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Functional expression and characterization of cytochrome P450 52A21 from *Candida albicans*[§] S:

FATTY ACID ω -HYDROXYLATION IN A NEW PROTEIN FRAMEWORK*

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Abstract

Candida albicans contains ten putative cytochrome P450 (CYP) genes coding for enzymes that appear to play important roles in fungal survival and virulence. Here, we report the characterization of CYP52A21, a putative alkane/fatty acid hydroxylase. The recombinant CYP52A21 protein containing a 6 × (His)-tag was expressed in *Escherichia coli* and was purified. The purified protein, reconstituted with rat NADPH-cytochrome P450 reductase, ω -hydroxylated dodecanoic acid to give 12-hydroxydodecanoic acid, but to a lesser extent also catalyzed (ω -1)-hydroxylation to give 11-hydroxydodecanoic acid. When 12,12,12-d₃-dodecanoic acid was used as the substrate, there was a major shift in the oxidation from the ω - to the (ω -1)-hydroxylated product. The regioselectivity of fatty acid hydroxylation was examined with the 12-iodo-, 12-bromo-, and 12-chlorododecanoic acids. Although all three 12-halododecanoic acids bound to CYP52A21 with similar affinities, the production of 12-oxododecanoic acid decreased as the size of the terminal halide increased. The regioselectivity of CYP52A21 fatty acid oxidation is thus consistent with presentation of the terminal end of the fatty acid chain for oxidation via a narrow channel that limits access to other atoms of the fatty acid chain. This constricted access, in contrast to that proposed for the CYP4A family of enzymes, does not involve covalent binding of the heme to the protein.

Keywords

cytochrome P450; *Candida albicans*; fatty acid hydroxylase; covalent heme binding; halogen oxidation

Introduction

The cytochrome P450 monooxygenases (CYPs or P450s) are members of a superfamily of heme-containing enzymes involved in the oxidative metabolism of endo- and xenobiotic chemicals [1]. These P450 enzymes are found throughout nature, from archaeobacteria to humans [2]. Despite differences in the phylogenetic origin, cellular localization, electron donor

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partners, and substrate specificities of P450 enzymes, their gross active sites and catalytic mechanisms appear to be remarkably conserved [3].

Candida albicans is a major pathogenic fungus that causes opportunistic oral and vaginal infections in humans [4,5]. Systemic fungal infections have emerged as important causes of morbidity and mortality in immunocompromised patients, such as those infected with HIV or undergoing cancer chemotherapy and/or organ/bone marrow transplantation [6,7]. The complete genome of *C. albicans*, was recently determined [8]. Previously, we reported the characterization of a fungal heme oxygenase, CaHmx1, from *C. albicans* and proposed that it might be a suitable target for agents directed at the treatment of *Candida* infections [9]. Furthermore, the *C. albicans* genome contains genes coding for approximately ten putative P450 enzymes, including CYP51 and CYP61, both of which are also found in *S. cerevisiae* (<http://www.candidagenome.org/>). It also possesses one member of the CYP52 family that is normally characterized by alkane-inducibility. CYP52 enzymes have been found in soil yeasts such as *Candida tropicalis* and *Candida maltosa*. These P450 enzymes are commonly responsible for the first and rate-limiting step in the ω -oxidation of *n*-alkanes and/or fatty acids [10-13]. The CYP52 from *C. albicans* was previously identified and was shown *in vivo* to possibly play a role in multidrug resistance, although it was not involved in azole drug resistance [14]. The oxidation of hydrocarbons or fatty acids by CYP52 thus appears to be important for lipid metabolism and for the pathogenesis and survival of *C. albicans*.

Fatty acids are commonly hydroxylated by the mammalian CYP4 enzymes, a family of proteins that are characterized by their unique ability to hydroxylate the thermodynamically disfavored terminal methyl group of fatty acids rather than the default, thermodynamically favored internal chain carbons that are hydroxylated by other P450 enzymes [1]. Clearly, the CYP4 enzymes have evolved a strategy that allows them to specifically direct the reaction to the terminal (ω -) carbon of the fatty acid chain. We have postulated that this strategy involves a constricted access channel through which the fatty acid terminus must be threaded for oxidation, have proposed that covalent binding of the heme to the protein facilitates the assembly of a sufficiently rigid access channel, and have used terminally halogenated fatty acids to obtain information on the effective diameter of the CYP4A channel [15-17].

In this study, we have overexpressed and purified CYP52A21, the first recombinant P450 from *C. albicans*, and have characterized the protein in terms of its structural, spectroscopic, and catalytic properties. The results indicate that covalent binding of the heme to the protein is not required for a narrow channel access to be used to control the fatty acid hydroxylation regiospecificity.

Materials and methods

Chemicals and enzymes

Dodecanoic acid, 12,12,12- d_3 -dodecanoic acid, sodium dithionite, glucose-6-phosphate, glucose-6-phosphate dehydrogenase, and NADP⁺ were purchased from Sigma (St. Louis, MO) or Aldrich Chemical Co. (Milwaukee, WI). The syntheses of 12-chlorododecanoic acid, 12-bromododecanoic acid, and 12-iodododecanoic acid have been reported elsewhere [17]. Other chemicals were of the highest grade commercially available. Recombinant rat NADPH-P450 reductase and CYP4A1, both expressed in *E. coli*, were kindly provided by Dr. Fengyun Xu (UCSF).

Construction of expression plasmids

The general approach has been described [9,18]. The genomic DNA from *C. albicans* was kindly provided by the laboratory of Alexander Johnson at UCSF. The open reading frame for

CYP52A21, and an added 6 × His-C-terminal tag, were amplified using PCR and the amplified PCR fragment was cloned into the pCW(Ori⁺) vector using the *Nde*I and *Xba*I restriction sites. The cloned vectors were verified by nucleotide sequencing and restriction digestion. The open reading frame of CYP52A21 contains a CTG encoding a Ser that was corrected by site-directed mutagenesis to enable expression of the correct recombinant protein in *E. coli* as previously described for CaHmx1 [9,19]. Site-directed mutagenesis to change Leu40 to a Ser was carried out using Quick-Change mutagenesis (Stratagene, La Jolla, CA). The change in nucleotides was confirmed by nucleotide sequencing.

