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Vibronic and Rydberg series assignments in the vacuum ultraviolet absorption spectrum of nitrous oxide



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ABSTRACT

We report a comprehensive photoabsorption study of nitrous oxide (N_2O) in the vacuum ultraviolet (45,000–95,000 cm^{-1}) region using synchrotron radiation. The observed spectrum comprises of a few valence transitions and low lying Rydberg series converging to the two spin–orbit components ($^2\Pi_{1/2,3/2}$) of the ground state of N_2O^+ . Spectral analysis is aided by extensive quantum chemical calculations of vertical excited states, oscillator strengths and potential energy curves using the time dependent density functional theory. Vibronic bands observed in the first absorption system (45,000–60,000 cm^{-1}) are assigned to hot band progressions in ν'_2 originating from $\nu''=1$ or 2. New insights into the assignment of the well-formed progression of bands in the $X^1\Sigma^+ \rightarrow C^1\Pi$ system (60,000–72,000 cm^{-1}) are afforded by consideration of the Renner–Teller interaction. A set of molecular vibrational parameters ($\omega_2=467 \text{ cm}^{-1}$, $x_{22}=-2.9$, $\epsilon=-0.24$) for the $C^1\Pi$ state are derived from a fitting of the experimental data. The $3p\pi^1\Sigma^+$ state at $\sim 77,600 \text{ cm}^{-1}$ shows a large quantum defect (0.96) which is explained as arising due to mixed valence–Rydberg character. In the 85,000–95,000 cm^{-1} region, a number of absorption features are observed with greater clarity than in earlier photoabsorption studies and assigned to Rydberg series of type $n\lambda$ ($n=3,4$; $l=s,p,d$; $\lambda=\sigma,\pi,\delta$) and accompanying vibronic bands. This work has resulted in clarification of several discrepancies in earlier Rydberg series assignments. Additionally, the $3p\pi^3\Sigma^-$ Rydberg state at 85,788 cm^{-1} , the valence transition $7\sigma \rightarrow 3\pi$ ($^1\Pi$) at 87,433 cm^{-1} and the $3d\lambda$ Rydberg series in the 91,700–92,600 cm^{-1} region are assigned for the first time.

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1. Introduction

There has been continued interest in the nitrous oxide (N_2O) molecule due to its important role in atmospheric and astrophysical processes [1–5]. For example, the isotopic fractionation of nitrogen atoms in the upper

atmosphere has been ascribed to the UV photodissociation dynamics of N_2O [3 and references therein]. Photodissociation of N_2O is known to produce metastable species such as O ($^1S, ^1D$) which participate in important atmospheric photochemical reactions. Recent studies show that nitrous oxide is the single largest anthropogenic threat to the ozone layer in the 21st century and also has a very high global warming potential [1]. It has been found to be involved in the catalytic destruction of ozone in the stratosphere and to be the main source of odd nitrogen which regulates the ozone layer, and also could be the only

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interstellar compound in which the N atom uses all five valence electrons [4]. The availability of synchrotron radiation and equivalent electron impact sources in recent times has served as an impetus for studies of photo-induced molecular processes, enabling measurements over wide spectral ranges and providing a wealth of data regarding the nature and properties of excited electronic states. Nitrous oxide is a case in point, understanding of its excited state structure and dynamics being greatly enhanced by studies carried out using a variety of experimental techniques [3–25] as well as theoretical calculations [26–35]. Nevertheless, a thorough understanding of the excited states of this molecule still remains a challenging problem.

The electronic absorption spectrum of N_2O spans a wide energy region, from the UV region ($45,000\text{ cm}^{-1}$) and extending deep into the vacuum ultraviolet (VUV) region. In the VUV region, a number of electronic transitions comprising of both valence transitions and Rydberg series converging to the lowest two ionization limits at ~ 12.89 and 12.91 eV ($103,963$ and $104,097\text{ cm}^{-1}$) [17] are observed. The first VUV absorption spectrum of nitrous oxide was reported by Duncan in 1936 [6] and subsequently revisited by Sponer and Teller [7], Zelikoff et al. [8] and Tanaka et al. [9]. Early work on nitrous oxide is well summarized in the review by Rabalais et al. [10]. The only synchrotron radiation based photoabsorption study of this molecule was by Nee et al. [11] who reported absolute cross sections up to the first ionization potential, but no new spectral assignments were given. Additionally, there have also been a number of reports on the electronic spectrum of N_2O , focusing on specific aspects in small energy regions. For instance, photodissociation of N_2O through its first absorption band ($45,000\text{--}60,000\text{ cm}^{-1}$) has received much attention in recent years due to its atmospheric implications [3,12–14,26–31]. In the $60,000\text{--}72,000\text{ cm}^{-1}$ region, after the initial work by Zelikoff et al. [8], there have been very few studies [15,16]. Vibronic assignments of the extensive and well formed progression of bands in this region are still incompletely understood. Towards the higher energy side of this progression ($> 75,000\text{ cm}^{-1}$), the absorption spectrum is dominated by Rydberg transitions. The Rydberg states of nitrous oxide have been studied by several authors using a variety of techniques like photoabsorption spectroscopy [9,11,15,17,18], resonance enhanced multiphoton ionization (REMPI) [19–22] and electron impact/electron energy loss spectroscopy [23–25]. Assignments of Rydberg series and accompanying vibronic structures reported in these papers are at wide variance with each other and a clear set of assignments is still not available. For example, the strong sharp band at $\sim 84,900\text{ cm}^{-1}$ has been assigned to the origin of the $3p\sigma^1\Pi$ transition in some works [17,24], and to the 3_0^1 vibronic band of $3p\sigma^1\Pi$ by others [19,20]. Although some of these discrepancies were resolved in the work by Cossart-Magos et al. [17], still several issues in assignments of transitions in the $85,000\text{--}95,000\text{ cm}^{-1}$ region remain unresolved. From a theoretical perspective, the electronic spectroscopy of nitrous oxide presents many challenges with additional complications introduced due to perturbations like Renner–Teller coupling and spin–orbit splitting. There

have been a wide range of theoretical studies covering various aspects like geometry and stability of its three isomers [26], photodissociation dynamics [27–31], excited electronic states [32–35], etc. A special mention must be made about the work by Hopper [34] which remains one of the most comprehensive theoretical studies of the electronic structure of nitrous oxide till date. Also worth citing are the frozen core calculations by Cossart-Magos et al. [17] which have helped in assigning many Rydberg transitions.

Despite this large volume of experimental and theoretical work, there remain several unresolved issues in the excited state electronic structure of nitrous oxide. The present work attempts to improve the understanding of the excited state structure of N_2O by carrying out a detailed reinvestigation of its VUV absorption spectrum. The photoabsorption spectrum of N_2O is recorded in the wavelength region $45,000\text{--}95,000\text{ cm}^{-1}$ using synchrotron radiation. Ab initio calculations are used to aid the spectral analysis. Details of the experiment, computations and results obtained are discussed in the ensuing sections.

