

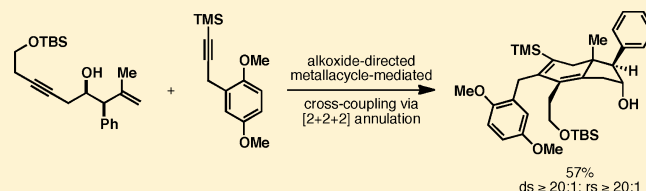
Asymmetric Synthesis of Dihydroindanes by Convergent Alkoxide-Directed Metallacycle-Mediated Bond Formation

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S Supporting Information

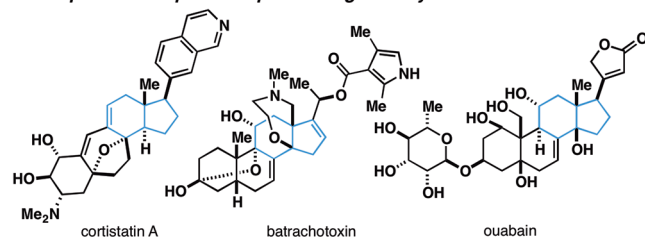
ABSTRACT: A convergent synthesis of highly substituted and stereodefined dihydroindanes is described from alkoxide-directed Ti-mediated cross-coupling of internal alkynes with substituted 4-hydroxy-1,6-enynes (substrates that derive from 2-directional functionalization of readily available epoxy alcohol derivatives). In addition to describing a new and highly stereoselective approach to bimolecular $[2 + 2 + 2]$ annulation that delivers products not available with other methods in this area of chemical reactivity, evidence is provided to support annulation by way of regioselective alkyne–alkyne coupling, followed by metal-centered $[4 + 2]$ rather than stepwise alkene insertion and reductive elimination. Overall, the reaction proceeds with exquisite stereochemical control and defines a convenient, convergent, and enantiospecific entry to fused carbocycles of great potential value in target-oriented synthesis and medicinal chemistry.



■ INTRODUCTION

Highly substituted and stereodefined hydroindanes define a structural motif that is encountered in natural and synthetic small molecules of broad pharmacological and biological relevance. Examples include complex natural products like cortistatin,¹ batrachotoxin,² ouabain,³ and cephalostatin⁴ (Figure 1A), as well as a

A. Complex natural products possessing densely functionalized indanes.



B. Common hydroindane-containing pharmaceutical agents.

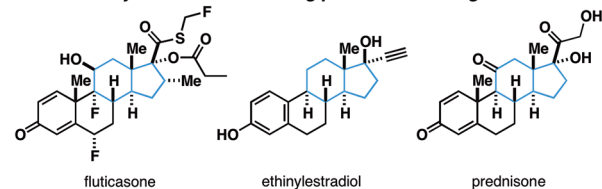


Figure 1. Dihydroindanes in natural products and pharmaceuticals.

large collection of less functionalized steroids that are prescribed daily for pharmaceutical intervention across a diverse landscape of therapeutic areas (Figure 1B).⁵ Robust chemical methods based on cycloaddition,⁶ cation–olefin cyclization,⁷ and Robinson annulation⁸ are typically embraced as strategies of choice for the assembly of such systems, yet these venerable methods can be challenged when confronted with the goal of accessing enantiodefined, highly substituted and/or oxygenated systems.⁹

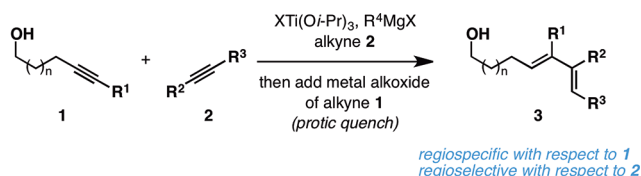
Pioneered by Vollhardt, metal-mediated $[2 + 2 + 2]$ annulation has surfaced as a useful strategy for the synthesis of dihydroindanes.¹⁰ First demonstrated in a wholly intramolecular fashion, recent studies have begun to describe the utility of such processes in intermolecular settings,¹¹ where the strategic benefit of convergency can dramatically enhance both the efficiency and flexibility of synthetic pathways to complex carbocyclic targets. This area of chemical reactivity, however, remains in its infancy, with intermolecular reactions being plagued by challenges associated with overcoming barriers associated with chemical reactivity (i.e., sluggish reactivity of substrates bearing substituted alkenes and/or sterically hindered internal alkynes), controlling regioselectivity (i.e., site of C–C bond formation on unsymmetrical systems) and stereoselection.¹² As such, the utility of $[2 + 2 + 2]$ annulation in target-oriented synthesis has been broadly constrained to the world of intramolecular cycloisomerization chemistry.¹⁰

In the context of a program aimed at the control of intermolecular metallacycle-mediated C–C bond formation, we have been actively engaged in the study of alkoxide-directed approaches to achieve fine control of metallacycle reactivity. These studies have led to the description of new intermolecular or convergent coupling reactions that have overcome the substantial limitations associated with prior art [including the difficulty in achieving sufficient reactivity and selectivity to accomplish bimolecular union of differentially substituted internal alkynes ($1 + 2 \rightarrow 3$; Figure 2A)¹³ and the cross-coupling of internal alkynes with substituted alkenes ($4 + 2 \rightarrow 5$; Figure 2B)¹⁴]. Here, we describe a general solution to the asymmetric synthesis of densely functionalized hydroindanes that builds on these earlier successes. In short, we report a highly stereoselective alkoxide-directed Ti-mediated intermolecular annulation between

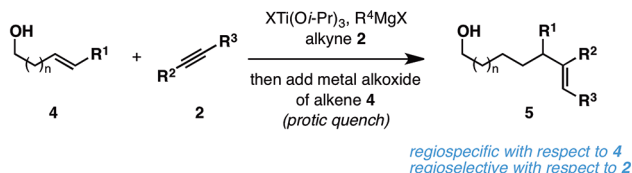
Received: November 8, 2011

Published: January 10, 2012

A. Alkoxide-directed metallacycle mediated coupling of internal alkynes:



B. Alkoxide-directed metallacycle mediated alkene-alkyne coupling:



C. Alkoxide-directed metallacycle mediated annulation for hydroindanes:

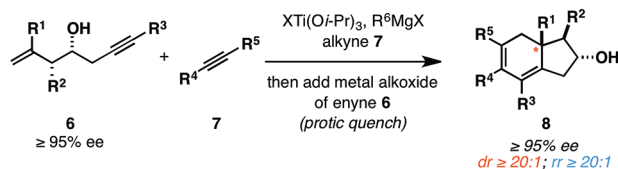


Figure 2. Metallacycle-mediated cross-coupling of alkenes and alkynes.

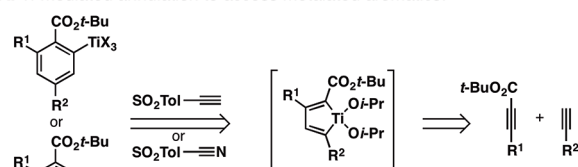
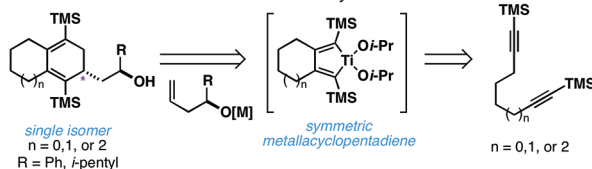
readily available 4-hydroxy-1,6-enynes and internal alkynes as a strategy to access densely functionalized hydroindanes (Figure 2C). This reaction, while related to previous metallacycle-mediated transformations (i.e., $[2 + 2 + 2]$ annulation chemistry), is to our knowledge the first that is capable of delivering stereo-defined angularly substituted hydroindanes in a convergent and highly selective manner using unsymmetric and nontethered reaction partners.

