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An End to the Search for the Ground State of C₈₄?

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Application of a simple symmetry rule for fullerene isomer distributions reveals that the 13 C NMR pattern of C_{84} in uniquely consistent with a 2:1 thermodynamic mixture of isoenergetic D_2 and D_{2d} symmetry isomers.

The search for the ground state structure of the fullerene C₈₄ has not been altogether straightforward. On the experimental side, an initial report of 4 dominant ¹³C NMR lines in the C₈₄ spectrum¹ soon gave way to a report of 25 lines,² and this has since been superseded by a reproducible count of 32.3.4 On the theoretical side, an initial proposal of $T_{\rm d}$, $D_{\rm 6h}$ and D2 symmetry candidate structures for C84 was based primarily on π -electronic stability.⁵ It has since become apparent that the stability of fullerenes is a delicate balance between π -electronic and σ -bonding strain effects, 6,7 and semi-empirical calculations have found these initial candidate structures to be more strained than several others.8,9 Developments such as these may seem surprising, since the initial structural proposals for C_{60} , 10 C_{70} , 11 and C_{76} were all confirmed by spectroscopic (IR and NMR) evidence as soon as the experiments became possible. 13-15 However, because the number of potentially stable C_n isomers increases rapidly with n, the ground-state structures of higher fullerenes such as C₈₄ are more difficult to determine.

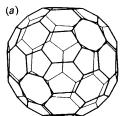
The current theoretical situation can be summarised as follows. Stable higher fullerenes are expected to have isolated pentagons, because this both decreases the σ -bonding strain energy¹⁶ and increases the π -bonding delocalisation.¹⁷ A complete tabulation of the 24 topologically distinct isolatedpentagon fullerene isomers of C₈₄ has been given by Manolopoulos and Fowler.⁷ This tabulation has been exploited by Zhang et al.⁸ who have performed semi-empirical calculations on all 24 isomers using a tight-binding Hamiltonian fit to the experimental properties of bulk carbon. These calculations suggest that two essentially isoenergetic isomers, with D_2 and D_{2d} symmetry, should be significantly (ca. 23 kJ mol⁻¹) more stable than any others. A similar conclusion has been reached independently by Raghavachari,9 who has performed semi-empirical (MNDO) calculations on a subset of the 24 isomers. The particularly stable D_2 and D_{2d} symmetry isomers are numbered 22 and 23 in the general tabulation, and are shown again here in Fig. 1.

A simple topological explanation for the special stability of these D_2 and D_{2d} isomers has also been proposed by Raghavachari. According to his argument, the steric stability of a fullerene cage is maximised when the hexagonal faces have equal numbers of adjacent hexagonal neighbours. Thus C_{60} has an ideal distribution of steric strain because each of its 20 hexagons has the same number (3) of hexagonal neighbours, all of which have equivalent local environments. Such an ideal situation cannot be realised by C_{84} , the most uniform neighbour distribution for which has 28 hexagons with 4 hexagonal neighbours and 4 hexagons with 5 neighbours. This

optimum distribution is only realised by one D_{2d} (23) and two D_2 (21 and 22) structures in the complete list of C_{84} possibilities.⁷ Moreover the four hexagons with five hexagonal neighbours are more evenly distributed on the surface of the fullerene in structures 22 and 23 than they are in structure 21, implying a secondary reduction in steric strain.⁹ Thus an elementary consideration of steric strain points uniquely to the special stability of the D_2 (22) and D_{2d} (23) isomers.

The current experimental situation is no less clear-cut. The $^{13}\mathrm{C}$ NMR spectrum of isolated C_{84} is dominated by 31 strong resonance lines and one additional line of half the intensity. In contrast to the case of C_{78} , for which a number of quite different NMR spectra have been observed in different laboratories, $^{3.18,19}$ this C_{84} NMR pattern has now been reproduced by several independent groups. 4,20,21 Since different fullerene generation and chromatographic separation conditions were used in these various studies, the 32 line spectrum would appear to be an intrinsic property of the ground state of C_{84} . The $^{13}\mathrm{C}$ NMR spectrum of chromatographically separated C_{84} obtained by the Sussex group is shown here in Fig. 2, and is essentially identical to those reported elsewhere. 3,20

The D_2 symmetry C_{84} structure in Fig. 1(a) has 21 sets of four equivalent carbon atoms. The D_{2d} symmetry C_{84} structure in Fig. 1(b) has 10 sets of eight equivalent carbon atoms, and one set of four. A 2:1 mixture of the two isomers is therefore consistent with the observed 13 C NMR pattern. However, since there are three other possible D_2 symmetry isolated-pentagon C_{84} isomers which also have 21 sets of four equivalent carbon atoms, 7 this interpretation of the spectrum is not unique. In order to provide stronger evidence for the assignment, we must also explain why the precise 2:1 ratio of D_2 and D_{2d} symmetry isomers is obtained. 3,4



