One Step Synthesis of Octalones or Indanones by Friedel-Crafts Acylation

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Summary Acylation of cyclohexene (or cyclopentene) by vinylacetyl bromide in the presence of AlCl₃ gave a mixture of the octalones (or indanones) (3) and the propenyl cyclohexenyl (or cyclopentenyl) ketones (4).

THE Robinson annulation reaction has long been employed as a convenient route to fused ring ketones. We have found that bicyclodecenones and bicyclononenones can be prepared by a new and efficient one-step synthesis.

Treatment of the cycloalkenes (1) with but-3-enoyl bromide (2) in CH₂Cl₂ in the presence of AlCl₃ at low tempertures (-78 to 0 °C) gave a mixture [70-80% overall yield from (2)] of the bicyclic ketones (3) and the dienones (4).

$$[CH_{2}]_{n} | + Br$$

$$(1)$$

$$[CH_{2}]_{n} | + [CH_{2}]_{n}$$

$$(3)$$

$$(4a,b)$$

$$(4c)$$

$$a; n = 1, R = Me$$

$$b; n = 2, R = H$$

$$c; n = 2, R = Me$$

At low temperatures, chloro-ketone side-products were formed in <5% yield, but when experiments were performed at higher temperatures (ca. 0 °C), the main products were β -chloro-ketones, see equation (1).

$$(1b) + (2) \longrightarrow \bigcup_{Cl} 0$$

The formation of compounds (3) and (4) probably involves a Friedel-Crafts acylation of the cycloalkenes (1), followed by enolisation of the ketones (5), and a thermally allowed cyclisation of the enolates (6). Thermal cyclisation of (6) should be a disrotatory process,2 but stereospecific nonconcerted cyclisation has been observed.3

In a typical experiment anhydrous AlCl₃ (10 mmol) was stirred with CH₂Cl₂ (25 ml) and (2) (10 mmol) at room temperature until dissolution. The mixture was cooled to -78 °C and cyclohexene (2 ml, ca. 20 mmol) was added. The mixture was kept at -78 °C for $30\,\mathrm{h}$ and was then hydrolysed (basic work-up). With 1-methyl-cyclopentene and -cyclohexene the temperature was raised from -78 to 0 °C over 24 h. Yields and ¹H n.m.r. data for compounds (3) and (4) were as follows: (3a) 55%, δ (CCl₄) 6.8 [1H, ddd (110.2, 4.4, and 3.8 Hz), 5.92 [1H, d (110.2 Hz), t (110.2 Hz)2 Hz)], 2.25 (2H, dd), and 1.13 (3H, s); (3b) 35%; (3c) 45%: trans-isomer 60%, δ (CDCl₃) 6.76 [1H, ddd (J 10, 5.0, and 2.75 Hz], 5.94 [1H, ddd (J 10, <math>2.5, and 1.25 Hz)], 2.54 [2H, t (J 2.8 Hz)], and 1.0 (3H, s); cis-isomer 40%, δ (CDCl₃) 6.74, 5.92, 2.73 (2H, t), and 0.91 (3H, s); (4a) 45%; (4b) 65%; and (4c) 55%.

We observed that the acyl bromide (2) reacts rapidly at -78 °C with the alkenes (1), but that several hours were required for the ketones (5) thus formed to be converted

$$(1) \xrightarrow{+} [CH_2]_n \xrightarrow{|H_2O|} (4)$$

$$(3) \xrightarrow{H_2O} [CH_2]_n \xrightarrow{|H_2O|} (4)$$

$$(7)$$

into the cyclised enolates (7). This result may be related to the observed stereoselectivity in the Robinson annulation reaction when the solvent used is Me₂SO.6

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