As depicted in Figure 3, Ti-alkoxide-based methods have been described for intermolecular $[2 + 2 + 2]$, yet these methods yield substituted aromatics¹⁵ (Figure 3A) or nearly symmetric cyclohexadienes¹⁶ (Figure 3B). In this latter case, intramolecular diyne cyclization delivers a symmetrical metallacyclopentadiene that is converted to a carbocyclic product by way of a process that avoids any challenges with regard to regioselection. Other metallacycle-mediated chemistry has been reported to access angularly substituted hydroindane products as depicted in Figure 3C,D.^{10c,17} These methods represent impressive demonstrations of intramolecular chemistry but, like many powerful modes of reactivity suitable for intramolecular C–C bond formation, these metallacycle-mediated bond constructions have not been demonstrated to be similarly useful in intermolecular settings. Given the significance of convergency in realizing step-economical pathways to complex molecules, this restriction to intramolecularity marks a significant limitation in the synthetic utility of these latter methods.

In short, the chemistry described herein marks a substantial advance in metallacycle-mediated annulation chemistry, offering a reaction process that enables bimolecular $[2 + 2 + 2]$ annulation as an entry to complex angularly substituted hydroindanes, structural motifs that are encountered in a diverse array of natural products and synthetic molecules of pharmacological relevance.

RESULTS

Initial investigation of intermolecular metallacycle-mediated coupling chemistry was focused on the union of internal alkynes with notoriously sluggish reaction partners, that is, other internal alkynes

I. Convergent $[2+2+2]$ annulation chemistry with Ti-alkoxides.A. Ti-mediated annulation to access metallated aromatics:¹⁵B. Ti-mediated annulation to access fused cyclohexadienes:¹⁶

II. Intramolecular metallacycle-mediated chemistry for the synthesis of angularly substituted hydroindanes.

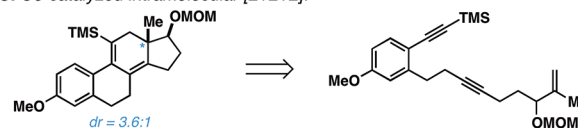
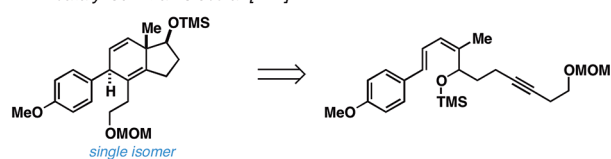
C. Co-catalyzed intramolecular $[2+2+2]$:^{10c}D. Ni-catalyzed intramolecular $[4+2]$:¹⁷

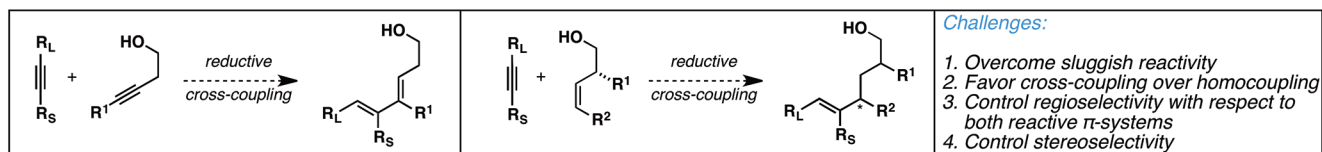
Figure 3. Established metallacycle-mediated annulation chemistry of relevance to the present advance.

and substituted alkenes. While a number of metallacycle-mediated reactions of this ilk are known, all available transformations are substantially limited in substrate scope. While reductive cross-coupling of alkynes is typically accomplished with one terminal alkyne,¹⁸ related union of internal alkynes with substituted alkenes has remained useful in only the simplest of cases, that is, with unhindered terminal mono-substituted alkenes.¹⁹ In fact, from a broader perspective, few reactions in organic chemistry are suitable for the regioselective carbometallation of an electronically unactivated alkene or alkyne, especially if such systems are 1,2-disubstituted.

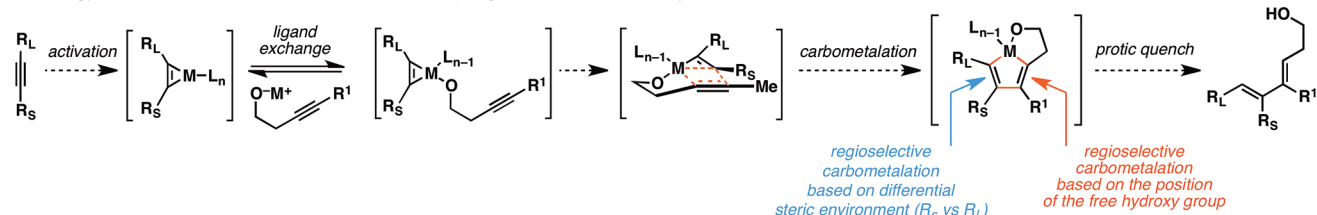
In an attempt to overcome the well established barrier to bimolecular reactivity of preformed metal–alkyne complexes with other electronically unactivated π -systems, we speculated that an alkoxide-directed approach as depicted in Figure 4A,B may offer a convenient solution. In this way, rapid and reversible ligand exchange would serve as a mechanism to render the carbometallation reaction intramolecular, and provide a means for stepwise encapsulation of the metal center by an orchestrated sequence of metal–ligand, metal–carbon, and carbon–carbon bond-forming processes. Importantly, in addition to providing a pathway to overcome the barriers to reactivity associated with scores of potential metallacycle-mediated cross-coupling processes, this strategy would define a mechanism to control regioselection that is distinct from established processes based on steric or electronic differentiation of the reacting π -systems.

To accomplish such a transformation, we accepted that catalytic versions of metallacycle-mediated coupling would not be reasonable as a general solution to the problem. This conclusion derived from the expected difficulty of favoring cross-coupling over homodimerization in a system where both

A. Reductive cross-coupling reactions of interest:



B. Strategy for alkoxide-directed reductive cross-coupling between internal alkynes:



C. Strategy for alkoxide-directed strategy for alkene–alkyne coupling:

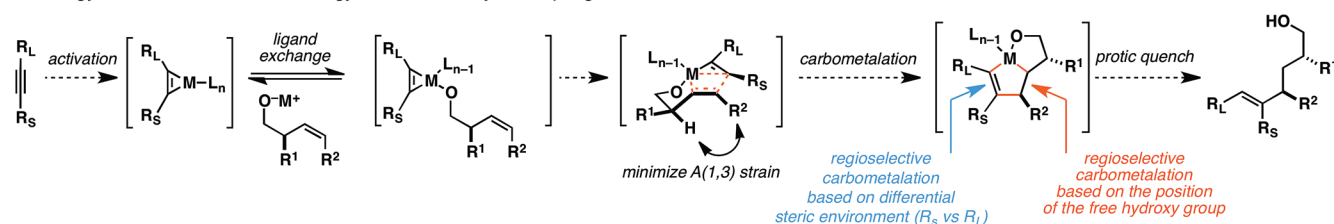


Figure 4. Alkoxide-directed alkyne–alkyne and alkene–alkyne reductive cross-coupling.

reactive π -systems are available (in large excess) to the catalytic metal center. Also, we aimed to achieve ligand-directed C–C bond formation with a ubiquitous free hydroxy group rather than with phosphines, thioethers, or terminal alkenes.²⁰ The combination of these factors led to the selection of established and inexpensive Ti(IV) alkoxides as the metal of choice to accomplish alkoxide-directed metallacycle-mediated cross-coupling as generalized in Figure 4B,C. In addition to Ti(Oi-Pr)₄ being easy to handle and purify in the absence of a glovebox, this reagent is inexpensive, nontoxic, and on aqueous workup delivers byproducts that are easily removed from product mixtures (TiO₂, *i*-PrOH, and MgX₂ salts—from the RMgX-mediated reduction of the Ti(IV) alkoxide). For these reasons, we moved forward with the development of stoichiometric Ti(IV)-mediated alkoxide-directed metallacycle-mediated cross-coupling chemistry.

