be complete after ca. 3 min (no further addition of HCl required). The mixture was stirred for an additional 50 min and poured into 800 mL of ice—water resulting in the formation of a precipitate. The mixture was cooled in an ice bath for an additional 4 h and filtered, and the solids were washed with additional water. After drying, 6d (4.87 g, 98%) was obtained as a colorless solid: mp 190–191 °C; $[\alpha]_D = +95^\circ$ (c = 1, methanol), [lit. mp 187–190 °C, $[\alpha]_D = +99^\circ$ (c = 0.26, DMF)]. Anal. ($C_{17}H_{17}NO_4S$) C, H, N, S.

(3R-cis)-3-(Acetyloxy)-1-[2-(dimethylamino)ethyl]-1,3,4,5-tetrahydro-4-(4-methoxyphenyl)-2H-1-benzazepin-2-one, Monohydrochloride (1f). A solution of (3R-cis)-3-(acetyloxy)-7-chloro-1-[2-(dimethylamino)ethyl]-1,3,4,5-tetrahydro-4-(4-methoxyphenyl)-2H-1-benzazepin-2-one monohydrochloride (1h, 170 mg, 0.36 mmol, see ref 1 for preparation) was neutralized in ether by washing with with saturated NaHCO₃. The ether solution of the free base was then concentrated and the residue redissolved in acetic acid (10 mL). Catalyst was then added (80

mg of 10% Pd/C), and the mixture was shaken overnight under 30 psi of hydrogen gas at room temperature. The mixture was then filtered through Celite and the filtrate, plus washes, was washed with saturated NaHCO₃. The organic layer was dried over MgSO₄ and concentrated. The residue was dissolved in ether and treated with HCl-saturated ether. The resulting white precipitate was filtered and dried to yield 0.13 g (83%) of the desired product: mp 63–65 °C; [α]_D = +117.2° (c = 1, MeOH); ¹H NMR (CD₃OD) δ 7.45 (m, 3 H), 7.35 (m, 1 H), 7.25 (d, 2 H), 6.90 (d, 2 H), 5.05 (d, 1 H), 4.30 (m, 2 H), 3.80 (m, 4 H), 3.65 (m, 1 H), 3.45 (m, 1 H), 3.00 (m, 8 H), 1.85 (s, 3 H); ¹³C NMR (CD₃OD) δ 20.35, 37.94, 44.05, 45.34, 51.53, 55.74, 56.46, 73.19, 114.71, 123.76, 128.88, 129.96, 130.67, 130.96, 132.63, 135.51, 141.38, 160.41, 171.07, 171.79. Anal. (C₂₃H₂₉N₂O₄·HCl·1.5H₂O) C, H, N.

Acknowledgment. Analytical data were supplied by the Bristol-Myers Squibb Analytical Research and Development Department.

Studies on Antidiabetic Agents. 11.1 Novel Thiazolidinedione Derivatives as Potent Hypoglycemic and Hypolipidemic Agents²

Takashi Sohda,* Katsutoshi Mizuno, Yu Momose, Hitoshi Ikeda, Takeshi Fujita, and Kanji Meguro

Research and Development Division, Takeda Chemical Industries, Ltd., 17-85, Jusohonmachi 2-chome, Yodogawa-ku, Osaka 532, Japan. Received January 13, 1992

In the course of further chemical modification of the novel antidiabetic pioglitazone (AD-4833, U-72, 107), a series of 5-[4-(2- or 4-azolylalkoxy)benzyl- or -benzylidene]-2,4-thiazolidinediones was prepared and evaluated for hypoglycemic and hypolipidemic activities in insulin-resistant, genetically obese, and diabetic KKA^y mice. Replacement of the 2-pyridyl moiety of pioglitazone by a 2- or 4-oxazolyl or a 2- or 4-thiazolyl moiety greatly enhanced in vivo potency. The corresponding 5-benzylidene-type compounds, in which a methine was used as a linker between the benzene ring and the thiazolidinedione ring, also had potent biological activity. Among the compounds synthesized, 5-[4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzyl]-2,4-thiazolidinedione (18) exhibited the most potent activity, more than 100 times that of pioglitazone. The synthesis and structure-activity relationships for this novel series of derivatives are detailed.

Introduction

Insulin resistance is a characteristic feature of noninsulin-dependent diabetes mellitus (NIDDM), in particular when it is associated with obesity. This insulin-resistant state at the peripheral tissue level causes impaired glucose utilization leading to hyperglycemia.³ Therefore, amelioration of insulin resistance with a drug would provide a novel and useful means of treating NIDDM. With regard to this concept^{3,4} few drugs have, however, been studied, and exercise and calorimetric restriction are still the fundamental modes of treating NIDDM patients. It has been reported that sulfonylureas, the most commonly used oral hypoglycemics, potentiate insulin action in peripheral tissues by increasing the number of insulin receptors; however, their main mechanism of action actually involves insulin secretion. This insulin secretion can bring about undesired effects such as induction of hypoglycemia.

In 1982, we reported a series of 5-(4-alkoxybenzyl)-2,4-thiazolidinediones^{6a} as novel antidiabetic agents which were shown to effectively reduce insulin resistance or potentiate insulin action in genetically diabetic and/or obese animals. Ciglitazone, a prototypical compound of the series (Chart I), was shown to normalize hyperglycemia, hyper-

 Part X: Momose, Y.; Meguro, K.; Ikeda, H.; Hatanaka, C.; Oi, S.; Sohda, T. Studies on Antidiabetic Agents. X. Synthesis and Biological Activities of Pioglitazone and Related Compounds. Chem. Pharm. Bull. 1991, 39, 1440-1445.

(2) This work has been presented at the 1st Annual Meeting of Division of Medicinal Chemistry, The Pharmaceutical Society of Japan (December 4-6, Okayama in Japan, 1991; Abstract pp 62-63). (a) Meguro, K.; Fujita, T. Thiazolidine Derivatives, Their Production and Use. European Patent 177 353, 1986. (b) Meguro, K.; Fujita, T. Thiazolidine Derivatives, Their Production and Use. U.S. Patent 4 775 687, 1988.

 (a) Reaven, G. M. Insulin-Independent Diabetes Mellitus: Metabolic Characteristics. Metabolism 1980, 29, 445-454. (b) DeFronzo, R. A.; Ferrannini, E.; Koivisto, V. New Concepts in the Pathogenesis and Treatment of Noninsulin-Dependent Diabetes Mellitus. Am. J. Med. 1983, 74 (Suppl. 1A), 52-81.
 (c) Olefsky, J. M.; Kolterman, O. G. Mechanisms of Insulin Resistance in Obesity and Noninsulin-Dependent (Type II) Diabetes. Am. J. Med. 1981, 70, 151-168.

(4) Reaven, G. M. Therapeutic Approaches to Reducing Insulin Resistance in Patients with Noninsulin-Dependent Diabetes Mellitus. Am. J. Med. 1983, 74 (Suppl. 1A), 109-112.

⁽⁵⁾ Gavin, J. R., III Dual Actions of Sulfonylureas and Glyburide. Receptor and Post-Receptor Effects. Am. J. Med. 1985, 79 (Suppl. 3B), 34-42.

^{(6) (}a) Sohda, T.; Mizuno, K.; Imamiya, E.; Sugiyama, Y.; Fujita, T.; Kawamatsu, Y. Studies on Antidiabetic Agents. II. Synthesis of 5-[4-(1-Methylcyclohexylmethoxy)benzyl]thiazolidine-2,4-dione (ADD-3878) and Ita Derivatives. Chem. Pharm. Bull. 1982, 30, 3580-3600. (b) Fujita, T.; Sugiyama, Y.; Taketomi, S.; Sohda, T.; Kawamatsu, Y.; Iwatsuka, H.; Suzuoki, Z. Reduction of Insulin Resistance in Obese and/or Diabetic Animals by 5-[4-(1-Methylcyclohexylmethoxy)benzyl]thiazolidine-2,4-dione (ADD-3878, U-63,287, Ciglitazone), a New Antidiabetic Agent. Diabetes 1983, 32, 804-810. (c) Chang, A. Y.; Wyse, B. M.; Gilchrist, B. J.; Peterson, T.; Diani, A. R. Ciglitazone, a New Hypoglycemic Agent. I. Studies in ob/ob and db/db Mice, Diabetic Chinese Hamsters, and Normal and Streptozotocin-Diabetic Rats. Diabetes 1983, 32, 830-838.

Method A (X = S, O)

^a(a) Aqueous HBr, NaNO₂ then CH₂=CHCOOMe, Cu₂O; (b) (NH₂)₂CS, NaOAc; (c) aqueous HCl.

insulinemia, and hypertriglyceridemia in various insulinresistant animal models without altering normoglycemia in nondiabetic animal models. Since our discovery of ciglitazone, many other structurally analogous thiazolidinediones have been reported, and drug development

(7) (a) Yoshioka, T.; Fujita, T.; Kanai, T.; Aizawa, Y.; Kurumada, T.; Hasegawa, K.; Horikoshi, H. Studies on Hindered Phenols and Analogues. 1. Hypolipidemic and Hypoglycemic Agents with Ability To Inhibit Lipid Peroxidation. J. Med. Chem. 1989, 32, 421-428. (b) Zask, A.; Jirkovsky, I.; Nowicki, J. W.; McCaleb, M. L. Synthesis and Antihyperglycemic Activity of Novel 5-(Naphthalenylsulfonyl)-2,4-thiazolidinediones. J. Med. Chem. 1990, 33, 1418-1423. (c) Clark, D. A.; Goldstein, S. W.; Volkmann, R. A.; Eggler, J. F.; Holland, G. F.; Hulin, B.; Stevenson, R. W.; Kreutter, D. K.; Gibbs, E. M.; Krupp, M. N.; Merrigan, P.; Kelbaugh, P. L.; Andrews, E. G.; Tickner, D. L.; Suleske, R. T.; Lamphere, C. H.; Rajeckas, F. J.; Kappeler, W. H.; McDermott, R. E.; Hutson, N. J.; Johnson, M. R. Substituted Dihydrobenzopyran and Dihydrobenzofuran Thiazolidine-2,4-diones as Hypoglycemic Agents. J. Med. Chem. 1991, 34, 319-325. (d) Iizima, I.; Ozeki, M.; Okumura, K.; Inamasu, M. Preparation of 5-[benzoxazolylmethyl (or methylene)]-2,4-thiazolidinediones as antidiabetics. European Patent 283 035, 1988. (e) Kees, K. L.; Cheeseman, R. S. Preparation of hypoglycemic 5-[[(hydroximinocyclohexyl)methoxy]benzyl]-2,4-thiazolidinediones. U.S. Patent 4728739, 1988. (f) Cantello, B. C. C.; Hindley, R. M. Urea and thiourea derivatives of thiazolidinedione, a process for their preparation, their pharmaceutical compositions, and their use in the therapy of hyperglycemia or hyperlipidemia. European Patent 295 828, 1988.

