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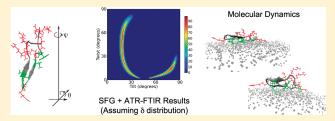
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Interfacial Orientation and Secondary Structure Change in Tachyplesin I: Molecular Dynamics and Sum Frequency Generation Spectroscopy Studies

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Supporting Information

ABSTRACT: Recent advances in the collection and interpretation of surface-sensitive vibrational spectroscopic measurements have made it possible to study the orientation of peptides and proteins in situ in a biologically relevant environment. However, interpretation of sum frequency generation (SFG) and attenuated total reflectance Fourier transform infrared (ATR-FTIR) vibrational spectroscopy is hindered by the fact that orientation cannot be inferred without some prior knowledge of the protein



structure. In this work, molecular dynamics simulations were used to study the interfacial orientation and structural deformation of the short β -sheet peptide tachyplesin I at the polystyrene/water interface. By combining these results with ATR-FTIR and SFG measurements, reasonable agreement was found with the simulation results, suggesting that tachyplesin I lies parallel to the surface, although the simulation results imply a broader distribution of peptide twist angles than could be characterized using available experimental measurements. The interfacial structure was found to be deformable even when disulfide bonds were preserved, and these local deviations from a purely extended β -sheet conformation may be of importance to future developments in the interpretation of SFG and ATR-FTIR spectra.

■ INTRODUCTION

Interfaces are ubiquitous in biology, and the interaction of peptides or proteins with polymers, lipids, and other materials must be understood to address questions such as biocompatibility, biofouling, enzyme immobilization, and the behavior of membrane-active peptides and proteins. However, few techniques possess the sensitivity and specificity to examine a thin layer of biomolecules at the interface in a biologically relevant environment, while also measuring quantitative information about the biomolecular internal conformation and overall orientation. Although indisputably valuable, many biophysical techniques for the study of proteins have limitations such as interfering signals due to water absorption, the requirement for high vacuum, large sample requirements, or a lack of inherent surface specificity. See

In recent years, sum frequency generation (SFG) vibrational spectroscopy has been applied to amino acids, peptides, and proteins, and numerous advantages have been demonstrated. ^{9–30} It has been found that SFG is capable of detecting even submonolayer amounts of protein in situ in aqueous environments at polymer or lipid interfaces. ³¹ Previously, it has been shown that polarized SFG signal intensity ratios can be used to quantitatively determine the orientation of α -helical peptides. ^{12,15,16} This was the

first protein secondary structure motif for which SFG orientation measurements became possible, due to the fact that the well-ordered structure leads to strongly detectable signals, and because only a single orientation angle (tilt) needs to be characterized due to its inherited symmetry properties. Characterizing a single angle is feasible even with a small set of measurements. A similar approach was also used to determine the orientation of 3_{10} helical structures, as in alamethicin. 32

Recently, it has been shown that SFG is similarly capable of studying the orientation of antiparallel β -sheet secondary structures, which consist of two hydrogen-bonded β -strands in an antiparallel orientation. ^{12,17,23-25,33-36} Since signal intensities are known to be related to the number of molecules at the interface, a recent study on β -sheet orientation ensured a high surface coverage of β -sheets by using the small peptide tachyplesin I (Figure 1), which consists of 17 residues (including six in the β -sheet conformation).

To apply SFG to the study of β -sheet orientation, a number of challenges must be addressed. First, characterizing two angles

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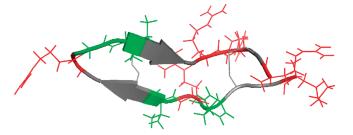


Figure 1. Tachyplesin I is a 17-residue β-sheet peptide with a mix of hydrophobic (green) and charged (red) residues.

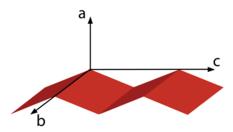


Figure 2. Molecular coordinate system used in this work. The long axis of the β -strands is oriented parallel to the c axis, while the plane of the β -sheet corresponds to the bc plane. The twist angle ψ represents rotation around the molecular c axis, and the tilt angle θ represents rotation around the molecular b axis.

(tilt and twist) with a limited set of measurements is only possible if one assumes that all molecules adopt the same orientation (δ distribution). In fact, the structure of tachyplesin I is not as inherently facially amphiphilic³⁷ as some previously studied helical peptides, ^{18,20} so it is unclear whether a δ orientation distribution is actually realistic at a model hydrophobic/hydrophilic boundary such as the polystyrene (PS)/water interface. Second, although a small β -sheet peptide was chosen as a means to increase the surface coverage of the secondary structure of interest, the actual region of tachyplesin I that adopts that secondary structure is fairly short: three residues per β -strand. As a result, it is quite possible that the structure could deform significantly when adsorbed. This deformation is not taken into account in the SFG orientation analysis relations, which assume that the strands of the β -sheet are in the fully extended conformation. Third and last, although a combination of attenuated total reflectance Fourier transform infrared (ATR-FTIR) and SFG measurements was used in the previous study, the orientation relations used for ATR-FTIR were not capable of extracting the full range of information in the spectrum. This is a consequence of the fact that FTIR is only capable of obtaining one measurement, yet the orientation of β -sheets would be more accurately described using two angles (tilt and twist, as shown in Figure 2). Previous orientation analysis methodologies have therefore required the assumption that one orientation angle can be averaged out ^{38,39} or held fixed. ^{17,40,41}

It is well-known that few β -sheet structures exist entirely in the extended conformation, and most sheets instead adopt a small twisting of the constituent strands. This inherent deformation (which is not to be confused with the twist angle used to characterize molecular orientation) can vary depending on the structure, as it depends on a variety of factors, including hydrogen bonding, the inherent chirality of the constituent amino acids, and side chain packing. To address questions about conformational

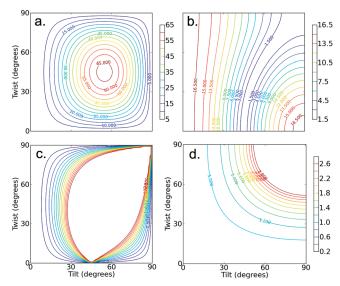


Figure 3. Calculated contour plots showing the relative magnitudes of (a) $\chi^{(2)}_{ssp}$, (b) $\chi^{(2)}_{ssp}$, and (c) the fitted ssp/spp ratios (these ratios incorporate the Fresnel factors directly) for the B_2 mode. (d) Contour plot showing the FTIR dichroic ratio for the B_2 mode.

