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Site-Specific ¹³C Chemical Shift Anisotropy Measurements in a Uniformly ¹⁵N,¹³C-Labeled Microcrystalline Protein by 3D Magic-Angle Spinning NMR Spectroscopy

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The dependence of NMR isotropic chemical shifts on protein conformation can be utilized to constrain protein secondary structure in solution^{1,2} and the solid state.³ Quantum chemical calculations^{4,5} and solution NMR measurements⁶ reveal smaller Cα chemical shift anisotropy (CSA) magnitude in α -helical versus β -sheet conformations, and solid-state studies of small peptides have quantified the dependence of tensor parameters on backbone dihedral angles $(\phi,\psi)^{4,7,8}$ The availability of theoretical shielding surfaces for all 20 common amino acids⁵ motivates the development and application of CSA measurement techniques, such as recoupling sequences by Tycko et al. that enabled scaled, undistorted powder patterns to be site-resolved in 2D experiments.9 Windowless sequences were developed to improve robustness with respect to frequency offsets and radio frequency (rf) inhomogeneity¹⁰ in order to study U-¹³Clabeled peptides. 11 Recently, Chang and Tycko utilized symmetry principles to devise the Recoupling of the Chemical Shift Anisotropy (ROCSA) sequence, 12 which suppresses homonuclear dipolar effects and reduces the rf field required, allowing application at > 10 kHz MAS rates, where resolution in uniformly ¹³C, ¹⁵N-labeled proteins is optimal.13

In this Communication, we report site-resolved CSA measurements in a microcrystalline protein of known structure (the 56 residue β 1 immunoglobulin binding domain of protein G, GB1),¹⁴ from a 3D experiment incorporating a ROCSA dimension (Figure 1). SPC-5 mixing,¹⁵ soft-pulse decoupling of scalar ¹³C-¹³C couplings, and TPPM decoupling of the protons¹⁶ yielded wellresolved and sensitive 3D spectra, from which 127 ROCSA line shapes were extracted, constituting 102 unique backbone and sidechain ¹³C sites. The GB1 sample (~10 mg) was prepared as a microcrystalline precipitate in a 3.2 mm rotor as described elsewhere. 17 The ROCSA time domain signal was fit to simulations performed with SIMPSON,18 to minimize the root-mean-squared deviation (RMSD) between experiment and simulation as a function of the tensor parameters and a phenomenological effective relaxation rate (see Supporting Information). Gly and Ala $C\alpha$ line shapes (Figure 2) provided a basis for initial evaluation since both small peptide SSNMR studies^{4,7,8} and solution NMR cross-correlation measurements⁶ are available. In agreement with quantum chemical calculations⁷ and previous solid-state measurements in ubiquitin, ¹⁹ our measurements showed smaller variations in the span ($\Omega = \delta_{11}$ $-\delta_{33}$) between helix and sheet than typical for solution crosscorrelation measurements.⁶ This difference is ascribed to the relative orientation of the CSA to the $C\alpha$ -H α dipole vector.^{4,19} As in the previous work of Hong,19 the ROCSA technique allows for the direct comparison of experimental results to calculated chemical shielding surfaces.^{4,5} The displayed "triple plot" (Figure 2b) illustrates the CSA tensor elements for the $C\alpha$ resonances of the Gly and Ala residues of GB1, compared to calculations based upon the GB1 crystal structure of the T2Q mutant of GB1 (A. M.

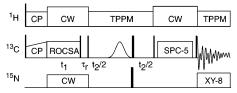


Figure 1. ROCSA-C-C-O-C-O-D recoupling pulse sequence. ROCSA was applied under CW decoupling on 15 N and 1 H. A soft (Gaussian) pulse was placed in the middle of the t_2 evolution period to remove the influence of the scalar (CO-C α and CO-C β) couplings. Narrow and wide rectangles represent $\pi/2$ and π pulses. Full experimental details, including phase cycles, are provided in the Supporting Information.

