A Mild Retro- $[_{\pi}4_{s}+_{\pi}2_{s}]$ Cleavage Route to Furans and Fulvenes

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Summary A new route to β -substituted furans and fulvenes has been developed which is based on the ready thermal fragmentation of 4,5-diazatricyclo[6,2,1,0^{2,5}]-undeca-3,5,9-trienes.

Retro- $[_{\pi}4_8 + _{\pi}2_8]$ (Diels-Alder) reactions are commonly used in synthesis.¹ These reactions are conducted under thermal conditions and extreme temperatures are often required to effect fragmentation.¹ This is one limitation where temperature sensitive products are the goal,² although this has been partly offset by flash vacuum pyrolysis techniques. This latter technique is elegantly demonstrated by the recent report by de Mayo and his co-workers on the synthesis of pentalene.³

We⁴ and others⁵ have observed that derivatives of the tricyclo[6,2,1,0^{2,5}]undeca-3,5,9-triene system (1) are particularly labile in respect to retro-[$_{\pi}4_{8}+_{\pi}2_{8}$] cleavage, especially when the $_{\pi}2_{8}$ component becomes part of an aromatic system. We have now developed a scheme for the synthesis of β -substituted furans and fulvenes which is based on the fragmentation of the 4,5-diaza-analogue of (1). The features of this scheme are: (i) simple formation of the diaza-intermediate, (ii) low temperatures involved, especially in the fragmentation step, (iii) essentially quantitative conversions in each step.

In order to evaluate the stability of the 4,5-diazatricyclo-[6,2,1,0^{2,5}]undeca-3,5,9-triene system, the 3,6-di-(2'-pyridyl) derivative (3) was prepared from the reaction of equimolar

(1)
$$R=2^{1}$$
-Pyridyl

(1) $R=2^{1}$ -Pyridyl

(2) $R=2^{1}$ -Pyridyl

(3) $R=2^{1}$ -Pyridyl

(4)

the literature.2,6 The standard method involves the following steps: (a) Formation of bicyclic adduct, (5)—(7); (b) Selective hydrogenation to the dihydro-derivative (11)—(13); (c) High temperature pyrolysis (195—400°) to the 5-membered product (8)—(10) plus ethylene.

Reaction of the 7-substituted norbornadienes (5)-(7) readily occurred with the s-tetrazine (2) under mild con-

Summary of conditions for conversion of 7-substituted norbornadienes into furans and fulvenes

TABLE

| $egin{array}{ccc} 	ext{Substituents} & 	ext{Y} \end{array}$ | | Dihydro- compound | Starting material | Reaction conditions | Product | ¹ H n.m.r. (δ p.p.m.) ^a of product |
|---|-------------|----------------------|----------------------|--|-------------------|---|
| CF ₃ | 0 | (11) | (5) | room temp., CDCl ₃ few h | (8) b | 7.90 (s, vinylic) |
| CO ₂ Me | $C = CPh_2$ | (12) | (6) | room temp., CDCl ₃ , overnight | (9) c | 7·35—7·65 (m, phenyl) 6·97 (s, vinylic) 3·85 (s, OMe) |
| CO ₂ Me | $C = CMe_2$ | (13) | (7) | 50°, CDCl ₃ , 2 h | (10) ^d | 7·16 (s, vinylic) 3·83 (s. OMe) 2·31 (s, Me) |

a 60 MHz, CDCl₃, MeSi as internal standard. b Low boiling liquid, isolated by g.l.p.c. (carbowax 20m, fluoroport, 58°) (m/e 204). c M.p. 166°, reported² 165—167°, isolated by preparative layer chromatography (silica, CHCl₃) (m/e 350). d Unstable low-melting solid, isolated by short path distillation (m/e 222).

amounts of 3,6-di-(2'-pyridyl)-s-tetrazine and norbornadiene in CDCl₃ at -20° . A rapid reaction occurred and nitrogen was eliminated to yield (3) in quantitative yield [1H n.m.r., δ 1·19, 1·28 (2H, AB q, J 9 Hz, 11-H₂), 3·40 (2H, m, 1-, 8-H), 3.54 (2H, s, 2-, 7-H), 6.50 (2H, t, 9-,10-H), 7.3—8.90 p.p.m. (8H, complex m, pyridyl)]. The lack of coupling between C-2(7) and C-1(8) supported the suggested stereochemistry, and is consistent with the s-tetrazine approaching from the less hindered exo-side. Intermediate (3) was stable below -10° , but above this temperature fragmentation to 3,6-di-(2'-pyridyl)pyridazine (4) and cyclopentadiene occurred.

In order to compare our method with the standard retro- $[\pi^4 + \pi^2]$ route used in synthesis of fulvenes² and furans, we have chosen three reactions already reported in

ditions (see Table) to produce the corresponding 5-membered products (8)—(10) plus the pyridazine (4). ¹H N.m.r. spectroscopy showed that the conversions were quantitative, although the isolated yields were only in the range 70-90%. The spectral properties of the products were in agreement with the assigned structures (see Table). Dimethyl 6,6-dimethylfulvene-2,3-dicarboxylate was only moderately stable, and this would account for the lack of success reported by Alder and Trimborn² using the standard retro- $[\pi 4_8 + \pi 2_8]$ route in attempts to prepare this

The generality of this method is demonstrated by the synthesis of isobenzofuran, isobenzofulvenes, and isoindoles.8b

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