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Impacts of Land Use Change on Nitrogen Cycling Archived in Semiarid Unsaturated Zone Nitrate Profiles, Southern High Plains, Texas

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Nitrate (NO₃) profiles in semiarid unsaturated zones archive land use change (LUC) impacts on nitrogen (N) cycling with implications for agricultural N management and groundwater quality. This study quantified LUC impacts on NO₃ inventories and fluxes by measuring NO₃ profiles beneath natural and rainfed (nonirrigated) agricultural ecosystems in the southern High Plains (SHP). Inventories of NO₃—N under natural ecosystems in the SHP normalized by profile depth are extremely low (2-10 kg NO₃-N/ha/m), in contrast to those in many semiarid regions in the southwestern U.S. Many profiles beneath cropland (9 of 19 profiles) have inventories at depth that range from $28-580 \text{ kg NO}_3-\text{N/ha/m}$ (median 135 kg/ha/m) that correspond to initial cultivation, dated using soil water Cl. These inventories represent 74% (median) of the total inventories in these profiles. This NO₃ most likely originated from cultivation causing mineralization and nitrification of soil organic nitrogen (SON) in old soil water (precultivation) and is attributed to enhanced microbial activity caused by increased soil wetness beneath cropland (median matric potential -42 m) relative to that beneath natural ecosystems (median -211 m). The SON source is supported by isotopes of NO₃ (δ^{15} N: +5.3 to +11.6; δ^{18} 0: +3.6 to +12.1). Limited data in South Australia suggest similar processes beneath cropland. Mobilization of the total inventories in these profiles caused by increased drainage/ recharge related to cultivation in the SHP could increase current NO₃—N levels in the underlying Ogallala aquifer by an additional 2-26 mg/L (median 17 mg/L).

1. Introduction

How can LUC impact nitrogen cycling and ultimately affect groundwater quality? Many studies have shown that changes from natural ecosystems to agricultural ecosystems decreased soil organic N (SON), which is highly correlated with reductions in soil organic carbon (SOC) (e.g., $r^2 = 0.7$ (1)).

Cultivation in prairie soils in western Canada caused large losses in ON (31-56%) and OC (41-53%) (2). Reductions in SON and SOC are attributed to decreased plant organic matter inputs and increased outputs through erosion, crop harvests, and decomposition of organic matter (3). Cultivation breaks up the aggregate structure of soil, which influences soil wetness and oxygenation and increases conversion of SON to NO₃ and SOC to the greenhouse gas, CO₂ (3). Experimental results indicate that net N conversion to NO₃ is linearly related to soil wetness (matric potential between -400 and -1 m) (4). Conversion of SON to NO₃ is a two step process involving mineralization (ammonification) which converts SON to NH₄ and nitrification which converts NH₄ through NO₂ to NO₃. Therefore, cultivation can result in creation of NO₃ reservoirs. Oxidation of SON during initial cultivation and swamp reclamation based on groundwater $\delta^{15}N_{NO_3}$ of +4.7 to +11.4%resulted in contamination of the Coastal Plain aquifer in Israel (5). In semiarid regions, increased drainage below the root zone and groundwater recharge generally occur when natural grassland and shrubland ecosystems are converted to rainfed (nonirrigated) agricultural ecosystems (6). Increased drainage also increases soil wetness which can enhance microbially mediated mineralization and nitrification, and create NO₃ reservoirs. Mineralization and nitrification of SON are also found in many agricultural areas today and are used as a source of NO_3 for crops (7).

In addition to creation of NO₃ reservoirs, LUC can also mobilize these NO₃ reservoirs, including any that existed prior to cultivation. Increased drainage/recharge beneath rainfed cropland in semiarid regions can displace NO₃ reservoirs into underlying aquifers. Irrigation also increases drainage/recharge from excess irrigation applications and results in mobilization of NO3 reservoirs. Examples of NO3 reservoirs that existed prior to LUC include NO3 accumulations from atmospheric deposition and evapotranspirative concentration in the southwestern U.S. (8) and from N fixation by leguminous natural vegetation, as evidenced by high NO₃/Cl ratios relative to those in precipitation in Senegal (9). Changes in land use from natural ecosystems to cropland are also associated with additional inputs of N from inorganic and organic fertilizers (10), from N fixation by leguminous crops (9), and from NO₃ in irrigation water (11).

