Highly Efficient Asymmetric Synthesis of β -Amino Acid Derivatives via Rhodium-Catalyzed Hydrogenation of β -(Acylamino)acrylates

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Enantiomerically pure β -amino acids and their derivatives are important building blocks for the synthesis of β -peptides, β -lactam antibiotics, and many important drugs (e.g., taxol side chain). Peptides containing β -amino acids show increased stability to enzymatic hydrolysis and are being evaluated as promising pharmaceutical products. Recent studies show that a variety of β -peptides can fold in a specific conformation, and these may lead design of molecules with unique properties.2 For these reasons, enantioselective synthesis of β -amino acids has attracted extensive interest. Although several stoichiometric and catalytic methods have been developed to make chiral β -amino acids,³ straightforward hydrogenation of 3-aminoacrylic acid derivatives represents one of the simplest and efficient routes. However, previous attempts at this asymmetric hydrogenation using Rh and Ru catalysts led to poor enantioselectivity.4 Herein, we report the first highly enantioselective hydrogenation of β -(acylamino)acrylates for the synthesis of β -amino acid derivatives using Rh-BICP and Rh-DuPhos catalytic systems (Figure 1).

We have previously demonstrated that Rh–BICP complexes are highly enantioselective catalysts for hydrogenation of dehydroamino acids and E/Z mixtures of

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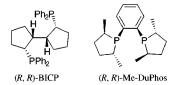


Figure 1.

^a Conditions: (a) NH₄OAc, MeOH, rt; (b) Ac₂O, py, THF, reflux.

enamides.⁵ Another catalytic system for hydrogenation of an isomeric mixture of (Z)- and (E)-enamides with high enantioselectivity is Burk's Rh–DuPhos catalyst.⁶ Prompted by these results, we have applied this catalyst to hydrogenation of 3-aminoacrylic acid derivatives 1. Substrates 1a-1h can be conveniently made from the corresponding β -keto esters **3** to give 3-amino-2-alkenoates 4, followed by acylation (Scheme 1).4 The E|Zmixture of enamides (1a-1g) can be separated by column chromatography except for substrate 1h. To search for the optimal conditions for the hydrogenation reaction, we have performed a number of experiments using methyl 3-acetamino-2-butenoate (1a) as a prototypical substrate (Table 1). Acylation of the parent β -amino crotonate with Ac_2O in THF gave a 1:1.6 mixture of (E)-1a and (Z)-1a, which can be separated on a silica gel column with EtOAc/hexanes as eluent. Unlike the results obtained with a Ru-BINAP catalyst in which (Z)-1a was hydrogenated faster than (*E*)-**1a** in a competitive experiment, ^{4a} (E)-1a is more reactive than (Z)-1a in the Ru-BICPcatalyzed hydrogenation. Under 40 psi of H₂, (E)-1a was completely reduced to (*R*)-2a in 24 h (entry 3), while (*Z*)-1a is not reactive under these conditions. Higher H₂ pressure (294 psi) is required for complete reduction of (Z)-1a (entry 14). Different catalyst precursors were also tested for hydrogenation of (*E*)-1a, and a cationic Rh(I) species gave better results than a neutral Rh(I) complex (entries 1−3). A change of H₂ pressure has no obvious effect on the enantioselectivity in the hydrogenation of (*E*)-**1a** (entries 3-5). A study of solvent effects indicates that toluene is a good solvent (entries 6-10 and entry 3), while the ee is significantly lower in MeOH (entry 10). Therefore, the optimal conditions for hydrogenation of (*E*)-**1a** using a Rh–BICP catalyst was as shown in entry 3. Under these experimental conditions, other bidentate chiral phosphines were tried for hydrogenation of (E)-**1a**. While hydrogenation of (*E*)-**1a** with Rh–DIOP (entry 11, 79.0% ee) and Rh-BINAP species (entry 12, 69.0% ee) gave lower ee's than the result with the Rh-BICP complex (entry 3, 96.1% ee), a slightly higher ee was obtained with a Rh-Me-DuPhos catalyst (entry 13,

