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Synthesis of polycyclic substituted vinylarenes via a one-pot intramolecular aryl alkylation-N-tosylhydrazone insertion reaction†

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Cite this: Chem. Commun., 2014, 50 3882

Received 30th January 2014, Accepted 14th February 2014

DOI: 10.1039/c4cc00809i

www.rsc.org/chemcomm

mild conditions.

A novel method of a palladium-catalyzed/norbornene-mediated intramolecular C-H activation/N-tosylhydrazones insertion reaction is developed. In this process, various bicyclic or tricyclic substituted vinylarenes are obtained with high efficiency under

Palladium-catalyzed cascades involving direct C-H bond activation have emerged as powerful tools for rapid access to complex highly functionalized structures. The Catellani reaction offers a unique approach to activate the ortho C-H bonds of arvl halides and to provide dual functionalizations at both the ipsoand ortho-positions.3 In the past years, many Catellani-type reactions have been reported. Seminal work by Catellani utilizing terminal alkynes⁴ or olefins⁵ as the coupling partners to trap the aryl-Pd complex provided various phenanthrenes and vinylbiphenyl compounds. After Catellani's work, some terminal reagents such as arylboronic acids,⁶ reductants⁷ were reported. In 2006, a palladiumcatalyzed tandem intramolecular ortho alkylation-cyanation reaction for the synthesis of annulated heterocycles using CN anions in the termination step was further developed by Lautens and co-workers.8 But almost all of these transformations involved the types of terminal couplings that are limited to Heck, Suzuki, Cassar-Sonogashira or hydrogenolysis reactions. Therefore, among the various new developments in modified Catellani cross-coupling reactions, the expansion of cross-coupling partners is particularly desirable.

On the other hand, numerous examples of palladiumcatalyzed multiple insertion of diazo compounds have been reported.^{9,10} In 2001, Van Vranken reported the catalytic crosscoupling using (trimethylsilyl)diazomethane as the coupling partner in Pd-catalyzed reactions. 11 Barluenga and co-workers demonstrated the application of N-tosylhydrazones in the

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Scheme 1 One pot Pd-catalyzed coupling of N-tosylhydrazones and aryl halides with tethered alkyl halides for the synthesis of polycyclic substituted vinylarenes. X, Y = I, Br. n = 1, 2, 3.

Pd-catalyzed cross-coupling with aryl halides affording the desired alkene products. 12 However, to our knowledge, applying diazo compounds to the palladium-catalyzed Catellani reaction has not been explored in the literature to date. In this context, we report a coupling of N-tosylhydrazones and aryl halides with tethered alkyl halides. This strategy allows rapid assembly of oxygen or nitrogen containing heterocyclic systems from alkyl halides under palladium and norbornene-mediated conditions, which then react with N-tosylhydrazones through an efficient sequence of Pd carbene migratory insertion and β-H elimination leading to the formation of two separate C-C bonds (Scheme 1).

After optimizing the reaction conditions, we employed aryl iodide (1a) (1.0 equiv.), N-tosylhydrazone (2a) (2.0 equiv.), Pd(OAc)₂ (10 mol%), PPh₃ (20 mol%), Cs₂CO₃ (5.0 equiv.), H₂O (5.0 equiv.), and norbornene (1.0 equiv.) in dioxane (0.1 M) at 80 °C (for details, see the ESI†).

We examined the scope of the reaction with a number of N-tosylhydrazones with different substitution patterns (Scheme 2). The reaction system displayed good tolerance toward a range of functional groups. Aromatic groups bearing electron-donating (products 3c, 3e, 3h, 3j) or electron-withdrawing substituents (products 3b, 3d, 3f, 3g, 3i, 3k) were all tolerated. Steric hindrance had little effect on this transformation, both orthoand meta-substituted N-tosylhydrazones exhibited good reactivity. 1,2-Diphenylethanyl and 1-(naphthalen-2-yl)ethanyl substituted N-tosylhydrazones were good substrates for this transformation, which smoothly afforded the desired products 3l and 3m in 92% and 66% yields, respectively. Additionally, the corresponding 3,4-dihydronaphthalenyl substituted N-tosylhydrazone

