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The lowest triplet state of 1,3,5-hexatrienes: Quantum chemical force field calculations and experimental resonance Raman spectra

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Theoretical and Raman spectroscopic studies are presented of *E* and *Z*-1,3,5-hexatriene and their 3,4- and 2,5-dideuteriated analogs in ground and excited triplet states. The T_1 potential energy surface is calculated from extended SCF-LCAO-MO-CI theory. Energy minima and equilibrium geometries are determined in T_1 . Frequencies and normal modes of vibration are calculated for the minima of the T_1 and S_0 states. Energies of higher triplet levels are computed and oscillator strengths for the transitions from T_1 to T_n are determined. The displacements in equilibrium geometries between the T_1 and the T_n level corresponding to the strongest $T_1 \rightarrow T_n$ transitions are calculated and are used to estimate the intensities of the resonance Raman spectra of the T_1 state under the assumption of a predominant Franck-Condon scattering mechanism. The results indicate that the planar *E* and *Z* forms of hexatriene and its analogs are the only ones contributing substantially to the $T_1 \rightarrow T_n$ absorption and the T_1 resonance Raman spectra found in the present experiments. The existence of a twisted form in the T_1 state cannot be ruled out, but its contribution to the resonance Raman spectra corresponding to an electronic $T_1 \rightarrow T_n$ transition around 315 nm is likely to be much weaker than that of the *E* or *Z* forms. Satisfactory agreement is found between the calculated and experimentally determined resonance Raman spectra. An assignment is obtained for the experimentally determined vibrational modes in T_1 . The theoretical results indicate a substantial rotation of normal modes from S_0 to T_1 .

I. INTRODUCTION

The electronic structure and spectroscopy of short polyenes has been studied widely during the past decade.¹ Most work has dealt with ground state and excited singlet state properties. These have been summarized recently by Hemley *et al.*² Less attention has been devoted to the triplet manifold.

A number of theoretical calculations of 1,3,5-hexatriene have appeared in the literature. These include both *ab initio*³⁻¹² and semiempirical^{12,12-14} methods.

The geometry of *E*- and *Z*-hexatriene in the ground state in the gas phase has been determined by electron diffraction.^{15,16} Vibrational spectroscopy has been used by a number of authors as a means to investigate the ground state of hexatriene.¹⁷⁻²⁵ These studies include Raman and infrared spectra of hexatriene- d_0 ¹⁷⁻²⁵ and tetradeuteriated¹⁹⁻²³ hexatrienes. Furthermore, vibrational spectra of less stable conformers have been reported.^{26,27}

The triplet states of 1,3,5-hexatriene (HT) and methylated derivatives have been studied by direct singlet-triplet absorption spectroscopy,^{28,29} electron energy loss spectroscopy,³⁰ time-resolved absorption spectroscopy,^{31,32} and time-resolved resonance Raman spectroscopy.³²⁻³⁵ The vertical energies of the lowest triplet state T_1 of *E*-1,3,5-hexatriene (EHT) and *Z*-1,3,5-hexatriene (ZHT) determined by electron energy loss spectroscopy are 2.03 and 2.07 eV, re-

spectively. A first maximum for the $T_1 \rightarrow T_n$ absorption spectrum of 1,3,5-heptatriene was found around 315 nm with a first-order decay rate of $k = 3.4 \times 10^6 \text{ s}^{-1}$.³² Similar first-order decay rates were found for EHT and ZHT from resonance Raman experiments.³⁵ The resonance Raman spectra of triplet EHT and ZHT were reported to be identical indicating equilibration with respect to twisting around the central C=C double bond in the T_1 state during the lifetime of the triplet state. In contrast the resonance Raman spectra of triplet *E*- and *Z*-2,5-dimethyl-1,3,5-hexatriene were reported to be different from each other, which was interpreted in terms of nonequilibration of excited rotamers (NEER) in the triplet state.³⁴ No detailed assignment of the triplet state resonance Raman spectra was given.

In order to give a more detailed assignment of the triplet vibrations we report in the present paper Raman spectra from the ground state of *E*-3,4- d_2 -1,3,5-hexatriene (*E*-3,4- d_2 -HT) and *E*-2,5- d_2 -1,3,5-hexatriene (*E*-2,5- d_2 -HT) as well as time-resolved resonance Raman spectra of these two compounds in their lowest excited triplet T_1 state. Furthermore, quantum chemical force field (QCFF/PI) and CNDO/S calculations are performed on the ground and excited triplet states of the *E* and *Z* isomers of HT, 3,4- d_2 -HT and 2,5- d_2 -HT. The previously reported triplet state resonance Raman spectra of EHT and ZHT³⁵ and the new experimental results presented here are then discussed in the light of the theoretical calculations.

II. EXPERIMENTAL METHODS

A. Materials

Acetonitrile- d_0 (Merck, LiChrosolv), acetonitrile- d_3 (Fluka, puriss., >99.8% D), and acetone (Ferak, p.a.) were used as received. The deuteriated 1,3,5-hexatrienes were prepared from the corresponding deuteriated 3-hexene-2,5-diols via pyrolysis of their acetates at 530 °C/0.1 mm, followed by separation of the reaction mixture using preparative gas chromatography (GC) (squalane column). 2,5-Dideuterio-3-hexene-2,5-diol was obtained by reduction of 3-hexene-2,5-dione³⁶ with LiAlD_4 (>98% D). Reduction of 3-hexyne-2,5-diol (Aldrich) with LiAlD_4 followed by quenching with D_2O ³⁷ afforded 3,4-dideuterio-3-hexene-2,5-diol. The diols were converted to the acetates by treatment with acetic anhydride and pyridine. ^1H NMR spectrometry (Bruker WM300, 300 MHz) of the deuteriated products indicated a high deuterium content (>95%). No isomeric or other impurities were found by capillary GC on the samples. The deuterium incorporation was judged to be high, since in the Raman spectra no bands were found that were attributable to species other than E -3,4- d_2 -HT and E -2,5- d_2 -HT. The purified samples were put in glass capillaries under nitrogen, and the capillaries were cooled briefly and sealed. Throughout the transient experiments the capillaries with trienes were opened and solutions prepared and transferred to sample cells under an Ar atmosphere. Prior to the addition of triene the solvents were purged with Ar for more than 35 min.

B. Methods

The details of experimental procedures for recording conventional Raman spectra²⁵ and for time-resolved experi-

ments³⁴ have been described previously. Briefly, the triplet state of the triene was produced by exciting acetone as sensitizer with a pump pulse from an excimer laser at 308 nm. The resonance Raman spectrum of hexatriene in its T_1 state was obtained by using the second harmonic at 317.5 nm from a Nd:YAG-pumped dye laser as probe pulse. The triplet state spectra were then obtained by subtraction procedures. The isomeric purity of the samples before and after the experiments was monitored by means of GC and by evaluating the preresonance Raman spectra of the solutions with respect to vibrational bands from the ground states of different hexatriene isomers.

III. EXPERIMENTAL RESULTS

A. Ground state

The Raman spectrum of neat E -3,4- d_2 -1,3,5-hexatriene excited at 514.5 nm is shown in Fig. 1 and the resulting wave numbers and polarizations are tabulated in Table I. Figure 2(a) shows the ground state Raman spectrum of an Ar-saturated solution of ≈ 9 mM E -3,4- d_2 -HT in acetonitrile- d_3 with pulsed excitation at 317.5 nm after subtraction of solvent. Figure 2(b) shows the difference spectrum obtained by subtraction of a spectrum of a solution before the transient experiments from that obtained of the same solution after irradiation with 2000 pump-and-probe pulses. Hence spectrum 2(B) is indicative³⁵ of the isomerization from the E to the Z isomer of 3,4- d_2 -HT, the negative bands being interpreted as a decrease in concentration of the E and the positive bands as an increase in concentration of the Z isomer. The solution used to obtain spectrum 2(B) was an Ar-saturated solution of 0.54 M acetone and ≈ 9 mM E -3,4- d_2 -HT in acetonitrile- d_3 . GC analysis showed no isomeric impuri-

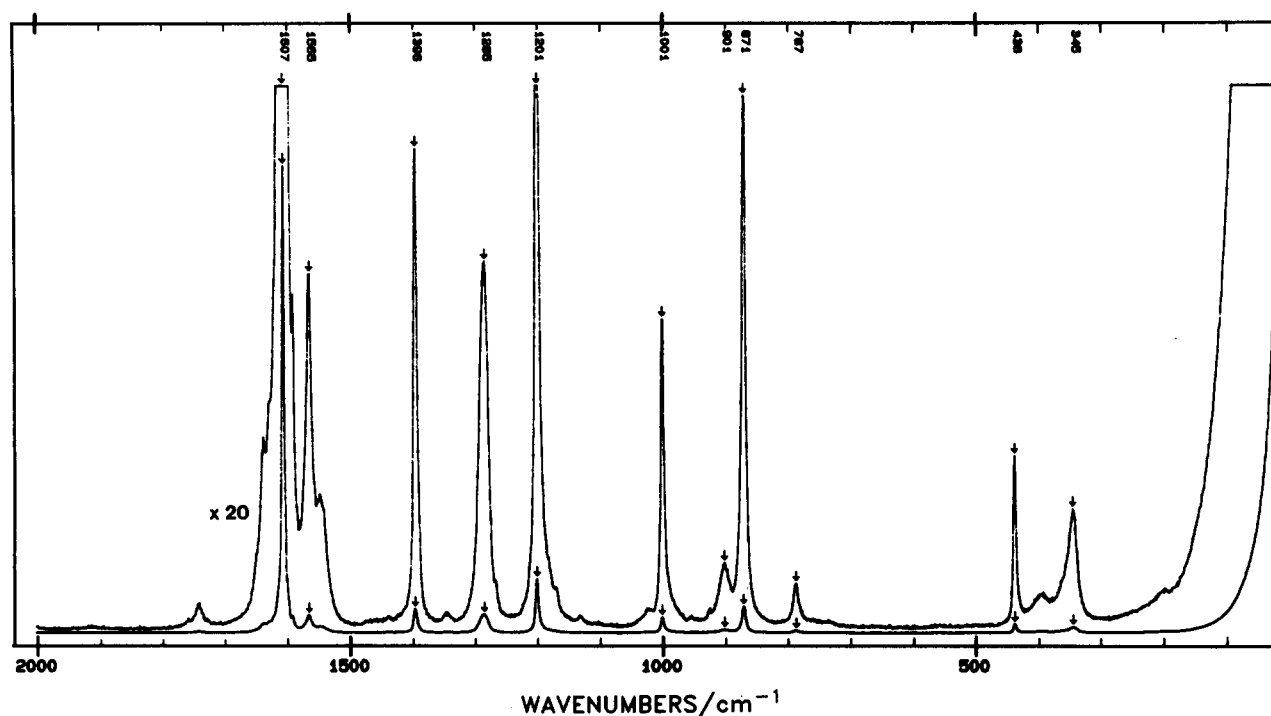


FIG. 1. Raman spectrum of neat E -3,4- d_2 -1,3,5-hexatriene excited at 514.5 nm.

TABLE I. Wave numbers (cm^{-1}) of Raman bands of the ground states of *E*-3, 4- d_2 -HT and *E*-2, 5- d_2 -HT excited at 514.5 and 317.5 nm.^a

<i>E</i> -3, 4- d_2 -HT		<i>E</i> -2, 5- d_2 -HT
514.5 nm	317.5 nm	317.5 nm
1638		
1629		
1607p	1610	1618
1591		
1565p	1563	1570
1548		
1424		
1396p	1397	1381
1344		
1285p	1290	1214
1267		
1201p	1204	1295
1171		
1131		
1025		
1001p	1002	1022
954		
925		
901dp		
871p	871	
787dp		
558dp		
438p		
395		
360		
345p		
201dp		

^ap polarized, dp depolarized.

ties before the transient experiments, and the ratio by GC of *E* to *Z* was 0.79/0.21 after the experiments of Fig. 2(B). The vibrational bands seen in Fig. 2(A) correspond to the strongest bands in the spectrum of Fig. 1; however, the relative intensities are changed due to preresonance with the optically allowed $1^1A_g \rightarrow 1^1B_u$ transition. In particular, the intensity of the C=C and C-C stretching bands at 1607 and 1201 cm^{-1} is strongly enhanced. This agrees with results on hexatriene- d_0 from the literature.³⁸ Three positive bands at 1320, 1132, and 897 cm^{-1} are observed in Fig. 2(B). Although the Raman spectrum of *Z*-3,4- d_2 -HT is not known, on the basis of our previous results with hexatriene- d_0 we are confident to assign these bands to *Z*-3,4- d_2 -HT. However, more caution is required in the C=C stretching region where bands from *E*-3,4- d_2 -HT and *Z*-3,4- d_2 -HT probably overlap.