Enzyme expression and purification

Expression and purification of CYP52A21 were carried out, with minor modifications, by previously described procedures [20]. Briefly, the *E. coli* strains transformed with the pCW (Ori⁺) vectors were inoculated into TB medium containing 100 µg/ml ampicillin and 1.0 mM IPTG. The expression cultures were grown in 1 liter Fernbach flasks at 37 °C for 3 h and then at 28 °C with shaking at 200 rpm for 24 h. Bacterial inner membrane fractions containing CYP52A21 were isolated and prepared from 1 liter TB (with ampicillin, 100 µg/ml) expression cultures of *E. coli* DH5α. Purification of the CYP52A21 protein using a Ni²⁺-nitrilotriacetate column was carried out as previously described for a different protein [20,21]. Briefly, the membrane was solubilized at 4 °C overnight in 100 mM potassium phosphate buffer (pH 7.4) containing 20% glycerol, 0.5 M NaCl, 10 mM β-mercaptoethanol, and 1.5% CHAPS (w/v). The solubilized fraction was then loaded onto a Ni²⁺-nitrilotriacetate column (Qiagen, Valencia, CA) and the purified protein was obtained with elution buffer that contained 400 mM imidazole. The eluted fraction containing the P450 was dialyzed at 4 °C against 100 mM potassium phosphate buffer (pH 7.4) containing 20% glycerol and 0.1 mM EDTA.

Spectroscopic characterization

Sodium dithionite was added to reduce the purified ferric P450. The CO-ferrous P450 complex was then generated by passing CO gas through a solution of the ferrous P450. UV-visible spectra were collected on a CARY Varian spectrophotometer in 100 mM potassium phosphate buffer (pH 7.4) at room temperature.

Measurement of CYP52A21 oxidation activity by GC-MS

The fatty acid ω/(ω-1)-hydroxylation activities of CYP52A21 were determined using a P450/NADPH-P450 reductase/phospholipid reconstituted systems [22]. The reaction mixture included 100 pmol of purified P450 protein, 250 pmol of rat NADPH-P450 reductase, and L-α-dilauroyl-*sn*-glycero-3-phosphocholine (45 µM) in 0.50 ml of 100 mM potassium phosphate buffer (pH 7.4), along with a specified amount of the substrate. An NADPH-generating system [22] was used to start the reactions. The incubations were generally done for 10-30 min at 37 °C and were terminated by addition of 0.5 ml of CH₂Cl₂. The CH₂Cl₂ extract was then dried under N₂ before the residue was trimethylsilylated by incubation with *N,O*-bis(trimethylsilyl) trifluoroacetamide (50 µl) at 70 °C for 10-15 min. The derivatized samples were allowed to cool, vortexed, and transferred to sealed Teflon-capped glass vials for either manual or autoinjection into the GC-MS. Analyses were performed on an Agilent 6850 gas chromatograph coupled to an Agilent 5973 Network Mass Selective Detector in the electron impact mode as previously described [17].

Heme staining on SDS-gel electrophoresis

Heme detection by SDS-PAGE was previously reported [23]. Briefly, SDS-gel electrophoresis was carried out for purified CYP52A21 and CYP4A1 (as a positive control). The gel was then immersed in the dark in a mixture of 3 parts of 3,3',5,5'-tetramethylbenzidine (1.5 mg/mL MeOH) and 7 parts of 250 mM sodium acetate buffer (pH 5.0) for 1-2 h before H₂O₂ was

added to a final concentration of 30 mM. Over the next 30 min, light blue bands emerged, indicating the presence of heme.

Spectral Binding Titrations

Purified CYP52A21 was diluted to 2 μ M in 100 mM potassium phosphate buffer (pH 7.4) and divided between two 0.5 ml glass cuvettes. Spectra (350 – 500 nm) were then recorded with subsequent additions of substrates (from a methanol stock [24]) using a CARY Varian spectrophotometer. The difference in absorbance between the absorption maximum (390 nm) and minimum (420 nm) was then plotted *versus* the substrate concentration [25].

Results

Sequence alignment with other P450s

The gene for CYP52A21 from *C. albicans* was named *ALK8* because it encodes a putative protein highly homologous to the alkane-inducible P450s from *C. maltosa* and *C. tropicalis* [14]. Previous *in vivo* experiments suggested that the protein encoded by *ALK8* could hydroxylate fatty acids [14]. The well-known P450 enzymes that catalyze fatty acid hydroxylation are the members of the mammalian P450 4A family and P450_{BM3} (CYP102) [1]. Amino acid sequence alignment of CYP52A21 with P450 4A1 and 4A3 revealed a high sequence similarity with a score of more than 70%, while the alignment with P450_{BM3} was only 30% (Fig. 1) [26]. The good sequence alignment of CYP52A21 with CYP4A3 suggests that they may have similar enzymatic specificities. However, it is significant that the conserved Glu residue responsible for covalent attachment of the heme to the protein in the P450 4A family is not found in CYP52A21 (Fig. 1) [27]. Indeed, heme staining by adding 3,3',5,5'-tetramethylbenzidine and H₂O₂ to an SDS-Page gel showed that CYP52A21 does not have a covalently bound heme, in contrast to CYP4A1 which, as expected, does contain the protein-bound heme (supplementary Fig. S1).

Expression and purification of CYP52A21

The typical P450 expression level spectrally determined in the whole cell culture was around 60 nmol P450 holoenzyme per liter of culture medium. After separation by ultracentrifugation, there was no detectable P450 in the soluble fraction and all the detectable P450 was located in the membrane fraction. CHAPS, an anionic surfactant, successfully solubilized the P450 protein from the membrane fraction. Purified CYP52A21, obtained after chromatography on a Ni-column, gave a single band on SDS-PAGE at 52 kDa, as expected from the open reading frame of the gene (Fig. 2).

