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Detection of Two Naturally Occurring Structural Isomers of Partially Hydrolyzed Fumonisin B₁ in Corn by On-Line Capillary Liquid Chromatography–Fast Atom Bombardment Mass Spectrometry

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Fifteen corn kernel and corn screening samples were analyzed for the presence of fumonisin toxins. Samples were extracted with acetonitrile/water (1:1) and cleaned up with C₁₈ solid phase extraction (SPE) cartridges. Analysis of fumonisins was carried out by on-line capillary liquid chromatography–fast atom bombardment mass spectrometry (capillary LC/FAB/MS). Two isomers of partially hydrolyzed fumonisin B₁ (PHFB₁) were detected in five samples, and *N*-acetyl fumonisin B₂ (FA₂) was detected in one sample. Hydrolyzed fumonisin B₁ (HFB₁) was also detected in one sample, but further confirmation is needed.

Keywords: *Mycotoxin; fumonisin; capillary LC/FAB/MS*

INTRODUCTION

Fumonisin (Figure 1) are a family of mycotoxins produced mainly by *Fusarium moniliforme* Sheldon (Bezuidenhout et al., 1988; Gelderblom et al., 1988, 1992; Thiel et al., 1991; Plattner et al., 1992; Branham and Plattner, 1993), a common fungal contaminant of corn worldwide (Marasas et al., 1984). Consumption of *F. moniliforme* contaminated corn has been proven to be the causal factor of equine leukoencephalomalacia (ELEM) in horses (Kriek et al., 1981) and also to be correlated with the high incidence of human esophageal cancer in South Africa (Marasas et al., 1984) and China (Yang, 1980). Fumonisin B₁ (FB₁), the most abundant naturally occurring analogue among several discovered fumonisins, causes ELEM in horses (Kellerman et al., 1990; Wilson et al., 1992) and pulmonary edema in swine (Harrison et al., 1990; Colvin and Harrison, 1992).

Chemically, FB₁ is a diester of propane-1,2,3-tricarboxylic acid (TCA) and 2-amino-12,16-dimethyl-3,5,10,-14,15-pentahydroxylcosane. The C-14 and C-15 hydroxyl groups are involved in ester formation with the terminal carboxy group of propane-1,2,3-tricarboxylic acid (Bezuidenhout et al., 1988). Other B-series fumonisins are fumonisin B₂ (FB₂), B₃ (FB₃), and B₄ (FB₄) (Bezuidenhout et al., 1988; Gelderblom, 1992; Plattner et al., 1992), which are the C-10 and C-5 dehydroxy and the C-5,10 didehydroxy analogues of FB₁, respectively. The A-series fumonisins include the *N*-acetates of FB₁ (FA₁) and FB₂ (FA₂). Fumonisin C₁ (FC₁), which lacks the amino-end terminal methyl group of FB₁, has been reported (Branham and Plattner, 1993).

Since the first report of the natural occurrence of FB₁ in corn in 1990 (Sydenham et al., 1990), contamination of corn and corn-based feed or food samples by FB₁ and FB₂ as well as FB₃ has been reported by many laboratories (Plattner et al., 1990; Sydenham et al., 1991; Bane

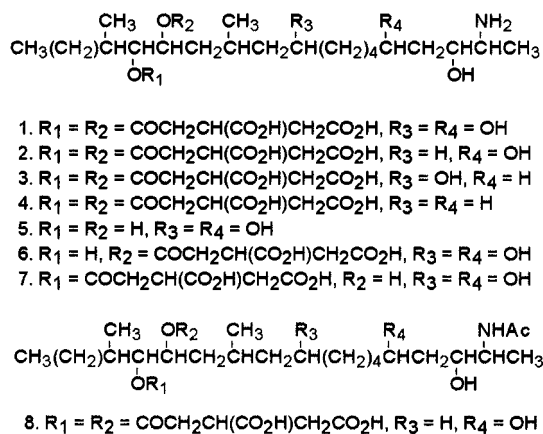


Figure 1. Chemical structures of (1) fumonisin B₁ (FB₁), (2) fumonisin B₂ (FB₂), (3) fumonisin B₃ (FB₃), (4) fumonisin B₄ (FB₄), (5) hydrolyzed FB₁ (HFB₁), (6 and 7) partially hydrolyzed FB₁ (PHFB₁), and (8) *N*-acetyl FB₂ (FA₂).