Studies that evaluate impacts of LUC on biogeochemical cycles are generally limited by the absence of monitoring data prior to cultivation, which occurred over decades to about a century ago in many regions. However, thick unsaturated zones in semiarid regions provide a historical record of system response to LUC that includes precultivation periods. Kinniburgh (12) used the phrase: "The unsaturated zone, key to the past, guide to the future." Soil physics and environmental tracer profiles provide a guide to past impacts of LUC on water and nutrient cycling and a key to the future by predicting groundwater concentrations of different elements from mobilization within the unsaturated zone. Pressure data and water content data in the unsaturated zone provide information on LUC impacts on water movement. Soil—water Cl concentrations provide a conservative tracer that essentially tracks water movement and can be used to date soil water and LUC (13). Typical Cl bulges in natural semiarid regions can be used to track conditions prior to cultivation, and the top of the Cl bulge generally represents the transition from natural to cropland conditions (14). Comparison of NO₃ and Cl data can be used to evaluate nonconservative processes (e.g., additional sources or sinks for NO₃).

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The objectives of this study were (1) to evaluate impacts of LUC on unsaturated zone NO3 reservoirs in a semiarid region, including possible creation and mobilization of NO₃ reservoirs, and (2) to assess the implications for groundwater quality. To achieve these objectives, NO₃ reservoirs beneath natural and agricultural ecosystems in the southern High Plains were quantified, processes related to the distribution of NO₃ reservoirs were evaluated, and impacts of mobilizing these NO₃ reservoirs on groundwater quality were estimated. This study represents an expansion of previous work that focused on impacts of LUC on the water cycle (11, 13). Novel $aspects\ of\ this\ study\ include\ the\ large\ number\ of\ unsaturated$ zone profiles in different land use settings (natural ecosystems, five profiles; rainfed agricultural ecosystems, 19 profiles); understanding of flow and transport processes in the unsaturated zone provided by data on soil water content, matric potential, and Cl concentrations; and availability of long-term records of groundwater NO₃ concentrations from the Texas Water Development Board database (www.twdb. state.tx.us). The results of this study should provide valuable information to agricultural communities and groundwater resource specialists to optimize N management for enhancing soil fertility and sustainability and minimizing future loading to underlying aquifers.

2. Materials and Methods

The study area in the SHP (75 000 km²) in Texas is underlain by the Ogallala or High Plains aquifer, the largest aquifer in the U.S. Sediments in the unsaturated zone include clays, silts, sands, and gravels of eolian and alluvial origin. The clay content in the upper $1.5-2~\mathrm{m}$ soil zone ranges from 1 to 68% (mean 28%) in the SHP and decreases from north to south (15). Land use in the SHP consists of 53% cropland and fallow, 45% grassland/shrubland, and 2% other (Figure 1). Cotton is the dominant crop in the region. Approximately 11% of the SHP (21% of cropland) is irrigated (16).

Previous studies in the SHP showed that groundwater recharge beneath natural ecosystems is restricted to ephemeral lakes, termed playas (17). Unsaturated flow in interplaya areas is characterized by net upward flow (no drainage/recharge) and accumulation of Cl with a bulge shape, peaking near the root zone, as a result of evapotranspirative enrichment of infiltrating precipitation during the past 10 000–15 000 years since the Pleistocene time (13). The time (t, year) required to accumulate chloride is as follows:

$$t = \int_{0}^{z} \rho_{b} \text{Cl}_{uz} dz / (P \times \text{Cl}_{p})$$
 (1)

where ρ_b is bulk density, Cl_{uz} is Cl concentration in the unsaturated zone (mg/kg of dry soil), dz is depth interval, P is mean annual precipitation, and Cl_p is Cl concentration in the precipitation. LUC from natural ecosystems to rainfed agricultural ecosystems increased drainage below the root zone and ultimately groundwater recharge (11, 13). Drainage (11) or recharge (11) rates were calculated using the chloride mass balance (11) approach by dividing 110 input by 111 concentrations in the unsaturated zone (111 input by 112 concentrations in the unsaturated zone (111 input by 112 concentrations in the unsaturated zone (111 input by 112 concentrations in the unsaturated zone (111 input by 112 concentrations in the unsaturated zone (112 is depth interval, 113 in the unsaturated zone (113 in the unsaturated zone (114 input by 115 in the unsaturated zone (114 input by 115 in the unsaturated zone (114 input by 115 in the unsaturated zone (115 input by 116 i

$$D = R = \frac{P \times \text{Cl}_{\text{p}}}{\text{Cl}_{\text{uz}}}$$
 (2)

Potassium chloride fertilizer was not applied at any of the cropland sites in this study. Results from 19 profiles beneath rainfed cropland throughout the SHP showed that drainage/recharge increased from 0 mm/year under natural ecosystems in interplaya settings to 24 mm/year (median) under cropland, which represents \sim 5% of mean annual precipitation (13). About half of the profiles (10 of 19) had low Cl concentrations throughout the profile with young pore water ages (\leq 15–62 years, estimated using the CMB method),