Figure 3 shows preresonance Raman spectra for *E*-2,5- d_2 -HT corresponding to Fig. 2. Unfortunately the amount of sample available was not sufficient to record the ordinary Raman spectrum of this compound with 514.5 nm excitation. The wave numbers of the strongest Raman bands observed in Fig. 3(A) are tabulated in Table I. The spectrum of Fig. 3(A) was obtained from an Ar-saturated solution of ≈ 5 mM *E*-2,5- d_2 -HT in acetonitrile- d_0 and that of Fig. 3(B) from an Ar-saturated solution of 0.54 M acetone and ≈ 5 mM *E*-2,5- d_2 -HT in acetonitrile- d_0 before and after irradiation with 2000 pump-and-probe pulses. In spectrum 3(B) negative bands are again seen at positions corresponding to ground state bands of *E*-2,5- d_2 -HT and only one posi-

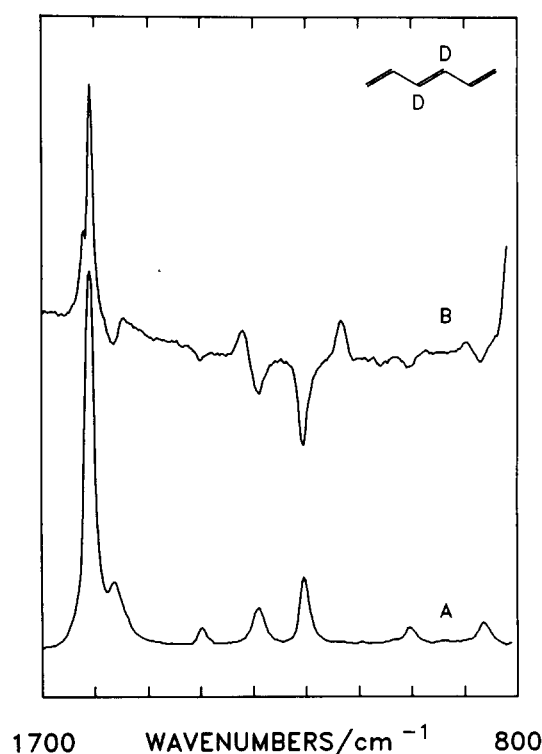


FIG. 2. Ground state (A) and difference (B) preresonance Raman spectra of *E*-3,4- d_2 -1,3,5-hexatriene in Ar-saturated CD_3CN solution excited at 317.5 nm. A: ≈ 9 mM *E*-3,4- d_2 -HT; B: 0.54 M acetone and ≈ 9 mM *E*-3,4- d_2 -HT after 2000 pump-and-probe pulses. Solvent and acetone bands subtracted.

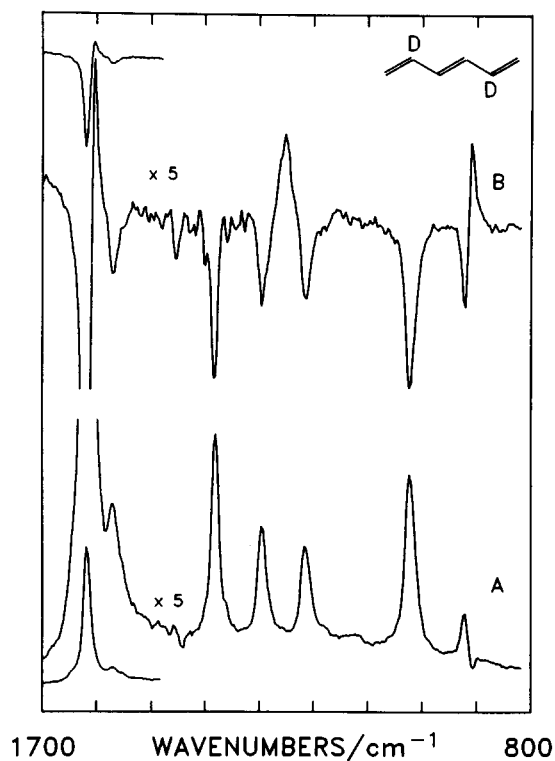


FIG. 3. Ground state (A) and difference (B) preresonance Raman spectra of *E*-2,5- d_2 -1,3,5-hexatriene in Ar-saturated CH_3CN solution excited at 317.5 nm. A: ≈ 5 mM *E*-2,5- d_2 -HT; B: 0.54 M acetone and ≈ 5 mM *E*-2,5- d_2 -HT after 2000 pump-and-probe pulses. Solvent and acetone bands subtracted.



FIG. 4. Triplet state time-resolved resonance Raman spectra of Ar-saturated solutions of 0.54 M acetone and A: ≈ 5 mM *E*-3,4- d_2 -HT in CH_3CN , B: ≈ 9 mM *E*-3,4- d_2 -HT in CD_3CN , and C: ≈ 5 mM *E*-2,5- d_2 -HT in CH_3CN . Solvent, acetone, and hexatriene ground state bands subtracted. Excitation wavelength 317.5 nm, pump-probe delay 60 ns.

tive band is observed at 1252 cm^{-1} and assigned to *Z*-2,5- d_2 -HT. Again, GC analysis showed no isomeric impurities before the transient experiments, and the ratio of *E* to *Z* by GC was 0.68/0.32 after the experiments of Fig. 3(B).

B. Excited triplet state

The spectra of *E*-3,4- d_2 -HT and *E*-2,5- d_2 -HT in their lowest excited triplet states obtained with a time delay of 60 ns between pump and probe pulses and excited at 317.5 nm are shown in Fig. 4. The corresponding wave numbers are tabulated in Table II together with previous data for hexatriene- d_0 . Spectrum 4(A) was obtained from an Ar saturated solution of 0.54 M acetone and ≈ 5 mM *E*-3,4- d_2 -HT in acetonitrile- d_0 , spectrum 4(B) from an Ar-saturated solution of 0.54 M acetone and ≈ 9 mM *E*-3,4- d_2 -HT in acetonitrile- d_3 and spectrum 4(C) from an Ar-saturated solution

of 0.54 M acetone and ≈ 5 mM *E*-2,5- d_2 -HT in acetonitrile- d_0 . Acetonitrile- d_3 was used as solvent in some experiments in order to establish the presence of transient bands in regions masked by strong solvent bands from acetonitrile- d_0 . All spectra shown in Fig. 4 are spectra after the subtraction of solvent bands and bands from ground state hexatriene obtained by procedures described in detail previously.³⁴ Such subtraction procedures can introduce negative bands due to the wavelength dependence of transient absorption. Such a band is seen in spectrum 4(C) at 917 cm^{-1} , the position of a strong band from acetonitrile- d_0 .

Outside the region shown in Fig. 4, spectra were recorded down to 300 cm^{-1} , but no additional transient bands were found.

The isomeric purity of the samples was monitored throughout the experiments. The amount of *Z* isomers formed in the experiments underlying Fig. 4 was determined by GC before and after the experiments. The ratio of *E* to *Z* was 1.00/0.00 before the experiments, whereas after the experiments it was 0.96/0.04 for spectra 4(A) and 4(B) and 0.93/0.07 for spectrum 4(C). Hence the average isomeric purity of the samples was better than 96% during the recording of the triplet state resonance Raman spectra.

IV. CALCULATIONS

A. Methods of calculation

Equilibrium geometries, normal coordinates, and vibrational frequencies were computed by the QCFF/PI program^{39,40} which has been shown to be quite reliable in describing conjugated and aromatic systems.^{13,14,41-43} With respect to the original program we have introduced the following modifications:

(a) Energies, gradients, and Hessians of ground and excited states were computed by use of a general CI procedure including multiple excitations.⁴⁴ Since the original QCFF/PI program was based on CI limited to singly excited configurations (SECs) with respect to the closed shell ground configuration, a number of changes had to be introduced in the routines computing energy gradients and Hessians. These modifications are very similar to those introduced by Zerbetto⁴⁵ in the upgrading of QCFF/PI with a CI procedure comprising SECs and doubly excited configurations (DECs) with respect to a closed shell configuration.

(b) In the computation of energy gradients and Hessians we have included the off-diagonal elements of the HF matrix, F_{ij} , expressed in the MO basis. These matrix elements vanish, when the MOs are obtained by the SCF treatment. For this reason they have been neglected in the original program.^{39,40} However, although the F_{ij} elements vanish for SCF MOs, their gradients do not. Hence their inclusion is necessary to obtain correct analytical gradients of excited electronic states.

(c) To compute energy gradients and Hessians, the energy of the N th electronic state is expressed as^{39,40}

$$V_N = \sum_r (R_r^\alpha \alpha_r + R_r^\gamma \gamma_{rr}) + \sum_{r \neq s} (R_{rs}^\beta \beta_{rs} + R_{rs}^\gamma \gamma_{rs}), \quad (1)$$

where r and s label atomic orbitals, α and β represent core integrals and γ Coulomb integrals, and R are coefficients

TABLE II. Wave numbers (cm^{-1}) of Raman bands of EHT- d_0 , *E*-3,4- d_2 -HT, and *E*-2,5- d_2 -HT in the T_1 state.

EHT- d_0 in CH_3CN	<i>E</i> -3,4- d_2 -HT in CH_3CN	<i>E</i> -3,4- d_2 -HT in CD_3CN	<i>E</i> -2,5- d_2 -HT in CH_3CN
1570 vs	1551 vs	1551 vs	1560 vs
1270 s		1352 w	1326 w
1234 w	1310 s	1309 s	1286 m
1200 m		1209 w	1226 w
1106 s	937 m	939 m	1140 sh
	890 m	891 m	1123 s

depending on the CI and MO coefficients. In the QCFF/PI program the hessian of this energy is computed analytically under the assumption that the coefficients R are constant. In this way the hessian of V_N is computed very efficiently but not entirely correctly. To take the derivatives of these coefficients into account, Warshel and Karplus³⁹ introduced a correction based on the polarizability theory in the MO scheme.⁴⁶ This correction leads to a correct force field for the (closed shell) ground state but is not satisfactory for excited states.

A straightforward procedure of computing the hessian of the energy V_N is numerical differentiation. Unfortunately this has the disadvantage of being tedious and in many cases unpracticable. However, we have noted that the gradients of V_N computed analytically, under the assumption of constant coefficients R , are quite close to the numerical gradients. Hence we have adopted the procedure of computing the hessian of V_N by numerical differentiation of analytical gradients. With this procedure we have been able to compute correctly the S_1 excited state frequencies of polyenes, benzene, and naphthalene.⁴⁷

Normal coordinates were computed by diagonalization of the mass weighted Cartesian force constant matrix. Correlation between normal modes of different isotopic species is made by computing the rotation matrix relating the normal coordinates of the two species.

The calculations on triplet states were performed by the half-electron SCF Hamiltonian⁴⁸ followed by a CI calculation in which all singly excited determinants (SEDs) with respect to the lowest triplet, belonging to the $\pi\pi^*$ space, were considered. This CI includes 33 $\pi\pi^*$ triplet determinants.

Electronic energies and wave functions were computed by the SCF and CI procedures outlined above, both with the QCFF/PI and the CNDO/S⁴⁹ Hamiltonian.

B. Results

1. Ground state

A large number of calculations of the equilibrium geometry and force field of *E*- and *Z*-hexatriene have been performed using *ab initio*,⁷⁻¹⁰ CNDO,²⁰ and QCFF/PI Hamiltonians.^{13,14} These calculations, together with the classical observations by Lippincott *et al.*^{17,18} and the more recent studies by Sabljic and McDiarmid²⁴ and Langkilde *et al.*²⁵ provide a reliable assignment of the infrared and Raman spectra of these compounds.

In this section we present the results of QCFF/PI calculations of equilibrium geometry and vibrational frequencies of 1,3,5-hexatriene and its 3,4-*d*₂- and 2,5-*d*₂-deuteriated derivatives. On the one hand, on the basis of these calculations we can assign the Raman spectra of the two deuteriated species shown in Sec. III. On the other hand, the comparison of observed and calculated wave numbers of the three isotopic species is a valuable test to assess the reliability of the computed force fields.

Beside their interest *per se*, ground state force fields, frequencies, and normal modes are a natural reference for the force fields, frequencies, and normal modes of the lowest triplet state T_1 . For this reason we have performed the

TABLE III. Geometry of the hexatriene carbon skeleton in the ground state S_0 , the triplet state T_1 , and the triplet state T_n ($n = 5$ or 6), (bond lengths in Å).

State geometry	C ₁ C ₂	C ₂ C ₃	C ₃ C ₄	C ₁ C ₂ C ₃	C ₂ C ₃ C ₄
S_0					
E^a	1.343	1.470	1.354	122.2°	121.8°
E^b	1.347	1.462	1.359	121.9°	121.6°
E^c	1.324	1.463	1.329	124.3°	124.0°
E^d	1.3407	1.4630	1.3454	124.1°	123.9°
E^e	1.3399	1.4520	1.3462	124.1°	124.0°
E^f	1.337	1.458	1.368	121.7°	124.4°
Z^a	1.343	1.471	1.355	121.8°	125.3°
Z^g	1.3290	1.4621	1.3366	123.49°	127.19°
Z^h	1.336	1.462	1.362	122.13°	125.94°
T_1					
E^a	1.397	1.386	1.470	122.5°	121.5°
Z^a	1.397	1.387	1.472	122.0°	124.8°
Tw^a	1.384	1.388	1.480	122.6°	121.5°
T_n					
$E(n=5)^a$	1.410	1.428	1.415	121.2°	121.4°
$Z(n=5)^a$	1.411	1.432	1.412	121.2°	123.9°
$Tw(n=6)^a$	1.411	1.415	1.470	121.2°	121.0°

^aThis work.

^bReference 14 (b) (theory).

^cReferences 8 and 11 (theory).

^dReference 9 (theory).

^eReference 10 (theory).

^fReference 15 (expt.).

^gReference 7 (theory).

^hReference 16 (expt.).

ground state calculations at the SCF level, without inclusion of interactions with doubly excited configurations (the contribution of singly excited configurations is automatically included in the SCF treatment), since T_1 calculations are based on a CI limited to singly excited configurations with respect to the dominant, HOMO-LUMO, configuration of the state T_1 .

a. E-hexatriene. The computed ground state geometry of the carbon skeleton is reported in Table III together with the results of other calculations and with experimental parameters. The numbering of the atoms is given in Fig. 5. The agreement between our calculations and the experimental results is satisfactory. This is especially true for the values of the CCC angles. The terminal C=C bonds are correctly predicted to be shorter than the central C=C bond.

