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### ORGANIC LETTERS

2010 Vol. 12, No. 19 4292–4295

# Monanchocidin: A New Apoptosis-Inducing Polycyclic Guanidine Alkaloid from the Marine Sponge *Monanchora pulchra*

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Received July 22, 2010

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#### **ABSTRACT**

Monanchocidin (1), a guanidine alkaloid with an unprecedented skeleton system derived from a polyketide precursor, ( $\omega$ -3)-hydroxy fatty acid, and containing a 2-aminoethyl- and 3-aminopropyl-substituted morpholine hemiketal ring, has been isolated from the sponge *Monanhora pulchra*. The structure of 1 was assigned on the basis of detailed analysis of 1D and 2D NMR spectra, mass spectrometry, and results of chemical transformations. Compound 1 shows pro-apoptotic and cytoxic activities.

Polycyclic guanidine alkaloids are a unique class of spongederived metabolites exhibiting a broad range of biological activities such as cytotoxic, <sup>1–10</sup> antifungal, <sup>1,11</sup> antiviral, <sup>1,2,6,12–14</sup> antimicrobial, <sup>15</sup> antiprotozoal, <sup>11,15</sup> and antima-

- larial<sup>7,11</sup> activities. Members of this class include the series of pentacyclic<sup>1-6,10,14,16-18</sup> and tricyclic<sup>7,10,12,13,15,18</sup> guanidine
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alkaloids bearing the (5,6,8b)-triazaperhydroacenaphthalene skeleton.

In continuation of our search for new physiologically active marine natural products, we have found that extracts from the Far-Eastern sponge *Monanchora pulchra* (Lambe, 1894)<sup>19</sup> were cytotoxic against a human acute monocytic leukemia cell line (THP-1). Monanchocidin (1),<sup>20</sup> a cytotoxic

constituent of *M. pulchra*, was isolated from the frozen sponge (0.02% of dry weight) after extraction with EtOH, evaporation, partition between H<sub>2</sub>O and *n*-BuOH, partition of the BuOH-soluble materials between aqueous EtOH and hexane, and repeated column chromatography of the ethanol-soluble fraction over Sephadex LH-20 (EtOH) and HPLC (YMC-ODS-A column, 75%EtOH/0.1% aqueous TFA).

Table 1. NMR for Data Monanchocidin (1) in DMSO-d<sub>6</sub> and CD<sub>3</sub>OD