et al., 1992; Pittet et al., 1992; Thiel et al., 1992; Hopmans et al., 1993; Rice et al., 1994), while the natural occurrence of FB₄, A-fumonisin, and C-fumonisin has not been reported. Hydrolyzed FB₁ (HFB₁) was detected in tortilla chips, masa, and canned yellow corn (Hopmans et al., 1993). It was suggested that the occurrence of HFB₁ in these food samples resulted from hydrolysis of both TCA groups from FB₁ during the food preparation process, which involved a step with high pH and heat. Two structural isomers of partially hydrolyzed FB₁ (PHFB₁), formed by the hydrolysis of one of the TCA groups, were detected in the feces of vervet monkeys dosed with FB₁ (Shephard et al., 1994) and corn cultural material of *F. moniliforme* treated with Ca(OH) (Sydenham et al., 1995). PHFB₁ and HFB₁ were also found in fecal samples collected from cattle, sheep, and rats fed with diets mixed with *F. moniliforme* cultural material (Rice et al., 1994).

Liquid chromatography (LC) with fluorescence detection has been commonly used for the analysis of fumonisins in contaminated food or feed products. Derivatization of fumonisins, however, is required for the LC analysis (Thiel et al., 1991; Stack and Eppley, 1992;

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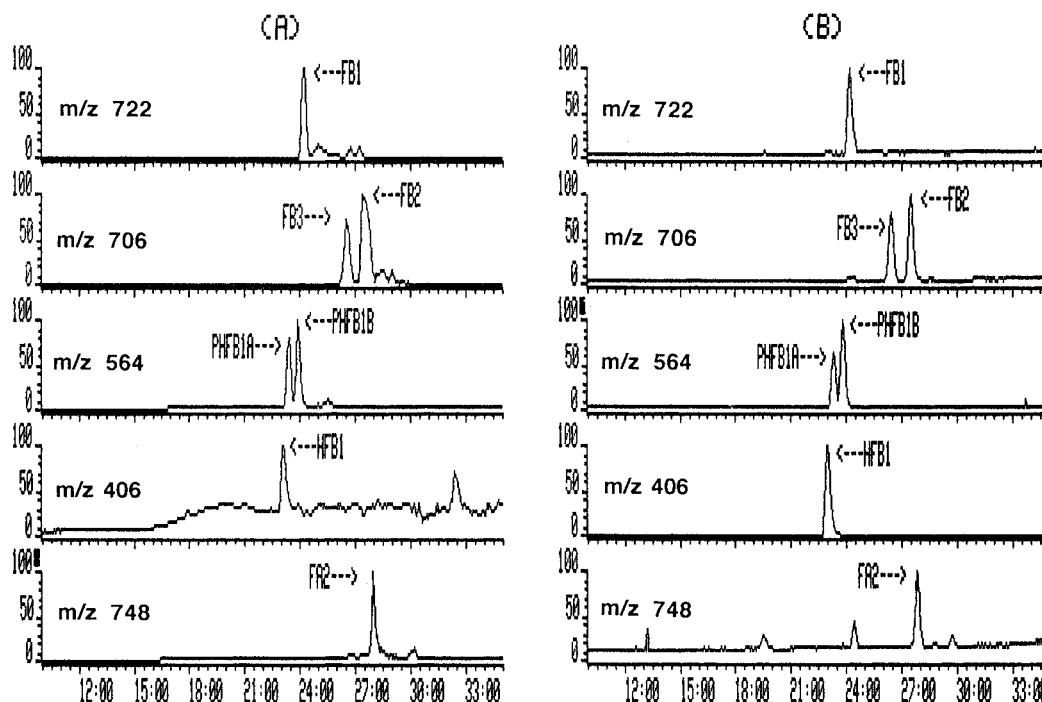


Figure 2. Capillary LC/FAB/MS mass chromatograms (m/z 722, 706, 564, 406, and 748) of (A) extract of sample FS870 and (B) fumonisin standard mixture containing FB₁, FB₂, FB₃, PHFB₁, HFB₁, and FA₂.

