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# Regioselective Cope Rearrangement and Prenyl Transfers on Indole Scaffold Mimicking Fungal and Bacterial Dimethylallyltryptophan Synthases †

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Supporting Information

ABSTRACT: Aromatic prenyltransferases are an actively mined enzymatic class whose biosynthetic repertoire is growing. Indole prenyltransferases catalyze the formation of a diverse set of prenylated tryptophan and diketopiperazines, leading to the formation of fungal toxins with prolific biological activities. At a fundamental level, the mechanism of C4-prenylation of Ltryptophan recently has surfaced to engage a debate between a "direct" electrophilic alkylation mechanism (for wt DMATS and FgaPT2) versus an indole C3-C4 "Cope" rearrangement followed by rearomatization (for mutant FgaPT2). Herein we provide the first series of regioselectively tunable conditions for a Cope rearrangement between C3 and C4 positions. Biomimetic conditions are reported that effect a [3,3]-sigmatropic shift whose two-step process is interrogated for intramolecularity and ratelimiting general base-promoted mechanism. Solvent polarity serves a crucial role in changing the regioselectivity, resulting in sole [1,3]-shifts under decalin. An intermolecular variant is also reported that effectively prenylates the C3 position of L-tryptophan, resulting in products that mimic the structures accessed by bacterial indole prenyltransferases. We report an elaborate investigation that includes screening various substituents and measuring steric and electronic effects and stereoselectivity with synthetically useful transformations.

# ■ INTRODUCTION

Prenyltransferases are an extensive family of enzymes responsible for activating and modifying isoprenoids (C<sub>5</sub>, C<sub>10</sub>, C<sub>15</sub>, C<sub>20</sub>, etc.) for the biosynthesis of diverse secondary metabolites. Aromatic prenyltransferases are an actively mined enzymatic class whose biosynthetic repertoire is growing. Tryptophan-containing alkaloids and peptidyl natural products are prenylated by a monophyletic subclass of aromatic prenyltransferases in fungi, cyanobacteria, and a few plants. Claviceps purpurea dimethylallyltryptophan synthase (DMATS) catalyzes the committed step toward formation of C4dimethylallyltryptophan (1) from L-tryptophan (L-Trp) and dimethylallyl diphosphate (DMAPP), as represented in Figure

This step is a crucial biosynthetic event leading to the formation of structurally diverse C4-prenylated alkaloid natural products. Late-stage modifications on 1 lead to divergent pathways toward ergot and clavine alkaloids, and their specificity is organism-dependent.<sup>6–8</sup> Evolutionarily related DMATS species, also referred to as indole prenyl transferases (PTs), have been identified and biochemically characterized

from multiple genera of fungi<sup>1,9-30</sup> and other microbial systems. 31-36 The underlying mechanistic dichotomy between a direct C4-prenylation on the indole nucleus (through an electrophilic aromatic substitution<sup>3,37</sup>) and a C3-reverse prenylation followed by a Cope rearrangement-tautomerization sequence (recently brought to light by Tanner et al. 27,28,38,39) still lingers, depending on the nature of the enzyme one investigates. Proposed biosynthetic formalisms predated even the initial discovery and biochemical characterization of Claviceps purpurea DMATS gene. 40,41 Mechanistic speculations have spurred experimental research for over three decades, with synthetic models originating from the laboratories of Arigoni, <sup>42</sup> Wenkert, <sup>43</sup> and Jackson, <sup>44–47</sup> leading up to recent reports of elegant mechanistic probes developed by Gaich and co-workers 48,49 (Scheme 1A, B).

In the pregenomic era, the structure of echinulin, a fungal prenylated tryptophan alkaloid, soon after its disclosure generated active interest from synthetic chemists attempting

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#### DMATS enzymology

Figure 1. Enzymatic reaction catalyzed by C4-dimethylallyltryptophan synthase (4-DMATS) and its biosynthetic context toward clavine and ergotamine alkaloid formation in multiple fungal organisms.

to offer a mechanistic rationale for its biosynthesis. The direct C4 electrophilic alkylation mechanism originally proposed (and experimentally verified) for the Claviceps 4-DMATS enzyme by Poulter et al. recently has received a rejuvenated interest. The position isotope exchange (PIX) experiments<sup>38,39</sup> along with the substitution studies on wild-type Claviceps 4-DMATS<sup>37</sup> added further evidence for a direct C4 electrophilic alkylation of L-Trp. The discovery of a C3-reverse prenylated pyrroloindoline product under a Lys174Ala mutant of FgaPT2 enzyme raised the possibility of a potential C3- to C4-Cope rearrangement as a mechanistic model.<sup>39</sup> In the postgenomic era, nearly 200 gene candidates that share significant sequence similarity to DMATS are identifiable in publicly available genomes. While the synthetic endeavors reported by Voute et al. 50 and Gaich et al. 48,49 (as shown in Scheme 1B) provided some insight into the feasibility of models proposed earlier by Arigoni, 42 Wenkert, 43 and Jackson, 44 these biosynthetic transformations continue to present special challenges for synthetic chemists attempting to mechanistically emulate their molecular construction steps.

A direct Cope rearrangement<sup>51</sup> on open-chain L-Trp to access 1 is still missing. Herein we report, for the first time, a detailed study of an efficient [3,3]-Cope rearrangement (from C3 to C4), under biomimetic and environmentally friendly conditions (Scheme 1C). Additionally, a regio- and stereoselective [1,3]-rearrangement and an intermolecular prenyl transfer reaction resulting in pyrroloindolines (with C3-normal prenyl substitution) are reported. These reactions are appealing additions to the synthetic toolkit, considering the tediousness that prefunctionalization of the C4 position of indole demands for synthesizing complex, bioactive natural products.

#### RESULTS AND DISCUSSION

# Access to Substrates for a [3,3]-Cope Rearrangement.

As shown in Table 1, in order to test the feasibility of a [3,3]-sigmatropic rearrangement on the open-chain indole scaffold, we prepared a variety of substrates using a Meerwein–Eschenmoser–Claisen rearrangement<sup>52–54</sup> on C3-substituted indoles (15). Treatment of nitrile-containing C3-substituted indole 15a with *N*-chlorosuccinimide (NCS) in the presence of 1,4-dimethylpiperazine, prenol (8), and trichloroacetic acid resulted in a 61% yield of oxindole 16a. The series of reverse-prenylated oxindoles synthesized included *N*-phthalyl-protected tryptamine (16b), its *N*-Me-Tosyl protected variant (16c),

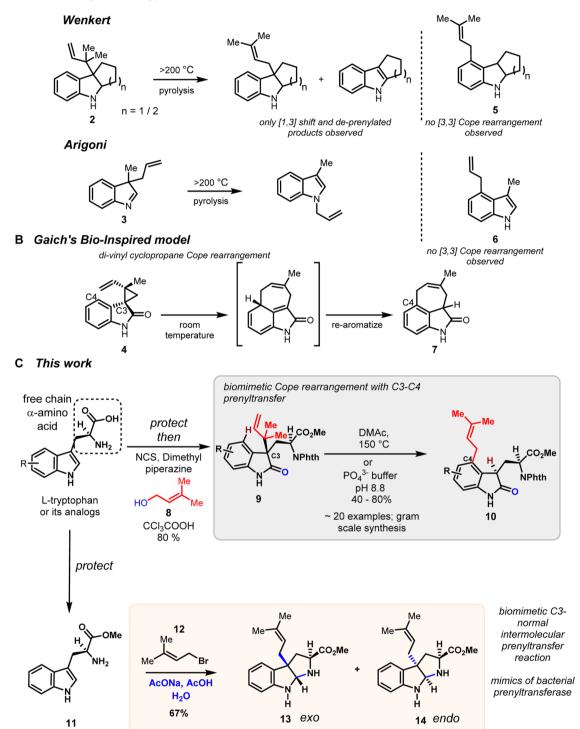
substituted indole C3-propionic acid carboxyethyl ester (16d) and its methyl ketone version (16e), and a simpler C3-methyl substitution (16f). The series of other substituted C3-reverse prenylated oxindoles prepared for this study included examples 16g-k possessing a C6-carboxymethyl ester substitution and a C6-fluoro-, C6-chloro-, C7-methyl-, and C5-methoxy-substituted oxindoles, respectively, all in moderate to high percent isolated yields. Technically, any allylic alcohol could participate in the one-step method involving activation with NCS as shown in eq 1, and therefore a series of nonprenyl alcohols were engaged to prepare oxindoles 16l-o, respectively. The extended prenyl chain compounds geraniol and nerol gave oxindoles 16p,q, respectively. Treatment of derivative 15r with (E,E)-farnesol resulted in the formation of 16r.

Heat- and Microwave-Promoted [3,3]- and [1,3]-Prenyl Transfer Rearrangements. Having synthesized **16a**—r as precursors for testing the [3,3]-sigmatropic rearrangement, we first attempted to identify the optimal set of solvent and temperature conditions to effect this transformation. A series of solvents ranging from those with very low dielectric constants such as toluene and decalin to highly polar solvents with very high dielectric constants such as dimethylacetamide (DMAc) and water were screened (see Table S3 in the Supporting Information). Bearing in mind that oxindole C3substitution drastically affects the conformational preferences and populations with respect to reactivity at the C4 position, 55 we chose to test the feasibility of the [3,3]-sigmatropic rearrangement with a C3-methyl substituent for reasons of least perturbation to the process of the two olefins engaging in the rearrangement. Therefore, 16f was subjected to thermal rearrangement conditions, it was observed that many solvents were suboptimal in effecting any rearrangement, and we recovered unreacted starting 16f. Upon employing DMAc and decalin, two solvents with drastically different polarities, we observed a dramatic shift in the regioselectivity of rearrangement products, as illustrated in Scheme 2. When it was heated to 160 °C in DMAc, oxindole 16f gave the Cope rearrangement product 17f in 32% yield (see the Supporting Information for additional details).

When decalin was employed as the solvent, under reflux temperature, we observed smooth and near-quantitative conversion to C3-dimethylallyl oxindole **18f**, which contained a *normal* prenyl substitution. A formal [1,3]-prenyl transfer process at the C3 position of oxindole had occurred under

Scheme 1. (A) Prior 4-DMATS Enzymatic Mimics That Failed, (B) C3–C4 Divinyl Cyclopropane Ring-Opening Cope Rearrangement Reported by Schwarzer and Gaich,<sup>48</sup> and (C) Description of Intra- and Intermolecular Prenyl Transfer Reactions under Biomimetic Conditions Reported Herein

# A Prior Biosynthetic Hypothesis



decalin as the nonpolar solvent. Thus, mechanistically divergent pathways were found to be operative under the two solvent conditions. Excited by the prospect of observing a [3,3]-sigmatropic (reverse-prenyl to prenyl) rearrangement occurring between the C3 and C4 positions of oxindole 16f, we next tested a series of oxindoles for estimating the synthetic efficiency of the Cope rearrangement. As shown in Table 2

(eq 2), various substitution patterns were studied, including substituents at the C3, C5, C6, and C7 positions. Cyanoethylcontaining **16a** rearranged to give the [3,3]-product **17a** in 31% yield. The protected tryptamine **16b** underwent a [3,3]-sigmatropic shift to the C4 position in 74% yield. A range of substitution patterns was tested, and uniformly, the efficiency of the [3,3]-rearrangement was evaluated as given in Table 2.

% yield

74

52

40

63

65

Table 1. Prenyl Transfer Reaction for Synthesis of Reverse-Prenylated Oxindoles

65

16f

C5-OMe MeC

Minor amounts of [1,3]-rearrangement products were formed in some cases. We probed the probable involvement of a radical-mediated process during the formation of products resulting from a [1,3]-prenyl rearrangement. On the basis of similar studies performed by Harwood et al. on related Claisen rearrangements, <sup>56–58</sup> we wondered if the radical process initiated by atmospheric oxygen could then be can inhibited under presence of hydroquinone. Accordingly, when **16e**,f were

Scheme 2. Shift of Regioselectivity for Prenyl Transfer Reactions with Drastically Different Solvent Dielectric Constants<sup>a</sup>

Table 2. Substrate Scope and Generality Given as a Function of R Group Variability at the C3 Position of Oxindoles<sup>a</sup>

	substituents						yield (%)	
substrate	R	$\mathbb{R}^1$	$R^2$	R <sup>3</sup>	time (h)	conversion (%)	[3,3]-17	[1,3]-18
16a	CH <sub>2</sub> CH <sub>2</sub> CN	Н	Н	Н	20	nd	31	<5
16b	CH <sub>2</sub> CH <sub>2</sub> NPhth	Н	Н	Н	17	62	74	4
16c	CH <sub>2</sub> CH <sub>2</sub> N(Tos)Me	Н	Н	Н	18	60	57	11 <sup>c</sup>
16d <sup>c</sup>	CH <sub>2</sub> CH <sub>2</sub> COOEt	Н	Н	Н	48	50	52	<5
16e	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	Н	Н	43	nd	43 $(50^b)$	ca. 10 <sup>c</sup>
16f	Me	Н	Н	Н	50	nd	$32 (36^b)$	16
$16g^d$	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	COOMe	Н	17	nd	43	<5
16h	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	F	Н	15	ns	<5	<5
16i	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	Cl	Н	15	ns	<5	<5
16j	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	Н	Me	18	45	60	10 <sup>c</sup>
16k	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	OMe	Н	Н	22	nd	$45 (50^b)$	<5
16s	CH <sub>2</sub> CH(NPhth)COOMe	Н	Н	Н	40	100	61	<5
16t	CH <sub>2</sub> CH <sub>2</sub> N(Nos)Me	Н	Н	Н	15	58	50	11 <sup>c</sup>

<sup>&</sup>quot;Sterically larger R groups encouraged [3,3]-rearrangement. Abbreviations: nd, not determined; n, —not stable under the conditions tested. "Isolated yield based on recovered starting material. "Calculated from H NMR data." (Unoptimized.

independently subjected to [1,3]-prenyl rearrangement conditions in decalin, we observed complete recovery of starting material, even after 72 h. However, the reaction proceeded well when hydroquinone was eliminated, suggesting that the radical-based mechanism was probable. Overall, in non-isatin-derived substrates (16e,f) as well as in isatin-derived cases (21a-c; vide infra), we observed a radical pathway to be operative under [1,3]-rearrangement processes.

The synthetic utility of examples 16b, c, s, t is particularly noteworthy, as many alkaloid natural products possess a tryptamine or tryptophan side chain and their C4-prenylated products are directly accessed through this [3,3]-rearrangement process. Convinced that the strategy was working on oxindoles with the tryptamine side chain, we studied the conditions to improve reaction rates and yields. Multiple boron- and

aluminum-based Lewis acids failed to catalyze the Cope rearrangement process. However, heating the oxindole in aqueous phosphate buffer with microwave assistance dramatically accelerated the rearrangement (Table 3).

For example, the cyanoethyl oxindole **16a**, when subjected to microwave irradiation at 150 W and 150 °C, in just 60 min (as opposed to several hours under nonmicrowave heating conditions) gave a 58% yield of the C4-prenylated **17a**. Phthalyl-protected tryptamine gave a 75% yield of the C4-prenylated product **17b** in 40 min. Substrates **16c**—t all underwent a [3,3]-Cope rearrangement in durations that ranged between 40 and 90 min, at pH 8.8 in phosphate buffer. Except in cases **16f**, **h**, **k** all other substrates gave the [3,3]-rearranged product **17** as the major product under these conditions. The three exceptions produced minor amounts of

<sup>&</sup>lt;sup>a</sup>% yield refers to isolated yield after chromatography.

