

See discussions, stats, and author profiles for this publication at: <https://www.researchgate.net/publication/262786045>

Anmindenols A and B, Inducible Nitric Oxide Synthase Inhibitors from a Marine-Derived *Streptomyces* sp.

ARTICLE in JOURNAL OF NATURAL PRODUCTS · MAY 2014

Impact Factor: 3.8 · DOI: 10.1021/np500285a · Source: PubMed

CITATIONS

5

READS

27

8 AUTHORS, INCLUDING:



Lee Jihye

University of Education, Lahore

38 PUBLICATIONS 214 CITATIONS

SEE PROFILE



Hiyoung Kim

University of Illinois at Chicago

19 PUBLICATIONS 118 CITATIONS

SEE PROFILE



Inho Yang

Ewha Womans University

23 PUBLICATIONS 113 CITATIONS

SEE PROFILE



Hyukjae Choi

Yeungnam University

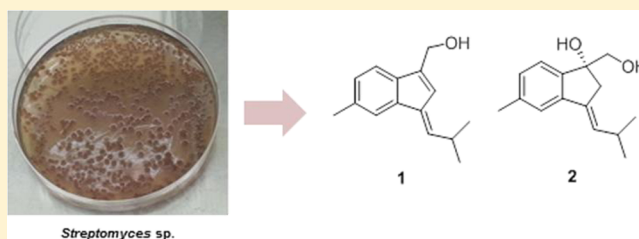
50 PUBLICATIONS 401 CITATIONS

SEE PROFILE

Anmindenols A and B, Inducible Nitric Oxide Synthase Inhibitors from a Marine-Derived *Streptomyces* sp.Jihye Lee,[†] Hiyoung Kim,[†] Tae Gu Lee,[†] Inho Yang,[†] Dong Hwan Won,[†] Hyukjae Choi,[‡] Sang-Jip Nam,^{*,§} and Heonjoong Kang^{*,†,⊥}[†]Center for Marine Natural Products and Drug Discovery, School of Earth and Environmental Sciences, Seoul National University, NS-80, Seoul 151-747, Korea[‡]College of Pharmacy, Yeungnam University, Gyeongsangbuk-do 712-749, Korea[§]Department of Chemistry and Nano Science, Global Top 5 Program, Ewha Womans University, Seoul 120-750, Korea[⊥]Research Institute of Oceanography, Seoul National University, Seoul 151-742, Korea

S Supporting Information

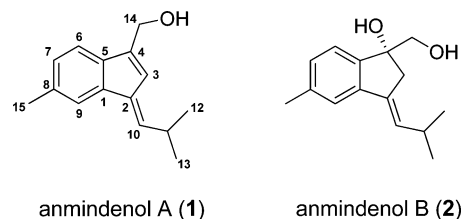
ABSTRACT: Anmindenols A (1) and B (2), inhibitors of inducible nitric oxide synthase (iNOS), were isolated from a marine-derived bacterium *Streptomyces* sp. Their chemical structures were elucidated by interpreting various spectroscopic data, including IR, MS, and NMR. Anmindenols A and B are sesquiterpenoids possessing an indene moiety with five- and six-membered rings derived from isoprenyl units. The absolute configuration of C-4 in anmindenol B was determined by electronic circular dichroism (ECD) of a dimolybdenum complex. Anmindenols A (1) and B (2) inhibited nitric oxide production in stimulated RAW 264.7 macrophage cells with IC₅₀ values of 23 and 19 μ M, respectively.



Tidal flats are spread widely over the west and southwest coasts of the Korean peninsula.¹ The physical and chemical properties of these tidal flats are unpredictable due to water exchange caused by tidal cycles.² The shallow water column in tidal flats with strong tidal currents and high winds stimulates transportation, dispersion, and mixing of nutrients for biological production.³ These physical vectors, along with the variability of salinity, temperature, pH, and nutrient composition, contribute to the great diversity of microbial communities in the tidal flats.^{2,4} The phylogenetic analysis based on 16S rRNA sequences of bacteria from tidal flat sediments collected at Dongmak, located on the west coast of Korea, showed an average sequence similarity to bacterial strains with sequences in GenBank of only 88.4%, ranging from 74.9 to 97.6%.⁵ These data suggested that tidal flat sediments could be a great resource for discovering new microorganisms that produce unique secondary metabolites.

