

From an Eight-Component Self-Sorting Algorithm to a Trisheterometallic Scalene Triangle

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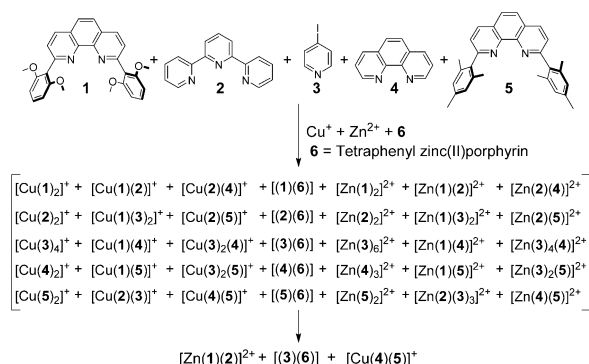
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Abstract: Using motifs from 3-fold complete self-sorting in an eight-component library, we report on the design and fabrication of a fully dynamic trisheterometallic scalene triangle, a demanding supramolecular structure that complements the so far known triangular structures.

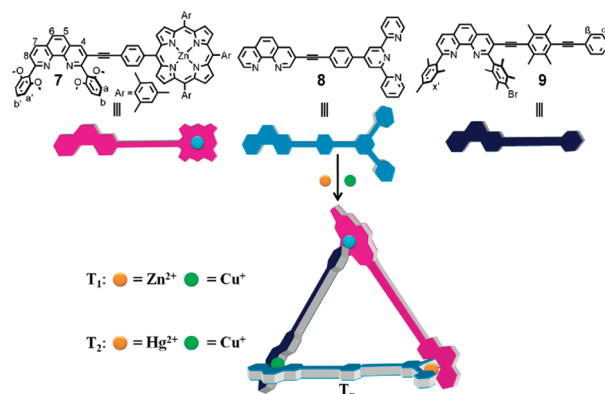
To fit with the trend of evolution (“*Evolutionary processes are anamorphic, or complexity-generating*”),¹ artificial self-assembly will need to master vastly enhanced complexity and diversity issues, for example by enlarging the number of different components and interactions.² While biological self-assembly follows intricate orthogonal self-sorting algorithms,^{3–5} artificial self-sorting is still limited in orthogonality. Indeed, most of the known self-sorting processes entail either multiple species or a single assembly along with excess ligand(s).^{3–5} In contrast, we seek to merge all members of a library in a single multicomponent aggregate employing *complete*⁵ and *integrative*⁴ self-sorting. As an example we elaborated the self-assembly of a dynamic bimetallic trapezoid from a six-component $A^1A^2D^1D^2D^3D^4$ (A = acceptor, D = donor) library.⁵ Extending the above conceptual insights to self-sorting in an eight-component $A^1A^2A^3D^1D^2D^3D^4D^5$ library (Scheme 1), we report herein on the fabrication of a trisheterometallic scalene triangle, a demanding and until now unrealized supramolecular structure (Scheme 2) that complements the so far known triangles (monometallic and equilateral,⁶ bisheterometallic,^{7,8} and isosceles^{9,10}).

Self-sorting in metallosupramolecular structures is managed by various factors, such as steric and electronic effects, π – π interactions, and metal-ion coordination specifics. For the self-sorting eight-component library depicted in Scheme 1, we chose to blend our previous $A^1A^2D^1D^2D^3D^4$ library⁵ with the pyridine–zinc porphyrin binding motif.¹¹ To our delight, full orthogonality of the pyridine–zinc porphyrin binding motif with the two other individual heteroleptic metal–ligand combinations was established. Only 3

Scheme 1. Self-Sorting in a $A^1A^2A^3D^1D^2D^3D^4D^5$ Library



Scheme 2. Synthesis of the Supramolecular Scalene Triangles T_n



out of 35 plausible combinations, *i.e.* $[\text{Zn}(\mathbf{1})(\mathbf{2})]^{2+}$, $[\text{Cu}(\mathbf{4})(\mathbf{5})]^+$, and $[(\mathbf{3})(\mathbf{6})]$, were afforded while making full use of all library members in a 3-fold complete self-sorting (readily derived from ^1H NMR analysis; see Supporting Information). Obviously, none of the available bi- or tridentate ligands can amalgamate readily with the zinc porphyrin **6** due to steric bulk, all the more as this would lead to uncoordinated nitrogen ligands in the overall mixture, which would violate the maximum site occupancy rule.^{3a}

