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Convenient Synthesis of ϵ -Halo- β -ketoesters and γ, γ' -Dibromoalkanones by Regio- and Chemoselective Reaction of 2-Alkylidenetetrahydrofurans with Boron Trihalides: A "Ring-Closure/Ring-Cleavage" Strategy

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$$R^{1}$$
 R^{2} R^{5} R^{5} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2} R^{3} R^{4} R^{4}

The reaction of boron tribromide and boron trichloride with 2-alkylidenetetrahydrofurans, readily available on the basis of cyclizations of free and masked dianions with 1,2-dielectrophiles, allowed an efficient synthesis of a variety of carbonyl compounds with remote halide functionality. This includes the chemo- and regioselective synthesis of 6-bromo- and 6-chloro-3-oxoalkanoates and 1,7-dibromoheptan-4-ones. The approach outlined herein can be regarded as a "ring-closure/ring-cleavage" strategy.

Introduction

Functionalized carbonyl compounds containing a halide group at a remote position represent versatile synthetic building blocks.¹⁻⁴ 1,3-Dicarbonyl derivatives containing a halide group at a remote position are more rare. Despite their preparative usefulness, syntheses of 6-halo-3-oxoalkanoates (ϵ -halo- β -ketoesters) and the related 6-halo-1,3-diketones have only scarcely been reported in the literature. For example, 7-bromoheptane-2,4-dione has been prepared by reaction of 5-bromopent-1-yne with acetic anhydride. The reaction of α-acetylγ-butyrolactone with HBr has been reported to give 5-bromopentan-2-one by ring-opening and subsequent decarboxylation.² The reaction of 3-oxoalkanoate dianions with 1-bromo-2-chloroethane has been reported to give 6-chloro-3-oxoalkanoates.3 In contrast, it has been reported that bis-1,3-diketones are not directly available by reaction of disodio 1,3-diketones with 1,2-dibromoethane. 4

We have recently reported⁵ a convenient synthesis of 6-bromo-3-oxoalkanoates by reaction of BBr₃ with 2-alkylidenetetrahydrofurans. The starting materials, 2-alkylidenetetrahydrofurans, are readily available based on cyclizations of 1,2-dielectrophiles with 1,3-dicarbonyl dianions or 1,3-bis-silyl enol ethers (masked dianions)reactions developed by us in the recent years.⁶ Thus, our approach to 6-bromo-3-oxoalkanoates can be regarded as a "ring-closure/ring-cleavage" strategy. Whereas the BBr₃mediated cleavage of methylaryl ethers is well-known and broadly used,⁷ reactions of other ethers are more rare. Known examples include the formation of ω -bromoalkanols by ring-opening of cyclic ethers with BBr₃/ MeOH 8 or the transformation of lactones into ω -halocarboxylic acids. Herein, we wish to report full details of our "ring-closure/ring-cleavage" strategy and studies related to the preparative scope.

Results and Discussion

The reaction of 2-alkylidenetetrahydrofuran **2a**, prepared by cyclization of the dianion **1a** of ethyl aceto-

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SCHEME 1a

OLi OLi
OEt

1a

ref. 10a

2a

$$ii$$
 4 BBr_3

Br₂B--O
OEt

Br
OEt

Br
OEt

Br
OEt

Br
3a (76%)

 a Key: (i) (1) 2.3 equiv of LDA, THF, 0 °C, 1 h, (2) Br(CH₂)₂Cl, $-78 \rightarrow 20$ °C, 14 h, then reflux, 12 h; (ii) (1) 4 equiv of BBr₃, CH₂Cl₂, 0 \rightarrow 20 °C, 12 h, 20 °C, 6 h, (2) H₂O.

SCHEME 2. Synthesis of 3a-v^a

$$R^{1}$$
 R^{5} R^{5} R^{5} R^{5} R^{2} R^{2} R^{2} R^{5} R^{5} R^{2} R^{5} R^{5

 a Key: (i) (1) 4 equiv of BX3, CH2Cl2, 0 \rightarrow 20 °C, 12 h, 20 °C, 6 h, (2) H2O.

acetate with 1-bromo-2-chloroethane, 10 with BBr $_3$ afforded ethyl 6-bromo-3-oxohexanoate (3a) in 76% yield (Scheme 1). The reaction presumably proceeds by activation of 2a (intermediate A), ring-cleavage (intermediate B), and subsequent protonation of the enolate. During the optimization, the use of an excess of BBr $_3$ (4 equiv) proved to be important.

The preparative scope of our methodology was studied. The required starting materials were prepared by literature methods. 10-12 The reaction of BBr₃ with 2-alkylidenetetrahydrofurans 2b-e and 2g-i afforded the 2-alkyl- and 2-aryl-6-bromo-3-oxoalkanoates **3b-e** and 3g-i (Scheme 2, Table 1). During the formation of 3i, the methylaryl ether of the starting material was cleaved. In addition to **3i** (72%), simple demethylation of methyl (dihydrofuran-2-ylidene)(4-methoxyphenyl)acetate (2i) and formation of methyl (dihydrofuran-2-ylidene)(4-hydroxyphenyl)acetate (2i') was observed. Treatment of 2-alkylidenetetrahydrofuran **2f** with boron trichloride (BCl₃) afforded the 6-chloro-2-phenyl-3-oxoalkanoate **3f**. This experiment showed that not only a bromide but also a chloride function can be introduced in good yield. The reaction of BBr₃ with 2-alkylidenetetrahydrofurans 2j-v afforded the 6-bromo-3-oxoalkanoates 3j-v (Table 1). All

TABLE 1. Products and Yields

3	\mathbb{R}^1	\mathbb{R}^2	\mathbb{R}^3	R^4	${ m R}^5$	X	% yield ^a
a	Н	Н	Н	H	OEt	Br	76
b	H	Η	H	Oct	OEt	Br	95
c	H	Η	H	Dec	OEt	Br	81
d	H	\mathbf{H}	H	$(CH_2)_3Cl$	OMe	Br	84
\mathbf{e}	H	Η	H	Bn	OMe	Br	96
f	H	Η	H	Ph	OMe	Cl	84
\mathbf{g}	H	Η	H	$4\text{-MeC}_6\mathrm{H}_4$	OMe	Br	89
h	H	Η	H	$4\text{-ClC}_6\mathrm{H}_4$	OMe	Br	77
i	H	Η	H	$4\text{-HOC}_6\text{H}_4$	OMe	Br	72^b
j	H	Η	Ph	Ph	OMe	Br	96
k	H	Η	\mathbf{Et}	H	OEt	Br	83
1	H	Η	\Pr	H	OEt	Br	96
m	H	Η	$(CH_2)_3Cl$	H	OMe	Br	86
n	H	\mathbf{H}	-(C	$H_2)_9-$	OEt	Br	87
0	Me	Η	H	H	OMe	Br	80
\mathbf{p}	Et	Η	H	H	OMe	Br	91
\mathbf{q}	Bu	Η	H	H	OMe	Br	75
r	$\mathrm{CH_{2}Br}$	Η	H	H	OEt	Br	93
\mathbf{s}	$\mathrm{CH_{2}Cl}$	Η	H	H	OMe	Br	80
t	H	\mathbf{Et}	H	H	OMe	Br	82
u	Η	Ph	H	H	OEt	Br	83
\mathbf{v}	H	Η	H	H	Ph	Br	98

^a Yields of isolated products. ^b From **2i** ($R^4 = 4$ -(MeO)C₆H₄).

SCHEME 3. Synthesis of 3w^a

 a Key: (i) (1) 4 equiv of BBr₃, CH₂Cl₂, 0 \rightarrow 20 °C, 12 h, 20 °C, 6 h, (2) H₂O.

reactions proceeded in good to very good yield and with very good chemo- and regioselectivity.

The reaction of BBr₃ with 5-vinyl-2-alkylidenetetrahydrofuran ($2\mathbf{w}$), readily available by cyclization of dilithiated ethyl acetoacetate with 1,4-dibromo-2-butene, ^{10a,13} resulted in formation of ethyl 8-bromo-3-oxooct-6-enoate ($3\mathbf{w}$) in 98% yield (Scheme 3). In this reaction, the cleavage of the tetrahydrofuran moiety proceeded by an S_N2' mechanism with migration of the double bond. Notably, a direct synthesis of $3\mathbf{w}$ by reaction of dilithiated ethyl acetoacetate with 1,4-dibromobut-2-ene was not possible.

