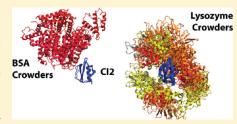


Variable Interactions between Protein Crowders and Biomolecular Solutes Are Important in Understanding Cellular Crowding

Michael Feig*,†,‡ and Yuji Sugita^{‡,§,||}

ABSTRACT: The effect of cellular crowding was examined from molecular dynamics simulations of chymotrypsin inhibitor 2 (CI2) in the presence of either lysozyme or bovine serum albumin (BSA) crowder molecules as a complement to recent experimental studies of the same systems (Miklos, A. C.; Sarkar, M.; Wang, Y.; Pielak, G. J. J. Am. Chem. Soc. 2011, 133, 7116). The simulations confirm a destabilization and significantly slowed diffusion of CI2 in the presence of lysozyme and indicate that this observation is a result of extensive, nonspecific protein—protein interactions between CI2 and lysozyme. CI2 interacts much less with BSA crowders corresponding to a weak effect of crowding. Energetic analysis suggests an



overall favorable crowding free energy in the presence of lysozyme, while weaker interactions with BSA appear to be unfavorable.

■ INTRODUCTION

Crowding in biological cells exerts a modulating effect on biomolecular structure and dynamics.^{1,2} Furthermore, cellular environments modulate kinetic properties such as diffusion³ and association rates^{4,5} that are of importance for understanding cellular dynamics.^{6,7} Past studies of crowding have focused largely on volume exclusion by crowder molecules. Theoretically, this effect is often modeled with hard sphere crowders,² while polymer-based crowding agents are used in corresponding experimental work.² The main finding from these studies is that volume exclusion due to crowding favors compact states over extended states^{8,9} thereby destabilizing the denatured state and stabilizing the folded state as well as aggregates and complexes.

Recently, this entropy-centered view of crowding has been challenged by NMR, fluorescence, and mass spectrometry measurements of protein stability in concentrated protein solutions $^{10-12}$ and biological cells. 13 The emerging view is that crowded biological environments tend to destabilize rather than stabilize native protein structures. This suggests that enthalpic contributions may be at least as important as the volume exclusion effect. Electrostatic interactions and effectively reduced dielectric response of the environment presumably play a dominant role, $^{14-16}$ but details remain unclear.

Particularly interesting is the recent study of chymotrypsin 2 (CI2) in concentrated solutions of bovine serum albumin (BSA) and lysozyme. ¹⁰ It was found that BSA marginally destabilized CI2, while lysozyme had a more pronounced destabilizing effect. It is not clear, though, why exactly CI2 is destabilized in the presence of the protein crowders and why there is a large difference between lysozyme and BSA crowders. Here, we follow up on this work with molecular dynamics (MD) simulations of the same

systems with the specific goal to understand the molecular origins of how CI2 is destabilized differentially in the presence of the protein crowders. In particular, we relate the different nature of protein—protein interactions between CI2 and lysozyme vs CI2 and BSA to different effects of crowding by proteins.

In the following, we will first describe the computational methodology followed by a presentation and discussion of the simulation results.

■ METHODS

Three MD simulations (at 298 K, 1 atm pressure) were set up to mimic the experimental conditions. In the first (referred to as CI2W), a single molecule of CI2 was simulated in explicit solvent, representative of infinite dilution. In the second and third simulations, CI2 was simulated in the presence of eight lysozyme (CI2LYS) and eight BSA (CI2BSA) molecules, respectively, also with explicit solvent and counterions for charge neutralization (cf. Figure 1). The box size was set to match the experimental conditions of 100 g of protein/L of solution (cf. Table 1). The crowder volume and weight fractions are 7% and 10%, respectively, which are at the low end of typical cellular conditions.

