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The Trimeric Periplasmic Chaperone Skp of *Escherichia coli* Forms 1:1 Complexes with Outer Membrane Proteins *via* Hydrophobic and Electrostatic Interactions

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The interactions of outer membrane proteins (OMPs) with the periplasmic chaperone Skp from *Escherichia coli* are not well understood. We have examined the binding of Skp to various OMPs of different origin, size, and function. These were OmpA, OmpG, and YaeT (Omp85) from *Escherichia coli*, the translocator domain of the autotransporter NalP from *Neisseria meningitidis*, FomA from *Fusobacterium nucleatum*, and the voltage-dependent anion-selective channel, human isoform 1 (hVDAC1) from mitochondria. Binding of Skp was observed for bacterial OMPs, but neither for hVDAC1 nor for soluble bovine serum albumin. The Skp trimer formed 1:1 complexes, OMP·Skp₃, with bacterial OMPs, independent of their size or origin. The dissociation constants of these OMP·Skp₃ complexes were all in the nanomolar range, indicating that they are stable. Complexes of Skp₃ with YaeT displayed the smallest dissociation constants, complexes with NalP the largest. OMP binding to Skp₃ was pH-dependent and not observed when either Skp or OMPs were neutralized at very basic or very acidic pH. When the ionic strength was increased, the free energies of binding of Skp to OmpA or OmpG were reduced. Electrostatic interactions were therefore necessary for formation and stability of OMP·Skp₃ complexes. Light-scattering and circular dichroism experiments demonstrated that Skp₃ remained a stable trimer from pH 3 to pH 11. In the OmpA·Skp₃ complex, Skp efficiently shielded tryptophan residues of the transmembrane strands of OmpA against fluorescence quenching by aqueous acrylamide. Lipopolysaccharide (LPS), a major component of the outer membrane of Gram-negative bacteria, bound to OmpA·Skp₃ complexes at low stoichiometries. Acrylamide quenching of fluorescence indicated that in this ternary complex, the tryptophan residues of the transmembrane domain of OmpA were located closer to the surface than in binary OmpA·Skp₃ complexes. This may explain previous observations that folding of Skp-bound OmpA into lipid bilayers is facilitated in presence of LPS.

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Keywords: Skp; outer membrane protein; membrane protein folding; molecular chaperone; protein-protein interactions

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Abbreviations used: BSA, bovine serum albumin; FomA, major outer membrane protein of *Fusobacterium nucleatum*; hVDAC1, voltage-dependent anion-selective channel human isoform-1; LPS, lipopolysaccharide; NalP, translocator domain of the autotransporter NalP from *Neisseria meningitidis*; OM, outer membrane; OMP, outer membrane protein; OmpA, outer membrane protein A of *E. coli*; OmpG, outer membrane protein G of *E. coli*; PPIases, peptidyl-prolyl *cis/trans* isomerases; SurA, the survival factor A; TMD-OmpA, transmembrane domain of OmpA, i.e. amino acid residues 0–176; WT-OmpA, wild type OmpA; YaeT (Omp85), 89 kDa outer membrane protein of *E. coli*.

Introduction

Outer membrane proteins (OMPs) of Gram-negative bacteria are synthesized in the cytosol and translocated in an unfolded form by the SecYEG translocon across the cytoplasmic membrane into the periplasm, before they insert into the outer membrane (OM). In recent years, several studies have demonstrated that the deletion of certain genes encoding periplasmic proteins in Gram-negative bacteria, e.g. deletion of the genes encoding the 17 kDa protein (Skp) or the survival factor A (SurA), results in reduced concentrations of OMPs in the OM, indicating that these periplasmic proteins serve as molecular chaperones in the assembly pathway of OMPs.¹ In *Escherichia coli*, expression of these chaperones is under control of either the σ^E ^{2,3} or the two-component CpxA/CpxR stress-response system,^{4,5} and upregulated when misfolded OMPs accumulate in the periplasm. Periplasmic chaperones are key factors to prevent aggregation and misfolding of bacterial OMPs. However, little is known about their interaction with integral membrane proteins in solution.

Studies with peptide libraries indicated that periplasmic SurA binds preferentially to peptides containing the amino acid sequence motif aromatic-random-aromatic with micromolar affinities. Such motifs are found frequently in OMPs.^{6,7} A binding specificity of Skp has not been reported. When periplasmic cell extracts were run over an affinity column containing unfolded OmpF that was covalently linked to Sepharose, Skp was the major protein that bound to the column.⁸ Upon deletion of the *skp* gene, reduced concentrations of OMPs, like outer membrane protein A (OmpA), OmpF, OmpC, or LamB were found in the OM of *E. coli*.⁸ Skp binds OmpA early after secretion through the cytoplasmic membrane.⁹ It also improves the functional expression of soluble antibody fragments in the periplasm of *E. coli*.¹⁰ Periplasmic chaperones do not require ATP, which is not found in the periplasm.¹¹

We previously demonstrated that the simultaneous presence of Skp and lipopolysaccharide (LPS) improves insertion and folding of OmpA into phospholipid membranes.¹² OmpA has been used extensively as a model to study membrane insertion and folding of β -barrel membrane proteins. Unfolded OmpA in 8 M urea inserts and folds spontaneously into preformed lipid bilayers upon dilution of the urea.¹³ In the presence of Skp, the kinetics of this folding reaction is retarded and folding yields are reduced.¹² A similar effect is observed, when OmpA folding experiments are performed in the presence of LPS, which is a major component of the OM. However, when Skp and LPS are present simultaneously at low stoichiometries (three Skp and two to seven LPS per OmpA), the folding rates of OmpA are faster than in the absence of both. The concurrent presence of Skp and LPS also leads to larger yields of folded OmpA in lipid bilayers.¹² The biosyntheses of Skp and LPS are

apparently correlated, since the gene *firA*, which codes for an acyltransferase involved in LPS biosynthesis^{14,15} is located only four bases downstream from the *skp* gene, mapped at the 4 min region on the chromosome.¹⁰ Pulse labeling and biochemical reconstitution experiments suggested that LPS may be required for efficient assembly of OMPs such as trimeric PhoE,¹⁶ and monomeric OmpA,¹⁷ into outer membranes.

Skp consists of 141 amino acid residues and forms a stable homotrimer,¹⁸ consistent with the previously determined stoichiometry of Skp in complex with unfolded OmpA.¹² The trimer resembles a jellyfish and is formed of a tentacle domain with α -helical tentacles that protrude about 60 Å from a β -barrel body termed the association domain. The tentacle domain defines a central cavity, which is accessible from the tip region of the tentacle domain and from the side, where the tentacles are separated by about 25 Å.^{19,20} The structural motif of a predicted LPS binding site²⁰ is similar to the one identified in the OMP FhuA of *E. coli*²¹ and is located in the middle of the Skp tentacles. This predicted binding site is composed of three basic residues (K77, R87, and R88) on the Skp surface.

Here, we have examined Skp binding to several OMPs, namely OmpA, OmpG, and YaeT (also called Omp85) from *E. coli*, the translocator domain of the autotransporter NalP from *Neisseria meningitidis*, FomA from *Fusobacterium nucleatum*, and hVDAC1, isoform 1 (*Homo sapiens*) from mitochondrial OMs. OmpA (35 kDa) is composed of a 171 residue 8 stranded β -barrel transmembrane domain and a 154 residue periplasmic domain. The β -barrel of the 301 residue OmpG (33 kDa) consists of 14 β -strands with a central pore, which is capable of transporting large solutes.^{22,23} The translocator domain of NalP (32 kDa) forms a 12-stranded TM β -barrel that contains an N-terminal α -helix.²⁴ The structure of the 89 kDa YaeT (also called Omp85) is not known, but YaeT consists of a C-terminal TM domain (predicted to contain 12 TM β -strands²⁵) and a large 500 residue periplasmic domain. The structure of the YaeT homolog FhaC of *Bordetella pertussis* was published recently. FhaC forms a 16-stranded β -barrel.²⁶ YaeT (Omp85) is a highly conserved protein that is essential for cell viability and for outer membrane biogenesis.^{25,27,28} FomA (40 kDa), a voltage-dependent general diffusion porin, is predicted to form a 14-stranded TM β -barrel.^{29,30} hVDAC1 (31 kDa) is the human isoform 1 of the important voltage-dependent anion-selective channel of the mitochondrial outer membrane.³¹ We investigated whether the interaction of Skp of *E. coli* is specific for OMPs of *E. coli* or whether Skp recognizes the amino acid sequence of β -strands independent of the origin of the OMP. For several OMPs, we estimated Skp binding stoichiometries and the free energies of their binding to Skp to determine whether these are affected by the size of the OMP transmembrane domain. We asked whether binding is exclusively caused by hydrophobic interactions or whether it is also mediated in

part by electrostatic interactions. Finally, we examined binding of LPS to complexes of Skp and OmpA.

