New Terpenoids from the Soft Corals Sinularia capillosa and Nephthea chabroli

ORGANIC LETTERS

XXXXVol. xx, No. x

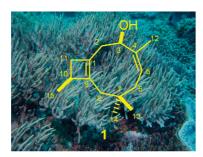
Shi-Yie Cheng,†,|| Ki-Jhih Huang,† Shang-Kwei Wang,‡ Zhi-Horng Wen,† Chi-Hsin Hsu,† Chang-Feng Dai,§ and Chang-Yih Duh*,†,||

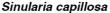
Department of Marine Biotechnology and Resources, National Sun Yat-Sen University, Kaohsiung, Taiwan, Department of Microbiology, Kaohsiung Medical University, Kaohsiung, Taiwan, Institute of Oceanography, National Taiwan University, Taipei, Taiwan, and Asia-Pacific Ocean Research Center, National Sun Yat-sen University, Kaohsiung, Taiwan

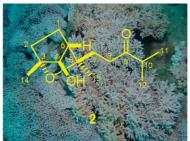
yihduh@mail.nsysu.edu.tw

Received August 11, 2009

ABSTRACT







Nephthea chabroli

Two new terpenoids, capillosanol (1) and chabranol (2), possessing unprecedented terpenoid skeletons, were isolated from the soft corals Sinularia capillosa and Nephthea chabroli, respectively. The structures of 1 and 2 were elucidated through extensive spectroscopic analyses. The cytotoxicities of these compounds were tested in vitro.

1833.

Soft corals, especially those of the genera Sinularia and Nephthea, have been well recognized as a rich source of sesquiterpenoids, providing a wide range of structural diversity¹⁻¹² and exhibiting various bioactivities such as

cytotoxic, 3,6-9 anti-inflammatory, 9,10 and antimicrobial properties.2,10

As part of a continuing search for bioactive substances from marine invertebrates, we explored the chemical investigations of the Formosan soft corals S. capillosa Tixier-

National Sun Yat-Sen University.

^{*} Kaohsiung Medical University.

[§] National Taiwan University.

^{II} Asia-Pacific Ocean Research Center.

⁽¹⁾ Blunt, J. W.; Copp, B. R.; Hu, W.-P.; Munro, M. H. G.; Northcote, P. T.; Prinsep, M. R. Nat. Prod. Rep. 2009, 26, 170.

⁽²⁾ Kamel, H. N.; Slattery, M. *Pharm. Biol.* **2005**, *43*, 253.
(3) Ahmed, A. F.; Kuo, Y.-H.; Dai, C.-F.; Sheu, J.-H. *J. Nat. Prod.*

⁽⁴⁾ Chao, C.-H.; Hsieh, C.-H.; Chen, S.-P.; Lu, C.-K.; Dai, C.-F.; Wu, Y.-C.; Sheu, J.-H. *Tetrahedron Lett.* **2006**, *47*, 2175.

⁽⁵⁾ Chao, C.-H.; Hsieh, C.-H.; Chen, S.-P.; Lu, C.-K.; Dai, C.-F.; Sheu, J.-H. Tetrahedron Lett. 2006, 47, 5889.

⁽⁶⁾ Cheng, S.-Y.; Dai, C.-F.; Duh, C.-Y. J. Nat. Prod. 2007, 70, 1449.

⁽⁷⁾ El-Gamal, A. A. H.; Wang, S.-K.; Dai, C.-F.; Duh, C.-Y. J. Nat. Prod. 2004, 67, 1455.

⁽⁸⁾ Wang, S.-K.; Duh, C.-Y. Chem. Pharm. Bull. 2007, 55, 762.

⁽⁹⁾ Cheng, S.-Y.; Wen, Z.-H.; Wang, S.-K.; Chiang, M. Y.; El-Gamal, A. A. H.; Dai, C.-F.; Duh, C.-Y. Chem. Biodiversity 2009, 6, 86.

