See discussions, stats, and author profiles for this publication at: https://www.researchgate.net/publication/8142664

Molecule-based photonically switched half-adder

ARTICLE in JOURNAL OF THE AMERICAN CHEMICAL SOCIETY · JANUARY 2005

Impact Factor: 12.11 · DOI: 10.1021/ja045577l · Source: PubMed

CITATIONS READS

136 64

9 AUTHORS, INCLUDING:



Joakim Andreasson

Chalmers University of Technology

58 PUBLICATIONS 2,364 CITATIONS

SEE PROFILE



Gerdenis Kodis

Arizona State University

75 PUBLICATIONS 2,399 CITATIONS

SEE PROFILE



Subhajit Bandyopadhyay

Indian Institute of Science Education and Re...

30 PUBLICATIONS 846 CITATIONS

SEE PROFILE



Thomas A Moore

Arizona State University

331 PUBLICATIONS 16,772 CITATIONS

SEE PROFILE



Published on Web 11/16/2004

Molecule-Based Photonically Switched Half-Adder

Joakim Andréasson,† Gerdenis Kodis,† Yuichi Terazono,† Paul A. Liddell,† Subhajit Bandyopadhyay,‡ Reginald H. Mitchell,[‡] Thomas A. Moore,*,[†] Ana L. Moore,*,[†] and Devens Gust*,[†]

Department of Chemistry and Biochemistry, Arizona State University, Tempe, Arizona 85287-1604, and Department of Chemistry, University of Victoria, P.O. Box 3065, Victoria, BC, Canada V8W 3V6

Received July 22, 2004; E-mail: gust@asu.edu

Molecule-based approaches to computing are scientifically intriguing and have potential technological utility.1-4 Molecular Boolean logic operators have been reported, 1-13 but mathematical calculations require several gates operating together. The half-adder, a combination of an AND gate and an XOR (exclusive OR) gate that share inputs, carries out binary addition; in combination, siliconbased half-adders perform the logic operations of modern computing. A few molecular half-adders have been studied previously. 14-16 As with most molecule-based logic elements, switching these pioneering constructs requires physical addition of chemical species that diffuse to a molecule and carry out reactions. Molecule-based logic gates with optical inputs and outputs would not require access for chemicals or wires and could operate on a fast time scale and in nonfluid media. Here, we report a molecule-based half-adder with all-photonic inputs and outputs that consists of two photochromic molecular switches and a third-harmonic-generating crystal (THG).

As shown in Table I, each input and output of a half-adder may be either on (designated 1) or off (0). The AND gate gives an on response only when both inputs, A and B, are on. The XOR gate generates an on output when either input is on, but not when the inputs are both on or both off. In binary addition, the XOR gate output is the sum digit, and the AND gate output is the carry digit.

Figure 1A shows a schematic diagram of the molecule-based half-adder. An optical cuvette contains a solution of AND gate 1 and XOR gate 2 molecules (Chart 1). Triad 1 consists of a porphyrin (P) linked to a fullerene electron acceptor (C_{60}) and a dihydropyrene photochrome (DHP). We have previously shown¹⁷ that, in the DHP-P-C₆₀ form (1c), laser excitation of the porphyrin (e.g., at 650 nm) initiates photoinduced electron transfer to give a DHP- $P^{\bullet+}-C_{60}^{\bullet-}$ charge-separated state, which evolves by charge shift to DHP $^{\bullet+}$ -P-C $_{60}$ $^{\bullet-}$. This final state, whose lifetime is 2 μ s, is detected optically via the transient absorption of $C_{60}^{\bullet-}$ at 1000 nm. When 1 is used in the AND gate, this transient absorbance signals the on state (vide infra). Visible (532 nm) irradiation of 1c leads to photoisomerization of the DHP to the cyclophanediene (CPD), yielding 10. Laser excitation of 10 produces a short-lived (<10 ns) CPD-P•+-C₆₀•- state, but charge shift does not occur due to the high oxidation potential of CPD. Long-lived charge separation is not observed (gate off). Irradiation of 10 with UV (355 nm) light converts the triad back to the DHP form 1c. Thermal interconversion is extremely slow.

When 1 is used in the photochemical AND gate, it is set initially in the CPD form. Irradiation of 10 with light from either input laser A (1064 nm) or input laser B (532 nm) does not lead to isomerization, and measurement of the transient absorbance generated by a readout laser shows no DHP $^{\bullet+}$ -P- $C_{60}^{\bullet-}$ (output *X off*). However, simultaneous irradiation by both lasers generates 355 nm

Table 1. Truth Table for the Half-Adder

input A	input B	output X AND gate (carry digit)	output YXOR gate (sum digit)	binary sum
0	0	0	0	00
1	0	0	1	01
0	1	0	1	01
1	1	1	0	10

