. for 2 h and concentrated. Purification by flash chromatography (CH₂Cl₂/MeOH, 85:15) afforded 21 (147 mg, 94%, white solid) as a mixture of epimers (d.r. = 52/48): R_f 0.36 $(CH_2Cl_2/MeOH = 85:15)$; ¹H NMR (DMSO-d₆) δ 7.69 (d, 1H, $J_{NH-H2'} = 7.7 \text{ Hz}, NH), 7.41-7.32 \text{ (m, 5H, H}_{ar}), 5.07-4.99 \text{ (m, 3H, }$ $CH_2Ph, OH_{4'}), 4.42 (ddd, 1H, J = 6.0 Hz, J_{OH6'-H6'a} = 5.8 Hz, J =$ 3.1 Hz, OH₆), 4.39 (ddd, 1H, $J_{\text{HI}'-\text{H3a}} = 9.1$ Hz, $J_{\text{HI}'-\text{H2}'} = 5.6$ Hz, $J_{\text{HI}'-\text{H3b}} = 4.6 \text{ Hz}, H_{\text{I}'}, 4.01-3.90 \text{ (m, 2H, H}_{\text{I}''}), 3.71 \text{ (ddd, 1H, }$ $J_{\text{H2'-H3'}} = 9.9 \text{ Hz}, J_{\text{H2'-NH}} = 7.7 \text{ Hz}, J_{\text{H2'-H1'}} = 5.6 \text{ Hz}, H_{2'}), 3.56 \text{ (ddd,}$ 1H, $J_{\text{H6'a-H6'b}} = 11.6 \text{ Hz}$, $J_{\text{H6'a-OH6'}} = 5.8 \text{ Hz}$, $J_{\text{H6'a-H5'}} = 2.5 \text{ Hz}$, $H_{6'a}$), 3.50-3.44 (m, 1H, H₆b), 3.43-3.27 (m, 4H, H₁, H₂, H₅), 3.17-3.11(m, 1H, H₄), 2.90, 2.89 (2dd, 1H, $J_{H3a-H3b} = 16.1$ Hz, $J_{H3a-H1'} = 9.1$ Hz, H_{3a}), 2.65, 2.64 (2dd, 1H, $J_{\text{H3b-H3a}} = 16.1$ Hz, $J_{\text{H3b-H1'}} = 4.6$ Hz, H_{3b}), 1.78, 1.78 (2 s, 3H, CH₃CO*, CH₃CO°), 1.62–1.54 (m, 1H, $H_{2''a}$), 1.52–1.43 (m, 1H, $H_{3''}$), 1.39–1.16 (m, 6H, $H_{2''b}$, $H_{4''a}$, $H_{5''}$, $H_{6"}$), 1.10–1.01 (m, 1H, $H_{4"b}$), 1.05, 1.05 (2 s, 6H, $H_{8"}$, $H_{9"}$), 0.82, 0.83 (2d, 3H, $J_{\text{HI0''}-\text{H3''}} = 6.6 \text{ Hz}$, $H_{\text{10''}}$); ¹³C NMR (DMSO-d₆) δ 200.6 (d, $J_{C2-P} = 5.0 \text{ Hz}$, C_2), 169.2 (CH₃CO), 136.4 (d, $J_{Cq-P} =$ $6.0 \text{ Hz}, \text{Cq}_{\text{ar}}$), 128.4, 128.2, 127.7 (CH $_{\text{ar}}$), 74.8 (C $_{\text{5'}}$), 70.6 (C $_{\text{4'}}$), 70.3 $(C_{3'})$, 69.4 $(C_{1'})$, 68.7 $(C_{7''})$, 66.9, 66.8 $(2d, J_{CH2Ph-P} = 6.0 \text{ Hz}, CH_2Ph)$, 64.2 (d, $J_{C1''-P}$ = 6.5 Hz, $C_{1''}$), 60.9 ($C_{6'}$), 52.6 ($C_{2'}$), 43.8 ($C_{6''}$), 41.8 (d, $J_{C1-P} = 120.0 \text{ Hz}$, C_1), 41.3 (C_3), 37.0, 36.9 ($C_{4''}^*$, $C_{4''}^\circ$), 36.8, 36.7 (2d, $J_{C2''-P} = 5.0$ Hz, $C_{2''}^*$, $C_{2''}^\circ$) 29.3, 29.2 ($C_{8''}$, $C_{9''}$), 28.7 $(C_{3''})$, 22.6 (CH₃CO), 21.1 (C_{5''}), 19.1, 19.1 (C_{10''}°, C_{10''}*); ³¹P NMR (DMSO-d₆) δ 21.2 (s, 0.48P, P°), 21.1 (s, 0.52P, P*); MS (ESI): $m/z = 610 \text{ [M + Na]}^+ 100\%; \text{ HRMS calcd for } C_{28}H_{46}NNaO_{10}P^+$ 610.2757, found 610.2779.

