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Stereoselective synthesis of 4-substituted-cyclic sulfamidate-5-carboxylates by asymmetric transfer hydrogenation accompanied by dynamic kinetic resolution and applications to concise stereoselective syntheses of (—)-epi-cytoxazone and the taxotere side-chain†

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Ò

racemic

2: R = Me

4: R = Ph

0.5 mol% (R,R)-1a

HCO₂H/Et₃N

EA. 25°C, 0.5 h

0.5 mol%

(R,R)-1a HCO₂H/Et₃N

R=Ph

Dynamic kinetic resolution driven, asymmetric transfer hydrogenation reactions of cyclic sulfamidate imine-5-carboxylate esters were developed. Applications of the new methodology to stereoselective syntheses of the taxotere side-chain and (–)-epi-cytoxazone are described.

1,2-Amino alcohol motifs, including those found in β -amino- α -hydroxy acids, are present in a vast range of natural products and pharmaceutically related compounds. In addition, the relative and absolute stereochemistry of the 1,2-amino alcohol moiety generally governs the biological activities of these substances. Therefore, the development of methods for stereoselective synthesis of members of this family has received considerable attention. 1a,2

Transition metal catalyzed-asymmetric transfer hydrogenation reactions (ATH)³ of carbonyl compounds containing configurationally labile stereogenic C–H centers, accompanied by dynamic kinetic resolution (DKR), have become efficient and powerful techniques for controlling the stereochemistry at two contiguous stereogenic centers. Examples of processes of this type include ATH of α -substituted- β -ketoesters, 4 β -ketoamides, 5 α -alkoxy- β -keto phosphonates, 6 1,2-diketones, 7 α -ketoesters, 8 and α -ketophosphonates. However, only a few reports exist describing ATH reactions of imines that are accompanied by DKR. In this context, we recently described a highly efficient procedure for ATH–DKR of prochiral cyclic sulfamidate imines, using HCO₂H/Et₃N as the hydrogen source and chiral Rh-catalysts (Scheme 1). $^{10\alpha,b}$ In this early effort,

R

CO₂Me

(4R.5S)-3

(4R,5S)-5

(4S,5R)-3

major (75% ee) (4S,5R)-5

major (98% ee)

HN

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While considering other strategies to improve the stereoselectivity of ATH-DKR reactions of cyclic imine 2, we envisioned that introduction of a carboxylate group at C-5 would also enhance the acidity of H-5 and, as a result, would promote high levels of stereoselectivity in the ATH-DKR reaction.

Below, we describe the results of an investigation exploring this proposal, which led to the first examples of highly efficient ATH reactions of cyclic imines **6**, which are accompanied by

⁽⁴S,5S)-7a racemic 6a (4R.5R)-7a - Chiral transition metal-catalysts 1a-1e 0:5:0 Rů N H₂ CI CI (R,R)-1e: R= p-Tolyl (R,R)-1a: M=Rh (R,R)-1c: R= p-Tolyl (R.R)-1b: M=Ir (R.R)-1d: R= F_Ph Scheme 1 we showed that ATH of 4,5-disubstituted cyclic sulfamidate imines 2,

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 $[\]dagger$ Electronic supplementary information (ESI) available: Experimental procedures and characterization data with the copies of $^1\text{H-},\ ^{13}\text{C-NMR}$ spectra, chiral HPLC chromatograms of all chiral compounds and X-ray crystallography data of (S,S)-7j. CCDC 1007235. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c4cc06395c

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Scheme 2 Synthesis of 4-substituted sulfamidate imine-5-carboxylates 6.

DKR and can be applied in the synthesis of stereochemically enriched, chiral cyclic sulfamidate-5-carboxylate esters 7.

The racemic cyclic imine-5-carboxylate esters 6, used in this work, were prepared from α -hydroxy- β -keto ester (B) and sulfamoyl chloride by modification of a previously described procedure (Scheme 2).11

Racemic 4-phenyl-5-methoxycarbonyl cyclic imine 6a was selected as the model substrate in initial efforts aimed at the identification of the most suitable catalyst systems for the ATH reactions. Reactions of 6a were performed using the known chiral transition metal catalysts 1a-e (0.5 mol%) and employing HCO₂H/Et₃N as the hydrogen source in EtOAc at rt (Table 1).