Chart I

pioglitazone(AD-4833, U-72,107)

along this line is becoming one of the major concerns in the field of antidiabetic agents.⁸

In a previous paper,9 we reported on the potent hypo-

⁽⁸⁾ For efforts related to non-thiazolidinedione ciglitazone-like agents, see: (a) Kees, K. L.; Cheeseman, R. S.; Prozialeck, D. H.; Steiner, K. E. Perfluoro-N-[4-(1H-tetrazol-5-ylmethyl)-phenyl]alkanamides. A New Class of Oral Antidiabetic Agents. J. Med. Chem. 1989, 32, 11-13. (b) Dow, R. L.; Bechle, B. M.; Chou, T. T.; Clark, D. A.; Hulin, B.; Stevenson, R. W. Benzyloxazolidine-2,4-diones as Potent Hypoglycemic Agents. J. Med. Chem. 1991, 34, 1538-1544. (c) Lombardo, L. J.; Alessi, T. R. Preparation of novel naphthalenyl-3H-1,2,3,5-oxathiadiazole 2-oxides as hypoglycemics. U.S. Patent 4895 861, 1990.

Scheme IIa

Method B (X = S, O)

a (a) NaH.

Scheme IIIa

Method C (X = S, O)

^a(a) 2,4-Thiazolidinedione, piperidine.

glycemic and hypolipidemic activity of pioglitazone (AD-4833, U-72, 107) in insulin-resistant animal models such as KKA^y mice¹⁰ and Wistar fatty rats.¹¹ The structure–activity relationship (SAR) studies on pioglitazone and related compounds revealed that the presence of a pyridyl ring on the p-alkoxy chain of the benzyl moiety potentiated biological activities and that location of the pyridine nitrogen α to the oxyethyl chain was a very important factor for the activity.^{1,9a} These findings led us to the syntheses of other heterocyclic analogues of pioglitazone having the ring nitrogen at the same relative position. This report details some SAR studies on this series.

Chemistry

Most of the 5-benzyl-2,4-thiazolidinediones in Tables I and II were synthesized starting from 4-[(2- or 4-)oxazolylor -thiazolylalkoxy]aniline derivatives 1-3 by the general procedure described in previous papers. 1,6a,9a Meerwein arylation of the aniline derivatives gave the 3-aryl-2-bromopropionates 4, which were then treated with thiourea to afford the iminothiazolidinones 5. Acid hydrolysis of 5 gave the desired 2,4-thiazolidinediones 6-32 and 44-48 in good yield (method A, Scheme I).

The requisite 4-(4-azolylalkoxy)aniline derivatives 1 were prepared by condensation of 4-azolyl alcohols with p-fluoronitrobenzene followed by reduction as described in

(9) (a) Sohda, T.; Momose, Y.; Meguro, K.; Kawamatsu, Y.; Sugiyama, Y.; Ikeda, H. Studies on Antidiabetic Agents. Synthesis and Hypoglycemic Activity of 5-[4-(Pyridylalkoxy)benzyl]-2,4-thiazolidinediones. Arzneim.-Forsch. 1990, 40, 37-42. (b) Ikeda, H.; Taketomi, S.; Sugiyama, Y.; Shimura, Y.; Sohda, T.; Meguro, K.; Fujita, T. Effects of Pioglitazone on Glucose and Lipid Metabolism in Normal and Insulin Resistant Animals. Arzneim.-Forsch. 1990, 40, 156-162. (c) Sugiyama, Y.; Taketomi, S.; Shimura, Y.; Ikeda, H.; Fujita, T. Effects of Pioglitazone on Glucose and Lipid Metabolism in Wistar Fatty Rats. Arzneim.-Forsch. 1990, 40, 263-267. (d) Sugiyama, Y.; Shimura, Y.; Ikeda, H. Effects of Pioglitazone on Hepatic and Peripheral Insulin Resistance in Wistar Fatty Rats. Arzneim.-Forsch. 1990, 40, 436-440.

(10) Iwatsuka, H.; Shino, A.; Suzuoki, Z. General Survey of Diabetic Features of Yellow KK Mice. Endocrinol. Jpn. 1970, 17, 23-35.

(11) Ikeda, H.; Shino, A.; Matsuo, T.; Iwatsuka, H.; Suzuoki, Z. A New Genetically Obese-Hyperglycemic Rat (Wistar Fatty). Diabetes 1981, 30, 1045-1050. the synthesis of the 4-[2-(substituted 2-pyridyl)ethoxy]-benzyl derivatives. The 2-(2-oxazolyl)ethoxy analogues 3 were also obtained by reduction of the corresponding nitro compounds, which were synthesized by cyclization of α -[3-(4-nitrophenoxy)propionyl]amino ketones with POCl₃. Coupling of 4-(bromoacetyl)azoles with p-acetamidophenol followed by NaBH₄ reduction and alkaline hydrolysis afforded the anilines 2 bearing a hydroxy group on the 4-ethoxy chain.

O-Alkylation of 5-(4-hydroxybenzyl)-2,4-thiazolidinedione with (2- or 4-chloromethyl)azoles 52 provided a simpler method for the preparation of 5-(2- or 4-azolylmethoxy)benzyl derivatives and was used as the general synthetic procedure for compounds 33-40 and 49-51 (method B, Scheme II).

Compound 41 possessing a hydroxymethyl group at the 5-position on the oxazole ring was obtained by bromination of 18 with NBS followed by hydrolysis (see Experimental Section). Oxidation of 30 and 31 was effected by DMSO-Ac₂O to give the corresponding 2-oxoethyl derivatives 42 and 43, respectively.

The 5-benzylidene-2,4-thiazolidinediones 54-64 were obtained by Knoevenagel condensation of the benzaldehydes 53 with 2,4-thiazolidinedione using piperidine as base (method C, Scheme III). The starting benzaldehydes 53 were readily prepared by the method described in the previous paper. The 2-oxoethyl derivatives 63 and 64 were reduced with NaBH₄ to the corresponding hydroxy compounds 65 and 66, respectively.

2- or 4-functionalized oxazoles and thiazoles used in methods A-C above were prepared following the procedure reported previously by Meguro et al.¹² The derivatives in Tables I-III are simple racemates or racemic mixtures of diastereomers, and no attempts at optical resolution have been carried out.

Biological Procedures

The biological activities of the compounds prepared were tested using genetically obese and diabetic KKA^y mice¹⁰ (8–11-week-old). After being fed a laboratory chow (CE-2, Clea Japan Inc., Tokyo, Japan) for 3 d, the mice were

⁽¹²⁾ Meguro, K.; Tawada, H.; Sugiyama, Y.; Fujita, T.; Kawamatsu, Y. Studies on Antidiabetic Agents. VII. Synthesis and Hypoglycemic Activity of 4-Oxazoleacetic Acid Derivatives. Chem. Pharm. Bull. 1986, 34, 2840-2851.

Table I. Physical Data and Yield of 5-[4-(4-Azolylalkoxy)benzyl]-2,4-thiazolidinediones

entry	\mathbb{R}^1	\mathbb{R}^2	Х	Y	n	$prep$ $method^a$	yield ^b (%)	mp (°C)	formula	anal.c
6	Н	Me	S	CH ₂	1	A	20	185-186	$C_{16}H_{16}N_2O_3S_2$	C,H,N
7	Me	H	\mathbf{s}	CH_2	1	Α	21	142-143	$C_{16}H_{16}N_2O_3S_2$	C,H,N
8	Et	H	S	CH_2	1	Α	40	148-149	$C_{17}H_{18}N_2O_3S_2$	C,H,N
9	i-Pr	H	\mathbf{s}	CH_2	1	Α	34	107-108	$C_{18}H_{20}N_2O_3S_2$	C,H,N
10	cyclohexyl	H	S	CH_2	1	Α	26	289-291	$C_{21}H_{23}N_2O_3S_2Na$	C,H,N
11	Ph	Н	\mathbf{s}	CH_2	1	Α	41	113-114	$C_{21}H_{18}N_2O_3S_2$	C,H,N
12	Me	H	0	CH_2	1	Α	22	184-185	$C_{16}H_{16}N_2O_4S$	C,H,N
13	Pr	Н	О	CH_2	1	Α	11	87-88	$C_{18}H_{20}N_2O_4S$	C,H,N
14	Me	Me	0	CH_2	1	Α	46	200-201	$C_{17}H_{18}N_2O_4S$	C,H,N
15	Me	$\mathbf{E}\mathbf{t}$	0	CH_2	1	Α	31	189-190	$C_{18}H_{20}N_2O_4S$	C,H,N
16	cyclohexyl	H	0	CH_2	1	Α	13	269-271	$C_{21}H_{23}N_2O_4SNa^{-1}/_2H_2O$	C.H.N
17	Ph	н	0	CH_2	1	Α	29	10 9- 110	$C_{21}H_{18}N_2O_4S$	C,H,N
18	Ph	Me	0	CH_2	1	Α	52	116-117	$C_{22}H_{20}N_2O_4S$	C,H,N
19	Ph	Et	0	CH_2	1	Α	21	10 9 –111	$C_{23}H_{22}N_2O_4S$	C,H,N
20	Ph	Me	0	CH_2	2	Α	34	130-131	$C_{23}H_{22}N_2O_4S$	C,H,N
21	cyclohexyl	Me	0	CH_2	1	Α	27	273-275	$C_{22}H_{25}N_2O_4SN_8$	C,H,N
22	2-furyl	Me	0	CH_2	1	Α	21	114-115	$C_{20}H_{18}N_2O_5S$	C,H,N
23	2-thienyl	Me	О	CH_2	1	Α	32	144-145	$C_{20}H_{18}N_2O_4S_2$	C,H,N
24	$4-(MeO)C_6H_4$	Me	0	CH_2	1	Α	51	167-168	$\mathbf{C_{23}H_{22}N_2O_5S}$	C,H,N
25	$3,4-(MeO)_2C_6H_3$	Me	0	CH_2	1	Α	29	167-168	$C_{24}H_{24}N_2O_6S$	C,H,N
26	$3-(Me)C_6H_4$	Me	0	CH_2	1	Α	37	99-100	$C_{23}H_{22}N_2O_4S$	C,H,N
27	$3-(MeS)C_6H_4$	Me	0	CH_2	1	Α	31	161-162	$C_{23}H_{22}N_2O_4S_2$	C,H,N
28	2-(Cl)C ₆ H ₄	Me	0	CH_2	1	Α	23	93-94	$C_{22}H_{19}N_2O_4SCl$	C,H,N
29	4-(OH)C ₆ H ₄	Me	0	CH_2	1	Α	45	213-214	$C_{22}H_{20}N_2O_5S$	C,H,N
30	Me	Me	0	CH(OH)	1	A	32	214-215	$C_{17}H_{18}N_2O_5S$	C,H,N
31	Ph	Me	0	CH(OH)	1	A	50	165-166	$C_{22}H_{20}N_2O_5S$	C,H,N
32	cyclohexyl	Me	0	CH(OH)	1	Α	19	oil	$C_{22}H_{26}N_2O_5S$	C,H,N
33	Me	H	\mathbf{s}	CH_2	0	В	40	181-182	$C_{15}H_{14}N_2O_3S_2$	C,H,N
34	Ph	H	S	CH_2	0	В	41	164-165	$C_{20}H_{16}N_2O_3S_2$	C,H,N
35	3-Py	Н	S	CH_2	0	В	13	205-206	$C_{19}H_{15}N_3O_3S_2$	C,H,N
36	Me	Н	0	CH_2	0	В	28	192-193	$C_{15}H_{14}N_2O_4S$	C,H,N
37	Pr	Н	0	CH_2	0	В	36	114-115	$C_{17}H_{18}N_2O_4S$	C,H,N
38	Ph	Н	0	CH_2	0	В	47	188-189	$C_{20}H_{16}N_2O_4S$	C,H,N
39	Ph	Me	0	CH_2	0	В	79	162-163	$C_{21}H_{18}N_2O_4S$	C,H,N
40	2-fu r yl	Me	0	CH_2	0	В	50	194-196	$C_{19}H_{16}N_2O_5S$	C,H,N
41	Ph	CH_2OH	0	CH_2	1	Ex	21	98-99	$C_{22}H_{20}N_2O_5S$	C,H,N
42	Me	Me	0	C=0	1	Ex	48	161-162	$C_{17}H_{16}N_2O_5S$	C,H,N
43	Ph	Me	0	C=0	1	Ex	81	168-169	$C_{22}H_{18}N_2O_5S$	C,H,N

