

Differential-pulse Voltammetric Determination of Phosphate as Molybdovanadophosphate at a Glassy Carbon Electrode and Assessment of Eluents for the Flow Injection Voltammetric Determination of Phosphate, Silicate, Arsenate and Germanate

The redox behaviour of molybdovanadophosphate at a glassy carbon electrode is described and a procedure is given for the voltammetric determination of phosphate as molybdovanadophosphate at a glassy carbon electrode in a static system. Procedures are also given for the voltammetric flow injection determination of phosphate, silicate, arsenate and germanate by the injection of heteropolyacids pre-formed in various aqueous, aqueous acetone and aqueous ethanolic reagents into eluents consisting of the reagent blank. This procedure effectively eliminates the background signal of the blank and allows precise determinations to be made. Plots of electrode potential against the current obtained are given.

Silicate and phosphate can be determined at 10^{-7} and 10^{-6} M levels, respectively. Arsenate has only been determined at the 10^{-5} M level, and the precise determination of germanate is difficult owing to adsorption at the glassy carbon electrode.

Keywords: Orthophosphate, silicate, arsenic and germanium determination; voltammetry; flow injection analysis

A. G. FOGG and N. K. BSEBSU

Chemistry Department, Loughborough University of Technology, Loughborough, Leicestershire, LE11 3TU.

Analyst, 1981, **106**, 1288–1295.

Use of Ascorbic and Thioglycollic Acids to Eliminate Interference from Iron in the Aluminon Method for Determining Aluminium

The use of ascorbic and thioglycollic acids as inhibitors for the interference of iron in the aluminon method of Hsu have been examined. The use of ascorbic acid, as proposed by Jayman and Sivasubramaniam, has been found to change iron interference from positive to negative causing aluminium to be underestimated. However, the addition of 0.2 ml of a 1% V/V solution of thioglycollic acid to solutions containing aluminium in amounts ranging from 10 to 50 μ g has been proved to suppress the interference from up to 900 μ g of iron.

Keywords: Aluminium determination; aluminon method; ascorbic acid; iron interference; thioglycollic acid

F. CABRERA, L. MADRID and P. DE ARAMBARRI

Centro de Edafología y Biología Aplicada del Cuarto, Apartado 1052, Seville, Spain.

Analyst, 1981, **106**, 1296–1301.

METHODS

Official Methods of Analysis of the AOAC

Thirteenth Edition—1980

Agricultural Liming Materials
Fertilizers
Plants
Disinfectants
Hazardous Substances
Pesticide Formulations
Animal Food
Baking Powders and
Baking Chemicals
Beverages: Distilled Liquors
Beverages: Malt Beverages
and Brewing Materials
Beverages: Wines
Beverages: Non-Alcoholic
and Concentrates
Cacao Bean and Its Products
Cereal Foods
Coffee and Tea
Dairy Products

Eggs and Egg Products
Fish and Other Marine Products
Flavors
Food Additives: Direct
Food Additives: Indirect
Fruits and Fruit Products
Gelatin, Dessert Preparations,
and Mixes
Meat and Meat Products
Metals and Other Elements
as Residues in Foods
Natural Poisons
Nuts and Nut Products
Oils and Fats
Pesticide Residues
Spices and Other Condiments
Sugars and Sugar Products
Vegetable Products, Processed
Waters; and Salt

Color Additives
Cosmetics
Drugs and Food Additives
in Animal Tissues
Drugs in Foods
Vitamins and Other Nutrients
Drugs: General
Drugs: Acidic
Drugs: Alkaloid and Related Bases
Drugs: Neutral
Drugs: Illicit
Extraneous Materials Isolation
Forensic Sciences
Microbiological Methods
Microchemical Methods
Radioactivity
Spectroscopic Methods
Standard Solutions and Materials

Association of Official Analytical Chemists, Dept. A
1111 N 19th St. Suite 210, Arlington, VA 22209

Please enter my order for the Thirteenth Edition of Official Methods of Analysis, 1980 at \$78.00 post paid, via book post.
-----number of copies

NAME -----

Please Print

ADDRESS -----

CITY ----- STATE or COUNTRY ----- ZIP CODE -----

ATTENTION: -----

PLEASE ENCLOSE REMITTANCE WITH ORDER.