The reaction of BBr₃ with tetrahydro[2,3']bifuranyliden-2'-one $\mathbf{5a}^{10}$ afforded 1,7-dibromoheptan-4-one ($\mathbf{6a}$) (Scheme 4, Table 2). The formation of 6a can be explained by BBr₃mediated ring-opening of the cyclic enol, cleavage of the lactone, decarboxylation, and final protonation of the enolate D. The novel unsymmetrical 1,7-dibromoheptan-4-ones **6b−d** were successfully prepared from the known 2-alkylidenetetrahydrofurans **5b**-**d**. ¹² Treatment of the novel 2-alkylidenetetrahydrofuran trans-5e with BBr₃ afforded a 1:1 diastereomeric mixture of syn- and anti-1,7-dibromo-6-methyloctan-4-one (**6e**) (Scheme 5). This experiment suggests that, in contrast to the cleavage of methylaryl ethers, the ring opening of **5e** proceeds by an $S_{N}1$ rather than an $S_{N}2$ mechanism and thus results in epimerization. Therefore, a racemization would be expected for the BBr₃-mediated cleavage of enantiomerically pure 2-alkylidenetetrahydrofurans **5b**-**e** and **2o**-

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SCHEME 4. Synthesis of 6a-da

 a Key: (i) (1) 4 equiv of BBr₃, CH₂Cl₂, 0 \rightarrow 20 °C, 12 h, 20 °C, 6 h, (2) H₂O.

TABLE 2. Products and Yields

6	R	$\%$ yield a
a	Н	73
b	${f Me}$	88 68 85
c	\mathbf{Et}	68
d	$\mathrm{CH_{2}Cl}$	85

SCHEME 5. Synthesis of $6e^a$

^a Yields of isolated products.

 a Key: (i) 2 equiv of TiCl₄, CH₂Cl₂, 4 Å molecular sieves, $-78 \rightarrow 20$ °C, 37%; (ii) (1) 4 equiv of BBr₃, CH₂Cl₂, 0 \rightarrow 20 °C, 12 h, 20 °C, 6 h, (2) H₂O, 87%.

s. 1,7-Dibromoheptan-4-ones represent versatile synthetic building blocks. ¹⁴ For example, **6a** has been used for the synthesis of medium-sized carba- and heterocycles. ^{14a,b} Notably, unsymmetrical 1,7-dibromoheptan-4-ones are not readily available by other methods. ^{14f}

The reaction of 2-acetyl- γ -butyrolactone with HBr has been reported to give 1-bromopentan-4-one (7). Treatment of α -acetyl- γ -butyrolactone with BBr $_3$ resulted in formation of 7 in 78% yield (Scheme 6). This result indicates that the employment of BBr $_3$ can be successfully employed for the synthesis of simple γ -bromoketones by cleavage of simple lactones.

SCHEME 6. Synthesis of 7^a

 a Key: (i) 1) 4 equiv of BBr₃, CH₂Cl₂, 0 \rightarrow 20 °C, 12 h, 20 °C, 6 h, (2) H₂O, 78%.

Conclusions

In summary, the reaction of borontribromide with 2-alkylidenetetrahydrofurans allowed an efficient synthesis of a variety of 6-bromo- and 6-chloro-3-oxo-alkanoates and 1,7-dibromoheptan-4-ones. The reactions proceeded with very good chemo- and regioselectivity.

Experimental Section

2-Alkylidenetetrahydrofurans. The required starting materials were prepared by published procedures: 2-Alkylidenetetrahydrofurans 2k-n and 2v were prepared in one step by cyclization of dilithiated 1,3-dicarbonyl compounds 1 with 1-bromo-2-chloroethane (Scheme 3).¹⁰ 2-Alkylidenetetrahydrofurans 2b-e, containing an alkyl group located at the exocyclic double bond, were prepared in two steps by synthesis of 2-alkylidenetetrahydrofurans and subsequent lithiation and alkylation (Scheme 3). 10b The aryl-substituted 2-alkylidenetetrahydrofurans **2f**-**j** were prepared by bromination of 2-alkylidenetetrahydrofurans and subsequent Suzuki reaction.¹¹ 2-Alkylidenetetrahydrofurans **2o**-**u** were obtained in one step by cyclization of 1,3-bis-silyl enol ethers with epoxides. 12 Compound 5a was prepared by cyclization of dilithiated α-acetyl-γ-butyrolactone with 1-bromo-2-chloroethane. 10 Compounds **5b**-**d** are available by cyclization of the α -acetyl- γ butyrolactone α-acetyl-γ-butyrolactone derived 1,3-bis-silyl enol ether with epoxides. 12a

General Procedure for the Reaction of 2-Alkylidenetetrahydrofurans with Boron Tribromide or Boron Trichloride. To a $\mathrm{CH_2Cl_2}$ solution (10 mL/mmol) of the 2-alkylidenetetrahydrofuran (2, 5) (1 equiv) was added BBr₃ (4 equiv) at 0 °C. The reaction mixture was allowed to warm to 20 °C over 12 h and was stirred for 6 h at 20 °C. Water (15 mL) was slowly added to the reaction mixture, and the organic layer was separated. The aqueous layer was extracted with $\mathrm{CH_2Cl_2}$ (3 × 30 mL). The combined organic extracts were dried $\mathrm{(Na_2SO_4)}$ and filtered, and the filtrate was concentrated in vacuo. The residue was purified by chromatography (silica gel, n-hexane/EtOAc = $100:1 \rightarrow 1:1$) to give the 6-bromo-3-oxoalkanoate 3 or 1,7-dibromoheptan-4-one 6.

6-Bromo-3-oxohexanoic Acid Ethyl Ester (3a). Starting with 2a (0.200 g, 1.28 mmol) and BBr₃ (1.283 g, 5.12 mmol) in CH₂Cl₂ (10 mL), **3a** was isolated after chromatography (silica gel, n-hexane/EtOAc = 50:1 \rightarrow 10:1) as a slightly yellow oil (0.229 g, 76%, 7% of enol form). ¹H NMR (CDCl₃, 300 MHz): $\delta=1.\overline{29}$ (t, J=7.2 Hz, 3 H), 2.16 (quint, J=6.6 Hz, 2 H), 2.77 (t, J = 6.9 Hz, 2 H), 3.45 (t, J = 7.2 Hz, 2 H), 3.46 (s, 2H), 4.21 (q, J = 7.2 Hz, 2 H), 5.03 (s, 1 H, CH of enol), 12.12(s, 1 H, OH of enol). 13 C NMR (CDCl₃, 150 MHz): $\delta_{\rm C} = 14.1$, 26.17, 32.9, 40.8, 49.3, 61.4, 90.0 (enol), 167.0, 172.5 (enol), 176.4 (enol), 201.6. IR (neat, cm⁻¹): $\tilde{\nu} = 2982$ (w), 2937 (w), 1743 (s), 1716 (s), 1649 (w), 1440 (m), 1369 (m), 1320 (s), 1247 (s), 1184 (m), 1116 (m), 1096 (m), 1031 (m), 975 (w), 942 (w), 851 (w), 806 (w), 557 (w). MS (EI, 70 eV): m/z = 237 (M⁺, 1), 191 (5), 157 (14), 148 (63), 130 (100), 115 (35), 87 (18). The exact molecular mass $m/z = 236.0048 \pm 2$ ppm [M⁺] for C₈H₁₃O₃Br was confirmed by HRMS (EI, 70 eV). Anal. Calcd for $C_8H_{13}O_3Br$ (237.093): C, 40.53; H, 5.53. Found: C, 40.13; H, 6.11.

2-(4-Bromobutyryl)decanoic Acid Ethyl Ester (3b). Starting with 2b (0.070 g, 0.26 mmol) and BBr₃ (0.264 g, 1.04

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mmol) in CH₂Cl₂ (5 mL), **3b** was isolated after chromatography (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 1:1$) as a slightly yellow oil (0.086 g, 95%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.89$ (t, J = 7.2 Hz, 3 H), 1.20–1.34 (m, 15 H), 1.80–187 (m, 2 H), 2.14 (quint, J = 6.6 Hz, 2 H), 2.62–2.82 (m, 2 H), 3.43 (t, J = 7.2 Hz, 1 H), 3.44 (t, J = 6.6 Hz, 2 H), 4.20 (q, J = 7.2 Hz, 2 H). ¹³C NMR (CDCl₃, 150 MHz): $\delta_{\rm C} = 14.2$, 14.3, 22.8, 26.4, 27.6, 28.4, 29.3, 29.4, 29.5, 32.0, 33.2, 40.0, 59.4, 61.5, 169.9, 204.5. IR (neat, cm⁻¹): $\tilde{\nu} = 2956$ (m), 2927 (s), 2856 (w), 1744 (s), 1716 (s), 1462 (w), 1441 (w), 1406 (w), 1371 (w), 1298 (w), 1250 (w), 1186 (w), 1163 (w), 1123 (w), 1095 (w), 1028 (w). MS (EI, 70 eV): m/z = 349 (M⁺, 3), 304 (3), 251 (7), 236 (26), 227 (8), 200 (100), 197 (5), 159 (94), 135 (8), 123 839), 121 826), 115 (16). The exact molecular mass $m/z = 348.1300 \pm 2$ ppm [M⁺] for C₁₆H₂₉O₃Br was confirmed by HRMS (EI, 70 eV).

