

## An Efficient Synthesis of Spiro[cyclohexane-1,3'-indol-2'(3'*H*)-ones] *via* Radical Cyclisation

Keith Jones,<sup>a\*</sup> Mervyn Thompson,<sup>b</sup> and Colin Wright<sup>a</sup>

<sup>a</sup> Department of Chemistry, King's College London, Strand, London WC2R 2LS, U.K.

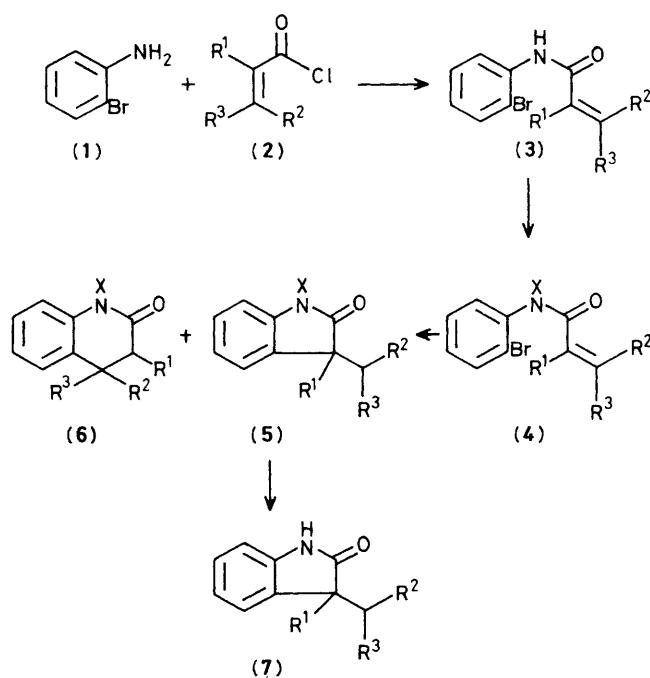
<sup>b</sup> Beecham Pharmaceuticals, Medicinal Research Centre, Coldharbour Road, The Pinnacles, Harlow, Essex CM19 5AD, U.K.

Treatment of the *o*-bromo-*N*-acryloylanilides (**4**) with tri-*n*-butylstannane leads to the formation of 3-substituted- and 3,3-disubstituted-2-oxindoles (**5**) in high yield.

In connection with our work directed towards the synthesis of the *Gelsemium* alkaloids,<sup>1</sup> we required a mild method for the formation of the 3-spiro-2-oxindole system late in the synthetic route. Although there are several methods available for the synthesis of 2-oxindoles in general<sup>2</sup> and 3-spiro-2-oxindoles in particular,<sup>3</sup> none of these methods seemed compatible with our planned synthetic approach. Recently,

radical cyclisation has been shown to be a powerful method for the construction of 5-membered rings under neutral, mild conditions<sup>4</sup> and we report now the application of this approach in a new synthesis of 2-oxindoles.

We envisaged forming the 2-oxindole ring by the cyclisation of the aryl radical generated from the *o*-bromo-*N*-acryloylanilide (**4**) onto the carbon-carbon double bond of the



- a; X = Me, R<sup>1</sup> = R<sup>2</sup> = R<sup>3</sup> = H  
 b; X = R<sup>3</sup> = Me, R<sup>1</sup> = R<sup>2</sup> = H  
 c; X = R<sup>2</sup> = R<sup>3</sup> = Me, R<sup>1</sup> = H  
 d; X = R<sup>1</sup> = Me, R<sup>2</sup> = R<sup>3</sup> = H  
 e; X = Me, R<sup>1</sup>, R<sup>3</sup> = -[CH<sub>2</sub>]<sub>4</sub>-, R<sup>2</sup> = H  
 f; X = SEM, R<sup>1</sup> = Me, R<sup>2</sup> = R<sup>3</sup> = H  
 g; X = SEM, R<sup>1</sup>, R<sup>3</sup> = -[CH<sub>2</sub>]<sub>4</sub>-, R<sup>2</sup> = H

Scheme 1. SEM = Me<sub>3</sub>SiCH<sub>2</sub>CH<sub>2</sub>OCH<sub>2</sub>.

$\alpha,\beta$ -unsaturated amide side chain as shown in Scheme 1. Reaction of *o*-bromoaniline (**1**) with the appropriately substituted acryloyl chloride (**2**) proceeded at 0 °C in ether to give the acryloylanilides (**3**).<sup>†</sup> Protection of the nitrogen<sup>‡</sup> was achieved with either methyl iodide [NaH, tetrahydrofuran (THF), room temp.] or 2-trimethylsilylethoxymethyl (SEM) chloride<sup>5</sup> (NaH, THF, reflux) to give the cyclisation substrates (**4**)<sup>†</sup> in high overall yields (Table 1).

