ResearchGate

See discussions, stats, and author profiles for this publication at: https://www.researchgate.net/publication/14606697

New Polyoxygenated ent-Manoyl Oxides Obtained by Biotransformation with Filamentous Fungi

ARTICLE in JOURNAL OF NATURAL PRODUCTS · DECEMBER 1995

Impact Factor: 3.8 · DOI: 10.1021/np50125a008 · Source: PubMed

| CITATIONS | READS |
|-----------|-------|
| 7 | 7 |

5 AUTHORS, INCLUDING:



Antonio Martínez

University of Granada

70 PUBLICATIONS 592 CITATIONS

SEE PROFILE

NEW POLYOXYGENATED ENT-MANOYL OXIDES OBTAINED BY BIOTRANSFORMATION WITH FILAMENTOUS FUNGI

ANDRÉS GARCÍA-GRANADOS,* EMILIO LIÑÁN, ANTONIO MARTÍNEZ, MARÍA ESTHER ONORATO,

Departamento de Química Orgánica e Instituto de Biotecnología, Universidad de Granada, Granada 18071, Spain

and José María Arias

Departamento de Microbiología, Facultad de Ciencias, Universidad de Granada, Granada 18071, Spain

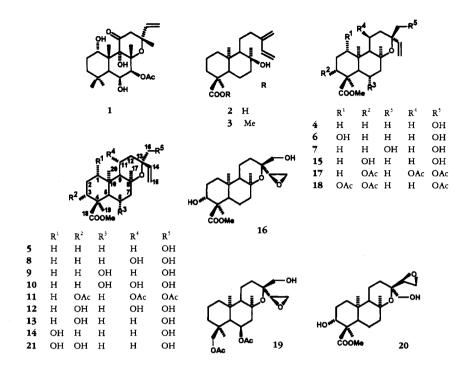
ABSTRACT.—Incubation of methyl (13R)-ent-16-hydroxy-8 α ,13-epoxylabd-14-en-18-oate [4] with Curvularia lunata yielded ent-1 β -hydroxy [6] and ent-6 β -hydroxy [7] derivatives, and that of methyl (13S)-ent-16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate [5] with the same organism gave ent-11 β -hydroxy [8], ent-6 β -hydroxy [9], and ent-6 β ,11 β -dihydroxy [10] derivatives. The incubation of substrates 4 and 5 with Fusarium moniliforme afforded ent-1 β -hydroxy derivatives (6 and 14, respectively). Cunninghamella elegans produced ent-3 β -hydroxy, ent-1 β -hydroxy and ent-1 β ,3 β -dihydroxy derivatives, and led to epoxidation of the double bond of the substrates. In addition, ent-3 β ,11 α -dihydroxy (as the acetoxy derivative 17) and ent-3 β ,11 β -dihydroxy [12] derivatives were isolated from incubations of substrates 4 and 5, respectively. Compounds 7, 9–11, 14, 16–18, and 21 were characterized as new polyoxy genated ent-manoyl oxides.

In recent years, a group of diterpenoids with an 11-oxo-manoyl oxide basic skeleton has been isolated from the medicinal plant *Coleus forskohlii* Briq. (Lamiaceae) (1). The major product, forskolin [1], shows remarkable biological activity and has emerged as a very attractive target for synthesis (2,3), structure-activity relationship studies (4,5) and biomedical research (6). An additional new antihypertensive labdane diterpenoid, 13-epiforskolin, was also isolated from *Coleus forskohlii* (7).

Biotransformation processes with microorganisms can be used to introduce hydroxyl groups at difficult positions on the substrates (8). Chemical-microbiological methods constitute an alternative way to obtain new polyoxygenated products from abundant natural products. Thus, ent-8α-hydroxylabda-13(16),14-dien-18-oic acid [2], isolated from an Andalusian species of Sideritis (10), was transformed to its methyl ester [3]. Methyl (13R)-ent-16-hydroxy-8α,13-epoxylabd-14-en-18-oate [4] and methyl (13S)-ent-16-hydroxy-8\alpha,13-epoxylabd-14-en-18-oate [5]) were then obtained from the methyl ester [3] (9). From substrates 4 and 5, we have accomplished a series of biotransformations with filamentous fungi to study the influence of these microorganisms on the nature of the transformations, and to obtain new manoyl oxides of the enantioseries in order to determine their biological activity. In previous papers we have described the biotransformation of ent-16-hydroxymanoyl oxides with R. nigricans (9,11,12) and reported that the configuration at C-13 is important for the observed effect of the microorganism. An ent-11β-hydroxy derivative isolated from the biotransformation of substrate 5 was found to be an inhibitor of adenylate cyclase (12). To complete this study we have incubated substrates 4 and 5 with the fungi Curvularia lunata, Fusarium moniliforme, and Cunninghamella elegans, and determined the biotransformation products, which are described herein.