2-(4-Bromobutyryl)undecanoic Acid Ethyl Ester (3c). Starting with 2c (0.100 g, 0.34 mmol) and BBr₃ (0.338 g, 1.35 mmol) in CH₂Cl₂ (5 mL), **3c** was isolated after chromatography (silica gel, n-hexane/EtAOc = $50:1 \rightarrow 20:1$) as a slightly yellow oil (0.103 g, 81%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.88$ (t, J $= 6.9 \text{ Hz}, \bar{3} \text{ H}), 1.19 - 1.13 \text{ (m, 19 H)}, 1.80 - 1.88 \text{ (m, 2 H)}, 2.14$ (quint, J = 6.6 Hz, 2 H), 2.64-2.82 (m, 2 H), 3.42 (t, J = 7.5Hz, 1 H), 3.44 (t, J = 6.6 Hz, 2 H), 4.20 (q, J = 7.2 Hz, 2 H). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 14.30, 14.32, 22.9, 26.4, 27.6,$ 28.4, 29.49, 29.51, 29.70, 29.73, 29.77, 32.1, 33.2, 39.8, 59.4, 61.6, 169.9, 204.5. IR (neat, cm $^{-1}$): $\tilde{\nu}=2956$ (m), 2927 (s), 2855 (m), 1744 (s), 1716 (s), 1645 (w), 1463 (m), 1407 (w), 1371 (w), 1300 (w), 1251 (m), 1184 (m), 1150 (m), 1128 (w), 1096 (w), 1031 (w). MS (EI, 70 eV): m/z = 377 (M⁺, 6), 256 (5), 252 (8), 229 (81), 199 (6), 183 (7), 150 (100), 121 (29). The exact molecular mass $m/z=376.1613\pm 2$ ppm [M⁺] for $\mathrm{C_{18}H_{33}O_{3}Br}$ was confirmed by HRMS (EI, 70 eV). Anal. Calcd for C₁₈H₃₃O₃-Br (377.361): C, 57.29; H, 8.81. Found: C, 56.92; H, 8.15.

6-Bromo-2-(3-chloropropyl)-3-oxohexanoic Acid Methyl Ester (3d). Starting with 2d (0.100 g, 0.46 mmol) and BBr₃ (0.463 g, 1.83 mmol) in CH_2Cl_2 (10 mL), **3d** was isolated after chromatography (silica gel, *n*-hexane/EtOAc = $50:1 \rightarrow 1:1$) as a slightly yellow oil (0.116 g, 84%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.74 - 1.83$ (m, 2 H), 2.01 (quint, J = 7.5 Hz, 2 H), 2.10–2.19 (m, 2 H), 2.63–2.88 (m, 2 H), 3.42–3.57 (m, 5 H, CH), 3.76 (s, 3 H). $^{13}\mathrm{C}$ NMR (CDCl $_3$, 150 MHz): $\delta_\mathrm{C}=25.6$, 26.3, 30.3, 33.1, 39.9, 44.4, 52.8, 58.2, 169.9, 203.7. IR (neat, cm⁻¹): $\tilde{\nu} = 2956$ (m), 1745 (s), 1715 (s), 1639 (w), 1438 (m), 1406 (w), 1369 (w), 1273 (m), 1254 (m), 1203 (m), 1178 (m), 1150 (m), 1102 (w), 1078 (w), 1045 (w). MS (EI, 70 eV): m/z = $300 (M^+, 1), 269 (2), 263 (3), 237 (2), 222 (3), 205 (2), 177 (5),$ 150 (98), 148 (100), 129 (2), 123 (26), 118 (5), 116 (11), 112 (6), 101 (4). The exact molecular mass $m/z = 297.9971 \pm 2$ ppm $[M^+]$ for $C_{10}H_{16}O_3BrCl$ was confirmed by HRMS (EI, 70 eV). Anal. Calcd for C₁₀H₁₆O₃BrCl (299.592): C, 40.09; H, 5.38. Found: C, 39.85; H, 5.95.

2-Benzyl-6-bromo-3-oxohexanoic Acid Methyl Ester (3e). Starting with 2e (0.100 g, 0.43 mmol) and BBr₃ (0.436 g, 1.72 mmol) in CH₂Cl₂ (5 mL), **3e** was isolated after chromatography (silica gel, n-hexane/EtOAc = $50:1 \rightarrow 1:1$) as a slightly yellow oil (0.130 g, 96%, 5% of enol form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.02 - 2.13$ (m, 2 H), 2.48 (dt, J = 18.3, 6.6 Hz, 1 H), 2.77 (dt, J = 18.3, 6.6 Hz, 1 H), 3.18 (d, J = 7.5 Hz, 2 H), 3.35 (dt, J = 6.6, 1.4 Hz, 2 H), 3.71 (s, 3 H), 3.82 (t, J = 7.5Hz), 7.15-7.30 (m, 5 H), 12.09 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): δ_C = 26.2, 32.9, 34.2, 41.0, 52.7, 60.5, 102.0 (enol), 126.9, 128.8, 128.9, 138.1, 169.5, 172.0 (enol), 203.7. IR (neat, cm⁻¹): $\tilde{v} = 3029$ (w), 3002 (w), 2953 (m), 1746 (s), 1715 (s), 1633 (m), 1604 (w), 1495 (w), 1437 (m), 1405 (w), 1369 (m), 1316 (m), 1255 (m), 1212 (s), 1170 (s), 1105 (m), 1078 (m), 1060 (m), 1031 (w), 983 (w), 751 (m), 701 (m). MS (EI, 70 eV): $m/z = 313 \; (\mathrm{M}^+, \, 2), \, 283 \; (1), \, 255 \; (6), \, 232 \; (10), \, 200 \; (100), \, 172$ (14), 163 (100), 148 (21), 131 (55), 103 (14), 91 (27), 77 (9). The exact molecular mass $m/z = 312.0361 \pm 2$ ppm [M⁺] for C₁₄H₁₇O₃Br was confirmed by HRMS (EI, 70 eV).

6-Chloro-3-oxo-2-phenylhexanoic Acid Methyl Ester (3f). Starting with 2f $(0.030~g,\,0.14~mmol)$ and BCl₃ $(0.55~mL,\,$

1.0 M in heptane, 0.55 mmol) in CH₂Cl₂ (5 mL), **3f** was isolated after chromatography (silica gel, n-hexane/EtOAc = 50:1 → 20:1) as a colorless oil (0.030 g, 84%). ¹H NMR (CDCl₃, 300 MHz): δ = 2.01 (quint, J = 6.6 Hz, 2 H), 2.65 (t, J = 6.9 Hz, 2 H), 3.53 (t, J = 6.3 Hz, 2 H), 3.76 (s, 3 H), 4.74 (s, 1 H), 7.27−7.38 (m, 5 H). ¹³C NMR (CDCl₃, 150 MHz): δ _C = 26.5, 38.7, 44.5, 52.8, 65.1, 127.3, 128.7, 129.0, 134.2, 169.0, 207.4. IR (neat, cm⁻¹): $\tilde{\nu}$ = 3064 (w), 3031 (w), 2963 (m), 2926 (m), 2870 (w), 2857 (w), 1743 (s), 1717 (s), 1644 (w), 1603 (w), 1495 (w), 1450 (s), 1440 (s), 1408 (m), 1370 (m), 1350 (m), 1309 (m), 1260 (s), 1205 (s), 1158 (s), 1113 (s), 1095 (s), 1026 (s), 798 (w), 734 (w), 702 (s), 650 (w). MS (EI, 70 eV): m/z = 254 (M⁺, 3), 222 (13), 196 (4), 164 (43), 149 (71), 118 (55), 104 (100), 91 (38), 77 (40). The exact molecular mass m/z = 254.0710 ± 2 ppm [M⁺] for C₁₃H₁₅O₃Cl was confirmed by HRMS (EI, 70 eV).

6-Bromo-3-oxo-2-p-tolylhexanoic Acid Methyl Ester (3g). Starting with 2g (0.090 g, 0.39 mmol) and BBr₃ (0.388 g, 1.55 mmol) in CH₂Cl₂ (10 mL), 3g was isolated after chromatography (silica gel, n-hex/EtOAc = 50:1 \rightarrow 5:1) as a colorless oil (0.109 g, 89%, 21% of enol form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.09$ (quint, J = 6.6 Hz, 2 H), 2.36 (s, 3 H), $2.67~({\rm t},\,J=6.6~{\rm Hz},\,2~{\rm H}),\,3.36~({\rm t},\,J=6.6~{\rm Hz},\,2~{\rm H}),\,3.75~({\rm s},\,3$ H), 4.69 (s, 1 H), 7.17-7.26 (m, 4 H), 13.06 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_{\rm C} = 21.3$ (CH₃), 26.7, 33.0, 39.6, 52.8, 64.8, 104.4 (enol), 129.4, 129.5, 129.9, 138.5, 169.2, 137.8 (enol), 175.6 (enol), 202.9. IR (neat, cm⁻¹): $\tilde{\nu} = 3132$ (w), 3090 (w), 3026 (w), 3004 (w), 2954 (m), 2923 (m), 2862 (w), 1749 (s), 1717 (s), 1644 (w), 1609 (w), 1513 (w), 1437 (m), 1409 (w), 1338 (w), 1307 (w), 1262 (m), 1224 (m), 1205 (s), 1185 (m), 1159 (s), 1114 (w), 1076 (m), 1045 (w), 1020 (w), 814 (w). MS (EI, 70 eV): m/z = 313 (M⁺, 4), 280 (3), 253 (2), 232 (29), 201 (4), 174 (11), 164 (100), 148 (36), 132 (72). The exact molecular mass $m/z = 312.0361 \pm 2$ ppm [M⁺] for $C_{14}H_{17}O_3Br$ was confirmed by HRMS (EI, 70 eV).

