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Conformational Flexibility of *meso*-Tetrakis(perfluoroalkyl)porphyrins: Energetics of Ruffling versus Saddling

Tebikie Wondimagegn and Abhik Ghosh*

Institute of Chemistry, University of Tromsø, N-9037 Tromsø, Norway Received: January 5, 2000; In Final Form: February 28, 2000

To better understand the existence of multiple nonplanar conformations of certain porphyrins, we have carried out geometry optimizations with nonlocal density functional theory and triple- ζ plus polarization basis sets for the ruffled and saddled conformations of four metalloporphyrins, M(TCF₃P) and M(TC₂F₅P), where M is Ni and Zn, TCF₃P is *meso*-tetrakis(trifluoromethyl)porphyrin, and TC₂F₅P is *meso*-tetrakis(pentafluoroethyl)porphyrin. Short F···H_{β} contacts lead to strongly nonplanar geometries for every optimized structure. Ruffling is significantly favored over saddling for Ni(II) complexes, whereas the ruffled and saddled geometries are essentially equienergetic for Zn(II) complexes, consistent with experimental observation of both predominantly ruffled and saddled conformations for *meso*-tetrakis(perfluoroalkyl)porphyrins. Contrastingly, for dodecaphenylporphyrins, which generally exhibit saddled conformations, ruffling and saddling appear to be most competitive for Ni(II) complexes. Despite shorter F···H_{β} contacts in the TC₂F₅P complexes, the TCF₃P and TC₂F₅P complexes exhibit comparable nonplanar distortions for a particular metal.

Introduction

In a paper entitled "Conformational Flexibility in Dodecasubstituted Porphyrins", 1 Smith and co-workers demonstrated that dodecaphenylporphyrins (DPPs) exhibit a variety of nonplanar conformations such as ruffled, saddled, and waved in the solid state. Although not recognized as such, the same phenomenon of conformational flexibility has also been found for meso-tetrakis(perfluoroalkyl)porphyrins. 2,3 Thus, crystallographic analyses revealed a ruffled conformation for free-base² meso-tetrakis(heptafluoropropyl)-porphyrin and a predominantly saddled conformation, with a mild degree of ruffling, for the cobalt(II)³ complex of the same porphyrin. On the basis of recent successes with first-principles quantum chemical studies of nonplanar porphyrins^{4,5} and hydroporphyrins,⁶ we present here an initial density functional theory-based study of the energetics of ruffling versus saddling distortions in meso-tetrakis(perfluoroalkyl)porphyrins.⁷

Methods

For the calculations, we used the ADF⁸ program system, PW91 functional, Slater-type triple- ζ plus polarization basis sets, a fine mesh for numerical evaluation of electron repulsion integrals, and tight convergence criteria for atomic forces and displacements in the geometry optimizations. Both ruffled (D_{2d}) and saddled (C_{2v}) conformations were optimized for M(TCF₃P) and M(TC₂F₅P), where M is Ni and Zn, TCF₃P is *meso*-tetrakis(trifluoro-methyl)porphyrin, and TC₂F₅P is *meso*-tetrakis(pentafluoroethyl)porphyrin. At this point, our computer resources have not permitted frequency analyses on the optimized structures. Indeed, in each case, the "pure" ruffled or saddled geometry may actually be a transition state, with some lower-symmetry conformation being the global minimum. The optimized geometries are shown in Figure 1.

Results

- (a) Energetics of Ruffling Versus Saddling. For Ni(TCF₃P) and Ni(TC₂F₅P), the ruffled conformation was found to be favored over the saddled form by the relatively significant margins of 10.73 and 7.93 kcal/mol, respectively. In contrast, the ruffled conformations of Zn(TCF₃P) and Zn(TC₂F₅P) are favored over the saddled conformations by only 2.20 and 3.23 kcal/mol, respectively. In general, a close correlation exists between ruffling distortions and short M–N_{porphyrin} distances. ^{5,9} It is thus reasonable that the small low-spin Ni(II) ion favors the ruffled conformation. However, the optimized data on the zinc porphyrins suggest that except for the smallest metal ions, the energies of the ruffled and saddled conformations are very delicately balanced.
- (b) Description of Optimized Geometries. For the ruffled conformations, the ruffling torsion angles⁷ for Ni(TCF₃P) and Ni(TC₂F₅P) are 45.6 and 44.7°, respectively. These values are significantly higher than the ruffling angles of 27.2 and 25.7° for Zn(TCF₃P) and Zn(TC₂F₅P), respectively. These ruffling angles are consistent with metal-nitrogen distances of 192, 191, 204, and 205 pm for Ni(TCF₃P), Ni(TC₂F₅P), Zn(TCF₃P), and Zn(TC₂F₅P), respectively. Although all of the optimized geometrical parameters are generally in excellent agreement with the relevant experimental data, a comment is in order on the short Ni-N distances, of around 190 pm, found in our calculations and experiments. In a study on metallohydroporphyrins, we have shown that peripheral substituents on the tetrapyrrole ring play a significant role in shortening the metalnitrogen distances and that Ni(II) complexes of unsubstituted hydroporphyrins exhibit significantly longer Ni-N distances of 195-197 pm.⁶