Spectral properties of CYP52A21

The reduced CO difference spectrum of CYP52A21 exhibited an absorption maximum at 449 nm (Fig. 2A). Examination of the absolute spectra showed that the ferric form of the protein was almost completely in the low spin state (Fig. 2B).

Binding of dodecanoic acid to CYP52A21

Titration of purified CYP52A21 with dodecanoic acid (lauric acid) showed a typical type I substrate binding spectral change (increase at 390 nm, and decrease at 420 nm). This is consistent with displacement of the distal iron water ligand upon binding of the dodecanoic acid to the protein (Fig. 3). The calculated K_d value is $5.1 \pm 0.4 \mu$ M (Fig. 3). In contrast, no spectral change was observed when the enzyme was titrated with either dodecane or hexadecane, which suggests that simple hydrocarbons are not bound by this enzyme (result not shown).

Catalytic activities of CYP52A21

A reconstituted system consisting of CYP52A21 and rat NADPH-P450 reductase catalyzed the oxidation of dodecanoic acid (Fig. 4). An NADPH generating system supported this reaction more efficiently than NADPH itself (data not shown). The major product of the reaction was the ω -hydroxylated 12-hydroxydodecanoic acid (12.26 min), although the (ω -1)-hydroxylated 11-hydroxydodecanoic acid (11.49 min) was also obtained as a minor product (Fig. 4). In general terms, this product pattern is similar to that obtained with the mammalian CYP4A enzymes. The concentration-dependence of dodecanoic acid oxidation is shown Fig. 5. The Hill equation instead of the Michaelis-Menten equation was used to obtain the best fit for the data, from which a k_{cat} of approximately 33 min^{-1} and a K_m of $57 \mu\text{M}$ were calculated (Fig. 5). Analysis of the data with the Hill equation indicated apparent cooperativity with $n = 2.3$. In accord with the finding that neither dodecane nor hexadecane caused a spectral shift, no detectable products were formed in incubations of these hydrocarbons with the reconstituted CYP52A21 system (data not shown).

ω -Hydroxylation by CYP52A21 was greatly attenuated while formation of the (ω -1)-hydroxylated product was increased when 12,12,12- d_3 -dodecanoic acid was used as the substrate instead of unlabeled dodecanoic acid (Table 1). This large shift in the hydroxylation regiospecificity caused by the replacement of the terminal methyl hydrogens by deuteriums indicates that both the ω - and (ω -1)-positions are accessible to the oxidizing species, although the intrinsically disfavored ω -hydroxylation is favored over the intrinsically easier (ω -1)-hydroxylation. However, as oxidation of a C-D bond has a higher energy barrier than that of a C-H bond, the consequent decrease in the rate of ω -hydroxylation with the deuterated substrate allows the (ω -1)-hydroxylation to compete more effectively, resulting in increased formation of 11-hydroxydodecanoic acid at the expense of 12-hydroxydodecanoic acid.

Oxidation of halododecanoic acids by CYP52A21

12-Halododecanoic acids have been used to investigate the dimensions of the putative channel through which substrates are presented for oxidation in the mammalian enzyme CYP4A1 [17]. The P450-catalyzed oxidation of halogen-substituted carbons normally proceeds via hydroxylation of the carbon followed by elimination of the halide ion to give the carbonyl product. However, in the CYP4A enzymes, which preferentially oxidize the terminal atom of the hydrocarbon chain, the oxidation results to a substantial extent in direct oxidation of the terminal halide atom. In the CYP4A1 study, it was found that the 12-chloro and 12-bromododecanoic acids underwent halide atom oxidation, but the 12-iodo substrate did not. As the inherent reactivity of the halogen atoms predicts the reverse reactivity order (i.e., $\text{I} > \text{Br} > \text{Cl}$), product formation was not governed by intrinsic halogen oxidizability. The oxidation order correlates, however, with the size of the halogen atom, as the chlorine is the smallest and the iodine the largest. From these results it was concluded that a "channel" limits substrate access to the oxidizing species, and that the channel was wide enough for presentation of a bromo but not iodo terminus.

As shown here, CYP52A21 also preferentially oxidizes the terminal carbon atom of fatty acid substrates. We have therefore investigated the ability of this enzyme to oxidize the 12-chloro-, 12-bromo-, and 12-iodododecanoic acids. Difference binding studies, which monitor the spin state change caused by displacement of the distal water ligand from the iron on binding of the fatty acid, first established that all three substrates are readily bound by the enzyme. The three 12-halododecanoic acids had comparable K_d values and were bound 1.2-2-fold tighter than the parent dodecanoic acid (Table 2). The catalytic studies established that the total hydroxylation (conversion) of the 12-halododecanoic acids by CYP52A21 decreased as the size of the terminal atom increased (Table 3). All three 12-halododecanoic acids underwent comparable oxidation of the halide atom, as indicated by formation of 12-hydroxydodecanoic acid (Table

3). Previous work has shown that the hydroxyl oxygen in this product derives from water, not molecular oxygen, and is introduced by hydrolysis of an initially formed oxohalonium ($R-X^+-O^-$) metabolite (17). However, the extent of “ ω -1” hydroxylation, which leads to the aldehyde 12-oxododecanoic acid, decreased in proportion to the size of the terminal halogen atom (i.e., $Cl > Br \gg I$). This suggests that the ability of the oxidizing species to reach around the terminal halogen atom to react with the last carbon in the chain is increasingly impaired as the terminal halogen atom gets larger. The results differ from those obtained with CYP4A1 in that the 12-iodosubstrate was oxidized to some extent by CYP52A21. The access channel in this enzyme thus appears to be modestly larger than in CYP4A1. It is not that large, however, as access to the last carbon of the chain (i.e., that adjacent to the halogen atom) was severely impaired as the halogen atom increased in size. As indicated in Table 4, the turnover numbers for the 12-halododecanoic acids decreased as the size of the halogen atom increased, and were much lower than the turnover numbers for the unhalogenated lauric acid, as might be expected from the greater difficulty in oxidizing substrates in which the available oxidation sites are a poorly oxidizable halogen atom or a carbon atom that is increasingly sterically masked as the size of the halogen atom increases.