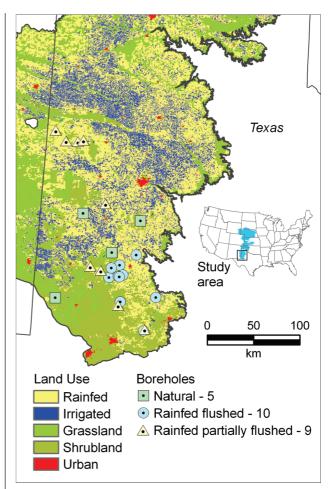


FIGURE 1. Location of sampled boreholes in natural ecosystems (five boreholes) and in rainfed agricultural ecosystems (10 profiles with young soil water, CI completely flushed and nine profiles with old water at the base, CI partially flushed). Inset shows location of southern High Plains in the U.S. Generalized land cover is based on National Land Cover Data (NLCD), 1992. Agricultural categories (rainfed and irrigated) represent a combination of NLCD classifications (row crops, 38%, small grains, 12%; pasture/hay, 2%; fallow and orchards, 1%). Irrigation coverage is based on ref 16.

corresponding to postcultivation, and high downward water fluxes (6–92 mm/year; median 42 mm/year) (13). These profiles are referred to as flushed because the precultivation Cl bulge has been completely displaced through the sampled profile. Although a Cl bulge could exist below the sampled depth, this is unlikely because 9 of the 10 "flushed" profiles are found in the southeastern SHP where the water table has risen several meters in the past few decades, indicating that the wetting front has reached the water table (13). High tritium levels in groundwater in this region also indicate recent recharge (18). The other nine partially flushed profiles have high Cl concentrations toward the base of the profiles (peaks 238–2103 mg/L), reflecting the transition from natural ecosystems (high Cl) at the base to cropland (low Cl) conditions at shallower depths.

The same boreholes that were used to evaluate impacts of LUC on the water cycle were also used to evaluate impacts of LUC on NO₃ inventories and fluxes in this study. Most of the methods used in this study are described in ref 13. A brief description is included in this section. Boreholes were drilled to depths of 2.9–29 m (Figure 1). Core samples were used for laboratory measurement of water content and matric potential. Concentrations of Cl, NO₂, and NO₃ were determined by adding double deionized water to the sediment

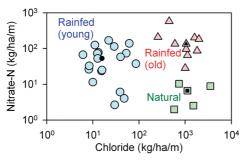


FIGURE 2. Relationship between NO_3 —N inventory and CI inventory in natural ecosystems, and young (postcultivation) and old (precultivation) soil water in rainfed agricultural ecosystems normalized by profile depths. Solid symbols represent median values.

sample in a 1:1 to 2:1 ratio by weight, shaking for 4 h, centrifuging the supernatant, and filtering through 0.2 μ m filters. Ion concentrations were analyzed by ion chromatography (Dionex ICS 2000; detection limit 0.01 mg/L). Ion concentrations are expressed on a mass basis as mg ion per kg of dry soil (= supernatant concentration multiplied by extraction ratio, g water/g soil and divided by water density) and as mg ion per L of soil pore water (= mg/kg divided by gravimetric water content and multiplied by water density). Concentrations on a mass basis are useful for interprofile comparisons and to reduce variations from differences in soil water contents due to textural variability. Inventories of ions (kg/ha) were calculated by multiplying depth-weighted salt concentrations (mg/kg) by the interval thickness (m), soil bulk density (kg/m³), and 10⁴ (m²/ha) for units conversion. Stable isotopes of nitrogen ($\delta^{15}N_{NO_2}$) and oxygen ($\delta^{18}O_{NO_2}$) in NO₃ were analyzed in 10 soil water extracts from six boreholes to determine whether variable sources of NO₃ could be distinguished. The isotopes were analyzed by the bacterial N₂O method at the U.S. Geological Survey Stable Isotope Laboratory in Reston VA (19).

3. Results and Discussion

Inventories of NO_3 and Cl, normalized by profile depth, vary by land use: low NO_3 and high Cl beneath natural ecosystems, moderate to high NO_3 and low Cl in young soil water beneath cropland, and high NO_3 and moderate to high Cl in old soil water beneath cropland (Figure 2). All inventories have been normalized by profile depth. All cropland sites are rainfed (nonirrigated).

3.1. Nitrate Profiles beneath Natural Ecosystems. Inventories of NO₃ in five profiles beneath natural ecosystems (grasslands and shrublands) provide information on background levels of NO₃ in the system prior to cultivation. Inventories of NO₃-N are all low (2-10 kg N/ha/m, median 7 kg N/ha/m, \geq 1 m depth) and profile concentration means range from 2 to 9 mg NO₃-N/L (median: 7 mg/L) (Table 1, Figures 2, 3c, and Supporting Information Figure S1a). The NO₃/Cl molar concentration ratios in soil water (0.001–1.6) are lower than the [NO₃ + NH₄]/Cl molar ratios in precipitation (3.8-18.2, median 10.9; 1985-2006) from the National Atmospheric Deposition Program (NADP) site at Muleshoe, indicating that no additional source of NO3 is required to explain the measured NO₃ in these profiles. In contrast to the low NO₃-N inventories in these profiles, Cl inventories are high (580-3500 kg Cl/ha/m, median 1120 kg/ha/m) (Figure 2). Typical Cl profiles are bulge shaped (Figure 3b) and represent accumulation times of 3000-29 000 years, as a result of long-term drying over these times (13). Most profiles did not extend deep enough to sample the entire Cl bulge, with the exception of the Lyn06-01 profile. The soils have low water contents and low matric potentials, indicating dry conditions (Figure 3a, Table 1).