During the course of our screening program designed to discover bacterial secondary metabolites as inhibitors of inflammation, inducible nitric oxide synthase (iNOS) was targeted. Nitric oxide (NO) is a free radical gas with diverse physiological and pathological functions for neurotransmission, host defense, and cardiovascular function in mammals.⁶ It is also important for regulating immune cell function and as an activator in the cell-mediated rejection of allergenic transplants.⁷ NO is produced by three isoforms of NO synthases (NOS), neuronal (nNOS), inducible (iNOS), and endothelial (eNOS), which catalyze the conversion of arginine to

citrulline.⁶ However, NO overproduction leads to numerous human diseases, such as asthma, diabetes, inflammation, septic shock, and chronic inflammatory diseases.⁶ Thus, the specific control of NO production offers great therapeutic value. In particular, the use of selective iNOS inhibitors could be beneficial in the inflammatory process.⁷



We investigated marine-derived bacterial strains isolated from tidal flat sediments at Anmyeon Island on the west coast of Korea to discover novel iNOS inhibitors, by evaluating the strain CMDD10D111. This strain shares 97.4% 16S rRNA gene sequence identity with *Streptomyces phaeopurpureus*, indicating it could be a new *Streptomyces* sp. LC-MS analysis of an extract of CMDD10D111 growing in Mar4 media to which was added 3% DMSO revealed the presence of *m/z* peaks at 215.1 and 237.0 with an unusual chromophore. Large-scale fermentation and flash chromatography followed by HPLC yielded

Received: March 31, 2014

Published: May 30, 2014

Table 1. NMR Data for Anmindenols A and B in CDCl₃

number	Anmindenol A ^a				Anmindenol B ^c	
	δ_C , mult. ^b	δ_H , (J in Hz)	COSY	HMBC	δ_C , mult. ^b	δ_H , (J in Hz)
1	138.1, C				134.6, C	
2	137.1, C				141.5, C	
3	120.6, CH	6.67, s	10	1, 2, 4, 5, 14	41.3, CH ₂	3.00, d (16.6), 2.62, d (16.9)
4	144.4, C				81.1, C	
5	138.5, C				142.9, C	
6	118.9, CH	7.18, d (7.6)		1, 2, 4, 7, 8	123.7, CH	7.24, d (7.4)
7	127.8, CH	7.03, d (7.6)	6, 9	1, 5, 6, 9, 15	129.0, CH	6.99, d (7.2)
8	135.3, C				139.1, C	
9	120.1, CH	7.38, s		1, 2, 5, 7, 15	120.5, CH	7.19, d (3.1)
10	138.8, CH	6.36, d (10.0)		1, 3, 11, 12, 13	129.0, CH	5.75, d (9.3)
11	29.8, CH	3.00, m	10	2, 5, 10, 12, 13	29.0, CH	2.48, m
12	23.5, CH ₃	1.13, d (6.7)	11	10, 11	22.9, CH ₃	0.98, d (4.8)
13	23.5, CH ₃	1.13, d (6.7)	11	10, 11	22.8, CH ₃	0.98, d (5.5)
14	60.2, CH ₂	4.73, s	3, 10	1, 3, 4, 5	69.1, CH ₂	3.67, d (10.7), 3.53, d (10.2)
15	21.8, CH ₃	2.37, s	7, 9	1, 7, 8, 9	21.5, CH ₃	2.29, s

^a600 MHz for ¹H NMR and 150 MHz for ¹³C NMR. ^bMultiplicity was determined by the analysis of 2D NMR spectroscopic data. ^c700 MHz for ¹H NMR and 175 MHz for ¹³C NMR.

anmindenols A (**1**, 3.2 mg) and B (**2**, 1.5 mg). We herein describe the isolation and structure elucidation of anmindenols A (**1**) and B (**2**) and their inhibitory activity against iNOS.fxl.

