# Mechanistic studies of the SufS-SufE cysteine desulfurase: evidence for sulfur transfer from SufS to SufE

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Abstract SufS is a cysteine desulfurase of the *suf* operon shown to be involved in iron–sulfur cluster biosynthesis under iron limitation and oxidative stress conditions. The enzyme catalyzes the conversion of L-cysteine to L-alanine and sulfide through the intermediate formation of a protein-bound cysteine persulfide in the active site. SufE, another component of the *suf* operon, has been previously shown to bind tightly to SufS and to drastically stimulate its cysteine desulfurase activity. Working with *Escherichia coli* proteins, we here demonstrate that a conserved cysteine residue in SufE at position 51 is essential for the SufS/SufE cysteine desulfurase activity. Mass spectrometry has been used to demonstrate (i) the ability of SufE to bind sulfur atoms on its cysteine 51 and (ii) the direct transfer of the sulfur atom from the cysteine persulfide of SufS to SufE. A reaction mechanism is proposed for this novel two-component cysteine desulfurase

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# 1. Introduction

Cysteine desulfurases are important enzymes of sulfur metabolism present in all living organisms. They are involved in the biosynthesis of a variety of sulfur-containing compounds such as thiamine, thionucleosides in tRNAs, molybdopterin and iron–sulfur clusters [1]. In *Escherichia coli* there are three cysteine desulfurases named IscS, SufS and CsdA [1,2]. They all are pyridoxal-phosphate-dependent enzymes which catalyze the decomposition of L-cysteine into L-alanine and sulfane sulfur. In all of them the active site contains a conserved catalytically essential cysteine residue, to which the sulfur atom of the cysteine substrate is transferred to generate alanine and a protein-bound persulfide intermediate [3,4]. This persulfide could then directly transfer its sulfur atom to target molecules or be reduced to release sulfide in solution.

SufS, a homodimeric protein, was first proposed to specif-

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Abbreviations: DTT, dithiothreitol; PCR, polymerase chain reaction

ically serve for mobilization of selenium from selenocysteine. Indeed, it had been shown to display a relatively much weaker cysteine desulfurase activity, absolutely dependent on the presence of a cysteine at position 364 (in *E. coli*) in the active site [2,5]. We also observed a very poor cysteine desulfurase activity using the highly homologous SufS protein from *Erwinia chrysanthemi* [6].

However, we also discovered that SufS from *E. chrysanthemi* tightly interacts with another homodimeric protein named SufE and that, in complex with SufE, it displays a drastically enhanced cysteine desulfurase activity. The activity of the complex is abolished upon substitution of the conserved cysteine residue of SufS (Cys369) to serine [6]. *sufS* and *sufE* belong to the same *suf* operon which has been shown to play an important role in iron–sulfur cluster synthesis/repair and to provide a protective mechanism under stressful conditions such as oxidative stress and iron limitation [7–10].

In order to better understand the mechanism of this new type of cysteine desulfurase enzyme, consisting of a complex of two homodimeric proteins SufS and SufE, and the molecular basis for the stimulatory effects of SufE on SufS activity, we have prepared histidine-tagged SufS and SufE from E. coli in pure form. We made the working hypothesis that these effects were dependent on sulfur atom transfer from the active site persulfide of SufS to a conserved cysteine of SufE. Examination of SufE amino acid sequences for conserved cysteine residues that might be involved in this function revealed a highly conserved cysteine at position 51 (Fig. 1). We thus also investigated the properties of a SufE mutant in which this cysteine was substituted by serine. Using mass spectrometry we have indeed demonstrated (i) the ability of SufE to bind sulfur atoms on its cysteine 51 residue, (ii) the direct transfer of sulfane sulfur from SufS to cysteine 51 of SufE. A mechanism for cysteine desulfuration by the SufS/SufE system is proposed.

#### 2. Materials and methods

2.1. Materials and plasmids

All chemicals were of reagent grade and obtained from Sigma-Aldrich Chemical Co. or Fluka unless otherwise stated. Cysteine was from Boehringer Mannheim.

*E. coli* strains were grown aerobically at 37°C in Luria–Bertani (LB) rich medium [11]. When necessary, antibiotics were added at the following concentration: 50 μg/ml ampicillin.