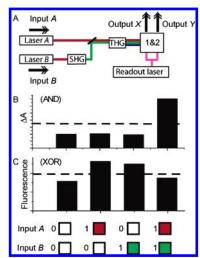


Figure 1. Half-adder and molecule-based logic gates. (A) The half-adder. Molecules 1 and 2 in 2-methyltetrahydrofuran reside in a cuvette labeled "1 & 2". Via pulsed (5 ns pulses at 10 Hz for 10 min) Nd:YAG lasers, the cuvette may be irradiated at either 1064 nm (input A) or 532 nm via a second-harmonic generator (SHG, input B). When both inputs are on, the sample is irradiated with 355 nm light from the third-harmonic generator (THG). Outputs X (AND gate) and Y (XOR gate) are the transient absorbance of the fullerene radical anion and porphyrin fluorescence, respectively. The readout laser (650 nm) provides light for the porphyrin excitation needed to produce these outputs. (B) Experimental output (absorbance change at 1000 nm) for AND gate triad 1 after exposure to inputs at 1064 nm (red squares) and/or 532 nm (green squares). The dashed line is a threshold level for detection of an on response. (C) Experimental output (fluorescence at 720 nm) from XOR gate dyad 2 after exposure to the inputs.

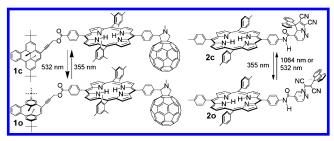
light via the third-harmonic-generating crystal, and ${\bf 1o}$ is switched to the DHP form, 1c. Long-lived $C_{60}^{\bullet-}$ absorbance due to DHP $^{\bullet+} P-C_{60}^{\bullet-}$ is detected via the readout laser system, and output X is on. Experimental observation of these logic operations within the half-adder is demonstrated in Figure 1B. After a logic operation, the state of the gate may be read out once or several times prior to resetting the gate to 10 by irradiation at 532 nm.

Molecular XOR gates are uncommon, and previous examples involve chemical diffusion. 1,6,9,18 Dyad 2, in combination with the THG, carries out the XOR function with only light as inputs and output. A porphyrin is linked to a photochromic dihydroindolizine (DHI) to form P-DHI dyad 2c. 19 Irradiation of the porphyrin of 2c

Arizona State University.

University of Victoria.

Chart 1. Structures of Molecular AND Gate 1 and XOR Gate 2



leads to typically strong fluorescence at 720 nm (on output). Irradiation of 2c at 355 nm opens the photochrome to the betaine (BT) form 20, which has a higher reduction potential than DHI. Light absorption by the porphyrin of P-BT is followed by rapid (50 ps) photoinduced electron transfer to yield P*+-BT*-, leading to strong quenching of the porphyrin fluorescence (output off). Irradiation of 20 at 532 nm (photoisomerization) or 1064 nm (thermal isomerization) causes reversion to 2c.

When dyad 2 acts as an XOR gate, the molecule is initially set in the 20 state. The fluorescence readout of the gate (output Y) is off (Figure 1C). Excitation with input laser A (1064 nm) thermally isomerizes 20 to DHI form 2c, electron transfer ceases, and the readout laser induces strong porphyrin fluorescence; output Y switches on. Similarly, irradiation of 20 with input laser B (532 nm) induces photochemical isomerization, and fluorescence output Y is turned on. However, if both input lasers are turned on, the 355 nm light does not induce net photoisomerization of 20, electron transfer still occurs, and the fluorescence output Y remains off. The requirements for an XOR gate are met. After a logic operation, the gate may be reset with 355 nm light.

The two molecular switches and the THG all work together as a half-adder (Figure 1A). The two shared inputs are laser beams at 1064 nm (input A) or 532 nm (input B). The output of the AND gate (output X, the carry digit) is the nanosecond transient absorbance of DHP•+-P-C₆₀•- in 1c. The output of the XOR gate (output Y, the sum digit) is the porphyrin fluorescence of ¹P-DHI in **2c**. (The porphyrin moiety in **1** is never significantly fluorescent.) Taken together, parts B and C of Figure 1 demonstrate the halfadder function experimentally. Initially, 1 is in the CPD-P-C₆₀ form 10, and 2 is in the P-BT state 20. When both input beams are off, the outputs of both gates are also off. The output of the half-adder is the binary sum 00 (first row of Table 1). When input A is on, molecule 1 (AND) is not affected, but 20 isomerizes to 2c. The molecule fluoresces, switching the XOR output on. The output of the half-adder is 01 (1 in base-10). If instead input B at 532 nm is switched on, molecule 10 is still not affected, and 20 is again isomerized to P-DHI, leading to an on response for output Y. Again, the output of the half-adder is 01. Finally, if both inputs A and B are switched on, the 355 nm light from the THG converts 10 to 1c, and long-lived charge separation gives an on response from output X (carry digit). Because the population of **20** is not significantly affected by the 355 nm light, the XOR output remains off. The corresponding output of the half-adder is the binary sum 10 (2 in base-10), completing the truth table. After readout, the half-adder is reset to the initial state using 355 nm (2) and 532 nm (1) light. The rate of isomerization at 532 nm is about 50 times higher for 1 than for 2, permitting the AND gate to be reset by the 532 nm light without significantly affecting the XOR gate. The results of photochemical cycling of 1 and 2 were investigated. Although triad 1 is unchanged after eight cycles, 2 undergoes some decomposition¹⁹ (~45%, see Supporting Information). Note that

the THG provides a transient signal that affects 1 and 2 differently. The photochromes of 1 and 2 are molecular switches that record the inputs for later readout, and the remaining parts of 1 and 2 provide the readouts.