6-Bromo-2-(4-chlorophenyl)-3-oxohexanoic Acid Methyl Ester (3h). Starting with 2h (0.100 g, 0.40 mmol) and BBr₃ (0.398 g, 1.58 mmol) in CH₂Cl₂ (10 mL), 3h was isolated chromatography (silica gel, n-hexane/EtOAc = $50:1 \rightarrow 10:1$) as a slightly yellow oil (0.102 g, 77%, 22% of enol form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.07-2.13$ (m, 2 H), 2.64-2.73 (m, 2 H), 3.32 - 3.43 (m, 2 H), 3.76 (s, 3 H), 4.72 (s, 1 H), 7.27 -7.38 (m, 4 H), 13.02 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_c = 26.5, 32.9, 39.9, 53.0, 64.2, 103.7 \text{ (enol.)}, 129.3,$ 130.9, 133.5, 134.7, 168.6, 172.9 (enol), 175.4 (enol), 202.1. IR (neat, cm⁻¹): $\tilde{v} = 3003$ (w), 2954 (m), 2921 (w), 1751 (s), 1717 (s), 1645 (s), 1610 (s), 1491 (s), 1438 (s), 1408 (s), 1491 (s), 1438 (s), 1335 (s), 1298 (s), 1272 (s), 1209 (s), 1162 (s), 1091 (s), 1047 (m), 1013 (s), 984 (w), 830 (s), 764 (w), 748 (w), 559 (w), 503 (w). MS (EI, 70 eV): m/z = 334 (M⁺, 6), 302 (21), 221 (5), 184 (89), 148 (100). The exact molecular mass $m/z = 331.9815 \pm 2$ ppm [M⁺] for C₁₃H₁₄O₃BrCl was confirmed by HRMS (EI, 70 eV).

6-Bromo-2-(4-hydroxyphenyl)-3-oxohexanoic Methyl Ester (3i). Starting with 2i (0.130 g, 0.5 mmol) and BBr₃ (0.525 g, 2.1 mmol) in CH₂Cl₂ (5 mL), 3i and 2i' were isolated after chromatography (silica gel, n-hexane/EtOAc = $30:1 \rightarrow 1:1$) as brownish and white solids (3i: 0.118 g, 72%, 10% of enol-form; 2i': 0.027 g, 22%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.09$ (quint, J = 6.9 Hz, 2 H), 2.68 (t, J = 6.9 Hz, 2 H), 3.35 (t, J = 6.9 Hz, 2 H), 3.76 (s, 3 H, OCH₃), 4.70 (s, 1 H), 5.65 (br, 1 H), 6.83 (d, J = 8.7 Hz, 2 H), 7.20 (d, J = 8.7Hz, 2 H), 13.03 (br, 1 H, OH of enol). ¹³C NMR (CDCl₃, 75 MHz): $\delta_C = 26.3$, 32.7, 39.4, 52.8, 63.9, 103.9 (enol), 116.0, $123.7, 130.6, 156.1, 169.6, 175.0 \text{ (enol)}, 203.6. \text{ IR (KBr, cm}^{-1})$: $\tilde{\nu} = 3396 \text{ (m)}, 3184 \text{ (w)}, 2957 \text{ (w)}, 1735 \text{ (s)}, 1703 \text{ (s)}, 1614 \text{ (w)},$ 1594 (w), 1516 (s), 1439 (m), 1359 (m), 1335 (w), 1302 (w), 1274 (m), 1249 (m), 1213 (s), 1161 (m), 1095 (w), 991 (w), 832 (w), 557 (w), 528 (w). MS (EI, 70 eV): m/z = 315 (M⁺, 1), 284 (14), 234 (5), 203 (1), 175 (5), 165 (85), 150 (29), 118 (3), 109 (100), 106 (55). Anal. Calcd for C₁₃H₁₅O₄Br (315.163): C, 49.54; H, 4.80. Found: C, 49.63; H, 5.03.

(Dihydrofuran-2-ylidene)(4-hydroxyphenyl)acetic Acid Methyl Ester (2i'). $^1{\rm H}$ NMR (CDCl_3, 300 MHz): $\delta=1.97$ (quint, J=7.2 Hz, 2 H), 2.52 (t, J=7.5 Hz, 2 H), 3.68 (s, 3 H), 4.47 (t, J=6.9 Hz, 2 H), 5.43 (broad s, 1 H), 6.79 (dd, J=6.6, 2.1 Hz, 2 H), 7.05 (dd, J=6.6, 2.1 Hz, 2 H). $^{13}{\rm C}$ NMR (CDCl_3, 75 MHz): $\delta_{\rm C}=13.4$, 32.1, 51.4, 74.2, 103.0, 115.2, 129.6, 131.8, 154.7, 167.2, 170.4. IR (KBr, cm $^{-1}$): $\tilde{\nu}=3404$ (br), 2955 (w), 2914 (w), 1684 (s), 1616 (s), 1515 (s), 1435 (m), 1338 (w), 1327 (w), 1305 (m), 1276 (s), 1243 (m), 1193 (s), 1170 (s), 1098 (w), 1051 (s), 1025 (w), 999 (m), 846 (m), 596 (w). MS (EI, 70 eV): m/z=234 (M $^+$, 100), 202 (56), 174 (18). The exact molecular mass $m/z=234.0892\pm2$ ppm [M $^+$] for $\rm C_{13}H_{14}O_4$ was confirmed by HRMS (EI, 70 eV).

6-Bromo-3-oxo-2,4-diphenylhexanoic Acid Methyl Es**ter** (3j). Starting with 2j (0.060 g, 0.20 mmol) and BBr₃ (0.204 g, 0.82 mmol) in CH₂Cl₂ (5 mL), 3j was isolated after chromatography (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 3:1$) as a slightly yellow oil (0.072 g, 96%, 33% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.10-2.33$ (m, 1 H), 2.42-2.65 (m, 1 H), 3.03-3.20 (m, 1 H), 3.21-3.43 (m, 1 H), 3.65 (s, 3 H), 3.93 (t, $J = 7.2 \; \mathrm{Hz}, \, 1 \; \mathrm{H}$), $4.69 \; (\mathrm{s}, \, 1 \; \mathrm{H}), \, 7.00 - 7.38 \; (\mathrm{m}, \, 10 \; \mathrm{H}), \, 13.22$ (s, 1 H). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 29.9$, 31.9, 34.9, 48.9, 55.9, 127.2, 128.0, 128.8, 128.9, 129.5, 129.7, 134.1, 137.3, 167.6, 206.8. IR (KBr, cm⁻¹): $\tilde{\nu} = 2956$ (w), 2922 (w), 1750 (s), 1716 (s), 1494 (w), 1453 (m), 1436 (m), 1346 (w), 1308 (m), 1263 (m), 1206 (m), 1158 (m), 1114 (w), 1098 (w), 1078 (w), 1027 (w), 1013 (w), 704 (s). MS (EI, 70 eV): m/z = 375 (M⁺, 1), 344 (2), 294 (2), 236 (1), 235 (2), 225 (12), 197 (64), 177 (4), 149 (31), 117 (63), 91 (100), 77 (13). HRMS (ESI): calcd. for $C_{19}H_{20}O_3Br$ ([M + 1]⁺) = 375.05959; found = 375.05880.

