## The Molecular Structure of an Isothiathiophthen

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Summary X-Ray crystallographic data for an isothiathiophthen support its formulation as a 1,3-dithiol-2ylidene thioketone rather than as a bicyclic system analogous to 6a-thiathiophthen.

X-Ray crystallographic studies have shown that the 6a-thiathiophthen system (I) is essentially planar, with the three sulphur atoms arranged almost linearly. The sum of the two S-S distances in each compound studied is nearly constant (ca. 4.7 Å) but, subject to this condition, the individual S-S distances show considerable variation, depending presumably on the substitution pattern. Nevertheless, 6a-thiathiophthens can be regarded as bicyclic

systems, admittedly of a special type, with a particularly interesting bonding situation at the central sulphur atom.<sup>2</sup>

No exact structural data have been reported for "isothiathiophthens", 3-5 the 1,3-dithiole analogues (II) of 6athiathiophthens, and the nature of the bonding in the isocompounds is unknown.

Certain similarities in spectra and in chemical behaviour between the two series may be noted. Thus, isothiathiophthens with aryl substituents [II;  $R^1 = R^3$  (or  $R^4$ ) = Ar,  $R^2 = R^4$  (or  $R^3$ ) = H] have maxima in their electronic absorption spectra at ca. 490, 340, and 265 nm,  $^3$ ,  $^6$  resembling 6a-thiathiophthens.  $^7$  The iso-compounds are thermally stable;  $^5$  they are converted  $^3$ ,  $^6$  into dithiolylidene ketones by

methods familiar in the 6a-thiathiophthen series, and with triethyloxonium fluoroborate they give S-ethyl derivatives,3 as do 6a-thiathiophthens. Nitrosation of the diphenylisothiathiophthen [II;  $R^1 = R^3$  (or  $R^4$ ) = Ph,  $R^2 = R^4$  (or R3) = H gives the nitroso-derivative of the corresponding 1,3-dithiolylidene ketone,6 exactly paralleling the behaviour of 2,5-diphenyl-6a-thiathiophthen.8

 $\beta = 110^{\circ}1(1)'$ ,  $\gamma = 89^{\circ}6(1)'$ ,  $V = 777\cdot3 \times 10^{-24} \text{ cm}^3$ ,  $\rho_{\underline{m}} = 1\cdot60 \text{ g cm}^{-3}$ , Z = 2,  $\rho_{c} = 1\cdot672 \text{ g cm}^{-3}$ . Space group P1. A total of 1144 non-zero independent reflexions with  $2 \theta < 90^{\circ}$ , was measured on an automatic diffractometer with  $Cu-K_{\alpha}$  radiation. While the shape and size of the crystal were not ideal for very accurate data, the structure, incorporating anisotropic temperature factors, has been refined to a conventional R-factor of 0.11 on the observed data. A view of the molecular structure is shown in the Figure. The e.s.d. for C-S and C-C lengths are 0.025 and 0.035 Å, respectively, while angles are  $\pm 2^{\circ}$ .

From the Figure, it can be seen that the p-bromophenyl group occupies the position corresponding to R4 in structure (II). The central group of eight atoms is approximately planar (maximum deviation 0.08 Å), with the phenyl ring and the p-bromophenyl ring making angles of 26° and 29° with this plane, respectively. The exocyclic  $S \cdots S$  distance is 2.91 (1) Å which, although much less than twice the van der Waals radius of sulphur (3.70 Å), 10 is considerably longer than any S...S distance found in a thiathiophthen; the longest such distance is 2.56 Å in one of the molecules of 5-pbromophenyl-3-benzoyl-2-methylthio-6a-thiathiophthen. 16,11 The transannular S · · · S distance in 1-acetonyl-1-thionia-5thiacyclo-octane perchlorate (V) is 3·13 Å, 12 although in this

FIGURE. Stereoscopic view of a single molecule along the b axis.

We now report structural data for the isothiathiophthen (II;  $R^1 = Ph$ ,  $R^2 = R^3 = H$ ,  $R^4 = p - Br \cdot C_6H_4$ ) prepared in the following way. p-Bromophenacyl bromide was condensed9 with the gem-dithiol (III), derived from acetophenone, to give the cyclic hemi-thioacetal (IV or geometrical isomer) and this product was converted directly, by the action of phosphorus pentasulphide in benzene, into the isothiathiophthen (dark purple needles from benzene, m.p. 200—203°),  $\lambda_{\text{max}}$  265, 343, and 490 nm ( $\epsilon$  24,880, 12,740, and 30,550).

The crystals are very fine needles elongated along the baxis. Crystal data:  $C_{17}H_{11}S_3Br$ , M=391.4, triclinic, a=17.775 (3), b = 7.190 (2), c = 6.474 (1) Å,  $\alpha = 90^{\circ}55(1)'$ , molecule, one of the sulphur atoms probably bears a substantial positive charge, while a  $S(3) \cdots S(4)$  distance of 2.966 Å was reported in (VI).13 The long S · · · S distance in (VI) is achieved by increasing the angles, a, b, and c from 120° to 127·1, 125·5, and 122·5°, respectively, whereas the corresponding angles in the isothiathiophthen are 131, 127, and 119°.

The C(11)-C(13), C(13)-C(14), and C(14)-S(15) lengths (Figure) are 1.26, 1.42, and 1.72 Å, respectively. These dimensions, and the S-S distance of 2.91 Å, are more in accord with a structure of type (II) than with a bicyclic structure analogous to structure (I) for 6a-thiathiophthen.

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