Benzyl (R)-citronellyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 22

To a solution of 10 (80 mg, 0.10 mmol) in THF (2 mL) was added water (2 mL) and trifluoroacetic acid (2 mL) at 0 °C. The reaction mixture was stirred at 0 °C for 2 h and at r.t. for 4 h. The mixture was then neutralized with aqueous NH₄OH solution. The aqueous phase was extracted with Et2O and the combined organic layers were dried (MgSO₄) and concentrated. Purification by flash chromatography (CH₂Cl₂/MeOH, 85:15) and lyophilization afforded 22 (50 mg, 84%, white solid): $R_{\rm f}$ 0.40 $(CH_2Cl_2/MeOH = 85:15)$; ¹H NMR (CD_3OD) δ 7.38–7.20 (m, 5H, H_{ar}), 5.10–4.95 (m, 3H, $H_{6''}$, CH_2Ph), 4.59–4.49 (m, 1H, $H_{1'}$), 4.03–3.88 (m, 2H, $H_{1''}$), 3.86 (dd, 1H, $J_{H2'-H3'} = 9.5$ Hz, $J_{H2'-H1'} =$ 5.4 Hz, H₂), 3.68–3.58 (m, 2H, H₆), 3.49 (dd, 1H, $J_{H3'-H2'} = 9.5$ Hz, $J_{\text{H3'-H4'}} = 8.3 \text{ Hz}$, $H_{\text{3'}}$), 3.45–3.38 (m, 1H, $H_{\text{5'}}$), 3.30–3.18 (m, 3H, $H_{4'}$, H_1), 3.00–2.90 (m, 0.6H, H_{3a}), 2.78–2.72 (m, 0.4H, H_{3b}), 1.95-1.78 (m, 2H, $H_{5"}$), 1.85 (s, 3H, CH_3CO), 1.62-1.50 (m, 1H, $H_{2a''}$), 1.56 (s, 3H, $H_{8''}$), 1.50 (s, 3H, $H_{9''}$), 1.47–1.38 (m, 1H, $H_{3''}$), 1.37-1.27 (m, 1H, $H_{2b''}$), 1.26-1.15 (m, 1H, $H_{4b''}$), 1.09-0.98 (m, 1H, $H_{4a''}$), 0.78, 0.77 (2d, $J_{H10''-H3''}$ = 6.4 Hz, 3H, $H_{10''}$); ¹³C NMR (CD₃OD) δ 201.8 (C₂), 174.0 (CH₃CO), 137.5 (d, $J_{Cq-P} = 5.5$ Hz, Cq_{ar}), 132.3 ($C_{7''}$), 129.9, 129.4 (CH_{ar}), 125.8 ($C_{6''}$), 76.5 ($C_{5'}$), 72.1,

72.1 ($C_{3'}$, $C_{4'}$), 70.2 ($C_{1'}$), 69.7 (d, $J_{CH2Ph-P} = 6.7$ Hz, CH_2Ph), 66.7 ($C_{1''}$), 62.7 ($C_{6'}$), 54.4 ($C_{2'}$), 45.0 (C_{1}), 42.8 (C_{3}), 38.4 (2d, $J_{C2''-P} = 6.8$ Hz, $C_{2''}$), 38.1 ($C_{4''}$), 30.2 ($C_{3''}$), 26.5, 26.0 ($C_{8''}$, $C_{5''}$), 22.9 (C_{4} H₃CO), 19.7 ($C_{10''}$ °, $C_{10''}$ *), 17.9 ($C_{9''}$); ³¹P NMR (CD_{3} OD) δ 22.3 (m); MS (ESI): m/z = 570 [M + H]⁺ 50, 592 [M + Na]⁺ 100%; HRMS calcd for C_{28} H₄₅NO₉P⁺ 570.2832, found 570.2806.

