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## Copper(I)-Catalyzed Asymmetric Monoborylation of 1,3-Dienes: Synthesis of Enantioenriched Cyclic Homoallyl- and Allylboronates

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The synthesis of enantioenriched α-chiral organoboron compounds has attracted considerable attention in organic synthesis because C-B bonds can be converted to C-O, C-N, or C-C bonds in a stereospecific manner. 1 Asymmetric hydroboration of alkenes is one of the most important methods used for the preparation of such compounds; however, the reaction with an asymmetric catalysis is limited to aromatic alkenes, alkenes with an amide directing group, and cyclopropenes.<sup>2,3</sup> Asymmetric hydroboration of 1,3-dienes provides homoallyllic (1,2-addition) or allylic (1,4-addition) boron compounds. These synthesized compounds are highly useful, if they can be prepared in a regioand enantioselective manner, because further derivatization using the remaining alkene moiety as well as the boron group easily leads to multifunctional chiral compounds. Although the stoichiometric asymmetric reaction of 1,3-diene using diisopinocamphenylborane (Ipc<sub>2</sub>BH) was pioneered by H. C. Brown a quarter of a century ago, 3b the asymmetric catalytic hydroboration of 1,3-dienes with high enantio- and diastereoselectivities has not been reported.<sup>4,5</sup>

Here, we report the first asymmetric catalytic monoborylation of 1,3-dienes with a copper(I) catalyst. This reaction produced unprecedented cyclic optically active homoallylboronates with high regio- and enantioselectivity.<sup>6</sup> In addition, a drastic change of product preference between homoallyl- and allylboronates was found for the reaction of 1,3-cyclopentadiene and 1,3-cycloheptadiene. Enantioenriched cyclic allylboronates, which are useful synthetic reagents but difficult to prepare, were also synthesized.<sup>6</sup>

A number of asymmetric copper(I)-catalyzed reactions with diboron have been reported.  $^{2d,7-9}$  Yun reported the acceleration effect of proton sources,8 and the importance of the low LUMO levels of the substrate has also been reported. 2d,9,10 Guided by these findings, we anticipated that 1,3-dienes can also be used in the copper(I)-catalyzed borylation in combination with an appropriate proton source. The reaction was first carried out by the addition of 2.0 equiv of methanol to a mixture of 1,3-cyclohexadiene (1a), 1.5 equiv of bis(pinacolato)diboron (2), 5 mol % of Cu(O-t-Bu), and (R,R)-Me-Duphos in THF at room temperature. The reaction reached completion within 3 h to afford homoallylboronate (S)-3a in 96% yield with 87% ee (Table 1, entry 1). No allylboronate (4a) and multiborylated products were detected. Using other chiral ligands resulted in moderate to high yields with lower enantio- and regioselectivities (69–97% yield, 61–86% ee, 3a/4a 96:4–>99:1, entries 2-5). The enantioselectivity was improved to 94% ee with (R,R)-Me-Duphos at -20 °C (92% yield, entry 6). The reaction with a mixture of CuCl and K(O-t-Bu), which is more readily available than Cu(O-t-Bu), gave almost comparable results, but the regioselectivity was slightly decreased (entry 7).

**Table 1.** Copper(I)-Catalyzed Monoborylation of 1,3-Cyclohexadiene<sup>a</sup>

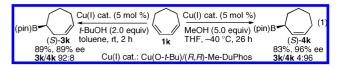
() 1a	Cu(O- <i>t</i> -Bu)–ligand (5 mol %) or CuCl–K(O- <i>t</i> -Bu)–ligand (5 mol %)	(pin)B= () +	(nin)B
	bis(pinacolato)diboron (2) (1.5 equiv) THF, MeOH (2.0 equiv)	(S)-3a	4a

entry	ligand	time (h)	temp	<b>3a/4a</b> ratio <sup>b</sup>	yield <sup>c</sup> (%)	ee <sup>b</sup> (%)
1	(R,R)-Me-Duphos	3	rt	>99:1	96	87
2	(R,R)-QuinoxP*	3	rt	98:2	97	86
3	(R)-BINAP	3	rt	>99:1	69	61
4	(R)-Segphos	3	rt	96:4	70	67
5	(R,R)-i-Pr-DuPhos	3	rt	97:3	92	83
6	(R,R)-Me-DuPhos	24.5	−20 °C	>99:1	92	94
$7^d$	(R,R)-Me-DuPhos	26	−20 °C	98:2	97	94