Results

Fluorescence spectroscopy indicates that Skp interacts with bacterial outer membrane proteins

Six OMPs of different origin, size and function, OmpA, OmpG, YaeT, NalP, FomA, and hVDAC1, were over-expressed and isolated from *E. coli* in their unfolded forms in 8 M urea. We used fluorescence spectroscopy to investigate their interactions with Skp in aqueous solution after dilution of the urea. For comparison, the interactions of Skp with a soluble protein, bovine serum albumin (BSA), were also examined. Intrinsic Trp fluorescence spectroscopy is an elegant tool to study complex formation of membrane proteins, since the fluorescence emission depends on the hydrophobicity of the micro-environment of the fluorophore. Most OMPs contain tryptophan residues in their hydrophobic transmembrane domains. Skp is only very weakly fluorescent when excited at 290 nm or above, since it does not contain tryptophan and has only a single tyrosine residue.

The unfolded forms of the OMPs and of BSA in 8 M urea showed fluorescence spectra with an emission maximum at $\lambda_U \approx 347$ nm (Figure 1; Table 1). Upon strong dilution of the urea, fluorescence maxima of the aqueous forms, λ_{AQ} , were shifted

Table 1. Effect of Skp on the fluorescence emission spectra of OMPs at pH 8 (Figure 1)

OMP	pI ^a	F_C/F_{AQ} ^b	λ_U ^c (nm)	λ_{AQ} ^c (nm)	λ_C ^c (nm)	$\Delta\lambda$ ^d (nm)
WT-OmpA	5.6	2.26	347.2	344.4	338.5	5.9
TMD-OmpA	5.5	2.20	347.3	344.3	338.6	5.7
OmpG	4.4	1.84	347.7	343.4	339.5	3.9
YaeT	4.9	1.29	347.4	338.9	338.1	0.8
NalP	8.0	1.86	348.0	344.5	339.6	4.9
FomA	8.6	1.34	347.4	339.8	339.0	0.8
hVDAC1	8.6	1.03	347.2	334.9	335.3	-0.4
BSA	5.6	1.03	345.4	331.1	331.5	-0.4

^a Calculated isoelectric point. The pI was calculated using ProtParam at the Swissprot database, based on published algorithms.⁴⁹

^b Ratio of the OMP fluorescence intensities at 330 nm in the presence and in the absence of Skp.

^c Wavelength of the fluorescence emission maxima of the unfolded OMP in 8 M urea (λ_U), after 100-fold dilution of urea with aqueous buffer (λ_{AQ}), and of OMP-Skp complexes (λ_C). The maxima were calculated by fitting a lognormal distribution to the experimental spectra.⁴⁵

^d $\Delta\lambda = \lambda_{AQ} - \lambda_C$.

towards shorter wavelengths, ranging from $\lambda_{AQ} \approx 344$ – 340 nm for the bacterial OMPs, to $\lambda_{AQ} \approx 335$ nm for hVDAC1 and to $\lambda_{AQ} \approx 331$ nm for BSA, indicating a more hydrophobic environment, caused either by hydrophobic collapse or by formation of folded structure (in the soluble BSA). Fluorescence intensities were higher for YaeT, FomA and BSA, but lower for WT-OmpA, TMD-OmpA, OmpG, NalP, and hVDAC1.

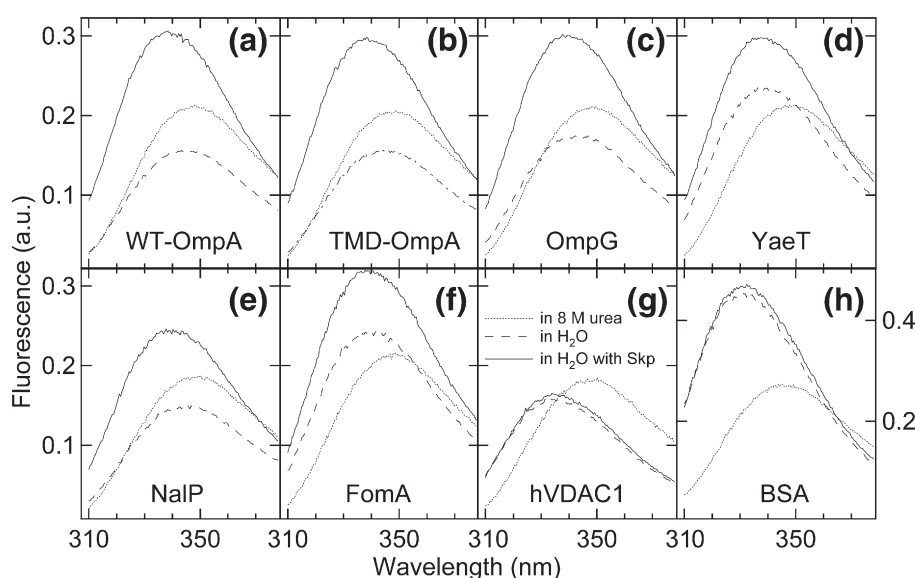


Figure 1. Skp of *E. coli* interacts with outer membrane proteins of different origin. The fluorescence spectra of: (a) 0.43 μ M WT-OmpA; (b) 0.41 μ M TMD-OmpA; (c) 0.18 μ M OmpG; and (d) 0.14 μ M YaeT of *E. coli*; (e) for 0.47 μ M NalP of *N. meningitidis*; (f) for 0.20 μ M FomA of *F. nucleatum*; (g) 0.65 μ M human VDAC isoform 1 of the mitochondrial outer membrane; and (h) for 0.60 μ M soluble bovine serum albumin were recorded for the denatured proteins (···) and for the aqueous forms after 100-fold dilution of the urea either in the absence (---) or in the presence of a fivefold molar excess of Skp (—). All spectra were normalized, dividing fluorescence by the concentration of the OMP and by the number of Trp residues in the OMP. The samples were in 10 mM Tris-HCl (pH 8.0). Spectra were recorded at 25 °C at an excitation wavelength of 295 nm.

When a fivefold molar excess of Skp was added to the aqueous forms of the OMPs after dilution of the denaturant, the fluorescence intensities of all bacterial OMPs increased compared to that of either their aqueous or urea-unfolded forms, indicating their binding to Skp. There was no change in the fluorescence maxima of hVDAC1 or BSA, suggesting that these two proteins do not bind to Skp. The wavelengths of the fluorescence maxima of the OMP:Skp complexes, λ_C , were even shorter, 338–340 nm (Figure 1; Table 1), which demonstrated that the environment of the fluorescent tryptophan residues of the OMPs is more hydrophobic in complex with Skp. The strongest increase in fluorescence (2.2–2.3-fold) and the largest shift of the emission maximum, 5.7–5.9 nm, was observed when Skp was reacted with TMD-OmpA or WT-OmpA, respectively (Table 1).

Trimeric Skp forms 1:1 complexes with outer membrane proteins

Our previous work¹² described a 3:1 stoichiometry for binding of Skp to aqueous OmpA, which is consistent with the more recent observation that Skp is a trimer in solution.¹⁸ However, the binding stoichiometry may depend on the size of the OMP or its transmembrane domain. To obtain stoichiometries and to estimate the strength of Skp binding, we recorded fluorescence spectra of the aqueous forms of each OMP after dilution of the urea as a function of the molar Skp/OMP ratio until no further in-

crease of the fluorescence intensity was observed (Figure 2). Skp formed complexes with five OMPs: OmpA (WT-OmpA and TMD-OmpA), OmpG, YaeT of *E. coli*, NalP of *N. meningitidis*, and FomA of *F. nucleatum*. When OMPs were titrated with Skp, their emission spectra were a superposition of the fluorescence spectra of the Skp-bound and of the free form of the OMP. Both the free and the bound OMP contributed to the fluorescence signal of the sample (Figure 1). Since changes in fluorescence at 330 nm ($F_{330 \text{ nm}}$) were relatively large and $F_{330 \text{ nm}}$ increased linearly for all OMPs upon initial titration with Skp (Figure 2), we monitored binding at this wavelength. $F_{330 \text{ nm}}$ was analyzed as a linear combination of the fluorescence contributions of bound and free OMP as described in Materials and Methods. For analysis, binding functions (equation (2)) were fit to the fluorescence intensities at 330 nm as a function of the Skp/OMP ratio. The fits to these binding data converged for four OMPs (Figure 2), but not for FomA (data not shown). The stoichiometries, association constants, dissociation constants, and free energies of Skp binding that were obtained from the fits are given in Table 2 for each of these OMPs. The stoichiometries were all similar and demonstrated formation of stable 1:3 complexes, OMP:Skp₃. The size of the OMPs varied from 19 kDa (TMD-OmpA) to 89 kDa (YaeT) and the number of transmembrane β -strands of the folded OMP ranged from 8 in OmpA to 14 in OmpG. These differences are obviously of no consequence for the binding stoichiometry. Since there was neither a