Sydenham et al., 1992; Hopmans et al., 1993; Rice and Ross, 1994). Analysis by gas chromatography/mass spectrometry (GC/MS) also requires hydrolysis followed by derivatization of fumonisins (Plattner et al., 1990). Nonderivatized fumonisins have been analyzed by mass spectrometry using various ionization techniques including fast atom bombardment (FAB) (Korfman et al., 1991; Chen et al., 1992; Mirocha et al., 1992; Holcomb et al., 1993), ion spray (IS) (Chen et al., 1992), thermospray (TS) (Korfman et al., 1991), liquid secondary ion (LSI) (Bezuidenhout et al., 1988; Plattner et al., 1990), and electrospray (ES) (Korfman et al., 1991; Caldas et al., 1995; Sydenham et al., 1995). Liquid chromatography coupled with ion spray mass spectrometry (LC/IS/MS) (Chen et al., 1992) and electrospray mass spectrometry (LC/ES/MS) (Doerge et al., 1994; Plattner, 1995; Musser, 1996) have been proven to be useful methods to analyze fumonisins in fungal cultural materials and contaminated food samples.

This paper reports the natural occurrence of the two isomers of PHFB₁ and of FA₂ in corn samples. Evidence for the natural occurrence of HFB₁ in corn is also presented. This paper also describes a microcapillary resolution and FAB detection of the fumonisins.

MATERIALS AND METHODS

Analytical Standards. FB₁ was provided by Dr. Robert M. Eppley of the Food and Drug Administration (Washington, DC). FB₂ and FB₃ were purchased from the Division of Food Sciences and Technology, Council for Scientific & Industrial Research, Pretoria, South Africa. FA₁ and FA₂ were provided by Dr. Hamed K. Abbas (USDA, ARS, Stoneville, MS).

Sample Preparation. Fifteen corn kernel or corn screening (broken corn kernels) samples collected in 1990 were ground to fine powder with a Stein Laboratories Mill (Fred Stein Laboratories, Atchison, KS). Twenty-five grams of each ground sample was extracted with 50 mL of acetonitrile/water (1:1) on a shaker for 1 h. A 2 mL aliquot of the filtered extract was diluted with 6 mL of distilled water and acidified to ca. pH 4 with 0.1 N HCl. The diluted and acidified extract was applied to a Waters C₁₈ Sep-Pak Classic solid phase extraction

(SPE) cartridge (Waters, Milford, MA), previously washed with 5 mL of acetonitrile followed by 5 mL of distilled water. The loaded cartridge was then washed with 2 mL of acetonitrile/water (3:17). Fumonisin were eluted with 2 mL of acetonitrile/water (7:3). Twenty-five grams of ground clean corn was spiked with 100 μ L of FB₁ standard solution (1 mg/mL) as a control. The same extraction and cleanup procedure was carried out for the control extract preparation.

Preparation of PHFB₁ and HFB₁ from FB₁. Five hundred milligrams of FB₁ was dissolved in 1 mL of 0.5 M NaOH and kept at room temperature for 15 min. The solution was then acidified to pH 4 with 1 N HCl and passed through Amberlite XAD-2 resin (Aldrich, Milwaukee, WI) packed in a disposable glass pipet and washed with acetone followed by distilled water. A mixture of FB₁, PHFB₁, and HFB₁ was eluted with 2 mL of MeOH.

Analysis of Fumonisin. The fumonisin mixture prepared from the hydrolysis of FB₁ was mixed with standards of FB₂, FB₃, FA₁, and FA₂ to make a standard mixture. The standard mixture and the cleaned up corn extracts were analyzed by on-line capillary liquid chromatography-fast atom bombardment mass spectrometry (capillary LC/FAB/MS) on a VG 7070EQ mass spectrometer (VG Analytical, Wythenshawe, Manchester, U.K.) equipped with a VG dynamic FAB ion source and an Ion Tech Xenon saddle field gas gun (Ion Tech, Teddington, Widdlesex, U.K.) operated at 8–9 kv. The mass spectrometer was interfaced with two Shimadzu LC-600 liquid chromatograph pumps (Shimadzu, Kyoto, Japan) through a 20 cm length, 320 μ m i.d. reverse phase capillary LC column. The column was packed with 3 μ m Phase Sep ODS2 C₁₈ (Phase Separations, Norwalk, CT). A 50 μ m i.d. fused silica capillary was used as delivery tubing between the column and the mass spectrometer. Samples were injected through a Valco CI4W internal sample injector (Valco Instruments, Houston, TX) with 0.5 μ L sample volume. The total flow rate (mixer flow rate) was 160 μ L/min, but a 124 mm length, 1.2 mm i.d. stainless steel tubing packed with 5 μ m Phase Sep ODS2 C₁₈ was used as a preinjector splitter to obtain a capillary column flow rate of 4 μ L/min. The Shimadzu LC-600 is capable of delivering 1 μ L/min; however, the higher total flow rate was necessary to achieve a repeatable gradient mixing in a relatively short time. The capillary column flow rate was measured at the end of the 50 μ m i.d. delivery capillary by measuring the time of filling a 20 μ L micropipet with a mobile