2. Methodology

2.1. Experimental

VUV photoabsorption studies were carried out using the Photophysics beamline coupled to the 450 MeV storage ring Indus-1 at Raja Ramanna Centre for Advanced Technology, Indore, India. Details of the experimental setup have been described in earlier papers [36,37]. Briefly, synchrotron radiation is dispersed by a 1 m Seya-Namioka monochromator with 2400 l/mm grating (resolving power ~ 1000) and made to pass through a 25 cm long stainless steel gas cell. The experimental resolution is $\sim 6\text{ meV}$ ($\sim 48\text{ cm}^{-1}$) at 6 eV and $\sim 12\text{ meV}$ ($\sim 97\text{ cm}^{-1}$) at 12 eV. The cell is fitted with lithium fluoride windows, which limit the transmission to $> 1050\text{ Å}$. Nitrous oxide of stated purity 99.9% procured from M/s. Alchemie Gases is used without further purification and introduced into the cell through a system of Swagelok valves. The transmitted intensity is detected using a sodium salicylate coated window and UV–visible photomultiplier tube. The measured intensity without sample serves as the reference or I_0 and the intensity with sample introduced serves as I . Using the Beer–Lambert law, a plot of $\ln(I_0/I)$ versus wavelength, scanned in steps of 0.5 Å generates the absorption spectrum. Spectra are recorded at several pressures in the range $0.001\text{--}10\text{ mbar}$, measured using capacitance gauges. Due to the widely varying absorption cross sections in different wavelength regions, optimum pressures used also differ from region to region. The optimum pressures for obtaining clear absorption features in the sub-regions $50,000\text{--}62,000\text{ cm}^{-1}$, $62,000\text{--}72,000\text{ cm}^{-1}$ and $72,000\text{--}95,000\text{ cm}^{-1}$ are found to be $\sim 10\text{ mbar}$, 0.1 mbar and 0.001 mbar respectively. We however do not observe any noticeable change in the spectra when the pressure is varied in the neighborhood of these optimum pressures. The synchrotron beam current is recorded simultaneously at every step in order to normalize the data with respect to the decay in beam

current. Standard atomic lines of Xenon are used for calibration of the spectra.

2.2. Computational

N₂O is a 16 valence electron system with a linear C_{∞v} geometry, the ground state molecular orbital configuration being given by $1\sigma^2 2\sigma^2 3\sigma^2 4\sigma^2 5\sigma^2 6\sigma^2 1\pi^4 7\sigma^2 2\pi^4: ^1\Sigma^+$. There have been several theoretical studies [26–35] on the electronic states of N₂O, the most exhaustive one being the work by Hopper in 1984 [34] reporting extensive MCSCF/CI calculations of several vertically excited singlet and triplet states as well as some potential energy surfaces. To the best of our knowledge there is no reported calculation of excited states up to the first ionization limit using the time dependent density functional theory (TD-DFT). Theoretical chemistry studies show that for excited state calculations, the TD-DFT method provides a good balance between numerical precision and computational cost [38]. In the present study we use the DFT and MP2 methods for geometry optimization and computation of vibrational wavenumbers of the neutral and ionic ground states, while TD-DFT calculations are used for prediction of vertical excited states. Basis sets used in the calculations include Pople's split valence basis set 6-31G(3d) with three d-type polarization functions [39], correlation consistent polarized basis sets of Dunning cc-pVnZ [40,41] and the polarization consistent basis sets of Jensen, pc-4 [42,43]. These families of basis sets are suitable for recovering the correlation energy, whereas the pc-4 basis is known to be particularly suited to density functional (DFT) calculations [43]. DFT computations are carried out using two hybrid exchange correlation GGA functionals viz. Perdew–Burke–Ernzerhof (PBE0) [44] and Becke–3–Lee–Yang–Parr (B3LYP) [45,46]. In particular, the PBE0 functional is known to perform well for excited state calculations [38]. All calculations are done using the GAMESS (USA) suite of programs using parallel mode of computation when necessary [47]. Molecular orbital contour plots generated using the software MacMolPlt [48] for the highest occupied molecular orbitals (HOMO) and lowest unoccupied molecular orbitals (LUMO) of N₂O have been given as [Supplementary material](#). The HOMO (2 π) has an N–N

bonding and N–O antibonding character while the LUMO (3 π) is antibonding with respect to both N–N and N–O.

3. Results and discussion

3.1. Computational results

3.1.1. Geometry optimization and computation of vibrational frequencies

Geometry optimization of the ground state of N₂O is carried out using DFT as well as MP2 methods with several basis sets. The theoretically predicted vibrational frequencies using the DFT functional PBE0 and various basis sets are listed in [Table 1](#) without any scaling factors. Difference between the calculated and observed [49] frequencies may be attributed to neglect of anharmonicity in the calculation. The geometry optimization of the ground state of N₂O⁺ is also tabulated in [Table 1](#). Calculated ionic frequencies often aid in the analysis of vibronic structure of Rydberg transitions as vibrational frequencies in excited Rydberg states are expected to be closer to vibrational frequencies of the corresponding ionic ground state rather than the ground state of the neutral molecule [50,51]. This approach which has been used successfully in some of our earlier works [36,37] to assign vibronic spectra is also employed in the present study. It is observed experimentally that in the ground state of the N₂O⁺ ion, the molecule remains linear with an increase in R_{NN} bond length by 0.027 Å while the R_{NO} bond length remains almost the same as in the neutral molecule. This is also borne out by the N–N bonding and N–O antibonding character of the HOMO, from which the electron is removed.

3.1.2. Vertical excited state energies

Vertical excited states are calculated at the ground state optimized geometry using the TDDFT method. Calculations are performed for several basis sets and DFT functionals. The best results in terms of overall agreement with current experimental data and earlier theoretical work [34] are obtained for the PBE0 functional using the pc-4 basis set. Predicted singlet and triplet excited states up to ~14 eV (96,800 cm^{−1}) along with calculated oscillator strengths (O.S.) and lambda diagnostic values λ ($0 \leq \lambda \leq 1$) are presented in [Table 2](#). The lambda diagnostic is a measure of

Table 1
Ground state geometry optimization and vibrational frequencies using DFT/PBE0.

Basis set	Energy (a.u.)	R _{NN} (Å)	R _{NO} (Å)	ν_1 (cm ^{−1})	ν_2 (cm ^{−1})	ν_3 (cm ^{−1})
Optimized parameters (N ₂ O: X ¹ Σ ⁺)						
6-311G(d)+	−184.5202	1.122	1.176	1376	593	2409
cc-pV5Z	−184.5555	1.117	1.175	1376	637	2404
pc-4	−184.5566	1.117	1.174	1376	637	2405
Experimental values ^a		1.127	1.185	1284.9	588.8	2223.8
Optimized parameters (N ₂ O ⁺ : ² Π)						
6-311G(d)+	−184.0583	1.135	1.178	1247	456	1890
cc-pV5Z	−184.0840	1.132	1.178	1246	498	1876
pc-4	−184.0850	1.132	1.177	1247	499	1876
Experimental values ^b		1.154	1.185	1127	457	1738

^a [49].

^b [53].

Table 2Singlet and triplet vertical excited states of N₂O using TDDFT; PBE0/pc-4.