flexibility, we used all-atom molecular dynamics (MD) simulations aimed at addressing the above concerns and provide a more complete understanding of β -sheet orientation at interfaces. MD simulations have been widely used to study peptides and proteins on surfaces of varying charge and hydrophilicity, using force fields optimized to reproduce biomolecular structure and dynamic behavior. In recent years, early work using simple model surfaces and implicit solvent or simplified coarse-grained models has given way to studies that increasingly employ polymer, inorganic, lipid, and self-assembled monolayer (SAM) interfaces as model systems. 47-54 Due to the large amounts of computer processing time required for extensive characterization, the use of more detailed model systems requires a number of trade-offs. Our simulations employed a simplified polystyrene (PS)/water interface in a spherical boundary that balances the competing requirements of detail and computational performance. This model hydrophobic/hydrophilic interface is also amenable to characterization using a combination of SFG and MD. We believe that combining these techniques makes it possible to understand the interfacial orientation and structural flexibility of tachyplesin I in more detail, and will facilitate future studies of other peptides and proteins with β -sheet structures at interfaces.

■ METHODS

SFG Data Analysis. It has only recently become possible to analyze the orientation of β -sheets on the basis of SFG measurements, following the derivation of the required molecular hyperpolarizability and surface susceptibility tensor components. ^{17,36} Although a range of polarizations and vibrational modes were found capable of producing SFG amide I signals, the B1 and B3 amide I vibrational modes are relatively weak. The most important SFG measurement available in SFG studies on β -sheets is the single intensity ratio $\chi^{(2)}_{xxz}/\chi^{(2)}_{yzx}$ of the B2 mode. ^{17,36} Plots of the most relevant experimental observables are shown in Figure 3, and the relevant equations are provided in the Supporting Information.

MD Simulations. The solvent/surface model system used for these simulations consists of an array of ethylbenzene molecules for the model PS surface and polymer bulk, solvated using TIP3 explicit solvent

Table 1. Initial Conditions Used for the Simulations^a

		PDB rotation (deg)		Euler angles (deg)	
identifier	length (ns)	tilt	twist	tilt	twist
tachx270*	12 (18*)	90	270	100	65
tachx0*	4.8 (18*)	90	0	110	-32
tachx180*	12	90	180	76	28
tachx315*	12	90	315	119	21
tachx235*	6 (18*)	90	235	109	42
tachx90	2	90	90	77	-75

^a PDB rotation indicates the angles used to rotate the entire peptide molecule from coordinates in the PDB file (based on the backbone atoms in all 17 residues). Angles are also listed in a convention consistent with SFG analysis, determined on the basis of only the six residues in the β-sheet portion of the peptide. The angles determined via these two methods differ because the sheet is not perfectly aligned with the plane of the peptide. Simulation identifiers marked with an asterisk were also performed a second time with disulfide bonds removed.

constrained in a spherical "droplet in a vacuum" system of 36 Å radius using a spherical quartic boundary potential. Previous studies have shown that the presence of a water/vacuum boundary primarily affects the first few layers of water, so that the structure of the water in the center of the droplet remains reasonable. More details and discussion of this model system may be found in a previous paper. 18 The structure of the tachyplesin I peptide was obtained from the Protein Databank (PDB ID 1ma2). Structure 9 of the 31-conformer NMR ensemble was extracted and used as the starting structure for all simulations. This starting structure was chosen due to the roughly parallel orientation of the two strands, particularly in the highly flexible disordered regions of the structure. This conformation therefore provides the best chance for all portions of the peptide to interact with the surface in a variety of initial simulation conditions. For a hydrophobic polymer surface, one would expect that hydrophobic interactions would dominate the adsorption behavior. 55-57 However, the structure of tachyplesin I does not possess the same strong facial amphilicity 37 as was present in magainin 2 or LK14 (studied previously). 18,20,22 Therefore, a variety of simulations were performed with the peptides placed in different starting orientations.

The peptide was placed into the model solvent/surface system above the surface, so that the closest point of the peptide was \sim 2.5 Å from the surface. The peptide was oriented along the x axis of the coordinate system and then rotated into the respective coordinates for a variety of starting orientations as specified in Table 1. Overlapping water molecules were removed, and peptide coordinates were held fixed as an additional 500 ps of equilibration was then performed for the solvent and surface in the presence of peptide. No counterions were included.

For production runs, dynamics were performed using a Nosé—Hoover thermostat at 298 K with the velocity-Verlet algorithm. The SHAKE algorithm was used to constrain all bonds involving hydrogen atoms, allowing the use of a 2 fs time step. All setup, equilibration, and dynamics were performed using version c34b1 of the CHARMM molecular dynamics package with the CHARMM22 all-atom parameter set. ^{58,59} These parameters were used for the water, protein, and surface residues. In the latter case, although not optimized for polymers, the CHARMM22 parameters should adequately represent the primarily hydrophobic interactions that dominate behavior of the polymer surface. A cutoff of 12.0 Å was used when calculating the nonbonded interactions.

Experiments have shown that the presence of the two disulfide bonds in tachyplesin I is key to retention of the β -sheet structure. ⁶⁰ The loss of the two disulfide bonds may greatly affect the tachyplesin I—PS surface interactions. For comparison purposes, MD simulations were also performed on peptides with the disulfide bonds removed.

The most relevant exploratory simulations are summarized in Table 1, where the angles specified represent rotation of the entire PDB coordinate file to the laboratory coordinate system. To facilitate comparison to results from SFG spectroscopy, these initial positions were also calculated in terms of the Euler angles required to rotate a β -sheet from the molecular coordinate system to the specified position defined in the laboratory coordinate system. In Table 1, not all tilt angles (the angle between the β -sheet c axis and the surface normal) determined using this latter procedure are exactly equal to 90° (peptide lying down). This is because the six β -sheet residues are not perfectly aligned with the overall peptide plane defined by all 17 total residues. The Euler angles θ and ψ are defined on the basis of two vectors defined using backbone atoms. These vectors are oriented parallel and perpendicular to the sheet axis in the molecular coordinate system, respectively, and may be calculated from the rotation matrix elements based on the components of each of those vectors:

$$\theta = \cos^{-1} \left(\frac{z_{\parallel}}{|\text{vec}_{\parallel}|} \right)$$

$$\psi = \sin^{-1} \left(\frac{z_{\perp}}{|\text{vec}_{\perp}| \sin \theta} \right)$$
(1)

where veq_{\parallel} refers to the vector running parallel to the long axis of the strands and vec_{\perp} refers to the vector running perpendicular to the long axis of the strands in the plane of the β -sheet.