As summarized in Figure 5, this strategy for the control of metallacycle-mediated C–C bond formation proved to be quite effective and general.^{13,14} In the case of alkyne–alkyne coupling (Figure 5A), successful reactions were achieved with homopropargylic and bis-homopropargylic alcohols, delivering 1,3-diene products as single regio- and stereoisomers. In the case of products **9** and **10**, the regioselectivity reported ($\geq 42:1$ and $\geq 65:1$) is based on the signal-to-noise ratio of each ¹H NMR spectrum of the crude product mixture, as no evidence could be found for presence of the other regioisomeric product. Symmetrical nonconjugated alkynes are also viable coupling partners in this diene synthesis (\rightarrow **11**; \rightarrow **12**). Further, while demonstrating a powerful means to overcome the barriers associated with reactivity for the cross-coupling process, this directing group strategy proved quite effective for overriding steric effects in the site-selective C–C bond forming process (i.e., for **13–15**, C–C bond formation occurs at the site distal to the free hydroxyl, independent of the nature of the proximal substituent: Et, *i*-Pr, *t*-Bu).

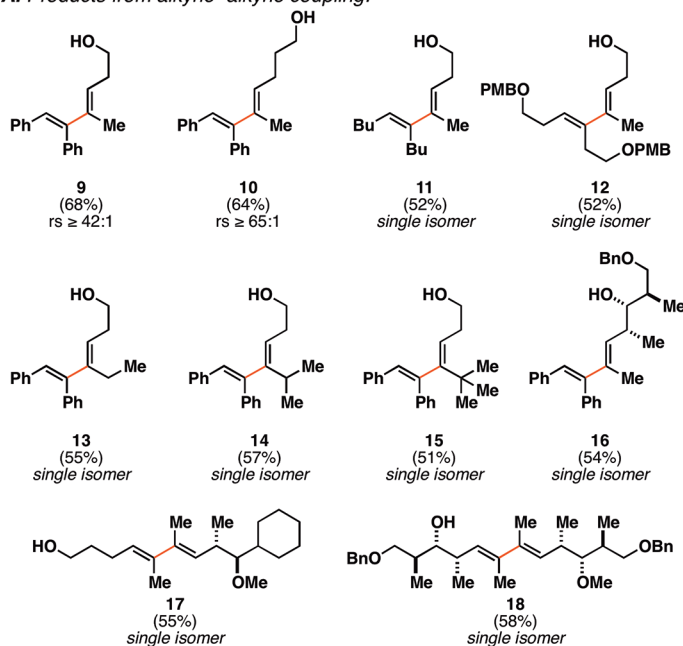
This general strategy also proved to be effective in coupling substrates that contain a hindered secondary alcohol directing

group (i.e., \rightarrow **16**). And, most interestingly, coupling of two unsymmetrical alkyne coupling partners proved to be highly selective, delivering tetrasubstituted 1,3-dienes **17** and **18** as single isomers. Here, exquisite regioselectivity is observed in the carbometallation of both π -systems. This observation was a welcomed surprise, as related coupling reactions of internal alkynes with carbonyl electrophiles do not typically proceed with such high levels of regioselectivity.²¹ While the mechanistic origins of this phenomenon remain to be determined, we suspect that the transition states associated with the present coupling reactions are significantly more product-like, as one would suspect that these processes are less exothermic than the analogous reactions of metal–alkyne complexes with aldehydes. As such, the closer distance between the reactive loci in the transition state should result in an enhancement of regioselectivity due to steric differences between the alkyne termini.

We note that these reductive cross-coupling reactions uniformly proceed with exceptional levels of selectivity, albeit in modest yield (51–68%). This compromise is associated with the frequent inability to push these reactions to completion and/or the observation of homodimer derived from the alkynyl alcohol coupling partner. This second observation likely arises from the use of a slight excess of Ti(Oi-Pr)₄ to efficiently convert the first alkyne coupling partner (2.5 equiv) to the metal–alkyne complex.

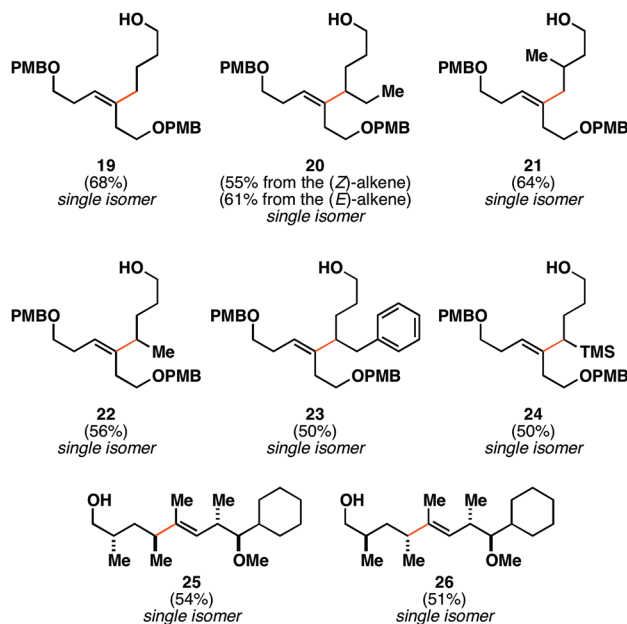
Moving on to the more demanding coupling reactions between alkynes and alkenes, we were delighted to find that the general approach taken with alkyne–alkyne coupling was also effective here. As illustrated in Figure 5B, a range of products can be generated from the union of internal alkynes with: (1) terminal- (\rightarrow **19**), (2) (*E*)-disubstituted- (\rightarrow **20**), (3) (*Z*)-disubstituted- (\rightarrow **20**), and (4) 1,1-disubstituted alkenes (\rightarrow **21**). In all cases, the site of C–C bond formation is completely controlled by the position of the free hydroxy group in the homoallylic alcohol coupling partners. Further, as seen in the alkyne–alkyne coupling process, the union of unsymmetrical alkynes with unsymmetrical homoallylic alcohols also proceeds

A. Products from alkyne–alkyne coupling:



Typical reaction conditions: 2.5 eq of alkyne, $\text{Ti}(\text{Oi-Pr})_4$, $c\text{-C}_5\text{H}_9\text{MgCl}$, PhMe , then introduce preformed Li-alkoxide of homopropargylic or bis-homopropargylic alcohol coupling partner.