6-Bromo-3-oxo-4-ethylhexanoic Acid Ethyl Ester (3k). Starting with **2k** (0.100 g, 0.54 mmol) and BBr₃ (0.549 g, 2.17 mmol) in CH₂Cl₂ (10 mL), 3k was isolated after chromatography (silica gel, *n*-hexane/EtOAc = $100:1 \rightarrow 1:1$) as a yellow oil (0.118 g, 83%, 29% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.92$ (t, J = 7.2 Hz, 3 H), 1.29 (t, J = 7.2 Hz, 3 H), 1.51-1.63 (m, 1 H), 2.84-2.91 (m, 1 H), 3.29-3.38 (m, 1 H), 3.40-3.48 (m, 1 H), 3.51 (d, J = 3.3 Hz, 2 H), 4.20 (q, J = 7.2Hz, 2 H), 5.06 (s, 1 H, enol), 12.09 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 11.2$, 14.2, 24.1, 31.5, 32.8, 48.9, 51.6, 61.5, 91.0 (enol), 167.0, 172.7 (enol), 178.6 (enol), 205.3. IR (neat, cm⁻¹): $\tilde{\nu} = 2968$ (m), 2935 (w), 1745 (s), 1712 (s), 1648 (m), 1631 (m), 1461 (w), 1424 (w), 1386 (w), 1369 (w), 1352 (w), 1311 (m), 1234 (s), 1176 (w), 1153 (m), 1117 (w), 1095 (w), 1031 (m). MS (EI, 70 eV): m/z = 265 (M⁺, 2), 236 (1), 186 (5), 179 (8), 158 (59), 143 (7), 130 (5), 114 (55), 97 (5), 87 (42), 70 (100), 55 (30). The exact molecular mass $m/z = 264.0361 \pm 1000$ 2 ppm $[M^+]$ for $C_{10}H_{17}O_3Br$ was confirmed by HRMS (EI, 70 eV).

4-(2-Bromoethyl)-3-oxoheptanoic Acid Ethyl Ester (31). Starting with **21** (0.100 g, 0.50 mmol) and BBr₃ (0.505 g, 2.02 mmol) in CH₂Cl₂ (5 mL), 31 was isolated after chromatography (silica gel, n-hex/EtOAc = 50:1 \rightarrow 20:1) as a slightly yellow oil (0.135 g, 96%, 12% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.92$ (t, J = 6.9 Hz, 3 H), 1.28 (t, J = 7.2 Hz, 3 H), 1.22-1.49 (m, 3 H), 1.58-1.68 (m, 1 H), 1.86-1.95 (m, 1 H), 2.24-2.33 (m, 1 H), 2.90-2.94 (m, 1 H), 3.29-3.37 (m, 1 H), 3.40-3.48 (m, 1 H), 3.51 (s, 2 H), 4.19 (q, J = 7.2 Hz, 2 H), 5.05 (s, 1)1 H, enol), 12.10 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 14.18, 14.24, 20.2, 31.5, 33.3, 35.5, 43.9, 49.0, 61.6,$ 90.9 (enol), 167.0, 172.7 (enol), 178.9 (enol), 205.4. IR (neat, cm⁻¹): $\tilde{\nu} = 2962$ (s), 2935 (s), 2873 (w), 1746 (s), 1712 (s), 1650 (s), 1629 (s), 1463 (m), 1446 (m), 1424 (m), 1382 (w), 1371 (w), 1311 (m), 1235 (s), 1153 (m), 1097 (w), 1034 (m), 842 (w), 808 (w). MS (EI, 70 eV): m/z = 279 (M⁺, 3), 264 (3), 250 (3), 236 $(30),\,200\,(5),\,172\,(25),\,165\,(7),\,163\,(11),\,157\,(64),\,143\,(75),\,130$ (10), 115 (100), 101 (11), 87 (79). Anal. Calcd for C₁₁H₁₉O₃Br (279.173): C, 47.33; H, 6.86. Found: C, 47.67; H, 6.54

4-(2-Bromoethyl)-7-chloro-3-oxoheptanoic Acid Methyl Ester (3m). Starting with **2m** (0.070 g, 0.32 mmol) and BBr₃ (0.921 g, 1.28 mmol) in CH₂Cl₂ (5 mL), **3m** was isolated after chromatography (silica gel, n-hex/EtOAc = $50:1 \rightarrow 5:1$)

as a slightly yellow oil (0.082 g, 86%, 33% of enol-form). 1H NMR (CDCl₃, 300 MHz): $\delta=1.62-198$ (m, 4 H), 2.16–2.33 (m, 2 H), 2.92–3.01 (m, 1 H), 3.31–3.54 (m, 4 H), 3.56 (s, 2 H), 3.75 (s, 3 H), 5.10 (s, 1 H, enol), 12.03 (s, 1 H, OH of enol). ^{13}C NMR (CDCl₃, 150 MHz): $\delta_C=27.9,$ 29.6, 31.2, 33.2, 35.36, 43.5, 44.6, 48.6, 52.6, 91.1 (enol), 167.4, 172.9 (enol), 177.8 (enol), 204.7. IR (neat, cm $^{-1}$): $\tilde{\nu}=2997$ (w), 2955 (s), 2868 (w), 1750 (s), 1711 (s), 1655 (s), 1629 (s), 1442 (s), 1404 (m), 1370 (m), 1317 (s), 1238 (s), 1155 (s), 1097 (m), 1050 (m), 1013 (m), 840 (w), 809 (m), 735 (w), 651 (m). MS (EI, 70 eV): m/z=299 (M $^+$, 2), 164 (2), 227 (s), 222 (4), 192 (18), 142 (15), 129 (18), 114 (3), 101 (100). The exact molecular mass $m/z=297.9971\pm2$ ppm [M $^+$] for C $_{10}H_{16}O_{3}BrCl$ was confirmed by HRMS (EI, 70 eV).

3-(2-Bromoethyl)-2-oxocyclododecanecarboxylic Acid Ethyl cester (3n). Starting with 2n (0.100 g, 0.36 mmol) and BBr₃ (0.357 g, 1.43 mmol) in CH₂Cl₂ (10 mL), **3n** was isolated after chromatography (silica gel, n-hex/EtOAc = $50:1 \rightarrow 15:1$) as a yellow solid (0.113 g, 87%, 5% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.95-1.40$ (m, 15 H), 1.41-1.69 (m, 3 H), 1.72-1.85 (m, 1 H), 1.88-2.05 (m, 1 H), 2.09-2.18 (m, 1 H), 2.28-2.52 (m 2 H), 3.21-3.42 (m, 2 H), 3.44-3.49 (m, 1 H), 4.03 (dd, J = 11.7, 3.0 Hz, 1 H), 4.17 (q, J = 7.2 Hz, 2 H), 13.01 (s, 1 H, OH of enol). 13 C NMR (CDCl₃, 150 MHz): $\delta_{\rm C} =$ 14.3, 20.4, 22.1, 22.2, 23.4, 23.5, 26.4, 26.6, 26.9, 27.9, 31.5, 32.4, 49.7, 53.7, 61.8, 102.0 (enol), 169.9, 174.5 (enol), 177.4 (enol), 206.7. IR (neat, cm⁻¹): $\tilde{\nu} = 2929$ (s), 2855 (s), 1735 (s), 1701 (s), 1637 (w), 1629 (w), 1619 (w), 1465 (s), 1342 (s), 1371 (w), 1340 (w), 1235 (s), 1185 (m), 1156 (m), 1147 (m), 1125 (m), 1096 (m), 1064 (m), 1027 (m), 982 (w), 944 (w), 854 (w), 836 (w), 746 (w), 728 (w), 702 (w). MS (EI, 70 eV): m/z = 361 $(M^+, 53), 316 (77), 281 (68), 255 (60), 238 (80), 236 (100), 208$ (57), 165 (61), 107 (59). The exact molecular mass m/z = 360.1300 ± 2 ppm [M⁺] for $C_{17}H_{29}O_3Br$ was confirmed by HRMS (EI, 70 eV).

6-Bromo-3-oxoheptanoic Acid Methyl Ester (30). Starting with **2o** (0.060 g, 0.38 mmol) and BBr₃ (0.385 g, 1.54 mmol) in CH₂Cl₂ (5 mL), **30** was isolated after chromatography (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 3:1$) as a slightly yellow oil (0.072 g, 80%, 8% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.74 \, (dd, J = 6.6, 1.5 \, Hz, 3 \, H), 1.96 - 2.04 \, (m, 1 \, H), 2.11 -$ 2.20 (m, 1 H), 2.80 (dt, J = 1.5, 6.6 Hz, 2 H), 3.49 (s, 2 H), 3.75(s, 3 H), 4.12-4.19 (m, 1 H), 5.07 (s, 1 H, enol), 12.04 (s, 1 H, OH of enol). 13 C NMR (CDCl₃, 150 MHz): $\delta_{\rm C} = 26.7, 34.4, 41.3,$ 49.2, 50.8, 52.6, 89.8 (enol), 167.6, 174.2 (enol), 178.1 (enol), 201.7. IR (neat, cm⁻¹): $\tilde{\nu} = 2957$ (m), 2926 (m), 1750 (s), 1717 (s), 1652 (m), 1633 (m), 1439 (s), 1408 (s), 1378 (s), 1321 (s), 1257 (s), 1233 (s), 1194 (s), 1172 (s), 1152 (s), 1075 (m), 1012 (m), 963 (w), 887 (w), 840 (w), 655 (w), 587 (w), 530 (w). MS (EI, 70 eV): m/z = 221 (M⁺-Me, 16), 163 (18), 148 (12), 135 (14), 121 (21), 108 (21), 93 (17), 80 (100). HRMS (ESI): calcd for $C_8H_{14}O_3Br$ ([M + 1]+) 237.01187, found 237.01187.

