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# Marianins A and B, Prenylated Phenylpropanoids from *Mariannaea camptospora*

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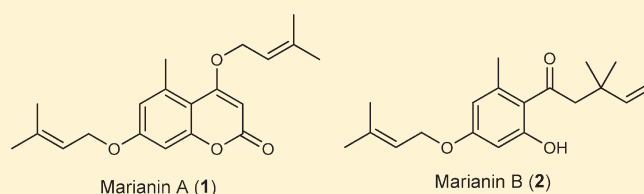
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**S** Supporting Information

**ABSTRACT:** Marianins A (1) and B (2), two new prenylated phenylpropanoids, were isolated from the culture extract of the fungus *Mariannaea camptospora*. Structures of marianins were elucidated by interpretation of NMR and other spectroscopic data. 1 is a 5-methylcoumarin bearing two prenyloxy groups, while 2 is an orcinol derivative substituted with a 3,3-dimethyl-4-pentenoyl chain. 2 is possibly derived from 1 through a Claisen rearrangement of the prenyl group, followed by lactone hydrolysis and decarboxylation. These compounds showed weak antibacterial activity against *Micrococcus luteus*.



Many of fungal species in the order Hypocreales show pathogenicity to higher organisms such as insects and plants. These pathogenic fungi are currently attracting substantial attention as a source of bioactive small molecules owing to their potential in secondary metabolite production.<sup>1</sup> As an example, members of the genus *Cordyceps* are host-specific entomopathogens, from which numerous structurally unique metabolites have been isolated.<sup>1b</sup> *Mariannaea* is also described as a pathogen to some insects<sup>2</sup> and reptiles,<sup>3</sup> and it has been recovered from soil or rotten wood, indicating its saprophytic property as well.<sup>4</sup> Members of this genus show high morphological similarity to the insect-pathogen *Paecilomyces*, and its teleomorph is phylogenetically close to the plant-pathogen *Nectria*.<sup>5</sup> Six species and one variety are included in the genus *Mariannaea*,<sup>6</sup> but only one metabolite, mariannaeapyrone, has been reported from this group to date.<sup>7</sup> In our investigation on chemically unexplored pathogenic fungi, HPLC/UV-based metabolite analysis of a *Mariannaea* strain led to the isolation of two prenylated phenylpropanoids, marianins A (1) and B (2). We herein describe the isolation and structure elucidation of these new compounds.

The producing strain *Mariannaea camptospora* TAMA 118 was isolated from a rotten wood sample collected in Tokyo, Japan. It was cultured in SGCH-X medium, and the whole culture broth was extracted with 1-butanol. The crude extract obtained after solvent removal (2.2 g from 1 L) was subjected to consecutive fractionation using silica gel and C-18 column chromatographies, followed by reversed-phase HPLC, to yield 4.0 and 1.8 mg, respectively, of marianins A (1) and B (2).

Marianin A (1) was obtained as a colorless, amorphous solid that gave an  $[M - H]^-$  peak at  $m/z$  327.1602 (calcd for  $C_{20}H_{23}O_4$ , 327.1602) in the negative ion HR-ESI/MS, consistent with the molecular formula  $C_{20}H_{24}O_4$  (nine degrees of unsaturation). The IR spectrum indicated the presence of a carbonyl functional group ( $1708\text{ cm}^{-1}$ ). NMR data of 1 showed the presence of 20 carbons including four oxygenated  $sp^2$  carbons, five olefinic or aromatic carbons, four quaternary  $sp^2$  carbons, two oxygenated methylenes, and five methyl groups (Table 1). The  $^1H-^1H$  COSY spectrum showed two cross-peaks, each connecting methylene protons and a vinyl proton to give two small fragments,  $H_2-11/H-12$  and  $H_2-17/H-18$ . The first fragment was expanded to include a three-carbon fragment C-14/C-13/C-15 on the basis of HMBC correlations from the methyl proton singlets  $H_3-14$  and  $H_3-15$  to one another, to C-13, and to C-12, establishing a prenyl group. Similarly, the second COSY-defined fragment ( $H_2-17/H-18$ ) and a three-carbon fragment C-20/C-19/C-21 were joined by a series of HMBC correlations from  $H_3-20$  and  $H_3-21$  to one another and to C-18 and C-19, to provide another prenyl unit. The aromatic part was constructed starting from the methyl protons  $H_3-16$ , which showed long-range couplings to C-10, C-5, and C-6. The *meta* relationship of C-6 and C-8 was indicated by a small coupling constant ( $J = 2.3\text{ Hz}$ ) between the protons bonding to these carbons. Chemical shifts of C-6, C-8, and C-10 were relatively upfield, suggesting that these carbons were located *ortho* to the oxygenated  $sp^2$  carbons C-7 and C-9. These data, along with