Inhibition of CYP52A21 by azole compounds

C. albicans infections are treated with azole antifungal agents that are generally considered to act by inhibiting P450 enzymes involved in steroid and possibly lipid metabolism. We did not find any inhibition of CYP52A21 dodecanoic acid hydroxylation by fluconazole or econazole, and ketoconazole only showed a weak inhibition with $IC_{50} = 104 \mu M$ (Fig. 6). CYP52A21 thus does not appear to be one of the targets of azole antifungal agents in *C. albicans*.

Discussion

Baker's yeast *S. cerevisiae* has only three P450 genes: CYP51, a lanosterol demethylase; CYP57, which is responsible for synthesis of the dityrosine required in cell wall assembly; and CYP61, which has been identified by proteomic studies as a sterol 22-desaturase [28]. Fungal P450s, unlike the bacterial enzymes but like the mammalian proteins, are generally membrane-associated and their turnover is supported by a single NADPH-dependent flavoprotein reductase. Analysis of the genomic database (<http://www.candidagenome.org/>) showed that the pathogenic fungus *C. albicans* possesses approximately ten putative P450 enzymes, including a CYP52 protein. The CYP52 family encompasses the alkane-oxidizing proteins found in soil/environmental yeasts like *C. tropicalis* and *C. maltosa*, but homologues of these enzymes are not found in *S. cerevisiae* or *S. pombe*. An analysis by David Nelson indicates that the CYP52 sequences most closely resemble a cluster of bacterial P450 enzymes called the eukaryote-like bacterial P450s (<http://drnelson.utmem.edu/CytochromeP450.html>). These include CYP102 (a fatty acid hydroxylase) and CYP110. This cluster next joins with the CYP4 clan that includes the mammalian fatty acid hydroxylases.

The metabolites of arachidonic acid produced by the mammalian CYP4 family have important physiological roles. 20-Hydroxyeicosatetraenoic acid (20-HETE), the ω -hydroxylated product of arachidonic acid, is a vasoconstrictor [29,30], whereas 19-HETE, the (ω -1)-hydroxylated product, as well as some of the arachidonic acid epoxides are vasodilators [30-32]. Therefore, the ω -hydroxylation regiospecificity of CYP4A enzymes is of critical physiological importance. We previously (a) postulated that a constricted access channel controls ω -hydroxylation, (b) discovered that the heme is covalently bound to the protein in most CYP4 enzymes, (c) identified the residue involved in the covalent linkage, and (d) proposed that covalent heme binding is important for the architecture of the narrow channel [17,33,34]. It is thus somewhat unexpected that CYP52A21 from *C. albicans* predominantly hydroxylates fatty acids at the ω -position (Figure 4) despite the fact that the Glu residue involved in heme-covalent

binding in CYP4A enzymes is replaced in CYP52A21 by an Ala and its heme is not covalently bound (Fig. 1 and supplementary Fig. S1).

Nevertheless, the basic strategy used to enforce ω -hydroxylation in CYP52A21 appears to be similar to that used by the CYP4A enzymes: sterically restricting access to the oxidizing species so that oxidation of carbons other than the relatively unreactive terminal carbon is difficult. The observation that replacement of the terminal methyl in the fatty acid by a trideuteriomethyl group shifts the oxidation to the ω -1 position demonstrates that ω - and (ω -1)-hydroxylation occur in competition, and as ω -hydroxylation becomes more difficult due to the deuterium isotope effect, (ω -1)-hydroxylation increases in importance (Table 1). The evidence provided by the oxidation of 12-halododecanoic acids supports the existence of a sterically restricted channel for presentation of the terminal atom of the substrate to the oxidizing species. The channel may be slightly larger than in CYP4 enzymes, as CYP52A21 oxidizes not only the 12-chloro- and 12-bromo-, but also the 12-iodododecanoic acids to 12-hydroxydodecanoic acid. CYP4A1 only oxidizes the chloro and bromo analogues, presumably because the 12-iodo is too large to enter the channel. The fact that the (ω -1)-hydroxylation that leads to the terminal aldehyde decreases as the halide becomes larger supports the inference of a sterically restricted channel. Thus, if heme covalent binding contributes to the assembly of a rigid access channel in the CYP4 enzymes, its contribution is not a prerequisite to the formation of such a channel.

CYP52A21 oxidizes myristic (C₁₄) and palmitic (C₁₆) acids preferentially at the ω -position (Table 5). The conversion of myristic acid to the hydroxylated product is comparable to that of dodecanoic acid, but the hydroxylation of palmitic acid was decreased by 70%. In contrast, CYP52A21 did not detectably oxidize dodecane or hexadecane. This observation, the fact that the rate of hydroxylation of dodecanoic acid was the same in the presence or absence of hexadecane or dodecane, and the absence of a binding spectral shift when the hydrocarbons were added to ferric CYP52A21, clearly establish that the hydrocarbons do not bind and are not substrates for this enzyme. Two conclusions follow from these results: (a) the enzyme can process fatty acids of different lengths, and (b) the fatty acid carboxyl group is absolutely required for binding to CYP52A21. A differential specificity for alkanes versus fatty acids has been reported for the CYP52 enzymes of *C. maltosa* [12]. Alk5A was found to preferentially hydroxylate fatty acids whereas Alk1A preferentially hydroxylated alkanes [12]. CYP52A21 thus resembles Alk5A more than it does Alk1A.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgements

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The abbreviations used are

CYP

	cytochrome P450
PCR	polymerase chain reaction
TB	Terrific Broth
K_d	A dissociation constant estimated by measurement of the UV-visible changes in the P450 heme spectrum

