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Møller-Plesset third order calculations with large basis sets

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The value of the Møller-Plesset third order calculations is examined. An efficient method for the evaluation of the gradient of the MP3 energy is reported, and it has been programmed for both restricted and unrestricted Hartree-Fock wave functions. Large basis set calculations (TZ2P or better) are reported for the optimization of geometries and the determination of harmonic frequencies (which are obtained by finite differences of analytic gradients). The molecules selected are NH₂, PH₂, AsH₂, H₂O, NH₃, H₂CO, HCN, and C₂H₂. For the closed shell systems, the RMP3 predictions for bond lengths are inferior ($\approx 0.006 \text{ Å}$) to RMP2 predictions ($\approx 0.003 \text{ Å}$) for single bonds, and for multiple bonds the RMP3 bond lengths are too short by approximately the same amount ($\approx 0.01 \text{ Å}$) that RMP2 are too long. For the open shell systems, the UMP3 geometrical parameters show only a marginal improvement over UMP2, except for PH2 where the bond length error is reduced to 0.003 Å. The results for harmonic frequencies show a similar comparison between MP2 and MP3. On the basis of this experience, it appears that large scale calculations at the MP3 level are not recommended; MP2 calculations with a large basis set are much cheaper and provide results with a similar, if not superior, accuracy.

I. INTRODUCTION

The accurate prediction of the properties of small and medium size molecules is now a goal within the grasp of the quantum chemist. He has available a range of methods, of a varying level of sophistication. In this paper we shall look at the perturbation approach for the inclusion of electron correlation, commonly called the Møller-Plesset (MP) method.1

The MP approach has been popularized by its availability in the GAUSSIAN series of programs, specifically at second, third, and fourth order,² MP2, MP3, and MP4. For closed shell systems, one proceeds from the restricted Hartree-Fock approximation, and thus the series RMP2, RMP3, and RMP4 is obtained. For open shell systems, the unrestricted Hartree-Fock³ approximation must be used, leading to the series UMP2, UMP3, and UMP4. For a review of the results which may be obtained with those methods, see Ref. 2.

Recently, it has been shown that at the RMP2 level, substantially improved accuracy is achieved if basis sets larger than the recommended 6-31G* (or 6-31G**) are used. If one performs calculations with a TZ2P basis set, then there are sufficient calculations to show that for single bonds an accuracy of ≈ 0.003 Å may be achieved, and for multiple bonds ≈ 0.01 Å. This represents an improvement by a factor of 2 or 3 over 6-31G* basis calculations. With this RMP2 method, automatic optimization of structures through analytic gradients is possible at little extra cost over the energy calculation. Simandiras et al.4 have also shown that RMP2 analytic second derivatives may be evaluated efficiently; these are now available in the quantum chemistry package CADPAC.⁵ The results of calculations by this approach have shown that it is possible to achieve a 2% average accuracy in predicted harmonic frequencies, when a TZ2P (+f)basis set is used. This is a substantial improvement over the

 \approx 10% accuracy which is obtained by the RHF method and brings quantum chemistry into the realm of useful predictive powers, without the need for any scaling procedures.⁶

MP2 is the simplest scheme for the inclusion of electron correlation. No iterative procedure is necessary. Its success outlined above is rather remarkable. The question must then be asked whether this accuracy can be maintained as the level of sophistication is improved. The next level of sophistication is MP3 which is an n^6 procedure. It will be cheaper than any of the other improved levels of sophistication, all of which require iterations; we may mention CISD,7 CEPA-n,8 CPF,9 and CCSD,10 which are single-reference based and limited to single and double excitations. With the exception of CISD, all these methods are size consistent 11 which is now recognized to be more important than obeying the variational upper bound principle.

The purpose here is an in-depth examination of MP3. We shall investigate both RMP3 and UMP3. Before commencing, we recall some evidence from the convergence of the MPn series. 12 The convergence of the RMPn series was erratic, especially at distorted geometries. In our calculations, we find that for H₂O, with a TZ2P basis set, the energy lowering from RMP2, RMP3, and RMP4, at equilibrium geometry, is 0.265, 0.003, and 0.018 hartree. In other words RMP3 does not appear to give the improvement that RMP2 and RMP4 achieve. At $1.5r_e$ with C_{2v} symmetry the MP3 contribution was found to raise the energy. 12 Of course, we note that MP2 and MP3 deal only with double excitations, whereas MP4 introduces single, triple, and quadruple excitations. On the other hand, the convergence of the UMPn series is much more smooth, although at distorted geometries the series is slowly convergent, predominantly it appears due to spin contamination. For NH₂, at r_e with the TZ2P basis set, we find that the energy lowering from the UMP2, UMP3, and UMP4 calculations is 0.196, 0.016, and 0.011 hartree. We thus anticipate for the later results in this paper that the UMP3 and RMP3 calculations exhibit a different behavior.

Before we can proceed with a discussion of our results, we shall outline the method we have used to evaluate MP3 gradients for the optimization of geometries. There have been two earlier papers on the evaluation of these gradients; the first by Gauss and Cremer, ¹³ which when their Appendix is included, gives an efficient way to proceed. The second paper by Bartlett and co-workers, ¹⁴ published almost simultaneously, outlines the same procedure, and presents some DZ basis set calculations on H₂O. There is no evidence in these papers to suggest that efficient algorithms have been written.

For a method to be viable and successful, it must be capable of being used with large basis sets (i.e., more than 100 basis functions). Even on today's computers, that means the programs must operate "out-of-core." The best algorithms will take advantage of available vector or array facilities. For these reasons we present in the next section our approach for the evaluation of UMP3 (and RMP3) gradients, and we discuss some of the complexities which were met in the efficient evaluation.

II. THE EVALUATION OF MP3 GRADIENTS

The notation we use is the same as that presented in our earlier papers on the derivation of MP2 analytic second derivatives¹⁵ and MP2 formulas for infrared intensities.¹⁶ Thus a perturbed orbital is represented, for example, through

$$\phi_i^{(x)} = \sum_p U_{pi}^x \phi_p + \phi_i^x, \qquad (1)$$

where ϕ_i^x denotes the evaluation of ϕ_i with differentiated basis functions, and U_{pi}^x are solutions of the coupled perturbed Hartree-Fock (CPHF) equations.¹⁷ The orthonormality constraint yields

$$U_{nq}^{x} + U_{qn}^{x} + S_{nq}^{x} = 0. (2)$$

We shall derive the theory for the UMP3 gradient, superscripts of α or β will identify the spin of the orbitals being discussed, i, j... denote occupied orbitals, a, b... denote virtual orbitals, and p, q... refer to all orbitals.