State	Predicted energy		Major transitions	O.S.	λ	Assignment/nature	Observed energy (eV)
	(cm ⁻¹)	(eV)					
³ Σ^+	42,296	5.244	2 $\pi \rightarrow 3\pi$	0.000	0.762	a ³ Σ^+ (V)	–
³ Δ	47,571	5.898	2 $\pi \rightarrow 3\pi$	0.000	0.481/0.768*	b ³ Δ (V)	–
³ Σ^-	55,305	6.857	2 $\pi \rightarrow 3\pi$	0.000	0.487	c ³ Σ^- (V)	–
¹ Σ^-	55,305	6.857	2 $\pi \rightarrow 3\pi$	0.000	0.487	A ¹ Σ^- (V)	5.6–7.7
¹ Δ	58,411	7.242	2 $\pi \rightarrow 3\pi$	0.000	0.488/0.779*	B ¹ Δ (V)	–
³ Π	67,557	8.376	2 $\pi \rightarrow 8\sigma$	0.000	0.390	3s $\sigma^3\Pi$ (R)	7.7–8.9
¹ Π	70,630	8.757	2 $\pi \rightarrow 8\sigma$	0.000	0.389	3s σ C ¹ Π (R)	–
³ Π	76,405	9.473	7 $\sigma \rightarrow 3\pi$	0.000	0.616	7 $\sigma \rightarrow \pi^*$ (V)	–
¹ Σ^+	80,930	10.034	2 $\pi \rightarrow 3\pi$	0.423	0.539	3p π D ¹ Σ^+ (R+V)	9.1–10.1
			2 $\pi \rightarrow 4$				
³ Σ^+	83,745	10.383	2 $\pi \rightarrow 4\pi$	0.000	0.391	(R)	–
³ Π	84,559	10.484	2 $\pi \rightarrow 9\sigma$	0.000	0.206	3p σ ³ Π (R)	10.47
³ Δ	84,898	10.526	2 $\pi \rightarrow 4\pi$	0.000	0.245/0.390*	(R)	–
¹ Π	85,301	10.576	2 $\pi \rightarrow 9\sigma$	0.078	0.205	3p σ ¹ Π (R)	10.53
¹ Δ	85,963	10.658	2 $\pi \rightarrow 4\pi$	0.000	0.241/0.384*	(R)	–
¹ Σ^-	86,245	10.693	2 $\pi \rightarrow 4\pi$	0.000	0.243	(R)	–
³ Σ^-	86,245	10.693	2 $\pi \rightarrow 4\pi$	0.000	0.243	3p π ³ Σ^- (R)	10.48
³ Σ^+	89,116	11.049	1 $\pi \rightarrow 3\pi$	0.000	0.788	(V)	–
¹ Π	92,568	11.477	7 $\sigma \rightarrow 3\pi$; 2 $\pi \rightarrow 10\sigma$	0.022	0.522	7 $\sigma \rightarrow \pi^*$ (V+R)	10.84
³ Π	93,173	11.552	2 $\pi \rightarrow 10\sigma$	0.000	0.234	4s σ ³ Π (R)	10.94
¹ Π	94,036	11.659	2 $\pi \rightarrow 10\sigma$; 7 $\sigma \rightarrow 3\pi$	0.050	0.320	4s σ ¹ Π (R)	11.05
³ Δ	94,730	11.745	1 $\pi \rightarrow 3\pi$	0.000	0.471/0.751*	(V+R)	–
³ Σ^+	97,617	12.103	2 $\pi \rightarrow 5\pi$	0.000	0.328	(R)	–
³ Π	98,553	12.219	2 $\pi \rightarrow 10\sigma$; 2 $\pi \rightarrow 11\sigma$	0.000	0.423	4p σ ³ Π (R)	11.74
¹ Σ^+	98,763	12.245	7 $\sigma \rightarrow 8\sigma$; 2 $\pi \rightarrow 4\pi$	0.267	0.426	4p π ¹ Σ^+ (R)	11.64
³ Δ	98,964	12.270	2 $\pi \rightarrow 5\pi$	0.000	0.237/0.378*	(R)	–
¹ Δ	99,045	12.280	2 $\pi \rightarrow 5\pi$	0.000	0.179/0.287*	(R)	–
³ Σ^-	99,400	12.324	2 $\pi \rightarrow 5\pi$	0.000	0.185	4p π ³ Σ^- (R)	11.74
¹ Σ^-	99,400	12.324	2 $\pi \rightarrow 5\pi$	0.000	0.188	(R)	–
³ Σ^+	100,303	12.436	7 $\sigma \rightarrow 8\sigma$	0.000	0.268	(R)	–
¹ Σ^+	101,505	12.584	2 $\pi \rightarrow 5\pi$	0.104	0.303	(R)	–
¹ Σ^+	102,634	12.725	7 $\sigma \rightarrow 8\sigma$	0.312	0.291	(R)	–
³ Σ^-	103,981	12.892	1 $\pi \rightarrow 3\pi$	0.000	0.519	(V)	–
¹ Σ^-	103,981	12.892	1 $\pi \rightarrow 3\pi$	0.000	0.519	(V)	–
¹ Π	104,578	12.962	1 $\pi \rightarrow 12\sigma$; 2 $\pi \rightarrow 12\sigma$	0.001	0.423	(R)	–
¹ Δ	107,247	13.297	1 $\pi \rightarrow 3\pi$	0.000	0.527/0.841*	(V)	–
³ Π	109,982	13.636	7 $\sigma \rightarrow 3\pi$	0.000	0.578	(V)	–
³ Σ^+	111,603	13.837	7 $\sigma \rightarrow 9\sigma$	0.000	0.268	(R)	–
¹ Π	112,434	13.940	2 $\pi \rightarrow 11\sigma$; 2 $\pi \rightarrow 12\sigma$	0.046	0.403	4p σ ¹ Π (R)	11.76

 λ : Lambda diagnostic, (R): Rydberg, (V): valence; and O.S.: oscillator strength.* Two values of λ are for the two components of the degenerate Δ states.

the spatial overlap in a given excitation [52]. A relatively large value of λ implies short range excitations, or a large overlap between the occupied and virtual orbitals, whereas a relatively small value of λ implies long range excitations. In the present study the lambda diagnostic value is used as an estimate of the valence or Rydberg character of the transition. In general, large values of λ (> 0.6) imply valence character, small values (< 0.3) imply Rydberg character and intermediate values (0.3–0.6) may be due to a mixed character (cf. Table 2). However, these values are used in conjunction with the nature of the final orbital involved in the transition. In the present work, we find that although most Rydberg transitions are predicted with lambda value of < 0.3 , there are a few transitions

which are clearly Rydberg type from the nature of the final orbital, but predicted with $\lambda=0.3$ –0.4. In the case of N₂O it is found that when the final orbital is 3 π , the transition is typically valence in nature, whereas for 4 π 8 σ , 9 σ , 10 σ ... the excited state is Rydberg in nature (cf. Table 2). Valence–Rydberg mixed transitions are those for which there are two excitations predicted with comparable amplitude, typically one of valence type and the other of Rydberg type.

3.1.3. Potential energy curves for excited singlet states

Some insight into the geometries and dynamics of the excited states can be gained by studying their potential

energy curves (PECs). PECs for excited singlet states are obtained with respect to change in bond angle, change in N–N bond length and change in N–O bond length, while keeping other geometrical parameters fixed to the ground state equilibrium values. Fig. 1 (a, b and c) shows the potential energy curves up to the first few singlet excited states plotted with respect to N–N bond length r_1 , N–O bond length r_2 and N–N–O bond angle θ respectively. It may be noted that the present study is reported in terms of θ , whereas some of the earlier studies [27,30,32] used the Jacobi angle γ . The two angles are related by a simple relation of the form: $\theta = \gamma + \sin^{-1}(r_1 \sin \gamma / 2r_2)$. To the best of our knowledge, this is the first report of PECs of N_2O generated using the TDDFT method although PECs of N_2O have been reported using other methods [30–32]. The PECs for the first few excited states agree well with earlier reports [30–32], while extending the work to higher excited states.

3.2. Experimental observations and correlation with computational results

The experimentally observed VUV absorption spectrum of N_2O recorded using the Photophysics beamline at Indus-1 is shown in Fig. 2 and may be distinguished into four major regions. Analysis and interpretation of the absorption bands are discussed in the following sections, aided by theoretical calculations wherever relevant. The correspondence of theoretically predicted excitation energies and experimentally observed bands is based on initial and final state molecular orbitals involved as well as comparison of the O–C values. We may mention here that the accuracy of the calculated values is expected to be very good in the energy region up to HOMO+1 eV [38], which in the present case corresponds to ~ 10.85 eV. Above this energy, the agreement with experimental values becomes somewhat poorer.

