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Highly Enantioselective Conjugate Additions of Aldehydes to Vinyl Sulfones

Aitor Landa,^[a] Miguel Maestro,^[b] Carme Masdeu,^[a] Ángel Puente,^[a] Silvia Vera,^[a]
Mikel Oiarbide,^[a] and Claudio Palomo*^[a]

Given the vast chemistry of the aldehyde and the sulfone groups,^[1] a combination of both functionalities through stereoselective C–C bond-forming reactions is highly appealing. Towards this goal, the 1,4-addition of aldehydes to unsaturated sulfones is a highly valuable tool. Unfortunately, catalytic and enantioselective variations of the reaction are seemingly elusive.^[2] First reports of catalytic enantioselective conjugate additions of aldehydes to vinyl sulfones involve secondary/tertiary 1,2-diamine organocatalysts **1** and **2**,^[3] which activate the aldehyde component through enamine formation.^[4,5] The behaviour of these catalysts is significant, but the method holds important limitations with regards to substrate scope and enantioselectivity: a large excess (10 equivalents) of the aldehyde is usually required and modest selectivities (typically 70% *ee*) are obtained. On the other hand, while the direct catalytic and asymmetric 1,4-addition of aldehydes to certain electron-deficient olefins,^[6,7] most particularly nitroolefins,^[7,8] have met success recently, one accompanying problem to the reaction with vinyl sulfone acceptors is the retroaddition, which causes formation of undesired dimeric side products.^[3] Herein we report the use of less basic chiral amine catalysts (see Figure 1) to provide a solution to these problems thus considerably expanding the utility of vinyl sulfones in organic synthesis.

In an initial screen, commercially available prolinol silyl ethers **3** and **4**,^[9] and analogue **5**, recently disclosed from our

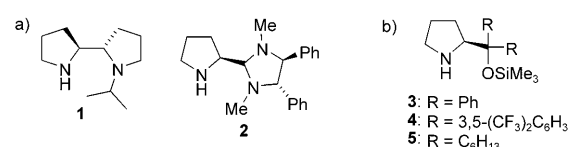


Figure 1. Amine catalysts explored for the conjugate addition of aldehydes to vinyl sulfones: a) Previous studies; b) this work.

laboratory,^[10] were tested for the reaction between aldehydes **6** and vinyl bis(sulfone) **7**. At the outset, it was not clear whether each catalyst would be equally effective. Besides the above problems, there is the fact that diaryl prolinol ethers may exhibit certain substrate specificity.^[11]

To our delight, by using catalyst **3**, products **10** were obtained (Table 1), after reduction of the 1,4-addition adduct, in yields typically over 90% and enantioselectivities greater than 95% for both linear as well as β -branched aldehydes. For example, propionaldehyde, which provides racemic adducts with catalyst **1**, affords **10a** with 95% *ee*. Similarly,

Table 1. Addition of aldehydes **6** to vinyl sulfones promoted by **3**.

Aldehyde 6 , R	Sulfone	Product	Yield [%] ^[a]	syn/anti ^[b]	ee [%] ^[c]
a , CH ₃	7	10a	95	–	95
	8	11a	60	85:15	99(99) ^[d]
b , CH ₃ CH ₂	7	10b	93	–	98
	9	12b	49	70:30	99(n.d.)
c , CH ₃ (CH ₂) ₂	7	10c	94	–	97
d , CH ₃ (CH ₂) ₃	8	11d	52	70:30	99(99) ^[d]
	9	12d	51	70:30	99(n.d.)
e , CH ₃ (CH ₂) ₄	7	10e	95	–	96
f , PhCH ₂	7	10f	92	–	95
g , (CH ₃) ₂ CH	7	10g	91	–	99

[a] Isolated yield of product alcohol after column chromatography (hexanes/EtOAc 60:40). [b] Determined by ¹H/¹³C NMR and chiral HPLC. [c] Determined by chiral HPLC. [d] *ee* of the minor diastereomer. n.d.: not determined.