Structures of CI2 and lysozyme were taken from the PDB structures 1YPC and 6LYZ, respectively. The structure for BSA was built via homology modeling based on the structure for human serum albumin (1AO6) using MODELER.¹⁷ Initial systems were setup by first placing CI2 at the origin. In the CI2LYS and CI2BSA, crowder molecules were placed at the eight corners of a cube far

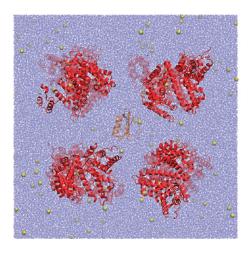
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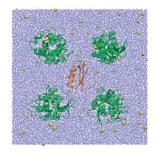




Figure 1. Simulated systems from left to right: CI2BSA, CI2LYS, and CI2W. CI2, orange; BSA, red; lysozyme, green; ions, yellow.

Table 1. Simulated Systems

	system solute(s)	waters	ions	average box length (cubic size) [Å]	$ crowder \\ concentration \left[g/L \right] $	crowder volume fraction [%]	simulation length [ns]
CI2W	CI2	5839	-	56.66	-	-	160
CI2LYS	CI2 + $8 \times lysozyme$	55656	64 Cl ⁻	121.55	105.9	6.97	244
CI2BSA	$CI2 + 8 \times BSA$	253302	136 Na ⁺	201.27	108.2	7.22	117

enough from CI2 to avoid close contacts either between the crowder molecules or between the crowders and CI2. Each of the systems was then solvated with explicit solvent in a cubic box large enough to achieve a crowder concentration of about 100 g/L (cf. Table 1). Water molecules were subsequently replaced randomly by counterions to achieve charge neutrality (cf. Table 1). The initial systems were then minimized with 1000 steps of conjugate gradient minimization and subsequently heated in multiple steps to 298 K (4 ps @ 50 K, 4 ps @ 100 K, 4 ps @ 200 K, 4 ps @ 250 K, and 10 ps @ 298 K). Of the following production runs, the first 30 ns were considered equilibration time.

All of the simulations were run in the NTP ensemble with a Langevin thermo- and barostat using a damping coefficient of 5 ps⁻¹. Periodic boundaries were used. Electrostatic interactions were calculated from particle-mesh Ewald¹⁸ summation using a direct space cutoff of 12 Å (with a switching function becoming effective at 10 Å). The same cutoff was applied to Lennard-Jones interactions. A time step of 2 fs was used in combination with SETTLE¹⁹ to constrain heavy atom—hydrogen bond distances. The CHARMM27 force field²⁰ was used to describe protein protein interactions in combination with a CMAP-based correction map for ϕ/ψ -torsions. ^{21,22} The CMAP used here was improved from the originally published map to reduce the sampling of α -helical conformations while enhancing sampling of PPII conformations and improving the agreement with NMR Jcoupling values for alanine-based peptides.²³ The modified CMAP is available from the authors upon request. The TIP3P model²⁴ was used to describe water interactions. The CI2W simulation was carried out for 160 ns, CI2BSA for 117 ns, and CI2LYS for 244 ns. Averages were calculated for 30-115 ns to exclude an initial equilibration time and cover the same amount of sampling from all trajectories.

The free energy of crowding is given as

$$\Delta \Delta G_{\text{crowding}} = \Delta \Delta H - T \Delta \Delta S + \Delta \Delta G_{\text{solvation}} \tag{1}$$

Changes in enthalpy and solvation free energies were estimated via an MMGB/SA approach^{2,5} based on snapshots taken at 100 ps intervals from the CI2LYS and CI2BSA simulations

$$\Delta \Delta H = \Delta H_{\text{lvs/BSA} + \text{CI2}} - \Delta H_{\text{lvs/BSA}} - \Delta H_{\text{CI2}}$$
 (2)

$$\begin{split} \Delta\Delta G_{\text{solvation}} &= \Delta G_{\text{lys/BSA} + \text{CI2}}^{\text{GB}} - \Delta G_{\text{lys/BSA}}^{\text{GB}} - \Delta G_{\text{CI2}}^{\text{GB}} \\ &+ \gamma \text{SASA}_{\text{lys/BSA} + \text{CI2}} - \gamma \text{SASA}_{\text{lys/BSA}} - \gamma \text{SASA}_{\text{CI2}} \end{split} \tag{3}$$

The enthalpy term was calculated from vacuum molecular mechanics force field energies. The solvation energies were calculated with the GBMV model²⁶ and $\gamma = 0.005$ cal/mol/Å². A dielectric constant of $\varepsilon = 65$ was used to account for a reduced dielectric response of water due to moderate crowding. ^{14,15} $\Delta\Delta G_{\rm solvation}$ between $\varepsilon = 80$ and $\varepsilon = 65$ is about 4 kcal/mol for CI2. SASA is the solvent-accessible surface area.

