

Gymnothelignans A-O: Conformation and Absolute Configuration Analyses of Lignans Bearing Tetrahydrofuran from Gymnotheca chinensis

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Supporting Information

ABSTRACT: Fifteen new lignans, gymnothelignans A-O (1-15), bearing tetrahydrofuran with variable conformations belonging to three potentially related skeletons were isolated from Gymnotheca chinensis Decne. The structures were elucidated by means of detailed spectroscopic analysis. Absolute configurations were assigned using X-ray singlecrystal diffraction and chemical transformations. Moreover, by the homology, compounds 1-11 and eupomatilones were

confirmed to have uniform R-configuration at C-5. However, a synthesized congener has long been mistaken as 5-epimer of eupomatilone-6. This work provides guidance for the absolute configuration establishment of the subeupomatilone family with trans-H-4-H-5 configuration.

■ INTRODUCTION

Gymnotheca chinensis Decne, as one of the endemic genera of seed plants in China, is a perennial herb of Saururaceae. The whole plants of G. chinensis have long been used as traditional herbal medicine to treat contusions and strains. Little phytochemical information about this genus is available except for G. involucrata Pei. Our recent investigation on the constituents of G. chinensis led to the isolation of 15 new lignans (1-15), together with five known compounds kaempferol-4',7-dimethyl-3-O-glucoside (16),1 kaempferol-7methyl-3-O-glucoside (17),2,3 blumenol A (18),4 1-bisabolon (19),⁵ and β -sitosterol (20).

The 15 new lignans, gymnothelignans A-O (1-15), bearing tetrahydrofuran (THF) with variable conformations belonged to three unusual lignan skeletons, namely, dibenzocyclooctene, eupomatilone, and eupodienone. The latter two rare types were only previously isolated from the Australian shrub Eupomatia bennettii^{6,7} and E. laurina,^{8,9} respectively. Owing to their attractive structures, there have been total syntheses of all of the members of eupomatilones $^{10-20}$ except eupomatilone-1. The absolute configuration of eupomatilone-6 has only just been recently proposed. 16 Moreover, it was reported that eupodienones could be rearranged to form dibenzocyclooctene derivatives.^{9,21} Herein, we report the isolation, structural elucidation, and relative and absolute configuration of these compounds. We also explored the relationship between eupomatilones and eupodienones.

■ RESULTS AND DISCUSSION

Dried and powdered whole plants of G. chinensis were extracted with ethanol at room temperature to give an extract, which was suspended in H2O and extracted with petroleum ether and ethyl acetate successively. The ethyl acetate extracts were subjected to silica gel column chromatography followed by reversed-phase HPLC to yield 15 new lignans and five known compounds. Structures of these compounds were elucidated by a combination of detailed spectroscopic analysis. The absolute configurations were determined by X-ray single-crystal diffraction and chemical conversions.

Gymnothelignan A (1) was obtained as a colorless crystal. Its molecular formula, C₃₀H₃₇NO₉, was established by HRESIMS, which requires 13 degrees of unsaturation. The IR (KBr) spectrum showed absorption bands due to hydroxyl (3437 cm⁻¹) and aromatic groups (1619, 1474 cm⁻¹). The ¹H and ¹³C NMR (Table 1) spectra displayed signals of three methoxy groups, two secondary methyl groups, one methylenedioxy group, five methylene groups, five methine groups, four sp² carbons, and 10 non-hydrogenated carbons (assigned by DEPT, Table 1).

The analyses of ¹H-¹H COSY and HMBC spectra (Figure 1) suggested that the gross structure has three subunits (units A-C), as shown in Figure 2. The connection of C-1' to C-1''was deduced from the HMBC correlations of H-2"/6" to C-1' (it possesses an unusual doubly attached ring system which

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exhibits hindered rotation about the biaryl bond⁶). The correlations of 3"-OCH3 to C-3" and 5"-OCH3 to C-5" positioned the two methoxy groups to C-3" and C-5", respectively. In addition, a hydroxyl group was connected to C-4" by analysis of the chemical shift of C-4", which was shifted $(\Delta \delta = -12.4 \text{ and } -12.3, \text{ respectively})$ in comparison with those of C-3" and C-5". The presence of a methylenedioxy group was assigned by the HMBC correlations of OCH₂O to C-4' and C-5', as well as H-3' to C-4' and C-5'. Similarly, a methoxy group at C-6' was also assigned. Thus, the gross structure of unit A was proposed as shown in Figure 1. In unit B, the ¹H-¹H COSY cross-peaks revealed the sequential connections of C-2-C-7. This was also supported by the HMBC correlations of H-2 to C-7 and H-5 to C-6. The downfield chemical shifts revealed that C-2 and C-5 were both oxygenated. HMBC correlations of H-2/C-5 and H-5/C-2 displayed the connection of C-2 to C-5 via an ether bond. The 13 C NMR chemical shift of C-2 ($\delta_{\rm C}$ 108.6) was indicative of a ketal or hemiketal. In unit C, the sequential connections of C-6"'-C-10" were confirmed by the $^{1}\text{H}-^{1}\text{H}$ COSY spectrum. HMBC correlations of H-4" at δ_{H} 2.80 to C-2"'/6"'/10"' as well as those of H-2"' at $\delta_{\rm H}$ 7.12 to C-4"'/C-5" are in association with the relatively downfield 13C NMR chemical shifts²² of C-2" and C-4", and HRMS confirmed that an oxazoline ring fused with a cyclohexanol at C-5‴.

The connection of C-2′ (unit A) to C-5 (unit B) was established by the analyses of HMBC correlations of H-5 to C-1′ and C-3′, and H-3′ to C-5. The connection between units B

and C was achieved by HMBC correlation of H-2 to C-8", giving rise to the connection of C-2 to C-8" through an ether bond. NOESY cross-peaks of H-2 to H-8" also supported the connection of units B and C (Figure 2).

The stereochemistry of different substituent groups on the THF (unit B) was designated using a NOESY experiment. The cross-peaks of H-2/H₃-7, H-5/H₃-6, and H-4/3' indicated that H-2, H-5, H₃-6, and H₂-7 were cofacial. Likewise, H-3, H-4, and units A and C were on the opposite face. The assignments were also buttressed by vicinal coupling constant values of H-2-H-3 (2.1 Hz) and H-4-H-5 (7.7 Hz) when the dihedral angle of H-2-H-3 took the value close to 90° and that of H-4-H-5 was close to 180°. In the molecule, the bulky biphenyl group unit A was in the equatorial position, whereas unit C was in the axial position (Figure 3). On the basis of single-crystal Cu K α X-ray diffraction (Figure S84, Supporting Information), C-7 was revealed to have α -orientation of the THF moiety. The absolute stereochemistry of 1 was also determined (Flack parameter, 0.2(3), calculated using 1217 Friedel pairs).²³ Gymnothelignan A (1) was then assigned as (2S,3S,4R,5R)-3,4-dimethyl-2-hydroxyl-5-(4"-hydroxyl-4',5'-methylenedioxy-6',3",5"-trimethoxylbiphenyl-2'-yl)tetrahydrofuranyl-8"'-oxo-1'''-oxa-3'''-azaspiro[4.5]dec-2'''-enyl-2,8'''-ether.

