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Antitubercular Dihydroagarofuranoid Sesquiterpenes from the Roots of Microtropis fokienensis

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Four new dihydroagarofuranoid sesquiterpenes (1–4) and a new hydroxybenzylsalicylaldehyde, forkienin (5), together with nine known compounds have been isolated from the roots of *Microtropis fokienensis*. The structures of the new compounds were determined through analyses of physical data. Compounds 3, 4, 7, and 8 exhibited potent antitubercular activities (MICs $\leq 26.0 \, \mu$ M) against *Mycobacterium tuberculosis* 90-221387 *in vitro*.

Microtropis fokienensis Dunn (Celastraceae) is a small shrub that grows in high-altitude forests throughout southern China and Taiwan. Various triterpenes, $^{2.3}$ sesquiterpene alkaloids, and dihydroagarofuranoid sesquiterpenes, are widely distributed in plants of the family Celastraceae. Many of these compounds exhibit antitumor, antiinflammatory, insecticidal, and anti-AIDS² activities. In our search for compounds with antitubercular activities, four new β-dihydroagarofuranoid sesquiterpenes (1–4), a new hydroxybenzylsalicylaldehyde, forkienin (5), and nine known compounds (6–14) have been isolated and identified from the root of M. *fokienensis*. Their antitubercular activities against M. *tuberculosis* 90-221387 were evaluated *in vitro*.

$$R_{2^{l_{1}}} = \begin{bmatrix} OBz & OBz$$

- $1 \quad R_1 = R_2 = OAc$
- 2 $R_1 = OAc, R_2 = H$
- 3 $R_1 = OAc, R_2 = OH$
- 4 $R_1 = OH, R_2 = OAc$

Results and Discussion

Extensive chromatographic purification of the EtOAc-soluble fraction of the roots of M. fokienensis on a silica gel column and preparative TLC afforded five new (1-5) and nine known compounds (6-14).

Compound **1** was isolated as an amorphous powder, $[\alpha]_D^{25}$ +34.7. The FABMS of **1** afforded an $[M+H]^+$ ion at m/z 699, implying a molecular formula of $C_{40}H_{43}O_{11}$, which was confirmed by HRFABMS. UV absorptions at 231, 274, and 281 nm were similar to those of mutangin⁷ and suggested the presence of aromatic moieties. Ester carbonyl groups in the molecule were indicated by the bands at 1746, 1721, and 1714 cm⁻¹ in the IR spectrum and were confirmed by resonances at δ 165.2, 165.4, 166.8, 169.5, and 170.0 in the 13 C NMR spectrum. The 1 H NMR spectrum of **1** was also similar to that of mutangin⁷ except that a C-6 benzoyloxy group $[\delta$ 8.04 (2H, d, J = 7.5 Hz, H-2′ and H-6′), 7.46 (2H, t, J = 7.5

$$R_{2^{1}} = \frac{R_{1}}{14^{2}} = \frac{R_{2}}{10^{9}} =$$

Hz, H-3' and H-5'), and 7.59 (1H, t, J=7.5 Hz, H-4')] of 1 replaced a C-6 acetoxy group [δ 2.12 (3H, s)] of mutangin.⁷ This was supported by the HMBC correlations between H-6 (δ 6.32) and PhCO₂-5 (δ 165.4). In the ¹H NMR spectrum of 1, resonances due to acylated oxymethine protons at δ 5.84 (1H, d, J=3.5 Hz), 5.67 (1H, q, J=3.5 Hz), 6.32 (1H, s), and 5.55 (1H, d, J=7.0 Hz) were assigned to H_{ax}-1, H_{eq}-2, H_{ax}-6, and H_{eq}-9, respectively, by ¹H-¹H COSY and NOESY (Figure 1) spectra. The axial orientation of the C-9 benzoate moiety was supported by NOESY experiments (Figure 1), which showed the interactions between H-2",6" (δ 8.06) of the C-9 benzoate and the C-12 methyl (δ 1.51) and H-1 (δ 5.84). NOESY correlations between the C-14 methyl and the AcO-2, H-6, and H-15 groups confirmed their axial orientations. The stereochemical assignments, which were based on the splitting patterns and coupling constants of H-1 [δ 5.84 (d, J=3.5 Hz)], H-2 [δ

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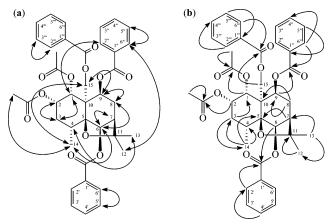


Figure 1. NOESY (a) and HMBC (b) correlations of 1.