RESULTS AND DISCUSSION

Incubation of substrate $\mathbf{4}$ with *Curvularia lunata* for 7 days gave 76% recovered $\mathbf{4}$ together with the metabolites $\mathbf{6}$ and $\mathbf{7}$. Both metabolites showed a molecular peak at m/z 366 in their cims $(m/z 367 [M+1]^+)$, which indicated that the microorganism had introduced an oxygen atom into substrate $\mathbf{4}$ in both cases. Metabolite $\mathbf{6}$ exhibited nmr



and cims data identical to those of methyl (13*R*)-ent-1 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate, obtained in a previous biotransformation from substrate **4** with *R. nigricans* (9). The ¹H-nmr spectrum of metabolite **7** showed a signal at 4.00 ppm (1H, ddd, $J_1 = J_2 = 3.2$ Hz, $J_3 = 2.4$ Hz) due to an equatorial proton, geminal to a hydroxy group on C-6. Comparison of the ¹H-nmr spectra of substrate **3** and metabolite **7** indicated that the ¹H-nmr spectrum of metabolite **7** showed deshielding effects of the protons at C-17 (δ 1.49, $\Delta\delta$ +0.34), C-19 (δ 1.47, $\Delta\delta$ +0.43) and C-20 (δ 1.11, $\Delta\delta$ +0.43), compatible with the presence of an axial hydroxy group at C-6. Analysis of the ¹³C-nmr spectra of metabolite **7** and substrate **4** confirmed that this hydroxylation occurred at C-6.

The biotransformation of substrate **5** with *C. lunata* for 7 days left 21% of substrate **5** unaltered and produced metabolites **8** (26%), **9** (18%), **10** (3%), and a polar mixture that was acetylated to yield triacetate **11** (3%). The spectroscopic data of metabolite **8** were nearly identical to those of the *ent*-11 β -hydroxy derivative isolated in a previous incubation of substrate **5** with *R. nigricans* (12). The cims of metabolite **9** had a molecular peak of m/z 366 (m/z 367 [M+1]⁺). Its ¹H-nmr spectrum showed a signal at 3.99 ppm (1H, ddd, $J_1 = J_2 = 3.2$ Hz, $J_3 = 2.7$ Hz) owing to an equatorial proton, geminal to a hydroxy group on C-6. Comparison of the ¹³C-nmr data of substrate **5** and metabolite **9** confirmed that **9** was methyl (13*S*)-*ent*-6 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate.

The cims spectrum of metabolite ${\bf 10}$ exhibited a molecular peak at m/2 383, compatible with being a dihydroxylated derivative of substrate ${\bf 5}$. Its $^1{\rm H}$ -nmr spectrum showed a signal at 4.51 ppm identical to that observed in the $^1{\rm H}$ -nmr spectrum of metabolite ${\bf 8}$. This signal was due to an equatorial proton, geminal to a hydroxy group on C-11. Another signal appeared at 4.00 ppm owing to a proton geminal to an axial hydroxy group at C-6. The $^1{\rm H}$ -nmr spectrum of ${\bf 10}$, compared with that of substrate ${\bf 5}$, showed deshielding of the protons of the hydroxymethylene group at C-16 as well as of the protons at C-17 and C-20 ($\Delta \delta$ +0.57 and +0.58, respectively). Analysis of the $^{13}{\rm C}$ -nmr spectra of metabolite ${\bf 10}$ and substrate ${\bf 5}$

showed variations in chemical shift similar to those observed in the 13 C-nmr spectra of metabolites **8** and **9**. Thus, metabolite **10** was assigned as methyl (13*S*)-*ent*-6 β ,11 β ,16-trihydroxy-8 α ,13-epoxylabd-14-en-18-oate.

After acetylation of the mixture of polar metabolites, triacetate **11** was isolated. The nmr and cims data for **11** were identical to those of methyl (13*S*)-ent-3 β ,11 β ,16-triacetoxy-8 α ,13-epoxylabd-14-en-18-oate obtained by acetylation from methyl (13*S*)-ent-3 β ,11 β ,16-trihydroxy-8 α ,13-epoxylabd-14-en-18-oate [**12**] isolated from the incubation of substrate **5** with *R. nigricans* (12).

Incubation of **5** with *R. nigricans* yielded the *ent*-3 β -hydroxy derivative [**13**, 36%] (12). Incubation of this product [**13**] with *C. lunata* for 10 days gave the trihydroxy derivative [**12**] (22%), which was also isolated from the incubation of substrate **5** with *R. nigricans* (12).

When substrate 4 was incubated with Fusarium moniliforme for 14 days, metabolite 6 was obtained. The incubation of substrate 5 with F. moniliforme for 14 days gave metabolite 14 (14%) with nmr and cims data very similar to those of metabolite 6. Its 1 H-nmr spectrum showed a signal at δ 3.45 (1H, dd, J_1 =9.3 Hz, J_2 =5.6 Hz), due to an axial proton, geminal to the hydroxyl group on C-1. The 13 C-nmr spectrum of 14, compared with that of substrate 5, indicated that 14 was methyl (13S)-ent-1 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate.