Benzyl hexadecanyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 23

To a suspension of 19 (65 mg, 76 µmol) in water (2 mL) was added trifluoroacetic acid (2 mL) at 0 °C. The reaction mixture was stirred at r.t. for 3 h and concentrated. Purification by flash chromatography (CH₂Cl₂/MeOH, 9:1) afforded 23 (45 mg, 90%, white solid) as a mixture of epimers (d.r. = 55/45): R_f 0.32 $(CH_2Cl_2/MeOH = 9:1)$; ¹H NMR δ 7.45–7.39 (m, 1H, NH), 7.38-7.28 (m, 5H, H_{ar}), 5.12-5.00 (m, 2H, CH_2Ph), 4.76-469 (m, 1H, $H_{1'}$), 4.09–4.01 (m, 1H, $H_{2'}$), 3.97–3.88 (m, 2H, $H_{1''}$), 3.90–3.82 $(m, 1H, H_{6'a}), 3.77-3.66 (m, 2H, H_{3'}, H_{6'b}), 3.63-3.52 (m, 2H, H_{4'}, H_{6'b})$ $H_{5'}$), 3.26–3.12 (m, 2H, H_{1}), 3.04–2.82 (m, 2H, H_{3}), 1.94, 1.93 (2 s, 3H, CH₃CO*, CH₃CO°), 1.58–1.50 (m, 2H, H_{2"}), 1.33–1.18 (m, 26H, H_{3"} to H_{15"}), 0.87 (t, 3H, $J_{\text{H16"-H15"}} = 7.0 \text{ Hz}$, H_{16"}); ¹³C NMR δ 200.1, 200.0 (2d, $J_{C2-P} = 5.5 \text{ Hz}$, C_2° , C_2^{*}), 172.2 (CH₃CO), 135.9 $(d, J_{Cq-P} = 5.5 \text{ Hz}, Cq_{ar}), 128.8, 128.8, 128.8, 128.2, 128.1 (CH_{ar}),$ $75.1 (C_{5'}), 71.2 (C_{3'}), 70.7 (C_{4'}), 68.6 (C_{1'}), 68.5, 68.4 (2d, J_{CH2Pb-P} =$ $6.5 \text{ Hz}, \text{CH}_2\text{Ph}^*, \text{CH}_2\text{Ph}^\circ), 67.3 \text{ (d, } J_{\text{Cl''-P}} = 6.5 \text{ Hz, } C_{\text{l''}}), 61.6 \text{ (C}_6),$ 52.9 (C_2), 42.9 (C_3), 42.2, 42.1 (2d, $J_{C1-P} = 127.0 \text{ Hz}$, C_1^* , C_1°), 32.1, 29.9, 29.8, 29.8, 29.7, 29.5, 29.3, 25.5, 23.1 ($C_{3''}$ to $C_{15''}$), 30.5, $30.4 (2d, J_{C2''-P} = 5.0 \text{ Hz}, C_{2''}^{\circ}, C_{2''}^{*}), 22.8 (\underline{CH}_{3}CO), 14.2 (C_{16''}); {}^{31}P$ NMR δ 21.1 (s); MS (ESI): $m/z = 1333 [2M + Na]^+ 100\%$; HRMS calcd for C₃₄H₅₈NNaO₉P⁺ 678.3747, found 678.3765.

5-Uracil-1-yl-pentyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2-oxopropylphosphonate 24