6-Bromo-3-oxooctanoic Acid Methyl Ester (3p). Starting with $\mathbf{2p}$ (0.150 g, 0.88 mmol) and BBr_3 (0.883 g, 3.53 mmol) in CH₂Cl₂ (15 mL), **3p** was isolated after chromatography (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 1:1$) as a slightly yellow oil (0.200 g, 91%, 8% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.05$ (t, J = 7.2 Hz, 3 H), 1.82 - 1.92 (m, 2 H), 1.94 - 1.922.06 (m, 1 H), 2.15-2.24 (m, 1 H), 2.80-2.87 (m, 2 H), 3.49 (s, 2 H), 3.75 (s, 3 H), 3.95-4.07 (m, 1 H), 5.07 (s, 1 H, enol), 12.08 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 75 MHz): $\delta_{\rm C}$ = 10.5, 27.7, 32.4, 41.1, 49.1, 52.3, 59.1, 89.8 (enol), 166.5, 201.3. IR (neat, cm⁻¹): $\tilde{v} = 2971$ (m), 2940 (m), 2881 (w), 1750 (s), 1717 (s), 1646 (w), 1639 (w), 1456 (m), 1439 (m), 1409 (m), 1378 (m), 1322 (s), 1262 (s), 1247 (s), 1216 (s), 1171 (m), 1116 (w), 1077 (w), 1051 (w), 1015 (w). MS (EI, 70 eV): m/z = 249 $(M^+, 6)$, 219 (3), 192 (37), 177 (21), 171 (3), 157 (26), 139 (25), 116 (9), 110 (7), 102 (95), 97 (86), 87 (20), 84 (24), 70 (100), 55 (95). Anal. Calcd for C₉H₁₅O₃Br (251.120): C, 43.05; H, 6.02. Found: C, 43.89; H, 6.54.

6-Bromo-3-oxodecanoic Acid Methyl Ester (3q). Starting with 2q (0.150 g, 0.76 mmol) and BBr₃ (0.766 g, 3.03 mmol)

in CH₂Cl₂ (15 mL), 3q was isolated after chromatography (silica gel, *n*-hexane/EtOAc = $75:1 \rightarrow 1:1$) as a slightly yellow oil (0.158 g, 75%, 24% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.89$ (t, J = 6.9 Hz, 3 H), 1.29–1.57 (m, 4 H), 1.77– 1.88 (m, 2 H, 1.91-2.04 (m, 1 H), 2.16-2.27 (m, 1 H), 2.83 (t, J = 7.2 Hz, 2 H, 3.48 (s, 2 H), 3.75 (s, 3 H), 3.99-4.08 (m, 1)H), 5.04 (s, 1 H, enol), 12.06 (s, 1 H, OH of enol). $^{13}\mathrm{C}$ NMR (CDCl₃, 150 MHz): $\delta_C = 14.1, 22.2, 29.8, 32.5, 39.3, 41.2, 49.3,$ 52.5, 57.6, 89.7 (enol), 167.6, 173.8 (enol), 177.8 (enol), 201.8. IR (neat, cm⁻¹): $\tilde{v} = 2958$ (m), 2933 (m), 2868 (w, C-H), 1750 (s, O=C-O), 1719 (s, C=O), 1654 (w), 1633 (w), 1438 (m), 1409 (m), 1377 (w), 1361 (w), 1320 (m), 1241 (m), 1197 (m), 1173 (m), 1151 (m), 1101 (w), 1081 (w), 1026 (w). MS (EI, 70 eV): $m/z = 279 \, (\mathrm{M}^+, 2), 205 \, (5), 199 \, (35), 167 \, (7), 139 \, (2), 129 \, (15),$ 116 (100), 101 49), 85 (34), 60 (45). Anal. Calcd for C₁₁H₁₉O₃-Br (279.173): C, 47.33; H, 6.86. Found: C, 47.25; H, 7.34.

6,7-Dibromo-3-oxoheptanoic Acid Ethyl Ester (3r). Starting with **2r** (0.100 g, 0.40 mmol) and BBr₃ (0.402 g, 1.61 mmol) in CH₂Cl₂ (10 mL), 3r was isolated after chromatography (silica gel, n-hex/EtOAc = $50:1 \rightarrow 10:1$) as a slightly yellow oil (0.123 g, 93%, 7% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.29$ (t, J = 7.2 Hz, 3 H), 1.93 - 2.05 (m, 1 H), 2.51 - $2.62 \text{ (m, 1 H)}, 2.80-2.86 \text{ (m, 2 H)}, 3.48 \text{ (s, 2 H)}, 3.62 \text{ (dd, } J = 0.62 \text{ (m, 1 H)}, 0.80-2.86 \text{ (m, 2 H)}, 0.48 \text{ (s, 2 H)}, 0.62 \text{ (dd, } J = 0.62 \text{ (m, 2 H)}, 0.80-2.86 \text{ (m$ 7.5, 6.6 Hz, 1 H), 3.86 (dd, J = 7.5, 4.5 Hz, 1 H), 4.21 (q, J =7.2 Hz, 2 H), 4.19-4.27 (m, 1 H), 5.06 (s, 1 H, enol), 12.10 (s, 1 H, OH of enol). $^{13}\mathrm{C}$ NMR (CDCl_3, 150 MHz): $\delta_\mathrm{C} = 14.3, 30.2,$ 36.2, 40.5, 49.5, 51.9, 61.9, 90.2 (enol), 167.1, 201.4. IR (neat, cm $^{-1}$): $\tilde{v} = 2983$ (w), 2935 (w), 1743 (s), 1716 (s), 1651 (w), 1467 (w), 1436 (w), 1410 (m), 1370 (m), 1318 (s), 1258 (s), 1239 (s), 1177 (m), 1157 (m), 1095 (w), 1029 (m), 566 (w). MS (EI, 70 eV): $m/z = 330 \, (M^+, 1), 251 \, (11), 243 \, (20), 215 \, (3), 205 \, (4),$ 169 (100), 163 (16), 143 (5), 135 (26), 115 (28), 97 (5), 95 (32), 87 (13). Anal. Calcd for C₉H₁₄O₃Br₂ (330.016): C, 32.76; H, 4.28. Found: C, 33.36; H, 4.01.

6-Bromo-7-chloro-3-oxoheptanoic Acid Methyl Ester (3s). Starting with 2s (0.100 g, $\bar{0}.52$ mmol) and BBr₃ (0.531 g, 2.10 mmol) in CH₂Cl₂ (10 mL), 3s was isolated after chromatography (silica gel, *n*-hexane/EtOAc = $75:1 \rightarrow 1:1$) as a slightly yellow oil (0.113 g, 80%, 8% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.94 - 2.06$ (m, 1 H), 2.44 - 2.57 (m, 1 H), 2.80 - 2.062.86 (m, 2 H), 3.50 (s, 2 H), 3.74 (dd, J = 11.4, 8.4 Hz, 1 H),3.76 (s, 3 H), 3.93 (dd, J = 11.4, 4.8 Hz, 1 H), 4.13-4.22 (m, 1)H), 5.07 (s, 1 H, enol), 12.07 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 29.3, 40.4, 48.3, 49.1, 52.3, 52.6, 167.5,$ 201.2. IR (neat, cm⁻¹): $\tilde{\nu} = 2955$ (w), 2929 (w, C-H), 1747 (s, O=C-O), 1717 (s, C=O), 1653 (w), 1633 (w), 1438 (s), 1408 (m), 1375 (w), 1322 (m), 1260 (m), 1165 (m), 1125 (w), 1087 (w), 1013 (w). MS (EI, 70 eV): m/z = 271 (M⁺, 2), 241 (7), 199 (35), 191 (9), 171 (5), 155 (100), 129 (6), 117 (18), 101 (71), 84 (6). The exact molecular mass $m/z = 269.9658 \pm 2$ ppm [M⁺] for C₈H₁₂O₃BrCl was confirmed by HRMS (EI, 70 eV).