^aA, B = general method, Ex = experimental procedure described. ^bOverall yield based on 1 or 2 (method A), 52 (method B). ^cCompounds gave satisfactory analyses (±0.4%).

Table II. Physical Data and Yield of 5-[4-(2-Azolylalkoxy)benzyl]-2,4-thiazolidinediones

$$R^1$$
 N
 R^2
 X
 $(CH_2)_n$
 N
 S
 N

entry	\mathbb{R}^1	\mathbb{R}^2	X	n	prep method ^a	yield ^b (%)	mp (°C)	formula	anal.c
44	i-Bu	Me	0	2	A	27	123-124	$C_{20}H_{24}N_2O_4S$	C,H,N
45	cyclohexyl	Me	0	2	Α	24	175-176	$C_{22}H_{26}N_2O_4S$	C,H,N
46	Ph	Me	0	2	Α	34	261-262	$C_{22}H_{19}N_2O_4SNa$	C,H,N
47	Ph	$\mathbf{E}\mathbf{t}$	0	2	Α	30	248-250	$C_{23}H_{21}N_2O_4SNa$	C,H,N
48	Me	Ph	0	2	A	21	168-169	$C_{22}H_{20}N_2O_4S$	C,H,N
49	Ph	H	S	1	В	49	164-165	$C_{20}H_{16}N_2O_3S_2$	C,H,N
50	2-benzo	thiazolyl		1	В	35	184-185	$C_{18}H_{14}N_2O_3S_2$	C,H,N
51	2-benz	oxazolyl		1	В	33	188-189	$C_{18}H_{14}N_2O_4S$	C,H,N

^aA, B = general method. ^bOverall yield based on 3 (method A) or 52 (method B). ^cCompounds gave satisfactory analyses (±0.4%).

divided into experimental groups of five mice each according to their blood glucose levels. The test compounds were given as a dietary admixture at 0.005% or 0.001% in the CE-2 powdered diet. The mice were fed the experimental diet and water ad libitum for 4 d. Blood samples were taken from the orbital vein. Blood glucose was determined using the glucose oxidase method¹³ and plasma

triglyceride using a commercially available assay kit (Cleantech TG-S, Iatron Laboratories Inc., Tokyo, Japan).

⁽¹³⁾ Werner, W.; Rey, H. G.; Wielinger, H. Über die Eigenschaften eines neuen Chromogens für die Blutzuckerbestimmung nach der GOD/POD-Methode. Fresenius' Z. Anal. Chem. 1970, 252, 224-228.

Table III. Physical Data and Yield of 5-[4-(4-Azolylalkoxy)benzylidene]-2,4-thiazolidinediones

entry	\mathbb{R}^1	\mathbb{R}^2	x	Y	n	prep method ^a	yield ^b (%)	mp (°C)	formula	anal.c
54	Me	Н	s	CH ₂	1	C	82	215-216	$C_{16}H_{14}N_2O_3S_2$	C,H,N
55	Ph	H	S	CH_2	1	C	91	210-211	$C_{21}H_{16}N_2O_3S_2$	C,H,N
56	Ph	Me	0	CH_2	1	C	95	214-215	$C_{22}H_{18}N_2O_3S$	C,H,N
57	Ph	$\mathbf{E}\mathbf{t}$	0	CH_2	1	C	72	175-176	$C_{23}H_{20}N_2O_3S$	C,H,N
58	4-(Cl)C ₆ H ₄	Me	0	CH_2	1	C	91	214-215	$C_{22}H_{17}CIN_2O_3S$	C,H,N
59	$3-(MeS)C_6H_4$	Me	0	CH_2	1	С	67	185-186	$C_{23}H_{20}N_2O_4S_2$	C,H,N
60	1-Me-cyclohexyl	Me	0	CH_2	1	С	51	172-175	$C_{23}H_{26}N_2O_4S$	C,H,N
61	2-thienyl	Me	0	CH_2	1	С	48	221-222	$C_{20}H_{16}N_2O_4S_2$	C,H,N
62	Ph	Me	0	CH ₂	0	C	81	225-226	$C_{21}H_{16}N_2O_4S$	C,H,N
63	Ph	Me	0	c <u>⊸</u> o	1	С	76	244-245	$C_{22}H_{16}N_2O_5S$	C,H,N
64	Me	Me	0	C=O	1	C	63	234-235	$C_{17}H_{14}N_2O_5S$	C,H,N
65	Ph	Me	0	CH(OH)	1	Ex	98	252-253	$C_{22}H_{18}N_2O_5S$	C,H,N
66	Me	Me	0	CH(OH)	1	Ex	98	223-224	$C_{17}H_{16}N_2O_5S$	C,H,N

 o C = general method, Ex = experimental procedure described. b Yield from 53 (method C). c Compounds gave satisfactory analyses ($\pm 0.4\%$).

The maximum decreases in blood glucose and plasma triglyceride levels were calculated as percentage change from the control value. Effective dose to reduce blood glucose and plasma triglyceride levels by 25% (ED $_{25}$) was determined using data from an experiment in which three different doses were used. The doses were selected according to the potency of compound. The dosages of test compounds (mg/kg/d) were calculated from food intake and body weight. ED $_{25}$ (mg/kg/d) was then derived by linear regression analysis of the data.

Results and Discussion

We have already reported a thiazole derivative 67 (Table IV) with an oxyethyl chain at the 5-position of the thiazole ring. 6a Detailed pharmacological evaluation of this compound revealed that 67 was slightly less active than ciglitazone (Table IV). A sharp increase in activity was, however, observed on shifting the side chain from the 5-to the 4-position of the thiazole ring, i.e. from β to α to the ring nitrogen, as seen in compounds 6 and 7. This finding parallels that observed in the case of the 3-pyridyl- and 2-pyridylethoxy compounds reported previously. 9a On the basis of this general trend between structure and activity, we focused our synthetic efforts on the preparation of various 5-(4-alkoxybenzyl)-2,4-thiazolidinediones bearing 4-oxazolyl and 4-thiazolyl moieties.

The 4-oxazolyl and 4-thiazolyl derivatives synthesized possessed, on the whole, superior activities over the compounds which had been studied previously. No significant difference was observed between the activities of the oxazole and thiazole derivatives (7 vs 12; 11 vs 17; 33 vs 36; 34 vs 38; etc.).

Variation of the distance between the benzene ring and the azole ring showed that two-carbon units were the most effective for eliciting the activities (7 > 33; 18 > 39 > 20; 11 > 34; etc.), as was the case with the phenylalkoxy or pyridylalkoxy derivatives. ^{6a,9a}

Among the compounds with two-carbon units (12-19) and (21-29), introduction of a methyl group at the 5-position of the oxazole ring potentiated the activity, whereas an ethyl group was not as effective, indicating that a methyl moiety was optimal (14 > 12) and (

The effect of a substituent at the 2-position was studied for both the two-carbon (6-19, 21-29) and one-carbon unit (33-40) compounds. For those bearing alkyl groups, the

number of carbon atoms present in the substituent seems to be unrelated to the activities (7 vs 8 and 9; 12 vs 13; 36 vs 37; etc.); however, conversion of the 2-alkyl group to a cycloalkyl group tended to increase the activity (7, 8 and 9 vs 10; 12 and 13 vs 16; 14 vs 21). Compound 21, in particular, showed very potent activity, being about 42–155 times more potent than ciglitazone. Compounds with an aryl or a heteroaryl group showed superior activities to those with an alkyl or a cycloalkyl group as can be seen in 11, 17–19, 22, 23, 34, 35, 38, 39, 40, etc. Introduction of substituent(s) into the 2-phenyl group of 18, an extremely active compound having a potency some 277–620 times that of ciglitazone, did not enhance activity but even resulted in a decrease in activity in some cases (e.g. 24 and 25).

