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2010 Vol. 12, No. 4 744-747

Total Synthesis of (—)-Exiguolide

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Received December 8, 2009

ABSTRACT

The first total synthesis of the naturally occurring enantiomer of exiguolide ((-)-1) has been completed. This very convergent synthesis features the following as main steps: (i) a Trost's ruthenium-catalyzed ene—yne cross-coupling reaction (this complex transformation allows the challenging control of the C5—C28 double bond geometry along with the stereoselective construction of the tetrahydropyran ring A) and (ii) a very efficient one-pot, two-step stereoselective conjugated allylic alcohol substitution that allowed the control of the C15 stereogenic center.

In 2006, the Ikegami group reported the isolation of (—)-exiguolide ((—)-1) from the marine sponge *Geodia exigua*. From a structural point of view, 1 is a macrolide that displays a number of salient motifs rendering this challenging target quite attractive for synthetic chemists. (—)-Exiguolide ((—)-1) is a 16-membered macrolactone bearing five C—C double bonds, and seven stereogenic centers. The macrocycle is fused with two tetrahydropyran rings A and B, ring A bearing a methoxycarbonylmethylidene function at C5 with a Z configuration. This target does not only represent a synthetic challenge, since biological tests revealed interesting properties. Thus, (—)-1 inhibits the fertilization of sea urchin gametes, which indicates that this compound could inhibit the fusion of viruses with cell membranes.

In 2008, Lee et al. reported the synthesis of *ent*-exiguolide ((+)-1), the enantiomer of the naturally occurring compound, thus establishing the right absolute configuration for this

product.³ They designed an approach featuring the macro-

cyclization by ring closing metathesis creating then the

C16-C17 double bond. The installation of the methoxycar-

bonylmethylidene function at C5 was made by a Horner-

in the abstract. We identified the control of the configuration of the methoxycarbonylmethylidene function at C5 as one of the main challenges of this synthesis. Focusing on this, it appeared from the literature that one of the most expeditious methodologies allowing full control of this double bond would be the ene—yne cross-coupling reaction described by Trost et al., which is catalyzed by the cationic Ru^{II} complex [CpRu(MeCN)₃]PF₆ (4). Furthermore, in the course of this reaction, the tetrahydropyran ring A would ideally be closed

Wadsworth—Emmons reaction but required a stoichiometric amount of an asymmetric phosphonoacetate to finally only deliver a *Z/E* mixture (5.8/1).

This original structure attracted our attention and teased our imagination leading to the retrosynthetic plan presented in the abstract. We identified the control of the configuration of the methoxycarbonylmethylidene function at C5 as one of the main challenges of this synthesis. Focusing on this, it

⁽¹⁾ Ohta, S.; Uy, M. M.; Yanai, M.; Ohta, E.; Hirata, T.; Ikegami, S. *Tetrahedron Lett.* **2006**, *47*, 1957–1960.

⁽²⁾ Ikegami, S.; Kobayashi, H.; Myotoishi, Y.; Ohto, S.; Kato, K. H. J. Biol. Chem. 1994, 269, 23262–23267.

⁽³⁾ Kwon, M. S.; Woo, S. K.; Na, S. W.; Lee, E. Angew. Chem., Int. Ed. 2008, 47, 1733–1735.

^{(4) (}a) Gill, T. P.; Mann, K. R. *Organometallics* **1982**, *1*, 485–488. (b) Trost, B. M.; Older, C. M. *Organometallics* **2002**, *21*, 2544–2546.

Scheme 1. Control of C15 Stereogenic Center by S_N2' and Synthesis of Fragment 3

stereoselectively in the course of an intramolecular oxa-Michael addition. One must notice that a similar framework occurs in some other important natural product families of high biological interest such as in the bryostatins.⁶ In 2008, as we were developing this study, Trost described the total synthesis of bryostatin 16 for which he made use of catalyst 4.⁷ For the construction of ring B we would use the classical reduction of the corresponding hemiketal. Securing the C15 stereochemistry was identified as another challenge, and here we thought it possible to perform a stereoselective S_N2′ reaction on an asymmetric Z-allylic alcohol activated as a phosphate 5. The macrolactonic ring would be closed by a classical Yamaguchi macrolactonization and we imagined bringing the C21–C27 side chain in the late steps of the synthesis by using a cross-metathesis reaction.