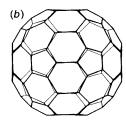


Fig. 1 Favoured C_{84} isomers with D_2 and D_{2d} symmetry.^{8,9} (a) A view down the principal axis that passes through two symmetry-equivalent pyracylene patches in the D_2 isomer. (b) Rotation of the central bond of either patch produces the D_{2d} isomer

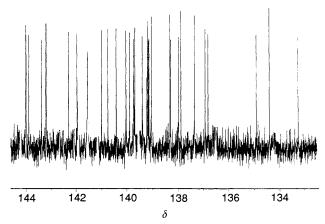


Fig. 2 ¹³C NMR spectrum of C₈₄, obtained relative to CDCl₃ as lock signal. Additional lines due to other minor isomer components have been deleted for clarity of presentation, and the full spectrum will be discussed in ref. 4. The pattern of 31 strong lines and one line at half the intensity is consistent with the observations of Kikuchi et al.3 Diederich and Whetten have reported 31 of the 32 lines seen here,²⁰ and have since also assigned the 32nd line²⁵

Assuming that the semi-empirical calculations are correct, 8,9 so that the D_2 and D_{2d} symmetry structures in Fig. 1 are essentially isoenergetic, the anticipated thermodynamic isomer distribution is given by a simple rule: The equilibrium mole fraction of each molecule in a set of isoenergetic and interconverting C_n isomers is inversely proportional to the order of its molecular point group.²² This rule has recently been derived²² for the general case of an arbitrary number of isoenergetic C_n fullerene isomers interconverting under the Stone-Wales²³ pyracylene transformation, but we shall now re-derive it for clarity in the context of C_{84} .

One interesting feature of the D_2 and D_{2d} symmetry fullerenes in Fig. 1 is that they are directly related by the Stone-Wales transformation.²⁴ Such a direct transformation is not actually required by the general argument,22 but it does make the present analysis easier. The transforming pyracylene patch of two hexagons and two pentagons has site symmetry C_2 in both isomers,²⁴ with the consequence that two equivalent patches in the D_2 isomer can transform to give the D_{2d} isomer and four equivalent patches in the D_{2d} isomer can transform back to give D_2 . [2 (= 4/2) is the ratio of the orders of the D_2 and C_2 point groups, whereas 4(=8/2) is the ratio of the orders of D_{2d} and C_2]. More detailed considerations reveal that two pyracylene patches in the D_{2d} isomer transform to give the left enantiomer, and two the right enantiomer, of the D_2 isomer.²²

According to the principle of microscopic reversibility the equilibrium concentrations of the D_2 and D_{2d} isomers are related by $K_c = [D_2]/[D_{2d}] = k_1/k_{-1}$, where k_1 and k_{-1} are the total forward $(D_{2d} \rightarrow \overline{D_2})$ and reverse $(D_2 \rightarrow D_{2d})$ reaction rate constants. Since the two isomers are assumed isoenergetic these total rate constants are simply $k_1 = 4k$ and $k_{-1} = 2k$, where k is the generic rate constant of a single $(D_{2d} \rightarrow D_2 \text{ or } D_2 \rightarrow D_{2d})$ pyracylene transformation. Hence $[D_2]/[D_{2d}] = 2/1$ at equilibrium as required. On a more detailed level the equilibrium consists of equal proportions of the D_{2d} isomer and the left and right enantiomers of D_2 .²² However, since the left and right D_2 enantiomers have identical ¹³C NMR patterns, the observed result is simply a 2:1 mixture of D_2 and D_{2d} .

When combined with the semi-empirical calculations described above, this elementary derivation of an entropic 2:1 isomer distribution suggests an unequivocal assignment of the C₈₄ NMR pattern in Fig. 2 to the (almost) isoenergetic D_2 and D_{2d} symmetry ground-state isomers in Fig. 1. More-

over the fact that the same NMR spectrum is obtained in several laboratories suggests that the experimental C₈₄ product is determined by thermodynamic considerations and is insensitive to the detailed kinetics of the formation process.

The C₈₄ result may also have important implications for other higher fullerenes. In general, since the number of potentially stable C_n isomers increases rapidly with n, one might expect entropic distributions of several essentially isoenergetic isomers to become the rule beyond C₈₄. Such distributions will be characterised by large numbers of ¹³C NMR lines of more or less equal intensity (because, in competition with the above symmetry rule for the isomer distribution, the average number of equivalent atoms in a molecule increases in proportion to its point-group order). But perhaps the most important conclusion we can draw from the C_{84} story is that scattered pieces of experimental, $^{2-4,20,25}$ theoretical 7,22 and semi-empirical 8,9 evidence have all contributed fruitfully to the solution of the structural puzzle.

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