Anmindenol A (**1**) was isolated as a yellow amorphous solid. Its molecular formula was determined to be C₁₅H₁₈O based on HRFABMS with seven degrees of unsaturation. The ¹H NMR spectrum of **1** displayed three aromatic protons [δ_H 7.18 (d, J = 7.6 Hz), 7.03 (d, J = 7.6 Hz), 7.38 (s)], two olefinic protons [δ_H 6.67 (s), 6.36 (d, J = 10.0 Hz)], and one methyl singlet [δ_H 2.37]. The ¹H NMR spectrum also showed a methyl doublet integrating for six hydrogens [δ_H 1.13, (d, J = 6.7 Hz)]. The COSY cross peaks of H₃-12 and H₃-13 to H-11 indicated that these methyls were coupled to a methine proton (Table 1). In addition, the ¹³C NMR and HSQC spectroscopic data revealed five olefinic carbons and five downfield shifted quaternary carbons. These data and the molecular formula indicated that **1** contained two rings. The presence of 15 carbon signals, including of one methylene carbon [δ_C 60.2], one upfield shifted quaternary carbon [δ_C 29.8], and three methyl carbon signals [δ_C 23.5, 23.5, 21.8], in the ¹³C NMR spectrum suggested that **1** was a sesquiterpenoid.

The COSY correlations of H-9 and H-6 to H-7, and the long-range HMBC correlations from a methyl singlet H₃-15 to carbons C-7, C-8, and C-9 allowed the establishment of a trisubstituted-benzene moiety bearing a methyl group at position C-8. The remaining unknown part of the structure with the indene moiety was identified by interpreting the COSY and HMBC spectroscopic data. An olefinic proton H-10 coupled to the methine proton H-11 in the COSY spectrum was observed to exhibit HMBC correlations to the carbons C-1, C-2, and C-3, which permitted the C-1/C-2/C-3 attachment. Lastly, HMBC correlations from the methylene protons H₂-14 to the carbons C-3, C-4, and C-5 and from H-3 to the carbons C-1, C-2, and C-4 allowed the assignment of anmindenol A to be completed, as shown in Figure 1. The 10*E* double-bond geometry was assigned by NOESY correlations between H-3 and H-11.

Anmindenol B (**2**) was isolated as a yellow amorphous solid and the molecular formula of **2** was determined to be C₁₅H₂₀O₂ based on HRFABMS. The ¹H NMR spectrum of **2** was almost identical to that of **1** except for the presence of one additional

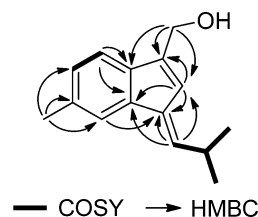


Figure 1. COSY and key HMBC correlations of anmindenol A (**1**).

methylene resonance. The ¹³C NMR data of **2** were also similar to those of **1** except for the upfield-shifted carbon signals for C-3 [δ_C 41.3] and C-4 [δ_C 81.1]. HMBC correlations from H-3 to carbons C-4, C-5, and C-14 and the carbon chemical shift for C-4 [δ_C 81.1] indicated that **2** had a hydroxy group at C-4. The interpretation of 2D NMR spectroscopic data permitted the identification of the structure of **2**.

To determine the absolute configuration for C-4, we conducted an ECD experiment by using the Mo₂(OAc)₄ circular dichroism (CD) method.⁸ A metal complex of a vicinal diol (cyclic and acyclic 1,2-diols) and Mo₂(OAc)₄ acts as an auxiliary chromophore. As a consequence, the conformational freedom of the flexible molecule gives an induced CD curve at 305 nm. The observed sign of the Cotton effect induced by the O–C–C–O torsion angle allows to assign the absolute configuration.⁹ On the basis of the empirical rule, an (*R*)-1,2-diol with Mo₂(OAc)₄ gives rise to a negative CD band at 305 nm, whereas a complex having the *S* configuration gives a positive CD band at 305 nm.⁸ The negative Cotton effect at 305 nm observed in the CD spectrum of the metal complex of **2** (2-Mo₂(OAc)₄) establishes the *R*-configuration for C-4 (Supporting Information).