Table 3. C4-Prenylation of Tryptophan Scaffold under Phosphate-Buffered Conditions<sup>a</sup>

	su						
substrate	R	$\mathbb{R}^1$	$\mathbb{R}^2$	R <sup>3</sup>	temp ( $^{\circ}$ C)	time (min)	yield [3,3]-product 17 (isolated) (%)
16a	CH <sub>2</sub> CH <sub>2</sub> CN	Н	Н	Н	150	60	58
16b	CH <sub>2</sub> CH <sub>2</sub> NPhth	Н	Н	Н	150	40	75
16c	$CH_2CH_2N(Tos)Me$	Н	Н	Н	150	65	40
16d	CH <sub>2</sub> CH <sub>2</sub> COOEt	Н	Н	Н	150	65	42
16e	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	Н	Н	150	70	46
16f	Me	Н	Н	Н	165	90	16 <sup>b</sup>
16g	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	COOMe	Н	150	60	40
16h	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	F	Н	150	60	$40^d$
16i	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	Cl	Н	150	60	45
16j	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	Н	Н	Me	150	62	50
16k	CH <sub>2</sub> CH <sub>2</sub> COCH <sub>3</sub>	OMe	Н	Н	150	70	48 <sup>c</sup>
16s	CH <sub>2</sub> CH(NPhth)COOMe	Н	Н	Н	150	60	54
16t	CH <sub>2</sub> CH <sub>2</sub> N(Nos)Me	Н	Н	Н	150	50	ns

<sup>&</sup>lt;sup>a</sup>Abbreviations: ns, not stable under conditions tested. <sup>b</sup>42% of [1,3]-product was obtained <sup>c</sup>13% of [3,3]-product without an R group was obtained. <sup>d</sup>22% of [1,3]-product was obtained.

Scheme 3. (A) [3,3]-Prenyl Transfer reaction on L-Trp-Derived 16s, Starting with C3-Reverse Prenylation Followed by DMAc or Phosphate-Buffer-Mediated Rearrangement To Yield 17s and (B) Chiral HPLC Analysis of a Single Diastereomer of 16s Converting to a 3:1 Mixture of Diastereomers of 17s

Scheme 4. [3,3]-Prenyl Transfer Reaction on  $\alpha$ -Amino Acid, Starting with C3-Reverse Prenylation To Yield 16s after Protection Followed by Heat- or Microwave-Mediated Rearrangement, (A) [3,3]-Cope Rearrangement of Geraniol-Derived 16p, (B) [3,3]-Cope Rearrangement of Nerol-Derived 16q, and (C) [3,3] Cope Rearrangement of (E,E)-Farnesol-Derived 16r

either a [1,3]-prenyl rearrangement product (like in Table 2) for 16f,k or showed loss of the C3-substituent altogether in the case of 16h.

Realizing that a worthy test of a Cope rearrangement to construct a C4-prenylated indole scaffold will be on a  $\alpha$ -amino acid scaffold of L-Trp, as shown in Scheme 3A, we attempted sequential prenylation on the amino acid L-Trp upon protection (as its methyl ester N-phthalyl counterpart) by subjecting it to a Meerwein–Eschenmoser–Claisen rearrangement with prenol, resulting in 16s in 80% yield. Thus, on subjecting 16s to either the thermal rearrangement condition in DMAc or microwave irradiation in phosphate buffer at pH 8.8, we were glad to observe formation of 17s in 58–61% and 54% yields, respectively. The product 17s was formed as a 1:1 mixture of diastereomers (Scheme 3A).

This observation prompted that the Cope rearrangement/tautomerization sequence of C3-reverse prenylation substrates 16 to products 17 is not necessarily stereospecific. In order to further test the stereoselective nature of this transformation, we sought to explore a control experiment wherein L-Trp-derived 16s, purified to a single diastereomeric composition (dr: >99:1), could be subjected to the Cope rearrangement/tautomerization sequence and the product stereochemistry be

evaluated. Scheme 3B illustrates the results of such an experiment. Upon chiral HPLC analysis of the product 17s, we clearly observed a 3:1 ratio of products varying likely at the C3 stereocenter. As the likelihood of epimerization at the  $\alpha$ -carbon of the  $\alpha$ -amino acid chain is low, we attribute this result to the presence of either antipode at the C3 center. Overall, this Cope rearrangement proved to be stereoselective with respect to the C3 oxindole center, and its stereochemical course could also be translated to linear isoprenoid containing oxindoles.

As shown in Scheme 4, a chain-elongated version of the prenyl transfer process was next tested on substrate 16p. Similar to the case for 16s, oxindole 16p was accessed, as a single diastereomer (in 58% yield), through a Meerwein–Eschenmoser--Claisen rearrangement on N-phthalyl tryptamine (Table 1). Subjecting 16p to a [3,3]-Cope rearrangement process in DMAc solvent at  $160\,^{\circ}$ C gave a 17:83 mixture (in 55% yield) of E and E geometric isomers that differed in their newly formed double-bond geometry. Similarly, when 16p was subjected to microwave irradiation conditions, 17p was obtained as a 23:77 mixture of E and E isomers, as shown in Scheme E A Zimmerman—Traxler-like transition state assembly, depicted by E 19, probably rationalizes the observation of formation of the E isomer as the major product, keeping in

Scheme 5. (A) [1,3]-Prenyl Transfer Reaction for C3-Hydroxy-Substituted Oxindoles and (B) Proposed Pathway with Radical Intermediate 23b Likely To Explain Regioselectivity Observed for Formation of 22 Based on Hydroquinone Inhibition Data in A

mind that the participating atoms of the oxindole portion are  ${\rm sp}^2$  hybridized. The oxindole **16q**, on the other hand, was obtained as an 8:1 mixture of diastereomers (in 60% yield) from the Meerwein–Eschenmoser–Claisen rearrangement on N-phthalyl tryptamine and nerol as the allylic alcohol (Table 1).

When 16q was subjected to the Cope rearrangement process, the C3 to C4 prenyl migration occurred smoothly, resulting in 17q, which possessed the *E* geometry across the newly formed olefin in 46% and 62% yields, under DMAc or phosphate-buffered conditions with microwave irradiation, respectively. The product displayed the geometric ratio of *E* and *Z* isomers reflective of a situation similar to 17p but opposite in magnitude due to the nature of the diastereomeric composition of 16q. Overall, we observed stereospecific C3 to C4 prenyl transfer migration under Cope rearrangement conditions, with respect to the olefinic portion of the isoprenoid chain, for 16p,q. Similarly, when oxindole 16r (previously synthesized

from (*E,E*)-farnesol in 48% yield) was subjected to Cope rearrangement conditions, **17r** was obtained in 52% yield as a single isomer containing the *Z,E* double-bond geometry. Thus, we see complete conservation of the double-bond geometry related from farnesol substrate to product **17r** after two pericyclic rearrangements.

Mechanistic Evaluation of [3,3]-Cope Rearrangement. Scheme 5 outlines the results we observed when C3-hydroxy oxindoles 21a-c were subjected to the Cope rearrangement conditions. Interestingly, as opposed to other examples where anionic oxy-Cope rearrangements are known to occur in a more accelerated manner, we observed no [3,3]-rearrangement in any of these examples. Instead, every substrate underwent a nearly exclusive [1,3]-prenyl rearrangement resulting in C3-normal prenyl substitution in 22a-c. The synthetic efficiency of these thermally activated [1,3]-prenyl migration processes was reasonable, and 22a-c were obtained in 50%, 50%, and 41%

Scheme 6. Mechanism of [3,3]-Prenyl Transfer Reaction: (A) Crossover Experiment; (B) Two-Step Mechanism; (C) pH Dependence of (B); (D) Electrophilic Aromatic Substitution Model of 4-DMATS Involving Deprotonation of C4-Hydrogen Similar to Intermediate 26

yields, respectively, with DMAc as the solvent. The corresponding efficiency for [1,3]-prenyl transfer under microwave-mediated conditions were considerably higher at 89%, 75%, and 87%, respectively, for **22a**–c. While the [3,3]-Cope rearrangement process was investigated mechanistically as discussed below, the 3-hydroxy examples provided a reasonable rationale, as represented in Scheme 5B, for their mechanism for the [1,3]-prenyl migration event. Similar to the case for **16e,f**,

the presence of 10 mol % hydroquinone during the [1,3]-prenyl migration event entirely abrogated the formation of 22a-c. On the basis of nature's biosynthetic repertoire involving  $C_5$  (and higher order) isoprenoid pyrophosphates (DMAPP, GPP, FPP, etc.) in numerous examples of chain elongation and branching events, it is reasonable to consider a prenyl cation involvement (as shown in 23a) for the transformation of 21 to 22. However, considering the fact that a radical inhibitor causes complete

Scheme 7. Intermolecular Prenyl Transfer Reaction To Construct a Pyrroloindoline Skeleton under Biomimetic Conditions

shutdown of the conversion, we propose that a radical-based pathway involving the tightly coupled radical pair **23b** as the likely intermediate toward **22**. Wenkert and Sliwa, in their reported study<sup>43</sup> on 6-5-6 (or 6-5-5) pyrroloindolines that were C3-reverse prenylated, observed this type of a [1,3]-prenyl shift consistently for most examples they attempted under pyrolysis conditions. The mechanistic explanation offered here also could be applied to examples seen in their studies.

We next probed the [3,3]-Cope process for a mechanistic understanding, specifically asking whether the process is strictly intramolecular or if there was any intermolecularity involved when the prenyl group migrates from the C3 to the C4 position of the oxindole ring system. Therefore, as shown in Scheme 6A, a competition experiment was conducted with 16f, q premixed in an equimolar ratio and were subjected to phosphate-buffered solution at pH 8.8 and microwave irradiation at 150 °C. At complete conversion of the starting materials, we observed only the formation of 17f,q. The product containing a prenyl  $(C_5)$ group on the C4 position of tryptamine scaffold represented by 24 and the product containing a geranyl  $(C_{10})$  unit at the C4 position of methyl-carrying oxindole represented by 25 were not observed by <sup>1</sup>H NMR. This observation confirmed that an intermolecular scrambling of prenyl groups was not operative. No prenyl-scrambled [1,3]-products were seen either, indicating overall that the Cope process is intramolecular under these conditions.

Biochemical characterizations of DMATS and FgaPT2 revealed the presence of Lys174 and Glu89 as crucially important and essential residues. <sup>3,37–39</sup> Their roles were clarified through mutation studies, which revealed that Glu89 was involved in activation of the indole nucleophile through engagement in hydrogen bonding and ensuing removal of the N1 hydrogen. Lys174, whose site-directed mutagenesis nearly abolished the production of 1, served as the base responsible for deprotonation and tautomerization of the intermediate formed immediately after the C4-alkylation.

In our studies, as represented in Scheme 6B, considering that a [3,3] rearrangement results in the nonaromatic intermediate 26 (equivalent to the biological case preceding formation of 1), we wondered if this formal two-step process leading to C4-prenylated 17b would be sensitive to varying pH. Due to the fact that this "rearomatization" is essential for formation of the eventual product 17, the influence of a basic or acidic medium (mimicking the presence or absence of Lys174) is expected to

accelerate or decelerate the overall process, if this rearomatization event is rate-limiting. Therefore, we conducted an investigation by estimating the percent conversion (and corresponding percent yield) of the C4-prenylated product (17b) formation, starting from 16b at various pHs in phosphate-buffered solutions (Scheme 6C). Between pH 6 and 6.5, there was no 17b observed, indicating that the deprotonation of the C4 proton to cause rearomatization was shut down in even a slightly acidic medium. Drastic change toward >60% conversion and yield was observed at pH  $\geq$ 7.0. The conversion and yield improved consistently with increase in pH and were maximized close to pH 8.8. From a biomimetic standpoint this result was a near replica of the enzyme active site. The first step of the two-step process, a [3,3]-Cope rearrangement, is not expected to involve polar or ionic transition states (though this is not unprecedented) and therefore is expected to be pH neutral. Though we treat this Cope rearrangement as an equilibrium process, we do expect the apparent equilibrium constant to be favoring 26, considering steric congestion at C3 of the precursor oxindole 16. Therefore, the presence of a basic medium (at pH 8.8) is seen to dramatically shift the overall efficiency toward 17b through expediting the proton transfer event, resulting in a fully aromatic and stabilized 17b. With 17b as a representative, we anticipate most other substrates undergoing this prenyl transfer process to conform to this biomimetic pH effect.

The Cope rearrangement was also found to be sensitive to the ring substituent and the steric composition at the C3 position. As shown in Figure S1 (Supporting Information), the duration of the thermal rearrangement for each substitution pattern revealed the effect that the individual substitutions had on the transformation. Electron-withdrawing substituents on the aromatic ring of oxindole (16i, g, h) promoted the rate of the prenyl transfer process to the C4 position. This correlated with the expectation that the formal second step of the process involving tautomerization of intermediate 26 is expected to be promoted by electron-withdrawing substituents by virtue of lowering the p $K_a$  of the proton at the C4 position of 26 (see the blue highlight in Scheme 6B). Conversely, the presence of an electron-donating methoxy substituent (16k) slowed down the rearrangement process, again by virtue of increasing the  $pK_a$  of the proton at the C4 position of 26. The presence of substituents on the C3 chain of the tryptamine or tryptophan scaffold to a general extent had a positive effect on promoting

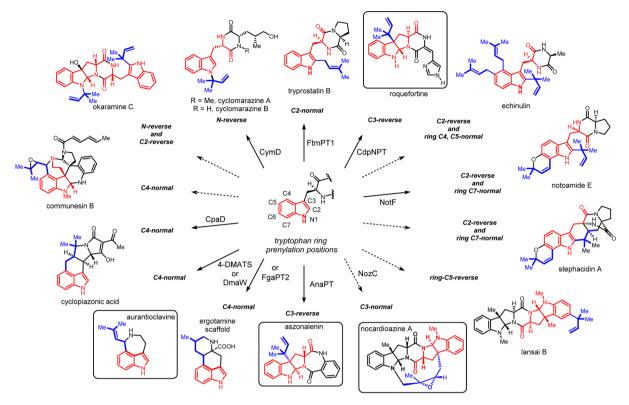


Figure 2. Indole prenyltransferases and their regioselectivity of prenylation (blue) on L-Trp (red). Bold arrows reflect the fact that the indicated enzyme has been biochemically characterized. Scaffolds presented in boxes represent examples for which biomimetic pathways are presented in this report.

the rate of the rearrangement process. Steric effects were manifested in two specific ways. The first type of steric effect that was observed was on examples that carried a C3 substituent. Among a cyanoethyl group (16a), an ethyl methyl ketone (16e), an ethyl carboxy ester (16d), and a protected tryptophan (16s), we found that the rate was influenced by a subtly balancing steric factor that tilted toward faster rearrangement for substituents that were too bulky. Second, a possible Thorpe-Ingold effect was apparent for substituents that forced the reactive olefins to engage in the rearrangement process. An extreme of this correlation can be seen in Gaich's reports that carry a cyclopropane to promote the rearrangement.<sup>48</sup> Additionally, the absence of a Cope rearrangement on any of the nonreverse prenyl oxindole examples (16l-o) corroborated the apparent presence of the Thorpe-Ingold effect imposed by the presence of the geminal dimethyl substitution of the reverse-prenyl group. The relatively slow and sluggish rate observed for C3-methyl substitution (16f) was attributable to a similar effect.

