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# Design and Synthesis of Pyrido[2,1-*b*][1,3,5]thiadiazine Library via Uncatalyzed Mannich-Type Reaction

Victor V. Dotsenko,<sup>\*,†,§</sup> Konstantin A. Frolov,<sup>†</sup> Tatyana M. Pekhtereva,<sup>‡</sup> Olena S. Papaianina,<sup>‡</sup> Sergey Yu. Suykov,<sup>‡</sup> and Sergey G. Krivokolysko<sup>†</sup>

<sup>†</sup>ChemEx Laboratory, Vladimir Dal' East Ukrainian National University, 91034 Lugansk, Ukraine

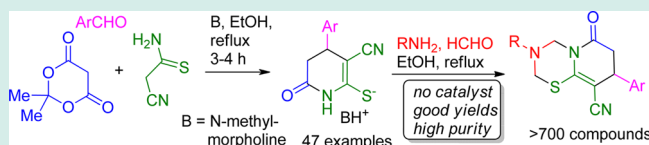
<sup>§</sup>Kuban State University, 350040 Krasnodar, Russian Federation

<sup>‡</sup>L. M. Litvinenko Institute of Physical-Organic Chemistry and Coal Chemistry NAS of Ukraine, 83114 Donetsk, Ukraine

**S** Supporting Information

**ABSTRACT:** This Research Article describes the synthesis of an over 700-member library of (8*R*/8*S*)-3-*R*-8-aryl-6-oxo-3,4,7,8-tetrahydro-2*H*,6*H*-pyrido[2,1-*b*][1,3,5]thiadiazin-9-carbonitriles by uncatalyzed Mannich-type reaction of *N*-methylmorpholinium (4*R*/4*S*)-4-aryl-3-cyano-6-oxo-1,4,5,6-tetrahydropyridin-2-thiolates with a set of primary amines and excessive HCHO. The scope and limitations of the reaction were studied. Starting thiolates were obtained in yields of 53–82% by multicomponent reaction of aromatic aldehydes, cyanothioacetamide, 2,2-dimethyl-1,3-dioxane-4,6-dione (Meldrum's acid), and *N*-methylmorpholine, followed by heterocyclization of the resulting Michael adducts.

**KEYWORDS:** uncatalyzed Mannich-type reaction, cyanothioacetamide, Meldrum's acid, tetrahydropyridine-2-thiolates, aminomethylation, pyrido[2,1-*b*][1,3,5]thiadiazines



## INTRODUCTION

1,3,5-Thiadiazines have attracted considerable attention over the years because of their biological activities and applications in medicine and agriculture (for reviews on the 1,3,5-thiadiazine chemistry, see refs 1–3). 1,3,5-Thiadiazines are known as antidermatophytes,<sup>4,5</sup> antifungal and fungistatic agents,<sup>6–14</sup> antimicrobials and bactericides,<sup>15–23</sup> antifibrinolytic agents,<sup>24–27</sup> tuberculostatics,<sup>28–31</sup> etc. Thiadiazines are widely used as insecticides.<sup>32</sup> 2-*tert*-Butylimino-3-isopropyl-5-phenyl-3,4,5,6-tetrahydro-2*H*-1,3,5-thiadiazin-4-one, also known as buprofezin or Applaud (Figure 1), was found to be the most active growth regulator on the greenhouse whitefly (*Trialeurodes vaporariorum*) and the brown planthopper (*Nilaparvata lugens*), which is regarded as one of the most serious insect pests in rice fields.<sup>33</sup> Buprofezin acts on the insects by strong suppression of oviposition because of the inhibition of

prostaglandin E2 biosynthesis<sup>34</sup> and by inhibition of chitin biosynthesis and integumentary cuticle deposition.<sup>35,36</sup> Recently, buprofezin has been found to be an acetylcholinesterase inhibitor in *B-biotype Bemisia tabaci*.<sup>37</sup> Another widely used compound is tetrahydro-3,5-dimethyl-1,3,5-thiadiazine-2-thione, also known under names Dazomet, Basamid, Mylone, Thiazone, Carbothialdine, or DMTT (Figure 1), which is a soil fumigant and used as a powerful insecticide, papermaking slimicide and nematocide, fungicide, and herbicide for cabbage, cucumber, maize, potato, and tomato plants.<sup>38–44</sup> Dazomet also behaves as a monodentate ligand toward some metal carbonyls.<sup>45</sup> In addition, Dazomet have been reported to possess an ovicidal effect on helminths eggs.<sup>46</sup> Another bioactive 1,3,5-thiadiazine, NIP-200 (3,5-dimethyl-4,6-diphenyl-tetrahydro-2*H*-1,3,5-thiadiazine-2-thione, Figure 1) is a potent hypolipidemic agent increasing the synthesis of bile acids as a result of the activation of cholesterol 7α-hydroxylase, the rate-limiting enzyme in the conversion of cholesterol to bile acids.<sup>47</sup> 3,5-Dibenzyltetrahydro-2*H*-1,3,5-thiadiazine-2-thione (D47, Dibenztion) is an antimycotic agent useful for treatment of dermatomycoses.<sup>48–51</sup>