As the results given in Table 1 show, the efficiencies and stereochemical outcomes of ATH reactions of 6a are strongly affected by both the nature of the transition metal and the diamine ligands. For example, reaction of this substrate using Ru-catalysts with diamine ligands bearing electron rich or electron deficient arylsulfonyl groups (1c,3c 1d,12 1e4a) proceed to very low conversions (Table 1, entries 3-5). However, ATH of 6a using Ir-catalyst 1b¹³ reaches completion in 0.5 h (conversion > 99%) but takes place with a low level enantioselectivity (30% ee). Finally, the results reveal that ATH-DKR of 6a with Rh-catalyst (R,R)-1a,14 which possesses TsDPEN and pentamethylcyclopentadienyl ligands, for 0.5 h at rt produces (S,S)-7a in the highest conversion (>99%) and level of stereoselectivity (>25:1 dr, 98% ee).

The influence of solvent on the ATH reaction of 6a catalyzed by (R,R)-1a was investigated. In most of the solvents tested (EtOAc, CH₂Cl₂, Cl(CH₂)₂Cl, CHCl₃, toluene, DMF, MeOH, THF, and 2-propanol), ATH of 6a takes place completely to form (S,S)-7a with high levels of stereoselectivity (>25:1 dr, 84-99% ee) (see, Table S2 in ESI†). For the purpose of experimental convenience and based on optimization of stereoselectivity, further ATH reactions were carried out in EtOAc as solvent.

Table 1 Optimization of chiral catalysts 1a-e for ATH-DKR of 6a^a

Entry	Cat.1	$\operatorname{Convn}^b\left(\%\right)$	dr (syn:anti)	ee^d (%)	Config.
1	(R,R)-1a	>99	>25:1 ^c	98	S,S
2	(R,R)-1 b	>99	$> 25:1^{c}$	30	S,S
3	(R,R)-1c	13	_	95	S,S
4	(R,R)-1d	6	_	_	_
5	(R,R)-1e	17	_	83	S,S

^a Reaction conditions: **6a** (0.5 mmol), cat-**1** (0.5 mol%), HCO₂H/Et₃N (5:2, 0.5 mL), EtOAc (5 mL), rt, 0.5 h. Determined by ¹H NMR analysis of crude products. ^c Only 4,5-cis products were detected by using ¹H NMR analysis of crude product mixtures. d Determined by chiral HPLC. e See Scheme S1 in ESI

Table 2 ATH-DKR of cyclic sulfamidate imine-5-carboxylates 6^a

0,_0		0,,0	0,,0
N, S, O	(R,R)-1a(0.5 mol%)	HN_g_O +	HN_g_O
4)\	HCO ₂ H/Et ₃ N(5:2)	4 5	4)—(5
R ₁ CO ₂ R ₂	EtOAc(0.1M), 25°C	R ₁ CO ₂ R ₂	R ₁ CO ₂ R ₂
6		(S,S)- 7	(R,R)- 7

		6		(S,S)-		(R,R)-7	
Entry	Substrate		Time	Conv.b	.b ee ^c	Conf.d	
	6, 7	R_1	R_2	(h)	(%)	(%)	
1	a		Ме	0.5	>99(92)	98	S,S^e
2	b	S. Y.	i-Pr	0.5	>99(85)	98	S,S
3	c	C A	Bn	0.5	>99(87)	98	S, S
4	d	C A	<i>t</i> -Bu	0.5	>99(87)	>99	<i>S</i> , <i>S</i>
5^f	a	<u>ک</u> ر	Me	0.5	>99(94)	98	R , R^f
6	e	Me X,	Me	12	20	_	_
7 ^g	e	Me	Me	12^g	>99(95) ^g	92 ^g	_
8	f	Me	Me	0.5	>99(99)	99	S,S
9	g	Me	Me	0.5	>99(99)	99	S,S
10	h	CI	Ме	0.5	>99(99)	97	S,S
11	i	CI	Ме	0.75	>99(92)	97	S,S
12	j	MeO	Me	0.5	>99(99)	>99	S,S^h
13 ⁱ	j	MeO	Me	3.5^{i}	>99 ⁱ	99 ⁱ	S,S^h
14	k	F	Me	0.5	>99(96)	97	S,S
15	1	F ₃ C	Ме	0.5	>99(88)	99	S,S
16	m	NC X	Me	0.5	>99(95)	96	S,S
17	n	MeO₂C	Me	0.5	>99(92)	97	S,S
18	o		Ме	0.5	>99(91)	97	S,S
19	p	0	Me	1.0	>99(92)	95	S,S
20	q	S	Me	2.0	>99(94)	99	S,S
21 22	r s	<i>n</i> -Pr– Ph(CH ₂) ₂ –	Me Me	1.5 12	>99(69) >99(54)	91 ^{<i>j</i>} 76	ND^k ND^k

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Table 2 (continued)