3.2. Nitrate Inventories in Young Soil Water Representing Rainfed Agriculture (Postcultivation). Concentrations of NO₃ in soil water, which infiltrated after the land was converted to cropland (postcultivation or young soil water; >15 to >89 years ago in different profiles (Table 1)), provide information on NO₃ levels associated with agriculture. A total of 10 of the 19 profiles had the precultivation Cl bulge completely flushed through the sampled profile and young soil water throughout, whereas the remaining nine profiles had young soil water restricted to the upper 1.6-8.9 m zone underlain by old (precultivation) water with high Cl concentrations representing the top of the Cl bulge. Inventories of NO₃-N in this young soil water range from 3 to 169 kg/ ha/m (median 53 kg/ha/m) (Table 1). Many of the profiles (seven profiles) have low NO₃-N concentrations below the root zone (1 m) with peak concentrations from 8 to 26 mg/L (Table 1). The remaining 12 profiles have higher NO₃-N concentrations with distinct peaks from 38 to 287 mg/L with peak depths from 1.0 to 4.6 m (Figure 3i and o).

Potential sources of NO₃ in these young soil waters include atmospheric deposition, fertilizers, biological fixation, and mineralization, and nitrification of soil organic nitrogen (SON). The dominant fertilizers applied in the SHP are ammonium based. Although pesticides and herbicides could be used to track a fertilizer source, previous studies of soils under two irrigated fields in the SHP indicate that pesticides and herbicides are retained or degraded in the upper 2 m of the soil zone (10). SON does not represent an original source and could ultimately be derived from atmospheric deposition, biological fixation, and/or fertilizers. Molar ratios of NO₃/Cl in soil water are up to 59 (Table 1) and exceed molar ratios found in precipitation (3.8–18.2) in many profiles indicating an additional source of NO₃ to the system, most likely fertilizer. Isotopic values in young soil water range from +4.7 to +8.8% for $\delta^{15}N_{NO_3}$ and from +7.1 to +9.1% for $\delta^{18}O_{NO_3}$ (Supporting Information, Figure S2 and Table S1). These $\delta^{15}N_{NO_3}$ values are slightly higher than typical NO₃ fertilizers and may be attributed to volatilization of ¹⁵N depleted ammonia during or after application of ammonium-based fertilizers and are within the range of $\delta^{15}N_{NO_3}$ values found in soil water in alluvial fans in Texas that were attributed to ammonia volatilization ($\delta^{15} N_{NO_3}$ 2–14; mean 9) (20). The $\delta^{15}N_{NO_3}$ values are also consistent with nitrification of reduced SON in the soil zone from other studies ($\delta^{15}N_{NO_3}$: +4 to +8‰ (21, 22)). The $\delta^{18}O_{NO_3}$ values are typical of those expected from microbial nitrification of reduced N in ammoniumbased fertilizers or of SON in the soil zone assuming the O was derived from a 2:1 proportion of unfractionated H₂O-O and atmospheric O₂-O (+23.8‰) (23). Estimates of $\delta^{18}O_{H_2O}$ from precipitation (-22.7 to +4.9%; median -6.2%) or groundwater (-9.1 to -4.2%; median -6.2%) in the SHP (18) would result in $\delta^{18}O_{NO_3}$ of -7.1 to +11.1% (median +3.8%). The $\delta^{18}O_{H_2O}$ values may also be more enriched in the unsaturated zone as a result of evaporation. Therefore, the isotopes cannot distinguish between mineralization/ nitrification of SON and nitrification of ammonium-based fertilizers.