A suspension of 20 (15 mg, 24 μ mol) and Pd/C 10% (15 mg) in methanol (2 mL) was purged from air and filled with H₂. The reaction was stirred at r.t. for 4 h then diluted with methanol, filtered through a Celite pad and concentrated. Lyophilization afforded **24** (12 mg, 94%, white solid); $[\alpha]_D^{20}$ +26 (c 1.0, DMSO); ¹H NMR (DMSO-d₆) δ 11.19 (s, 0.8H, H_{3u}), 9.99 (s, 0.2H, H_{3u}), 7.65 (d, 1H, $J_{H6u-H5u} = 7.6$ Hz, H_{6u}), 7.65 (d, 1H, $J_{H5u-H6u} = 7.6$ Hz, H_{5u}), 7.28 (bs, 1H, NH), 5.01 (s, 1H, OH₄), 4.80 (s, 1H, OH₃), 4.50 (m, 1H, OH₆), 4.36 (ddd, 1H, $J_{H1'-H3a} = 7.3$ Hz, $J_{H1'-H3b} =$ 6.0 Hz, $J_{\text{H}'-\text{H}'} = 5.7$ Hz, H₁), 3.75 (ddd, 1H, $J_{\text{H}'-\text{H}'} = 9.8$ Hz, $J_{\text{H2'-NH}} = 7.8 \text{ Hz}, J_{\text{H2'-H1'}} = 5.7 \text{ Hz}, H_{\text{2'}}), 3.73-3.67 \text{ (m, 2H, } H_{\text{1''}}),$ 3.65 (t, 2H, $J_{\text{H5"-H4"}} = 7.0 \text{ Hz}$, $H_{\text{5"}}$), 3.58 (ddd, 1H, $J_{\text{H6'a-H6'b}} = 11.4$ Hz, $J_{\text{H6'a-OH6'}} = 5.7$ Hz, $J_{\text{H6'a-H5'}} = 1.5$ Hz, $H_{\text{6'a}}$), 3.47–3.35 (m, 2H, $H_{6'b}$, $H_{5'}$), 3.34–3.25 (m, 1H, $H_{3'}$), 3.16–3.00 (m, 2H, $H_{4'}$, H_{3a}), 2.86 (dd, 1H, J_{H1a-P} = 19.7 Hz, $J_{H1a-H1b}$ = 11.6 Hz, H_{1a}), 2.71 (dd, 1H, $J_{\text{H1b-P}} = 19.7 \text{ Hz}, J_{\text{H1b-H1a}} = 11.6 \text{ Hz}, H_{\text{1b}}), 2.61 \text{ (dd, 1H, } J_{\text{H3b-H3a}} =$ 16.3 Hz, $J_{\text{H3b-H1'}} = 6.0$ Hz, H_{3b}), 1.76 (s, 3H, CH₃CO), 1.65–1.45 (m, 4H, $H_{2''}$, $H_{4''}$), 1.36–1.20 (m, 2H, $H_{3''}$); ^{13}C NMR (DMSO- d_6) δ 203.5 (C₂), 169.8 (CH₃CO), 163.8(C_{4u}), 151.0 (C_{2u}), 145.8 (C_{6u}), $100.8 (C_{5u}), 75.0 (C_{3'}), 71.3 (C_{4'}), 70.7 (C_{5'}), 69.4 (C_{1'}), 63.4 (d, J_{C-P} =$ 4.1 Hz, $C_{1''}$), 61.4 ($C_{6'}$), 52.7 ($C_{2'}$), 47.5 ($C_{5''}$), 45.7 (C_{1}), 41.4 (C_{3}), 29.3 (d, $C_{2''}$, $J_{C-P} = 6.6$ Hz), 28.2 ($C_{4''}$), 22.7 (CH_3CO), 21.4 ($C_{3''}$); ³¹P NMR (DMSO-d₆) δ 7.91 (m); MS (ESI): $m/z = 520 \text{ [M - H]}^-$ 100%; HRMS calcd for $C_{20}H_{31}N_3O_{11}P^-$ 520.1696, found 520.1722.

(R)-7-Hydroxy-3,7-dimethyloctyl 3-(2-acetamido-2-deoxy-α-Dglucopyranosyl)-2-oxopropylphosphonate 25