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Table 1.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Data for Marianin A (**1**) in  $\text{CDCl}_3$ 

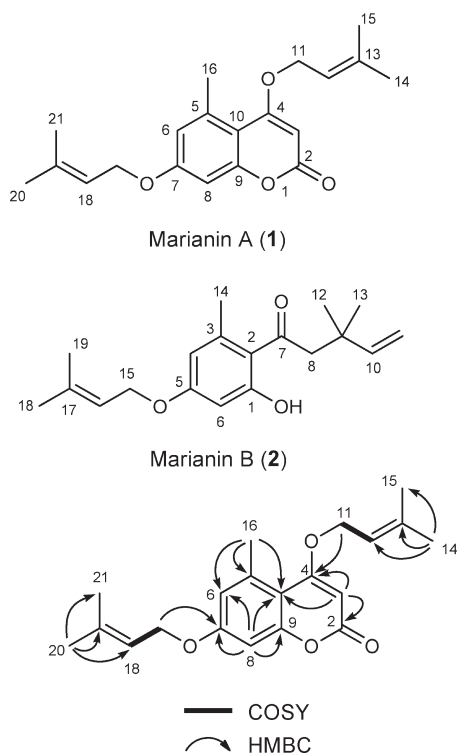
position	$\delta_{\text{C}}$ , mult. <sup>a</sup>	$\delta_{\text{H}}$ (J in Hz) <sup>b</sup>	HMBC <sup>b,c</sup>
2	163.4, qC		
3	88.0, CH	5.51, s	2, 4, 5, 10
4	168.9, qC		
5	138.5, qC		
6	116.2, CH	6.61, d (2.3)	7, 8, 10, 16
7	161.1, qC		
8	99.4, CH	6.66, d (2.3)	4, 6, 7, 9, 10
9	156.7, qC		
10	108.0, qC		
11	66.1, $\text{CH}_2$	4.61, d (6.7)	4, 12, 13
12	117.5, CH	5.50, m	
13	140.0, qC		
14	25.75 <sup>d</sup> , $\text{CH}_3$	1.82, s	12, 13, 15
15	18.29 <sup>e</sup> , $\text{CH}_3$	1.76, s	12, 13, 14
16	23.6, $\text{CH}_3$	2.60, s	5, 6, 10
17	65.1, $\text{CH}_2$	4.57, d (6.8)	7, 18, 19
18	118.8, CH	5.46, m	
19	139.1, qC		
20	25.83 <sup>d</sup> , $\text{CH}_3$	1.80, s	18, 19, 21
21	18.34 <sup>e</sup> , $\text{CH}_3$	1.76, s	18, 19, 20

<sup>a</sup> Recorded at 100 MHz. <sup>b</sup> Recorded at 500 MHz. <sup>c</sup> HMBC correlations are from proton to the indicated carbon. <sup>d,e</sup> Interchangeable.

Table 2.  $^1\text{H}$  and  $^{13}\text{C}$  NMR Data for Marianin B (**2**) in  $\text{CDCl}_3$ 

position	$\delta_{\text{C}}$ , mult. <sup>a</sup>	$\delta_{\text{H}}$ (J in Hz) <sup>b</sup>	HMBC <sup>b,c</sup>
1	165.2, qC		
2	117.1, qC		
3	140.4, qC		
4	112.3, CH	6.27, d (2.5)	2, 5, 6, 14
5	163.1, qC		
6	99.7, CH	6.30, d (2.5)	1, 2, 4, 5
7	206.2, qC		
8	54.5, $\text{CH}_2$	2.93, s	2, 7, 9, 10, 12, 13
9	37.7, qC		
10	147.3, CH	5.94, dd (17.5, 10.7)	9, 12, 13
11	110.5, $\text{CH}_2$	4.91, dd (10.7, 0.8)	9, 10
		4.95, dd (17.5, 0.8)	
12	27.4, $\text{CH}_3$	1.15, s	8, 9, 10, 11, 13
13	27.4, $\text{CH}_3$	1.15, s	8, 9, 10, 11, 12
14	25.3, $\text{CH}_3$	2.53, s	2, 3, 4, 6, 8
15	64.9, $\text{CH}_2$	4.50, d (6.5)	5, 16, 17
16	118.8, CH	5.46, m	
17	139.0, qC		
18	25.8, $\text{CH}_3$	1.80, s	19, 16, 17
19	18.2, $\text{CH}_3$	1.74, s	18, 16, 17
1-OH		12.6, s	1, 2, 5, 6