To estimate the change in entropy for CI2, sampling in CI2W was compared with CI2LYS and CI2BSA. To estimate the change in entropy upon crowding for CI2, vibrational, translational, and rotational contributions were analyzed separately. Vibrational contributions were determined from quasi-harmonic analysis after removing translations and rotations of CI2. Only little difference was found between CI2 in water and in the presence of the crowders (cf. Table 5). Translational entropies were estimated by comparing sampling in CI2W and CI2LYS/CI2BSA. To focus on the remaining translational freedom in the presence of the crowder molecules, a procedure similar to previous calculations of translational entropies in the context of protein—ligand binding²⁷ was followed. More specifically, for each snapshot the following

Table 2. Root-Mean-Square Deviations^a

		CI2		lysozyme		BSA	
		Са	heavy	Сα	heavy	Са	heavy
CI2W	30-115 ns	1.02 (0.19)	1.85 (0.19)	N/A	N/A	N/A	N/A
	30-max	1.11 (0.21)	1.95 (0.21)				
CI2LYS	30-115 ns	1.18 (0.19)	1.92 (0.17)	1.06 (0.22)	1.62 (0.20)	N/A	N/A
	30-max	1.22 (0.16)	2.01 (0.16)	1.10 (0.32)	1.68 (0.27)		
CI2BSA	30-115 ns	1.01 (0.13)	1.86 (0.14)	N/A	N/A	3.45 (0.45)	3.82 (0.41)
	30-max	1.01 (0.13)	1.86 (0.14)			3.45 (0.44)	3.83 (0.41)

^a Average root-mean-square deviations (rmsd's) in Å for CI2, lysozyme, and BSA with respect to initial structures. CI2 data are averaged over time. Lysozyme and BSA data are averaged over both time and multiple copies present in the CI2LYS and CI2BSA simulations, respectively. Standard deviations are given in parentheses.

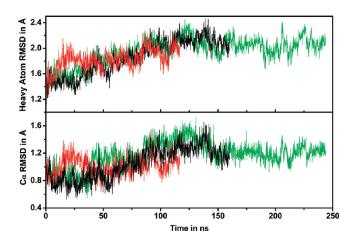


Figure 2. Time series of $C\alpha$ and heavy atom root-mean-square deviations (rmsd's) for CI2 from initial structure after superposition for CI2W (black), CI2LYS (green), and CI2BSA (red) simulations.

steps were carried out: (1) the closest crowder molecule from CI2 was identified based on heavy atom distances; (2) the crowder/ CI2 pair was aligned to a reference crowder structure; (3) the center of mass and orientation of CI2 after superposition of the crowder structure was counted with a histogram to determine sampling of space and angular directions. This procedure provides the sampling of CI2 relative to the crowders. An accurate estimate of absolute entropies requires complete conformational sampling, which would require much longer simulations. Instead, a differential approach was used here, where results for the CI2LYS and CI2BSA simulations were compared with results from artificial trajectories, where the CI2 coordinates were replaced with the coordinates from the CI2W trajectory. The latter essentially represents a simulation of CI2 in the presence of the crowders but with no effective interaction or volume exclusion. Hence, a comparison between the two gives a direct estimate of the effect of crowding on translational entropies. Volumes were obtained from counting visited positions of the CI2 center of mass on a 3 Å grid. Rotational entropies were analyzed with the same approach after removing translational motion and using the area on the unit sphere traced out by the normalized vector between the $C\alpha$ atom of residue 17—at the center of the long helix of CI2—and the center of mass of CI2. The visited volume was found to decrease by about half for both lysozyme and BSA. This is in part a result of excluded volume and in part due to association with the crowders.