Dehydration of **2** will form the more highly conjugated **1**. It is possible that **1** is an artifact formed during the extraction and isolation procedure, although **1** was detected in the original extract by LC-MS and there was no indication that **2** was converted to **1** during characterization of **2**.

Anmindenols A and B are sesquiterpenoids possessing a unique indene moiety. Terpenoids consist of important cell components such as steroids, carotenoids, and vitamins, and also have been reported to possess a broad variety of biological

activities with a diversity of structures.¹⁰ Traditionally, the producers of terpenoid secondary metabolites are plants, insects, fungi, and some marine invertebrates.¹¹ Actinomycetes are also now known as a source of producing terpenoid natural products including sesquiterpenoids. In particular, nitro-pyrrolins A–E¹² and neomarinone¹³ are meroterpenoids bearing a sesquiterpenoid moiety isolated from marine-derived actinomycetes and caryolane-1,7 α -diol,¹⁴ 1,6,11-eudesmanetriols,¹⁴ and 11-eudesmene-1,6-diol¹⁴ are sesquiterpenoids discovered from *Streptomyces* sp. However, indene-containing sesquiterpenoids, such as gloeophyllols A, B, and C isolated from the mushroom *Gloeophyllum* sp. 97022., are rare.¹⁵ In addition, there are other reported sesquiterpenoids with a saturated 6,5 ring system. 2-Octahydro-3a,7,7,7a-tetramethyl-1-methylene-1-H-indenemethanol, its ferulic ester and 3b,4,4,7a-tetramethyl-1-H-decahydroindeno[1,2-c]furan-3-ol are hydro-indenes that have been isolated from the plant *Thapsia villosa*.¹⁶ Calenzanol¹⁷ and illudins S¹⁸ and M¹⁸ were discovered from the red alga *Laurencia microcladia* and from the mushroom *Clitocybe illudens*, respectively. Anmindenols A and B are the first sesquiterpenoid natural products containing an indene moiety discovered from actinomycetes.

Anmindenols A and B were tested for their effects against NO production in lipopolysaccharide (LPS)-activated mouse macrophage RAW264.7 cells. Anmindenols A and B inhibited NO production with IC₅₀ values of 23 μ M and 19 μ M, respectively. Anmindenols A and B did not display any significant cytotoxicities against a human renal cancer cell line (A498) and two human pancreatic cancer cell lines (MIA-paca and PANC-1) up to a compound concentration of 100 μ M. Therefore, anmindenols A and B can be novel skeletons offering a new platform for iNOS inhibitors.

EXPERIMENTAL SECTION

General Experimental Procedures. The optical rotation was measured using a Rudolph Research Autopol III polarimeter with a 5 cm cell. The UV spectrum was recorded in a Scinco UVS-2100 with a path length of 1 cm. CD spectra were collected in an Applied Photophysics Chirascan plus CD spectrometer with a 0.5 mm path-length rectangular cuvette. Infrared spectra were recorded on a Thermo Electron Corporation spectrometer. NMR spectral spectroscopic data of anmindenols A and B were obtained using Bruker Avance 600 and 700 MHz spectrometers, respectively [CDCl₃ (δ_{H} 7.26; δ_{C} 77.0) was used as an internal standard]. EI-MS and FAB-MS spectra were measured on a JEOL JMS-AXS05WA mass spectrometer. Low resolution LC-MS data were measured using an Agilent Technologies 6120 quadrupole LC/MS system with a reversed-phase C₁₈ column (Phenomenex luna Su (2), 4.6 mm \times 50 mm, 5 μ m) at a flow rate of 1.0 mL/min. The extracts were separated by HPLC WATERS 1525 binary HPLC pump, WATERS 2489 UV/visible detector using an MG2 C₁₈ (250 mm \times 10 mm, 5 μ m) reversed-phase HPLC column.