Plasmids pET-Shis and pET-Ehis, encoding the His-tagged SufS and SufE proteins, respectively, were derived from pET22b+ (Novagen). sufS insert, obtained by polymerase chain reaction (PCR) amplification using oligonucleotides Ndel-5s/XhoI-5as (see below) and Ndel-XhoI digestion, was introduced into pET22b+ digested by the same enzymes, yielding pET-Shis. The same method was used for pET-Ehis with oligonucleotides NdeI-6s/XhoI-6as and NdeI-XhoI digestion. The oligonucleotides are: NdeI-5s, 5'-CCGCATATGATTTTTTCCGTCGACAAAGTG-3'; XhoI-5as, 5'-GTTCTCGAGTCCC-AGCAAACGGTGAATACG-3'; NdeI-6s, 5'-CCGCATA-TGGCTT-TATTGCCGG-3'; XhoI-6as, 5'-GTTCTCGAGGCTAAGTGCAGCGGCTTTGGC-3'; XhoI-6as, 5'-GTTCTCGAGGCTAAGTGCAGCGGCTTTGGC-3'.

Plasmid pET-E<sub>C51S</sub>his was obtained as follows. Mutagenesis of sufE was performed in two steps. First, two complementary mutagenic oligonucleotides, 3'-sufE C51S and 5'-sufE C51S were each used in combination with an oligonucleotide complementary to the 5' end (NdeI-6s) and 3' end (XhoI-6as) of sufE, respectively. The mutagenic oligonucleotides were designed such as cysteine 51 is changed to a serine residue. Oligonucleotides 3'-sufE C51S and NdeI-6s in one hand and oligonucleotides 5'-sufE C51S and XhoI-6as in another hand yielded the DNA fragments sufE1, sufE2, respectively. The matrix used was pET-Ehis. In the second step sufE1 and sufE2 were hybridized and used as templates for PCR amplification with the pair of oligonucleotides NdeI-6s/XhoI-6as. The resulting NdeI-XhoI fragment was cloned into pET22b+ yielding the plasmid pET-E<sub>C51S</sub>his. The oligonucleotides are: 3'-sufE C51S, 5'-CCACACCT-GACTCTGGCTGCCCTGAATGC-3'; 5'-sufE C51S, 5'-AGCATT-CAGGGCAGCCAGAGTVAGGTGTGG-3'.

#### 2.2. Purification of SufS, SufE and SufE<sub>C51S</sub>

E. coli, BL21(DE3) cells were transformed with plasmids pET-Shis or pET-Ehis and expression was induced by adding 0.5 mM isopropyl β-D-thiogalactoside (IPTG) at  $A_{600} = 0.5$ . After 4 h at 30°C (SufS) or 3 h at 37°C (SufE), the pellet obtained from a 500 ml (SufS) or 600 ml (SufE) culture was immediately resuspended in buffer A (100 mM Tris-HCl, 50 mM NaCl, pH 8). Cell disruption was obtained after two passages through a French press. After centrifugation at 12 000 rpm for 30 min at 4°C, soluble proteins were loaded onto a 5 ml nickel Hi-trap column (Amersham Pharmacia Biotech). Elution was achieved with a 4-500 mM imidazole linear gradient in buffer A. Eluted fractions were analyzed by sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE) and Coomassie blue staining. The purified fractions were pooled and imidazole was removed using a BIOMAX-5K device (Millipore) equilibrated with buffer A. In the case of SufE, the resulting protein solution was loaded onto a Superdex-75 (Pharmacia Amersham Biotech) gel filtration column and elution was carried out at 4°C with buffer A, at 30 ml/h. Fractions containing pure protein were pooled and concentrated onto a BIO-MAX-5K. Samples were then aliquoted and stored at -80°C. Fast protein liquid chromatography (FPLC) gel filtration with an analytical Superdex-75 at a flow rate of 0.5 ml/min equilibrated with buffer A was used for size determination. A gel filtration calibration kit (molecular weight marker kit, Sigma) was used as molecular weight standards. The same purification procedure was used for SufE and  $SufE_{C51S}$ .