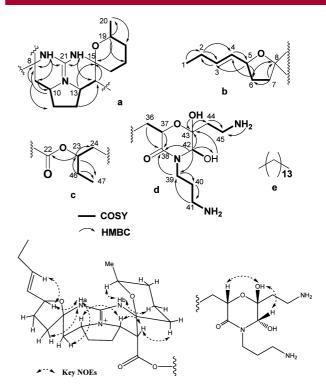
			1, CD <sub>3</sub> OD			
position	$\delta_{\mathrm{H}}$ (mult, $J$ in Hz)	$\delta_{\mathrm{C}}$ (DEPT)	COSY	HMBC (5 Hz)	$\delta_{\mathrm{H}}$ (mult, $J$ in Hz)	$\delta_{ m C}$
1	0.95 (t, 7.5)	$13.2~\mathrm{CH_3}$	H2	C2, C3	0.99 (t, 7.5)	14.3
2	2.01 (m)	$24.5~\mathrm{CH_2}$	H1, H3	C1, C3, C4	2.05 (m, 2H)	26.8
3	5.72 (dt, 6.4, 15.3)	$134.2~\mathrm{CH}$	H2, H4	C1, C2, C4, C5,	5.77 (dt, 6.4, 15.3)	136.9
4	5.42 (ddt, 1.5, 7.2, 15.3)	$129.2~\mathrm{CH}$	H3, H5	C2, C5	5.45 (ddt, 1.5, 7.2, 15.3)	130.7
5	4.48 (br. q, 7.2)	$79.5~\mathrm{CH}$	H4, H6a, H6b	C3	4.57 (br. q, 7.2)	82.4
6	1.70 (m)	$31.2~\mathrm{CH_2}$	H5, H6b, H7	C4, C5, C7, C8	1.78 (m)	33.5
	2.21 (m)		H5, H6a	C4, C7, C8	2.22 (m)	
7	2.12 (m)	$35.8~\mathrm{CH_2}$	H6a	C5, C6, C8		38.4
8		88.3 C				90.6
9	1.50 (m)	$37.6~\mathrm{CH_2}$	H9b, H10	C8, C10, C11	1.69 (m)	39.9
	2.25 (m)		H9a, H10	C8, C10, C11	2.27 (dd, 4.0, 13.0)	
10	3.91 (m)	$53.2~\mathrm{CH}$	H9a, H9b	C8, C9, C11, C21	4.02 (m)	55.4
11	1.44 (m)	$29.8~\mathrm{CH_2}$	H10, H11b	, , ,	1.64 (m)	
	2.22 (m)	2	H10, H11a	C12, C13	2.29 (m)	
12	1.61 (m)	$26.2~\mathrm{CH_2}$	H12b, H13	C14	1.77 (m)	28.2
	2.28 (m)		H12a, H13	C11, C13	2.29 (m)	
13	4.19 (m)	52.5 CH	H14	C12, C14	4.32 (m)	55.3
14	2.99 (d, 5.0)	49.3 CH	H13	C13, C15, C22	3.04 (d, 5.0)	51.5
15	2.55 (a, 5.0)	80.7 C	1110	010, 010, 022	5.04 (d, 5.0)	83.2
16	1.64 (m)	31.3			1.69 (m)	33.2
17	1.93 (m)	$17.6~\mathrm{CH_2}$	H16, H18a		1.82 (m)	20.0
18	1.35 (m) 1.18 (m)	$31.5 \text{ CH}_2$	1110, 1110a		1.27 (m)	33.2
10	1.62 (m)	31.5 C112	H18a, H19		2.24 (m)	55.2
19		CC E CH	*	C15 C17 C20		60.9
	3.74 (m)	66.5 CH	H18, H20	C15, C17, C20	3.86 (m)	69.2
20	1.08 (d, 6.2)	$21.6 \text{ CH}_3$	H19	C15, C17, C18, C19	1.13 (d, 6.4)	22.5
21	0.01()	148.5 C		014 015 001		151.1
21-N <u>H</u>	9.21 (s)			C14, C15, C21		
22	9.55 (s)	10000		C8, C9, C21		1500
22	4.55 ( )	168.3 C	TT 4.0	G00 G04 G46 G48	4.00 ( )	170.8
23	4.75 (m)	76.5 CH	H46	C22, C24, C46, C47	4.82 (m)	79.2
24	1.48 (m)	32.4	H23		1.58	34.8
25 - 34	1.20 - 1.25  (br  s)	28.9 - 29.3				
35	1.22 (m)	$28.9~\mathrm{CH_2}$			1.45 (m)	27.0
36	1.73 (m)	$31.9~\mathrm{CH_2}$	H37	C35, C38	1.79 (m)	34.0
					1.86 (m)	
37	4.08 (dd, 3.6, 7.8)	$70.8~\mathrm{CH}$	H36	$C36, C38, C43^a$	4.27 (dd, 3.6, 8.2)	73.5
38		$169.2~\mathrm{C}$				173.6
39	3.25 (m)	$41.7~\mathrm{CH_2}$	H39b, H40	C38, C40, C41, C42	3.46 (dt, 6.0, 14.2)	43.3
	3.45 (m)		H39a, H40	C38, C40, C41, C42	3.66 (m)	
40	1.80 (m)	$25.5~\mathrm{CH_2}$	H39a, H39b, H41	C39, C41	1.98 (m)	27.4
41	2.77 (m)	$36.7~\mathrm{CH_2}$	H40, $N_{\underline{H}_2}$ 41		2.95 (m)	38.6
$41-NH_2$	7.73 (br. s)		H41			
42	4.41 (d, 6.4)	80.9 CH	OH42	C38, C39, C43	4.59 (br. s)	83.3
42-OH	6.52 (d, 6.9)		$\overline{\text{H42}}$	C42, C43		
43		94.4 C				96.7
43-OH	6.61 (s)			C42, C43		
44	1.94 (m)	$34.7~\mathrm{CH_2}$	H45	C43, C45	2.12 (m)	36.5
	2.04 (m)	2	H45	C43, C42, C45	2.22 (m)	
45	2.93 (m)	$34.3~\mathrm{CH_2}$	H44a, H44b, NH <sub>2</sub> 45	-, - ,	3.18 (m)	36.7
$45-NH_2$	7.76 (br. s)		,, 11 <u></u> 210	C45	()	
46	1.52 (m)	$26.2~\mathrm{CH_2}$	H23, H47	C23, C47	1.60 (m)	28.3
- 0						
47	0.83 (t, 7.4)	$9.5~\mathrm{CH_3}$	H46	C23, C46	0.90 (t, 7.4)	10.6