[†] Electronic supplementary information (ESI) available. CCDC 981261 (30). For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/ c4cc00809j

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Scheme 2 Pd-catalyzed synthesis of five-membered heterocycles. General reaction conditions: $\bf 1$ (0.2 mmol), $\bf 2$ (0.4 mmol), Pd(OAc)₂ (10% mmol), PPh₃ (20% mmol), Cs₂CO₃ (5.0 equiv.), norbornene (1.0 equiv.), dioxane (2 ml), H₂O (5.0 equiv.), 80 °C, 16 h.

was compatible with the reaction conditions and was converted into 3n in 78% yield. Notably, the synthesis of nitrogen-containing heterocyclic products also proved to be very efficient which were obtained in moderate to good yields (products 3o–s).

The scope of the aryl halides was investigated next (Table 1). We also investigated the feasibility of using aryl bromides under the optimized reaction conditions. To our delight, substrate **1a** and **1b** were transformed into the five-membered ring oxacycle **3a** in 89% and 61% yields, respectively. Considering the reaction, maybe **1b** is less competitive toward Pd(0) oxidative addition than aryl iodide **1a**. Extension to larger ring systems was also possible under these conditions affording the desired six- and seven-membered rings in moderate to good yields. Interestingly, the methyl containing aryl iodides exhibited higher reactivity than aryl bromides except in the case of entry 3. However, the methoxy containing aryl iodides with tethered alkyl iodide were less reactive than tethered alkyl bromide (entries 4–6).

Following the success of the formation of the bicyclic substrates, we explored the possibility of tricyclic products through a double *ortho*-alkylation–carbene migratory insertion and β -H elimination sequence (Table 2). Gratefully, the 5, 6, 5-ring system product 5a was prepared in 81% yield (entry 1), While the larger 6, 6, 6- and 7, 6, 7-ring systems were only generated in 46% and 42% yields, respectively (entries 2 and 3). In addition, tricyclic compound 5d was prepared in 79% yield from the corresponding *p*-methoxy *N*-tosylhydrazone (entry 4).

Table 1 Synthesis of bicyclic vinylarenes *via* a palladium-catalyzed/norbornene-mediated tandem reaction^a

R_	n = 1,2,3 X, Y = Br, I 2a	Pd(OAc) ₂ , PPh ₃ norbornene, H ₂ O Cs ₂ CO ₃ , dioxane 80 °C, 16 h	3
Entry	Substrates	Products	Yield ^b (%)
1	1a, Y = I 1b, Y = Br	3a	89 61
2	1c, Y = I 1d, Y = Br	31	60 49
3	1e, Y = I 1f, Y = Br	340	55 60
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^a Conditions: **1** (0.2 mmol), **2a** (0.4 mmol), Pd(OAc)₂ (10% mmol), PPh₃ (20% mmol), Cs₂CO₃ (5.0 equiv.), norbornene (1.0 equiv.), dioxane (2 ml), H₂O (5.0 equiv.), 80 °C, 16 h. ^b Yield of the isolated product.

1k, X = Br 1l. X = I

To fully demonstrate the applicability of this methodology, we attempted using a substrate lacking an *ortho* substituent **1t**, tosylhydrazone **2a** and alkyl iodide **6a** for the multicomponent reaction. We were able to obtain the product in satisfactory yield (Scheme 3).

In summary, we have developed a distinct strategy to synthesize polycyclic substituted vinylarenes using a palladium-catalyzed/norbornene-mediated tandem *ortho* alkylation-metal carbene insertion reaction system. This method has been applied toward the synthesis of a variety of synthetically useful bicyclic and tricyclic vinylarenes products. The reaction features good yields, excellent functional-group tolerance, and a wide compatibility of structurally and electronically varied substrates, making it applicable to structurally complex compounds. In addition, this modified Catellani-type C-H activation provides

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Table 2 Synthesis of tricyclic vinylarenes *via* a palladium-catalyzed/norbornene-mediated tandem reaction^a