We commenced with the synthesis of the C6–C21 fragment 3 (Scheme 1). First, we accessed the enantio-enriched epoxide (+)-7 using the Jacobsen hydrolytic kinetic resolution (HKR)⁸ on the corresponding racemic epoxide *rac*-7 obtained by *m*-CPBA epoxidation of alkene 6. Alkyne 8 was furnished by reaction of epoxide (+)-7 with lithium trimethylsilylacetylide, followed by removal of the TMS alkyne protective group and protection of the alcohol function. The lithium acetylide of 8 was condensed with the known Weinreb amide 9.9 This efficient cross-coupling step afforded propargylic ketone 10 in 95% yield. The ketone function of 10 was reduced into alcohol 11 with a good diastereoselectivity by using Luche conditions. The semireduction of the alkyne function of 11, using the Lindlar

Next, the TBDPS protective group was selectively removed under basic conditions¹³ affording alcohol **14**, which was oxidized into aldehyde **15** under Swern conditions.¹⁴ The latter was transformed with use of the Luche¹⁵ procedure into homoallylic alcohol **16** (1/1 mixture of diastereoisomers)

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catalyst, led to Z allylic alcohol 12. Our S_N2' strategy for the control of the allylic C15 stereogenic center is based on the assumption that the constrained geometry of Z allylic alcohols blocks the rotation of the allylic C-C bonds, which is not the case with the corresponding E isomers. As a consequence, S_N2' on activated Z allylic alcohols would furnish only one product by attack of the nucleophile anti to the leaving group allowing then a good transfer of chirality (E isomers usually give mixtures). A similar strategy has been reported in the literature that made use of perfluorobenzoate esters as leaving groups. 11 Unfortunately we met troubles in synthesizing the required perfluorobenzoate ester from alcohol 12 as this ester appeared unstable and was obtained in poor yields. After many experiments, screening various leaving groups and cuprates, we finally found simple and efficient one-pot, two-step conditions allowing a direct transformation of the Z allylic alcohol 12 into the desired alkene 13 in high yield (96%) and a very good transfer of chirality (dr around 95:5 by ¹H NMR spectroscopy). ¹² In this protocol, alcohol 12 was first activated as phosphate 5 (KHMDS in Et₂O, and addition of ClPO(OEt)₂) and subsequently substituted by methyl cuprate (Me₂CuLi) formed in Et_2O . This protocol has been applied to various kinds of Z allylic alcohols with success.

⁽⁵⁾ Trost, B. M.; Frederiksen, M. U.; Rudd, M. T. Angew. Chem., Int. Ed. 2005, 44, 6630–6666.

⁽⁶⁾ Cossy, J. C. R. Chimie 2008, 11, 1477–1482.

⁽⁷⁾ Trost, B. M.; Dong, G. Nature 2008, 456, 485-488.

⁽⁸⁾ Tokunaga, M.; Larrow, J. F.; Kakiuchi, F.; Jacobsen, E. N. Science 1997, 936–938.

⁽⁹⁾ Funel, J.-A.; Prunet, J. J. Org. Chem. 2004, 69, 4555–4558.

^{(10) (}a) Luche, J.-L. J. Am. Chem. Soc. 1978, 100, 2226–2227. (b) Gemal, A. L.; Luche, J.-L. J. Am. Chem. Soc. 1981, 103, 5454–5459.

⁽¹¹⁾ Harrington-Frost, N.; Leuser, H.; Calaza, M. I.; Kneisel, F. F.; Knochel, P. *Org. Lett.* **2003**, *5*, 2111–2114.

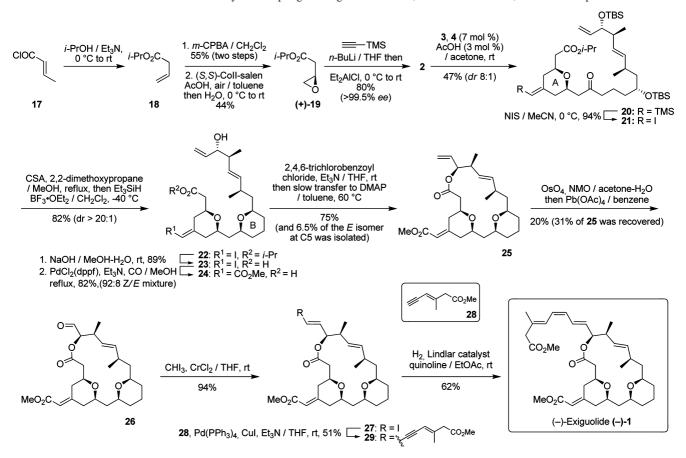
⁽¹²⁾ We have synthesized the unwanted diastereosomer of 13 in order to locate it in ¹H NMR spectra of 13 (see the Supporting Information for the full reference).