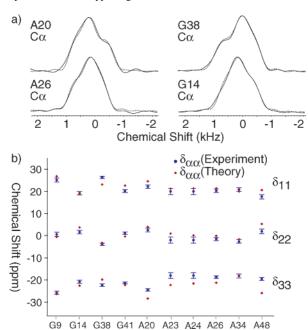


Figure 2. (a) ROCSA Cα powder patterns (solid line) with best-fit simulations (dotted). The label indicates the $^{13}C^{-13}C$ cross-peak from which CSA data were extracted. Data were sampled for 22 hypercomplex points with a dwell of 90 μ s. (b) Plot of the principal components of Cα CSA tensor in the traceless representation for all Gly and Ala residues in GB1 (blue) compared to theoretical values assuming conformation from the crystal structure (red). Error bars were computed using a Monte Carlo algorithm as discussed in Supporting Information. The theoretical values shown here are not empirically scaled.

Gronenborn, unpublished). While there is little change in the span of these tensors, there are important overall shifts of the principal elements. G14, for example, resides in a turn, whereas G38 is positioned at the transition between the α -helix and the β 3-strand; the downfield $C\alpha$ isotropic shift of G38 arises primarily from a change in the δ_{11} tensor element. These results agree with the study of Wittebort et al. for Gly residues with similar secondary structure.

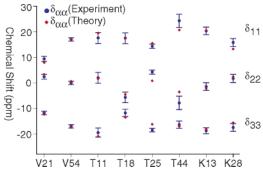


Figure 3. Plot of the principal components of the $C\alpha$ CSA tensors (traceless) for selected Val, Lys, and Thr residues in GB1 (blue) versus theoretical values (red) based upon the crystal structure. Error bars were calculated using a Monte Carlo algorithm as discussed in Supporting Information. The theoretical values shown here are not empirically scaled.

A20, in a reverse β -turn, displays an upfield C α isotropic shift due to δ_{33} , which shows a large variation relative to the helical Ala residues. Least-squares analysis of theory versus experiment shows a slope of -1.02 with a R^2 value of 0.99 for Gly, and a slope of -0.80 and a R^2 value of 0.98 for Ala. The RMSD between experiment and theory was 3.6 ppm for all Ca carbons studied here. When the theoretical values were scaled using the empirical factors reported by Havlin et al., the RMSD improved to 2.0 ppm.

 $C\alpha$ CSA tensors of $C\beta$ -branched chain amino acids are especially sensitive to both backbone and side-chain conformation (χ_1) .^{4,5} For example, V21 (helical geometry) has a span of 21.2 ± 1.8 ppm, whereas V54 (in strand β 4) has $\Omega = 35.1 \pm 1.5$ ppm. Thr C α tensor elements demonstrate interesting conformational dependence, as the shielding surfaces exhibit multiple minima.^{4,5} T11 C α (in a near helical conformation) has a larger span (37.1 \pm 3.9) than T18 (29.3 \pm 3.6), which is located in strand β 2; T25 (helical) has a span similar to that of T18 (32.0 \pm 1.5), but with a positive δ_{22} (like T11). Both tensors are \sim 10 ppm narrower than T44 (41.0 \pm 4.0), which is in the middle of a β -sheet. The relatively poor agreement with theory for T44 may be due to small changes in side-chain conformation, or that contributions from hydrogen bonding are not fully accounted for in the quantum chemical shielding surfaces. Differences as a function of conformation among long-chain residues, such as Lys, are less pronounced, but still much greater than the experimental error (e.g., between K13 (β -sheet) and K28 (α -helix)).

The precision of ROCSA for measuring smaller tensors also enables us to report (in the Supporting Information) values for 81 $C\beta$ and other side-chain line shapes, reporting on 59 unique sites. For example, the C β tensor span of A20 is nearly twice as broad as those of the helical Ala residues, a more pronounced dependence upon (ϕ, ψ) conformation than the isotropic $C\beta$ -shifts.²⁰ For Lys $C\beta$, large variations in the individual tensor elements arise despite similar secondary structures, demonstrating sensitivity to side-chain conformation and electrostatics. Thr C β line shapes vary substantially, and methyl 13C tensors for Val, Ile, Met, and Thr vary in span from ~18 to more than 40 ppm and sample the full range of asymmetry, in a manner that depends little (if at all) upon backbone geometry.