Fertilizer application rates are generally based on expected crop yield. Typical yields for rainfed cotton lint are 150 to 300 kg N/ha and recommended N fertilizer application rates for these yields range from 15–30 kg N/ha (24). Nitrogen application rates reported by land owners for the sampled sites range from 10–30 kg N/ha. Inputs from atmospheric deposition are much lower (0.3–1.8 kg NO₃–N/ha; 0.1–2.8 kg NH₄–N/ha). Estimates of field-measured net mineralization rates (subtracting inputs from irrigation water and harvests) from previous experiments in the SHP range from 12–51 kg N/ha per growing season in this region and are generally higher in more clayey soils (7). The original source of the mineralized N could be fertilizer N, biological fixation,

TABLE 1. Soil Physics (Matric Potential and Water Content) and Concentrations and Inventories of Nitrate and Chloride in Sampled Boreholes^a

peak	depth (m)		2.2	2.2	1.1	4.3	2.7	2.2		10.1	6.2	3.7	4.0	9.2	5.3	2.8	4.0	7.1	5.3		3.4	1.0	2.5	1.3	2.2	2.8	2.2	3.4	2.2	2.2		3.5	1.1	1.0	1.3	1.0	3.4	4.6	2.2	3.4	1.0	 8.	
concentration inventory pe	N (mg/L)		25	34	29	8.3	36	29		62	36	303	52	111	168	156	88	145	111		51	9.4	92	22	231	90	175	87	287	90		28	10	16	38	26	91	78	254	26	8.0	32	
	N (kg/ha(-m))		76 (10)	(8.8)	24 (6.6)	69 (2.5)	45 (2.0)	(9'9)	•	219 (65)	86 (28)	1057 (578)	265 (87)	309 (203)	824 (183)	(19 (97)	103 (135)	191 (209)	265 (135)		165 (24)	13 (6.1)	2127 (114)	34 (38)	640 (78)	224 (74)	212 (139)	160 (58)	874 (169)	212 (74)		288 (73)	9 (2.6)	8 (4.0)	85 (25)	80 (26)	552 (67)	366 (53)	647 (125)	214 (33)	63 (11)	149 (29)	
	CI (kg/ha(-m))		5381 (736)	25643 (3505)	4012 (1120)	41046 (1507)	13200 (577)	13200 (1120)		3753 (1119)	3047 (1000)	820 (448)	5508 (1807)	2430 (1594)	8904 (1981)	6731 (1052)	455 (597)	235 (257)	3047 (1052)											108 (39)		50 (13)	100 (30)	91 (48)	54 (16)	25 (8.2)	50 (6.1)	88 (13)	57 (11)	55 (8.4)	55 (10)	55 (12)	
	N (mg/L)		9.1	7.2	6.1	1.6		9.9	ıtion)	45	15	268	32	79	06	67	80	120	79	vation)	12	3.0	63	17	47	33	77	32	83	33	on)	30	2.2	3.1	12	13	37	34	63	15	4.8	4	
	CI (mg/L)		477	2,514	1,072	889		980	(precultiva	785	557	210	377	550	1,220	755	385	155	557	r (postculti	5.3	19	16	41	8.2	20	34	21	10	19	ostcultivatio	5.5	25	36	7.0	4.4	3.5	8.4	5.3	4.1	4.7	5.4	
	N/CI (mol/mol)	tems	0.002 - 1.6	0.002 - 0.38	0.003 - 1.4	0.001 - 1.1		0.002 - 1.3	hed-old water	0.05 - 0.53	0.04 - 0.11	2.0 - 4.5	0.09 - 0.14	0.16 - 1.1	0.006 - 1.0	0.08-7.2	0.24 - 0.81	1.5-2.4	0.09 - 1.0	ed-young wate	0.79 - 16	0.16 - 4.0	2.0 - 15	0.72 - 1.7	3.0 - 45	1.6 - 9.6	3.7 - 5.9	2.6-8.5	3.9-57	2.0 - 9.6	oung water (po	3.8-31	0.06 - 0.6	0.11 - 1.4	1.8-10	4.2 - 33	1.1-61	3.6 - 27	3.1 - 59	5.3-40	0.51 - 13	2.5–29	
	CMB age (year)	natural ecosystems	4183	17291	3029	29100	10780	10780	ride partially flus	3614	2914	712	4150	2206	7600	4370	378	300	2914	de partially flushe	82	64	43	44	22	88	29	71	74	64	ıloride flushed—γ	34	62	62	36	16	33	28	35	23	15	34	
Historical DTW	WC (g/g)		0.08	60.0	0.07	0.12		211 0.08	gricultu	ricultu		0.13	0.14	0.17	rO	0.11	0.09	0.10	0.11	0.11	culture,	0.12	0.13	0.11	0.13	0.10	0.15	0.09	0.10	0.13	0.12	culture, ch	0.15	0.08	60.0	0.14	0.13	0.12	0.10	0.13	0.13	0.15	0.13
	MP (-m)		145	144	278	283	! !					18	33	16		52	29	42	78	42		13	6.3	26	5.0	4.2	6.4	99	25	26	13	rainfed agri	2.1			2.2	1.0	6.1	9.9	26	12	21	6.3
	DTW 2000 (m)		28	30	28	27	23		rai		52	21	െ	21	36	34	32	28		rainfe	28	52	21	ത	21	36	34	32	28		_	6	7	14	വ	10	10	10	14	27	33		
	Ε		56	31	27	56	31			32	45	23	13	30	38	23	31	32			32	45	23	13	30	38	23	31	32			22	22	30	14	22	28	22	20	20	34		
	date		1958 1938			1979 1969		ļ		1979		1979	1963 1938		1938	1979	1947	1953	1969	2		1979	1979	1963	1938	1938	1979	1947	1953	1969			1938	1938	1938	1937	1938	1938	1938	1936	1947	1953	
	depth (m)		8.3	8.3	4.7	29.0	24.0			11.4	0.9	4.7	2.0	10.7	8.5	8.9	4.5	7.1			7.7	2.8	2.5	1.6	8.9	3.4	2.2	3.4	5.9			2.0	4.3	2.9	4.4	4.0	9.5	7.9	6.2	7.5	6.5		
	setting borehole		And05-02	Daw06-01	Daw03-12	Lyn06-01	Ter06-03	Median		Bai05-01	Bai05-02	Bai06-01	Gai05-01	Gai05-02	Lam05-01	Mar05-02	Mar05-04	Ter05-01	Median		Bai05-01	Bai05-02	Bai06-01	Gai05-01	Gai05-02	Lam05-01	Mar05-02	Mar05-04	Ter05-01	Median		Daw03-02	Daw03-03	Daw03-04	Daw03-05	Daw05-01	Daw06-02	Daw06-03	How05-01	Mar05-01	Mar05-03	median	

