206. (25 S)-1α, 25, 26-Trihydroxycholecalciferol, a New Vitamin D₃ Metabolite: Synthesis and Absolute Stereochemistry at C (25)

Preliminary Communication

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Summary

The $1\alpha,25,26$ -trihydroxy metabolite of vitamin D_3 , isolated from bovine serum, was shown to possess the $(25\,S)$ -configuration by HPLC. comparison of the 1,3,26-triacetate derivative with authentic $(25\,R)$ - and $(25\,S)$ -samples. The convergent synthesis of $(25\,R)$ - $1\alpha,25,26$ - and $(25\,S)$ - $1\alpha,25,26$ -trihydroxycholecalciferols (10a) and (10b) has been accomplished.

As part of a synthetic program on the vitamin D_3 metabolites, we recently prepared $(25\,R)$ -1a, 25, 26- and $(25\,S)$ -1a, 25, 26-trihydroxycholecalciferols, anticipating that one of these substances could be an undiscovered metabolite [1]. Since the absolute configuration of $(25\,S)$ -25, 26-dihydroxycholecalciferol has been conclusively determined [2] [3], 1a-hydroxylation would afford $(25\,S)$ -1a, 25, 26-trihydroxycholecalciferol. However, the 26-hydroxylating enzyme is not well characterized [1]. Thus, hydroxylation of 1a, 25-dihydroxycholecalciferol could in principle give either $(25\,R)$ - or $(25\,S)$ -1a, 25, 26-trihydroxycholecalciferol. In 1981, Horst et al. isolated a new metabolite of vitamin D_3 from the plasma of vitamin D_3 -treated cows [4]. We have compared the cow metabolite with pure $(25\,R)$ -1a, 25, 26- and $(25\,S)$ -1a, 25, 26-trihydroxycholecalciferols and have shown, by HPLC. co-migration of the triacetate derivatives, that the natural metabolite possesses the $(25\,S)$ -configuration.

The synthesis of the key intermediates, $(25\,R)$ -1a, 25, 26-and $(25\,S)$ -1a, 25, 26-trihydroxycholesterols utilizes the alkylation-reduction method of *Wicha & Bal* [5] [6], which we have also employed in the synthesis of $(25\,R)$ -25, 26- and $(25\,S)$ -25, 26-dihydroxycholecalciferols [2]. 1a, 3β -Dihydroxy-androst-5-en-17-one (1a-

hydroxy-5-dehydro-3-epiandrosterone) (1) was converted to the bis (tetrahydro-pyranyloxy) compound 2 in over 90% yield with 3,4-dihydro-2 H-pyran and p-toluene-sulfonic acid catalyst. This substance was treated with triethyl phosphonoacetate and sodium ethoxide in ethanol to afford cleanly the (E)-unsaturated ester 3. Catalytic hydrogenation with platinum oxide in ethanol then afforded ester 4 in 80% overall yield from 2. This substance was alkylated (lithium diisopropylamide, THF/HMPTA (hexamethylphosphorotriamide), -40°) [5] [6] with the iodide 5a of established stereochemistry and optical purity [2] to yield the monoalkylated ester 6a in 85% yield. The ester 6a was sequentially reduced (LiAlH₄, THF, 60°), esterified (TsCl, pyridine, 0°), and hydrogenolyzed (LiAlH₄, THF, 60°) to oily acetonide 7a. Exposure of 7a to acidic methanol/2,2-dimethoxypropane 1:1 at 0° afforded crystalline acetonide 8a; m.p. 151-153°, $[a]_D^{25} = -28^{\circ}$ (c = 1, CHCl₃) in 54% overall yield from 4. Treatment of either 7a or 8a with acidic methanol gave (25 R)-1a, 25, 26-trihydroxycholesterol; m.p. 235-238°, $[a]_D^{25} = -24^{\circ}$ (c = 1, DMF).

In similar fashion, the ester 4 was alkylated with iodide 5b to give monoalkylated ester 6b. This substance was submitted to the hydrogenolysis sequence to yield oily acetonide 7b, which was treated with methanol/2,2-dimethoxypropane

1:1 and p-toluenesulfonic acid to yield the crystalline acetonide **8b**; m.p. 175-177°, $[a]_D^{25} = -40^\circ$ (c = 1, CHCl₃). Both compounds **7b** and **8b** then afforded (25 S)-1a, 25, 26-trihydroxycholesterol; m.p. 197-199°, $[a]_D^{25} = -29^\circ$ (c = 1, DMF) on stirring with acidic methanol.

The cholesterol acetonides **8a** and **8b** were converted into 5,7-diene derivatives **9a** and **9b** by sequential acetylation (acetic anhydride, pyridine), allylic bromination (1,3-dibromo-5,5-dimethylhydantoin), and dehydrobromination (s-collidine). These dienes were individually photolyzed using a mercury lamp to give the corresponding previtamins which were saponified and thermolyzed at 90°. The acetonide groups were then removed with acidic methanol. HPLC.-purification and recrystallization afforded pure (25 R)-1a, 25, 26-trihydroxycholecalciferol **10a**; m.p. 146-148°, $[a]_D^{25} = +64^\circ$ (c=0.5, CH₃OH) in 30% overall yield from **8a** and (25 S)-1a, 25, 26-trihydroxycholecalciferol **10b**; m.p. 162-164°, $[a]_D^{25} = +59^\circ$ (c=0.5, CH₃OH) in 33% overall yield from **8b**.

To prove unequivocally the absolute configuration of the natural bovine metabolite, we compared the HPLC. elution times of the 1,3,26-triacetates of the natural substance and of the synthetic $(25\,R)$ -1a,25,26- and $(25\,S)$ -1a,25,26-trihydroxycholecalciferols. A 1:1 mixture of the synthetic triacetates, individually prepared with acetic anhydride in pyridine, gave the HPLC. trace A of the Figure. This tracing demonstrated that the $(25\,S)$ -epimer is eluted before the $(25\,R)$ -epimer. A sample of bovine metabolite was prepared by multi-step purification of eight liters of bovine plasma [4]. This sample was purified further by HPLC. (μ Porasil® column; hexane/2-propanol 4:1 as eluant) and then acetylated. The bovine metabolite and its triacetate were identified by their characteristic UV. and high-resolution mass spectra. Addition of the triacetylated bovine metabolite to the ca. 1:1 mixture of triacetylated synthetic epimers previously described gave the HPLC. trace B shown in the Figure. The co-migration of the bovine metabolite with the faster-eluting epimer shows that bovine 1a,25,26-trihydroxycholecalciferol possesses the $(25\,S)$ -configuration.

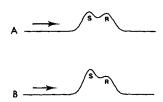


Figure. A) HPLC. tracing of a 1:1 mixture of the synthetic triacetates; B) HPLC. tracing of a 1:1 mixture of the synthetic triacetates upon addition of the triacetylated bovine metabolite.

 $R = (25R) - 1\alpha$, 25, 26-Trihydroxycholecalciferol 1, 3, 26-triacetate $S = (25S) - 1\alpha$, 25, 26-Trihydroxycholecalciferol 1, 3, 26-triacetate

Chromatograph: Waters 244

Eluant: 2:1 hexane/ethyl acetate
Flow Rate: 2 ml/min - 5 recycles
Column: 3× μ Porasil®
Detector: UV., 254 nm

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