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Table 1. Ligand Effect on Rh-Catalyzed Asymmetric Hydrogenation of an Enamide^a

| entry | ligand | substrate | Rh(I) precursor | solvent | H ₂ pressure, psi | conv, % | % ee ^b |
|-------|-----------------|-----------|--|------------|------------------------------|---------|-------------------|
| 1 | (R,R)-BICP | (E)-1a | [Rh(COD)Cl] ₂ | toluene | 40 | 94.5 | 63.2 |
| 2 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]BF ₄ | toluene | 40 | 100 | 95.1 |
| 3 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | toluene | 40 | 100 | 96.1 |
| 4 | (R,R)-BICP | (E)-1a | [Rh(COD)2]OTF | toluene | 14.7 | 82.8 | 94.9 |
| 5 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | toluene | 735 | 100 | 95.9 |
| 6 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | hexanes | 40 | 100 | 91.0 |
| 7 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | benzene | 40 | 100 | 95.9 |
| 8 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | THF | 40 | 100 | 94.4 |
| 9 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | CH_2Cl_2 | 40 | 100 | 93.5 |
| 10 | (R,R)-BICP | (E)-1a | [Rh(COD) ₂]OTf | MeOH | 40 | 100 | 85.4 |
| 11 | (+)-DIOP | (E)-1a | [Rh(COD) ₂]OTf | toluene | 40 | 100 | 79.0 |
| 12 | (R)-BINAP | (E)-1a | [Rh(COD) ₂]OTf | toluene | 40 | 100 | 69.0 |
| 13 | (R,R)-Me-DuPhos | (E)-1a | [Rh(COD) ₂]OTf | toluene | 40 | 100 | 99.3 |
| 14 | (R,R)-BICP | (Z)-1a | [Rh(COD) ₂]OTf | toluene | 294 | 100 | 88.6 |
| 15 | (R,R)-Me-DuPhos | (Z)-1a | [Rh(COD) ₂]OTf | toluene | 294 | 100 | 63.7 |

^a The reaction was conducted at rt under H_2 for 24 h. The catalyst was made in situ by stirring a solution of Rh(I) precursor and ligand for 30 min {[substrate (0.5 mmol, 0.125 M):[Rh(COD)₂]OTf:ligand = 1:0.01:0.011]}. ^b Enantiomeric excesses were determined by GC using a CHIRAL SELECT 1000 column.

Table 2. Rh-Catalyzed Asymmetric Hydrogenation of Enamide for the Synthesis of β -Amino Acid Derivatives^a