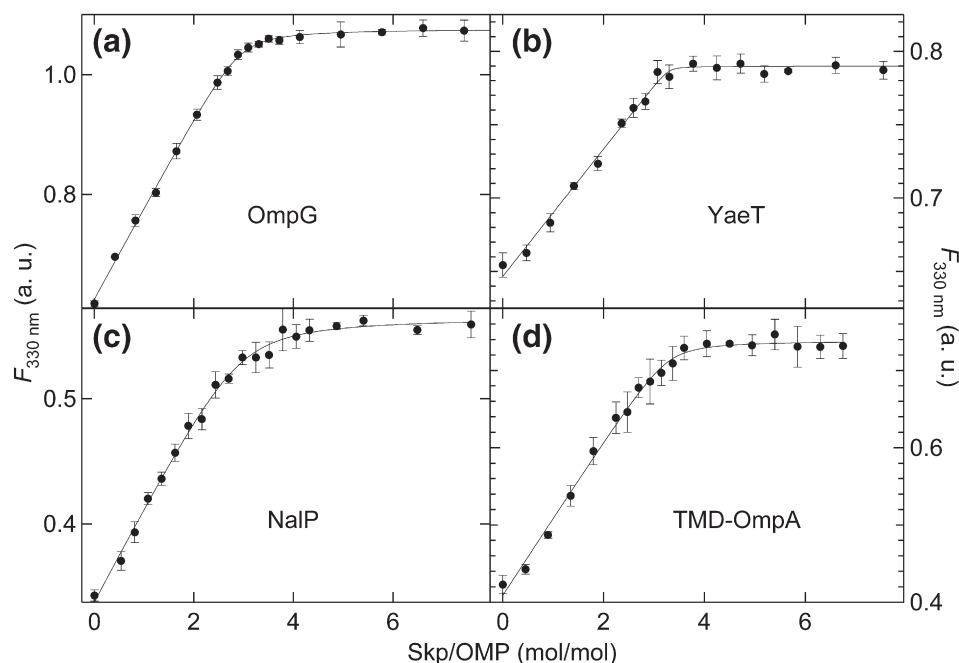


Figure 2. Binding of Skp to the aqueous forms of various outer membrane proteins in solution. The fluorescence signals at 330 nm of the aqueous forms of (a) OmpG, (b) YaeT, (c) NalP, and (d) TMD-OmpA increased as a function of the molar Skp/OMP ratio until saturation levels were reached. At each Skp/OMP ratio, three separate samples were prepared and fluorescence spectra were recorded to obtain averaged fluorescence intensities. To calculate stoichiometries and binding constants, binding isotherms given by equation (2) were fit to the data (—). Spectra were recorded at 25 °C and at an excitation wavelength of 290 nm.

Table 2. Stoichiometry and free energy of Skp binding to OMPs

OMP	M^a	$Z_{\text{TM}\beta\text{S}}^b$	n^c	$K_{\text{ass}} (\mu\text{M}^{-1})^d$	$K_D (\text{nM})^d$	$\Delta G_{\text{ass}} (\text{kJ/mol})^e$
WT-OmpA	35	8	2.8 ± 0.1	$46. \pm 35$	$22. \pm 16$	-43.0 ± 2
WT-OmpA (1 M NaCl)	35	8	3.1 ± 0.2	$12. \pm 4$	$83. \pm 30$	-40.3 ± 1.0
TMD-OmpA	19	8	3.3 ± 0.1	$49. \pm 35$	$20. \pm 15$	-43.9 ± 3.2
OmpG	33	14	2.9 ± 0.1	$82. \pm 20$	$12. \pm 3$	-45.2 ± 0.7
OmpG (0.1 M NaCl)	33	14	2.9 ± 0.2	$19. \pm 9$	$53. \pm 17$	-41.6 ± 1.6
OmpG (0.5 M NaCl)	33	14	2.8 ± 0.3	$12. \pm 6$	$83. \pm 27$	-40.3 ± 1.8
YaeT ^f	89	u^f	3.3 ± 0.1	$2900. \pm 900$	0.3 ± 0.1	-54.0 ± 0.9
NalP ^g	32	12	3.0 ± 0.1	$20. \pm 8$	$50. \pm 20$	-42.4 ± 1.3

^a Molecular mass, M in kDa.^b Number of transmembrane β -strands, $Z_{\text{TM}\beta\text{S}}$, of the folded form of the OMP.^c Stoichiometry, n , estimated from fits to the data shown in Figures 2 and 5.^d Estimated binding constants for association K_{ass} and dissociation $K_D = 1/K_{\text{ass}}$.^e Estimated free energy of binding for the association reaction.^f A structure of the YaeT TM domain has not been published to date and the number of strands in the TM domain is unknown.^g Translocator domain of NalP, residues 776–1083.

difference in the stoichiometry or in the free energy of binding for TMD-OmpA (176 residues) and WT-OmpA (325 residues),¹² Skp recognizes the transmembrane domain of OmpA specifically. Dissociation constants were in the nanomolar range, indicating stable complexes. The free energy of complex formation, ΔG_{ass} , ranged from -42 kJ/mol to -45 kJ/mol (-10 kcal/mol to -11 kcal/mol) for TMD-OmpA, WT-OmpA, NalP and OmpG, but binding was stronger for YaeT with $\Delta G \approx -54$ kJ/mol (-13 kcal/mol).

Skp binding to outer membrane proteins is pH-dependent

OMPs develop their hydrophobic surface upon formation of their amphipathic β -strands. The anti-parallel strands form a β -barrel, in which the polar residues are oriented towards the interior lumen and the hydrophobic residues are oriented towards the fatty core of the lipid bilayer. The average hydrophobicity of OMPs is low when compared to membrane proteins of the cytoplasmic membrane and closer to the average hydrophobicity of soluble proteins. In complex with Skp, OmpA remained largely unfolded.¹² It is therefore conceivable that the interactions between Skp and aqueous OMPs are not entirely of a hydrophobic nature. Skp is a very basic protein and positively charged (theoretical pI ~ 9.6 – 10.5), while OMPs of *E. coli* have isoelectric points ranging from pI 4.5 to pI 6 and are negatively charged. These opposite net charges suggest electrostatic attraction between Skp and OMPs. Skp has a large cavity inside its tentacle domain that contains hydrophobic patches and could shield the hydrophobic residues of OMPs from the aqueous space while preserving them in their monomeric forms for subsequent membrane insertion. For these reasons, the formation of stable and stoichiometric complexes of unfolded aqueous OMPs with Skp may be driven by electrostatic interactions, or by the hydrophobic effect, or by both.

To investigate whether the charged residues of Skp and OMPs participate in complex formation, we recorded the fluorescence spectra of four aqueous

OMPs in the absence and in the presence of Skp as a function of pH, including the regions above and below the isoelectric points of Skp and OMPs, respectively. The fluorescence intensities at 330 nm, $F_{330 \text{ nm}}$, and the wavelengths, λ_{max} , of the emission maxima of OMPs and OMP-Skp₃ complexes were pH-dependent (Figure 3).

Skp did not interact with aqueous hVDAC1 (pI 8.6). In the absence of Skp, $F_{330 \text{ nm}}$ and λ_{max} of the fluorescence spectra of hVDAC1 changed only slightly between pH 4 and pH 9 (Figure 3(d)). At both ends of the pH range, in particular at pH ≥ 10 , fluorescence decreased and λ_{max} increased, indicating a more polar environment and increased water accessibility of the fluorescent tryptophan residues. This is caused by side-chain deprotonation (at pH > 9) or protonation (at pH < 5), which both increase the net charge of hVDAC1 (pI 8.6). Independent of pH, the addition of Skp had little or no effect on the fluorescence spectra of hVDAC1.

The fluorescence properties of the aqueous forms of the bacterial OMPs (Figure 3(a)–(c)) displayed a different dependence on pH in the absence of Skp. From pH 6 to pH 10, the fluorescence maxima of these OMPs were located at a longer wavelength than the emission maxima of hVDAC1, indicating at least partial exposure of the tryptophan residues of OmpA, NalP, and FomA to the aqueous solution. When the pH was lowered to 4 or 5, $F_{330 \text{ nm}}$ increased sharply, and λ_{max} decreased strongly, for OmpA, NalP, and FomA, indicating that the fluorescent tryptophan residues became more shielded from polar interactions and had a more hydrophobic environment at this pH. In the absence of Skp, WT-OmpA and NalP displayed one major titration (around pH 5.5) of Trp exposure to the aqueous environment. The pH-dependent changes of the FomA spectra indicated two titrations, one between pH 5 and pH 7, and a second between pH 9 and pH 11. This suggested that there are two different conformations of FomA in different regions of pH that display a different surface exposure of tryptophan to the aqueous solution.

