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Au(I)-Catalyzed Enantioselective 1,3-Dipolar Cycloadditions of Münchnones with Electron-Deficient Alkenes

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Synthetic methods relying on gold complexes as catalysts have recently been the focus of intense development.[1] Despite numerous advances, relatively few enantioselective gold-catalyzed transformations have been described. The earliest example of an enantioselective gold(I)-catalyzed transformation, the Hayashi–Ito aldol reaction, was proposed to rely on activation of the nucleophile as a chiral monophosphineAu(I) enolate.[2] In contrast, the majority of recently reported methods rely on the electrophilic nature of cationic bisphosphinegold(I) complexes to activate π-bonds toward addition of nucleophiles.[3] Therefore, the utility of chiral bisphosphinegold-(I) complexes would be signficantly extended if they could be employed as catalysts for enantioselective transformations that are not predicated on π-bond activation. To this end, herein we describe the development of a bisphosphinegold(I)-catalyzed enantioselective 1,3-dipolar cycloaddition[4] reaction of mesoionic azomethine ylides (münchnones) with alkenes.

We were inspired by Tepe's recent report of silver(I)acetate-catalyzed münchnone/alkene 1,3-dipolar cycloadditions,[5,6] to consider the use of our recently developed bisphosphinegold(I) carboxylate complexes as catalysts for this transformation.[3c] Gratifyingly, treatment of a THF solution of azlactone **1a** and 1.5 equiv of *N*-phenylmaleimide (**3**) with 2 mol % triphenylphosphinegold(I) benzoate at room temperature, followed by in situ esterification, afforded the desired Δ^1-pyrroline ($\pm$)-**2a** in 85% yield with excellent diastereoselectivity (Table 1, entry 1).

Having established an achiral phosphinegold(I) benzoate as a catalyst for the formation for **2a**, we next focused on the enantioselective reaction of **1a** and **3** (Table 1). In general, the diphenyl-substituted biarylphosphinegold(I) benzoate complexes successfully employed in the hydroamination gave modest selectivity for the reaction of **1a** and **3** (entries 2−5). While substitution on the phosphine aryl ring resulted in improved selectivity (entry 8), we were pleased to find the (*S*)-Cy-SEGPHOS(AuOBz)$_2$ (**4**)[7]-catalyzed cycloaddition gave **2a** with a notable increase in enantioselectivity (88% ee) (entry 9). Further optimization of reaction conditions revealed that the reaction performed similarly in various solvents;[8] however, employing fluorobenzene (PhF) as the solvent produced a further improvement and provided **2a** in 76% yield and 95% ee (entry 10).[9] Notably, in all cases only the *exo*-adduct was observed.
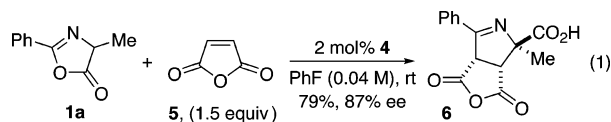
A variety of electron-deficient alkenes were found to be viable dipolarophiles in the gold(I)-catalyzed münchnone cycloaddition. For example, 2 mol % gold(I) benzoate **4**-catalyzed the reaction of **1a** with maleic anhydride (**5**), under conditions similar to those used with *N*-phenylmaleimide to afford acid **6** in 79% yield and

Table 1. Development of the Au(I)-Catalyzed Cycloaddition[a]

entry	catalyst	time (h)	yield (%)[b]	ee (%)[c]
1	PPh$_3$AuOBz	0.5	85	−
2	(*R*)-BINAP(AuOBz)$_2$	2	78	−8
3	(*R*)-SEGPHOS(AuOBz)$_2$	3	81	−40
4	(*R*)-DIFLUOROPHOS(AuOBz)$_2$	3	56	−44
5	(*R*)-Cl−MeO-BiPHEP(AuOBz)$_2$	3	80	−5
6	(*R*)-SYNPHOS(AuOBz)$_2$	3	62	−55
7	(*R*)-3,5-xylyl-BINAP(AuOBz)$_2$	3	63	−24
8	(*R*)-DTBM-SEGPHOS(AuOBz)$_2$	7	81	−83
9	(*S*)-Cy-SEGPHOS(AuOBz)$_2$, (**4**)	2	70	88
10	**4**	5	76	95[d]

