See discussions, stats, and author profiles for this publication at: https://www.researchgate.net/publication/244135062

High-lying Rydberg states of vinyl bromide studied by two-photon resonant ionization spectroscopy

ARTICLE in CHEMICAL PHYSICS LETTERS · AUGUST 2004

Impact Factor: 1.9 · DOI: 10.1016/j.cplett.2004.06.128

CITATIONS	READS
3	18

4 AUTHORS, INCLUDING:



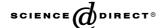
University of Colorado Boulder

20 PUBLICATIONS 478 CITATIONS

SEE PROFILE



Available online at www.sciencedirect.com



Chemical Physics Letters 394 (2004) 287-292



www.elsevier.com/locate/cplett

High-lying Rydberg states of vinyl bromide studied by two-photon resonant ionization spectroscopy

Chi-Hung Chuang, Chun-Cing Chen, Hsing-Chen Wu, Yit-Tsong Chen *

Department of Chemistry, National Taiwan University, No. 1, Sec. 4, Roosevelt Road, Taipei 106, Taiwan Institute of Atomic and Molecular Sciences, Academia Sinica, P.O. Box 23-166, Taipei 106, Taiwan

Received 26 April 2004; in final form 29 June 2004 Available online 27 July 2004

Abstract

High-lying Rydberg states of vinyl bromide at $70\,000-79\,500$ cm⁻¹ have been investigated using 2+1 resonance-enhanced multiphoton ionization (REMPI) spectroscopy. Seven Rydberg series, including one s, three p, two d, and one f, have been identified from fitting the term values of observed electronic states to Rydberg formula rendering the ionization energy of $79\,194\pm17$ cm⁻¹ (9.819 ±0.002 eV). All of the seven series converge to the same ionization-energy limit, corresponding to the ground state of vinyl bromide cation. The adiabatic ionization energy of vinyl bromide determined from this study is in excellent agreement with the values reported recently by other methods.

© 2004 Elsevier B.V. All rights reserved.

1. Introduction

Spectroscopic investigations of the excited electronic states and potential energy surfaces of polyatomic molecules are of fundamental importance to study molecular photochemical reactions. The spectroscopic [1–12] and dissociation dynamical [13–18] studies of vinyl bromide (C₂H₃Br, abbreviated as VB hereafter) have been extensively studied in the past decades. Following our previous investigations of vinyl chloride (C₂H₃Cl) ranging from its first excited electronic state to the first ionization energy (IE) [19–21], we are now extending the spectroscopic study to VB, an analogue to vinyl chloride with a halogen substitution.

In an early spectroscopic study, Chadwick et al. [1] used photoelectron spectroscopy (PES) to obtain the spectra of VB, 1,2-dibromoethylene, and 2-bromopropene, from which the interaction between lone-pair electrons located at the Br atom and π -bonding electrons

* Corresponding author. Fax: +886223620200. E-mail address: ytchen@pub.iams.sinica.edu.tw (Y.-T. Chen). was studied. Mines and Thompson [2] also applied PES in the range of 6–21 eV to determine the IEs of vinyl halides and allyl halides. Schander and Russell [3] reported the vacuum ultraviolet (VUV) absorption spectrum of VB at 110–210 nm, and identified two Rydberg series, $\pi \rightarrow ns$ and $\pi \rightarrow np$, convergent to the first IE of 79 860 cm⁻¹. In the same work, they also assigned three Rydberg series, $n \rightarrow ns$, $n \rightarrow np$, and $n \rightarrow nd$, with the same convergence limit to the second IE of 87 880 cm⁻¹. Later, Sze et al. [4] obtained valence-shell electron energy lost (VSEEL) spectra of monohaloethylenes, in which s, p, and d Rydberg states were tentative assigned.