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CYP52A21    ....MLEEIS .....FE SWYLYLIICG VLYFCTNYLY IQYLNKKLGA KSPGSCKSDG CFGFRSGFEM LKAKKIGRQV
CYP4A1      MSVSALSSTR FTGSIISGFLQ VASVLGLLLL LVKAVQFYLO RQWLLKAF.. QQFSPSPFFHW FFG...HKQ FQGDKEQQI
CYP4A3      MGFSVTFPTR SLDGVSGGFQ GAFLLSLFLV LFKAVQFYLR RQWLLKAL.. EKFPSTPSHW LWG...HD.. LKDRFQQV
CYP102      MTIKEMPQPK TFGE..... ..LKNLPLN TDKPVQALMK IADELGEI.. FKFEA..... ..
CYP52A21    DLLHERFNDL SNPEVNDTFT SYNFGFFMII TKDPENIKAL LATQFNDFSL GRRYDFFKPL LGKGIFT..L DGEGWKHSRS
CYP4A1      MTCVENFPFA FPRWFWSKA .....YLI VYDPDYMKVI LGR..SDPKA NGVYRLLAPW IGYGLLL..L NGQPWFQHR
CYP4A3      LTWVEKFPGA CLQWLSGSKT .....RVL LYDPDYVKVV LGR..SDPKA SGYQFLAPW IGYGLLL..L NGKWFQHR
CYP102      .....PGR VTRYLSSQRL .....IKE ACDES..... ..R..FDKNL SQALKFVRDF AGDGLFTSWT HEKNWKAHN
CYP52A21    MLRPQFAREQ IA.HVKALEP HFQLKKHIT KNKGQ..FFD IQELFFRFTV DSATEFLFGE SVSSLKDESI GYNQDEDFDA
CYP4A1      MLTPAFHYDI LKPYVKIMAD SIRLMLDKWE QLAGQDSSIE IFQHISLMTL DTVMKCAFSH N.GSVQVD.. ..GNYKSYIQ
CYP4A3      MLTPAFHYGI LKPYVKIMAD SVNIMLDKWE KLDDQDHPLE IFHYVSLMTL DTVMKCAFSH Q.GSVQLD.. ..VNSRSYTK
CYP102      ILLPSFSQQA MKGYHAMMVD IAVQLVQKWE RLNADE.HIE VPEDMTRLTL DTIGLCGFNY RFNSFYRD.. ..QHPFFITS
CYP52A21    GRKDFPE.AF NKSQIYLSTR SLLQNLVWL NPKDFQNCNA IVHKFSDYV N..KALNATP EEELEKNSGYV FL...YELV
CYP4A1      AIGNLNDLFH SRVRNIF..H QNDTIYNFSS NGHLEFRACQ LAHDHTDGI KLRKDQLQNA GELEKVKKKR RLDLDILL
CYP4A3      AVEDLNNLTF FRVRSF..Y GNSIIYNMSS DGRLSRRACQ IAHEHTDGI KMRKAQLQNE EELQKARKKR HLDLDILL
CYP102      MVRALDE... .AMNKLQ..R ANPDDPAYDE NKRQFQEDIK VMNDLVDKII ADRKASGEQS DDL..... ..LTHMLNG
CYP52A21    KQ....TRD PQVLRDQALN ILLAGRDTTA GLLSFAVFEL ARNPQVWKKL REEIIERYAD GVTEISFETL KQCEYLKAVI
CYP4A1      ARMENGDSL DKDLRAEVD FMEEGHDTTA SGVSMIFYAL ATHPEHQRC REEVQSVLGD G.SSITWDHL DQIPYTTMCI
CYP4A3      AKMEDGKSL DEDLRAEVD FMEEGHDTTA SGISWVFYAL ATHPEHQRC REEVQSILGD G.TSVTWDHL DQIPYTTMCI
CYP102      KPPTGTEPLD DENIRYQIIT FLIAGHEHETS GLLSFALYFL VKNPHVLQKA AEEAARVLVD P.VP.SYKQV KQLKVVGMVL
CYP52A21    NETLRLYPSV PRNGRFANKN TTLPRGGGPD GMSTILIKQG QAVMYSIYST QRDPKYYGKD ADVFRPERWF EPETRKLQWA
CYP4A1      KEALRLYPPV PGIVRELSTS VTFPDGRS.. .....LPKG IQVTLISYGL HHNPVKVWNP .EVFDPSR.F APDSPRHS
CYP4A3      KEALRLYPPV PSVRELSSP VTFPDGRS.. .....IPKG ITTILIYGL HHNPVWNP .KVFDPSR.F SPDSPRHS
CYP102      NEALRLWPTA PAFSLYAKED TVLGGEYP.. .....LEKG DELMVLIPQL HRDKTIWGDD VEEFRPER.F ENPSAIPQHA
CYP52A21    FLFPNGGPRI CLGQQFALTE ASYVLVRLAQ CCENLQLTEG IEYPPAKLTH LTMCMFDGVP IKVD..... ..
CYP4A1      FLFPNGGARN CIGKQFAMSE MKVIVALTLL RFE...LLPD PTKVPIPLR LVLKSKNGIY LYLKKLH... ..
CYP4A3      YLPFSGGARN CIGKQFAMNE LKVAVALTLL RFE...LLPD PTRIPVPMAR LVLKSKNGIH LRLKKLR... ..
CYP102      FKFPNGQRA CIGQQFALHE ATLVLMMLK HFD...F.ED HTNYELDIKE TLTLKPEGFV VKAKSKKIPL GGIPSPSTEQ
CYP52A21    .....
CYP4A1      .....
CYP4A3      .....
CYP102      SAKKVRKAE NAH

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Fig 1.

Sequence alignment of CYP52A21 with CYP4A1, CYP4A3, and CYP101 (P450_{BM3}). The amino acid sequences were aligned using the software T-Coffee from the Swiss Institute of Bioinformatics (<http://www.ch.emblnet.org/software/TCoffee.html>). The residues corresponding to the glutamate involved in heme-covalent binding in the CYP4 family are shown in bold.