Prenylating the C3 position of indoles directly with alkyl electrophiles is a challenging process under aqueous conditions. To address this challenge, an intermolecular prenyl transfer reaction was attempted next, as outlined in Scheme 7. L-Trp methyl ester (11) was subjected to prenylation at the C3 position in the presence of prenyl bromide in sodium acetate—acetic acid buffer and pH 2.7 at room temperature and gave rise to products 13 and 14 in a 4:1 ratio favoring the *exo* product. The corresponding N1-methylated substrate (27) underwent the C3-prenylation event with prenyl bromide and resulted in a mixture of *exo* and *endo* diastereomers 28 and 29. The use of prenyl bromide to activate the C3 position of the indole ring system concomitantly resulted in formation of the pyrroloindo-

line ring system. Both in the case of nonmethylated 13 and in the case of methylated 29, the products were characterized for their relative stereochemistry through measurement of the NOESY correlations around the protons of the pyrroloindoline skeleton. The prenylation of indoles under aqueous conditions is challenging. We derived our inspirations from the report of Bocchi et al., though we fortuitously arrived at drastic improvements over the percent yields of the pyrroloindoline products.<sup>59</sup> Additionally, the indole C3-prenylation conditions we report herein results in approaches to natural product biosynthetic studies of the nocardioazine alkaloids. A high degree of biomimeticity is evident between our prenylation strategy (11 to 13 + 14; and 27 to 28 + 29) and those of the indole prenyltransferases observed in the literature leading to multiple alkaloid pathways encoded by the Aspergillus sp. 21 and the *Bacillus subtilis* ComX pheromonone synthesizing ComQ geranyl transferase enzyme. 60,61

**Tryptophan Prenylations in Biosynthesis.** Tryptophan prenylation (with  $C_5$  isoprenoid) is a well-documented enzymatic phenomenon. Every possible site for enzymatic prenylation on the indole scaffold (N1–C7) has been observed, with the individual enzyme being functionally active on select prenylation position(s) (Figure 2). Among the individual pathways, indole prenyltransferases arguably are one of the most intricate enzymes involved in the biosynthesis of complex natural products. Fungal biosynthetic pathways leading to a large collection of prenylated indole alkaloids (a few representative examples are presented in Figure 2) have spurred an extensive body of work on their structure elucidation, asymmetric synthesis, medicinal chemistry, and related studies. Gene clusters and their characterization have evolved in the postgenomic era and have actively inspired

further chemoenzymatic syntheses, mechanistic enzymology, 7,8,13,14,62 and structural biology of these unique biocatalysts. 63 For example, aurantioclavines are representative of the synthetic challenges that continue to spur newer methodologies to emerge. 64,65 Well into the 21st century, the only known L-Trp prenyltransferase was from Claviceps purpurea. Genome sequencing projects on Aspergillus spurred the identification of a series of indole prenyltransferases that were found to prenylate a variety of substrates such as L-Trp, diketopiperazines containing tryptophan units, and L-Trpconjugated benzodiazepines. As summarized in Figure 2, a spectrum of prenylation patterns can now be observed. Specifically, the work presented herein mimics the indole prenyltransferases operative toward the biosynthesis of four specific alkaloid classes. The C3-reverse prenylation reactions used herein for the synthesis of a series of oxindoles 16 mimic the products seen from AnaPT-catalyzed C3-reverse prenylation, leading to the formation of aszonalenin and CdpNPTcatalyzed formation of roquefortines. The C3-C4 prenyl migrating Cope rearrangement reported herein for the series of compounds 16 transforming to 17 mimics the catalytic action carried out by mutant FgaPT2. The product resulting from the Cope rearrangement is identical with that formed from a direct alkylation seen in Claviceps purpurea DMATS. The mimicry shown by the intramolecular [1,3]-prenyl migration event (16 to 18 or 21 to 22) results in products that share structural similarity to C3-normal prenylating enzymes. The intermolecular prenylation reaction resulting in the C3-normal pattern shown in Scheme 6 also mimics these indole prenyltransferases. One such pathway leads to the formation of nocardioazines, biosynthesized by Nocardiopsis sp. CMB M0232. Ongoing activity in our laboratory is currently elucidating the enzyme action of NozC, a bacterial homologue that prenylates C3 positions of L-Trp diketopiperazine substrates.

Interestingly, the catalysis affected by indole prenyltransferases is now beginning to be better understood through X-ray crystallography. As represented in those with known threedimensional structures show remarkable similarity in their biophysical folds leading to overlapping substrate-product relationships with surprising levels of promiscuity. Overall, this work is anticipated to enrich the synthetic toolkit<sup>63</sup> and add novel directions for future investigations in alkaloid total synthesis and biosynthesis. Moreover, in conjunction with earlier work published by Schwarzer et al., 48,49 these reactions together offer feasibility for a Cope rearrangement mechanism, at least under nonenzymatic conditions. In a related manner, a possible extension of this work could be directly applicable to the synthesis of compounds that may throw light on the mechanisms of FgaPT2 and DMAW synthase enzymes from Aspergillus and Claviceps, respectively.

#### CONCLUSION

L-Trp presents a dazzling array of nucleophilic positions that enzymes take full advantage of by prenylating various positions in natural product biosynthetic pathways. Synthetic mimicry of these selective biocatalysts has been rare, primarily due to the challenge in taming regio- and stereoselectivities of such transformations. Herein we present a facile [3,3]-prenyl group migrating Cope rearrangement that was evaluated for its synthetic efficiency, substituent diversity, biomimeticity, and possible mechanistic underpinnings. We find that the phosphate-buffered microwave-mediated conditions uniformly resulted in good yields of C4-prenylated products. Mechanis-

tically, pH variations and the ensuing sensitivity revealed a slower step involving tautomerization of the C4 proton to the C3 position. In turn, this implied that the Cope rearrangement itself could be the reversible faster step. Parallel to the Cope rearrangement, we found that use of a nonpolar solvent such as decalin promoted a regioselective [1,3]-prenyl transfer reaction resulting in C3-normal prenyl substituted oxindoles. C3hydroxylation also enforced a stricter selectivity for the [1,3]prenyl transfer process. Direct prenylation of the L-Trp scaffold was also effected under biomimetic conditions, resulting in pyrroloindolines with C3-normal prenyl substitution. Overall, these conditions are expected to find use in natural product total synthesis, considering the urgent need for stereoselective methods for complex alkaloid construction. The design and synthesis of mechanistic probes for DMATS and related enzymes may also be possible through these prenyl transfer reactions. These results may prompt synthetic chemists in the alkaloid field to explore additional methods of elaborating the indole ring utilizing prenylations and sigmatropic rearrangements.

#### **■ EXPERIMENTAL SECTION**

Representative Procedures. General Procedure A: Reverse Prenylation of C3-Substituted Indoles. A two-neck round-bottomed flask with a magnetic stir bar (capped with septa) was dried under vacuum. Once cooled to ambient temperature under inert atmosphere, the flask was charged with 3-cyanoethylindole (15a; 256 mg, 1.5 mmol) that was mixed with 5 mL of dry dichloromethane and stirred. Freshly distilled 1,4-dimethylpiperazine (0.13 mL, 0.96 mmol) was added, and the reaction flask was cooled to 0 °C on ice. Recrystallized NCS (221 mg, 1.65 mmol) was added, and the stirring continued at the same temperature for 2 h under nitrogen. In a separate roundbottomed flask, CCl<sub>3</sub>COOH (60 mg, 0.36 mmol) and prenyl alcohol (0.3 mL, 3.0 mmol) were mixed, diluted with dichloromethane (2 mL), and stirred at room temperature. After 2 min, this mixture was cannulated into the reaction flask at 0 °C and the ensuing mixture was gradually warmed to room temperature and stirred for 12 h. Wet dichloromethane (3 mL with 5 drops of water) was added dropwise to the mixture to quench the reaction. The reaction mixture was washed with 0.5 M NaHCO<sub>3</sub> (20 mL), 0.5 M HCl (20 mL), and brine (20 mL). The organic layers were dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated to yield the product mixture as a yellow oil. The mixture was subjected to silica gel chromatography (18% ethyl acetate in hexanes) to yield the purified product 16a (as a representative example) in 61% isolated yield (236 mg, 0.93 mmol) as a pale oil that solidified upon standing.

General Procedure B: Cope Rearrangement under Thermal Conditions. The C3 reverse-prenylated oxindole 16s (1.0 g, 2.31 mmol) was taken in a vial and capped with a septum. Under nitrogen, 16s was dissolved in DMAc to form a 0.1 M solution. The solution was transferred to a Schlenk tube using a syringe and capped under nitrogen and heated in an oil bath at 155 °C for 40 h. The reaction mixture was cooled to room temperature, and DMAc was removed in vacuo. The reaction mixture was diluted with ethyl acetate (15 mL) and washed with brine (15 mL  $\times$  3). The organic layer was dried with anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated to give the crude product, which was subjected to silica gel chromatography (refer to general experimental details in the Supporting Information) with ca. 1/4 v/v ethyl acetate/hexane as eluent, and the product oxindole 17s (as a representative example) was isolated as a pale yellow solid in 61% yield (610 mg, 1.41 mmol).

General Procedure C: Cope Rearrangement under Microwave Irradiation. A 10 mM phosphate buffer (pH 8.8) was prepared by dissolving 3 mg of monosodium phosphate hydrate and 262 mg of disodium hydrogen phosphate in 100 mL of doubly deionized water. C3-reverse prenylated oxindole (30 mg, 0.08 mmol) 16b was dissolved in 0.75 mL of buffer in a microwave-compatible sealed tube and subjected to irradiation of 150 W of 150 °C over 40 min. The reaction

mass was extracted with ethyl acetate (10 mL) and washed with brine (10 mL). The organic layer was dried with anhydrous  $Na_2SO_4$  and evaporated to give the crude reaction mixture, which upon purification by preparative TLC (refer to general experimental details in the Supporting Information) gave the product 17b in 75% yield (22.5 mg, 0.06 mmol).

General Procedure D: [1,3]-Prenyl Group Rearrangement. The C3 reverse-prenylated oxindole (50 mg, 0.13 mmol) 16b was transferred to a Schlenk tube as a dichloromethane solution. The solvent was removed by purging nitrogen followed by high vacuum to remove the last traces of dichloromethane. The oxindole was added to decalin to form a 0.1 M solution (turbidity observed). The Schlenk tube was sealed and heated in an oil bath at 185 °C for 2 days. Decalin was evaporated in vacuo. The crude sample was directly loaded into the silica gel column using toluene as solvent. The product 18b was obtained in 65% yield (32.5 mg, 0.087 mmol) after column chromatography with 1/4 v/v ethyl acetate/hexane.

General Procedure E: Intermolecular C3-Prenylation of Tryptophan. To a magnetically stirred solution of L-Trp methyl ester 11 (400 mg, 1.83 mmol) in sodium acetate/acetic acid buffer (pH 2.7, 30 mL) was added prenyl bromide (635  $\mu$ L, 5.50 mmol) over a period for 45–50 min at room temperature. The resulting mixture was stirred at the same temperature overnight. Evaporation of acetic acid under reduced pressure resulted in a solid residue which was dissolved in ethyl acetate. The solution was neutralized by addition of sodium carbonate solution, and the ester was extracted three times with ethyl acetate. The combined organic layers were dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, giving a diastereomeric mixture of the cyclic products 13 and 14 (in a 4/1 ratio) in 67% overall yield (358 mg, 1.25 mmol) (based on recovered starting material).

*C3-Reverse Prenylated Oxindole Substrates.* 3-(3-(2-Methylbut-3-en-2-yl)-2-oxoindolin-3-yl) propanenitrile (16a). Following the general procedure A, treatment of 3-cyanoethylindole 15a gave product 16a (236 mg, 0.93 mmol) as a white solid (mp 86 °C) in 61% yield. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.21 (br s, 1H), 7.25 (td, J = 7.6, 1.2 Hz, 1H), 7.16 (m, 1H), 7.03 (td, J = 7.6, 0.8 Hz, 1H), 6.93 (d, J = 8.0 Hz, 1H), 6.03 (dd, J = 17.6, 10.8 Hz, 1H), 5.13 (dd, J = 10.8, 1.2 Hz, 1H), 5.02 (dd, J = 17.6, 1.2 Hz, 1H), 2.47 (ddd, J = 13.6, 11.2, 5.6 Hz, 1H), 2.17 (ddd, J = 13.6, 11.2, 4.8 Hz, 1H), 1.91 (ddd, J = 16.8, 11.2, 5.6 Hz, 1H), 1.71 (ddd, J = 16.4, 11.2, 4.8 Hz, 1H), 1.15 (s, 3H), 1.03 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 180.3, 142.6, 141.9, 128.8, 128.5, 125.6, 122.1, 119.1, 114.5, 110.1, 58.1, 42.1, 27.7, 22.1, 21.9, 13.4. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3275, 2923, 1737, 1620, 1491, 1370 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>16</sub>H<sub>18</sub>N<sub>2</sub>O [M]<sup>+</sup> 254.1419, found 254.1415.

*N,4-Dimethyl-N-(2-(3-(2-methylbut-3-en-2-yl)-2-oxoindolin-3-yl)-ethyl)benzenesulfonamide* (*16c*). Following the general procedure A, treatment of *N*-Me-Tos-ethylindole (500 mg, 1.52 mmol) **15c** gave product **16c** as a pale yellow solid (mp 104 °C), in 66% yield (414 mg, 1.01 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.78 (br s, 1H) 7.46–7.41 (m, 2H), 7.26–7.18 (m, 4H), 7.05 (td, J = 7.6, 1.2 Hz, 1H), 6.85–6.82 (m, 1H), 6.01 (dd, J = 17.2, 10.8 Hz, 1H), 5.08 (dd, J = 10.8, 1.2 Hz, 1H), 4.96 (dd, J = 17.6, 1.2 Hz, 1H), 2.6 (s, 3H), 2.59–2.51 (m, 1H), 2.43 (ddd, J = 13.4, 10.2, 5.2 Hz, 1H), 2.37 (s, 3H), 2.32 (ddd, J = 13.2, 10.3, 5.9 Hz, 1H), 2.06 (ddd, J = 13.4, 10.3, 4.8 Hz, 1H), 1.1 (s, 3H), 1.0 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 180.1, 143.2, 143.1, 141.5, 134.4, 129.7, 129.6 (2 × C), 128.2, 127.4 (2 × C), 126.0, 121.9, 114.0, 109.4, 56.7, 47.1, 42.1, 35.3, 29.5, 22.2, 21.8, 21.6. FT-IR (KBr,  $\nu_{\text{max}}$ ): 2923, 1737, 1620, 1491 cm<sup>-1</sup>. HRMS (EI): calcd for  $C_{23}H_{28}N_2O_3S$  [M]<sup>+</sup> 412.1821, found 412.1818.

