# Fragment Based Design of New H<sub>4</sub> Receptor—Ligands with Anti-inflammatory Properties in Vivo

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Using a previously reported flexible alignment model we have designed, synthesized, and evaluated a series of compounds at the human histamine  $H_4$  receptor ( $H_4R$ ) from which 2-(4-methyl-piperazin-1-yl)-quinoxaline (3) was identified as a new lead structure for  $H_4R$  ligands. Exploration of the structure—activity relationship (SAR) of this scaffold led to the identification of 6,7-dichloro 3-(4-methyl-piperazin-1-yl)quinoxaline (VUF 10214, 57) and 2-benzyl-3-(4-methyl-piperazin-1-yl)quinoxaline (VUF 10148, 20) as potent  $H_4R$  ligands with nanomolar affinities. In vivo studies in the rat reveal that compound 57 has significant anti-inflammatory properties in the carrageenan-induced paw-edema model.

### Introduction

Histamine is an endogenous amine that exerts a wide array of physiological activities by a group of four G-protein coupled receptors (GPCR's). The histamine H<sub>1</sub> receptor (H<sub>1</sub>R) mediates inflammation and vasodilatation, and H<sub>1</sub>R antagonists have long been used for the treatment of allergic disorders such as hay fever. The H<sub>2</sub>R is well-known for its regulation of gastric acid secretion and H<sub>2</sub>R antagonists have been successfully used for the treatment of gastric ulcers. The H<sub>3</sub>R is being studied intensively and has been shown to play an important role in several cognitive processes as well as in the regulation of food intake and the regulation of sleep and wakefulness. Currently H<sub>3</sub>R antagonists are being evaluated for their clinical efficacy in ADHD<sup>a</sup>, dementia and narcolepsy. The H<sub>3</sub>R is the processes as well as in the regulation of food intake and the regulation of sleep and wakefulness.

The H<sub>4</sub>R is the most recently discovered histamine receptor subtype. The DNA sequence was discovered in the human genome by several groups, and subsequent pharmacological characterization identified it as a novel histamine receptor.<sup>4-8</sup> Although the sequence identity with the H<sub>1</sub>R and H<sub>2</sub>R is relatively low, the H<sub>4</sub>R protein has a considerable sequence identity with the H<sub>3</sub>R receptor (31% overall, 54% in the transmembrane region).9 Nevertheless, the H<sub>4</sub>R has a distinct pharmacological profile. <sup>9</sup> The H<sub>4</sub>R plays a role in immunological and inflammatory processes and is predominantly expressed on hematopoetic and immune cells such as eosinophils, mast cells, and macrophages as well as in peripheral tissues such as spleen, thymus, and bone marrow. 10 Currently, the H<sub>4</sub>R is considered a promising target for the treatment of various chronic inflammatory diseases such as inflammatory bowel disease, asthma, and rheumatoid arthritis. 11-13 Quite recently, its role has also been postulated in the proliferation of colon carcinoma cells, in the modulation of angiogenesis, and in mediating pruritis. 14,15

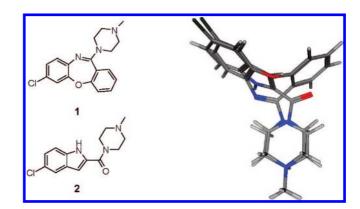


Figure 1. Previously reported H<sub>4</sub>R ligands 1 and 2.