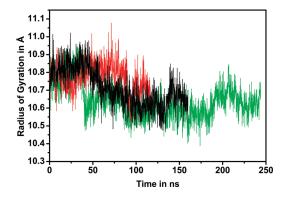


Figure 3. Time series of radius of gyration for CI2 for CI2W (black), CI2LYS (green), and CI2BSA (red) simulations.

However, with both crowders the full rotational freedom was maintained (cf. Table 5). The change in entropy of the crowder molecules was neglected.

■ RESULTS AND DISCUSSION

The CI2 structure remained stable during all of the simulations with average C\$\alpha\$-RMSDs of about 1 Å and heavy atom RMSDs of about 2 Å (cf. Table 2 and time series in Figure 2). There was little difference between CI2 in dilute solvent and in the presence of BSA. The rmsd was slightly higher with lysozyme crowders (1.2 vs 1.0 Å) and remained near that value during the entire CI2LYS simulation (cf. Figure 2).

CI2 features an extended flexible loop for residues 35–45 that could potentially assume more compact conformations if constrained by the environment. To measure compaction of CI2, the radius of gyration was calculated. Without crowders, the average radius of gyration was 10.71 Å. We found a decrease in the presence of lysozyme to 10.64 Å but an increase with BSA crowders to 10.78 Å (cf. Figure 3). The changes are small and essentially within the uncertainty given a standard deviation of 0.1 Å. Therefore, we cannot conclude clearly that crowding leads to compaction for this system.

We subsequently examined root-mean-square fluctuations (rmsf's) around the average structures from each of the simulations. Lysozyme crowding led to increased fluctuations in parts of the structure, in particular at residues 5-8, 16-26, and 52-55 (cf. Figure 4). These residues map onto parts of the two

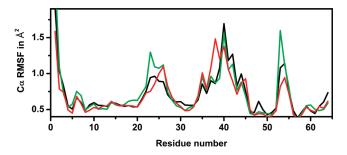


Figure 4. $C\alpha$ root-mean-square fluctuations (rmsf's) for CI2 during 30-115 ns in CI2W (black), CI2LYS (green), and CI2BSA (red).

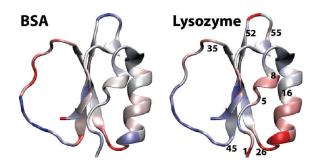


Figure 5. Increased (red) and decreased (blue) $C\alpha$ -rmsf of CI2 in the presence of BSA (left) and lysozyme (right).

Table 3. Average Number of Contacted Crowder Molecules^a

	CI2LYS	CI2BSA
CI2	2.59 (0.51)	1.11 (0.59)
Lys.—Lys.	1.36 (0.50)	N/A
BSA-BSA	N/A	1.85 (0.78)

^a Averaged over 30–115 ns with standard deviations in parentheses. Contacts have a minimum heavy atom distance of less than 10 Å.

N-terminal helices of CI2 (cf. Figure 5) as well as the nearby loop connecting the strands of the central β -sheet. Interestingly, many of the same residues showed decreased thermodynamic stability in NMR experiments of the same system (cf. Figure 1 in that work 10). Hence, it appears that increased structural fluctuations around the native structure may partially explain the observed decrease in thermodynamic stability of the folded vs the unfolded state. Additional residues of CI2 with decreased stability involve the central strand of the β -sheet. Since we did not see any significant increase in rmsf for this well-packed part of the CI2 structure, we hypothesize that reduced stability in this region is a result of partial unfolding on time scales not accessible by our submicrosecond simulations.