Ethyl 3-(3-(2-Methylbut-3-en-2-yl)-2-oxoindolin-3-yl)propanoate (16d). Following the general procedure A, treatment of 3-ethylcarboxyethylindole (281 mg, 1.29 mmol) 15d gave product 16d as a colorless oil, in 55% yield (214 mg, 0.71 mmol).  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.86 (br s, 1H), 7.23- 7.14 (m, 2H), 6.99 (td, J = 7.6, 1.2 Hz, 1H), 6.86–6.82(m, 1H), 6.05 (dd, J = 17.6, 10.8 Hz, 1H), 5.1 (dd, J = 10.8, 1.2 Hz, 1H), 5.0 (dd, J = 17.6, 1.2 Hz, 1H), 3.98 (qq, J = 10.8, 7.2 Hz, 2H), 2.37 (ddd, J = 13.6, 11.1, 5.2 Hz, 1H), 2.2 (ddd, J = 13.6, 11.6, 4.8 Hz, 1H), 1.96 (ddd, J = 16.4, 11.6, 5.2 Hz, 1H), 1.62 (ddd, J = 17.2, 11.6, 4.8 Hz, 1H), 1.17 (s, 3H), 1.14(t, J = 7.2 Hz, 3H), 1.06 (s,

3H).  $^{13}\mathrm{C}$  NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  181.1, 173.2, 143.2, 141.8, 130.1, 128.2, 125.9, 121.8, 114.0, 109.3, 60.5, 58.2, 42.0, 30.1, 26.5, 22.4, 21.9, 14.2. FT-IR (KBr,  $\nu_{\mathrm{max}}$ ): 3275, 2956, 1713, 1620, 1491, 1148 cm $^{-1}$ . HRMS (EI): calcd for  $\mathrm{C_{18}H_{23}NO_{3}}$  [M]+ 301.1678, found 301.1677.

3-(2-Methylbut-3-en-2-yl)-3-(3-oxobutyl)indolin-2-one (16e). Following the general procedure A, treatment of 3-(2-oxobutyl)indole (450 mg, 2.41 mmol) 15e gave product 16e as a semisolid (mp 98 °C) in 50% yield (324 mg, 1.19 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.17 (br s, 1H), 7.21 (td, J=7.6, 1.2 Hz, 1H), 7.15–7.12 (m, 1H), 6.99 (td, J=7.6, 1.2 Hz, 1H), 6.89–6.85 (m, 1H), 6.03 (dd, J=17.6, 10.8 Hz, 1H), 5.08 (dd, J=10.8, 1.2 Hz, 1H), 4.99 (dd, J=17.6, 1.2 Hz, 1H), 2.31–2.07 (m, 3H), 1.91 (s, 3H), 1.70–1.63 (m, 1H), 1.16 (s, 3H), 1.05 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 208.1, 181.6, 143.2, 141.8, 130.4, 128.1, 125.9, 121.9, 113.9, 109.6, 58.1, 41.9, 39.1, 30.1, 25.2, 22.5, 21.9. FT-IR (KBr,  $\nu_{\rm max}$ ): 2923, 1713, 1620, 1462, 1370, 752 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>17</sub>H<sub>21</sub>NO<sub>2</sub> [M]<sup>+</sup> 271.1572, found 271.1580.

*Methyl 3-(2-Methylbut-3-en-2-yl)-2-oxo-3-(3-oxobutyl)indoline-6-carboxylate* (*16g*). Following the general procedure A, treatment of 6-carboxymethyl-3-(2-oxobutyl)indole (411 mg, 1.67 mmol) **15g** gave product **16g** as a pale yellow oil in 74% yield (407 mg, 1.24 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.51 (br s, 1H), 7.68 (dd, J = 7.9, 1.5 Hz, 1H), 7.57 (d, J = 1.3 Hz, 1H), 7.19 (d, J = 7.9 Hz, 1H), 5.98 (dd, J = 17.4, 10.8 Hz, 1H), 5.05 (dd, J = 10.8, 0.8 Hz, 1H), 4.94 (dd, J = 17.4, 0.8 Hz, 1H), 3.89 (s, 3H), 2.37–2.02 (m, 3H), 1.89 (s, 3H), 1.71–1.54 (m, 1H), 1.15 (s, 3H), 1.01 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.7, 181.1, 166.7, 142.6, 142.2, 135.8, 130.2, 125.5, 123.4, 114.3, 110.2, 58.3, 52.4, 42.1, 38.9, 29.9, 25.0, 22.4, 21.8. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3259, 2957, 1716, 1628, 1458, 1286 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>19</sub>H<sub>23</sub>NO<sub>4</sub> [M]<sup>+</sup> 329.1627, found 329.1626.

6-Fluoro-3-(2-methylbut-3-en-2-yl)-3-(3-oxobutyl)indolin-2-one (16h). Following the general procedure A, treatment of 6-carboxymethyl-3-(2-oxobutyl)indole (453 mg, 2.20 mmol) 15h gave product 16h as a pale yellow oil in 52% yield (331 mg, 1.15 mmol).  $^1$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.46 (br s, 1H), 7.10–6.98 (m, 1H), 6.71–6.60 (m, 2H), 5.99 (dd, J = 17.4, 10.8 Hz, 1H), 5.06 (dd, J = 10.8, 1.1 Hz, 1H), 4.96 (dd, J = 17.4, 1.1 Hz, 1H), 2.32–2.01 (m, 3H), 1.91 (s, 3H), 1.76–1.63 (m, 1H), 1.14 (s, 3H), 1.01 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>): δ 208.0, 182.0, 163.9, 161.5, 143.2, 142.9, 126.8, 125.6, 114.2, 108.3, 98.3, 57.8, 42.0, 39.0, 30.0, 25.2, 22.4. FT-IR (KBr,  $\nu_{\rm max}$ ): 3262, 2968, 1709, 1600, 1488, 1206 cm $^{-1}$ . HRMS (EI): calcd for C<sub>17</sub>H<sub>20</sub>FNO<sub>2</sub> [M] $^+$  289.1478, found 289.1473.

6-Chloro-3-(2-methylbut-3-en-2-yl)-3-(3-oxobutyl)indolin-2-one (16i). Following the general procedure A, treatment of 6-chloro-3-(2-oxobutyl)indole (291 mg, 1.31 mmol) 15i gave product 16i as a pale yellow solid (mp 114 °C) in 40% yield (160 mg, 0.53 mmol).  $^1$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.30 (br s, 1H), 7.05 (dd, J = 8.3, 3.7 Hz, 1H), 7.00–6.88 (m, 2H), 5.99 (dd, J = 17.4, 10.8 Hz, 1H), 5.08 (dd, J = 10.8, 1.0 Hz, 1H), 4.97 (dd, J = 17.4, 1.0 Hz, 1H), 2.34–2.02 (m, 3H), 1.93 (s, 3H), 1.75–1.63 (m, 1H), 1.16 (s, 3H), 1.02 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.9, 181.5, 142.9, 142.7, 133.8, 128.8, 126.6, 121.8, 114.4, 110.2, 57.9, 42.1, 39.0, 30.1, 25.1, 22.5, 21.9. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3262, 2968, 1709, 1600, 1488, 1206 cm $^{-1}$ . HRMS (EI): calcd for C<sub>17</sub>H<sub>20</sub>ClNO<sub>2</sub> [M] $^+$  305.1183, found 305.1185.

7-Methyl-3-(2-methylbut-3-en-2-yl)-3-(3-oxobutyl) indolin-2-one (16j). Following the general procedure A, treatment of 7-methyl-3-(2-oxobutyl)indole (291 mg, 1.45 mmol) 15j gave product 16j as a pale yellow oil in 63% yield (259 mg, 0.91 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.96 (br s, 1H), 7.02 (m, 1H), 6.95 (d, J = 6.9 Hz, 1H), 6.92–6.83 (m, 1H), 6.02 (dd, J = 17.4, 10.8 Hz, 1H), 5.03 (dd, J = 10.8, 1.2 Hz, 1H), 4.96 (dd, J = 17.4, 1.2 Hz, 1H), 2.31 (s, 3H), 2.29–2.23 (m, 1H), 2.21–2.02 (m, 2H), 1.88 (s, 3H), 1.69 (ddd, J = 15.6, 11.3, 3.4 Hz, 1H), 1.15 (s, 3H), 1.04 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 208.0, 182.3, 143.3, 140.7, 129.9, 129.4, 123.1, 121.7, 118.9, 113.7, 58.4, 41.8, 39.1, 30.0, 25.2, 22.5, 21.8, 16.6. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3262, 2968, 1709, 1713, 1600, 1488, 1206 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>18</sub>H<sub>13</sub>NO<sub>2</sub> [M]<sup>+</sup> 285.1729, found 285.1722.

5-Methoxy-3-(2-methylbut-3-en-2-yl)-3-(3-oxobutyl)indolin-2-one (16k). Following the general procedure A, treatment of 5-

methoxy-3-(2-oxobutyl)indole (455 mg, 2.09 mmol) **15k** gave product **16k** as a white solid (mp 136 °C) in 65% yield (410 mg, 1.36 mmol). 

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.11 (br s, 1H), 6.85–6.77 (m, 1H), 6.76–6.68 (m, 2H), 6.05 (dd, J=17.4, 10.8 Hz, 1H), 5.07 (dd, J=10.8, 1.2 Hz, 1H), 4.97 (dd, J=17.4, 1.2 Hz, 1H), 3.75 (s, 3H), 2.31–2.18 (m, 1H), 2.17–2.01 (m, 2H), 1.90 (s, 3H), 1.77–1.63 (m, 1H), 1.16 (s, 3H), 1.02 (s, 3H). 

<sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 208.0, 181.4, 155.1, 143.1, 135.2, 131.9, 114.0, 113.1, 112.3, 109.7, 58.5, 55.7, 41.9, 39.1, 30.0, 25.2, 22.4, 21.8. FT-IR (KBr,  $\nu_{\rm max}$ ): 3262, 2968, 1709, 1713, 1600, 1488, 1206 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>18</sub>H<sub>23</sub>NO<sub>3</sub> [M]<sup>+</sup> 301.1678, found 301.1674.

2-(2-(3-(But-3-en-2-yl)-2-oxoindolin-3-yl)ethyl)isoindoline-1,3dione (161). Following the general procedure A, treatment of Nphthalyltryptamine (400 mg, 1.38 mmol) 15l gave product 16l as a mixture of major and minor diastereomers (dr = 5.6:1) as a pale yellow oil in 32% yield (159 mg, 0.44 mmol) (alongside full recovery of unreacted starting material). Data for the major diastereomer are as follows. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.59 (br s, 1H), 7.71–7.66 (m, 2H), 7.64-7.58 (m, 2H), 7.11-7.05 (m, 1H), 7.01-6.98 (m, 1H),  $6.87 \, 6.74 \, (m, 2H), \, 5.83 \, (ddd, J = 16.9, \, 10.3, \, 9.0 \, Hz, \, 1H), \, 5.20 - 5.09$ (m, 2H), 3.57–3.48 (m, 2H), 2.71 2.67 (m, 1H), 2.55–2.43 (m, 1H), 2.23 2.18 (m, 1H), 0.73 (d, J = 6.8 Hz, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  181.1, 168.0 (2 × C), 141.5, 138.0, 133.8 (2 × C), 132.0 (2 × C), 129.6, 127.9, 124.2, 123.0 (2 × C), 122.2, 117.6, 110.1, 54.9, 45.5, 34.4, 33.6, 14.8. FT-IR (KBr,  $\nu_{\rm max}$ ): 3257, 3065, 2972, 2927, 1772, 1707, 1619, 1469, 720 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>22</sub>H<sub>20</sub>N<sub>2</sub>O<sub>3</sub> [M]+ 360.1474, found 360.1469.

2-(2-(2-Oxo-3-(1-phenylallyl)indolin-3-yl)ethyl)isoindoline-1,3-dione (16m). Following the general procedure A, treatment of N-phthalyltryptamine (400 mg, 1.38 mmol) 15m gave product 16m as a mixture of major and minor diastereomers (dr = 8:1) as a pale yellow oil in 47% yield (274 mg, 0.65 mmol). Data for the major diastereomer are as follows. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.85 (br s, 1H), 7.72 7.56 (m, 4H), 7.30 (d, J = 7.4 Hz, 1H), 7.06 6.90 (m, 4H), 6.85 (t, J = 7.5 Hz, 1H), 6.67 (d, J = 7.2 Hz, 2H), 6.59 (d, J = 7.6 Hz, 1H), 6.25 (dt, J = 16.3, 10.3 Hz, 1H), 5.35–5.19 (m, 2H), 3.69 (d, J = 10.0 Hz, 1H), 3.48 (t, J = 6.6 Hz, 2H), 2.63 (dt, J = 14.6, 7.4 Hz, 1H), 2.38–2.26 (m, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 179.9, 168.0 (2 × C), 141.6, 138.4, 135.4, 133.8 (2 × C), 132.0 (2 × C), 129.1, 128.4 (2 × C), 128.3, 127.8 (2 × C), 127.0, 124.3, 123.1 (2 × C), 122.1, 119.2, 110.1, 58.0, 55.9, 34.4, 33.3. FT-IR (KBr,  $\nu_{max}$ ): 3420, 2926, 2855, 1716, 1470, 1265 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>27</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub> [M]<sup>+</sup> 422.1630, found 422.1626.

3-Allyl-3-methylindolin-2-one (16n). Following the general procedure A, treatment of C3-methylindole (260 mg, 1.98 mmol) 15n gave product 16n as a pale yellow oil in 47% yield (174 mg, 0.93 mmol).  $^1$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.88 (br s, 1H), 7.23–7.14 (m, 2H), 7.03 (td, J = 7.5, 1.0 Hz, 1H), 6.95–6.91 (m, 1H), 5.51 (ddt, J = 17.1, 10.1, 7.3 Hz, 1H), 5.08–4.98 (m, 1H), 4.97 4.92 (m, 1H), 2.64–2.44 (m, 2H), 1.41 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>): δ 183.1, 140.5, 134.2, 132.6, 127.9, 123.3, 122.4, 118.9, 109.9, 48.9, 42.5, 22.9. FT-IR (KBr,  $\nu_{max}$ ): 3371, 2253, 1709, 1470, 1160 cm $^{-1}$ . HRMS (EI): calcd for  $C_{12}H_{13}$ NO [M] $^+$  187.0997, found 187.0998.

3-Methyl-3-(2-methylallyl)indolin-2-one (16o). Following the general procedure A, treatment of C3-methylindole (500 mg, 3.82 mmol) 15o gave product 16o as a pale yellow oil in 53% yield (406 mg, 2.02 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.22 (br s, 1H), 7.23–7.16 (m, 2H), 7.07–7.00 (m, 1H), 6.92–6.84 (m, 1H), 4.61 4.59 (m, 1H), 4.56 4.53 (m, 1H), 2.73 (dd, J = 13.5, 0.8 Hz, 1H), 2.49 (d, J = 13.5 Hz, 1H), 1.41 (s, 3H), 1.39 (dd, J = 1.4, 0.9 Hz, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 182.7, 141.3, 140.4, 134.3, 127.8, 123.7, 122.4, 114.5, 109.8, 49.3, 45.7, 25.1, 23.8. FT-IR (KBr,  $\nu_{\rm max}$ ): 3214, 2971, 2925, 1708, 1621, 1473, 1232 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>13</sub>H<sub>15</sub>NO [M]\* 201.1154, found 201.1155.