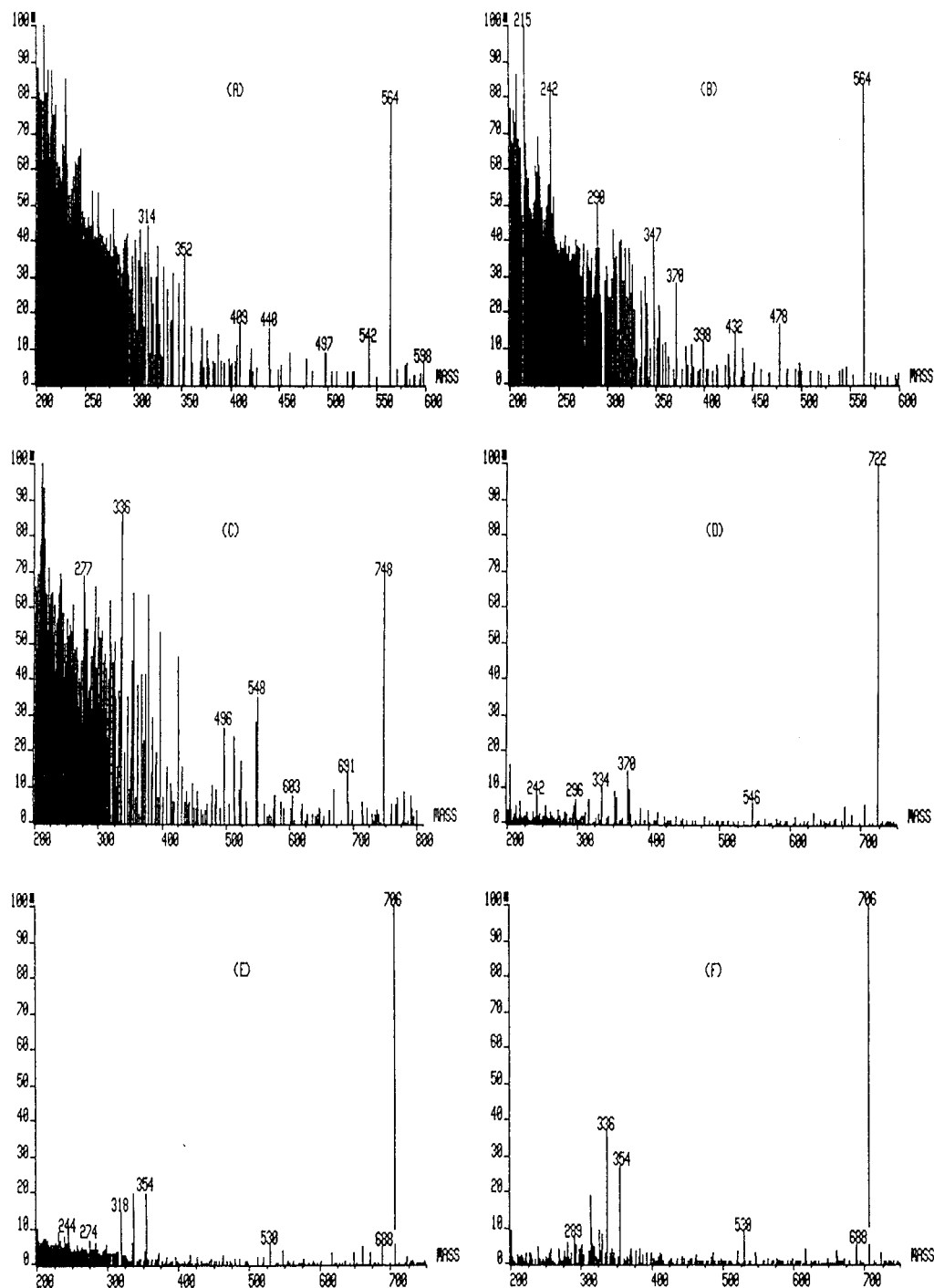


Figure 3. Background-subtracted FAB mass spectra of (A) PHFB₁a with [MH]⁺ at *m/z* 564, (B) PHFB₁b with [MH]⁺ at *m/z* 564, (C) FA₂ with [MH]⁺ at *m/z* 748, (D) FB₁ with [MH]⁺ at *m/z* 722, (E) FB₂ with [MH]⁺ at *m/z* 706, and (F) FB₃ with [MH]⁺ at *m/z* 706 in the extract of sample FS870.