Recently, Hoxha et al. [5,6] applied synchrotron radiation to observe the VUV absorption spectrum of VB, and identified three Rydberg series, $2a'' \rightarrow ns$, $2a'' \rightarrow np$, and $2a'' \rightarrow nd$, convergent to the first IE of 9.804 eV. Spectroscopic analysis for the transitions with excitation energies at 5.0–12.0 eV revealed that the electronic transitions involve the excitations of valence to virtual valence states and Rydberg series. Lee and Kim [7] reported a very accurate IE=9.8171±0.0006 eV (79180±5 cm⁻¹) of VB by one-photon VUV mass

analysis threshold ionization (MATI) spectroscopy, and assigned most of the observed vibronic transitions to the cationic ground state based on calculated vibrational frequencies and Franck–Condon factors. Qian et al. [8] employed high-resolution pulsed field ionization-photoelectron (PFI-PE) and threshold photoelectron (TPE) spectroscopies to measure the first IE of 9.8200 ± 0.0015 eV $(79\,204\pm12~{\rm cm}^{-1})$ and second IE of 10.9156 ± 0.0010 eV $(88\,040\pm8~{\rm cm}^{-1})$ in VB. In the same study, members of four Rydberg series $(7a'\rightarrow ns, 7a'\rightarrow nd,$ and two $7a'\rightarrow np$) convergent to the second IE were observed. Other spectroscopic studies of VB also include the measurements of vibrational and rotational features in the infrared and microwave spectral regions [9–11].

In the present work, we have focused on the study of the high-lying Rydberg states of VB at $70\,000-79\,500$ cm⁻¹ by 2+1 REMPI spectroscopy. From the observed 2+1 REMPI spectrum of VB, we have been able to identify seven Rydberg series, and to determine the first IE to be $79\,194\pm17$ cm⁻¹, in excellent agreement with the recently reported values of $79\,180\pm5$ cm⁻¹ by MATI spectroscopy [7] and $79\,204\pm12$ cm⁻¹ by PFI-PE spectroscopy [8].

2. Experimental

The experimental details for obtaining 2+1 REMPI molecular spectra in our laboratory have been described in several earlier publications [22–25]. Briefly, VB (Aldrich, 98%) of 60 Torr partial pressure was seeded in 2 atm helium to form a mixture, which was then expanded into a vacuum chamber through a pulsed valve. A focused laser beam was led to cross the molecular beam and to ionize VB. The VB ions, repelled by an electric field, moved across a field-free time-of-flight (TOF)

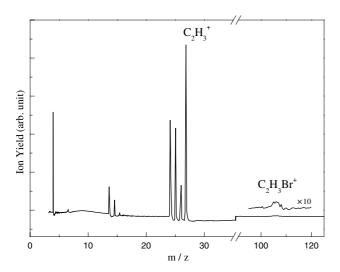


Fig. 1. Time-of-flight mass spectrum of vinyl bromide excited at 292.5 nm by two-photon resonant ionization.

mass spectrometer and were finally detected by a microsphere plate. An optogalvanic hollow-cathode lamp filled with Ne was employed simultaneously in the spectral scan to calibrate the laser frequency.

A typical TOF mass spectrum of VB via 2+1 REMPI is shown in Fig. 1. Based on a similar experience in our previous study of vinyl chloride [19,20], the signal of m/z = 27 ($C_2H_3^+$), the strongest one in the observation (Fig. 1), was monitored as a function of laser frequency due to the weak parent ion ($C_2H_3Br^+$) signal. Care was taken to ensure that the 2+1 REMPI spectra are identical by monitoring either the fragmentary $C_2H_3^+$ or the parent ion $C_2H_3Br^+$.

3. Results and discussion

The 2+1 REMPI spectrum of VB at 70000–79500 cm⁻¹ is shown in the lower trace of Fig. 2. The sharp edge at ~78750 cm⁻¹ marks the onset of ionization continuum of VB. The adiabatic IE=79180 cm⁻¹ of VB determined by MATI spectroscopy [7] is marked by an arrow. The upper trace in Fig. 2 from 70000 to 78750 cm⁻¹ has been magnified five times with respect to the lower one for clearer visualization of the spectral features. The well-resolved peaks with their transition frequencies and relative intensities are tabulated in Table 1.