| | RO'OC | | _COOR' | | | | |
|----------------|---------------------|--------------------------|--------------|--------------|------------------------------|-------------------|---------|
| | Ì | (<i>E</i>)-1 or | (Z)-1 | [Rh(CO | D)2]OTf + L | 2 | |
| | R ^{''} N | • • • • | R NHAc | toluene | , rt, H ₂ R NHAc | 2 | |
| entry | ligand | substrate | R | R' | H ₂ pressure, psi | % ee ^b | conv, % |
| 1 | (R.R)-BICP | (E)- 1b | Et | Me | 40 | 96.8 (R) | 100 |
| $\overline{2}$ | (R.R)-Me-DuPhos | (E)- 1b | Et | Me | 40 | 99.6 (R) | 100 |
| 3 | (R,R)-BICP | (Z)-1b | Et | Me | 294 | 86.9 (R) | 100 |
| 4 | (R,R)-Me-DuPhos | (Z)-1b | Et | Me | 294 | 21.2 (R) | 100 |
| 5 | (R,R)-BICP | (<i>É</i>)- 1 c | <i>i</i> -Bu | Me | 40 | 90.9(R) | 100 |
| 6 | (R,R)-Me-DuPhos | (<i>É</i>)- 1 c | <i>i</i> -Bu | Me | 40 | 98.5 (R) | 100 |
| 7 | (R,R)-BICP | (Z)-1c | <i>i</i> -Bu | Me | 294 | 92.9(R) | 93 |
| 8 | (R,R)-Me $-$ DuPhos | (Z)-1c | <i>i</i> -Bu | Me | 294 | 62.4 (R) | 96 |
| 9 | (R,R)-BICP | (<i>E</i>)-1d | Me | Et | 40 | 96.0 (R) | 100 |
| 10 | (R,R)-Me $-$ DuPhos | (<i>E</i>)-1d | Me | Et | 40 | 98.7 (R) | 100 |
| 11 | (R,R)-BICP | (Z)-1d | Me | Et | 294 | 88.0 (R) | 100 |
| 12 | (R,R)-Me $-$ DuPhos | (Z)-1d | Me | Et | 294 | 62.3 (R) | 100 |
| 13 | (R,R)-BICP | (<i>E</i>)- 1e | Pr | Et | 40 | 96.6 (R) | 100 |
| 14 | (R,R)-Me $-$ DuPhos | (E)-1e | Pr | Et | 40 | 99.6 (R) | 100 |
| 15 | (R,R)-BICP | (Z)-1e | Pr | Et | 294 | 90.7 (R) | 100 |
| 16 | (R,R)-Me $-$ DuPhos | (Z)-1e | Pr | Et | 294 | 34.4 (R) | 100 |
| 17 | (R,R)-BICP | (E)-1f | <i>i</i> -Pr | Et | 40 | 97.0 (S) | 100 |
| 18 | (R,R)-Me $-$ DuPhos | (E)-1f | <i>i</i> -Pr | Et | 40 | 97.6 (S) | 100 |
| 19 | (R,R)-BICP | (Z)- 1f | <i>i</i> -Pr | Et | 294 | 91.0 (S) | 100 |
| 20 | (R,R)-Me $-$ DuPhos | (Z)-1f | <i>i</i> -Pr | Et | 294 | 42.1 (S) | 100 |
| 21 | (R,R)-BICP | (E)-1g | Me | <i>i</i> -Pr | 40 | 95.6 (R) | 100 |
| 22 | (R,R)-Me $-$ DuPhos | (E) -1 \mathbf{g} | Me | <i>i</i> -Pr | 40 | 98.1 (R) | 100 |
| 23 | (R,R)-BICP | (Z)-1g | Me | <i>i</i> -Pr | 294 | 86.4 (R) | 100 |
| 24 | (R,R)-Me $-$ DuPhos | (Z)-1g | Me | <i>i</i> -Pr | 294 | 61.9 (R) | 100 |
| 25 | (R,R)-BICP | (E/Z)-1h | Ph | Me | 294 | 66.0 (S) | 100 |
| 26 | (R,R)-Me $-$ DuPhos | (E/Z)- 1h | Ph | Me | 294 | 65.1 (S) | 100 |

^a The reaction was conducted at rt under H_2 for 24 h. The catalyst was made in situ by stirring a solution of [Rh(COD)₂]OTf and ligand for 30 min {[substrate (0.5 mmol, 0.125 M):[Rh(COD)₂]OTf:ligand = 1:0.01:0.011]}. ^b Enantiomeric excesses were determined by GC using a CHIRAL SELECT 1000 column or γ-DEX-225.

99.3% ee). However, enantioselectivity in hydrogenation of (Z)-1a with the Rh-BICP catalyst (entry 14, 88.6% ee) is significantly higher than that achieved with the Rh-Me-DuPhos complex (entry 15, 63.7% ee).

On the basis of the hydrogenation conditions for (E)-**1a** and (Z)-**1a**, a variety of 3-aminoacrylic acid derivatives were reduced using Rh-BICP and Rh-Me-DuPhos catalysts (Table 2). In general, hydrogenation of an E isomer of enamides gave a higher ee than the corresponding Z isomer (an exception is with substrate **1a** and a Rh-BICP catalyst, entries 5 and 7), and hydrogenation of the E isomer is faster than with the corresponding Z

isomer. Changing the ester groups and alkyl substituents of 3-aminoacrylic acid derivatives shows minor variations in both enantioselectivity and reactivity with both Rh–BICP and Rh–Me–DuPhos catalysts (entries 1–24).