phase. The desired capillary column flow rate was obtained by adjusting the total flow. The solvent system consisted of (A) water/glycerol/trifluoroacetic acid (TFA) (98:2:0.1) and (B) water/acetonitrile/glycerol/TFA (20:78:2:0.1). Two water/acetonitrile gradient programs were used for fumonisin analysis. All the samples and the standard mixture were run in the gradient system A: isocratic solvent A for 1 min followed by ramping the solvent B concentration from 0% to 80% in 14 min and maintaining the solvent B concentration at 80% for another 5 min. To achieve better separation for all the fumonisins, the extract of one corn screening sample FS870 and the standard mixture were also run in a longer gradient program B: isocratic solvent A for 1 min followed by ramping the solvent B concentration from 0% to 80% in 29 min and maintaining the concentration for another 5 min. All the samples and the standard mixture were analyzed by capillary

LC/FAB/MS in single ion monitoring (SIM) mode with sampling time set at 80 ms and interchannel time at 20 ms. FAB mass spectra of sample FS870 and fumonisin standards were obtained in full scan mode. The cleaned up sample extracts were directly analyzed in SIM mode without further sample concentration. For full scan FAB analysis, a 500 μ L aliquot of the extract of sample FS870 was evaporated to dryness in N₂ and redissolved in 200 μ L of acetonitrile/water (7:3) to make a concentrated sample. FAB spectra of FB₁, FB₂, FB₃, PHFB₁, and FA₂ were also obtained in full scan mode through capillary LC/FAB/MS.

RESULTS AND DISCUSSION

The protonated molecular ions ([MH]⁺) for FB₁, FB₂, FB₃, PHFB₁, HFB₁, and FA₂ are *m/z* 722, 706, 706, 564,

406, and 748, respectively. Figure 2A shows the mass chromatogram of those masses in the capillary LC/FAB/MS analysis of the corn screening sample FS870 by the gradient program B, suggesting the existence of FB₁, FB₂, FB₃, two isomers of PHFB₁, HFB₁, and FA₂. The peak retention time at each mass (24:10 for FB₁ at *m/z* 722, 26:25 for FB₃ at *m/z* 706, 27:29 for FB₂ at *m/z* 706, 23:16 and 23:47 for isomers of PHFB₁ at *m/z* 564, 22:58 for HFB₁ at *m/z* 406, and 27:51 for FA₂ at *m/z* 748) agreed with the corresponding fumonisin standard peak in the standard mixture analysis (Figure 2B). The mass chromatogram of *m/z* 564 in Figure 2A shows two peaks indicating the presence of two isomers of PHFB₁ in the corn screening sample. The two isomers of PHFB₁ with retention times of 23:16 and 23:47 were designated PHFB₁a and PHFB₁b, respectively. In this gradient program, the two PHFB₁ isomers eluted before FB₁ with a near base line separation from FB₁. In the short gradient program A, however, the retention time of FB₁ fell between the retention times of PHFB₁a and PHFB₁b, and the peaks were partially resolved from the FB₁ peak.

The identification of PHFB₁ and FA₂ as well as FB₁, FB₂, and FB₃ in the sample FS870 was confirmed by their full FAB spectra (Figure 3). Panels A and B show spectra of PHFB₁a and PHFB₁b, respectively. Both spectra give [MH]⁺ at *m/z* 564. Panel C shows the FAB spectrum of FA₂ with the [MH]⁺ at *m/z* 748. No fragment ions from the [MH]⁺ were observed in the spectra due to the relatively small amounts of the compounds and high background signals in the low mass area. The FAB spectrum of FB₁ shown in the panel D has a strong [MH]⁺ at *m/z* 722 and minor fragment ions at *m/z* 546 and 370, corresponding to loss of one and two TCA groups from [MH]⁺, respectively. Fragment ions resulting from sequential losses of H₂O from the ion at *m/z* 370 were also observed in the spectrum. All those fragment ions in the spectrum of FB₁ have been observed in the collisionally activated dissociation (CAD) daughter ion spectrum of *m/z* 722 from a FAB/MS/MS analysis of FB₁ by Korfmacher et al. (1991). Similar fragmentation patterns were seen in the spectra of FB₂ (panel E) and FB₃ (panel F). Both spectra show a strong [MH]⁺ at *m/z* 706 and a minor fragment ion [MH - TCA]⁺ at *m/z* 530 as well as [MH - 2TCA]⁺ at *m/z* 354. Water molecule loss from [MH]⁺ and sequential H₂O losses from the fragment ion at *m/z* 354 were also seen in the spectra. HFB₁ was not detected in the full scan analysis, apparently due to the very small quantity of the compound in the sample.