Due to equipment limitations, the experiments were performed with a single input laser. After input irradiation, the molecular response was determined using suitable instrumentation (fluorescence and transient absorbance spectrometers). This was possible because each state of the half-adder has a lifetime of ≥ 1 h at ambient temperatures. The photochemical events leading to the readouts for 1 and 2 have time constants $\leq 2 \mu s$. The pulsed laser inputs A and B are applied for ≤ 10 min. Total irradiation time is $\leq 30 \,\mu s$ per input, and the half-adder is capable of relatively rapid cycle times.

Although the current device is not suitable for the construction of useful computers, it points the way to possible applications. The inputs and outputs are all optical, and no diffusional processes or bimolecular chemical reactions are required. Thus, this approach to molecule-based logic is suitable for nonfluid media, and the inputs and outputs do not require physical access, except by light. After receiving inputs, the half-adder remains latched in the resulting state for hours, facilitating single or multiple readouts, and can then be reset with light. The device is capable of cycle times on the order of tens of microseconds or less. Although half-adder function has been demonstrated here with optical outputs, both outputs are based on intramolecular electron-transfer reactions, and in principle could be detected electronically if the molecules were suitably "wired" into circuits.

Acknowledgment. This work was supported by the NSF (CHE-0352599, D.G., T.M., A.M.) and NSERC of Canada (R.H.M.). J.A. thanks the Carl Trygger Foundation for postdoctoral support.

Supporting Information Available: Details and examples of data. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- (1) Balzani, V.; Credi, A.; Venturi, M. Chem. Phys. Chem. 2003, 3, 49-59. de Silva, A. P.; Fox, D. B.; Moody, T. S.; Weir, S. M. Pure Appl. Chem. **2001**. 73. 503-511.
- (3) Aviram, A. J. Am. Chem. Soc. 1988, 110, 5687-5692.
- (4) Ball, P. Nature **2000**, 406, 118–120.
- (5) Kuciauskas, D.; Liddell, P. A.; Moore, A. L.; Moore, T. A.; Gust, D. J. Am. Chem. Soc. 1998, 120, 10880-10886.
- (6) Langford, S. J.; Yann, T. J. Am. Chem. Soc. 2003, 125, 11198-11199. (7) Gunnlaugsson, T.; MacDonaill, D. A.; Parker, D. J. Am. Chem. Soc. 2001,
- 123, 12866-12876. (8) Ghosh, P.; Bharadwaj, P. K. J. Am. Chem. Soc. 1996, 118, 1553-1554.
- (9) de Silva, A. P.; McClenaghan, N. D. Chem. Eur. J. 2002, 8, 4935-4945.
- (10) Xu, H.; Xu, X.; Dabestani, R.; Brown, G. M.; Fan, L.; Patton, S.; Ji, H. F. J. Chem. Soc., Perkin Trans. 2002, 2, 636-643
- (11) Raymo, F. M.; Giordani, S. J. Am. Chem. Soc. 2002, 124, 2004-2007.
- (12) Saghatelian, A.; Volcker, N. H.; Guckian, K. M.; Lin, V. S. Y.; Ghadiri, M. R. J. Am. Chem. Soc. 2003, 125, 346-347 (13) Kompa, K. L.; Levine, R. D. Proc. Natl. Acad. Sci. 2001, 98, 410-414.
- (14) de Silva, A. P.; McClenaghan, N. D. J. Am. Chem. Soc. 2000, 122, 3965-(15) Stojanovic, M. N.; Stefanovic, D. J. Am. Chem. Soc. 2003, 125, 6673-
- 6676. (16) Guo, X.; Zhang, D.; Zhang, G.; Zhu, D. J. Phys. Chem. B 2004, 108, 11942-11945.
- (17) Liddell, P. A.; Kodis, G.; Andréasson, J.; de la Garza, L.; Bandyopadhyay, S.; Mitchell, R. H.; Moore, T. A.; Moore, A. L.; Gust, D. *J. Am. Chem. Soc.* **2004**, *126*, 4803–4811.
- (18) Stojanovic, M. N.; Mitchell, T. E.; Stefanovic, D. J. Am. Chem. Soc. 2002, 124, 3555-3561
- Terazono, Y.; Kodis, G.; Andréasson, J.; Jeong, G.; Brune, A.; Hartmann, T.; Dürr, H.; Moore, A. L.; Moore, T. A.; Gust, D. J. Phys. Chem. B **2003**, 108, 1812–1814.

JA045577L