In identifying the Rydberg series of VB, we have taken the previous REMPI spectroscopic analysis of vinyl chloride for comparison [19–21]. The Br atom could cause large quantum defects owing to its effective hindrance to Rydberg electrons when they travel close to the molecular core of VB. More congested spectral features might also be expected because of a possibly higher density of quantum states due to the heavier Br atom. Observed transitions from the 2+1 REMPI spectrum are analyzed using the Rydberg formula [26]

$$v = IE - R/(n - \delta^2),$$

where v is transition frequency, IE is ionization energy, $R = 109736.7526 \text{ cm}^{-1}$ is the Rydberg constant taking the mass of VB into account, n is a principal quantum number, and δ is the corresponding quantum defect. The value of n- δ can also be denoted as an effective quantum number of n^* , which represents quite distinctively for different Rydberg orbitals.

In the present study, we have been able to identify seven Rydberg series of VB (Table 1) from the observed REMPI spectrum as due to the promotion of a π -electron to the s, p, d, and f Rydberg states. A series of transitions with regular and strong peak intensities have been fitted into the Rydberg formula. Fig. 3 shows the best fit of this series to the Rydberg formula rendering IE=79195 \pm 14 cm⁻¹. This series has been assigned as

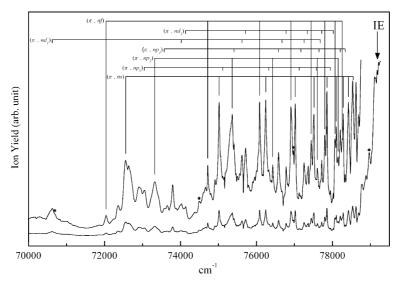


Fig. 2. 2+1 REMPI spectra and spectroscopic labelings for the transitions as due to the promotion of a π -electron to the s, p, d, and f Rydberg series of vinyl bromide at $70\,000-79\,500$ cm⁻¹. The upper trace has been magnified five times with respect to the lower one.

Table 1 Peak positions (ν), effective quantum numbers (n^*), ionization energies (IE) for the identified Rydberg series of vinyl bromide at 70 000–79 500 cm^{-1a}

(π, ns)		(π, np_1)		(π, np_2)		(π, np_3)		(π, nd_1)		(π, nd_2)		(π, nf)	
n^*	ν	n^*	ν	n*	ν	n^*	ν	n^*	ν	n^*	ν	n^*	ν
2.124	54757 ^b					2.523	61830°			2.852	65 581 ^d		
2.090	54735 ^e					2.469	$61862^{\rm f}$						
3.102	67670 ^b					3.438	69791°	3.582	70624(m)	3.934	71985 ^d	3.917	72041(m
3.017	67801 ^e					3.438	70 577 ^f						•
4.065	72555(s)	4.223	73051(m)	4.327	73321(s)	4.417	73 572(m)	4.614	74023(m)	4.651	74134(m)	4.953	74718(s)
4.093	72525 ^b					4.434	73493°			4.839	74388 ^d		
4.040	73 137 ^e					4.434	74278 ^f						
5.126	75018(s)	5.178	75112(w)	5.360	75362(s)	5.384	75410(s)	5.555	75621(s)	5.605	75715(s)	5.941	76083(s)
5.096	74848 ^b					5.330	75211 ^c			5.825	75840 ^d		
5.052	75 560 ^e					5.392	$76086^{\rm f}$						
6.100	76246(s)	6.173	76325(w)	6.312	76428(s)	6.479	76582(s)	6.617	76671(w)	6.723	76780(s)	6.932	76908(s)
6.148	76171 ^b					6.572	76534°			6.931	76784 ^d		
6.053	76865 ^e					6.425	77202^{f}						
7.106	77022(s)	7.226	77123(w)			7.360	77170(w)	7.552	77253(s)	7.667	77341(m)	7.912	77439(s)
						7.452	77099°			7.882	77 308 ^d		
						7.481	77899 ^f						
8.080	77514(s)	8.210	77 577(w)	8.334	77602(m)	8.420	77648(w)	8.565	77681(w)	8.588	77720(s)	8.879	77800(s)
9.056	77857(s)	9.329	77944(w)							9.619	78022(w)	9.876	78067(s)
10.015	78 101(s)			10.302	78 148(m)	10.523	78 205(s)					10.839	78 258(s)
10.975	78 284(s)					11.303	78337(m)						
11.923	78423(s)												
12.973	78 543(s)												
79 195 ± 14		79205 ± 27		79182 ± 19		79196 ± 20		79177 ± 25		79208 ± 27		79192 ± 13	

