

Enantioselective Construction of Spiroindolenines by Ir-Catalyzed Allylic Alkylation Reactions

Qing-Feng Wu, Hu He, Wen-Bo Liu, and Shu-Li You*

State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032, China

Received June 20, 2010; E-mail: slyou@mail.sioc.ac.cn

Abstract: With 2-methyl-1,2,3,4-tetrahydroquinoline-derived phosphoramidite ligand (*R,R*)-**L**₆, Ir-catalyzed intramolecular C-3 allylic alkylation of indoles has been realized to afford highly enantioenriched spiroindolenine derivatives in 92–98% yields with up to >99/1 *dr* and 97% *ee*.

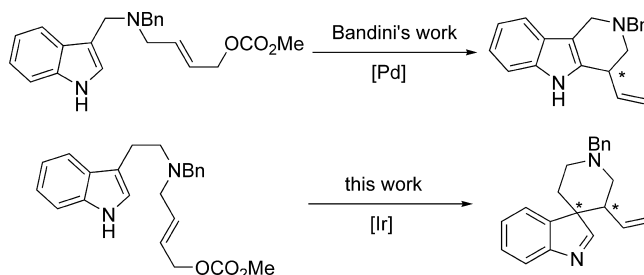
The spiroindolenine and spiroindoline units are privileged heterocyclic motifs that form the structural core for a large family of alkaloid natural products such as koumine, perophoramidine, and communesins.¹ Consequently, enormous efforts have been devoted to the development of efficient synthetic protocols for the construction of these skeletons.²

The synthesis of indolenines from indoles features a straightforward route and ready availability of starting materials but is faced with the difficulty of constructing a quaternary stereocenter *via* a dearomatization process.³ In 2005, Tamaru and co-workers reported a C-3 selective Pd-catalyzed allylation of 1*H*-indoles promoted by triethylborane using allylic alcohols.⁴ Soon after, Trost and Quancard described an enantioselective Pd-catalyzed C-3-alkylation of various 3-substituted indoles to construct a range of indolenine and indoline derivatives bearing quaternary stereocenters.^{3b} Rawal and co-workers recently extended the reaction to 2,3-disubstituted indoles.^{3c} Since Ir-catalysts have been successfully applied to the allylic alkylation of indoles at C-3 and N-1 by us and Hartwig,^{5,6} we envisaged that the spiroindolenines might be accessed via Ir-catalyzed intramolecular allylic alkylation of indoles. As an inspiration, Bandini and co-workers reported a highly enantioselective Pd-catalyzed intramolecular allylic alkylation for the synthesis of tetrahydro- β -carboline and tetrahydro- γ -carboline (Scheme 1).⁷ Recently, we found the spiroindolenine substructures could be formed through intramolecular allylic alkylation by extending the linker. Herein, we describe the first highly enantioselective synthesis of spiroindolenines via Ir-catalyzed asymmetric allylic alkylation reaction.

At the outset, we utilized a well-developed Ir-catalytic system including [Ir(COD)Cl]₂ and phosphoramidite **L**₁ (Table 1) as catalyst.⁸ In the presence of 2 mol % of [Ir(COD)Cl]₂, 4 mol % of **L**₁, and 2 equiv of Cs₂CO₃, reaction of **1a** in THF for 1.5 h gave spiroindolenine **2a** in 70% yield, 76/24 *dr*, and 95% *ee* (entry 1, Table 1). Screening various bases, DABCO, K₃PO₄, DBU, Et₃N, DIEA, DBN, and BSA (entries 2–8, Table 1), confirmed that Cs₂CO₃ was optimal. Next, we examined several chiral ligands, as summarized in Table 1. Ligands **L**₂, **L**₃, and **L**₅ could catalyze the reaction in excellent *ee*'s, but with only moderate *dr* ratios. To our delight, after screening two ligands developed in our group, **L**₆ and **L**₇,^{5b} we found that catalyst derived from **L**₆ gave satisfactory results in terms of yield, *dr*, and *ee* (entry 13, Table 1). Varying the solvent (DCM, toluene, dioxane, DME, and Et₂O) influenced the outcome considerably. Refluxed DCM gave the best result, affording **2a** in 95% yield, >99/1 *dr*, and 96% *ee* (entry 15, Table 1).