Overall, Rh—Me—DuPhos is a slightly more enantioselective catalyst for hydrogenation of the *E* isomers of enamides **1b**—**1g** than is the Rh—BICP catalyst. On the other hand, Rh—BICP is a superior catalyst for hydrogenation of the *Z* isomers of **1b**—**1g** compared to the Rh—Me—DuPhos complex. For example, Rh—Me—DuPhoscatalyzed hydrogenation of (*Z*)-**1b** only gave 21.2% ee (entry 4) compared with 86.9% ee achieved with the Rh—

BICP catalyst (entry 3). From a practical point of view, hydrogenation of E/Z mixtures of β -(acylamino)acrylates with the Rh–BICP catalyst offers the most convenient access to corresponding β -amino acid derivatives because high enantiomeric excesses can be obtained from both isomers. Our results with Rh–BICP represent the highest enantioselectivity achieved for the hydrogenation of β -(acylamino)acrylates with alkyl substituents. A limitation of the reaction is that only moderate ee's were obtained with an aryl substituent in the β -(acylamino)acrylates using both Rh–BICP and Rh–Me–DuPhos catalysts (entries 25 and 26).

In summary, highly enantioselective hydrogenation of the E isomer of β -(acylamino)acrylates with a Rh–Me–DuPhos catalyst and E/Z isomers of β -(acylamino)acrylates with a Rh–BICP catalyst has been achieved in this study. This method provides an efficient approach to important β -amino acid building blocks. Future research will explore the scope of this reaction with other enamides, especially β -(acylamino)acrylates with an α -substituent or containing a ring.

Experimental Section

General Methods. All reactions and manipulations were performed in a nitrogen-filled glovebox or using standard Schlenk techniques. Toluene, benzene, and THF were distilled from sodium benzophenone ketyl under nitrogen. Methylene chloride was distilled from CaH₂. Methanol, ethanol, and 2-propanol were distilled from Mg under nitrogen. Chiral BICP ligand was prepared using previously described procedures. (R)-BINAP and (+)-DIOP were purchased from Aldrich. (R,R)-Me—DuPHOS was purchased from Strem. Column chromatography was performed using EM Merck Silica Gel 60 (230–400 mesh).

Synthesis of Substrates (Scheme 1): Step a. A solution of β -keto ester **3** (12 mmol) and NH₄OAc (4.6 g, 60 mmol) in MeOH (15 mL) was stirred at rt for 3 d. After the solvent was evaporated under reduced pressure, the residue was diluted with CHCl₃ (30 mL). The resulting solid was filtered off and washed with CHCl₃ (2 × 30 mL). The combined filtrate was washed with water and brine, and dried over sodium sulfate. Evaporation of the solvent gave the 3-amino-2-alkenoate **4**, which was used for the next step without purification.

Step b. To a solution of 3-amino-2-alkenoate **4** in THF (12 mL) were added pyridine (2 mL) and acetic anhydride (6 mL). The reaction mixture was then heated under reflux for 24 h. After the mixture was cooled to rt, the volatile was evaporated. The residue was dissolved in EtOAc (20 mL), and the solution was washed with water (10 mL), 1 N HCl (10 mL), 1 M KH $_2$ PO $_4$ (10 mL), NaHCO $_3$ (saturated, 10 mL), and brine (15 mL). After the solution was dried over sodium sulfate, the solvent was evaporated under reduced pressure. Chromatography of the residue on silica gel with a gradient solvent of EtOAc in hexanes (15–70%) as eluant gave the Z isomer, followed by a byproduct and then the E isomer.

- (*Z*)-Methyl 3-Acetamido-2-butenoate:^{4a} 1 H NMR (360 MHz, CDCl₃) δ 1.89 (s, 3 H), 2.24 (s, 3H), 3.57 (s, 3H), 4.76 (s, 1H), 10.9 (br, 1H).
- (*E*)-Methyl 3-Acetamido-2-butenoate: 4a 1 H NMR (360 MHz, acetone- d_{6}) δ 2.02 (s, 3 H), 2.29 (s, 3H), 3.58 (s, 3H), 6.85 (s, 1H), 8.72 (br, 1H).
- (*Z*)-Ethyl 4-Methyl-3-acetamido-2-pentenoate: $^{1}\mathrm{H}$ NMR (360 MHz, CDCl₃) δ 1.10–1.24 (m, 6 H), 1.25–1.30 (m, 3H), 2.13 (s, 3H), 3.87 (m, 1H), 4.13–4.19 (m, 2H), 5.04 (s, 1 H), 11.10 (br, 1 H); $^{13}\mathrm{C}$ NMR (90.5 MHz, CDCl₃) δ 13.9, 14.0, 21.0, 25.2, 29.0, 59.6, 92.4, 165.2, 168.0, 169.4; MS m/z 199, 184, 156, 111, 96, 70, 43; HRMS calcd for $C_{10}H_{18}\mathrm{NO}_3$ 200.1287, found 200.1295.
- (*E*)-Ethyl 4-Methyl-3-acetamido-2-pentenoate: 1 H NMR (200 MHz, acetone- d_{6}) δ 1.12 (d, J = 7.08 Hz, 6 H), 1.20 (t, J = 7.08 Hz, 3H), 2.08 (s, 3H), 4.05 (q, J = 7.09 Hz, 2H), 4.30 (m, 1H), 6.97 (s, 1 H), 8.14 (br, 1 H); 13 C NMR (50.3 MHz, acetone- d_{6}) δ 14.5, 19.6, 24.6, 27.7, 59.4, 100.8, 158.7, 168.3, 170.6; MS