⁽¹⁰⁾ Cheng, S.-Y.; Huang, Y.-C.; Wen, Z.-H.; Chiou, S.-F.; Wang, S.-K.; Hsu, C.-H.; Dai, C.-F.; Duh, C.-Y. Tetrahedron Lett. 2009, 50, 802.

⁽¹¹⁾ Anjaneyulu, A. S. R.; Gowri, P. M.; Krishna Murthy, M. V. R. J. Nat. Prod. 1999, 62, 1600. (12) Bowden, B. F.; Coll, J. C.; Mitchell, S. J. Aust. J. Chem. 1980, 33,

В

Durivault and *N. chabroli* Audouin, which were collected from the Dongsha Atoll and Siaoliouciou Island, respectively.

Chromatographic separation on the acetone extracts of the soft corals *S. capillosa* and *N. chabroli* resulted in the isolation of two new terpenoids, named as capillosanol (1) and chabranol (2), respectively (Figure 1).

Figure 1. Structures of metabolites 1 and 2.

The acetone extract of *S. capillosa* was concentrated to a brown gum, which was partitioned between EtOAc and H₂O. The EtOAc-soluble residue (60 g) was subjected to CC on silica gel using *n*-hexane—EtOAc mixtures of increasing polarity to yield 40 fractions. Fraction 14 (0.8 g) was applied to a C₁₈ gel column to obtain a mixture (72 mg) that was further purified by HPLC (LiChrosorb RP-18, 7 μ m, 25 × 250 mm), eluting with MeOH—H₂O (85:15) to yield **1** (2.0 mg). In the same manner, the EtOAc fraction (100 g) of the other soft coral *N. chabroli* was subjected to CC on silica gel to furnish 40 fractions. Fraction 16 (1.5 g) was fractionated over Sephadex LH-20 eluting with MeOH followed by RP-18 HPLC purification using MeOH—H₂O (75:25) as eluent to give **2** (1.0 mg).

Capillosanol (1)¹³ was obtained as a white amorphous powder. The positive HRESIMS of 1 exhibited a pseudomolecular ion peak at m/z 243.1727 [M + Na]⁺, consistent with the molecular formula of C₁₅H₂₄O, implying four degrees of unsaturation. Its IR spectrum absorptions at 3426 cm⁻¹ indicated the presence of a secondary hydroxyl, which was supported by the ¹H and ¹³C NMR signals (Table 1) resonating at $\delta_{\rm H}$ 4.77 (1H, br s, H-3) and $\delta_{\rm C}$ 71.0 (CH, C-3). Meanwhile, the HMBC correlations (Figure 2) observed from H₃-12 to C-3, C-4, and C-5 led to the position of the hydroxyl at C-3. The NMR spectra of 1 contained resonances for a trisubstituted double bond [$\delta_{\rm H}$ 5.59 (br s, 1H); $\delta_{\rm C}$ 139.5 (qC) and 122.3 (CH)] and a tetrasubstituted double bond [$\delta_{\rm C}$ 130.2 (qC) and 142.5 (qC)]. The above moieties accounted for two of the four degrees of unsaturation, indicating a bicyclic structure for metabolite 1.

From the COSY spectrum (Figure 2) of **1**, it was possible to establish the proton connects from H-3 to H_3 -15 through H_2 -2, H_2 -11, and H-10, and from H_2 -6 to H_3 -12 through H-5, as well as a long-range COSY correlation between H_2 -2 and