NNHTs
$$Pd(OAc)_2, PPh_3$$
 norbornene, H_2O Cs_2CO_3 , dioxane $O(Cs_2CO_3)$ $O(Cs_2$

	4 2	80 C, 16 h	5
Entry	Substrates	Products	Yield ^b (%
1	4a	Ph 5a	81
2	4b	Ph 5b	46
3	4c	Ph	42
4		OMe	79
	4b	5d	

 a Conditions: 4 (0.2 mmol), 2 (0.4 mmol), Pd(OAc) $_2$ (10% mmol), PPh $_3$ (20% mmol), Cs $_2$ CO $_3$ (5.0 equiv.), norbornene (2.0 equiv.), dioxane (4 ml), H $_2$ O (5.0 equiv.), 80 °C, 16 h. b Yield of the isolated product.

Scheme 3 Three component reaction. Conditions: **1t** (0.2 mmol), **2a** (0.4 mmol), **6a** (0.6 mmol), Pd(OAc)₂ (10% mmol), PPh₃ (20% mmol), Cs₂CO₃ (5.0 equiv.), norbornene (2.0 equiv.), dioxane (4 ml), H_2O (5.0 equiv.), 80 °C, 16 h.

broad implications for developing various dual functionalizations of aryl halides.

We thank the National Science Foundation (NSF 21072080 and 21272101), National Basic Research Program of China (973 Program) 2010CB833203 and "111" program of MOE and PCSIRT: IRT1138 for financial support.

Notes and references

(a) L. Ackermann, V. Rubén and R. K. Anant, *Angew. Chem., Int. Ed.*,
2009, 48, 9792; (b) T. W. Lyons and M. S. Sanford, *Chem. Rev.*, 2010,
110, 1147; (c) G. P. Chiusoli, M. Catellani, M. Costa, E. Motti,