Effect of Temperature on the Structural Rearrangements of Polyesters (LAC-series) when Used as Liquid Stationary Phases in Gas - Liquid Chromatography

This study deals with the effect of temperature on the gas-chromatographic behaviour of thiols, alkanes, branched-chain alkanes and cyclic alkanes on polyester (LAC-446, LAC-745, LAC-772, LAC-860, LAC-886, LAC-841 and LAC-935) liquid stationary phases. Increases were observed in specific retention volumes as well as improvements in the separations at certain temperatures. The critical temperatures were different for the various liquid stationary phases; these were 110 °C for LAC-446 and LAC-886 and 120 °C for LAC-860 and LAC-772. Differential-thermal analysis of two of the liquid stationary phases indicated a change in the physical properties of the polymers during heating.

Keywords: Gas - liquid chromatography; liquid stationary phases; effect of temperature; LAC-series of polyesters; structural rearrangements

ALBERTINE E. HABBOUSH, SABRI M. FARROHA, AL'A K. ABDUL AL-SADA and ABDUL MASSEH N. KITTO

College of Science, University of Baghdad, Adhamiya, Baghdad, Iraq.

Analyst, 1981, **106**, 1302–1308.

Rapid Automated Enzymatic Method for the Determination of Alcohol in Blood and Beverages Using Flow Injection Analysis

An enzymatic method for the determination of alcohol using flow injection analysis is described. Samples are suitably diluted and introduced directly into the system. Blood alcohol is analysed by the stop-flow technique at a rate of 70–80 samples per hour and the results are compared with those from headspace gas chromatography. Alcohol in several beverages is analysed by the continuous-flow technique at a rate of up to 120 samples per hour. In both instances the result is available less than 30 s after injection.

Keywords: Flow injection analysis; alcohol determination; enzymatic method; whole blood; stop-flow technique

P. J. WORSFOLD, J. RŮŽIČKA and E. H. HANSEN

Chemistry Department A, Technical University of Denmark, Building 207, DK-2800 Lyngby, Denmark.

Analyst, 1981, **106**, 1309–1317.

A Copper(I) Iodide Paper for the Detection and Determination of the Concentration of Mercury Vapour in the Workplace Atmosphere

The development and preparation of a test paper for mercury in air is described. A measured volume of air is drawn through a silica gel-loaded filter-paper coated with a mixture of copper(I) iodide and sodium carboxymethylcellulose, and in the presence of mercury a pink stain is produced. The paper has been validated in the laboratory and a factory for the reliable determination of mercury vapour in air at concentrations up to 100 $\mu\text{g m}^{-3}$.

Keywords: Mercury vapour; copper(I) iodide test paper; workplace atmosphere; air

S. CRISP, D. W. MEDDLE, J. M. NUNAN and A. F. SMITH

Department of Industry, Laboratory of the Government Chemist, Cornwall House, Stamford Street, London, SE1 9NQ.

Analyst, 1981, **106**, 1318–1325.

TUCK IN UNDER FLAP A

**THE ANALYST
READER ENQUIRY SERVICE**

December, 1981

For further information about any of the products featured in the advertisements in this issue, please write the appropriate A number in one of the boxes below.

Postage paid if posted in the British Isles but overseas readers must affix a stamp.

FIRST FOLD

--	--	--	--	--	--	--	--

(Please use BLOCK CAPITALS)

NAME

OCCUPATION

ADDRESS

SECOND FOLD

Postage
will be
Paid by
Licensee

Do not affix Postage Stamps if posted in
Gt. Britain, Channel Islands or N. Ireland

