See discussions, stats, and author profiles for this publication at: https://www.researchgate.net/publication/23471520

# N-Acyldihydropyridones as Synthetic Intermediates. A Stereoselective Synthesis of Acyclic Amino Alcohols Containing Multiple Chiral Centers

ARTICLE in THE JOURNAL OF ORGANIC CHEMISTRY · DECEMBER 2008

Impact Factor: 4.72 · DOI: 10.1021/jo802029y · Source: PubMed

CITATIONS	READS
15	17

#### 3 AUTHORS, INCLUDING:



Daniel L Comins
North Carolina State University

**251** PUBLICATIONS **5,916** CITATIONS

SEE PROFILE



J Org Chem. Author manuscript; available in PMC 2009 December 19.

Published in final edited form as:

J Org Chem. 2008 December 19; 73(24): 9744–9751. doi:10.1021/jo802029y.

# N-Acyldihydropyridones as Synthetic Intermediates. A Stereoselective Synthesis of Acyclic Amino Alcohols Containing Multiple Chiral Centers

W. Stephen McCall, Teresa Abad Grillo, and Daniel L. Comins\*
Department of Chemistry, North Carolina State University, Raleigh, North Carolina 27695-8204

## **Abstract**

Various multisubstituted piperidines containing a phenyl group at C-2 can be opened regio- and stereoselectively with cyanogen bromide. The ring-opened products contain useful cyanamide and benzylic bromide functional groups. The benzyl bromide can be cleanly reduced, or substituted with various nucleophiles via an  $S_{\rm N}2$  process to add additional heteroatoms stereoselectively. This methodology is useful for the stereoselective synthesis of uniquely substituted alkylamine derivatives containing multiple chiral centers and various functionality. Diastereomerically pure amino alcohols containing three to five contiguous stereocenters were prepared using this strategy.

## Keywords

Dihydropyridones; piperidines; von Braun reaction; cyanogen bromide; piperidinols; amino alcohols; cyanamides

#### Introduction

Compared to acyclic intermediates, the stereocontrolled introduction of stereocenters into cyclic systems is generally easier to accomplish. This concept has been used often by organic chemists in designing routes to stereoselectively synthesize acyclic synthetic intermediates containing multiple chiral centers. Substituents are introduced into a cyclic building block with stereocontrol and then the ring is opened to yield the open chain intermediate with translation of stereochemistry. Historically, this approach to stereodefined acyclic intermediates has been designed and utilized for the synthesis of a specific target molecule, and few general methodologies have been developed to prepare chiral acyclic intermediates in this manner. This is particularly evident in piperidine chemistry and in the construction of chiral alkylamines. The stereoselective preparation of chiral amino alcohols is of particular importance to medicinal chemistry as they are found as fragments in numerous antiviral drugs, natural products, and various biologically active compounds. The opening of chiral piperidinols appeared to have considerable potential for the facile formation of these valuable functionalized intermediates.

Over the years, several methodologies have been developed in our labs in support of ongoing efforts to synthesize complex alkaloid natural products utilizing the dihydropyridone ring system as a building block.<sup>2</sup> These versatile heterocycles are readily synthesized from 1-acylpyridinium salts of 4-methoxypyridines and a nucleophile. If a chiral chloroformate is used in the formation of the 1-acylpyridinium salt, the dihydropyridones can be obtained in a highly diastereoselective fashion providing ready access to either antipode for synthetic purposes.<sup>3</sup> The functionality and conformational bias present in these heterocycles allow regio and stereoselective substitutions to be carried out at every carbon on the ring. This chemistry is powerful for the stereocontrolled synthesis of piperidines with multiple chiral centers.<sup>2</sup>

We felt that the development of a general piperidine ring-opening reaction would broaden the scope of our dihydropyridone chemistry by allowing chiral acyclic amino alcohol derivatives with 2–5 contiguous or discontinuous stereocenters to be prepared from stereodefined piperidines with control of relative and absolute stereochemistry.

As previously described in our preliminary publication,<sup>4</sup> we discovered that the von Braun tertiary amine cleavage reaction<sup>5</sup> gave both stereo and regiocontrolled ring opening in high yield at the 2-position of substituted 2-phenyl-*N*-methylpiperidines **1** (Scheme 1). Herein we describe the synthesis of various piperidinols containing multiple chiral centers, and their conversion to the corresponding acyclic intermediates. This methodlogy gives rapid entry into chiral acyclic amino alcohol derivatives of the type **2** with functionality that may be manipulated in a myriad of interesting directions depending on the subsequent chemistry that is desired. The work reported herein illustrates that the methodology is general and may be applied to the preparation of various chiral amino alcohols with excellent stereocontrol.

#### **Results and Discussion**

The first system we chose to investigate was a piperidine containing 3 stereocenters. Treatment of commercially available 4-methoxypyridine, phenyl chloroformate, and phenylmagnesium bromide followed by acidic workup gave the desired dihydropyridone 3 in 83% recrystalized yield<sup>6</sup> (Scheme 2). Compound 3 was then subjected to copper-mediated conjugate addition of methyl Grignard to afford preferentially the *cis* substituted piperidone 4.<sup>7</sup> The diasteroselectivity was 5:1 favoring the *cis* isomer which was separated by column chromatography. LAH reduction provided the 4-piperidinol 5 as a single diastereomer. The complete facial selectivity observed during the reduction step can be explained by examining the low-energy conformation of piperidone 4 (Figure 1). Due to A<sup>(1,3)</sup> strain, both the C-2 and C-6 substituents are held in the axial orientation. Facile reduction of the ketone carbonyl with LAH on the less hindered face occurs prior to carbamate reduction. This result is an illustrative example of how the conformational bias of these heterocycles can be effectively used for stereocontrol. The piperidinol was protected to give 6 as either an acetate or a TBS derivative.

The ring-opening reactions of piperidines  $\bf 6$  were carried out under standard von Braun conditions using cyanogen bromide in refluxing  $CH_2Cl_2$ . The only products observed were the targeted acyclic cyanamides  $\bf 7$  as single diastereomers, due to both regio- and stereospecific ring opening of the intermediate cyanoammonium salt. The regioselectivity is a result of activation of the C-2, C-N bond by the phenyl substituent at the C-2 position. Structure elucidation of the precursor  $\bf 6b$  through 2-D NOESY studies gave a tentative assignment of relative stereochemistry. We anticipated that the reaction would proceed through an  $S_N2$  mechanism and assumed as such based on literature precedent; the resulting relative stereochemical assignment for products  $\bf 7$  would be as shown in Scheme 2. Isolation of the von Braun product  $\bf 7b$  as an X-ray quality crystal confirmed the assignment and proved unequivocally that the reaction did in fact proceed through an  $S_N2$  mechanism with inversion of configuration.

The functionalities resulting from the ring-opening von Braun reaction are a benzylic bromide as well as a cyanamide. The benzylic bromide is synthetically useful due to its ability to be readily substituted by various nucleophiles with inversion of stereochemistry resulting in a preserved stereocenter. The cyanamide can act as an amine protecting group, or it can be modified under a variety of conditions. This functional group versatility broadens the scope of the methodology for use in the synthesis of diverse, chiral 1,3-amino alcohols. Scheme 3 shows some of the chemistry that we carried out on the ring-opened products 7 and illustrates the utility and general reactivity of these intermediates.