Table 1. ¹H and ¹³C NMR Spectroscopic Data of 1^a

	1		
C/H	¹³ C	$^{1}\mathrm{H}$	
1	$130.2~({ m qC})^b$		
2	$34.5 (CH_2)$	a: 1.76 m; b: 1.54 m	
3	71.0 (CH)	4.77 br s	
4	139.5 (qC)		
5	122.3 (CH)	5.59 br s	
6	53.1 (CH)	a: 2.31 d (15.5); ^c b: 2.02 d (15.5)	
7	37.2 (qC)		
8	54.0 (CH)	a: 2.20 d (15.0); b: 1.92 d (15.0)	
9	142.5~(qC)		
10	37.9 (CH)	2.25 m	
11	$32.1 (CH_2)$	α: 1.82 d (13.0); β: 1.40 d (13.0)	
12	$17.7 (CH_3)$	1.68 s	
13	$29.5 (CH_3)$	1.02 s	
14	$29.6 (CH_3)$	1.03 s	
15	$23.1 (CH_3)$	0.97 d (7.0)	
$_{3-\mathrm{OH}}$		3.78 d (4.0)	

 $[^]a$ Spectra were measured in CD₃COCD₃ ($^1\mathrm{H},500$ MHz and $^{13}\mathrm{C},125$ MHz). b Multiplicities are deduced by HSQC and DEPT experiments. cJ values (in Hz) are in parentheses.

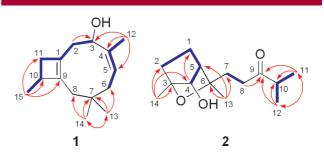


Figure 2. Selected ${}^{1}H-{}^{1}H$ COSY (—) and key HMBC (\rightarrow) correlations of 1 and 2.

H₂-11; H-5 and H₃-12 (Figure 2). The connectivities between C-9 and C-10; C-3 and C-4 were elucidated on the basis of the HMBC correlations from H₃-15 to C-9, C-10, and C-11 and from H₃-12 to C-3, C-4, and C-5. Moreover, the HMBC spectrum showed correlations from H₃-13/H₃-14 to C-6, C-7, and C-8, proving the attachment between C-6 and C-8 through C-7. Although there were no direct HMBC correlations available, the remaining one unsaturation indicated that C-8 should be linked to C-9. This assumption was further supported by the NOESY correlation from H-8a ($\delta_{\rm H}$ 2.21) to H₃-15 (Figure 3). The long-range COSY correlation between H₂-2 and H₂-11 is attributed to the W-type coupling $({}^{4}J_{2,11})$ of the highly strained ring system, which was further identified by the crucial NOESY correlation between H-11 β and H-2a and absence of the NOESY correlations between H₂-2 and Me-15. Accordingly, the planar structure of metabolite 1, possessing a bicyclo[7.2.0]undecane moiety, was proposed decidedly.

The geometry of the trisubstituted olefin was assigned as Z based on the NOESY correlations (Figure 3) between H-5 and H₃-12. The crucial NOE correlations between H-10 with

Org. Lett., Vol. xx, No. x, XXXX

⁽¹³⁾ Capillosaniol (1): white amorphous powder; $[\alpha]_{D}^{25} + 114$ (c 0.1, CHCl₃); IR (KBr) v_{max} 3426, 2928, 1655, 1449, 1373 cm⁻¹; ¹H NMR and ¹³C NMR data, see Table 1; ESIMS m/z 243 [M + Na]⁺; HRESIMS m/z 243.1727 [M + Na]⁺ (calcd for $C_{15}H_{24}ONa$, 243.1725).

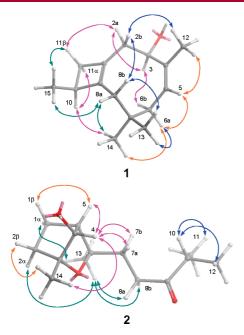


Figure 3. Key NOE correlations of 1 and 2.