Considering the structure-activity relationships mentioned above, it seems that the 2-position of the oxazole ring might play an important role in hydrophobic binding in the drugs active site. It seems likely that a phenyl group would fit best into the hydrophobic pocket of the binding site, and a cyclohexyl group would seem next best.

Derivatives of 18 hydroxylated at the 5-methyl (41) and on the 2-phenyl ring (29) showed only reduced activities. In contrast, compounds 30–32 having a hydroxy group on the ethoxy chain, which were expected to have the same effect as in the previous study with the 2-pyridylethoxy series, ^{9a} exhibited activities either stronger than or comparable to those of the parent compounds (14 vs 30, 18 vs 31, 21 vs 32). These facts support the suggestion obtained from the previous study, ^{9a} that a steric effect rather than a net increase in hydrophilicity of the molecule is the more important factor to consider when trying to rationalize how the hydroxy group induces favorable biological activity.

The activity of the oxo derivatives 42 and 43 seem to be slightly lower than those of the hydroxy derivatives (42 vs 30, 43 vs 31); however, these activities were at quite a high level.

Potent antidiabetic activity was also found in the 5-[4-(2-azolylalkoxy)benzyl]-2,4-thiazolidinediones (Table II), which have the oxyethyl chain α to the azole ring nitrogen. In particular, those having a bulky group at the 4-position of the ring had superior activities as can be seen in 44-47 and 49. On the other hand, compound 48 with only a small substituent (Me) at the 4-position did not show any noticeable activities at the doses employed. These facts as well as the findings with 18 and its analogues clearly

Table IV. Hypoglycemic and Hypolipidemic Activities of 2,4-Thiazolidinediones

		oglycemic activit	<u> </u>	hypolipidemic activity		
		e (%)		dose		
	0.001	0.005	$\mathrm{ED}_{25}{}^{b}$	0.001	0.005	$\mathrm{ED}_{25}{}^{b}$
6		52****			45***	
7	17	42****	4	15	52***	4
8		55****			37***	
9	104444	56****	0.0	2244	45****	
10	43****		0.8	33**		3
11	39***	52***	0.5	54***	63****	0.3
12	13	37***	5	20	42***	3
13	29**	52****	2	26***	65****	2
14	27* 15***	55****	2	10	56****	3 5
15 16	31***	46****	3 1	0 14	45***	9
17	41****	58****	0.3	51****	68***	0.5
18	55****	59****	0.05	67****	66****	0.09
19	49***	55****	0.3	55****	79***	0.6
20	70	21***	0.0	00	29**	0.0
21	51****	52***	0.2	58****	71****	0.6
22	50****	50****	0.4	41****	69****	0.9
23	50****	51****	0.09	41****	72****	0.3
24	53***	55****	0.3	63****	72****	0.6
25	52****		0.5	54***		0.6
26	49****		0.05	61****		0.06
27		52****		~ -	67****	0.00
28	52****	* -		57****	•	
29	22***			26****		
30	55****	51****	0.2	52****	62****	0.5
31	54***	58***	0.05	62****	81****	0.1
32		49****			67****	
33		19			30	
34	11**	28***	6	3	32*	8
35		25**			30**	
36		19**			1	
37		12			29*	
38	16**	48****	3	12	51***	4
39	23***	41****	3	17**	44***	5
40		43****			35***	
41		31***			39**	
42		39***			27***	
43		52****			67****	
44		53****			43***	
45		54****			51****	
46		54****			60****	
47		46****			51****	
48		18*			7	
49		48****			44**	
50 51		43*** 35**			43**	
54		34**			24	
55	49***	57***	0.6	44***	5 57****	1.3
56	55****	01	0.4	65****	01	0.2
57	00	54****	V-3	00	59***	0.2
58		54***			58***	
59		52***			58***	
60		24****			9	
61		55****			57****	
62		37***			40***	
63		45****			51****	
64		33***			29	
65	36****	56****	0.9	54***	60****	0.3
66	26****	59****	2.2	37*	61****	0.8
67						
	Ņ	—√Me △	NH			
	Ų) p—(0			
	(0.02%)	(0.01%)		(0.02%)	(0.01%)	
ciglitazone	31**	15*	40	25*	8	40
			31			25

^a Maximum reductions in blood glucose and plasma triglyceride levels in KKA^y mice at a dosage of 0.001 or 0.005% in the diet were calculated as percentage reduction with respect to the control value. ^b Effective dose (mg/kg/d) of 25% reduction, estimated from dose-response curve at three doses. ^c Statistically significant at (*) p < 0.05, (**) p < 0.02, (***) p < 0.01, (****) p < 0.001.

suggest that a structure having a bulky group (e.g. phenyl) and an oxyalkyl chain at the α and α' positions, respectively, of the oxazole or thiazole ring is favorable for activity. Judging from the activities of compounds 50 and 51, benzoxazole and benzothiazole would also seem to fit well into the binding site.

A series of 5-benzylidene-2,4-thiazolidinediones were prepared on the basis of our previous findings in the pyridylalkoxy series. 1 Most compounds exhibited potent hypoglycemic and hypolipidemic activities, showing the effectiveness of the methine moiety as a linker between the benzene and the thiazolidinedione rings. The potency shown was slightly low compared to that shown by the corresponding methylene counterparts, but a similar SAR was observed. Concerning the substituent at the 2-position of the azole ring, an aryl group was superior to an alkyl or cycloalkyl group. Compounds hydroxylated on the ethoxy chain were equipotent, and the oxo derivatives were slightly less potent than the parent compound.

In summary, extremely potent antidiabetic activites were attained in a series of 5-[4-[2-(4-oxazolyl)ethoxy]benzylor -benzylidene]-2,4-thiazolidinediones. Compound 18, the most potent compound in this series, did not alter plasma glucose in normal rats, which suggests that this series of compounds possess the same biological profile as ciglitazone. Further pharmacological evaluation is now in progress in order to select a compound for clinical use.

Experimental Section

Melting points were taken on a Yanagimoto micro melting point apparatus and are uncorrected. Elemental analyses (C, H, and N) were carried out by the Analytical Department of the Takeda Chemical Industries, Ltd. ¹H NMR spectra of deuteriochloroform or DMSO- d_6 solutions (internal standard TMS, δ 0) were recorded on a Varian EM-390 or a Varian Gemini-200 spectrometer. Infrared spectra were recorded on a Hitachi IR-215 spectrometer. All compounds exhibited ¹H NMR, IR, and analytical data consistent with the proposed structures. Column chromatography was done with E. Merck silica gel 60 (0.063-0.200 mm)

Method A. 5-[4-[2-(5-Methyl-2-phenyl-4-oxazolyl)ethoxy]benzyl]-2,4-thiazolidinedione (18). A solution of NaNO₂ (16.8 g, 0.24 mol) in H₂O (30 mL) was added dropwise to a stirred and ice-cooled mixture of 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]aniline (65.0 g, 0.22 mol), aqueous HBr (47%, 151.5 g, 0.88 mol), MeOH (200 mL), and acetone (500 mL) below 5 °C. The whole was stirred at 5 °C for 30 min and methyl acrylate (117 mL, 1.32 mol) was added and the temperature was raised to 38 $^{\circ}$ C. Powdered Cu₂O (2.0 g, 0.014 mol) was added in small portions to the vigorously stirred mixture. After a N2 gas evolution had ceased, the reaction mixture was concentrated in vacuo. The residue was diluted with H₂O, made alkaline with concentrated NH4OH, and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO4), and concentrated in vacuo to give methyl 2-bromo-3-[4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]phenyl]propionate as a crude oil (89.5 g, 92%): ¹H NMR $(CDCl_3)$ δ 2.33 (3 H, s), 2.93 (2 H, t, J = 7 Hz), 3.15 (1 H, dd, J= 14, 7 Hz), 3.40 (1 H, dd, J = 14, 7 Hz), 3.65 (3 H, s), 4.21 (2 H, t, J = 7 Hz), 4.32 (1 H, t, J = 7 Hz), 6.81 (2 H, d, J = 9 Hz), 7.08 (2 H, d, J = 9 Hz), 7.4 (3 H, m), 7.98 (2 H, m).

A mixture of methyl 2-bromo-3-[4-[2-(5-methyl-2-phenyl-4oxazolyl)ethoxy]phenyl]propionate (a crude oil, 38.6 g, 87 mmol), thiourea (6.6 g, 87 mmol), NaOAC (7.1 g, 87 mmol), and EtOH (350 mL) was stirred under reflux for 3 h and concentrated in vacuo. The residue was neutralized with aqueous NaHCO3 and Et₂O (150 mL)-hexane (150 mL) was added. The whole was stirred at room temperature for 15 min, and the crystals were collected by filtration to give 2-imino-5-[4-[2-(5-methyl-2phenyl-4-oxazolyl)ethoxy]benzyl]-4-thiazolidinone (19.1 g, 54%): mp 212-213 °C (EtOH). Anal. (C₂₂H₂₁N₃O₃S) C, H, N.

A mixture of 2-imino-5-[4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzyl]-4-thiazolidinone (18.8 g, 46 mmol), 2 N HCl (200 mL), and EtOH (200 mL) was stirred under reflux for 12 h. The reaction mixture was concentrated in vacuo. The residue was

diluted with H₂O, neutralized with saturated aqueous NaHCO₃, and extracted with CHCl₃. The CHCl₃ extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to give the title compound (18.0 g, 96%): mp 116-117 °C (EtOH); ¹H NMR $(CDCl_8) \delta 2.34 (3 H, s), 2.94 (2 H, t, J = 7 Hz), 3.02 (1 H, dd, J)$ = 14, 9 Hz), 3.41 (1 H, dd, J = 14, 4 Hz), 4.18 (2 H, t, J = 7 Hz), 4.42 (1 H, dd, J = 9, 4 Hz), 6.80 (2 H, d, J = 9 Hz), 7.09 (2 H, d, J = 9 Hz), 7.4 (3 H, m), 7.95 (2 H, m). Anal. ($C_{22}H_{20}N_2O_4S$)

Method B. 5-[4-(2-Phenyl-4-oxazolylmethoxy)benzyl]-2.4-thiazolidinedione (38). A solution of 5-(4-hydroxybenzyl)-2,4-thiazolidinedione^{6a} (9.4 g, 42 mmol) in DMF (80 mL) was treated with NaH (60% in oil, 3.4 g, 84 mmol) at room temperature for 30 min, and then a solution of 4-(chloromethyl)-2-phenyloxazole¹⁴ (9.6 g, 50 mmol) in DMF (20 mL) was added. The whole was stirred at 70 °C for 1 h, poured into H2O, and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to give the title compound (9.1 g, 47%): mp 188-189 °C (EtOH); ¹H NMR $(d_{\rm gr} {\rm DMSO}) \delta 3.03$ (1 H, dd, J = 14, 9 Hz), 3.34 (1 H, dd, J = 14, 4 Hz), 4.88 (1 H, dd, J = 9, 4 Hz), 5.03 (2 H, s), 6.98 (2 H, d, J= 9 Hz), 7.18 (2 H, d, J = 9 Hz), 7.5 (3 H, m), 8.0 (2 H, m). Anal. $(C_{22}H_{16}N_2O_4S)$ C, H, N.