Collection and Phylogenetic Analysis of Strain CMDD10D111. Strain CMDD10D111 was isolated from marine sediment from Anmyeon Island, Chungcheongnam-do, South Korea in 2010. The 16S rRNA gene sequence using primers 27f and 1492r for this strain has been deposited with GenBank (accession number KC136293). It shares 97.4% sequence identity with the type strain for *Streptomyces phaeoauripureus* (EU274371.1).

Cultivation and Extraction. Strain CMDD10D111 was cultured in 40 4-L Pyrex flasks each containing 1 L of the medium Mar 4 (2 g of kelp meal, 2 g of D-mannitol, 1 g of fish meal, 20 g/L of KBr, 8 g/L of Fe₂(SO₄)₃·4H₂O, and 30 mL of DMSO dissolved in 970 mL natural seawater) at 25 °C with shaking at 150 rpm. After 10 days, the broth was extracted with EtOAc and evaporated to yield an organic extract of CMDD10D111 (3.8 g).

Isolation of Anmindenols A and B. The extract (3.8 g) was subjected to silica flash column chromatography using step-gradient elution of MeOH in CH₂Cl₂ (0%, 1%, 2%, 5%, 10%, 50%, 100%) to afford seven fractions (Fr 1–Fr 7). Fr 1 (740.9 mg), which contained the mixture of anmindenols, was further purified by C₁₈ HPLC using 55% CH₃CN in H₂O to obtain anmindenols A (1, 3.2 mg) and B (2, 1.5 mg).

Anmindenol A (1). yellow, amorphous solid; UV (MeOH) λ_{max} (log ϵ) 202 (3.07), 262 (3.16), 305 (2.21) nm; IR (film) ν_{max} 3444, 1633 cm⁻¹; ¹H NMR data, see Table 1; ¹³C NMR data, see Table 1; LRMS m/z 215.1 [M + H]⁺; HRFABMS m/z 213.2179 [M – H][–] calcd for C₁₅H₁₇O, 213.1279.

Anmindenol B (2). yellow, amorphous solid; [α]_D²⁵ + 80 (c 0.4, MeOH); UV (MeOH) λ_{max} (log ϵ) 210 (3.39), 260 (3.33), 295 (2.67) nm; CD (8.62 mM, DMSO), λ_{max} ($\Delta\epsilon$) 425 (0.52), 303 (–0.47), and 260 (0.32) nm; IR (film) ν_{max} 3418, 2926, 1710, 1607, 1384 cm⁻¹; ¹H NMR data, see Table 1; ¹³C NMR data, see Table 2; HRFABMS m/z 231.1377 [M – H][–] calcd for C₁₅H₁₉O₂, 231.0577.

Determinations of the Absolute Configuration of C-4 in Anmindenol B Using Mo₂(OAc)₄. Anmindenol B (0.5 mg) was dissolved in 250 μ L of DMSO and the solution was divided into two 125 μ L aliquots. A 125 μ L aliquot of 8.62 mM DMSO solution of Mo₂(OAc)₄ was added to 125 μ L of prepared anmindenol B solution to make 4.32 mM anmindenol B and Mo₂(OAc)₄ mixture solution. The mixture was kept for 30 min to form a stable metal complex, after which the ECD spectrum was recorded for induced CD. The observed band of the ECD curve at 305 nm was used to determine the absolute configuration of C-4.

Nitric Oxide Assay. The NO assay was performed for compounds by measuring NO production in LPS-induced RAW 264.7 mouse macrophage cells according to a previously published protocol.¹⁹

ASSOCIATED CONTENT

Supporting Information

Copies of NMR spectroscopic data for 1–2 and CD spectrum of 2. This material is available free of charge via the Internet at <http://pubs.acs.org>.