H-11α ($\delta_{\rm H}$ 1.83) suggested that the two protons were oriented on the same side of the cyclobutene moiety, while H₃-15 was oriented on the opposite side. Moreover, H-2b ($\delta_{\rm H}$ 1.54) was found to show NOE correlations with H-8b ($\delta_{\rm H}$ 1.93) and H₃-12, and H-3 exhibited NOE correlations with H-6b ($\delta_{\rm H}$ 2.00) and H-2a ($\delta_{\rm H}$ 1.76), indicating the β-orientation of 3-OH. The above findings indicated the 3*R** and 10*S** configurations as depicted in Figure 3. The results, together with other detailed NOESY correlations (Figure 3) of 1, determined the structure of capillosanol as shown in the formula 1

Chabranol (2)¹⁴ was isolated as a colorless, viscous oil. HRESIMS of metabolite 2 exhibited a pseudomolecular ion peak at m/z 263.1625 [M + Na]⁺ and established a molecular formula of C₁₄H₂₄O₃, indicating three degrees of unsaturation. The ¹³C NMR (Table 1) displayed 14 carbon signals, which were identified by the assistance of the DEPT spectrum as four methyls, four methylenes, three methines, and three quaternary carbons. The ¹H NMR signal $[\delta_{\rm H} 4.13 \text{ (br s, 1H)}]$ (Table 2) and a broad IR absorption at 3437 cm⁻¹, together with the observation of one oxygen-bearing carbon resonance $(\delta_{\rm C}$ 79.3) in ¹³C NMR spectrum, revealed the presence of one hydroxyl. Furthermore, a keto-carbonyl carbon was recognized as being present in 2 from its ¹³C NMR signal at $\delta_{\rm C}$ 212.2 (qC, C-9), as well as from a strong IR absorption at 1714 cm⁻¹. By interpretation of COSY correlations (Figure 2), it was possible to establish three partial structures from H_2 -2 to H-4 through H_2 -1 and H-5 and from H_3 -11 to H_3 -12 through H-10, as well as COSY correlation between H₂-7 and H₂-8. The connectivities of these partial structures were

Table 2. ¹H and ¹³C NMR Spectroscopic Data of **2**^a

	2	
C/H	¹³ C	$^{1}\mathrm{H}$
1	$22.5 \; (\mathrm{CH_2})^b$	α: 1.84 m; β: 2.64 m
2	$33.0 (CH_2)$	α: 1.44 m; β: 1.62 m
3	84.5 (qC)	
4	79.3 (CH)	4.13 br s
5	50.2 (CH)	$2.00 \mathrm{\ br\ s}$
6	78.7 (qC)	
7	$36.0 (CH_2)$	a: 1.91 m; b: 1.57 m
8	$35.4 (CH_2)$	a: 2.63 m; b: 2.35 m
9	212.2 (qC)	
10	41.7 (CH)	2.58 m
11	$19.5 (CH_3)$	$1.09 d (7.2)^c$
12	$19.3 (CH_3)$	1.11 d (7.2)
13	$24.0 (CH_3)$	1.16 s
14	$18.1 (CH_3)$	1.20 s

 $[^]a$ Spectra were measured in CDCl₃ (1 H, 400 MHz and 13 C, 100 MHz). b Multiplicities are deduced by HSQC and DEPT experiments. cJ values (in Hz) are in parentheses.

further established by the HMBC correlations (Figure 2). Moreover, the HMBC correlations observed from H₂-8/H₃-11/H₃-12 to C-9 indicated the position of the keto-carbonyl group at C-9. To confirm the position of the ether linkage, **2** was submitted to acetylation with Ac₂O in pyridine at room temperature overnight. Formation of monoacetylated derivative **2a**¹⁵ proved the presence of a secondary hydroxyl in the original structure. The HMBC correlations observed from H₃-14 to C-2/C-3/C-4 led to the assignment of the hydroxyl at C-4. Indeed, the position of the ether linkage at C-3/C-6 was confirmed by the above observations. Thus, the gross structure of **2**, possessing a cyclopentane ring fused to a tetradrofuran ring at C-3 and C-6, was elucidated unmistakably.