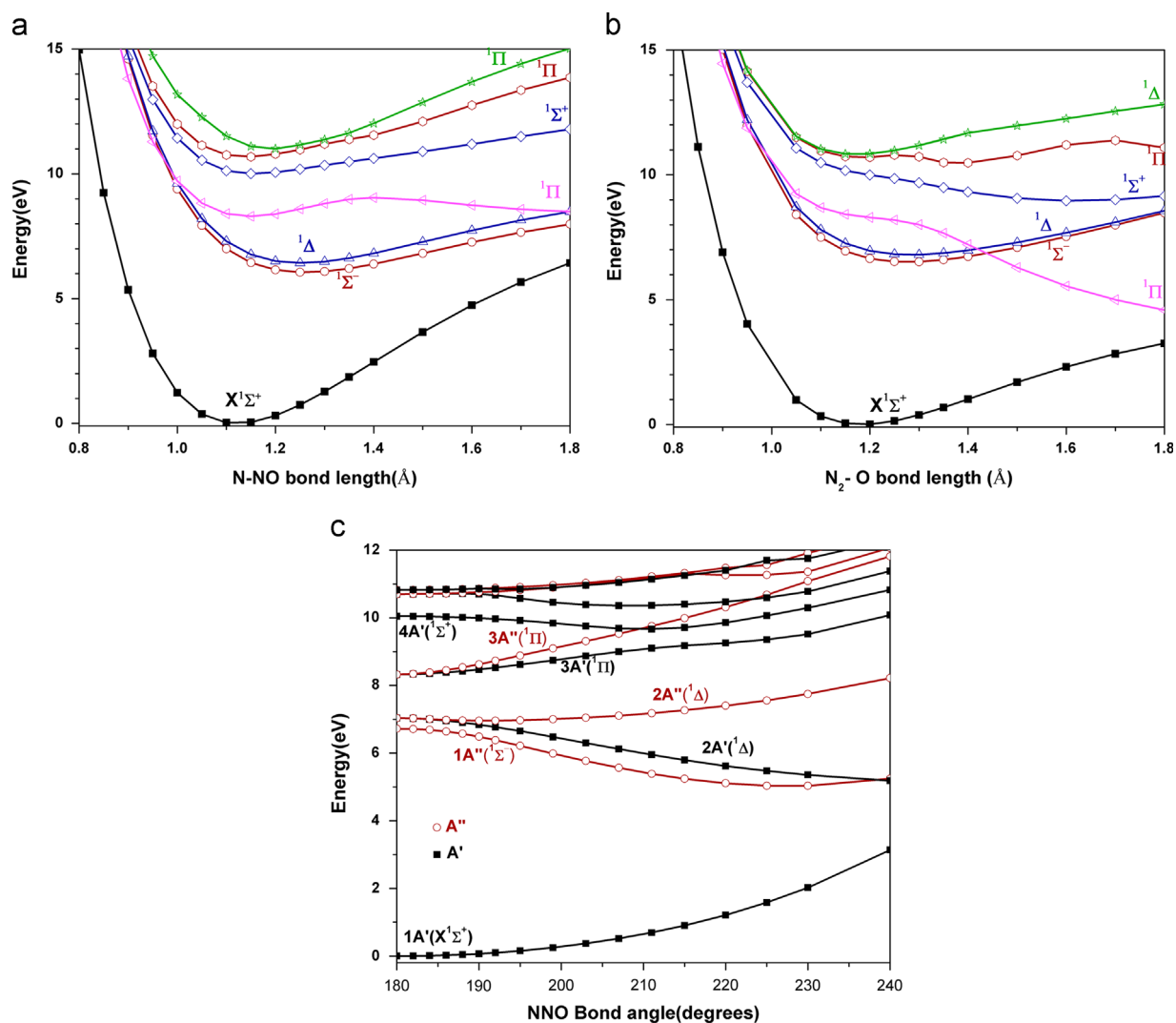


Fig. 1. Potential energy curves of the first few excited states of N_2O . (a, b and c) Plotted with respect to N–NO bond length, N_2 –O bond length and NNO bond angle respectively.

3.2.1. The energy region 5.6–7.7 eV (45,000–60,000 cm⁻¹)

At relatively low pressures (< 1 mbar), the absorption system observed in this region appears as a continuum, but as the pressure is increased (> 1 mbar), vibronic bands appear, superimposed on the continuum (cf. Fig. 3). There have been several recent experimental and theoretical studies of this region covering various aspects such as temperature and isotopic substitution effects, product angular distributions, potential energy surfaces, etc. [14,27–29,31]. In an earlier work by us, a partial listing of prominently observed bands had been reported [15]. The first two singlet excited states of N₂O are A'¹Σ⁺ and B'¹Δ (cf. Table 2), transitions to both of which are forbidden under C_{∞v} symmetry. In the Franck-Condon region, the molecule is non-linear in the excited state and exhibits C_s symmetry. The observed absorption spectrum is ascribed to transitions from the linear ground state X ¹Σ⁺ to the ¹A' (¹Σ⁺)

state and the ¹A' and ¹A'' Renner–Teller components of the ¹Δ state, both of which are allowed under C_s symmetry [3,14,27]. The underlying continuum is due to the dissociative natures of the ¹Σ⁺ state and A' component of the ¹Δ state, as evident from the potential energy curve with respect to bond angle (cf. Fig. 1c).

It has been established [27] that the vibronic structures in this region arise predominantly from A' (¹Δ) state, with a smaller contribution from the A'' (¹Δ) state. In this context, it is instructive to examine the variation of oscillator strength with the N–N–O bond angle as plotted in Fig. 4a. One sees that there is an appreciable gain in oscillator strength for the A' (Δ) state, whereas there is a very small increase in oscillator strength for the A'' (Σ⁺) and A'' (Δ) states (cf. inset in Fig. 4a). This could be the reason why the continuum arising from the repulsive A' (Δ) state is much stronger than the overriding vibrational structure. The observed vibronic features correspond to excitation of the ν₂ bending mode in the excited electronic state, although the diffuseness of the bands has been attributed to large amplitude NN bond stretching [27]. It has also been discussed earlier [3,14,27] that the observed structure is dominated by hot band transitions from the (0,1,0) and (0,2,0) levels with the ratios of oscillator strengths for vibronic transitions going as σ(ν₂'=1)/σ(ν₂'=0) = 2 and σ(ν₂'=2)/σ(ν₂'=1) = 2.9 [27]. In the present work, two major series are identified (cf. Table 3). On the basis of the observed separation of ~588 cm⁻¹ (ground state value of ν₂) between the two progressions and earlier theoretical simulations [27], we infer that these are hot band progressions from (0,1,0) and (0,2,0) (cf. Fig. 3), rather than the fundamental. A few weaker and diffuse features on the higher energy side could be members of the fundamental progression from (0,0,0). The extreme diffuseness of the bands in this region prevents a clear assignment of the series; hence these bands are excluded from Table 3. A measured average separation of ~478 cm⁻¹ between successive members of

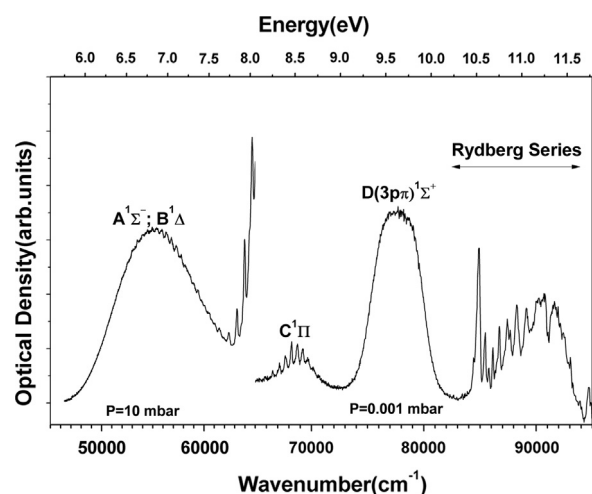


Fig. 2. The VUV photoabsorption spectrum of N₂O in the 45,000–95,000 cm⁻¹ region.

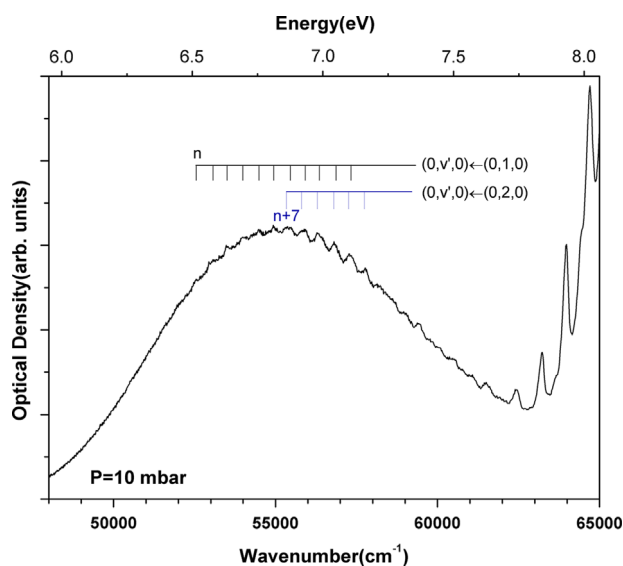


Fig. 3. Part of the VUV absorption spectrum of N₂O showing vibronic assignments in the 45,000–65,000 cm⁻¹ region. *n* Refers to the quanta of bending vibration in the excited state ν₂'.