In the presence of 2 mol % of [Ir(COD)Cl]₂, 4 mol % of **L**₆, and 2 equiv of Cs₂CO₃ under DCM reflux, various 3-indolyl allyl carbonates were tested to examine the generality of the reaction. The

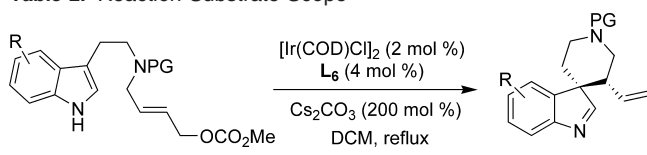
Scheme 1. Pd- and Ir-Catalyzed Intramolecular Allylic Alkylation

Table 1. Optimization of the Reaction Conditions^a

entry	ligand	solvent	base	temp (°C)	t (h)	yield (%) ^b	<i>dr</i> ^c	<i>ee</i> (%) ^d
1	L ₁	THF	Cs ₂ CO ₃	50	1.5	75	76/24	95
2	L ₁	THF	DABCO	50	1.5	48	61/39	95
3	L ₁	THF	K ₃ PO ₄	50	1.5	62	74/26	95
4	L ₁	THF	DBU	50	1.5	68	68/32	95
5	L ₁	THF	Et ₃ N	50	1.5	63	77/23	95
6	L ₁	THF	DIEA	50	1.5	67	66/34	95
7	L ₁	THF	DBN	50	1.5	41	70/30	95
8	L ₁	THF	BSA	50	1.5	56	75/25	95
9	L ₂	THF	Cs ₂ CO ₃	50	1.5	74	84/16	96
10	L ₃	THF	Cs ₂ CO ₃	50	1.5	72	81/19	93
11	L ₄	THF	Cs ₂ CO ₃	50	24	trace		
12	L ₅	THF	Cs ₂ CO ₃	50	24	40	>99/1	82
13	L ₆	THF	Cs ₂ CO ₃	50	1.5	95	96/4	91
14	L ₇	THF	Cs ₂ CO ₃	50	24	30	97/3	80 ^e
15	L ₆	DCM	Cs ₂ CO ₃	reflux	1.5	95	>99/1	96
16	L ₆	toluene	Cs ₂ CO ₃	50	1.5	84	>99/1	93
17	L ₆	dioxane	Cs ₂ CO ₃	50	1.5	73	90/10	86
18	L ₆	DME	Cs ₂ CO ₃	50	1.5	60	94/6	90
19	L ₆	Et ₂ O	Cs ₂ CO ₃	reflux	1.5	85	>99/1	93

^a Reaction conditions: 0.2 mmol of **1a**, 0.4 mmol of base in solvent (1.0 mL). ^b Isolated yield of the major diastereoisomer. ^c Determined by ¹H NMR of the crude reaction mixture. ^d Determined by chiral HPLC analysis. ^e Product with opposite configuration was obtained.

results are summarized in Table 2. Reaction of allylic carbonates with a varying protecting group on the linking N atom (Bn, *p*-Br-C₆H₄CH₂, Me, allyl) all gave the spiroindolenine products in good yields with

Table 2. Reaction Substrate Scope


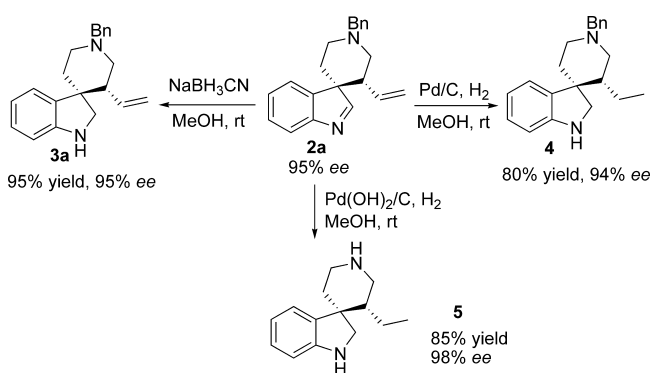
entry	substrate	product	yield (%) ^a	dr ^b	ee (%) ^c
1	1a (R = Bn)	2a	95	>99/1	96
2	1b (R = <i>p</i> -Br-C ₆ H ₄ CH ₂)	2b	92	96/4	95
3	1c (R = Me)	2c	95	>99/1	96
4	1d (R = allyl)	2d	93	96/4	96
5	1e	2e	97	>99/1	93
6	1f	2f	98	>99/1	94
7	1g	2g	93	96/4	94
8	1h	2h	95	>99/1	93
9	1i	2i	93	97/3	88
10	1j	2j	68/ 20 ^d	75/25	93/91
11	1k	2k	50/ 40 ^d	58/42	97/97