<sup>a</sup> Recorded at 100 MHz. <sup>b</sup> Recorded at 500 MHz. <sup>c</sup> HMBC correlations are from proton to the indicated carbon.

Figure 1. COSY and key HMBC correlations for **1**.

HMBC correlations from H-6 and H-8 to one another, to C-7, and to C-10, and from H-8 to C-9, established the benzenoid substructure. To this unit was connected a three-carbon fragment C-2/C-3/C-4 on the basis of HMBC correlations from H-3 to

C-2, C-4, and C-10 and a four-bond correlation from H-8 to C-4. HMBC correlations from H<sub>2</sub>-11 to C-4 and from H<sub>2</sub>-17 to C-7 linked the prenyl groups to these carbons through ether linkages. The remaining three degrees of unsaturation were assigned to the C-2 carbonyl functionality, the C-3–C-4 double bond, and a lactone ring connected between C-2 and C-9, to complete the structure of **1** (Figure 1).

Marianin B (**2**) was obtained as a colorless, amorphous solid that analyzed for the molecular formula  $\text{C}_{19}\text{H}_{26}\text{O}_3$  on the basis of an  $[\text{M} - \text{H}]^-$  peak at  $m/z$  301.1803 observed in the HR-ESITOFMS. The IR spectrum showed absorption bands for hydroxyl ( $3261\text{ cm}^{-1}$ ) and carbonyl ( $1609\text{ cm}^{-1}$ ) functionalities.  $^1\text{H}$  and  $^{13}\text{C}$  NMR analysis of **2** revealed the presence of 19 carbons including one carbonyl, two oxygenated  $\text{sp}^2$  carbons, one  $\text{sp}^2$  methylene, four olefinic or aromatic carbons, three quaternary  $\text{sp}^2$  carbons, two  $\text{sp}^3$  methylenes (one is oxygenated), one quaternary  $\text{sp}^3$  carbon, and five methyl groups (Table 2). **2** also possessed a prenyl group, as confirmed by a COSY correlation between H<sub>2</sub>-15 and H-16 and HMBC correlations from H<sub>3</sub>-18 and H<sub>3</sub>-19 to one another, to C-16, and to C-17. Typical coupling patterns for a vinyl group were recognized in the  $^1\text{H}$  NMR spectrum of **2**. Specifically, deshielded protons at  $\delta$  4.91 and 4.95 bonding to a single carbon at  $\delta$  110.5 were mutually coupled with a small geminal coupling constant ( $J = 0.8\text{ Hz}$ ), and these protons (H<sub>2</sub>-11) had COSY correlations to a vinyl proton, H-10. This proton showed correlations to C-9, C-12, and C-13, and two equivalent singlet methyl protons, H<sub>3</sub>-12 and H<sub>3</sub>-13, in turn, showed a series of HMBC correlations to C-9, C-10, and methylene carbon C-8. Furthermore, H<sub>2</sub>-8 was correlated to carbonyl carbon C-7 and quaternary  $\text{sp}^2$  carbon C-2. These correlation data established a 3,3-dimethyl-4-pentenoyl chain connecting to the aromatic core. The 1,2,3,5-tetrasubstituted benzene was elucidated by HMBC correlations from an

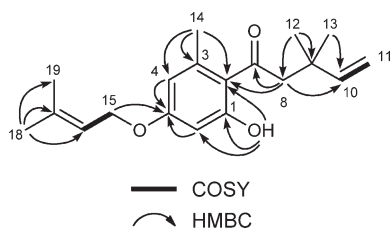


Figure 2. COSY and key HMBC correlations for 2.

exchangeable proton at  $\delta$  12.6 to C-1, C-2, and C-6, from methyl protons H<sub>3</sub>-14 to C-2, C-3, and C-4, and from aromatic protons H-4 and H-6 to C-5. Strong hydrogen bonding of the phenolic proton to the C-7 carbonyl was suggested by the IR absorption band at 1609 cm<sup>-1</sup>, which was significantly low as a wavenumber for keto carbonyls.<sup>8</sup> The prenyloxy group was attached to C-5 by an HMBC correlation from H<sub>2</sub>-15 to C-5, to complete the structure of 2 (Figure 2).