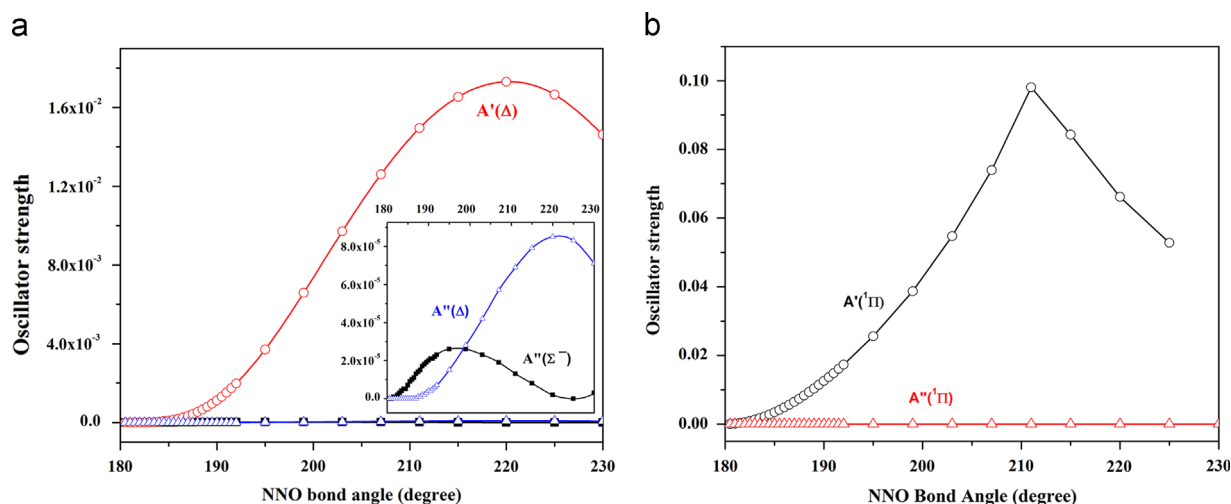


Fig. 4. Variation of oscillator strength with bending angle for low lying excited states of N_2O .

Table 3

Vibronic bands observed (cm^{-1}) in the photoabsorption spectrum of N_2O in the 52,000–60,000 cm^{-1} region.

ν'	$(\nu''=1)$	$(\nu''=2)$
n	52,545	
$n+1$	53,060	
$n+2$	53,504	
$n+3$	53,978	
$n+4$	54,477	
$n+5$	54,939	
$n+6$	55,453	
$n+7$	55,909	55,322
$n+8$	56,340	55,810
$n+9$	56,850	56,288
$n+10$	57,330	56,797
$n+11$	57,779	57,264
$n+12$	–	57,752
$n+13$	–	58,184
ω	479	477

a progression is consistent with the interpretation of these structures arising predominantly from excitation of the ν'_2 mode.

3.2.2. The energy region 7.7–8.85 eV ($\sim 60,000$ – $72,000 \text{ cm}^{-1}$)

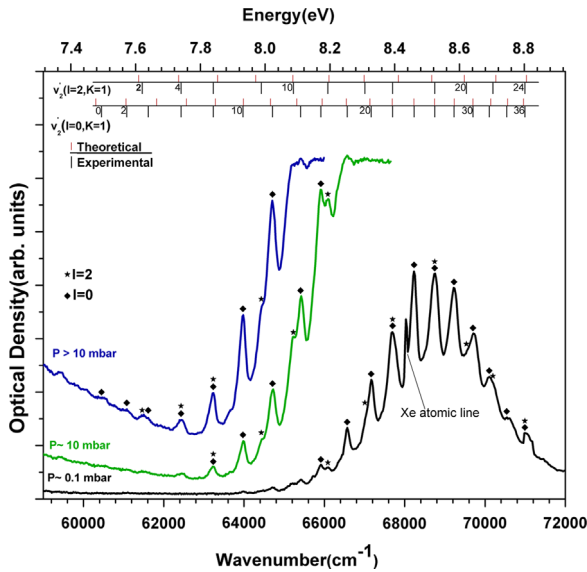
The electronic transition observed in this region is accompanied by a well formed progression of bands and corresponds to the classically designated transition $X^1\Sigma^+ - C^1\Pi$. This transition is correlated with the theoretically predicted excited state at 8.759 eV ($70,646 \text{ cm}^{-1}$) which arises from the orbital excitation $2\pi \rightarrow 8\sigma$ (cf. Table 2). In concurrence with the interpretation by Hopper [34], this excited state may be assigned to the first member of the ns Rydberg series $3s\sigma^1\Pi$ converging to the (adiabatic) IP $104,097 \text{ cm}^{-1}$ ($^2\Pi_{1/2}$), with a quantum defect of ~ 1.37 .

Tentative assignments of the observed vibronic bands have been proposed in earlier literature [6–8,16]. The earliest work by Duncan [6] reported a few bands of average spacing 621 cm^{-1} , proposing that the value of 621 cm^{-1} may either represent a reduction of the

stretching frequency ν_1 or an increase in bending frequency ν_2 in the excited state. As early as 1941, the possibility of two progressions arising from a degenerate upper state split due to interaction between electronic and vibrational motions was also suggested by Sponer and Teller [7]. Later, Zelikoff et al. [8] reported 21 bands which they fitted to single progression assuming a fundamental stretching frequency of 1005 cm^{-1} and an unusually high value of anharmonicity in the upper state. A recent paper by Zhen et al. [16] reports a jet-cooled study of six bands, restricted to the region above $67,000 \text{ cm}^{-1}$ assigning the observed bands to the ν'_2 bending mode. The bands observed in the present work are listed in Table 4 and shown in Fig. 5. The separation between successive bands is observed to be irregular and decreases rather rapidly, from a value of about 1000 cm^{-1} to $\sim 500 \text{ cm}^{-1}$ (cf. Table 4). This unusual structure cannot be satisfactorily explained in terms of any of the known vibrational frequencies of nitrous oxide either as a single ν_2 progression, for which the initial value of 1000 cm^{-1} seems too high, or as a single ν_1 progression, for which one has to assume a very high anharmonicity. While the possibility of multiple progressions cannot be completely ruled out, we propose that most of the observed bands can be fitted to a ν_2 progression, provided one assumes a Renner–Teller interaction in the excited $^1\Pi$ (degenerate) electronic state. Although the predicted oscillator strength for a vertical transition to the $C^1\Pi$ state at equilibrium geometry is zero (cf. Table 2), as the molecule executes a bending vibration the $^1\Pi$ state splits into $^1A'$ and $^1A''$ states, of which the A' component is seen to gain considerable oscillator strength as the molecule deviates from linearity (cf. Fig. 4b). Therefore, it is very likely that the electronic transition is accompanied primarily by excitation of the bending mode (which is also responsible for its observed intensity), with a possibility of smaller contribution from stretching modes. The Renner–Teller interaction couples the vibrational angular momentum l of the bending mode with the electronic orbital angular momentum Λ [51] as described below.