For final data analysis, description of the resulting orientation is hampered by the fact that a significant portion of the peptide may be disordered. (At best, only 6 of the 17 residues comprise the β -sheet region.) Thus, in all results reported, the orientation of the molecule was defined solely from the β -sheet portion of the molecule (residues 6-8 and 11-13) using the latter Euler angle convention described above.

■ RESULTS

Molecular Dynamics. Including simulations in which the disulfide bonds were removed, a total of 11 molecular dynamics simulations spanning a total of \sim 120 ns were evaluated here. Of these, two initial conditions (tachx90 and tachx235) showed no or poor interaction with the surface and were therefore excluded from the analysis. Since the peptide was allowed to move and diffuse freely to better accommodate formation of peptide—surface interactions, all windows presented were monitored carefully to ensure that the peptide did not randomly diffuse too close to the edges of the spherical model system (at the water/vacuum boundary) during the time scales studied.

In general, all simulations ended with the long axis of the peptide lying roughly parallel to the plane of the surface (tilt of 90°), but a variety of twist angles were observed; the two most common local minima in time-dependent plots of orientation were twist angles centered around 0° and 45° (Figure 4). The changing angle between the β -strands (measured from vectors running through the backbone atoms of the two individual β -strands) also indicates that structural deformation of the β -sheet region occurs. Since analysis of molecular orientation in MD trajectories depends on the positions of selected backbone atoms (residues 6–8 and 11–13), it is important to note that the orientations listed will only have physical meaning if the peptide retains the β -sheet conformation at the end of the simulations.

Adsorption and Orientation. To understand the factors that affect adsorption of tachyplesin I, a variety of molecular dynamics simulations were performed in which the peptide was placed at different initial orientations. The results (Table 2) are striking for the variety of final orientations and structures observed. The

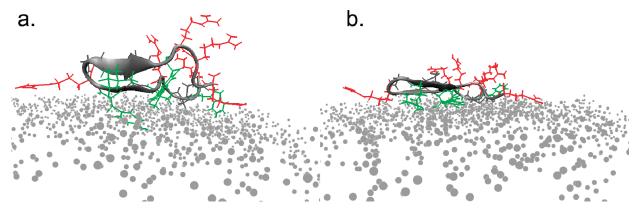


Figure 4. Two candidate final orientations of the peptide show qualitatively similar trends in surface area burial. Hydrophobic side chains are shown in green, and charged side chains are drawn in red. Surface residues are shown as gray dots, and water molecules are not shown. The peptide adopts a twist angle of (a) \sim 50° (simulation tachx270) or (b) \sim 10° (simulation tachx315).

Table 2. Final Orientation of the β -Sheet Section of the Peptide When Disulfide Bonds Were Retained^a

	with three residues			with two residues		
name	tilt (deg)	twist (deg)	interstrand angle (deg)	tilt (deg)	twist (deg)	interstrand angle (deg)
tachx270	107 ± 2.8	52 ± 5.3	12	106 ± 3.3	28 ± 3.9	34
tachx0	94 ± 5.3	60 ± 6.9	39	92 ± 5.5	48 ± 6.5	12
tachx180	88 ± 4.2	-8 ± 7.2	28	102 ± 5.3	-5 ± 7.5	28
tachx315	85 ± 3.5	12 ± 4.8	9	83 ± 3.3	13 ± 4.7	23

^a Angles are from the average of the last 50 frames in a rotation convention consistent with SFG data analysis. To account for deformation of the sheet structure, results for the entire sheet (residues 6–8 and 11–13) were compared to angles for a shorter segment (residues 6 and 7 and 12 and 13). The error bars quoted on tilt and twist angles represent 1 standard deviation.

simulations reported in Table 2 focus on the subset of ten trajectories that showed some interaction with the surface: five in which the disulfide bonds were kept intact and five additional windows in which the disulfide bonds were removed initially. It is interesting to note that, in one case (tachx235), the peptide only adsorbed when disulfide bonds were broken. This suggests that in this case reorganization of the backbone may help to accommodate rearrangements of side chains and alleviate unfavorable contacts with the surface.

Side chain rearrangements were also seen to dictate adsorption orientation and behavior in other simulation windows on the time scales studied. Reorganization of the bulky tyrosine side chains at residues 8 and 13 was found to influence the total disruption of secondary structure in the window tachx180. In the simulation window tachx270, bad side chain contacts were more transient, as the loss and recovery of secondary structure correlated to initial burial of the aliphatic chain of an arginine at the hairpin turn (residue 9). The adsorbed peptide eventually forms more (and more favorable) hydrophobic contacts, and recovery of the β -sheet structure in tachx270 coincides with release of arginine from the surface. In the analogous simulation without disulfide bonds, no such disruption of secondary structure was seen on the time scales studied, suggesting that flexibility of the peptide backbone allows side chains to reorganize more readily.

Structural Deformation. A variety of structural deformation metrics were applied in an attempt to better understand the behavior of the peptide at the interface:

(a) Root-mean-squared displacement (rmsd) was evaluated as a metric for deformation of the β -sheet portion of the

- peptide (backbone atoms of residues 6-8 and 11-13), as well as for the backbone atoms of all residues.
- (b) To distinguish strands in the extended conformation from fully hydrogen-bonded β -sheets, the number of hydrogen bonds in the β -sheet and backbone was quantified over time. Hydrogen bonds were quantified using the default CHARMM cutoffs (4.5 Å donor—acceptor distance and a donor—acceptor angle above 90°).
- (c) The secondary structure of each residue over time was monitored using the STRIDE algorithm for secondary structure assignment and the VMD Timeline plugin 61,62 to determine whether the peptide remained in the β -sheet conformation. STRIDE employs bond angles, hydrogen bonds, and other criteria to assign secondary structure.

These results are presented in Table 3, with the most direct visual comparison of secondary structure deformation presented in Figure 5 (and the expanded comparison in the Supporting Information).

