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Ptilomycalin D, a Polycyclic Guanidine Alkaloid from the Marine Sponge *Monanchora dianchora*

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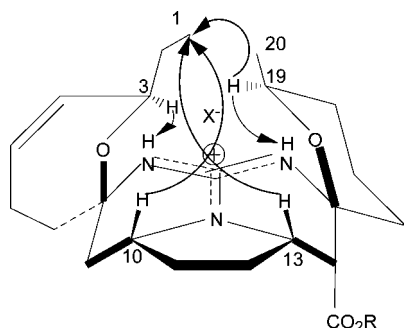


Figure 1. Key NOE for ptilomycalin D (**2**) and crambescidic acid (**3**).

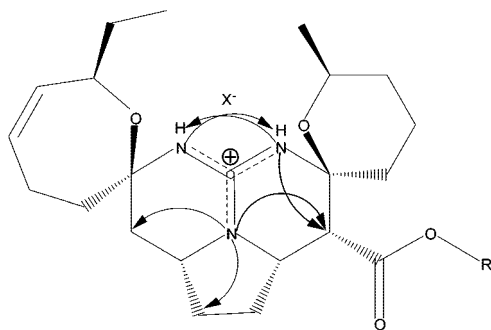


Figure 2. ^{15}NH HMBC correlations for ptilomycalin D (**2**) and crambescidic acid (**3**).

resonances at δ 3.8–4.2 revealed the presence of several protons near nitrogen- and oxygen-linked groups. The presence of nitrogen was also borne out by a ^{15}NH HMBC experiment (Figure 2), in which correlations to three carbon atoms was observed. Moreover, the downfield-shifted carbon resonance at δ 148.9 was attributed to a quaternary carbon of an imine. Therefore the observed values were similar to those of ptilomycalin A. We noted a major difference in the NMR data (Table 1) with the presence of a methyl group at δ_{H} 0.82 and δ_{C} 13.74 in addition to the two methyl groups of the pentacyclic portion, CH_3 -1 at δ 0.77 and CH_3 -20 at δ 1.00. Thus, compound **2**, designated as ptilomycalin D, possessed a terminal methyl group instead of the spermidine moiety in ptilomycalin A.¹

The determination of the second guanidine alkaloid (**3**) was established by comparison of the 1- and 2-D NMR spectra of **3** with those of ptilomycalins A (**1**) and D (**2**). Thus, the “vessel” part was readily established. Moreover, the comparison of the carbon resonances of the pentacyclic part (C_1 – C_{23}) suggested that **3** has the same stereochemistry as ptilomycalins A and D. Both ESMS and FABMS showed two strong peaks at m/z 658. The molecular formula was established to be $\text{C}_{38}\text{H}_{63}\text{N}_3\text{O}_6$ [$\text{M} + \text{H}$, 658.5] $^+$. The determination of the long aliphatic chain was realized by corroboration of the MS spectrum fragmentation pattern and comparison with the NMR spectra of **2**. The ^1H and ^{13}C NMR resonances were similar to those of ptilomycalin D (**2**). We noted differences in the NMR data: the absence of a methyl group, δ_{H} 0.82 and δ_{C} 13.74, and the presence of a carbonyl group at δ 175.4. Moreover, HMBC correlations from H-37 (δ_{H} 2.37) to the methylene chain and a quaternary carbon at δ 175.4 suggested the presence of a carboxylate moiety. This was confirmed by the mass spectrum, which indicated two main mass fragments at m/z 612 ($\text{C}_{37}\text{H}_{62}\text{N}_3\text{O}_4$) and m/z 640 ($\text{C}_{38}\text{H}_{62}\text{N}_3\text{O}_5$) corresponding respectively to $[\text{MH} - \text{CO} - \text{H}_2\text{O}]$ and $[\text{MH} - \text{H}_2\text{O}]$ and suggesting the appendix of the “vessel” to be a C_{16} ω -hydroxycarboxylic acid. Compound **3** corresponds therefore to crambescidic acid,¹¹ and this is the first report of this acid in the sponge *M. dianchora*.