With the complexes $[\text{Zn}(\mathbf{1})(\mathbf{2})]^{2+}$, $[\text{Cu}(\mathbf{4})(\mathbf{5})]^+$, and $[(\mathbf{3})(\mathbf{6})]$ representing the three corners of a scalene triangle, all tools are available for designing the three different sides along with their binding units (Scheme 2). Thus, we instated **1** and **6** as the termini of the phenanthroline–porphyrin hybrid **7** being readily accessible *via* Sonogashira cross-coupling (Supporting Information). Along a known procedure,⁵ **2** and **4** were merged in the terpyridine–phenanthroline ligand **8**, while the complexation properties of **3** and **5** were amalgamated within the phenanthroline–pyridine hybrid **9**. Details of the synthesis are contained in the Supporting Information. For all ligands, spacers were chosen to render the building blocks **7–9** unequal in length.

In a first set of experiments all components (**7**, **8**, **9**, Zn^{2+} , and Cu^+) were mixed in 1:1:1:1:1 ratio and refluxed for 2 h in acetonitrile/DCM (2:1). After obtaining a clear dark-violet solution the reaction mixture was characterized by electrospray ionization mass spectroscopy (ESI-MS), ^1H NMR, COSY, diffusion-ordered spectroscopy (DOSY), elemental analysis, and differential pulse voltammetry (DPV). ESI mass spectra (Figure 1) showed only peaks that were in full agreement with $T_1 = [\text{CuZn}(\mathbf{7})(\mathbf{8})(\mathbf{9})](\text{OTf})_2(\text{PF}_6)$. Most importantly, the full integrity of the scalene triangle T_1 was unambiguously proven by intense signals at $m/z = 894.9$, associated with $[\text{CuZn}(\mathbf{7})(\mathbf{8})(\mathbf{9})]^{3+}$ and at $m/z = 1415.5$, associated with $[\text{CuZn}(\mathbf{7})(\mathbf{8})(\mathbf{9})](\text{PF}_6)^{2+}$.

Data from ^1H NMR and DOSY NMR (Supporting Information) further supported the structural assignment of T_1 . A single diffusion

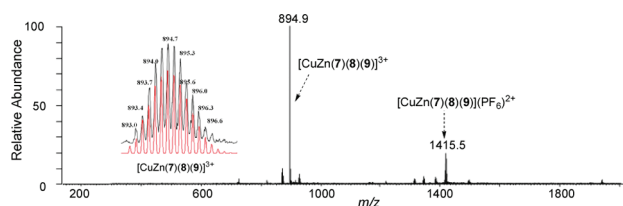


Figure 1. ESI-MS of the scalene triangle **T**₁. Inset: Experimental (black line) and calculated (red line) isotopic distribution of [CuZn(7)(8)(9)]³⁺.

coefficient as well as a single set of signals provided evidence for the clean formation of the scalene triangle **T**₁ in solution. In order to evaluate the connectivity of the ligands in **T**₁ we paid special attention to several characteristic proton resonances. For example, the pyridine protons (α , β) of **9** in **T**₁ experienced a diagnostic upfield shift from 8.61 to 2.62 ppm and from 7.38 to 5.66 ppm in the ¹H NMR, respectively, a typical shift of pyridine protons upon axial coordination to a zinc porphyrin.¹¹ Thus, following our design, ligand **7** is indeed connected to **9** by a zinc porphyrin–pyridine interaction (Scheme 2). Diagnostically shifted b-H, b'-H protons of **7** yielded further information regarding the connectivity in **T**₁. The ca. 0.60 ppm upfield shifts of b-H and b'-H protons (from 7.40 and 7.50 ppm in **7** to 6.78 and 6.95 ppm in **T**₁) are indicative of a [Zn(7_{phenAr2})(8_{terpy})]²⁺ complex.^{5,12}

The suggested structure requires that **T**₁ is chiral due to the stereogenic [Cu(8_{phen})(9_{phenAr2})]⁺ unit. As a result, several groups being homo- or enantiotopic in the individual ligands become diastereotopic in **T**₁. For example, the four methoxy groups of **7** show up as four singlets at 2.85–2.93 ppm. Their shift is indicative of a [Zn(7_{phenAr2})(8_{terpy})]²⁺ complex.⁵ Likewise, the two mesityl protons (χ') become diastereotopic in **T**₁ (δ = 5.92, 6.10 ppm). As these protons show up at δ = 6.92 ppm in **9**, their characteristic upfield shift in **T**₁ confirms the [Cu(8_{phen})(9_{phenAr2})]⁺ complexation.⁷