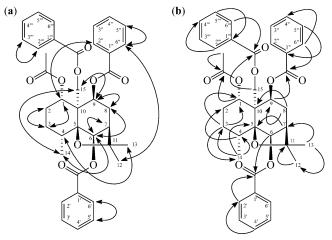


Figure 2. NOESY (a) and HMBC (b) correlations of 2.

5.67 (q, J = 3.5 Hz)], H-6 [δ 6.32 (s)], and H-9 [δ 5.55 (d, J = 7.0Hz)], are in agreement with the relative configurations observed at these positions in this class of natural products.^{8,9} Thus, the structure of 1 was elucidated as $1\alpha,2\alpha$ -diacetoxy- $6\beta,9\beta,15$ -tribenzoyloxy- β -dihydroagarofuran. The location of the ester groups was confirmed by the HMBC spectrum, which exhibited cross-peaks between H-1 (δ 5.84), AcO-1 (δ 1.58), and the carbonyl of one acetate ester (δ 169.5); H-2 (δ 5.67), AcO-2 (δ 2.12), and the carbonyl of the second acetate ester (δ 170.0); H-6 (δ 6.32), H-2'/ 6' (δ 8.04), and the carbonyl of one benzoate ester (δ 165.4); H-9 $(\delta 5.55)$, H-2"/6" $(\delta 8.06)$, and the carbonyl of the second benzoate ester (δ 165.2); and H-15 (δ 4.96, 5.12), H-2" '/6" ' (δ 8.27), and the carbonyl of the third benzoate ester (δ 166.8). Assignments of carbon resonances are shown in Table 2 based on HSQC and HMBC (Figure 1) techniques.

Compound 2 was isolated as an amorphous powder. The ESIMS of 2 afforded an $[M + Na]^+$ ion at m/z 663, implying a molecular formula of C₃₈H₄₀O₉Na, which was confirmed by the HRESIMS. The UV absorptions of 2 at 231, 274, and 281 nm were similar to those of 1. The presence of ester carbonyl groups was revealed by the bands at 1748 and 1722 cm⁻¹ in the IR spectrum and was confirmed by resonances at δ 165.3, 165.4, 166.7, and 169.7 in the ¹³C NMR spectrum. The ¹H and ¹³C NMR (Tables 1 and 2) spectra of 2 were similar to those of 1 except that a hydrogen replaced the *O*-acetyl group (δ 2.12) at C-2 of **1**. From the ${}^{1}H^{-1}H$ COSY and NOESY spectra of 2, the resonances at δ 5.69 (1H, dd, J = 12.5, 4.5 Hz), 6.24 (1H, s), and 5.57 (1H, d, J = 7.5 Hz) were assigned as Hax-1, Hax-6, and Heq-9, respectively. The axial orientation of the C-9 benzoate was supported by NOESY experiments (Figure 2), which showed the interactions between H-2",6" (δ 8.08) of the C-9 benzoate and the C-12 methyl (δ 1.50) and H-1 (δ 5.69). NOESY correlations between the C-14 methyl and

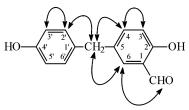


Figure 3. NOESY correlations of 5.

H-6 and H-15 confirmed their axial orientations. The splitting patterns and coupling constants of H-1 [δ 5.69 (dd, J = 12.5, 4.5Hz)], H-6 [δ 6.24 (s)], and H-9 [δ 5.57 (d, J = 7.5 Hz)] are in agreement with the relative configuration observed at these positions in other members of this class of natural products,⁹ and thus, the configurational assignments of 2 were also established. On the basis of the above data, the structure of 2 was elucidated as 1α -acetoxy- 6β ,9 β ,15-tribenzoyloxy- β -dihydroagarofuran. Assignment of the $^{13}\mathrm{C}$ NMR resonances was confirmed by the DEPT, HSQC, and HMBC (Figure 2) techniques.