[a] Dr. A. Landa, Dr. C. Masdeu, Á. Puente, S. Vera,
Prof. Dr. M. Oiarbide, Prof. Dr. C. Palomo
Departamento de Química Orgánica I, Facultad de Química
Universidad del País Vasco, Apdo. 1072
20080 San Sebastián (Spain)
Fax: (+34) 943-015-270
E-mail: claudio.palomo@ehu.es

[b] Dr. M. Maestro
(X-ray analyses)
Departamento de Química Fundamental, Facultad de Ciencias, Universidade da Coruña, Campus Zapateira s/n, 15071A Coruña (Spain)

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valeraldehyde affords **10c** with 97% *ee*, whilst only 54% *ee* is obtained with catalyst **1** (74% *ee* with **2**). Among the solvents methylene chloride and toluene gave the best results. No reaction was observed in protic solvents such as ethanol, methanol or isopropyl alcohol, whilst in DMF—a typical solvent for other enamine-based reactions—low *ee* values are obtained. Catalyst **4** behaved similarly, but **5**, which performs very well in other 1,4-addition reactions,^[10] gave rise to 60–75% *ee* at best.^[12] Formation of dimeric side product was determined to be below the limit of detection of ¹H NMR spectroscopy. On the other hand, β -substituted sulfones **8** and **9** were also good electrophiles, giving rise adducts **11** and **12** with good yields, diastereomeric ratios in the range from 70:30 to 80:20, and essentially complete enantioselectivity.^[13]

Table 2. Aldehyde addition to *E*- α -ethoxycarbonyl vinyl sulfones promoted by **3/4**.^[a]

13: R¹ = Ph
14: R¹ = 4-MeC₆H₄
15: R¹ = 2-naphthyl

Product 16		Cat.	<i>t</i> [h]	Yield [%] ^[b]	dr ^[c]	<i>ee</i> [%] ^[d]
R	R ¹					
a Me	Ph	3	3	85	80:20	99
		4	3	54	80:20	98
b Me	2-naphthyl	3	3	75	75:25	97
c Pr	4-MeC ₆ H ₄	3	20	60	75:25	99
		4	20	30	70:30	97
d Pn	Ph	3	16	78	75:25	98
		4	16	35	75:25	n.d.

[a] Reactions conducted at 0.5 mmol scale in CH₂Cl₂ at RT overnight. Aldehyde/vinyl sulfone 3:1 ratio. [b] Isolated yield of lactone product after column chromatography (Hex/EtOAc 90:10). [c] Determined by ¹³C NMR; relative and absolute configuration of the minor diastereomer not determined. [d] Determined by chiral HPLC. n.d.: not determined.

The present catalytic reaction can also be applied to other vinyl sulfones with maintained levels of stereoselectivity. For example, as shown in Table 2, α -ethoxycarbonyl vinyl sulfones **13**, **14**, and **15** reacted with aldehydes in the presence of catalysts **3/4** (10 mol%) to give, after reduction and cyclisation, the corresponding lactone adducts **16** with three contiguous stereocenters. Nearly perfect enantiocontrol is observed in most cases for both catalysts, but catalyst **3** provided better yields (up to 85% yield over three steps) than **4**. It is partic-

ularly noteworthy the fact that the sulfone group is key for the 1,4-addition reaction to proceed satisfactorily. For instance, whilst vinyl sulfones **13a** and **13c**, upon reaction with propanal and pentanal, respectively, afforded products **16a** and **16c** in 85 and 60% yield, the respective arylidene ethyl malonates, which would generate analogous δ -lactones, were unreactive regardless of the reaction conditions employed.^[14] In this respect, δ -lactones are common structural units of natural products as well as versatile building blocks of several classes of biologically active compounds.^[15]

A third variation concerns α,β -unsaturated nitriles, a recalcitrant class of Michael acceptors^[16] that remain completely unreactive under the present catalytic conditions. However, addition of aldehydes to α -cyano vinyl sulfones **17–20** took place smoothly at –40 °C (Table 3) to give, after reduction of the resulting adduct, alcohols **21**. The latter compounds are versatile intermediates which allow, for example, access to lactones **22** or cyano alcohols **23** in good yields, diastereomeric ratio, and again essentially perfect enantioselectivity. This approach constitutes a new enantioselective entry to building blocks otherwise difficult to access from unsaturated nitriles.^[16] In the above reactions a threefold excess of aldehyde substrate is employed regularly, but nearly equimolar amounts suffice for achieving equal efficiency. Configuration of the products was established by X-ray analyses of *syn*-**12b**, **16a**, and *trans*-**22b**,^[17] and by assuming a uniform reaction mechanism.^[18]