In the UHF framework, the coupled equations (CPUHF) take the form

$$(\epsilon_a^{lpha} - \epsilon_i^{lpha})U_{ai}^{lpha x} + \sum_{bj}^{lpha} A_{aibj}^{lpha a}U_{bj}^{lpha x} + \sum_{bj}^{eta} A_{aibj}^{lpha eta}U_{bj}^{eta x}$$

$$= -\epsilon_{ai}^{\alpha x} + S_{ai}^{\alpha x} \epsilon_i + \frac{1}{2} \sum_{kj}^{\alpha} A_{aikj}^{\alpha \alpha} S_{kj}^{\alpha x} + \frac{1}{2} \sum_{kj}^{\beta} A_{aikj}^{\alpha \beta} S_{kj}^{\beta x},$$
(3)

where

$$A_{pqrs}^{\alpha\alpha} = 2(pq|rs)^{\alpha\alpha} - (pr|qs)^{\alpha\alpha} - (ps|qr)^{\alpha\alpha},$$

$$A_{pqrs}^{\alpha\beta} = 2(pq|rs)^{\alpha\beta},$$
(4)

and

$$\epsilon_{pq}^{\alpha} = h_{pq}^{\alpha} + \sum_{i}^{\alpha} \left[(pq|jj)^{\alpha\alpha} - (pj|qj)^{\alpha\alpha} \right] + \sum_{i}^{\beta} (pq|jj)^{\alpha\beta}.$$
(5)

Equation (3), together with its β analogs, may be summar-

ized as

$$\begin{pmatrix} \mathbf{H}^{\alpha\alpha} & \mathbf{H}^{\alpha\beta} \\ \mathbf{H}^{\beta\alpha} & \mathbf{H}^{\beta\beta} \end{pmatrix} \begin{pmatrix} \mathbf{U}^{\alpha x} \\ \mathbf{U}^{\beta x} \end{pmatrix} = \begin{pmatrix} \mathbf{b}^{\alpha x} \\ \mathbf{b}^{\beta x} \end{pmatrix}$$
(6)

and they may be solved for the independent occupied-virtual block of the U^x matrix. The dependent members (occupied-occupied or virtual-virtual) may be determined through

$$U_{pq}^{\alpha x} = \frac{1}{(\epsilon_{p}^{\alpha} - \epsilon_{q}^{\alpha})} \left(-\epsilon_{pq}^{\alpha x} + S_{pq}^{\alpha x} \epsilon_{q}^{\alpha} + \sum_{bj}^{\alpha} A_{pqbj}^{\alpha \alpha} U_{bj}^{\alpha x} + \sum_{bj}^{\beta} A_{pqbj}^{\alpha \beta} U_{bj}^{\beta x} - \frac{1}{2} \sum_{kj}^{\alpha} A_{pqkj}^{\alpha \alpha} S_{kj}^{\alpha x} - \frac{1}{2} \sum_{kj}^{\beta} A_{pqkj}^{\alpha \beta} S_{kj}^{\beta x} \right)$$

$$(7)$$

but this is not satisfactory in the case of nearly degenerate orbitals ($\epsilon_p=\epsilon_q$). We use the alternative scheme

$$U_{pq}^{x} = U_{qp}^{x} = -\frac{1}{2}S_{pq}^{x} \tag{8}$$

but we must then take care that the matrix ϵ is not diagonal at the perturbed geometry; i.e., $\epsilon_{pq}^{(x)} \neq 0$.

In a spin-orbital formalism, and a nondiagonal representation of the Fock matrix (i.e., not canonical orbitals) the MP3 energy may be written

$$E_{3} = -\sum_{nmpq} \langle 0|H|D_{n}\rangle (H_{0})_{nm}^{-1}$$

$$\times \langle D_{m}|H|D_{q}\rangle (H_{0})_{qp}^{-1}\langle D_{p}|H|0\rangle$$

$$= -\mathbf{I}\cdot\mathbf{H}_{0}^{-1}\cdot\mathbf{H}\cdot\mathbf{H}_{0}^{-1}\cdot\mathbf{I},$$
(9)

where

$$(H_0)_{nm}^{-1} = \langle D | H_0 - E_0 | D \rangle_{nm}^{-1}, \tag{10}$$

 D_m , D_n denote double replacement determinants. Direct differentiation of Eq. (9) yields

$$\frac{dE_3}{dx} = E_3^{(x)} = -2\mathbf{I}^{(x)} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{H} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{I}$$
$$-\mathbf{I} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{H}^{(x)} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{I}$$
$$+2\mathbf{I} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{H}_0^{(x)} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{H} \cdot \mathbf{H}_0^{-1} \cdot \mathbf{I}. \quad (11)$$

We can now expand Eq. (11) to yield the following expression:

$$E_{3}^{(x)} = -2 \sum_{ijab} (ia|jb)^{(x)} p_{ij}^{ab} - \sum_{abcd} (ac|bd)^{(x)} t_{ij}^{ab} t_{ij}^{cd}$$

$$- \sum_{kilj} (ki|lj)^{(x)} t_{ij}^{ab} t_{kl}^{ab}$$

$$- \sum_{ikac} t_{ij}^{ab} t_{kj}^{cb} [(ia|kc)^{(x)} - (ac|ki)^{(x)}]$$

$$+ \sum_{ikac} Y_{ac} \epsilon_{ac}^{(x)} - \sum_{ik} Y_{ik} \epsilon_{ik}^{(x)}, \qquad (12)$$

where

$$p_{ij}^{ab} = 2r_{ij}^{ab}/D_{ij}^{ab},$$

$$r_{ij}^{ab} = \frac{1}{8} \sum_{cd} t_{ij}^{cd} (ab \| cd) + \frac{1}{8} \sum_{kl} t_{kl}^{ab} (kl \| ij)$$

$$+ \frac{1}{4} \sum_{kc} \left[t_{ik}^{ac} (bk \| jc) + t_{ik}^{cb} (ak \| jc) + t_{kl}^{ab} (bk \| ic) + t_{kl}^{cd} (ak \| ic) \right],$$

$$(14)$$

$$Y_{ac} = 4 \sum_{ijb} r_{ij}^{cb} t_{ij}^{ab} / D_{ij}^{cb},$$

 $Y_{ik} = 4 \sum_{i=1}^{n} r_{kj}^{ab} t_{ij}^{ab} / D_{kj}^{ab},$

and

$$t_{ij}^{ab} = [(ia|jb) - (ib|ja)]/D_{ij}^{ab},$$

$$D_{ii}^{ab} = \epsilon_a + \epsilon_b - \epsilon_i - \epsilon_i.$$
(16)