Binding of Skp to WT-OmpA, NalP, and FomA was pH-dependent, as indicated by the changes in

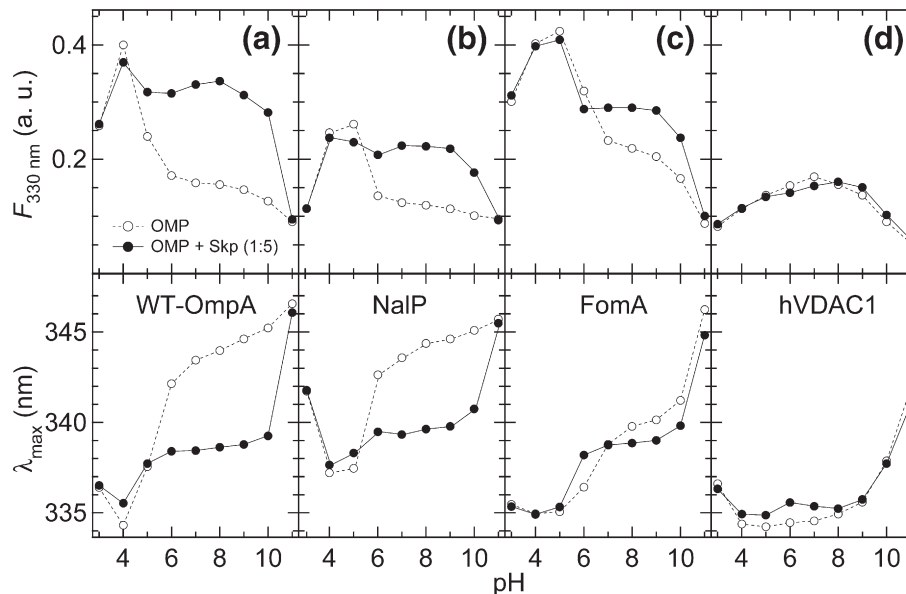


Figure 3. Effect of pH on the binding of Skp to outer membrane proteins. To examine the pH-dependence of binding of Skp to OMPs (a) WT-OmpA, (b) NalP, (c) FomA and (d) hVDAC1, fluorescence spectra of these OMPs were recorded for the aqueous forms in the absence (○) and in the presence (●) of a fivefold molar excess of Skp as a function of pH between pH 3 and pH 11. Urea was diluted 100-fold to a residual concentration of 80 mM. The upper panels show the fluorescence intensities at 330 nm and the lower panels show the wavelengths of the intensity maxima of the spectra upon excitation at 295 nm.

the Trp fluorescence spectra upon addition of Skp (Figure 3(a)–(c)). From pH 6 to pH 10, Skp binding resulted in strongly increased OMP fluorescence and in a shift of λ_{\max} towards shorter wavelength, indicating a more hydrophobic environment of the tryptophan residues of these OMPs in the complexes. Binding of Skp to FomA led to smaller fluorescence changes than Skp binding to OmpA or NalP. In the absence of Skp, the tryptophan residues of the aqueous form of FomA are apparently less exposed to water than those of OmpA and NalP and, therefore, fluorescence spectra of FomA were less affected by Skp binding.

Above pH ~ 10.5 , the positively charged lysine residues of Skp deprotonate and become neutral. Skp did not bind to OMPs at pH 11 or above, indicating a contribution of charge interactions to the free energy of OMP binding to Skp. Binding of OmpA or NalP to Skp was not detected below pH 5. While this is close to the pI of OmpA, NalP and FomA have much higher theoretical isoelectric points, namely at pI 8.0 and at pI 8.6, respectively. The charge distribution in the aqueous forms of these OMPs after dilution of the denaturant is unknown and it cannot be excluded that it is asymmetric. The titrations of both OMPs in the absence of Skp indicated conformational transitions below pH 6 in both NalP and FomA as a consequence of protonation.

Skp is a stable trimer over a wide range of pH values

We next performed dynamic light-scattering experiments to exclude the possibility that the ob-

served pH-dependence of OMP binding to Skp may be caused by a loss of the trimeric structure of Skp at very acidic or very basic pH. The hydrodynamic radius of Skp remained unaltered at $R_H = 3.5$ nm over the entire pH range from 3 to 11 (Figure 4). The experimentally obtained molecular mass of the Skp particles $M_{\text{exp}}(\text{Skp}_3)$ was ≈ 54 kDa as calculated using equation (1). This is consistent with the trimeric structure of Skp (calculated from the amino acid sequence as $M(\text{Skp}_3) = 47$ kDa). Skp contains a central cavity within its tentacle domain and therefore the specific volume of Skp is likely larger than that of a typical protein. An overestimation of $M_{\text{exp}}(\text{Skp}_3)$ is therefore not surprising. The pH-independence of R_H and of CD spectra of Skp (Figure 4(b)) indicated that the Skp trimer is stable and resists changes of pH.

High ionic strength reduces the stability of Skp₃–OMP complexes

To further examine the electrostatic contribution to the binding of Skp₃ to OMPs, we titrated OmpA (Figure 5(a)) and OmpG (Figure 5(b)) with Skp at different concentrations of salt. The fluorescence intensities of OmpA and OmpG increased at higher concentrations of NaCl, but λ_{AQ} decreased only slightly to about 340 nm (data not shown), indicating that the tryptophan residues are still partially exposed to the aqueous environment at high concentrations of salt. Skp binds to these OMPs even at 1 M NaCl. At high ionic strength, a more gradual saturation of the binding functions was observed with OmpA (Figure 5(a), □) and with OmpG (Figure 5(b), △, ◇). At a low concentration of

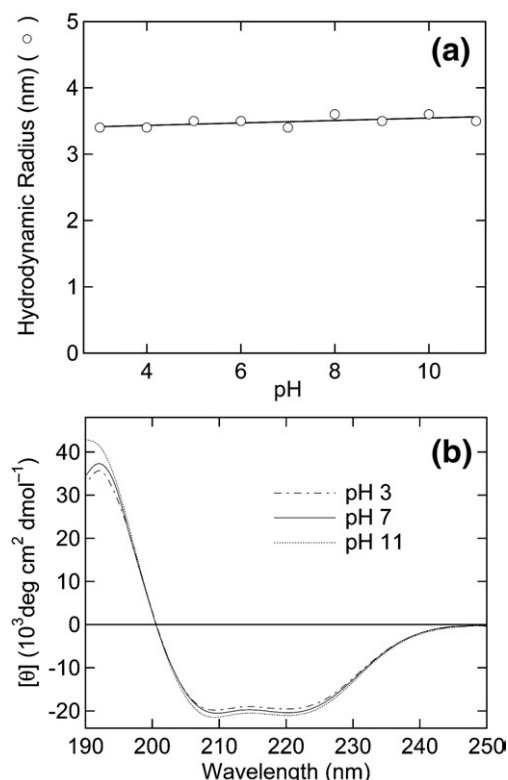


Figure 4. Skp forms a stable trimer from pH 3 to pH 11. (a) Light-scattering experiments demonstrate that the hydrodynamic radius of Skp shows no variation between pH 3 and pH 11. Experiments were performed at 25 °C at an Skp concentration of 0.13 mM. (b) CD spectra of Skp at pH 3 (---), pH 7 (—), and pH 11 (···), and similar CD spectra at pH 4, 5, 6, 8, 9, and 10 (data not shown), indicate that the secondary structure of Skp does not change over the entire pH range.

salt, the binding data leveled off sharply (Figure 5(a) and (b), ○). The more gradual leveling at high ionic strength already suggested weaker interactions between the binding partners. As expected, fits of the binding functions to the data resulted in lower association constants and the free energies of binding (association) were reduced by ~3 kJ/mol for OmpA and by ~5 kJ/mol for OmpG (Table 2). Although the difference in the free energies was small, the association constant for OmpG binding was reduced two- to fourfold. The actual contribution of electrostatic interactions to the binding of Skp to OMPs was likely much larger, since the presence of salt does not completely shield the charged residues of the proteins. In comparison, the protonation of the anionic residues at low pH or the deprotonation of cationic residues at high pH completely altered the charge-state of the proteins.

LPS binds to OmpA·Skp₃ complexes

LPS facilitated folding of Skp-bound OmpA into lipid bilayers,¹² and the crystal structure of *E. coli* Skp^{19,20} indicated a possible LPS-binding site in the tentacle domain of each Skp monomer, which is

conserved in Skp orthologs from other bacteria.²⁰ To examine LPS binding to OmpA·Skp₃ complexes, we gradually titrated these complexes with LPS and recorded the fluorescence spectra. Increased LPS binding to OmpA·Skp₃ resulted in an initial steep reduction of the fluorescence intensity (Figure 6(a), ●) and in an initial steep increase of the wavelength of the intensity maximum λ_{max} (Figure 6(b), ●) up to a molar ratio of about 3 LPS/(OmpA·Skp₃). At higher ratios, between 5 and 15 LPS/(OmpA·Skp₃), the decrease in intensity leveled off at ~65% of the original intensity and the increase in λ_{max} approached ~342 nm. The LPS binding stoichiometry to OmpA·Skp₃ is a far lower than the molar ratio of 25 LPS/OmpA observed previously for LPS binding to OmpA in absence of Skp.¹² In absence of Skp, LPS binding to unfolded OmpA led to a fluorescence increase and a λ_{max} decrease to ~335 nm (Figure 6, ○). These data suggest a different mode of interaction of LPS with the OmpA·Skp₃ complex compared to the interaction of LPS with OmpA alone, which is reflected in the fluorescence properties of OmpA.