concentrations in mg/L), CI and NO₃—N concentrations and inventories, and peak NO₃—N concentrations and associated depths. CMB ages for natural and partially flushed old water are for the base of the CI-flushed profile and do not include the root zone age (1 m depth). MP, WC, and concentration values represent depth-weighted means below the root zone (1 m). Inventory values in parentheses are normalized by the sampled borehole depth. Depths refer to borehole depths, with the exception of partially flushed young water where depths refer to the flushed zone, historical depth to water (DTW) and associated balance (CMB) ages, range of NO₃-N/Cl molar ratios (based on a Borehole settings include natural ecosystems, rainfed agricultural ecosystems (chloride partially flushed: old water and young water, and chloride flushed: young water). measurement year, DTW (~year 2000), matric potentials (MP), water contents (WC), chloride mass

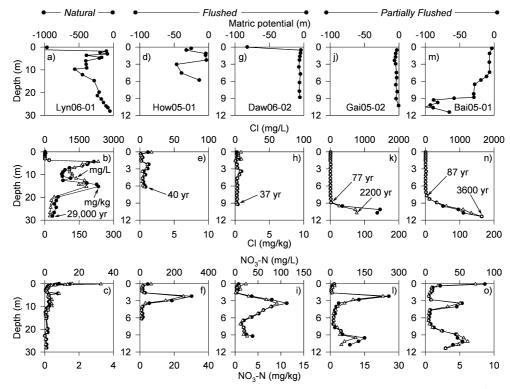


FIGURE 3. Representative matric potential, CI and NO_3 —N profiles in a natural grassland/shrubland ecosystem (Lyn06-01), rainfed agricultural ecosystems where CI is flushed (young soil water only; How05-01 and Daw06-02), and CI is partially flushed (young and old soil water; Gai05-02 and Bai05-01). Concentrations are shown in mg/L soil pore water and mg/kg of dry sediment. Chloride mass balance ages for soil pore water are shown at the base of the CI flushed zone and the base of the profile.

and/or N from long-term atmospheric deposition stored in organic matter in the soil. Previous studies in this region reported ^{15}N recoveries in plants ranging from 19 to 28%, resulting in $\sim\!70-80\%$ of the N partially volatilizing and/or remaining in the soil (7). Therefore, rates of fertilizer application and SON mineralization are sufficient to account for NO3 inventories measured in these profiles.

3.3. Nitrate Inventories in Old Soil Water Representing Land Use Change Impacts. The partially flushed Cl profiles (nine profiles) have old (precultivation) soil water toward the base of the profiles with CMB ages up to 7600 years (Table 1). High NO₃ concentrations are found toward the base of these profiles, surrounding the zone where Cl concentrations increase and where soil water ages predate agricultural fertilizer applications (Figure 3l and o). The NO₃-N inventories in this zone range from 28 to 578 kg N/ha/m (median: 135 kg N/ha/m) with peak NO₃-N concentrations of 36-303 mg/L and peak depths from 2.8-10.1 m (Figure 3l and o, Table 1). Some of the profiles did not extend deep enough to sample the entire NO₃-N inventory. The NO₃-N inventories in the mineralized zone (86–1057 kg N/ha) represent 18–89% (median 74%) of total inventories in these profiles below the root zone (1 m). There are several possible explanations for the high NO₃ in this transition zone.