⁽¹³⁾ Kabat, M. M.; Lange, M.; Wovkulich, P. M.; Uskoković, M. R. *Tetrahedron Lett.* **1992**, *33*, 7701–7704.

⁽¹⁴⁾ Omura, K.; Swern, D. Tetrahedron 1978, 34, 1651-1660.

⁽¹⁵⁾ Pétrier, C.; Luche, J.-L. J. Org. Chem. 1985, 50, 910-912.

Scheme 2. Ruthenium-Catalyzed Coupling of Fragments 2 and 3, Macrolactonization, and Final Steps



and readily oxidized by Dess–Martin periodinane¹⁶ into β , γ -unsaturated ketone 3, the key "ene" coupling partner of the Ru^{II} catalyzed ene—yne coupling (53.4% over 11 steps from (+)-7).

On the other hand, the known "yne" partner 2^{17} was accessed from *trans*-crotonoyl chloride 17, which was transformed into β , γ -unsaturated ester 18 (Scheme 2). ¹⁸ An epoxidation by *m*-CPBA furnished *rac*-19, which was enantio-enriched by using the Jacobsen HKR method furnishing (+)-19. Finally, 2 was cleanly obtained after reaction with an aluminum acetylide nucleophile, ¹⁹ the more classical reaction of the corresponding lithium acetylide in the presence of BF₃ led to decomposition.

We have already used Ru^{II} catalyst **4** to build the tetrahydropyran ring of (+)-neopeltolide,²⁰ and discovered that a catalytic amount of acetic acid accelerated the reaction also allowing a remarkable enhancement of the diastereoselectivity. The same conditions of cross-coupling of alkyne **2** with alkene **3** led to tetrahydropyran **20** in a 47% yield as

a 8:1 mixture of diastereomers easily separated by HPLC. Anyway, this result compares favorably with the 34% yield (dr not given) obtained by Trost for his bryostatin 16 synthesis, on comparable substrates. Furthermore one should notice that in our case, the reaction was regioselective as the C20-C21 terminal double bond was not engaged in the cross-coupling. The TMS group of 20 being particularly acid sensitive, we performed its iodolysis leading to 21, prior to the acid-promoted removal of the two TBS protective groups and subsequent cyclization into the hemiketal precursor of cycle B. The crude mixture of hemiketals was subsequently reduced by Et₃SiH in the presence of BF₃OEt₂ at -40 °C affording compound 22 (dr >20:1) now featuring the two tetrahydropyran rings A and B. After saponification we obtained acid 23. Then, the methoxycarbonyl function was cleanly installed at C28 by a Pd⁰-catalyzed carbonylation²¹ of iodinated derivative 23 in MeOH furnishing the desired (Z)- α , β -unsaturated ester **24** (92:8 Z/E mixture). We proceeded then to the macrolactonization step using the Yamaguchi conditions²² and obtained macrolactone 25 in a good 74% yield (9.2% over 17 steps). The order of these steps (saponification—carbonylation—macrolactonization) is slightly

746 Org. Lett., Vol. 12, No. 4, 2010

^{(16) (}a) Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4156-4158.
(b) Dess, D. B.; Martin, J. C. J. Am. Chem. Soc. 1991, 113, 7277-7287.
(17) Shen, R.; Lin, C. T.; Bowman, E. J.; Bowman, B. J.; Porco, J. A., Jr. J. Am. Chem. Soc. 2003, 125, 7889-7901.

^{(18) (}a) Ozeki, T.; Kusaka, M. Bull. Chem. Soc. Jpn. 1966, 39, 1995–1998. (b) Roulland, E.; Ermolenko, M. S. Org. Lett. 2005, 7, 2225–2228.

⁽¹⁹⁾ Larchevêque, M.; Petit, Y. Bull. Chim. Soc. Fr. 1989, 130-139.