2-(2-(3-(3,7-Dimethylocta-1,6-dien-3-yl)-2-oxoindolin-3-yl)ethyl)-isoindoline-1,3-dione (16p). Following the general procedure A, treatment of *N*-phthalyltryptamine (616 mg, 2.12 mmol) 15p gave product 16p as a single diastereomer (dr > 19:1) as a pale yellow oil in 58% yield (544 mg, 1.23 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.98 (br s, 1H), 7.74–7.66 (m, 2H), 7.66–7.59 (m, 2H), 7.18 (d, J = 7.0

Hz, 1H), 7.03 (td, J = 8.0, 1.2 Hz, 1H), 6.86–6.76 (m, 2H), 5.82 (dd, J = 17.5, 10.9 Hz, 1H), 5.24 (dd, J = 10.9 and 1.1 Hz, 1H), 5.01 (dd, J = 17.5, 1.1 Hz, 1H), 4.94–4.84 (m, 1H), 3.42–3.21 (m, 2H), 2.48–2.32 (m, 2H), 1.82–1.64 (m, 2H), 1.57 (s, 3H), 1.46 (s, 3H), 1.35 (ddd, J = 19.3, 12.9, 6.5 Hz, 1H), 1.27–1.16 (m, 4H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 180.0, 168.0 (2 × C), 141.7, 141.5, 133.8 (2 × C), 132.1 (2 × C), 131.4, 130.0, 127.9, 125.9, 124.6, 123.1 (2 × C), 121.7, 116.4, 109.7, 57.7, 45.8, 34.8, 34.4, 29.5, 25.8, 23.1, 17.7, 15.5. FT-IR (KBr,  $\nu_{\rm max}$ ): 3319, 2923, 2253, 1773, 1712, 1620, 1475, 1402, 908, 733 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>28</sub>H<sub>30</sub>N<sub>2</sub>O<sub>3</sub> [M]<sup>+</sup> 442.2256, found 442.2259.

2-(2-(3-(3,7-Dimethylocta-1,6-dien-3-yl)-2-oxoindolin-3-yl)ethyl)isoindoline-1,3-dione (16q). Following the general procedure A, treatment of N-phthalyltryptamine (475 mg, 1.64 mmol) 15q gave product 16q as a mixture of major and minor diastereomers (dr = 8:1) as a pale vellow oil in 60% yield (434 mg, 0.98 mmol). Data for the major diastereomer are as follows.  $^1$ H NMR (400 MHz, CDCl $_3$ ):  $\delta$ 8.61 (br s, 1H), 7.71-7.64 (m, 2H), 7.63-7.56 (m, 2H), 7.18-7.13 (m, 1H), 6.98 (td, J = 7.7, 1.2 Hz, 1H), 6.85–6.79 (m, 1H), 6.75 (td, J)= 7.6, 1.1 Hz, 1H), 6.01 (dd, J = 17.5, 10.8 Hz, 1H), 5.27–5.14 (m, 1H), 5.01-4.90 (m, 2H), 3.46-3.22 (m, 2H), 2.52-2.34 (m, 2H), 1.75-1.62 (m, 2H), 1.57 (s, 3H), 1.51-1.41 (m, 5H), 1.06 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  180.6, 168.0 (2 × C), 142.0, 141.1, 133.7 (2 × C), 132.0 (2 × C), 131.3, 130.0, 127.9, 125.9, 124.7, 123.0  $(2 \times C)$ , 121.5, 115.9, 109.8, 58.00, 46.0, 34.8, 34.0, 29.4, 25.8, 22.9, 17.7, 16.5. FT-IR (KBr,  $\nu_{\rm max}$ ): 3428, 2923, 2354, 1713, 1563, 1462, 1370 cm $^{-1}$ . HRMS (EI): calcd for  $C_{28}H_{30}N_2O_3$  [M] $^+$  442.2256, found 442.2252.

(E)-2-(2-(2-oxo-3-(3,7,11-trimethyldodeca-1,6,10-trien-3-yl)indolin-3-yl)ethyl)isoindoline-1,3-dione (16r). Following the general procedure A, treatment of *N*-phthalyltryptamine (900 mg, 3.10 mmol) 15r gave product 16r as a single diastereomer (dr > 19:1) as a pale yellow oil in 48% yield (760 mg, 1.49 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.28 (br s, 1H), 7.72–7.66 (m, 2H), 7.66–7.59 (m, 2H), 7.18 (d, J = 7.4 Hz, 1H), 7.03 (td, J = 7.7, 1.2 Hz, 1H), 6.88-6.76 (m, 2H), 5.84 (dd, *J* = 17.5, 10.9 Hz, 1H), 5.24 (dd, *J* = 10.9, 1.1 Hz, 1H), 5.05-4.97 (m, 2H), 4.91-4.87 (m, 1H), 3.46-3.21 (m, 2H), 2.50-2.29 (m, 2H), 2.01-1.97 (m, 2H), 1.86-1.83 (m, 1H), 1.82-1.71 (m, 1H), 1.70-1.60 (m, 4H), 1.54 (s, 3H), 1.46 (s, 3H), 1.43-1.33 (m, 1H), 1.32–1.18 (m, 5H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  180.3,  $168.0 \ (2 \times C)$ , 141.9, 141.5, 134.9,  $133.8 \ (2 \times C)$ ,  $132.1 \ (2 \times C)$ , 131.4, 130.0, 127.9, 125.8, 124.5, 124.4, 123.1 (2  $\times$  C), 121.7, 116.4, 109.9, 57.7, 45.8, 39.7, 34.7, 34.4, 29.5, 26.7, 25.8, 23.0, 17.8, 16.1, 15.6. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3358, 2974, 2891, 2362, 1777, 1714, 1624, 1563, 1479, 1406, 1370, 1047 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>33</sub>H<sub>38</sub>N<sub>2</sub>O<sub>3</sub> [M]<sup>+</sup> 510.2882, found 510.2880.

Methyl-2-(1,3-Dioxoisoindolin-2-yl)-3-(3-(2-methylbut-3-en-2-yl)-2-oxoindolin-3-yl)propanoate (16s). Following the general procedure A, treatment of N-phthalyl-L-tryptophan carboxymethyl ester (12.5 g, 0.035 mol) 15s gave product 16s as a pale yellow solid (mp 120 °C) in 80% yield (12.4 g, 0.029 mol) as a 1:1 mixture of two diastereomers. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.34 (br s, 1H), 7.85–7.79 (m, 2H), 7.72 (br s, 1H), 7.67-7.61 (m, 2H), 7.60-7.51 (m, 4H), 7.33-7.24 (m, 2H), 7.09 (td, J = 7.6, 1.2 Hz, 1H), 6.87 (dd, J = 16.5, 7.6 Hz, 2H), 6.48 (d, J = 4.0 Hz, 2H), 6.26 (dt, J = 7.4, 4.4 Hz, 1H), 6.07 (t, J = 11.2Hz, 1H), 6.02 (t, J = 11.2 Hz, 1H), 5.13 (dd, J = 7.4, 1.1 Hz, 1H), 5.10(dd, *J* = 7.4, 1.1 Hz, 1H), 5.03 (dd, *J* = 15.1, 1.1 Hz, 1H), 4.98 (dd, *J* = 15.1, 1.1 Hz, 1H), 4.79 (dd, *J* = 10.3, 4.6 Hz, 1H), 4.25 (dd, *J* = 11.1, 2.1 Hz, 1H), 3.68 (s, 3H), 3.63 (s, 3H), 3.21 (ABX, J = 14.8, 11.1 Hz, 1H), 3.05-2.95 (m, 2H), 2.90 (ABX, J = 14.8, 2.2 Hz, 1H), 1.15 (s, 3H), 1.12 (s, 3H), 1.03 (s, 3H), 0.99 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.9, 179.6, 169.8, 169.4, 167.4 (2 × C), 167.0 (2 × C), 142.9, 142.7, 142.1, 141.2, 134.1 (2  $\times$  C), 133.7 (2  $\times$  C), 131.8 (2  $\times$ C), 131.6 (2  $\times$  C), 130.1, 129.1, 128.7, 127.1, 125.9, 124.9, 123.6 (2  $\times$ C), 122.9 (2 × C), 122.0, 121.2, 114.4, 114.4, 110.0, 109.7, 56.9, 56.7,  $53.1 (2 \times C)$ , 49.5, 49.1, 42.7, 42.4, 30.3, 28.9, 22.5, 22.1, 21.9, 21.5. FT-IR (KBr,  $\nu_{\text{max}}$ ): 2923, 1737, 1713, 1620, 1462, 1370 cm<sup>-1</sup>. HRMS (EI): calcd for  $C_{25}H_{24}N_2O_5$  [M]<sup>+</sup> 432.1685, found 432.1666. HPLC: t<sub>R</sub> for single diastereomer 10.8 min (Chiralpak AS column, 30% <sup>i</sup>PrOH/Hex). The assignments for proton NMR data are provided in Table S1 (Supporting Information).

C4-Prenylated Products of Cope Rearrangement. 3-(4-(3-Methylbut-2-en-1-yl)-2-oxoindolin-3-yl)propanenitrile (17a). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (53 mg, 0.21 mmol) 16a gave product 17a in 31% yield (16.4 mg, 0.06 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (7.5 mg, 0.029 mmol) 16a gave product 17a in 58% yield (4.4 mg, 0.017 mmol) as a colorless solid (mp 120 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.63 (br s, 1H), 7.20 (t, J = 7.8 Hz, 1H), 6.89 (d, J = 7.8 Hz, 1H), 6.77 (d, J = 7.3 Hz, 1H), 5.21 (dddd, J = 6.9, 5.5, 2.8, 1.4 Hz, 1H), 3.62 (dd, I = 8.2, 3.1 Hz, 1H), 3.42–3.23 (m, 2H), 2.59–2.43 (m, 2H), 2.41-2.31 (m, 1H), 2.25-2.14 (m, 1H), 1.77 (d, J = 1.3 Hz,3H), 1.75 (d, I = 0.8 Hz, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  178.9, 141.6, 138.9, 134.1, 129.0, 124.8, 123.8, 121.4, 119.1, 108.1, 44.1, 31.5, 25.8, 25.6, 18.2, 13.4. FT-IR (KBr,  $\nu_{\rm max}$ ): 3275, 2956, 1620, 1563, 1458 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>16</sub>H<sub>18</sub>N<sub>2</sub>O [M]<sup>+</sup> 254.1419, found

2-(2-(4-(3-Methylbut-2-en-1-yl)-2-oxoindolin-3-yl)ethyl)isoindoline-1,3-dione (17b). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (500 mg, 1.34 mmol) 16b gave product 17b in 74% yield (370 mg, 0.99 mmol) based on recovered starting material. Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (30 mg, 0.08 mmol) **16b** gave product **17b** in 75% yield (22.5 mg, 0.06 mmol) as a yellow solid (mp 184 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.97 (br s, 1H), 7.76–7.70 (m, 2H), 7.64–7.56 (m, 2H), 7.02 (t, I = 7.8 Hz, 1H), 6.7 (d, I = 8.1 Hz, 1H), 6.7 (d, I = 8.0 Hz, 1H), 5.21 (t, J = 7.0 Hz, 1H), 3.71 (ABX, J = 13.5, 6.7 Hz, 1H), 3.66 (ABX, J = 13.5, 7.0 Hz, 1H), 3.61 (t, J = 5.0 Hz, 1H), 3.41 (ABX, J = 1.016.0, 7.4 Hz, 1H), 3.29 (ABX, I = 16.0, 6.8 Hz, 1H), 2.55 (AB, I = 6.8Hz, 1H), 2.51 (AB, J = 6.8 Hz, 1H), 1.72 (s, 3H), 1.71 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.6, 168.1 (2 × C), 142.0, 138.5, 133.8  $(2 \times C)$ , 133.7, 132.1  $(2 \times C)$ , 128.2, 125.7, 123.1  $(3 \times C)$ , 121.5, 108.1, 44.0, 34.7, 31.4, 27.2, 25.8, 18.1. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3275, 2923, 1713, 1620, 1462, 1370, 752 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>23</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub> [M]+ 374.1630, found 374.1633.

N,4-Dimethyl-N-(2-(4-(3-methylbut-2-en-1-yl)-2-oxoindolin-3-yl)ethyl)benzenesulfonamide (17c). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (100 mg, 0.24 mmol) 16c gave product 17c in 57% yield (57 mg, 0.14 mmol) based on recovered starting material. Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (7.0 mg, 0.017 mmol) 16c gave product 17c in 40% yield (2.8 mg, 0.007 mmol) as a pale yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.94 (br s, 1H), 7.63-7.49 (m, 2H), 7.28 7.23 (m, 2H), 7.17 (t, J = 7.7 Hz, 1H), 6.88 (d, J = 7.5 Hz, 1H), 6.72 (d, J = 7.6 Hz, 1H), 5.29–5.21 (m, 1H), 3.59 (dd, J = 7.7, 3.2 Hz, 1H), 3.44 3.28 (m, 2H), 3.08–2.90 (m, 2H), 2.70 (s, 3H), 2.50-2.40 (m, 1H), 2.38 (s, 3H), 2.21-2.06 (m, 1H), 1.74 (d, J = 1.2 Hz, 3H), 1.73 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.3, 143.4, 141.6, 138.8, 134.4, 133.9, 129.7 (2 × C), 128.5, 127.6 (2 × C), 125.8, 123.4, 121.6, 107.6, 46.9, 43.3, 35.5, 31.4, 27.8, 25.9, 21.6, 18.2. FT-IR (KBr,  $\nu_{\rm max}$ ): 3275, 2923, 1713, 1620, 1462, 1370, 1148, 752 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>23</sub>H<sub>28</sub>N<sub>2</sub>O<sub>3</sub>S [M]+ 412.1821, found 412.1826.

Ethyl 3-(4-(3-Methylbut-2-en-1-yl)-2-oxoindolin-3-yl)propanoate (17d). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (7.0 mg, 0.023 mmol) 16d gave product 17d in 52% yield (3.6 mg, 0.012 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (6.0 mg, 0.02 mmol) 16d gave product 17d in 42% yield (2.5 mg, 0.008 mmol) as a pale yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$ 8.52 (br s, 1H), 7.17-7.11 (m, 1H), 6.85 (d, J = 7.2 Hz, 1H), 6.74 (d, J = 7.6 Hz, 1H), 5.22 (tdd, J = 5.7, 2.8, 1.4 Hz, 1H), 4.12–3.94 (m, 2H), 3.61 (dd, J = 6.8, 3.4 Hz, 1H), 3.36 (ddd, J = 23.6, 16.1, 7.1 Hz, 2H), 2.52-2.42 (m, 1H), 2.40-2.24 (m, 2H), 2.23-2.14 (m, 1H), 1.73 (d, J = 1.6 Hz, 6H), 1.19 (t, J = 7.1 Hz, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.8, 173.0, 141.7, 138.9, 133.6, 128.5, 126.1, 123.8, 121.7, 107.6, 60.6, 44.7, 31.4, 29.6, 25.8, 24.6, 18.1, 14.3. FT-IR (KBr,  $\nu_{\rm max}$ ): 3275, 2956, 2855, 1713, 1620, 1563, 1458, 1370 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>18</sub>H<sub>23</sub>NO<sub>3</sub> [M]<sup>+</sup> 301.1678, found 301.1677.