One of the most effective approaches to 1,3,5-thiadiazines is based on the double Mannich-type reaction of thioamides, dithiocarbamates or related *S,N*-binucleophilic species with primary amines and formaldehyde.<sup>1–3</sup> Cyclic thioamides (γ- and δ-thiolactams, 2-mercaptoazoles, -azines, or their 2-

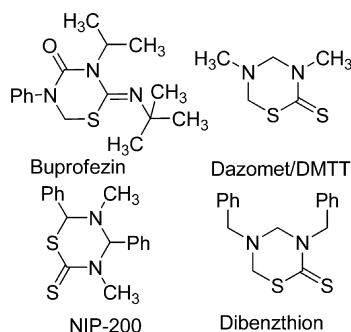
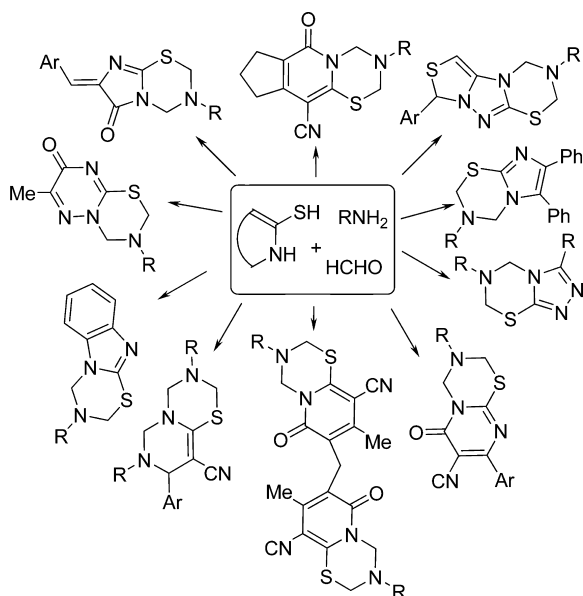


Figure 1. Biologically active 1,3,5-thiadiazines.



thioxotautomers) can also be successfully employed in this reaction, giving rise to a variety of ring-condensed 1,3,5-thiadiazines. In fact, the related syntheses of 1,2,4-triazolo[3,4-*b*][1,3,5]thiadiazines,<sup>52–60</sup> imidazo[2,1-*b*][1,3,5]thiadiazines,<sup>61,62</sup> 1,2,4-triazino[3,2-*b*][1,3,5]thiadiazines,<sup>63</sup> thiazolo[3',4':1,5][1,2,4]triazolo[3,4-*b*][1,3,5]thiadiazines,<sup>64</sup> 1,3,5-thiadiazino[3,2-*a*]benzimidazoles,<sup>64</sup> cyclopenta[*g*]pyrido[2,1-*b*][1,3,5]thiadiazines,<sup>65</sup> bis(pyrido[2,1-*b*][1,3,5]thiadiazin-7-yl)methanes,<sup>66</sup> pyrimido[2,1-*b*][1,3,5]thiadiazines,<sup>67</sup> and pyrimido[4,3-*b*][1,3,5]thiadiazines<sup>68–70</sup> have been reported (Scheme 1).

**Scheme 1. Diversity of Ring-Condensed 1,3,5-Thiadiazines**

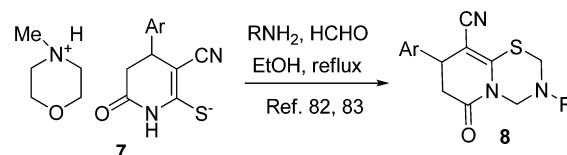


However, despite the diversity and availability of ring-condensed 1,3,5-thiadiazines, their use and applicability is much less studied.<sup>3</sup> Thus, pyrazolo[1,5-*c*][1,3,5]thiadiazine-2-diones **1** (Figure 2) have been reported as effective fungicides<sup>71–73</sup> and photosynthetic electron transport inhibitors.<sup>74,75</sup> 1,2,4-Triazolo[3,4-*b*][1,3,5]thiadiazines **2** were recognized as antibacterial agents,<sup>52–55</sup> while their oxo-analogs **3** showed moderate insecticidal activity.<sup>76</sup> In addition, 3-azacephalosporins **4** showed antibacterial activity,<sup>77</sup> and imidazo- and pyrimido[2,1-*b*][1,3,5]thiadiazines **5** were recognized as insecti-

cides.<sup>78–80</sup> Recently we found that pyrido[2,1-*b*][1,3,5]-thiadiazine **6** and related compounds showed significant inhibition against tick-borne encephalitis virus and Powassan virus.<sup>81</sup> Encouraged by this success, we turned our attention to the synthesis of related pyrido[2,1-*b*][1,3,5]thiadiazines in regard to library construction.