B. Products from alkene–alkyne coupling:



Typical reaction conditions: 2 eq of alkyne, $\text{Ti}(\text{Oi-Pr})_4$, $c\text{-C}_5\text{H}_9\text{MgCl}$, PhMe , then introduce preformed Li-alkoxide of homoallylic alcohol coupling partner.

Figure 5. Examples of alkoxide-directed reductive cross-coupling.

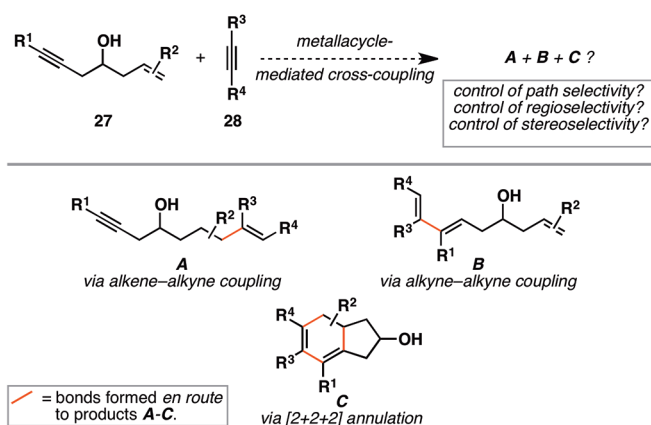
with exquisite levels of regioselection with respect to both reacting π -systems (\rightarrow 25 and 26). In these final cases, we observed a general lack of influence of double asymmetric relationships on the stereochemical course of bond formation.

With a firm foundation of preliminary data that confirms the ability to employ a hydroxy-group directing strategy to enable intermolecular metallacycle-mediated union of preformed metal–alkyne complexes with internal alkynes or substituted alkenes, we moved on to investigate the inherent reactivity of substrates that could react by either mode of reactivity in isolation, or by a process that engaged all three π -systems in an annulation process.

As illustrated in Figure 6A, preformation of a Ti-alkyne complex of 28 ($\text{Ti}(\text{Oi-Pr})_4$, $c\text{-C}_5\text{H}_9\text{MgCl}$, Et_2O , -78 to -30 $^\circ\text{C}$), followed

by addition of the lithium alkoxide of an enyne coupling partner (27) may result in either: (1) alkene–alkyne coupling (to deliver A),¹⁴ (2) alkyne–alkyne coupling (to furnish B),¹³ or (3) $[2 + 2 + 2]$ annulation (to produce the functionalized carbocycle C).^{15,16} As illustrated in Figure 6B, alkene substitution plays a dominant role in affecting product distribution.²² If the alkene is unsubstituted, reaction proceeds by alkene–alkyne coupling to deliver 30 (eq 1). Alternatively, substrates that contain terminal substitution on the alkene react by alkyne–alkyne coupling (eqs 2 and 3). Finally, a substrate bearing a simple 1,1-disubstituted alkene (35) undergoes reaction by yet a different course; here, convergent union engages all three π -systems and proceeds by way of a highly stereoselective $[2 + 2 + 2]$

A. Coupling of enynes with alkynes may proceed by different paths:



B. Alkene substitution dictates path selectivity in Ti-mediated coupling:

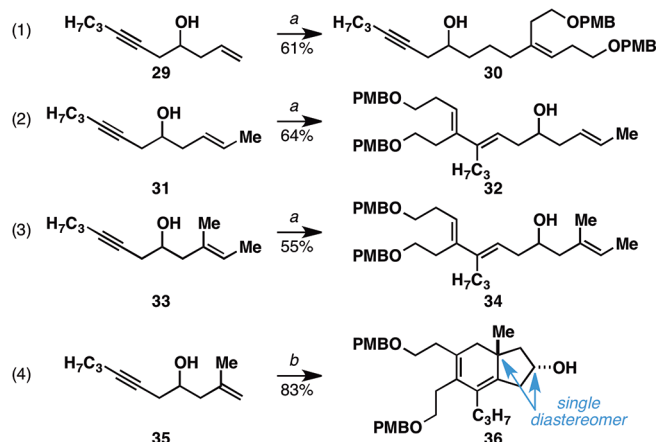


Figure 6. Path selectivity in alkoxide-directed metallacycle-mediated cross-coupling. Reaction conditions: (a) $(\text{PMBOCH}_2\text{CH}_2\text{C})_2$, $\text{Ti}(\text{Oi-Pr})_4$, $c\text{-C}_5\text{H}_9\text{MgCl}$, (-78 to -30 $^\circ\text{C}$), then add enyne as the corresponding Li-alkoxide (-30 $^\circ\text{C}$); (b) identical procedure as described for “a”, except that this reaction was warmed to 0 $^\circ\text{C}$. If this reaction (35 \rightarrow 36) was quenched at -30 $^\circ\text{C}$, little change in overall yield was observed (77%). For entries 2 and 3, if these reactions were warmed to 0 $^\circ\text{C}$ prior to quenching, no hydroindane could be identified from the product mixture.

annulation, delivering the functionalized dihydroindane product **36** in 83% yield as a single stereoisomer.

At this point, we speculated that the unique sequence of bond-forming events inherent to these alkoxide-directed coupling reactions that was previously described as a “stepwise encapsulation of the metal center by metal–ligand, metal–carbon, and carbon–carbon bond-forming reactions” may define a powerful new direction for the control of metal-centered [2 + 2 + 2] chemistry. On the basis of this perspective, we pursued the study of regio- and stereoselection in this unique annulation process in hopes of achieving a robust strategy for the synthesis of densely functionalized and stereodefined hydroindane systems that are not accessible with current [2 + 2 + 2] annulation methods.

As illustrated in Table 1, a range of 4-hydroxy-1,6-enynes and substituted alkynes are effective coupling partners in this

Table 1. Initial Studies Regarding Regioselection and Enantiospecificity

entry	enyne	alkyne	yield (%)	dr	rs	product ^{a,b}
						alkoxide-directed Ti-mediated convergent coupling
1			74 ^c	≥20:1	2:1	
	37 (95% ee)	38				39 (95% ee)
2			75 ^c	≥20:1	3:1	
	40	41				42
3			53 ^d	≥20:1	4:1	
	43	44				45
4			55 ^d	≥20:1	5:1	
	46	47				48

^aReaction conditions: alkyne, Ti(Oi-Pr)₄, *c*-C₅H₅MgCl, toluene (−78 to −30 °C). ^bReaction conditions: cool to −78 °C and add lithium alkoxide of enyne, warm to a final temperature of between −15 and 23 °C. ^cYield reported is for the mixture of regioisomers. ^dYield reported is for the major regioisomer. See Supporting Information for additional information.