Compound 3 was also isolated as an amorphous powder. The molecular formula of 3 was established as C₃₈H₄₀O₉ by HRESIMS. The IR spectrum of 3 showed absorption bands at 3530, 1752, and 1725 cm⁻¹ ascribable to the hydroxy and carbonyl functions. The ¹H NMR spectrum of **3** showed that it was the 2-O-deacetyl derivative of 1. Thus, compound 3 is 1α -acetoxy- 2α -hydroxy- 6β , 9β , 15-tribenzovloxy- β -dihydroagarofuran. This was confirmed by ¹H-¹H COSY and NOESY experiments. The assignments of ¹³C NMR resonances of 3 were confirmed by DEPT, HSQC, and HMBC techniques.

Compound 4 was obtained as an amorphous powder. The molecular formula C₃₈H₄₀O₁₀ was deduced from the pseudomolecular ion at m/z 679.2523 [M + Na]⁺ in the HRESI mass spectrum. The IR spectrum of 4 showed absorption bands at 3442, 1750, and 1726 cm⁻¹ ascribable to hydroxyl and carbonyl functions. The ¹H NMR spectrum of **4** showed that it was the 1-O-deacetyl derivative of 1. According to the above data, the structure of 4 was elucidated as 2α -acetoxy- 1α -hydroxy- 6β , 9β , 15-tribenzoyloxyβ-dihydroagarofuran, which was further confirmed by ¹H⁻¹H COSY, NOESY, DEPT, HSQC, and HMBC experiments.

Compound 5 was isolated as an amorphous powder. The HREIMS gave an $[M]^+$ ion at m/z 228.0781 (calcd 228.0781), consistent with a molecular formula of C₁₄H₁₂O₃. The presence of hydroxy groups was revealed by the band at 3469 (br) cm⁻¹ in the IR spectrum, which was confirmed by the resonances at δ 4.85 (br s, D₂O exchangeable, OH-4') and 10.89 (s, D₂O exchangeable, OH-2) in the ¹H NMR spectrum. The ¹H NMR spectrum of 5 showed an AA'BB' spin system at δ 6.78 (2H, d, J = 8.4 Hz, H-3' and H-5') and 7.04 (2H, d, J = 8.4 Hz, H-2' and H-6') and an ABX spin system at δ 6.92 (1H, d, J = 8.4 Hz, H-3), 7.31 (1H, d, J =2.0 Hz, H-6), and 7.35 (1H, dd, J = 8.4, 2.0 Hz, H-4), along with a benzylic methylene group at δ 3.90 (2H, s). The hydrogen-bonded hydroxy group resonated at δ 10.89 (1H, s, OH-2), and the adjacent formyl group at δ 9.83 (1H, s, CHO-1). According to the above data, the structure of 5 is 2-hydroxy-5-(4-hydroxybenzyl)benzaldehyde, named forkienin. This was further confirmed by the ¹H-¹H COSY and NOESY (Figure 3) experiments.

The known compounds including three β -dihydroagarofuranoid sesquiterpenes, mutangin (6), orbiculin G (7), 10 and triptogelin G-2 (8),8 four benzenoids, syringic acid (9),11 vanillic acid (10),12 p-hydroxybenzoic acid (11), 13 and p-hydroxybenzaldehyde (12), 14 and two steroids, β -sitosterol (13)¹⁵ and β -sitostenone (14), ¹⁶ were readily identified by comparison of physical and spectroscopic data with corresponding authentic samples or literature values.