As we have shown in preliminary communications,<sup>82,83</sup> these pyrido[2,1-*b*][1,3,5]thiadiazines could be easily prepared by Mannich-type reaction of *N*-methylmorpholinium 4-aryl-6-oxo-3-cyano-1,4,5,6-tetrahydropyridine-2-thiolates **7** with primary amines and excess formaldehyde (Scheme 2). This effective and

**Scheme 2. Synthesis of Pyrido[2,1-*b*][1,3,5]thiadiazines **8****

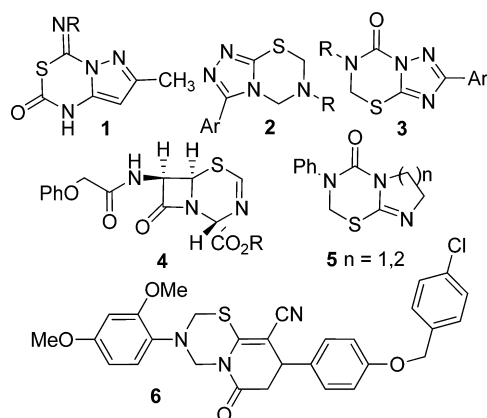


time-saving protocol is based on the use of inexpensive, readily available building blocks. In contrast to related double Mannich reactions,<sup>52–60</sup> no catalyst is required for this process. In most cases, the obtained pyrido[2,1-*b*][1,3,5]thiadiazines **8** were pure enough for analytical purposes that allowed us to exclude any purification steps. These results prompted us to study the scope and limitations of the reaction.

## RESULTS AND DISCUSSION

In the current study, we report an optimized protocol for the synthesis of 700+ membered library of 3-*R*-8-aryl-6-oxo-3,4,7,8-tetrahydro-2*H*,6*H*-pyrido[2,1-*b*][1,3,5]thiadiazine-9-carbonitriles **8** {1–47, 1–31}, employing a small library of *N*-methylmorpholinium 4-aryl-6-oxo-3-cyano-1,4,5,6-tetrahydropyridine-2-thiolates **7** {1–47} (Table 1), 37% aq. HCHO and a set of various primary amines **9** {1–31} (Figure 3). The required thiolates **7** {1–47} are easily accessible by one-pot condensation of aromatic aldehydes, cyanothioacetamide **10** and Meldrum's acid (2,2-dimethyl-1,3-dioxane-4,6-dione) **11** in the presence of *N*-methylmorpholine following the consequent cyclization of the isolable Michael adduct, *N*-methylmorpholinium 5-(3-amino-1-aryl-2-cyano-3-thioxopropyl)-2,2-dimethyl-4-oxo-4*H*-1,3-dioxin-6-olate **12**, as outlined in Scheme 3.<sup>81–90</sup> As the starting 1,4,5,6-tetrahydropyridine-2-thiolates **7** have been prepared as racemic mixtures of (4*R*)- and (4*S*)-enantiomers, all the pyrido[2,1-*b*][1,3,5]thiadiazine-9-carbonitriles **8** also were obtained as mixtures of (8*R*)- and (8*S*)-isomers.

We found that 1,4,5,6-tetrahydropyridine-2-thiolates **7** {1–47} easily react with primary amines **9** {1–31} and excess 37% aqueous formaldehyde under short-term heating in EtOH to give 3-*R*-8-aryl-6-oxo-3,4,7,8-2*H*,6*H*-pyrido[2,1-*b*][1,3,5]thiadiazine-9-carbonitriles **8** {1–47, 1–31} (Scheme 3). Presumably, the reaction proceeds through the formation of non-isolable intermediate **13**. The order of mixing of the reactants does not exert a noticeable influence on the yields of the final products. Thus, when a suspension of thiolate **7** {2} in EtOH was treated consecutively with 4.2 equiv of 37% HCHO and 1.1 equiv of PhCH<sub>2</sub>NH<sub>2</sub> **9** {3}, thiadiazine **8** {2,3} was obtained in 83% yield. Alternatively, when a hot solution of thiolate **7** {2} in aq. EtOH was added to the mixture of HCHO and benzylamine in EtOH, thiadiazine **8** {2,3} yielded in 77%. EtOH is the solvent of choice since thiolates **7** were found to be



**Figure 2. Biologically active ring-condensed 1,3,5-thiadiazines.**

Table 1. Diversity and Yields of 4-Aryl-6-oxo-3-cyano-1,4,5,6-tetrahydropyridine-2-thiolates 7