[a] Reactions run with 0.33 mmol **1a**; for conditions concerning in situ ester formation see Supporting Information (SI). [b] Isolated yield. [c] Absolute configuration determined by X-ray crystallography on the corresponding α-methylbenzylamide (see SI). [d] Reaction performed at 0.5 M in PhF.
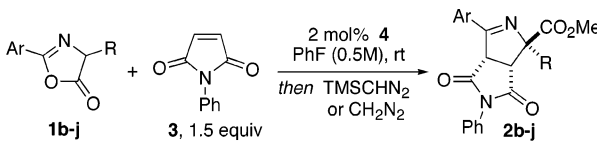
Table 2. Reaction of **1a** with Various Acyclic Alkenes[a]

entry	product		time (h)	yield (%)[b]	ee (%)
1	**7a**, X = CO$_2$^{t}Bu	R = OMe	24	56	99[c]
2	**7b**, X = CO$_2$^{t}Bu	R = NHCH$_2$Ph	14	74	95
3	**7c**, X = CO$_2$Et	R = OMe	14	66	90
4	**7d**, X = CO$_2$Me	R = OMe	14	89[d]	93
5	**7e**, X = CN	R = NHCH$_2$Ph	14	68	76

[a] Reactions run with 0.33 mmol **1a** at 0.5 M; for conditions concerning in situ ester/amide formation see SI. [b] Isolated yield unless otherwise noted. [c] Reaction run with 10 equiv of alkene and 2% mol **4**. [d] Yield determined by ^{1}H NMR.

87% ee (eq 1).[10] Acyclic alkenes could also be employed as partners in the gold(I)-catalyzed cycloaddition (Table 2). Generally, the reactions performed best using a 3:1 mixture of THF and PhF and slightly higher catalyst loading (3.5%). Notably, in all cases the reactions proceeded with excellent diastereo- and regioselectivity.

Various azlactones (**1b**−**j**) were prepared, and the results of their gold(I) benzoate-catalyzed enantioselective cycloaddition with *N*-phenylmaleimide are shown in Table 3. Substitution at the para position of the azlactone aromatic ring was well tolerated (entries 1−4). While increasing the steric demand at the azlactone C2 or C4 position resulted in decreased reactivity in PhF, switching the solvent to a mixture of THF and PhF permitted isolation of the cycloadducts in good yields (entries 5, and 7−9). The use of this solvent mixture allowed for smooth reaction of azlactone **1f**, bearing

Table 3. Reaction of *N*-Phenylmaleimide (**3**) with Azlactones[a]

entry		product	time (h)	yield (%)[b]	ee (%)
1	**b**, R = Me	Ar = *p*-MeO−C₆H₄−	18	77	95
2	**c**, R = Me	Ar = *p*-Br−C₆H₄−	15	75	93
3	**d**, R = Me	Ar = *p*-Cl−C₆H₄−	15	72	92
4	**e**, R = Me	Ar = *p*-NO₂C₆H₄−	1.5	98	91
5	**f**, R = Me	Ar = *o*-Me−C₆H₄−	4	73	86[c]
6	**g**, R = H	Ar = Ph	24	84	−98[d]
7	**h**, R = allyl	Ar = Ph	8	86	87[c]
8	**i**, R = Ph	Ar = Ph	1.5	35	78[e]
9	**j**, R = Bn	Ar = Ph	36	71	68[c]

[a] Reactions run with 0.33 mmol azlactone; for conditions concerning in situ ester formation see SI. [b] Isolated yield. [c] Run at 0.5 M in 3:1 THF/PhF. [d] At 0.5 M in acetone, using (*R*)-DTBM-SEGPHOS(AuOBz)₂; note also the change in the sense of ligand chirality. [e] Run at 0.25 M in 3:1 THF/PhF at 0 °C with 5% mol **4**.