A suspension of 21 (75 mg, 0.13 mmol) and Pd/C 10% (75 mg) in methanol (4 mL) was purged from air and filled with H₂. The reaction was stirred at r.t. for 3 h then diluted with methanol, filtered through a Celite pad and concentrated. Lyophilization afforded **25** (63 mg, 100%, white solid): $[\alpha]_D^{20}$ +51 (c 1.0, MeOH); H NMR (DMSO-d₆) δ 7.88 (d, 1H, $J_{NH-H2'}$ = 8.0 Hz, NH), 4.35 (ddd, 1H, $J_{\text{HI'-H3b}} = 7.3 \text{ Hz}$, $J_{\text{HI'-H2'}} = 6.0 \text{ Hz}$, $J_{\text{HI'-H3a}} = 5.5 \text{ Hz}$, J_{II} , 3.91– 3.78 (m, 2H, $H_{1''}$), 3.71 (ddd, 1H, $J_{H2'-H3'} = 10.0$ Hz, $J_{H2'-NH} = 8.0$ Hz, $J_{\text{H2'-H1'}} = 6.0 \text{ Hz}$, $H_{\text{2'}}$), 3.56 (bd, 1H, $J_{\text{H6'a-H6'b}} = 11.6 \text{ Hz}$, $H_{\text{6'a}}$), 3.48-3.37 (m, 2H, $H_{3'}$, $H_{6'b}$), 3.36-3.29 (m, 1H, $H_{5'}$), 3.09 (dd, 1H, $J = 8.9 \text{ Hz}, J = 8.3 \text{ Hz}, H_{4'}, 3.01, 2.97 \text{ (AB from ABX, 1H, } J_{AB} =$ 13.1 Hz, $J_{AX} = 21.6$ Hz, $J_{BX} = 20.5$ Hz, H_{1a} , H_{1b}), 2.84, 2.78 (AB) from ABX, 1H, $J_{AB} = 16.3$ Hz, $J_{AX} = 5.5$ Hz, $J_{BX} = 7.3$ Hz, H_{3a} , H_{3b}), 1.77 (s, 3H, CH₃CO), 1.64–1.47 (m, 2H, $H_{2''a}$, $H_{3''}$), 1.40– $1.15\ (m,\ 6H,\ H_{2''b},\ H_{4''a},\ H_{5''},\ H_{6''}),\ 1.12-1.04\ (m,\ 1H,\ H_{4''b}),\ 1.06$ (s, 6H, H_{8"}, H_{9"}), 0.85 (d, 3H, $J_{\text{H}10''-\text{H}3''} = 6.4 \text{ Hz}$, $H_{10''}$); ¹³C NMR (DMSO-d₆) δ 201.9 (d, J_{C2-P} = 5.0 Hz, C_2), 169.4 (CH₃CO), 74.8 $(C_{5'})$, 71.0 $(C_{4'})$, 70.4 $(C_{3'})$, 69.5 $(C_{1'})$, 68.7 $(C_{7''})$, 62.7 $(d, J_{C1''-P} =$ 6.0 Hz, $C_{1''}$), 61.1 ($C_{6'}$), 52.7 ($C_{2'}$), 44.3 (d, $J_{Cl-P} = 121.0$ Hz, C_{1}), $43.9 (C_{6''}), 41.3 (C_3), 37.2, 37.1 (C_{2''}, C_{4''}), 29.3, 29.2 (C_{8''}, C_{9''}), 28.9$ $(C_{3''})$, 22.6 (CH₃CO), 21.1 (C_{5''}), 19.2 (C_{10''}); ³¹P NMR (DMSO-d₆) δ 14.1 (s); MS (ESI): $m/z = 993 [2M - H]^{-} 100\%$; HRMS calcd for C₂₁H₃₉NO₁₀P⁻ 496.2312, found 496.2314.