#### 2.3. Cysteine and selenocysteine desulfurase activity assays

The activity was assayed from the amount of alanine formed from L-cysteine or L-selenocysteine. Selenocysteine solution was obtained after reduction of a 60 mM selenocystine-HCl solution with 100 mM dithiothreitol (DTT). The standard reaction mixture in a final volume of 100 µl buffer B (0.1 M Tris-HCl pH 8, 30 mM KCl) contained 0.5 nmol SufS, 1 molar equivalent of SufE and 4 mM cysteine. The reaction was carried out at 37°C during 10 min and stopped by the addition of 10% (v/v) 1 M TCA. After centrifugation, the supernatant was analyzed for alanine production. For that procedure, the supernatant was dried on a speed-vac and the residue dissolved into 150 µl of a citrate buffer, pH 2.2. A ninhydrin derivative was generated and analyzed at 570 nm by high performance liquid chromatography (HPLC) on a 7300 Beckman apparatus working with an ion exchange column S101036 calibrated with pure amino acid standards according to a published procedure [12]. The cysteine desulfurase activity is in units (µmol alanine/min)/mg of protein. Protein concentration (by monomer) was determined by the method of Bradford standardized with bovine serum albumin [13].

#### 2.4. Sulfur transfer assays

All the experiments were done under anaerobic conditions inside a glove box (<2 ppm  $O_2$ , 18°C). Both SufS and SufE were pretreated with 10 mM DTT and then repurified by gel filtration prior to use. Sulfur transfer reactions were carried out at 37°C for 20 min with 100  $\mu$ M SufS, 100  $\mu$ M SufE (wild-type or mutant protein) and 4 mM cysteine in a final volume of 50  $\mu$ l buffer C (50 mM Tris–HCl, pH 7.5). Reactions were initiated by the addition of cysteine and were stopped by centrifugation at  $2000\times g$  through a size exclusion column (Micro Bio-spin 6 (Bio-Rad)). The spin column eluate was then analyzed by (ESI)-quadrupole mass spectrometry.

For the one turnover experiments, SufS (500  $\mu M$ ) was incubated with 4 mM cysteine for 30 min at 37°C in order to load the protein with sulfur and to generate the persulfide intermediate. Excess of cysteine was removed by desalting the protein over a Micro Biospin 6 column. The resulting protein (100  $\mu M$ ) was incubated with SufE (100  $\mu M$ , wild-type or mutant protein) for different reaction times at 37°C in a final volume of 50  $\mu l$  buffer C. The reaction was stopped by freezing in liquid nitrogen and proteins were analyzed by mass spectrometry.

#### 2.5. Mass spectrometry

Mass spectra of proteins were obtained by ESI-mass spectrometry on a Q-TOF Micro mass spectrometer (Micromass, Manchester, UK) equipped with a Z-spray ion source, operating with a needle voltage of 3 kV. Sample cone and extraction voltages were 70 and 3.5 V, respectively. Samples were infused continuously at a 5 µl/min flow rate with a concentration between 400 and 900 nM in water/acetonitrile (1/1, v/v) with 0.2% formic acid. The mass spectra were recorded in the 700–1600 range of mass-to-charge ratio (m/z) with a 1 s scan time. A 1 µM solution of Glu-fibrinopeptide B was used to calibrate the instrument in the MS/MS mode. Spectra were acquired and processed with MassLinx 4.0 (Micromass).

#### 3. Results

# 3.1. Purification and characterization of SufS, SufE and SufE<sub>C51S</sub> from E. coli

The three proteins SufS, SufE and SufE<sub>C51S</sub> were obtained in the C-terminal His-tagged forms easily purified using Ni-NTA columns. About 13.5 mg of SufS, in a more than 95% pure as judged by Coomassie blue stained SDS–PAGE were obtained from a 500 ml culture. The yields were significantly inferior for proteins SufE and SufE<sub>C51S</sub> (1–1.2 mg from 600 ml culture). Further purification was achieved using an additional Superdex-75 gel filtration chromatography step. The latter experiment demonstrated that both wild-type and mutant SufE behaved each as a dimer of roughly 35 kDa in solution. Details of the purification procedures are described in Section 2.