 $^a$  Conditions: **1a** (0.25 mmol), **2** (0.375 mmol), Cu(O-*t*-Bu) (5 mol %, 0.0125 mmol), ligand (5 mol %, 0.0125 mmol), THF (0.25 mL), and methanol (2.0 equiv).  $^b$  The ee and **3/4** ratio values were determined by HPLC analysis of the benzoate ester of the homoallylic alcohol obtained after H<sub>2</sub>O<sub>2</sub>/NaOH oxidation of **3a**.  $^c$  Isolated yield.  $^d$  The reaction was carried out at 0.5 mmol scale with 5 mol % CuCl and 20 mol % of K(O-*t*-Bu) in THF (0.7 mL).

The scope of the reaction was subsequently examined (Table 2). 1,3-Cyclohexadiene derivatives **1b**—**e** were converted into the corresponding optically active homoallylboronates **3b**—**e** in high yields with high ee values (90–97%, 93–95% ee, entries 1–4). The reaction of **1f** also gave **3f** (99%, 88% ee, entry 5), which has silyl enol ether functionality. The attempted reaction of **1g** failed probably due to the large steric hindrance (entry 6). The reaction with acyclic 1,3-dienes gave homoallylboronates with high regioselectivity (entries 7 and 8), but a poor ee value was observed in **3i** (42% ee).

The reactions of 1a predominantly afforded the homoallylboronate **3a** at room temperature and at -20 °C (Table 1). Conversely, the regioisomeric preferences in the reaction of 1,3-cyclopentadiene (1j) and 1,3-cycloheptadiene (1k) were switched by changing the reaction temperature (Table 3).<sup>10</sup> The copper(I)-catalyzed reaction of **1j** at a low temperature of -40 °C using methanol in THF predominantly gave allylboronate (S)-4j with a high ee (entry 1, 87%, 96% ee, 3j/4j 7:93). At room temperature, the major product was homoallylboronate 3j (entry 2, 73%, 3j/4j 71:29). The regioselectivity of 3j at room temperature was improved by using tert-butanol in toluene (entry 3, 77%, 3j/4j 92:8). The reaction with *tert*-butanol in toluene at -40 °C gave predominantly (S)-4j, indicating the reaction temperature is the most important factor in controlling the 3j/4j selectivity (entry 4, 86%, 97% ee, 3j/4j 6:94). A similar condition-dependent selectivity profile was also observed in the reaction of 1k with high regio- and enantioselectivities (eq 1).



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Table 2. Substrate Scope<sup>a</sup>

$$\begin{array}{c} \text{Cu(O-$t$-Bu) (5 mol \%)} \\ (R,R)\text{-Me-DuPhos (5 mol \%)} \\ \hline \textbf{2} \text{ (1.5 equiv), } -20 \,^{\circ}\text{C} \\ \text{THF, MeOH (2.0 equiv)} \\ \end{array} \begin{array}{c} \text{(pin)B} \\ \hline \textbf{3} \end{array} + \begin{array}{c} \text{(pin)B} \\ \hline \textbf{4} \end{array}$$

		time		3/4	yield <sup>c</sup>	$ee^d$
entry	diene	(h)	product	ratio <sup>b</sup>	(%)	(%)
1	—Me 1b	17	(pin)B————Me 3b	>20:1	90	95
2	Bu 1c	17	(pin)B► Bu 3c	>20:1	97	95
3e	—Ph 1d	25	(pin)B————————————————————————————————————	94:6	95	95
4	Bu 1e	22.5	(pin)B 3e	>20:1	91	93
5	1f OTMS	16	(pin)B ST ST OTMS	>20:1	99	88
6	MeMe	24	(pin)B Me 3g Me	_	0	-
<b>7</b> <sup>f</sup>	Me 1h	1.5	(pin)B————— 3h	>20:1	97	_
8 <sup>g</sup>	C <sub>6</sub> H <sub>13</sub> Me	20	(pin)B C <sub>6</sub> H <sub>13</sub> Me <b>3i</b> Me	91:9	78	42