Table 1. NMR Spectroscopic Data (CDCl_3) for Ptilomycalin D (**2**)^a

position	δ_{C} (ppm)	δ_{H} (J in Hz)	HMBC
1	9.76, CH_3	0.77, t (6.7)	
2a	28.84, CH_2	1.50, m	1, 4
2b		1.42, m	
3	70.83, CH		1, 4, 5, 2b
4	133.20, CH	5.43, m	2a, 6b
5	129.72, CH	5.60, m	6a, 6b, 7a, 7b
6	23.20, CH_2	2.28, m	4, 5
7a	36.55, CH_2	2.40, m	5, NH
7b		1.92, m	
8	83.40, qC		6a, 6b, 7a, 7b, NH_2
9a	36.55, CH_2	2.56, m	
9b		1.46, m	
10	53.75, CH	3.94, m	9a, 9b, 11b, 12a
11a	30.30, CH_2	2.15, m	
11b		1.60, m	
12a	26.47, CH_2	2.26, m	
12b		1.72, m	
13	51.81, CH	4.24, m	11a, 12a, 12b, 14
14	49.35, CH	2.92, d (5)	
15	80.43, qC		14, NH
16	30.60, CH_2	1.64, m	
17a	18.30, CH_2	2.15, m	
17b		1.76, m	
18a	31.90, CH_2	1.60, m	20
18b		1.19, m	
19	67.05, CH	3.86, m	20
20	21.07, CH_3	1.00, d (6.7)	
21	148.90, qC		
22	168.32, qC		13, 14, 24
23	65.30, CH_2	4.04, m	24
24	26.80, CH_2	1.57, m	
25	25.54, CH_2	1.44, m	
26	29.50, CH_2	1.19, m	
27	28.19, CH_2	1.56, m	
28 to 35	29.40, CH_2	1.21, m	
36	25.20, CH_2	1.46, m	
37	32.00, CH_2	2.36, m	
38	13.74, CH_3	0.82, t (6.7)	

^a In CDCl_3 , 500 MHz for ^1H and 100 MHz for ^{13}C .

The occurrence of the new ptilomycalin D (**2**), along with crambescidic acid (**3**), may be explained by a biological ω -oxidation process of the hexadecanol moiety yielding the ω -hydroxycarboxylic acid found in crambescidic acid (**3**), and ptilomycalin D may be considered as the biogenetic precursor of **3**, which is the biogenetic precursor of ptilomycalin A.¹¹

Experimental Section

General Experiment Procedures. Optical rotations were obtained with a Jasco P-1010 polarimeter. ^1H and ^{13}C NMR spectra were recorded on Bruker ARX-500 and Avance-400 spectrometers. ^1H , ^{13}C , COSY, HSQC, and HMBC were recorded using standard Bruker pulse sequences. EIMS, CIMS, and HRMS measurements were recorded on a Fisons, Autospec Q instrument.

Biological Material. The sponge, *M. dianchora* (voucher sample Museum d'Histoire Naturelle de Marseille MHN.15615), was collected at a depth of about 18 m (site gorgones) in Nosy-Be. The living sponge is 2–3 cm thick with cylindrical oscules of 10 mm, red and white near the oscules. The identification to *M. dianchora*, from the Philippines and Central Pacific, is tentative, as our specimen, possibly an undescribed species, differs by the presence of spinose microxas and by the shape of the small chelae (with two teeth instead of three and a slight swelling on the shaft).

Cytotoxicity Assays. Cells were maintained in logarithmic growth in EMEM/naaa, supplemented with 5% FCS, 2.0 mM L-glutamine, 10^{-2} M NaHCO_3 , and 0.1 g/L penicillin G + 0.1 g/L streptomycin sulfate. Cytotoxic activity was screened, using an adapted form of the method described by Bergeron et al.¹² against the following cell lines: P-388 (ATCC CCL46). P-388 cells were seeded into 16 mm wells at 1×10^4 cells/well in 1-mL aliquots of EMEM 5% FCS containing different concentrations of the corresponding crude extract. A separate set of

cultures, without drug, was seeded as growth control to ensure that cells remained in the exponential phase of growth. All determinations were duplicated. After three days incubation at 37 °C, 10% CO₂, in 98% humidity, an approximate IC₅₀ value (drug concentration causing a 50% reduction in cell survival) was determined by comparison of the growth in wells with drugs to growth in control wells.