stereoselective annulation process. Further, this reaction proceeds in an enantiospecific fashion.^{23a} As depicted in entry 1, union of the optically active enyne **37** (95% ee) with Me₂PhSi-alkyne **38** occurs in 74% yield to deliver a mixture of regioisomeric hydroindane products, the major isomer of which is the stereodefined hydroindane **39** (95% ee; dr ≥ 20:1). While regiochemical control in the functionalization of alkyne **38** was not particularly high in this

case (rs = 2:1), the stereochemically defined product **39** possesses substitution of significance in the context of de novo steroid synthesis [see C8, C9 and C11 (steroid numbering)]. In efforts to explore the scope of this annulation process, we observed enhancements in regioselection as a function of the coupling partners employed. First, regioselection was seen to increase slightly when using the TMS-substituted alkyne **41** that contains an aliphatic substituent (CH₂CH₂OPMB) rather than Ph (as in **38**). Coupling of **40** with alkyne **41** proceeds in 75% yield to deliver a 3:1 mixture of regioisomeric carbocycles, the major isomer of which is the hydroindane product **42** (dr ≥ 20:1; entry 2). Regioselection also improved in the coupling reaction of TMS-substituted enyne **44** with 4-hydroxy-1,6-enyne **43** (entry 3). Here, the annulation event proceeds with 4:1 regioselection, ≥20:1 diastereoselection, and the major isomer **45** can be isolated in 53% yield as a single isomer. Finally, when the alkene of the 4-hydroxy-1,6-enyne possesses a more sterically demanding substituent (as compared to Me), regioselection also increases (entry 4). Here, union of the benzyl-substituted substrate **46** with TMS-alkyne **47** proceeds with 5:1 regioselection, ≥20:1 diastereoselection, and the major product **48** can be isolated in 55% yield as a single isomer.^{23b} While in entries 3 and 4 of Table 1 the major isomer could be separated from the minor regioisomer, it is important to note that in all cases, purification of the major isomer is facile after chemoselective protodesilylation of the major products' vinylsilane (see Supporting Information for details).

Notably, this annulation process can be employed to generate hydroindane products possessing C17 β-substitution (Table 2). As depicted in entry 1, coupling of enyne **49** with alkyne **50** results in the formation of hydroindane products possessing a C17 β-Me substituent in 63% yield and ≥20:1 diastereoselection, albeit as a 3:1 mixture of regioisomers. Interestingly, when exploring the utility of this annulation process for the synthesis of hydroindanes possessing C17 β-aryl substitution (a prominent architectural feature of the cortistatins), we discovered that this process can proceed with exquisite regiochemical control. As illustrated in entry 2, coupling of hydroxy-enyne **52** with alkyne **41** delivers hydroindane product **53** in 64% yield, with ≥20:1 ds and 13:1 rs. While the mechanistic origin of this long-range effect on regioselection remains unclear, all enyne substrates bearing an allylic Ph-substituent underwent Ti-mediated annulation reactions with high levels of regioselectivity (typically ≥20:1), and consistently exquisite levels of diastereoselection (uniformly ≥20:1) (entries 2–7).

While this annulation process is quite effective at generating 17-β-aryl substituted hydroindanes as essentially single regio and stereoisomers (i.e., **53**, **56**, **58**, **59**, **61** and **63**), the process is sensitive to the relative stereochemistry of the enyne starting material. As illustrated in entry 8 of Table 2, annulation of enyne **64** with TMS-phenylacetylene (**50**) produced a complex mixture of products from which only 22% yield of the C17 α-substituted hydroindane product (**65**) could be obtained. While this stereochemical pattern certainly represents a preparative limitation to the current chemistry, product **65** could easily be isolated as a single regio- and stereoisomer.

Mechanistic Hypothesis. Metal-centered [2 + 2 + 2] annulation chemistry has long been proposed to occur by initial formation of a metallacyclopentadiene, followed by either:

- (1) [4 + 2]/cycloreversion (Figure 7A), or
- (2) alkene insertion/reductive elimination (Figure 7B).

In our efforts to explore the scope of this Ti-mediated annulation, we have come upon an experiment that distinguishes between these two pathways, and further provides evidence in

Table 2. C17 β -Aryl Substitution and Regioselection

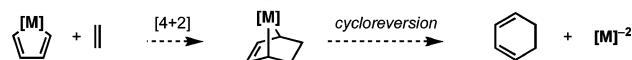
entry	enyne	alkyne	yield (%)	dr	product ^{a,b}
					alkoxide-directed Ti-mediated convergent coupling
1			63 ^c	≥20:1 3:1	
2			64	≥20:1 13:1	
3			52	≥20:1 ≥20:1	
4			57	≥20:1	
5			57	17:1 ≥20:1	
6			54	≥20:1 ≥20:1	
7			54	≥20:1 ≥20:1	
8			22	- ^d - ^d	

^aReaction conditions: alkyne, Ti(Oi-Pr)₄, *c*-C₃H₉MgCl, toluene (−78 to −30 °C). ^bReaction conditions: cool to −78 °C and add lithium alkoxide of enyne, warm to a final temperature of between −15 and 23 °C. ^cYield reported is for the mixture of regioisomers. ^dPrecise selectivities could not be determined from the ¹H NMR spectrum of the crude product; hydroindane product **65** was isolated as a single regio- and stereoisomer. See Supporting Information for additional information

support of a sequential [4 + 2]/cycloreversion in preference to alkene insertion/reductive elimination.

As illustrated in Figure 8A, coupling of enyne **66** with alkyne **55** delivers a carbocyclic product (**67**) that lacks the PMB ether of the starting material and contains an exocyclic alkene. This observation is consistent with the empirical model depicted in Figure 8B, where initial alkyne–alkyne coupling delivers metal-lacyclopentadiene **68** in a regioselective fashion.¹³ sequent stereo-

A. Annulation by sequential [4+2] and cycloreversion:



B. Annulation by sequential alkene insertion and reductive elimination:

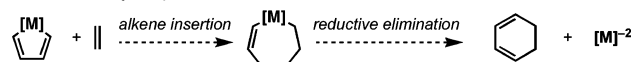
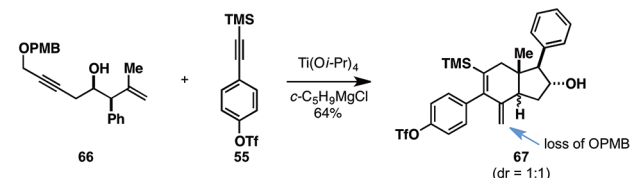
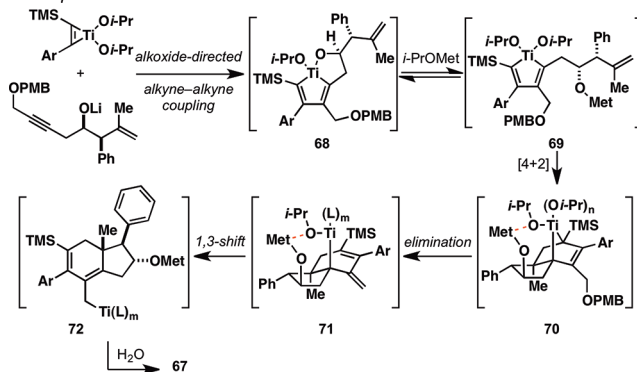


Figure 7. Potential pathways for metal-centered annulation.

A. Ti-mediated coupling/annulation followed by reductive cleavage:



B. Empirical model:



C. Insertion/elimination pathway:

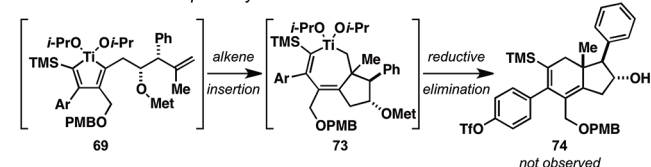


Figure 8. Development of an empirical model for the Ti-mediated intermolecular hydroindane synthesis.