Hexadecanyl 3-(2-acetamido-2-deoxy-α-D-glucopyranosyl)-2oxopropylphosphonate 26

A suspension of 23 (40 mg, 61 µmol) and Pd/C 10% (40 mg) in ethanol (10 mL) was purged from air and filled with H2. The reaction was stirred at r.t. for 3 h then diluted with methanol. Filtration through a Celite pad and concentration afforded 26 (30 mg, 86%, white solid): $[\alpha]_{D}^{20}$ +41 (c 1.0, MeOH); ¹H NMR (DMSO d_6) δ 8.64 (d, 1H, $J_{NH-H2'}$ = 9.0 Hz, NH), 4.34–4.26 (m, 1H, H_{1'}), 3.81-3.70 (m, 1H, H₂), 3.68-3.52 (m, 2H, H_{1"}), 3.50-3.38 (m, 1H, $H_{3'}$), 3.38–3.22 (m, 4H, H_{3a} , $H_{5'}$, $H_{6'}$), 3.04–2.96 (m, 1H, $H_{4'}$), 2.81 (dd, 1H, $J_{\text{HIa-P}} = 21.5 \text{ Hz}$, $J_{\text{HIa-HIb}} = 11.5 \text{ Hz}$, J_{Ia}), 2.60–2.35 (m, 2H, H_{1b}, H_{3b}), 1.73 (s, 3H, CH₃CO), 1.50–1.40 (m, 2H, H_{2"}), 1.32– 1.18 (m, 26H, $H_{3''}$ to $H_{15''}$), 0.85 (t, 3H, $J_{H16''-H15''} = 7.0$ Hz, $H_{16''}$); ¹³C NMR (DMSO-d₆) δ 203.3 (d, J_{C2-P} = 6.0 Hz, C_2), 169.6 (CH₃CO), 75.0 ($C_{5'}$), 71.7 ($C_{4'}$), 70.8 ($C_{3'}$), 69.3 ($C_{1'}$), 63.1 (d, $J_{C1''-P} = 6.0$ Hz, $C_{1''}$), 61.6 ($C_{6'}$), 52.5 ($C_{2'}$), 46.3 (d, $J_{C1-P} = 104.0 \text{ Hz}$, C_1), 41.8 (C_3), 31.2, 29.0, 28.9, 28.8, 28.6, 25.4, 22.6 ($C_{3''}$ to $C_{15''}$), 30.6 ($C_{2''}$), 22.0 (<u>C</u>H₃CO), 13.9 (C_{16"}); ³¹P NMR (DMSO-d₆) δ 7.9 (s); MS (ESI): $m/z = 564 \text{ [M - H]}^- 100\%$; HRMS calcd for $C_{27}H_{51}NO_9P^-$ 564.3301, found 564.3304.

Enzymatic assays

The activities of the compounds against MraY transferase were tested as previously described.36,16 The assay was performed in a reaction mixture of 10 µL containing, in final concentrations, 100 mM Tris-HCl, pH 7.5, 40 mM MgCl₂, 1.1 mM C₅₅-P, 250 mM NaCl, 0.25 mM UDP-MurNAc-[14C]pentapeptide (337 Bq), and 8.4 mM N-lauroyl sarcosine. The reaction was initiated by the addition of MraY enzyme purified from *Bacillus subtilis* (50 ng), and the mixture was incubated for 30 min at 37 °C under shaking with a thermomixer (Eppendorf). The reaction was stopped by

heating at 100 °C for 1 min. The compounds were also tested against MurG purified from E. coli as previously described. 36,37 Reaction mixtures contained, in a final volume of 12.5 µL, 200 mM Tris-HCl, pH 7.5, 10 mM MgCl₂, 16 µM UDP-[14C]GlcNAc (1.7 kBq), 16 µM lipid I analogue, 30% (v/v) dimethyl sulfoxide and MurG (7 ng). After 30 min at 37 °C, it was stopped by boiling for 3 min. Then, the mixture was lyophilised and taken up in 10 µL of 2-propanol/ammonium hydroxide/water (6:3:1; v/v/v). In both cases, the radiolabeled substrate (UDP-MurNAc-pentapeptide in the case of MraY, UDP-GlcNAc in the case of MurG) and reaction product (lipid I, product of MraY, and lipid II, product of MurG) were separated by TLC on silica gel plates LK6D (Whatman) using 2-propanol/ammonium hydroxide/water (6:3:1; v/v/v) as the mobile phase. The radioactive spots were located and quantified with a radioactivity scanner (model Multi-Tracemaster LB285; EG&G Wallac/Berthold). Residual activities were calculated with respect to a control assay without inhibitors. IC50 values were determined with 7 inhibitor concentrations. Data represent the mean of independent triplicate determinations, and the standard deviations were less than 10%.

Acknowledgements

We gratefully acknowledge the European Community for the financial support of the Eur-INTAFAR integrated project within the 6th PCRDT framework (contract No. LSHM-CT-2004-512138) and for a post-doctoral fellowship granted to B. A. We thank the "Région Ile de France" for a doctoral grant to N.A. and the "Ville de Paris" for a post-doctoral fellowship to R. F.

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