Substitution of the bromide 7b with sodium azide by  $S_N2$  gave the intermediate azide 8 in high yield. Selective reduction using Pearlman's catalyst ( $Pd(OH)_2$ ) and  $H_2$  in EtOAc in the presence of  $Boc_2O$  afforded the Boc protected amine  $9.^{11}$  The bromide 7b was carried through a double inversion sequence involving reaction with sodium picolinate and mild coppermediated methanolysis to provide 10, and a subsequent Mitsunobu reaction to give alcohol  $11.^{12}$  Direct substitution of 7b with sodium benzoate in DMPU at room temperature gave the ester 12.

Catalytic hydrogenation of **7a** and **7b**, again with Pearlman's catalyst in EtOAc, provided **13a** and **13b** via hydrogenolysis of the benzylic bromide while the rest of the molecule remained intact. If the reaction solvent for reduction of **7b** was changed to methanol instead of EtOAc, not only did hydrogenolysis of the benzylic bromide occur, but concomitant reduction of the cyanamide to the formamidine **14** was observed. This functionality proved to be very stable, and under several conditions attempted, no further reduction of the formamidine occurred. Following the method of Meyers<sup>13</sup>, the formamidine was converted to the Cbz protected counterpart **15**, which on treatment with hydrazine under mild conditions afforded amine **16**.

Since the number and type of substituents on the piperidine could affect the ring-opening step, examples with four and five stereocenters were studied. The synthesis of the four-center precursor began with 2-phenyl-2,3-dihydropyridone 3 which was alkylated at the 3-position to give, as a single diastereomer, compound 17 (Scheme 4). Michael addition of methyl cuprate under our standard conditions afforded 4-piperidone 18. It is worthy of note at this point that the conjugate addition, with an axial substituent at the 3-position of 17, gives a single diastereomer via axial attack. A similar result was observed during a study of the Mukaiyama-Michael reaction of certain N-acyl-2,3-dihydro-4-pyridones. <sup>14</sup> LAH reduction proceeded smoothly to give a 7:1 mixture of piperidinols which on subsequent protection and purification provided the TBS ether 19 in 55% yield for the two steps. The relative stereochemistry assignment for 19 was confirmed through 2-D NOESY experiments. As in the reaction of 4 with LAH, the reduction of ketone 18 gave the axial alcohol intermediate as the major isomer. The von Braun reaction was carried out on substrate 19. Once again, total regioselectivity imparted by the 2-phenyl substituent occurred, as well as inversion of configuration indicative of an S<sub>N</sub>2 reaction to afford bromide **20** in 88% yield. It was found at this time that the von Braun ring-opening reaction of our 2-phenylpiperidines proceeds smoothly at room temperature. As before, S<sub>N</sub>2 substitution of the bromide with sodium azide provided the corresponding azide 21.

The possibility of opening a fully substituted piperidine ring was investigated next. Dihydropyridone **18** was methylated at C-5 with LiHMDS and methyl iodide to afford the all axial tetrasubstituted piperidone **22** (Scheme 5). Stereospecific reduction of the ketone carbonyl with L-Selectride provided the equatorial alcohol **23** as a single diastereomer. This time the facial selectivity observed is a result of the presence of axial C-3 and C-5 methyl groups in the low energy conformation (Figure 2). Reduction with LAH gave the *N*-methylpiperidinol **24** in 98% yield. Since the sterically hindered hydroxyl of **24** proved difficult to protect, the von Brawn ring-opening was carried out directly. It was hoped that alcohol

protection would not be needed, since an example of the von Braun reaction proceeding in the presence of hydroxyl groups has been reported. <sup>15</sup>

The fully substituted piperidine **24** was treated with cyanogen bromide at room temperature to afford **25** in good yield. X-ray analysis of minor byproduct **26** confirmed the stereochemical assignments for **23** and **24**.<sup>4</sup>

To broaden the scope of this methodology, the preparation of polyhydroxy piperidines via a tetrahydropyridine intermediate was examined. <sup>16</sup> The 2-alkyldihydropyridone **27** was reduced under Luche conditions to the corresponding *trans*-alcohol which was methylated to provide ether **28**. <sup>17</sup> Lewis acid mediated phenylzinc addition, via an intermediate iminium ion, gave tetrahydropyridine **29**. <sup>18</sup> With the C-2 and C-6 substituents of **29** pseudo axial, dihydroxylation with OsO<sub>4</sub> would certainly lead to the β-diol. Since flexible stereocontrol in the incorporation of functionality is important to the scope of this methodology, we were interested to see if the all cis diol **31** could be prepared. To this end, **29** was subjected to the Woodward dihydroxylation reaction. <sup>19</sup> To our delight, the all *cis* piperidine **30** was obtained in 60% yield. Deprotection and reductive methylation of **30** using a one-pot procedure gave the *N*-methyl derivative **31**. <sup>20</sup> The hydroxyl groups were protected as benzyl ethers to provide **32**. Cyanogen bromide ring-opening was again highly efficient affording a 96% yield of cyanamide **33**. The bromine of **33** was reductively removed in the presence of the benzyl ethers and cyanamide group using mild catalytic hydrogenation conditions to give **34** in near quantitative yield.

#### Conclusion

This methodology is of broad scope as it has the potential for use in the preparation of uniquely substituted amine derivatives containing multiple chiral centers and various functionality. The relative stereochemistry can be introduced with a high degree of control and with considerable variability. Contiguous or skip stereocenters can be introduced via appropriate conversions of the *N*-acyldihydropyridone precursor. Since introduction of substituents onto the dihydropyridone or tetrahydropyridine intermediates can be accomplished stereoselectively through direct addition<sup>2</sup> or epimerization, <sup>2h,21</sup> most desired diastereomers can be prepared efficiently in a few number of steps. The potential is there for incorporating quaternary centers, <sup>2h,21</sup> and for using other aryl or non-aryl ring-opening activating groups at the C-2 position. Although this study used racemic materials, enantiopure diastereomeric products of either antipode can be prepared by starting with readily available nonracemic dihydropyridones.<sup>3</sup> This methodology should be amenable to the asymmetric synthesis of natural products and other biologically active compounds.

# **Experimental Section**

The synthesis and characterization of compounds **3–5**, **6b**, **7b**, **8–9**, **12**, **17–18**, and **22–26** have been previously reported.<sup>4</sup>

#### (2S\*,4R\*,6S\*)-1,2-Dimethyl-6-phenylpiperidin-4-yl acetate (6a)

To a solution of **5** (115 mg, 0.56 mmol) in 4 mL of THF was added acetic anhydride (62.8 mg, 0.62 mmol), TEA (75.3 mg, 0.616 mmol), and DMAP (2.8 mg, 0.03 mmol). The mixture was allowed to stir at rt for 18 h. After the reaction was deemed complete by TLC, the mixture was filtered through Celite and the filtrate was concentrated under reduced pressure. The crude residue was purified by radial PLC (30% EtOAc/hexanes) to give 109 mg (79%) of **6a** as a yellow oil.  $^{1}$ H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.18 (d, 1H, J = 8 Hz), 1.54 (q, 1H, J = 15.6 Hz), 1.68 (q, 1H, J = 15.6 Hz), 1.96 (s, 3H), 1.98 (s, 3H), 1.91–2.04 (m, 2H), 2.17–2.27 (m, 1H), 3.03 (dd, 1H, J = 12.4, 3.6 Hz), 4.76–4.87 (m, 1H), 7.20–7.32 (m, 5H);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  21.5, 39.9, 40.2, 41.4, 58.0, 68.9, 71.0, 127.4, 127.6, 128.7, 144.4, 170.7; IR (neat)

1434, 1453, 1494, 1602, 1740, 2779, 2845, 2970, 3027, 3062 cm $^{-1}$ ; HRMS:  $(M+H)^+$  calcd for  $C_{15}H_{21}NO_2$  248.1651; found 248.1647.