5-Bromomethyl-3-oxoheptanoic Acid Methyl Ester (3t). Starting with 2t (0.100 g, 0.59 mmol) and BBr₃ (0.589 g, 2.35 mmol) in CH₂Cl₂ (10 mL), 3t was isolated after chromatography (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 5:1$) as a slightly yellow oil (0.122 g, 82%, 12% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 0.91$ (t, J = 7.2 Hz, 3 H), 1.03–1.09 (m, 2 H), 2.18-2.21 (m, 1 H), 2.56 (dd, J = 18, 5.4 Hz, 1 H), 2.78 (dd, J= 18, 7.2 Hz, 1 H), 3.47 (s, 2 H), 3.50-3.58 (m, 2 H), 3.75 (s, 3 H), 5.07 (s, 1 H, enol), 12.05 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_c = 11.2, 25.5, 36.4, 38.6, 46.9, 49.6, 52.6,$ 90.8 (enol), 167.6, 201.7. IR (neat, cm⁻¹): $\tilde{\nu} = 2966$ (m), 2936 (w), 2879 (w), 1749 (s), 1717 (s), 1654 (w), 1633 (w), 1454 (m), 1439 (m), 1407 (m), 1377 (w), 1322 (m), 1245 (m), 1156 (m), 1128 (w), 1076 (w), 1012 (w). MS (EI, 70 eV): m/z = 249 (M⁺) 10), 219 (5), 192 (43), 177 (27), 171 (3), 157 (38), 139 (30), 116 (3), 110 (8), 102 (91), 87 (89), 87 (21), 84 (20), 70 (97), 55 (100). Anal. Calcd for $C_9H_{15}O_3Br$ (251.120): C, 43.05; H, 6.02. Found: C, 43.11; H, 6.08.

6-Bromo-3-oxo-5-phenylhexanoic Acid Ethyl Ester (3u). Starting with 2u (0.200 g, 0.86 mmol) and BBr₃ (0.862 g, 3.44 mmol) in CH₂Cl₂ (15 mL), 3u was isolated after chromatog-

raphy (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 1:1$) as a slightly yellow oil (0.223 g, 83%, 9% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.25$ (t, J = 7.2 Hz, 3 H), 3.01 (dd, J = 17.7, 7.2 Hz, 1 H), 3.24 (dd, J = 17.7, 5.7 Hz, 1 H), 3.88 (s, 2 H), 3.56-3.66 (m, 3 H), 4.16 (q, J=7.2 Hz, 2 H), 4.93 (s, 1 H, enol), 7.19-7.32 (m, 5 H), 12.09 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 14.3, 38.1, 42.4, 46.9, 49.9, 61.6, 91.9$ (enol), 127.6, 127.9, 128.9, 141.2, 166.9, 173.8 (enol), 176.7 (enol), 200.7. IR (neat, cm⁻¹): $\tilde{\nu} = 3105$ (w), 3086 (w), 3063 (w), 3030 (w), 2983 (m), 2934 (w), 2907 (w, C-H), 1743 (s), 1717 (s), 1650 (m), 1606 (w), 1495 (w), 1470 (w), 1451 (m), 1408 (m), 1369 (m), 1318 (s), 1247 (s), 1197 (m), 1157 (m), 1096 (m), 1075 (m), 1030 (s), 942 (w), 851 (w), 760 (m), 702 (s), 652 (w), 542 (w). MS (EI, 70 eV): m/z = 313 (M⁺, 1), 267 (1), 232 (88), 225 (9), 203 (6), 199 (2), 187 (6), 185 (26), 158 (15), 145 (100), 141 (4), 129 (2), 114 (69), 103 (74), 87 (11). Anal. Calcd for C₁₄H₁₇O₃Br (313.191): C, 53.69; H, 5.47. Found: C, 54.10; H,

6-Bromo-1-phenylhexane-1,3-dione (3v). Starting with **2v** (0.100 g, 0.53 mmol) and BBr₃ (0.532 g, 2.13 mmol) in CH₂-Cl₂ (10 mL), **3v** was isolated after chromatography (silica gel, *n*-hexane/EtOAc = 100:1 → 50:1) as a yellow solid (0.140 g, 98%, 84% of enol-form, data are listed for enol). ¹H NMR (CDCl₃, 300 MHz): δ = 2.25 (quint, J = 6.9 Hz, 2 H), 2.64 (t, J = 7.2 Hz, 2 H), 3.50 (t, J = 6.6 Hz, 2 H), 6.21 (s, 1 H), 7.43 − 7.56 (m, 3 H), 7.87 − 7.90 (m, 2 H), 16.02 (s, 1 H, OH). ¹³C NMR (CDCl₃, 150 MHz): δ _C = 28.4, 33.2, 37.6, 96.7, 127.2, 128.8, 132.6, 135.0, 183.1, 195.7. IR (neat, cm⁻¹): \tilde{v} = 3439 (br), 1617 (s), 1573 (s), 1490 (m), 1458 (m), 1433 (m), 1359 (w), 1298 (w), 1264 (m), 767 (m), 694 (m). MS (EI, 70 eV): m/z = 269 (M⁺, 15), 189 (3), 175 (5), 162 (53), 147 (100), 120 (3), 105 (58), 77 (48). The exact molecular mass m/z = 268.0099 ± 2 ppm [M⁺] for C₁₂H₁₃O₂Br was confirmed by HRMS (EI, 70 eV).

8-Bromo-3-oxooct-6-enoic Acid Ethyl Ester (3w). Starting with **2w** (0.150 g, 0.82 mmol) and BBr₃ (0.833 g, 3.29 mmol) in CH₂Cl₂ (15 mL), **3w** was isolated after chromatography (silica gel, *n*-hexane/EtOAc = $50:1 \rightarrow 1:1$) as a slightly yellow oil (0.211 g, 98%, 10% of enol-form). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.29$ (t, J = 7.2 Hz, 2 H), 3.44 (s, 2 H), 3.92 (d, J = 6.0 Hz, 2 H), 4.20 (q, J = 7.2 Hz, 2 H), 4.98 (s, 1 H, enol), 5.72–5.77 (m, 2 H), 12.10 (s, 1 H, OH of enol). ¹³C NMR (CDCl₃, 150 MHz): $\delta_{\rm C} = 14.2$, 25.8, 33.0, 41.9, 49.4, 61.5, 89.9 (enol), 127.6, 134.0, 167.1, 201.7. IR (neat, cm⁻¹): $\tilde{\nu} = 2983$ (w), 2935 (w), 1742 (s), 1716 (s), 1654 (w), 1438 (w), 1410 (w), 1369 (w), 1316 (m), 1247 (m), 1206 (m), 1160 (w), 1113 (w), 1100 (w), 1031 (m), 968 (w). MS (EI, 70 eV): m/z = 263 (M⁺, 3), 218 (1), 183 (100), 176 (7), 148 (2), 137 (42), 135 (1), 130 (9), 119 (6), 114 (19), 110 (6), 95 (61), 87 (23).

1,7-Dibromoheptan-4-one (6a). ^{14a} Starting with **5a** (0.300 g, 1.95 mmol) and BBr₃ (1.950 g, 7.8 mmol) in CH₂Cl₂ (15 mL), **6a** was isolated after chromatography (silica gel, n-hexane/EtOAc = $50:1 \rightarrow 1:1$) as a slightly yellow oil (0.365 g, 73%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.14$ (quint, J = 6.6 Hz, 4 H), 2.65 (t, J = 6.6 Hz, 4 H), 3.45 (t, J = 6.6 Hz, 4 H). ¹³C NMR (CDCl₃, 75 MHz): $\delta_{\rm C} = 26.4$, 33.4, 40.8, 208.1. IR (neat, cm⁻¹): $\tilde{\nu} = 2964$ (m), 2918 (w), 1714 (s), 1436 (m), 1409 (m), 1375 (m), 1296 (w), 1282 (w), 1253 (m), 1197 (w), 1151 (w), 1091 (w), 1046 (w), 558 (w). MS (EI, 70 eV): m/z = 272 (M⁺, 1), 193 (4), 164 (24), 150 (100), 121 (38), 106 (3), 93 (7).

1,7-Dibromooctan-4-one (6b). Starting with **5b** (0.020 g, 0.12 mmol) and BBr₃ (0.119 g, 0.48 mmol) in CH₂Cl₂ (3 mL), **6b** was isolated after chromatography (silica gel, *n*-hexane/ EtOAc = 100:1 → 50:1), as a colorless oil (0.030 g, 88%). ¹H NMR (CDCl₃, 300 MHz): δ = 1.73 (d, J = 6.6 Hz, 3 H), 1.92 – 2.03 (m, 1 H), 2.08 – 2.18 (m, 1 H), 2.18 (quint, J = 6.6 Hz, 2 H), 2.65 (t, J = 6.6 Hz, 2 H), 2.66 – 2.70 (m, 2 H), 3.45 (t, J = 6.6 Hz, 2 H), 4.12 – 4.18 (m, 1 H). ¹³C NMR (CDCl₃, 75 MHz): δ _C = 26.3, 26.6, 33.2, 34.4, 40.8, 40.9, 51.0, 208.4 IR (neat, cm⁻¹): $\tilde{\nu}$ = 2965 (m), 2923 (m), 1714 (s), 1637 (w), 1439 (m), 1410 (m), 1377 (m), 1336 (w), 1283 (w), 1250 (m), 1215 (w), 182 (m), 1142 (m), 1079 (m). MS (EI, 70 eV): m/z = 286 (M⁺, 3), 205 (4), 178 (1), 166 (45), 164 (45), 148 (49), 121 (16), 83

(30), 55 (100). HRMS (ESI): calcd. for $C_8H_{14}OBr$ ([M - Br]⁺) = 205.02281, found = 205.02196.