The QCFF/PI force field, computed with respect to Cartesian coordinates, was transformed into a force field expressed in terms of the internal coordinates defined in Table

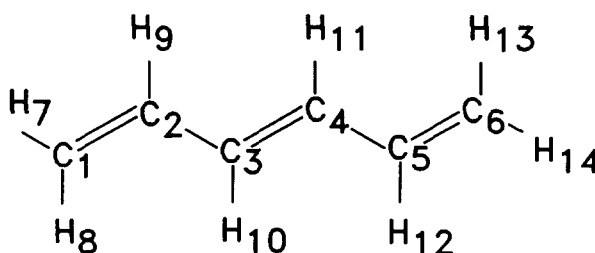


FIG. 5. Numbering of atoms in 1,3,5-hexatriene.

No.	Type	Definition
1	CC stretch	C_1C_2
2	CC stretch	C_2C_3
3	CC stretch	C_3C_4
4	CC stretch	C_4C_5
5	CC stretch	C_5C_6
6	vinyl CH rock	$-C_1C_2H_9 + C_3C_2H_9$
7	trans CH rock	$-C_2C_3H_{10} + C_4C_3H_{10}$
8	trans CH rock	$-H_{11}C_4C_5 + C_3C_4H_{11}$
9	vinyl CH rock	$-C_6C_5H_{12} + C_4C_5H_{12}$
10	CH ₂ scissor	$2H_7C_1H_8 - H_7C_1C_2 - H_8C_1C_2$
11	CH ₂ scissor	$2H_{13}C_6H_{14} - H_{13}C_6C_5 - H_{14}C_6C_5$
12	CH ₂ rock	$-H_7C_1C_2 + H_8C_1C_2$
13	CH ₂ rock	$-H_{14}C_6C_5 + H_{13}C_6C_5$
14	CCC terminal	$C_1C_2C_3$
15	CCC central	$C_2C_3C_4$
16	CCC central	$C_3C_4C_5$
17	CCC terminal	$C_4C_5C_6$
18	CH ₂ twist	$H_8C_1C_2C_3 + H_8C_1C_2H_9 + H_7C_1C_2C_3 + H_7C_1C_2H_9$
19	C-C torsion	$C_1C_2C_3H_{10} + C_1C_2C_3C_4 + H_9C_2C_3H_{10} + H_9C_2C_3C_4$
20	C-C torsion	$H_{10}C_3C_4C_5 + H_{10}C_3C_4H_{11} + C_2C_3C_4C_5 + C_2C_3C_4H_{11}$
21	C-C torsion	$C_3C_4C_5H_{12} + C_3C_4C_5C_6 + H_{11}C_4C_5H_{12} + H_{11}C_4C_5C_6$
22	CH ₂ twist	$H_{12}C_5C_6H_{14} + H_{12}C_5C_6H_{13} + C_4C_5C_6H_{14} + C_4C_5C_6H_{13}$
23	CH ₂ wag	$H_9H_7C_1C_2$
24	vinyl CH wag	$C_1H_9C_2C_3$
25	trans CH wag	$C_2H_{10}C_3C_4$
26	trans CH wag	$C_3H_{11}C_4C_5$
27	vinyl CH wag	$C_4H_{12}C_5C_6$
28	CH ₂ wag	$C_5H_{13}C_6H_{14}$

In Tables VIII and IX we report our computed wave numbers together with some of the previous theoretical results^{7,9,10,14(b)} and with the most recent experimental

1	8.88																		
2	0.62	5.42																	
3	-0.08	0.60	8.50																
4	0.01	-0.02	0.60	5.42															
5	-0.01	0.01	-0.08	0.62	8.88														
6	-0.16	0.20	0.02	0.53													
7	-0.02	-0.20	0.19	0.02	..	-0.02	0.55												
8	..	0.02	0.19	-0.20	-0.02	0.01	0.02	0.55											
9	0.02	0.20	-0.16	..	0.01	-0.02	0.53										
10	-0.20	-0.01	0.46									
11	-0.01	-0.20	0.46								
12	0.02	0.02	-0.04	-0.01	-0.01	..	0.52							
13	0.02	0.02	-0.01	-0.04	..	-0.01	..	0.52						
14	0.22	0.20	-0.01	-0.01	-0.02	..	-0.04	..	1.37					
15	-0.01	0.21	0.21	-0.02	0.03	1.39				
16	..	-0.02	0.21	0.21	-0.01	0.03	1.39			
17	-0.01	0.20	0.22	-0.02	..	-0.04	0.03	1.37		

TABLE VI. The out-of-plane diagonal force constants of hexatriene in the E , Z , and twisted (Tw) geometry and in the S_0 and T_1 states. (Units see Table V.)

	E, S_0	Z, S_0	E, T_1	Tw, T_1	Z, T_1
18	0.14	0.14	0.08	0.10	0.08
19	0.03	0.03	0.10	0.09	0.09
20	0.13	0.12	0.03	0.03	0.02
21	0.03	0.03	0.10	0.09	0.09
22	0.14	0.14	0.08	0.10	0.08
23	0.18	0.18	0.12	0.13	0.12
24	0.23	0.22	0.24	0.25	0.24
25	0.22	0.24	0.17	0.15	0.19
26	0.22	0.24	0.17	0.15	0.19
27	0.23	0.22	0.24	0.25	0.24
28	0.18	0.18	0.12	0.13	0.12

data.^{24,25} The assignment given is qualitative and indicates the main components; in fact all the modes result from mixing of many internal coordinates, in particular the modes assigned as CH rock and C–C stretch. All theoretical calculations are in substantial agreement with each other.

The experimental assignments are in large part confirmed by calculations. There is a discrepancy between McDiarmid and Sabljic²⁴ and Langkilde *et al.*²⁵ in the assignment of CH₂ rock (b_u) and CH₂ wag (b_g). All the calculations are concordant in the following points: the frequency of CH₂ rock (b_u) is about 950 cm⁻¹ and that of CH₂ wag (b_g) is higher than that of the lowest CH wag (b_g) and certainly higher than 900 cm⁻¹.

In Table X we report the wave numbers and the dominant components of the normal modes of EHT- d_0 , E -3,4- d_2 -HT, and E -2,5- d_2 -HT. The correlation between the wave numbers of different isotopomers is drawn on the basis of the rotation matrix between the modes of the respective isotopomers.

Deuteration changes the wave numbers of CH rock and CH wag vibrations. Hence it modifies both in-plane and out-of-plane normal modes. The mixed character of the in-plane modes with wave numbers in the region 1172–1350 cm⁻¹ is seen by comparing with the corresponding modes of the deu-

teriated compounds, where a change of the dominant contributions is observed.

Upon deuteration in positions 3,4, one of the a_g CH rock wave numbers drops to 1003 cm⁻¹, i.e., below the wave number of the C–C stretch, which is then pushed upwards compared to EHT- d_0 by the interaction with the 1003 cm⁻¹ mode. The other CH rock remains nearly unchanged. The 1003 cm⁻¹ mode gets close to the position of the CH₂ rock mode and pushes it down in wave number.

Similar effects are observed for the b_u modes as well. The CH rock at 1310 cm⁻¹ is lowered upon 3,4-deuteration to 1034 cm⁻¹ and by near-resonant interaction pushes the CH₂ rock vibration from 946 to 896 cm⁻¹.

Among the out-of-plane modes, the only significant change noted is the lowering of one of the CH wags, namely of the a_u mode at 990 cm⁻¹ and the b_g mode at 858 cm⁻¹.

Essentially similar effects are observed for E -2,5- d_2 -HT. Among the a_g modes, that at 1305 cm⁻¹ (CH rock) is lowered to 1027 cm⁻¹, which in turn pushes the CH₂ rock down from 941 to 864 cm⁻¹. Among the b_u modes, the CH rock at 1289 cm⁻¹ is lowered to a value below 1000 cm⁻¹ and interacts strongly with the CH₂ rock mode. At the same time, the mode at 1172 cm⁻¹ which is predominantly C–C stretch is pushed up to 1214 cm⁻¹ by interaction with the CH₂ and CD rocks. As far as the out-of-plane modes are concerned, the effect of deuteration is simply that of decreasing the CH wag frequencies.

b. Z-hexatriene. Computed and experimental geometries of Z -hexatriene are reported in Table III. The QCFF/PI computed geometry is in good agreement with experiment.¹⁶

The force field of Z -hexatriene appears very similar to that of E -hexatriene and is not given here. The main difference is a higher value of the two central CCC bending force constants and of the coupling between them.

In Table IX we report the computed in-plane and out-of-plane wave numbers together with the results of previous theoretical work^{7,14(b)} and with the most recent spectroscopic observations.^{24,25} As can be seen from the tables, our results are in substantial agreement with those of the two previous calculations. Experimental assignments appear in

TABLE VII. The in-plane force field of E -hexatriene in the T_1 state (CH stretch force constants omitted).

1	5.97																			
2	1.33	6.76																		
3	0.25	0.48	5.06																	
4	-0.24	0.13	0.48	6.76																
5	0.36	-0.24	0.25	1.33	5.97															
6	-0.17	0.19	0.02	0.53														
7	-0.02	-0.19	0.20	0.02	..	0.02	0.55													
8	..	0.02	0.20	-0.19	-0.02	0.01	0.02	0.55												
9	0.02	0.19	-0.17	..	0.01	-0.02	0.53											
10	-0.20	-0.01	0.45											
11	-0.01	-0.20	0.45										
12	0.02	0.02	-0.04	0.52									
13	0.02	0.02	-0.04	..	-0.01	..	0.52						
14	0.20	0.19	-0.02	-0.01	-0.02	..	-0.04	..	1.36					
15	-0.01	0.21	0.20	-0.01	0.03	1.38				
16	..	-0.01	0.20	0.21	-0.01	0.04	1.38			
17	-0.02	0.19	0.20	-0.01	..	-0.02	..	-0.04	0.03	1.36		

TABLE VIII. Computed and experimental vibrational wave numbers (cm^{-1}) of *E*-hexatriene in the S_0 state.

	This work	HBK ^a	FSLBP ^b	BKPP ^c	SKL ^d	Expt. ^e	Expt. ^f
a_g	1664 C=C	1619	1662	1664	1644	1623	1627
	1594 C=C	1577	1583	1559	1578	1574	1579
	1419 CH ₂ sciss	1413	1414	1410	1403	1397	1400
	1350 CH rock	1350	1313	1299	1310	1288	1320
	1305 CH rock	1294	1288	1267	1292	1283	1288
	1206 C-C	1192	1198	1193	1202	1188	1192
	941 CH ₂ rock	931	926	932	934	930	934
	461 CCC bend	457	432	429	443	444	449
	393 CCC bend	375	340	346	357	353	355
b_u	1645 C=C	1617	1634	1618	1626	1629	1629
	1445 CH ₂ sciss	1463	1445	1449	1438	1433	1432
	1310 CH rock	1302	1301	1298	1296	1296	1295
	1289 CH rock	1271	1270	1235	1269	1255	1255
	1172 C-C	1179	1126	1121	1127	1132	1186
	946 CH ₂ rock	944	957	953	958	966	1129
	589 CCC bend	578	530	536	553	541	543
	179 CCC bend	175	143	146	151	152	...
a_u	1082 CH wag	1057	1026	1015		1008	1013
	990 CH wag	961	953	948		938	937
	949 CH ₂ wag	943	922	916		900	901
	643 C=CH ₂ twist	619	679	677		683	683
	241 C=C twist	242	242	234		248	(317)
	101 C-C twist	106	94	88		94	90
b_g	1050 CH wag	1020	1000	995		985	988
	952 CH ₂ wag	948	929	925		901	872
	858 CH wag	829	874	870		868	906
	611 C=CH ₂ twist	594	590	582		428	433
	214 C-C twist	228	211	189		215	217

^a Reference 14(b), PPP-DCI calculation.^b Reference 9, scaled force field from *ab initio* 4-21 G calculation.^c Reference 7, scaled force field from *ab initio* 6-31 G calculation.^d Reference 10, scaled force field from *ab initio* TZ + P calculation.^e Reference 25, experimental.^f Reference 24, experimental.

general to be consistent with each other and with those proposed by calculations. Two small discrepancies are noted. One pertains to the CH rock modes of b_1 type, the lowest of which is assigned to the 1187 cm^{-1} band by Langkilde *et al.*,²⁵ but to the 1280 cm^{-1} band by Sabljic *et al.*²⁴ This band is predicted theoretically to be close to or above 1300 cm^{-1} , which is more compatible with the assignment of Sabljic *et al.* In the b_2 species, the theoretical results concur with the assignment of CH and CH₂ wags of Langkilde *et al.* rather than with that of Sabljic *et al.*

The vibrational wave numbers and assignments of *Z*-hexatriene and of its 3,4- and 2,5-deuteriated isomers are given in Table XI. The effects of deuteriation are essentially limited to the region $1100\text{--}1400\text{ cm}^{-1}$ of the C-C stretches and CH rocks and to the CH wags. Furthermore, deuteriation induces a substantial rotation of the modes. These effects are similar to those found for the *E* isomer.

The experimentally observed wave numbers for the *Z* isomers of 3,4-*d*₂-HT and 2,5-*d*₂-HT (*vide supra*) can be compared with the calculated positions of a_1 modes. For *Z*-3,4-*d*₂-HT the modes at 1320 , 1132 , and 897 cm^{-1} are as-

signed to the calculated bands at 1324 , 1170 , and 899 cm^{-1} , respectively, and for *Z*-2,5-*d*₂-HT the mode at 1252 cm^{-1} is assigned to the calculated one at 1268 cm^{-1} .