Treatment of the *N*-substituted *o*-bromoacryloylanilides (**4**)<sup>†</sup> with tri-*n*-butylstannane and a catalytic amount of azoisobutyronitrile (AIBN) in toluene under reflux gave the 2-oxindoles (**5**)<sup>†</sup> in high yields (Table 1). As expected, cyclisation of compounds (**4a–c**) gave the 3-alkyl-2-oxindoles (**5a–c**) as the sole products confirming the high bias for 5-membered ring formation in such radical cyclisations.<sup>6</sup> Cyclisation of (**4d**) and (**4e**) gave an inseparable mixture of (**5d**)–(**6d**) and (**5e**)–(**6e**) respectively in which the 2-oxindole products predominated over the 2-dihydroquinolone products by 4:1 and 3:1 respectively.

<sup>†</sup> All new compounds gave satisfactory spectral data. All compounds were elaborated to known oxindoles for comparison.

<sup>‡</sup> Treatment of the *N*-unsubstituted acryloylanilides (**3**) with tri-*n*-butylstannane under the usual conditions gave no cyclisation products.

Table 1. % Yields<sup>a</sup> of compounds (**3**)–(**6**).

	(3)	(4)	(5)	(6)
a	73	87	79	0
b	84	94	72	0
c	89	98	80	0
d	90	80	72 <sup>b</sup>	18 <sup>b</sup>
e	95	74	69 <sup>b</sup>	22 <sup>b</sup>
f		81	70 <sup>b</sup>	15 <sup>b</sup>
g		85	80 <sup>b</sup>	9 <sup>b</sup>

<sup>a</sup> Yields are for chromatographically homogeneous material. <sup>b</sup> Yields by <sup>1</sup>H n.m.r. integration of inseparable mixture.

In view of the steric hindrance around the cyclisation terminus in (**4d**) and (**4e**), the predominant formation of the 5-membered ring cyclisation product is unusual.<sup>6</sup> This is probably caused by the shorter C–N bond lengths which, coupled with the conformational rigidity of the acryloylanilides (**4**), favour the 5-*exo-trig* cyclisation pathway.<sup>7</sup>

Interestingly, replacement of the *N*-methyl group by the *N*-SEM group [(**4f**) and (**4g**)] followed by radical cyclisation leads to formation of oxindoles and dihydroquinolones in a different ratio. Thus, cyclisation of (**4f**) under the usual conditions gives a 5:1 mixture of oxindole (**5f**) and dihydroquinolone (**6f**) in 85% yield. Similar reaction of (**4g**) gives a 8:1 mixture of spiro-2-oxindole (**5g**) and dihydroquinolone (**6g**) in 89% yield. It is apparent from the <sup>1</sup>H n.m.r. spectra of the cyclisation substrates (**4f**) and (**4g**) that the SEM group has a considerable effect on the conformation of these molecules and this no doubt contributes to the increased selectivity for 5-*exo-trig* cyclisation. Purification (by recrystallisation) and removal of the SEM group<sup>5</sup> gave the *N*-unsubstituted-2-oxindoles (**7f**) and (**7g**) in 81 and 80% yield respectively. Thus we have achieved the synthesis of the 3-spiro-2-oxindole (**7**) in 51% overall yield from *o*-bromoaniline (**1**).

We thank the S.E.R.C. and Beecham Pharmaceuticals for a C.A.S.E. award (C. W.).

Received, 21st October 1985; Com. 1502

## References

- J. A. Joule, in 'The Monoterpenoid Indole Alkaloids,' vol. 25, Part 4, of 'Heterocyclic Compounds,' ed. J. E. Saxton, Wiley Interscience, London, 1983, pp. 239–243.
- M. O. Terpko and R. F. Heck, *J. Am. Chem. Soc.*, 1979, **101**, 5281; J. F. Wolfe, M. C. Sleeve, and R. R. Goehring, *ibid.*, 1980, **102**, 3646, and references therein.
- R. F. Moore and S. G. P. Plant, *J. Chem. Soc.*, 1951, 3475; R. S. Johnson, T. O. Lovett, and T. S. Stevens, *J. Chem. Soc. C*, 1970, 796; R. J. Owellen, *J. Org. Chem.*, 1974, **39**, 69; N. A. Jonsson and P. Moses, *Acta Chem. Scand., Ser. B*, 1974, **28**, 225; A. S. Kende and J. C. Hodges, *Synth. Commun.*, 1982, **12**, 1; I. Fleming, M. A. Loreto, J. P. Michael, and I. H. M. Wallace, *Tetrahedron Lett.*, 1982, **23**, 2053.
- D. J. Hart, *Science*, 1984, **223**, 883.
- J. M. Muchowski and D. R. Solas, *J. Org. Chem.*, 1984, **49**, 203.
- A. L. J. Beckwith, *Tetrahedron*, 1981, **37**, 3073.
- A. Padwa, H. Nimmesgern, and G. S. K. Wong, *Tetrahedron Lett.*, 1985, **26**, 957; A. L. J. Beckwith and W. B. Gara, *J. Chem. Soc., Perkin Trans. 2*, 1975, 795.