## (1R\*,3R\*,5S\*)-1-Bromo-5-(isocyano(methyl)amino)-1-phenylhexan-3-yl acetate (7a)

Protected piperidinol **6a** (270 mg, 1.09 mmol) was dissolved in CH<sub>2</sub>Cl<sub>2</sub> (40 mL). A solution of BrCN (1.8 mL, 5.4 mmol, 3.0 M in CH<sub>2</sub>Cl<sub>2</sub>) was added and the reaction mixture was brought to reflux. The reaction was monitored by TLC analysis. After 3 h, the reaction mixture was cooled and concentrated under reduced pressure. The crude residue was purified by radial PLC (20–30% EtOAc/hexanes) to give compound **7a** (364 mg, 95%) as a yellow oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.1 (s, 6H), 0.9 (s, 9H), 1.21 (d, 3H, J = 8.8 Hz), 1.53 (m, 1H), 1.89 (m, 1H), 2.17 (m, 1H), 2.56 (m, 1H), 5.03 (dd, 1H, J = 6.8 Hz), 7.26–7.40 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.2, –3.7, 18.2, 36.9, 41.9, 47.7, 51.1, 54.5, 68.2, 117.3, 127.4, 128.8, 129.1, 142.1; IR (neat) 1508, 1560, 1655, 1685, 2209, 2362, 2857, 2896, 2930, 2954, 3032 cm<sup>-1</sup>; HRMS: (M+H)<sup>+</sup> calcd for C<sub>20</sub>H<sub>33</sub>BrN<sub>2</sub>OSi: 425.1624; found, 425.1629.

# *N*-((2*S*\*,4*R*\*,6*S*\*)-4-(*tert*-Butyldimethylsilyloxy)-6-hydroxy-6-phenylhexan-2-yl)-*N*-methylcyanamide (10)

To bromide **7b** (80 mg, 0.19 mmol) in 2.7 mL of HMPA was added freshly prepared sodium picolinate (32 mg, 0.23 mmol). The reaction was allowed to stir at rt for 18 h at which time 3 mL of brine was added along with 3 mL of Et<sub>2</sub>O. The layers were separated and the aqueous layer was extracted with Et<sub>2</sub>O ( $3 \times 3$  mL). The organic extracts were combined, washed with  $H_2O$  (2 × 2 mL) and brine (1 × 2 mL), and dried over MgSO<sub>4</sub>. Filtration and concentration gave an oil that was dissolved in 5 mL of CHCl<sub>3</sub>. MeOH (49 µL, 1.197 mmol) and Cu (OAc)<sub>2</sub> (27 mg, 0.15 mmol) were added and the mixture was stirred at rt for 18 h. Concentration afforded a crude oil that was dissolved in CH<sub>2</sub>Cl<sub>2</sub>. The solution was washed with 50:50 saturated NH<sub>4</sub>Cl/20% NH<sub>4</sub>OH solution and the layers separated. The organic layer was dried over MgSO<sub>4</sub> and concentrated. Purification by radial PLC (EtOAc/hexanes) gave 45 mg of benzyl alcohol 10 (63% yield overall) as an oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz) δ 0.13 (d, 6H, J = 24.4 Hz), 0.93 (s, 9H), 1.27 (d, 3H, J = 6.4 Hz), 1.75–2.02 (m, 4), 2.84 (s, 3H), 3.09 (m, 1H), 4.12 (sextet, 1H, J = 4.4 Hz), 4.98 (dd, 1H, J = 2.8, 10.4 Hz), 7.36 (m, 5H); <sup>13</sup>C NMR  $(CDCl_3, 75\ MHz)\ \delta\ -4.6, \ -4.0, \ 18.2, \ 26.1, \ 37.4, \ 41.8, \ 45.8, \ 54.8, \ 68.4, \ 71.5, \ 117.2, \ 125.9,$ 127.7, 128.7, 144.8; IR (neat) 1063, 1256, 1458, 2211, 2858, 2932, 2954, 3445 cm<sup>-1</sup>; HRMS  $(M+H)^+$  calcd for  $C_{20}H_{34}N_2O_2Si$  363.2468, found 363.2462.

# $N-((2S^*,4R^*,6R^*)-4-(tert-Butyldimethylsilyloxy)-6-hydroxy-6-phenylhexan-2-yl)-N-methylcyanamide (11)$

To benzyl alcohol **10** (45 mg, 0.12 mmol) in 2 mL of THF at -20 °C was added triphenylphosphine (65 mg, 0.25 mmol) and picolinic acid (31 mg, 0.25 mmol). After stirring for 20 min, DEAD (39 µl, 0.25 mmol) was added dropwise over 5 min and then the mixture was allowed to warm to rt over a 4 h period. The reaction mixture was concentrated under reduced pressure. The residue was triturated with hot hexanes and filtered. The filtrate was concentrated to give an oil that was taken directly on to the next step. To the crude ester was added 6 mL of CHCl<sub>3</sub>, MeOH (39 µL, 2.7 mmol), and Cu(OAc)<sub>2</sub> (23 mg, 0.12 mmol), and the mixture was stirred at rt for 20 h. The reaction was quenched with 50:50 saturated NH<sub>4</sub>Cl/25% NH<sub>4</sub>OH solution, and the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic extracts were washed with H<sub>2</sub>O and brine, dried over MgSO<sub>4</sub>, and concentrated under reduced pressure. Purification by radial PLC (EtOAc/hexanes) gave **11** (23 mg, 53% over 2 steps) as a white solid, mp 82–83 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.11 (d, 6H, J = 5.2 Hz), 0.9 (s, 9H), 1.23 (d, 3H, J = 6.0 Hz), 1.65 (m, 1H), 1.77 (m, 1H), 1.99 (m, 2H), 2.73 (bs, 1H), 2.84 (s, 3H), 3.12 (m, 1H), 4.14 (m, 1H), 4.83 (d, 1H, J = 10 Hz), 7.26–7.35 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.3, –3.8, 18.4, 26.1, 37.3, 42.4, 47.0, 54.4, 67.8, 71.7, 117.6, 125.8, 127.7, 128.7,

145.1; IR (neat) 775, 836, 1061, 2209, 2855, 2929, 2952 cm $^{-1}$ ; HRMS (M+H) $^+$  calcd for  $C_{20}H_{34}N_2O_2Si$  363.2468, found 363.2457.