Incubation of substrate 4 with Cunninghamella elegans for five days gave metabolites 15, 16, and a mixture of polar metabolites which was acetylated to give triacetates 17 and 18. Spectroscopic data for metabolite 15 were identical to those of the ent-3βhydroxy derivative isolated from the incubation of substrate 4 with R. nigricans (9). Metabolite 16 was the result of epoxidation of the vinyl moiety together with an equatorial hydroxylation at C-3. The structure of 16 was deduced from its nmr and cims data and from comparison with the spectroscopic data for (135,14R)-ent-6\alpha, 18-diacetoxy-16-dihydroxy-8α,13:14,15-diepoxylabdane [19], described previously (11). Thus, metabolite 16 was the result of epoxidation of the double bond of metabolite 15 from the Re-face, to give methyl (13S,14R)-ent- 3β , 16-dihydroxy- 8α , 13: 14, 15-diepoxylabdan-18-oate. The ¹H-nmr spectra of the triacetoxy derivatives 17 and 18 indicated that two of the acetoxy groups were at C-3 (equatorial) and C-16 in both products. The third acetoxy group showed an equatorial disposition at C-11 in product 17 (geminal proton at δ 5.21, ddd, J_1 =9.1 Hz, J_2 =5.5 Hz, J_3 =4.2 Hz) and an equatorial disposition at C-1 in product **18** (geminal proton at δ 4.80, dd, J_1 =11.3 Hz, J_2 =4.7 Hz). ¹³C-Nmr data confirmed the positions of these functionalizations. Thus, products 17 and 18 were assigned as methyl (13R)-ent- 3β , 11α , 16-triacetoxy- 8α , 13-epoxylabd-14-en-18-oate and methyl (13R)-ent- 1β , 3β , 16-triacetoxy- 8α , 13-epoxylabd-14-en-18-oate, respectively.

The incubation of substrate **5** with *C. elegans* for five days gave two metabolites, the spectroscopic data of which were identical to those **12** and **13** isolated in a previous incubation of substrate **5** with *R. nigricans* (12). Furthermore, two new metabolites [**20** and **21**] were isolated. Metabolite **20** was characterized as methyl (13R,14S)-ent-3 β ,16-dihydroxy-8 α ,13:14,15-diepoxylabd-14-en-18-oate, also isolated from the incubation of substrate **5** with *R. nigricans* (12). The spectroscopic data of metabolite **21** indicated that the microorganism had introduced two hydroxy groups into substrate **5**. Its ¹H-nmr spectrum showed two signals for equatorial protons, geminal to a hydroxy group at C-1 and C-3, respectively. The ¹³C-nmr data were also compatible with the structure, methyl (13*S*)-ent-1 β ,3 β ,16-trihydroxy-8 α ,13-epoxylabd-14-en-18-oate [**21**].

Analysis of the incubations of substrates 4 and 5 (epimers at C-13) allowed one to observe the influence of this chiral center in the biotransformation processes of these diterpenoids with filamentous fungi. Thus, C. lunata biotransformed more efficiently substrate 5 than

substrate **4**. F. moniliforme biotransformed both substrates in low yields. However, C. elegans biotransformed substrate **5** (73%) more efficiently than substrate **4** (50%).

ent-1 β -Hydroxy and ent-6 β -hydroxy derivatives were obtained in the incubation of substrate 4 with C. lunata. The incubation of substrate 5 with this microorganism gave rise to ent-11 β -hydroxylation followed by ent-6 β -hydroxylation and small quantities of ent-6 β ,11 β -dihydroxy and ent-3 β ,11 β -dihydroxy derivatives. We observed that 11-hydroxylation is characteristic of substrates that belong to the 13S configuration (13-normal). C. lunata produced 6-hydroxy derivatives in both incubations, suggesting that this type of hydroxylation is characteristic of this microorganism regardless of the configuration at C-13.

The action of *F. moniliforme* was similar in both substrates and ent-1 β -hydroxy derivatives were obtained. The effects of *C. elegans* were also independent of the configuration at C-13 since, in both cases, ent-3 β -hydroxylation and ent-1 β ,3 β -dihydroxylation were observed. The microorganism also led to epoxidation of the double bond in both substrates. The only difference in the action of *C. elegans* on substrates 4 and 5 was the functionalization at C-11 of the ent-3 β -hydroxy derivatives. Thus, in the incubation of substrate 4, the action occurred on the ent- α face to give an equatorial hydroxylation. However, in the incubation of substrate 5, the reaction occurred on the ent- β face to give an axial hydroxylation at C-11.

These incubations have provided a wide range of new polyoxygenated manoyl oxides. The incubation of substrate 5 with C. lunata improved the yield of an ent-11 β -hydroxy derivative [8, 26%], obtained in a previous incubation from substrate 5 with R. nigricans (10%) (12). Compound 8 is known to inhibit the activity of the adenylate cyclase enzyme (12). The ent-6 β -hydroxy derivatives were also obtained by incubations of other ent-13-epi-manoyl oxides with C. lunata, and may also mediate the activity of the adenylate cyclase enzyme (13,14). In view of these results, it has been observed that the use of different microorganisms can facilitate greatly the accumulation of new diterpene oxides, which can then be tested for possible biological activity.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—The nmr spectra (300 MHz, ¹H and 75.4 MHz, ¹³C) were determined in CDCl₃ solution (which also provided the lock signal) on a Bruker AM-300 spectrometer. Assignments of ¹³C-nmr chemical shifts were made with the aid of DEPT techniques using a flip angle of 135°. Ir spectra were obtained on a Perkin-Elmer 983 G spectrometer, and cims spectra on a Hewlett-Packard 5988 A spectrometer in CH₄. Specific rotations were measured in CHCl₃ (1 dm tube) with a Jasco DIP-370 polarimeter at 20°. Cc employed Merck 60 Si gel (230–400 mesh ASTM), with CH₂Cl₂ and increasing amounts of Me₂CO as eluent. Tlc was performed using Si gel (Merck G), with visualization by spraying with H₂SO₄/HOAc/H₂O followed by heating at 120°.