In the case of BSA crowders, structural fluctuations were more similar to the noncrowded case and involved both increased and decreased fluctuations at different parts of CI2, which is generally in agreement with the experimental data. ¹⁰ The largest changes in the presence of BSA were seen in the extended loop, for which experimental data are not available. ¹⁰

We now turn to an analysis of CI2—crowder interactions. On average, CI2 interacted with 2.6 lysozyme molecules but only 1.1 BSA molecules (cf. Table 3). CI2—crowder interactions involved different molecules coming on and off during the course of

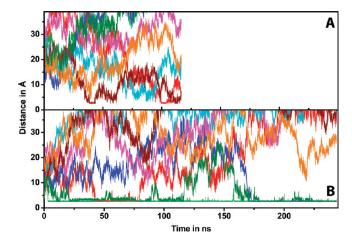


Figure 6. Minimum distance between CI2 and BSA (A) or lysozyme (B) crowders based on heavy atom distances as a function of simulation time. Different colors indicate different crowder atoms.

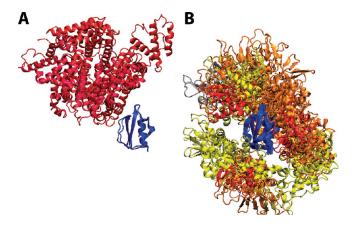


Figure 7. Representative structures of CI2 contacts with BSA (A) and lysozyme (B) crowder molecules obtained via K-means clustering (15 Å radius) of complexes with a minimum contact distance of less than 10 Å during 30—115 ns. Structures closest to the cluster centers for clusters with contact distances below 6 Å are shown. Coloring indicates cluster size relative to total number of structures (red, >5%; orange, <5%, >1.5%; yellow, <1.5%).

the simulations (cf. Figure 6), but contacts were maintained much longer with lysozyme (>20 ns) than with BSA (<20 ns). A smaller number of interacting BSA molecules may simply be a consequence of the larger size of BSA, but the shorter contact times with BSA indicate that CI2—BSA interactions are inherently weaker. Moreover, BSA—BSA interactions were more extensive than lysozyme—lysozyme interactions with 1.9 average BSA—BSA contacts vs 1.4 lysozyme—lysozyme contacts (cf. Table 3). Taken together, these data suggest that BSA favors other BSA molecules over CI2, while lysozyme seems to favor CI2 over interactions with other lysozyme molecules.

Representative CI2—crowder interactions are shown in Figure 7. There are extensive close-range interactions with lysozyme in contrast to only a few interactions with BSA. This is further illustrated with time-averaged radial distributions of crowder heavy atoms around each residue of CI2 (cf. Figure 8). Again, there are extensive interactions between CI2 and lysozyme

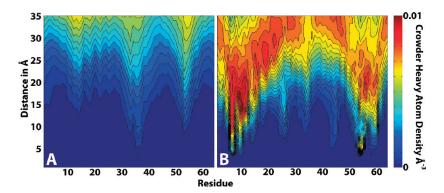


Figure 8. BSA (A) and lysozyme (B) crowder heavy atom radial densities around $C\alpha$ atoms of CI2 residues during 30–115 ns. Color coding indicates number densities according to color bar.

that were strongest near the N- and C-termini of CI2 but involved the entire structure of CI2. The picture is entirely different for CI2—BSA interactions with few close contacts, mostly near the extended loop. Comparing Figure 8 with zFigure 4, one finds that most residues with the closest crowder contacts also exhibited increased rmsf. This suggests that direct protein—protein contacts are responsible for the destabilization of the CI2 structure in the presence of lysozyme.

The free energy of crowding (i.e., transfer from aqueous solvent to a crowded environment) was calculated as described in the Methods section. There are significant changes in enthalpy and solvation free energies due to crowding, especially in the case of lysozyme due to more extensive interactions (cf. Table 4), but enthalpy and solvation changes largely cancel each other. The CI2 entropy is reduced modestly, mostly resulting from a partial loss of translational entropy (cf. Table 5). The overall crowding free energy is estimated to be negative with lysozyme and positive with BSA with a difference of 9.1 kcal/mol. An intriguing conclusion from these data is that in a heterogeneous cellular environment a solute would be drawn to an environment where more extensive protein-protein contacts are possible even if there is no strong specific binding. This may suggest a thermodynamic driving force toward spatial organization of proteins in cellular environments.