Scheme 1. Proposed Mechanism of Au(I)-Catalyzed 1,3-DCR
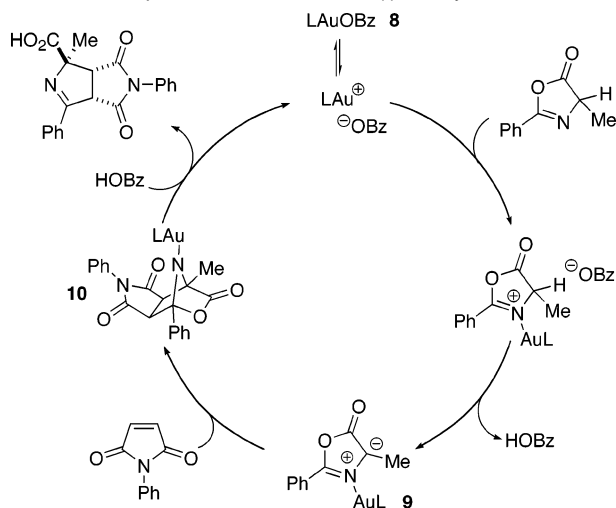
a 2-methylphenyl C2 substituent, providing product **2f** in good yield with 86% ee.[11] Similarly, C4 allyl-substituted azlactone **1h** underwent gold(I)-catalyzed cycloaddition to furnish **2h** in 86% yield and 87% ee; however, a decrease in enantioselectivity was observed with a further increase in the size of the C4 substituent to benzyl (entry 9). The reaction of **3** with glycine-derived azlactone **1g** catalyzed by **4** produced **2g** with only 81% ee. Fortunately, switching the catalyst to (*R*)-DTBM-SEGPHOS(AuOBz)₂ allowed for the isolation of cycloadduct **2g** in 84% yield and 98% ee (entry 6).

A proposed catalytic mechanism, paralleling those postulated for reactions of acyclic azomethine ylides, is shown in Scheme 1.[6] Dissociation of a carboxylate counterion from **8** provides an open coordination site for azlactone binding. Deprotonation of the activated substrate, presumably by benzoate, generates *N*-aurated-dipole **9**. Reaction of **9** with the dipolarophile produces initial cycloadduct **10**. Subsequent C−O bond cleavage and protonation followed by dissociation of the Δ¹-pyrroline regenerates the catalyst.[12]

In summary, we have developed the first catalytic enantioselective reaction of azlactones with alkenes to provide Δ¹-pyrrolines.[13,14] Notably, the gold-catalyzed cycloadditions proceed with excellent diastereo- and regioselectivity. The reaction is proposed to proceed through a 1,3-dipole[15] generated by deprotonation of a gold(I)-activated azlactone and therefore represents an important departure from the mechanistic paradigm of π-activation most commonly proposed in contemporary asymmetric catalysis with gold complexes. The development of enantioselective reactions relying on gold(I)-catalyzed generation of nucleophiles is ongoing and will be reported in due course.

Supporting Information Available: Experimental procedures and compound characterization data; X-ray crystallgraphic data. This material is available free of charge via the Internet at http://pubs.acs.org.

References

(1) For recent reviews see: (a) Jiménez-Núñez, E.; Echavarren, A. M. *Chem. Commun.* **2007**, 333. (b) Furstner, A.; Davies, P. W. *Angew. Chem., Int. Ed.* **2007**, *46*, 2. (c) Gorin, D. J.; Toste, F. D. *Nature* **2007**, *446*, 395.

(2) Ito, Y.; Sawamura, M.; Hayashi, T. *J. Am. Chem. Soc.* **1986**, *108*, 6405.

(3) (a) Hamilton, G. L.; Kang, E. J.; Mba, M.; Toste, F. D. *Science* **2007**, *317*, 496. (b) Liu, C.; Widenhoefer, R. A. *Org. Lett.* **2007**, *9*, 1935. (c) LaLonde, R. L.; Sherry, B. D.; Kang, E. J.; Toste, F. D. *J. Am. Chem. Soc.* **2007**, *129*, 2452. (d) Zhang, Z.; Widenhoefer, R. A. *Angew. Chem., Int. Ed.* **2007**, *46*, 283. (e) Johansson, M. J.; Gorin, D. J.; Staben, S. T.; Toste, F. D. *J. Am. Chem. Soc.* **2005**, *127*, 18002. (f) Munoz, M. P.; Adrio, J.; Carretero, J. C.; Echavarren, A. M. *Organometallics* **2005**, *24*, 1293.

(4) For reviews on asymmetric 1,3-dipolar cycloaddition reactions see: (a) Pellissier, H. *Tetrahedron* **2007**, 3235. (b) Kanemassa, S. *Synlett* **2002**, 1371. (c) Gothelf, K. V. *Synthesis* **2002**, 211. (d) Gothelf, K. V.; Jorgensen, K. A. *Chem. Rev.* **1998**, *98*, 863.

(5) Peddibhotla, S.; Tepe, J. *J. Am. Chem. Soc.* **2004**, *126*, 12776.

