

Platinum metals

I 7700

47- 030

Structure and Properties of Ordered Li₂IrO₃ and Li₂PtO₃. — Li₂MO₃ (M: Ru, Ir, Pt) are prepared from mixtures of Li₂CO₃ and Ru, Ir, or Pt (1025 K, 12 h) and characterized by powder XRD, TGA, UV/VIS diffuse reflectance spectroscopy, electrical measurements, and DFT electronic structure calculations. The compounds crystallize in the monoclinic space group C2/m. They are thermally stable against reduction in air, O₂, and N₂ below 1375 K, which is an attractive feature for sensor operation over a wide range of temperatures. The thermal stability is significantly reduced in forming gas (N₂/H₂). The Ir compound is a metallic conductor, while the Pt compound is a semiconductor. — (O'MALLEY*, M. J.; VERWEIJ, H.; WOODWARD, P. M.; J. Solid State Chem. 181 (2008) 8, 1803-1809; Dep. Chem., Ohio State Univ., Columbus, OH 43210, USA; Eng.) — W. Pewestorf