SufS displays a large selenocysteine deselenase activity, leading to an efficient mobilization of selenium from selenocysteine (3.5 units/mg) (data not shown). In contrast, as shown in Table 1, it displays a very weak cysteine desulfurase activity which was greatly increased upon addition of SufE (about 30-fold). No stimulation of the selenocysteine deselenase activity could be observed (data not shown). Comparable effects were observed with the SufS and SufE proteins from *E. chrysanthemi* [6]. As in the case of *E. chrysanthemi* [6], SufE from *E. coli* has by itself no cysteine desulfurase activity (data not shown).

Finally, results in Table 1 clearly show that the mutated  $SufE_{C51S}$  protein has absolutely no effect on the cysteine desulfurase activity of SufS. This demonstrates the importance of cysteine 51 for SufE function. We checked, by using the yeast two-hybrid methodology, that  $SufE_{C51S}$  makes a complex with SufS in a way similar to wild-type SufE (data not shown)

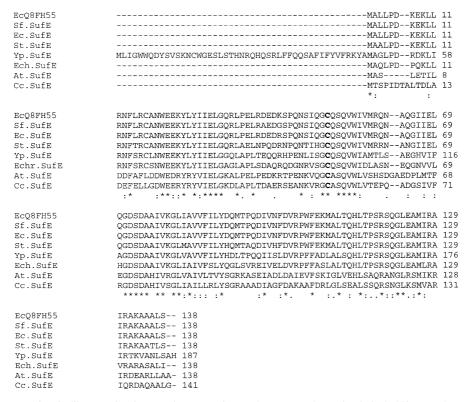


Fig. 1. Sequence alignment of SufE-like proteins from various organisms. The conserved cysteine is in boldface, and conserved amino acids are marked by asterisks. Ec, E. coli; Ff, Shigella flexneri; St; Salmonella thyphimurium; Yp, Yersinia pestis; Ech, E. chrysanthemi; At, Agrobacterium tumefaciens; Cc, Colobacter crescentus.

## 3.2. Sulfur transfer from SufS to SufE

The stimulation of the SufS cysteine desulfurase activity by SufE involves cysteine residues on both proteins (Cys364 on SufS and Cys51 on SufE), suggesting sulfur transfer between these two proteins. We used mass spectrometry, following the methodology by Dean and Johnson [14], to investigate the sulfur transfer mode from SufS to SufE.

The deconvoluted mass spectrum of SufE treated with DTT only comprises the monomer molecular ion peak at 16734.6 Da in agreement with the theoretical mass based on primary sequence (16865.44 Da minus 131 Da, corresponding to the cleavage of the N-terminal methionine) (Fig. 2A) whereas that of SufS pretreated with DTT comprises the monomer molecular ion peak at 45499.9 Da in agreement with the theoretical mass based on primary sequence (45498.89 Da) (Fig. 3A).

Evidence for the capacity of SufE to accept sulfur atoms was provided by the mass spectrum of SufE (Fig. 2B) after a reaction in which SufE was incubated with a stoichiometric amount of SufS in the presence of an excess of the substrate, cysteine, and in the absence of DTT. Fig. 2B shows a major peak at 16736.4 Da and peaks at 16769, 16800 and 16832 Da corresponding to the addition of one, two and three sulfur atoms, respectively. A small peak corresponding to the addition of four sulfur atoms is also clearly observed at 16864 Da.

Table 1 Effect of SufE and SufE<sub>C51S</sub> on the cysteine desulfurase activity of SufS (SufE is added in equimolar amount to SufS)

	SufS	SufS+SufE	SufS+SufE <sub>C51S</sub>
V <sub>m</sub> (units/mg)	0.025	0.70	0.03

These peaks are attributed to SufE with covalently attached polysulfides or a mixture of polysulfides and persulfides, since the addition of DTT to the reaction mixture converts SufE back to the initial as prepared form (Fig. 2C). At the end of the reaction, the deconvoluted mass spectrum of SufS only comprises the monomer molecular ion peak showing that in the presence of SufE, sulfur atoms do not accumulate on SufS (data not shown).

