369 Number 15, 1965

## Diels-Alder Reactions of 9,10-Anthraquinodimethane

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REACTION of an o-quinodimethane with a dienophile usually gives a 1:1 Diels-Alder adduct.1,2,3 Investigation of the addition reactions of 9,10anthraquinodimethane (I), a p-quinodimethane, with various dienophiles, shows that this reactive intermediate reacts with two molecules of the Other cases of p-quinodimethanes dienophile. reacting in this way do not appear to have been reported. The intermediate (I) is liberated from cis- or trans-9,10-dimethyl-9,10-dihydro-9,10-dihydroxyanthracene, by dehydration in a suitable

solvent, in the presence of an excess of the dienophile.4 Even with a large excess of the intermediate (I) no product other than a di-adduct was isolated; or, in nitrobenzene as solvent, the product of its aromatization.

Reaction with dimethyl acetylenedicarboxylate in boiling nitrobenzene gives tetramethyl 4,5benzopyrene-1,2,7,8-tetracarboxylate (II) (m.p. 321°, 23% yield). Alkaline hydrolysis gives 4,5-benzopyrene-1,2,7,8-tetracarboxylic acid which readily gives the corresponding dianhydride

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(m.p. >360°). Decarboxylation of the acid gives 4,5-benzopyrene (37% yield), thus excluding the

$$\begin{array}{c} CH_2 \\ CH_2 \\ CH_2 \\ CO_2Me \\ CO_2Me \\ \end{array}$$

$$(IV)$$

<sup>5</sup> E. Clar and M. Zander, Tetrahedron, 1963, 19, 521.

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possibility that the initial reaction gives a perylene derivative.

In boiling acetic ahydride, 9,10-anthraquinodimethane reacts with excess of maleic anhydride 4,5-benzo-1,2,3,6,7,8,8a,10a-octahydropyrene-1,2:7,8-dicarboxylic anhydride (m.p. 230°, 60% yield) which on dehydrogenation gives the fully aromatic dianhydride identical with that prepared as described above.

With 1,4-naphthaquinone, in boiling nitrobenzene, 9,10-anthraquinodimethane gives 4,5benzodinaphtho[2',3':1,2][2'',3'':7,8]pyrene-1',4':1'', 4''-diquinone (III) (m.p. >360°, 46% yield). If the reaction is stopped after 10 minutes a mixture of (III) and its hydrogenated derivatives is obtained. The progress of aromatization may be followed by the increase in intensity of a very strong absorption band at 707 cm.-1 which we attribute to the out-of-plane deformation absorption of the four adjacent aromatic C-H bonds on the terminal rings adjacent to the pquinone rings. The frequency of this band is outside that normally attributed to four adjacent aromatic C-H bonds (770 to 735 cm.-1), but we base our assignment on analogy with 5,12-dihydro-3,4-benzonaphtho-[1',2':1,2]naphthacene-6,11-quinone (IV) (m.p.  $322-324^{\circ}$ ; prepared in 56%yield by the reaction of 5,6-chrysaquinodimethane with 1,4-naphthaquinone in diethyl phthalate) which shows a similar very strong band at 707 cm.<sup>-1</sup> (The infrared spectrum of 9,10anthraquinone shows an absorption at 692 cm.<sup>-1</sup> due to aromatic C-H). Reduction of the diquinone (III) gives 4,5-benzodinaphtho[2',3':1,2]pyrene thus confirming the structure previously assigned to this hydrocarbon by Clar.<sup>5</sup>

(Received, July 8th, 1965; Com. 427.)