 $\it m/z$ 199, 184, 156, 113, 111, 96, 70, 43; HRMS calcd for $\it C_{10}H_{18}$ - $\it NO_3$ 200.1287, found 200.1286.

- (*Z*)-Ethyl 3-Acetamido-2-hexenoate: $^1\mathrm{H}$ NMR (360 MHz, CDCl₃) δ 0.94 (t, J=7.39 Hz, 3H), 1.27 (t, J=7.11 Hz, 3H), 1.60 (m, 2H), 2.12 (s, 3H), 2.72 (t, 2H), 4.13 (t, J=7.12 Hz, 2H), 4.92 (s, 1H), 11.10 (br, 1H); $^{13}\mathrm{C}$ NMR (90.5 MHz, CDCl₃) δ 13.4, 14.0, 21.3, 25.1, 35.8, 59.6, 95.7, 158.7, 168.1, 169.1; MS m/z 199, 184, 170, 156, 129, 113, 96, 83, 43; HRMS calcd for $\mathrm{C_{10}H_{18}NO_3}$ 200.1287, found 200.1290.
- (*E*)-Ethyl 3-Acetamido-2-hexenoate: 1 H NMR (360 MHz, acetone- d_{6}) δ 1.15 (t, J=7.39 Hz, 3H), 1.44 (t, J=7.13 Hz, 3H), 1.53 (m, 2H), 2.28 (s, 3H), 2.95 (m, 2H), 4.30 (q, J=7.14 Hz, 2H), 7.14 (s, 1H), 8.91 (br, 1H); 13 C NMR (90.6 MHz, acetone- d_{6}) δ 14.4, 15.1, 23.2, 25.1, 34.1, 59.9, 101.8, 155.4, 168.9, 170.8; MS m/z 199, 184, 170, 156, 156, 129, 112, 96, 83, 43; HRMS calcd for $C_{10}H_{18}NO_{3}$ 200.1287, found 200.1289.
- (*Z*)-Isopropyl 3-Acetamido-2-butenoate: ¹H NMR (360 MHz, CDCl₃) δ 1.42–1.45 (m, 6H), 2.32 (s, 3H), 2.55 (m, 3H), 5.05 (s, 1H), 5.20 (m, 1H), 11.3 (br, 1H); ¹³C NMR (90.6 MHz, CDCl₃) δ 21.7, 21.8, 25.1, 67.0, 96.8, 154.5, 168.5, 168.7; MS m/z 185, 142, 126, 101, 83, 57, 43; HRMS calcd for $C_9H_{16}NO_3$ 186.1130, found 186.1130.
- (*E*)-Isopropyl 3-Acetamido-2-butenoate: 1 H NMR (360 MHz, acetone- d_{6}) δ 1.10 (d, J = 6.33 Hz, 6H), 1.94 (s, 3H), 2.20 (s, 3H), 4.85 (m, 1H), 6.74 (s, 1H), 8.55 (br, 1H); 13 C NMR (90.6 MHz, acetone- d_{6}) δ 18.5, 22.6, 25.1, 66.9, 102.5, 151.1, 168.7, 170.5; MS m/z 185, 142, 126, 110, 83, 57, 43; HRMS calcd for $C_{9}H_{16}NO_{3}$ 186.1130, found 186.1130.
- (*Z*)-Ethyl 3-Acetamido-2-butenoate: $^1{\rm H}$ NMR (360 MHz, CDCl₃) δ 1.05–1.09 (m, 3H), 1.93 (s, 3H), 2.17 (s, 3H), 3.93–3.95 (m, 2H), 4.68 (s, 1H), 10.9 (br, 1H); $^{13}{\rm C}$ NMR (90.6 MHz, CDCl₃) δ 14.1, 21.7, 25.1, 59.7, 96.3, 154.9, 168.8, 169.0; MS m/z 171, 129, 98, 84, 69, 57, 43; HRMS calcd for $C_8H_{14}{\rm NO}_3$ 172.0974, found 172.0976.