The assortment of the metal ions in the two metal exchanging corners of the scalene triangle was interrogated by DPV probing the Cu⁺ oxidation wave. Due to the diagnostically different redox potentials of [Cu(4)(5)]⁺ ($E_{1/2}$ = 0.44 V_{SCE}), [Cu(1)(4)]⁺ (0.29 V_{SCE}) and [Cu(1)(2)]⁺ (−0.21 V_{SCE}),⁵ a mixture of copper(I) complexes would show several copper(I) oxidation waves. A single oxidation wave at 0.76 V_{SCE} in **T**₁ (Supporting Information) confirmed the presence of only one type of copper(I) complex, pointing persuasively to the formation of [Cu(8_{phen})(9_{phenAr2})]⁺. A combination of ESI-MS, ¹H NMR, DPV, DOSY, and elemental analysis thus unambiguously provided evidence for the clean formation of scalene triangle **T**₁.

Despite the many different entities potentially arising from five donor and three acceptor units, the exclusive formation of **T**₁ based on thermodynamic equilibration is no surprise in light of the 3-fold *completive self-sorting* described in Scheme 1 and the design criteria applied to ligands **7**–**9**. Thus, it seems to be a promising strategy for future multicomponent structure design to first probe *completive self-sorting* in a library of mononuclear cornerstones and then to merge the motifs in multiligand building blocks for *integrative self-sorting*.

In another experiment, we replaced Zn²⁺ by Hg²⁺ following a similar protocol to fabricate the trimetallic scalene triangle **T**₂. **7**, **8**, **9**, Hg²⁺, and Cu⁺ were mixed in a 1:1:1:1:1 ratio and refluxed for 2 h in acetonitrile/DCM = 2:1. As for **T**₁, ¹H NMR and ESI-MS data supported the clean formation of the trisheterometallic scalene triangle **T**₂. As Hg²⁺ binds more strongly than Zn²⁺ in [M(terpy)(phenAr₂)]ⁿ⁺ complexes,¹³ we tested metal exchange in **T**₁ upon addition of Hg²⁺ (1 equiv). Complete transformation of **T**₁ to **T**₂ was noticed within one day as evidenced from ESI-MS.

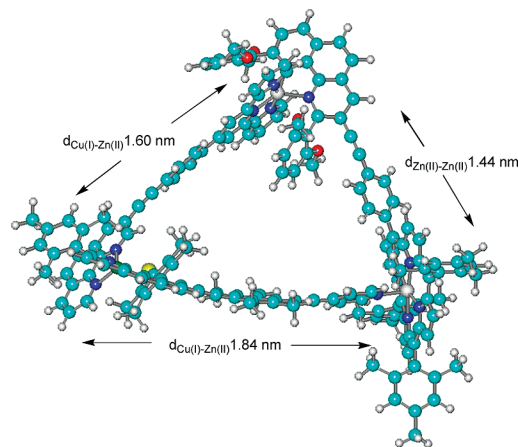


Figure 2. Energy minimized structure of the scalene triangle **T**₁. Counteranions are not included.

As all attempts to obtain a crystal structure of **T**_n were met with failure, MM⁺ force field computations and molecular dynamics on **T**_n provided some insight about their structure as scalene triangles. Taking the metal–metal distance as a measure, the three metal corners of **T**₁ are separated by 1.44, 1.60, and 1.84 nm in the energy minimized structure (Figure 2) and by 1.44, 1.66, and 1.84 nm in **T**₂ (Supporting Information), nicely illustrating the geometrical scalene arrangement of **T**_n.

In conclusion, we report on the fabrication of two scalene triangles **T**_n that were designed along the eight components of a 3-fold *completive self-sorting* library. The triangles are scalene from both a geometrical and a compositional point of view. Precise tuning of steric and electronic effects, π – π interactions, and metal-ion specifics led to the formation of a single species in solution excluding other aggregates. To the best of our knowledge, **T**₁ and **T**₂ are the first supramolecular scalene triangles with three different self-assembled corners. Furthermore, **T**₂ is the first trisheterometallic scalene triangle. Such structural diversity in a rather simple supramolecular architecture points the way to promising devices with electronically different subunits.¹⁴

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Supporting Information Available: Experimental procedures and spectroscopic data are provided for **7**, **9**, and all triangular assemblies **T**_n. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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