On substitution of the explicit form for the differentiated orbitals, one obtains

$$E_3^{(x)} = \sum_{i} U_{ai}^x L_{ai}^{MP3} + W^x, \tag{17}$$

where W^* involves derivative one and two electron integrals and the derivative overlap matrix. The "Lagrangian" L is given by

$$L_{ai}^{MP3} = \sum_{bc} Y_{bc} A_{bcai} - \sum_{jk} Y_{jk} A_{jkai}$$

$$-4 \sum_{jb} \left[\sum_{c} p_{ij}^{cb} (ac|jb) + \sum_{k} p_{kj}^{ab} (ki|jb) \right]$$

$$+4 \sum_{cbd} f_{ab}^{cd} (ic|bd) - 4 \sum_{jkl} f_{ij}^{kl} (ka|lj)$$

$$+2 \sum_{kc} \left[\sum_{j} f_{jk}^{ac} \{ (ij|kc) - (ic|kj) \} \right]$$

$$-\sum_{kc} f_{ik}^{bc} \{ (ba|kc) - (bc|ka) \} , \qquad (18)$$

where

$$f_{ab}^{cd} = \frac{1}{4} \sum_{kl} t_{kl}^{ab} t_{kl}^{cd}, \quad f_{ij}^{kl} = \frac{1}{4} \sum_{cd} t_{ij}^{cd} t_{kl}^{cd},$$

$$f_{jk}^{ac} = \sum_{ll} t_{jl}^{ad} t_{kl}^{cd}.$$
(19)

The evaluation of L can be arranged in terms of matrix multiplication operations, hence the code can be vectorized, making the evaluation highly efficient. This is the reason for a full 4-index transformation, so that r_{ij}^{ab} could be efficiently processed. Within the unrestricted framework, the spin-orbital formula (12) can be separated into its space-spin components, using

$$(t_{ij}^{ab})^{\alpha\alpha} = \frac{\left[(ia|jb)^{\alpha\alpha} - (ib|ja)^{\alpha\alpha} \right]}{\epsilon_a^{\alpha} + \epsilon_b^{\alpha} - \epsilon_i^{\alpha} - \epsilon_j^{\alpha}},$$

$$(t_{ij}^{ab})^{\alpha\beta} = \frac{(ia|jb)^{\alpha\beta}}{\epsilon_a^{\alpha} + \epsilon_b^{\beta} - \epsilon_i^{\alpha} - \epsilon_j^{\beta}}$$
(20)

and similarly for the β terms.

The result is a formula

$$E_{3}^{(x)} = \sum_{ai}^{\alpha} L_{ai}^{\alpha} U_{ai}^{\alpha x} + \sum_{ai}^{\beta} L_{ai}^{\beta} U_{ai}^{\beta x} + W^{x}, \qquad (21)$$

where L_{ai}^{α} and L_{ai}^{β} are the α and β forms of the Lagrangian matrix (18).

The final form for the MP3 gradient is obtained from Eq. (21) through application of the idea of Handy and Schaefer, ¹⁸ viz.,

$$\mathbf{L}^{T}\mathbf{U}^{x} = \mathbf{L}^{T}(\mathbf{H}^{-1} \cdot \mathbf{b}^{x})$$

$$= [(\mathbf{H}^{-1})^{T} \cdot \mathbf{L}]^{T} \cdot \mathbf{b}^{x}$$

$$= \mathbf{Z}^{T} \cdot \mathbf{b}^{x}, \tag{22}$$

where

(15)

$$\mathbf{H}^T \cdot \mathbf{Z} = \mathbf{L}. \tag{23}$$

In unrestricted terms this becomes

$$\begin{pmatrix} \mathbf{H}^{\alpha\alpha} & \mathbf{H}^{\alpha\beta} \\ \mathbf{H}^{\beta\alpha} & \mathbf{H}^{\beta\beta} \end{pmatrix} \begin{pmatrix} \mathbf{Z}^{\alpha} \\ \mathbf{Z}^{\beta} \end{pmatrix} = \begin{pmatrix} \mathbf{L}^{\alpha} \\ \mathbf{L}^{\beta} \end{pmatrix}. \tag{24}$$

This only requires the solution of one set of equations to obtain \mathbb{Z} , instead of the 3N sets, since all the matrices involved are independent of the nuclear perturbations. The UMP3 gradient can therefore be rewritten as

$$E_{3}^{(x)} = \sum_{ai}^{\alpha} Z_{ai}^{\alpha} b_{ai}^{\alpha x} + \sum_{ai}^{\beta} Z_{ai}^{\beta} b_{ai}^{\beta x} + W^{x}.$$
 (25)

The importance of this is that the derivative integrals involved with the b* matrix do not have to be stored or transformed. After collecting terms in Eq. (25), the gradient expression can be written in terms of one and two particle coefficient matrices ("effective density matrices") running over the molecular orbitals. In terms of spin orbitals this yields

$$E_{3}^{(x)} = \sum_{pq} g_{pq} S_{pq}^{x} + \sum_{pq} \gamma_{pq} h_{pq}^{x} + \sum_{pqrs} \Gamma_{pqrs} (pq|rs)^{x}.$$
(26)

The one and two particle coefficient matrices can be defined from

$$\sum_{bc} Y_{bc} h_{bc}^{x} - \sum_{jk} Y_{jk} h_{jk}^{x} - \sum_{ai} Z_{ai} h_{ai}^{x}$$

$$\equiv \sum_{pq} \gamma_{pq} h_{pq}^{x}, \quad -2 \sum_{ijab} p_{ij}^{ab} (ia|jb)^{x} - \sum_{abcd} f_{ab}^{cd} (ac|bd)^{x}$$

$$- \sum_{ijkl} f_{ij}^{kl} (ki|lj)^{x} - \sum_{ikac} f_{ik}^{ac} [(ia|kc)^{x} - (ac|ki)^{x}]$$

$$+ \sum_{pq} \gamma_{pq} \sum_{j} [(pq|jj)^{x} - (pj|qj)^{x}]$$

$$\equiv \sum_{pqrs} (pq|rs)^{x},$$
(28)

$$2\sum_{ijab} \left[\sum_{k} p_{ij}^{ab}(ka|jb) S_{ki}^{x} + \sum_{c} p_{ij}^{ab}(ic|jb) S_{ac}^{x} + 2\sum_{k} p_{kj}^{ab}(ki|jb) S_{ai}^{x} \right] + 2\sum_{abcd} \left[\sum_{e} f_{ab}^{cd}(ac|bd) S_{ea}^{x} + 2\sum_{i} f_{ab}^{cd}(ic|bd) S_{ai}^{x} \right]$$

$$+ 2\sum_{ijkl} \sum_{m} f_{ij}^{kl}(km|lj) S_{mi}^{x} + \sum_{ikac} \left[\sum_{b} f_{ik}^{ac} \{ (bi|kc) - (bc|ki) \} S_{ba}^{x} + \sum_{j} f_{ik}^{ac} \{ (ja|kc) - (ac|kj) \} S_{ji}^{x} \right]$$

$$+ 2\sum_{j} f_{jk}^{ac} \{ (ij|kc) - (ic|kj) \} S_{ai}^{x} \right] - \frac{1}{2}\sum_{pqkj} \gamma_{pq} A_{pqkj} S_{kj}^{x} - \frac{1}{2}\sum_{pq} \gamma_{pq} (\epsilon_{q} + \epsilon_{p}) S_{pq}^{x} \equiv \sum_{pq} g_{pq} S_{pq}^{x}.$$

$$(29)$$

Backtransforming the density matrices to the AO basis yields the final form for the UMP3 gradient:

$$E_{3}^{(x)} = \sum_{\mu\nu} g_{\mu\nu} S_{\mu\nu}^{x} + \sum_{\mu\nu} \gamma_{\mu\nu} h_{\mu\nu}^{x} + \sum_{\mu\nu\rho\sigma} \Gamma_{\mu\nu\rho\sigma} (\mu\nu|\rho\sigma)^{x}.$$
(30)