$\begin{array}{c ccccccccccccccccccccccccccccccccccc$							
Entry	6, 7	Substrate R ₁	R ₂	Time (h)	Conv. ^b (%)	ee ^c (%)	Conf.
	0, .	1	112	(11)	(,0)	(70)	
23	t	C St.	Me	24	50	_	ND^k
24^g	t	\(\sigma_{\frac{1}{2}\cdot_{\frac{1}{2}}}\)	Me	48 ^g	>99(67) ^g	47^{j}	ND^k

^a Reaction conditions: 6 (0.5 mmol), (R,R)-1a (0.5 mol%), HCO₂H/Et₃N (5:2, 0.5 mL), EtOAc (5 mL), rt. ^b Determined by ¹H NMR analysis of the crude product mixtures (isolated yields in parentheses). ^c Determined by using chiral HPLC. Only 4,5-cis products were detected by using ¹H NMR analysis of crude product mixtures. ^d Absolute configuration of **7b-i**, **7k-q** was determined by analogy to **7a** and **7j**. ^e See, Scheme S1 in ESI. f (g, g)-**1a** (0.5 mol%) was used. g 1:1 mixture of HCO₂H/Et₃N was used as the hydrogen source. ^h Determined by using X-ray crystallographic analysis (CCDC 1007235). i 0.1 mol% of (R,R)-1a (S/C = 1000)was used. j ee of ring-opened derivatives derived from 7r and 7t respectively (see, Scheme S2 in ESI). k Not determined.

The scope and limitations of the ATH-DKR reaction were explored using a variety of cyclic sulfamidate imine-5-carboxylates (6). All reactions were carried out in EtOAc (25 °C) under the optimized reaction conditions employing (R,R)-1a (0.5 mol%) as the catalyst and a 5:2 mixture of HCO₂H/Et₃N as the hydrogen source. The results are summarized in Table 2.

ATH of 6a with (R,R)-1a under the optimized reaction conditions produces a mixture of stereoisomeric 4,5-cis sulfamidates, in which the (4S,5S)-7a isomer predominates (98% ee, 92% yield, Table 2, entry 1). None of the 4,5-trans sulfamidates are detected in the crude product mixture by using ¹H-NMR spectroscopic analysis. These results show that hydrogen addition to 6a occurs exclusively from the less hindered face of the cyclic imine moiety. 10a In addition, ATH-DKR reactions of 4-phenyl-cyclic imine-5-carboxylates containing different ester moieties, such as isopropyl (6b) and benzyl (6c), also produce the corresponding cyclic sulfamidates (7b, 7c) with excellent efficiencies and stereoselectivities. Moreover, ATH-DKR of the t-butyl ester 6d forms nearly a single stereoisomer of the corresponding cyclic sulfamidate 7d.

ATH reaction of 6a under the optimized conditions, except in this case using the (S,S)-1a as catalyst, produces the antipodal sulfamidate (R,R)-7a with efficiency and stereoselectivity (98% ee, 94% yield) that match those of accompanying reaction using (R,R)-1a (Table 2, entry 5). These observations demonstrate that the source of dynamic kinetic resolution in this process is the configurational ability of the C-5 stereogenic center in 6a caused by rapid racemization under the reaction conditions ^{10a} (Scheme 3). As a result, the absolute stereochemistry of the major reduction product depends on the enantiomer of the Rh-catalyst employed in a manner such that (5S)-6a is preferentially reduced with (R,R)-1a to form sulfamidate (S,S)-7a and (5R)-6a is preferentially reduced with (S,S)-1a to give sulfamidate (R,R)-7a.

ATH-DKR of cyclic sulfamidate imines possessing either electron-withdrawing or -donating groups at the meta- or