Strikingly, we found that, after 12-18 ns of simulation, the very short β -sheet region of the peptide can be deformed even when disulfide bonds are present (as in tachx180). Given that distinct B_2 mode signals from β -sheets are observed experimentally, the observed deformation may be an artifact of side chain reorganizations on the short time scales studied in these simulations. When the disulfide bonds were removed, the β -sheet structure was only retained in window tachx270. In runs tachx0 and tachx235, only a single residue remained transiently in the bridged conformation. Experimentally, no β -sheet signals were observed in the absence of disulfide bonds.

When disulfide bonds were retained, secondary structure was generally retained in these simulations to a much higher degree. Only in one of the trajectories was secondary structure completely lost during ongoing reorganizations of side chain residues as the peptide reoriented from a nonoptimal initial position (tachx180). Partial retention or even recovery of the secondary structure was observed in two of the four trajectories for which the peptide was found to adsorb to the surface (tachx270 and tachx0). The results in Figure 5 are in reasonable agreement with the hydrogen bond and rmsd quantification in Table 3. It is interesting to note that neither metric alone is sufficient to unequivocally indicate peptide folding, however. This is likely due to the fact that the β -sheet portion of tachyplesin I is very short, so small local deformations are possible without loss of secondary structure. This highlights the importance of backbone dynamics in interpreting molecular orientation, as most orientation analysis methods require a

Table 3. Structural Deformation Metrics: Average Number of Hydrogen Bonds and rmsd over the Last 1 ns of Each Simulation Trajectory

	no. of hy	no. of hydrogen bonds		nsd (Å)
name	β sheet	backbone	β sheet	backbone
tachx270	2.6	3.5	0.6	2.5
no disulfide bonds	4.8	5.4	0.5	4.6
tachx0	1.5	1.6	1.1	3.0
no disulfide bonds	1.5	2.1	1.1	5.2
tachx180	2.8	4.0	1.9	2.1
no disulfide bonds	0.6	0.8	3.7	5.6
tachx315	4.1	4.3	0.75	3.2
no disulfide bonds	1.3	0.3	2.8	4.1
tachx235 (no disulfides only)	1.8	2.7	1.1	3.5

known and fixed structure. Even when the β -sheet secondary structure is preserved (as for tachx315), our simulations show that the strands are not arranged in a completely ideal and extended conformation, with interstrand angles of 10° or higher.

Stability of the Final Adsorbed State. The highly deformable peptide backbone of tachyplesin I prevents the use of constrained sampling methods to calculate a free energy profile as a function of peptide orientation. It appears that backbone deformation is important to the initial adsorption process, but addition of an artificial constraint might lead to other less meaningful disruption of the structure rather than restraining peptide orientation. On the basis of the assumption that hydrophobic surface area burial drives a change in free energy upon adsorption, 55-57 solvent-accessible surface area was calculated for all windows in the simulation on the basis of the surface area of the selected residues plus the polystyrene surface minus the surface area of the selected residues in solvent and the surface area of the polystyrene surface in the absence of peptide. 63 The surface area module in CHARMM did not directly provide the surface area of the residue as it contributes to the surface area of the overall protein, and as a result, the values measured are a relative metric that may not add up to 100%.

In all but one simulation window (tachx315), the removal of disulfide bonds leads to greater surface area burial (Table 4), yet since simulation windows that showed high hydrophobic surface area burial were accompanied by higher burial of other residues as well, it is unclear how much of an enthalpic driving force is involved, and calculations of the total potential energy of the system were inconclusive. By contrast, the strong facial segregation of residues in previously studied peptides^{20,22} created a clear driving force for adsorption via burial of hydrophobic surface area. In tachyplesin I, residues are poorly segregated, and side chains

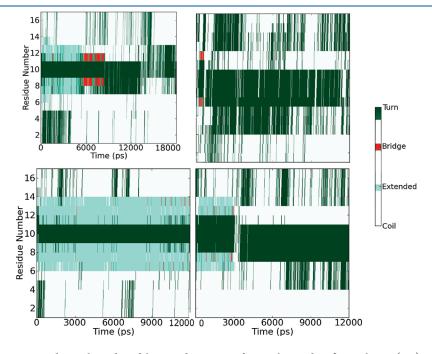


Figure 5. Secondary structure assigned to each residue of the peptide over time for simulation identifiers tachx180 (top) and tachx315 (bottom). The left panels show the structure over time with disulfide bonds present, and the right panels are for the same initial conditions with disulfide bonds removed. The color scheme used is presented as a sidebar.

Table 4. Burial of Solvent-Accessible Surface Area by Residue $Type^a$

chan	change in solvent-accessible surface area (SASA), \mathring{A}^2			
overall	hydrophobic	hydrophilic	charged	
887	639	466	277	
1659	850	1110	894	
1273	800	765	719	
1582	862	852	651	
1295	614	696	632	
1411	931	772	769	
1458	792	786	593	
1147	589	690	670	
1287	304	872	726	
	overall 887 1659 1273 1582 1295 1411 1458 1147	overall hydrophobic 887 639 1659 850 1273 800 1582 862 1295 614 1411 931 1458 792 1147 589	(SASA), Å ² overall hydrophobic hydrophilic 887 639 466 1659 850 1110 1273 800 765 1582 862 852 1295 614 696 1411 931 772 1458 792 786 1147 589 690	

^a Individual tallies do not add up to 100% due to limitations in the algorithm used.

are free to reorganize. This leads to a variety of candidate orientations with trade-offs between favorable and unfavorable peptide—surface interactions. Other factors, such as structural deformation and entropic considerations, likely therefore play a significant role in determining the overall free energy of adsorption.⁶⁴ A single unambiguously most stable twist angle or backbone arrangement could not be determined on the time scales studied by MD simulations, suggesting that both enthalpic and entropic considerations contribute to a diverse free energy landscape.

SFG and ATR-FTIR Spectra. Spectra obtained from tachyplesin I at the PS/water interface were presented previously. As discussed above, SFG data analysis and orientation determination was focused on the B₂ mode peak at 1635 cm⁻¹, since other signals were too weak to ensure reliable ratios for analysis. These spectra were fit as described in a previous paper. For total internal reflection geometry, and taking $n' = (n_1 + n_2)/2$ as the corrected index of refraction at the interface, the normalized Fresnel factors L_{xxz} , L_{yxz} , and L_{yzx} are 10.1, 3.5, and 1.0, respectively.