4-(3-Methylbut-2-en-1-yl)-3-(3-oxobutyl)indolin-2-one (17e). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (92 mg, 0.34 mmol) 16e gave product 17e in 43% yield (39.6 mg, 0.15 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (10 mg, 0.04 mmol) 16e gave product 17e in 46% yield (4.6 mg, 0.016 mmol) as a colorless solid (mp 123 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.52 (br s, 1H), 7.16 (t, J = 7.8 Hz, 1H), 6.85 (d, J = 7.9 Hz, 1H), 6.72 (d, J = 7.8 Hz, 1H), 5.24–5.16 (m, 1H), 3.56 (dd, J = 7.6, 3.5 Hz, 1H), 3.35 (ddd, J = 24.0, 16.8 Hz, 2H), 2.61–2.48 (m, 1H), 2.46–2.26 (m, 2H), 2.23–2.12 (m, 1H), 2.06 (s, 3H), 1.73 (br s, 6H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.9, 180.1, 141.6, 139.1, 133.6, 128.4, 126.5, 123.4, 121.7, 107.6, 44.5, 38.6, 31.3, 30.2, 25.9, 23.5, 18.1. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3275, 2956, 1713, 1620, 1563, 1458, 1148 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>17</sub>H<sub>21</sub>NO<sub>2</sub> [M]<sup>+</sup> 271.1572, found 271.1571.

3-Methyl-4-(3-methylbut-2-en-1-yl)indolin-2-one (17f). C3-reverse-prenyl methyloxindole 16f was prepared according to published methods.<sup>8</sup> Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (56 mg, 0.26 mmol) 16f gave product 17f in 32% yield (16.6 mg, 0.08 mmol). 12 Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (7.0 mg, 0.023 mmol) 16f gave product 17f in 16% yield (1.1 mg, 0.005 mmol)<sup>12</sup> as a colorless solid (mp 124 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.35 (br s, 1H), 7.14 (t, J = 7.8 Hz, 1H), 6.84 (d, J =7.8 Hz, 1H), 6.75 (d, J = 7.8 Hz, 1H), 5.27 5.21 (m, 1H), 3.47 (q, J =7.6 Hz, 1H), 3.40 (ABX, J = 16.0, 7.4 Hz, 1H), 3.30 (ABX, J = 16.0, 6.6 Hz, 1H), 1.74 (d, J = 1.2 Hz, 3H), 1.73 (br s, 3H), 1.53 (d, J = 7.6 Hz, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  181.2, 141.2, 138.6, 133.4, 129.0, 128.1, 123.2, 121.9, 107.6, 40.8, 31.3, 25.8, 18.1, 15.5. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3162, 3028, 2984, 2915, 1717, 1684, 1620, 1455 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>14</sub>H<sub>17</sub>NO [M]<sup>+</sup> 215.1310, found 215.1314.

*Methyl* 4-(3-Methylbut-2-en-1-yl)-2-oxo-3-(3-oxobutyl)indoline-6-carboxylate (17g). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (94.0 mg, 0.29 mmol) 16g gave product 17g in 43% yield (40.4 mg, 0.12 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (31 mg, 0.09 mmol) 16g gave product 17g in 40% yield (12.4 mg, 0.04 mmol) as a pale yellow solid (mp 90 °C).  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.34 (br s, 1H), 7.58 (d, J = 0.8 Hz, 1H), 7.39 (d, J = 1.4 Hz, 1H), 5.23–5.16 (m, 1H), 3.91 (s, 3H), 3.60 (ddd, J = 8.0, 3.7 Hz, 1H), 3.44 (ddd, J = 18.5, 15.8, 6.9 Hz, 1H), 2.60 (ddd, J = 14.0, 9.8, 6.0 Hz, 1H), 2.50–2.38 (m, 1H), 2.38 2.28 (m, 1H), 2.20–2.08 (m, 1H), 2.08 (s, 3H), 1.74 (d, J = 0.9 Hz, 6H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.6, 179.4, 166.9, 141.8, 139.2, 134.3, 131.8, 130.5, 125.2, 121.1, 108.1, 52.4, 44.4, 38.5, 31.3, 30.2, 25.9, 23.4, 18.2; HRMS (EI): calcd for  $C_{19}H_{23}NO_4$  [M] $^+$  329.1627, found 329.1622.

6-Fluoro-4-(3-methylbut-2-en-1-yl)-3-(3-oxobutyl)indolin-2-one (17h). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (22 mg, 0.076 mmol) 16h gave product 17h in 40% yield (8.8 mg, 0.03 mmol) as a pale oil.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.92 (br s, 1H), 6.56 (dd, J = 10.7, 2.3 Hz, 1H), 6.46 (dd, J = 8.3, 2.3 Hz, 1H), 5.26–5.12 (m, 1H), 3.54 3.48 (m, 1H), 3.40–3.25 (m, 2H), 2.63–2.52 (m, 1H), 2.45–2.28 (m, 2H), 2.17–2.09 (m, 1H), 2.08 (s, 3H), 1.74 (d, J = 1.1 Hz, 3H), 1.71 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.8, 179.8, 164.3, 161.8, 142.3, 140.9, 134.5, 121.8, 120.9, 109.4, 96.2, 43.8, 38.5, 31.2, 25.9, 23.7, 18.1. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3262, 2923, 2253, 1715, 1614, 1450, 1123 cm $^{-1}$ . HRMS (EI): calcd for C<sub>17</sub>H<sub>20</sub>FNO<sub>2</sub> [M] $^+$  289.1478, found 289.1478.

6-Chloro-4-(3-methylbut-2-en-1-yl)-3-(3-oxobutyl)indolin-2-one (17i). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (83 mg, 0.27 mmol) 16i gave product 17i in 45% yield (37.4 mg, 0.12 mmol) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.75 (br s, 1H), 6.84 (d, *J* = 1.8 Hz, 1H), 6.72 (d, *J* = 1.8 Hz, 1H), 5.17 (ddd, *J* = 7.1, 4.2, 1.5 Hz, 1H), 3.52 (dd, *J* = 7.9, 3.5 Hz, 1H), 3.4 3.24 (m, 2H), 2.64–2.53 (m, 1H), 2.46–2.28 (m, 2H), 2.16–2.06 (m, 4H), 1.74 (d, *J* = 1.3 Hz, 3H), 1.72 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.7, 179.3, 142.3, 140.6, 134.5, 133.9, 124.9, 123.2, 120.8, 108.0, 43.9, 38.5, 31.1, 30.2, 25.9, 23.6, 18.2. HRMS (EI): calcd for C<sub>17</sub>H<sub>20</sub>ClNO<sub>2</sub> [M]<sup>+</sup> 305.1183, found 305.1182.

7-Methyl-4-(3-methylbut-2-en-1-yl)-3-(3-oxobutyl)indolin-2-one (17j). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (100 mg, 0.35 mmol) 16j gave product 17j in 60% yield (60 mg, 0.21 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (8.0 mg, 0.028 mmol) 16j gave product 17j in 50% yield (4.0 mg, 0.014 mmol) as a pale oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.95 (br s, 1H), 6.98 (d, J = 7.9 Hz, 1H), 6.77 (d, J = 7.9 Hz, 1H), 5.24-5.14 (m, 1H), 3.57 (dd, J = 7.8, 3.4 Hz, 1H), 3.31 (ddd, J = 22.0, 15.9, 5.6 Hz, 2H), 2.60-2.50 (m, 1H), 2.37 (dddd, I = 30.0, 17.1, 9.5, 4.0 Hz, 2H), 2.22 (s, 3H), 2.20-2.10 (m, 1H), 2.06 (s, 3H), 1.72 (d, J = 1.3 Hz, 6H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  207.9, 180.4, 140.3, 136.2, 133.4, 129.8, 126.1, 123.3, 122.0, 116.6, 44.8, 38.6, 31.1, 30.2, 25.9, 23.5, 18.1, 16.4. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3162, 3049, 2972, 2915, 2851, 1692, 1630, 1595, 1426 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>18</sub>H<sub>23</sub>NO<sub>2</sub> [M]<sup>+</sup> 285.1729, found 285.1727.

5-Methoxy-4-(3-methylbut-2-en-1-yl)-3-(3-oxobutyl)indolin-2-one (17k). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (62 mg, 0.21 mmol) 16k gave product 17k in 45% yield (28 mg, 0.09 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (15 mg, 0.05 mmol) 16k gave product 17k in 48% yield (7.2 mg, 0.023 mmol) as a pale yellow solid (mp 140 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.99 (s, 1H), 6.72 (d, J = 8.4 Hz, 1H), 6.66 (d, J = 8.3 Hz, 1H), 5.10 5.02 (m, 1H), 3.79 (s, 3H), 3.57 (dd, J = 7.3, 3.5 Hz, 1H), 3.38 3.29 (m, 2H), 2.55–2.34 (m, 2H), 2.32–2.22 (m, 1H), 2.21–2.10 (m, 1H), 2.05 (s, 3H), 1.75 (s, 3H), 1.66 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 207.8, 179.6, 153.9, 134.8, 132.7, 128.7, 128.6, 121.7, 109.9, 107.2, 56.2, 44.9, 38.6, 30.1, 26.0, 25.9, 23.8, 18.1. FT-IR (KBr,  $\nu_{\rm max}$ ): 3275, 2933, 1737, 1620, 1563, 1491, 1370, 1244 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>18</sub>H<sub>23</sub>NO<sub>3</sub> [M]<sup>+</sup> 301.1678, found 301.1674.

(Z)-2-(2-(4-(3,7-Dimethylocta-2,6-dien-1-yl)-2-oxoindolin-3-yl)ethyl)isoindoline-1,3-dione (17p). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (60 mg, 0.27 mmol) 16p gave product 17p in 55% yield (33 mg, 0.075 mmol) as a mixture of E and Z isomers (E:Z = 17:83). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (50 mg, 0.113 mmol) 16p gave product 17p in 50% yield (25 mg, 0.056 mmol) as a mixture of E and Z isomers (E:Z = 23:77) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.45 (br s, 1H), 7.78-7.70 (m, 2H), 7.67 - 7.59 (m, 2H), 7.01 (t, J = 7.8 Hz, 1H), 6.74 - 6.63(m, 2H), 5.26 5.20 (m, 1H), 5.16-5.02 (m, 1H), 3.76-3.58 (m, 3H), 3.47-3.26 (m, 2H), 2.61-2.45 (m, 2H), 2.20-1.98 (m, 4H), 1.72 (m, 3H), 1.67 (s, 3H), 1.59 (m, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$ 179.2, 168.1 (2  $\times$  C), 141.8, 138.7, 137.6, 133.8 (2  $\times$  C), 132.1 (2  $\times$ C), 131.9, 128.3, 125.7, 124.2, 123.3, 123.2  $(2 \times C)$ , 121.9, 107.9, 43.9, 34.7, 32.1, 30.9, 27.3, 26.6, 25.9, 23.5, 17.8. FT-IR (KBr,  $\nu_{\rm max}$ ): 2919, 1769, 1716, 1616, 1450, 1398, 718 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>28</sub>H<sub>30</sub>N<sub>2</sub>O<sub>3</sub> [M]<sup>+</sup> 442.2256, found 442.2258.

(Z)-2-(2-(4-(3,7-Dimethylocta-2,6-dien-1-yl)-2-oxoindolin-3-yl)ethyl)isoindoline-1,3-dione (17q). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (140 mg, 0.32 mmol) 16q gave product 17q in 46% yield (64.4 mg, 0.145 mmol) as a mixture of E and Z isomers (E:Z = 83:17). Following general procedure C, under microwave conditions, C3reverse prenyl oxindole (24 mg, 0.054 mmol) 16q gave product 17q in 62% yield (14.9 mg. 0.034 mmol) as a mixture of E and Z isomers (E:Z = 77:23) as a colorless oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.90 (br s, 1H), 7.76-7.70 (m, 2H), 7.64-7.57 (m, 2H), 7.01 (t, J = 7.8 Hz, 1H), 6.76-6.60 (m, 2H), 5.27-5.18 (m, 1H), 5.16-5.00 (m, 1H), 3.77-3.57 (m, 3H), 3.48-3.26 (m, 2H), 2.58-2.46 (m, 2H), 2.17-1.99 (m, 4H), 1.74-1.69 (m, 3H), 1.66 (d, J = 1.0 Hz, 3H), 1.59 (m, J = 1.0 Hz, J = 1.0 Hz3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.6, 168.0 (2 × C), 141.9, 138.5, 137.4, 133.8 (2  $\times$  C), 132.1 (2  $\times$  C), 131.7, 128.2, 125.7, 124.2, 123.1 (2  $\times$  C), 121.4, 108.1, 44.0, 39.7, 34.7, 31.3, 27.1, 26.6, 25.8, 17.8, 16.4. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3246, 2968, 2923, 2851, 2366, 2334, 1773, 1733, 1705, 1616, 1596, 1453, 1400, 1113, 1018, 721 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>28</sub>H<sub>30</sub>N<sub>2</sub>O<sub>3</sub> [M]<sup>+</sup> 442.2256, found 442.2251.

2-(2-(2-Oxo-4-((2Z,6E)-3,7,11-trimethyldodeca-2,6,10-trien-1-yl)-indolin-3-yl)ethyl)isoindoline-1,3-dione (17r). Following general

procedure C, under microwave conditions, C3-reverse prenyl oxindole (75 mg, 0.147 mmol) **16r** gave product **17r** in 52% yield (39 mg, 0.076 mmol) as a colorless oil.  $^{1}$ H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.77–7.70 (m, 3H), 7.65 7.62 (m, 2H), 7.00 (t, J = 7.8 Hz, 1H), 6.72–6.61 (m, 2H), 5.26 5.04 (m, 3H), 3.78–3.58 (m, 3H), 3.47–3.26 (m, 2H), 2.63–2.45 (m, 2H), 2.21–1.91 (m, 8H), 1.75–1.70 (m, 3H), 1.67 (s, 3H), 1.61–1.56 (m, 6H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.2, 168.0 (2 × C), 141.8, 138.7, 137.7, 135.6, 133.8 (2 × C), 132.1 (2 × C), 131.5, 128.3, 125.7, 124.5, 124.0, 123.3, 123.2 (2 × C), 121.9, 107.9, 43.9, 39.9, 34.7, 32.2, 31.0, 27.3, 26.9, 26.5, 25.9, 23.6, 17.9, 16.2. FT-IR (KBr,  $\nu_{\rm max}$ ): 3258, 2921, 2851, 1773, 1716, 1620, 1596, 1452, 1399, 1107, 718.8 cm $^{-1}$ . HRMS (EI): calcd for C $_{33}$ H $_{38}$ N $_{2}$ O $_{3}$ [M] $^{+}$  510.2882, found 510.2885.