## (3S\*,5S\*)-5-(N-Methylcyanamido)-1-phenylhexan-3-yl acetate (13a)

To bromide **7a** (51 mg, 0.145 mmol) in EtOAc (3 mL) was added Pd(OH)<sub>2</sub> (40 mg, 0.06 mmol) and NaOAc (30 mg, 0.36 mmol), and the mixture was placed under a balloon pressurized atmosphere of H<sub>2</sub> with good stirring. After 15 h, the reaction mixture was filtered through Celite and concentrated to give 33 mg (83%) of **13a** which was used without further purification.  $^{1}$ H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  1.24 (d, 3H, J = 8.8 Hz), 1.65–1.74 (m, 1H), 1.83–1.99 (m, 2H), 2.04 (s, 3H), 2.64 (t, 2H, J = 10.8 Hz), 2.81 (s, 3H), 2.87–2.94 (m, 1H), 5.05–5.13 (m, 1H), 7.15–7.30 (m, 5H);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  18.4, 21.4, 31.8, 36.6, 37.9, 39.7, 54.9, 71.0, 116.9, 126.3, 128.5, 128.7, 141.4, 170.9; IR (neat) 701, 749, 939, 1023, 1041, 1135, 1167, 1239, 1374, 1434, 1455, 1496, 1557, 1603, 1652, 1735, 2207, 2360, 2933, 2969, 3027, 3646, 3673, 3748, 3851 cm<sup>-1</sup>; HRMS (M+H)<sup>+</sup> calcd for C<sub>16</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub> 275.1760, found 275.1750.

## N-((2S\*,4S\*)-4-(tert-Butyldimethylsilyloxy)-6-phenylhexan-2-yl)-N-methylcyanamide (13b)

To benzyl bromide **7b** (28 mg, 0.06 mmol) in 3 mL of dry EtOAc was added Pd(OH)<sub>2</sub> (19 mg, 0.03 mmol) and NaOAc (11 mg, 0.13 mmol). The mixture was stirred under an atmosphere of H<sub>2</sub> at balloon pressure for 12 h. The mixture was filtered through a Celite pad with an EtOAc wash. The solvent was removed under reduced pressure to yield **13b** (23 mg, 100%) as an oil which did not need further purification. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.07 (d, 6H, J = 6.9 Hz), 0.90 (s, 9H), 1.23 (d, 3H, J = 6.6 Hz), 1.55–1.64 (m, 1H), 1.76–1.89 (m, 2H), 2.56–2.71 (m, 2H), 2.83 (s, 3H), 3.07–3.16 (m, 1H), 3.87–3.94 (m, 1H), 7.17–7.31 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.4, –3.7, 18.3, 26.1, 19.9, 31.0, 37.4, 39.9, 42.0, 54.7, 68.5, 117.4, 126.0, 128.6, 128.7, 142.4; IR (neat) 776, 1063, 1254, 1458, 2208, 2857, 2932, 2952 cm<sup>-1</sup>; HRMS (M+H)<sup>+</sup> calcd for C<sub>20</sub>H<sub>34</sub>N<sub>2</sub>OSi 347.2519, found 347.2528.

## N-((2S\*,4S\*)-4-(tert-Butyldimethylsilyloxy)-6-phenylhexan-2-yl)-N-methylformimidamide (14)

To bromocyanamide **7b** (30 mg, 0.07 mmol) in 2.5 mL of EtOH was added Pearlman's Catalyst (Pd(OH)<sub>2</sub>, 4 mg, 0.03 mmol) and NaOAc (6 mg, 0.07 mmol). The mixture was placed under a balloon pressurized atmosphere of H<sub>2</sub> and stirred at rt for 15 h at which time the reaction was deemed complete by TLC. The mixture was filtered through a pad of Celite and then concentrated. The crude solid was triturated with CH<sub>2</sub>Cl<sub>2</sub>. The hot mixture was filtered and the remaining solids washed with hot CH<sub>2</sub>Cl<sub>2</sub>. The filtrate was concentrated under reduced pressure and the crude solid residue was recrystallized from EtOAc to give **14** (24.2 mg, 99%) as a white solid, mp 184–185 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.07 (d, 6H, J = 7.2 Hz), 0.90 (s, 9H), 1.32 (d, 3H, J = 6.6), 1.65–1.82 (m, 4H), 2.61 (m, 1H), 3.14 (s, 3H), 3.64 (bs, 1H), 3.81 (m, 1H), 7.14–7.33 (m, 5H), 7.62 (s, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –3.9, –3.4, 20.3, 26.2, 31.1, 32.0, 39.0, 40.9, 59.2, 68.9, 126.4, 128.5, 128.5, 153.7; IR (neat) 836.1, 1063, 1255, 1699, 2358, 2857, 2954 cm<sup>-1</sup>; HRMS (M+H)<sup>+</sup> calcd for C<sub>19</sub>H<sub>19</sub>NO<sub>3</sub> 349.2675, found 349.2680.

# (E)-Benzyl(( $(2S^*,4S^*)$ -4-(*tert*-butyldimethylsilyloxy)-6-phenylhexan-2-yl)(methyl)amino) methylenecarbamate (15)

To formamidine 14 (30 mg, 0.087) in 3 mL of  $Et_2O$  was added 3 mL of saturated aqueous NaHCO<sub>3</sub>. The biphasic reaction mixture was cooled to 0 °C and benzyl chloroformate (16.2 mg, 0.1 mmol) was added dropwise. The reaction mixture was stirred at 0 °C for 10 min and then warmed to rt. After stirring for 1 h, the layers were separated and the aqueous layer was extracted with  $Et_2O$ . The combined extracts were washed with water and brine, dried over MgSO<sub>4</sub>, and concentrated under reduced pressure. The residue was purified by radial PLC

(EtOAc/hexanes) to give **15** (40 mg, 97%) as an oil.  $^{1}$ H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.29 (d, 6H, J= 15.6 Hz), 0.91 (s, 9H), 1.28 (d, 3H, J= 6 Hz), 1.58–1.84 (m, 14H), 2.60 (m, 2H), 2.93 (s, 3H), 3.62 (m, 1H), 3.80 (m, 1H), 4.71 (s and 5.19 s, due to rotamers, 2H), 7.14–7.43 (m, 10H), 8.44 (s and 8.55 s, due to rotamers, 1H);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.4, –3.4, 18.3, 20.3, 26.2, 29.0, 31.1, 39.5, 41.0, 56.5, 67.6, 68.7, 126.2, 127.2, 12.4, 128.6, 128.8, 137.0, 141.9, 163.1, 164.4; IR (neat) 1062, 1211, 1378, 1657, 2285, 2928, 2949, 3027 cm<sup>-1</sup>. HRMS (M+H)<sup>+</sup> calcd for  $C_{28}H_{42}N_2O_3Si$  483.3043, found 483.3033.

# (2S\*,4S\*)-4-(tert-Butyldimethylsilyloxy)-N-methyl-6-phenylhexan-2-amine (16)

To the protected formamidine **15** (19 mg, 0.039 mmol) in 2 mL of a 6/4 EtOH/H<sub>2</sub>O solution was added one drop of glacial acetic acid along with N<sub>2</sub>H<sub>4</sub>•H<sub>2</sub>O (5  $\mu$ l, 0.11 mmol). The reaction was stirred at rt for 15 h and then 1 mL of water and 3 mL of Et<sub>2</sub>O were added. The layers were separated and the water layer was extracted with Et<sub>2</sub>O. The combined extracts were washed with brine, dried over K<sub>2</sub>CO<sub>3</sub>, and concentrated under reduced pressure. The residue was purified by radial PLC (EtOAc/hexanes) to give 11 mg (88%) of **16** as an oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.07 (s, 6H), 0.90 (s, 9H), 1.04 (d, 3H, J = 6.0 Hz), 1.25 (s, 1H), 1.48–1.81 (m, 4H), 2.38 (s, 3H), 2.63 (m, 3H), 3.86 (m, 1H), 7.17–7.36 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.2, 18.3, 20.6, 26.1, 31.9, 33.9, 39.3, 44.0, 51.8, 70.1, 126.0, 128.6, 142.7; IR (neat) 1057, 1256, 1373, 1472, 1604, 2793, 2856, 2929, 2955 cm<sup>-1</sup>; HRMS (M +H)<sup>+</sup> calcd for C<sub>19</sub>H<sub>35</sub>NOSi 322.2566, found 322.2574.

