## EFFECT OF Li2O ON THE WHITENESS

## OF ENAMELS FOR ALUMINUM

Yu.I. Bulavin

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The whiteness of opaque titanium-dioxide silicate enamels for aluminum depends not only on the content of TiO<sub>2</sub> but also on the content of Li<sub>2</sub>O which is an opacifying activator.

The activation effect of Li<sub>2</sub>O has quite a complex character.

Above all Li<sub>2</sub>O is an excellent flux and seriously reduces the viscosity of the glass in the enamel during firing of it on to aluminum, creating thereby optimum conditions for intensive transition of part of the TiO<sub>2</sub> into the opacifying phase.

Furthermore, the strong polarizing effect of the cations  $\mathrm{Li}^+$  with relatively low polarization of the  $\mathrm{Ti}^{+4}$  also is important.

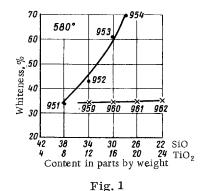
The effect of  $\text{Li}_2\text{O}$  on the opacification of leadless silicate enamels for aluminum with titanium dioxide is seen from the following experiment. If in enamel of composition (in weight part): 28.96 SiO<sub>2</sub>, 17.0 TiO<sub>2</sub>, 6.96 B<sub>2</sub>O<sub>3</sub>, 2.0 Al<sub>2</sub>O<sub>3</sub>, 6.0 SnO<sub>2</sub>, 4.5 Li<sub>2</sub>O, 13.08 K<sub>2</sub>O, 18.12 Na<sub>2</sub>O, developed at a firing temperature of 580°, whiteness about 78%, all the Li<sub>2</sub>O is replaced in equivalent quantities by Na<sub>2</sub>O or K<sub>2</sub>O, then in the same conditions of firing for aluminum enamel the coating becomes clear, although the quantity of opacifier (TiO<sub>2</sub> and SnO<sub>2</sub>) remain as previously. The firing of this nonlithium enamel at higher temperatures of 600°, 620°, and 630° does not give rise to opacification.

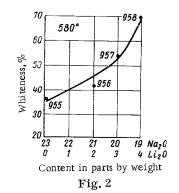
The above position was confirmed also in experiments with other enamels (Fig. 1 and 2). In Fig. 1 enamel No. 959 is nonlithium. Its composition in weight parts is: 36 SiO<sub>2</sub>, 10 TiO<sub>2</sub>, 6 SnO<sub>2</sub>, 8 B<sub>2</sub>O<sub>3</sub>, 2 P<sub>2</sub>O<sub>5</sub>, 23 Na<sub>2</sub>O, 12 K<sub>2</sub>O. In enamel No. 951 there are four parts by weight of Li<sub>2</sub>O. The composition is as follows: 38 SiO<sub>2</sub>, 8 TiO<sub>2</sub>, 6 SnO<sub>2</sub>, 8 B<sub>2</sub>O<sub>3</sub>, 2 P<sub>2</sub>O<sub>5</sub>, 19 Na<sub>2</sub>O, 12 K<sub>2</sub>O, 4 Li<sub>2</sub>O. The composition of enamel No. 955 (see Fig. 2) is 28 SiO<sub>2</sub>, 18 TiO<sub>2</sub>, 6 SnO<sub>2</sub>, 8 B<sub>2</sub>O<sub>3</sub>, 2 P<sub>2</sub>O<sub>5</sub>, 23 Na<sub>2</sub>O, 12 K<sub>2</sub>O.

An increase in the nonlithium enamel of the content of  $TiO_2$  at the expense of  $SiO_2$  to the maximum possible quantities has very little influence on the coefficient of diffuse reflection and permits us to obtain on the aluminum only gray enamel coatings.

The introduction into the enamel (instead of  $\text{Li}_2\text{O}$ ) of other oxides, such as MgO, CaO, BaO, SiO,  $\text{P}_2\text{O}_5$  did not give the expected positive results.

Without Li<sub>2</sub>O it is impossible to obtain opacified (TiO<sub>2</sub>) leadless silicate enamels for aluminum with high percentage of whiteness.





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Since a high percentage of whiteness of leadless boron-titanium silicate enamels proved to be possible only in the presence of both  $\text{TiO}_2$  and  $\text{Li}_2\text{O}$ , then naturally the question arises as to the optimum ratio of these oxides.

Investigation showed that the high percentage of whiteness (70-80%) with excellent development of other properties in leadless silicate enamels for aluminum can be obtained with a ratio of  $1 \text{ Li}_2\text{O}$ : (1.5-2)  $\text{TiO}_2$  and a total content of  $\text{TiO}_2$  of 18-21% by weight.