ISOLATION OF STARTING MATERIAL AND PREPARATION OF SUBSTRATES **4**, **5**, AND **13**.—The starting material used in this study was *ent*-8α-hydroxylabd-13(16),14-dien-18-oic acid [**2**, 6-deoxyandalusoic acid] isolated from *Sideritis varoi* subsp. *cuatrecasassii* (10). Methyl (13*R*)-*ent*-16-hydroxy-8α,13-epoxylabd-14-en-18-oate [substrate **4**] and methyl (13*S*)-*ent*-16-hydroxy-8α,13-epoxylabd-14-en-18-oate [substrate **5**] were obtained from acid **2**, via methyl ester **3**, as reported previously (12). Substrate **13** [methyl (13*S*)-*ent*-3β,16-dihydroxy-8α,13-epoxylabd-14-en-18-oate] was obtained previously by biotransformation of substrate **5** with *Rbizopus nigricans* (12).

ORGANISMS, MEDIA, AND CULTURE CONDITIONS.—Curvularia lunata CECT 2130, Fusarium moniliforme CECT 2152, and Cunninghamella elegans CECT 2123 were obtained from the Colección Española de Cultivos Tipo (CECT), Departamento de Microbiología, Universidad de Valencia, Spain. The bacterial culture was stored in YEPGA medium containing 1% yeast extract, 1% peptone, 2% glucose, and 2% agar, at pH 5. A medium composed of peptone (0.1%), yeast extract (0.1%), beef extract (0.1%), and glucose (0.5%) at pH 5.7 in $\rm H_2O$ was used in all transformation experiments. Erlenmeyer flasks (250 ml) containing 80 ml of medium were inoculated with a suspension of *R. nigricans*. Incubations were maintained at 28° with gyratory shaking (150 rpm) for 6 days, after which the substrates 4, 5, and 13 in EtOH were added.

RECOVERY AND PURIFICATION OF METABOLITES.—Cultures were filtered and pooled, cells were washed with H_2O and the liquid was saturated with NaCl and extracted with C H_2Cl_2 . These extracts were mixed, dried over MgSO₄, and evaporated at 40° in vacuo. Mixtures of products obtained were chromatographed on Si gel columns.

BIOTRANSFORMATION OF SUBSTRATE 4 WITH C. LUNATA.—Substrate 4 (200 mg) was dissolved in EtOH (5 ml) and the solution distributed among five Erlenmeyer flask cultures. These were incubated for seven days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product 4 [153 mg, 76%], methyl (13R)-ent-1 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate [6, 6 mg, 3%] (9), and methyl (13R)-ent-6 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate [7, 17 mg, 8%].

Methyl (13R)-ent-6β, 16-dihydroxy-8α, 13-epoxylabd-14-en-18-oate [7].—Gum; [α]D -10° (c=1, CHCl₃); ir ν max 3489 (OH), 3088 (C=C), 1718, 1250 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.91 (1H, dd, X part of an ABX system, $J_{AX}+J_{BX}=28.3$ Hz, H-14), 5.08 (2H, AB part of an ABX system, H₂-15), 4.00 (1H, ddd, $J_1=J_2=3.1$ Hz, $J_3=2.4$ Hz, H-6), 3.67 (3H, s, MeOCO-), 3.29 and 2.96 (2H, AB system, J=10.8 Hz, H₂-16), 1.49, 1.47 (3H each, s, Me-17, Me-19), 1.11 (3H, s, Me-20); ¹³C nmr (CDCl₃, 75.4 MHz) δ 179.12 (C-18), 144.11 (C-14), 113.37 (C-15), 76.12 (C-13), 76.03 (C-8), 71.02 (C-6), 69.71 (C-16), 59.14 (C-9), 52.74 (C-5), 52.06 (COOMe), 49.95 (C-7), 48.24 (C-4), 40.72 (C-1), 38.83 (C-3), 36.86 (C-10), 28.47 (C-12), 25.12 (C-17), 18.79 (C-19), 17.88 (C-2), 17.35 (C-20), 15.15 (C-11); cims (methane) m/z 367 ([M+1]⁺, 11), 349 ([M+1-H₂O]⁺, 100), 331 ([M+1-2H₂O]⁺, 92).

BIOTRANSFORMATIONOF SUBSTRATE **5** WITH *C. LUNATA*.—Substrate **5** (300 mg) was dissolved in ErOH (10 ml) and the solution distributed among ten Erlenmeyer flask cultures. These were incubated for seven days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product [**5**] (62 mg, 21%), methyl (13S)-ent-11 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate [**9**, 55 mg, 18%], methyl (13S)-ent-6 β ,11 β ,16-trihydroxy-8 α ,13-epoxylabd-14-en-18-oate [**10**, 10 mg, 3%], and a mixture of polar metabolites which was acetylated with Ac₂O/pyridine (1:2 ml) for 24 h at room temperature to give methyl (13S)-ent-3 β ,11 β ,16-triacetoxy-8 α ,13-epoxylabd-14-en-18-oate [**11**, 12 mg, 3%].