The antitubercular effects of the isolates from the roots of M. fokienensis were tested in vitro against M. tuberculosis 90-221387. The antitubercular activity data are shown in Table 3. The clinically used antitubercular agent ethambutol was used as the positive

Table 1. ¹H NMR Data of Compounds 1-4^a

Н	1	2	3	4
1	5.84 d (3.5)	5.69 dd (12.5, 4.5)	5.76 d (3.2)	5.83 d (3.6)
2	5.67 q (3.5)	1.67 m 2.02 m	4.49 q (3.2)	5.40 q (3.6)
3 (ax)	1.89 dd (15.0, 2.0)	2.02 m 1.60 m	1.97 br d (15.2)	1.92 br d (14.8)
3 (eq)	2.60 ddd (15.0, 6.8, 3.5)	2.39 m	2.48 m	2.48 ddd (14.8, 6.4, 3.6)
4	2.66 m	2.59 m	2.62 m	2.61 m
6	6.32 s	6.24 s	6.36 s	6.17 s
7	2.57 dd (3.5, 3.0)	2.52 dd (3.5, 2.5)	2.54 dd (3.2, 2.8)	2.54 dd (3.6, 2.8)
8 (ax)	2.31 dd (16.0, 3.0)	2.31 dd (16.0, 2.5)	2.28 (16.0, 2.8)	2.38 dd (16.0, 2.8)
8 (eq)	2.74 ddd (16.0, 7.0, 3.5)	2.69 ddd (16.0, 7.5, 3.5)	2.70 ddd (16.0, 7.2, 3.2)	2.74 ddd (16.0, 7.2, 3.6)
9	5.55 d (7.0)	5.57 d (7.5)	5.53 d (7.2)	5.64 d (7.2)
12	1.51 s	1.50 s	1.49 s	1.55 s
13	1.50 s	1.49 s	1.49 s	1.49 s
14	1.30 d (7.5)	1.10 d (7.5)	1.36 d (7.6)	1.35 (7.6)
15	4.96 d (13.7)	4.74 d (12.5)	5.12 d (12.8)	4.92 d (12.8)
	5.12 d (13.7)	5.06 d (12.5)	5.18 d (12.8)	5.03 d (12.8)
OAc-1	1.58 s	1.54 s	1.54 s	
OAc-2	2.12 s			2.10 s
2'/6'	8.04 d (7.5)	8.05 d (7.5)	8.05 d (7.5)	8.04 d (7.5)
3'/5'	7.46 t (7.5)	7.47 t (7.5)	7.47 t (7.5)	7.48 t (7.6)
4'	7.59 t (7.5)	7.59 t (7.5)	7.59 t (7.5)	7.58 t (7.6)
2"/6"	8.06 d (7.6)	8.08 d (7.5)	8.07 d (7.5)	8.09 d (7.6)
3"/5"	7.46 t (7.6)	7.46 t (7.5)	7.46 t (7.5)	7.48 t (7.6)
4"	7.59 t (7.6)	7.59 t (7.5)	7.59 t (7.5)	7.59 t (7.6)
2" '6" '	8.27 d (7.5)	8.28 d (7.5)	8.29 d (7.5)	8.21 d (7.6)
3"'/5"'	7.58 t (7.5)	7.58 t (7.5)	7.58 t (7.5)	7.55 t (7.6)
4"'	7.66 t (7.5)	7.65 t (7.5)	7.65 t (7.5)	7.63 t (7.6)

^a Recorded in CDCl₃ at 400 MHz. Values in ppm (δ). J (in Hz) in parentheses.

Table 2. 13 C NMR Data of Compounds $1-4^a$

Table 2.	C INMIN Data of	Compounds	, 1 7	
С	1	2	3	4
1	71.7	73.6	74.4	69.6
2	69.6	22.6	68.4	73.2
2 3	30.9	26.5	32.4	30.8
4	33.7	34.0	33.5	33.6
5	89.4	89.7	89.4	89.5
6	79.4	79.6	79.3	79.4
7	48.8	48.7	48.7	48.7
8	34.7	34.6	34.7	34.7
9	69.2	69.6	69.2	69.3
10	53.7	53.5	53.5	54.5
11	82.8	82.6	82.8	82.8
12	26.0	26.0	26.0	26.0
13	30.6	30.6	30.6	30.6
14	18.1	16.9	18.0	18.2
15	66.0	65.6	65.8	65.8
1'	129.7^{b}	129.6^{b}	129.7^{b}	129.6^{b}
2'/6'	129.6	129.5	129.6	129.5
3′/5′	128.8	128.8	128.8	128.7
4'	133.4	133.4	133.4	133.4
1"	129.9^{b}	129.8^{b}	129.9^{b}	129.8^{b}
2"/6"	130.1	130.2	130.2	130.1
3"/5"	128.3	128.3	128.3	128.3
4"	133.5	133.5	133.5	133.5
1"′	129.1^{b}	129.2^{b}	129.1^{b}	129.2^{b}
2"'/6"'	130.0	129.9	130.0	129.7
3" '/5" '	128.7	128.7	128.7	128.5
4"'	133.4	133.3	133.4	133.3
CO_2 -1	169.5	169.7	169.4	
CO_2 -2	170.0			171.1
CO ₂ -6	165.4	165.4	165.4	165.5
CO_2 -9	165.2	165.3	165.3	165.3
CO_2 -15	166.8	166.7	166.8	166.7
$MeCO_2$ -1	20.4	20.8	20.7	
MeCO ₂ -2	21.4			21.4