^a v and IE are in cm⁻¹. Average FWHM is \sim 30 cm⁻¹ for the observed transitions in this work. Relative intensities are indicated in parentheses with s (strong), m (medium), and w (weak).

 (π, ns) based on the identified $\pi \rightarrow ns$ [3] and $2a'' \rightarrow ns$ [6] in two earlier studies, which have also been listed in Table 1 for comparison.

For other Rydberg series of VB, there were several reports in previous studies. Schander and Russell [3] as-

signed $\pi \rightarrow ns$ and $\pi \rightarrow np$ transitions convergent to the first IE with δ values of 2.97 and 2.57, respectively. In the same study, they also assigned $n \rightarrow ns$ ($\delta = 2.96$), $n \rightarrow np$ ($\delta = 2.55$), and $n \rightarrow nd$ ($\delta = 2.11$) series converging to the second IE. In addition, their Rydberg-formula

^b Rydberg series of $2a'' \rightarrow ns$ taken from [6] with average FWHM ~ 100 cm⁻¹.

^c Rydberg series of $2a'' \rightarrow np$ taken from [6].

^d Rydberg series of $2a'' \rightarrow nd$ taken from [6].

^e Transition of $\pi \rightarrow ns$ taken from [3] with average FWHM ~250 m⁻¹.

f Transition of $\pi \rightarrow np$ taken from [3].

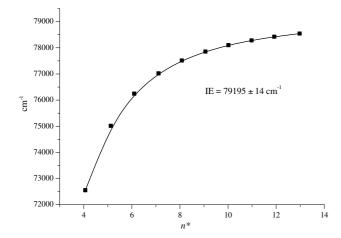


Fig. 3. The best fitting curve of the (π, ns) Rydberg series of vinyl bromide.

fittings for the transitions of bromomethanes and dibromoethylenes render the δ values between 2.0 and 3.0 for s and p Rydberg series [3,27,28]. Sze et al. [4] obtained the VSEEL spectrum of VB, in which Rydberg states of 5-10s (5-11s), 5-8p (5-7p), and 4d (4-6d) convergent to the first (second) IE were tentatively assigned. Based on their assignment, we have tried to fit their observed transitions to Rydberg formula and obtained the δ values of ~2.9 (~3.2), ~2.5 (~2.5), and ~1.4 (~1.1), respectively, for s, p, and d Rydberg orbitals convergent to the first (second) IE. Recently, Qian et al. [8] employed high-resolution PFI-PE and TPE spectroscopies to measure the first and second IE's of VB, and to identify one $7a' \rightarrow ns$ ($\delta = 1.075$), two $7a' \rightarrow np$ ($\delta = 0.465$ and 0.816), and one $7a' \rightarrow nd$ ($\delta = 0.235$) Rydberg series converging to the second IE. It is noted that their determined δ values for the Rydberg series of VB differ significantly from those reported earlier [3,4].

