## Generalized Synthesis of Pentaco-ordinated Tungsten(IV) Carbene Complexes

## A. Aguero, J. Kress, and J. A. Osborn\*

Laboratoire de Chimie Inorganique Moléculaire et de Catalyse, U.A. au C.N.R.S. n° 424, Institut Le Bel, 4 rue Blaise Pascal, 6700 Strasbourg, France

General methods enabling the synthesis of a large variety of pentaco-ordinate tungsten( $\nu$ ) carbene complexes are described permitting notably the isolation of the first unsaturated (<18e) metal carbenes containing  $\beta$ -hydrogens.

Although several metal carbene complexes have been found to catalyse olefin metathesis, <sup>1,2</sup> only one type<sup>2</sup> seems to induce the very high activity found for classical catalyst systems.<sup>3</sup> We report here the synthesis of further members of this important family, thus enabling a clearer understanding of their chemical and catalytic properties.

The previously reported complexes (1) and (2) (Table 1) resulted from the reaction of WO(OCH<sub>2</sub>Bu<sup>1</sup>)<sub>2</sub>(CH<sub>2</sub>Bu<sup>1</sup>)<sub>2</sub> with AlX<sub>3</sub>.<sup>2</sup> Although (3)—(5) can be prepared similarly, this synthetic route is not general. Fortunately, we find that ligand substitution reactions can be conveniently carried out on (1)—(5), allowing replacement of the anionic ligands X or OR, as well as of the neopentylidene group.

The stepwise substitution of the axial halide ligands X has been achieved in several cases. For example, treatment of (2) with the appropriate number of equivalents of LiOCH<sub>2</sub>Bu<sup>t</sup> (in pentane) or AgSO<sub>3</sub>CF<sub>3</sub> (in benzene) yields straightforwardly (6) and (7), or (8) and (9), respectively. The <sup>1</sup>H and <sup>13</sup>C n.m.r. chemical shifts (Table 1) are consistent with the expected  $\pi$ -donor strength of the axial ligands, and show clearly that the usually labile trifluoromethanesulphonate anion is strongly co-ordinated to the metal and unfortunately the cationic species such as [W(CHBu<sup>t</sup>)(OCH<sub>2</sub>Bu<sup>t</sup>)<sub>2</sub>Br]<sup>+</sup> which are extremely active in metathesis are not formed.<sup>2</sup> No changes are observed in the  $C_{\alpha}$ -H<sub> $\alpha$ </sub> coupling constant, whose high value indicates that only a weak direct W-H<sub> $\alpha$ </sub> interaction<sup>4</sup> occurs in all these compounds, despite their formally unsaturated nature.

The equatorial OR ( $R = CH_2Bu^t$ ,  $Pr^i$ ) ligand in (1)—(5) can also be substituted: e.g. (10) and (11) are formed from (2) and (3) by treatment with  $BX_3$  (0.5 equiv.). Certain alcohols e.g.  $Ph_3SiOH$  or phenols react with (2), displacing  $Bu^tCH_2OH$ , to yield complexes such as (12) and (13). Compound (12) exists as two isomers (12a) and (12b) (ca. 3:1 in  $C_6H_6$  at 25 °C) differing by the respective positions of the  $OCH_2Bu^t$  and  $OSiPh_3$  ligands in the equatorial plane of the molecule. We find for these complexes that rapid intermolecular alkoxo exchange occurs and hence the (12a):(12b) ratio is under

thermodynamic control. Homologous compounds such as  $W(CHBu^i)(OCH_2Bu^i)(OR)Br_2$  ( $R=Ph, Pr^i$ ) or  $W(CH-Bu^i)(OCH_2Bu^i)(OEt)I_2$  behave similarly. In contrast with axial substitution, the replacement of the equatorial neopent-oxo ligands by poorer  $\pi$ -donors is followed by the decrease of the  $J(C_\alpha H_\alpha)$  value (Table 1) which indicates the strengthening of the direct  $W-H_\alpha$  interaction<sup>4</sup> and explains the high field shift of the  ${}^1H_\alpha$  n.m.r. signal (Table 1).

The neopentylidene complexes (1)—(5) react with terminal olefins  $CH_2=CR^1R^2$  (10—20 equiv.,  $CH_2Cl_2$ , 25 °C, h) in a metathesis reaction (L = ligand), equation (1). Good yields of the new carbene complexes are obtained by displacement of the equilibrium by vacuum evaporation of the Bu<sup>1</sup>CH=CH<sub>2</sub>. Hence reaction of e.g. (3) with hex-1-ene yields (14), with

Table 1. Characteristic n.m.r. data for the tungsten carbene complexes.