The relative configuration of 2 was determined through inspection of the NOESY spectrum as well as a computergenerated lower energy conformation using MM2 force field calculations (Figure 3). From the NOESY spectrum of 2, $H-1\beta$ was found to show an NOE correlation with H-5, and $H-2\beta$ exhibited an NOE correlation with H_3 -14, indicating the β -orientations of H-5 and H₃-14. In addition, H-4 was determined as α on the basis of the analysis of coupling constants and splitting patterns of H-5. This finding was supported by the observation of a very small coupling constant (close to zero) between H-4 and H-5, implying the dihedral angle between the above two protons was almost 90°, consistent with the observation in the computer-modeled structure of 2. Furthermore, the NOE correlations could be observed between H-1 α /H₃-13 and H-2 α /H₃-13. Thus, H₃-13 should be placed on the α face. The above findings indicated the $3R^*,4R^*,5S^*,6R^*$ configuration as depicted in

Org. Lett., Vol. xx, No. x, XXXX

⁽¹⁴⁾ Chabranol (2): colorless, viscous oil; $[\alpha]^{25}_D$ –56 (c 0.1, CHCl₃); IR (KBr) ν_{max} 3437, 2969, 2937, 1714, 1458, 1374 cm⁻¹; ¹H NMR and ¹³C NMR data, see Table 1; ESIMS m/z 263 $[M+Na]^+$; HRESIMS m/z 263.1625 $[M+Na]^+$ (calcd for $C_{14}H_{24}O_3Na$, 263.1623).

⁽¹⁵⁾ $4(R^*)$ -Acetoxychabranol (**2a**): colorless, viscous oil; $[\alpha]^{25}_D$ -38 (c 0.1, CHCl₃); selected ¹H NMR (CDCl₃, 400 MHz) δ 4.92 (1H, br s, H-4), 2.63 (1H, m, H-8a), 2.39 (1H, m, H-8b), 2.60 (1H, m, H-10), 2.44 (1H, m, H-1 β), 2.28 (1H, br s, H-5), 2.08 (3H, s, 4-OAc), 1.21 (3H, s, Me-14), 1.17 (3H, s, Me-13), 1.11 (3H, d, J = 6.8, Me-12), 1.10 (3H, J = 6.8, Me-11); ESIMS m/z 305.4 [M + Na]⁺.

Figure 3. On the basis of the above observations and other detailed NOESY correlations (Figure 3), the structure of chabranol (2) was established unambiguously.

It is worthwhile to mention that metabolite 1 has a previously unknown carbon skeleton. We propose the name "capillosane" for this new skeleton. Farnesyl pyrophosphate may be involved in the biosynthesis of compound 1 through cyclization, oxidation, 1,3-hydrogen shift, 1,2-methyl migration, 1,2-hydrogen shift, and deprotonation to result in the formation of a capillosane-type skeleton (see the Supporting Information). A possible biosynthetic pathway for the loss of a carbon fragment from cyclopentane sesquiterpene by enzymatic oxidative modifications that could provide a cyclopentane norsesquiterpene skeleton of 2 was postulated.

Compounds 1, 2, and 2a were evaluated for cytotoxicity assays against P-388 (mouse lymphocytic leukemia), A-459 (human lung carcinoma), and HT-29 (human colon adenocarcinoma) cancer cell lines. Compounds 2 and 2a displayed moderate cytotoxicity against P-388, with an ED₅₀ of 1.81

and 3.03 μ g/mL, respectively. With the exception of the above findings, the obtained negative results showed that they were not cytotoxic against these cancer cell lines (ED₅₀ > 50 μ g/mL). The in vitro cytotoxic assays were carried out according to the procedure described previously.¹⁶

Acknowledgment. Financial support was provided by the National Science Council (NSC96-2320-B-110-003-MY3) and Ministry of Education (97C031703) of the Republic of China (Taiwan) awarded to C.-Y.D.

Supporting Information Available: ¹H NMR, ¹³C NMR, COSY, NOESY, HMQC, and HMBC spectra for **1** and **2**. This material is available free of charge via the Internet at http://pubs.acs.org.

OL901864D

(16) Hou, R.-S.; Duh, C.-Y.; Chiang, M. Y.; Lin, C.-N. J. Nat. Prod. 1995, 58, 1126.