## Microbial Hydroxylations of β-Carboline Derivatives

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Ethyl β-carboline-3-carboxylate (1a) and its 4-alkyl derivatives (1b—d) are hydroxylated by Sporotrichum sulfurescens at C-6 and C-8 of the aromatic nucleus; side chain hydroxylation of (1c) and (1d) occurs with Streptomyces lavendulae and Streptomyces griseus.

Recently, ethyl  $\beta$ -carboline-3-carboxylate (1a) was isolated from human urine by Braestrup and co-workers, who demonstrated that (1a) possessed high affinity for benzodiazepinebinding proteins.<sup>1</sup> Although it remains doubtful whether (1a) represents an endogenous ligand of the benzodiazepine receptor, chemical interest in  $\beta$ -carboline derivatives has been restimulated by Braestrup's observations.

Microbial functionalization in this alkaloid class had not been investigated; we, therefore, started a screening programme involving 92 commonly used micro-organisms of which three fungi were found to effect preparatively useful conversions of (1a) and some 4-alkyl substituted derivatives of (1a): Sporotrichum sulfurescens ATCC 7195, Streptomyces lavendulae ATCC 8664, and Streptomyces griseus ATCC

Sporotrichum sulfurescens ATCC 7195, although hitherto uncommon in aromatic hydroxylation,3 turned out to be the most useful species. Fermentation of (1a) with ATCC 7195. using standard procedures,4 resulted in the formation of the 6-hydroxy-derivative (2a) accompanied by small amounts of

ÇO,Et CO<sub>2</sub>Et (1) (2)

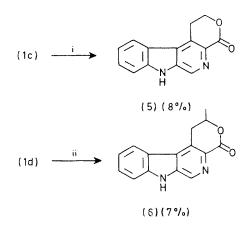
Reactant	6-hydroxylation	6-glucoside	8-glucoside
(1a)	62%	8%	7%
(1b)		20%	18%
(1c)		-	70%
(1d)	****	_	68%

Figure 1

the glucosides (3a) and (4a) (Figure 1).† The unusual formation of 4'-O-methyl- $\beta$ -glucosides is a characteristic feature of ATCC 7195, as previously mentioned by Kieslich et al.4

The introduction of 4-alkyl substituents into the carboline skeleton essentially influenced the regioselectivity of microbial attack. Whereas (1a) was predominantly affected at C-6, compound (1b) was converted into an almost equal mixture of 6- and 8-hydroxylated derivatives, isolated as their 4'-Omethyl- $\beta$ -glucosides (3b) and (4b); functionalization at C-6 was completely suppressed with compounds (1c) and (1d), which, in good yields, were transformed into their 8-hydroxyderivatives, again isolated as the glucosides (4c) and (4d).

Streptomyces lavendulae ATCC 8664 and Streptomyces griseus ATCC 10 137 are capable of hydroxylating the side



i, Streptomyces lavendulae; ii, Streptomyces griseus.

† Structural assignments were made on the basis of <sup>1</sup>H n.m.r., i.r.,

and u.v. spectra as well as elemental analysis. (2a): m.p. 248—250 °C;  $^{1}$ H n.m.r. ( $^{1}$ H $_{0}$ Me $_{2}$ SO)  $^{\delta}$  1.40 (3H, t, J7 Hz, CO $_{2}$ CH $_{2}$ Me), 4.42 (2H, q, J7 Hz, CO $_{2}$ CH $_{2}$ Me), 7.18 (1H, dd, J9 and 2 Hz, H-7), 7.56 (1H, d, J9 Hz, H-8), 7.70 (1H, d, J2 Hz, H-5), 8.86 (1H, s, H-4), 8.93 (1H, s, H-1), 9.36 (1H, s, 6-OH),

H-5), 8.86 (1H, S, H-4), 8.93 (1H, S, H-1), 9.36 (1H, S, 6-OH), and 11.88 (1H, S, NH).

(3a): m.p. 270—272 °C; ¹H n.m.r. ([²H<sub>6</sub>]Me<sub>2</sub>SO) δ 1.40 (3H, t, J 7 Hz, CO<sub>2</sub>CH<sub>2</sub>Me), 3.01—3.84 (6H, m, H-2', H-3', H-4', H-5', and H-6'), 3.52 (3H, s, 4'-OMe), 4.42 (2H, q, J 7 Hz, CO<sub>2</sub>CH<sub>2</sub>Me), 4.77 (1H, t, J 5 Hz, 6'-OH), 5.04 (1H, d, J 7.5 Hz, H-1'), 5.30 (1H, d, J 4.5 Hz, 3'-OH), 5.46 (1H, d, J 4.5 Hz, 2'-OH), 7.38 (1H, dd, J 9 and 2 Hz, H-7), 7.64 (1H, d, J 9 Hz, H-8), 8.10 (1H, d, J 2 Hz, H-5), 8.91 (1H, s, H-4), 8.98 (1H, s, H-1), and 11.96 (1H s. NH). 11.96 (1H, s, NH).

(4a): m.p. 263—265 °C; ¹H n.m.r. ([²H<sub>8</sub>]Me<sub>2</sub>SO) δ 1.40 (3H, t, J 7 Hz, CO<sub>2</sub>CH<sub>2</sub>Me), 3.04—3.73 (6H, m, H-2', H-3', H-4', H-5', and H-6'), 4.41 (2H, q, J 7 Hz, CO<sub>2</sub>CH<sub>2</sub>Me), 4.90 (1H, m, 6'-OH), 5.05 (d, 1H, J7.5 Hz, H-1'), 5.38 (2H, m, 2'-OH and 3'-OH), 7.25 (1H, t, J7.8 Hz, H-6), 7.44 (1H, dd, J7.8 and 1.5 Hz, H-7), 8.09 (1H, dd, J7.8 and 1.5 Hz, H-7), 8.09 (1H, s, H-4), 9.02 (1H, s, H-1), and 11.88 (1H, s, NH).

chains of (1c) and (1d) to form, with concomitant ester saponification, the lactones (5) and (6), respectively.‡ Although these transformations proceeded in poor yields, most of the

‡ Formation of the lactone (6) (m.p. 325—328 °C) is a highly enantioselective process, the absolute configuration of (6), however, is undetermined. C.d. spectrum (Me<sub>2</sub>SO)  $\lambda$  274 ( $\Delta\epsilon$  –9.27), 308 (-0.881), 338 (+0.878), and 349 nm (+1.18).

starting material was recovered unchanged from the cultures and could be recycled.

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