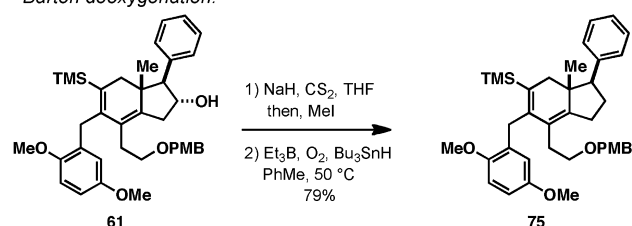
selective annulation is then proposed to occur through a sequence of steps that includes alkoxide-promoted cleavage of the oxametallacyclopentane (**68** → **69**), followed by stereoselective intramolecular [4 + 2] (**69** → **70**), a highly stereoselective process that is thought to be controlled by intramolecular coordination of the pendant alkoxide to a ligand on Ti. Unlike previous examples depicted in Table 1, intermediate **70** is suitably functionalized to undergo vinylogous elimination of the PMB ether to generate an unstable tertiary allyltitanium species **71**. Finally, 1,3-metallotropic shift generates allyltitanium species **72** which, on protonolysis with allylic transposition, is poised to deliver the observed product **67**. Alternatively, if annulation proceeded by a sequence defined by alkene insertion and reductive elimination (Figures 7B and 8C), a suitable intermediate is not readily apparent to support the facile elimination of the PMB ether [i.e., **73** (Figure 8B) vs **70** (Figure 8C)].²⁴

Overall, alkoxide-directed intermolecular [2 + 2 + 2] annulation between an internal alkyne and a 1,6-enyne defines a unique and powerful approach to controlling regio- and stereoselection in this class of chemical reactivity. The virtues of stepwise encapsulation of the metal center (through a process of sequential ligand exchange and intramolecular carbometala-

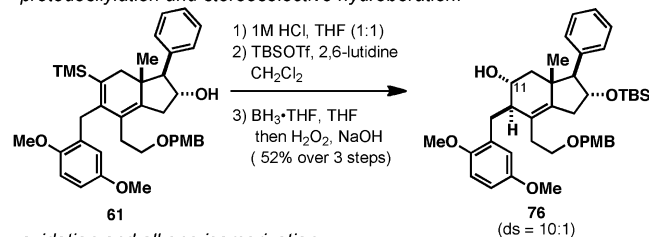
tion) set the stage for a highly controlled series of bond forming events en route to a functionalized carbocyclic system of great potential utility in natural product synthesis. Further, the mechanistic course of this process (which is more consistent with $[4 + 2]$ /cycloreversion than alkene insertion/reductive elimination) may afford the opportunity to divert chemical reactivity away from the expected $[2 + 2 + 2]$ products (i.e., Table 1) to stereodefined products that take advantage of the unique reactivity of bridged polycyclic allylic metal intermediates (as demonstrated in Figure 8A).

Utility of the Carbocyclic Products. The present annulation process delivers complex carbocycles that can be differentially functionalized to address a range of common substitution patterns about the hydroindane core of natural products and small molecules of pharmacological relevance. As depicted in Figure 9, deoxygenation of the D-ring hydroxy group proceeds

• **Barton deoxygenation:**



• **protodesilylation and stereoselective hydroboration:**



• **oxidation and alkene isomerization:**

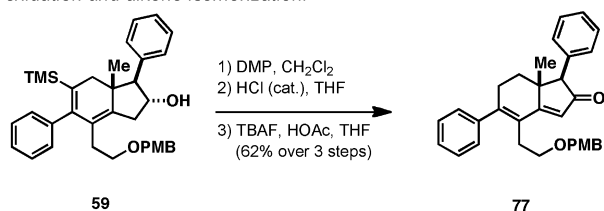


Figure 9. Some functionalization processes suitable for elaboration of the dihydroindane products.

readily by the Barton procedure (**61** → **75**).²⁵ Alternatively, C11 functionality (steroid numbering) can easily be introduced by protodesilylation²⁶ and hydroboration (**61** → **76**). Finally, a sequence of oxidation, isomerization, and protodesilylation is suitable to access a CD-ring system bearing a dienone (**59** → **77**). While certainly not comprehensive, the transformations depicted in Figure 9 serve to introduce the great potential utility of the functionalized dihydroindane products derived from the Ti-mediated annulation reaction in the stereoselective synthesis of complex carbocycles.

CONCLUSION

We have described a convenient and highly stereoselective pathway to a range of functionalized hydroindanes by the union of simple 4-hydroxy-1,6-enynes with silyl-substituted alkynes. Since the enyne-containing starting materials are easily derived

from two-directional functionalization of chiral epoxy alcohol derivatives (including epichlorohydrin), the synthetic pathway secured is also amenable to the production of optically active hydroindanes that bear a variety of functionality of likely utility in target-oriented synthesis. While mechanistic ambiguity has traditionally surrounded related $[2 + 2 + 2]$ annulation chemistry, our studies have led to the conclusion that sequential $[4 + 2]$ cycloaddition/cycloreversion is likely operative in preference to alkene insertion/reductive elimination. From a preparative perspective, the convergent annulation described is efficient (isolated yields ranging from 48 to 83%), regioselective (up to $\geq 20:1$), and highly stereoselective (dr is uniformly $\geq 20:1$). We are unaware of another *intermolecular* metallacycle-mediated annulation reaction capable of delivering stereo-defined products of related structure to that observed here—structural motifs that are core skeletal components of a vast array of pharmacologically active natural and synthetic substances. On the basis of the considerations discussed, the current science defines a significant advance in metallacycle-mediated annulation chemistry, and marks the establishment of an exceptionally concise means for the convergent assembly of densely functionalized hydroindanes. As such, we anticipate that the current findings will be of great potential utility in chemical synthesis and medicinal chemistry.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures and tabulated spectroscopic data for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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ACKNOWLEDGMENTS

We gratefully acknowledge financial support of this work by the National Institutes of Health—NIGMS (GM80266 and GM80266-04S1). The authors also acknowledge Hayley Browdy for assistance in synthesizing the panel of enynes depicted in Figure 6B.

REFERENCES

- (a) Narayan, A. R. H.; Simmons, E. M.; Sarpong, R. *Eur. J. Org. Chem.* **2010**, 3553–3567. (b) Flyer, A. N.; Si, C.; Myers, A. G. *Nat. Chem.* **2010**, 2, 886–892. (c) Aoki, S.; Watanabe, Y.; Sanagawa, M.; Setiawan, A.; Kotoku, N.; Kobayashi, M. *J. Am. Chem. Soc.* **2006**, 128, 3148–3149. (d) Shenvi, R. A.; Guerrero, C. A.; Shi, J.; Li, C.-C.; Baran, P. S. *J. Am. Chem. Soc.* **2008**, 130, 7241–7243. (e) Lee, H. M.; Nieto-Oberhuber, C.; Shair, M. D. *J. Am. Chem. Soc.* **2008**, 130, 16864–16866. (f) Nicolaou, K. C.; Peng, X.-S.; Sun, Y.-P.; Polet, D.; Zou, B.; Lim, C. S.; Chen, D. Y.-K. *J. Am. Chem. Soc.* **2009**, 131, 10587–10597. For an approach to the core skeleton of the cortistatins by coupling of a functionalized hydroindane with an aromatic A-ring precursor, see: (g) Dai, M.; Danishefsky, S. J. *Tetrahedron Lett.* **2008**, 49, 6610–6612.
- (a) Daly, J. W.; Witkop, B.; Bommer, P.; Biemann, K. *J. Am. Chem. Soc.* **1965**, 87, 124–126. (b) Dumbacher, J. P.; Beehler, B. M.; Spande, T. F.; Garraffo, H. M.; Daly, J. W. *Science* **1992**, 258, 799–801. (c) Tokuyama, T.; Daly, J. *J. Am. Chem. Soc.* **1969**, 91, 3931–3938. (d) Albuquerque, E. X.; Daly, J. W.; Witkop, B. *Science* **1971**, 172, 995–1002. (e) Kurosu, M.; Marcin, L. R.; Grinstainer, T. J.; Kishi, Y. *J. Am. Chem. Soc.* **1998**, 120, 6627–6628.