#### (2S\*,3R\*,4R\*,6S\*)-4-(tert-Butyldimethylsilyloxy)-1,3,6-trimethyl-2-phenylpiperidine (19)

To piperidone 18 (347 mg, 1.07 mmol) in 20 mL of THF was added portionwise LAH (241 mg, 6.44 mmol), and the mixture was heated at reflux for 4 h. The reaction was cooled to 0  $^{\circ}$ C and quenched by careful addition of 10% NaOH (0.72 mL) and H<sub>2</sub>O (0.5 mL). Celite (3.5 g) was added and the mixture was allowed to warm to rt with stirring for 1.5 h. The mixture was filtered through a pad of Celite and concentrated under reduced pressure. A portion of the crude alcohol (115 mg, 0.52 mmol) was stirred in 10 mL of dry CH<sub>2</sub>Cl<sub>2</sub> with TBSOTf (277 mg, 1.05 mmol), Et<sub>3</sub>N (83 mg, 0.68 mmol), and a catalytic amount of imidazole (3 mg, 0.03 mmol) at rt for 1.5 h. The reaction mixture was quenched with 8 mL of saturated aqueous NaHCO<sub>3</sub> solution. The aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organics were combined and washed with H<sub>2</sub>O and brine, and dried over MgSO<sub>4</sub>. Filtration and concentration under reduced pressure gave a crude oil which was purified by radial PLC (EtOAc/hexanes) to afford 96 mg (55%) of **19** as a clear oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.08 (d, 6H, J = 6.3 Hz), 0.6 (d, 3H, J = 6.3 Hz), 0.90 (s, 9H), 1.17 (d, 3H, J = 6 Hz), 1.63 (m, 2H), 1.89 (s, 3H), 2.14 (m, 2H)1H), 2.54 (d, 1H, J = 9.9 Hz), 3.3 (dt, 1H, J = 4.5, 5.4 Hz), 7.73 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.4, –3.7, 15.8, 18.3, 21.9, 26.1, 40.7, 44.5, 46.1, 58.2, 74.9, 127.2, 128.0, 128.5, 128.7, 143.7, 149.9; IR (neat) 702, 835, 873, 1086, 1249, 1375, 2777, 2850, 2933, 2964  $cm^{-1}$ ; HRMS (M+H)<sup>+</sup> calcd for C<sub>20</sub>H<sub>35</sub>NOSi 334.2566, found 334.2560.

# *N*-((2*S*\*,4*R*\*,5*R*\*,6*R*\*)-6-Bromo-4-(*tert*-butyldimethylsilyloxy)-5-methyl-6-phenylhexan-2-yl)-*N*-methylcyanamide (20)

To a solution of piperidine **19** (96 mg, 0.288 mmol) in 7 mL of CHCl<sub>3</sub> was added BrCN (0.47 mL, 1.4 mmol, 3.0 M in CH<sub>2</sub>Cl<sub>2</sub>). The mixture was allowed to stir at rt for 5 h and then concentrated under reduced pressure to give a yellow oil. The crude product was freed of residual BrCN under high vacuum. The crude yield was quantitative, but purification by radial PLC (EtOAc/hexanes) gave 110 mg (88%) of **20** as a yellow oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  –0.09 (d, 6H, J = 12.6 Hz), 0.89 (s, 9H), 1.09 (d, 3H, J = 6.3 Hz), 1.28 (d, 3H, J = 6.9 Hz), 1.63 (m, 1H), 1.88 (m, 1H), 2.23 (m, 1H), 2.74 (s, 3H), 2.85 (m, 1H), 3.58 (m, 1H), 5.04 (d, 1H, J = 8.7 Hz) 7.32 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  –4.1, –3.7, 13.1, 17.9, 18.4, 26.1, 37.3, 40.5, 45.8, 55.3, 59.5, 69.8, 117.0, 128.0, 128.5, 128.9, 142.0; IR (neat) 775, 836, 1078,

 $1254,1384, 1454, 2208, 2856, 2931, 2953, 3464 \text{ cm}^{-1}$ ; HRMS (M+H)<sup>+</sup> calcd for C<sub>21</sub>H<sub>35</sub>BrN<sub>2</sub>OSi 439.1780, found 439.1765.

# $N-((2S^*,4R^*,5S^*,6S^*)-6-Azido-4-(tert-butyldimethylsilyloxy)-5-methyl-6-phenylhexan-2-yl)-<math>N-$ methylcyanamide (21)

To a solution of bromocyanamide **20** (43 mg, 0.10 mmol) in 2 mL of freshly distilled DMSO was added NaN<sub>3</sub> (19 mg, 0.29 mmol). The mixture was stirred at rt for 18 h and then diluted with saturated aqueous NaHCO<sub>3</sub> (2 mL). The aqueous phase was extracted with Et<sub>2</sub>O, and the combined extracts were washed with H<sub>2</sub>O and brine. The mixture was dried (MgSO<sub>4</sub>), filtered, and concentrated under reduced pressure. The crude product was purified by radial PLC (EtOAc/hexanes) to give 31 mg (77%) of **21** as a clear oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  0.09 (m, 1H), 0.16 (d, 6H, J = 15.3 Hz), 0.60 (d, 3H, J = 6.9 Hz), 0.9 (m, 3H), 0.94 (s, 9H), 1.27 (d, 3H, J = 6.3 Hz), 1.58–1.86 (m, 4H), 2.86 (s, 3H), 2.96 (m, 1H), 4.22 (m, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  -4.5, -3.7, 18.2, 26.1, 37.2, 41.6, 42.6, 54.6, 68.2, 117.4, 134.1, 134.2, 136.5, 181.4; HRMS (M+H)<sup>+</sup> calcd for C<sub>21</sub>H<sub>35</sub>N<sub>5</sub>OSi 402.2689, found 402.2689.

#### 2-Nonyl-4-oxo-3,4-dihydro-2H-pyridine-1-carboxylic acid benzyl ester (27)

Prepared as an oil (76%) according to the literature procedure.  $^6$  <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.76 (d, 1H, J = 4.8 Hz), 7.37 (s, 5H), 5.32-5.21 (m, 3H), 4.58 (m, 1H), 2.79 (dd, 1H, J = 6.4, 16.4), 2.45 (d, 1H, J = 16.4), 1.61 (m, 2H), 1.25 (m, 13H), 0.8 (t, 3H, J = 5.4 Hz);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  193.3, 141.7, 141.6, 135.1, 128.9, 128.8, 128.6, 107.3, 69.1, 53.6, 39.8, 31.9, 30.6, 29.6, 29.4, 25.8, 22.8, 14.3; HRMS (M+H)<sup>+</sup> calcd for C<sub>22</sub>H<sub>31</sub>NO<sub>3</sub> 358.2382, found 358.2380.