Our calculation of the free energy of crowding based on simulations of a solute in the presence of crowder molecules accounts for conformational responses resulting from solute-crowder interactions but is limited to folded states of the solute and may suffer from poor statistics due to slow kinetics of sampling different solute—crowder interactions. Alternatively, estimates of free energies of crowding can also be obtained with a postprocessing approach where only the crowder molecules are simulated first and free energies for insertion of solute molecules are then obtained with a Widom-type particle insertion method. 16,28 The advantage of this approach is that very different solute conformations can be readily tested, in particular folded and unfolded states, but at the expense of neglecting changes in solute or crowder conformations due to solute-crowder interactions because the systems are not simulated together. However, the general conclusions of such studies that nonspecific proteinprotein interactions are capable of destabilizing native solutes 16 are in agreement with our findings.

Finally, we will discuss translational diffusion constants as a function of crowding. A comparison of MSD as a function of time shows that diffusion was slowed down in the presence of the

Table 4. Estimated Free Energies of Crowding^a

	$\Delta\Delta H$	$\Delta\Delta G_{\rm solv}$	$-T\Delta\Delta S$	$\Delta\Delta G_{80/65}$	$\Delta\Delta G_{ m crowding}$		
BSA	-21.3 (1.6)	22.9 (1.54)	0.3 (0.1)	4.2 (0.9)	6.1 (0.3)		
Lys	-194.0(2.0)	186.2 (1.95)	0.5 (0.1)	4.3 (0.6)	-3.0(0.5)		
^a [kcal/mol] with statistical errors given in parentheses; $T = 298 \text{ K}$							
Averages over the trajectories exclude the first 30 ns.							

crowder molecules (cf. Figure 9). Furthermore, diffusion constants with BSA appear to be significantly different for short times (<10 ns) vs longer times (>15 ns). Hence, we determined diffusion constants from the slope of MSD vs time according to the Einstein relationship for two regimes: 1–10 ns and 15–30 ns (cf. Table 6). The long-time diffusion constant is reduced by half in the presence of BSA and to about one-fourth with lysozyme. These results are in reasonable agreement with experimental data, ²⁹ but there is a general overestimation of diffusion rates, especially in dilute solvent. This is likely a result of the TIP3P water model used here, which is known to result in overestimated diffusion rates. ³⁰

A significant difference in diffusion rates for different time regimes is an indication of anomalous diffusive behavior. Anomalous diffusion is expected for dynamics in crowded and confined media,³¹ where motions over short distances are presumed to be less affected by the crowder obstacles than long-range diffusion. Diffusion of CI2 with BSA crowders fits this picture. The shorttime diffusion rate is close to the rate in aqueous solvent, but long-time diffusion is significantly slowed down. In contrast, diffusion in the presence of lysozyme does not seem to involve significant anomalous behavior. The diffusion rates were reduced significantly already for short time motions and changed little between the short- and long-time regimes. This fundamental difference in diffusive behavior in the presence of BSA and lysozyme can be explained by the difference in CI2 interactions between lysozyme and BSA. With BSA, there are few interactions so that CI2 can essentially move around freely until occasionally colliding with a BSA molecule. This is similar to the diffusive behavior seen in the presence of noninteracting hard sphere crowders. With lysozyme, CI2 is always involved in a complex with one or more crowder molecules. The larger complexes are expected to diffuse slower than CI2 alone. Furthermore, since there is little evidence of anomalous diffusion, it appears that these complexes act effectively like a larger molecule diffusing in dilute solvent.

Table 5. Estimated Changes in Entropy Due to Crowding^a

		volume [ų]	$\Delta\Delta S_{\mathrm{trans}}$	area on unit sphere	$\Delta\Delta S_{ m rot}$	ΔS_{vib} QHA	$\Delta\Delta S_{\mathrm{vib}}$	$\Delta\Delta S_{ ext{total}}$
CI2W	30-160 ns					10.82		
CI2W/CI2LYS ^b	30-160 ns	14985		12.6				
CI2W/CI2BSA ^b	30-115 ns	7128		11.6				
CI2LYS	30-160 ns	7668	-1.31	12.3	\approx 0	10.50	-0.32	-1.64
CI2BSA	30-115 ns	4158	-1.05	10.8	\approx 0	10.76	-0.06	-1.11

^a Estimated change in entropy of CI2 between dilute solvent and crowded environments. All entropies are given in units of [cal/mol/K]. ^b Conformations from CI2W relative to crowder positions in CI2LYS/CI2BSA.