Coumarins are the phenylalanine-derived secondary metabolites widely distributed in plants and are also produced by some fungi and bacteria.<sup>9</sup> These aromatic lactones are often modified by prenylation,<sup>10</sup> but those bearing more than two prenyloxy groups are very rare. Except for 1, only two plant-derived coumarins are known to be *O*-prenylated at two sites.<sup>11</sup> 2 features an unprecedented 3,3-dimethyl-4-pentenoyl chain attaching to the prenylated orcinol. This unique metabolite could be derived from 1 as illustrated in Figure 3. Migration of the 4-*O*-prenyl group to C-3 can occur by Claisen rearrangement (Figure 3, path A). Involvement of this type of rearrangement has been shown in the biogenesis of plant phenylpropanoids,<sup>12,13</sup> while the direct introduction of the dimethylallyl group at C-3 is also possible by reverse-prenylation (Figure 3, path B).<sup>14</sup> The C-2 carbonyl carbon is likely removed by lactone hydrolysis, followed by decarboxylation, as an analogous sequence of reactions has been demonstrated to proceed during alkaline hydrolysis of a plant coumarin.<sup>15</sup>

Marianins A (1) and B (2) showed weak antimicrobial activity against *Micrococcus luteus* with an MIC value of 15 and 30  $\mu$ g/mL, respectively, while both compounds had no activity against *Escherichia coli* or *Candida albicans* at 30  $\mu$ g/mL. Marianins lacked significant activity in a cancer cell cytotoxicity assay. Marianin A (1) was slightly active against HeLa and MCF7 cells with IC<sub>50</sub> values of 34 and 39  $\mu$ M, respectively, and marianin B (2) was inactive against these cell lines (IC<sub>50</sub> > 100  $\mu$ M).

## EXPERIMENTAL SECTION

**General Experimental Procedures.** UV spectra were recorded on a Hitachi U-3210 spectrophotometer. IR spectra were measured on a Perkin-Elmer Spectrum 100. NMR spectra were recorded on a Bruker AVANCE 400 or a Bruker AVANCE 500 spectrometer and referenced to the signals of tetramethylsilane as an internal standard. HR-ESI-TOFMS were recorded on a Bruker microTOF focus spectrometer. Silica gel 60 (Kanto Chemical Co., Inc., 63–210 mesh) and silica gel 60-C18 (Nacalai Tesque, 250–350 mesh) were used for silica gel and ODS column chromatographies, respectively. HPLC separation was performed using a Capcell Pak C18 MGII S5 (Shiseido Co., Ltd., 20  $\times$  150 mm) with a photodiode array detector.

**Microorganism.** Strain TAMA 118 was isolated from a rotten wood sample collected at Tamagawa University, Machida, Tokyo, by direct isolation under microscope. The strain was identified as *Mariannaea camptospora* Samson on the basis of morphological and cultural

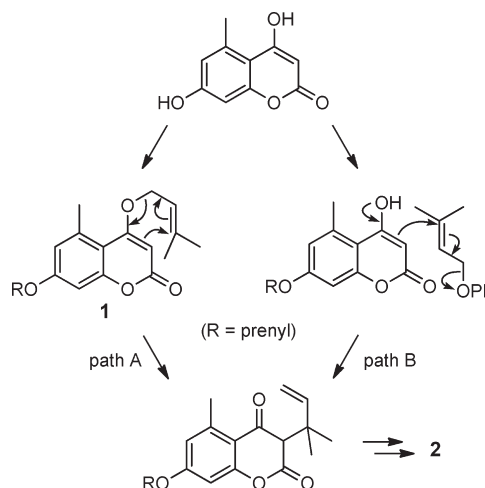


Figure 3. Proposed biogenesis of 2.

characteristics and 99% similarity of internal transcribed spacer (ITS) sequence (562 nucleotides; GenBank accession number AB587666) to *M. camptospora* NBRC 33106 (accession number AB112029) and 94% similarity to *M. camptospora* CBS 209.73 (accession number AY624202).