2. Lowest triplet state T_1

In this section we report the computed equilibrium geometries, force fields, and vibrational wave numbers of hexatriene and its deuteriated isomers in the *E*, *Z* and twisted (Tw) forms in the lowest triplet state T_1 . The corresponding experimental results for hexatriene-*d*₀ have been published previously³⁵ and those for the 3,4- and 2,5-dideuteriated hexatrienes are given in Sec. III of this paper.

The potential energy curve for the twisting around the central C=C bond (coordinate φ) in the ground state exhibits typically a high barrier ($\approx 50\text{ kcal/mol}$) at the geometry $\varphi = 90^\circ$. This is the basis for the existence of two distinct and stable forms, *E* and *Z*, in the ground state at room temperature. In the T_1 state the potential energy curve along φ is rather flat. This is a well known feature, that applies to all the ethylene derivatives and can be shown by Woodward-Hoffmann correlation diagrams. If the T_1 potential energy sur-

TABLE IX. Computed and experimental vibrational wave numbers (cm^{-1}) of Z -hexatriene in the S_0 state.

	This work	HBK ^a	BPKP ^b	Expt. ^c	Expt. ^d
a_1					
1664 C=C	1619	1646	1623	1643	
1585 C=C	1565	1549	1578	1540	
1424 CH ₂ sciss	1423	1411	1397	1398	
1330 CH rock	1325	1322	1315	1320	
1254 CH rock	1241	1239	1246	1249	
1120 C-C	1118	1078	1082	1084	
882 CH ₂ rock	884	859	884	885	
422 CCC bend	417	384	394	393	
199 CCC bend	201	164	170	182	
b_1					
1642 C=C	1615	1612	1623	1623	
1465 CH ₂ sciss	1468	1455	1450	1451	
1399 CH rock	1397	1334	1278	...	
1303 CH rock	1297	1289	1187	1280	
1225 C-C	1217	1187	1138	1187	
944 CH ₂ rock	937	943	954	(979)	
707 CCC bend	705	676	680?	(508)	
420 CCC bend	410	347	353	355	
a_2					
1072 CH wag	1049	1010	1032?	...	
987 CH wag	961	974	952	954	
949 CH ₂ wag	943	921	907	...	
670 C=CH ₂ twist	649	704	707	707	
362 C=C twist	350	326	332	335	
153 C-C twist	168	126	155	160	
b_2					
1046 CH wag	1017	997	989	990	
952 CH ₂ wag	948	928	906	825	
777 CH wag	764	794	815	908	
615 C=CH ₂ twist	597	572	585	586	
104 C-C twist	113	88	100	110	

^a Reference 14(b).^b Reference 7.^c Reference 25.^d Reference 24.

face has two minima separated by an energy barrier ΔE , the rate constant k for the process leading from one minimum to the other can be expressed as

$$k = \frac{k_b T}{h} e^{-\Delta E/k_b T}, \quad (2)$$

where k_b is the Boltzmann constant. At room temperature ($T = 298 \text{ K}$), with a typical $\Delta E \leq 8 \text{ kcal/mol}$, this results in $k \geq 8.36 \times 10^6 \text{ s}^{-1}$. Hence equilibration can take place within the lifetime ($\sim 100 \text{ ns}^{35}$) of the T_1 state. Thus in T_1 HT molecules can be regarded as moving almost freely along φ between the E and the Z geometries and are distributed statistically among the different geometries according to the Boltzmann rule. This property is the basis for their ability to isomerize in T_1 . The actual $T_1 \rightarrow S_0$ decay takes place close to the Tw geometry, where the T_1-S_0 energy gap is smallest; once on the top of the S_0 barrier the molecule must choose one of the two stable forms of the S_0 state, thus eventually undergoing isomerization. If in T_1 the Tw geometry has the lowest energy, then according to the Boltzmann rule this is the dominant species in T_1 and the decay $T_1 \rightarrow S_0$ is very fast. If on the other hand the Tw geometry has a (moderately)

higher energy than the E and Z forms in the T_1 state, the latter two are the dominant species. The decay then will still take place from a geometry near the Tw geometry, albeit with a slower rate, due to the larger energy gap between T_1 and S_0 at the Tw geometry. It is thus important to establish the energies of these three key geometries and to examine how many minima are found on the T_1 potential energy curve along φ .

According to the QCFF/PI calculations, relative minima are found at the E and Z geometries. After geometry optimization the Z form and the Tw form are computed at 1.1 and 7.1 kcal/mol, respectively, above the energy of the E form. The force field and the normal coordinates were evaluated at the three geometries E , Z and Tw. Previous calculations of the T_1 energy curve include *ab initio*^{3,4,12} work and a MINDO/3 study.¹² One of the *ab initio* calculations³ leads to results similar to ours: there are small barriers between the E and Z geometries and the E and Z forms have almost the same energy. The other *ab initio* and the MINDO/3 calculation lead to somewhat different results indicating a minimum with an energy of -6^{12} or -1.9 kcal/mol^4 at the Tw geometry. However, MINDO/3, like most NDO methods, is known to overestimate conjugation and thus to prefer twisted geometries.

Accepting our calculated energy differences, the full thermal equilibration among the three forms is established with a rate constant $k = 3.8 \times 10^7 \text{ s}^{-1}$ according to Eq. (2). Hence the equilibrium is established in a time of $\sim 30 \text{ ns}$, a time considerably shorter than the lifetime ($\sim 100 \text{ ns}^{35}$) of the T_1 state. The relative equilibrium populations corresponding to the calculated energies are 1, 0.16, and 10^{-5} for the E , Z , and Tw forms, respectively. These results will be discussed in Sec. V in the light of the experimentally observed resonance Raman spectra.

a. E-hexatriene. The computed equilibrium geometry of the E form in the T_1 state is given in Table III. No direct experimental parameters are available for comparison. The differences of bond lengths with respect to the ground state are in agreement with expectations based on the changes in bond orders. It is interesting to note that, while both the C_1-C_2 and the C_3-C_4 bonds lengthen, the lengthening of the latter is considerably larger. In T_1 all the CCC angles become similar, roughly equal to 121.5° . The force field is given in Tables VI (diagonal force constants only) and VII. The main changes with respect to the force field of S_0 are observed in the CC stretch submatrix. The diagonal elements have changed in agreement with the changes of bond length. However, the largest force constant is considerably smaller than its counterpart in S_0 . The vicinal coupling between $C_1=C_2$ and C_2-C_3 has become larger and the coupling between C_2-C_3 and $C_3=C_4$ has decreased. At the same time the couplings between nonadjacent stretches, which are small in S_0 , have all become sizable in T_1 . Also twisting force constants have changed in the way suggested by the change in bond length: the force constants of the twistings around the bonds $C_1=C_2$ and $C_3=C_4$ have decreased, while the force constants of the twisting around C_2-C_3 has increased.

The computed vibrational wave numbers of E -hexatriene in the T_1 state and of its deuteriated forms are reported

TABLE X. Computed and experimental ground state wave numbers (cm^{-1}) of *E*-hexatriene and its dideuterated forms.^a

EHT- d_0	<i>E</i> -3, 4- d_2 -HT	Expt. ^b	<i>E</i> -2, 5- d_2 -HT	Expt. ^c
a_g				
1664 C=C	1647 C=C	1607	1651 C=C	1618
1594 C=C	1577 C=C	1565	1592 C=C	1570
1419 CH ₂ sciss	1419 CH ₂ sciss	1396	1403 CH ₂ sciss	1381
1350 CH rock	1255 C-C (CH rock)	1201	1343 CH rock	1295
1305 CH rock	1303 CH rock	1285	1245 C-C (CH ₂ rock)	1214
1206 C-C	1003 CD (CH ₂) rock	1001	1027 CD (CH ₂) rock	1022
941 CH ₂ rock	879 CH ₂ (CD) rock	871	864 CH ₂ (CD) rock	
461 CCC bend	449 CCC bend	438	442 CCC bend	
393 CCC bend	390 CCC bend	345	389 CCC bend	
b_u				
1645 C=C	1643 C=C		1630 C=C	
1445 CH ₂ sciss	1441 CH ₂ sciss		1428 CH ₂ sciss	
1310 CH rock	1034 CD (CH ₂) rock		1302 CH rock	
1289 CH rock	1295 CH (CH ₂ , CD) rock		1214 C-C (CH ₂ rock)	
1172 C-C	1173 C-C (CH rock)		984 CH ₂ (CD) rock	
946 CH ₂ rock	896 CH ₂ (CD) rock		887 CD, CH ₂ rock	
589 CCC bend	560 CCC bend		582 CCC bend	
179 CCC bend	177 CCC bend		175 CCC bend	
a_u				
1082 CH wag	1052 CH wag		1031 CH wag	
990 CH wag	750 CD wag		870 CD wag	
949 CH ₂ wag	949 CH ₂ wag		952 CH ₂ wag	
643 C=CH ₂ twist	641 C=CH ₂ twist		624 C=CH ₂ twist	
241 C=C twist	238 C=C twist		218 C=C twist	
101 C-C twist	100 C-C		99 C-C twist	
b_g				
1050 CH wag	1048 CH wag		893 CH, CD wag	
952 CH ₂ wag	951 CH ₂ wag	901	954 CH ₂ wag	
858 CH wag	765 CD wag	787	845 CD, CH wag	
611 C=CH ₂ twist	579 C=CH ₂ twist	558	585 C=CH ₂ twist	
214 C-C twist	197 C-C twist	201	211 C-C twist	

^a The correspondence between the modes of deuterated forms and those of hexatriene- d_0 has been obtained from the normal modes rotation matrix. Only dominant contributions to normal modes are given; internal coordinates in parentheses are less important. The labeling C=C and C-C is that pertinent to the ground state.

^b This work; 514.5 nm excitation.

^c This work; 317.5 nm excitation.

in Table XII. We discuss first hexatriene- d_0 , starting with the in-plane vibrations. We note a decrease in the frequencies of the CC stretching modes compared to S_0 : they are roughly 1550, 1200, and 1150 cm^{-1} of a_g symmetry and 1430 and 1140 cm^{-1} of b_u symmetry. Some of these modes are in the range of the CH rock wave numbers and accordingly mix strongly with them. Also the mixing between the three types of CC stretches is extensive and much larger than in S_0 : This is due to the larger off-diagonal matrix elements found in the force fields. The highest wave number mode of b_u symmetry corresponds to CH₂ scissoring rather than CC stretching. Among the out-of-plane modes, the CH₂ wag wave numbers are lower than those in S_0 by 200 cm^{-1} . As expected, the order of the wave numbers of C₂C₃ and C₃C₄ twists is reversed compared to S_0 .

Deuteriation in positions 3,4 or 2,5 produces a lowering of one of the CH rocks with an extensive mixing of the five a_g normal modes in the region 1369–939 cm^{-1} , namely of the modes based on the two CH rocks, two CC stretches, and the CH₂ rock. The CD rock mixes heavily with the CH₂ rock in both deuteriated species.

The effects of deuteriation on out-of-plane modes depend on the position of the isotopic substitution. Deuteriation in the 3,4 positions causes a decrease of the 830 cm^{-1} (a_u) and 805 cm^{-1} (b_g) CH wag wave numbers to 661 and 683 cm^{-1} , respectively, without significant perturbations on the other modes. Deuteriation in the 2,5 positions lowers the CH wags and, as a consequence these modes couple strongly with the CH₂ wags.

b. Z-hexatriene. The computed equilibrium geometry of Z-hexatriene in T_1 is reported in Table III. Bond lengths are almost identical to those of *E*-hexatriene. In contrast, CCC angles are found to be slightly different and this is probably related to different nonbonded interactions in the two geometries.

The force field is very similar to that of *E*-hexatriene and is not shown here. The only differences, which are rather small, are limited to the CCC submatrix.