^a Isolated yields of the major diastereoisomer. ^b Determined by ¹H NMR of the crude reaction mixture. ^c Determined by chiral HPLC analysis. ^d Isolated yields of the minor diastereoisomer.

excellent *dr* and *ee* (92–95% yields, 96/4–>99/1 *dr*, 95–96% *ee*, entries 1–4, Table 2). Notably, the allyl protecting group in **1d** did not interfere with the alkylation process. The stereochemistry of the products was determined by X-ray structural determination of their enantiopure bromine-containing derivatives.⁹ Substrates bearing either an electron-donating group (4-Me, 5-MeO, 6-BnO) (entries 5–7, Table 2) or electron-withdrawing group (6-Br, 5-F) (entries 8–9, Table 2) on the indole core all led to their corresponding products in excellent yields, *dr*, and *ee* (93–98% yields, 96/4–>99/1 *dr*, 88–94% *ee*). The 2-substituted indolyl allyl carbonates were also tolerated and afforded the corresponding products in excellent yields and *ee* (88–90% yields, 91–97% *ee*, entries 10–11, Table 2), although in moderate *dr*.

To test the five-member ring spiroindolenine formation, Bandini's substrate, by shortening one carbon in **1a** (as shown in Scheme 1),⁷ was tested in the current catalytic system. Interestingly, tetrahydro- γ -carboline product was obtained,⁹ which suggests the formation of spiroindolenine is likely caused by the favorable six-member ring product by reacting at the C-3 position over the seven-member ring product at the C-2 position.

The multifunctionalized spiroindolenine products obtained here could undergo versatile transformation. As shown in Scheme 2,

Scheme 2. Transformation of Product **2a**

treatment of **2a** with sodium cyanoborohydride afforded the spiroindoline **3**. The C=C and C=N could be reduced by Pd/C catalyzed hydrogenation to give spiroindoline **4**. Interestingly, for Pd(OH)₂/C catalyzed hydrogenation, the Bn group was also removed. In all cases, there was no notable loss of the optical purity.¹⁰

In summary, we have developed a highly enantioselective synthesis of spiroindolenine derivatives via Ir-catalyzed intramolecular C-3 allylic alkylation of indoles. The spiroindolenine derivatives were obtained in excellent yields with up to >99/1 *dr* and 97% *ee*. Further extension of the reaction scope and development of more efficient catalytic systems are currently underway in our laboratory.

Acknowledgment. We thank the NSFC (20872159, 20821002, 20923005) and National Basic Research Program of China (973 Program 2009CB825300) for generous financial support.

Supporting Information Available: Experimental procedures and characterization of the products. This material is available free of charge via the Internet at <http://pubs.acs.org>.