Methyl (13S)-ent-6β,16-dibydroxy-8α,13-epoxylabd-14-en-18-oate [9].—Gum; [α]D -11° (ϵ =0.5, CHCl₃); ir ν max 3480 (OH), 3087 (C=C), 1715, 1246 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.78 (1H, dd, J_1 =17.4 Hz, J_2 =10.9 Hz, H-14), 5.20 (1H, dd, J_1 =17.4 Hz, J_2 =1.5 Hz), 5.08 (1H, J_1 =10.8 Hz, J_2 =1.5 Hz) (H₂-15), 3.99 (1H, ddd, J_1 = J_2 =3.2 Hz, J_3 =2.7 Hz, H-6), 3.67 (3H, s, MeOCO-), 3.32, 3.26 (2H, AB system, J=11.0 Hz, H₂-16), 1.51, 1.48 (3H each, s, Me-17, Me-19), 1.18 (3H, s, Me-20); ¹³C nmr (CDCl₃, 75.4 MHz) δ 179.20 (C-18), 144.03 (C-14), 113.99 (C-15), 76.29 (C-13), 74.87 (C-8), 70.85 (C-6), 68.50 (C-16), 53.53 (C-5), 52.74 (C-9), 52.07 (COOMe), 50.74 (C-7), 48.13 (C-4), 40.37 (C-1), 38.75 (C-3), 37.12 (C-10), 27.27 (C-12), 26.96 (C-17), 18.85 (C-19), 17.74 (C-2), 16.61 (C-20), 14.30 (C-11); cims (methane) m/z 367 ([M+1]⁺, 11), 349 ([M+1-H₂O]⁺, 100), 331 ([M+1-2H₂O]⁺, 92).

Methyl (13S)-ent-6β,11β,16-tribydroxy-8α,13-epoxylabd-14-en-18-oate {**10**}.—Gum; ir ν max 3425 (OH), 3080 (C=C), 1717, 1248 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.78 (1H, dd, J_1 =17.4 Hz, J_2 =10.8 Hz, H-14), 5.21 (1H, dd, J_1 =17.4 Hz, J_2 =1.3 Hz), 5.10 (1H, dd, J_1 =10.8 Hz, J_2 =1.3 Hz) (H₂-15), 4.51 (1H, ddd, J_1 = J_2 =6.8 Hz, J_3 =4.4 Hz, H-11), 4.00 (1H, ddd, J_1 = J_2 =3.2 Hz, J_3 =2.7 Hz, H-6), 3.67 (3H, s, MeOCO-), 3.54, 3.35 (2H, AB system, J=11.1 Hz, H₂-16), 1.80 (3H, s, Me-17), 1.58 (3H, s, Me-20), 1.53 (3H, s, Me-19); ¹³C nmr (CDCl₃, 75.4 MHz) δ 179.16 (C-18), 143.57 (C-14), 114.32 (C-15), 75.82 (C-13), 74.99 (C-8), 70.85 (C-6), 68.62 (C-16), 64.25 (C-11), 56.40 (C-9), 53.09 (C-5), 52.22 (C-7), 52.12 (COOMe), 48.26 (C-4), 40.80 (C-1), 38.75 (C-3, C-12), 28.81 (C-17), 18.98, 18.92 (C-19, C-20), 17.71 (C-2); cims (methane) m/2 383 ([M+1]⁺, 51), 365 ([M+1-H₂O]⁺, 100), 347 ([M+1-2H₂O]⁺, 62), 329 ([M+1-3H₂O]⁺, 39).

Methyl(13S)-ent-3β,11β,16-triacetoxy-8α,13-epoxylabd-14-en-18-oate[11].—Gum; [α]D -56° (c=1, CHCl₃); ir ν max 3083 (C=C), 1740, 1239 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.81 (1H, dd, J_1 =17.3 Hz, J_2 =10.8 Hz, H-14), 5.39 (1H, ddd, J_1 = J_2 =6.6 Hz, J_3 =4.5 Hz, H-11), 5.26 (1H, dd, J_1 =17.3 Hz, J_2 =1.3 Hz), 5.10 (1H, dd, J_1 =10.8 Hz, J_2 =1.3 Hz) (H₂-15), 5.11 (1H, dd, J_1 =11.8 Hz, J_2 =5.0 Hz, H-3), 4.16, 4.00 (2H, AB system,J=11.1 Hz, H₂-16), 3.60 (3H, s, MeOCO-), 2.06, 2.03, 1.95 (3H each, s, 3×AcO), 1.52 (3H, s, Me-17), 1.15 (3H, s, Me-20), 1.03 (3H, s, Me-19); ¹³C nmr (CDCl₃, 75.4 MHz) δ 176.25 (C-18), 170.82, 170.07, 170.03 (3×MeCOO), 142.35 (C-14), 114.41 (C-15), 74.89, 73.53 (C-8, C-13), 68.70 (C-16), 66.34 (C-11), 55.44 (C-9), 52.44 (COOMe), 51.90 (C-4), 51.72 (C-5), 43.73 (C-7), 36.83 (C-10), 36.55 (C-1), 35.94 (C-12), 26.80 (C-17), 22.83, 22.23 (C-2, C-6), 21.75, 21.09, 21.04 (3×MeCOO), 17.07 (C-20), 11.69 (C-19); cims (methane) m/z 509 ([M+1]⁺, 21), 449 ([M+1-AcOH]⁺, 100), 389 ([M+1-2AcOH]⁺, 37).