**Fermentation.** Strain TAMA 118 grown on a PDA slant was inoculated into 150 mL polypropylene flasks each containing 20 mL of the SGCH-X medium [10 g of sodium glutamate, 30 g of sucrose, 0.5 g of yeast extract (Difco Laboratories), 0.4 g of KCl, 2 g of CaCO<sub>3</sub>, 0.015 mg of KH<sub>2</sub>PO<sub>4</sub>, 0.005 mg of MgSO<sub>4</sub>·7H<sub>2</sub>O, 2.5 mL of metal solution, and 1 L of ion exchanged water (pH was adjusted to 6.5 before addition of CaCO<sub>3</sub>)], supplemented with 0.02 g of XAD1180 resin (Organo Co., Ltd.). Metal solution was prepared as containing 15 mg of FeSO<sub>4</sub>·7H<sub>2</sub>O, 9 mg of ZnSO<sub>4</sub>·7H<sub>2</sub>O, 4 mg of MnSO<sub>4</sub>·5H<sub>2</sub>O, 5.5 mg of CuSO<sub>4</sub>·5H<sub>2</sub>O, 6 mg of Co(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O, 2.5 mg of H<sub>3</sub>BO<sub>3</sub>, and 2 mg of Na<sub>2</sub>MoO<sub>4</sub>·2H<sub>2</sub>O in 100 mL of 1 M H<sub>2</sub>SO<sub>4</sub>. After sterilization, the inoculated flasks were placed on a rotary shaker (225 rpm) at 25 °C for 21 days.

**Extraction and Isolation.** At the end of the fermentation period, 20 mL of 1-butanol was added to each flask, and they were allowed to shake on a rotary shaker (225 rpm) for 30 min. The mixture was centrifuged at 3000 rpm for 5 min, and the organic layer was separated from the aqueous layer containing the mycelium. Evaporation of the organic solvent gave approximately 2.2 g of extract from 1 L of culture. The crude extract was subjected to silica gel column chromatography with a step gradient of CHCl<sub>3</sub>/MeOH (1:0, 20:1, 10:1, 4:1, 2:1, 1:1, and 0:1 v/v). Fraction 4 was further purified by C-18 reversed-phase HPLC with MeCN/0.1% HCO<sub>2</sub>H (80:20) to give 4.0 mg of 1. Fractions 2 and 3 were combined and concentrated to provide semipure 2 (15 mg), which was further purified by C-18 reversed-phase HPLC with MeCN/0.1% HCO<sub>2</sub>H (75:25) to give 1.8 mg of 2.

**Marianin A (1):** colorless, amorphous solid; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 208 (4.52), 222 (4.30), 288 (3.91), 308 (4.07), 319 (3.99) nm; IR (ATR)  $\nu_{\max}$  2913, 2855, 1708, 1594, 1155 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; HR-ESI-TOFMS [M – H]<sup>–</sup> 327.1602 (calcd for C<sub>20</sub>H<sub>23</sub>O<sub>4</sub>, 327.1602).

**Marianin B (2):** colorless, amorphous solid; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 220 (3.85), 275 (3.48) nm; IR (ATR)  $\nu_{\max}$  3261, 2924, 2855, 1609, 1159 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 2; HR-ESI-TOFMS [M – H]<sup>–</sup> 301.1803 (calcd for C<sub>19</sub>H<sub>25</sub>O<sub>3</sub>, 301.1809).

**Biological Assays.** Antimicrobial assay was carried out using *Escherichia coli* NIH-JC2, *Micrococcus luteus* ATCC9343, and *Candida albicans* IFO1594 according to the procedures previously described.<sup>16</sup> An MIC value of the standard antibiotic tetracycline hydrochloride