entry	compound	Ar	yield	entry	compound	Ar	yield
1	7{1}	Ph	67 <sup>a</sup>	24	7{24}	2-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	57 <sup>d</sup>
2	7{2}	2-MeC <sub>6</sub> H <sub>4</sub>	71 <sup>b</sup>	25	7{25}	3-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	64
3	7{3}	4-MeC <sub>6</sub> H <sub>4</sub>	71 <sup>a</sup>	26	7{26}	4-NO <sub>2</sub> C <sub>6</sub> H <sub>4</sub>	73
4	7{4}	4-EtC <sub>6</sub> H <sub>4</sub>	68	27	7{27}	4-MeSC <sub>6</sub> H <sub>4</sub>	83
5	7{5}	4- <i>i</i> -PrC <sub>6</sub> H <sub>4</sub>	70	28	7{28}	3-PhCH <sub>2</sub> OC <sub>6</sub> H <sub>4</sub>	68 <sup>g</sup>
6	7{6}	2-ClC <sub>6</sub> H <sub>4</sub>	73 <sup>c</sup>	29	7{29}	4-PhCH <sub>2</sub> OC <sub>6</sub> H <sub>4</sub>	78 <sup>g</sup>
7	7{7}	4-ClC <sub>6</sub> H <sub>4</sub>	79	30	7{30}	5-Br-2-MeOC <sub>6</sub> H <sub>3</sub>	76
8	7{8}	2-FC <sub>6</sub> H <sub>4</sub>	76	31	7{31}	3-Br-4-MeOC <sub>6</sub> H <sub>3</sub>	73
9	7{9}	3-FC <sub>6</sub> H <sub>4</sub>	69	32	7{32}	2-furyl	55 <sup>f</sup>
10	7{10}	4-FC <sub>6</sub> H <sub>4</sub>	71 <sup>d</sup>	33	7{33}	5-Me-2-furyl	53
11	7{11}	3-BrC <sub>6</sub> H <sub>4</sub>	61	34	7{34}	2-thienyl	71 <sup>e</sup>
12	7{12}	2-Cl-6-FC <sub>6</sub> H <sub>3</sub>	75	35	7{35}	3-Me-2-thienyl	57
13	7{13}	2,6-Cl <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	62	36	7{36}	4-HO-3-MeOC <sub>6</sub> H <sub>3</sub>	76
14	7{14}	2-MeOC <sub>6</sub> H <sub>4</sub>	78 <sup>h</sup>	37	7{37}	4-HO-3-EtOC <sub>6</sub> H <sub>3</sub>	82
15	7{15}	4-MeOC <sub>6</sub> H <sub>4</sub>	72 <sup>e</sup>	38	7{38}	4-EtO-3-MeOC <sub>6</sub> H <sub>3</sub>	69
16	7{16}	2-EtOC <sub>6</sub> H <sub>4</sub>	80 <sup>f</sup>	39	7{39}	3-Br-4-HO-5-MeOC <sub>6</sub> H <sub>2</sub>	74
17	7{17}	3,4-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	70 <sup>a</sup>	40	7{40}	3-Br-4-HO-5-EtOC <sub>6</sub> H <sub>2</sub>	69
18	7{18}	2,5-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	70 <sup>a</sup>	41	7{41}	1-naphthyl	72 <sup>a</sup>
19	7{19}	2,4-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	82 <sup>d</sup>	42	7{42}	2-MeO-1-naphthyl	65
20	7{20}	2,3-(MeO) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	70	43	7{43}	4-(4-ClC <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> O)C <sub>6</sub> H <sub>4</sub>	77 <sup>g</sup>
21	7{21}	3,4-(OCH <sub>2</sub> O)C <sub>6</sub> H <sub>3</sub>	79 <sup>a</sup>	44	7{44}	4-PhCH <sub>2</sub> O-3-MeOC <sub>6</sub> H <sub>3</sub>	75 <sup>g</sup>
22	7{22}	2,4,5-(MeO) <sub>3</sub> C <sub>6</sub> H <sub>2</sub>	71	45	7{45}	4-(2-ClC <sub>6</sub> H <sub>4</sub> CH <sub>2</sub> O)C <sub>6</sub> H <sub>4</sub>	57
23	7{23}	3,4,5-(MeO) <sub>3</sub> C <sub>6</sub> H <sub>2</sub>	67 <sup>a</sup>	46	7{46}	4-EtOC <sub>6</sub> H <sub>4</sub>	65 <sup>a</sup>
				47	7{47}	3-MeOC <sub>6</sub> H <sub>4</sub>	68

<sup>a</sup>Ref 89. <sup>b</sup>Ref 82. <sup>c</sup>Ref 84. <sup>d</sup>Ref 83. <sup>e</sup>Ref 85. <sup>f</sup>Ref 90. <sup>g</sup>Ref 81. <sup>h</sup>Ref 86.