Combined Spectroscopies for Orientation Determination. Orientation analysis may be performed by considering the ratio of the SFG experimental observables $\chi_{\rm eff}^{(2)}$ in the polarizations of interest. The SFG-measured quantity for the B_2 mode peak centered at 1635 cm $^{-1}$ is

$$\frac{\chi_{\rm ssp}^{(2)}}{\chi_{\rm spp}^{(2)}} = \frac{L_{\rm yyz}}{L_{\rm yxz}} \frac{N_{\rm s} \langle \cos^3 \theta \cos \psi \sin \psi - \cos \theta \cos \psi \sin \psi \rangle \beta_{\rm acb}}{-\frac{1}{2} N_{\rm s} \langle \cos^2 \theta - \sin^2 \theta \cos^2 \psi \rangle \beta_{\rm acb}}$$
(2)

To characterize both of the required orientation angles (tilt and twist), a second measurement was obtained from polarized ATR-FTIR spectroscopy. Orientation relations for β -sheet structures have been published previously for ATR-FTIR ^{40,68} and were employed in a previous study to characterize the orientation of tachyplesin I by considering only the tilt angle (θ) :¹⁷

$$R^{\text{ATR}}(\text{amideI}) = \frac{E_x^2}{E_y^2} + \frac{2\langle \cos^2 \theta \rangle}{3 - \langle \cos^2 \theta \rangle} \frac{E_z^2}{E_y^2}$$
(3)

In the above equation, the twist angle can be neglected due to the assumption that the strands adopt a symmetrical β -barrel

structure in which the twist angle is held fixed. 40,69 For a simple β -sheet structure consisting of two hydrogen-bonded antiparallel strands, the twist angle should be considered. The necessary orientation relations can be derived by considering the known vibrational modes of a β -sheet in the extended conformation. In general, the dichroic ratio from s- and p-polarized IR absorption is given 40 by

$$R^{\text{ATR}} = \frac{A_{\parallel}}{A_{\perp}} = \frac{E_x^2}{E_y^2} + \frac{\langle M_z^2 \rangle}{\langle M_y^2 \rangle} \frac{E_z^2}{E_y^2}$$
(4)

The angular brackets indicate orientational averaging, and M_x , M_y , and M_z refer to the respective components of the dipole moment in the laboratory coordinate system with axes x, y, and z. These dipole moment components can be calculated on the basis of the known axial orientations of the three IR-active antiparallel β -sheet vibrational modes: B_1 , B_2 , and B_3 , which are known to orient along the molecular c, b, and a axes of the β -sheet, respectively. A correction for the orientations of individual transition moments of the individual peptide units 37 was, therefore, found to be unnecessary in our analysis.

If one assumes that the entire dipole moment of the molecule lies along the c axis of its coordinate system initially, and is projected by an angle Θ into the sheet coordinate frame, then the expected dipole moment components for the β -sheet in the laboratory frame required by eq 4 can be derived. As shown in Figure 2, the molecular coordinate system (a,b,c) for the peptide is defined so that the plane of the sheet corresponds to the bc plane; this coordinate system was chosen so as to be consistent with previously published relations for SFG analysis. ¹⁷

$$\mu_{\text{lab},B_{1}} = R_{z}(\phi) R_{y}(\theta) R_{z}(\psi) \begin{bmatrix} 0 \\ 0 \\ |\mu| \end{bmatrix}$$

$$\mu_{\text{lab},B_{2}} = R_{z}(\phi) R_{y}(\theta) R_{z}(\psi) R_{x}(\Theta) \begin{bmatrix} 0 \\ 0 \\ |\mu| \end{bmatrix}$$

$$\mu_{\text{lab},B_{3}} = R_{z}(\phi) R_{y}(\theta) R_{z}(\psi) R_{y}(\Theta) \begin{bmatrix} 0 \\ 0 \\ |\mu| \end{bmatrix}$$

$$(5)$$

where the rotation matrices about the intrinsic molecular axes for an arbitrary rotation angle α are of the form

$$R_{x}(\alpha) = \begin{bmatrix} 1 & 0 & 0 \\ 0 & \cos \alpha & \sin \alpha \\ 0 & -\sin \alpha & \cos \alpha \end{bmatrix}; \quad R_{y}(\alpha) = \begin{bmatrix} \cos \alpha & 0 & -\sin \alpha \\ 0 & 1 & 0 \\ \sin \alpha & 0 & \cos \alpha \end{bmatrix}$$
$$R_{z}(\alpha) = \begin{bmatrix} \cos \alpha & \sin \alpha & 0 \\ -\sin \alpha & \cos \alpha & 0 \\ 0 & 0 & 1 \end{bmatrix}$$
(6)

The transition moment orientation $\Theta = 90^{\circ}$ corresponds to the amide I band, and averaging is performed over the azimuthal angle ϕ to reflect that individual molecules are randomly oriented in the plane of the surface. Then from eqs 4–6,

orientation relations for the B₁, B₂, and B₃ amide I modes may be derived:

$$R^{\text{ATR}}(B_1) = \frac{E_x^2}{E_y^2} + 2\langle \cot^2 \theta \rangle \frac{E_z^2}{E_y^2}$$

$$R^{\text{ATR}}(B_2) = \frac{E_x^2}{E_y^2} + \frac{-2\langle \sin^2 \psi \rangle \langle \sin^2 \theta \rangle}{\langle \sin^2 \psi \rangle \langle \sin^2 \theta \rangle - 1} \frac{E_z^2}{E_y^2}$$

$$R^{\text{ATR}}(B_3) = \frac{E_x^2}{E_y^2} + \frac{2\langle \cos^2 \psi \rangle \langle \sin^2 \theta \rangle}{1 + \langle \sin^2 \psi \rangle \langle \sin^2 \theta \rangle - \langle \sin^2 \theta \rangle} \frac{E_z^2}{E_y^2}$$

$$(7$$

Using these equations, it is possible to extract the full range of orientation information from the combination of SFG and ATR-FTIR measurements. Expressions for the electric field amplitudes E_x^2 , E_y^2 , and E_z^2 have been derived previously, and depend on the indices of refraction for the substrate, interfacial layer (in this case, the PS film), and solvent. These indices are taken to be 4.00, 1.55, and 1.33, respectively. For the polymer interface, this leads to values for E_x^2 , E_y^2 , and E_z^2 of 1.97, 2.25, and 1.36, respectively.