Method C. 5-[4-[2-(5-Methyl-2-phenyl-4-oxazolyl)ethoxy]benzylidene]-2,4-thiazolidinedione (56). A mixture of 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzaldehyde (40.3 g, 0.131 mol), 2,4-thiazolidinedione (23.0 g, 0.196 mol), piperidine (2.6 mL, 26 mmol), and EtOH (700 mL) was refluxed for 9 h. After cooling, the precipitated crystals were collected by filtration to give the title compound (42.6 g, 80%): mp 214-215 °C (CHCl₃-EtOH); ¹H NMR (d_6 -DMSO) δ 2.35 (3 H, s), 2.94 (2 H, t, J = 6.5 Hz), 4.30 (2 H, t, J = 6.5 Hz), 7.07 (2 H, d, J = 9 Hz), 7.4-7.65 (5 H, m), 7.72 (1 H, s), 7.8-8.1 (2 H, m). Anal. (C₂₂- $H_{18}N_2O_4S)$ C, H, N.

5-[4-[2-[5-(Hydroxymethyl)-2-phenyl-4-oxazolyl]ethoxy]benzyl]-2,4-thiazolidinedione (41). NBS (2.75 g, 15 mmol) was added in small portions to a refluxing mixture of 18 (6.0 g, 15 mmol), α,α' -azobis(isobutyronitrile) (0.5 g, 3 mmol), and CCl₄ (150 mL). The mixture was refluxed for 10 min, washed with H₂O, dried (MgSO₄), and concentrated in vacuo to leave 5-[4-[2-[5-(bromomethyl)-2-phenyl-4-oxazolyl]ethoxy]benzyl]-2,4-thiazolidinedione as a crude oil (ca. 8.0 g): ¹H NMR (CDCl₈) δ 3.03 (2 H, t, J = 7 Hz), 2.9–3.2 (1 H, m), 3.48 (1 H, dd, J = 14, 5 Hz), 4.24 (2 H, t, J = 7 Hz), 4.45 (1 H, dd, J = 9, 5 Hz), 4.61 (2 H, s), 6.81 (2 H, d, J = 9 Hz), 7.10 (2 H, d, J = 9 Hz), 7.4 (3 H, m), 8.0(2 H, m), 8.70 (1 H, br s). The oil was dissolved in dioxane (100 mL)-2 N HCl (100 mL). The mixture was refluxed for 7 h, poured into H₂O, and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to leave an oil, which was chromatographed on SiO₂ (200 g) with Et₂O-hexane (1:1, v/v) to give the title compound (1.3 g, 21%): mp 98-99 °C (acetone-hexane); ¹H NMR (CDCl₃) δ 2.36 (1 H, br s), 3.04 (2 H, t, J = 6 Hz), 3.06 (1 H, dd, J = 14, 9 Hz), 3.39(1 H, dd, J = 14, 9 Hz), 4.6-4.9 (2 H, m), 6.80 (2 H, d, J = 9 Hz),7.10 (2 H, d, J = 9 Hz), 7.4 (3 H, m), 8.0 (2 H, m), 8.50 (1 H, br)s). Anal. $(C_{22}H_{20}N_2O_5S)$ C, H, N.

5-[4-[2-(5-Methyl-2-phenyl-4-oxazolyl)-2-oxoethoxy]benzyl]-2,4-thiazolidinedione (43). A mixture of 5-[4-[2hydroxy-2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzyl]-2,4thiazolidinedione (31) (0.5 g, 1.2 mmol), Ac₂O (1.0 mL, 10.6 mmol), and DMSO (10 mL) was stirred at room temperature for 1 h and allowed to stand overnight. The mixture was poured into H₂O and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to leave an oil, which was chromatographed on SiO₂ (40 g) with C₆H₆-acetone (9:1, v/v) to give the title compound (0.41 g, 81%): mp 168–169 °C (AcOEt-hexane); ¹H NMR (CDCl₃) δ 2.69 (3 H, s), 3.02 (1 H, dd, J = 14, 9 Hz), 3.35 (1 H, dd, J = 14, 4 Hz), 4.82 (1 H, dd, J= 9, 4 Hz), 5.38 (2 H, s), 6.88 (2 H, d, J = 9 Hz), 7.17 (2 H, d, J = 9 Hz), 7.5 (3 H, m), 8.0 (2 H, m). Anal. (C₂₂H₁₈N₂O₅S) C, H, N.

⁽¹⁴⁾ Simiti, I.; Chindris, E. Die Delépine-Reaktion bei 2-Aryl-4chloromethyloxazolen 15. Mitt. Heterozyklische Verbindungen. Arch. Pharmacol. 1971, 304, 425-429.

5-[4-[2-Hydroxy-2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzylidene]-2,4-thiazolidinedione (65). NaBH₄ (0.33 g, 8.7 mmol) was added portionwise to a stirred and ice-cooled solution of 5-[4-[2-(5-methyl-2-phenyl-4-oxazolyl)-2-oxoethoxy]benzylidene]-2,4-thiazolidinedione (63) (3.6 g, 8.9 mmol) in DMF-MeOH (1:1, v/v, 80 mL). After stirring at room temperature for 10 min, the reaction mixture was diluted with H₂O and acidified with AcOH to give the title compound (3.56 g, 98%): mp 252-253 °C (CHCl₃-MeOH); ¹H NMR (d_6 -DMSO) δ 2.43 (3 H, s), 4.27 (2 H, d, J = 7 Hz), 4.85-5.1 (1 H, m), 5.61 (1 H, br s), 7.10 (2 H, d, J = 9 Hz), 7.4-7.65 (5 H, m), 7.73 (1 H, s), 7.85 (2 H, m), 12.47 (1 H, br s). Anal. ($C_{22}H_{18}N_2O_5$ S) C, H, N.

General Procedure for 4-[2-(4-Azolyl)ethoxy]anilines (1). 4-[2-(5-Methyl-2-phenyl-4-oxazolyl)ethoxy]aniline. A stirred and ice-cooled solution of 2-(5-methyl-2-phenyl-4-oxazolyl)ethanol (57.0 g, 0.28 mol) and 4-fluoronitrobenzene (39.6 g, 0.28 mol) in DMF (500 mL) was treated with NaH (60% in oil, 14.0 g, 0.35 mol). The reaction mixture was stirred at room temperature for 1 h and poured into $\rm H_2O$ to give 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]nitrobenzene (74.7 g, 82%): mp 97-98 °C (MeOH); ¹H NMR (CDCl₃) δ 2.37 (3 H, s), 2.98 (2 H, t, J = 6 Hz), 4.32 (2 H, t, J = 6 Hz), 6.92 (2 H, d, J = 9 Hz), 7.4 (3 H, m), 7.9 (2 H, m), 8.18 (2 H, d, J = 9 Hz). Anal. ($\rm C_{18}H_{16}N_2O_4$) C, H, N.

A mixture of 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]-nitrobenzene (74.0 g, 0.23 mol), 10% Pd–C (2.5 g), and MeOH (500 mL) was hydrogenated at ambient temperature and atmospheric pressure. The insoluble catalyst was removed by filtration and the filtrate was concentrated in vacuo to give the title compound (65.5 g, 98%): mp 93–94 °C (AcOEt-hexane); ¹H NMR (CDCl₃) δ 2.33 (3 H, s), 2.90 (2 H, t, J = 6 Hz), 3.0 (2 H, br), 4.14 (2 H, t, J = 6 Hz), 6.56 (2 H, d, J = 9 Hz), 6.74 (2 H, d, J = 9 Hz), 7.4 (3 H, m), 7.9 (2 H, m). Anal. (C₁₈H₁₈N₂O₂) C, H, N.

General Procedure for 4-[2-Hydroxy-2-(4-oxazolyl)ethoxy]anilines (2). 4-[2-Hydroxy-2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]aniline. A solution of Br₂ (23.2 g, 0.15 mol) in CHCl₃ (25 mL) was added dropwise to a stirred solution of 4-acetyl-5-methyl-2-phenyloxazole¹⁵ (26.5 g, 0.13 mol) in CHCl₃ (300 mL) at 50 °C. The mixture was stirred at 55 °C for 30 min and poured into saturated aqueous NaHCO₃ (500 mL). The CHCl₃ layer was separated, washed with brine, dried (MgSO₄), and concentrated in vacuo to give 4-(bromoacetyl)-5-methyl-2-phenyloxazole (33.9 g, 92%): mp 88–89 °C (Et₂O-hexane); ¹H NMR (CDCl₃) δ 2.70 (3 H, s), 4.56 (2 H, s), 7.4–7.6 (3 H, m), 8.0 (2 H, m). Anal. (C₁₂H₁₀NO₂Br) C, H, N.

A mixture of 4-(bromoacetyl)-5-methyl-2-phenyloxazole (33.8 g, 0.12 mol), 4-acetamidophenol (17.3 g, 0.12 mol), $\rm K_2CO_3$ (27.6 g, 0.2 mol), and methyl ethyl ketone (400 mL) was stirred under reflux for 3 h and concentrated in vacuo. The residue was diluted with $\rm H_2O$ (300 mL) and $\rm Et_2O$ (300 mL)-hexane (100 mL) was added to the mixture. The whole was stirred at room temperature for 10 min, and the crystals were collected by filtration to give 4-[2-(5-methyl-2-phenyl-4-oxazolyl)-2-oxoethoxy]acetanilide (23.5 g, 58%): mp 175-176 °C (EtOH); ¹H NMR (d_6 -DMSO) δ 2.11 (3 H, s), 2.70 (3 H, s), 4.34 (2 H, s), 6.92 (2 H, d, J = 9 Hz), 7.30 (2 H, d, J = 9 Hz), 7.45 (3 H, m), 8.1 (2 H, m). Anal. ($\rm C_{20}H_{18}N_2O_4$) C, H, N.