EtOH (2 ml) and the solution distributed between two Erlenmeyer flask cultures. These were incubated for ten days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product [5] (30 mg, 60%) and methyl (13S)-ent-3 β ,11 β ,16-trihydroxy-8 α ,13-epoxylabd-14-en-18-oate [12, 12 mg, 22%] (12).

ACETYLATION OF METABOLITE 12.—Metabolite 12 (10 mg) was acetylated with Ac_2O -pyridine (0.5:1) for 24 h at room temperature to yield 8 mg of triacetate 11 (60 %).

BIOTRANSFORMATION OF SUBSTRATE 4 WITH F. MONILIFORME.—Substrate 4 (100 mg) was dissolved in EtOH (2 ml) and the solution distributed between two Erlenmeyer flask cultures. These were incubated for 14 days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product [4] (59 mg, 59%) and metabolite 6 (24 mg, 23%) (9).

BIOTRANSFORMATION OF SUBSTRATE **5** WITH *F. MONILIFORME*.—Substrate **5** (100 mg) was dissolved in ErOH (2 ml) and the solution distributed between two Erlenmeyer flask cultures. These were incubated for 14 days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product [**5**] (61 mg, 61%) and methyl (13*S*)-*ent*-1β,16-dihydroxy-8α,13-epoxylabd-14-en-18-oate [**14**, 15 mg, 14%]: gum; [α]D -10° (c=0.7, CHCl₃); ir ν max 3452 (OH), 3085 (C=C), 1725, 1246 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.81 (1H, dd, J_1 =17.4 Hz, J_2 =10.8 Hz, H-14), 5.29 (1H, dd, J_1 =17.4 Hz, J_2 =1.5 Hz), 5.13 (1H, J_1 =10.8 Hz, J_2 =1.5 Hz) (H₂-15), 3.66 (3H, s, MeOCO-), 3.45 (1H, dd, J_1 =9.3 Hz, J_2 =5.6 Hz, H-1), 3.26 (2H, s, H₂-16), 1.25, 1.12, 0.88 (3H each, s, 3×Me); ¹³C nmr (CDCl₃, 75.4 MHz) δ 178.83 (C-18), 143.83 (C-14), 114.39 (C-15), 79.05 (C-1), 76.41 (C-8), 75.40 (C-13), 68.83 (C-16), 52.95 (C-9), 52.15 (COOMe), 50.21 (C-5), 47.35 (C-4), 43.11 (C-7), 42.58 (C-10), 34.99 (C-3), 28.40 (C-12), 26.73 (C-2), 26.13 (C-13), 22.65 (C-6), 17.98 (C-11), 16.09 (C-19), 11.49 (C-20); cims (methane) m/z 367 ({M+1}⁺, 100), 349 ({M+1-H,O})⁺, 57), 331 ({M+1-2H,O})⁺, 94).

BIOTRANSFORMATION OF SUBSTRATE 4 WITH C. ELEGANS.—Substrate 4 (100 mg) was dissolved in EtOH (2 ml) and the solution distributed between two Erlenmeyer flask cultures. These were incubated for five days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product [4] (2 mg, 2%), methyl (13R)-ent-3 β ,16-dihydroxy-8 α ,13-epoxylabd-14-en-18-oate [15, 17 mg, 15%](9), methyl (13S,14R)-ent-3 β ,16-dihydroxy-8 α ,13;14,15-diepoxylabdan-18-oate [16, 35 mg, 29%], and a mixture of polar metabolites which was acetylated with Ac₂O-pyridine (1:2 ml) for 24 h at room temperature to give methyl (13R)-ent-3 β ,11 α ,16-triacetoxy-8 α ,13-epoxylabd-14-en-18-oate [17, 12 mg, 3%] and methyl (13R)-ent-1 β ,3 β ,16-triacetoxy-8 α ,13-epoxylabd-14-en-18-oate [18, 5 mg, 3%].

Methyl (13S, 14R)-ent-3β, 16-dihydroxy-8α, 13; 14,15-diepoxylabdan-18-oate [16].—Gum; [α]D \sim 26° (c=0.9, CHCl₃); ir ν max 3449 (OH), 1718, 1246 (COO), 1051 (CO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 3.92 (1H, dd, J_1 =11.5 Hz, J_2 =4.4 Hz, H-3), 3.64 (3H, s, MeOCO), 3.43, 3.19 (2H, AB system, J=10.9 Hz), (H₂-16), 2.86 (1H, dd, J_1 = J_2 =3.0 Hz), 2.77 (1H, J_1 = J_2 =4.7 Hz), 2.71 (1H, dd, J_1 =4.7 Hz, J_2 =3.0 Hz), (H-14 and H₂-15), 1.20, 1.03, 0.71 (3H each, s, 3×Me), ¹³C nmr (CDCl₃, 75.4 MHz) δ 177.76 (C-18), 75.18 (C-3, C-8), 73.22 (C-13), 68.21 (C-16), 57.02 (C-9), 54.95 (C-14), 53.67 (C-4), 52.10 (COOMe), 51.00 (C-5), 45.93 (C-15), 41.93 (C-7), 37.10 (C-1), 35.96 (C-10), 26.28 (C-2), 24.22 (C-12), 23.43 (C-17), 22.13 (C-6), 16.02 (C-20), 15.95 (C-11), 10.46 (C-19); cims (methane) m/z 383 ([M+1]⁺, 25), 365 ([M+1-H,O]⁺, 100), 347 ([M+1-2H₂O]⁺, 37).