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The molecular formula of monanchocidin (1) was established as  $C_{47}H_{83}N_6O_8$  on the basis of HRESIMS data (m/z859.6267, M<sup>+</sup>, calcd 859.6237,  $C_{47}H_{83}N_6O_8$ ) and  $^{13}C$  NMR data (Table 1). The mass of the fully deuterium-exchanged molecule (m/z 867) indicated the presence of 8 exchangeable protons. The <sup>1</sup>H and <sup>13</sup>C NMR data (DMSO-d<sub>6</sub>, Table 1) for 1 revealed the presence of a guanidine group ( $\delta_{\rm C}$  148.5 and  $\delta_{\rm H}$  9.21 and 9.55), three methyl groups ( $\delta_{\rm H}$  0.83, 0.95, 1.08;  $\delta_{\rm C}$  9.5, 13.2, 21.6), one disubstituted double bond ( $\delta_{\rm H}$  5.72, 5.42;  $\delta_{\rm C}$  134.2, 129.2), two *N*-substituted CH carbons ( $\delta_{\rm H}$ 3.91, 4.19;  $\delta_{\rm C}$  53.2, 52.5), five oxymethines ( $\delta_{\rm H}$  4.48, 3.74, 4.75, 4.08, 4.41;  $\delta_C$  79.5, 66.5, 76.5, 70.8, 80.9), two carbonyl groups ( $\delta_{\rm C}$  168.3 and 169.2), one carbonyl-linked methine  $(\delta_{\rm H} 2.99; \delta_{\rm C} 49.3)$ , three quaternary carbons  $(\delta_{\rm C} 88.3, 80.7,$ and 94.4) and an aliphatic long chain ( $\delta_{\rm H}$  1.20–1.25;  $\delta_{\rm C}$ 28.9 - 29.3).

Substructures **a**–**e** of **1** were established by COSY, HSQC, and HMBC experiments. Fragment **a** has been seen previously in many pentacyclic guanidine alkaloids  $^{1-6,10,14,16-18}$  isolated from marine sponges and starfish. It was revealed starting from signals of the methyl group in the tetrahydropyran moiety ( $\delta_{\rm H}$  1.13,  $\delta_{\rm C}$  22.5, CH<sub>3</sub>-20, CD<sub>3</sub>OD) and characteristic signals of the (5,6,8b)-triazaperhydroacenaphthalene core ( $\delta_{\rm C}$  151.1, C-21;  $\delta_{\rm H}$  4.02,  $\delta_{\rm C}$  55.4, CH-10;  $\delta_{\rm H}$  4.32,  $\delta_{\rm C}$  55.3, CH-13, CD<sub>3</sub>OD).

Interpretation of the COSY spectrum, in conjunction with the HSQC data, starting from the lower field methyl triplet ( $\delta_{\rm H}$  0.95; CH<sub>3</sub>-1) indicated substructure **b**, unusual in guanidine alkaloids (Figure 1), in which the  $\Delta^3$ -olefin was assigned as E on the basis of the coupling constant between H-3 and H-4 (J=15.3 Hz). The NMR data in DMSO- $d_6$  showed the absence of OH groups at C-5 and C-8, which



**Figure 1.** Partial structures of **1** with selected COSY, HMBC, and NOE correlations.

was also confirmed by the peracetylation of 1 (Ac<sub>2</sub>O, pyridine).  $^1H$  NMR chemical shifts of the characteristic signals of tetrahydrofuran moiety in 1 and monanchocidin peracetate (1a) were the same. Substructures "c" and "d" were established in the same manner. The position of the ethyl group in the polymethylene chain of 1 was assigned by HMBC experiment, which indicated that the CH<sub>3</sub>-47 signal at  $\delta_{\rm H}$  0.83 was correlated to C-46 (26.2) and C-23 (76.5) signals. The H-23 proton at 4.75 was also correlated to C-22 (168.3), C-24 (32.4), C-46 (26.2), and C-47 (9.5). Thus, the ethyl group is located at C-23.

The chemical shift of C-23 suggested an ester linkage at that point. The C-22 ester carbonyl showed correlations with H-14 as well as H-23. Thus, 1 consists of a pentacyclic guanidinium ring system (vessel part) and unusual, containing morpholine ring unit d (anchor part) were connected to each other through an ester linkage and a long-chain hydrocarbon moiety. The HRESIMS data of 1 show 13 methylene groups in the connecting chain (substructure e). The chemical shifts of the protons and carbons at 39, 40, 41, and 45 positions in the fragment d were comparable to those of the spermidine residue of ptilomycalin A.<sup>21</sup> The proton at  $\delta_{\rm H}$  4.08 (H-37) correlated with the carbonyl carbon at  $\delta_{\rm C}$  169.2 (C-38), the carbon signal at  $\delta_{\rm C}$  169.2 correlated with the protons on C-39 and a HMBC correlation between a hydroxyl proton at  $\delta_{\rm H}$ 6.61 and quaternary C-43 (94.4) indicated that C-43 is a hemiketal carbon. Analysis of key HMBC correlations then led to the construction of a morpholine hemiketal ring.