BUSINESS REPLY SERVICE
Licence No. W.D. 106

2

CUT ALONG THIS EDGE

**Reader Enquiry Service
The Analyst
The Royal Society of Chemistry
Burlington House
Piccadilly London W1E 6WF
ENGLAND**

THIRD FOLD

FIRST FOLD

Open-cell Polyurethane Foam as a Sorbent in the Extraction of Iodine-131

Open-cell polyurethane foams have proved to be effective as sorbents for inorganic and organic species and as supporting materials for various hydrophobic organic phases. They possess outstanding sorption, mass-transfer and hydrodynamic properties, which enable them to be used at relatively high flow-rates of aqueous solutions during column operation. Iodine-131 can be extracted from samples of water and milk using resilient open-cell polyurethane foams in static and pulsed column beds, when in the form of a cylindrical packing impregnated with a tri-alkylamine containing dissolved inactive iodine. Pulsed column beds, which have some advantages over static column beds, can be conveniently automated. This may lead to the development of a new type of environmental monitoring device for radioactive iodine.

Keywords: Iodine-131 extraction; polyurethane foam; water; milk; sorbent

Š. PALÁGYI

Institute of Radioecology and Applied Nuclear Techniques, P.O. Box A-41, 04061 Košice, Czechoslovakia.

and T. BRAUN

Institute of Inorganic and Analytical Chemistry, L. Eötvös University, P.O. Box 123, 1443 Budapest, Hungary.

Analyst, 1981, **106**, 1326–1333.

Reactive Ion-exchange Precipitation Procedure for the Determination of Trace Amounts of Oxalate

The feasibility of using reactive ion-exchange for the determination of oxalate in aqueous solution has been demonstrated. The procedure consists of pH adjustment of the analyte solution, concentration of oxalate on a lead(II)-loaded cation exchange mini-column in the form of lead oxalate precipitate, reactive elution of oxalate by sulphuric acid dissolution and analysis of the concentrated eluate by permanganate titration. Almost quantitative yields of oxalate are obtained for concentrations typically found in human urine. The procedure represents a simple and reliable technique to concentrate oxalate for determination by classical methods.

Keywords: Oxalate trace determination; reactive ion exchange; precipitation; titration

SUSAN M. ANDEL, GILBERT E. JANAUER and WILLIAM E. BERNIER

Department of Chemistry, State University of New York at Binghamton, Binghamton, N.Y. 13903, USA.

Analyst, 1981, **106**, 1334–1337.

Sample Fusion at Low Temperature for the Potentiometric Determination of Fluorine in Silicate Materials

Short Paper

Keywords: Fluorine determination; silicates; alkaline fusion; fluoride ion-selective electrode

B. FABBRI and F. DONATI

C.N.R., Research Institute for Ceramics Technology, Via Granarolo 6, 48018 Faenza, Italy.

Analyst, 1981, **106**, 1338–1341.

Determination of Fluoride Ion in Bovine Milk Using a Fluoride Ion-selective Electrode*Short Paper**Keywords: Fluoride determination; milk; ion-selective electrode***CLIFFORD G. BEDDOWS**

School of Health and Applied Sciences, Leeds Polytechnic, Leeds, LS1 3HA.

and DAVID KIRK

Department of Hotel and Catering Studies, Sheffield City Polytechnic, Sheffield, S1 1WB.

Analyst, 1981, **106**, 1341–1344.**Use of a Carbon Fibre Indicator Electrode for the Potentiometric Titration of Glycine in Glacial Acetic Acid***Short Paper**Keywords: Glycine determination; potentiometric titration; carbon fibre indicator electrode***V. J. JENNINGS**

Department of Applied Chemistry, Coventry (Lanchester) Polytechnic, Priory Street, Coventry, CV1 5FB.

Analyst, 1981, **106**, 1344–1347.**Reagent for the Spectrophotometric Determination of Iron(II) in Alkaline Solution***Short Paper**Keywords: Iron(II) determination; pyridine-2-hydrazide reagent; spectrophotometry***KHALID A. ABDULLAH and YOUNIS I. HASSAN**

Department of Chemistry and Biology, College of Education, University of Mosul, Mosul, Iraq.

and A. M. AL-DAHER and W. A. BASHIR

Department of Chemistry, College of Science, University of Mosul, Mosul, Iraq.

Analyst, 1981, **106**, 1348–1351.