## (2R,\*4R\*)-4-Methoxy-2-nonyl-3,4-dihydro-2H-pyridine-1-carboxylic acid benzyl ester (28)

To compound 27 (733 mg, 2.05 mmol) in 80 mL of MeOH was added solid CeCl<sub>3</sub>•7H<sub>2</sub>O (1.53 g, 4.10 mmol) at rt and the mixture was allowed to stir for 15 min. After cooling to -40 °C, NaBH<sub>4</sub> (232 mg, 6.15 mmol) was added and the reaction progress was monitored by TLC. After disappearance of starting material showed completion of the reaction, acetone was added to quench excess NaBH<sub>4</sub>. Solvent was removed in vacuo, and water was added to the crude residue. The aqueous mixture was extracted with EtOAc. The combined organic layers were washed with H<sub>2</sub>O and brine, dried over MgSO<sub>4</sub>, and concentrated under reduced pressure to give an oil. The oil was taken on to the next step without further purification. The oil from above was dissolved in THF (5 mL) and cooled to -40 °C and then t-BuOK (1.0 M, THF, 2.46 mmol) was added. After stirring for 20 min, MeI (0.76 mL, 12.3 mmol) was added and the mixture was allowed to slowly warm to rt. After 1 h, saturated aqueous NaHCO<sub>3</sub> was added. The aqueous layer was extracted with 3 × 10 mL of EtOAc. The combined extracts were washed with H<sub>2</sub>O and brine. The organic layer was dried over MgSO<sub>4</sub> and concentrated to give an oil (28, 675 mg, 87% yield) that was used without further purification. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.36 (s, 5H), 6.82 (brs, 1H), 5.20 (d, 3H, J = 12.6 Hz), 4.96 (brs, 1H), 4.29 (brs, 1H),  $3.97 \text{ (m, 1H)}, 3.36 \text{ (s, 3H)}, 2.20 \text{ (brs, 1H)}, 1.72 \text{ (m, 16H)}, 0.88 \text{ (t, 3H, } J = 6.6 \text{ Hz)}; ^{13}\text{C NMR}$ 31.4, 26.1, 25.8, 22.8, 14.3; HRMS (M+H)<sup>+</sup> calcd for C<sub>23</sub>H<sub>34</sub>NO<sub>3</sub> 374.2695, found 374.2709.

#### (2R\*,6R\*)-4-Methoxy-2-nonyl-3,4-dihydro-2H-pyridine-1-carboxylic acid benzyl ester (29)

To anhydrous ZnBr $_2$  (1.85 g, 8.2 mmol) in 10 mL of toluene at rt was added PhMgBr (8.2 mL, 8.2 mmol, 1.0 M in THF), and the mixture was stirred for 1 h at rt. Methyl ether **28** (675 mg, 1.81 mmol) in 10 mL of toluene at 0 °C was added followed immediately by addition of BF $_3$ ·OEt $_2$  (2.7 mmol). The mixture was stirred at 0 °C for 1 h and then allowed to slowly warm to rt and quenched with 10% HCl. The crude reaction mixture was extracted with Et $_2$ O. The combined organic layers were washed with H $_2$ O and brine, dried over MgSO $_4$ , and

concentrated under reduced pressure. Purification by radial PLC (5–10% EtOAc/hexanes) gave upon reiterative purification 535 mg (73%) of the *cis* diastereoemer **29** as an oil.  $^{1}$ H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.45 (brs, 1H), 7.31-7.21 (m, 10H), 5.95 (s, 2H), 5.71 (brs, 1H), 5.19 (s, 2H), 4.50 (m, 1H), 2.44 (ddd, 1H, J = 3.0, 5.2, 13.1 Hz), 2.01 (d, 1H, J = 13.1), 1.44 (m, 1H), 1.28-0.80 (m, 14H), 0.87 (t, 3H, J = 6.9 Hz);  $^{13}$ C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  156.5, 142.4, 136.9, 128.6, 128.4, 128.2, 128.0, 127.2, 125.9, 124.4, 67.4, 53.6, 49.4, 34.2, 32.0, 29.8, 29.6, 29.4, 28.9, 28.8, 26.7, 22.8, 14.3; HRMS (M+H) $^{+}$  calcd for C<sub>28</sub>H<sub>34</sub>NO<sub>2</sub> 419.2824, found 419.2792.

# $(2S^*,3R^*,4S^*,6R^*)$ -3,4-Dihydroxy-6-nonyl-2-phenyl-piperidine-1-carboxylic acid benzyl ester (30)

To olefin **29** (336 mg, 0.83 mmol) in 4 mL of AcOH was added AgOAc (288 mg, 1.64 mmol) and  $I_2$  (208 mg, 0.83 mmol). The mixture was allowed to stir at rt for 12 h and then AcOH:H<sub>2</sub>O (24:1) was added and stirring was continued for 2 h. Saturated brine was added and the reaction mixture was filtered through Celite. The Celite pad was washed with EtOAc and MeOH. The filtrate was concentrated to remove solvent and then a solution of 2.0 M KOH/MeOH was added. The mixture was stirred at rt for 4 h. Brine was added and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>, and the combined extracts were dried over MgSO<sub>4</sub>. Concentration and purification by radial PLC (20–50% EtOAc/hexanes) gave 218 mg (60%) of diol **30** as an oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.38-7.19 (m, 8H), 6.95 (m, 2H), 5.10 (d, 1H, J = 5.2 Hz), 4.99 (d, 1H, J = 12.4 Hz), 4.92 (d, 1H, J = 12.4 Hz), 4.32 (m, 1H), 4.16 (m, 1H), 3.93 (m, 1H), 2.31 (ddd, 1H, J = 6.4, 10.8, 12.4 Hz), 1.93 (m, 1H), 1.84 (m, 1H), 1.75 (ddd, 1H, J = 5.2, 10.8, 12.8 Hz), 1.33-1.25 (m, 14H), 0.89 (t, 3H, J = 6.8 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  156.5, 140.0, 136.5, 128.9, 128.4, 127.9, 127.8, 127.4, 126.8, 71.2, 68.0, 67.5, 63.1, 51.0, 38.4, 32.1, 31.1, 29.9, 29.7, 29.5, 26.9, 22.9, 14.3.

## (2S\*,3R\*,4S\*,6R\*)-1-Methyl-6-nonyl-2-phenyl-piperidine-3,4-diol (31)

A mixture of piperindiol **30** (40 mg, 0.09 mmol), formaldehyde (0.4 mL, 37% in H<sub>2</sub>O), and Pd(OH)<sub>2</sub> in 9 mL of MeOH was placed under a hydrogen atmosphere (50 psi, Parr shaker overnight) at rt. The reaction mixture was filtered through Celite and the filter pad was washed with copious amounts of methanol. The filtrate was concentrated *in vacuo*, and residue was purified by column chromatography (0–20% EtOAc/hexanes) to give 20 mg (70%) of the desired *N*-methylpiperidinol **31** as an oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.35-7.27 (m, 5H), 3.66 (brs, 2H), 3.14 (s, 1H), 2.21-2.19 (m, 2H), 2.11 (m, 1H), 2.03 (s, 3H), 1.87 (m, 1H), 1.69 (dd, 1H, J = 9.0, 18.6, 12.4 Hz), 1.60-1.40 (m, 4H), 1.29-1.25 (m, 22H), 0.87 (t, 3H, J = 5.4 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  140.9, 128.7, 128.5, 128.3, 127.7, 73.3, 72.8, 70.6, 62.4, 40.1, 34.3, 33.8, 32.1, 30.3, 29.9, 29.8, 29.6, 24.6, 22.9, 14.3.