(1) It could reflect high NO_3 concentrations that accumulated under natural ecosystems prior to cultivation. However, the fact that none of the profiles under contemporary natural ecosystems have high NO_3 concentrations does not seem to be consistent with this scenario (Figures 2, 3c; Table 1).

(2) It could result from preferential flow of water with NO_3 from fertilizers or mineralization of SON from shallow zones. However, the gradual changes in NO_3 concentrations would not be expected if preferential flow was the transport mechanism (Figure 3l and o). Uniformly low Cl concentrations in the flushed portion of these profiles provide strong

evidence for predominantly piston-type flow in these settings (Figure 3k and n) (13).

(3) The most likely source of this high NO₃ zone in old water beneath cropland is mineralization and nitrification of SON that occurred during initiation of cultivation. Isotopes in soil water extracts from the proposed mineralized zones of profiles range from +5.3 to +11.6% for $\delta^{15}N_{NO_3}$ and from +3.6 to +12.1% for $\delta^{18}O_{NO_3}$, which are consistent with nitrification of reduced SON (Supporting Information, Figure S2, Table S1). The similarity in $\delta^{15}N_{NO_3}$ and $\delta^{18}O_{NO_3}$ values in deep NO₃ peaks in old soil water and shallow NO₃ peaks in young soil water indicates that the source of the NO₃ in these two zones is not distinguishable using isotopes. Both old and young NO₃ may be derived from mineralization and nitrification of SON. SON does not represent the original source of the NO₃, which could be atmospheric deposition, N fixation by vegetation or bacteria, and/or a geologic source. Molar ratios of NO_3/Cl in the old soil water are ≤ 7 (Table 1), which are generally less than ratios found in precipitation (3.8–18.2) and could be derived from atmospheric deposition. Much larger total inventories of NO3-N are found in other semiarid regions from atmospheric deposition in the southwest U.S. and central High Plains ($\leq 13600 \text{ kg N/ha}$ (8)). The lack of high NO₃ under natural ecosystems in the SHP could indicate that the NO₃ from atmospheric deposition over the past 10 000 to 30 000 years was assimilated and immobilized in organic matter in the SHP, or the N as NH₄ could have been adsorbed onto clays, or partially adsorbed onto clays and volatilized as NH₃. Much more enriched N isotopes in paleogroundwater in the SHP ($\delta^{15}N_{NO_3}+12.3\%$) than in the northern high plains (+1.3%) was attributed to loss of N from the system (25), possibly through ammonia volatilization.

How do SON mineralization and nitrification occur in old soil water? The most likely mechanism is related to increased soil wetness beneath cropland. Mineralization is a microbially mediated process and is thus affected by soil

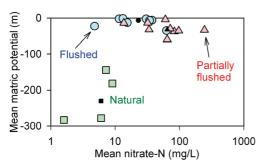


FIGURE 4. Mean matric potential versus mean NO_3 — NO_3

water content and temperature (4, 26). Experimental results from previous studies indicate that net N mineralization is linearly related to matric potential between -400 and -1 m (4). Median matric potentials in the mineralized zones (-2.5 to -78 m; median for all profiles: -42 m) are much higher than median matric potentials in profiles beneath natural ecosystems (-144 to -283 m; median -211 m), indicating wetter conditions that would enhance microbial mineralization and nitrification (Table 1; Figure 4). Cultivation results in downward movement of a wetting (pressure) front and solute (chloride) front (13, 27) (Supporting Information, Figure S3). The wetting front is marked by an increase in moisture content and matric potential and migrates ahead of the chloride front. The velocity of the wetting front $(v_{\rm wf})$ and chloride front $(v_{\rm cf})$ are as follows:

$$v_{\rm wf} = \frac{q_{\rm w}}{\theta_{\rm w} - \theta_{\rm d}}; \ v_{\rm cf} = \frac{q_{\rm w}}{\theta_{\rm w}}; \ \frac{v_{\rm wf}}{v_{\rm cf}} = \frac{\theta_{\rm w}}{\theta_{\rm w} - \theta_{\rm d}}$$
(3)

where $q_{\rm w}$ is water flux (drainage or recharge rate), $\theta_{\rm w}$, and $\theta_{\rm d}$ are water contents above and below the wetting front, and $v_{\rm wf}$ / $v_{\rm cf}$ is the ratio of the velocities of the wetting and chloride fronts. The chloride front represents the interface between new water above and old water below (Supporting Information Figure S3). The zone between the chloride front and wetting front does not represent newly infiltrated water but downward displacement of stored water in the profile. This is the main zone where SON mineralization occurs. The mineralization proposed in this study at the beginning of cultivation may be similar to N mineralization flushes associated with drying/rewetting of soils that occurs at much shorter timescales (28).