Scheme 3 Proposed mechanism for DKR in ATH of 6a

para-positions on the phenyl ring leads to production of the corresponding sulfamidates in high yields and stereoselectivites. However, ATH of the cyclic imine 6e possessing an ortho-methyl substituted phenyl group is sluggish, reaching only 20% conversion even after 12 h. Based on the results of recent studies which show that the HCO₂H/Et₃N (F/T) ratio has a significant effect on both the ATH rate and the level of enantioselectivity, 5b,6,15 we employed the 1:1 instead of a 5:2 mixture of F/T as the hydrogen source for ATH of 6e. This reaction proceeds to completion in 12 h and is attended by a slightly decreased level of stereoselectivity (92% ee) (Table 2, entries 6 and 7). Cyclic imines containing heteroaromatic moieties also serve as suitable substrates for the ATH-DKR reaction, as exemplified by the results of reactions of furan 6p and thiophene 6q (Table 2, entries 19 and 20). Importantly, we also found that the catalyst loading can be reduced to 0.1 mol% (S/C = 1000) in the ATH-DKR reaction of 6j without deterioration of optical purity when the process is carried out using a longer reaction time (Table 2, entry 13). ATH reaction of 4-alkyl substituted cyclic sulfamidate imine-5-carboxylates was also explored. The results show that the efficiencies and stereoselectivities of the processes are sensitive to the steric bulkiness of the 4-alkyl group. Thus, ATH-DKR reaction of 4-(n-propyl) cyclic imine 6r is complete in 1.5 h (91% ee) but that of the 4-phenethyl containing cyclic imine 6s requires 12 h for completion and occurs with a lower level of stereoselectivity (76% ee) (Table 2, entries 21 and 22). ATH-DKR reaction of 4-cyclohexyl-substituted cyclic imine 6t is more sluggish resulting in only 50% conversion after 24 h. However, by employing a 1:1 mixture of HCO₂H/Et₃N as the hydrogen source, reaction of 6t reaches completion in 48 h but it takes place with a lower level of stereoselectivity (47% ee) (Table 2, entries 23 and 24).

The cyclic sulfamidates 7 produced in these reactions are valuable intermediates for the synthesis of various chiral β-amino-αhydroxy carboxylic acids or 1,2-functionalized amines16 such as those present in the side chain of the anticancer drug taxotere (10)17 and the potent cytokine modulator (-)-epi-cytoxazone $(12)^{2b,18}$ (Fig. 1).

Fig. 1 Examples of biologically important 1,2-amino alcohol compounds.

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$$(S,S)\text{-}7a \xrightarrow{a} \xrightarrow{BocN^{'}S \circ CO_2Me} \xrightarrow{Boc} \xrightarrow{B$$

Scheme 4 Reaction conditions: (a) (Boc)₂O, Et₃N, cat. DMAP, CH₂Cl₂, 100%; (b) (i) PhCO₂NH₄, DMF, 55 °C, 12 h; (ii) 1N HCl, CH₂Cl₂, 6 h, rt, 82%; (c) KCN, MeOH, 65 °C, 85%; (d) 1N NaOH, MeOH-THF, rt, 88%.

$$(S,S)-7] \stackrel{a}{=} \stackrel{\text{Boch}}{\overset{(S,S)-7]}{\overset{a}{=}}} \stackrel{\text{Boch}}{\overset{(S,S)-7]}{\overset{a}{=}}} \stackrel{\text{Boch}}{\overset{(S,S)-7}{\overset{($$

Scheme 5 Reaction conditions: (a) (Boc)₂O, Et₃N, cat. DMAP, CH₂Cl₂, 94%; (b) (i) PhCO₂NH₄, DMF, 55 °C, 12 h; (ii) 1N HCl, CH₂Cl₂, 6 h, rt, 100%; (c) KCN, MeOH, 65 °C, 86%; (d) NaBH₄, MeOH, rt, 92%; (e) NaH, THF, rt, 95%

In order to demonstrate the utility of the methodology developed in this effort, we employed it in the synthesis of the taxotere side-chain 10¹⁷ (Scheme 4).

Accordingly, (S,S)-7a formed by ATH-DKR reaction of 6a is converted to its N-Boc derivative, which upon treatment with PhCO₂NH₄ undergoes ring opening^{10a,19} to form (2R,3S)-8a. Selective removal of the O-benzoyl group in 8a using KCN¹¹ in MeOH and subsequent hydrolysis of methyl ester 9a produce the taxotere side-chain 10¹⁷ (ca. 61% overall yield over 4 steps from (S,S)-7a).

An additional example demonstrating the usefulness of the methodology is found in the synthesis of (-)-epi-cytoxazone (12) starting with (S,S)-7j (Scheme 5) 2b,18 (ca. 70% overall yield over 5 steps from (S,S)-7i).

In summary, a convenient and highly stereoselective method for the preparation of 4-substituted cyclic sulfamidate-5-carboxylate esters 7 was developed in this investigation. The process, involving asymmetric transfer hydrogenation accompanied by dynamic kinetic resolution (ATH-DKR), uses HCO₂H/Et₃N as the hydrogen source and chiral Rh catalysts (S,S)- or (R,R)-Cp*RhCl(TsDPEN). Most of the ATH-DKR reactions probed in this study occur rapidly (30 min) and highly stereoselectively under mild and experimentally convenient conditions (rt, without the need for solvent degassing or an inert atmosphere). The utility of this methodology was demonstrated by its application to stereoselective syntheses of the taxotere side-chain and (-)-epi-cytoxazone.

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