Only signals from the strong B_2 mode peak of $\sim 1635~{\rm cm}^{-1}$ were observed. Hence, the possible orientations of the β -sheet structure may be determined from the available experimental measurements by considering all orientations that provide a match for the following experimentally measured quantities:

$$\begin{cases} \frac{\chi_{\text{ssp}}^{(2)}}{\chi_{\text{spp}}^{(2)}} = \frac{L_{\text{yyz}}}{L_{\text{yzx}}} \frac{N_{\text{s}} \langle \cos^3 \theta \cos \psi \sin \psi - \cos \theta \cos \psi \sin \psi \rangle \beta_{\text{acb}}}{-\frac{1}{2} N_{\text{s}} \langle \cos^2 \theta - \sin^2 \theta \cos^2 \psi \rangle \beta_{\text{acb}}} = 1.04 \\ R^{\text{ATR}}(B_2) = \frac{E_x^2}{E_y^2} + \frac{-2 \langle \sin^2 \psi \rangle \langle \sin^2 \theta \rangle}{\langle \sin^2 \psi \rangle \langle \sin^2 \theta \rangle - 1} \frac{E_z^2}{E_y^2} = 0.95 \end{cases}$$

$$(8)$$

If one assumes that all molecules of peptide on the surface adopt exactly the same orientation, then the angles θ and ψ can be solved for directly from the above as a system of nonlinear equations. However, this may result in multiple unique solutions, and the relationship between results is not always clear—particularly when large experimental error bars are involved. To better illustrate the effect of possible larger uncertainties in the experiments and data analysis (due to factors such as uncertainty in the Fresnel coefficients and interfacial electric field amplitudes), the scoring function and graphical presentation developed previously ⁷⁴ (see the Supporting Information) was applied to illustrate all matches within $\pm 40\%$ of the target criteria (Figure 6).

Upon combining the measurements from SFG and ATR-FTIR, it may be seen that two broad arcs of high-scoring matches for experimental measurements exist. One region of high-scoring matches occurs at low tilt angles, which would correspond to the peptide roughly standing up. However, such low tilt angles would provide extremely poor contact between the peptide and the surface, and could only be adopted if either the charged residues of the disordered tail or the charged arginine at the β -turn (residue 9) were unfavorably buried on the surface while hydrophobic contacts with the surface were kept minimal. In our simulations, we explored initial conditions where contact between the peptide and surface was initially poor. In these cases, the peptide either quickly collapsed down to lie flat on the surface, failed to adsorb (tachx235), or (in the case of tachx180) adsorbed, but only upon significant distortion of the peptide backbone as dramatic side chain reorganizations took place. The

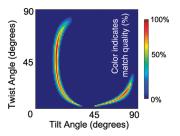


Figure 6. Final orientations of tachyplesin I that best satisfy all three experimental constraints: a match for the SFG ssp/spp and ATR-FTIR dichroic ratios ($\pm 40\%$) and B_2 signal intensity greater than the B_1 signal intensity. Within the allowed error bars, a color scheme is applied to indicate overall match quality based on closeness of the calculated and experimental values.

second region of high-scoring matches is more physically reasonable, as the highest scores are assigned to positions where the plane of the sheet lies nearly parallel to the surface ($\theta=80^\circ,\psi=10^\circ$). Such a final position is consistent with at least some of the simulation windows considered. Although not all simulation windows converge to this position, it is possible that the lack of convergence to a final sheet orientation parallel to the surface in the time scales studied is due to the need for side chains to reorganize prior to the peptide lying flat on the surface. This is supported by the structural deformations observed in some windows, as well as the observation that the process of peptide adsorption could occur differently when disulfide bonds are preserved or removed. Thus, some conclusions may be hampered by limitations in simulation length.

Both simulations and experiments support the observation that the long axis of the sheet lies parallel to the surface in a narrow range of tilt angles ($\theta \approx 90^{\circ}$). If SFG experiments are interpreted by assuming only a δ distribution in tilt and twist angles, then it would appear that the plane of the peptide also lies flat on the surface ($\psi \approx 10^{\circ}$). Given the orientation of the B₂ mode transition moment and the assumption that individual molecules are randomly oriented in the plane of the surface, such a twist angle might lead to very weak, or even undetectable, SFG signals, which would be at odds with the experimental observation of B₂ mode signals originating from the peptide in a β -sheet conformation. However, our MD simulation results do point to the feasibility of adopting a range of twist angles (such as the orientations shown in Figure 4), since driving factors for adsorption—such as surface area burial appear to be similarly satisfied for windows with other small (45°) twist angles (Table 4). Due to the limited number of experimental measurements, a distribution of angles could not be directly characterized, but we believe that advances in computer simulation performance and new experimental techniques will make such characterization possible in the future.

■ CONCLUSIONS

New experimental techniques provide a powerful means to probe the orientation of peptides at interfaces, but often require key assumptions. In this paper, we explored several of those assumptions with regard to the adsorption behavior of the small β -sheet peptide tachyplesin I. The orientation and secondary structure change was characterized using a variety of metrics. General agreement was found between simulations and experiments with regard to the role of disulfide bonds in preserving the secondary structure, but simulations revealed a flexible interfacial

structure that may affect data analysis methodologies which depend on the assumption of a linear and extended β -sheet conformation (or of any single rigid conformation). For the short β -sheet region in tachyplesin I, our results suggest that ignoring structural deformation is a reasonable approximation, but our simulations highlight the need to consider dynamic behavior as well as the well-known twisting of β -strands for future studies of larger β -sheet proteins.

Our findings from molecular dynamics were used to inform the interpretation of results from SFG and ATR-FTIR by deriving orientation relations for ATR-FTIR in a consistent rotation convention with SFG. The result was a more complete picture of molecular orientation and adsorption behavior. It was found that the long axis of the sheet will orient parallel to the plane of the surface to maximize favorable interactions, and that further rearrangements of the backbone will occur upon removal of the disulfide bonds. Our simulation results suggest that the peptide adopts a range of twist angles that would provide equally effective peptide-surface interactions, which would better explain the observed SFG signal intensities despite the lack of sufficient measurements to characterize a distribution of tilt angles directly from experiments. Lastly, time-dependent trends in adsorption seen in our simulation suggest the importance of side chain reorganization in determining the final orientation and adsorption time scale.