The same experiment repeated with SufE<sub>C51S</sub> did not result in covalent attachment of sulfane sulfur (polysulfides or persulfides) to the protein. The mass spectrum of SufE<sub>C51S</sub> displays only one peak corresponding to the molecular mass (Fig. 2D) whereas a major peak at 45 532 Da was obtained for SufS (data not shown). Since the latter occurs at +32 Da and is dramatically decreased by the addition of DTT (data not shown), it is likely to correspond to the addition of one sulfur atom presumably attached to cysteine 364 as a stable persulfide. These results not only show that cysteine 51 in SufE is absolutely required for sulfur binding to SufE but also suggest that sulfur atoms on SufE originate from sulfane sulfur located on SufS protein. Mutation of cysteine 51 in SufE results in storage of sulfur on SufS.

Direct evidence for sulfur transfer from SufS to SufE was provided by mass spectrometry analysis of SufE and SufS after a single turnover. In such a reaction, the persulfide intermediate on SufS was first generated by incubating SufS with an excess of cysteine in the absence of DTT. After desalting over a Microspin column, an equimolar amount of SufE was added and the reaction mixture analyzed by mass spectrometry at different reaction times. Fig. 3 shows the mass spectrum of SufS before (Fig. 3A) and after reaction with

cysteine (Fig. 3B). In Fig. 3B, a peak at 45 532.0 Da is observed in addition to the peak at 45 500.0 Da again assigned to the persulfide-containing form of SufS, generated by desulfuration of cysteine. Fig. 4 shows the mass spectrum of SufE before (Fig. 4A) and after (Fig. 4B, C) incubation with the persulfide-containing form of SufS. After a few seconds incubation (Fig. 4B) a major peak at 16734.9 Da is observed, corresponding to the monomeric form of SufE, together with two other peaks at 16767 and 16800 Da corresponding to the addition of one and two sulfur atoms, respectively. Fig. 4C shows that after 5 min incubation SufE has the ability to bind additional sulfur atoms, up to four sulfur atoms (peaks at 16832 and at 16864 Da for three and four sulfur atoms incorporated, respectively). SufE mass spectra after 10 and 20 min reactions were identical to Fig. 4C showing that sulfur transfer is fast and complete after a few minutes. Incubation of SufE with sodium sulfide (2 mM final concentration) for 20 min did not result in sulfur binding to the protein (data not shown). These experiments suggest a fast and direct transfer of sulfur atoms from SufS persulfide intermediate to SufE.

#### 4. Discussion

The experiments reported here provide new insights into the mechanism of cysteine desulfuration by the SufS/SufE system.

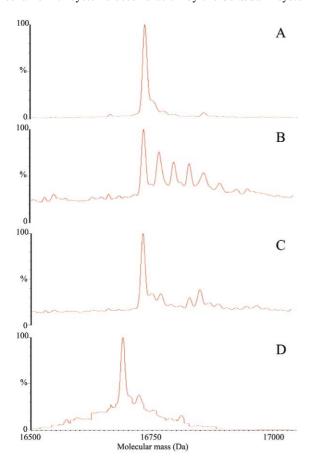


Fig. 2. Reconstructed ESI mass spectra of *E. coli* SufE. A: SufE (100  $\mu$ M). B: Reaction mixture containing SufS (100  $\mu$ M), SufE (100  $\mu$ M) and cysteine (4 mM) after 20 min reaction. C: Reaction mixture as in B after addition of DTT (10 mM). D: Reaction mixture containing SufS (100  $\mu$ M), SufE<sub>C51S</sub> (100  $\mu$ M) and cysteine (4 mM) after 20 min reaction. Mass spectrometry analysis was achieved after dilution of the reaction mixtures (see Section 2).

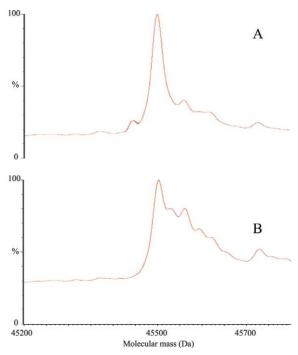


Fig. 3. Reconstructed ESI mass spectra of *E. coli* SufS (A) before and (B) after incubation of SufS ( $100~\mu M$ ) with 4 mM cysteine and desalting. Mass spectrometry analysis was achieved after dilution of the samples.