To estimate the binding stoichiometry and the binding constant, we again assumed a simple mass action law with identical LPS-binding sites within

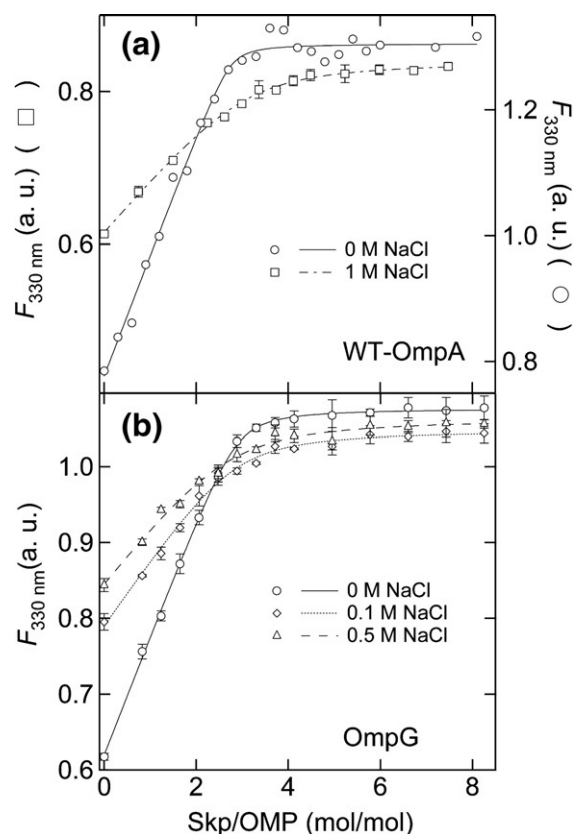


Figure 5. Skp binding to unfolded WT-OmpA and to unfolded OmpG at different concentrations of salt. (a) Fluorescence intensity increase upon addition of Skp to unfolded OmpA in the absence (○) and in the presence of 1 M NaCl (□). (b) Fluorescence intensity increase upon addition of Skp to unfolded OmpG in the absence of NaCl (○), in 0.1 M (Δ) and 0.5 M NaCl (◇).

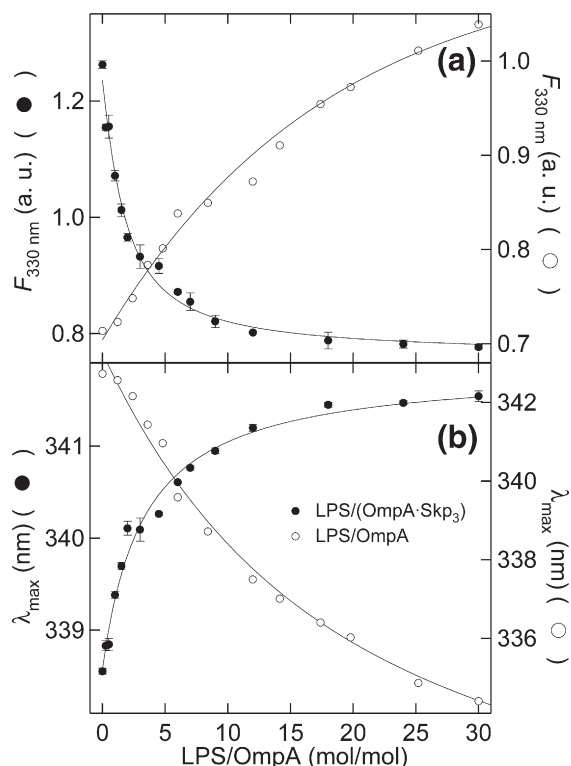


Figure 6. LPS binds to the OmpA-Skp₃ complex. (a) The intrinsic fluorescence intensity at 330 nm and (b) the wavelength of maximum fluorescence of the complex of OmpA and Skp were recorded after addition of LPS at molar LPS/(OmpA·Skp₃) ratios ranging from 0 to 30. Three separate samples were measured at each concentration of LPS and an average value was calculated. F_{330} and λ_{\max} of OmpA after addition of LPS at molar LPS/OmpA ratios ranging from 0 to 30 are shown for comparison (○). The data for LPS binding to WT-OmpA were taken from Bulieris *et al.*¹²

the Skp₃·OmpA complex. The fits to equation (2) suggested a low binding stoichiometry of 1.5(±2) LPS/OmpA·Skp₃ and a dissociation constant of $K_D \approx 500$ nM. These were crude estimates, since LPS likely bound to both OmpA and Skp and therefore the assumption of equivalent binding sites may not be correct.

Skp and LPS shield the tryptophan residues of OmpA against fluorescence quenching by acrylamide

To test the average exposure of the five tryptophan residues of OmpA to the aqueous quencher acrylamide for both aqueous and Skp-bound forms of OmpA, we performed fluorescence quenching studies with acrylamide. We examined the effect of LPS on tryptophan accessibility for aqueous and Skp-bound OmpA (Figure 7). The fluorescence of aqueous OmpA was heavily quenched and a Stern-Volmer constant of $K_{SV} = 6.0$ M⁻¹ was calculated, consistent with previous results.³² In the presence of LPS, but in the absence of Skp, the Stern-Volmer constant is 4.5 M⁻¹, indicating LPS binding to

aqueous OmpA, which partially shields it against quenching by acrylamide. In the presence of Skp, but in the absence of LPS, the quenching was weakest with a Stern-Volmer constant of 2.7 M⁻¹. In the presence of both Skp and LPS, the Stern-Volmer constant increased to 3.6 M⁻¹. Upon addition of LPS, the tryptophan residues of OmpA, which were well-shielded in the complex with Skp, became more accessible to the fluorescence quencher, but their fluorescence was still less quenched than for the aqueous OmpA in the absence of both Skp and LPS, or in the presence of LPS alone. These fluorescence-quenching results supported the formation of a ternary complex between LPS, Skp and OmpA at low stoichiometries. They also indicated a conformational change in the OmpA·Skp₃ complex, partially exposing hydrophobic tryptophan residues to the aqueous space.

Discussion

Our present results on the interactions of periplasmic Skp with OMPs provide detailed new insights into the properties and function of this membrane protein chaperone. After dilution of the urea, the aqueous forms of bacterial OMPs were bound by the Skp trimer with nanomolar affinity. Stable complexes were formed independent of the origin of the bacterial OMP. Binding did not occur with hVDAC1 from the OM of mitochondria,

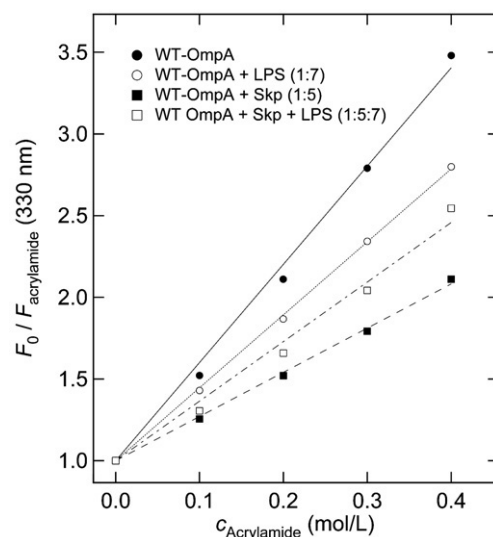


Figure 7. In the presence of LPS or Skp, the tryptophan residues of OmpA are less accessible to aqueous fluorescence quenchers. Fluorescence spectra of OmpA were recorded at different concentrations of the fluorescence quencher acrylamide to obtain the Stern-Volmer plots for OmpA in aqueous solution (●), in the presence of Skp (■), in the presence of LPS (○) or in the presence of Skp and LPS (□). F and F_0 are fluorescence intensities in the presence and in the absence of acrylamide, respectively. Stern-Volmer constants of 6.0 M⁻¹ (OmpA), 4.5 M⁻¹ (OmpA + LPS), 2.7 M⁻¹ (OmpA + Skp) and 3.6 M⁻¹ (OmpA + Skp + LPS) were obtained from the slopes of linear fits.

although hVDAC1 also forms a transmembrane β -barrel similar to bacterial OMPs. Independent of the size of their transmembrane domains, a 1:1 binding stoichiometry was observed for the bacterial OMPs. The binding affinity was highest for YaeT (Omp85), which is essential for the assembly of OMPs in bacteria. Both hydrophobic and electrostatic interactions contributed to the formation of OMP-Skp₃ complexes. We further observed binding of the negatively charged LPS to OmpA-Skp₃ complexes. LPS induced a partial exposure of OmpA to the aqueous phase. The LPS-induced topological change of the OmpA-Skp₃ complex could explain the facilitated folding of OmpA into lipid membranes in the presence of both Skp and LPS that was described earlier.¹²

The Skp trimer binds bacterial OMPs at 1:1 stoichiometry

The five bacterial OMPs (NalP, OmpG, OmpA, FomA, and YaeT) that ranged from 19 kDa to 89 kDa in molecular mass and from 8 to 14 in the number of their transmembrane strands, all bound Skp₃ at 1:1 stoichiometry in aqueous solution after dilution of the denaturant. Other stoichiometries may be prevented because of steric constraints by the size of the proteins or because there is only one specific Skp₃-binding site in each of the OMPs. The observed 1:1 stoichiometry for the interaction of the Skp trimer with OMPs does not completely exclude the formation of larger aggregates (OMP-Skp₃)_n. Here, the Skp₃ binding stoichiometries were determined for four bacterial OMPs that differed in size by a factor >4. It is far more likely that a large OMP would require two Skp₃ for binding, leading to an OMP-Skp₆ complex, which is not observed here, rather than the smallest OMP would form an (OMP-Skp₃)₂ complex. The facts that only one binding stoichiometry was obtained, independent of the size of the bound OMP, and that Skp alone forms trimers but not higher aggregates suggest strongly that the OMP-Skp₃ complexes do not form larger aggregates.