^a Recorded in CDCl₃ at 100 MHz. Values in ppm (δ) . ^b Values superscripted with b are interchangeable in every column.

control. From the results of our antitubercular tests, the following conclusions can be drawn: (a) Among the β -dihydroagarofuranoid sesquiterpene analogues (1–8), compounds 3, 4, 7, and 8 exhibited potent antitubercular activities (MICs \leq 26.0 μ M) against *M. tuberculosis* 90-221387 *in vitro*. (b) Among the analogues (1–4), compound 3, with a 1-acetoxy-2-hydroxy moiety, and 4, with a

Table 3. Antitubercular Effects

compound	MICs $(\mu M)^a$
1	96.0
2	78.1
3	19.5
4	15.8
5	276
6	51.8
7	14.6
8	26.0
9	318
10	375
11	362
12	>400
13	>400
14	>400
ethambutol ^b	30.6

^a Data were means of 3 or 4 replicates. ^bEthambutol was used as a positive control.

2-acetoxy-1-hydroxy group, exhibited more effective antitubercular activities than **1**, with a 1,2-diacetoxy group, and **2**, with a 1-acetoxy functionality, against *M. tuberculosis* 90-221387 *in vitro*. (c) Orbiculin G (7) is the most effective among the isolates, with an MIC of 14.6 μ M against *M. tuberculosis* 90-221387 *in vitro*. (d) The benzenoids **9–12** and the steroids **13** and **14** showed no antitubercular activities against *M. tuberculosis* 90-221387 *in vitro*.

Experimental Section

General Experimental Procedures. All melting points were determined on a Yanaco micro-melting point apparatus and were uncorrected. Optical rotations were measured using a Jasco DIP-370 polarimeter in CHCl₃. UV spectra were obtained on a Jasco UV-240 spectrophotometer. IR spectra (KBr or neat) were recorded on a Perkin-Elmer system 2000 FT-IR spectrometer. NMR spectra, including COSY, NOESY, HMBC, and HSQC experiments, were recorded on a Varian Unity 400 or a Varian Inova 500 spectrometer operating at 400 and 500 MHz (1 H) and 100 and 125 MHz (13 C), respectively, with chemical shifts given in ppm (δ) using TMS as an internal standard. EI, ESI, and HRESI mass spectra were recorded on a Bruker APEX II mass spectrometer. HREI, FAB, and HRFAB mass spectra were recorded on a JEOL JMX-HX 110 mass spectrometer. Silica gel (70–230, 230–

400 mesh) (Merck) was used for CC. Silica gel 60 F-254 (Merck) were used for TLC and preparative TLC.

Plant Material. The roots of M. fokienensis were collected from Chunrih, Pingtung County, Taiwan, in November 2004 and identified by Dr. I. S. Chen. A voucher specimen (Sheng-Zehn Yang 023531) was deposited in the herbarium of the Department of Forest Resource, Management and Technology, National Pingtung University of Science and Technology, Pingtung, Taiwan.