In conclusion, we have introduced a 3D experiment to correlate CSA powder line shapes with isotropic chemical shifts, enabling 127 site-specific measurements in a protein. More efficient acquisition of CSA data will help to develop a database of experimental tensors from proteins of known structure to complement ongoing theoretical efforts for purposes of CSA-based structure refinement. CSA values measured directly in proteins will also aid in the design and interpretation of solution NMR experiments that utilize CSAdipole cross-correlation, either to constrain torsion angles or to enhance resolution and sensitivity, such as in TROSY.21

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Supporting Information Available: Details of data acquisition, processing and fitting procedures; examples of fitted line shapes; a table of measured tensor elements (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) Spera, S.; Bax, A. J. Am. Chem. Soc. 1991, 113, 5490-5492.
- (2) Wishart, D. S.; Sykes, B. D.; Richards, F. M. J. Mol. Biol. 1991, 222,
- (3) Luca, S.; Filippov, D. V.; van Boom, J. H.; Oschkinat, H.; de Groot, H.
- J. M.; Baldus, M. *J. Biomol. NMR* **2001**, *20*, 325–331. (4) Havlin, R. H.; Laws, D. D.; Bitter, H. M. L.; Sanders, L. K.; Sun, H. H.; Grimley, J. S.; Wemmer, D. E.; Pines, A.; Oldfield, E. J. Am. Chem. Soc. **2001**, 123, 10362–10369.
- (5) Sun, H. H.; Sanders, L. K.; Oldfield, E. J. Am. Chem. Soc. 2002, 124, 5486-5495
- Tjandra, N.; Bax, A. *J. Am. Chem. Soc.* **1997**, *119*, 9576–9577. Heller, J.; Laws, D. D.; Tomaselli, M.; King, D. S.; Wemmer, D. E.; Pines, A.; Havlin, R. H.; Oldfield, E. *J. Am. Chem. Soc.* **1997**, *119*, 7827–
- (8) Chekmenev, E. Y.; Xu, R. Z.; Mashuta, M. S.; Wittebort, R. J. J. Am. Chem. Soc. 2002, 124, 11894-11899.
- Tycko, R.; Dabbagh, G.; Mirau, P. A. J. Magn. Reson. 1989, 85, 265-
- (10) Liu, S. F.; Mao, J. D.; Schmidt-Rohr, K. J. Magn. Reson. 2002, 155, 15-28.
- (11) Hong, M.; Yao, X. L. J. Magn. Reson. 2003, 160, 114-119.
- (12) Chan, J. C. C.; Tycko, R. J. Chem. Phys. 2003, 118, 8378-8389. McDermott, A.; Polenova, T.; Bockmann, A.; Zilm, K. W.; Paulsen, E.
- K.; Martin, R. W.; Montelione, G. T. J. Biomol. NMR 2000, 16, 209-
- (14) Gallagher, T.; Alexander, P.; Bryan, P.; Gilliland, G. L. Biochemistry 1994, *33*, 4721–4729.
- (15) Hohwy, M.; Rienstra, C. M.; Jaroniec, C. P.; Griffin, R. G. J. Chem. Phys. **1999**, 110, 7983-7992.
- (16) Bennett, A. E.; Rienstra, C. M.; Auger, M.; Lakshmi, K. V.; Griffin, R. G. *J. Chem. Phys.* **1995**, *103*, 6951–6958.
 (17) Franks, W. T.; Zhou, D. H.; Wylie, B. J.; Money, B. G.; Graesser, D. T.;
- Frericks, H. L.; Sahota, G.; Rienstra, C. M. J. Am. Chem. Soc. 2005, in
- (18) Bak, M.; Rasmussen, J. T.; Nielsen, N. C. J. Magn. Reson. 2000, 147, 296 - 330.
- (19) Hong, M. J. Am. Chem. Soc. 2000, 122, 3762-3770.
 (20) Saito, H.; Tabeta, R.; Sjohi, A.; Ozaki, T.; Ando, I. Macromolecules 1983, 16, 1050-1057.
- (21) Pervushin, K.; Riek, R.; Wider, G.; Wuthrich, K. Proc. Natl. Acad. Sci. U.S.A. 1997, 94, 12366-12371.

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