Gymnothelignan B (2) had the same molecular formula of C₃₀H₃₇NO₉ of 1 by HRESIMS. The ¹H and ¹³C NMR (Table 1) spectra indicated that it was a diastereoisomer of 1. By comparing the ¹H NMR spectra with those of 1, the signals of H-6"'-H-10" were shifted ($\Delta\delta$ = -0.10 to -0.20), which indicated that unit C at C-2 (unit B) of 2 should have α orientation because the bulk effects from unit A could be substantially relieved (the alternative explanation that epimerization occurred at C-5 was less likely due to the chemical shift of C-5 and bulk biphenyl group⁶). Moreover, the ¹³C NMR chemical shift of C-2 was shifted ($\Delta \delta = -2.6$), and the chemical shift of C-5 was shifted ($\Delta\delta$ = +0.6) instead (it was probably a result of epimerization occurring at C-2). Similarly, the signals of C-3 ($\delta_{\rm C}$ 40.8) and C-7 ($\delta_{\rm C}$ 9.4) were shifted ($\Delta \delta$ = -3.4 and -1.8, respectively). This indicated that epimerization(s) occurred in the molecule.²⁴ Correlation observed in the NOESY (Figure 2) spectrum of H-2/H₃-7 suggested that H-2 and H_3 -7 were cofacial. The similar correlation of H-4/H-3', which was also observed in the molecule of 1, was indicative of H-4 being in the β -orientation. Similarly, H-5 and H₃-6 were on the opposite face. Obviously, epimerizations occurred at C-2 and C-3 in comparison to 1. Since the vicinal coupling constants of H-2-H-3 and H-4-H-5 took moderate value (4-5 Hz), the conformation of the THF moiety should be a result of interconversion of two conformational isomers (conformers 1 and 2, Figure 3). In summary, the structure of 2 was determined to be the 2,3-bisepimer of 1.

Gymnothelignans C (3) and H (8) only differed from unit C by a methyl group at C-2 based on the HRESIMS and NMR data analyses (Table 1 for 3, Tables 2 and 3 for 8). The structure of unit C of 3 was established by comparing its ¹³C NMR with that of 2. The main seco-lignan part was assigned as follows (take 8 for example). Correlations of H-2/H-3, H-3/H-3', and H-5/H-4 were observed in the NOESY spectrum (Figure S36, Supporting Information). This showed that H-2, H-3, H₃-6, and biphenyl were cofacial, whereas H-4, H-5, H₃-7, and 2-OCH₃ were on the opposite face. The method aforementioned was applied for the determination of THF (Figure 3). Except for 2-OCH₃, all of the substituents were in the equatorial position. The configuration of 8 was finally

Table 1. NMR Data for Compounds 1-3 (CD₃COCD₃)^a

		1		2		3
no.	$\delta_{\rm C}$, mult	$\delta_{ m H}$	$\delta_{ m C}$	$\delta_{ m H}$	$\delta_{ extsf{C}}$	$\delta_{ m H}$
2	108.6, CH	4.86 d (2.1)	106.0	5.21 d (4.7)	104.1	5.17 d (3.9)
3	44.2, CH	2.15 m ^b	40.8	2.38 m ^b	42.7	1.82 m*,b
4	44.1, CH	2.35 m ^b	44.1	2.01 m ^b	47.2	1.79 m*, ^b
5	84.3, CH	4.51 d (7.7)	84.9	4.56 d (4.3)	79.4	5.07 d (8.1)
6	12.2, CH ₃	0.71 d (7.0)	15.5	0.74 d (7.2)	16.5	0.69 d (6.5)
7	11.2, CH ₃	0.79 d (7.3)	9.4	0.90 d (7.2)	12.8	0.94 d (6.4)
2'	137.1, C		137.7		134.4	
3'	103.0, CH	7.03 s	101.6	6.69 s	103.1	6.63 s
4'	149.3, C		149.4		149.0	
5'	136.6, C		137.0		136.9	
6'	141.5, C		141.8		141.8	
1'	129.7, C		128.8		128.8	
1"	127.4, C		127.6		127.8	
2"	110.0, CH	6.42 d (1.7)	110.0	6.44 br s	109.4	6.40 br s
3"	148.3, C		148.5		148.5	
4"	135.9, C		135.9		135.9	
5"	148.2, C		148.3		148.3	
6"	108.4, CH	6.45 d (1.7)	108.3	6.45 br s	108.1	6.43 br s
OCH ₂ O	102.1, CH ₂	6.03, 6.01, each 1H, d (1.0)	102.0	6.01, 6.00, each 1H, br s	102.0	6.01, 2H, br s
6'-OCH ₃	60.1, CH ₃	3.74 s	60.1	3.74 s	60.1	3.75 s
3"-OCH ₃	56.8, CH ₃	3.82 s	56.8	3.82 s	56.7	3.81 s
5"-OCH ₃	56.7, CH ₃	3.81 s	56.7	3.82 s	56.7	3.81 s
2‴	146.5, CH	7.12 t (1.7)	146.5	7.09 br s	146.5	7.09 br s
4‴	46.2, CH ₂	2.80 d (1.7)	46.5	2.69 d (1.4)	46.4	2.69 dd (1.7)
5‴	84.1, C		83.8		83.9	
6‴	32.9, CH ₂	1.62-2.02 m*	32.6	1.52-1.82 m*	32.7	1.51-1.82 m³
7‴	28.3, CH ₂	1.62-2.02 m*	28.3	1.52-1.82 m*	28.6	1.51-1.82 m ³
8‴	73.0, CH	3.81 m*,b	72.6	3.70 m ^b	72.5	3.71 m ^b
9‴	30.3, CH ₂	1.62-2.02 m*	30.3	1.52-1.82 m*	30.3	1.51-1.82 m ²
10‴	33.0, CH ₂	1.62-2.02 m*	32.7	1.52-1.82 m*	32.8	1.51-1.82 m ³

^aSpectra recorded at 600 MHz for ¹H NMR and 150 MHz for ¹³C NMR. Chemical shifts (δ) are in ppm and coupling constants (J) in Hz. ^bAssigned by ¹H–¹H COSY. *Signals overcharged.

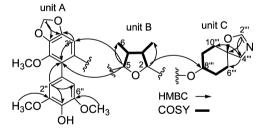


Figure 1. Key ${}^{1}H-{}^{1}H$ COSY and HMBC correlations supporting the structures of units A–C and their final assembly into gymnothelignan A.

confirmed by single-crystal X-ray diffraction (Figure S85, Supporting Information). The relative stereochemistry was thus determined as $2R^*$, $3S^*$, $4S^*$, $5R^*$.