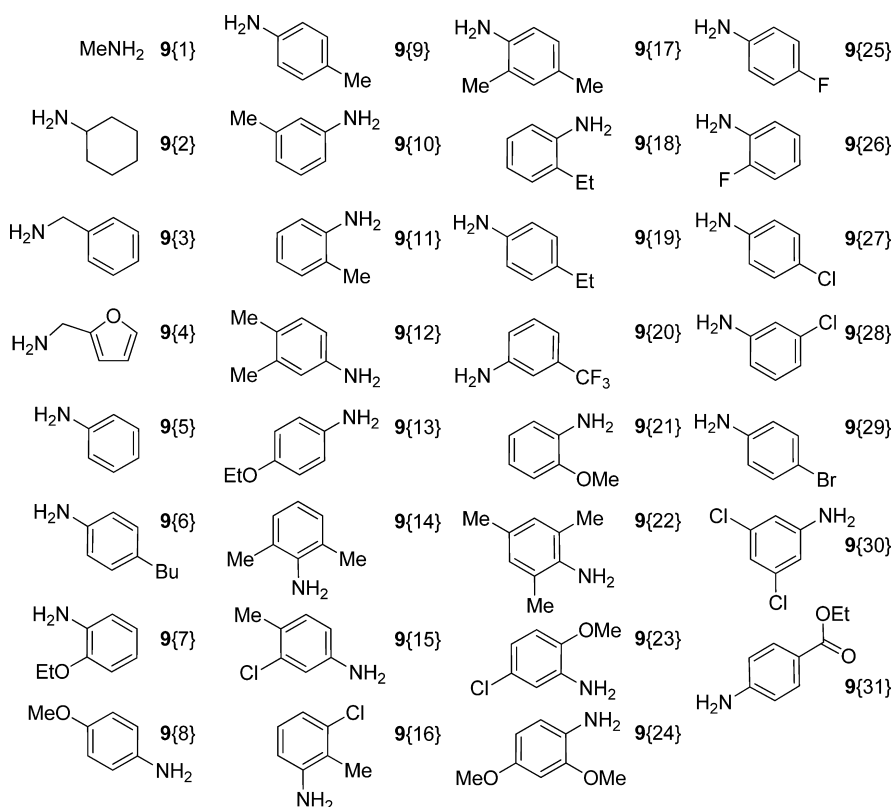
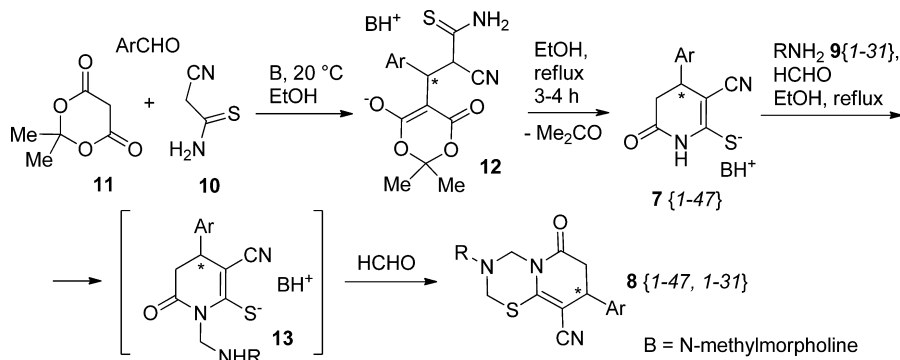


Figure 3. Diversity of primary amines 9.

less soluble in higher alcohols, whereas lower yields of thiadiazines **8** were obtained with MeOH. Both aliphatic and aromatic primary amines reacted under these conditions. However, we did not succeed to obtain pyridothiadiazines **8**

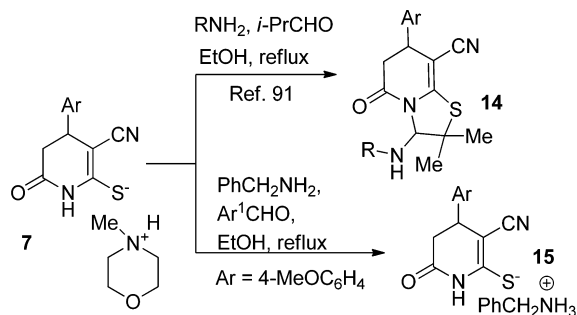
from anilines bearing strong electron-withdrawing substituents (e.g., NO<sub>2</sub>, CN, Ac, PhC(O)) in ortho- or para-position to amino group, heterocyclic amines (e.g., Gewald's 2-aminothiophenes, 2-aminopyridines, 2-aminothiazoles) and amino

Scheme 3. Synthesis of Starting Thioliates 7 and Pyrido[2,1-*b*][1,3,5]thiadiazine-9-carbonitriles 8

144 acids ( $\alpha$ - and  $\beta$ -alanines). Unsatisfactory results were also  
 145 obtained in the case of most sterically hindered amines, such as  
 146 2,6-dimethylaniline **9**{14}, 2-MeOC<sub>6</sub>H<sub>4</sub>NH<sub>2</sub> **9**{21}, 5-Cl-2-  
 147 MeOC<sub>6</sub>H<sub>3</sub>NH<sub>2</sub> **9**{23}, 2-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>NH<sub>2</sub>, *tert*-butylamine, and 2-  
 148 ethyl-6-methylaniline. Meanwhile, 2-EtOC<sub>6</sub>H<sub>4</sub>NH<sub>2</sub> **9**{7}, mesi-  
 149 dine **9**{22}, and 2-alkylanilines **9**{11,16,17,18} showed good  
 150 reactivity.