- (3) (a) Prassas, I.; Diamandis, E. P. *Nat. Rev. Drug Discovery* **2008**, *7*, 926–935. (b) Newman, R. A.; Yang, P.; Pawlus, A. D.; Block, K. I. *Mol. Interventions* **2008**, *8*, 36–49. (c) Riganti, C.; Campia, I.; Kopecka, J.; Gazzano, E.; Doublier, S.; Aldieri, E.; Bosia, A.; Ghigo, D. *Curr. Med. Chem.* **2011**, *18*, 872–885. (d) Overman, L. E.; Rucker, P. V. *Heterocycles* **2000**, *52*, 1297–1314. (e) Reddy, M. S.; Zhang, H.; Phoenix, S.; Deslongchamps, P. *Chem. Asian J.* **2009**, *4*, 725–741.
- (4) (a) Lee, S.; LaCour, T. G.; Fuchs, P. L. *Chem. Rev.* **2009**, *109*, 2275–2314. (b) LaCour, T. G.; Guo, C.; Bhandaru, S.; Boyd, M. R.; Fuchs, P. L. *J. Am. Chem. Soc.* **1998**, *120*, 692–707. (c) Moser, B. R. *J. Nat. Prod.* **2008**, *71*, 487–491. (d) Rudy, A.; Lopez-Anton, N.; Dirsch, V. M.; Vollmar, A. M. *J. Nat. Prod.* **2008**, *71*, 482–486. (e) Fortner, K. C.; Kato, D.; Tanaka, Y.; Shair, M. D. *J. Am. Chem. Soc.* **2010**, *132*, 275–280.
- (5) (a) Zeleen, F. J. Medicinal Chemistry of Steroids. In *Principles of Medical Biology, Volume 8B, Molecular and Cellular Pharmacology*; JAI Press, Inc.: Stamford, 1997; pp427–463. For reviews on the synthesis of steroids, see: (b) Chapelon, A.-S.; Moraleda, D.; Rodriguez, R.; Ollivier, C.; Santelli, M. *Tetrahedron* **2007**, *63*, 11511–11616. (c) Jankowski, P.; Marczak, S.; Wicha, J. *Tetrahedron* **1998**, *54*, 12071–12150.
- (6) (a) Parker, K. A.; Iqbal, T. *J. Org. Chem.* **1982**, *47*, 337–342. (b) Wilson, S. R.; Jacob, L. *J. Org. Chem.* **1992**, *57*, 4380–4385. (c) Wender, P. A.; Smith, T. E. *Tetrahedron* **1998**, *54*, 1255–1275. (d) Taber, T. F.; Song, Y. *J. Org. Chem.* **1996**, *61*, 7508–7512.
- (7) (a) Snider, B. B.; Kirk, T. C. *J. Am. Chem. Soc.* **1983**, *105*, 2364–2368. (b) Johnson, W. S.; Elliot, J. D.; Hanson, G. J. *J. Am. Chem. Soc.* **1984**, *106*, 1138–1139. (c) Hatakeyama, S.; Numata, H.; Osanai, K.; Takano, S. *J. Chem. Soc., Chem. Commun.* **1989**, 1893–1895. (d) Browder, C. C.; West, F. G. *Synlett* **1999**, *9*, 1363–1366.
- (8) (a) Hajos, Z. G.; Parrish, D. R. *J. Org. Chem.* **1974**, *39*, 1612–1615. (b) Stork, G.; Shiner, C. S.; Winkler, J. D. *J. Am. Chem. Soc.* **1982**, *104*, 310–312. (c) Stork, G.; Winkler, J. D.; Shiner, C. S. *J. Am. Chem. Soc.* **1982**, *104*, 3767–3768. (d) Yamamoto, K.; Iijima, M.; Ogimura, Y.; Tsuji, J. *Tetrahedron Lett.* **1984**, *25*, 2813–2816. (e) Kotoku, N.; Sumii, Y.; Hayashi, T.; Kobayashi, M. *Tetrahedron Lett.* **2008**, *49*, 7078–7081.
- (9) For example, see: refs 1d, 5b, 5c, 6b, 6d, 7b, 7c, 8d, 8e, and (a) Corey, E. J.; Huang, A. X. *J. Am. Chem. Soc.* **1999**, *121*, 710–714. (b) Rychnovsky, S. D.; Mickus, D. E. *J. Org. Chem.* **1992**, *57*, 2732–2736. (c) Clasby, M. C.; Craig, D.; Jaxa-Chamiec, A. A.; Lai, J. Y. Q.; Marsh, A.; Slawin, A. M. Z.; White, A. J. P.; Williams, D. J. *Tetrahedron* **1996**, *52*, 4769–4802. (d) Nemoto, H.; Matsuhashi, N.; Imaizumi, M.; Nagai, M.; Fukumoto, K. *J. Org. Chem.* **1990**, *55*, 5625–5631.
- (10) (a) Sternberg, E. D.; Vollhardt, K. P. C. *J. Am. Chem. Soc.* **1980**, *102*, 4839–4841. (b) Sternberg, E. D.; Vollhardt, K. P. C. *J. Org. Chem.* **1982**, *47*, 3447–3450. (c) Clinet, J.-C.; Duñach, E.; Vollhardt, K. P. C. *J. Am. Chem. Soc.* **1983**, *105*, 6710–6712. (d) Butenschön, H.; Winkler, M.; Vollhardt, K. P. C. *J. Chem. Soc., Chem. Commun.* **1986**, 388–390. (e) Grotjahn, D. B.; Vollhardt, K. P. C. *J. Am. Chem. Soc.* **1986**, *108*, 2091–2093. (f) Germanas, J.; Aubert, C.; Vollhardt, K. P. C. *J. Am. Chem. Soc.* **1991**, *113*, 4006–4008. (g) Johnson, E. P.; Vollhardt, K. P. C. *J. Am. Chem. Soc.* **1991**, *113*, 381–382. For reviews, see: (h) Vollhardt, K. P. C. *Acc. Chem. Res.* **1977**, *10*, 1–8. (i) Aubert, C.; Buisine, O.; Petit, M.; Slowinski, F.; Malacria, M. *Pure Appl. Chem.* **1999**, *71*, 1463–1470. (j) Rubin, M.; Sromek, A. W.; Gevorgyan, V. *Synlett* **2003**, 2265–2291. (k) Gandon, V.; Aubert, C.; Malacria, M. *Chem. Commun.* **2006**, 2209–2217. For examples of catalytic asymmetric intramolecular [2 + 2 + 2] annulation, see: (l) Shibata, T.; Kurokawa, H.; Kanda, K. *J. Org. Chem.* **2007**, *72*, 6521–6525. (m) Shibata, T.; Tahara, Y. *J. Am. Chem. Soc.* **2006**, *128*, 11766–11767. For recent reviews of [2 + 2 + 2] annulations, see: (n) Shibata, T.; Tsuchikama, K. *Org. Biomol. Chem.* **2008**, *6*, 1317–1323. (o) Tanaka, K. *Chem. Asian J.* **2009**, *4*, 508–518. (p) Leboeuf, D.; Gandon, V.; Malacria, M. Transition Metal-Mediated [2 + 2 + 2] Cycloadditions. In *Handbook of Cyclization Reactions*; Ma, S., Ed.; Wiley-VCH: Weinheim, 2009; Vol. 1, p 367.
- (11) (a) Evans, P. A.; Lai, K. W.; Sawyer, J. R. *J. Am. Chem. Soc.* **2005**, *127*, 12466–12467. (b) Shibata, T.; Arai, Y.; Tahara, Y. *Org. Lett.* **2005**, *7*, 4955–4957.
- (12) For recent reviews that describe these challenges, see: (a) Reichard, H. A.; McLaughlin, M.; Chen, M. Z.; Micalizio, G. C. *Eur. J. Org. Chem.* **2010**, 391–409. (b) Reichard, H. A.; Micalizio, G. C. *Chem. Sci.* **2011**, *2*, 573–589.