Gymnothelignan D (4) was found to have the molecular formula $C_{24}H_{32}O_8$ by HRESIMS. The 1H and ^{13}C NMR spectra (Table 4) were very similar to those of 1 except for the OCH₃ replacing unit C at C-2. Single-crystal²⁵ X-ray diffraction (Figure S86, Supporting Information) revealed that 4 had two conformers (conformers 4a and 4b, Figure 3) on the THF moiety, but only a single conformer in solution (acetone) was observed. The chemical shifts values of H-2 and H-5 in acetone- d_6 were distinguishable, and the vicinal coupling constants of H-4–H-5 (9.9 Hz) and H-2–H-3 (0 Hz) were

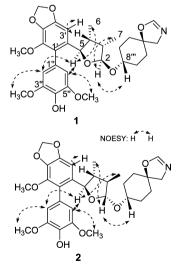


Figure 2. Key NOESY spectra of compounds 1 and 2.

in favor of the conformation of THF in 4 (conformer 4b, Figure 3) when dihedral angle of H-4—H-5 was near 180° and that of H-2—H-3 was near 90°. Finally, the absolute configuration was established by single-crystal X-ray diffraction, as (2S,3S,4R,5R)-3,4-dimethyl-2-methoxyl-5-(4"-hydroxyl-4',5',6',3",5"-pentamethoxylbiphenyl-2'-yl)THF.

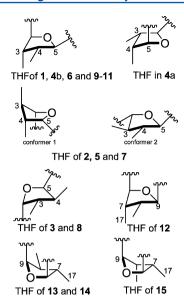


Figure 3. Conformations of THF in compounds 1-15.

Gymnothelignan E (5) was assigned as a diastereomer of 4 by comparing its HRESIMS and NMR (Table 4) with those of 4. The 13 C NMR chemical shifts of C-2 ($\delta_{\rm C}$ 108.3) and C-3 ($\delta_{\rm C}$ 40.3) shifted ($\Delta\delta$ = -3.3 and -3.9, respectively) in comparison to that of 4, indicating that epimerization(s) had occurred in the molecule. Characteristic NOESY (Figure S24, Supporting Information) cross-peaks of H-2/H₃-7, H₃-7/H-4, 2-OCH₃/H-5, and H-3/5 distinctly confirmed the proposed relative stereochemistry structure of 5. The vicinal coupling constants (H-2–H-3 and H-4–H-5) along with the relative configuration of different substituents on the THF suggested that THF had two interconvertible conformers (Figure 3). Henceforth, the structure of 5 was elucidated as the 2,3-bisepimer of 4.

Gymnothelignans F (6) and G (7) had the same molecular formulas as 8, as shown by HRESIMS. The NMR spectra (Tables 2 and 3) revealed that 6, 7, and 8 were diastereomers. Both 6 and 7 demonstrated *trans*-H-4-H-5 configuration by comparing their 13 C NMR chemical shifts of C-5 with that of 4 (the alternative explanation that the γ -gauche interaction 26 had been substantially relieved by comparing the 13 C NMR

Table 3. 13 C NMR Data for Compounds 6–9 and 11 (Acetone- d_6)^a

2 111.5 108.2 106.2 104.3 3 44.0 40.4 42.6 44.1 4 44.4 44.3 46.6 45.1 5 84.1 84.9 79.5 83.8 6 11.8 15.2 16.2 12.2 7 11.3 9.1 12.5 11.2 2' 137.2 137.7 134.3 137.0 3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	105.9 44.0 44.6 84.3 11.8 11.4
4 44.4 44.3 46.6 45.1 5 84.1 84.9 79.5 83.8 6 11.8 15.2 16.2 12.2 7 11.3 9.1 12.5 11.2 2' 137.2 137.7 134.3 137.0 3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	44.6 84.3 11.8
5 84.1 84.9 79.5 83.8 6 11.8 15.2 16.2 12.2 7 11.3 9.1 12.5 11.2 2' 137.2 137.7 134.3 137.0 3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	84.3 11.8
6 11.8 15.2 16.2 12.2 7 11.3 9.1 12.5 11.2 2' 137.2 137.7 134.3 137.0 3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	11.8
7 11.3 9.1 12.5 11.2 2' 137.2 137.7 134.3 137.0 3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	
2' 137.2 137.7 134.3 137.0 3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	11.4
3' 102.8 101.4 102.9 103.4 4' 149.5 149.3 148.9 149.3	
4' 149.5 149.3 148.9 149.3	137.3
	102.8
5/ 13/3 13/0 13/0 1371	149.5
5' 136.3 136.9 136.9 137.1	136.1
6' 141.4 141.9 141.7 141.4	141.5
1' 130.1 128.5 128.6 129.5	130.2
1" 127.4 127.5 127.5 127.6	127.4
2" 110.1 110.0 109.3 110.2	110.0
3" 148.2 148.5 148.5 148.3	148.3 ^b
4" 135.9 135.9 135.9 135.9	135.9
5" 148.1 148.4 148.2 148.2	148.1 ^b
6" 108.5 108.3 108.2 108.5	108.5
2-OCH ₃ 55.0 54.9 54.5	
OCH ₂ O 102.1 102.0 102.0 102.0	102.1
6'-OCH ₃ 60.1 60.1 60.1 60.1	60.1
3"-OCH ₃ 56.8 56.7 56.7 56.8	56.7
5"-OCH ₃ 56.7 56.8 56.7 56.8	56.7

^aSpectra recorded at 150 MHz. Chemical shifts (δ) are in ppm. ^bSignals may be exchangeable.

chemical shifts of C-2′ with that of 8, in which C-2′ was shielded by C-6 when it demonstrated *cis*-H-4–H-5). NOESY (Figures S28 and S32, diagnostic cross-peaks of H-2/H₃-7, H-4/H-3′, H-5/H₃-6, and 2-OCH₃/H-3′ for 6, H-2/H₃-7/H-4, H-4/H-3′, and H-5/H₃-6 for 7; Supporting Information) and vicinal coupling constants (H-2–H-3 and H-4–H-5) allowed the assignment of relative stereochemistry and conformation of THF (Figure 3). Compounds 6 and 7 were subsequently established to be the 2,4- and 3,4-bisepimer of 8, respectively.

Gymnothelignans I (9) and J (10) were found to be demethyl analogues of 6 and 4, respectively, based on the HRESIMS and NMR data (Tables 2, 3, and 4) analyses, with major differences in the ¹³C NMR chemical shifts of C-2.