151 The scope of the reaction is limited to the use of  
 152 formaldehyde only. Under these conditions, aliphatic aldehydes  
 153 are known to react in different ways: thus, as we have shown  
 154 before,<sup>91</sup> the reaction of thioliates **7** with primary amines **9** and  
 155 isobutyraldehyde leads to thiazolo[3,2-*a*]pyridines **14** (Scheme  
 156 4). The reaction of thiolate **7**{6} with acetaldehyde and alkyl

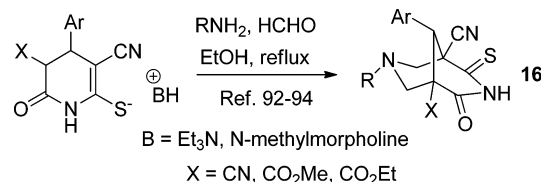
## Scheme 4. Reaction of Thioliates 7 with Amines 9 and Various Aldehydes



157 amines led to appearance of cherry red coloration and resulted  
 158 in the formation of tars, probably due to aldol-type  
 159 condensations of CH<sub>3</sub>CHO promoted by amines. The product  
 160 of the reaction of thiolate **7**{15} with *p*-toluidine and  
 161 benzaldehyde consists of a red tarry mass, while the reaction  
 162 with benzylamine and 4-ClC<sub>6</sub>H<sub>4</sub>CHO, 4-MeOC<sub>6</sub>H<sub>4</sub>CHO or  
 163 furfural afforded colorless crystalline solids, which appeared to  
 164 be the same compound, benzylammonium salt **15**. The  
 165 prolonged heating did not result in the formation of any  
 166 Mannich-type products with thioliates **7**.

167 The structure of thiolate component is another key factor  
 168 which has strong influence on the reaction outcome. As we  
 169 have reported earlier,<sup>92-94</sup> those 1,4,5,6-tetrahydropyridine-2-  
 170 thioliates that bear an electron-withdrawing group (C≡N,  
 171 CO<sub>2</sub>R) at C-5 position under the same conditions gave no  
 172 1,3,5-thiadiazine derivatives but readily underwent the double  
 173 Mannich-type reaction at C-3 and C-5 to afford 3,7-  
 174 diazabicyclo[3,3,1]nonanes **16** (Scheme 5). In contrast, the  
 175 aminomethylation of thioliates **7** proceeds with high regiose-  
 176 lectivity to give no C-3 attacked products, even in trace

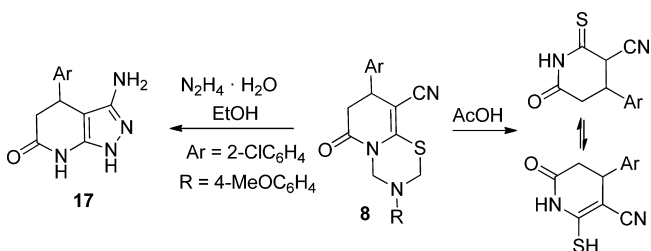
## Scheme 5. Aminomethylation of C-5 Substituted 1,4,5,6-Tetrahydropyridine-2-thioliates



177 amounts. We were a bit surprised by the fact that the yields of  
 178 pyrido[2,1-*b*][1,3,5]thiadiazines **8** are dependent on the nature  
 179 of an aromatic substituent at C-4 position of starting thiolate **7**.  
 180 The worst results were obtained with thioliates **7**{24-26,32,34}  
 181 (Ar = 2-NO<sub>2</sub>Ph, 3-NO<sub>2</sub>Ph, 4-NO<sub>2</sub>Ph, 2-furyl, 2-thienyl).

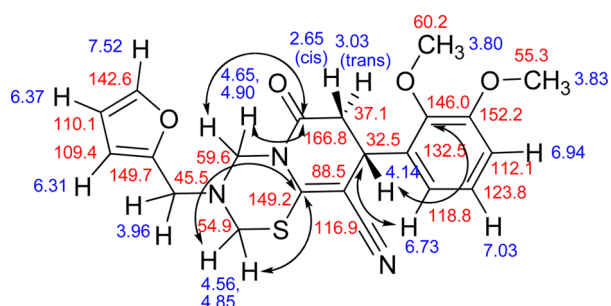
182 Overall, pyrido[2,1-*b*]thiadiazines **8** were obtained in yields ranging  
 183 from poor to excellent, depending mostly on the nature of  
 184 primary amine and thiolate. Table S1 (Supporting Information)  
 185 gives a few representative examples of how the yields of  
 186 pyrido[2,1-*b*]thiadiazines **8** depend on the nature of the reagents.

