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Electric field dependence of the capture rate constant of electrons by SF₆ and O₂ dissolved in either liquid Ar or Xe^{a)}

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Bakale $et\ al.^1$ measured the capture rate k of electrons by SF_6 and O_2 in liquid Ar and Xe near the triple point. They interpreted the dependence of k on the electric field F using hot electrons. We will show that this is unnecessary.

The kinetics for the capture of an electron by an impurity is determined by the trapping in its excited state. $^{2-4}$ This should be valid for electrons captured by O_2 or SF_6 dissolved in Ar or Xe. This state will have a binding energy ϵ , a probability of capturing an electron per unit time α , that of re-emission into the conduction band β , and that of making a transition to the ground state γ . Call N the impurity density, n_1 the density of electrons in the conduction band, n_2 those in the excited state, and n_3 those in the ground state.

We can then write

$$\frac{dn_1}{dt} = -\alpha Nn_1 + \beta n_2 , \quad \frac{dn_2}{dt} = \alpha Nn_1 - (\beta + \gamma) n_2 ,$$

$$\frac{dn_3}{dt} = \gamma n_2 .$$
 (1)

These equations have a solution $n_1 = n_1^0 \exp{-kt}$, where

$$k = \frac{1}{2} (\alpha N + \beta + \gamma) \left\{ 1 \pm \left[1 - 4 \frac{\alpha N \gamma}{(\alpha N + \beta + \gamma)^2} \right]^{1/2} \right\} \quad . \quad (2)$$

Since we know that when $\alpha \to 0$, $k \to 0$, the minus sign is chosen. Expanding the radical, we write

$$k = \alpha N \gamma / (\alpha N + \beta + \gamma)$$
.

For a Maxwellian electron distribution, we have^{2,5}

$$\beta/\alpha = (2\pi m_0 m^* k_B T/h^2)^{3/2} \exp{-\epsilon(F)/k_B T}$$
.

When the electron is bound in a hydrogenic orbit, the Schrödinger equation can be separated in parabolic coordinates (η, ξ) and the quantity (in atomic units)^{6,7} $\phi(\eta) = -(E/2) + (\rho_1/\eta K) - (m^2 - 1)/4 m^* \eta^2 + (F\eta/4)$ plays the role

role of the potential. Here m^* , K, m, and ρ_1 , are, respectively, the electron effective mass in units of m_0 , the relative dielectric constant, the azimuthal quantum number, and one of the parabolic quantum numbers. The extremum of $\phi(\eta)$ is at η_m . Because large η are important, the third term in $\phi(\eta)$ is neglected. The binding energy is $\epsilon = E/2 - 2(\rho_1 F/K)^{1/2}$, and

$$\begin{split} \frac{1}{k} &= \left(\frac{1}{\gamma} + \frac{1}{\alpha N}\right) + \left[\frac{1}{N\gamma} \left(\frac{2\pi m_0 m^* k_B T}{h^2}\right)^{3/2} \exp\left(-\frac{E}{2k_B T}\right)\right] \\ &\times \exp\frac{2}{k_B T} \sqrt{\frac{\rho_1 F}{K}} = B + A \exp c \sqrt{F} \quad . \end{split}$$

A least squares analysis determines A, B, and c. We find B < 0, and we assign this artifact to the approximation done in going from Eq. (3) to Eq. (3a) in Reference 1. The values of c are compared with the H atom theory in Table I. When n = 2, the possible values of ρ_1 are $\frac{3}{4}$, $\frac{1}{2}$, and $\frac{1}{4}$, for states that transform as (S - z), x or y, and (S + z), respectively. For a quantum number n, ρ_1 varies between (n - 1)/2n and 1/2n.

No model for the excited states of SF_6 and O_2 in Ar or Xe exists. In the gas the electron affinity of O_2 is 9 ~ 0.45 eV. There are no known excited states. That of SF_6 is between 10 1.1 and 1.5 eV.

In SF₆ the electron can only be bound if it approaches the six F nuclei. When K increases, the extra electron density near the F decreases and c decreases more rapidly than as $K^{-1/2}$. The charge density in O_2^- ground state is rather localized between the atoms^{11,12} and the extra electron that approaches an O sees an effective charge > e; accordingly, c is larger than for H. Excited states of O_2^- in Ar or Xe should have a similar behavior; this is necessary to explain the observed values of c.

We have shown that the electric field dependence of the capture of electrons by SF_6 or O_2 in Ar and Xe is explained without reference to hot electrons. The line

TABLE I. Comparison of the experimental and the calculated values of the coefficient c multiplying \sqrt{F} . The theoretical values are calculated for a hydrogenic state with either principal quantum number 2 or larger, whose symmetry is given in the rightmost column. K is calculated from the polarizability using the Clausius-Mosotti equation.

Solvent	Dopant	c_{exp}	c_{th} $(n=2)$	c_{th} (n large)	Symmetry
Ar	SF ₆	3. $65 \times 10^{-3} \pm 1 \times 10^{-4}$	4.06×10^{-3} 5.74×10^{-3}	$8.12 \times 10^{-3}/n$	(S + z)
	O_2	$1.74 \times 10^{-2} \pm 1 \times 10^{-3}$	7×10^{-3}	8.12×10^{-3}	(S-z)
Xe	\mathbf{SF}_{6}	$1.07 \times 10^{-4} \pm 1 \times 10^{-5}$	1.58×10^{-3}	$3.16 \times 10^{-3}/n$	(S+z)
	02	$8.15 \times 10^{-3} \pm 1 \times 10^{-4}$	2.23×10^{-3} 2.74×10^{-3}	3.16×10 ⁻³	x, y $(S-z)$

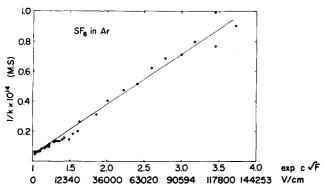


FIG. 1. Experimental dependence of the inverse of the trapping rate constant of electrons by SF_6 in Ar as a function of $\exp c\sqrt{F}$. The value of c is given in Table I.

in Fig. 1 is very sensitive to the parameter c whose value agrees embarassingly well with what is expected from a theory of the Schottky ionization of an excited state of an H like impurity.

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ERRATA

Erratum: Dynamics of fluctuations and spinodal decomposition in polymer blends [J. Chem. Phys. 72, 4756 (1980)]

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The structure of the mobility coefficient $\Lambda(q)$ described in this paper is wrong at large wave vectors q. A direct calculation of the coherent scattering function $S_q(t)$ (for one labeled chained in a homopolymer melt) shows that (for $qR_0>1$) $S_q(t)$ splits into two parts: one associated with the local distribution of kinks (fast relaxation) and one associated with the overall drift of the chain in its tube (slow relaxation). In the paper a single relaxation time was introduced, while two times are required in reality. The corrected $S_q(t)$ is analyzed in a paper submitted to Journal de Physique (Paris).

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