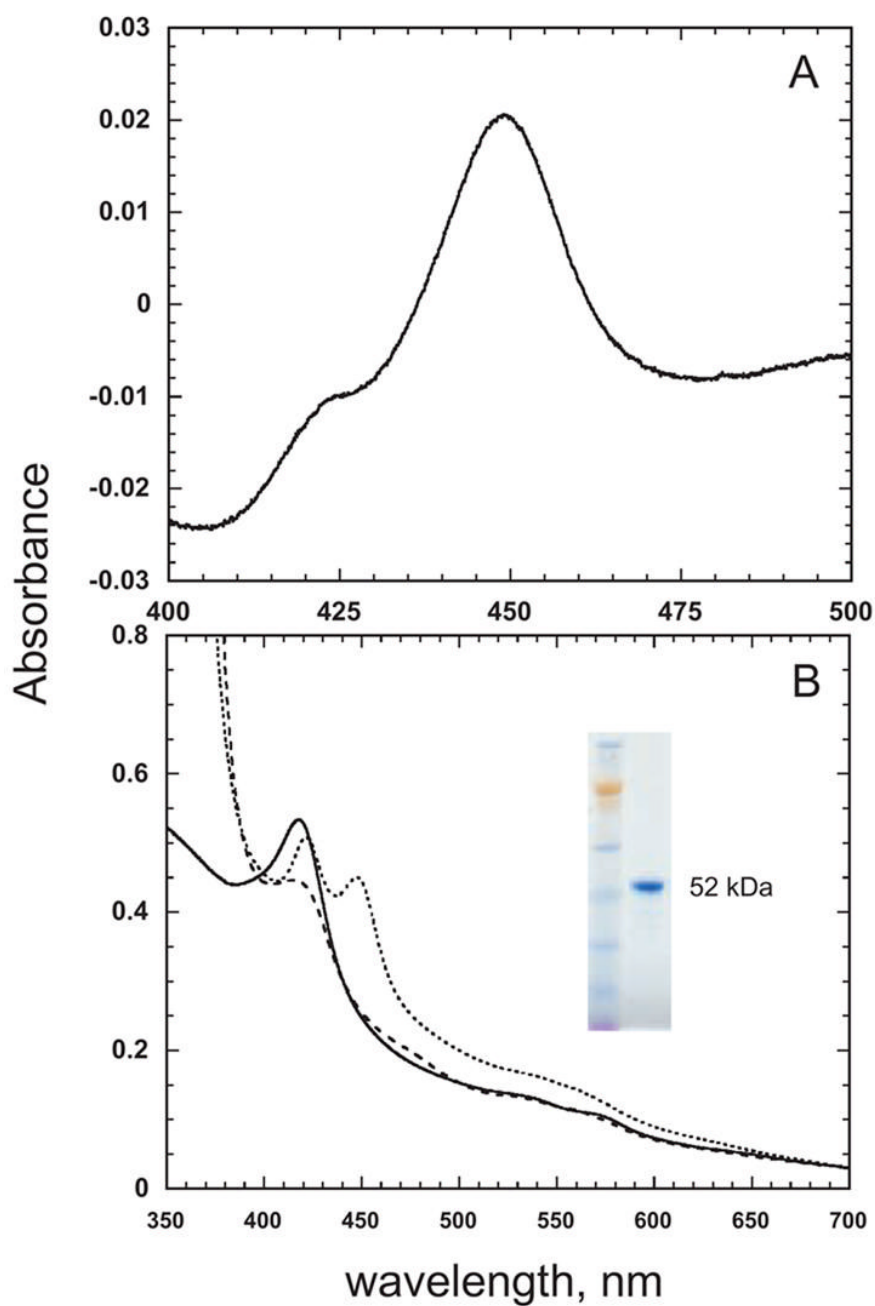


Fig 2. Spectra of purified CYP52A21. A, The $\text{Fe}^{2+} \cdot \text{CO}$ vs Fe^{2+} difference spectrum. B, The absolute spectra of the ferric (—), ferrous (---), and ferrous carbon monoxide protein (----). The *inset* shows an SDS-PAGE gel of the purified CYP52A21 protein.

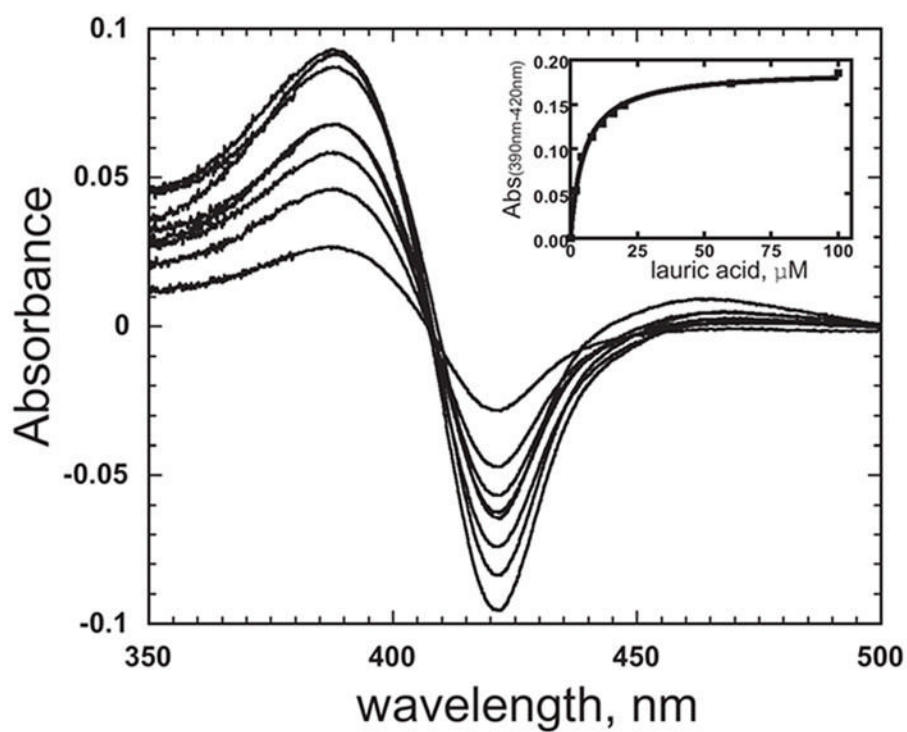


Fig 3. Titration of CYP52A21 with dodecanoic acid. Increasing concentrations of dodecanoic acid were added to both the sample and reference cuvettes. The inset shows the plot of $\Delta A_{390-420nm}$ vs concentration of dodecanoic acid.

