## **Additions and Corrections**

Vol. 3, 2001

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Ethyl(methyl)dioxirane as an Efficient Reagent for the Oxidation of Nucleoside Phosphites into Phosphates under Anhydrous Conditions.

Page 815. The reagent used for the oxidation was 2-butanone peroxide, not ethyl(methyl)dioxirane. Therefore, all occurrences of "ethyl(methyl) dioxirane" in the title and text should be replaced with "2-butanone peroxide".

The correct abstract graphic is

The following corrections should be made to the text in the second paragraph because the molar concentration of the peroxide solution cannot be defined, since 2-butanone peroxide is a mixture of several compounds with different molecular weights:

Delete the molar concentration (7.3 M) listed in the first sentence. In the second sentence, replace "a 0.1 or 0.01 M solution" with "a 0.67 or 0.067% solution". In sentence 5 (beginning "First, the utility..."), delete "(1 equiv)" appearing after ethyl(methyl) dioxirane. In sentence 7 (beginning "Noteworthy here ..."), "1 equiv" should be changed to "a low concentration solution".

In footnote 16, change "0.1 M ethyl(methyl) dioxirane" to "0.67% 2-butanone peroxide".

OL010160V

10.1021/ol010160v

**Published on Web 08/01/2001** 

Vol. 3, 2001

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*N*-(*tert*-Butoxycarbonyl)-*N*-[4-(dimethylazaniumylidene)-1,4-dihydropyridin-1-ylsulfonyl] Azanide: A New Sulfamoylating Agent. Structure and Reactivity toward Amines.

Page 2243, Scheme 2. The R' moiety was omitted in compound 3. The correct structure is given below.

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OL010155R

10.1021/ol010155r **Published on Web 07/26/2001**