The computed vibrational wave numbers of Z-hexatriene- d_0 and the two deuteriated species of Z-hexatriene are given in Table XIII. For a_1 symmetry the dominant feature is the strongly mixed character of most of the planar vibra-

TABLE XI. Computed ground state wave numbers (cm^{-1}) of *Z*-hexatriene and its dideuteriated forms.^a

	ZHT- d_0	Z-3, 4- d_2 -HT	Z-2, 5- d_2 -HT
a_1	1664 C=C 1585 C=C 1424 CH ₂ sciss (CH rock) 1330 CH rock 1254 CH rock, C-C 1120 C-C, CH rock 882 CH ₂ rock 422 CCC bend 199 CCC bend	1653 C=C 1572 C=C 1423 CH ₂ sciss (CH rock) 1324 CH rock 1170 C-C 837 CD rock 899 CH ₂ rock 421 CCC bend 198 CCC bend	1649 C=C 1584 C=C 1405 CH ₂ sciss 1268 CH rock, C-C 952 CD rock 1154 CH rock 865 CH ₂ (CD) rock 418 CCC bend 194 CCC bend
b_1	1642 C=C 1465 CH ₂ sciss 1399 CH rock 1303 CH rock 1225 C-C (CH rock) 944 CH ₂ rock 707 CCC bend 420 CCC bend	1639 C=C 1411 CH ₂ sciss 1323 C-C (CCC bend) 1273 CH rock, C-C 1077 CD, CH ₂ rock 881 CH ₂ (CD) rock 685 CCC bend 406 CCC bend	1627 C=C 1459 CH ₂ sciss 1383 CH rock (CH ₂ sciss) 1032 CD (CH ₂) rock 1248 C-C (CH ₂ rock) 841 CH ₂ (CD) rock 670 CCC bend 410 CCC bend
a_2	1072 CH wag 987 CH wag 949 CH ₂ wag 670 CH ₂ =C (C=C) twist 363 C=C twist 153 C-C twist	1043 CH wag 801 CD wag 950 CH ₂ wag 648 CH ₂ = C twist 338 C=C twist 162 C-C twist	1026 CH wag 865 CD wag 953 CH ₂ wag 656 CH ₂ =C twist 349 C=C twist 136 C-C twist
b_2	1046 CH wag 952 CH ₂ wag 777 CH wag 615 CH ₂ =C twist 104 C-C twist	1044 CH wag 952 CH ₂ wag 666 CD wag, CH ₂ = C twist 558 CH ₂ =C twist, CD wag 104 C-C twist	881 CH, CD wag 954 CH ₂ wag 774 CD, CH wag 588 CH ₂ =C twist 98 C-C twist

^a See footnote a of Table X.

tions. In *Z*-hexatriene- d_0 the 1539 cm^{-1} band corresponds to a mixture of the different types of CC stretch. The four modes at 1331 , 1290 , 1207 , and 1089 cm^{-1} correspond to a combination of the two CH rocks and the two CC stretches. Upon deuteration the mixing involves the CH₂ rock as well, besides the four modes noted above. In the case of 2,5- d_2 -HT even the CH₂ scissoring becomes involved in the mixing.

In the b_1 space, we notice the same kind of mixing of internal coordinates. Interestingly, the mode with highest wave number has C-C stretch and CH₂ scissor components and its dominant character depends on the specific substitution.

In the out-of-plane modes we notice the same features as in EHT.

c. Twisted hexatriene. The optimized geometry of Tw hexatriene is given in Table III. It can be seen that the carbon skeleton parameters are essentially similar to those of the *E* geometry. Small differences are the slightly longer central C₃-C₄ bond and the slightly shorter C₁-C₂ bond.

At the twisted geometry, the molecular symmetry group is C_2 and it is not possible to distinguish rigorously between in-plane and out-of-plane coordinates. However, since the interactions between formerly in-plane and out-of-plane coordinates are very small, the distinction is still practically feasible.

The "in-plane" force field (not shown here) is very similar to that of the *E* form. The main difference pertains to off-diagonal elements between CC stretch coordinates: here only the couplings between stretching of adjacent bonds are non-zero.

The computed vibrational wave numbers are reported in Table XIV. The highest wave number of "a" type is about 1500 cm^{-1} , that is, very similar to the wave number of CH₂ scissor. Hence the modes with the two highest wave numbers are mixed and are of CC stretch and CH₂ scissor parentage. Furthermore, similarly to the *E* and *Z* isomers, we notice extensive mixing among CC stretch and CH rocks. A difference with respect to the *E* and *Z* forms is the inversion of the order of the CH₂ wag and the lowest CH wag mode.

The effect of deuteration is manifested in the lowering of a CH rock mode, which then mixes with a CH₂ rock mode, and in the lowering of a CH wag mode which mixes with CH₂ wag. The same comments hold for the "b"-type vibrations.

d. Rotation of normal coordinates, Duschinsky effect. The normal modes of T_1 are not the same as in S_0 . Electronic excitation induces a mixing of coordinates, as it was first pointed out by Duschinsky.⁵¹ The linear transformation relating the vibrational modes of S_0 and T_1 can be written as

$$Q^g = JQ' + \Delta Q^g, \quad (3)$$

TABLE XII. Computed wave numbers (cm^{-1}) of *E*-hexatriene and its deuterated forms in the T_1 state.^{a,b}

	EHT- d_0	<i>E</i> -3, 4- d_2 -HT	<i>E</i> -2, 5- d_2 -HT
a_g	1554 C-C, C=C _c	1525 C-C, C=C _c	1539 C-C, C=C _c
	1483 CH ₂ sciss	1479 CH ₂ sciss (C=C)	1481 CH ₂ sciss
	1369 CH rock (C=C)	1340 CH rock, C=C	1310 C=C (CCC bend)
	1321 CH rock (C-C, CCC bend)	1260 CH rock, CCC bend, C=C _c	1290 CH (CD) rock
	1193 C=C (C-C, CH rock)	964 CH ₂ , CD rock	1023 CH ₂ , CD rock
	1154 C=C _c (C=C)	1158 C=C _c , C=C	1153 C=C _c (C=C)
	939 CH ₂ rock	896 CD, CH ₂ rock	857 CD, CH ₂ rock
	448 CCC bend	438 CCC bend	431 CCC bend
	385 CCC bend	381 CCC bend	382 CCC bend
b_u	1473 CH ₂ sciss	1473 CH ₂ sciss	1472 CH ₂ sciss
	1430 CH rock, C-C	1413 CH rock, C-C (C=C)	1374 CH rock, C-C
	1323 CH rock (CCC bend)	1027 CD, CH ₂ rock	1304 C-C, CH rock
	1243 CH rock, C-C	1258 C-C (C=C, CH rock)	1170 C=C (CD rock)
	1139 C=C (C-C)	1167 C=C (C-C)	1010 CH ₂ , CD rock
	963 CH ₂ rock	907 CH ₂ , CD rock	879 CD, CH ₂ rock
	599 CCC bend	567 CCC bend	592 CCC bend
	177 CCC bend	175 CCC bend	173 CCC bend
a_u	1032 CH wag	996 CH wag	898 CH (CH ₂) wag
	830 CH wag	661 CD wag	780 CD, CH ₂ wag
	757 CH ₂ wag	769 CH ₂ wag	737 CH ₂ , CD wag
	565 C=CH ₂ twist	551 C=CH ₂ twist	563 C=CH ₂ twist
	193 C-C twist	186 C-C twist	173 C-C twist
	110 C=C twist	107 C=C twist	109 C=C twist
b_g	1012 CH wag	994 CH wag	883 CH (CH ₂) wag
	805 CH wag	683 CD wag	785 CD, CH ₂ wag
	754 CH ₂ wag	767 CH ₂ wag	745 CH ₂ , CD wag
	564 C=CH ₂ twist	534 C=CH ₂ twist	561 C=CH ₂ twist
	297 C-C twist	285 C-C twist	295 C-C twist

^a See footnote a of Table X.^b C=C_c denotes the central CC bond.

where Q^g and Q' are the normal modes of S_0 and T_1 , respectively, ΔQ^g is the displacement of T_1 equilibrium geometry projected on the S_0 normal coordinates and J is the rotation matrix expressing the Duschinsky effect.

To illustrate this, the calculated Duschinsky matrices for the a_g modes of the *E* isomers of hexatriene- d_0 and 3,4- d_2 -HT in the T_1 state with respect to the S_0 state are given in Table XV. It can be seen that indeed a considerable rotation of normal coordinates takes place. This rotation is not limited to the region of strong mixing in the ground state (1000 – 1300 cm^{-1}) but includes the C=C stretching modes of the ground state around 1600 cm^{-1} and the scissoring modes at 1419 cm^{-1} as well. As a result of the Duschinsky effect, the mixing of internal coordinates is considerably more extensive in the normal coordinates of T_1 than in those of S_0 . One of the effects of deuteration is seen to be a partial decoupling of the CD rock mode (1003 cm^{-1} in S_0) from the other ones and consequently a considerably smaller rotation of this mode.

3. Electronic energies of triplet states

Computed electronic energies, wave functions, and oscillator strengths for the lowest triplet states are given in

Tables XVI, XVII, and XVIII for the *E*, *Z*, and Tw forms. These calculations have been performed with both CNDO/S and QCFF/PI Hamiltonians. The results obtained by QCFF/PI are shown in Fig. 6. The geometries adopted were those of Table III, obtained by energy minimization.

The lowest triplet state T_1 is well described by a single configuration, namely the HOMO → LUMO (or $3 \rightarrow 4$) with respect to the ground configuration. The HOMO → LUMO configuration of *E*-hexatriene is considered the reference configuration, and in Tables XVI, XVII, and XVIII the other configurations are expressed relative to this reference. The energy of the T_1 state of *E*-hexatriene is found at 1.08 eV (24.9 kcal/mol) by QCFF/PI. It should be noted that experimentally the T_1 – S_0 energy gap is found at 2.03 eV .^{28,29} This underestimate of the T_1 – S_0 energy gap is due to our choice of using a different CI for the T_1 and the S_0 states in order to treat the two states on equal footing as far as potential energy surfaces are concerned. QCFF/PI and CNDO/S are parametrized in order to give correct vertical energy gaps within the same CI. Using the same CI for S_0 and T_1 we find a vertical T_1 – S_0 energy gap of 2.06 eV (QCFF/PI) and 2.33 eV (CNDO/S), in good agreement with the experimental results.

TABLE XIII. Computed wave numbers (cm^{-1}) of Z-hexatriene and its deuterated forms in the T_1 state.^a

	ZHT- d_0	Z-3, 4- d_2 -HT	Z-2, 5- d_2 -HT
a_1	1539 C-C, C=C _c (C=C) 1475 CH ₂ sciss 1331 CH rock (C=C _c) 1290 CH rock (C-C) 1207 C=C (C=C _c , CH rock) 1089 C=C _c (CH ₂ , CH rock) 893 CH ₂ rock (C=C _c) 416 CCC bend 194 CCC bend	1531 C-C, C=C _c , CH rock 1475 CH ₂ sciss 1254 C=C (C-C, CCC bend) 1294 CH rock, C=C _c 1160 C=C _c (C=C) 822 CD rock 914 CH ₂ , CD rock 415 CCC bend 192 CCC bend	1512 C-C, C=C _c , CH ₂ sciss 1470 CH ₂ sciss (C-C) 1308 CH rock (C-C) 968 CD rock (C=C) 1212 C=C, C=C _c 1110 C=C _c , C=C, CH ₂ rock 864 CH ₂ , CD rock 411 CCC bend 184 CCC bend
b_1	1504 C-C (CH ₂ sciss) 1460 CH ₂ sciss (C-C, C=C) 1362 CH rock (C=C) 1310 CH rock (C-C) 1112 C=C (C-C, CH rock) 955 CH ₂ rock 718 CCC bend 405 CCC bend	1480 CH ₂ sciss 1427 C-C, CH rock 1345 CH rock, C-C 1173 C=C (CD rock) 1042 CD, CH ₂ rock 893 CH ₂ , CD rock 694 CCC bend 392 CCC bend	1496 C-C (CH ₂ sciss) 1456 CH ₂ sciss (C-C, C=C) 1226 C=C (C-C, CD rock) 1291 CH rock (CCC bend) 1006 CH ₂ (CD) rock 894 CD, CH ₂ rock 687 CCC bend 395 CCC bend
a_2	1017 CH wag 830 CH wag 757 CH ₂ wag 575 C=CH ₂ twist 306 C-C twist 143 C=C twist	991 CH wag 693 CD (CH ₂) wag 763 CH ₂ (CD) wag 549 C=CH ₂ twist 293 C-C twist 136 C=C twist	875 CH wag 784 CD (CH ₂) wag 747 CH ₂ (CD) wag 575 C=CH ₂ twist 306 C-C twist 127 C=C twist
b_2	1015 CH wag 799 CH wag 744 CH ₂ wag 556 C=CH ₂ twist 149 C-C twist	992 CH wag 766 CH ₂ (CD) wag 643 CD (CH ₂) wag 538 C=CH ₂ twist 148 C-C twist	850 CH wag 777 CH ₂ , CD wag 729 CD, CH ₂ wag 554 C=CH ₂ twist 142 C-C twist

^a See footnote a of Tables X and XII.

In the *E* form, the second triplet is found at ~ 2.2 eV above T_1 and the transition $T_1 \rightarrow T_2$ is very weakly allowed. From ~ 2.0 eV up to ~ 6.5 eV we find only one triplet state, T_5 , with a large transition moment to T_1 . This state is located at about 4.5 eV (4.32 eV with QCFF/PI and 4.69 eV with CNDO/S) above T_1 . The T_5 state is obtained essentially by the same two configurations appearing in the wave functions of T_2 but with opposite phase. We wish to stress that the two methods used yield the same results and hence support each other. From the physical point of view these results are interesting, since they indicate that both transient absorption experiments and transient resonance Raman spectra of T_1 can be explained in terms of only one electronic transition, namely the $T_1 \rightarrow T_5$ transition. Essentially the same results have been obtained for the *Z* form.

At the Tw geometry the situation is somewhat different. In fact the system is composed of two allylic moieties interacting weakly and, accordingly, some of the excited states can be considered to be due to allylic excitations that are

degenerate. The state with the largest oscillator strength to T_1 is T_6 computed to be at ~ 5.5 eV above T_1 . This theoretical result is comparable with the experimentally found absorption spectrum for the allyl radical in an Ar matrix at 10 K.⁵⁰ A weak electronic absorption band was found at 408.5 nm and a strong one with a maximum at 213 nm, corresponding to energies of 3.03 and 5.81 eV, respectively. These values are rather close to the calculated values 3.40 and 5.49 eV (QCFF/PI) and 3.27 and 5.75 eV (CNDO/S). Hence at 90°, the state active in absorption and in resonance Raman scattering has shifted toward the far UV. Even for this geometry, CNDO/S and QCFF/PI methods yield equivalent results.