As previously shown with SufS from E. chrysanthemi [6], SufS from E. coli has a very weak cysteine desulfurase activity which is greatly increased upon addition of SufE. There are several possible explanations for the low activity of SufS. First, as shown by the three-dimensional structure of SufS [15] the distance between the critical active site cysteine residue (cysteine 364) and the PLP substrate Schiff base intermediate is too large to allow an efficient sulfur transfer to SufS. Alternatively, a low activity could result from the limited accessibility of the intermediate protein-bound persulfide and its slow decomposition to sulfide and cysteine by DTT which closes the catalytic cycle in the in vitro assay. A stable persulfide in SufS is consistent with the observation reported here of a peak at 45532 Da in the mass spectrum of the protein treated with cysteine in the absence of DTT (Fig. 3B) and the slow decay of this peak upon treatment with DTT. It should be also noted that the enzyme form in which the active cysteine is in the persulfide state has been crystallized and its three-dimensional structure determined [16].

The mechanism of the reaction catalyzed by the SufS/SufE complex is depicted in Scheme 1.

In a first step, cysteine binds to SufS and transfers its sulfur atom to cysteine 364 to generate a persulfide on the protein. This is consistent with the following observations: (i) incubation of SufS with cysteine results in binding of a sulfur atom



Scheme 1. Cysteine desulfuration catalyzed by the SufS/SufE system.  $R\text{-}CH_2SH$ : cysteine,  $R\text{-}CH_3$ : alanine.

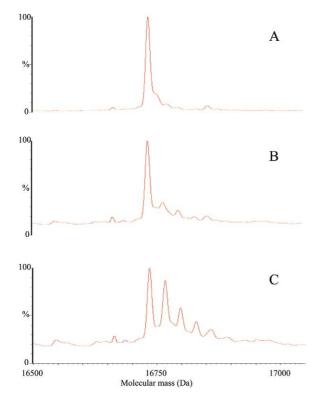


Fig. 4. Sulfur transfer from SufS to SufE. Reconstructed ESI mass spectra of *E. coli* SufE. A: SufE (100  $\mu$ M). Reaction mixture containing the persulfide-containing form of SufS (100  $\mu$ M), SufE (100  $\mu$ M) after a few seconds (B) and 5 min reaction (C). Mass spectrometry analysis was achieved after dilution of the reaction mixtures.

to the protein as shown by mass spectrometry (our results); (ii) the activity of the SufS/SufE-dependent reactions is abolished upon substitution of the critical cysteine of SufS to serine [6]; (iii) the  $K_{\rm m}$  value for cysteine is the same in the SufS alone and in the SufS/SufE-dependent reactions [6]; (iv) SufE alone does not display a cysteine desulfurase activity [6].

In a second step, the sulfur atom of the persulfide in SufS is transferred to cysteine 51 of SufE presumably through a direct transpersulfuration reaction. This is consistent with the following observations: (i) in a one-turnover experiment in the absence of both substrate and DTT, the sulfur atom attached to SufS is transferred to SufE as shown by mass spectrometry; (ii) in the absence of the reducing agent DTT, incubation of SufS/SufE with cysteine results in the accumulation of several sulfur atoms (polysulfides) selectively in SufE (and not in SufS) which can be removed upon further treatment with DTT; (iii) this is abolished upon substitution of cysteine 51 of SufE to serine.

There are precedents of such transpersulfuration reaction: for example the transfer of sulfur from IscS to IscU, during iron–sulfur cluster biosynthesis, and to ThiI, during 4-thiouridine synthesis [14,17,18].

In the last step, sulfur is liberated upon reaction of SufE persulfide/polysulfide with an acceptor compound: DTT in the in vitro assay or presumably a physiological target compound in vivo.

The great acceleration of the cysteine desulfurase activity of the SufS/SufE complex is thus proposed to result from a fast sulfur transfer from SufS to a specific cysteine residue of SufE and a relatively increased accessibility of the sulfur atom in SufE as compared to SufS. Whether this also implies considerable conformational changes of both SufS and SufE which would occur upon binding of SufE to SufS is a possibility.

Further kinetic and structural studies are planned to better understand how the enzyme activity of this new type of cysteine desulfurase is tuned.

## 5. Note added in proof

The results of a study reported in a paper in press in the Journal of Biological Chemistry (F.W. Outten, M.J. Wood, F.M. Munoz and G. Storz) have led to similar conclusions.

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