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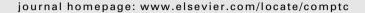
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### Corrigendum

Corrigendum to "Theoretical study of the electronic (hyper)polarizabilities of amino acids in gaseous and aqueous phases" [Comput. Theor. Chem. 976 (2011) 188–190]

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The author regrets that the range of  $\langle \alpha \rangle$  values in Fig. 4 was incorrectly presented in the published article. Fig. 4 is correctly

gas phase
\*--\* water solution

3

1

O gas phase
\*--\* water solution

Ala

Gly

Ala

Gly

**Fig. 4.** Relationship between  $\langle \gamma \rangle$  and  $\langle \alpha \rangle$  values of the aliphatic amino acids computed at the MP2/aug-cc-pVDZ level. Gas phase  $\langle \alpha \rangle$  values are taken from Ref. [1]. Gas phase:  $\langle \gamma \rangle$  =  $-0.186 + 0.020 \cdot \langle \alpha \rangle$ , r = 0.99. Water solution:  $\langle \gamma \rangle$  =  $-0.594 + 0.034 \cdot \langle \alpha \rangle$ , r = 0.99.

80

 $\langle \alpha \rangle$  (a.u.)

100

120

60

reproduced below (with the range for the  $\langle\alpha\rangle$  values at 40–130 a.u. instead of 0–100 a.u.).

### Reference

[1] S. Millefiori, A. Alparone, A. Millefiori, A. Vanella, Electronic and vibrational polarizabilities of the twenty naturally occurring amino acids, Biophys. Chem. 132 (2008) 139–147 (and references therein).