- (*E*)-Ethyl 3-Acetamido-2-butenoate: ¹H NMR (360 MHz, acetone- d_6) δ 1.39 (t, J=7.04 Hz, 3H), 2.23 (s, 3H), 2.49 (s, 3H), 4.25 (q, J=7.27 Hz, 2H), 7.04 (s, 1H), 8.91 (br, 1H); ¹³C NMR (90.6 MHz, acetone- d_6) δ 15.1, 18.5, 25.1, 59.9, 102.0, 151.4, 169.1, 170.6; MS m/z 171, 156, 129, 98, 84, 57, 43; HRMS calcd for C₈H₁₃NO₃ 171.0895, found 171.0900.
- (*Z*)-Methyl 3-Acetamido-2-pentenoate: ^1H NMR (360 MHz, CDCl₃) δ 1.03 (t, J=7.35 Hz, 3H), 2.07 (s, 3H), 2.72 (m, 2H), 3.62 (s, 2H), 4.89 (s, 1H), 11.0 (br, 1H); ^{13}C NMR (90.6 MHz, CDCl₃) δ 12.1, 24.8, 26.9, 50.5, 93.9, 160.3, 167.9, 169.4; MS m/z 171, 140, 129, 98, 84, 69, 43; HRMS calcd for C₈H₁₃NO₃ 171.0895, found 171.0898.
- (*E*)-Methyl 3-Acetamido-2-pentenoate: 1H NMR (360 MHz, acetone- d_6) δ 1.21 (t, J=7.52 Hz, 3H), 2.15 (s, 3H), 2.84 (q, J=7.47 Hz, 2H), 3.69 (s, 2H), 6.96 (s, 1H), 8.80 (br, 1H); ^{13}C NMR (90.6 MHz, acetone- d_6) δ 13.2, 24.4, 25.0, 50.4, 99.9, 156.6, 168.5, 170.3; MS m/z 171, 140, 129, 112, 98, 84, 69, 43; HRMS calcd for $C_8H_{13}NO_3$ 171.0895, found 171.0890.
- (*E*)-Methyl 5-Methyl-3-acetamido-2-hexenoate: 4a 1 H NMR (360 MHz, acetone- d_{6}) δ 1.01 (d, J=6.72 Hz, 6H), 2.13–2.15 (m, 3H), 2.79 (d, J=7.44 Hz, 2H), 3.68 (s, 3H), 7.07 (s, 1H), 8.75 (br, 1H).

Methyl 3-Acetamido-3-phenyl-2-propenoate:^{4a} ¹H NMR (200 MHz, CDCl₃) δ (*Z* isomer) 2.17 (s, 3H), 3.77 (s, 3H), 5.29 (s, 1H), 7.37–7.45 (m, 5H); (*E* isomer) 2.38 (s, 3H), 3.77 (s, 3H), 6.65 (s, 1H), 7.37–7.45 (m, 5H).

General Procedure for Asymmetric Hydrogenation. To a solution of $[Rh(COD)_2]OTf~(2.0~mg,~4.27~\times~10^{-3}~mmol)$ in toluene (3.4 mL) in a glovebox was added the chiral ligand (0.047 mL of a 0.1 M solution in toluene, 4.7 $\times~10^{-3}~mmol)$. After the mixture was stirred for 30 min, substrate 1 (0.427 mmol) was added. The hydrogenation was performed at rt under 10 atm of hydrogen for 24 h. The hydrogen pressure was released carefully in a hood, and the reaction mixture was passed through a short silica gel column to remove the catalyst. The enantiomeric excesses were measured by GC without further purification.