We believe that our simulation results, combined with the improved combination of ATR-FTIR and SFG orientation relations, thus provide a more complete picture of adsorbed orientation behavior, as well as representing an important step toward future vibrational spectroscopic studies of more complex β -sheet-containing proteins.

ASSOCIATED CONTENT

Supporting Information. Details of SFG orientation analysis, expanded version of Figure 5, and procedure for creating the graphical display of matches for the experimental results shown in Figure 6. This material is available free of charge via the Internet at http://pubs.acs.org.

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■ REFERENCES

- (1) Gray, J. J. Curr. Opin. Struct. Biol. 2004, 14, 110–115.
- (2) Yebra, D. M.; Kiil, S.; Dam-Johansen, K. Prog. Org. Coat. 2004, 50, 75–104
- (3) Lee, J. M.; Park, H. K.; Jung, Y.; Kim, J. K.; Jung, S. O.; Chung, B. H. *Anal. Chem.* **2007**, *79*, 2680–2687.

- (4) Holland-Nell, K.; Beck-Sickinger, A. G. ChemBioChem 2007, 8, 1071–1076.
 - (5) Tamm, L. K.; Tatulian, S. A. Q. Rev. Biophys. 1997, 30, 365-429.
 - (6) Barth, A.; Zscherp, C. Q. Rev. Biophys. 2002, 35, 369-430.
- (7) Laederach, A.; Andreotti, A. H.; Fulton, D. B. Biochemistry 2002, 41, 12359–12368.
 - (8) Whitmore, L.; Wallace, B. A. Biopolymers 2008, 89, 392-400.
- (9) Wang, J.; Buck, S. M.; Even, M. A.; Chen, Z. J. Am. Chem. Soc. **2002**, 124, 13302–13305.
- (10) Wang, J.; Clarke, M. L.; Zhang, Y.; Chen, X.; Chen, Z. Langmuir **2003**, *19*, 7862–7866.
- (11) Wang, J.; Even, M. A.; Chen, X.; Schmaier, A. H.; Waite, J. H.; Chen, Z. J. Am. Chem. Soc. 2003, 125, 9914–9915.
- (12) Chen, X.; Wang, J.; Sniadecki, J. J.; Even, M. A.; Chen, Z. Langmuir 2005, 21, 2662–2664.
- (13) Wang, J.; Clarke, M. L.; Chen, X.; Even, M. A.; Johnson, W. C.; Chen, Z. Surf. Sci. **2005**, 587, 1–11.
- (14) Wang, J.; Paszti, Z.; Clarke, M. L.; Chen, X.; Chen, Z. J. Phys. Chem. B 2007, 111, 6088–6095.
- (15) Wang, J.; Lee, S.-H.; Chen, Z. J. Phys. Chem. B 2008, 112, 2281–2290.
- (16) Nguyen, K. T.; Le Clair, S. V.; Ye, S.; Chen, Z. J. Phys. Chem B **2009**, 113, 12169–12180.
- (17) Nguyen, K. T.; King, J. T.; Chen, Z. J. Phys. Chem. B 2010, 114, 8291–8300.
- (18) Boughton, A. P.; Andricioaei, I.; Chen, Z. *Langmuir* **2010**, *26*, 16031–16036.
- (19) Chen, Z.; Ward, R.; Tian, Y.; Malizia, F.; Gracias, D. H.; Shen, Y. R.; Somorjai, G. A. *J. Biomed. Mater. Res.* **2002**, *62*, 254–264.
- (20) Mermut, O.; Phillips, D. C.; York, R. L.; McCrea, K. R.; Ward, R. S.; Somorjai, G. A. J. Am. Chem. Soc. **2006**, 128, 3598–3607.
- (21) Phillips, D. C.; York, R. L.; Mermut, O.; McCrea, K. R.; Ward, R. S.; Somorjai, G. A. *J. Phys. Chem. C* **2007**, *111*, 255–261.
- (22) York, R. L.; Browne, W. K.; Geissler, P. L.; Somorjai, G. A. Isr. J. Chem. 2007, 47, 51–58.
- (23) Weidner, T.; Apte, J. S.; Gamble, L. J.; Castner, D. G. *Langmuir* **2010**, *26*, 3433–3440.
- (24) Baugh, L.; Weidner, T.; Baio, J. E.; Nguyen, P.-C.; Gamble, L. J.; Stayton, P. S.; Castner, D. G. *Langmuir* **2010**, *26*, 16434–16441.
- (25) Fu, L.; Ma, G.; Yan, E. C. J. Am. Chem. Soc. 2010, 132, 5405–5412.
 - (26) Anglin, T. C.; Liu, J.; Conboy, J. C. Biophys. J. 2007, 92, L01-3.
- (27) Anglin, T. C.; Brown, K. L.; Conboy, J. C. J. Struct. Biol. 2009, 168, 37–52.
- (28) Jung, S. Y.; Lim, S. M.; Albertorio, F.; Kim, G.; Gurau, M. C.; Yang, R. D.; Holden, M. A.; Cremer, P. S. *J. Am. Chem. Soc.* **2003**, *125*, 12782–12786.
- (29) Chen, X.; Sagle, L. B.; Cremer, P. S. J. Am. Chem. Soc. 2007, 129, 15104–15105.
- (30) Hall, S. A.; Jena, K. C.; Trudeau, T. G.; Hore, D. K. J. Phys. Chem. C 2011, 110516065918031.
- (31) Wang, J.; Paszti, Z.; Even, M. A.; Chen, Z. J. Phys. Chem. B 2004, 108, 3625–3632.
- (32) Ye, S.; Nguyen, K. T.; Chen, Z. J. Phys. Chem. B 2010, 114, 3334–3340.
- (33) Pauling, L.; Corey, R. B. Proc. Natl. Acad. Sci. U.S.A. 1951, 37, 251–256.
- (34) Pauling, L.; Corey, R. B. Proc. Natl. Acad. Sci. U.S.A. 1951, 37, 729–740.
- (35) Pauling, L.; Corey, R. B. Proc. Natl. Acad. Sci. U.S.A. 1953, 39, 247–252.
- (36) Wang, J.; Chen, X.; Clarke, M. L.; Chen, Z. Proc. Natl. Acad. Sci. U.S.A. 2005, 102, 4978–4983.
- (37) Oishi, O.; Yamashita, S.; Nishimoto, E.; Lee, S.; Sugihara, G.; Ohno, M. *Biochemistry* **1997**, *36*, 4352–4359.
- (38) Matsuzaki, K.; Shioyama, T.; Okamura, E.; Umemura, J.; Takenaka, T.; Takaishi, Y.; Fujita, T.; Miyajima, K. *Biochim. Biophys. Acta* **1991**, *1070*, 419–428.

