

# Additions and Corrections

1964, Volume 3

**G. P. Haight, Jr., David C. Richardson, and Nancy Hall Coburn:** A Spectrophotometric Study of Equilibria Involving Mononuclear Chromium(VI) Species in Solutions of Various Acids.

Page 1778. The correct labels for Figure 1 are: . . .,  $\text{CrO}_4^{2-}$ ; ---,  $\text{HCrO}_4^-$ ; —,  $\text{H}_2\text{CrO}_4$ ; - · - · -,  $\text{Cr(VI)}$  in 3 *M*  $\text{HClO}_4$ . A corrected figure is printed below.

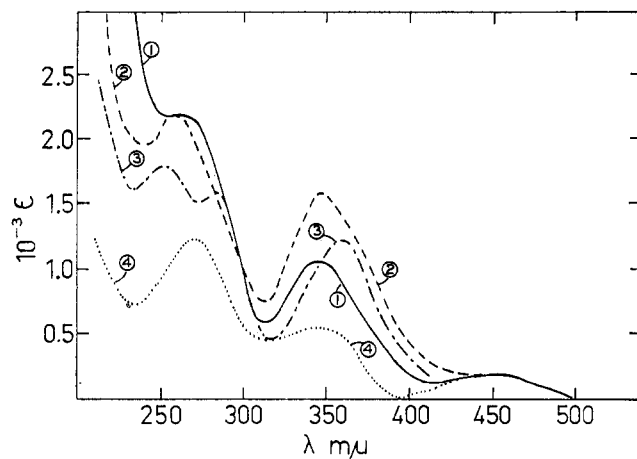


Figure 1.—Spectra of Cr(VI) species ( $\epsilon$  is molar absorptance): 1, solid line,  $\text{CrSO}_7^{2-}$ , measured directly; 2, dashed line,  $\text{HCrO}_4^-$ , measured directly; 3, dash-dot line,  $\text{CrO}_3\text{Cl}^-$ , measured directly; 4 dotted line,  $\text{H}_2\text{CrO}_4$ , calculated from mixed spectrum and  $K_D$ .

G. P. HAIGHT, JR.

1965, Volume 4

**Robert F. Pasternack and Robert A. Plane:** Solvation of Cobalt(II) and Nickel(II) Ions in Acetone-Water and Ethanol-Water Solutions.

Page 1172. The fifth entry in Table IV for  $K_1$  should read 0.55 and the fourth entry in Table V for  $K_2$  should read  $4.9 \times 10^{-2}$ .—R. A. PLANE

1966, Volume 5

**E. O. Schlemper and Doyle Britton:** The Crystal Structure of Trimethyltin Cyanide.

Page 507. In column 2, lines 1–3, the two infrared frequencies should be interchanged.—DOYLE BRITTON.

**James R. Preer and G. P. Haight, Jr.:** Solubility Studies of Tetramethylammonium Salts of Complex Halides. V. Tris-(tetramethylammonium) Enneabromodibismuthate(III).

Page 657. The ordinate for Figure 1 should be labeled  $[A^+]^{-3/2}$  and eq 6 should read  $S \propto [(\text{CH}_3)_4\text{N}^+]^{-3/2}$ .—G. P. HAIGHT, JR.

**Chin Hsuan Wei, Glenn R. Wilkes, Paul M. Treichel, and Lawrence F. Dahl:** Preparation and Structure of a Tetrameric Cyclopentadienyliron Sulfide,  $[\text{C}_5\text{H}_5\text{FeS}]_4$ .

Page 901. In Table I, the value of  $x$  for  $\text{C}_7$  should be 0.3971 (not 0.3791).

Page 903. In Table V(d), the distance of  $\text{C}_{12}$  from the plane is  $-0.02$  (not  $-0.20$ ).—L. F. DAHL