Extraction and Separation. The dried roots (3.5 kg) were extracted with cold MeOH, and the extract was concentrated under reduced pressure. The MeOH extract (362 g), when partitioned between H₂O-EtOAc (1:1), afforded an EtOAc-soluble fraction (fraction A, 41 g). The H₂O-soluble fraction was further extracted with n-BuOH to afford an n-BuOH-soluble fraction (B, 88 g) and an H2O-soluble fraction (C, 173 g). Fraction A (41 g) was chromatographed on silica gel (70–230 mesh, 1.65 kg), eluting with CH₂Cl₂, gradually increasing the polarity with MeOH to give 15 fractions: A1-3 (each 6 L, CH₂Cl₂), A4 and A5 (each 12 L, CH₂Cl₂-MeOH, 20:1), A6-8 (each 4 L, CH₂Cl₂-MeOH, 10:1), A9-11 (each 2 L, CH₂Cl₂-MeOH, 5:1), A12 and A13 (each 2 L, CH₂Cl₂-MeOH, 2:1), A14 (3 L, CH₂Cl₂-MeOH, 1:1), A15 (5 L, MeOH). Fraction A2 (2.5 g) was chromatographed further on silica gel (230–400 mesh, 76 g) eluting with *n*-hexane–EtOAc (10:1) to give 13 fractions. Fraction A2-4 (223 mg) was purified further by preparative TLC (CHCl₃-EtOAc, 30:1) to obtain 14 (23.4 mg). Fraction A2-5 (202 mg) was purified further by preparative TLC (CHCl₃acetone, 50:1) to obtain 8 (3.3 mg). Fraction A2-6 (211 mg) was purified further by preparative TLC (n-hexane-acetone, 3:1) to obtain 2 (4.0 mg). Fraction A2-8 (245 mg) was purified further by preparative TLC (CH₂Cl₂) to obtain 7 (3.5 mg). Fraction A3 (3.5 g) was chromatographed further on silica gel (230-400 mesh, 108 g) eluting with n-hexane-EtOAc (10:1) to give 11 fractions. Fraction A3-3 (151 mg) was purified further by preparative TLC (CH₂Cl₂-EtOAc, 20:1) to obtain 5 (2.1 mg). Fraction A3-4 (142 mg) was purified further by preparative TLC (CH₂Cl₂-acetone, 30:1) to obtain 1 (3.2 mg). Fraction A3-7 (267 mg) was purified further by preparative TLC (CH₂Cl₂-acetone, 30:1) to obtain 6 (5.2 mg). Fraction A3-10 (155 mg) was purified further by preparative TLC (n-hexane-EtOAc, 3:2) to obtain 3 (3.3 mg), 4 (2.4 mg), and 12 (2.6 mg). Fraction A5 (2.5 g) was chromatographed further on silica gel (230–400 mesh, 74 g) eluting with n-hexane–EtOAc (10: 1) to give eight fractions. Fraction A5-4 (141 mg) was purified further by preparative TLC (n-hexane-EtOAc, 30:1) to obtain 13 (14.1 mg). Fraction A9 (2.45 g) was chromatographed further on silica gel (230-400 mesh, 86 g) eluting with CH₂Cl₂-MeOH (10:1) to give nine fractions. Fraction A9-5 (180 mg) was purified further by preparative TLC (EtOAc-MeOH, 8:1) to obtain 9 (4.5 mg) and 10 (2.3 mg). Fraciton A10 (2.33 g) was chromatographed further on silica gel (230-400 mesh, 81 g) eluting with CH₂Cl₂-MeOH (10:1) to give 10 fractions. Fraction A10-7 (180 mg) was purified further by preparative TLC (EtOAc-MeOH, 8:1) to obtain 11 (3.7 mg).

Antitubercular Activity Assay. The antitubercular activity of each test compound was evaluated and compared with the minimal inhibitory concentration (MIC) using the clinical susceptible isolate of M. tuberculosis (90-221387). Middlebrook 7H10 agar was used to determine the MICs as recommended by the proportion method.¹⁷ Briefly, each test compound was added to Middlebrook 7H10 agar supplemented with OADC (oleic acid-albumin-dextrose-catalase) at 50-56 °C by a serial dilution to yield a final concentration of 100 to 0.8 µg/mL. Ten milliliters of each concentration of test compoundcontaining medium was dispensed into plastic quadrant Petri dishes. The inoculum of test isolate of *M. tuberculosis* was prepared by diluting the initial inoculum in Middlebrook 7H9 broth until the turbidity was reduced to that of an equivalent of McFarland no. 1 standard. Final suspensions were performed by adding Middlebrook 7H9 broth and preparing 10⁻² dilutions of the standardized suspensions. After solidification of the Middlebrook 7H10 medium, 33 µL portions of the dilutions were placed on each quadrant of the agar plates, and the agar plates were incubated at 35 °C with 10% CO2 for 2 weeks.