Table 4Assignment of vibronic bands observed (cm^{-1}) in the $X^1\Sigma^+-C^1\Pi$ system of N_2O including the Renner–Teller interaction in the excited state.

v_2'	$l=0, K=1$			$l=2, K=1$		
	Calculated	Observed	O–C	Calculated	Observed	O–C
0	60,308	60,450	142			
2	61,057	61,078	21	61,374	61,462	88
4	61,822	61,612	–210	62,371	62,429 ^a	58
6	62,566	62,429 ^a	–137	63,342	63,232 ^a	–110
8	63,287	63,232 ^a	–55	64,290	64,430	140
10	63,986	63,972	–14	65,214	65,227	13
12	64,662	64,707	45	66,114	66,087	–27
14	65,314	65,406	92	66,991	67,007	16
16	65,943	65,913	–30	67,845	67,698 ^a	–147
18	66,549	66,565	16	68,676	68,746	70
20	67,132	67,163	31	69,483	69,535	52
22	67,692	67,698 ^a	6	70,268	70,195	–73
24	68,229	68,236	7	71,029	70,990 ^a	–39
26	68,742	68,746	4	Molecular parameters of $C^1\Pi$ state		
28	69,233	69,228	–5	Parameter	Value	Error
30	69,700	69,701	1	ω_0^0	$59,851 \text{ cm}^{-1}$	± 25
32	70,144	70,106	–38	ω_2	467 cm^{-1}	± 1.5
34	70,564	70,540	–24	x_{22}	–2.9	± 0.08
36	70,962	70,990 ^a	28	ε	–0.24	± 0.008

^a Blended lines, O–C – difference in observed and calculated energy.**Fig. 5.** VUV absorption spectrum of N_2O showing Renner–Teller coupled vibronic band assignments in the $60,000\text{--}72,000 \text{ cm}^{-1}$ region.

The general expression for vibronic term values of a linear triatomic molecule is given by

$$G(v_1, v_2, v_3) = \sum_i \omega_i \left(v_i + \frac{d_i}{2} \right) + \sum_i \sum_{k>i} x_{ik} \left(v_i + \frac{d_i}{2} \right) \left(v_k + \frac{d_k}{2} \right) + \sum_i \sum_{k>i} g_{ik} l_i l_k \quad (1)$$

where $G(v_1, v_2, v_3)$ is the vibronic energy in cm^{-1} , ω_i are the vibrational frequencies, x_{ik} and g_{ik} are the anharmonicity constants, l_i is the vibrational angular momentum and d_i is the degeneracy of the i th mode. For the bending

vibration ω_2 , the vibrational energy level expression (1) reduces to

$$G(v_2, l) = \{\omega_2(v_2 + 1) + g_{22}l^2\} + x_{22}(v_2 + 1)^2 = G^\pm(v_2, K) + x_{22}(v_2 + 1)^2 \quad (2)$$

Due to the Renner–Teller interaction, the resultant angular momentum is given by $K = l + \Lambda$ and the interaction terms $G^\pm(v_2, K)$ may be written as

$$K = 0; \quad G^\pm(v_2, 0) = \omega_2 \sqrt{1 \pm \varepsilon(v_2 + 1)}, \quad v_2 = 1, 3, 5, \dots \quad (3)$$

$$K \neq 0, \quad v_2 = K - 1; \quad G(v_2, K) = \omega_2 \left[(v_2 + 1) - \frac{1}{8} \varepsilon^2 K(K + 1) \right] \quad (4)$$

$$K \neq 0, \quad v_2 > K - 1;$$

$$G^\pm(v_2, K) = \omega_2 \left(1 - \frac{1}{8} \varepsilon^2 \right) (v_2 + 1) \pm \frac{1}{2} \omega_2 \varepsilon \sqrt{(v_2 + 1)^2 - K^2} \quad (5)$$

where ε is the Renner–Teller interaction parameter. Selection rules for the vibronic transitions are given in terms of the standard group theoretical notation for dipole allowed transitions [51]; viz. $\Gamma(\psi_{e'v'}) \otimes \Gamma(\psi_{e''v''}) \supseteq \Gamma(M)$. Further, for the case of vibronic coupling, we consider only the dipole moment components characterizing the allowed electronic transition $\Sigma^+ \rightarrow \Pi$ [51]. Thus, only those vibronic transitions are allowed for which the direct product of the symmetry species of the initial and final vibronic wavefunctions transforms as (x, y) . Since the initial electronic state is of Σ^+ symmetry, only vibronic levels of Π symmetry are allowed. Further, since levels of Π symmetry arise only from the Renner–Teller interaction between Π electronic state and vibrational levels of even quantum number v_2' , the selection rules effectively allow only transitions with even vibrational quantum number v_2'

Table 5Transitions observed in the VUV photoabsorption spectrum of N₂O in the 75,000–95,000 cm^{−1} region.

Peak ^a	Assignment	Q.D	Peak ^b	Assignment	Peak ^c	Assignment	Peak ^d	Assignment	Peak ^e	Assignment	Peak ^f	Assignment
77,600	3pπ ¹ Σ ⁺	0.96	–	3pπ ¹ Σ ⁺	77,520	D ¹ Σ ⁺						
							82,725	3pσ ³ Π	83,026		82,743	3 ₁ ¹ 3pσ ¹ Π
											83,042	2 ₁ ¹ 3pσ ¹ Π
					83,264	3pσ ¹ Π	83,178	3pσ ¹ Π	83,188	0 ₀ ⁰ 3pπ ¹ Δ	83,200	0 ₀ ⁰ 3pσ ¹ Π
									83,356	2 ₁ ¹ Δ(N20 ⁺ ?)		
83,433	2 ₂ ⁰ 3pσ ³ Π								83,621	2 ₁ ² Σ ⁺	83,628	2 ₀ ¹ 3pσ ¹ Π
83,915	2 ₁ ⁰ 3pσ ³ Π								83,911	2 ₀ ¹ Δ		
									84,057	2 ₀ ² Φ	84,061	2 ₀ ² 3pσ ¹ Π
									84,330	1 ₀ ¹ 3pσ ¹ Π	84,339	1 ₀ ¹ 3pσ ¹ Π
84,439	0 ₀ ⁰ 3pσ ³ Π	0.63	84,379	3pσ ³ Π								
			84,631	3pπ ³ Σ [−]					84,780	3 ₀ ¹ 2 ₁ ¹		
84,910	0 ₀ ⁰ 3pσ ¹ Π	0.61	84,918	0 ₀ ⁰ 3pσ ¹ Π	84,818	3 ₀ ¹ 3pσ ¹ Π			84,937	3 ₀ ¹ 3pσ ¹ Π	84,927	3 ₀ ¹ 3pσ ¹ Π
85,464	2 ₀ ¹ 3pσ ¹ Π		85,482	2 ₀ ¹ 3pσ ¹ Π	85,397	1 ₀ ¹ 3pσ ¹ Π			85,441		85,465	1 ₀ ¹ 3pσ ¹ Π
85,788	3pπ ³ Σ [−]	0.55										
86,152	3 ₀ ¹ 3pσ ³ Π		86,122	3 ₀ ¹ 3pσ ³ Π							86,012	0 ₀ ⁰ 3pπ ¹ Σ ⁺ ; ¹ Δ
86,362(w)												
86,512(sh)									86,478			
86,714	3 ₀ ¹ 3pσ ¹ Π		86,670	3 ₀ ¹ 3pσ ¹ Π	86,655	3 ₀ ² 3pσ ¹ Π			86,643		86,700	1 ₀ ³ 3pσ ¹ Π
86,979(w)											86,900	3 ₀ ² 3pσ ¹ Π
87,198(sh)											87,122	1 ₀ ¹ 3pπ ¹ Σ ⁺ ; ¹ Δ
87,433	7σ → 3π ¹ Π				87,336	1 ₀ ² 3pσ ¹ Π					87,438	0 ₀ ⁰ 3dσ ¹ Π
87,685	2 ₁ ⁰ 4sσ ³ Π										87,712	3 ₀ ¹ 3pπ ¹ Σ ⁺ ; ¹ Δ
88,004(sh)											88,027	0 ₀ ⁰ 3dπ ¹ Σ ⁺ ; ¹ Δ
88,265	4sσ ³ Π	1.36	88,262	4sσ ³ Π	88,183	3 ₀ ³ 3pσ ¹ Π						
88,436(sh)	2 ₁ ⁰ 4sσ ¹ Π											
88,778(w)												
89,130	4sσ ¹ Π	1.29	89,150	4sσ ¹ Π								
89,243(w)												
					88,968	3dδ ¹ Π ¹ Φ					88,972	0 ₀ ⁰ 3dδ ¹ Π ¹ Φ
89,512(w)												
89,750	2 ₀ ¹ 4sσ ¹ Π											
90,036	3 ₀ ¹ 4sσ ³ Π											
90,230	1 ₀ ¹ 4sσ ¹ Π											
90,510												
90,737	3 ₀ ¹ 4sσ ¹ Π				90,662							
91,050	1 ₀ ¹ 3 ₀ ¹ 4sσ ³ Π											
91,382	1 ₀ ² 4sσ ¹ Π							91,283	4sσ ³ Π			
91,493	2 ₀ ¹ 3 ₀ ¹ 4sσ ¹ Π							91,443	4sσ ¹ Π			
91,623	3 ₀ ² 4sσ ³ Π				91,575							
91,784	3d δ ³ Π	−0.002										
91,952	3d δ ¹ Π	−0.006										
92,142	3dσ ³ Π	−0.05										
92,395	3dσ ¹ Π	−0.06										
	3dπ ¹ Σ ⁺	−0.08										
92,563(sh)	3dπ ³ Σ [−]	−0.08										
92,861	2 ₀ ¹ 3dπ ³ Σ [−]											
93,030	2 ₀ ¹ 3dπ ¹ Σ ⁺											
93,236	1 ₀ ¹ 3dσ ³ Π											
93,880	4pπ ¹ Σ ⁺	0.70						93,904	4pσ ³ Π			
94,040 (w)			94,085	4pπ ¹ Σ ⁺	94,158	4pσ ¹ Π		94,075	4pσ ¹ Π			
									4pπ ³ Σ ⁺			
94,696	4pσ ³ Π	0.56	94,696	4pσ ³ Π								
	4pπ ³ Σ [−]	0.58										
			94,755	4pπ ³ Σ [−]				94,705	4pπ ³ Σ ⁺			
94,872	4pσ ¹ Π	0.55	94,896	2 ₁ ¹ 4pσ ¹ Π				94,903	4pπ ¹ Σ ⁺			
			94,910	4pσ ¹ Π								