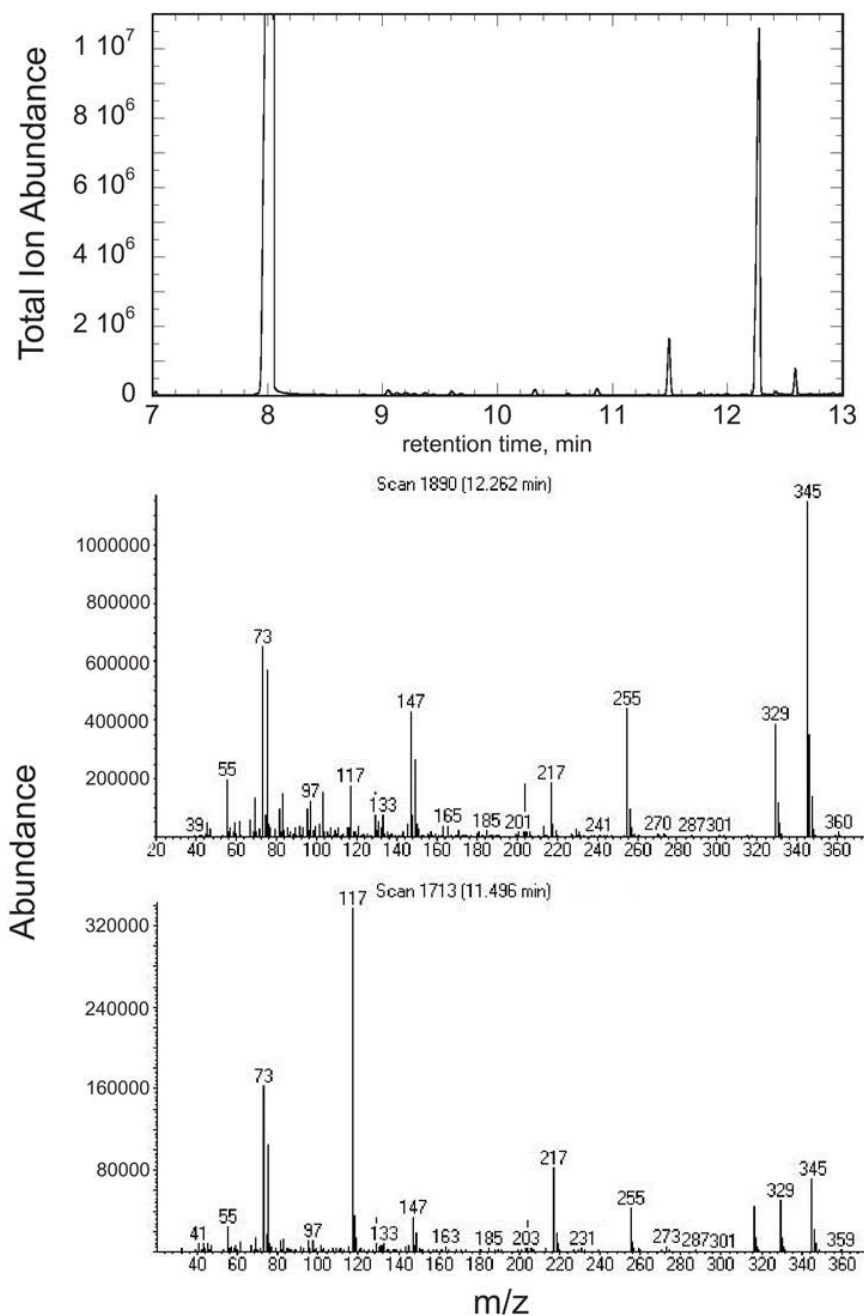


Fig 4. GC-MS of products formed in the reaction of CYP52A21 with dodecanoic acid. A. GC-MS total ion trace of the reaction products. B. Mass-spectrum of the product at 12.26 min, C. Mass-spectrum of the product at 11.49 min.