To summarize our procedure for the evaluation of a UMP3 gradient:

- (1) UHF-SCF calculation.
- (2) Full 4-index transformation of the integrals into the MO basis.
- (3) Construction L^{α} , L^{β} and the solution of the single set of equations to give \mathbf{Z}^{α} , \mathbf{Z}^{β} .
- (4) Construction of the effective density matrices g, γ , and Γ (both α and β forms) in the MO basis.
- (5) These matrices are back transformed to the AO basis; α and β dependence has now disappeared. They are added to the equivalent matrices from the UHF and UMP2 gradient theory.
- (6) The derivative AO integrals are evaluated, multiplied by the AO density matrices, then summed to give the final result.

As described above, we believe there are advantages in doing a full 4-index transformation, because this enables the r matrix to be defined, and hence allows the vectorized construction of L. The integrals are sorted onto external files such that for a particular p and q, all integrals of the form $J_{(rs)\alpha}^{(pq)\alpha} = (pq|rs)^{\alpha\alpha}$, $K_{(rs)\alpha}^{(pq)\alpha} = (pr|qs)^{\alpha\alpha}$, $J_{(rs)\beta}^{(pq)\alpha} = (pq|rs)^{\alpha\beta}$, and the equivalent β lists, are available.

The results of doing this are:

- (a) It facilitates the use of matrix multiplication on these integral blocks to perform the intermediate summations in the formation of L.
- (b) It avoids the problem of storing some large vector in core, for example, the complete t_{ij}^{ab} vector, because it can be formed as required. This obviously becomes significant for large calculations when memory requirements need to be kept to a minimum.

The UMP2 and UMP3 gradient programs use the derivative integrals and general transformation and symmetry facilities of the Cambridge Analytic Derivatives Package, CADPAC. The timings for our current program for the gradients of UHF, UMP2, UMP3 are in the ratio 1:2.5:6.0.

III. APPLICATIONS

We shall discuss calculations at the RMP2 and RMP3 level on H_2O , NH_3 , H_2CO , and HCN, and at the UMP2 and UMP3 level on NH_2 , PH_2 , and AsH_2 . For these doublet open shell systems $\langle \hat{S}^2 \rangle_0 = 0.76$, so spin contamination is

not important. For the RMP2 calculations, the harmonic frequencies are obtained analytically; for all other (RMP3, UMP2, UMP3) calculations, these frequencies are obtained from finite difference calculations using the gradient codes.

The basis sets used for the first row atoms and H are by now fairly standard. The DZP basis involves the 4s2p for first row atoms, or 2s for H, Dunning¹⁹ contraction of the 9s5p or 4s Huzinaga²⁰ primitive sets, with one set of polarization functions. The TZ2P basis involves the 5s4p for first row atoms, or 3s for H, Dunning contractions of the 10s6p or 5s Huzinaga primitive sets. The larger basis sets are constructed from the 8s6p contraction of van Duijneveldt's 13s8p primitive set, ²¹ and the 6s contraction of the 10s primitives for H. The polarization function exponents are for H: 1p, 1.0; 2p, 1.5, 0.5; 3p, 1.8, 0.6, 0.2. For C 1d 0.8; 2d, 1.2, 0.4; 3d, 1.8, 0.6, 0.2; 1f, 1.0. For N 1d 0.8; 2d, 1.35, 0.45; 3d, 2.4, 0.8, 0.266; 1f; 1.0. For O 1d 0.9; 2d, 1.35, 0.45; 3d, 2.7, 0.9, 0.3; 1f 1.0.

For phosphorus, the DZP basis involves the Dunning and Hay²² 6s4p contraction of the 11s7p Huzinaga²³ primitive set. The TZ2P basis involves the 9s6p contraction of the 12s9p McClean and Chandler²⁴ primitive set. The 11s8p contraction is built from the 9s6p contraction by adding more diffuse s and p functions formed by following the approximate geometrical progression of the exponents. The polarization function exponents are 1d, 0.5; 2d, 0.75, 0.25; 3d, 0.85, 0.3, 0.1; 1f, 0.6.

The arsenic calculations use the DZP basis which involves the 10s6p2d contraction of a Dunning 13s9p5d primitive set. The TZ2P basis involves the 13s9p3d contraction of a Dunning 20s14p6d primitive set.²⁵ The exponents of the f polarization functions are 1f, 0.5; 2f, 0.75 0.25.

A. NH₂

Previous studies of NH₂ using unrestricted techniques have generally not used large basis sets. For example, Pople and co-workers² have published optimized geometries of NH₂ at UHF, UMP2, UMP3, and UMP4 levels of theory using the 6-31G* basis set and also at UMP4 with a 6-311G** basis set. Their results are in reasonable agreement with experiment, but the studies presented here show the effect of increasing the basis set size and also calculate the corresponding harmonic vibrational frequencies.