|   | $\delta H_{\alpha}$ | $\delta C_{\alpha}$ | ${}^{1}J(C_{\alpha}H_{\alpha})$ |
|---|---------------------|---------------------|---------------------------------|
| Complexes                                     | $(C_6D_5Br)$        | $(C_6D_6)$          | /Hz                             |
| $W(CHBu^t)(OCH_2Bu^t)_2Cl_2(1)^a$             | 10.78               | 289.9               | 136                             |
| $W(CHBu^{t})(OCH_{2}Bu^{t})_{2}Br_{2}(2)^{a}$ | 11.12               | 297.2ь              | 135                             |
| $W(CHBu^t)(OCH_2Bu^t)_2I_2(3)$                | 11.38               | 305.2               | 135                             |
| $W(CHBu^t)(OPr^i)_2Cl_2(4)$                   | 10.64               | 288.6               |                                 |
| $W(CHBu^{t})(OPr^{i})_{2}Br_{2}(5)$           | 10.99               | 295.2               | 135                             |
| $W(CHBu^t)(OCH_2Bu^t)_3Br(6)$                 | 9.76                | 273.7ь              | 135                             |
| $W(CHBu^t)(OCH_2Bu^t)_4$ (7)                  | 7.90                | 252.6               | 137                             |
| $W(CHBu^t)(OCH_2Bu^t)_2(CF_3SO_3)Br(8)$       | 11.50               | 302.5ь              | 137                             |
| $W(CHBu^t)(OCH_2Bu^t)_2(CF_3SO_3)_2$ (9)      | 11.78               | 306.2ь              |                                 |
| $W(CHBu^t)(OCH_2Bu^t)Br_3(10)^a$              | 8.54c               | 308.2c,d            | 127                             |
| $W(CHBu^t)(OCH_2Bu^t)I_3(11)$                 | $6.66^{d}$          | 321.5c,d            | 114                             |
| $W(CHBu^t)(OCH_2Bu^t)(OSiPh_3)Br_2$ (12a)     | 10.64               | 299.2               | 134                             |
| (12b)   | 10.85               | 300.4               |                                 |
| $W(CHBu^t)(OSiPh_3)_2Br_2$ (13)               | 10.21               | 302.1               |                                 |
| $W(CHBu^n)(OCH_2Bu^t)_2I_2$ (14)              | 11.89e              | 302.1               |                                 |
| $W(CHPh)(OCH_2Bu^t)_2I_2(15)$                 | 11.88               | 285.8               |                                 |

a Ref. 2. b In C<sub>6</sub>D<sub>5</sub>Br. c At 240 K. d In CD<sub>2</sub>Cl<sub>2</sub>. c Triplet, <sup>3</sup>J(CH) 8 Hz.

styrene (15), and with the exomethylene carbocycles  $CH_2 = \overline{C(CH_2)_n}CH_2$  (n = 3,4) (16) and (17).† This reaction is quite general and provides for instance the first examples of non-18e metal carbene complexes possessing  $\beta$ -hydrogens, permitting a systematic study of their properties e.g. thermal stability, modes of decomposition, reactivity, which differ notably from (1)—(5).

$$LW=CHBu^{t} + CH_{2}=CR^{1}R^{2} \rightleftharpoons LW=CR^{1}R^{2} + Bu^{t}CH=CH_{2}$$
(1)

The combination of the three synthetic methods above provides access to a large number of simple pentaco-ordinate tungsten carbene complexes and further reports will be concerned with studies on their reactivity as metathesis catalysts in the presence or absence of Lewis acids. For example, using (3), functionalised olefins such as CH<sub>2</sub>=CH(CH<sub>2</sub>)<sub>2</sub>CO<sub>2</sub>Me can be metathesised at 60 °C into C<sub>2</sub>H<sub>4</sub> and MeO<sub>2</sub>C(CH<sub>2</sub>)<sub>2</sub>CH=CH(CH<sub>2</sub>)<sub>2</sub>CO<sub>2</sub>Me in chlorobenzene. Further, by such suitable ligand modifications,

certain of these carbenes show Wittig<sup>5</sup> type reactivity with carbonyl functional groups (e.g. in esters, lactones), the details of which will be reported separately.

We thank the C.N.R.S. for supporting this work. A. A. thanks the Venezuelan Government for financial support.

Received, 25th February 1985; Com. 256

## References

- C. P. Casey and T. J. Burkhardt, J. Am. Chem. Soc., 1974, 96, 7808; K. Weiss and P. Kindl, Angew. Chem., 1984, 96, 616; R. R. Schrock, S. M. Rocklage, J. H. Wengrovius, G. A. Rupprecht, and J. D. Fellmann, J. Mol. Catal., 1980, 8, 73; S. M. Rocklage, J. D. Fellman, G. A. Rupprecht, L. W. Messerle, and R. R. Schrock, J. Am. Chem. Soc., 1981, 103, 1440; F. N. Tebbe, G. W. Parshall, and D. W. Ovenall, ibid., 1979, 101, 5074.
- 2 J. Kress, M. Wesolek, and J. A. Osborn, J. Chem. Soc., Chem. Commun., 1982, 514; J. Kress and J. A. Osborn, J. Am. Chem. Soc., 1983, 105, 6346.
- 3 R. H. Grubbs, *Prog. Inorg. Chem.*, 1978, **24**, 1; J. J. Rooney and A. Stewart, *Catalysis (London)*, 1977, **1**, 277.
- 4 R. R. Schrock, Acc. Chem. Res., 1979, 12, 98.
- 5 G. Wittig, J. Organomet. Chem., 1975, 100, 279.

<sup>†</sup> Satisfactory analyses of (17) were obtained and the mass spectrum showed the expected parent ions and fragmentation, as for most of the reported compounds.