To get information on the energy of the lowest electronic states and in particular of the absorbing state, we have performed additional QCFF/PI calculations at angles $\varphi = 45^\circ$ and $\varphi = 60^\circ$ ($\varphi = 0^\circ$ is the *E* form). From the results (not shown here) it appears that the energy gap between the lowest triplet and the absorbing state increases smoothly go-

TABLE XIV. Computed wave numbers (cm^{-1}) of 90° twisted (Tw) hexatriene and its deuteriated forms in the T_1 state.^a

	HT- d_0	3, 4- d_2 -HT	2, 5- d_2 -HT
<i>a</i>	1497 C=C, C=C _c , C-C (CH ₂ sciss) 1483 CH ₂ sciss (C=C _c , C-C) 1317 CH rock, C-C (C=C) 1309 CH rock (CCC bend) 1192 C=C (C-C, CH rock) 1136 C=C _c (CH rock) 1035 CH wag 919 CH ₂ rock 821 CH ₂ wag 731 CH wag 582 C=CH ₂ twist 411 CCC bend 367 CCC bend 140 C-C twist - 168 C=C twist	1491 CH ₂ sciss 1472 C-C, C=C _c , CH rock 1310 C=C, C=C _c (C-C, CH rock) 1248 C=C _c (CH rock, CCC bend) 1170 C=C, C=C _c (C-C) 886 CD rock 1022 CH wag 913 CH ₂ rock 814 CH ₂ wag 654 CD wag 542 C=CH ₂ twist 404 CCC bend 352 CCC bend 139 C-C twist - 166 C=C twist	1488 CH ₂ sciss (C=C) 1459 C=C _c , C-C, CH rock 1275 C=C, CH rock 1285 C-C, C=C, CCC bend 999 CH ₂ , CD rock 1135 C=C _c (CD rock) 830 CD (CH ₂) wag 852 CD (CH ₂) rock 819 CH ₂ wag 726 CH wag 575 C=CH ₂ twist 405 CCC bend 360 CCC bend 135 C-C twist - 153 C=C twist
<i>b</i>	1493 CH ₂ sciss (C=C) 1460 C-C, CH rock 1311 CH rock (C-C, CCC bend) 1300 CH rock (C-C, C=C) 1194 C=C, C-C (CH rock) 1040 CH wag 957 CH ₂ rock 827 CH ₂ wag 779 CH wag 641 CCC bend (C=CH ₂ twist) 587 C=CH ₂ twist (CCC bend) 369 CCC bend 155 C-C twist	1491 CH ₂ sciss 1442 C-C, CH rock 1053 CD (CH ₂) rock 1296 C=C, C-C 1212 CH rock (C=C, C-C) 1023 CH wag 898 CH ₂ (CD) rock 816 CH ₂ wag 719 CD wag (CCC bend) 548 C=CH ₂ twist (CCC bend) 579 CCC bend (C=CH ₂ twist) 353 CCC bend 154 C-C twist	1488 CH ₂ sciss (C=C) 1425 C-C, CH rock 1287 CH rock, C-C 1267 C=C (CH rock) 1009 CH ₂ , CD rock 846 CD, CH wag 895 CD, CH ₂ rock 823 CH ₂ , CD wag 760 CD, CH ₂ wag 631 CCC bend (C=CH ₂ twist) 581 C=CH ₂ twist (CCC bend) 363 CCC bend 148 C-C twist

^a See footnote a of Tables X and XII.

ing from E to 45° and 60° geometries. However, even at $\varphi = 60^\circ$ this energy gap is still closer to that to the E form than to that of the Tw form.

4. Activity of totally symmetric modes in transient resonance Raman spectra

In this section we wish to determine, on the basis of the present calculations, the most active modes in the resonance Raman scattering of the transient T_1 state with an exciting wavelength in the region 315–317.5 nm as reported previously.³⁵ At the same time we can identify the modes that are most prominent in the vibrational structure of the intense $T_1 \rightarrow T_n$ (with $n = 5$ in the E and Z forms and $n = 6$ in the Tw form) transition.

A wavelength of 315 nm corresponds to an energy of 3.94 eV. Since this energy is sufficiently close to the calculated energy gap of the intense $T_1 \rightarrow T_5$ transition, found to be at ~ 4.3 eV, the scattering tensor is dominated by the contribution of this strongly allowed transition, and the experimental wavelength corresponds to a resonance situation. Therefore, it is a good approximation to restrict our attention to this

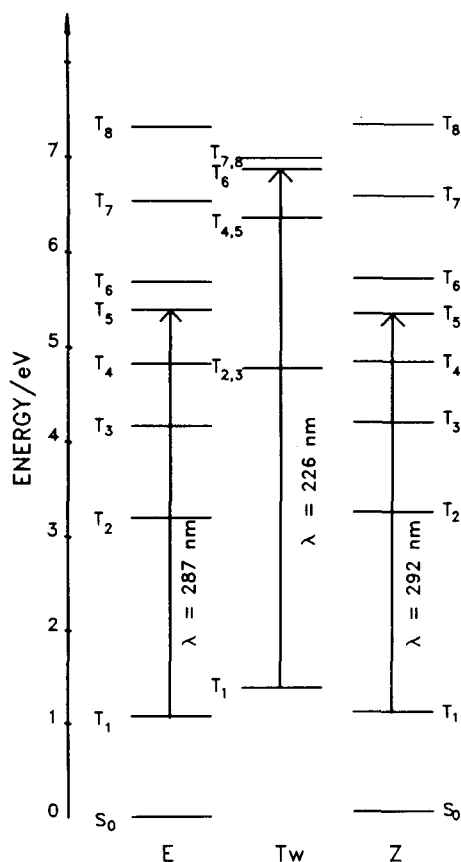
transition. In this case the most active modes are those gaining their scattering intensity from the Franck–Condon mechanism, namely the totally symmetric (TS) modes. Nontotally symmetric (NTS) modes, whose fundamentals gain intensity by vibronic coupling, are likely to show a relatively small activity.

The activity of TS modes may be due to displacement between the equilibrium geometries of T_1 and T_5 , or to the difference between vibrational wave numbers in the two states. It is well known that displacement usually gives rise to a much larger Franck–Condon activity than wave number differences. In the case of hexatriene, according to QCFF/PI calculations the vibrational wave numbers in T_5 never differ from the wave numbers in T_1 by more than 4%; in most cases the difference is $\leq 1\%$. Hence displacement is by far the most important factor governing Franck–Condon factors. For simplicity we thus assume the wave numbers of vibrational modes to be equal in the two states and we consider only the contribution of displacement.

The parameter governing the contribution of displacement to the Franck–Condon factor is γ , which is the ratio

TABLE XV. Duschinsky matrices of the rotation of the totally symmetric vibration modes (a_g) upon excitation from S_0 to T_1 .

		<i>E</i> -hexatriene- <i>d</i> ₀ Triplet state <i>T</i> ₁								
	<i>ν</i>	1554	1483	1369	1154	1321	1193	939	448	385
Ground state	1664	− 0.86	0.32	− 0.20	− 0.10	− 0.24	− 0.18			
	1594	− 0.21	− 0.66	0.27	− 0.61	− 0.20	0.16			
	1419	− 0.14	− 0.63	− 0.64	0.40					
	1350	0.35	0.19	− 0.65	− 0.62		− 0.16			
	1305	0.24		0.12	0.20	− 0.83	− 0.44			
	1206		− 0.12	0.22	− 0.12	0.45	− 0.84	− 0.11		
	941				0.12			− 0.98		
	461								− 0.99	− 0.10
	393								0.10	− 0.99
		<i>E</i> -3, 4- <i>d</i> ₂ -hexatriene Triplet state <i>T</i> ₁								
	<i>ν</i>	1525	1158	1479	1260	1340	964	896	438	381
Ground state	1647	− 0.86	− 0.12	0.34	− 0.25	0.26				
	1577	− 0.28	0.60	− 0.60	− 0.33	− 0.22	0.18			
	1419	0.11	0.41	0.69	− 0.19	− 0.55				
	1303	0.38	− 0.12		− 0.84	0.34				
	1255	0.13	0.63	0.20	0.27	0.68				
	1003		− 0.19		0.10		0.97			
	879							− 0.99		
	449								− 0.99	− 0.11
	390								− 0.11	0.99

FIG. 6. Calculated energy level diagram for the triplet states of 1,3,5-hexatriene in the planar *E*, *Z*, and twisted (*Tw*) forms and the strongest optically allowed transitions.

between half of the Stokes shift and the vibrational quantum. Thus if $Q_i(1)$ and $Q_i(n)$ are the equilibrium geometries along the (mass-weighted) normal coordinates Q_i in the states T_1 and T_n , respectively, the parameter γ_i associated to Q_i is given by

$$\gamma_i = (\omega_i/2\hbar) [Q_i(1) - Q_i(n)]^2, \quad (4)$$

where ω_i is in rad/s. The parameter $Q_i = [Q_i(1) - Q_i(n)]$ is simply the projection onto the normal coordinate Q_i of the change of equilibrium Cartesian coordinates in the two states. That is, Q_i can be written as¹³

$$Q_i = [\mathbf{x}(1) - \mathbf{x}(n)] \mathbf{M}^{1/2} \mathbf{L}_i(T_1), \quad (5)$$

where $\mathbf{x}(m)$ is the $3N$ dimensional vector of the equilibrium Cartesian coordinates in the state T_m , \mathbf{M} is the $3N \times 3N$ diagonal matrix of the atomic masses and \mathbf{L}_i is the $3N$ vector describing normal coordinates Q_i in terms of mass-weighted Cartesian coordinates.

We have computed the parameters γ_i for each TS normal coordinate at the *E*, *Z*, and *Tw* geometries and for the three isotopic species. The results are reported in Tables XIX, XX, and XXI. We consider first hexatriene- d_0 .

In the *E* geometry the largest γ 's are associated with the modes at 1554 and 1154 cm^{-1} ; smaller γ 's are found for the modes at 1369, 1321, and 447 cm^{-1} . In the *Z* geometry the largest γ 's are due to the modes at 1539 and 1290 cm^{-1} ; smaller values are found for the modes at 1475, 1207, and 194 cm^{-1} . At the *Tw* geometry the modes with the largest γ 's are found at 1483, 1317, and 1309 cm^{-1} . The γ 's for this geometry are smaller than the largest values found in the *E* and *Z* geometries. Hence the three geometries have each a

TABLE XVI. Relative energies (E , eV), wave functions, and $T_1 \rightarrow T_n$ oscillator strengths (f) for the lowest triplet states of *E*-hexatriene.

Type		CNDO/S		QCFF/PI		Wave function
		E	f	E	f	
T_1	B_u	0.00	...	0.00	...	0.96 (ground)
T_2	A_g	2.25	0.01	2.12	0.00	$0.69(2 \rightarrow 3) + 0.63(4 \rightarrow 5)$
T_3	B_u	3.15	0.00	3.09	0.00	$-0.64(1 \rightarrow 3) + 0.58(4 \rightarrow 6)$
T_4	B_u	3.93	0.00	3.75	0.00	$-0.54(2 \rightarrow 4) - 0.51(2,4 \rightarrow 3,5) - 0.50(3 \rightarrow 5)$
T_5	A_g	4.69	0.60	4.32	0.85	$-0.67(2 \rightarrow 3) + 0.69(4 \rightarrow 5)$
T_6	A_g	4.77	0.32	4.61	0.01	$0.42(1 \rightarrow 4) + 0.47(1,4 \rightarrow 3,5) - 0.40(3 \rightarrow 6)$
T_7	B_u	6.25	0.00	5.46	0.00	$0.67(1 \rightarrow 3) + 0.71(4 \rightarrow 6)$
T_8	B_u	6.97	0.00	6.24	0.00	$0.76(2 \rightarrow 4) - 0.56(3 \rightarrow 5)$

* MOs are labeled in the order of increasing energy. We have included with singly excited configurations also excited configurations in which one excitation is internal to the half-occupied orbital space.

TABLE XVII. Relative energies (E , eV), wave functions, and $T_1 \rightarrow T_n$ oscillator strengths (f) for the lowest triplet states of *Z*-hexatriene.

Type		CNDO/S		QCFF/PI		Wave functions ^a
		E	f	E	f	
T_1	B_2	0.00	...	0.00	...	0.96 (ground)
T_2	A_1	2.20	0.01	2.14	0.00	$0.68(2 \rightarrow 3) - 0.63(4 \rightarrow 5)$
T_3	B_2	3.14	0.01	3.09	0.00	$-0.65(1 \rightarrow 3) - 0.59(4 \rightarrow 6)$
T_4	B_2	3.87	0.01	3.73	0.00	$0.53(2 \rightarrow 4) + 0.54(2,4 \rightarrow 3,5) - 0.50(3 \rightarrow 5)$
T_5	A_1	4.57	0.61	4.24	0.66	$-0.67(2 \rightarrow 3) - 0.70(4 \rightarrow 5)$
T_6	A_1	4.75	0.07	4.61	0.01	$0.43(1 \rightarrow 4) + 0.41(3 \rightarrow 6)$
T_7	B_2	6.25	0.10	5.47	0.13	$0.67(1 \rightarrow 3) - 0.71(4 \rightarrow 6)$
T_8	B_2	6.96	0.09	6.23	0.04	$-0.77(2 \rightarrow 4) - 0.53(3 \rightarrow 5)$

* See footnote a of Table XVI.

TABLE XVIII. Relative energies (E , eV), wave functions, and $T_1 \rightarrow T_n$ oscillator strengths (f) for the lowest triplet states of 90° twisted (Tw) hexatriene.