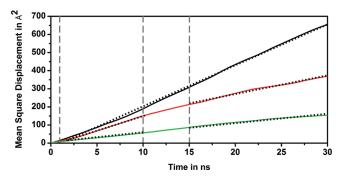


Figure 9. Mean square displacement (MSD) for the CI2 center of mass in CI2W (black), CI2LYS (green), and CI2BSA (red) simulations using all data after 30 ns. Vertical lines indicate short (1-10 ns) and long (15-30 ns) time regimes. Dashed lines indicate linear fits used to estimate diffusion coefficients.

Table 6. Diffusion Constants of CI2^a

	1-10 ns	15-30 ns	experiment ²⁹
water	31.8	31.7	15.5
BSA	26.6	14.2	8.2
lysozyme	8.8	7.4	5.1

 $^{^{}a}$ [μ m 2 /s] obtained according to Einstein relationship from slope of mean square displacement vs time.

CONCLUSIONS

The MD simulations provide a detailed view of how protein—protein interactions give rise to crowding effects in biological environments. The results directly complement recent experimental studies of the same systems. ¹⁰ The use of atomistic detail and full explicit solvent provides a more realistic description of crowding effects than in previous studies of crowding. In particular, protein—protein and protein—water interactions are described in a realistic fashion, and hydrodynamic effects are fully included through the behavior of the explicit water molecules. However, the use of full atomistic detail limits accessible time scales to the submicrosecond time regime.

The results presented here challenge the traditional view of biological crowding as being primarily a volume exclusion effect. Instead we propose that variable interactions between protein crowders and a given biomolecular solute are the key factor for understanding the effect of crowding in cellular environments. At one end of the spectrum, if a solute does not interact strongly with protein crowders, as in the CI2—BSA example, the solute structure and dynamics remain largely unaffected. The only significant effect was a moderate reduction in long-time diffusion

rates due to the crowders acting as obstacles. On the other hand, strong interactions with protein crowders, as in the CI2—lysozyme example, lead to perturbations of the solute structure, experimentally measurable destabilization, and significant reduction in diffusion rates due to long-time association with crowders. The effect of crowding in real biological environments, where protein—protein interactions span a wide range, is therefore expected to depend strongly on the local environment of a particular solute. Future studies will expand on the present work to a wider range of systems and different concentrations of crowder molecules.

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■ ABBREVIATIONS

BSA, bovine serum albumin; CHARMM, Chemistry at Harvard molecular mechanics program; CI2, chymotrypsin inhibitor 2; CI2W, CI2 simulation in water; CI2LYS, CI2 simulation with lysozyme; CI2BSA, CI2 simulation with BSA; CMAP, cross-correlation map for torsion angles; MD, molecular dynamics; MSD, mean-square displacement; NMR, nuclear magnetic resonance; NTP, constant number, temperature, and pressure; rmsd, root-mean-square deviation; rmsf, root-mean-square fluctuation; SASA, solvent-accessible surface area; TIP3P, transferable intermolecular potential 3 point water model

■ REFERENCES

- (1) Minton, A. P. Biopolymers 1981, 20, 2093.
- (2) Zhou, H.-X.; Rivas, G.; Minton, A. P. Annu. Rev. Biophys. 2008, 37, 375.
 - (3) Dix, J. A.; Verkman, A. S. Annu. Rev. Biophys. 2008, 37, 247.
 - (4) Kozer, N.; Schreiber, G. J. Mol. Biol. 2004, 336, 763.
 - (5) Wieczorek, G.; Zielenkiewicz, P. Biophys. J. 2008, 95, 5030.
- (6) Ridgway, D.; Broderick, G.; Lopez-Campistrous, A.; Ruaini, M.; Winter, P.; Hamilton, M.; Boulanger, P.; Kovalenko, A.; Ellison, M. *Biophys. J.* **2008**, *94*, 3748.
- (7) Takahashi, K.; Arjunan, S. N. V.; Tomita, M. FEBS Lett. 2005, 579, 1783
- (8) Cheung, M. S.; Klimov, D.; Thirumalai, D. *Proc. Natl. Acad. Sci. U.S.A.* **2005**, 102, 4753.