Table 2. ¹H NMR Data for Compounds 6-9 and 11 (Acetone-d₆)^a

no.	6	7	8	9	11
2	4.55 s	4.97 d (4.9)	4.91 d (4.4)	5.04 d (1.3)	5.07 br s
3	2.15 m ^b	1.97 m ^b	1.86 m ^b ,*	2.02 m ^b ,*	2.36 m ^c
4	2.35 m ^b	2.36 m ^b	1.82 m ^b ,*	2.34 m ^b	2.47 m ^c
5	4.53 d (8.9)	4.54 d (4.2)	5.01 d (8.6)	4.48 d (8.0)	4.55 d (8.9)
6	0.72 d (7.0)	0.64 d (7.2)	0.69 d (6.9)	0.69 d (7.0)	0.79 d (6.9)
7	0.76 d (7.3)	0.84 d (7.2)	0.91 d (6.6)	0.77 d (7.3)	0.85 d (7.3)
3′	6.86 s	6.68 s	6.64 s	7.16 s	6.91 s
2"	6.42 d (1.6)	6.42 d (1.7)	6.39 br s	6.42 d (1.6)	6.44 br s
6"	6.45 d (1.7)	6.45 d (1.7)	6.44 br s	6.44 d (1.4)	6.44 br s
2-OCH ₃	3.38 s	3.22 s	3.21 s		
OCH ₂ O	6.02, 6.02 (each 1H, d, 0.9)	5.99, 5.98 (each 1H, d, 0.9)	6.01, 6.00 (each 1H, d, 1.0)	6.00, 6.00 (each 1H, br s)	6.02, 6.01 (each 1H, br s)
6'-OCH ₃	3.74 s	3.73 s	3.74 s	3.73 s	3.74 s
3"-OCH ₃	3.82 s	3.81 s	3.81 s	3.81 s	3.82 s
5"-OCH ₃	3.80 s	3.79 s	3.80 s	3.81 s	3.81 s

[&]quot;Spectra recorded at 600 MHz. Chemical shifts (δ) are in ppm and coupling constants (J) in Hz. "Assigned by NOESY. "Assigned by $^1H-^1H$ COSY. "Signals overcharged.

Table 4. NMR Data for Compounds 4, 5, and 10 (Acetone- d_6)^a

	4		5		10	
no.	Н	С	Н	С	Н	С
2	4.58 s	111.6	5.00 d (4.9)	108.3	5.07 d (2.0)	104.4
3	2.15 m ^b	44.2	2.01 m ^b	40.3	2.11 m^{b}	44.3
4	2.40 m ^b	44.5	2.38 m ^{b,*}	44.3	2.40 m ^b	45.0
5	4.57 d (9.9)	84.2	4.62 d (4.0)	85.0	4.55 d (7.7)	83.9
6	0.71 d (7.0)	11.9	0.65 d (7.2)	15.3	0.70 d (7.0)	12.4
7	0.77 d (7.3)	11.3	0.85 d (7.2)	9.1	0.79 d (7.3)	11.3
2'		137.4		138.7		138.2
3'	7.08 s	107.5	6.89 s	106.1	7.41 s	110.1
4'		153.9		153.6		153.7
5'		142.2		142.1		142.0
6'		151.6		152.1		151.6
1'		130.3		129.0		129.7
1"		127.5		127.7		127.7
2"	6.44 br s	110.0	6.45 d (1.4)	109.9	6.45 br s	108.4
3"		148.2		148.4		148.2
4"		135.8		135.9		135.8
5"		148.1		148.4		148.2
6"	6.45 br s	108.4	6.48 br s	108.2	6.47 br s	108.2
2-OCH ₃	3.44 s	55.1	3.25 s	54.9		
4'-OCH ₃	3.87 s	61.2	3.87 s	61.3	3.86 s	61.2
5'-OCH ₃	3.80 s	60.8	3.83 s	60.8	3.82 s	60.7
6'-OCH ₃	3.58 s	56.1	3.58 s	56.3	3.58 s	56.1
3"-OCH ₃	3.81 s	56.7	3.81 s	56.8	3.82 s	56.8
5"-OCH ₃	3.81 s	56.7	3.81 s	56.8	3.82 s	56.8

^aSpectra recorded at 600 MHz for 1 H NMR and 150 MHz for 13 C NMR. Chemical shifts (δ) are in ppm and coupling constants (J) in Hz. b Assigned by NOESY. *Signals overcharged.

NOESY (Figures S40 and S44, Supporting Information) correlations (H-2/H₃-7, H-4/H-3', and H-5/H₃-6 for **9**, H-2/H₃-7, H-4/H-3', and H-5/H₃-6 for **10**) were similar to those of corresponding compounds **6** and **4**, suggesting the same relative stereochemistry of different substituents on THF in the two pairs of compounds. The conformations of THF of **9** and **10** are shown in Figure 3. Their structures were established as shown

Gymnothelignan K (11) was shown to have the molecular formula $C_{44}H_{50}O_{15}$ by HRESIMS. Its 1H and ^{13}C NMR spectra (Tables 2 and 3) were in good agreement with those of 9. Hence, compound 11 was presumed to be a dimer of 9 with a C_2 symmetry axis. Key HMBC (Figure S48, Supporting Information) correlation from H-2b to C-2a (or H-2a to C-2b) confirmed the linkage via an oxygen bridge between C-2a of unit A and C-2b of unit B. NOESY (Figure S49, Supporting Information) cross-peaks of H-2/H₃-7, H-4/H-3', and H-5/H₃-6 were in agreement with those of 9. Therefore, it was concluded that the conformations of THF (Figure 3) of both 11 and 9 were identical despite different substituents on the THF moiety. Thus, the absolute configuration of 11 was established as $2S_3S_5A_8S_7$.

Gymnothelignan L (12) was found to have molecular formula $C_{22}H_{24}O_7$ by HRESIMS. The 1H and ^{13}C NMR spectra (Tables 5 and 6) indicated that it had a dibenzocyclooctene skeleton. The HMBC (Figure 4) correla-

Table 5. ¹H NMR Data for Compounds 12–15 (Acetone- d_6)^a

no.	12	13	14	15
1	7.24 s	7.28 s	6.42s	6.50 s
6	5.09 d (1.9)	5.08 d (5.7)	3.43 d (2.4)	3.41 d (2.7)
7	2.22 m ^b	1.84 m ^b	2.03 m ^b ,*	2.76 m ^b
8	2.33 m ^b	2.08 m ^b ,*	1.97 m ^b ,*	2.31 m ^b
9	4.48 d (5.5)	4.99 d (6.2)	4.85 d (5.6)	4.71 s
11	6.50 s	6.36 s		
12			6.20 d (2.3)	6.19 d (2.3)
16			6.08 d (2.3)	6.09 d (2.3)
17	1.03 d (7.1)	1.16 d (6.7)	1.11 d (6.9)	1.02 d (7.4)
18	1.01 d (7.1)	0.59 d (6.8)	0.85 d (6.8)	1.05 d (7.2)
19	6.02, 5.96 (each 1H, d, 1.0)	6.06, 5.99 (each 1H, d, 0.8)	5.97, 5.96 (each 1H, d, 0.9)	5.96, 5.94 (each 1H, d, 0.9)
20	3.68 s	3.68 s	3.65 s	3.65 s
21	3.81 s	3.81 s	3.56 s	3.56 s
22	3.85 s	3.83 s	3.67 s	3.69 s

^aSpectra recorded at 600 MHz. Chemical shifts (δ) are in ppm and coupling constants (J) in Hz. ^bAssigned by NOESY. *Signals overcharged.