The potential of this catalytic methodology can further be shown by the reductive double desulfonation of **10f** and *syn*-**11d** to afford alcohols **24** and **25** (Scheme 1). This achievement represents a formal catalytic enantioselective α -alkylation of aldehydes with secondary alkyl halides, a process that still bears a considerable challenge.^[19] When the desulfonation step is preceded with a prior alkylation reaction under standard conditions, as in the transformation of **10f** into alcohols **26**, the above approach serves to access to

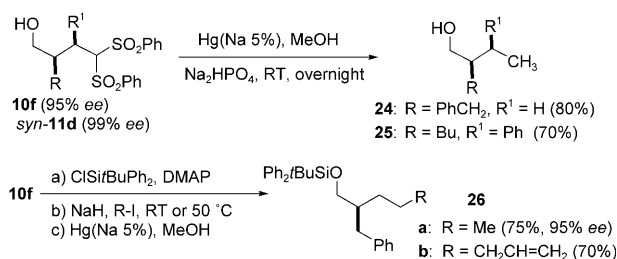
Table 3. Addition to *E*- α -cyano vinyl sulfones promoted by **3**.^[a]

17: R¹ = Ph
 18: R¹ = 4-ClC₆H₄
 19: R¹ = 4-MeC₆H₄
 20: R¹ = 2-naphthyl

i) **3** (20 mol %), CH₂Cl₂, -40 °C
 ii) NaBH₄, EtOH, -40 °C, 30 min

Compound		Product 22			Product 23			
R	R ¹	Yield ^[b] [%]	dr ^[c] <i>trans/cis</i>	<i>ee</i> ^[d] [%]	Yield ^[b] [%]	dr ^[c] <i>syn/anti</i>	<i>ee</i> ^[d] [%]	
a	Me	Ph	57	75:25	99(99)	70	75:25	99(98)
b		4-ClC ₆ H ₄	59	80:20	99(99)	—	—	—
c		4-MeC ₆ H ₄	65	80:20	99(n.d.)	70	75:25	99(97)
d		2-naphthyl	65	70:30	99(99)	52 ^[e]	80:20	99(90)
e	Et	Ph	65	70:30	99(99)	65	60:40	99(99)
f	Pr	Ph	72	70:30	99(99)	70	85:15	99(n.d.)
g	Bu	Ph	70	75:25	99(99)	—	—	—

[a] Ratio of aldehyde/vinyl sulfone 3:1. [b] Combined yield of diastereomers over the three steps. [c] Determined by ¹H NMR and/or HPLC. [d] Determined by HPLC. *ee* values in brackets refer to the minor diastereomer *cis*-**22** and *anti*-**23**, respectively. n.d.: not determined.



Scheme 1. Elaboration of adducts via desulfonation/alkylation standard protocols.

longer chain alkylated products in a practical way. These examples demonstrate how the present organocatalytic conjugate addition to vinyl bis(sulfone)s, in conjunction with a subsequent alkylation step, opens up new opportunities for the asymmetric synthesis of α -branched aldehydes and products thereof.

In summary, although further optimisation is needed to improve reaction diastereocontrol,^[20] the present catalytic system introduces an operationally simple protocol for accessing a variety of structural motifs from readily available vinyl sulfones and unmodified aldehydes as starting materials. The trick of success is the use of a diarylprolinol silyl ether as reaction promoter that leads to the highest enantioselectivity reported to date for this class of Michael acceptors. Further investigations on the utility of gem-disulfone adducts as alkyl carbanion surrogates are currently underway in our laboratory and will be reported in due course.

Experimental Section

Catalytic conjugate addition of aldehydes to vinyl sulfones: To a mixture of the catalyst **3** or **4** (10–20 mol %) and vinyl sulfone (1 mmol) in toluene (1 mL) (for sulfones **7–9**) or CH₂Cl₂ (1 mL) (for **13–15** and **17–20**) at –40 °C was added aldehyde **6** (1.5–3.0 equiv), and the mixture stirred overnight (16 h) at the same temperature. The resulting solution was diluted with EtOH (1 mL) and a suspension of NaBH₄ (2 equiv) in EtOH (2 mL) was then added drop-wise. The mixture was stirred at –40 °C for 30 min and quenched with H₂O (10 mL). The aqueous phase was extracted with CH₂Cl₂ (3 × 10 mL). The combined organic layers were dried over MgSO₄, filtered, concentrated under reduced pressure and the crude product purified by flash column chromatography (hexane/EtOAc 60:40).

Acknowledgements

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Keywords: alkylation • asymmetric catalysis • Michael addition • organocatalysis • vinyl sulfones

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