Skp of *N. meningitidis* and Skp of *F. nucleatum* share about 22% and 23% sequence identity with Skp of *E. coli*. The differences between these Skp homologs are apparently not very important for the binding of OMPs, since NalP of *N. meningitidis* and FomA of *F. nucleatum* both formed stable 1:1 complexes with *E. coli* Skp₃, although the dissociation constant for NalP-Skp₃ was twice as high as those of OmpA and OmpG complexes at the same ionic strength (Table 1).

Trimeric Skp binds OMPs with nanomolar affinity

The binding affinities of Skp₃ for various OMPs ranged from 0.3 nM to 50 nM (Table 2) at low ionic strength and were therefore about three orders of magnitude higher than the binding affinities of another periplasmic chaperone, SurA of *E. coli*, to

OmpG and OmpF.³³ In thermodynamic binding equilibria of OMPs in the presence of both, Skp and SurA, formation of OMP-Skp₃ complexes is favored over formation of OMP-SurA complexes and Skp should be the primary binding partner for OMPs if both Skp and SurA are present at the same sufficiently high concentration. It is possible that the expression levels of a single chaperone species are not high enough to solubilize all OMPs. Therefore, more than one species would be necessary to facilitate passage of all OMPs through the periplasm. Different binding affinities of the various chaperones in conjunction with regulated expression of chaperones and OMPs could lead to a preferred sequence of the assembly of OMPs into the OM.

It is interesting that the dissociation constant for complexes of Skp₃ with YaeT is much smaller, by about two orders of magnitude, than the dissociation constants of all other OMP-Skp₃ complexes (Table 2), indicating that the binding affinity of Skp₃ is strongest for YaeT. The periplasmic domain of YaeT, which binds polypeptides,^{26,34,35} may contribute to the increased stability. For folded YaeT in the outer membrane, the periplasmic domain may bind Skp to receive the unfolded OMP from the OMP-Skp₃ complex after passage through the periplasm. The possibility of selective binding of OMPs by the various periplasmic chaperones was proposed in a genetic study.³⁶ Interestingly the *skp* gene follows the *yaeT* gene immediately downstream on the chromosome (e.g. see the online supplement to Voulhoux *et al.*²⁵). Both Skp and YaeT are involved in the assembly pathway of OMPs. A preferred binding of Skp to YaeT in the periplasm and the possibility that expression levels of Skp may not be high enough to bind all the OMPs might explain why depletion of other chaperones like SurA led to reduced levels of OMPs like OmpA in the OM of *E. coli*. Each periplasmic chaperone, in addition to a general chaperone-function, may selectively bind a certain class of OMPs with higher affinities, similar to what is observed here for Skp binding to the aqueous form of YaeT. Preliminary data from our laboratory indicate that SurA can facilitate folding of OmpA into preformed phospholipid bilayers in the absence of both Skp and LPS, while Skp required the presence of LPS to facilitate folding of OmpA.¹² This difference very likely is a consequence of the stability of the OMP-Skp₃ complexes, which is weakened by specific interactions with the negatively charged LPS. SurA binds OmpA with much weaker affinity, similar to OmpG and OmpF,³³ and the far less stable complexes do not require much activation to deliver the aqueous forms of OMPs to a membrane.

Skp recognizes unfolded structures

There are no homologs of Skp in eukaryotic cells and therefore it did not surprise us that neither BSA nor hVDAC1 bound to Skp. However, Skp prevented aggregation of lysozyme from hen eggs²⁰

and promoted expression of soluble antibody fragments,¹⁰ indicating a more general role of Skp₃ as a molecular chaperone. Soluble BSA was not bound by Skp₃, because it folded too fast upon dilution of the urea.³⁷ For the same reason, no Skp₃ binding was observed for the soluble 154 residue periplasmic domain of OmpA, which folds independently of the transmembrane domain.⁸ Skp recognizes unfolded and misfolded structures. Although hVDAC1 required a lipid bilayer for correct folding, our previous work indicated that this OMP rapidly developed large amounts of β -sheet secondary structure in aqueous solution,³⁸ preventing the binding of Skp in our present study. In contrast, OmpA developed most of its β -sheet secondary structure not in aqueous solution, but when it inserted into the hydrophobic core of the membrane.³⁹

Skp interacts with hydrophobic residues of the transmembrane domain of OMPs

Skp₃ requires a significant content of non-native structure in its binding partner to form stable complexes. The binding partner must have either a high level of conformational flexibility or surface-exposed residues that are inaccessible when folded, or both. For folded β -barrels of transmembrane proteins, the polar residues are oriented towards the barrel lumen, while hydrophobic residues form the outer surface. The hydrophobic residues include the fluorescent tryptophan residues that are exposed to the lipids predominantly in their membrane-inserted, folded forms. The aqueous forms of OMPs, which are obtained after dilution of the urea, have, at least partially, an inside-out conformation, in which hydrophobic residues are predominantly buried inside a collapsed conformation, while the charged and polar residues are surface-oriented and water-exposed. The fluorescence data obtained for the bacterial OMPs indicated that the tryptophan residues are partially shielded from the aqueous space in the hydrophobically collapsed OMPs. In OMP-Skp₃ complexes, they are much less accessible, which is obvious from fluorescence intensity increases (F_C/F_{AQ}), shifts in λ_{max} of the spectra, and from the quenching experiments with acrylamide. The tryptophan residues of the OMPs interact with hydrophobic residues of Skp₃. An interpretation could be that Skp₃ binding turns around the partial inside-out conformation of the aqueous forms of the OMPs, requiring their conformational flexibility. This could then be consistent with a lack of Skp binding to hVDAC1, which developed most of the β -sheet structure already in aqueous solution. The exposed hydrophobic surface of a largely preformed antiparallel β -sheet of hVDAC1 could be too rigid to interface well with the hydrophobic surface of Skp inside the Skp tentacle basket.

The flexible three-pronged forceps-like structure of Skp₃ suggests that OMPs are bound inside the basket defined by the six α -helices of Skp₃ (Figure 8). As pointed out previously,^{19,20} the abundance of

positively charged residues at the end of the tentacles and the flexibility of the tentacles likely ensure that the inside of the tentacle domain is accessible in solution. Charged and hydrophobic areas form patches over the entire surface of the tentacle domain, which has a high positive net charge (Figure 8). Inside the tentacle basket, there are hydrophobic patches formed between the helices of each monomer. Two of them are composed of residues F13, A17, V22, and L26 or of residues F30, A34, L37, F74, and F81. These hydrophobic residues also form the interface between the two helices of the Skp monomer (Figure 8(c)). It is quite possible that hydrophobic residues of the transmembrane domains of the OMPs are bound at these locations.

Skp binds OMPs by electrostatic interactions

Skp₃ bound OmpA at a pH between the isoelectric points of OmpA and Skp, i.e. when OmpA was negatively charged and Skp positively charged. NalP (or FomA) binding to Skp₃ was observed down to pH 6, at which both NalP and FomA are positively charged. Since the positively charged Skp₃ still bound, complex formation may be caused either by hydrophobic interactions or by charge-charge interactions with specific regions of NalP (or FomA), in which the residues aspartate and glutamate dominate. Skp₃ binding to NalP (or FomA) was not observed when Skp₃ lost its positive net charge at very basic pH. Therefore, we concluded that a partially asymmetric distribution of positive and negative charges in OMPs such as NalP, results in the binding of positively charged Skp₃ at regions of the NalP or FomA, where their negative charges dominate. An asymmetric charge distribution in NalP and FomA also explains the pH-dependence of the fluorescence spectra of NalP and FomA, which indicated strong conformation changes in both OMPs due to protonation below pH 6 and pH 7, i.e. well below their calculated isoelectric points (Figure 3(b) and (c)). Below this pH region, NalP and FomA become more compact, as indicated by increased fluorescence and by shifts in λ_{max} towards shorter wavelengths. This is very likely a consequence of reduced charge-charge repulsion and reduced hydration of side-chains upon neutralization by protonation.

