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Mild and Efficient Method for Oxathioacetalization of Carbonyl Compounds

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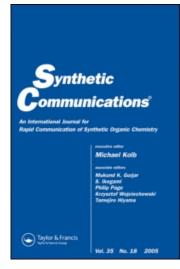
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Mild and Efficient Method for Oxathioacetalization of Carbonyl Compounds

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Abstract: An efficient procedure for the protection of carbonyl compounds into the corresponding 1,3-oxathioacetal has been achieved using PAS as catalyst.

Keywords: 2-Marcaptoethanol, PAS, protection

INTRODUCTION

Oxathioacetals are prepared from the corresponding carbonyl compounds by reaction with 2-marcaptoethanol by an equimolar amount of Lewis acid such as $BF_3 \cdot OEt_2^{[1]}$ and $ZnCl_2^{[2]}$ Other methods in the literature employ TMSOTf, $^{[3]}$ $ZrCl_4$, $^{[4]}$ $LiBF_4$, $^{[5]}$ $HClO_4$, $^{[6]}$ OTAB, $^{[7]}$ NBS, $^{[8]}$ $Sc(OTf)_3$, $^{[9]}$ $In(OTf)_3$, $^{[10]}$ TBAB, $^{[11]}$ and $Me_2S^+BrBr^{-[12]}$ as catalysts. We have observed that carbonyl compounds and 2-marcaptoethanol react conveniently in the presence of polyaniline sulphate salt (PAS) as a catalyst to give the corresponding 1,3-oxathiolanes in good yields Scheme 1.

The results are shown in Table 1.

In a typical reaction procedure, the carbonyl compound (1 mmol), 2-mar-captoethanol (1.2 mmol), and a catalytic amount of PAS (0.01 mmol) in dichloromethane (3 mL) were stirred at room temperature as required for

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O HO SH
$$\rightarrow$$
 O S R R' $\stackrel{PAS (Cat.)}{CII_2CI_2}$ R R' Scheme 1.

Table 1. Preparation of oxathioacetals from the carbonyl compounds

Entry	Substrate	Time (h)	Product	Yield (%) ^a
1	МеО-СНО	3	MeO S	72
2	MeO CHO	5	MeO S	60
3	но-Сно	4	HO S	60
4	PhH ₂ CO—CHO	4	PhCH ₂ O	70
5	О—СНО	4	o s	65
6	PhCOO CHO	2	PhCOO S	75
7	NO ₂ —CHO	4	NO ₂ O S	70
8	СНО	6		65
9	Ph CHO	4	Ph	75

(continued)

Table 1. Continued

Entry	Substrate	Time (h)	Product	Yield (%) ^a
10	O Ph Me	6	O S Ph Me	60
11	Ph	4.5	O S Ph	65
12	0	4	os	60

^aYields are pure isolated product, characterized by IR and ¹H NMR.

completion as monitored by thin-layer chromatography, (TLC). Evaporation of the solvent from the reaction mixture under reduced pressure gives the crude product. The pure product is obtained by column chromatography.

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