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<sup>(19)</sup> The sponge was collected by dredging during 36 scientific cruises of R/V "Academic Oparin", August 2008, near Urup Island (45°057,9 N; 150°44,9 E, depth 66 m).

<sup>(20)</sup> Monanchocidin (1): colorless oil;  $[\alpha]_D - 12$  (c 0.4, EtOH);  $^1H$ ,  $^13C$  NMR data, Table 1; HRESIMS m/z 859.6260  $[M]^+$  (calcd for  $C_{47}H_{83}N_6O_8$  859.6267). HRESIMS/MS of the ion  $[M]^+$  at m/z 859.6260: 758.5718  $[M-C_4H_9NO_2+2H]^+$ , 404.2338  $[M-C_{25}H_{50}N_3O_4+H]^+$ .

<sup>(21)</sup> Ohtani, I.; Kusumi, T.; Kakisawa, H.; Kashman, Y.; Hirsh, S. J. Am. Chem. Soc. 1992, 114, 8472–8479.

The relative stereochemistry of **1** was assigned by NOESY and ROESY. Diagnostic NOE correlations between the resonances of NHa 21 (9.55) and H-5, H-7, H-9 and NOEs from H-3 to H-5 were indicative of the relative configurations as  $5S^*$  and  $8R^*$  in substructure **b**. In addition,  $15S^*$  and  $19S^*$  relative configurations were suggested by diagnostic NOE correlations between NHb 21 (9.21) and H-19, H-17. An NOE between H-10 and H-13 in the NOESY and ROESY, together with the observed coupling constants between H-13 and H-14 (J = 5.0 Hz) located them all on the same side of the molecule. NOEs between OH-43 (6.61) and H-37 and between OH-43 and H-42 were observed, confirming the *trans*-position of hydroxyl groups and *cis*-position of H-37 and OH-43 in the anchor part. Configuration at C-23 was not determined.

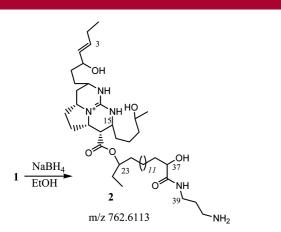


Figure 2. Reductive degradation of 1.

Treatment of 1 (NaBH<sub>4</sub>, EtOH, 60 °C, Figure 2) resulted in 2 by reduction of the hemiaminal groups at C-8, C-15

and C-42 with concomitant loss of C-42–C-45. The structure of **2** was established, using NMR, HRESIMS, and HRES-IMS/MS data. The <sup>1</sup>H NMR spectrum of **2** (DMSO- $d_6$ ) indicated the presence of a three new hydroxyl groups at C-5, C-19, and C-37 ( $\delta_{\rm H}$  4.77, 4.30 and 5.48), respectively.

The structure of monanchocidin (1) possesses a collection of uncommon features, including a vicinal hemiketal in a substituted 2-morpholinone ring formed by fusion of an  $\alpha$ -hydroxy acid and a highly oxidized spermidine unit C-41–C-45. The combination of two contiguous spiroring systems, a 1-oxa-6-azaspiro[4.5]decane and 1-oxa-7-azaspiro[5.5]undecane, is unprecedented among guanidine alkaloids. Finally, the long-chain substituted 2-morpholinone unit in 1 is unified with the complex bis-spiro-cyclic moiety through an ester linkage at the  $\omega$ -3 position of a hydroxy fatty acid residue and not the  $\omega$ -position, as encountered in related natural products.

Compound 1 demonstrates cytotoxicity against human leukemia THP-1 (IC<sub>50</sub> 5.1  $\mu$ M), human cervix epithelioid carcinoma HeLa (IC<sub>50</sub> 11.8  $\mu$ M), and mouse epidermal JB6 Cl41 (IC<sub>50</sub> 12.3  $\mu$ M) cell lines. It also induces 66% of early apoptosis in THP-1 cells at 3.0  $\mu$ M concentration.

**Acknowledgment.** We thank T. Molinski of the University of California, San Diego for reading this manuscript. The research described in this publication was supported by Grant NSS 3531.2010.4 from the President of RF and the Program of Presidium of RAS "Molecular and Cell Biology", Grant 09-04-00015-a from RFBR.

**Supporting Information Available:** Experimental procedures and full spectroscopic data for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

OL101716X

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