#### (2S\*,3R\*,4S\*,6R\*)-3,4-Bis-benzyloxy-1-methyl-6-nonyl-2-phenylpiperidine (32)

Diol **31** (50 mg, 0.15 mmol) was dissolved in 1.5 mL of THF and cooled to -40 °C. A solution of *t*-BuOK (1.0 M in THF, 0.45 mL) was added dropwise and the mixture was allowed to stir at -40 °C for 30 min. Benzyl bromide (0.053 mL, 0.45 mmol) was added and the reaction was allowed to warm slowly to rt over 2 h. Saturated NH<sub>4</sub>Cl was added and the aqueous layer was extracted with Et<sub>2</sub>O. The combined extracts were dried over MgSO<sub>4</sub> and concentrated under reduced pressure. Purification by radial PLC (0–20% EtOAc/hexanes) gave 64 mg (84%) of compound **32** as an oil. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.41 (m, 2H), 7.33-7.24 (m, 9H) 7.16 (m, 2H), 7.03 (m, 2H), 4.55 (s, 2H), 4.48 (d, 1H, J = 11.2 Hz), 4.06 (d, 3H, J = 11.2 Hz), 3.71 (s, 1H), 3.54 (ddd, 1H, J = 2.0, 4.0, 6.4 Hz), 3.06 (s, 1H), 2.13 (m, 1H), 2.01 (s, 3H), 1.90 (m, 1H), 1.67 (m, 2H), 1.52 (m, 2H), 1.40-1.20 (m, 13H), 1.88 (t, 3H, J = 6.8 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  129.3, 128.6, 128.3, 128.1, 128.0, 127.6, 127.4, 127.2, 127.1, 79.9, 78.1, 74.6, 73.0, 70.3, 63.5, 40.8, 34.1, 32.1, 31.5, 30.3, 29.8, 29.6, 25.7, 22.9, 14.4. EIMS (M +1)+514.

# *N*-((1*S*\*,2*S*\*,3*S*\*,5*S*\*)-2,3-Bis(benzyloxy-1-bromo-1-phenyltetradecan-5-yl)-*N*-methylcyanamide (33)

The protected alcohol **32** (20 mg, 0.039 mmol) was dissolved in 2 mL of CHCl<sub>3</sub>. To this was added a 3.0 M solution of BrCN (0.15 mL) and the mixture was heated to reflux. After 2 h, the reaction mixture was cooled and filtered through a Celite plug with a CHCl<sub>3</sub> wash. The filtrate was concentrated *in vacuo* to give **33** as an oil (23 mg) in 96% yield. <sup>1</sup>H NMR analysis showed complete conversion and no further purification was necessary. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 300 MHz)  $\delta$  7.50-7.25 (m, 10H), 7.19 (m, 3H), 6.82 (m, 2H), 4.76 (d, 1H, J = 7.0 Hz), 4.74 (d, 1H, J = 8.4 Hz), 4.62 (d, 1H, J = 8.4 Hz), 4.54 (d, 1H, J = 7.8 Hz), 4.30 (t, 2H, J = 9.6 Hz), 4.13 (d, 1H, J = 7.8 Hz), 3.11 (m, 1H), 2.62 (s, 3H), 1.86 (ddd, 1H, J = 2.1, 7.5, 9.9 Hz), 1.70 (ddd, 1H, J = 1.5, 8.1, 9.9 Hz), 1.50-1.20 (m, 15H), 0.88 (t, 3H, J = 6.9 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 75 MHz)  $\delta$  139.9, 138.2, 137.8, 128.9, 128.8, 128.7, 128.5, 128.4, 128.2, 127.9, 118.3, 82.5, 77.3, 75.4, 72.0, 59.2, 52.5, 35.8, 33.4, 32.5, 32.1, 29.8, 29.7, 29.6, 29.5, 26.4, 22.9, 14.3; LCMS (M +1)<sup>+</sup> 621.

## N-((2R\*,3S\*,5S\*)-2,3-Bis(benzyloxy)-1-phenyltetradecan-5-yl)-N-methylcyanamide (34)

Crude bromide **33** (23 mg, 0.037 mmol) was dissolved in 2 mL of EtOAc. Pd(OH)<sub>2</sub> (10 mol %) and NaOAc (15 mg, 0.186 mmol) were added and the reaction was placed under a H<sub>2</sub> atmosphere (balloon pressure) and allowed to stir at rt for 12 h. The reaction mixture was then filtered through Celite with EtOAc. The filtrate was concentrated under reduced pressure to give 19 mg (99%) of pure **34** as an oil.  $^{1}$ H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  7.40-7.20 (m, 15H), 4.66 (d, 1H, J = 11.6 Hz), 4.64 (d, 1H, J = 11.6 Hz), 4.50 (d, 1H, J = 12.0 Hz), 4.29 (d, 1H, J = 11.6 Hz), 3.90 (td, 1H, J = 1.2, 7.2 Hz), 3.57 (d, 1H, J = 10.0 Hz), 2.96 (dd, 1H, J = 7.2, 14.0 Hz), 2.90 (m, 1H), 2.75 (dd, 1H, J = 6.4, 14.0 Hz), 2.50 (s, 3H), 1.77 (m, 2H), 1.58 (m, 1H), 1.43 (m, 1H), 1.40-1.20 (m, 14H), 0.89 (t, 3H, J = 6.4 Hz);  $^{13}$ C NMR (CDCl<sub>3</sub>, 100 MHz)  $\delta$  139.0, 138.7, 138.3, 129.7, 128.8, 128.7, 128.6, 128.5, 128.0, 127.7, 126.5, 117.8, 81.1, 76.8, 72.9, 71.7, 59.3, 37.5, 33.7, 32.1, 29.8, 29.7, 29.6, 29.5, 26.3, 22.9, 14.3.

# **Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

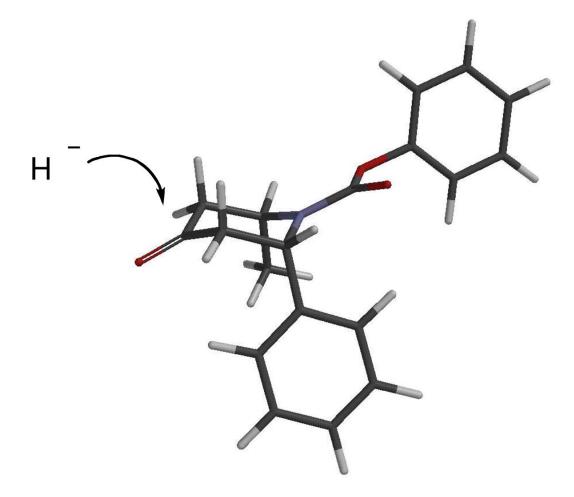
## **Acknowledgments**

This work was supported in part by the National Institutes of Health (Grant No. GM 34442). We thank the Spanish Government (Ministerio de Educación, Cultura y Deportes) for fellowship support to T. A. G.