(*R*)-Methyl 3-Acetamidobutanoate: 4a,b [α] 25 ${}_{D}$ 25 = +21.4 (*c* 1.4, MeOH); 1 H NMR (360 MHz, CDCl₃) δ 1.16 (d, J = 6.81 Hz, 3 H), 1.90 (s, 3H), 2.47 (m, 3H), 3.63 (s, 3H), 4.28 (m, 1H), 6.14 (br. 1H)

(*S*)-Ethyl 4-Methyl-3-acetamidopentanoate: $[\alpha]^{25}_D = +52.8$ (*c* 1.2, CHCl₃); ¹H NMR (360 MHz, CDCl₃) δ 0.83–0.86 (m, 6 H), 1.15–1.20 (m, 3H), 1.75 (m, 1H), 1.90 (s, 3H), 2.40–2.43 (m,

2H), 4.00–4.08 (m, 3H), 639 (br, 1 H); $^{13}\mathrm{C}$ NMR (90.5 MHz, CDCl₃) δ 13.9, 18.5, 19.0, 23.0, 31.3, 36.5, 51.3, 60.3, 169.4, 171.8; MS $\it m/z$ 202, 186, 158, 142, 116, 97, 70, 43; HRMS calcd for $C_{10}H_{20}NO_3$ 202.1443, found 202.1452.

(*R*)-Ethyl 3-Acetamidohexanoate: $[\alpha]^{25}_{\rm D} = +42.8$ (c 1.86, CHCl₃); $^1{\rm H}$ NMR (200 MHz, CDCl₃) δ 0.85 (t, J=6.92 Hz, 3H), 1.16–1.45 (m, 7H), 1.91 (s, 3H), 2.43–2.46 (m, 2H), 4.06 (q, J=7.14 Hz, 2H), 4.14–4.28 (m, 1H), 6.25 (br, 1H); $^{13}{\rm C}$ NMR (50.3 MHz, CDCl₃) δ 13.7, 14.0, 19.3, 23.2, 36.1, 38.5, 45.7, 60.4, 169.5, 171.8; MS m/z 201, 186, 172, 158, 142, 116, 97, 72, 43; HRMS calcd for $C_{10}{\rm H}_{20}{\rm NO}_3$ 202.1443, found 202.1450.

(*R*)-Isopropyl 3-Acetamidobutanoate: $[\alpha]^{25}_D=+35.5~(c~1.91,~CHCl_3);~^1H~NMR~(360~MHz,~CDCl_3)~\delta~1.13-1.18~(m,~6H),~1.88~(s,~3H),~2.37-2.47~(m,~2H),~4.24-4.28~(m,~1H),~4.92-4.96~(m,~1H),~6.59~(br,~1H);~^{13}C~NMR~(90.6~MHz,~CDCl_3)~\delta~19.8,~21.47,~21.53,~23.0,~40.4,~42.0,~67.7,~169.2,~170.9;~MS~{\it m/z}~187,~144,~128,~102,~86,~69,~43;~HRMS~calcd~for~C_9H_{18}NO_3~188.1288,~found~188.1287.$

(*R*)-Ethyl 3-Acetamidobutanoate: $[\alpha]^{25}_{D} = +41.9$ (*c* 1.69, CHCl₃); ¹H NMR (200 MHz, CDCl₃) δ 1.15 (t, J = 6.78 Hz, 3H), 1.18 (t, J = 7.08 Hz, 3H), 1.88 (s, 3H), 2.41–2.44 (m, 2H), 4.06 (q, J = 7.15 Hz, 2H), 4.22–4.30 (m, 1H), 6.42 (br, 1H); ¹³C NMR (50.3 MHz, CDCl₃) δ 14.0, 19.8, 23.2, 40.0, 41.9, 60.4, 169.3, 171.5; MS m/z 173, 158, 130, 116, 86, 70, 43; HRMS calcd for $C_8H_{15}NO_3$ 174.1130, found 174.1136.