The periplasmic turns are regions of the OMPs that are rich in negatively charged aspartate and glutamate residues (Table 3). Similarly, some outer loops of certain OMPs may also carry a negative net charge although they are often positively charged. For instance, FhuA displayed selectivity for negatively charged lipids in lipid bilayers, because positive charges dominated in the outer loops near the lipid headgroup region.⁴⁰ The loops of NalP have a net charge of +2, while the turns have a net charge of -3. In the OM, the OMP loops face the negatively charged LPS. The loops may contain positively charged binding sites for the phosphate groups of LPS. In the short periplasmic turns,

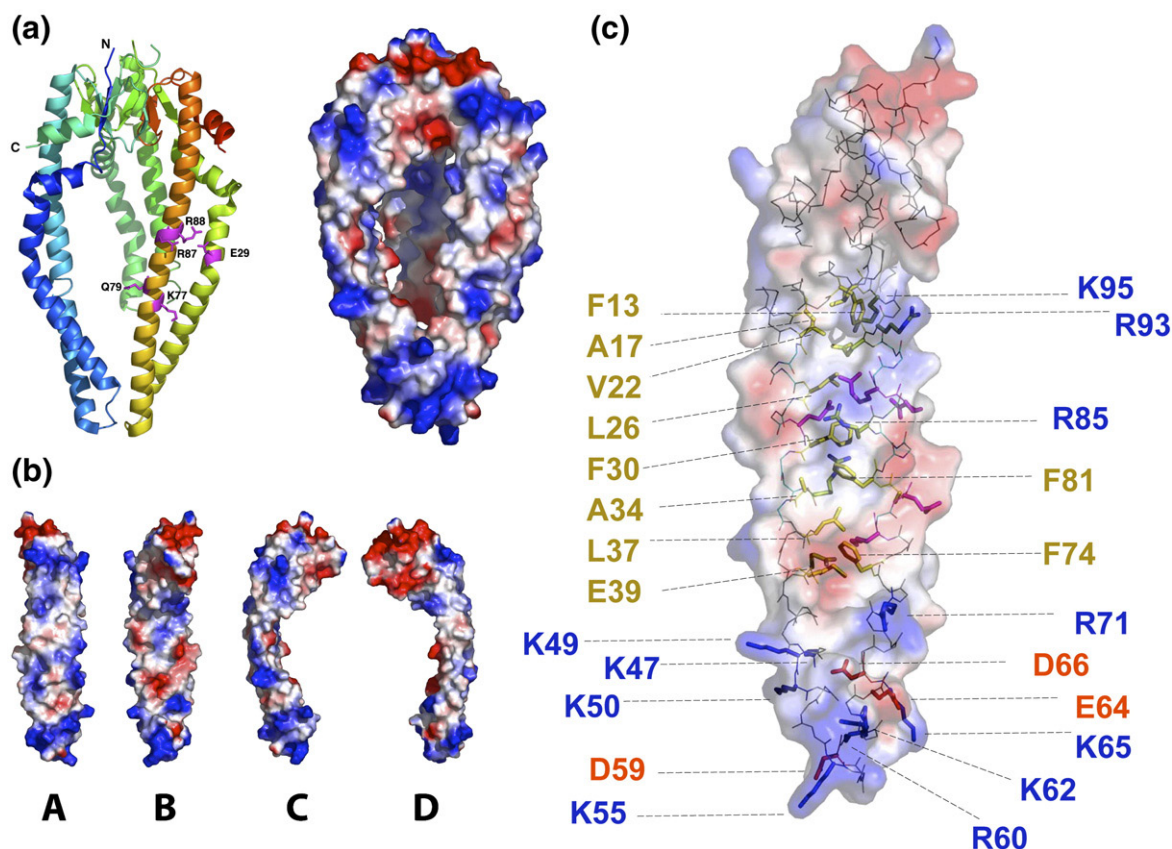


Figure 8. Crystal structure of Skp (PDB1SG2¹⁹ and 1U2M²⁰). (a) The Skp trimer is composed of a tentacle domain and an association domain. The association domain contains a nine-stranded β -barrel. This domain is surrounded by three pairs of α -helices that are about 65 Å long and form the tentacle domain. Each 141 residue Skp monomer contains a predicted LPS-binding site composed of residues K77, R87, and R88. The spatial orientation of these residues matches a conserved LPS-binding motif previously described for the OMP FhuA⁴¹ at a root-mean-square (rms) deviation of 1.75 Å.²⁰ The surface of the trimer is strongly positively charged, because each monomer contains 28 lysine and arginine residues, but only 19 aspartate and glutamate residues. (b) The charged residues are distributed asymmetrically, and most of the negatively charged aspartate and glutamate residues are found in the association domain, while the positively charged residues are found at the other end at the tips of the tentacle domain, leading to a large dipole moment of ~ 3700 Debye (770 eÅ).¹⁹ (c) The tip of each tentacle contains eight positively charged residues (K and L), but only three negatively charged residues (D and E) and is the region of the highest positive charge density.

negatively charged residues are found sometimes adjacent in pairs, which increases the local charge density and the attraction to the positively charged surface of Skp. It is expected that in the inside-out water-collapsed state of OMPs, the charged residues would be water-exposed at the OMP surface. Negatively charged turn or loop regions may therefore interact with the most basic regions of Skp₃. The Skp trimer has the highest positive charge density at the tips of the three tentacles (Figure 8) which each contain eight basic (six lysine and two arginine), but only three acidic residues (Figure 8(c)). Possibly, these three tip regions are sites in Skp₃ that contribute most strongly to the observed electrostatic interactions, in particular if the hydrophobic residues of the transmembrane strands are bound inside the tentacle basket as suggested.^{19,20} Interactions of the positively charged tentacle tips of Skp₃ with negatively charged and polar residues of the periplasmic turns in combination with hydrophobic interactions of the strand regions of the OMPs inside the Skp₃ tentacle basket could then also mean that

the OMPs adopt some kind of orientation upon binding to Skp₃.

LPS modulates the topology of the OmpA·Skp₃ complex

It has been shown that Skp alone inhibits folding of OmpA and that LPS reverses the retarding effect of Skp₃ by interacting with the OmpA·Skp₃ complex.¹² Our acrylamide quenching experiments indicated that LPS changes the topology of the OmpA·Skp₃ complex and partially exposes the fluorescent tryptophan residues of OmpA to the aqueous environment (Figures 6 and 7) even when present at only low molar ratios of LPS per OmpA·Skp₃ complex. The stoichiometry estimated for LPS binding to OmpA·Skp₃ complexes (Figure 6) corresponds well to our previous observations of optimal OmpA folding kinetics and folding yields at an LPS content of 2–7 mol per mol OmpA·Skp₃ complex.¹² The fluorescence quenching experiments with acrylamide suggested that LPS is not able to

Table 3. Acidic and basic residues in the periplasmic turns of some OMPs

OMP	PDB	Z _{TMPS} ^a	Z _T ^b	D ^c	E ^c	K ^c	R ^c	T-charge ^d
OmpA	1BXW	8	3	2	1	0	0	-3
NalP ^e	1UYN	12	5	2	1	0	0	-3
OmpG	2IWW	14	6	5	1	0	0	-6
OmpW	2F1V	8	3	3	0	0	1	-2
OmpT	1I78	10	4	1	2	0	0	-3
FadL	1T16	14	6	6	0	0	2	-4
OmpF	2OmF	16	7	4	1	1	0	-5
LamB	1MAL	18	8	3	2	1	0	-4
FhuA	2FCP	22	10	11	1	2	2	-8

^a Number of transmembrane β -strands of the OMP.^b Number of periplasmic β -turns of the OMP.^c Numbers of aspartate (D), glutamate (E), lysine (K), and arginine residues (R) in the periplasmic turns of these OMPs.^d Total charge of all turns.^e Translocator domain of NalP, residues 776–1083.

completely displace Skp₃ from OmpA (Figure 7). The LPS core region contains two phosphate and two diphosphate groups, which are negatively charged. Since the outer surface of Skp is very polar and positively charged in the tentacle domain, the binding of individual LPS molecules to Skp₃ involves electrostatic interactions with the OmpA·Skp₃ complex. Such interactions would be expected in the predicted LPS binding-site of Skp that contains two positively charged arginine and lysine residues,^{20,41} or in the tip regions of Skp, or both. Although covalent bonds are not formed, the interaction of LPS with OmpA·Skp₃ is exergonic and provides a significant amount of free energy for a conformation change in the complex. This weakens the interaction of Skp₃ with bound OmpA and results in a topology change of the complex that partially exposes the tryptophan residues of OmpA to the aqueous environment. In the fluorescence experiments, this was reflected by a decrease of intensity and by an increase of λ_{\max} of the tryptophan fluorescence of OmpA upon LPS binding (Figure 6). In the proximity of a lipid bilayer, the LPS-induced topology change led to the release of the OMP from the complex for insertion and folding into the membrane.¹² In conclusion, our results suggest that charge-charge interactions are necessary for binding and release of OMPs by the periplasmic chaperone Skp₃.