#### References

- (a)De Oliveira LF, Costa VEU. Tetrahedron: Asymmetry 2004;15:2583.(b) For synthesis of 1,3-aminoalcohols, see: Raghavan S, Rajender A, Joseph SC, Rasheed MA, Kumar KR. Tetrahedron: Asymmetry 2004;15:365.(c)Menche D, Arikan F, Li J, Rudolph S. Org Lett 2007;9:267. and references cited therein. [PubMed: 17217281]
- Comins, DL.; Joseph, SP. Advances in Nitrogen Heterocycles. Moody, CJ., editor. Vol. 2. JAI Press Inc.; Greenwich, CT: 1996. p. 251Comins, DL.; Joseph, SP. Comprehensive Heterocyclic Chemistry. Vol. 2. McKillop, A., editor. Vol. 5. Pergamon Press; Oxford, England: 1996. p. 37 (c) Comins DL. J Heterocycl Chem 1999;36:1491. (d) Joseph SP, Comins DL. Curr Opin Drug Discovery Dev 2002;5:870. (e) Kuethe JT, Comins DL. J Org Chem 2004;69:5219. [PubMed: 15287764] (f) Comins DL, Sahn JJ. Org Lett 2005;7:5227. [PubMed: 16268544] (g) Young DW, Comins DL. Org Lett 2005;7:5661. [PubMed: 16321016] (h) Gotchev DB, Comins DL. J Org Chem 2006;71:9393. [PubMed: 17137366] (i) Comins DL, Higuchi K. Beil J Org Chem 2007;3(42)
- 3. (a) Comins DL, Joseph SP, Goehring RR. J Am Chem Soc 1994;116:4719. (b) Comins DL, LaMunyon DH. Tetrahedron Lett 1994;35:7343. (c) Comins DL, Guerra-Weltzien L. Tetrahedron Lett

- 1996;37:3807. (d) Waldmann H. Synthesis 1994:535.Comins, DL.; O'Connor, S.; Alawar, RS. Comprehensive Heterocyclic Chemistry III. Black, D., editor. Vol. 7. Elsevier Ltd; Oxford, England: 2008. p. 41and references cited therein
- 4. McCall WS, Abad Grillo T, Comins DL. Org Lett 2008;10:3255. [PubMed: 18582075]
- 5. (a)von Braun J. Chem Ber 1900;33:1438.Review: (b)Hageman HA. Org React 1953;7:198.
- 6. Comins DL, Brown JD. Tetrahedron Lett 1986;27:4549.
- 7. Brown JD, Foley MA, Comins DL. J Am Chem Soc 1988;110:7445.
- (a) Albright JD, Goldman L. J Am Chem Soc 1969;91:4317.
   (b) Fodor G, Abidi S, Carpenter TC. J Org Chem 1974;39:1507.
- For examples, see: (a)Demko ZP, Sharpless KB. Org Lett 2001;3:4091. [PubMed: 11735592](b)
   Nekrasov DD. Russ J Org Chem (Engl Transl) 2004;40:1387.(c)Nekrasov DD. Chem Heterocycl
   Compd 2004;40:1107.
- For recent syntheses of 1,3-amino alcohols, see: (a)Broustal G, Ariza X, Campagne JM, Garcia J, Georges Y, Marinetti A, Robiette R. Eur J Org Chem 2007:4293.(b)Raghavan S, Rajender A, Joseph SC, Rasheed MA, Kumar KR. Tetrahedron: Asymmetry 2004;15:365.
- 11. Saito S, Nakajima H, Inaba M, Moriwake T. Tetrahedron Lett 1989;30:87.
- 12. Sammukia T, Jacobs JS. Tetrahedron Lett 1999;40:2685.
- 13. Meyers AI, Boes MB, Dickman DA. Org Synth 1988;67:60.
- 14. Kuethe JT, Comins DL. Org Lett 1999;1:1031. [PubMed: 10825955]
- 15. Casy AF, Hassan MMA. Tetrahedron 1967;23:2075.
- 16. For the preparation of polyhydroxy piperidines from dihydropyridones, see: (a)Kitazume T, Murata K, Okabe A, Takahashi Y, Yamazaki T. Tetrahedron: Asymmetry 1994;5:1029.(b)Comins DL, Fulp AB. Tetrahedron Lett 2001;42:6839.(c)Tzanetou EN, Kasiotis KM, Magiatis P, Haroutounian SA. Molecules 2007;12:735. [PubMed: 17851426]
- 17. Comins DL, Foley MA. Tetrahedron Lett 1988;29:6711.
- 18. Comins DL, Chung G, Foley MA. Heterocycles 1994;37:1121.
- 19. (a) Woodward RB, Brutcher FV Jr. J Am Chem Soc 1958;80:209. (b) Mangoni L, Adinolfi M, Barone G, Parrilli M. Tetrahedron Lett 1973:4485.
- Rosenberg SH, Spina KP, Condon SL, Polakowski J, Yao Z, Kovar P, Stein HH, Cohen J, Barlow JL, Klinghofer V, Egan DA, Tricarico KA, Perun TJ, Baker WR, Kleinert HD. J Med Chem 1993;36:460. [PubMed: 8474102]
- 21. Ege M, Wanner KT. Tetrahedron 2008;64:7273.



**Figure 1.** Calculated lowest energy conformation of **4** (MMFF)

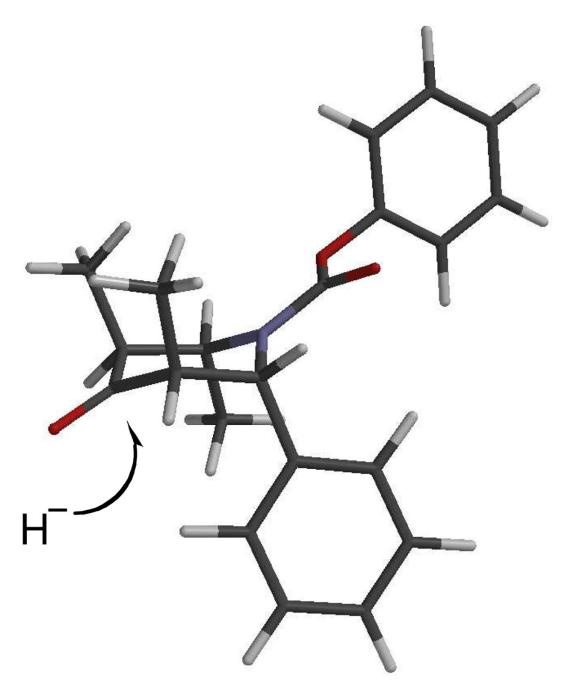


Figure 2. Calculated lowest energy conformation of 22 (MMFF).

 $R_1$ ,  $R_2$ ,  $R_4$ = H, alkyl or OBn  $R_3$ = protecting group

#### SCHEME 1.

The von Braun ring-opening reaction

OMe

1. PhoCocl,
THF, -23 °C

2. PhMgBr
3. H<sub>3</sub>O<sup>+</sup>
CO<sub>2</sub>Ph
62%

CO<sub>2</sub>Ph
83%

3

CO<sub>2</sub>Ph
62%

CO<sub>2</sub>Ph
62%

CO<sub>2</sub>Ph
62%

CO<sub>2</sub>Ph
64

CO<sub>2</sub>Ph
65

CO<sub>2</sub>Ph
66

CO<sub>2</sub>Ph
67

CH<sub>2</sub>Cl<sub>2</sub>, 
$$\Delta$$
95-99%

Ta R = Ac (95%)
b R = TBS (99%)

**SCHEME 2.** Synthesis of cyanamides 7

**SCHEME 3.** Reactions of Bromide 7

**SCHEME 4.** Synthesis of Cyanamide 21

**SCHEME 5.**Synthesis of Cyanamide 25

$$C_9H_{19}$$
 Ph  $C_9H_{19}$  Ph  $C_9H$ 

**SCHEME 6.**Synthesis of Cyanamide 34