The molecular properties of NH_2 at various levels of theory using a selection of basis sets are given in Table I. The RHF and UHF results using a large 8s6p4d 1f/6s3p basis set (close to the HF basis set limit) are very similar and display the typical errors for SCF optimized geometries. The NH bond length, $r_{\rm NH}$, is 0.016 Å too short and the angle consid-

TABLE I. SCF, UMP2, and UMP3 optimized geometries and harmonic frequencies for NH₂.

| Method | Basis set | r _{NH} | θ_{HNH} | ω_1 | ω_2 | ω_3 |
|--------|----------------|-----------------|----------------|------------|------------|------------|
| RHF | 8s6p4d 1f/6s3p | 1.008 | 104.8 | 3620 | 1643 | 3708 |
| UHF | 8s6p4d 1f/6s3p | 1.009 | 104.8 | 3605 | 1639 | 3693 |
| UMP2 | 6-31G** | 1.029 | 103.3 | | | |
| | DZP | 1.026 | 103.3 | 3492 | 1569 | 3620 |
| | 5s4p2d /3s2p | 1.020 | 103.3 | 3465 | 1555 | 3571 |
| | 8s6p3d /6s2p | 1.019 | 103.3 | 3454 | 1555 | 3558 |
| | 8s6p4d 1f/6s3p | 1.020 | 103.4 | 3455 | 1544 | 3563 |
| UMP3 | 6-31G** | 1.031 | 103.2 | | • • • | |
| | DZP | 1.027 | 103.2 | 3477 | 1576 | 3592 |
| | 5s4p2d /3s2p | 1.020 | 103.2 | 3453 | 1565 | 3547 |
| | 8s6p3d/6s2p | 1.019 | 103.2 | 3443 | 1561 | 3535 |
| | 8s6p4d 1f/6s3p | 1.020 | 103.4 | 3447 | 1556 | 3542 |
| UMP4 | 6-31G** | 1.034 | 102.9 | | • • • | ••• |
| | 6-311G** | 1.030 | 101.6 | | • • • | • • • |
| CISD | DZPb | 1.029 | 103.1 | ••• | • • • | |
| Expt.c | | 1.024 | 103.3 | ••• | ••• | ••• |

^a From Ref. 2.

erably too large by 1.5° compared to the experimental values. At the UMP2 level, the 6-31G* geometries are much improved over UHF (the bond length is now in error by + 0.005 Å), with the DZP basis the bond length is reduced to an error of + 0.002 Å, and with the large basis, 8s6p4d 1f/6s3p, the bond is 0.004 Å too short, with bond angle being correct to within 0.1°. At the UMP3 level the geometries are very similar to the equivalent UMP2 values with the same basis set. It is also noted that the CISD geometry²⁶ with a DZP basis set is similar to the UMP3 DZP geometry. The UMP4 6-31G* geometry² is different to that at UMP2 6-31G* in such a way that if this difference is carried over to large basis sets the UMP4 geometry will almost be coincident with experiment.

Experimental harmonic frequencies are not available, however from previous calculations, SCF harmonic frequencies close to the basis set limit are typically 10% greater than the "true" harmonic values. It can be seen from Table I that the UMP2 DZP frequencies are significantly lower than the UHF values calculated close to the basis set limit. The UMP2 values reduce as the basis set increases and it is expected that large basis set values will be good estimates of the harmonic frequencies. We base this on our experience with RMP2.4 The UMP3 harmonic frequencies are similar to the UMP2 values using the same basis set. There is a small reduction in the symmetric and asymmetric stretching modes ω_1 and ω_3 at UMP3 compared to UMP2, while the ω_2 bending frequency is increased by an even smaller magnitude. For example, employing an 8s6p3d /6s2p quality basis set, ω_1 and ω_3 are reduced by 11 and 23 cm⁻¹, respectively, and ω_2 is increased by 6 cm⁻¹. For the case of NH₂ the results predicted at the UMP3 level of theory do not display any significant differences compared to those predicted by UMP2.

TABLE II. SCF, UMP2, and UMP3 optimized geometries and harmonic frequencies for PH..

| Method | Basis set | r_{PH} | $\theta_{	ext{HPH}}$ | ω_1 | ω_2 | ω_3 |
|--------|---|-------------------|----------------------|------------|------------|------------|
| RHF | 11s8p3d 1f/6s3p | 1.409 | 93.7 | 2517 | 1223 | 251 |
| UHF | 11 <i>s</i> 8 <i>p</i> 3 <i>d</i> 1 <i>f</i> /6 <i>s</i> 3 <i>p</i> | 1.409 | 93.7 | 2511 | 1220 | 251 |
| UMP2 | DZP | 1.411 | 92.3 | 2502 | 1188 | 251 |
| | 9s6p2d /3s2p | 1.412 | 92.1 | 2480 | 1150 | 248 |
| | 11s8p2d/6s2p | 1.411 | 92.0 | 2471 | 1142 | 248 |
| | 11s8p3d/6s3p | 1.411 | 92.0 | 2461 | 1154 | 246 |
| | 11 <i>s</i> 8 <i>p</i> 3 <i>d</i> 1 <i>f</i> /6 <i>s</i> 3 <i>p</i> | 1.412 | 91.9 | 2459 | 1151 | 246 |
| UMP3 | DZP | 1.413 | 92.2 | 2488 | 1194 | 249 |
| | 9s6p2d/3s2p | 1.414 | 92.0 | 2455 | 1146 | 246 |
| | 11s8p2d / 6s2p | 1.414 | 91.9 | • • • | • • • | |
| | 11s8p3d/6s3p | 1.414 | 91.8 | | | |
| | $11s8p3d\ 1f/6s3p$ | 1.415 | 91.9 | ••• | • • • | •• |
| Expt.* | | 1.418 | 91.7 | | | • • |

^{*}From Ref. 35.

B. PH₂

Previous theoretical studies of PH₂ ²⁷ have not used large basis sets. The molecular properties of PH₂ at various levels of theory using a selection of basis sets are given in Table II. The large basis set UHF optimized geometry predicts the equilibrium $r_{\rm PH}$ to be 0.009 Å too short and $\theta_{\rm HPH}$ 2.0° too large compared to experiment. At the UMP2 level the values are improved at the DZP level compared to UHF, and using the large 11s8p3d 1f/6s3p basis set $r_{\rm PH}$ is too short by 0.006 Å and $\theta_{\rm HPH}$ is too large by 0.2°. The UMP3 geometries improve upon the UMP2 geometries using the same basis set. We see that for the largest basis set, 11s8p3d 1f/6s3p, the $r_{\rm PH}$ bond length is only 0.003 Å too short.

For the harmonic frequency predictions, we observe that the trends are rather similar to those for NH₂, there being no significant change between the UMP2 and UMP3 values. We suggest that our best values, from the 9s6p2d/3s2p basis, of 2455, 1146, and 2463 cm⁻¹ will be very close to the exact harmonic frequencies.

C. AsH₂

The molecular properties of AsH₂ are shown in Table III. The equilibrium structures calculated at the RHF and UHF levels are similar for the three basis sets employed. At the UMP2 level, the bond length decreases from 1.520 Å using a DZ basis set to 1.492 Å with the TZ2P quality basis set, the equilibrium bond angle similarly closing up from 92.3° at DZ to 90.6° at TZ2P. The unusual feature of the optimized geometries is that the inclusion of electron correlation has the effect of reducing the bond length. CISD calculations²⁸ also show the same effect. At the TZ2P level there is a reduction of 0.016 Å from UHF to UMP2. The bond angle also closes by 2.2°, however this follows the trend of NH₂ and PH₂ where there was a 1.5° reduction. For NH₂ there was a bond length increase of 0.01 Å and for PH₂ of 0.003 Å in line with UHF bond length predictions being too short and UMP2 correcting for this. However, in the case of AsH_2 the involvement of a full complement of d electrons

^bFrom Ref. 26.