A stirred and ice-cooled mixture of 4-[2-(5-methyl-2-phenyl-4-oxazolyl)-2-oxoethoxy]acetanilide (23.3 g, 67 mmol) and MeOH (250 mL) was treated with NaBH₄ (2.5 g, 67 mmol) for 30 min, and then AcOH (8 mL) was added. The mixture was poured into H₂O and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to give 4-[2-hydroxy-2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]acetanilide (23.0 g, 98%): mp 166–167 °C (AcOEt); ¹H NMR (d_6 -DMSO) δ 1.99 (3 H, s), 2.43 (3 H, s), 4.15 (2 H, d, J = 6 Hz), 4.91 (1 H, t, J = 6 Hz), 5.5 (1 H, br), 6.82 (2 H, d, J = 9 Hz), 7.5 (5 H, m), 7.9 (2 H, m), 9.7 (1 H, br s). Anal. ($C_{20}H_{20}N_2O_4$) C, H, N

A mixture of 4-[2-hydroxy-2-(5-methyl-2-phenyl-4-oxazolyl)-ethoxy]acetanilide (22.7 g, 64 mmol), 4 N KOH (300 mL), and EtOH (300 mL) was refluxed for 24 h and poured into H_2O . The

4-Acetyl-2,5-dimethyloxazole. Concentrated H_2SO_4 (8.8 mL, 0.32 mol) was added dropwise to a stirred mixture of 3-(acetyl-amino)-2,4-pentanedione (50 g, 0.32 mol) in Ac_2O (250 mL). The mixture was stirred at room temperature for 1 h and at 60 °C for 1 h, concentrated in vacuo, poured into H_2O , and extracted with AcOEt. The AcOEt extract was dried (MgSO₄) and concentrated in vacuo to give the title compound (28.8 g, 65%): mp 40–41 °C (hexane); 1H NMR (CDCl₃) δ 2.42 (3 H, s), 2.48 (3 H, s), 2.58 (3 H, s). Anal. ($C_7H_9NO_2$) C, H, N.

General Procedure for 4-[2-(2-Oxazolyl)ethoxy]anilines (3). 4-[2-(4-Methyl-5-phenyl-2-oxazolyl)ethoxy]aniline. A mixture of 3-(4-nitrophenoxy) propionic acid16 (13.1 g, 62 mmol), SOCl₂ (12.8 g, 108 mmol), DMF (0.3 g, 4 mmol), and toluene (100 mL) was stirred at 90 °C for 1 h and concentrated in vacuo. The residual oil was dissolved in AcOEt (30 mL). The solution was added dropwise to a stirred mixture of α -aminopropiophenone hydrochloride (11.5 g, 62 mmol), Na_2CO_3 (10.6 g, 100 mmol), and H₂O (100 mL) at room temperature. The organic layer was separated, washed with H₂O, dried (MgSO₄), and concentrated in vacuo to give α -[[3-(4-nitrophenoxy)propionyl]amino]propiophenone (16.0 g, 76%): mp 134-135 °C (AcOEt-hexane); ¹H NMR $(CDCl_3)$ δ 1.43 (3 H, d, J = 7 Hz), 2.74 (2 H, t, J = 6 Hz), 4.38 (2 H, t, J = 6 Hz), 5.58 (1 H, t, J = 7 Hz), 6.99 (2 H, d, J = 9 Hz),7.4-7.6 (5 H, m), 8.0 (2 H, m), 8.23 (2 H, d, J = 9 Hz). Anal. $(C_{18}H_{18}N_2O_5)$ C, H, N.

A mixture of α -[[3-(4-nitrophenoxy)propionyl]amino]propiophenone (15.5 g, 45 mmol), POCl₃ (20.8 g, 136 mmol), and toluene (150 mL) was stirred under reflux for 1 h. The mixture was poured into ice—H₂O, neutralized with aqueous NaHCO₃, and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to leave an oil, which was chromatographed on SiO₂ (150 g) with AcOEt-hexane (1:4, v/v) to give 4-[2-(4-methyl-5-phenyl-2-oxazolyl)ethoxy]nitrobenzene (13.0 g, 89%): mp 117-118 °C (AcOEt-hexane); ¹H NMR (CDCl₃) δ 2.37 (3 H, s), 3.29 (2 H, t, J = 6 Hz), 4.48 (2 H, t, J = 6 Hz), 6.97 (2 H, d, J = 9 Hz), 7.2-7.7 (5 H, m), 8.18 (2 H, d, J = 9 Hz). Anal. (C₁₈H₁₆N₂O₄) C, H, N.

A mixture of 4-[2-(4-methyl-5-phenyl-2-oxazolyl)ethoxy]-nitrobenzene (12.5 g, 39 mmol), Pd–C (5%, 3.0 g), and MeOH (150 mL) was hydrogenated at room temperature and atmospheric pressure. The catalyst was removed by filtration and the filtrate was concentrated in vacuo to give the title compound (11.0 g, 97%): mp 106–107 °C (EtOH); ¹H NMR (CDCl₃) δ 2.40 (3 H, s), 3.22 (2 H, t, J = 6 Hz), 3.30 (2 H, br s), 4.33 (2 H, t, J = 6 Hz), 6.62 (2 H, d, J = 9 Hz), 6.82 (2 H, d, J = 9 Hz), 7.2–7.7 (5 H, m). Anal. (C₁₈H₁₈N₂O₂) C, H, N.

General Procedure for 4-[2-(4-Azolyl)ethoxy]benzaldehydes (55, Y = CH₂, n = 1). 4-[2-(5-Methyl-2-phenyl-4-oxazolyl)ethoxy]benzaldehyde. A mixture of 2-(5-methyl-2-phenyl-4-oxazolyl)ethanol (57.0 g, 0.28 mol), 4-fluorobenzonitrile (40.8 g, 0.34 mol), and THF (600 mL) was treated with NaH (60% in oil, 13.5 g, 0.34 mol) with ice-cooling for 1 h and at room temperature for 12 h. The reaction mixture was poured into ice-H₂O and acidified with AcOH to give 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzonitrile (77.3 g, 91%): mp 119-120 °C (Et₂O-hexane); ¹H NMR (CDCl₃) δ 2.36 (3 H, s), 2.98 (2 H, t, J = 6.5 Hz), 4.29 (2 H, t, J = 6.5 Hz), 6.92 (2 H, d, J = 9 Hz), 7.3-7.65 (5 H, m), 7.85-8.05 (2 H, m). Anal. (C₁₉H₁₆N₂O₂) C, H,

A mixture of 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]-benzonitrile (76.5 g, 0.25 mol), Ra-Ni alloy (\sim 50%, 25.0 g), and aqueous HCOOH (70%, 600 mL) was stirred at 100 °C for 3 h.

precipitated crystals were collected by filtration and recrystallized from EtOH to give the title compound as colorless prisms (18.7 g, 96%): mp 166–167 °C; ¹H NMR (d_6 -DMSO) δ 2.42 (3 H, s), 4.05 (2 H, d, J = 6 Hz), 4.56 (2 H, br s), 4.87 (1 H, q, J = 6 Hz), 5.45 (1 H, d, J = 6 Hz), 6.83 (2 H, d, J = 9 Hz), 7.03 (2 H, d, J = 9 Hz), 7.5 (3 H, m), 7.9 (2 H, m). Anal. ($C_{18}H_{18}N_2O_3$) C, H, N.

⁽¹⁵⁾ Allan, A. W.; Walter, B. H. The Synthesis of Some Substituted 4-Acetyloxazoles and the Corresponding Acids. J. Chem. Soc. C 1968, 1397-1399.

⁽¹⁶⁾ Motiu-DeGrood, R.; Hunt, W.; Wilde, J.; Hupe, D. J. Rates and Equilibria for the Inactivation of Muscle Aldolase by an Active Site Directed Michael Reaction. J. Am. Chem. Soc. 1979, 101, 2182–2190.

After cooling, the insoluble solid was removed by filtration and the filtrate was concentrated in vacuo. The residue was diluted with H_2O , made alkaline with aqueous KOH, and extracted with AcOEt. The AcOEt extract was washed with brine, dried (Mg-SO₄), and concentrated in vacuo to give the title compound (50.6 g, 65%): mp 82-84 °C (CH₂Cl₂-hexane); ¹H NMR (CDCl₃) δ 2.37 (3 H, s), 2.99 (2 H, t, J = 6.5 Hz), 4.33 (2 H, t, J = 6.5 Hz), 6.97 (2 H, d, J = 9 Hz), 7.3-7.55 (3 H, m), 7.79 (2 H, d, J = 9 Hz), 7.85-8.1 (2 H, m), 9.85 (1 H, s). Anal. (C₁₈H₁₇NO₃) C, H, N.

4-(2-Phenyl-4-thiazolylmethoxy)benzaldehyde. A mixture of 4-(chloromethyl)-2-phenylthiazole (2.1 g, 10 mmol), 4-hydroxybenzaldehyde (1.22 g, 10 mmol), $K_2\text{CO}_3$ (2.76 g, 20 mmol), and DMF (30 mL) was stirred at 120 °C for 1 h. The reaction mixture was poured into $H_2\text{O}$ to give the title compound (2.8 g, 95%): mp 88–90 °C (Et₂O-hexane); ¹H NMR (CDCl₃) δ 5.32 (2 H, s), 7.10 (2 H, d, J = 9 Hz), 7.28 (1 H, s), 7.82 (2 H, d, J = 9 Hz), 7.0–8.1 (5 H, m), 9.87 (1 H, s). Anal. ($C_{17}H_{13}\text{NO}_2\text{S}$) C, H, N

General Procedure for 2-(4-Azolyl)ethanols. 2-(5-Methyl-2-phenyl-4-oxazolyl)ethanol. Et_3N (61.7 g, 0.61 mol) was added dropwise to a stirred and ice—salt cooled mixture of L-aspartic acid β -methyl ester hydrochloride (31.2 g, 0.17 mol), benzoyl chloride (23.9 g, 0.17 mol), and CH_2Cl_2 (700 mL). The mixture was stirred at the same temperature for further 1 h and poured into 1 N HCl (1 L). The organic layer was separated and the aqueous layer was extracted with CH_2Cl_2 . The combined organic layer was washed with brine, dried (MgSO₄), and concentrated in vacuo to give N-benzoyl-1-aspartic acid β -methyl ester (34.8 g, 82%): mp 153–154 °C (EtOH).