(6) For selected reports of transition-metal-catalyzed asymmetric azomethine ylide cycloadditions see: (a) Saito, S.; Tsubogo, T.; Kobayashi, S. *J. Am. Chem. Soc.* **2007**, *129*, 5364. (b) Zeng, W.; Chen, G.-Y.; Zhou, Y.-G.; Li, Y.-X. *J. Am. Chem. Soc.* **2007**, *129*, 705. (c) Cabrera, S.; Arrayas, R. G.; Carretero, J. C. *J. Am. Chem. Soc.* **2005**, *127*, 16394. (d) Zeng, W.; Zhou, Y.-G. *Org. Lett.* **2005**, *7*, 5055. (e) Gao, W.; Zhang, X.; Raghunath, M. *Org. Lett.* **2005**, *7*, 4241. (f) Oderaotoshi, Y.; Cheng, W.; Fujimoto, S.; Kasano, Y.; Minkata, S.; Komatsu, M. *Org. Lett.* **2003**, *5*, 5043. (g) Chen, C.; Li, X.; Schreiber, S. *J. Am. Chem. Soc.* **2003**, *125*, 10174. (h) Longmire, J.; Wang, B.; Zhang, X. *J. Am. Chem. Soc.* **2002**, *124*, 13400. For a review see: (i) Pandey, G.; Banerjee, P.; Gadre, S. R. *Chem. Rev.* **2006**, *106*, 4484.

(7) See Supporting Information (SI) for an X-ray structure of (*S*)-Cy-SEGPHOS(AuCl)₂.

(8) THF (2 h, 70%, 88% ee), acetone (2 h, 82%, 85% ee), DME (3 h, 79%, 87% ee), CH₂Cl₂ (5 h, 59%, 88% ee), benzene (5 h, 64%, 88% ee), toluene (5 h, 62%, 87% ee).

(9) Under these conditions, the choice of carboxylate counterion had no notable effect on the reaction. (benzoate, 5 h, 76%, 95% ee; *p*-nitrobenzoate, 5 h, 70%, 95% ee; acetate, 5 h, 71%, 95% ee).

(10) Cycloadduct **6** was not isolated; yield determined by ¹H NMR (see SI).

(11) Reaction of 2,4-dimethyl-2-oxazolin-5-one with **3**, catalyzed by **4**, gave the cycloadduct in 39% ee.

(12) Maryanoff, C. A.; Turchi, I. J. *Heterocycles* **1993**, 649.

(13) Reaction of **1a** and **3** catalyzed by 10% AgOAc, 10% (*S*)-QUINAP[6g] afforded **2a** in 14% yield and 5% ee after 12 hr at −20 °C in THF.

(14) For recent examples of metal-catalyzed processes proceeding through münchnones, see: (a) Dhawan, R.; Arndtsen, B. *J. Am. Chem. Soc.* **2004**, *126*, 468. (b) Dhawan, R.; Dghaym, R. D.; Arndtsen, B. *J. Am. Chem. Soc.* **2003**, *125*, 1474.

(15) For examples of gold-catalyzed processes proposed to proceed via reaction of a C-aurated 1,3-dipole generated by activation of an alkyne, see: (a) Asao, N.; Aikawa, H. *J. Org. Chem.* **2006**, *71*, 5253. (b) Kim, N.; Kim, Y.; Park, W.; Sung, D.; Gupta, A. K.; Oh, C. H. *Org. Lett.* **2005**, *7*, 5289. (c) Asao, N.; Sato, K.; Menggenbateer; Yamamoto, Y. *J. Org. Chem.* **2005**, *70*, 3682. (d) Asao, N.; Aikawa, H.; Yamamoto, Y. *J. Am. Chem. Soc.* **2004**, *126*, 7458. (e) Straub, B. F. *Chem. Commun.* **2004**, 1726. (f) Asao, N.; Nogami, T.; Lee, S.; Yamamoto, Y. *J. Am. Chem. Soc.* **2003**, *125*, 10921. (g) Asao, N.; Takahashi, K.; Lee, S.; Kasahara, T.; Yamamoto, Y. *J. Am. Chem. Soc.* **2002**, *124*, 12650. For the sole example concerning azomethine ylides, see: (h) Kusama, H.; Miyashita, Y.; Takaya, J.; Iwasawa, N. *Org. Lett.* **2006**, *8*, 289.

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