Methyl(13R)-ent-3β,11α,16-triacetoxy-8α,13-epoxylabd-14-en-18-oate [17].—Gum; [α]D = 15° (c=0.6, CHCl₃); ir ν max 3090 (C=C), 1735, 1244 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.84 (1H, dd, J_1 =17.5 Hz, J_2 =10.9 Hz, H-14), 5.29 (1H, dd, J_1 =17.5 Hz, J_2 =1.1 Hz), 5.10 (1H, dd, J_1 =10.9 Hz, J_2 =1.1 Hz), (H₂-15), 5.21 (1H, ddd, J_1 =9.1 Hz, J_2 =5.5 Hz, J_3 =4.2 Hz, H-11), 5.13 (1H, dd, J_1 =11.7 Hz, J_2 =4.6 Hz, H-3), 4.03, 3.92 (2H, AB system, J=10.8 Hz, H₂-16), 3.66 (3H, s, MeOCO-), 2.30 (1H, dd, J_1 =15.4 Hz, J_2 =5.5 Hz, H-12a), 2.04, 2.00, 1.96 (3H each, s, 3×AcO), 1.88 (1H, dd, J_1 =15.4 Hz, J_2 =4.2 Hz, H-12), 1.77 (1H, d, J=9.1 Hz, H-9), 1.25, 1.15, 0.89 (3H each, s, 3×Me); ¹³C nmr (CDCl₃, 75.4 MHz) δ 176.24 (C-18), 170.63, 170.48, 170.12 (3×MeCOO), 142.43 (C-14), 114.07 (C-15), 76.82 (C-3), 75.53 (C-8), 73.81 (C-13), 69.58 (C-16), 66.94 (C-11), 58.67 (C-9), 52.47 (COOMe), 52.04 (C-4), 50.75 (C-5), 42.66 (C-7), 37.24 (C-10), 36.85 (C-1), 34.45 (C-12), 26.51 (C-17), 22.97 (C-6), 21.99 (C-2), 21.72, 21.14, 21.06 (3×MeCOO), 16.83 (C-20), 11.59 (C-19); cims (methane) m/z 509 ([M+1]⁺, 12), 449 ([M+1-AcOH]⁺, 100), 389 ([M+1-2AcOH]⁺, 51), 329 ([M+1-3×AcOH]⁺, 19).

Methyl(13R)-enr-1β,3β,16-triacetoxy-8α,13-epoxylabd-14-en-18-oate [18].—Gum; [α]D -26° (c=0.5, CHCl₃); ir ν max 3088 (C=C), 1741, 1246 (COO) cm⁻¹; ¹H nmr (CDCl₃, 300 MHz) δ 5.88 (1H, dd, X part of an ABX system, $J_{AX}+J_{BX}=29.0$ Hz, H-14), 5.26 (1H, dd, $J_1=12.6$ Hz, $J_2=4.7$ Hz, H-3), 5.12–5.06 (2H, AB part of an ABX system, H₂-15), 4.80 (1H, dd, $J_1=11.3$ Hz, $J_2=4.7$ Hz, H-1), 3.79, 3.72 (2H, AB system, J=11.0 Hz, H₂-16), 3.65 (3H, s, MeOCO-), 2.05, 2.01, 1.95 (3H each, s, 3×AcO), 1.21, 1.14, 0.93 (3H each, s, 3×Me); ¹³C nmr (CDCl₃, 75.4 MHz) δ 175.89 (C-18), 170.96, 170.01, 169.51

 $(3 \times MeCOO)$, 143.04 (C-14), 113.97 (C-15), 78.53 (C-1), 75.77 (C-8), 74.38 (C-13), 73.10 (C-3), 70.83 (C-16), 57.63 (C-9), 52.59 (COOMe), 51.89 (C-4), 48.59 (C-5), 42.00 (C-7), 40.73 (C-10), 29.78 (C-2), 29.40 (C-12), 24.19 (C-17), 21.73 (MeCOO), 21.67 (C-6), 21.16, 20.91 (2×MeCOO), 17.54 (C-11), 13.37 (C-20), 11.17 (C-19); cims (methane) m/z 509 ([M+1]⁺, 4), 449 ([M+1-AcOH]⁺, 16), 389 ([M+1-2AcOH]⁺, 26), 329 ([M+1-3AcOH]⁺, 52), 311 (100).