A mixture of N-benzoyl-L-aspartic acid β -methyl ester (25.1 g, 0.1 mol), Ac₂O (95 mL, 1.0 mol), Et₃N (95 mL, 0.68 mol), and DMAP (1.2 g, 10 mmol) was stirred at room temperature for 1 h and at 90 °C for 1 h and poured into H₂O (1 L). The mixture was extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to give methyl 3-(benzoylamino)-4-oxovalerate as an oil (15.6 g, 63%): ¹H NMR (CDCl₃) δ 2.27 (3 H, s), 2.85 (1 H, dd, J = 20, 5 Hz), 3.08 (1 H, dd, J = 20, 5 Hz), 3.67 (3 H, s), 4.93 (1 H, dt, J = 8, 5 Hz), 7.5 (3 H, m), 7.8 (2 H, m).

A mixture of methyl 3-(benzoylamino)-4-oxovalerate (15.0 g, 60 mmol), POCl₃ (27.7 g, 180 mmol), and toluene (250 mL) was stirred under reflux for 2 h and concentrated in vacuo. The residue was diluted with $\rm H_2O$, neutralized with $\rm K_2CO_3$, and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to give methyl 2-(5-methyl-2-phenyl-4-oxazolyl)acetate as a crude oil (12.6 g, 91%): ¹H NMR (CDCl₃) δ 2.31 (3 H, s), 3.49 (2 H, s), 3.68 (3 H, s), 7.3 (3 H, m), 7.9 (2 H, m).

A solution of methyl 2-(5-methyl-2-phenyl-4-oxazolyl)acetate (crude oil, 12.2 g, 53 mmol) in Et₂O (50 mL) was added to a stirred and ice-cooled suspension of LiAlH₄ (2.0 g, 53 mmol) in Et₂O (200 mL). The mixture was stirred at ambient temperature for 30 min, and H₂O (11 mL) was added dropwise with ice-cooling. The insoluble solid was removed by filtration and the filtrate was concentrated in vacuo to give the title compound (9.8 g, 91%): mp 73-74 °C (AcOEt-hexane); ¹H NMR (CDCl₃) δ 2.30 (3 H, s), 2.70 (2 H, t, J = 6 Hz), 3.20 (1 H, br s), 3.90 (2 H, t, J = 6 Hz), 7.4 (3 H, m), 7.9 (2 H, m). Anal. (C₁₂H₁₃NO₂) C, H, N.

Ethyl 2-(2-Phenyl-4-oxazolyl)acetate. A mixture of ethyl 4-chloroacetoacetate (49.4 g, 0.3 mol) and benzamide (60.6 g, 0.5 mol) was heated at 120 °C for 2 h and diluted with saturated aqueous NaHCO₃. The mixture was extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to leave an oil, which was chromatographed on SiO₂ (700 g) with Et₂O-hexane (1:9, v/v) to give the title compound as an oil (26.4 g, 38%): ¹H NMR (CDCl₃) δ 1.27 (3 H, t, J = 7 Hz), 3.68 (3 H, s), 4.15 (2 H, q, J = 7 Hz), 7.4 (3 H, m), 7.67 (1 H, s), 8.0 (2 H, m).

Ethyl 2-(2-Cyclohexyl-4-thiazolyl)acetate. A mixture of ethyl 4-chloroacetoacetate (5.74 g, 35 mmol), cyclohexanethiocarboxamide (5.0 g, 35 mmol), and EtOH (50 mL) was refluxed for 1 h, poured into H_2O , and extracted with AcOEt. The AcOEt extract was washed with brine, dried (MgSO₄), and concentrated in vacuo to leave an oil, which was chromatographed on SiO₂ (50 g) with AcOEt-hexane (1:4, v/v) to give the title compound as an oil (6.3 g, 71%): ¹H NMR (CDCl₃) δ 1.28 (3 H, t, J = 7 Hz),

1.2-2.3 (10, H, m), 2.97 (1 H, m), 3.77 (2 H, s), 4.17 (2 H, q, J = 7 Hz), 7.0 (1 H, s).

Acknowledgment. The authors are grateful to Drs. H. Iwatsuka and Y. Kawamatsu for encouragement throughout this work.

Registry No. 1 ($R^1 = H$, $R^2 = Me$, x = S, n = 1), 141399-30-6; 1 (R^1 = Me, R^2 = H, x = S, n = 1), 141399-31-7; 1 (R^1 = Et, R^2 = H, x = S, n = 1), 141399-32-8; 1 (R¹ = i-Pr, R² = H, x = S, n= 1), 141399-33-9; 1 (R^1 = cyclohexyl, R^2 = H, x = S, n = 1), 141399-34-0; 1 (R¹ = Ph, R² = H, x = S, n = 1), 141399-35-1; 1 $(R^1 = Me, R^2 = H, x = O, n = 1), 141399-36-2; 1 (R^1 = Pr, R^3 = R^3)$ H, x = 0, n = 1), 141399-37-3; 1 (R¹ = Me, R² = Me, x = 0, n= 1), 141399-38-4; 1 (R¹ = Me, R² = Et, x = O, n = 1), 141399-39-5; 1 (R^1 = cyclohexyl, R^2 = H, x = O, n = 1), 141399-40-8; 1 (R^1 = Ph, $R^2 = H$, x = O, n = 1), 141399-41-9; 1 ($R^1 = Ph$, $R^2 = Me$, x = 0, n = 1), 103788-87-0; 1 (R¹ = Ph, R² = Et, x = 0, n = 1), 141399-42-0; 1 ($\mathbb{R}^1 = \mathbb{P}h$, $\mathbb{R}^2 = \mathbb{M}e$, x = 0, n = 2), 141399-43-1; 1 (R^1 = cyclohexyl, R^2 = Me, x = O, n = 1), 141399-44-2; 1 (R^1 = 2-furaryl, R^2 = Me, x = O, n = 1), 141399-45-3; 1 (R^1 = 2-thienyl, $R^2 = Me$, x = O, n = 1), 141399-46-4; 1 ($R^1 = 4$ -MeOC₆H₄, $R^2 =$ Me, x = 0, n = 1), 141399-47-5; $1 (R^1 = 3,4-(MeO)_2C_6H_4, R^2 =$ Me, x = 0, n = 1), 141399-48-6; 1 (R¹ = p-tolyl, R² = Me, x = 0O, n = 1), 141399-49-7; 1 (R¹ = m-MeSC₆ \hat{H}_4 , R² = Me, x = O, n= 1), 103789-38-4; $1 (R^1 = o-ClC36H_4, R^2 = Me, x = O, n = 1), 103789-15-7$; $1 (R^1 = p-OHC_6H_4, R^2 = Me, x = O, n = 1)$ 103789-19-1; 2 ($\mathbb{R}^1 = Me$, $\mathbb{R}^2 = Me$, $\mathbb{x} = O$), 103793-55-1; 2 (\mathbb{R}^1 = Ph, R^2 = Me, x = O), 103789-07-7; 2 (R^1 = cyclohexyl, R^2 = Me, x = 0), 103789-33-9; 3 (R¹ = i-Bu, R² = Me, x = 0), 107325-07-5; 3 (R¹ = cyclohexyl, R² = Me, x = O), 141399-50-0; $3 (R^1 = Ph, R^2 = Me, x = O), 141399-51-1; 3 (R^1 = Ph, R^2 = Et,$ x = O), 107325-06-4; 3 (R¹ = Me, R² = Ph, x = O), 107325-05-3; 4 (R¹ = Ph, R² = Me, x = O, n = 1), 103814-35-3; 5 (R¹ = Ph, R² = Me, x = 0, n = 1), 103788-55-2; 6, 141399-27-1; 7, 103787-98-0; 8, 103788-03-0; 9, 103788-04-1; 10, 103788-18-7; 11, 103788-00-7; 12, 103787-99-1; 13, 103788-02-9; 14, 103788-01-8; 15, 103788-08-5; 16, 103788-20-1; 17, 103926-56-3; 18, 103787-97-9; 19, 103788-06-3; **20**, 103788-31-4; **21**, 103788-22-3; **22**, 103788-09-6; **23**, 103788-10-9; 24, 103788-07-4; 25, 103788-13-2; 26, 141399-28-2; 27, 103788-28-9; 28, 103788-14-3; 29, 103788-15-4; 30, 103788-11-0; 31, 103788-05-2; **32**, 103788-30-3; **33**, 103787-88-8; **34**, 103787-86-6; **35**, 103787-91-3; **36**, 103787-89-9; **37**, 103787-87-7; **38**, 103787-85-5; **39**, 103787-90-2; 40, 141399-29-3; 41, 103788-24-5; 41 bromo derivative, 103788-58-5; **42**, 103788-33-6; **43**, 103788-34-7; **44**, 107324-84-5; **45**, 107324-86-7; 46, 107325-11-1; 47, 107324-88-9; 48, 107324-83-4; 49, 107342-77-8; **50**, 107324-90-3; **51**, 107324-89-0; **52** ($R^1 = Me$, $R^2 = H$, x = S), 39238-07-8; $52 (R^1 = Ph, R^2 = H, x = S)$, 4771-31-7; $52 (R^1 = 3-py, R^2 = H, x = S)$, 141399-52-2; $52 (R^1 = Me, R^2 = H, x = O)$, 141399-53-3; 52 (R¹ = Pr R² = H, x = O), 103788-63-2; 52 (R¹ = Ph, $R^2 = H$, x = O), 30494-97-4; 52 ($R^1 = Ph$, $R^2 = Me$, x = O), 103788-61-0; 52 (R¹ = 2-furyl, R² = Me, x = O), 141399-54-4; 52 $(R^1 = H, R^2 = Ph, x = S), 65385-00-4; 52 (R^1, R^2, x = 2-benzo$ thiazolyl), 37859-43-1; **52** (R¹, R², x = 2-benzoxazolyl), 41014-43-1; 53 (R¹ = Me, R² = H, x = S, 4 = CH₂, n = 1), 103789-55-5; 53 $(R^1 = Ph, R^2 = H, x = S, y = CH_2, n = 1), 103789-56-6; 53 (R^1$ = Ph, R^2 = Me, x = O, y = CH₂, n = 1), 103788-59-6; 53 (R^1 = Ph, R^2 = Et, x = O, y = CH₂, n = 1), 103789-57-7; 53 (R^1 = $4-\text{ClC}_6\text{H}_4$, R^2 = Me, x = O, y = CH₂, n = 1), 103789-59-9; 53 (R¹ = 3-MeSC_6H_4 , R^2 = Me, x = O, y = CH_2 , n = 1), 103789-60-2; 53 (R¹ = 1-Me-cyclohexyl, R² = Me, x = 0, $y = CH_2$, n = 1), 103789-63-5; **53** ($R^1 = 2$ -thienyl, $R^2 = Me$, x = O, $y = CH_2$, n = 01), 103789-62-4; 53 (R¹ = Ph, R² = Me, x = O, y = CH₂, n = O), 103789-66-8; 53 ($R^1 = Ph, R^2 = Me, x = O, y = CO, n = 1$), 103789-69-1; 53 ($R^1 = Me$, $R^2 = Me$, x = O, y = CO, n = 1), 103789-70-4; 54, 103788-36-9; 55, 103788-38-1; 56, 103788-35-8; **57**, 103788-40-5; **58**, 103788-42-7; **59**, 103788-43-8; **60**, 103788-48-3; **61**, 103788-45-0; **62**, 103788-50-7; **63**, 103788-51-8; **64**, 103788-46-1; 65, 103788-47-2; 66, 103788-52-9; 5-(p-hydroxybenzyl)-2,4-thiazolidinedione, 74772-78-4; 2-(5-methyl-2-phenyl-4-oxazolyl)ethanol, 103788-65-4; 1-fluoro-4-nitrobenzene, 350-46-9; 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]nitrobenzene, 103788-72-3; 4-acetyl-5-methyl-2-phenyloxazole, 2940-19-4; 4-(bromoacetyl)-5methyl-2-phenyloxazole, 103788-62-1; 4-acetamidophenol, 103-90-2; 4-[2-(5-methyl-2-phenyl-4-oxazolyl)-2-oxoethoxy]acetanilide, 103789-05-5; 4-[2-hydroxy-2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]acetanilide, 103789-06-6; 3-(acetylamino)-2,4-pentanedione, 5440-23-3; 4-acetyl-2,5-dimethoxazole, 23000-12-6; 3-(4-nitrophenoxy)propionic acid, 10572-16-4; α -aminopropiophenone hydrochloride, 16735-19-6; α -[[3-(4-nitrophenoxy)propionyl]-amino]propiophenone, 141399-55-5; 4-[2-(4-methyl-5-phenyl-2-oxazolyl)ethoxy]nitrobenzene, 107325-00-8; 4-fluorobenzonitrile, 1194-02-1; 4-[2-(5-methyl-2-phenyl-4-oxazolyl)ethoxy]benzonitrile, 103789-44-2; 4-hydroxybenzaldehyde, 123-08-0; 4-(2-phenyl-4-