Table 6. ¹³C NMR Data for Compounds 12–15 (Acetone- d_6)^a

no. 12 13 14 1 113.4 114.7 102.1 2 146.6 146.7 148.7 3 138.8 138.5 137.7 4 144.5 144.9 144.3	15 101.1 149.2 137.5 144.0
2 146.6 146.7 148.7 3 138.8 138.5 137.7	149.2 137.5
3 138.8 138.5 137.7	137.5
4 144.5 144.9 144.3	144.0
1 1110 1110	177.0
5 130.6 132.6 49.5	46.7
6 85.2 82.2 92.3	92.4
7 42.6 53.5 42.6	36.7
8 49.4 46.1 46.8	45.8
9 90.6 87.9 83.1	85.7
10 140.6 133.7 131.0	134.5
11 103.7 103.9 120.6	120.1
12 147.6 147.6 118.8	118.6
13 139.4 139.3 150.3	149.7
14 143.7 143.4 175.6	175.6
15 124.6 125.7 154.0	154.0
16 123.7 124.0 122.0	121.7
17 13.7 17.5 19.4	15.5
18 13.9 13.8 14.0	15.1
19 102.2 102.3 102.4	102.1
20 56.4 56.7 59.7	59.7
21 60.2 60.4 55.2	55.2
22 60.4 61.5 55.5	55.5

^aSpectra recorded at 150 MHz. Chemical shifts (δ) are in ppm.

tions of H-6/C-9 and H-9/C-6 presented the 6,9-epoxide. The correlations of H-1 to C-2, C-3, C-5, C-15, and C-16 corroborated the proton $\delta_{\rm H}$ 7.24 at C-1, differing from that of a congener reported in the literature. In addition, a strong correlation of H₃-22/H-6 was observed in the NOESY spectrum (Figure 4, excluding the possibility of $\delta_{\rm H}$ 7.24 at C-4). The correlations of H-6 to H₃-17 and H-9 to H₃-18 suggested H-6/H₃-17 and H-9/H₃-18 to be cofacial. The vicinal coupling constants (H-6 and H-9), along with assigned relative stereochemistry of H₃-17 and H₃-18, allowed the conformation assignment of THF, as shown in Figure 3. Compound 12 was established as $6S^*$, $7S^*$, $8R^*$, $9R^*$ and R^* -biphenyl. $2S^*$

Figure 4. Selected NOESY and HMBC spectra of compounds 12 and 14

Gymnothelignan M (13) was assigned as a diastereomer of 12 by comparison of the HRMS and NMR (Tables 5 and 6) data with that of 12. Diagnostic cross-peaks in the NOESY (Figure S57, Supporting Information) of H-11/H-9, H-11/H₃-18, H-7/H₃-18, H-7/H₃-17, and H-6/H₃-17 supported that H₃-17 and H₃-18 were in the α - and β -orientation, respectively. Consequently, compound 13 was proposed to be an epimer of 12 at C-8. The vicinal coupling constants of H-6 and H-9 along with substituents on the THF allowed the conformation assignment of THF as shown in Figure 3. The structure of 13 was therefore elucidated as the 8-epimer of 12.

Gymnothelignan N (14) was assigned to have molecular formula C₂₂H₂₄O₇ on the basis of HRESIMS. The ¹H and ¹³C NMR spectrum (Tables 5 and 6) showed that it probably belonged to the eupodienone⁸ family. The HMBC (Figure 4) correlations of H-6/C-9 and H-9/C-6 confirmed an epoxide at C-6 and C-9. The stereochemistry of epoxide at C-6 and C-9 was proposed to be in α -orientation due to the α -orientation of a substituent at C-9 of eupodienone (it was favorable for cyclization occurring at C-6 and C-9 without epimerizing). Two trans-methyl groups (H₃-17 and H₃-18) were determined by NOESY (Figure 4, cross-peaks of H-9/H-8 and H-6/H₃-17). The conformation of THF (Figure 3) favored the proposed relative configuration. Finally, the absolute configuration was established by single-crystal X-ray diffraction (Figure S85, Supporting Information) as (6S,7S,8S,9R)-6,9-epoxy-7,8-dimethyl-2,3-methylenedioxy-4,13,15-trimethoxy-10,11benzospiro[5.6]dodec-13,15-dien-14-one.

Gymnothelignan O (15) was found to have molecular formula C₂₂H₂₄O₇ by HRESIMS. Its ¹H and ¹³C NMR spectra (Tables 5 and 6) were quite close to those of 14. Detailed analysis of ¹H and ¹³C NMR spectra suggested that compounds 14 and 15 were diastereomeric at C-8. The vicinal coupling constants (H-9/H-8) value was 5.6 Hz for 14, whereas that of 15 was 0 Hz, indicating an epimerization at C-8. NOESY (Figure S65, Supporting Information) correlations of H-6/H₃-17 and H-9/H₃-18 supported that H-6, H-9, H₃-17, and H₃-18 existed in the same orientation. The conformation of THF is shown in Figure 3.

In order to determine the absolute configurations of compounds 3 and 8, compound 14 underwent nucleophilic substitution reactions^{29,30} to afford 8/14a and 14b.³¹ The possible mechanism of the conversion of 14 to 8 is shown in Scheme 1.²⁹ Therefore, a total of four chiral centers of 3/8 were undoubtedly determined to be 2R, 3S, 4S, 5R.

Similarly, treatment 15 afforded 15a (Scheme 2; a possibly minor product could not be detected probably due to the steric

Scheme 1. Possible Mechanism of 14 to 8^a

"The major product could arise from either inside attack on the diequatorial oxocarbenium ion 1 or outside attack on the diaxial cation 2.²⁹ Actually, the minor product was also elucidated by comparison of the NMR spectra with those of 8.

hindrance). The structure of 15 was thus determined to be the 8-epimer of 14.

Scheme 2. Chemical Transformation of 15 to 15a

The chemical transformation of 14 to yield 8 allowed us to establish the absolute configuration of the eupomatilone family, 32 which was consistent with that of eupomatilone-6 assigned previously. 12,16 It was reported that eupomatilone was derived from eupodienone (Scheme 3). 6 Chemical transformation in vitro buttressed the plausible biosynthetic pathways of eupodienone to eupomatilone mentioned above. The absolute configuration of eupomatilone-3 was therefore proposed as 3S,4S,5R. Considering that compounds 1–11

Scheme 3. Plausible Biosynthetic Pathways of Eupomatilone-3⁶

Scheme 4. Plausible Biosynthetic Pathways of 1, 9, 12, and 15

uniformly demonstrated *R*-configuration at C-5 regardless of whether its ¹³C NMR chemical shift was near δ 79 ppm, the revised structure of synthesized 5-*epi*-eupomatilone- $6^{19,20}$ to 3,5-bis-*epi*-eupomatilone- 6^{18} should again be 4-*epi*-eupomatilone- 6^{15} by our establishment of the absolute configuration.

The discovery of the three arrays of new lignans is an example of chemical diversity, extending the lignan family by derivatives formed by ring cleavage, oxidation, and esterification. Compounds 1, 9, 12, and 15 were considered to be derived from the same parent compound (Scheme 4). Compound 1 may biogenetically be derived from precursor A via intermediates 9 and 15 (two pathways, paths 1a and 1b). 8,33 Compound 12 was probably derived from precursor A via intermediates B or 15 (paths 12a and 12b) by dehydoxyl reaction and rearrangement.8 These compounds were tested for cytotoxic activity on HepG2 and Bcl7404 cell lines using MTT³⁴ methods, while 3 exhibited moderate cytotoxicity against HepG2 and Bcl7404 cells, with IC50 values of 15 and 17.5 μ g/mL, respectively (Table S1, Supporting Information). The seco-lignans 1-11 are rare natural products. Compounds 1-11, except 3 and 8, demonstrate trans-H-4-H-5, which were isolated as natural products for the first time and were mistaken as enantiomers of eupomatilones according to previous studies. 19,20 From now on, it is feasible to determine the absolute configuration for C-5 of the eupomatilone family with trans-H-4-H-5. In summary, the three skeletons of lignans are proposed to be derived from the same parent compound.