Type		CNDO/S		QCFF/PI		Wave functions ^a
		E	f	E	f	
T_1	B	0.00	...	0.00	...	0.97 (ground)
T_2	A	3.27	0.03	3.40	0.01	$0.55(2 \rightarrow 3) + 0.53(1 \rightarrow 4) + 0.46(4 \rightarrow 5) - 0.44(3 \rightarrow 6)$
T_3	B	3.34	0.00	3.40	0.00	$-0.53(2 \rightarrow 4) - 0.53(1 \rightarrow 3) + 0.47(4 \rightarrow 6) - 0.46(3 \rightarrow 5)$
T_4	B	5.40	0.00	4.98	0.00	$-0.42(2,4 \rightarrow 3,5) + 0.41(1,4 \rightarrow 3,6) + 0.40(2,3 \rightarrow 4,5)$
T_5	A	5.40	0.00	4.98	0.00	$0.42(1,4 \rightarrow 3,5) - 0.42(2,4 \rightarrow 3,6) - 0.40(1,3 \rightarrow 4,5)$
T_6	A	5.75	0.85	5.49	0.73	$-0.55(2 \rightarrow 3) - 0.34(1 \rightarrow 4) + 0.57(4 \rightarrow 5) - 0.50(3 \rightarrow 6)$
T_7	B	5.61	0.00	$-0.68(2 \rightarrow 4) + 0.68(1 \rightarrow 3)$
T_8	A	5.62	0.02	$0.59(2 \rightarrow 3) - 0.75(1 \rightarrow 4)$

* See footnote a of Table XVI.

TABLE XIX. Stokes shift parameters γ for the $T_1 \rightarrow T_5$ transition of totally symmetric normal modes (a_g) of *E*-hexatriene and its deuteriated forms.^a

EHT- d_0		<i>E</i> -3, 4- d_2 -HT		<i>E</i> -2, 5- d_2 -HT	
ν (cm ⁻¹)	γ	ν (cm ⁻¹)	γ	ν (cm ⁻¹)	γ
1554	0.24	1525	0.24	1539	0.27
1483	0.04	1479	0.08	1481	0.04
1369	0.05	1340	0.18	1310	0.06
1321	0.06	1260	0.01	1290	0.04
1193	0.03	964	0.04	1023	0.02
1154	0.25	1158	0.07	1153	0.28
939	0.01	896	0.08	857	0.00
447	0.06	438	0.04	431	0.06
385	0.04	381	0.04	382	0.05

* Computed by QCFF/PI.

TABLE XX. Stokes shift parameters γ for the $T_1 \rightarrow T_5$ transition of totally symmetric normal modes (a_i) of *Z*-hexatriene and its deuteriated forms.^a

ZHT- d_0		<i>Z</i> -3, 4- d_2 -HT		<i>Z</i> -2, 5- d_2 -HT	
ν (cm ⁻¹)	γ	ν (cm ⁻¹)	γ	ν (cm ⁻¹)	γ
1539	0.21	1531	0.22	1512	0.21
1475	0.06	1475	0.07	1470	0.14
1331	0.00	1254	0.22	1308	0.07
1290	0.37	1294	0.22	968	0.02
1207	0.11	1160	0.06	1212	0.32
1089	0.04	822	0.00	1110	0.06
893	0.03	914	0.04	864	0.01
416	0.04	415	0.04	411	0.04
194	0.10	192	0.10	189	0.10

* Computed by QCFF/PI.

TABLE XXI. Stokes shift parameters γ for the $T_1 \rightarrow T_0$ transition of totally symmetric normal modes (a) of 90° twisted (Tw) hexatriene and its deuteriated forms.^a

HT- d_0		3, 4- d_2 -HT		2, 5- d_2 -HT	
ν (cm ⁻¹)	γ	ν (cm ⁻¹)	γ	ν (cm ⁻¹)	γ
1497	0.00	1491	0.03	1488	0.04
1483	0.09	1472	0.06	1459	0.05
1317	0.12	1310	0.26	1275	0.06
1309	0.15	1248	0.02	1285	0.22
1192	0.00	1170	0.00	999	0.01
1136	0.00	886	0.01	1135	0.00
1035	0.00	1022	0.00	830	0.00
919	0.04	913	0.04	852	0.04
821	0.00	814	0.00	819	0.00
731	0.02	654	0.02	726	0.02
582	0.00	542	0.01	575	0.00
411	0.04	404	0.03	405	0.04
367	0.01	352	0.02	360	0.01
140	0.00	139	0.00	135	0.00
- 168	...	- 166	...	- 153	...

^aComputed by QCFF/PI.

different pattern for the γ values and the Tw form shows a markedly different pattern from the other two forms.

For the deuteriated species the values of the γ 's depend on the position of deuteration. The high γ values for the modes above 1500 cm⁻¹ persist for both the 3,4- and 2,5-deuteriated species in the *E* and *Z* forms. The largest γ 's of the *E* form of 2,5- d_2 -HT resemble those of *E*-hexatriene- d_0 whereas the γ 's of *E*-3,4- d_2 -HT are markedly changed. The *Z* form of 3,4- d_2 -HT shows two large γ 's for the modes at 1254 and 1294 cm⁻¹ and that of 2,5- d_2 -HT shows particularly one large γ at 1212 cm⁻¹. In the Tw geometry deuteration at either 3,4- or 2,5-positions induces a pattern with essentially one large γ around 1300 cm⁻¹.

V. DISCUSSION OF RAMAN SPECTRA

A. Ground state

The ground state Raman and infrared spectra of the *E* and *Z* isomers of hexatriene- d_0 have been discussed in detail previously.²⁵ As the present calculations do not change that general picture this shall not be discussed further. The theoretical calculations on the 3,4- and 2,5-deuteriated species agree in general quite well with the experimental data. An exception is the b_g (EHT) or b_2 (ZHT) C=CH₂ twist mode computed in the region 585–615 cm⁻¹, which experimentally is found at much lower wave numbers.

B. Triplet state

We here discuss the experimental resonance Raman results of hexatriene- d_0 and the deuteriated hexatrienes in their triplet states in the light of the results from the theoretical calculations.

For heptatriene³² and hexatriene⁵² the triplet–triplet difference absorption spectrum extends experimentally from 350 to at least 300 nm with a first maximum in the region 300–315 nm. No experimental data below 300 nm are available. Calculations with QCFF/PI as well as CNDO/S Ham-

iltonians agree qualitatively in the prediction of a strongly allowed electronic $T_1 \rightarrow T_3$ transition for the planar *E* and *Z* geometries with a transition energy in the region 4.24–4.69 eV corresponding to 264–292 nm. In the 90° twisted geometry the lowest transition with an oscillator strength comparable to that of this strong transition is predicted around 5.49–5.75 eV, i.e., 216–226 nm. From these results it seems likely that the molecules responsible for the strong absorbance around 315 nm are in or close to the *E* or *Z* geometries. Furthermore, a much stronger resonance enhancement in Raman scattering must be expected from molecules being in the *E* and *Z* than in the twisted geometry when exciting with a wavelength around 315 nm. In fact, the distance in energy of the exciting laser frequency from the calculated Tw transition is about three times larger than from that of the *E* or *Z* transitions. This indicates that Raman intensity should be about an order of magnitude smaller for molecules in the Tw than for those in the *E* or *Z* geometries.

We have plotted the calculated values of the γ 's for the *E*, *Z* and Tw forms of HT- d_0 , 3,4- d_2 -HT and 2,5- d_2 -HT as stick diagrams as a function of wave number together with the experimentally obtained resonance Raman spectra in the T_1 state in Figs. 7–9. In the comparison of the stick diagrams the different $T_1 \rightarrow T_n$ transition energies for the *E* and *Z* forms compared to the Tw form should be kept in mind. Furthermore, any assignment of calculated vibrational bands to experimentally observed ones must be considered qualitative, when the computed γ values or the observed intensities are not large. We recall that besides the approximations mentioned above we have assumed no Duschinsky ef-

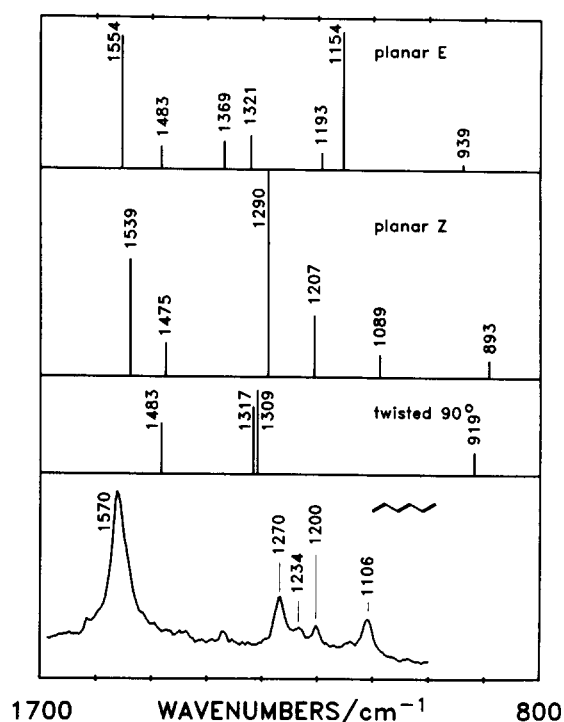


FIG. 7. Calculated wave numbers and relative γ -values for the totally symmetric vibrational modes of the *E* form in resonance with the $T_1 \rightarrow T_3$ transition, the *Z* form ($T_1 \rightarrow T_5$) and the Tw form ($T_1 \rightarrow T_6$) of 1,3,5-hexatriene- d_0 together with the experimental triplet state time-resolved resonance Raman spectrum.³⁵

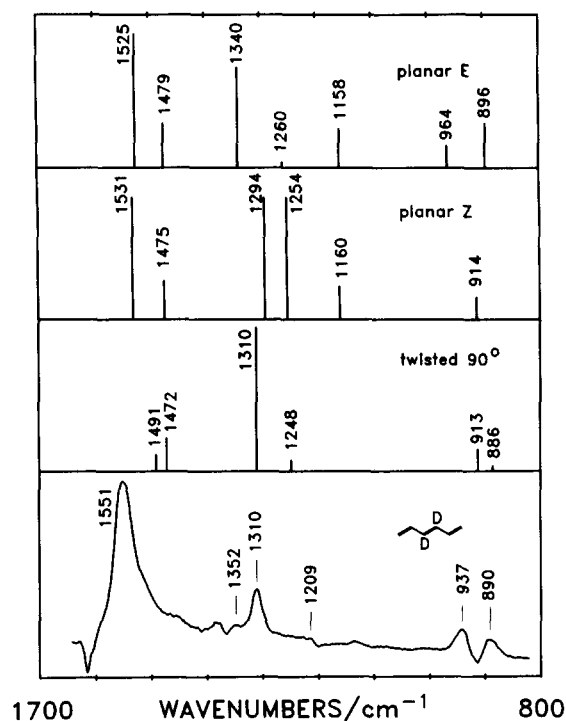


FIG. 8. Calculated wave numbers and relative γ -values for the totally symmetric vibrational modes of the *E* form in resonance with the $T_1 \rightarrow T_5$ transition, the *Z* form ($T_1 \rightarrow T_3$) and the Tw form ($T_1 \rightarrow T_6$) of 3,4-*d*₂-1,3,5-hexatriene together with the experimental triplet state time-resolved resonance Raman spectrum [see Fig. 4(A)].

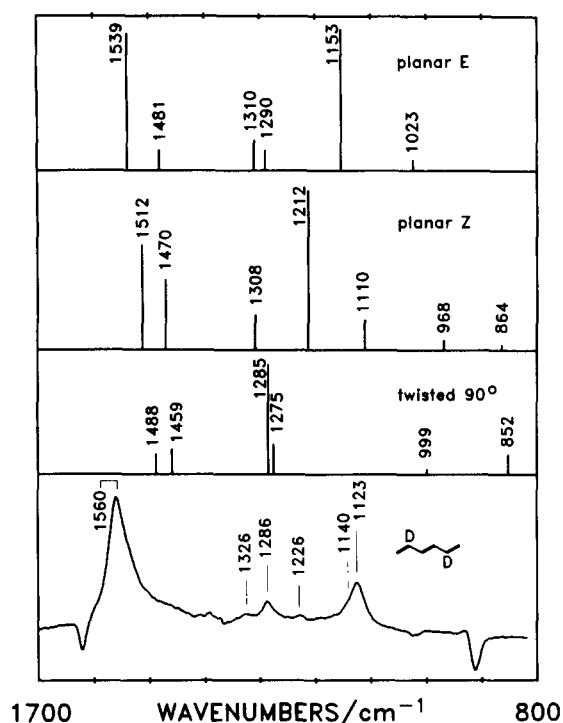


FIG. 9. Calculated wave numbers and relative γ -values for the totally symmetric vibrational modes of the *E* form in resonance with the $T_1 \rightarrow T_5$ transition, the *Z* form ($T_1 \rightarrow T_3$) and the Tw form ($T_1 \rightarrow T_6$) of 2,5-*d*₂-1,3,5-hexatriene together with the experimental triplet state time-resolved resonance Raman spectrum [see Fig. 4 (C)].

fect between T_1 and T_n . We have checked that such an effect is not very important in planar forms since diagonal terms of the rotation matrix are ≥ 0.90 . We first discuss hexatriene-*d*₀.