thiazolylmethoxy)benzaldehyde, 103789-67-9; β -methyl L-aspartate hydrochloride, 16856-13-6; β -methyl N-benzoyl-L-aspartate, 39741-26-9; methyl 3-(benzoxylamino)-4-oxovalerate, 54819-26-0; methyl 2-(5-methyl-2-phenyl-4-oxazolyl)acetate, 103788-64-3; ethyl 4-chloroacetoacetate, 638-07-3; benzamide, 55-21-0; ethyl 2-(2-phenyl-4-oxazolyl)acetate, 84446-03-7; ethyl 2-(2-cyclohexyl-4-thiazolyl)acetate, 24087-96-5; cyclohexanethiocarboxamide, 7390-42-3.

N-[4-[[(3,4-Dihydro-4-oxo-1,2,3-benzotriazin-6-yl)methyl]amino]benzoyl]-L-glutamic Acid, a Novel A-Ring Analogue of 2-Desamino-5,8-dideazafolic Acid¹

Andre Rosowsky,*,† Ronald A. Forsch,† and Richard G. Moran[‡]

Dana-Farber Cancer Institute and Department of Biological Chemistry and Molecular Pharmacology, Harvard Medical School, Boston, Massachusetts 02115, and USC Comprehensive Cancer Center, Los Angeles, California 90033.

Received October 9, 1991

N-[4-[[(3,4-Dihydro-4-oxo-1,2,3-benzotriazin-6-yl)methyl]amino]benzoyl]-L-glutamic acid ("2-aza-2-desamino-5,8-dideazafolic acid", ADDF) was synthesized from 2-amino-5-methylbenzamide via a four-step sequence consisting of diazotization, benzylic bromination, condensation with dimethyl N-(4-aminobenzoyl)-L-glutamate, and ester hydrolysis. ADDF was an inhibitor of recombinant mouse thymidylate synthase; inhibition was competitive with 5,10-methylenetetrahydrofolate as variable substrate ($K_{\rm i}=2.3~\mu{\rm M}$). It was a substrate for murine follypolyglutamate synthetase with kinetic characteristics ($K_{\rm m}=28~\mu{\rm M}$) comparable to those of aminopterin, and it inhibited the growth of L1210 cells in culture (IC50 = 0.52 $\mu{\rm M}$). The structural modification of the A-ring embodied in ADDF appears to offer a novel, heretofore unexplored approach to the design of TS inhibitors.

The potential therapeutic significance of folic acid analogues targeted against thymidylate synthase (TS) as opposed to dihydrofolate reductase (DHFR) was predicted more than 20 years ago by Borsa and Whitmore.² Shortly thereafter, Bird and co-workers3 reported the potent biological activity of 5,8-dideazafolic acid (1).4 This led to an extensive program of synthesis of quinazoline analogues substituted at the N^{10} position,⁵⁻⁷ on the phenyl ring,⁸ and in ring B,^{9,10} and to selection of N^{10} -propargyl-5,8-dideazafolic acid (CB3717) as a suitable candidate for biochemical¹¹⁻¹³ and clinical^{14,15} evaluation. While CB3717 had many desirable pharmacological properties, such as the ability to enter cells by a transport mechanism distinct from that of reduced folates and the classical DHFR inhibitor methotrexate (MTX), its clinical usefulness was hampered by low solubility at physiological pH, which gave rise to hepatic and renal toxicity.16

COOH

$$HN$$
 R
 CH_2NH
 CH_2NH
 $COOH$
 $COOH$

The toxicity encountered during clinical trials with CB3717 prompted a vigorous search for more soluble congeners, culminating in the discovery of a family of second-generation TS inhibitors that included 2-desamino-5,8-dideazafolic acid (2), 2-desamino-2-methyl-5,8-dideazafolic acid (3), and the corresponding N^{10} -methyl and N^{10} -propargyl analogues. Piochemical studies n^{20-22} revealed that, in general, replacement of the NH₂ group

at C² by H or Me resulted in weaker binding to purified TS but increased inhibition of the growth of cultured cells.

- (1) Paper 47 in this series. For previous paper, see: Rosowsky, A.; Forsch, R. A.; Reich, V. E.; Freisheim, J. H.; Moran, R. G. Side chain modified 5-deazafolate and 5-deazatetrahydrofolate analogues as mammalian folylpolyglutamate synthetase and glycinamide ribonucleotide formyltransferase inhibitors: synthesis and in vitro biological evaluation. J. Med. Chem. 1992, 35, 1578-1588.
- (2) Borsa, J.; Whitmore, G. F. Cell killing studies on the mode of action of methotrexate on L-cells in vitro. Cancer Res. 1969, 29, 737-744.
- (3) Bird, S. D.; Vaitkus, J. W.; Clarke, J. 2-Amino-4-hydroxy-quinazolines as inhibitors of thymidylate synthetase. Mol. Pharmacol. 1970, 6, 573-575.
- (4) Davoll, J.; Johnson, A. M. Quinazoline analogues of folic acid. J. Chem. Soc. 1970, 997-1002.
- (5) Jones, T. R.; Calvert, A. H.; Jackman, A. L.; Brown, S. J.; Jones, M.; Harrap, K. R. A potent antitumor quinazoline inhibitor of thymidylate synthetase: synthesis, biological properties and therapeutic results in mice. Eur. J. Cancer 1981, 17, 11-19.
- (6) Jones, T. R.; Calvert, A. H.; Jackman, A. L.; Eakin, M. A.; Smithers, M. J.; Betteridge, R. F.; Newell, D. R.; Hayter, A. J.; Stocker, A.; Harland, S. J.; Davies, L. C.; Harrap, K. R. Quinazoline antifolates inhibiting thymidylate synthase: Variation of the N¹⁰ substituent. J. Med. Chem. 1985, 28, 1468-1476.
- (7) Nair, M. G.; Nanavathi, N. T.; Nair, I. G.; Kisliuk, R. L.; Gaumont, Y.; Hsiao, M. C.; Kalman, T. I. Folate analogues. 26. Synthesis and antifolate activity of 10-substituted derivatives of 5,8-dideazafolic acid and of the poly- γ -glutamyl metabolites of N^{10} -propargyl-5,8-dideazafolic acid (PDDF). J. Med. Chem. 1986, 29, 1754–1760.
- (8) Jones, T. R.; Smithers, M. J.; Taylor, M. A.; Jackman, A. L.; Calvert, A. H.; Harland, S. J.; Harrap, K. R. Quinazoline antifolates inhibiting thymidylate synthase: Benzoyl ring modifications. J. Med. Chem. 1986, 29, 468-472.
- (9) Hynes, J. B.; Kumar, A.; Tomazic, A.; Washtien, W. L. Synthesis of 5-chloro-5,8-dideaza analogues of folic acid and aminopterin targeted for colon adenocarcinoma. J. Med. Chem. 1987, 30, 1515-1519; also see earlier papers from this group.
- (10) Hynes, J. B.; Patil, S. A.; Tomazic, A.; Kumar, A.; Pathak, A.; Tan, X.; Xianqiang, L.; Ratnam, M.; Delcamp, T. J.; Freisheim, J. H. Inhibition of murine thymidylate synthase and human dihydrofolate reductase by 5,8-dideaza analogues of folic acid and aminopterin. J. Med. Chem. 1988, 31, 449-454.

[†] Dana-Farber Cancer Institute.

USC Comprehensive Cancer Center.