- (9) Zhou, H. X. J. Mol. Recognit. 2004, 17, 368.
- (10) Miklos, A. C.; Sarkar, M.; Wang, Y.; Pielak, G. J. J. Am. Chem. Soc. 2011, 133, 7116.
- (11) Ignatova, Z.; Krishnan, B.; Bombardier, J. P.; Marcelino, A. M. C.; Hong, J.; Gierasch, L. M. Biopolymers (Pept. Sci.) 2007, 88, 157.
 - (12) Ghaemmaghami, S.; Oas, T. G. Nat. Struct. Biol. 2001, 8, 879.
- (13) Inomata, K.; Ohno, A.; Tochio, H.; Isogai, S.; Tenno, T.; Nakase, I.; Takeuchi, T.; Futaki, S.; Ito, Y.; Hiroaki, H.; et al. *Nature* **2009**, 458, 106.
- (14) Tanizaki, S.; Clifford, J. W.; Connelly, B. D.; Feig, M. *Biophys. J.* **2008**, 94, 747.
 - (15) Despa, F.; Fernandez, A.; Berry, R. S. Phys. Rev. Lett. 2004, 93.
- (16) McGuffee, S. R.; Elcock, A. H. PLoS Comput. Biol. 2010, 6, e1000694.
 - (17) Sanchez, R.; Sali, A. Proteins 1997, Supplement 1, 50.
- (18) Darden, T. A.; York, D.; Pedersen, L. G. J. Chem. Phys. 1993, 98, 10089.
 - (19) Miyamoto, S.; Kollman, P. A. J. Comput. Chem. 1992, 13, 952.
- (20) MacKerell, A. D., Jr.; Bashford, D.; Bellott, M.; Dunbrack, J. D.; Evanseck, M. J.; Field, M. J.; Fischer, S.; Gao, J.; Guo, H.; Ha, S.; et al. *J. Phys. Chem. B* **1998**, *102*, 3586.
- (21) MacKerell, A. D., Jr.; Feig, M.; Brooks, C. L., III J. Comput. Chem. 2004, 25, 1400.
- (22) MacKerell, A. D., Jr.; Feig, M.; Brooks, C. L., III J. Am. Chem. Soc. 2004, 126, 698.
- (23) Graf, J.; Nguyen, P. H.; Stock, G.; Schwalbe, H. J. Am. Chem. Soc. 2007, 129, 1179.
- (24) Jorgensen, W. L.; Chandrasekhar, J.; Madura, J. D.; Impey, R. W.; Klein, M. L. J. Chem. Phys. 1983, 79, 926.
 - (25) Gohlke, H.; Case, D. A. J. Comput. Chem. 2003, 25, 238.
- (26) Lee, M. S.; Feig, M.; Salsbury, F. R., Jr.; Brooks, C. L., III. J. Comput. Chem. 2003, 24, 1348.
- (27) Swanson, J. M. J.; Henchman, R. H.; McCammon, J. A. Biophys. J. 2004, 86, 67.
- (28) Qin, S. B.; Minh, D. D. L.; McCammon, J. A.; Zhou, H. X. J. Phys. Chem. Lett. **2010**, 1, 107.
 - (29) Wang, Y.; Li, C.; Pielak, G. J. J. Am. Chem. Soc. 2010, 132, 9392.
 - (30) Makarov, V. A.; Feig, M.; Pettitt, B. M. Biophys. J. 1998, 75, 150.
 - (31) Saxton, M. J. Biophys. J. 1994, 66, 394.