⁽²⁰⁾ Guinchard, X.; Roulland, E. Org. Lett. 2009, 11, 4700-4703.

⁽²¹⁾ Schoenberg, A.; Bartoletti, I.; Heck, R. F. J. Org. Chem. 1974, 39, 3318–3326.

⁽²²⁾ Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. Bull. Chem. Soc. Jpn. 1979, 52, 1989–1993.

unusual but it was the way we found to make the carbonylation reaction work in a reproducible manner.

For the last step of the synthesis it was first envisioned to introduce the C21–C27 side chain directly by a cross-metathesis reaction²³ on alkene **25**. Unfortunately, none of the catalysts (Grubbs II and Hoveyda–Grubbs II (Figure 1)),

Figure 1. Catalysts and alkenes used for the cross-metathesis reaction at C20.

cross-coupling partners (i, ii, 24 and iii 25 and allylic alcohol (Figure 1)), and solvents (CH₂Cl₂, toluene, rt or reflux) we tried gave any product, in all cases the starting material **25** was totally recovered and not even dimers of **25** were formed. We then reconsidered our retrosynthetic analysis, and we thought it reasonable to expect selectivity from the OsO₄ promoted dihydroxylation reaction²⁶ toward the less hindered and not electron-deficient C20–C21 double bond of **25**. Surprisingly we observed a poor selectivity and identified various diols and tetraols, some resulting from the unexpected dihydroxylation of the α , β -unsaturated ester at C5–C28, the C16–C17 double bond remaining untouched. A mild reaction with Pb(OAc)₄²⁷ finally delivered aldehyde **26** from this

mixture of diols with 20% yield over two steps and partial recovery of the starting material **25** (31%). The Takai—Utimoto²⁸ olefination reaction on aldehyde **26** furnished vinylic iodide derivative (-)-**27** in high yield, the enantiomer of which was described by Lee.³ In the two final steps of this synthesis we followed the Lee strategy. Thus we introduced the C22–C27 side chain by Sonogashira crosscoupling of **27** with alkyne **28**,^{3,29} which furnished dienyne **29**. The final step of semihydrogenation led to the targeted (-)-exiguolide (**1**), the characterization data of which are identical with those reported for the naturally occurring compound ($[\alpha]^{20}_D$ –95.0 (c 0.28, CHCl₃), lit.¹ [α]²⁰_D –92.5 (c 0.069, CHCl₃)).

In summary the first total synthesis of the naturally occurring enantiomer of exiguolide ((-)-1) has been completed. This very convergent synthesis furnished the target in 21 steps as the longest linear path (from (+)-7) in a 0.52% yield. This total synthesis features as main steps a Trost's ruthenium-catalyzed ene—yne cross-coupling reaction, allowing for control of the challenging C5—C28 double bond geometry, and a very efficient one-pot, two-step stereoselective conjugated allylic alcohol substitution used to control the C15 stereogenic center.

Acknowledgment. This work has been financially supported by the Institut de Chimie des Substances Naturelles (ICSN) and the Centre National pour la Recherche Scientifique (CNRS).

Note Added after ASAP Publication. Reference 15 was updated on January 22, 2010.

Supporting Information Available: Experimental procedures and characterization data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

OL902829E

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⁽²³⁾ Chatterjee, A. K.; Choi, T.-L.; Sanders, D. P.; Grubbs, R. H. J. Am. Chem. Soc. **2003**, 125, 11360–11370.

⁽²⁴⁾ Uno, B. E.; Gillis, E. P.; Burke, M. D. Tetrahedron 2009, 6, 3130–3138.

⁽²⁵⁾ Kujawa-Welten, M.; Pietraszuk, C.; Marciniec, B. Organometallics 2002, 21, 840–845.

⁽²⁶⁾ Schroeder, M. Chem. Rev. 1980, 80, 187-213.

⁽²⁷⁾ Bunton, C. A. In *Oxidation in Organic Chemistry Part A*; Wiberg K., Ed.; Academic: New York. 1965; pp 398–405.

⁽²⁸⁾ Takai, K.; Nitta, K.; Utimoto, K. J. Am. Chem. Soc. 1986, 108, 7408–7410.

⁽²⁹⁾ For the synthesis of the precursor of **28** see: Kigoshi, H.; Kita, M.; Ogawa, S.; Itoh, M.; Uemura, D. *Org. Lett.* **2003**, *5*, 957–960.