- (13) Ryan, J.; Micalizio, G. C. *J. Am. Chem. Soc.* **2006**, *128*, 2764–2765.
- (14) (a) Reichard, H. A.; Micalizio, G. C. *Angew. Chem., Int. Ed.* **2007**, *46*, 1440–1443. (b) Canterbury, D. P.; Micalizio, G. C. *J. Am. Chem. Soc.* **2010**, *132*, 7602–7604.
- (15) (a) Suzuki, D.; Urabe, H.; Sato, F. *J. Am. Chem. Soc.* **2001**, *123*, 7925–7926. (b) Suzuki, D.; Tanaka, R.; Urabe, H.; Sato, F. *J. Am. Chem. Soc.* **2002**, *124*, 3518–3519. (c) Tanaka, R.; Yuza, A.; Watai, Y.; Suzuki, D.; Takayama, Y.; Sato, F.; Urabe, H. *J. Am. Chem. Soc.* **2005**, *127*, 7774–7780. For earlier examples demonstrating the ability of Ti-aryloxides to promote [2 + 2 + 2] annulation en route to substituted aromatics, see: (d) Hill, J. E.; Balaich, G.; Fanwick, P. E.; Rothwell, I. P. *Organometallics* **1993**, *12*, 2911–2924.
- (16) (a) Sung, M. J.; Pang, J.-H.; Park, S.-B.; Cha, J. K. *Org. Lett.* **2003**, *5*, 2137–2140. For early examples demonstrating the ability of Ti-aryloxides to generate cyclohexadienes from the coupling of two alkynes and one alkene, see: (b) Balaich, G. J.; Rothwell, I. P. *J. Am. Chem. Soc.* **1993**, *115*, 1581–1583. (c) Johnson, E. S.; Balaich, G. J.; Rothwell, I. P. *J. Am. Chem. Soc.* **1997**, *119*, 7685–7693.
- (17) Wender, P. A.; Smith, T. E. *Tetrahedron* **1998**, *54*, 1255–1275.
- (18) (a) Hamada, T.; Suzuki, D.; Urabe, H.; Sato, F. *J. Am. Chem. Soc.* **1999**, *121*, 7342–7344. (b) See ref 15c. (c) Buchwald, S. L.; Watson, B. T.; Huffman, J. C. *J. Am. Chem. Soc.* **1987**, *109*, 2544–2546. (d) Buchwald, S. L.; Nielsen, R. B. *J. Am. Chem. Soc.* **1989**, *111*, 2870–2874. (e) Shimp, H. L.; Micalizio, G. C. *Org. Lett.* **2005**, *7*, 5111–5114. (f) Perez, L. J.; Shimp, H. L.; Micalizio, G. C. *J. Org. Chem.* **2009**, *74*, 7211–7219. For Zr-mediated cross-coupling of two different internal, yet symmetrical, alkynes, see: (g) Xi, Z.; Hara, R.; Takahashi, T. *J. Org. Chem.* **1995**, *60*, 4444–4448.
- (19) For addition reactions of preformed benzyne–ZrCp₂ complexes to simple alkenes, see: (a) Aoki, K.; Peat, A. J.; Buchwald, S. L. *J. Am. Chem. Soc.* **1998**, *120*, 3068–3073. For addition reactions of preformed Ti-vinylsilane complexes to alkynes, see: (b) Mizojiri, R.; Urabe, H.; Sato, F. *J. Org. Chem.* **2000**, *65*, 6217–6222. For a review of Co-catalyzed coupling chemistry of alkynes with alkenes, see: Jeganmohan, M.; Cheng, C.-H. *Chem.—Eur. J.* **2008**, *14*, 10876–10886.
- (20) For a review of directed reactions in organic chemistry, see: (a) Hoveyda, A. H.; Evans, D. A.; Fu, G. C. *Chem. Rev.* **1993**, *93*, 1307–1370. For a recent example of an alkene-directed metallacycle-mediated coupling, see: (b) Moslin, R. M.; Miller, K. M.; Jamison, T. F. *Tetrahedron* **2006**, *62*, 7598–7610. For examples of heteroatom-directed intermolecular Pauson–Khand reactions, see: (c) Gibson, S. E.; Mainolfi, N. *Angew. Chem., Int. Ed.* **2005**, *44*, 3022–3037.
- (21) (a) Harada, K.; Urabe, H.; Sato, F. *Tetrahedron Lett.* **1995**, *36*, 3203–3206. (b) Bahadoor, A. B.; Flyer, A.; Micalizio, G. C. *J. Am. Chem. Soc.* **2005**, *127*, 3694–3695. (c) Bahadoor, A. B.; Micalizio, G. C. *Org. Lett.* **2006**, *8*, 1181–1184.
- (22) For an example of how alkene substitution plays a significant role in dictating site selectivity in the reductive cross-coupling of 1,5-dienes with alkynes, see: Diez, P. S.; Micalizio, G. C. *J. Am. Chem. Soc.* **2010**, *132*, 9576–9578.
- (23) (a) See the Supporting Information for the opposite antipode—a demonstration that supports the enantiospecific nature of the process. (b) Overall, isolated yields of the annulation products typically range from 52–83%. These less than quantitative yields reflect: (i) incomplete consumption of the enyne, (ii) alkyne–alkyne coupling without annulation (the product of this is a functionalized triene, i.e., Figure 6B, eqs 2 and 3), and occasionally (iii) partial deprotection of the PMB/silyl ether of the annulation product.
- (24) Metallacycle-mediated [2 + 2 + 2] annulation is generally thought to proceed through initial formation of a metallacyclopentadiene

followed by either alkene insertion/reductive elimination or $[4 + 2]$ /cycloreversion. We have opted to invoke a model based on $[4 + 2]$ /cycloreversion, where stereochemistry is controlled by a coordination event between the homoallylic alkoxide and an alkoxide bound to Ti. This proposition does not preclude the involvement of an alkene insertion mechanism, and only intends to offer an empirical model to predict the stereochemical course of annulation and that is consistent with the observations summarized in Figure 8A. Future studies are aimed at deconvoluting these mechanistically distinct pathways.

(25) Crich, D.; Quintero, L. *Chem. Rev.* **1989**, 89, 1413–1432.

(26) Fleming, I.; Dunogues, J.; Smithers, R. *Org. React.* **1989**, 37, 57–193.