The Chemistry of Fungi. Part LII. (-)-Sclerotiorin 1334. By G. A. Ellestad and W. B. Whalley

(-)-Sclerotiorin has been converted into aposclerotioramine (II; R = H) from which it follows that (+)- and (-)-sclerotiorin differ only in the relative configuration at C-7.

From a comparison of the C_{10} dienoic acids formed upon alkaline hydrolysis, it has been concluded 2 that (-)-sclerotiorin (I; X = 0), produced by Penicillium hirayamae,3 and (+)-sclerotiorin, isolated as a metabolite of P. sclerotiorum, P. multicolor, and P. implictum, have the same configuration at the asymmetric centre in the side-chain. In view of the relatively low specific rotation associated with the asymmetric centre in this side-chain it seemed desirable to provide collateral evidence for this conclusion.

Thus, we have converted (—)-sclerotiorin successively into (—)-sclerioramine (I; X = NH) and thence into aposclerotioramine (II; R = H). This was characterised as

- Part LI, preceding Paper.
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the di-O-acetate (II; R = Ac) which was identical (m. p., mixed m. p., infrared spectrum, and o.r.d. curve) with a specimen prepared from (+)-sclerotiorin. Hence, in agreement with the work of Gregory and Turner, (+)- and (-)-sclerotiorin differ only in the relative configuration at C-7.

Since ozonolysis of (+)-sclerotiorin yields (+)-\alpha-methylbutyraldehyde which has the same absolute configuration as (-)-2-methylbutanol (cf. Birkinshaw⁵), and this has been shown 6 to possess the (S) absolute configuration, it follows that the absolute configurations of (+)- and (-)-sclerotiorin and of aposclerotioramine are as shown in (I; X =O) and (II), respectively.

EXPERIMENTAL

(-)-Sclerotiorin. This was obtained from the mycelium of P. hirayamae when grown upon a modified Czapek-Dox medium as described for the production of (+)-sclerotiorin, and had m. p. $204-205^{\circ}$, $[\alpha]_{0}^{26}-480^{\circ}$ (c 0.01 in ethanol). (-)-Sclerotioramine, prepared as for (+)-sclerotioramine, was reductively aromatised with zinc and alkali to yield aposclerotioramine 8 which formed the di-O-acetate, 8 m. p. 153° (Found: C, 66·0; H, 6·3; N, 3·4. Calc. for C₂₃H₂₄ClNO₄: C, 66·3; H, 6·3; N, 3·4%).

We are indebted to Professor W. Klyne for the o.r.d. determination, and to the National Institutes of Health of the United States for the award of a Post-Doctoral Fellowship to one of us (G. A. E.).

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[Received, January 18th, 1965.]

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