^c From Ref. 34.

TABLE III. Optimized geometries and harmonic frequencies for AsH₂.

| Method | Basis set | r _{AsH} | $	heta_{	extsf{HAsH}}$ | ω_1 | ω_2 | ω_3 |
|--------|-----------|------------------|------------------------|------------|------------|------------|
| RHF | DZ | 1.511 | 93.1 | 2297 | 1126 | 2291 |
| | DZP | 1.503 | 93.3 | 2349 | 1129 | 2340 |
| | TZ2P | 1.508 | 92.9 | 2344 | 1115 | 2340 |
| UHF | DZ | 1.511 | 93.0 | 2291 | 1126 | 2284 |
| | DZP | 1.503 | 93.2 | 2348 | 1131 | 2338 |
| | TZ2P | 1.508 | 92.8 | 2346 | 1125 | 2341 |
| UMP2 | DZ | 1.520 | 92.3 | 2218 | 1048 | 2210 |
| | DZP | 1.486 | 92.2 | 2371 | 1068 | 2371 |
| | TZ2P | 1.492 | 90.6 | 2293 | 1145 | 2367 |
| CISD | DZP | 1.488 | 92.3 | 2368 | 1084 | 2365 |

has a significant effect. These d orbitals cause the bond angle to be closer to 90° as predicted by UMP2 TZ2P.

The harmonic frequencies of AsH₂ show a rather erratic behavior. There is certainly evidence in Table III, that with a TZ2P basis set the RHF and UHF calculations have converged to ω_1 (symmetric stretch) ≈ 2340 , ω_2 (bend) ≈ 1120 , and ω_3 (asymmetric stretch) ≈ 2345 cm⁻¹. Introduction of electron correlation has a varied effect on these frequencies, dependent on the basis set, e.g., ω_2 changes by -63 cm⁻¹ with DZP and by +20 cm⁻¹ with TZ2P set. Our only option is to suggest that our TZ2P values of 2293, 1145, and 2367 cm⁻¹ are nearest the truth, but in this instance we do not believe these values have converged with respect to either basis set or method. However, we report these calculations because there appears to be no other calculations on AsH₂ in the literature.

D. H₂O

H₂O has been studied extensively experimentally and theoretically. Experimental geometries and harmonic frequencies are well known and believed to be highly accurate. A complete list of theoretical H₂O studies will not be given, but in view of the current investigations reference is given to results predicted by Møller-Plesset perturbation theory^{2,29,30} and CI³¹ methods.

The molecular properties of H₂O at various levels of theory using a series of basis sets are given in Table IV. The RMP2 geometrical parameters are well known and have been shown to be highly accurate.30 At the RMP3 level values predicted via use of a DZP basis set show an equilibrium bond length r_{OH} that is closer to experiment than the corresponding RMP2 DZP bond length, and a bond angle θ_{HOH} that is slightly inferior. Increasing the basis set size has the effect of reducing the RMP3 bond length and in the limit of saturating the s, p, and d shells and incorporating an f function the RMP3 equilibrium bond length of 0.954 Å is too short by 0.004 Å and the bond angle of 104.9° is too large by 0.4°. This in in considerably poorer agreement with experiment than the RMP2 limiting values of 0.959 Å and 104.4°. It should be noted that the CISD DZP optimized geometry as predicted by Lee et al.31 is similar to the RMP3 DZP geometry. They also performed a CISDTQ DZP calculation resulting in a predicted geometry that is different to the

TABLE IV. Optimized geometries and harmonic frequencies for H₂O.

| Method | Basis set | r _{OH} | θ_{HOH} | ω_1 | ω_2 | ω_3 |
|--------------------|---------------------|-----------------|-----------------------|------------|------------|------------|
| RHF | 8s6p4d 2f /6s3p2d* | 0.940 | 106.3 | 4130 | 1746 | 4231 |
| RMP2 | 6-31G*b | 0.969 | 104.0 | 3772 | 1737 | 3916 |
| | DZP | 0.962 | 104.5 | 3913 | 1665 | 4059 |
| | 5s4p2d/3s2p | 0.958 | 104.5 | 3859 | 1641 | 3985 |
| | 8s6p3d /6s2p | 0.957 | 104.5 | 3842 | 1628 | 3968 |
| | 8s6p4d 1f/6s3p | 0.959 | 104.4 | 3839 | 1629 | 3966 |
| RMP3 | 6-31G*b | 0.967 | 104.3 | ••• | ••• | • • • |
| | DZP | 0.959 | 104.8 | 3966 | 1688 | 4089 |
| | 5s4p2d/3s2p | 0.954 | 104.9 | 3930 | 1673 | 4032 |
| | 8s6p3d/6s2p | 0.952 | 104.9 | 3918 | 1662 | 4019 |
| | 8s6p4d 1f/6s3p | 0.954 | 104.9 | 3916 | 1664 | 4017 |
| RMP4 | 6-31G*b | 0.970 | 103.9 | • • • | • • • | • • • |
| RMP4 | 6-311G**b | 0.959 | 102.4 | • • • | • • • | |
| RMP4 | 39-STO ^c | 0.960 | 104.4 | 3823 | 1674 | 3936 |
| CISD | DZP^{d} | 0.959 | 104.8 | 3959 | 1690 | 4082 |
| CISD | TZ2Pd | 0.961 | 104.6 | 3901 | 1668 | 4006 |
| CISDTQ | DZP^{d} | 0.964 | 104.4 | 3887 | 1674 | 4017 |
| Expt. ^e | | 0.958 | 104.5 | 3832 | 1649 | 3942 |

^{*}From Ref. 37.

RMP3 geometry; the bond length is 0.005 Å longer and the angle is reduced by 0.4°. Although the DZP basis set does not contain sufficient functions to allow definite conclusions to be made, this does suggest that higher order excitations are important. Indeed Pople's RMP4 6-31G* geometry² is different from the RMP2 6-31G* geometry such that if this is carried over to large basis sets the RMP4 optimized geometry would be almost coincident with experiment. Unfortunately, there are not enough large basis set MP4 calculations in the literature to be certain that this is valid. However the RMP4 calculation performed by Bartlett et al.²⁹ employing a 39 Slater type orbital basis gives an $r_{\rm OH}$ of 0.960 Å and a θ_e of 104.4° which is a significant improvement on any of the RMP3 results.