Materials and Methods

Purification of Skp, WT-OmpA, TMD-OmpA, FomA and hVDAC1

Wild-type OmpA,¹² TMD-OmpA,⁴² FomA,³⁰ and hVDAC1³⁸ were over-expressed and purified from *E. coli* in their unfolded forms as described. Skp was isolated from *E. coli* as described.¹²

Purification of OmpG, NalP and YaeT

The *ompG* gene (lacking the nucleotide sequence encoding the signal peptide) was amplified by PCR

(60 °C annealing temperature) using 50 ng of *E. coli* MG1655 genomic DNA as template and the primers

5'-TAGGGCCATATGGAGGAAAGGAACGACTGG-3'
5'-CTACTCGAGTCAAAGCTTGAACGAGTAATT-
TACGCCG-3'

The PCR product was cloned into the pET29b vector (Novagen) using the NdeI/XhoI restriction sites, yielding pET29OmpGm2. Plasmid pET29OmpGm2 was transformed into *E. coli* BL21 (DE3) (Stratagene) to express OmpG in the form of inclusion bodies. A culture (20 ml) was grown for 18 h and used to inoculate 2 l of LB medium. After 3 h, IPTG was added to a final concentration of 0.1 mM. After 4 to 6 h of induction, cells were harvested by centrifugation for 30 min at 1500 g at 4 °C. The wet cell paste was resuspended in 40 ml of 20 mM Tris-HCl (pH 8.0), 0.1% (v/v) β -mercaptoethanol while cooling in an ice/water bath. Lysozyme was added to a concentration of 50 μ g/ml and the mixture was stirred for 30 min at 22 °C. The solution was then sonicated for 30 min using a Branson ultrasonifier W-450D (20% power, 50% pulse cycle) with a macrotip while cooling in an ice/water bath. The buffer and soluble proteins were removed by centrifugation at 3000g for 30 min at 4 °C. The pellet was washed in 20 ml of 20 mM Tris-HCl (pH 8.0), 0.1% β -mercaptoethanol, 1 M urea. The supernatant was removed by centrifugation at 5000 g for 30 min at 25 °C. The pellet was then dissolved in 40 ml of 20 mM Tris-HCl (pH 8.0), 0.1% β -mercaptoethanol, 8 M urea. The solution was loaded onto a Q-Sepharose FF column (Amersham) and the proteins were eluted by a NaCl gradient (0–100 mM). The yields of OmpG were about 50 mg from 1 l of culture.

Restriction sites for NcoI and BamHI were used for cloning *yaeT* into the pET15b vector (Novagen). The resulting plasmid, pET15_EcOMP85, was transformed into *E. coli* BL21 (DE3). His₆-tagged YaeT was expressed in the form of inclusion bodies in *E. coli*. YaeT was purified from these inclusion bodies as described for OmpG.

For expression of the translocator domain of NalP (residues D776 to F1083 of NalP), plasmid pPU320 was transformed into *E. coli* BL21 (DE3) (Stratagene) and the NalP protein was purified as described.²⁴

Purification of R-LPS

E. coli rough mutant F576 was cultivated as described,⁴³ and its LPS (R2 core type, $M \approx 3900$ g/mol) was isolated as described.⁴⁴

Fluorescence spectroscopy

Fluorescence spectra were recorded as described¹² on a Spex Fluorolog-3 spectrofluorimeter with double monochromators in the excitation and emission pathways. The excitation wavelength was 295 nm, and the bandwidths of the excitation monochromators were 2.5 nm. The bandwidths of the emission monochromators were 5 nm. The integration time was 0.05 s, and an increment of 0.5 nm was used to scan spectra in the range of 310–380 nm. Background intensities of Skp in the absence of OMPs were subtracted. Emission maxima of the spectra were calculated by fitting a lognormal distribution to the experimental spectra using Igor Pro 6.0 (WaveMetrics, Oregon).⁴⁵ All experiments were done at 25 °C. Each experiment was performed three times, and average intensities and emission maxima were calculated.

Binding of Skp to OMPs monitored by fluorescence spectroscopy

To monitor binding of Skp to OMPs, the background spectrum of Skp at a selected concentration was first recorded in 990 μL of urea-free 10 mM glycine buffer (pH 9.0). The concentrations of Skp were chosen to obtain Skp/OMP molar ratios ranging from 0 to 8. Then 10 μL of a stock solution of an unfolded OMP in 8 M urea was added and the fluorescence spectrum was recorded after 30 s of incubation. The background spectra of Skp were subtracted. Separate sets of fluorescence spectra were recorded for each OMP at final concentrations of 0.37 μM OmpG, 0.16 μM YaeT, 0.55 μM NalP, and 0.83 μM TMD-OmpA. In these experiments, the OMPs were diluted from concentrated stock solutions to reduce the urea concentration 100-fold.

Binding functions were fit to the experimental data assuming one class of identical binding sites in each OMP. In this case, the average concentration of bound Skp is given by:⁴⁶

$$[B]/[{}^t\text{OMP}] = n K_{\text{ass}}[F]/(1 + K_{\text{ass}}[F]) \quad (1)$$

where n is the number of binding sites, K_{ass} is the association constant, $[{}^t\text{OMP}]$ is the total concentration of the OMP, and $[F]$ is the concentration of the free ligand. Substitution of the free ligand with the total ligand concentration, $[L_0] = [B] + [F]$ and some rearrangements led to:

$$[B] = 0.5\{K_{\text{ass}}^{-1} + [L_0] + n[{}^t\text{OMP}]\} - ((K_{\text{ass}}^{-1} + [L_0] + n[{}^t\text{OMP}])^2 - 4n[{}^t\text{OMP}][L_0])^{1/2} \quad (2)$$

where $[B]$ and $[{}^t\text{OMP}] - [B]$ are the concentrations of bound and free OMP, respectively. The fluorescence signal of the OMP in binding experiments is a linear combination of the concentrations of bound and free OMP:

$$F = f_b[B] + f_f([{}^t\text{OMP}] - [B]) \quad (3)$$

where f_b and f_f are constants that describe the fluorescence contributions of bound and free OMP, respectively.

Skp binding to unfolded OMPs as a function of pH

To determine whether binding of Skp is pH-dependent, spectra of OMPs were recorded as described above, but in other buffers (10 mM) at various pH values: citrate, pH 3–6; Hepes, pH 7; Tris, pH 8; glycine, pH 9–10; and Caps, pH 11. A fivefold molar excess of Skp was used. Background spectra of Skp were recorded and subtracted. The OMP concentrations were 0.43 μM OmpA, 0.47 μM NalP, 0.65 μM hVDAC1, and 0.20 μM FomA.

Skp binding to unfolded OmpA or to unfolded OmpG as a function of ionic strength

To determine binding at different ionic strengths, experiments were performed in 10 mM Tris (pH 8.0), containing either 0, 0.1 M, 0.2 M, 0.5 M or 1 M NaCl. Titrations with Skp were done with 0.43 μM OmpA and 0.37 μM OmpG as described above.

Dynamic light-scattering

The hydrodynamic radius and particle mass of Skp in solution were measured by dynamic light scattering using

a Dynapro instrument (Wyatt Technology Corp.). Solutions of Skp (2 g/L) in 12 μL of buffer were first passed through aluminum oxide filters (Whatman) of 0.02 μm pore size. The intensity of the scattered light was measured and the hydrodynamic radius (R_H) was analyzed using the software Dynamics, Ver. 6. R_H was calculated from the theoretical hydrodynamic radius of an ideally spherical particle, R_{HTH} , and the ratio of the frictional coefficients, f/f_0 for a hydrated *versus* non-hydrated sphere according to:

$$R_H/R_{\text{HTH}} = f/f_0$$

The molecular mass of the Skp particles can be calculated from the hydrodynamic radius:

$$M = (4/3)\pi[(f_0/f)R_H]^3/[(V_P + H_{\text{AQ}})N_A] \quad (4)$$

where V_P is the specific volume of the particle (for a protein, an average of 0.73 cm^3/g is normally assumed⁴⁷), H_{AQ} is the hydration volume of the protein (typically assumed to be 0.35 cm^3/g protein). For a spherical protein, $f/f_0 = 1.2$. For Skp, which resembles a prolate ellipsoid, the Perrin factor is 1.02, leading to $f/f_0 = 1.22$. N_A is Avogadro's number.

CD spectroscopy

Far-UV CD spectra were recorded at room temperature with a Jasco 715 CD spectrometer (Jasco, Tokyo, Japan) using a 0.5 mm path length cuvette. Three scans were accumulated from 190–250 nm with a response time of 8 s, a bandwidth of 1 nm and a scan speed of 50 nm/min. Background spectra without Skp were subtracted. The concentration of Skp (30 μM) was determined for each sample using the method of Lowry.⁴⁸ The recorded CD spectra were normalized to the mean residue molar ellipticity $[\Theta](\lambda)$, given by:

$$[\Theta](\lambda) = 100 \frac{\Theta(\lambda)}{c \cdot n \cdot l} \quad (5)$$

where l is the path length of the cuvette in cm, $\Theta(\lambda)$ is the recorded ellipticity in degrees at wavelength λ , c is the concentration in mol/L, and n is the number of amino acid residues of Skp (141).

LPS binding to the OmpA·Skp₃ complex

The fluorescence spectra of OmpA·Skp₃ complexes were recorded in the absence and in the presence of increasing concentrations of LPS in 10 mM Tris-HCl (pH 8.0). The concentrations of OmpA and Skp were 0.43 μM and 1.3 μM , respectively. The LPS concentration ranged from 0 to 13 μM . Background spectra of Skp were subtracted.

Fluorescence quenching experiments with acrylamide

Acrylamide quenching experiments were performed in 10 mM Tris-HCl (pH 8.0) with 0, 0.1 M, 0.2 M, 0.3 M and 0.4 M acrylamide. The concentrations of OmpA, Skp and LPS were 0.85 μM , 4.25 μM and 6 μM , respectively. The data were fit to the Stern-Volmer function: $F_0/F = 1 + K_{\text{sv}}[Q]$. K_{sv} is the Stern-Volmer constant and $[Q]$ is the concentration of acrylamide. F was corrected for the inner filter effect for absorption to obtain $F_{\text{corr}} = F_{\text{meas}} \times 10^{1/2 \times 0.55 \times [Q]}$.

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