- N. Della Ca' and G. C. Maestri, *Chem. Rev.*, 2010, **254**, 456; (d) D. Alberico, M. E. Scott and M. Lautens, *Chem. Rev.*, 2007, **107**, 174; (e) D. Kalyani, A. D. Satterfield and M. S. Sanford, *J. Am. Chem. Soc.*, 2010, **132**, 8419; (f) N. D. Ball, J. W. Kampf and M. S. Sanford, *J. Am. Chem. Soc.*, 2010, **132**, 2878; (g) D. C. Powers and T. Ritter, *Nat. Chem.*, 2009, **1**, 302; (h) T. Tsujihara, K. Takenaka, K. Onitsuka, M. Hatanaka and H. Sasai, *J. Am. Chem. Soc.*, 2009, **131**, 3452
- 2 For reviews of Catellani reaction, see: (a) R. Ferraccioli, Synthesis, 2013, 581; (b) M. Catellani and M. C. Fagnola, Angew. Chem., Int. Ed., 1994, 33, 2421; (c) C. Amatore, M. Catellani, S. Deledda, A. Jutand and E. Motti, Organometallics, 2008, 27, 4549; (d) M. Catellani, E. Motti and N. Della Ca', Acc. Chem. Res., 2008, 41, 1512.
- 3 For recent achievements, see: (a) D. A. Candito and M. Lautens, Angew. Chem., Int. Ed., 2009, 48, 6713; (b) G. Maestri, N. Della Ca' and M. Catellani, Chem. Commun., 2009, 4892; (c) M. Blanchot, D. A. Candito, F. Larnaud and M. Lautens, Org. Lett., 2011, 13, 1486; (d) L. Jiao and T. Bach, J. Am. Chem. Soc., 2011, 133, 12990; (e) L. Jiao, E. Herdtweck and T. Bach, J. Am. Chem. Soc., 2012, 134, 14563; (f) L. Jiao and T. Bach, Angew. Chem., Int. Ed., 2013, 52, 6080.
- 4 M. Catellani, E. Motti and S. Baratta, Org. Lett., 2001, 23, 3611.
- (a) E. Motti, G. Ippomei, S. Deledda and M. Catellani, Synthesis, 2003, 2671; (b) M. Catellani, S. Deledda, B. Ganchegui, F. Hénin, E. Motti and J. Muzart, J. Organomet. Chem., 2003, 687, 473; (c) E. Motti, N. Della Ca', S. Deledda, E. Fava, F. Panciroli and M. Catellani, Chem. Commun., 2010, 46, 4291; (d) S. Pache and M. Lautens, Org. Lett., 2003, 5, 4827; (e) E. Motti, F. Faccini, I. Ferrari, M. Catellani and R. Ferraccioli, Org. Lett., 2006, 8, 3967; (f) H. Liu, M. E. Salfiti, D. I. Chai, J. Auffret and M. Lautens, Org. Lett., 2012, 14, 3648.
- 6 E. Motti, A. Mignozzi and M. Catellani, J. Mol. Catal. A: Chem., 2003, 204–205, 115.
- 7 (a) T. Wilhelm and M. Lautens, Org. Lett., 2005, 7, 4053; (b) A. Martins and M. Lautens, Org. Lett., 2008, 10, 5095; (c) A. Martins, D. A. Candito and M. Lautens, Org. Lett., 2010, 12, 5186.
- 8 (a) B. Mariampillai, D. Alberico, V. Bidau and M. Lautens, J. Am. Chem. Soc., 2006, 128, 14436; (b) B. Mariampillai, J. Alliot, M. Li and M. Lautens, J. Am. Chem. Soc., 2007, 129, 15372.
- (a) S. Chen and J. Wang, Chem. Commun., 2008, 4198; (b) C. Peng, Y. Wang and J. Wang, J. Am. Chem. Soc., 2008, 130, 1566; (c) R. Kudirka, S. K. J. Devine, C. S. Adams and D. L. Van Vranken, Angew. Chem., Int. Ed., 2009, 48, 3677; (d) W.-Y. Yu, Y.-T. Tsoi, Z. Zhou and A. S. C. Chan, Org. Lett., 2009, 11, 469; (e) Y.-T. Tsoi, Z. Zhou, A. S. C. Chan and W.-Y. Yu, Org. Lett., 2010, 12, 4506; (f) Z.-S. Chen, X.-H. Duan, P.-X. Zhou, S. Ali, J.-Y. Luo and Y.-M. Liang, Angew. Chem., Int. Ed., 2012, 51, 1370.
- 10 (a) B. Tréguier, A. Hamze, O. Provot, J.-D. Brion and M. Alami, Tetrahedron Lett., 2009, 50, 6549; (b) S. Messaoudi, B. Tréguier, A. Hamze, O. Provot, J.-F. Peyrat, J. R. De Losada, J.-M. Liu, J. Bignon, J. Wdzieczak-Bakala, S. Thoret, J. Dubois, J.-D. Brion and M. Alami, J. Med. Chem., 2009, 52, 4538; (c) S. K. J. Devine and D. L. Van Vranken, Org. Lett., 2007, 9, 2047; (d) C. Peng, J. Cheng and J. Wang, J. Am. Chem. Soc., 2007, 129, 8708; (e) S. K. J. Devine and D. L. Van Vranken, Org. Lett., 2008, 10, 1909For reviews, see: (f) Y. Zhang and J. Wang, Eur. J. Org. Chem., 2011, 1015; (g) J. Barluenga and C. Valdés, Angew. Chem., Int. Ed., 2011, 50, 7486; (h) Z. Shao and H. Zhang, Chem. Soc. Rev., 2012, 41, 560; (i) Z.-S. Chen, X.-H. Duan, L.-Y. Wu, S. Ali, K.-G. Ji, P.-X. Zhou, X.-Y. Liu and Y.-M. Liang, Chem.-Eur. J., 2011, 17, 6918; (j) Q. Xiao, Y. Zhang and J. Wang, Acc. Chem. Res., 2013, 46, 236; (k) Y. Xia, Y. Zhang and J. Wang, ACS Catal., 2013, 3, 2586.
- 11 K. L. Greenman, D. S. Carter and D. L. Van Vranken, *Tetrahedron*, 2001, 57, 5219.
- 12 J. Barluenga, P. Moriel, C. Valdés and F. Aznar, Angew. Chem., Int. Ed., 2007, 46, 5587.