Methyl 2-(1,3-Dioxoisoindolin-2-yl)-3-(4-(3-methylbut-2-en-1-yl)-2-oxoindolin-3-yl)propanoate (17s). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (1.0 g, 2.31 mmol) 16s gave product 17s in 61% yield (610 mg, 1.41 mmol). Following general procedure C, under microwave conditions, C3-reverse prenyl oxindole (50.0 mg, 0.12 mmol) 16s gave product 17s in 54% yield (27.0 mg, 0.06 mmol) as a 1:1 mixture of diastereomers as a colorless solid (mp 220 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.89 (br s, 1H), 8.68 (br s, 1H), 7.72 (dd, I = 5.5, 3.0 Hz, 2H), 7.67-7.58 (m, 4H), 7.47 (dd, J = 5.5, 3.0 Hz, 2H), 7.20 (t, J = 7.7Hz, 1H), 6.89 (d, J = 7.8 Hz, 1H), 6.67 (d, J = 7.7 Hz, 1H), 6.61 (t, J =7.8 Hz, 1H), 6.41 (d, I = 7.7 Hz, 1H), 6.38 (d, I = 7.9 Hz, 1H), 5.24 (t, J = 6.9 Hz, 1H), 5.12 (t, J = 6.9 Hz, 1H), 5.05 (dd, J = 9.4, 5.6 Hz, 1H), 4.79 (dd, J = 11.8, 2.5 Hz, 1H), 3.75-3.70 (m, 1H), 3.71 (s, 3H), 3.69-3.65 (m, 1H), 3.68 (s, 3H), 3.58 (dd, I = 7.6, 2.8 Hz, 1H), 3.43(ABX, J = 15.7, 6.8 Hz, 1H), 3.39–3.27 (m, 4H), 3.22 (ABX, J = 16.0, 6.8 Hz, 2H), 1.74 (s, 6H), 1.73 (s, 6H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  179.3, 178.7, 169.5, 169.3 (2 × C), 167.2, 167.1 (2 × C), 142.1, 141.5, 138.8, 138.3, 134.0, 133.9 (2  $\times$  C), 133.8 (2  $\times$  C), 131.9  $(2 \times C)$ , 131.7  $(2 \times C)$ , 128.9, 127. 7  $(2 \times C)$ , 125.7, 124.8, 123.5  $(2 \times C)$  $\times$  C), 123.2, 123.2 (2  $\times$  C), 123.1, 121.4 (2  $\times$  C), 108.4, 108.0, 53.1, 53.0, 48.9, 48.7, 43.4, 43.3, 31.5, 31.2, 27.8, 27.5, 25.9, 25.8, 18.1, 18.1. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3428, 2923, 1713, 1620, 1563, 1462, 1370, 752 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>25</sub>H<sub>24</sub>N<sub>2</sub>O<sub>5</sub> [M]<sup>+</sup> 432.1685, found 432.1684. HPLC:  $t_R$  for each diastereomer 8.8 min (major) and 12.9 min (minor) (Chiralpak AS column, 30% PrOH/Hex). A detailed assignment of proton NMR signals is provided in Table S2 (Supporting Information).

*N-Methyl-N-(2-(4-(3-Methylbut-2-en-1-yl)-2-oxoindolin-3-yl)-ethyl)-4-nitrobenzenesulfonamide* (17t). Following general procedure B, under thermal rearrangement conditions, C3-reverse prenyl oxindole (100 mg, 0.225 mmol) 16t gave product 17t in 50% yield (50 mg, 0.112 mmol)<sup>12</sup> as a pale yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.38–8.28 (m, 2H), 7.94 (s, 1H), 7.90–7.84 (m, 2H), 7.18 (t, J = 7.8 Hz, 1H), 6.89 (d, J = 7.8 Hz, 1H), 6.73 (d, J = 7.7 Hz, 1H), 5.29–5.21 (m, 1H), 3.60 (dd, J = 7.7, 3.2 Hz, 1H), 3.44 3.28 (m, 2H), 3.16–2.97 (m, 2H), 2.78 (s, 3H), 2.53–2.38 (m, 1H), 2.23–2.11 (m, 1H), 1.76 (d, J = 1.1 Hz, 3H), 1.74 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 179.02, 150.1, 143.5, 141.5, 138.9, 134.0, 128.7, 128.6 (2 × C), 125.5, 124.5 (2 × C), 123.6, 121.4, 107.8, 46.9, 43.2, 35.4, 31.4, 27.7, 25.9, 18.2. FT-IR (KBr,  $\nu_{max}$ ): 3444, 2923, 1707, 1596, 1531, 1455, 1350, 1163, 738 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>22</sub>H<sub>25</sub>N<sub>3</sub>O<sub>5</sub>S [M]<sup>+</sup> 443.1515, found 443.1511.

*C3-n-Prenylated Products of [1,3]-Prenyl Transfer Reaction. 2-(2-(3-(3-Methylbut-2-en-1-yl)-2-oxoindolin-3-yl)ethyl)isoindoline-1,3-dione (18b).* Following general procedure D, under thermal rearrangement conditions in decalin, C3-reverse prenyl oxindole (50 mg, 0.13 mmol) **16b** gave product **18b** in 65% yield (32.5 mg, 0.087 mmol) as a colorless solid (mp 167 °C). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.72–7.66 (m, 2H), 7.65–7.59 (m, 2H), 7.54 (s, 1H), 7.11–7.05 (m, 1H), 6.98–6.91 (m, 1H), 6.79–6.71 (m, 2H), 4.88–4.80 (m, 1H), 3.68–3.43 (m, 2H), 2.53–2.41 (m, 3H), 2.33–2.24 (m, 1H), 1.56 (s, 3H), 1.47 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 180.7, 168.0 (2 × C), 140.7, 136.0, 133.8 (2 × C), 132.1 (2 × C), 131.6, 127.7, 123.4, 123.1 (2 × C), 122.3, 117.1, 109.8, 52.0, 37.4, 34.5, 33.6, 26.0, 18.1. HRMS (EI): calcd for C<sub>23</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub> [M]<sup>+</sup> 374.1630, found 374.1633.

3-Methyl-3-(3-methylbut-2-en-1-yl)indolin-2-one (18f). Following general procedure D, under thermal rearrangement conditions in decalin, C3-reverse prenyl oxindole (56 mg, 0.26 mmol) 16f gave product 18f in 98% yield (55 mg, 0.25 mmol) as a pale yellow oil.  $^1$ H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.19 (s, 1H), 7.22–7.14 (m, 2H), 7.02 (td, J = 7.5, 1.0 Hz, 1H), 6.91–6.84 (m, 1H), 4.92–4.82 (m, 1H), 2.49 (qd, J = 14.2, 7.5 Hz, 2H), 1.58 (d, J = 1.0 Hz, 3H), 1.52 (s, 3H), 1.39 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>): δ 182.8, 140.2, 135.4, 134.7, 127.7, 123.5, 122.4, 118.2, 109.5, 49.0, 36.8, 26.0, 22.8, 18.1. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3210, 2916, 2850, 1710, 1620, 1463, 1377 cm $^{-1}$ . HRMS (EI): calcd for  $\text{C}_{14}\text{H}_{17}\text{NO}$  [M] $^+$  215.1310, found 215.1308.

*6-Fluoro-3-(3-methylbut-2-en-1-yl)-3-(3-oxobutyl)indolin-2-one* (*18h*). In an attempt to obtain the [3,3]-rearrangement product 17h following general procedure B, under thermal rearrangement conditions in DMAc, C3-reverse prenyl oxindole (22 mg, 0.076 mmol) 16h gave product 18h (as a minor product) in 20% yield as a colorless oil. This product was verified as the [1,3]-rearrangement product through spectral analysis as follows. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): *δ* 7.81 (s, 1H), 7.06 (dd, J = 8.2, 5.3 Hz, 1H), 6.78–6.69 (m, 1H), 6.63 (dd, J = 8.7, 2.3 Hz, 1H), 4.86 4.78 (m, 1H), 2.50 (ddd, J = 20.9, 14.1, 7.6 Hz, 2H), 2.34–2.23 (m, 1H), 2.15 (dd, J = 8.9, 6.7 Hz, 2H), 2.03–1.91 (m, 4H), 1.58 (s, 3H), 1.50 (s, 3H). HRMS (EI): calcd for  $C_{17}H_{20}FNO_2$  [M] $^+$  289.1478, found 289.1473.

(*E*)-2-(2-(3-Cinnamyl-2-oxoindolin-3-yl)ethyl)isoindoline-1,3-dione (18l). Following general procedure B, under thermal rearrangement conditions in DMAc, C3-reverse prenyl oxindole (45 mg, 0.107 mmol) 16l gave product 18l in 46% yield (20.7 mg, 0.05 mmol) as a colorless oil as the *only* product, and no [3,3]-rearrangement was observed. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.81 (s, 1H), 7.71–7.65 (m, 2H), 7.64–7.60 (m, 2H), 7.24–7.11 (m, 6H), 6.97 (td, J = 7.7, 1.2 Hz, 1H), 6.83–6.74 (m, 2H), 6.33 (d, J = 15.8 Hz, 1H), 5.86 (dt, J = 15.5, 7.6 Hz, 1H), 3.57 (tdd, J = 14.0, 10.7, 6.2 Hz, 2H), 2.71–2.60 (m, 2H), 2.54–2.44 (m, 1H), 2.40–2.31(m, 1H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 180.3, 168.0 (2 × C), 140.7, 137.2, 134.4, 133.8 (2 × C), 132.0 (2 × C), 131.1, 128.5 (2 × C), 128.0, 127.4, 126.4 (2 × C), 123.4, 123.1 (2 × C), 122.5, 110.1, 52.2, 42.3, 34.4, 33.7; FTIR (KBr,  $\nu_{\text{max}}$ ): 3432, 2253, 1713, 1620, 1470, 911.7 cm<sup>-1</sup>. HRMS (EI): calcd for  $C_{27}H_{22}N_2O_3$  [M]<sup>+</sup> 422.1630, found 422.1627.

C3-n-Prenylated Products of [1,3]-Prenyl Transfer Reaction on C3-Hydroxy Substrates. 5-Bromo-3-hydroxy-3-(2-methylbut-3-en-2yl)indolin-2-one (21b). Prenyl bromide (400  $\mu$ L, 3.55 mmol) was added to a magnetically stirred solution of 5-bromoisatin (400 mg, 1.78 mmol), indium powder (255 mg, 2.22 mmol), and sodium iodide (531 mg, 3.55 mmol) in DMF (5 mL) at 0 °C. The resulting mixture was stirred at the same temperature until complete disappearance of starting material (TLC). Saturated aqueous sodium hydrogen carbonate (2.5 mL) was added, and the mixture was warmed to room temperature before being extracted with ethyl acetate (3 × 10 mL). The organic extract was washed with brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated under reduced pressure. The resulting mass was subjected to silica gel chromatography (17% ethyl acetate in hexanes) to give the purified product 21b in 57% isolated yield as a pale oil. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.41 (dd, J = 2.0, 0.4 Hz, 1H), 7.37 (dd, J = 8.2, 2.0 Hz, 1H), 6.75 (dd, J = 8.2, 0.4 Hz, 1H), 6.09(dd, *J* = 17.6, 10.9 Hz, 1H), 5.06 (dd, *J* = 10.9, 1.4 Hz, 1H), 4.95 (dd, *J* = 17.6, 1.4 Hz, 1H), 1.22 (s, 3H), 1.03 (s, 3H). <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  181.3, 143.5, 142.5, 134.4, 133.0, 130.1, 114.9, 114.2, 112.2, 81.4, 44.2, 21.9, 20.6. FT-IR (KBr,  $\nu_{\rm max}$ ): 3347, 3230, 2988, 1702, 1616, 1475, 1442, 1313, 1170 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>13</sub>H<sub>14</sub>BrNO<sub>2</sub> [M]<sup>+</sup> 295.0208, found 295.0214.

5-Chloro-3-hydroxy-3-(2-methylbut-3-en-2-yl)indolin-2-one (21c). Colorless solid (mp 230 °C), 60%. <sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ): δ 10.31 (s, 1H), 7.20 (dd, J = 8.2, 2.2 Hz, 1H), 7.13 (d, J = 2.2 Hz, 1H), 6.75 (d, J = 8.2 Hz, 1H), 5.98 (s, 1H, –OH), 5.98 (dd, J = 17.6, 11.2 Hz, 1H), 4.99 (dd, J = 10.8, 1.4 Hz, 1H), 4.86 (dd, J = 17.6, 1.5 Hz, 1H), 1.14 (s, 3H), 0.93 (s, 3H). <sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ): δ 178.4, 142.5, 141.1, 133.1, 128.5, 125.6, 124.6, 113.1, 110.4, 79.1, 42.7, 21.2, 19.8. FT-IR (KBr,  $\nu_{\rm max}$ ): 3347, 3230, 2988, 1702, 1616, 1475, 1442, 1313, 1170 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>13</sub>H<sub>14</sub>ClNO<sub>2</sub> [M]<sup>+</sup> 251.0713, found 251.0714.

3-Hydroxy-3-(3-methylbut-2-en-1-yl)indolin-2-one (22a). Precursor 21a was synthesized according to published methods.<sup>6</sup> Following general procedure D, under thermal rearrangement conditions in decalin, C3-reverse prenyl oxindole (78 mg, 0.36 mmol) 21a gave product 22a in 50% yield (39 mg, 0.18 mmol) as a colorless oil. Microwave irradiation (at the same power levels as in procedure C) of C3-reverse prenyl oxindole (6.0 mg, 0.03 mmol) 21a in phosphate buffer provided product 22a in 89% yield (5.3 mg, 0.025 mmol). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.62 (s, 1H), 7.39–7.34 (m, 1H), 7.28– 7.23 (m, 1H), 7.07 (td, J = 7.5, 0.9 Hz, 1H), 6.86 (d, J = 7.8 Hz, 1H), 5.05-5.01 (m, 1H), 2.85 (s, 1H, exchangeable with  $D_2O$ ), 2.65 (d, J =7.6 Hz, 2H), 1.67 (s, 3H), 1.56 (s, 3H). <sup>13</sup>C NMR (150 MHz,  $CDCl_3$ ):  $\delta$  179.8, 140.3, 137.7, 130.6, 129.7, 124.6, 123.1, 115.7, 110.1, 76.5, 37.5, 26.1, 18.1. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3238, 2919, 1705, 1616, 1475, 1115 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>13</sub>H<sub>15</sub>NO<sub>2</sub> [M]<sup>+</sup> 217.1103, found 217.1106.

5-Bromo-3-hydroxy-3-(3-methylbut-2-en-1-yl)indolin-2-one (22b). Precursor 21b was synthesized according to published methods. Following general procedure B/D, under thermal rearrangement conditions in DMAc or decalin, C3-reverse prenyl oxindole (28 mg, 0.095 mmol) 21b gave product 22b in 50% yield (14 mg, 0.047 mmol) as a colorless oil. Microwave irradiation (at the same power levels as in procedure C) of C3-reverse prenyl oxindole (15.0 mg, 0.05 mmol) 21b in phosphate buffer provided product 22b in 75% yield (11.3 mg, 0.038 mmol). H NMR (400 MHz, CD<sub>3</sub>OD): δ 7.44–7.42 (m, 1H), 7.38 (dd, J = 8.2, 2.1 Hz, 1H), 6.80 6.76 (m, 1H), 4.87–4.80 (m, 1H), 2.61 (d, J = 7.6 Hz, 2H), 1.59 (d, J = 1.0 Hz, 3H), 1.49 (d, J = 0.9 Hz, 3H). NMR (100 MHz, DMSO- $d_6$ ): δ 178.6, 140.9, 134.8, 134.3, 131.4, 126.9, 116.8, 113.2, 111.4, 75.7, 36.3, 25.7, 17.8. HRMS (EI): calcd for C<sub>13</sub>H<sub>14</sub>BrNO<sub>2</sub> [M]+ 295.0208, found 295.0207.