BIOTRANSFORMATION OF SUBSTRATE 5 WITH C. ELEGANS.—Substrate 5 (100 mg) was dissolved in EtOH (2 ml) and the solution distributed between two Erlenmeyer flask cultures. These were incubated for five days and the metabolites were recovered and chromatographed on a Si gel column to obtain the starting product **5** (3 mg, 3%), methyl (13*S*)-ent-3β,11β,16-trihydroxy-8α,13-epoxylabd-14-en-18-oate [**12**, 26 mg, 21%] (12), methyl (13S)-ent-3β,16-dihydroxy-8α,13-epoxylabd-14-en-18-oate [13, 32 mg, 29%] (12), methyl (13R,14S)-ent-3β,16-dihydroxy-8α,13;14,15-diepoxylabdan-18-oate [20, 14 mg, 19%] (12), and methyl (13S)-ent-1\(\beta\), 3\(\beta\), 16-trihydroxy-8\(\alpha\), 13-epoxylabd-14-en-18-oate [21, 5 mg, 4\(\pi\)]; gum; $[\alpha]D - 19^{\circ} (c=0.3, CHCl_3); ir \nu \max 3425 (OH), 3090 (C=C), 1722, 1257 (COO) cm⁻¹; ¹H nmr (CDCl₃),$ 300 MHz) δ 5.80 (1H, dd, J_1 = 17.4 Hz, J_2 = 10.8 Hz, H-14), 5.20 (1H, dd, J_1 = 17.4 Hz, J_2 = 1.5 Hz), 5.13 $(1H, J_1 = 10.8 \text{ Hz}, J_2 = 1.5 \text{ Hz})$ $(H_2 = 1.5 \text{ Hz})$ $(H_3 = 1.5 \text{ Hz})$ $(H_4 = 1.5 \text{ Hz})$ $(H_4$ MeOCO-), 3.53 (1H, dd, J_1 =11.3 Hz, J_2 =4.5 Hz, H-1), 3.26 (2H, s, H₂-16), 1.24, 1.10, 0.88 (3H each, $s, 3 \times Me$), $^{13}C nmr(CDCl_3, 75.4 MHz) \delta 177.61 (C-18), 143.68 (C-14), 114.55 (C-15), 77.34 (C-1), 75.90,$ 75.22 (C-8, C-13), 72.67 (C-3), 68.77 (C-16), 53.71 (C-4), 52.62 (C-9), 52.47 (COOMe), 49.10 (C-5), 42.94 (C-7), 42.24 (C-10), 36.93 (C-2), 26.57 (C-12), 26.04 (C-17), 22.09 (C-6), 17.80 (C-11), 11.58 (C-20), 10.34 (C-19); cims (methane) m/z 383 ([M+1]⁺, 12), 365 ([M+1-H₂O]⁺, 100), 347 ([M+1-2H₂O]⁺) 86), 329 ($[M+1-3H_2O]^+$, 56).

ACKNOWLEDGMENTS

This work was supported by a grant from the Comisión Interministerial de Ciencia y Tecnología. We thank Karen Shashok for her assistance in the translation of the text.

LITERATURE CITED

- 1. S.V. Bhat, Prog. Chem. Org. Nat. Prod., 63, 1 (1993).
- 2. M.I. Colombo, J. Zinczuk, and E.A. Rúveda, Tetrahedron, 48, 963 (1992).
- 3. G. Jordine, S. Bick, U. Möller, P. Welzel, B. Daucher, and G. Maas, Tetrahedron, 50, 139 (1994).
- K.B. Seamon, J.W. Daly, H. Metzger, N.J. de Souza, and N.J. Reden, J. Med. Chem., 26, 436 (1983).
- J.D. Robbins, A. Laurenza, R.W. Kosley, G.J. O'Malley, B. Spahl, and K.B. Seamon, J. Med. Chem., 34, 3204 (1991).
- 6. K.B. Seamon and J.W. Daly, Adv. Cyclic Nucl. Res., 20, 1 (1986).
- 7. J.S. Tandon, R. Roy, S. Balachandran, and R.A. Vishwakarma, Bioorg. Med. Chem. Lett., 2, 249 (1992).
- 8. J.R. Hanson, Nat. Prod. Rep., 139 (1992).
- J.M. Arias, A. García-Granados, M.B. Jiménez, A. Martínez, F. Rivas, and M.E. Onorato, J. Chem. Res. (S), 277 (M), 2064 (1988).
- A. García-Granados, A. Martínez, A. Molina, M.E. Onorato, M. Rico, A. Sáenz de Buruaga, and J.M. Sáenz de Buruaga, *Phytochemistry*, 24, 1789 (1985).
- A. García-Granados, A. Martínez, J.P. Martínez, M.E. Onorato, and J.M. Arias, J. Chem. Soc., Perkin Trans. I, 1261 (1990).
- A. García-Granados, E. Liñán, A. Martínez, M.E. Onorato, A. Parra, and J.M. Arias, *Phytochemistry*, 38, 287 (1995).
- 13. A. García-Granados, A. Martínez, M.B. Jiménez, M.E. Onorato, F. Rivas, and J.M. Arias, J. Chem. Res., (S), 94 (1990).
- A. García-Granados, M.B. Jiménez, A. Martínez, A. Parra, F. Rivas, and J.M. Arias, *Phytochemistry*, 37, 741 (1994).

Received 2 June 1995