Similar processes may have occurred in semiarid regions in southern Australia, where shallow rooted crops and pasture replaced native eucalyptus vegetation in the early 1900s. High NO₃ concentrations extend into the zone of high Cl concentrations at depth in partially flushed profiles (Figure 5) (Supporting Information, Figure S4, Table S2) (29, 30), beneath rainfed cropland, similar to the relationship between NO₃ and Cl in the SHP. The BEM23 profile is from a sandy area at the Borrika site, with a postcultivation drainage rate at least 23 mm/year (29, 31). The 6HC profile is also from a sandy location in the Riverlands region, near the Murray River, with a postcultivation recharge rate of 14.8 mm/year (30). However, similar NO₃ bulges were also found in the shallow zone of two natural profiles at the Borrika site (29); therefore, additional profiles would be required to better understand processes related to LUC in this region.

Many previous studies have shown that SON decreases with cultivation along with SOC (1, 32, 33). Bronson (33) reported that total N in the 0–0.3 m sampled soil in the SHP was 2500 kg N/ha for native vegetation and 2100 kg N/ha for rainfed cropland, respectively, equivalent to a 400 kg N/ha decrease with cultivation. The reduction in SON is within

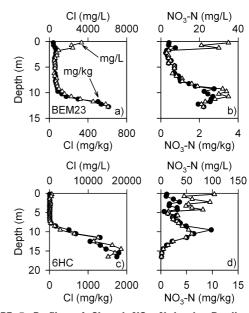


FIGURE 5. Profiles of CI and NO_3 —N in the Borrika region (BEM23) (29) and in the Murray Riverlands region (6HC) in rainfed agricultural ecosystems with partially flushed CI in S. Australia (Supporting Information, Table S2).

the range of increases in NO_3 —N from mineralization and nitrification of SON in old water with cultivation in the current study (86–1057 kg N/ha). The organic carbon (OC) reduction (0.25%) reported by Bronson (35) from natural ecosystems (0.76% OC) to rainfed cropland (0.51%) in the upper 0.3 m is also consistent with the amount of OC required to explain mineralized N inventories in this study (0.2% OC required for largest measured N inventory, 1000 kg N/ha, Table 1) (Supporting Information).

3.4. Impact of Mobilization of Unsaturated Zone Nitrate on Groundwater Contamination. Contamination of the Ogallala aquifer with NO $_3$ is widespread in the SHP. Most NO $_3$ contamination is concentrated in the southern half of the SHP (SHP-S) with 25% of wells exceeding the MCL (10 mg/L NO $_3$ —N) whereas only 1% of wells exceed the MCL in the northern half of the SHP (SHP-N) (Supporting Information, Figure S5). High NO $_3$ —N concentrations in the SHP-S region are attributed in part to the shallow water table (median depth \sim 30 m) relative to the SHP-N region (median depth \sim 63 m) (Figure 5) and to a lower aquifer saturated thickness (median \sim 15 m) in the SHP-S relative to the SHP-N region (median \sim 21 m) providing less water to dilute incoming contaminants.

The potential impact of mobilizing NO₃ by increasing recharge beneath cropland areas was evaluated by estimating the groundwater concentration from transport of the unsaturated zone NO₃ inventory and mixing within the aquifer (median saturated thickness, 15 m in the SHP-S, porosity 0.3) (Supporting Information). Mobilizing the entire NO₃-N inventories (100-1270 kg/ha) in the partially flushed profiles into the aquifer would increase groundwater NO3-N concentrations by 2-26 mg/L (median 17 mg/L). This calculation ignores dispersion. In addition, boreholes were not drilled to the water table; therefore, measured inventories underestimate total inventories. The NO₃-N inventory in the mineralized zone represents 74% (median) of the total sampled inventories in the partially flushed profiles and should continue to contribute substantially to groundwater NO₃ contamination in the SHP. Mobilizing NO₃-N inventories (8-874 kg/ha) in young water (postcultivation) should increase groundwater concentrations relative to current values by $0.2-18\,mg/L$ (median $4.4\,mg/L$). Unsaturated zone NO₃ profiles provide an excellent archive of past impacts of

LUC on N cycling and indicate that mineralization and nitrification of SON associated with the beginning of cultivation represents a substantial source of NO_3 in this system. Mineralization and nitrification also represent a continuous source of NO_3 that needs to be considered when estimating crop fertilizer requirements.

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Supporting Information Available

Plots of water content, matric potential, chloride, and nitrate for all sampled profiles (Figure S1), stable isotopes of nitrate (Figure S2, Table S1), schematic of relative movement of wetting and chloride fronts (Figure S3), Australian profiles and summary data (Figure S4, Table S2), and example calculations for organic carbon requirements for nitrogen mineralization and for estimating impacts of mobilizing unsaturated zone nitrate into the underlying aquifer. This material is available free of charge via the Internet at http://pubs.acs.org.

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