The nonplanar distortions, given by z displacements in pm in Figure 1, refer to the mean N_4 plane for the ruffled porphyrins. For saddled porphyrins, they refer to a plane containing the metal and perpendicular to the molecular C_2 axis. As in the case of the ruffled conformations, the nonplanar distortions are signifi-

^{*} To whom correspondence should be addressed. E-mail: abhik@chem.uit.no. Fax: +47 77644765.

Figure 1. Optimized geometries (pm, deg) of different metalloporphyrins studied.

cantly smaller for the Zn(II) complexes than for the Ni(II) complexes. Note the unusual mild doming in the saddled conformation of Zn(TCF₃P). It remains to be seen whether this is an artifact of the symmetry constraint used in the optimization.

The optimized structures (Figure 1) clearly demonstrate that the nonplanar distortions of meso-tetrakis(perfluoroalkyl)porphyrins result from short $F \cdots H_{\beta}$ contacts. The $F \cdots H_{\beta}$ contact distances are roughly comparable between the ruffled and the saddled conformations for any of the compounds studied. In other words, ruffling and saddling distortions appear to be comparably effective in alleviating $F \cdots H_{\beta}$ steric repulsions relative to a hypothetical planar conformation. Note also that each C₂F₅ group in the TC₂F₅P complexes has four relatively short $F \cdots H_{\beta}$ contacts compared to only two for each CF_3 group in the TCF₃P complexes. Yet the TCF₃P and TC₂F₅P complexes exhibit comparable nonplanar distortions for a particular metal. The clue to this paradox lies in the directionality or vectorial character of the $F \cdots H_{\beta}$ repulsions: the nonplanar distortions are governed by not only the number of short $F \cdots H_{\beta}$ contacts but also their exact orientations in space.

(c) Comparison of *meso*-Tetrakis(heptafluoropropyl)porphyrins and DPPs. How do the above results compare with relevant experimental results on conformational flexibility of porphyrins? Experimentally, the largest collection of data on conformational flexibility are available for the DPPs. Although we have not studied multiple conformations of DPPs theoretically, we take the liberty of making some speculative remarks on the experimental DPP structures.

Crystallographic analysis of unsubstituted NiDPP revealed a strongly ruffled conformation, a deformation mode not previously observed for sterically crowded dodecasubstituted porphyrins.1 In contrast, a fluorinated analogue of NiDPP, NiDP-PF₂₄, exhibited a saddled conformation. For another fluorinated derivative, NiDPPF₂₀, with meso-pentafluorophenyl groups, two crystalline forms were characterized that were found to have different proportions of ruffling and saddling, with one form more ruffled and the other more saddled.1 The more ruffled and more saddled forms featured mean Ni-N distances of 189.2(3) and 191.6(3) pm, respectively. In other words, although short Ni-N bonds around 190 pm can occur for both ruffled and saddled porphyrins, the Ni-N bonds are slightly shorter in the ruffled conformation than in the saddled one for a particular peripheral substitution pattern. Thus, in contrast to Ni(II) mesotetrakis(perfluoroalkyl)porphyrins, which appear to have the strongest preference for ruffled conformations, NiDPP derivatives appear to exhibit remarkable conformational flexibility. How can we explain this?

The fact that the majority of DPP derivatives exhibit saddled conformations¹ suggests that DPPs are inherently predisposed toward saddling rather than toward ruffling distortions. Only the very small Ni(II) ion appears to make ruffling energetically competitive with saddling. In contrast, *meso*-tetrakis(perfluoroalkyl)porphyrins do not seem to have an inherent preference for ruffling over saddling or vice versa and, thus, ruffling is significantly preferred for very small metal ions such as Ni(II). Additional calculations are underway to verify these speculative remarks

Finally, to put the results in a broader perspective, we note that conformational flexibility is also known for other classes of nonplanar aromatics, the premier example being the bowl-to-bowl inversion of corannulene.¹⁰

In summary, *meso*-tetrakis(perfluoroalkyl)porphyrins are conformationally flexible, and ruffled, saddled, and intermediate conformations may be expected in crystallographic analyses of different metal complexes. Although they are well known for dodecasubstituted porphyrins, multiple nonplanar conformations are less known for *meso*-tetrasubstituted β -unsubstituted porphyrins.

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