References

- (1) Chou, T. Q. *Chin. J. Physiol.* **1931**, 5, 345. (b) Liu, C.-T.; Wang, Q.-W.; Wang, C.-H. *J. Am. Chem. Soc.* **1981**, 103, 4634. (c) Verbitski, S. M.; Mayne, C. L.; Davis, R. A.; Concepcion, G. P.; Ireland, C. M. *J. Org. Chem.* **2002**, 67, 7124. (d) Numata, A.; Takahashi, C.; Ito, Y.; Takada, T.; Kawai, K.; Usami, Y.; Matsumura, E.; Imachi, M.; Ito, T.; Hasegawa, T. *Tetrahedron Lett.* **1993**, 34, 2355. (e) Jadulco, R.; Edrada, R. A.; Ebel, R.; Berg, A.; Schaumann, K.; Wray, V.; Steube, K.; Proksch, P. *J. Nat. Prod.* **2004**, 67, 78.
- (2) Selected examples in total synthesis: (a) Magnus, P.; Mugrage, B.; DeLuca, M. R.; Cain, G. A. *J. Am. Chem. Soc.* **1989**, 111, 786. (b) Magnus, P.; Mugrage, B.; DeLuca, M.; Cain, G. A. *J. Am. Chem. Soc.* **1990**, 112, 5220. (c) Fuchs, J. R.; Funk, R. L. *J. Am. Chem. Soc.* **2004**, 126, 5068. (d) Sabahi, A.; Novikov, A.; Rainier, J. D. *Angew. Chem., Int. Ed.* **2006**, 45, 4317. (e) Yang, J.; Wu, H.; Shen, L.; Qin, Y. *J. Am. Chem. Soc.* **2007**, 129, 13794. Selected examples of construction of spiroindole skeletons: (f) Ibaceta-Lizana, J. S. L.; Jackson, A. H.; Prasitpan, N.; Shannon, P. V. R. *J. Chem. Soc., Perkin Trans. 2* **1987**, 1221. (g) Linnepe, P.; Schmidt, A. M.; Eilbracht, P. *Org. Biomol. Chem.* **2006**, 4, 302.
- (3) (a) Austin, J. F.; Kim, S.-G.; Sinz, C. J.; Xiao, W.-J.; MacMillan, D. W. C. *Proc. Natl. Acad. Sci. U.S.A.* **2004**, 101, 5482. (b) Trost, B. M.; Quancard, J. *J. Am. Chem. Soc.* **2006**, 128, 6314, and references therein. (c) Kagawa, N.; Malerich, J. P.; Rawal, V. H. *Org. Lett.* **2008**, 10, 2381. (d) Eastman, K.; Baran, P. S. *Tetrahedron* **2009**, 65, 3149.
- (4) Kimura, M.; Futamata, M.; Mukai, R.; Tamaru, Y. *J. Am. Chem. Soc.* **2005**, 127, 4592.
- (5) (a) Liu, W.-B.; He, H.; Dai, L.-X.; You, S.-L. *Org. Lett.* **2008**, 10, 1815. (b) Liu, W.-B.; He, H.; Dai, L.-X.; You, S.-L. *Synthesis* **2009**, 2076. (c) Stanley, L. M.; Hartwig, J. F. *Angew. Chem., Int. Ed.* **2009**, 48, 7841.
- (6) For reviews: (a) Takeuchi, R.; Kezuka, S. *Synthesis* **2006**, 3349. (b) Helmchen, G.; Dahnz, A.; Dübon, P.; Schelwies, M.; Weihofen, R. *Chem. Commun.* **2007**, 675. For the first Ir-catalyzed asymmetric allylic alkylation reaction: (c) Janssen, J. P.; Helmchen, G. *Tetrahedron Lett.* **1997**, 38, 8025.
- (7) Bandini, M.; Melloni, A.; Piccinelli, F.; Sinisi, R.; Tommasi, S.; Umami-Ronchi, A. *J. Am. Chem. Soc.* **2006**, 128, 1424.
- (8) (a) Ohmura, T.; Hartwig, J. F. *J. Am. Chem. Soc.* **2002**, 124, 15164. (b) López, F.; Ohmura, T.; Hartwig, J. F. *J. Am. Chem. Soc.* **2003**, 125, 3426. (c) Kiener, C. A.; Shu, C.; Incarvito, C.; Hartwig, J. F. *J. Am. Chem. Soc.* **2003**, 125, 14272. (d) Raskatov, J. A.; Spiess, S.; Gnam, C.; Brödner, K.; Rominger, F.; Helmchen, G. *Chem.—Eur. J.* **2010**, 16, 6601.
- (9) See the Supporting Information (SI).
- (10) The *ee* of **4** and **5** were determined after their conversion to N-Ts derivatives (see the SI for details).

JA105111N