In the past, δ values were often used to identify the types of Rydberg states. For instance, typical δ values accompanying with the Rydberg series of a hydrocarbon molecule are ~ 1 , ~ 0.5 , ~ 0.3 , and ≤ 0.1 for s, p, d, and f orbitals, respectively [26,28]. To date, definite δ values for the Rydberg series of VB, however, are still lacking. Inconsistency in the δ values of VB from different reports [3,4,8] is evidenced as discussed above. While the determination of δ values requires a definitive assignment for the n quanta, an effective quantum number n^* , on the other hand, can be obtained quite straightforwardly from a Rydberg-formula fitting. In general, definite identifications of n and δ in a Rydberg series can be facilitated with an unambiguous assignment starting from the low-lying members in the Rydberg series, either from spectroscopic assignment or with the aid of theoretical calculations, e.g. our previous study of vinyl chloride [19–21].

Since the information about the lowest-lying member in each Rydberg series of VB is so far scarce, we will only determine the corresponding n^* values for the observed Rydberg states in this study. The type of Rydberg series, nevertheless, can still be figured out from the decimal of n^* . For instance, when the Rydberg states have the decimals of n^* to be ~ 0.1 , 0.2-0.5, 0.5-0.8, and ~ 0.9 , they correspond, respectively, to s, p, d, and f Rydberg series [6]. In the following analyses for the p, d, and f series of VB, the decimal of n^* will be used as a guideline for spectral assignments.

Three Rydberg series from our REMPI observation have been assigned to $\pi \rightarrow np$ transitions, denoted as (π, np_1) , (π, np_2) , and (π, np_3) , with IE=79205±27, 79182±19, and 79196±20 cm⁻¹, respectively. The observed transitions for the (π, np_1) series are either congested with other bands and located at peak shoulders or showing up with weak intensities. Transitions in the (π, np_2) series, on the other hand, are of medium to strong spectral intensities as shown in Fig. 2, although peaks corresponding to $n^* \sim 7.3$ and 9.3 were not observed. Peaks belonging to the (π, np_3) series are relatively weak at $n^* = 7.360$ and 8.420, but have medium to strong intensities elsewhere. Except the (π, np_3) series, which has been reported before [3,6], both (π, np_1) and (π, np_2) are identified for the first time.

As listed in Table 1, Rydberg series with IE=79177±25 and 79208 ± 27 cm⁻¹ are, respectively, assigned to (π, nd_1) and (π, nd_2) . The (π, nd_2) series has also been reported before by Hoxha et al. [6] and is listed in Table 1 for comparison. Comparing to the strong (π, ns) series, transitions in the (π, nd) series appear fewer in number. Peak intensities in the two (π, nd) series are widely distributed from strong to weak, and fluctuate along the series.

In the $\pi \rightarrow nf$ transitions of VB, a long series with sharp peaks and strong spectral intensities has been observed to converge to 79 192 ± 13 cm⁻¹. This series is a newly discovered one, whose peak positions appear near the (π, ns) series. Two factors have brought us to assign this series as (π, nf) . First, based on the decimal of n^* close to 1, it is legitimate to assign the concerned series as (π, nf) . Second, in view of the transitions to f Rydberg states of vinyl chloride [18,19], ethylene [29], and chloroethylenes [30], narrow lineshapes with strong peak intensities are common features in all of these observations. Moreover, because the relatively higher-lying f orbitals could have less perturbations from valence states, the long (π, nf) series can be expected to provide more accurate IE values. Indeed, IE = 79 192 ± 13 cm⁻¹ resulting from the Rydberg-formula fit of the (π, nf) series is very close to $IE = 79195 \pm 14$ cm⁻¹ obtained from another long (π, ns) series and both consist with the recently reported values by MATI [7] and PFI-PE [8] spectroscopies.