Q.D.: quantum defect; ¹Σ⁺ and ³Π converge to ²Π_{3/2} at 103,963 cm^{−1}; ³Σ[−] and ¹Π converge to ²Π_{1/2} at 104,097 cm^{−1}.^a Present study.^b [17].^c [11].^d [21].^e [20].^f [19].

whereas $^3\Pi_0$ and $^1\Pi_1$ converge to the ionic state $^2\Pi_{1/2}$ [54,17]. In the case of Σ states arising from a $\pi \rightarrow \pi^*$ excitation, the spin orbit operator couples states with opposite behavior with respect to the σ_v operator, i.e. Σ^+ is coupled to Σ^- [54]. Thus, transitions from the $^1\Sigma^+$ ground state become allowed to excited states of $^1\Sigma^+$ and $^3\Sigma^-$ symmetry through the spin–orbit mixing mechanism and $^1\Sigma^+$ series converge to $^2\Pi_{3/2}$, whereas the $^3\Sigma^-$ series converge to $^2\Pi_{1/2}$ [53,17]. It is worth mentioning here that, for most of the Rydberg transitions observed, triplet states appear with intensity comparable to that of singlets. This is an indication of strong spin–orbit coupling (Hund's case c) which one would normally expect only for high n Rydberg states. In the present study which is restricted to the region below $95,000\text{ cm}^{-1}$ we have observed $n=3$ and 4 members of these series. Assignments have been largely made based on the calculated quantum defect values reported by Cossart-Magos et al. [17]. It may be noted that the adiabatic ionization energies $103,963\text{ cm}^{-1}$ and $104,097\text{ cm}^{-1}$ from pulsed field ionization studies [55] are used throughout in the Rydberg series analysis.

3.2.4.1. $ns\sigma$ Series. The first member of this series, viz. $3s\sigma$ $^1\Sigma^+$ lies in the $7.7\text{--}8.85\text{ eV}$ ($62,000$ to $70,000\text{ cm}^{-1}$) region and has been discussed in Section 3.2.2. The next higher member of this series, viz. $4s\sigma$ $^1\Sigma^+$, is observed at $88,265$ and $89,130\text{ cm}^{-1}$ (cf. Fig. 6) with quantum defects of 1.36 and 1.29 respectively. These states arise from the $2\pi \rightarrow 10\sigma$ orbital excitation and are theoretically predicted at $11,552$ and $11,659\text{ eV}$ ($93,173$ and $94,036\text{ cm}^{-1}$) (cf. Table 2). The peak at $87,685\text{ cm}^{-1}$ may be assigned to the hot band transition $2_0^0 4s\sigma^3\Pi$ due to population of the ν_2 bending mode at room temperature. The corresponding hot band of the $4s\sigma$ $^1\Pi$ transition ($88,436\text{ cm}^{-1}$) appears as a shoulder to the intense $4s\sigma^3\Pi$ peak.

3.2.4.2. $np\sigma$ and $np\pi$ Series. The $3p\sigma$ configuration gives rise to the $3p\sigma$ $^3\Pi$ states which are observed at $84,439$ and $84,910\text{ cm}^{-1}$ correspondingly with a quantum defect of ~ 0.6 . The intensity of the triplet transition is considerably less than that of the corresponding singlet transition, thus indicating a weak spin–orbit coupling. These states can be identified with the theoretically predicted vertical excited states at 10.484 and 10.576 eV ($84,559$ and $85,301\text{ cm}^{-1}$) corresponding to the orbital excitation $2\pi \rightarrow 9\sigma$. The assignment of the origin of the $3p\sigma$ $^1\Pi$ transition in literature has been controversial. In some of the earlier REMPI works [19,20], the $84,910\text{ cm}^{-1}$ band has been assigned to $3_0^1 3p\sigma$ $^1\Pi$ with the origin of the Rydberg transition being located at $83,200\text{ cm}^{-1}$, whereas other works [17,24] have assigned the $84,910\text{ cm}^{-1}$ peak to the origin of $3p\sigma$ $^1\Pi$. From the current photoabsorption study, we observe that the peak at $84,910\text{ cm}^{-1}$ is the most intense in the entire $83,000\text{--}95,000\text{ cm}^{-1}$ region, whereas no line is observed at $83,200\text{ cm}^{-1}$ at any of the pressures used, reaffirming the assignment of the origin to $84,910\text{ cm}^{-1}$. The large value of predicted oscillator strength (cf. Table 2) for the $3p\sigma$ $^1\Pi$ transition supports the validity of this assignment. Two weak but distinct bands at $83,433\text{ cm}^{-1}$ and $83,915\text{ cm}^{-1}$ are observed to

the lower wavenumber side of the strong $84,910\text{ cm}^{-1}$ band, at relatively higher pressures. These bands are assigned to the hot band transitions $2_0^0 3p\sigma^3\Pi$ and $2_2^0 3p\sigma^3\Pi$. Vibronic bands corresponding to excitation of the ν_2 and ν_3 modes in the excited $3p\sigma$ $^3\Pi$ states are observed at $85,464$, $86,152$ and $86,714\text{ cm}^{-1}$ as listed in Table 5. The next pair of this series, $4p\sigma$ $^3\Pi$, appear at $94,696\text{ cm}^{-1}$ and $94,872\text{ cm}^{-1}$ with a quantum defect of ~ 0.55 . Theoretically, the triplet component is predicted at 12.219 eV ($98,553\text{ cm}^{-1}$), while the singlet component is predicted at 13.94 eV ($112,434\text{ cm}^{-1}$), based on the orbital excitations involved. This discrepancy in experimental and theoretical values may be due the spin–orbit coupling which is not accounted for in the theoretical model used.