Compound 1: amorphous powder; $[\alpha]^{25}_D$ +34.7 (c 0.2, CDCl₃); UV (MeOH) λ_{max} (log ϵ) 231 (4.12), 274 (3.13), 281 (3.06) nm; IR (KBr) ν_{max} 1746 (C=O), 1721 (C=O), 1714 (C=O) cm⁻¹; ¹H NMR data, see Table 1; ¹³C NMR data, see Table 2; FABMS m/z (rel int) 699 ($[M + H]^+$, 100); HRFABMS m/z 699.7729 $[M + H]^+$ (calcd for $C_{40}H_{43}O_{11}$, 699.7731).

Compound 2: amorphous powder; $[\alpha]^{25}_D$ +45.5 (c 0.18, CDCl₃); UV (MeOH) λ_{max} (log ϵ) 231 (4.17), 274 (3.14), 281 (3.08) nm; IR (KBr) ν_{max} 1748 (C=O), 1722 (C=O) cm⁻¹; ¹H NMR data, see Table 1; 13 C NMR data, see Table 2; ESIMS m/z (rel int) 663 ([M + Na]⁺, 100); HRESIMS m/z 663.2567 [M + Na]⁺ (calcd for $C_{38}H_{40}O_9Na$, 663.2570).

Compound 3: amorphous powder; $[\alpha]^{25}$ _D +40.6 (c 0.14, CDCl₃); UV (MeOH) λ_{max} (log ϵ) 231 (4.18), 275 (3.13), 281 (3.05) nm; IR (KBr) ν_{max} 3530 (OH), 1752 (C=O), 1725 (C=O) cm⁻¹; ¹H NMR data, see Table 1; ¹³C NMR data, see Table 2; ESIMS m/z (rel int) 663 ([M $+ \text{ Na}]^+$, 100); HRESIMS m/z 663.2572 [M + Na]⁺ (calcd for C₃₈H₄₀O₉-Na, 663.2570).

Compound 4: amorphous powder; $[\alpha]^{25}_D$ +41.8 (*c* 0.16, CDCl₃); UV (MeOH) λ_{max} (log ϵ) 232 (4.12), 275 (3.12), 281 (3.04) nm; IR (KBr) ν_{max} 3442 (OH), 1750 (C=O), 1726 (C=O) cm⁻¹; ¹H NMR data, see Table 1; ¹³C NMR data, see Table 2; ESIMS m/z (rel int) 679 ([M $+ \text{ Na}]^+$, 84); HRESIMS m/z 679.2523 [M + Na]⁺ (calcd for C₃₈H₄₀O₁₀-Na. 679.2519).

Fokienin (5): amorphous powder; UV (MeOH) λ_{max} (log ϵ) 225 (4.46), 284 (4.04), 335 (3.58) nm; IR (KBr) ν_{max} 3469 (OH), 1653 (C=O) cm⁻¹; ¹H NMR (400 MHz) δ 4.85 (br s, D₂O exchangeable, OH-4'), 3.90 (2H, s, $-CH_2-$), 6.78 (2H, d, J = 8.4 Hz, H-3'/5'), 6.92 (1H, d, J = 8.4 Hz, H-3), 7.04 (2H, d, J = 8.4 Hz, H-2'/6'), 7.31 (1H, d)d, J = 2.0 Hz, H-6), 7.35 (1H, dd, J = 8.4, 2.0 Hz, H-4), 9.83 (1H, s, CHO-1), 10.89 (1H, s, D₂O exchangeable, OH-2); EIMS m/z (rel int) 228 ([M]⁺, 100), 199 (50), 181 (28), 178 (51), 162 (250, 149 (30), 107 (82), 91 (95), 77 (38); HREIMS m/z 228.0781 [M]⁺ (calcd for $C_{14}H_{12}O_3$, 228.0781).

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