Due to the better spectral resolution of laser-assisted REMPI technique and the narrower population distribution of jet-cooled molecules in the present work, detailed features in the congested spectrum of VB can be discerned. The much better spectral resolution in this work (FWHM ~ 30 cm⁻¹ for the observed REMPI peaks), comparing to those in some earlier observations (FWHM's of \sim 250 and \sim 100 cm⁻¹ in the VUV absorption spectra by Schander and Russell [3] and Hoxha et al. [6], respectively), has allowed us to identify four new Rydberg series, (π, np_1) , (π, np_2) , (π, nd_1) , and (π, nf) . In an early study of $\pi \rightarrow ns$ and $\pi \rightarrow np$ by Schander and Russell [3], the series converge to IE = $79\,860 \text{ cm}^{-1}$ which is $\sim 700 \text{ cm}^{-1}$ away from the accurate values determined recently [7,8]. Nevertheless, based on similar transition frequencies, these two series of $\pi \rightarrow ns$ and $\pi \rightarrow np$ [3] should correspond, respectively, to (π, ns) and (π, np_3) of this work. In the study by Hoxha et al. [6], three series of VB, $2a'' \rightarrow ns$, $2a'' \rightarrow np$, and $2a'' \rightarrow nd$, were identified. In their assignment, IE=9.804 eV determined from their PES data (\sim 120 cm⁻¹ away from the recently measured IEs [7,8]) was used in the Rydberg-formula fit of their observed transitions. The $2a'' \rightarrow ns$, $2a'' \rightarrow np$, and $2a'' \rightarrow nd$ series [6] should also correspond, respectively, to the (π, ns) , (π, np_3) , and (π, nd_2) series of this work.

Most of the features observed in our 2+1 REMPI spectrum have been assigned to the Rydberg states converging to the first IE. However, some unassigned transitions are still remained in Fig. 2. These unassigned transitions might belong to either some other unidentified Rydberg series convergent to the first IE or the Rydberg series with convergence limit to the second or higher IE. For example, several peaks marked with asterisks in our 2+1 REMPI spectrum (Fig. 2) are found close/coincident with the transitions convergent to the second IE observed by Hoxha et al. [6], e.g. the REMPI peak at 70708 cm⁻¹ possibly corresponding to the assigned $7a' \rightarrow 5p_{\perp}$ transition at 70.824 cm⁻¹ in [6], and peaks at 74500, 76980, and 78971 cm⁻¹ to the $7a' \rightarrow 5d$, $7a' \rightarrow 6s$, and $7a' \rightarrow 6p_{\perp}$ transitions, respectively. In our measurements, we have not been able to obtain the transitions higher than the first IE, which however are crucial for the assignment of such Rydberg series convergent to the second or higher IE. In summary, the seven identified Rydberg series observed in this study fit very well to the Rydberg formula. We have taken the average IE value of the seven series to be $79194\pm17 \text{ cm}^{-1}$ (9.819±0.002 eV) as the adiabatic IE, which is in excellent agreement with the values obtained previously from various methods [1,2,4,5,7,8,12].

4. Conclusion

Using 2+1 REMPI spectroscopy, we have observed seven Rydberg series of VB at $70\,000-79\,500$ cm⁻¹, which have been assigned as due to the promotion of a π -electron to the s, p, d, and f Rydberg orbitals.

Among the seven Rydberg series, four of them are analyzed for the first time. The adiabatic IE of VB determined to be $79\,194\pm17~{\rm cm}^{-1}~(9.819\pm0.002~{\rm eV})$ is in excellent agreement with the recently reported values by MATI and PFI-PE spectroscopies.

Acknowledgements

The authors thank Prof. Jia-Lin Chang (Department of Science Education, National Taichung Teachers College) for his fruitful discussions during the course of this work. Research grants supported from National Science Council of ROC (Grant Nos. 92-2113-M-001-037 and 92-2113-M-002-039) are gratefully acknowledged.