From the appearance of strong Raman bands above 1500 cm⁻¹ we must conclude that a considerable part of the hexatriene molecules is in the *E* or *Z* geometry or in both. The Tw form cannot account for any band with strong intensity above 1500 cm⁻¹. From the appearance of the experimental spectrum we feel confident to assign the experimental band at 1106 cm⁻¹ to the calculated band of the *E* form at 1154 cm⁻¹ and the experimental band at 1270 cm⁻¹ to that of the *Z* form at 1290 cm⁻¹ and finally the strong experimental band at 1570 cm⁻¹ to a superposition of calculated bands from the *E* form at 1554 and the *Z* form at 1539 cm⁻¹. Finally the experimental band at 1200 cm⁻¹ is tentatively assigned to the *Z* form at 1207 cm⁻¹. Thus the resonance Raman spectrum can be explained as a superposition of the spectra of the *E* and *Z* forms without any contribution of the Tw form. It follows that the *E* and *Z* forms are present in T_1 with comparable weight. Although no indication of the Tw form is found in the resonance Raman spectrum, caution should be exerted in ruling out its presence, considering its less favorable resonance conditions, compared to the other two forms, under the present experimental conditions.

Turning now to the deuteriated species, the experimental band of *E*-3,4-*d*₂-HT at 1551 cm⁻¹ can be assigned to either the *E* or the *Z* form. It should be noted that the calculations correctly predict a lower wave number of this band

for the 3,4-deuteriated species than for EHT-*d*₀. The other rather strong experimental band at 1310 cm⁻¹ can be assigned to either a calculated band of the *E* form at 1340 cm⁻¹ or one of the two bands of the *Z* form at 1254 and 1294 cm⁻¹. Furthermore, a weak contribution from the mode calculated at 1310 cm⁻¹ from the twisted form to this band cannot be ruled out completely. The experimental bands at 937 and 890 cm⁻¹ can be assigned either to the two bands of the *E* form at 964 and 896 cm⁻¹ or to the band of the *Z* form at 914 cm⁻¹. Finally the weak experimental bands at 1209 and 1352 cm⁻¹ can probably be explained by some of the weaker calculated bands. Hence the experimental resonance Raman spectrum of *E*-3,4-*d*₂-HT indicates clearly the presence of the *E* form but does not allow firm conclusions on that of the *Z* form. Again the presence of the Tw form cannot be excluded from the present experiments. The experimentally found triplet state resonance Raman spectrum of *E*-2,5-*d*₂-HT leads to similar conclusions. Again the strong band at 1560 cm⁻¹ can be assigned to the *E* form (1539 cm⁻¹) and/or the *Z* form (1512 cm⁻¹). The other strong band at 1123 cm⁻¹ has a shoulder at ~ 1140 cm⁻¹. We assign the 1123 cm⁻¹ band to the calculated band of the *E* form at 1153 cm⁻¹ and the shoulder to the 1212 cm⁻¹ band of the *Z* form. The weaker experimental band at 1286 cm⁻¹ can then either be assigned to calculated bands at 1290 or 1310 cm⁻¹ of the *E* form or to the band at 1308 cm⁻¹ of the *Z* form. Again, a contribution from the band calculated at 1285 cm⁻¹ from the twisted form cannot be ruled out definitively. Hence the presence of the *E* form of *E*-2,5-*d*₂-HT is clearly indicated,

whereas that of the *Z* form seems likely in smaller concentration.

In conclusion, the calculated wave numbers of the totally symmetric normal modes in the T_1 state and the calculated γ values for the $T_1 \rightarrow T_n$ transitions from the planar *E* and *Z* forms of hexatriene- d_0 and dideuteriated hexatrienes provide a good basis for the understanding of the triplet state resonance Raman spectra. A comparison between the calculated and experimentally found resonance Raman spectra of EHT- d_0 , *E*-3,4- d_2 -HT, and *E*-2,5 d_2 -HT in their triplet state T_1 clearly indicates that the predominant form in the triplet state of these molecules probed by the present resonance Raman experiments is the planar *E* form. The presence of the *Z* form is indicated as well, particularly by the spectra of EHT- d_0 , which cannot be explained by the *E* form only. The twisted Tw form may be present, but its Raman spectrum is expected to be weaker by about an order of magnitude under the present experimental conditions.

VI. CONCLUSION

We have recorded the time-resolved resonance Raman spectra of *E*-3,4- d_2 -hexatriene and *E*-2,5- d_2 -hexatriene in their lowest T_1 states. These spectra together with the spectrum of *E*-hexatriene- d_0 obtained previously³⁵ are compared with results from theoretical calculations, based on QCFF/PI and CNDO/S Hamiltonians, in an attempt to determine the stable structures of these flexible molecules. To this end equilibrium geometries and minimum energies are computed for the planar *E* and *Z* forms in the S_0 and T_1 states and for the Tw form in the T_1 state. The relative energies of Tw and *Z* optimized geometries are found to be 7.1 and 1.1 kcal/mol, respectively, taking the energy of the *E* geometry as reference.

To allow interpretation of resonance Raman spectra, energies of higher triplet states up to T_8 are calculated for the three geometries together with oscillator strengths for the $T_1 \rightarrow T_n$ transitions. Only one strong transition is found up to 6 eV above T_1 for each of the three forms (*E*, *Z* and Tw). For the *E* and *Z* forms this transition is the $T_1 \rightarrow T_5$ transition at about 4.3 eV, while for the Tw form the $T_1 \rightarrow T_6$ transition at about 5.5 eV is by far the strongest. A comparison of the computed energy gaps with the maximum in the $T_1 \rightarrow T_n$ difference absorption spectrum, located around 300 nm,⁵² indicates that both planar forms (*E* and *Z*) are populated in the T_1 state.

The resonance Raman intensities of a_g vibrational modes in T_1 can then be estimated under the assumption of a dominant Franck-Condon mechanism for each of the three forms *E*, *Z*, and Tw and for HT- d_0 , 3,4- d_2 -HT and 2,5- d_2 -HT. The observed resonance Raman spectra can be satisfactorily attributed to the *E* and *Z* forms. This result, which modifies a previous suggestion,³⁵ confirms that in T_1 the *E* and *Z* forms are substantially populated. While there is no direct evidence for Tw population in T_1 , the weakness of the $T_1 \rightarrow T_n$ transition near 4.3 eV for this form, compared to the *E* and *Z* forms, cautions against the conclusion that this population is negligible.

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- ¹B. S. Hudson, B. E. Kohler, and K. Schulten, in *Excited States*, edited by E. C. Lim (Academic, New York, 1982), Vol. 6, pp. 1-95.
- ²R. J. Hemley, A. C. Lasaga, V. Vaida, and M. Karplus, *J. Phys. Chem.* **92**, 945 (1988).
- ³V. Bonačić-Koutecký and S. Ishimaru, *J. Am. Chem. Soc.* **99**, 8134 (1977).
- ⁴M. Said, D. Maynau, and J.-P. Malrieu, *J. Am. Chem. Soc.* **106**, 580 (1984).
- ⁵M. A. C. Nascimento and W. A. Goddard III, *Chem. Phys. Lett.* **60**, 197 (1979).
- ⁶M. A. C. Nascimento and W. A. Goddard III, *Chem. Phys.* **36**, 147 (1979).
- ⁷C. W. Bock, Y. N. Panchenko, S. V. Krasnoshchiokov, and V. I. Pupyshchev, *J. Mol. Struct. (Theochem)* **148**, 131 (1986).
- ⁸B. A. Hess, Jr. and L. J. Schaad, *J. Am. Chem. Soc.* **105**, 7500 (1983).
- ⁹G. Fogarazi, P. Szalay, P. P. Liescheski, J. E. Boggs, and P. Pulay, *J. Mol. Struct.* **151**, 341 (1987).
- ¹⁰P. G. Szalay, A. Karpfen, and H. Lischka, *J. Chem. Phys.* **87**, 3530 (1987).
- ¹¹C. W. Bock, P. George, and M. Trachtman, *J. Mol. Struct. (Theochem)* **109**, 1 (1984).
- ¹²I. Ohmine and K. Morokuma, *J. Chem. Phys.* **73**, 1907 (1980).
- ¹³(a) A. Warshel and M. Karplus, *Chem. Phys. Lett.* **17**, 7 (1972); (b) A. Warshel and P. Dauber, *J. Chem. Phys.* **66**, 5477 (1977).
- ¹⁴(a) A. C. Lasaga, R. J. Aerni, and M. Karplus, *J. Chem. Phys.* **73**, 5230 (1980); (b) R. J. Hemley, B. R. Brooks, and M. Karplus, *J. Chem. Phys.* **85**, 6550 (1986).
- ¹⁵M. Traetteberg, *Acta Chem. Scand.* **22**, 628 (1968).
- ¹⁶M. Traetteberg, *Acta Chem. Scand.* **22**, 2294 (1968).
- ¹⁷L. R. Lippencott, C. E. White, and J. P. Sibilis, *J. Am. Chem. Soc.* **80**, 2926 (1958).
- ¹⁸L. R. Lippencott and T. E. Kenney, *J. Am. Chem. Soc.* **84**, 3641 (1962).
- ¹⁹Yu. N. Panchenko, Yu. A. Pentin, and E. B. Rusach, *Russ. J. Phys. Chem.* **49**, 1536 (1975).
- ²⁰Yu. N. Panchenko, P. Czászár, and F. Török, *Acta Chim. Hung.* **113**, 149 (1983).
- ²¹Yu. N. Panchenko and E. B. Rusach, *Moscow Univ. Chem. Bull.* **20**, 221 (1979).
- ²²D. Raković, S. A. Stepanyan, L. A. Gribov, and Yu. N. Panchenko, *J. Mol. Struct.* **90**, 363 (1982).
- ²³B. U. Curry, Ph.D. thesis, University of California, Berkeley, 1983.
- ²⁴R. McDiarmid and A. Sabljic, *J. Phys. Chem.* **91**, 276 (1986).
- ²⁵F. W. Langkilde, R. Wilbrandt, O. F. Nielsen, D. H. Christensen, and F. M. Nicolaisen, *Spectrochim. Acta Part A* **43**, 1209 (1987).
- ²⁶Y. Furukawa, H. Takeuchi, I. Harada, and M. Tasumi, *J. Mol. Struct.* **100**, 341 (1983).
- ²⁷M. Tasumi and M. Nakata, *J. Mol. Struct.* **126**, 111 (1985).
- ²⁸D. F. Evans, *J. Chem. Soc.* **1960**, 1735.
- ²⁹N. G. Minnaard and E. Havinga, *Recl. Trav. Chim.* **92**, 1179 (1973).
- ³⁰R. McDiarmid, A. Sabljic, and J. P. Doering, *J. Am. Chem. Soc.* **107**, 826 (1985).
- ³¹A. A. Gorman, I. Hamblett, M. Irvine, P. Raby, M. C. Standen, and S. Yeates, *J. Am. Chem. Soc.* **107**, 4404 (1985).
- ³²F. W. Langkilde, R. Wilbrandt, and N.-H. Jensen, *Chem. Phys. Lett.* **111**, 372 (1984).
- ³³F. W. Langkilde, N.-H. Jensen, and R. Wilbrandt, *Chem. Phys. Lett.* **118**, 486 (1985).
- ³⁴F. W. Langkilde, N.-H. Jensen, R. Wilbrandt, A. M. Brouwer, and H. J. C. Jacobs, *J. Phys. Chem.* **91**, 1029 (1987).

- ³⁵F. W. Langkilde, N.-H. Jensen, and R. Wilbrandt, *J. Phys. Chem.* **91**, 1040 (1987).
- ³⁶J. Levisalles, *Bull. Soc. Chim. France* **1957**, 997.
- ³⁷A. M. Brouwer, J. Cornelisse, and H. J. C. Jacobs, *J. Photochem. Photobiol. A* **42**, 313 (1988).
- ³⁸A. B. Myers, R. A. Mathies, D. J. Tannor, and E. J. Heller, *J. Chem. Phys.* **77**, 3857 (1982).
- ³⁹A. Warshel and M. Karplus, *J. Am. Chem. Soc.* **94**, 5612 (1972).
- ⁴⁰A. Warshel and M. Levitt, QCPE No. 247, Indiana University 1974.
- ⁴¹A. Warshel, in *Modern Theoretical Chemistry*, edited by A. Segal (Plenum, New York, 1977), Vol. 7, Part A p. 133.
- ⁴²B. Hudson and J. Andrews, *Chem. Phys. Lett.* **63**, 493 (1979).
- ⁴³G. Orlandi and F. Zerbetto, *Chem. Phys.* **123**, 175 (1988).
- ⁴⁴D. R. Armstrong, R. Fortune, P. G. Perkins, and J. J. P. Stewart, *J. Chem. Soc. Faraday Trans. 2* **68**, 1839 (1972).
- ⁴⁵F. Zerbetto, *Dissertazione di Dottorato di Ricerca*, Bologna, 1986.
- ⁴⁶C. A. Coulson and H. C. Longuet-Higgins, *Proc. R. Soc. London Ser. A* **191**, 39 (1947).
- ⁴⁷F. Negri, G. Orlandi, F. Zerbetto, and M. Zgierski (unpublished results).
- ⁴⁸H. C. Longuet-Higgins and J. A. Pople, *Proc. Phys. Soc. London Sect. A* **68**, 591 (1955).
- ⁴⁹J. Del Bene and H. H. Jaffe, *J. Chem. Phys.* **48**, 1807 (1968).
- ⁵⁰G. Maier, H. P. Reisenauer, B. Rohde, and K. Dehnicke, *Chem. Ber.* **116**, 732 (1983).
- ⁵¹F. Duschinsky, *Acta Physicochim. USSR* **7**, 551 (1937).
- ⁵²A. A. Gorman (private communication).

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