EXPERIMENTAL SECTION

General Experimental Procedures. NMR spectra were recorded at 300 K (600 MHz for $^1\mathrm{H}$ and 150 MHz for $^{13}\mathrm{C})$ with the CD₃COCD₃ (δ 2.05/29.8) solvent as internal standard. The 1D and 2D NMR spectra were performed using standard software. The HRESIMS was performed on a Q-TOF mass spectrometer. Preparative HPLC was performed on a liquid chromatograph equipped with an UV detector and semipreparative column (C₁₈, 5

 μ m, 19 × 250 mm). Column chromatography was carried out on silica gel (160–200 mesh), MCI CHP-20 gel (75–150 μ m), ODS (40–63 μ m), and Sephadex LH-20. TLC was performed on precoated plates (GF254), and spots were detected on TLC under UV and by heating after spraying with color reagents: vanillic aldehyde (15 g) + ethanol (250 mL) + H₂SO₄ (2.5 mL). X-ray crystallographic analyses were carried out using Cu Kα radiation (λ = 1.54178 Å) at 298 K and Mo Kα radiation (λ = 0.71073 Å) at 153 K. The structures were solved by direct methods using the SHELXS-97 program.

Plant Material. The plant was collected from Jinfoshan in Chongqing City, People's Republic of China, in July 2010 and identified as *G. chinensis* by Prof. S. R. Yi. A voucher specimen (T57) has been deposited at the herbarium of Chengdu Institute of Biology, Chinese Academy of Science.

Extraction and Isolation. Dried and powdered whole plants of *G*. chinensis (1.4 kg) were extracted with ethanol at room temperature (3 \times 7 days) to give an extract (146 g), which was suspended in H₂O and extracted with petroleum ether and ethyl acetate $(3 \times 0.5 \text{ L}, 3 \text{ h each})$ successively. The ethyl acetate extract (16 g) was separated by column chromatography on MCI CHP-20 (6×30 cm) with a gradient system of aqueous methanol (7:3, 1 L, 8:2, 2 L, 9:1, 1 L, and 10:0, 1 L) to yield three fractions (A-C). Fraction B (8 g) was subjected to silica gel chromatography (4 × 60 cm) with chloroform/methanol mixtures of increasing polarity (35:1 to 1:1) to give fractions (BA-BH) and 20. Fraction BG was separated on a semipreparative column (2-13 min, 40-95%, 13-16 min, 95-100%) to afford 17 (13 mg, t_R 9.4 min) and 16 (18 mg, t_R 11.7 min). Fraction BD was purified on a semipreparative column (2-15 min, 40-95%, 15-18 min, 100%) to afford 18 (21 mg, t_R 9.1 min). Fraction BB (1.2-1.5 L, 3 g) was subjected to repeated silica gel chromatography with petroleum ether/ acetone mixtures of increasing polarity to give a fraction, which was separated on a semipreparative column (2-16 min, 60-95% aqueous methanol, flow rate, t_R 14.3 min) to afford 1 (15 mg), as well as a mixture (2-8 min, 70-80%, 8-16 min, 80%, 16-18 min, 95% aqueous methanol) which afforded 2 (6 mg, t_R 13.2 min) and 3 (2 mg, t_R 12.8 min). Thereafter, fraction BBD was separated on a semipreparative column (2-8 min, 20-85%, 8-12 min, 85-95%, 12-18 min, 95% aqueous methanol, flow rate, 16 mL/min) to afford 14 (25 mg, t_R 11.5 min), 15 (30 mg, t_R 11.8 min), and 12 (8 mg, t_R

12.6 min), as well as a mixture, which was further purified on a semipreparative column (2-15 min, 85-95%) to afford 19 (15 mg, t_R 9.7 min). Fraction BBE was then subjected to silica gel chromatography (3 × 60 cm) eluted with gradient polarity of petroleum ester/acetone (8:1 to 1:1) to give fraction BBEC and was further separated on a semipreparative column (2-15 min, 63-82%, 15–19 min, 82%, 19–23 min, 82–90%) to afford 9 (5 mg, t_R 6.4 min), 10 (3 mg, t_R 10.0 min), 14 (13 mg, t_R 11.3 min), 15 (9 mg, t_R 12.5 min), 6 (8 mg, t_R 13.4 min), 8 (8 mg, t_R 15.2 min), 7 (8 mg, t_R 15.5 min), and 12 (2 mg, t_R 16.1 min). Fraction BBF was subjected to silica gel chromatography (3 × 60 cm) eluted with 80/20 petroleum ether/ ethyl acetate (5% i-PrOH) to give 8 and a mixture, which was further separated on a semipreparative column (2-15 min, 60-95%, 15-18 min, 95-100%) to afford 4 (8 mg, t_R 11.4 min) and 5 (7 mg, t_R 12.1 min). Fraction BBG was separated by silica gel chromatography with petroleum ether/ethyl acetate mixtures of increasing polarity to give three fractions (BBGA-BBGC). Compound 13 (6 mg, t_R 13.6 min) was obtained from fraction BBGB (2-15 min, 60-95%, 15-18 min, 95%), whereas 11 (4 mg, t_R 15.6 min) was obtained from fraction BBGA (2-6 min, 30-60%, 6-6.2 min, 60-80%, 6.2-18 min, 80-95%). Semipreparative column chromatography solvent, aqueous MeOH; flow rate, 16 mL/min.

Crystal Structure Determination of 1. Diffraction data (φ and ω scans) were collected using Cu K α radiation (λ = 1.54178 Å). C₃₀H₃₇NO₉, MW = 555.61, orthorhombic, a = 10.3937(13) Å, b = 10.5073(13) Å, c = 25.914(3) Å, α = β = γ = 90.00°, V = 2830.0(6) ų, T = 298(2) K, space group $P2_12_12_1$, Z = 4, d = 1.304 g/cm³, F(000) = 1184, 6996 reflections measured, 4123 independent reflections ($R_{\rm int}$ = 0.1332). The final R_1 values were 0.0603 (I > $2\sigma(I)$). The final $wR(F^2)$ values were 0.1795 (I > $2\sigma(I)$). The final R_1 values were 0.1954 (all data). The final $wR(F^2)$ values were 0.2478 (all data). The goodness of fit on F^2 was 1.175. Flack parameter = 0.2(3). Crystallographic data for gymnothelignan A (1) have been deposited at the Cambridge Crystallographic Data Center (deposition number CCDC 857186). These data can be obtained free of charge from the Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/data_request/cif.