References

- D. Chadwick, D.C. Frost, A. Katrib, C.A. McDowell, A.N. McLean, Can. J. Chem. 50 (1972) 2642.
- [2] 2.G.W. Mines, H.W. Thompson, Spectrochim. Acta, Part A 29 (1973) 1377.
- [3] J. Schander, B.R. Russell, J. Am. Chem. Soc. 98 (1976) 6900.
- [4] K.H. Sze, C.E. Brion, A. Katrib, B. El-Issa, Chem. Phys. 137 (1989) 369.
- [5] A. Hoxha, R. Locht, B. Leyh, D. Dehareng, K. Hottmann, H. Baumgartel, Chem. Phys. 256 (2000) 239.
- [6] A. Hoxha, R. Locht, B. Leyh, D. Dehareng, K. Hottmann, H.W. Jochims, H. Baumgartel, Chem. Phys. 260 (2000) 237.
- [7] M. Lee, M.S. Kim, J. Chem. Phys. 119 (2003) 5085.
- [8] X.M. Qian, K.C. Lau, C.Y. Ng, J. Chem. Phys. 120 (2004) 11031.
- [9] A.P. Charmet, P. Stoppa, A. Baldacci, S. Giorgianni, S. Ghersetti, J. Mol. Struct. 612 (2002) 213.
- [10] S. Marchetti, R. Simili, M. Giorgi, Infrared Phys. Technol. 42 (2001) 529.
- [11] M. Hayashi, C.A. Ikeda, T. Inagusa, J. Mol. Spectrosc. 139 (1990) 299.
- [12] W. Lohr, H.W. Jochims, H. Baumgartel, Ber. Bunsenges. Phys. Chem. 79 (1975) 901.
- [13] M.E. Ebert, S.A. Abrash, L.M. Raff, J. Phys. Chem. 99 (1995) 17691.
- [14] H. Katayanagi, N. Yonekuira, T. Suzuki, Chem. Phys. 231 (1998) 345.
- [15] S.-R. Lin, S.-C. Lin, Y.-C. Lee, Y.-C. Chou, I.-C. Chen, Y.-P. Lee, J. Chem. Phys. 114 (2001) 7396.
- [16] D.-K. Liu, L.T. Letendre, H.-L. Dai, J. Chem. Phys. 115 (2001) 1734.
- [17] A. Rahaman, L.M. Raff, J. Phys. Chem. A 105 (2001) 2156.
- [18] S. Kunsagi-Mate, E. Vegh, G. Nagy, L. Kollar, J. Phys. Chem. A 106 (2002) 6319.
- [19] J.-L. Chang, J.-C. Shieh, J.-C. Wu, R. Li, Y.-T. Chen, Chem. Phys. Lett. 325 (2000) 369.
- [20] J.-L. Chang, R. Li, J.-C. Wu, J.-C. Shieh, Y.-T. Chen, J. Chem. Phys. 115 (2001) 5925.
- [21] J.-L. Chang, Y.-T. Chen, J. Chem. Phys. 116 (2002) 7518.
- [22] J.-C. Shieh, J.-L. Chang, J.-C. Wu, R. Li, A.M. Mebel, N.C. Handy, Y.-T. Chen, J. Chem. Phys. 112 (2000) 7384.
- [23] J.-C. Wu, R. Li, J.-L. Chang, Y.-T. Chen, J. Chem. Phys. 113 (2000) 7286.
- [24] C.-W. Liang, C.-C. Chen, C.-Y. Wei, Y.-T. Chen, J. Chem. Phys. 116 (2002) 4162.

- [25] C.-C. Chen, H.-C. Wu, C.-M. Tseng, Y.-H. Yang, Y.-T. Chen, J. Chem. Phys. 119 (2003) 241.
- [26] A.B.F. Duncan, Rydberg Series in Atoms and Molecules, Academic Press, New York, 1971.
- [27] G.C. Causley, B.R. Russell, J. Chem. Phys. 62 (1975) 848.
- [28] M.B. Robin, Higher Excited States of Polyatomic Molecules, Vol. 1, Academic Press, New York, 1974.
- [29] B.A. Williams, T.A. Cool, J. Chem. Phys. 94 (1991) 6358
- [30] B.A. Williams, T.A. Cool, J. Phys. Chem. 97 (1993) 1270.