The $3p\pi$ $^1\Sigma^+$ state has been assigned to the strong and broad band observed at $\sim 77,600\text{ cm}^{-1}$ as discussed in Section 3.2.3. The corresponding $3p\pi$ $^3\Sigma^-$ state is assigned to a weak feature at $84,631\text{ cm}^{-1}$ by Cossart-Magos et al. [17]. In the present study no peak is observed at $84,631\text{ cm}^{-1}$. The $3p\pi$ $^3\Sigma^-$ state is theoretically predicted at 10.693 eV ($86,245\text{ cm}^{-1}$), to the higher energy side of the $3p\sigma$ $^1\Pi$ state by about 940 cm^{-1} (cf. Table 2). We therefore assign the peak observed at $85,788\text{ cm}^{-1}$ (cf. Fig. 6) to $3p\pi$ $^3\Sigma^-$. It may be noted that this peak was not observed in the work by Cossart-Magos et al. [17], while other more intense spectral features in this region are reported as weak features by them. The obtained quantum defect of 0.55 confirms the assignment of this peak to a p state. The next pair of this series, viz. $4p\pi$ $^1\Sigma^+$ and $4p\pi$ $^3\Sigma^-$, are observed at $93,880$ and $94,696\text{ cm}^{-1}$ with quantum defects of 0.70 and 0.58 respectively. Correspondence of $4p\pi$ $^3\Sigma^-$ to the theoretically predicted transition at 12.324 eV ($99,400\text{ cm}^{-1}$) is straightforward. On the other hand, it is seen that $4p\pi$ $^1\Sigma^+$ can be correlated with the theoretically predicted energies at 12.245 eV ($98,763\text{ cm}^{-1}$) or 12.584 eV ($101,505\text{ cm}^{-1}$) based on orbital excitations. (cf. Table 2). As one goes to higher Rydberg states, complications due to multiple orbital excitations and spin–orbit interaction render unambiguous correspondence of experimental and theoretical values difficult.

3.2.4.3. $nd\sigma$, $nd\pi$ and $nd\delta$ Series. There is considerable discrepancy in assignment of the 3d states in earlier works. Although Cossart-Magos et al. have predicted the $3d\lambda$ transitions to lie in the region $\sim 90,000\text{--}92,500\text{ cm}^{-1}$, they have not been able to experimentally identify any clear peaks [17]. In the earlier REMPI-PES study by Scheper et al. [21] observed peaks at $91,283$ and $91,443\text{ cm}^{-1}$ have been assigned to $4s\sigma$ $^3\Pi$. However, the theoretical calculations of Cossart-Magos et al. [17] suggest that this region corresponds to excitation to the 3d Rydberg series. Moreover, as explained by them, the PES experiments may not correctly predict the symmetry of the excited Rydberg state. Patsilina et al. [19] have assigned $3d\lambda$ ($\lambda=\sigma,\pi,\delta$) origins to peaks at $87,438$, $88,027$ and $88,972\text{ cm}^{-1}$. This implies a quantum defect of $\sim 0.29\text{--}0.42$, which seems too high for a d state. In the present studies, several clear and distinct peaks are observed in this region as seen in Fig. 6. Based on the calculated quantum defect values and the observed positions of the higher members ($n=4, 5$) [17], we reassign the observed peaks in this region to singlet and triplet components of the $3d\lambda$ states. The five peaks

from 91,784 cm^{-1} to 92,563 cm^{-1} are assigned to $3d\lambda$ states as given in Table 5. The quantum defects obtained are in the range -0.002 to -0.08 , which is in agreement with the theoretical values of Cossart-Magos et al. [17], thus confirming the validity of the present assignments. Most of the remaining bands in this region which could be assigned to vibronic features accompanying the Rydberg series $4s\sigma$ and $3d\pi$ (cf. Table 5).

3.2.4.4. Valence excitations. The distinct feature observed at 87,433 cm^{-1} has been reported in a few earlier works [11,18,19]; however assignments made by these authors are not in agreement with each other. Assuming the frozen core calculations of Cossart-Magos et al. [17] to be the best theoretical estimate of quantum defects available, these features cannot be satisfactorily assigned to any Rydberg transition. We therefore propose that this peak is a valence transition $X^1\Sigma^+ \rightarrow ^1\Pi$ arising from the orbital excitation $7\sigma \rightarrow 3\pi^*$. This transition is theoretically predicted at 11.477 eV (92,568 cm^{-1}) with a lambda diagnostic value of 0.522, confirming its valence nature. The triplet component arising from this orbital excitation is predicted at 9.473 eV (76,405 cm^{-1}), which overlaps with the strong, broad peak at $\sim 77,700 \text{ cm}^{-1}$. It may be noted that our assignments are consistent with the prediction of valence transitions at 78,900 ($^3\Pi$) and 91,300 ($^1\Pi$) cm^{-1} by Cossart-Magos et al. [17] although they have not reported any experimental peaks corresponding to these transitions.

4. Conclusions

In this paper we have presented a photoabsorption study of nitrous oxide in the VUV region (45,000–95,000 cm^{-1}) using synchrotron radiation. Quantum chemical calculations of excited state energies, oscillator strengths and potential energy curves are performed in order to gain a better understanding of the excited state structure. The observed spectrum consists of a few valence transitions and several Rydberg series ($n=3, 4$) converging to the two spin-orbit components ($^2\Pi_{3/2,1/2}$) of the ground state of N_2O^+ . The first absorption system (45,000–60,000 cm^{-1}) is ascribed to the forbidden states $^1\Sigma^-$ and $^1\Delta$, made allowed by change in geometry of the molecule to a bent configuration in the excited states. The underlying strong continuum observed is attributed to the large gain in oscillator strength of the transition to the repulsive A' (Δ) state as the molecule bends. A pressure dependent study of this region enabled identifying and assigning several hot band transitions (originating from $\nu_2''=1, 2$) involving the ν_2' bending mode whose frequency is measured to be $\sim 478 \text{ cm}^{-1}$ in the excited state. The interpretation of the $C^1\Pi$ electronic state (60,000–72,000 cm^{-1}) as the $3s$ Rydberg transition is supported by theoretical predictions. The irregular spacing of vibronic features of the $X^1\Sigma^+ \rightarrow C^1\Pi$ absorption system is explained for the first time as arising due to a Renner–Teller interaction. Molecular parameters ($\omega_2=467 \text{ cm}^{-1}$, $x_{22}=-2.9$, $\epsilon=-0.24$) obtained by fitting the observed bands to the Renner–Teller expressions seem to be reasonable considering the value of ω_2 in the ionic state and

other excited states of N_2O are close to 467 cm^{-1} and the Renner–Teller parameter in the ionic ground state is ~ -0.2 .

To the higher energy region, the broad and intense $3p\pi \ ^1\Sigma^+$ transition at $\sim 77,600 \text{ cm}^{-1}$ is proposed to have a mixed valence–Rydberg character, explaining its unusually high quantum defect value of 0.96. In the 84,000–95,000 cm^{-1} region dominated by Rydberg series, several features are observed and assigned for the first time. In this region, the comprehensive experimental data supported by extensive theoretical calculations have resulted in unambiguous assignments and clarification of discrepancies in earlier work. The $3p\pi \ ^3\Sigma^-$ Rydberg state at 85,788 cm^{-1} , the valence transition $7\sigma \rightarrow 3\pi$ ($^1\Pi$) at 87,433 cm^{-1} and the $3d\lambda$ Rydberg series in the 91,700–92,600 cm^{-1} region as well as several accompanying vibronic transitions are assigned for the first time. The intensity of singlets and triplets being comparable for most of the observed Rydberg states leads to the conclusion that the spin–orbit coupling is quite large in Rydberg states of N_2O , even for low n (3,4). A few transitions in the 90,000–95,000 cm^{-1} region remain unassigned. A clearer picture may emerge by carrying out synchrotron radiation based photoabsorption studies at higher resolution and using better theoretical models. This paper provides a consolidated view and at the same time presents fresh insights into the VUV photoabsorption spectrum of nitrous oxide, which could form useful inputs for future work related to the excited state dynamics of this atmospherically important molecule.

Appendix A. Supporting information

Supporting data associated with this article can be found in the online version at <http://dx.doi.org/10.1016/j.jqsrt.2014.05.017>.

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