Korfmacher et al. (1991) detected an impurity component in a purified extract of corn culture material of *F. moniliforme* containing FB₁ and proposed the structure of PHFB₁ for the compound based on the CAD daughter ion spectrum of the ion at *m/z* 564 from a FAB/MS/MS analysis. This compound was suggested to be a new fumonisin-like compound in the corn culture material or an artifact caused by the sample purification procedure. In our case, the existence of PHFB₁ as an artifact is unlikely since a simple and mild procedure was used in the sample preparation. In order to eliminate the possibility of an artifact, a clean corn sample spiked with FB₁ (4 ppm) was extracted and cleaned up with the same procedure as fumonisin-contaminated corn samples. Capillary LC/FAB/MS analysis of the extract of this FB₁-spiked sample did not detect any PHFB₁ or HFB₁.

Table 1. Comparison of Peak Area of PHFB₁ (Combination of Two Isomers) and FB₁ in Corn Samples Analyzed by Capillary LC/FAB/MS

sample	PHFB ₁	FB ₁	PHFB ₁ :FB ₁
FS886	862	919	1:1.1
FS875	155	565	1:3.6
FS870	2155	4010	1:1.9
FS869	524	5320	1:10.2
FS867	998	7012	1:1.7

A total of 15 samples including corn kernels and corn screening were analyzed for FB₁, FB₂, FB₃, and PHFB₁ by capillary LC/FAB/MS in SIM mode. All the samples contained FB₁. FB₂ was detected in nine samples and FB₃ in seven samples. Four corn screening samples and one corn kernel sample contained two isomers of PHFB₁. Approximately equal amounts of PHFB₁a and PHFB₁b were detected in each of the five samples. Quantitation of PHFB₁ was not done since a quantitative standard was not available. Peak areas of FB₁ and PHFB₁ were compared to obtain an estimate for the relative levels of PHFB₁ in the samples (Table 1). Ratios of PHFB₁ (combination of two isomers) to FB₁ based on peak area ranged from 1:1.1 to 1:10.2. Samples in which PHFB₁ was not detected contained less than 0.1 µg/g FB₁.

We concluded that the two structural isomers of partially hydrolyzed FB₁ can be found in naturally contaminated corn samples. The existence of these two compounds in corn could be significant on the basis of their quantities relative to that of FB₁. The toxicology of these two compounds has not been determined. FA₂, the *N*-acetyl derivative of FB₂, was also reported as a natural contaminant of corn screening for the first time. FA₂ but not FA₁ was found in the sample and this agrees with the fact that the sample contained more FB₂ than FB₁, even though FB₁ usually is the most abundant fumonisin found in both fungal cultures and field samples. The possible existence of a very small amount of HFB₁ in the corn screening samples is also suggested. Further confirmation for this compound is needed, however.

This study also indicates that on-line reverse phase capillary LC/FAB/MS is a suitable method to analyze fumonisins in corn or corn-based samples. Samples can be directly analyzed after a simple extraction and cleanup procedure without derivatization. Sample concentration is not needed when analyzed in SIM mode because of its high sensitivity.

ABBREVIATIONS USED

CAD, collisionally activated dissociation; ES, electrospray; ELEM, leukoencephalomalacia; FA₁, *N*-acetyl fumonisin B₁; FA₂, *N*-acetyl fumonisin B₂; FAB, fast atom bombardment; FB₁, fumonisin B₁; FB₂, fumonisin B₂; FB₃, fumonisin B₃; FB₄, fumonisin B₄; FC₁, fumonisin C₁; GC/MS, gas chromatography/mass spectrometry; HFB₁, hydrolyzed fumonisin B₁; IS, ion spray; LSI, liquid secondary ion; PHFB₁, partially hydrolyzed fumonisin B₁; SIM, single ion monitoring; SPE, solid phase extraction; TCA, tricarboxylic acid; TFA, trifluoroacetic acid; TS, thermospray.

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