Extraction and Isolation. The freeze-dried sponge (46 g) was exhaustively extracted with CHCl₃/MeOH (1:1). The crude extract (2.8 g) was subjected to solvent partition (*n*-hexane, CCl₄, CHCl₃, and *n*-butanol), and the CHCl₃ fraction (1.7g) subsequently was purified on Sephadex LH-20 (30 × 1.5 cm), eluted with *n*-hexane/CHCl₃/MeOH (2:1:1), to afford ptilomycalin D (**2**) (85 mg, 0.002%, dry weight) and crambescidic acid (**3**) (90 mg, 0.002%, dry weight).

Ptilomycalin D (2): [α]_D²⁵ −7.2 (*c* 0.26, CHCl₃); ¹H and ¹³C NMR (CDCl₃) see Table 1; HRFABMS *m/z* 627.4994 (calcd for C₃₈H₆₄N₃O₄, 626.4975).

Crambescidic acid (3): white, amorphous solid; [α]_D²⁵ +4.4 (*c* 0.20, MeOH); ¹H NMR (CDCl₃) δ 5.61 (H-5, m), 5.43 (H-4, m), 4.23 (H-13, m), 3.94 (H-10, m), 3.86 (H-19, m), 2.90 (H-14, d, *J* = 5 Hz), 2.51 (H-9, m), 2.37 (H-7 and H-37, m), 2.28 (H-6, m), 2.26 (H-12, m), 2.15 (H-11 and H-17, m), 1.90 (H-7, m), 1.74 (H-12, m), 1.71 (H-17, m), 1.60 (H-16 and H-18, m), 1.57 (H-24, m), 1.56 (H-27, m), 1.56 (H-11, m), 1.48 (H-2, m), 1.40 (H-2, m), 1.37 (H-9, m), 1.22 (H-25, m), 1.19 (H-18 and H-26 and H-28 to H-35, m), 1.01 (Me-20, d, *J* = 6.7 Hz), 0.79 (Me-1, t, *J* = 6.7 Hz); HRFABMS *m/z* 658.4811 (calcd for C₃₈H₆₃N₃O₆, 657.4795).

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References and Notes

- (1) Kashman, Y.; Hirsch, S.; Mc Connell, O.; Ohtani, I.; Kusumi, T.; Kakisawa, H. *J. Am. Chem. Soc.* **1989**, *111*, 8925–8926.
- (2) Ohtani, I.; Kusumi, T.; Kakisawa, H.; Kashman, Y.; Hirsch, S. *J. Am. Chem. Soc.* **1992**, *114*, 8472–8479.
- (3) Jares-Erijman, E. A.; Sakai, R.; Rinehart, K. L. *J. Org. Chem.* **1991**, *56*, 5712–5715.
- (4) Jares-Erijman, E. A.; Ingram, A.; Carney, J.; Rinehart, K. L.; Sakai, R. *J. Org. Chem.* **1993**, *58*, 4805–4808.
- (5) Tavares, R.; Daloze, D.; Braekman, J. C.; Hajdu, E.; Van Soest, R. W. M. *Biochem. Syst. Ecol.* **1994**, *22*, 645–646.
- (6) Tavares, R.; Daloze, D.; Braekman, J. C.; Hajdu, E.; Van Soest, R. W. M. *J. Nat. Prod.* **1995**, *58*, 7, 1139–42.
- (7) Braekman, J. C.; Daloze, D.; Tavares, R.; Hadju, E.; Muriey, G.; Van Soest, R. *J. Nat. Prod.* **2000**, *63*, 193–196.
- (8) Palagiano, E.; De Marino, S.; Minale, L.; Riccio, R.; Zollo, F.; Iorizzi, M.; Carre, J.; Debitus, C.; Lucarain, L.; Provost, J. *Tetrahedron* **1995**, *51*, 3675–3682.
- (9) van Soest, R. W. M.; Braekman, J. C.; Faulkner, D. J.; Hajdu, E.; Harper, M. K.; Vacelet, J. *Bull. Inst. R. Sci. Nat. Belg.* **1996**, *66 Suppl.*, 89–101.
- (10) Jares-Erijman, E. A.; Ingram, A.; Sun, F.; Rinehart, K. L. *J. Nat. Prod.* **1993**, *56*, 2186–2188.
- (11) Gallimore, W. A.; Kelly, M.; Scheuer, P. J. *J. Nat. Prod.* **2005**, *68*, 1420–1423.
- (12) Raymond, J.; Bergeron, P. *Biochem. Biophys. Res. Commun.* **1984**, *121*, 3.

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