187 N-Methylmorpholinium 4-aryl-6-oxo-3-cyano-1,4,5,6-tetrahy-  
 188 dropyridine-2-thioliates **7**{1-47} are yellow or orange crystal-  
 189 line solids, soluble in hot aqueous EtOH but insoluble in  
 190 acetone and cold EtOH. Pyrido[2,1-*b*]thiadiazines **8** are colorless or  
 191 slightly yellowish crystalline solids, stable in neutral media,  
 192 soluble in hot acetone, EtOAc, DMF or DMSO, but sparingly  
 193 soluble in ether or alcohols. Compounds **8** are quite stable but  
 194 decompose when treated with AcOH or diluted strong acids to  
 195 form a complex mixture of *retro*-Mannich products. On the  
 196 other hand, 1,3,5-thiadiazine ring may be cleaved with  
 197 hydrazine hydrate to give after further heterocyclization the  
 198 known<sup>95</sup> pyrazolopyridines of general structure **17**. Thus, when  
 199 compound **8**{6,9} was reacted with excessive N<sub>2</sub>H<sub>4</sub>·H<sub>2</sub>O in hot  
 200 EtOH, pyrazolopyridine **17a** (Ar = 2-ClC<sub>6</sub>H<sub>4</sub>) was obtained in  
 201 36% yield (Scheme 6).

Scheme 6. Reactions of Pyrido[2,1-*b*]thiadiazines 8



Starting 1,4,5,6-tetrahydropyridine-2-thiolates **7** and pyrido[2,1-*b*][1,3,5]thiadiazines **8** were characterized by  $^1\text{H}$  NMR and IR spectroscopy. In the IR spectra of compounds **8**, strong absorptions at 1675–1690 and 2190–2205  $\text{cm}^{-1}$  were detected because of the  $\text{C}=\text{O}$  and conjugated  $\text{C}\equiv\text{N}$  groups, respectively. 1D NMR ( $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR,  $^{13}\text{C}$  DEPT NMR) and 2D NMR experiments ( $^1\text{H}$ – $^1\text{H}$  COSY,  $^1\text{H}$ ,  $^{13}\text{C}$ -HMBC, and  $^1\text{H}$ ,  $^{13}\text{C}$ -HSQC) were used for the complete and unambiguous  $^1\text{H}$  and  $^{13}\text{C}$  chemical shift assignments for selected compound, 3-(2-furylmethyl)-8-(2,3-dimethoxyphenyl)-6-oxo-3,4,7,8-tetrahydro-2*H*,6*H*-pyrido[2,1-*b*][1,3,5]-thiadiazine-9-carbonitrile **8**{20,4} (Figure 4). Full set of data of homo- and heteronuclear correlations is given in the Supporting Information, Table S2.



**Figure 4.** Assignment of signals and key  $^1\text{H}$ – $^{13}\text{C}$  HMBC correlations for compound **8**{20,4}.

In conclusion, an efficient and simple method for the preparation of 3-*R*-8-aryl-6-oxo-3,4,7,8-tetrahydro-2*H*,6*H*-pyrido[2,1-*b*][1,3,5]thiadiazine-9-carbonitriles **8**{1–47,1–31} using readily available starting materials by Mannich-type reaction is reported. A small library of *N*-methylmorpholinium 4-aryl-6-oxo-3-cyano-1,4,5,6-tetrahydropyridine-2-thiolates **7**{1–47}, 37% aq. HCHO and primary amines **9**{1–31} were used as starting compounds. The developed method requires no catalyst and usually gives acceptable yields of pure pyridothiadiazines.

## EXPERIMENTAL PROCEDURES

The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra of thiolates **7** and pyridothiadiazines **8** were performed on Bruker DRX-500 instrument (500.13 and 125.76 MHz for  $^1\text{H}$  and  $^{13}\text{C}$ , respectively) in  $\text{DMSO}-d_6$  using residual solvent peak ( $\delta$  2.49 ppm; 39.50 ppm for  $^1\text{H}$  and  $^{13}\text{C}$ , respectively) as reference or with  $\text{Me}_4\text{Si}$  as the internal standard. The  $^1\text{H}$  NMR spectra of benzylammonium salt **15** were recorded on a Bruker DRX-400 instrument (400.40 MHz) and  $^1\text{H}$  NMR spectrum of pyrazolopyridine **17a** was recorded on a Varian Gemini 200 instrument (199.975 MHz) in  $\text{DMSO}-d_6$ . NMR experiments for compound **8**{20,4} were performed on a Bruker Avance II 400 instrument (400.13 and 100.62 MHz for  $^1\text{H}$  and  $^{13}\text{C}$  respectively) in  $\text{DMSO}-d_6$  or  $\text{CCl}_4$ – $\text{DMSO}-d_6$  with  $\text{Me}_4\text{Si}$  as the internal standard. Data are reported as follows: chemical shift, multiplicity (*s* = singlet, *br. s* = broad singlet, *d* = doublet, *dd* = doublet of doublets, *t* = triplet, *q* = quartet, *m* = multiplet), coupling constants (Hz), integration and assignment of peak.

FT-IR spectra of thiolates **7** were recorded in KBr pellets using Thermo Nicolet Avatar 370 FT-IR Spectrometer. IR spectra of thiadiazines **8**, benzylammonium salt **15**, and

pyrazolopyridine **17a** were recorded on an IKS-29 IR-spectrometer (LOMO, USSR).