(*R*)-Methyl 3-Acetamidopentanoate: $[\alpha]^{25}_D = +52.0$ (*c* 1.26, CHCl₃); ¹H NMR (360 MHz, CDCl₃) δ 0.85 (t, J = 7.49 Hz, 3H), 1.45–1.54 (m, 2H), 1.91 (s, 3H), 2.46–2.50 (m, 2H), 3.61 (s, 3H), 4.06–4.12 (m, 1H), 6.20 (br, 1H); ¹³C NMR (90.6 MHz, CDCl₃) δ 10.5, 23.2, 27.0, 37.9, 47.4, 51.5, 169.6, 172.1; FAB-MS 174 (M⁺ + 1); HRMS calcd for C₈H₁₅NO₃ 174.1130, found 174.1133.

(*R*)-Methyl 5-Methyl-3-acetamidohexanoate:^{4a} $[\alpha]^{25}_D = +44.6$ (*c* 1.56, CHCl₃); ¹H NMR (360 MHz, CDCl₃) δ 0.89–0.91 (m, 6H), 1.20–1.35 (m, 1H), 1.45–1.70 (m, 2H), 1.96 (s, 3H), 2.44–2.60 (m, 2H), 3.68 (s, 3H), 4.28–4.35 (m, 1H), 5.98 (br, 1H).

(*S*)-Methyl 3-Acetamido-3-phenylpropanoate:^{4a} [α]²⁵_D = -40.5 (c 2.15, MeOH); ¹H NMR (360 MHz, CDCl₃) δ 1.92 (s, 3H), 2.76–2.83 (m, 2H), 3.53 (s, 3H), 5.34 (m, 1H), 6.65 (br, 1H), 7.18–7.27 (m, 5H).

Determination of Enantiomeric Excesses: Chiral Capillary GC Column. Chiral Select-1000 column (dimensions 15 m \times 0.25 mm (i.d.)) or γ -DEX 225 column (dimensions 30 m \times 0.25 mm (i.d.)). Carrier gas: He (1 mL/min). The racemic products were obtained by hydrogenation of substrates with an achiral catalyst. The following are the retention times for the racemic products.

Methyl 3-Acetamidobutanoate: (capillary GC, Chiral Select-1000 column, 130 °C, isothermal) (S) $t_1 = 6.94$ min, (R) $t_2 = 8.29$ min

Ethyl 4-Methyl-3-acetamidopentanoate: (capillary GC, γ -DEX 225 column, 145 °C, isothermal) (*S*) $t_1 = 18.81$ min, (*R*) $t_2 = 19.38$ min.

Methyl 4, 4-Dimethyl-3-acetamidopentanoate: (capillary GC, γ -DEX 225 column, 150 °C, isothermal) (*S*) $t_1=14.13$ min, (*R*) $t_2=14.78$ min.

Ethyl 3-Acetamidohexanoate: (capillary GC, γ -DEX 225 column, 140 °C, isothermal) (*S*) $t_1 = 25.72$ min, (*R*) $t_2 = 26.24$ min

Isopropyl 3-Acetamidobutanoate: (capillary GC, Chiral Select-1000 column, 140 °C, isothermal) (*S*) $t_1 = 6.63$ min, (*R*) $t_2 = 7.33$ min.

Ethyl 3-Acetamidobutanoate: (capillary GC, Chiral Select-1000 column, 140 °C, isothermal) (*S*) $t_1 = 6.05$ min, (*R*) $t_2 = 6.67$ min

Methyl 3-Acetamidopentanoate: (capillary GC, γ -DEX 225 column, 145 °C, isothermal) (R) $t_1=12.28$ min, (S) $t_2=12.63$ min

Methyl 5-Methyl-3-acetamidohexanoate: (capillary GC, γ -DEX 225 column, 145 °C, isothermal) (*S*) $t_1 = 17.20$ min, (*R*) $t_2 = 17.70$ min.

Methyl 3-Acetamido-3-phenylpropanoate: (capillary GC, Chiral Select-1000 column, 180 °C, isothermal) (*S*) $t_1 = 8.88$ min, (*R*) $t_2 = 9.32$ min.

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Supporting Information Available: NMR spectra for the compounds used in this study. This material is available free of charge via the Internet at http://pubs.acs.org.

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