Turning now to the harmonic frequencies the RMP2 values using the largest basis set are in error compared to experiment by $+7 (0.2\%), -20 (1.2\%), \text{ and } +24 \text{ cm}^{-1}$ (0.6%) for the symmetric stretch ω_1 , bend ω_2 , and asymmetric stretch ω_3 . Pulay suggests that the small errors in the stretching frequencies are a reflection of the highly accurate geometry.32 The RMP3 harmonic frequencies are higher and in greater error compared to experiment than the RMP2 frequencies with the same basis set. For example using the 8s6p3d/6s3p basis set the RMP3 frequencies are in error by $86 (2.2\%), 13 (0.8\%), and 77 cm^{-1} (2.0\%)$ compared to RMP2 errors of 10 (+0.3%), -19 (-1.2%), and 26 cm^{-1} (+ 0.7%). It could be argued that this reflects the poorer MP3 geometry. The CISD DZP frequencies are similar to the RMP3 DZP values and the CISDTQ frequencies at the DZP level are a significant improvement on either of these, suggesting once again the importance of higher order excitations. Bartlett's full MP429 calculation using the 39

^b From Ref. 2.

^c From Ref. 29.

d From Ref. 31.

^e From Ref. 36.

TABLE V. Optimized geometries for NH₃.

| Method | Basis set | r _{NH} | θ_{HNH} |
|--------|--------------------|-----------------|----------------|
| RHF | 8s6p4d 1f/6s3p1d** | 0.998 | 108.1 |
| RMP2 | 6-31G*b | 1.017 | 106.3 |
| | DZP | 1.014 | 106.7 |
| | 5s4p2d /3s2p | 1.009 | 107.2 |
| | 8s6p3d/6s2p | 1.008 | 107.1 |
| | 13s8p2d /8s2p1d | 1.007 | 107.3 |
| RMP3 | 6-31G*c | 1.017 | 106.2 |
| | DZP | 1.013 | 106.6 |
| | 5s4p2d /3s2p | 1.006 | 107.1 |
| | 8s6p3d /6s2p | 1.006 | 107.0 |
| RMP4 | 6-31G*c | 1.021 | 105.8 |
| Expt.d | | 1.012 | 106.7 |

^{*}From Ref. 37.

Slater type orbital basis yields the most accurate harmonic frequencies in error compared to experiment by -9 (-0.2%), +25 (+1.5%), and -6 cm⁻¹ (+0.2%). This investigation suggests that RMP3 yields poorer results than RMP2 and for highly accurate results there is a need to include the effects of higher order excitations, for example in a full RMP4 calculation. Of course, this conclusion is not definitive since it is based on limited MP4 data.

E. NH₃

The geometrical predictions for NH₃ in Table V show a very similar pattern to those for H₂O. On going from RMP2 to RMP3 with the larger basis sets, the bond length is reduced by ≈ 0.003 Å, so that it is in error by -0.006 Å. The RMP4 results indicate that if it was possible to perform RMP4 with the large basis sets, the predictions would have high accuracy.

F. H₂CO

This molecule is interesting in so far as there is a considerable change in the predicted CO bend length when com-

TABLE VI. Optimized geometries for H₂CO.

| Method | Basis set | r _{co} | r _{CH} | θ_{HCH} |
|--------|---------------|-----------------|-----------------|-----------------------|
| RHF | 5s4p2d /3s2p* | 1.178 | 1.092 | 116.1 |
| RMP2 | 6-31G*b | 1.221 | 1.104 | 115.6 |
| | DZP | 1.223 | 1.102 | 116.2 |
| | 5s4p2d /3s2p | 1.211 | 1.097 | 116.5 |
| | 8s6p3d /6s2p | 1.210 | 1.096 | 116.6 |
| RMP3 | 6-31G*b | 1.210 | 1.104 | 116.0 |
| | DZP | 1.211 | 1.101 | 116.4 |
| | 5s4p2d /3s2p | 1.198 | 1.097 | 116. |
| Expt.° | | 1.203 | 1.099 | 116.5 |

^{*}From Ref. 30.

TABLE VII. Optimized geometries and harmonic frequencies for HCN.

| Method | Basis set | r _{CN} | r _{CH} | ω_{i} | ω_2 | ω_3 |
|--------|----------------|-----------------|-----------------|--------------|------------|------------|
| RHF | 5s4p2d/3s2pa | 1.124 | 1.057 | 3600 | 869 | 2408 |
| RMP2 | 6-31G*b | 1.177 | 1.070 | 3517 | 703 | 2038 |
| | DZP | 1.184 | 1.069 | 3506 | 719 | 1999 |
| | 5s4p2d /3s2p | 1.164 | 1.064 | 3461 | 668 | 2033 |
| | 8s6p3d /6s2p | 1.162 | 1.063 | 3461 | 642 | 2030 |
| | 5s4p2d 1f/3s2p | 1.164 | 1.065 | 3473 | 738 | 2043 |
| RMP3 | 6-31G*b | 1.158 | 1.067 | • • • • | | |
| | DZP | 1.163 | 1.067 | 3539 | 750 | 223€ |
| | 5s4p2d/3s2p | 1.144 | 1.063 | 3488 | 694 | 2260 |
| | 8s6p3d/6s2p | 1.143 | 1.062 | • • • | • • • | ••• |
| | 5s4p2d 1f/3s2p | 1.144 | 1.062 | ••• | • • • | • • • |
| RMP4 | DZP | 1.182 | 1.071 | ••• | | |
| | TZ2P | 1.165 | 1.067 | ••• | • • • | |
| CISD | DZP | 1.165 | 1.068 | 3541 | 755 | 221 |
| Expt.c | | 1.153 | 1.065 | 3442 | 727 | 2129 |

^{*}From Ref. 40.

paring RMP2 and RMP3. The best RMP2 values for r_e are ≈ 0.008 Å too long, but our equivalent RMP3 calculations in Table VI indicate a bond length which is ≈ 0.005 Å too short. This change is similar to that observed with the 6-31G* basis set.

G. HCN

This behavior for multiple bonded systems is again reflected for HCN. We see in Table VII that the CN bond is ≈ 0.01 Å too long for RMP2 with large basis sets, but for RMP3 it is an equivalent amount too short. We have also performed RMP4 calculations, using the GAUSSIAN 82³³ program. The RMP4 predicted value of r_e is close to the RMP2 value using the TZ2P basis set and is 0.012 Å longer than the experimental value! These numbers display an erratic convergence of the bond length in HCN as the level of perturbation expansion increases from RMP2 through RMP3 to RMP4. This is reflected in the CN stretching frequency predictions: with a TZ2P basis the RMP2, RMP3 and experimental values are 2033, 2260, and 2129 cm⁻¹. The behavior for ω_1 (CH) and ω_3 (bend) are similar to those found and discussed above for H_2O .

H. C₂H₂

This is a well-studied molecule, and only a selection of results are included in Table VIII. The CC bond length is reduced, as now expected, by 0.012 Å on going from RMP2 to RMP3 with the TZ2P basis set, with the result that the RMP3 value is 0.007 Å too short. It is noted that more sophisticated methods, such as CPF, give a much better prediction.