5-Chloro-3-hydroxy-3-(3-methylbut-2-en-1-yl)indolin-2-one (22c). Precursor 21c was synthesized according to published methods.<sup>6</sup> Following general procedure B/D, under thermal rearrangement conditions in DMAc or decalin, C3-reverse prenyl oxindole (33 mg, 0.13 mmol) 21c gave product 22c in 41% yield (13.5 mg, 0.053 mmol) as colorless oil as the *only* product, and no [3,3]-rearrangement was observed. Microwave irradiation (at the same power levels as in procedure C) of C3-reverse prenyl oxindole (9.0 mg, 0.035 mmol) 21c in phosphate buffer provided product 22a in 87% yield (7.8 mg, 0.03 mmol). <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  7.30 (d, J = 2.2 Hz, 1H), 7.24 (dd, J = 8.3, 2.2 Hz, 1H), 6.83 (d, J = 8.3 Hz, 1H), 4.88–4.77 (m, 1H), 2.61 (d, J = 7.6 Hz, 2H), 1.58 (s, 3H), 1.49 (s, 3H);  $^{13}$ C NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  181.7, 141.5, 137.7, 134.8, 130.1, 128.7, 125.6, 117.2, 112.1, 77.9, 37.6, 26.0, 18.0. FT-IR (KBr,  $\nu_{\rm max}$ ): 3358, 2493, 2245, 2209, 2072, 1123, 976 cm<sup>-1</sup>. HRMS (EI): calcd for C<sub>13</sub>H<sub>14</sub>ClNO<sub>2</sub> [M]<sup>+</sup> 251.0713, found 251.0705.

C3-n-Prenyl-<sub>1</sub>-Trp-pyrroloindoline Methyl Ester (13 and 14). Following general procedure E, treatment of 1-trp-methyl ester (400 mg, 1.83 mmol) 11 gave products 13 and 14 (in a 4:1 ratio) as mixture of diastereomers in 67% overall yield (358 mg, 1.25 mmol) (based on recovered starting material) as a pale oil. Data for major diastereomer 13 are as follows. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.06–7.01 (m, 2H), 6.72 (dt, J = 7.4, 0.9 Hz, 1H), 6.56 (d, J = 7.9 Hz, 1H), 5.09 (t, J = 7.3 Hz, 1H), 4.91 (s, 1H), 3.71 (dd, J = 10.5, 5.8 Hz, 1H), 3.70 (s, 3H), 3.42 (br s, 2H), 2.47–2.41 (m, 2H), 2.38 (dd, J = 12.0, 5.8 Hz, 1H), 2.00 (dd, J = 12.0, 10.7 Hz, 1H), 1.68 (s, 3H), 1.55 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  174.4, 150.0, 134.7, 133.1, 128.2, 123.6, 119.6, 118.8, 109.1, 82.2, 59.4, 58.8, 52.2, 44.2, 36.9, 26.0, 18.2. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3366, 3044, 2933, 1823, 1737, 1672, 1605, 1477, 1365, 1225, 1110, 1024, 942, 833, 747 cm<sup>-1</sup>. HRMS (ESI+): calcd for C<sub>17</sub>H<sub>23</sub>O<sub>2</sub>N<sub>2</sub> [M + H]<sup>+</sup> 287.1681, found 287.1683.

N1-Me-C3-n-Prenyl-L-Trp-pyrroloindoline Methyl Ester (28 and 29). Following general procedure E, treatment of N-Me L-Trp-methyl ester 27 (680 mg, 2.93 mmol) gave product 28 in 21% yield and 29 in 11% yield as a diastereomeric mixture as a pale oil along with a 40% recovery of 27 amounting to a 53% overall yield (466 mg, 1.55 mmol). Data for diastereoisomer 28 are as follows. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.08 (dt, J = 7.7, 1.2 Hz, 1H), 7.00 (dd, J = 7.3, 1.2 Hz, 1H), 6.63 (dt, J = 7.4, 0.9 Hz, 1H), 6.34 (d, J = 7.8 Hz, 1H), 5.09 (qt, J = 6.7, 1.3 Hz, 1H), 4.67 (s, 1H), 3.71 (s, 3H), 3.66 (dd, J = 10.3, 6.2

Hz, 1H), 3.08 (br s, 1H), 2.83 (s, 3H), 2.45–2.38 (m, 2H), 2.35 (dd, J = 12.2, 6.2 Hz, 1H), 2.00 (t, I = 11.2 Hz, 1H), 1.68 (s, 3H), 1.56 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  174.6, 151.20, 134.6, 133.4, 128.3, 123.0, 119.7, 117.0, 105.6, 88.8, 59.5, 57.1, 52.2, 44.0, 36.7, 31.7, 26.0, 18.2. FT-IR (KBr,  $\nu_{\text{max}}$ ): 3366, 3044, 2933, 1823, 1737, 1672, 1605, 1477, 1365, 1225, 1110, 1024, 942, 833, 747, 662 cm<sup>-1</sup>. HRMS (EI): calcd for  $C_{18}H_{24}O_2N_2$  [M]<sup>+</sup> 300.1838, found 300.1838. Data for diastereoisomer 29 are as follows. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$ 7.05 (dt, J = 7.6, 1.2 Hz, 1H), 6.98 (dd, J = 7.3, 0.9 Hz, 1H), 6.60 (dt, J = 7.3) = 7.4, 0.8 Hz, 1H), 6.29 (d, J = 7.8 Hz, 1H), 5.13 (qt, J = 6.7, 1.2 Hz, 1H), 4.63 (s, 1H), 3.91 (dd, *J* = 7.8, 3.1 Hz, 1H), 3.50 (br s, 1H), 3.33 (s, 3H), 2.86 (s, 3H), 2.49–2.36 (m, 3H), 2.33 (ABX, J = 12.8, 7.9 Hz, 1H), 1.69 (s, 3H), 1.58 (s, 3H).  $^{13}$ C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$ 174.6, 150.8, 134.5, 133.6, 128.3, 123.2, 119.9, 117.0, 105.6, 88.7, 60.0, 56.0, 52.0, 41.0, 36.2, 31.4, 26.1, 18.2. FT-IR (KBr,  $\nu_{\rm max}$ ): 3366, 3044, 2933, 1823, 1737, 1672, 1605, 1477, 1365, 1225, 1110, 1024, 942, 833, 747, 662 cm<sup>-1</sup>. HRMS (EI): calcd for  $C_{18}H_{24}O_2N_2$  [M]<sup>+</sup>: 300.1838, found 300.1838.

#### ASSOCIATED CONTENT

# **S** Supporting Information

Text, figures, and tables giving general methods and procedures for all known precursors along with their syntheses, NMR spectra (<sup>1</sup>H and <sup>13</sup>C) of all new compounds, and conversion of compounds **16** and **17** as a function of substituent effects. This material is available free of charge via the Internet at http://pubs.acs.org.

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#### Notes

The authors declare no competing financial interest.

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# DEDICATION

 $^\dagger$ Authors dedicate this manuscript to Prof. Jeffrey N. Johnston (VICB) on the occasion of his 2014 Arthur C. Cope Scholar Award.

## REFERENCES

- (1) Li, S. M. Phytochemistry 2009, 70, 1746.
- (2) Gebler, J. C.; Poulter, C. D. Arch. Biochem. Biophys. 1992, 296, 308.
- (3) Gebler, J. C.; Woodside, A. B.; Poulter, C. D. J. Am. Chem. Soc. 1992, 114, 7354.
- (4) Tsai, H. F.; Wang, H.; Gebler, J. C.; Poulter, C. D.; Schardl, C. L. Biochem. Biophys. Res. Commun. 1995, 216, 119.
- (5) Shibuya, M.; Chou, H. M.; Fountoulakis, M.; Hassam, S.; Kim, S. U.; Kobayashi, K.; Otsuka, H.; Rogalska, E.; Cassady, J. M.; Floss, H. G. J. Am. Chem. Soc. 1990, 112, 297.
- (6) Schardl, C. L.; Panaccione, D. G.; Tudzynski, P. Alkaloids: Chem. Biol. 2006, 63, 45.

- (7) Wallwey, C.; Li, S. M. Nat. Prod. Rep. 2011, 28, 496.
- (8) Coyle, C. M.; Cheng, J. Z.; O'Connor, S. E.; Panaccione, D. G. Appl. Environ. Microbiol. 2010, 76, 3898.
- (9) Williams, R. M.; Stocking, E. M.; Sanz-Cervera, J. F. Biosynthesis: Aromatic Polyketides, Isoprenoids, Alkaloids 2000, 209, 97.
- (10) Unsold, I. A.; Li, S. M. ChemBioChem 2006, 7, 158.
- (11) Steffan, N.; Unsold, I. A.; Li, S. M. ChemBioChem 2007, 8, 1298.
- (12) Yin, W. B.; Ruan, H. L.; Westrich, L.; Grundmann, A.; Li, S. M. ChemBioChem 2007, 8, 1154.
- (13) Ding, Y. S.; Williams, R. M.; Sherman, D. H. J. Biol. Chem. 2008, 283, 16068.
- (14) Rigbers, O.; Li, S. M. J. Biol. Chem. 2008, 283, 26859.
- (15) Schneider, P.; Weber, M.; Hoffmeister, D. Fungal Genet. Biol. 2008, 45, 302.
- (16) Zou, H.; Zheng, X.; Li, S. M. J. Nat. Prod. 2009, 72, 44.
- (17) Li, S. M. Appl. Microbiol. Biotechnol. 2009, 84, 631.
- (18) Li, S.-M. Nat. Prod. Rep. 2010, 27, 57.
- (19) Zou, H. X.; Xie, X. L.; Linne, U.; Zheng, X. D.; Li, S. M. Org. Biomol. Chem. **2010**, *8*, 3037.
- (20) Yu, X.; Xie, X.; Li, S. M. Appl. Microbiol. Biotechnol. 2011, 92, 737.
- (21) Wollinsky, B.; Ludwig, L.; Xie, X.; Li, S. M. Org. Biomol. Chem. **2012**, 10, 9262.
- (22) Yu, X.; Liu, Y.; Xie, X.; Zheng, X.-D.; Li, S. M. J. Biol. Chem. **2012**, 287, 1371.
- (23) Yu, X.; Yang, A.; Lin, W.; Li, S. M. Tetrahedron Lett. 2012, 53, 6861.
- (24) Chen, J.; Morita, H.; Kato, R.; Noguchi, H.; Sugio, S.; Abe, I. Acta Crystallogr., Sect. F 2012, 68, 355.
- (25) Chen, J.; Morita, H.; Wakimoto, T.; Mori, T.; Noguchi, H.; Abe, I. Org. Lett. **2012**, *14*, 3080.
- (26) Liebhold, M.; Xie, X.; Li, S. M. Org. Lett. 2012, 14, 4882.
- (27) Mahmoodi, N.; Qian, Q.; Luk, L. Y. P.; Tanner, M. E. Pure Appl. Chem. 2013, 85, 1935.
- (28) Mahmoodi, N.; Tanner, M. E. ChemBioChem 2013, 14, 2029.
- (29) Yin, S.; Yu, X.; Wang, Q.; Liu, X.-Q.; Li, S. M. Appl. Microbiol. Biotechnol. 2013, 97, 1649.
- (30) Tsunematsu, Y.; Ishikawa, N.; Wakana, D.; Goda, Y.; Noguchi, H.; Moriya, H.; Hotta, K.; Watanabe, K. *Nat. Chem. Biol.* **2013**, *9*, 818.
- (31) Edwards, D. J.; Gerwick, W. H. J. Am. Chem. Soc. 2004, 126, 11432.
- (32) Schultz, A. W.; Oh, D. C.; Carney, J. R.; Williamson, R. T.; Udwary, D. W.; Jensen, P. R.; Gould, S. J.; Fenical, W.; Moore, B. S. J. Am. Chem. Soc. 2008, 130, 4507.
- (33) Yin, W. B.; Grundmann, A.; Cheng, J.; Li, S.-M. J. Biol. Chem. **2009**, 284, 100.
- (34) Schultz, A. W.; Lewis, C. A.; Luzung, M. R.; Baran, P. S.; Moore, B. S. *J. Nat. Prod.* **2010**, *73*, 373.
- (35) Qian, Q.; Schultz, A. W.; Moore, B. S.; Tanner, M. E. Biochemistry 2012, 51, 7733.
- (36) McIntosh, J. A.; Lin, Z.; Tianero, M. D. B.; Schmidt, E. W. ACS Chem. Biol. 2013, 8, 877.
- (37) Rudolf, J. D.; Wang, H.; Poulter, C. D. J. Am. Chem. Soc. 2013, 135, 1895.
- (38) Luk, L. Y. P.; Tanner, M. E. J. Am. Chem. Soc. 2009, 131, 13932.
- (39) Luk, L. Y. P.; Qian, Q.; Tanner, M. E. J. Am. Chem. Soc. 2011, 133, 12342.
- (40) Floss, H. G. Tetrahedron 1976, 32, 873.
- (41) Kozikowski, A. P.; Wu, J. P.; Shibuya, M.; Floss, H. G. J. Am. Chem. Soc. 1988, 110, 1970.
- (42) Pachlatko, P.; Tabacik, C.; Acklin, W.; Arigoni, D. Chimia 1975, 29, 526.
- (43) Wenkert, E.; Sliwa, H. Bioorg. Chem. 1977, 6, 443.
- (44) Jackson, A. H.; Smith, A. E. Tetrahedron 1965, 21, 989.
- (45) Jackson, A. H.; Smith, P. Tetrahedron 1968, 24, 2227.
- (46) Biswas, K. M.; Jackson, A. H. Tetrahedron 1969, 25, 227.
- (47) Jackson, A. H.; Lynch, P. P. J. Chem. Soc., Perkin Trans. 2 1987, 1215.

- (48) Schwarzer, D. D.; Gritsch, P. J.; Gaich, T. Angew. Chem., Int. Ed. **2012**, *51*, 11514.
- (49) Schwarzer, D. D.; Gritsch, P. J.; Gaich, T. Synlett 2013, 24, 1025.
- (50) Voute, N.; Philp, D.; Slawin, A. M. Z.; Westwood, N. J. Org. Biomol. Chem. 2010, 8, 442.
- (51) Cope, A. C.; Hardy, E. M. Y. J. Am. Chem. Soc. 1940, 62, 441.
- (52) Booker-Milburn, K. I.; Fedouloff, M.; Paknoham, S. J.; Strachan, J. B.; Melville, J. L.; Voyle, M. *Tetrahedron Lett.* **2000**, *41*, 4657.
- (53) Linton, E. C.; Kozlowski, M. C. J. Am. Chem. Soc. 2008, 130, 16162.
- (54) Ignatenko, V. A.; Zhang, P.; Viswanathan, R. Tetrahedron Lett. 2011, 52, 1269.
- (55) Viswanathan, R. Ph.D. Dissertation, Indiana University, Bloomington, IN, 2005.
- (56) Harwood, L. M. J. Chem. Soc., Chem. Commun. 1983, 530.
- (57) Cairns, N.; Harwood, L. M.; Astles, D. P. J. Chem. Soc., Chem. Commun. 1986, 1264.
- (58) Cairns, N.; Harwood, L. M.; Astles, D. P.; Orr, A. J. Chem. Soc., Chem. Commun. 1986, 182.
- (59) Bocchi, V.; Casnati, G.; Marchelli, R. Tetrahedron 1978, 34, 929.
- (60) Magnuson, R.; Solomon, J.; Grossman, A. D. Cell 1994, 77, 207.
- (61) Okada, M.; Sato, I.; Cho, S. J.; Dubnau, D.; Sakagami, Y. Tetrahedron 2006, 62, 8907.
- (62) Liu, X.; Walsh, C. T. Biochemistry 2009, 48, 11032.
- (63) Jost, M.; Zocher, G.; Tarcz, S.; Matuschek, M.; Xie, X.; Li, S. M.; Stehle, T. *J. Am. Chem. Soc.* **2010**, *132*, 17849.
- (64) Brak, K.; Ellman, J. A. Org. Lett. 2010, 12, 2004.
- (65) Liu, Q.; Li, Q.; Ma, Y.; Jia, Y. Org. Lett. 2013, 15, 4528.