Crystal Structure Determination of 4. Diffraction data (φ and ω scans) were collected using Cu K α radiation (λ = 1.54178 Å). Compound 4, C₂₄H₃₂O₈, MW = 448.50, was monoclinic and space group $P2_1$ with a = 10.8975(4) Å, b = 18.7065(8) Å, c = 11.7408(5) Å, $\alpha = \gamma = 90^{\circ}, \beta = 92.968(3)^{\circ}, V = 2390.20(17) \text{ Å}^3, Z = 4, d = 1.246 \text{ g/}$ cm³, F(000) = 960; crystal size $0.22 \times 0.15 \times 0.13$ mm; 14004reflections measured, 6996 reflections unique, $\theta_{\rm max}$ = 67.49°. The final R_1 values were 0.0482 $(I > 2\sigma(I))$. The final $wR(F^2)$ values were 0.1253 ($I > 2\sigma(I)$). The final R_1 values were 0.0960 (all data). The final $wR(F^2)$ values were 0.1870 (all data). The goodness of fit on F^2 was 0.990. Flack parameter = 0.1(3). The structure was solved by direct methods (SHELXS-97) and expanded using SHELXL-97. Crystallographic data for gymnothelignan D (4) have been deposited at the Cambridge Crystallographic Data Center (deposition number CCDC 857185). These data can be obtained free of charge from the Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/ data request/cif.

Crystal Structure Determination of 8. Diffraction data (φ and ω scans) were collected using Mo K α radiation (λ = 0.71073 Å). Compound 8, $C_{23}H_{28}O_8$, MW = 432.45, was orthorhombic and space group $P2_12_12_1$ with a = 6.9969(17) Å, b = 11.484(3) Å, c = 27.382(7)Å, $\alpha = \beta = \gamma = 90^{\circ}$, V = 2200.1(9) Å³, Z = 4, d = 1.306 g/cm³, F(000)= 920; crystal size $0.48 \times 0.35 \times 0.08$ mm; 19484 reflections measured, 3370 reflections unique ($R_{int} = 0.0477$). The final R_1 values were 0.0438 ($I > 2\sigma(I)$). The final $wR(F^2)$ values were 0.0906 (I > $2\sigma(I)$). The final R_1 values were 0.0511 (all data). The final $wR(F^2)$ values were 0.0944 (all data), $\theta_{\rm max}$ = 29.12°, goodness of fit was 0.999. The structure was solved by direct methods (SHELXS-97) and expanded using SHELXL-97. Crystallographic data for gymnothelignan H (8) have been deposited at the Cambridge Crystallographic Data Center (deposition number CCDC 870519). These data can be obtained free of charge from the Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/data request/cif.

Crystal Structure Determination of 14. Diffraction data (φ and ω scans) were collected using Cu K α radiation (λ = 1.54178 Å). Compound 14, $C_{24}H_{24}O_7$, MW = 400.41, was monoclinic and space group $P2_1$ with a = 10.068(3) Å, b = 9.308(2) Å, c = 11.584(3) Å, $\alpha =$ $\gamma = 90^{\circ}$, $\beta = 108.829(10)^{\circ}$, V = 1027.5(5) Å³, Z = 2, d = 1.294 g/cm³, F(000) = 424; crystal size $0.28 \times 0.22 \times 0.19$ mm; 4103 reflections measured, 3642 independent reflections ($R_{int} = 0.0266$). The final R_1 values were 0.0714 ($\overline{I} > 2\sigma(I)$). The final $wR(F^2)$ values were 0.2309 (I $> 2\sigma(I)$). The final R_1 values were 0.1246 (all data). The final $wR(F^2)$ values were 0.2943 (all data). The goodness of fit on F^2 was 1.156. Flack parameter = -0.3(6). The structure was solved by direct methods (SHELXS-97) and expanded using SHELXL-97. Crystallographic data for gymnothelignan N (14) have been deposited at the Cambridge Crystallographic Data Center (deposition number CCDC 870520). These data can be obtained free of charge from the Cambridge Crystallographic Data Center via www.ccdc.cam.ac.uk/ data request/cif.

Gymnothelignan A (1): colorless crystals; $[\alpha]_{\rm D}^{20}$ –9 (*c* 0.09, CH₃OH); mp 171–172 °C; UV $\lambda_{\rm max}$ (CH₃OH) 284 nm; IR (KBr) $\nu_{\rm max}$ 3437, 1619 cm⁻¹; ¹H and ¹³C NMR data (Table 1); HRESIMS m/z 578.2368 [M + Na]⁺ (calcd for C₃₀H₃₇NO₉Na 578.2361).

Gymnothelignan B (2): off-white powder; $[\alpha]_2^{20}$ –65 (c 0.04, CH₃OH); UV $\lambda_{\rm max}$ (CH₃OH) 281 nm; IR (KBr) $\nu_{\rm max}$ 3436, 1615 cm⁻¹; ¹H and ¹³C NMR data (Table 1); HRESIMS m/z 578.2382 [M + Na]⁺ (calcd for C₃₀H₃₇NO₉Na 578.2361).

Gymnothelignan C (3): off-white powder; $[\alpha]_2^{20}$ –30 (c 0.03, CH₃OH); UV $\lambda_{\rm max}$ (CH₃OH) 280 nm; IR (KBr) $\nu_{\rm max}$ 3401, 1635 cm⁻¹; ¹H and ¹³C NMR data (Table 1); HRESIMS m/z 578.2385 [M + Na]⁺ (calcd for C₃₀H₃₇NO₉Na 578.2361).

Gymnothelignan D (4): colorless crystals; $[\alpha]_{20}^{20}$ -12 (c 0.03, CH₃OH); mp 133–134 °C; UV $\lambda_{\rm max}$ (CH₃OH) 276 nm; IR (KBr) $\nu_{\rm max}$ 3392, 2960, 2933, 1607, 1490, 1461, 1097 cm⁻¹; ¹H and ¹³C NMR data (Table 4); HRESIMS m/z 471.1978 [M + Na]⁺ (calcd for C₂₄H₃₂O₈Na 471.1989).

Gymnothelignan E (5): white powder; $[\alpha]_D^{20} - 7$ (c 0.35, Me₂CO); UV λ_{max} (CH₃OH) 277 nm; IR (KBr) ν_{max} 3429, 2935, 1608, 1488, 1463, 1102 cm⁻¹; ¹H and ¹³C NMR data (Table 4); HRESIMS m/z 471.1976 [M + Na]⁺ (calcd for C₂₄H₃₂O₈Na 471.1989).

Gymnothelignan F (6): yellowish powder; $[\alpha]_{20}^{20}$ –2 (c 0.20, Me₂CO); UV λ_{max} (CH₃OH) 287 nm; IR (KBr) ν_{max} 3430, 2962, 2936, 1614, 1519, 1476, 1112 cm⁻¹; ¹H and ¹³C NMR data (Tables 2 and 3); HRESIMS m/z 455.1690 [M + Na]⁺ (calcd for C₂₃H₂₈O₈Na 455.1676).