It has recently been recognized that there is an anomalous behavior of the ω_4 (π_g) vibration, with respect to basis set.⁴ It is worthwhile commenting on the changes in the frequencies as one proceeds from RMP2 to RMP3. The greatest change, as expected, is in the ω_2 (CC) vibration, from

^b From Ref. 30.

^c From Ref. 2.

^d From Ref. 38.

^b From Ref. 2.

[°] From Ref. 39.

^bFrom Ref. 2.

Geometry from Ref. 41 and frequencies from Ref. 42.

TABLE VIII. Optimized geometries and harmonic frequencies for C₂H₂.

| Method | Basis set | $r_{\rm cc}$ | r _{CH} | ωι | ω_2 | ω_3 | ω_4 | ω_5 |
|--------|-------------------------|--------------|-----------------|-------|------------|------------|------------|------------|
| RHF | 8s6p3d2f/6s3p1d* | 1.179 | 1.059 | 3665 | 2210 | 3556 | 814 | 868 |
| RMP2 | 6-31G*b | 1.218 | 1.066 | 3593 | 2006 | 3516 | 444 | 783 |
| | DZP | 1.226 | 1.067 | 3555 | 1956 | 3470 | 560 | 732 |
| | 5s4p2d /3s2p | 1.208 | 1.060 | 3528 | 1984 | 3438 | 523 | 723 |
| | $8s6p3d\ 1f(=0.8)/6s3p$ | 1.207 | 1.061 | 3537 | 1982 | 3447 | 588 | 746 |
| RMP3 | 6-31G*b | 1.206 | 1.066 | • • • | , | • • • | | |
| | DZP | 1.214 | 1.066 | 3578 | 2074 | 3482 | 605 | 751 |
| | 5s4p2d/3s2p | 1.196 | 1.060 | 3547 | 2096 | 3446 | 566 | 745 |
| CPF | DZPa | 1.222 | 1.069 | 3533 | 2006 | 3443 | 547 | 719 |
| | 5s4p2d/3s2p | 1.204 | 1.063 | 3498 | 2028 | 3404 | 473 | 706 |
| | $5s4p2d\ 1f/3s2p$ | 1.203 | 1.063 | 3522 | 2038 | 3429 | 606 | 754 |
| Expt.c | | 1.203 | 1.062 | 3497 | 2011 | 3415 | 624 | 747 |

^{*}From Ref. 4.

1984 to 2096 cm⁻¹, with the TZ2P basis, compared to the experimental value of 2011 cm⁻¹. The next greatest change is in the ω_4 (π_g) vibration, from 523 to 566 cm⁻¹, compared to the experimental value of 624 cm⁻¹. However, we see that the corresponding value with the CPF method is 473 cm⁻¹. Inclusion of an f function in the basis set has been shown to being the RMP2 and CPF values of ω_4 into good agreement with experiment, suggesting that the corresponding RMP3 value would then be too high. Other frequency changes on proceeding from RMP2 to RMP3 are not significant.

IV. DISCUSSION

All these calculations using the RMP2, UMP2, RMP3, and UMP3 methods with TZ2P quality basis sets allow us to make some tentative conclusions on the value of such calculations.

(i) Our study on the open shell system NH_2 shows very little change in either geometrical parameters or frequencies on proceeding from UMP2 to UMP3. For PH_2 the bond length prediction is improved by ≈ 0.003 Å at UMP3, with a consequential change in the frequencies of ≈ 25 cm⁻¹. Our conclusion has to be that some improvement is achieved with these UMP3 calculations, but it is questionable whether the level of increased accuracy is in proportion to the cost of such calculations. UMP2 with a TZ2P basis seems a reliable method (provided there is no spin contamination!).

(ii) Our study on H_2O reflects one of the major difficulties of MP calculations. Our earlier comments have discussed the strange convergence characteristics of the energy series MP2, MP3, MP4 for H_2O , and it appears that this behavior is present for geometrical parameters and molecular frequencies. The TZ2P, RMP3, bond length is in error by -0.004 Å, compared to the +0.001 Å RMP2 error. The frequency errors for RMP3 are +98, +24, +90 cm⁻¹ compared to the RMP2 errors of +27, -8, +43 cm⁻¹ with the TZ2P basis. Clearly such calculations at the RMP3 level are not as successful as those at the RMP2 level. These comments are supported by the calculations on NH₃.

(iii) There is a considerable difference in the bond length predictions, for RMP2 and RMP3, for mutliple bonds. With the TZ2P basis set the respective errors in the predictions are CO (in H₂CO): +0.008, -0.005 Å; CN (in HCN); +0.011; -0.009 Å; CC (in C₂H₂): +0.005, -0.007 Å. In other words, there is no improvement in the predictive value on proceeding from RMP2 to RMP3, if large basis sets are used. Note that this is in contrast to the general conclusion of Pople and co-workers,² who observe that with a 6-31G* basis, for most molecules MP3 bond lengths are closer to the experimental values than are those at either Hartree-Fock or MP2.

So we are forced from the evidence of these calculations to conclude that the predictive powers of the MP3 approach are no better, and often inferior to those of MP2, if large basis sets are used. Also, MP3 is more expensive than MP2 and it has proved possible to program analytic MP2 second derivatives, whereas analytic MP3 second derivatives would be very difficult to program, and the results here definitely show that it is not worthwhile.

The remaining part of the discussion must be one of philosophy. All our numerical evidence in earlier publications 12 is that the MP method is convergent, that is the energy series MP2, MP3, MP4 ... ultimately converges to the exact eigenvalue of the Schrödinger equation. We must therefore expect properties to converge, both with respect to MPn level and basis set. The RMPn energy series does not converge smoothly, especially at distorted geometries. The UMPn series appears to converge smoothly, but often slowly because of spin contamination. We must therefore not be surprised if there is an erratic convergence of other properties, especially for RMPn.

We believe that it is better to select a method, or level of calculation, and then use as large a basis set as possible with that method. Such an approach appears to remove one uncertainty in the results obtained. The other approach, which is to use a more sophisticated method, with a consequently smaller basis set, leaves a greater uncertainty, especially when it is realized that more sophisticated methods neces-

^bFrom Ref. 2.

From Ref. 43.

sarily need larger basis sets for meaningful results. For all these reasons, we continue to believe that the RMP2 and UMP2 methods have much to offer: the method is simple, large basis sets can be used, and the predictive powers on the strength of many reported calculations are very good, although this may be fortuitous. It is admittedly slightly uncomfortable that RMP3 and UMP3 do not provide better predictions, but at the end of the day, quantum chemistry is designed to serve the chemical community, and if there is a good *ab initio* method which needs no scaling or other factors and which is successful for very many systems, then it must be of value.

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