LCMS analyses were obtained on a PE SCIEX API 150EX mass spectrometer (API-ES) following separation on a Shimadzu LC-10AD liquid chromatography system with Waters XBridge C18 3.5  $\mu\text{m}$  ( $4.6 \times 150$  mm) column, equipped with Shimadzu SP D-10A UV-vis detector (220 and 254 nm) and Sedex 75 ELSD detector.

Elementary analyses were taken on a Carlo Erba Strumentazione 1106 Analyzer.

Melting points were measured on a Kofler hot stage and are uncorrected. The purity of the compounds were checked by TLC (thin layer chromatography) on Silufol UV 254 plates (sorbent, Silpearl, large-pore silicagel after Pitra with luminiscent indicator for UV 254 on the aluminum foil; binder, starch) in the acetone–hexane (1:1) system; spots were visualized with iodine vapors and UV light.

**Synthesis of Starting *N*-Methylmorpholinium 4-Aryl-3-cyano-6-oxo-1,4,5,6-tetrahydropyridine-2-thiolates **7**{1–47}. General Procedure.** The thiolates **7**{1–47} were obtained in a manner analogous to reported procedures<sup>81–90</sup> as follows: A 0.5 L round-bottom flask fitted with an overhead stirrer was charged with the corresponding aromatic aldehyde (0.1 mol), cyanothioacetamide **10** (10.0 g, 0.10 mol) and EtOH (100 mL). *N*-Methylmorpholine (0.8–1.0 mL) was added, and the mixture was stirred for 1 h at 20 °C (yellow/orange crystalline 3-aryl-2-cyanoprop-2-enethioamides may precipitate from the solution). Then Meldrum's acid (2,2-dimethyl-1,3-dioxane-4,6-dione) **11** (15.0 g, 0.104 mol) and *N*-methylmorpholine (16.5 mL, 0.15 mol) were added, and the solution was stirred vigorously until the mixture became pale yellow and a white solid of the Michael adduct **12** precipitated. (If the precipitate does not appear within 20 min, the mixture was refluxed for 2–4 h and worked up as shown below.) The obtained slurry was stirred for 0.5 h. Then the flask was fitted with a reflux condenser. The mixture was refluxed to complete dissolution of the Michael adduct **12** and until evolution of  $\text{CO}_2$  ceased ( $\sim$ 2–4 h). The solution was evaporated to a syrupy consistency and treated with dry acetone (100 mL). The crystalline precipitate of the corresponding thiolate **7** separates upon cooling to 15 °C and stirring (or seeding). The mixture was allowed to stand overnight, after which the solid was filtered off, washed with cold EtOH and acetone to give *N*-methylmorpholinium 4-aryl-3-cyano-6-oxo-1,4,5,6-tetrahydropyridine-2-thiolates **7**{1–47} in 53–83% yields. The compounds were used without further purification.

**Synthesis of (8*R*/8*S*)-3-*R*-8-Aryl-6-oxo-3,4,7,8-tetrahydro-2*H*,6*H*-pyrido[2,1-*b*][1,3,5]thiadiazine-9-carbonitriles **8**{1–47,1–31}. General Procedure.** Pyrido[2,1-*b*][1,3,5]-thiadiazines **8** were prepared as follows: the corresponding thiolate **7**{1–47} (2.5 mmol) was dissolved in 15–20 mL of warm EtOH; water (3–5 mL) may be added if appropriate. The obtained solution may be filtered through a paper filter to remove trace solids. To the solution, a primary amine **9**{1–31} (2.6–2.7 mmol, 1.04–1.08 equiv) and an excess of 37% aq. HCHO (5.0 mL, *d* = 1.08 g/mL, 66.6 mmol) were added, and the mixture was refluxed for 2–4 min under vigorous stirring until the product began to separate from the boiling solution. If no solid separated, the solution was allowed to cool to room temperature and left for 24–72 h. The crystalline solid was collected and washed with water, cold EtOH, ether, and then purified (if appropriate) by recrystallization. Selected data on the yields of pyrido[2,1-*b*][1,3,5]thiadiazines **8** are given in 310

311 Table S1 (Supporting Information). Spectra of selected  
312 pyridothiadiazines **8** are given as PDF files in the archive  
313 (Supporting Information).

## 314 ■ ASSOCIATED CONTENT

### 315 ● Supporting Information

316 Further details on the experimental procedures and spectra.  
317 This material is available free of charge via the Internet at  
318 <http://pubs.acs.org>.

## 319 ■ AUTHOR INFORMATION

### 320 Corresponding Author

321 \*E-mail: Victor\_Dotsenko@bigmir.net.

### 322 Author Contributions

323 The study was initiated and designed by V.V.D. Compounds  
324 were synthesized and characterized by V.V.D. and K.A.F. NMR  
325 study was performed by S.Y.S., T.M.P., and O.S.P.; V.V.D.  
326 wrote the manuscript and Supporting Information. The study  
327 was supervised by S.G.K. All authors discussed and approved  
328 the publication of the manuscript.

### 329 Notes

330 The authors declare no competing financial interest.

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