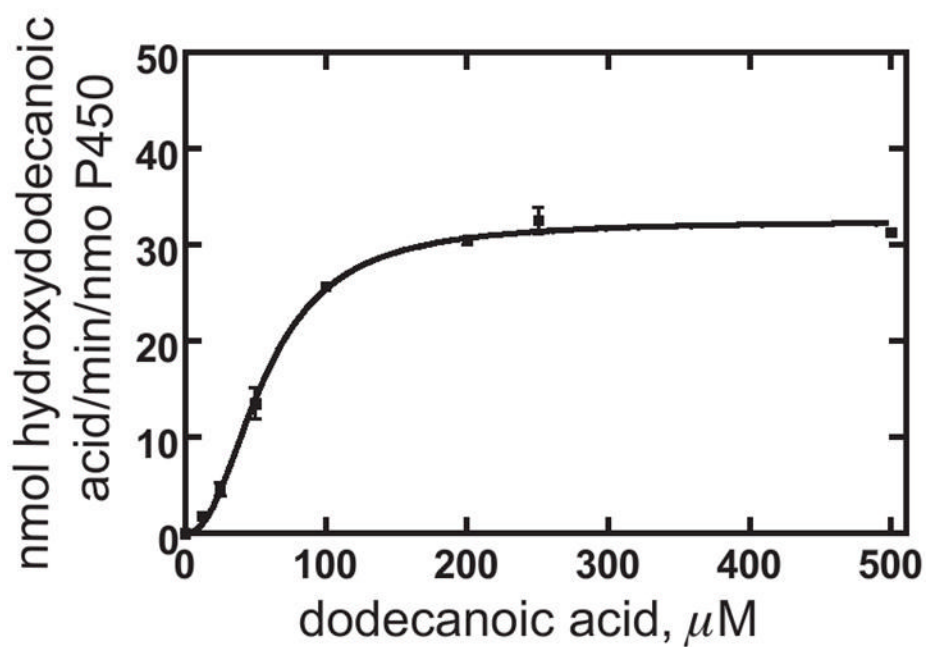


Fig 5. Steady-state kinetics of dodecanoic acid hydroxylation by CYP52A21. Each point is a mean \pm SD of triplicate assays. The estimated parameters were $k_{\text{cat}} 33 \pm 1$ nmol of total hydroxylated dodecanoic acid formed/min/nmol of P450, $K_{\text{m}} 57 \pm 2 \mu\text{M}$, and $k_{\text{cat}}/K_{\text{m}} 0.58 \pm 0.03$.

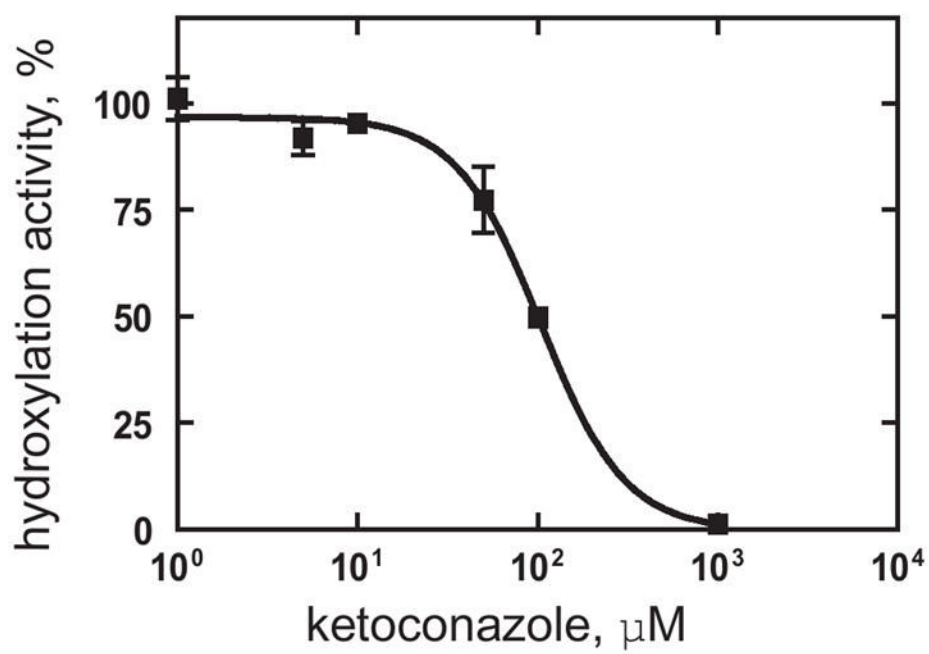


Fig 6.
Inhibition of CYP52A21 dodecanoic acid hydroxylation by ketoconazole.

Table 1
Isotope-dependent Shift of Dodecanoic Acid Hydroxylation Catalyzed by CYP52A21

Substrates	12-hydroxydodecanoic acid	11-hydroxydodecanoic acid conversion, %	Total	$\omega/(\omega-1)$
d ₀ -dodecanoic acid	49.0 ± 7.6	4.4 ± 0.9	53.4 ± 8.5	11.1 ± 2.9
d ₃ -dodecanoic acid	19.5 ± 1.5	22.3 ± 1.9	41.8 ± 3.5	0.87 ± 0.10

Table 2
Estimated Substrate Binding Affinities to CYP52A21

Substrates	K_d (μ M)
	CYP52A21
dodecanoic acid	5.1 ± 0.4
12-chlorododecanoic acid	2.7 ± 0.2
12-bromododecanoic acid	4.1 ± 0.2
12-iodododecanoic acid	2.5 ± 0.2

Table 3

Conversion of 12-Halododecanoic Acids by CYP52A21

Substrates	12-hydroxydodecanoic acid	12-oxododecanoic acid	11-hydroxydodecanoic acid	Total conversion	$\omega/(\omega-1)$
			%		
12-chlorododecanoic acid	1.5 ± 0.4	4.2 ± 0.2		5.7 ± 0.2	0.36 ± 0.27
12-bromododecanoic acid	1.5 ± 0.1	1.1 ± 0.1		2.6 ± 0.1	1.36 ± 0.11
12-iodododecanoic acid	1.1 ± 0.2	< 0.1		1.1 ± 0.2	

Table 4

Turnover Numbers for Halododecanoic Acid Oxidation by CYP52A21

Substrates	12-hydroxydodecanoic acid	12-oxododecanoic acid	11- hydroxydodecanoic acid	Total
dodecanoic acid	28		2	30
12-chlorododecanoic acid	0.25	0.70		0.95
12-bromododecanoic acid	0.25	0.18		0.43
12-iodo-dodecanoic acid	0.18			0.18

Table 5

Oxidation of Longer Fatty Acids by CYP52A21

Substrates	ω -hydroxylation	(ω -1)-hydroxylation conversion, %	Total	$\omega/(\omega-1)$
lauric acid (C ₁₂)	49.0 ± 7.6	4.4 ± 0.9	53.4 ± 8.5	11.1 ± 2.9
myristic acid (C ₁₄)	42.3 ± 0.4	4.4 ± 0.4	46.7 ± 4.6	9.6 ± 0.9
palmitic acid (C ₁₆)	14.2 ± 1.1	1.0 ± 0.1	15.3 ± 1.1	14.2 ± 1.8