 $^a$  Conditions: **1** (0.5 mmol), **2** (0.75 mmol), Cu(O-t-Bu) (5 mol %, 0.025 mol), ligand (5 mol %, 0.025 mmol) in THF (0.5 mL).  $^b$  Determined by  $^1$ H NMR or HPLC.  $^c$  Isolated yield.  $^d$  Determined by HPLC analysis.  $^c$  0.25 mmol scale.  $^f$  1.1 equiv of **2** was used.  $^g$  The reaction was carried out on 0.25 mmol scale at 0  $^o$ C with 10 mol % of Cu(O-t-Bu) and ligand.

 $\it Table 3.$  Product Switch in Copper(I)-Catalyzed Monoborylation of 1,3-Dienes<sup> $\it a$ </sup>

١ ١	· // · · ·			· ·			\—/
1j 2 (1.5 equiv)		.5 equiv) 3j		3j	j ( <i>S</i> )-4j		
entry	solvent	alcohol	temp	time (h)	<b>3/4</b> ratio <sup>b</sup>	yield <sup>c</sup> (%)	ee <sup>d</sup> (%)
1 <sup>e</sup>	THF	МеОН	−40 °C	27	7:93	87	96 ( <b>4j</b> )
$2^f$	THF	MeOH	rt	5	71:29	73	_
$3^f$	toluene	t-BuOH	rt	5	92:8	77	_

<sup>a</sup> Conditions: **1** (0.5 mmol), **2** (0.75 mmol), Cu(O-*t*-Bu) (5 mol %, 0.025 mol), ligand (5 mol %, 0.025 mmol), solvent (0.5 mL), and alcohol. <sup>b</sup> Determined by GC or HPLC. <sup>c</sup> Isolated yield. <sup>d</sup> Determined by HPLC analysis. <sup>e</sup> 5.0 equiv of alcohol were used. <sup>f</sup> 2.0 equiv of alcohol were used.

-40 °C

26

86

6:94

97 (4i)

Scheme 1. Possible Reaction Mechanism

t-BuOH

Cu(O-*t*-Bu) (5 mol %) (*R*,*R*)-Me-DuPhos (5 mol %)

4

A possible reaction mechanism is depicted in Scheme 1.<sup>11,12</sup> The borylcopper(I) formed from copper(I) alkoxide, and diboron 2 undergoes a *syn*-addition across one of the carbon—carbon double bonds to form a  $\sigma$ -allylcopper species **A**. Protonation of the kinetic

product  $\bf A$  at the lower temperature proceeded in an  $S_E2'$  fashion to provide allylboronate  $\bf 4$ . At higher temperatures,  $\bf A$  rapidly isomerizes to thermodynamically stable  $\bf B$  and slow protonation of  $\bf B$  gives homoallylboronate  $\bf 3$ . The slower protonation rate under the *tert*-butanol/toluene conditions is assumed to be responsible for the higher regioselectivity (Table 3).  $\bf 3a$  was only obtained in the reaction of  $\bf 1a$  (Table 1). This observation can be explained by the very rapid isomerization from  $\bf A$  to  $\bf B$  even at a low temperature.

In summary, we have shown the regio- and enantioselective monoborylation of 1,3-dienes through a copper(I)-catalyzed reaction. This should be a versatile method to obtain optically active cyclic homoallyl- and allylboronates, which were inaccessible by other catalytic methods. Further studies concerning the reaction mechanism and the expansion of the reaction scope are underway.

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**Supporting Information Available:** Experimental procedures and compound characterization data. This material is available free of charge via the Internet at http://pubs.acs.org.

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