- (13) Dreisbach, R. R. Physical Properties of Chemical Compounds; Advances in Chemistry Series; American Chemical Society: Washington,
- DC, 1959; Vol. II.

  (14) McGlashan, M. L.; Potter, D. J. B. *Proc. R. Soc. London, Ser. A*1962, 267, 478–484.

  (15) Guggenheim, E. A.; Wormald, C. J. *J. Chem. Phys.* 1965, 42, 3775–3780.
- (16) Oweimreen, G. A.; Hassan, M. *Mol. Cryst. Liq. Cryst.* **1983**, *100*, 351–371.
- (17) Oweimreen, G. A.; Lin, G. C.; Martire, D. E. J. Phys. Chem. 1979, 83,
- (18) Martire, D. E.; Riedl, P. J. Phys. Chem. 1968, 72, 3478-3488.
  (19) Liao, H. L.; Martire, D. E. Anal. Chem. 1972, 44, 498-502.
  (20) Ashworth, A. J. J. Chem. Soc., Faraday Trans. 1 1973, 69, 481-462.
- 459-466. (21) Ashworth, A. J.; Hooker, D. M. J. Chem. Soc., Faraday Trans. 1 1976, 72, 2240-2246.

Received for review April 3, 1989. Accepted January 12, 1990. We gratefully acknowledge financial support for this work from King Fahd University of Petroleum and Minerals.

## Correction

Simple Apparatus for Vapor-Liquid Equilibrium Measurements with Data for the Binary Systems of Carbon Dioxide with n-Butane and Isobutane

Lloyd A. Weber (J. Chem. Eng. Data 1989, 34, 171-175). Data from Table III were inadvertently omitted from the published article. The additional data are as follows:

Table III. Experimental Results fo the CO<sub>2</sub> + n-C<sub>4</sub>H<sub>10</sub> System (Component 1, CO<sub>2</sub>)

P/bar	$x_1/\%$	$y_1/\%$
	T = 344.26  K	
76.98	63.2	78.4
79.44	66.4	77.4
80.46	68.1	75.8
81.1	69.4	74.6