(39) Rodionova, N. A.; Tatulian, S. A.; Surrey, T.; Jaehnig, F.; Tamm, L. K. *Biochemistry* **1995**, *34*, 1921–1929.

- (40) Marsh, D. Biophys. J. 1997, 72, 2710-2718.
- (41) Páli, T.; Marsh, D. Biophys. J. 2001, 80, 2789-2797.
- (42) Chothia, C. J. Mol. Biol. 1973, 75, 295-302.
- (43) Salemme, F. J. Mol. Biol. 1981, 146, 143-156.
- (44) Salemme, F. R. Prog. Biophys. Mol. Biol. 1983, 42, 95-133.
- (45) Nesloney, C. Bioorg. Med. Chem. 1996, 4, 739-766.
- (46) Wang, L.; O'Connell, T.; Tropsha, A.; Hermans, J. J. Mol. Biol. 1996, 262, 283–293.
- (47) Collier, G.; Vellore, N. A.; Latour, R. A.; Stuart, S. J. Biointer-phases 2009, 4, 57–64.
- (48) Sun, Y.; Dominy, B. N.; Latour, R. A. J. Comput. Chem. 2007, 28, 1883–1892.
- (49) Sun, Y.; Welsh, W. J.; Latour, R. A. Langmuir 2005, 21, 5616-5626.
- (50) Vellore, N. A.; Yancey, J. A.; Collier, G.; Latour, R. A.; Stuart, S. J. Langmuir **2010**, 26, 7396–7404.
- (51) Wang, Q.; Zhao, C.; Zhao, J.; Wang, J.; Yang, J. C.; Yu, X.; Zheng, J. *Langmuir* **2010**, *26*, 3308–3316.
- (52) He, Y.; Hower, J.; Chen, S.; Bernards, M. T.; Chang, Y.; Jiang, S. Langmuir 2008, 24, 10358–10364.
- (53) Kandasamy, S. K.; Larson, R. G. Chem. Phys. Lipids 2004, 132, 113-132.
 - (54) Lee, H.; Larson, R. G. Molecules 2009, 14, 423-438.
 - (55) Wei, Y.; Latour, R. A. Langmuir 2009, 25, 5637-5646.
- (56) Vallone, B.; Miele, A. E.; Vecchini, P.; Chiancone, E.; Brunori, M. Proc. Natl. Acad. Sci. U.S.A. 1998, 95, 6103–6107.
- (57) Chipot, C.; Maigret, B.; Pohorille, A. Proteins: Struct., Funct., Bioinf. 1999, 36, 383–399.
- (S8) Brooks, B.; Bruccoleri, R.; Olafson, B.; States, D.; Swaminathan, S.; Karplus, M. J. Comput. Chem. 1983, 4, 187–217.
- (59) MacKerell, A. D., Jr.; Bashford, D.; Bellott, M.; Dunbrack, R. L., Jr.; Evanseck, J. D.; Field, M. J.; Fischer, S.; Gao, J.; Guo, H.; Ha, S.; Joseph-McCarthy, D.; Kuchnir, L.; Kuczera, K.; Lau, F. T. K.; Mattos, C.; Michnick, S.; Ngo, T.; Nguyen, D. T.; Prodhom, B.; Reiher, W. E., III; Roux, B.; Schlenkrich, M.; Smith, J. C.; Stote, R.; Straub, J.; Watanabe, M.; Wiórkiewicz-Kuczera, J.; Yin, D.; Karplus, M. J. Phys. Chem. B 1998, 102, 3586–3616.
- (60) Matsuzaki, K.; Nakayama, M.; Fukui, M.; Otaka, A.; Funakoshi, S.; Fujii, N.; Bessho, K.; Miyajima, K. *Biochemistry* 1993, 32, 11704–11710.
 - (61) Humphrey, W. J. Mol. Graphics 1996, 14, 33-38.
 - (62) Frishman, D.; Argos, P. Proteins 1995, 23, 566-579.
 - (63) Lee, B.; Richards, F. M. J. Mol. Biol. 1971, 55, 379-400.
 - (64) Yang, A. S.; Honig, B. J. Mol. Biol. 1995, 252, 366-376.
- (65) Lambert, A.; Davies, P.; Neivandt, D. Appl. Spectrosc. Rev. 2005, 40, 103–145.
- (66) Zhuang, X.; Miranda, P. B.; Kim, D.; Shen, Y. R. Phys. Rev. B 1999, 59, 12632–12640.
 - (67) Hecht, E. Optics, 4th ed.; Addison Wesley: Boston, MA, 2001.
- (68) Zbinden, R. Infrared Spectroscopy of High Polymers; Academic Press: New York, 1964.
- (69) Ramakrishnan, M.; Qu, J.; Pocanschi, C. L.; Kleinschmidt, J. H.; Marsh, D. *Biochemistry* **2005**, 44, 3515–3523.
 - (70) Miyazawa, T. J. Chem. Phys. 1960, 32, 1647-1652.
- (71) Hu, X.; Shin, K.; Rafailovich, M.; Sokolov, J.; Stein, R.; Chan, Y.; Wlwu, K. W.; Wu, W. L.; Kolb, R. *High Perform. Polym.* **2000**, 12, 621–629.
 - (72) Malitson, I. H. Appl.Opt. 1963, 2, 1103–1107.
- (73) CRC Handbook of Chemistry and Physics, 91st ed.; CRC Press: New York, 2010.
- (74) Boughton, A. P.; Yang, P; Tesmer, V; Ding, B; Tesmer, J. J. G.; Chen, Z. Proc. Natl. Acad. Sci. U.S.A. 2011, 108, E667–E673.