AUTHOR INFORMATION

Corresponding Authors

*H. Kang. Tel.: 82 2 880 5730. Fax: 82 2 883 9289. E-mail: hjkang@snu.ac.kr.

*S.-J. Nam. Tel: 82 2 3277 6805. E-mail: sjnam@ewha.ac.kr.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This work was supported by the Marine Biotechnology Program, the Ministry of Oceans and Fisheries, Korea and by the National Research Foundation of Korea grants (NRF-2012M1A5A1054307), the Ministry of Education, Science and Technology, Korea.

REFERENCES

- (1) Lee, T. L.; Lee, J.; Sul, W. J.; Iwai, S.; Chai, B.; Tiedje, J. M.; Park, J. *Appl. Environ. Microbiol.* **2011**, *77*, 3888–3891.
- (2) Wilms, R.; Sass, H.; Köpke, B.; Köster, J.; Cypionka, H.; Engelen, B. *Appl. Environ. Microbiol.* **2006**, *72*, 2756–2764.
- (3) Poremba, K.; Tillmann, U.; Hesse, K. J. *Helgol. Mar. Res.* **1999**, *53*, 19–27.
- (4) Pierre, G.; Graber, M.; Rafiliposon, B. A.; Dupuy, C.; Orvain, F.; Cringnis, M. D.; Maugard, T. *Microb. Ecol.* **2012**, *63*, 157–169.
- (5) Kim, B. S.; Oh, H. M.; Kang, H.; Park, S. S.; Chun, J. J. *Microbiol. Biotechnol.* **2004**, *14*, 205–211.
- (6) Surup, F.; Wagner, O.; Frieling, J. v.; Schleicher, M.; Oess, S.; Müller, P.; Grond, S. J. *Org. Chem.* **2007**, *72*, 5085–5090.

- (7) Hobbs, A.; Higgs, A.; Moncada, S. *Annu. Rev. Pharmacol. Toxicol.* **1999**, *39*, 191–220.
- (8) Frelek, J.; Ikekawa, N.; Takatsuto, S.; Snatzke, G. *Chirality* **1997**, *9*, 578–582.
- (9) Liu, H.-B.; Zhang, C.-R.; Dong, S.-H.; Yang, S.-P.; Sun, Q.; Geng, M.-Y.; Yue, J.-M. *J. Asian Nat. Prod. Res.* **2012**, *14*, 224–234.
- (10) Kuzuyama, T.; Seto, H. *Nat. Prod. Rep.* **2003**, *20*, 171–183.
- (11) Gallagher, K. A.; Fenical, W.; Jensen, P. R. *Curr. Opin. Biotechnol.* **2010**, *21*, 794–800.
- (12) Kwon, H. C.; Espindola, A. P. D. M.; Park, J. S.; Prieto-Davó, A.; Rose, M.; Jensen, P. R.; Fenical, W. *J. Nat. Prod.* **2010**, *73*, 2047–2052.
- (13) Hardt, I. H.; Jensen, P. R.; Fenical, W. *Tetrahedron Lett.* **2000**, *41*, 2073–2076.
- (14) Yang, Z.; Yang, Y.; Yang, X.; Zhang, Y.; Zhao, L.; Xu, L.; Ding, Z. *Chem. Pharm. Bull.* **2011**, *59*, 1430–1433.
- (15) Rasser, F.; Anke, T.; Sterner, O. *Phytochemistry* **2000**, *54*, 511–516.
- (16) Smitt, U. W.; Cornett, C.; Norup, E.; Christensen, S. B. *Phytochemistry* **1990**, *29*, 873–875.
- (17) Guella, G.; Skropeta, D.; Breuils, S.; Mancini, I.; Pietra, F. *Tetrahedron Lett.* **2001**, *42*, 723–725.
- (18) McMorris, T. C.; Anchel, M. *J. Am. Chem. Soc.* **1965**, *87*, 1594–1600.
- (19) Han, A.-R.; Kang, Y.-J.; Windono, T.; Lee, S. K.; Seo, E.-K. *J. Nat. Prod.* **2006**, *69*, 719–721.