1,7-Dibromononan-4-one (6c). Starting with **5c** (0.600 g, 3.29 mmol) and BBr₃ (3.300 g, 13.2 mmol) in CH₂Cl₂ (20 mL), **6c** was isolated after chromatography (silica gel, *n*-hex/EtOAc = 100:1 → 50:1) as a colorless oil (0.668 g. 68%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.05$ (t, J = 7.2 Hz, 3 H), 1.40–1.51 (m, 2 H), 1.82-1.92 (m, 2 H), 2.14 (quint, J = 6.6 Hz, 2 H),2.65 (t, J = 6.6 Hz, 2 H), 2.68-2.73 (m, 2 H), 3.45 (t, J = 6.6 Hz)Hz, 2 H), 3.94–4.03 (m, 1 H). ^{13}C NMR (CDCl₃, 75 MHz): δ_C = 12.1, 26.3, 32.2, 32.5, 33.2, 40.7, 40.8, 59.5, 208.4. IR (neat, cm $^{-1}$): $\tilde{v} = 2965$ (s), 2935 (m), 2879 (w), 1717 (s), 1604 (w), 1440 (s), 1410 (w), 1375 (m), 1312 (s), 1261 (m), 1227 (m), 1199 (s), 1154 (s), 1100 (m), 1066 (m), 1034 (m), 942 (w), 795 (w), 750 (w). MS (EI, 70 eV): m/z = 300 (M⁺, 1), 219 (6), 177 (19), 164 (10), 150 (24), 139 (18), 135 (7), 121 (15), 108 (12), 95 (15), 70 (61), 55 (63), 41 (100). HRMS (ESI): calcd for C₉H₁₆OBr $([M - Br]^+) = 219.03846$, found = 219.03779.

1,7-Dibromo-8-chlorooctan-4-one (6d). Starting with 5d (0.200 g, 0.99 mmol) and BBr₃ (0.999 g, 3.95 mmol) in CH₂Cl₂ (15 mL), **6d** was isolated after chromatography (silica gel, n-hexane/EtOAc = 75:1 \rightarrow 1:1) as a colorless oil (0.269 g, 85%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.92-2.05$ (m, 1 H), 2.15 (quint, J = 6.6 Hz, 2 H), 2.42-2.53 (m, 1 H), 2.66 (t, J = 6.6Hz, 2 H), 2.61-2.74 (m, 2 H), 3.46 (t, J = 6.6 Hz, 2 H), 3.73(dd, J = 11.4, 8.7 Hz, 1 H), 3.91 (dd, J = 11.4, 4.8 Hz, 1 H),4.13–4.22 (m, 1 H). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 26.5$, 29.6, 33.4, 40.5, 40.9, 48.4, 52.7, 208.0. IR (neat, cm $^{-1}$): $\tilde{\nu} =$ 2963 (w), 1714 (s), 1435 (m), 1410 (m), 1375 (m), 1287 (w), 1249 (w), 1180 (w), 1158 (w), 1093 (w), 774 (w). MS (EI, 70 eV): m/z = 321 (M⁺, 4), 285 (5), 241 (16), 213 (11), 199 (47), 171 (10), 164 (12), 161 (6), 150 (100), 133 (32), 121 (35). HRMS (ESI): calcd for $C_8H_{13}OBrCl$ ([M - Br]⁺) = 238.98384, found = 238.98305. Anal. Calcd for $C_8H_{13}OBr_2Cl$ (320.451): C, 29.99; H, 4.09. Found: C, 29.95; H, 4.74.

1,7-Dibromo-6-methyloctan-4-one (6e). Starting with trans-5e (0.100 g, 0.55 mmol) and BBr₃ (0.550 g, 2.2 mmol) in CH₂Cl₂ (5 mL), **6e** was isolated after chromatography (silica gel, n-hex/EtOAc = 100:1 \rightarrow 5:1) as a colorless oil (0.144 g, 87%). The product was obtained as a 1:1 mixture of diastereomers. The analytical data are given for a separated sample of a pure diaster eomer. 1 H NMR (CDCl₃, 300 MHz): δ = 0.98 (d, J = 6.6 Hz, 3 H), 1.68 (d, J = 6.9 Hz, 3 H), 2.13(quint, J = 6.6 Hz, 2 H), 2.17-2.24 (m, 1 H), 2.41 (dd, J =17.4, 7.0 Hz, 1 H), 2.62 (t, J = 6.6 Hz, 2 H), 2.69 (dd, J = 17.4, 6.1 Hz, 1 H), 3.45 (t, J = 6.6 Hz, 2 H), 4.31 (dq, J = 6.9, 2.8 HzHz, 1 H). ¹³C NMR (CDCl₃, 150 MHz): $\delta_C = 15.1$, 23.9, 26.5, 33.4, 35.9, 41.5, 48.7, 57.8, 208.6. IR (neat, cm $^{-1}$): $\tilde{v}=2969$ (s), 2927 (m), 1713 (s), 1444 (m), 1409 (m), 1376 (s), 1328 (m), 1273 (m), 1257 (m), 1208 (m), 1143 (m), 1081 (m), 1014 (w), 970 (w), 602 (w). MS (EI, 70 eV): m/z = 300 (M⁺, 6), 219 (7), 177 (21), 164 (66), 150 (65), 148 (74), 121 (26), 97 (31), 85 (7), 70 (100). HRMS (ESI): calcd for $C_9H_{16}OBr$ ([M - Br]⁺) = 219.03846, found = 219.03767.

5-Bromopentan-2-one (7).² Starting with α -acetyl- γ -butyrolactone (0.25 mL, 2.34 mmol) and BBr₃ (2.346 g, 9.37 mmol) in CH₂Cl₂ (15 mL), 7 was isolated after chromatography (silica gel, n-hex/EtOAc = $50:1 \rightarrow 1:1$) as a slightly yellow oil (0.206 g, 59%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 2.12$ (quint, J = 6.6 Hz, 2 H, 2.17 (s, 3 H), 2.65 (t, J = 6.9 Hz, 2 H), 3.45 (t, $J=6.3~\mathrm{Hz},\,2~\mathrm{H}).$ IR (neat, cm $^{-1}$): $\,\tilde{\nu}=2965$ (w), 1715 (s), 1432 (m), 1413 (m), 1367 (m), 1302 (w), 1282 (w), 1248 (m), 1217 (w), 1196 (w), 1179 (m), 1185 (m).

Synthesis of trans-4,5-Dimethyltetrahydro[2,3']bifuranyliden-2'-one (5e). The synthesis was carried out according to a known procedure for related products. 11 Starting with the 1,3-bis-silyl enol ether of α -acetyl- γ -butyrolactone (2.725 g, 10 mmol), cis-2,3-epoxybutane (0.883 g, 12 mmol), and TiCl₄ (2.2 mL, 20 mmol) in CH₂Cl₂ (75 mL), 5e was isolated after chromatograpy (silica gel, n-hexane/EtOAc = $75:1 \rightarrow 5:1$) as a colorless oil (0.669 g, 37%). ¹H NMR (CDCl₃, 300 MHz): $\delta = 1.11$ (d, J = 6.7 Hz, 3 H), 1.38 (d, J = 6.2 Hz, 3 H), 1.92-2.03 (m, 1 H), 2.54 (ddt, J = 18.0, 9.9, 3.0 Hz, 1H), 2.83-2.90 (m, 2 H), 3.55 (ddt, J = 18.0, 7.7, 1.7 Hz, 1 H), $4.01-4.10~(\mathrm{m},~1~\mathrm{H}),~4.30~(\mathrm{t},~J=7.7~\mathrm{Hz},~2~\mathrm{H}).~^{13}\mathrm{C}$ NMR (CDCl₃, 150 MHz): $\delta_C = 16.1$, 18.8, 25.0, 37.7, 39.5, 65.3, 86.7, 92.8, 168.7, 173.3. IR (neat, cm $^{-1}$): $\tilde{\nu} = 2975$ (w), 2930 (w), 1733 (s), 1680 (s), 1454 (w), 1389 (w), 1377 (w), 1306 (m), 1262 (m), 1164 (w), 1057 (s), 1023 (s), 1001 (s), 851 (w). MS (EI, 70 eV): m/z $= 182 (M^+, 46), 167 (100), 152 (6), 139 (4), 124 (10), 111 (52),$ 95 (17). HRMS (ESI): calcd. for $C_{10}H_{15}O_3$ ([M + 1]⁺) = 183.10212, found = 183.10127.

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Supporting Information Available: Experimental procedures, spectroscopic data, copies of NMR spectra, and details of crystal structure analyses. This material is available free of charge via the Internet at http://pubs.acs.org.

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