Registry No. Carbon, 7440-44-0.

Literature Cited

- (1) Porter, W. R. Environ. Sci. Technol., preceding paper in this issue.
- (2) Cornish-Bowden, A. "Principles of Enzyme Kinetics"; Butterworths: Boston, 1979.
- Weber, W. J., Jr., "Physicochemical Processes for Water Quality Control"; Wiley: New York, 1972.
- (4) Gaudy, A. F., Jr.; Gaudy, E. T. "Microbiology for Environmental Scientists and Engineers"; McGraw-Hill: New York, 1980.
- (5) Corwin, D. L.; Farmer, W. J. Environ. Sci. Technol. 1984, 18, 507-514.
- (6) Dalang, F.; Buffle, J.; Haerdl, W. Environ. Sci. Technol. **1984**, 18, 135-141.

Richard W. Walters*

Department of Civil Engineering University of Maryland College Park, Maryland 20742

Richard G. Luthy

Department of Civil Engineering Carnegie-Mellon University Pittsburgh, Pennsylvania 15213

Comment on "Acute Toxicity Screening of Water Pollutants Using a Bacterial Electrode"

SIR: This concerns an article by Dorward and Barisas (1) that was recently published in ES&T.

First, I do not understand the basis for the apparent inconsistent comparison of three methods of measuring acute toxicity of materials in aquatic systems. For two of the methods (fish tests and bioluminescent bacteria tests), results are reported as median-effect concentrations (LC50's and EC50's, respectively). For the authors' proposed method with Escherichia coli, however, "inhibition levels of 40% were chosen..." It seems to me an arbitrary and illogical choice, one that serves only to bring the \tilde{E} . coli test results more in line with fish and bioluminescent bacteria test results.

Second, because only 7 of the 12 substances tested with E. coli "...yielded dose [concentration]-effect relationships which have provided useful toxicological information", I question the validity of the proposed method. A test that produces useful information <60% of the time is not practical.

Literature Cited

(1) Dorward, E. J.; Barisas, B. G. Environ. Sci. Technol. 1984, 18, 967-972.

Rod Parrish

Environmental Research Laboratory U.S. Environmental Protection Agency Gulf Breeze, Florida 32561

SIR: I should like to respond to Parrish's letter (1) concerning our recent article which appeared in Environmental Science & Technology (2).

In response to Parrish's first query, we compared results from the various toxicity assays in as direct a way as possible. For our bacterial electrode assay, we report toxicant levels that inhibit Escherichia coli CO2 production by 40%. This level was chosen because it represents approximately half the maximum inhibition observed among various toxicants in short-term experiments. The Beckman Microtox assay results are directly comparable quantities, namely, toxicant levels that reduce bacterial bioluminescence 50% after 15 min. LC50 values obtained in rainbow trout bioassays are, of course, difficult to compare directly with results of either of the above methods. In any case, we present, for all assays, toxicant levels that reduce the parameter of interest—CO₂ production, light production, or organismal viability—by 40-50% in the test

With regard to Parrish's second point, we do not propose the E. coli electrode as a practical substitute for any existing toxicity assay. Our data show, as he notes above, that various substances highly toxic to higher organisms have little effect on the respiration of this bacterium. Our motivation in presenting the E. coli electrode was rather, as we stated, to provide data "to guide the development of other instrumental toxicity bioassays".

Literature Cited

- (1) Parrish, R. Environ. Sci. Technol., preceding paper in this
- (2) Dorward, E. J.; Barisas, B. G. Environ, Sci. Technol. 1984. 18, 967-972.

B. George Barisas

Department of Chemistry Colorado State University Fort Collins, Colorado 80523