Gymnothelignan G (7): off-white powder; $[\alpha]_{2}^{20}$ –10 (c 0.25, Me₂CO); UV λ_{max} (CH₃OH) 296 nm; IR (KBr) ν_{max} 3442, 2928, 1606, 1469, 1115 cm⁻¹; ¹H and ¹³C NMR data (Tables 2 and 3); HRESIMS m/z 455.1673 [M + Na]⁺ (calcd for C₂₃H₂₈O₈Na 455.1676).

Gymnothelignan H (8): colorless crystals; $[\alpha]_{2}^{20}$ -80 (c 0.03, Me₂CO); mp 184–185 °C; UV $\lambda_{\rm max}$ (CH₃OH) 301 nm; IR (KBr) $\nu_{\rm max}$ 3393, 2958, 2934, 1607, 1475, 1465, 1115, 1026 cm⁻¹; ¹H and ¹³C NMR data (Tables 2 and 3); HRESIMS m/z 455.1675 [M + Na]⁺ (calcd for C₂₃H₂₈O₈Na 455.1676).

Gymnothelignan I (9): yellowish powder; $[\alpha]_D^{20} - 8$ (c 0.10, Me₂CO); UV λ_{max} (CH₃OH) 302 nm; IR (KBr) ν_{max} 3436, 2964, 2937, 1614, 1476, 1113 cm⁻¹; ^1H and ^{13}C NMR data (Tables 2 and 3); HRESIMS m/z 441.1522 [M + Na]⁺ (calcd for C₂₂H₂₆O₈Na 441.1520).

Gymnothelignan J (10): yellowish powder; $[\alpha]_D^{20}$ –4 (c 0.10, Me₂CO); UV λ_{max} (CH₃OH) 298 nm; IR (KBr) ν_{max} 3426, 2962, 2936, 1608, 1460, 1100 cm⁻¹; ¹H and ¹³C NMR data (Table 4); HRESIMS m/z 457.1852 [M + Na]⁺ (calcd for C₂₃H₃₀O₈Na 457.1833).

Gymnothelignan K (11): white powder; $[\alpha]_D^{20}$ –10 (c 0.09, CH₃OH); UV $\lambda_{\rm max}$ (CH₃OH) 280 nm; IR (KBr) $\nu_{\rm max}$ 3435, 2935, 1615, 1476, 1114 cm⁻¹; ¹H and ¹³C NMR data (Tables 2 and 3); HRESIMS m/z 841.3069 [M + Na]⁺ (calcd for C₄₄H₅₀O₁₅Na 841.3043).

Gymnothelignan L (12): yellow powder; $[\alpha]_D^{20}$ -2 (c 0.20, Me₂CO); UV $\lambda_{\rm max}$ (CH₃OH) 307 nm; IR (KBr) $\nu_{\rm max}$ 3429, 2961,

2934, 1616, 1479, 1101 cm $^{-1}$; 1 H and 13 C NMR data (Tables 5 and 6); HRESIMS m/z 401.1595 $[M + H]^{+}$ (calcd for $C_{22}H_{25}O_{7}$ 401.1594).

Gymnothelignan M (13): yellow powder; $[\alpha]_{20}^{D0}$ +8 (c 0.03, CH₃OH); UV λ_{max} (CH₃OH) 300 nm; IR (KBr) ν_{max} 3428, 2957, 2928, 1617, 1451, 1079 cm⁻¹; ^{1}H and ^{13}C NMR data (Tables 5 and 6); HRESIMS m/z 401.1597 $[M + H]^{+}$ (calcd for $C_{22}H_{25}O_{7}$ 401.1594).

Gymnothelignan N (14): colorless crystals; $[\alpha]_{20}^{20}$ +13 (c 0.04, CH₃OH); mp 201–202 °C; UV $\lambda_{\rm max}$ (CH₃OH) 290 nm; IR (KBr) $\nu_{\rm max}$ 2958, 2936, 1668, 1614, 1475, 1112 cm⁻¹; ¹H and ¹³C NMR data (Tables 5 and 6); HRESIMS m/z 423.1423 [M + Na]⁺ (calcd for C₂₂H₂₄O₇Na 423.1414).

Gymnothelignan O (15): white powder; $[\alpha]_{0}^{20}$ +15 (c 0.04, CH₃OH); UV $\lambda_{\rm max}$ (CH₃OH) 283 nm; IR (KBr) $\nu_{\rm max}$ 2936, 1668, 1615, 1477, 1113 cm⁻¹; ¹H and ¹³C NMR data (Tables 5 and 6); HRESIMS m/z 423.1435 [M + Na]⁺ (calcd for C₂₂H₂₄O₇Na 423.1414).

Treatment of 14 and 15 with H₂SO₄. Gymnothelignan N (14, 11 mg) was dissolved in methanol (2 mL) and then added 5 mL of 2 M H₂SO₄ and heated at 80 °C under reflux for 1 h. After cooling to room temperature, the solution was neutralized with excess NaHCO3 and filtered, and the filtrate was extracted three times each with 7 mL of ethyl acetate. The combined organic layer was evaporated to dryness and then separated by HPLC with a C18 column, using a mixed solvent of methanol/water (0-2 min, 70%, 2-16 min, 70-95%, 16–18 min, 95%) to yield the corresponding 14a (1.7 mg, t_R 10.8 min, 15% yield) and 14b (0.5 mg, t_R 9.7 min, 5% yield). 14a: white powder, $[\alpha]_{\rm D}^{20}$ -80 (c 0.03, Me₂CO); ¹H NMR (600 MHz, acetone- d_6 , $\delta_{\rm H}$ 2.05, Figure S66) was in good agreement with that of 8. 14b: yellowish powder, $[\alpha]_D^{20}$ +100 (c 0.02, Me₂CO); ¹H NMR (600 MHz, acetone d_6 , $\delta_{\rm H}$ 2.05, Figure S67) $\delta_{\rm H}$ 6.90 (s, H-3'), 6.45, 6.36 (br s, each 1H, H-6", and 2"), 6.01, 6.00 (br s, each 1H, OCH₂O), 5.11 (d, J = 7.1 Hz, H-5), 4.54 (d, J = 2.6 Hz, H-2), 3.82, 3.80, 3.74, 3.43 (s, each 3H, 3", 5'', 6', and 2-OCH₃), 1.69, 1.67 (m, 2H, H-4, and H-3), 0.90 (d, J =7.1 Hz, H₃-7), 0.71 (d, J = 7.2 Hz, H₃-6). Treatment of 15 (17 mg) as described for 14 afforded corresponding 15a (2.8 mg, t_R 16.2 min, 16% yield, solvent, 60-95% aqueous MeOH, flow rate, 16 mL/min). 15a: white powder, 1 H NMR (600 MHz, acetone- d_6 , $\delta_{\rm H}$ 2.05, Figure S68) was in good agreement with that of 6.

ASSOCIATED CONTENT

Supporting Information

IR, 1D, and 2D NMR spectra of compounds **1–15**, ¹H NMR spectra of **14***a*, **14***b*, and **15***a*, NMR free induction decay (FID) files of compounds **1**, **4**, and **8**, CIFs of compounds **1**, **4**, **8**, and **14**, cytotoxic activity of these compounds on HepG2 and Bcl7404 cell lines. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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