(Sigma-Aldrich Co.) against *M. luteus* was 0.1  $\mu\text{g/mL}$ . Cytotoxic assay was carried out using HeLa human cervical cancer cells and MCF7 human breast cancer cells. Cancer cells were suspended in RPMI medium containing 10% FBS (Sigma-Aldrich, Inc.) and 2 mM L-glutamine and seeded into the wells of a 96-well culture plate ( $1 \times 10^4$  cells/ $50 \mu\text{L}$ /well). Then, test compounds at various concentrations in DMSO/RPMI medium (0.8:92.2 v/v,  $50 \mu\text{L}$ ) were added to the wells. After incubation for 48 h in a humidified 5%  $\text{CO}_2$  incubator at  $37^\circ\text{C}$ , MTT (0.25 mg, Sigma-Aldrich, Inc.) in PBS (–) ( $50 \mu\text{L}$ ) was added to each well, and the plates were placed in the incubator at  $37^\circ\text{C}$  for 4 h. Medium in the wells was removed by suction, and DMSO ( $100 \mu\text{L}$ ) was added to each well. After 10 min, the absorbance at 570 nm was read by a microplate reader.  $\text{IC}_{50}$  values of the positive control staurosporin (Wako Pure Chemical Industries, Ltd.) against HeLa and MCF7 cells were 4 pM and 50 nM, respectively.

## ■ ASSOCIATED CONTENT

**S Supporting Information.** 1D and 2D NMR spectra of **1** and **2**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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## ■ REFERENCES

- (1) (a) Tan, R. X.; Zou, W. X. *Nat. Prod. Rep.* **2001**, *18*, 448–459. (b) Isaka, M.; Kittakoop, P.; Kirtikara, K.; Hywel-Jones, N. L.; Thebtaranonth, Y. *Acc. Chem. Res.* **2005**, *38*, 813–823. (c) Gunatilaka, A. A. L. *J. Nat. Prod.* **2006**, *69*, 509–526.
- (2) Liu, H.; Skinner, M.; Parker, B. L.; Brownbridge, M. J. *Econ. Entomol.* **2002**, *95*, 675–681.
- (3) Banning, J. L.; Weddle, A. L.; Wahl, G. W., III; Simon, M. A.; Lauer, A.; Walters, R. L.; Harris, R. N. *Oecologia* **2008**, *156*, 423–429.
- (4) (a) Samson, R. A. *Stud. Mycol.* **1974**, *6*, 1–119. (b) Okuda, T.; Yamamoto, K. *Mycoscience* **2000**, *41*, 411–414.
- (5) Samuels, G. J.; Samson, R. A. *Sydowia* **1991**, *43*, 249–263.
- (6) <http://www.indexfungorum.org/names/names.asp>.
- (7) Fabian, K.; Anke, T.; Sterner, O. Z. *Naturforsch., C: Biosci.* **2001**, *56*, 106–110.
- (8) (a) Pretsch, E.; Buhlmann, P.; Affolter, C. *Structure Determination of Organic Compounds*; Springer: New York, 2000. (b) Tabuchi, H.; Tajimi, A.; Ichihara, A. *Biosci. Biotechnol. Biochem.* **1994**, *58*, 1956–1959.
- (9) (a) *Dictionary of Natural Products on DVD*; Chemical Database, Version 18.1; Chapman & Hall, 2009. (b) Heide, L. *Nat. Prod. Rep.* **2009**, *26*, 1241–1250.
- (10) Yazaki, K.; Sasaki, K.; Tsurumaru, Y. *Phytochemistry* **2009**, *70*, 1739–1745.
- (11) Rashid, M. A.; Armstrong, J. A.; Gray, A. I.; Waterman, P. G. *Phytochemistry* **1992**, *31*, 1265–1269.
- (12) Chamberlain, T. R.; Collins, J. F.; Grundon, M. F. *Chem. Commun.* **1969**, 1269–1270.
- (13) Donnelly, W. J.; Grundon, M. F.; Ramachandran, V. N. *Proc. R. Soc. B* **1977**, *77B*, 443–447.
- (14) Stocking, E. M.; Williams, R. M.; Sanz-Cervera, J. F. *J. Am. Chem. Soc.* **2000**, *122*, 9089–9098.
- (15) Irie, H.; Kinoshita, K.; Mizutani, H.; Takahashi, K.; Ueo, S.; Yamamoto, K. *Yakugaku Zasshi* **1968**, *88*, 627–634.
- (16) Igarashi, Y.; Yu, L.; Miyanaga, S.; Fukuda, T.; Saitoh, N.; Sakurai, H.; Saiki, I.; Alonso-Vega, P.; Trujillo, M. E. *J. Nat. Prod.* **2010**, *73*, 1943–1946.