Previously we reported the discovery of VUF6884 (1, Figure 1), a tricyclic clozapine analogue with high affinity for the  $H_4R$  (p $K_i$  7.6, full agonist  $\alpha = 1$ ), and saturation binding analysis demonstrated that this ligand bound to the orthosteric binding site of the  $H_4R$ .<sup>16</sup>

We also showed that the  $H_4R$  antagonist JNJ7777120 (2)<sup>17,18</sup> (p $K_i$  7.8, antagonist  $\alpha = 0$ ) displaces 1 from its  $H_4R$  binding site. On the basis of these findings and on the structural similarities between 1 and 2, we proposed that they had overlapping binding modes and created a model for the future design of  $H_4R$  ligands (Figure 1). In this paper, we will discuss the design and synthesis, structure–activity relationship, and pharmacological evaluation of a new class of potent  $H_4R$  ligands, which was discovered on the basis of the flexible alignment model of 1 and 2.<sup>16</sup>

# Chemistry

Compounds 3–7 (Table 1) were prepared in one step from their commercially available chloro precursors by substitution with *N*-methylpiperazine under microwave irradiation. Starting from 2,4-dichloroquinazoline (8), quinazoline fragment 9 was synthesized according to patent literature (Scheme 1).<sup>19</sup> Starting from phenylenediamine (10) and 4,5-dichlorophenylenediamine (11) (Scheme 2), intermediate quinoxalinones 12–17 were prepared as described in literature, <sup>20–22</sup> except for intermediate

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<sup>&</sup>lt;sup>a</sup> Abbreviations: ADHD, attention deficit hyperactivity disorder; SEM, standard error of the mean; PEI, pectin esterase inhibitor; DIPEA, diisopropylethylamine, PMB, *para*-methoxybenzylamine.

Table 1. Structures and Affinity of Several Heterocyclic Fragments for the H<sub>4</sub>R

Nr	Structure	$pK_i (\mu M) \pm SEM^a$	Ligand Efficiency <sup>b</sup>
3	N N N N N N N N N N N N N N N N N N N	$6.05 \pm 0.07$	0.488
4		$5.16 \pm 0.06$	0.416
5		$6.23 \pm 0.04$	0.475
6		$7.04 \pm 0.01$	0.536
7	N N N N N N N N N N N N N N N N N N N	$4.69 \pm 0.07$	0.378
9	N N N N N N N N N N N N N N N N N N N	$5.12 \pm 0.06$	0.413

<sup>&</sup>lt;sup>a</sup> Measured by displacement of [<sup>3</sup>H]histamine binding using membranes of HEK cells transiently expressing the human H<sub>4</sub>R. pK<sub>i</sub>'s are calculated from at least three independent measurements as the mean  $\pm$  SEM. <sup>b</sup> Ligand efficiency ( $\Delta g$ ) is calculated as the binding energy per non-hydrogen atom ( $\Delta g = \Delta G/N_{\text{non-hydrogen atoms}}$  with  $\Delta G = -RT \ln K_i$ ).

#### Scheme 1<sup>a</sup>

$$\begin{bmatrix}
N & CI & a \\
N & N
\end{bmatrix}$$

$$\begin{bmatrix}
N & CI \\
N & N
\end{bmatrix}$$

$$\begin{bmatrix}
N & N & N \\
N & N
\end{bmatrix}$$

$$\begin{bmatrix}
N & N & N \\
N & N
\end{bmatrix}$$

#### Scheme 2<sup>a</sup>

15, which was synthesized from 4,5-dichlorophenylenediamine (11) and ethyl 2-oxo-4-phenylbutyrate. The different quinoxaline-2-ones were converted to their corresponding chlorides with

neat phosphorus oxychloride. After removal of excess reagent and no further work up, microwave-assisted heating was used to couple the crude chlorides with neat *N*-methylpiperazine to yield quinoxalines **18–23** (Scheme 2).

Commercially available 2,3-dichloroquinoxaline (24) and 2,3,6,7-tetrachloroquinoxaline (25) were reacted with N-methylpiperazine in the presence of diisopropylethylamine in tetrahydrofuran under microwave irradiation to give key intermediates 26 and 27 in high yields (Scheme 3). Unsubstituted quinoxaline 26 was then added to solutions of alcohols and sodium hydride in dimethylformamide to give compounds 28-50 (Table 2).<sup>23</sup> Thioether **51** and phenolether **52** were synthesized by the same procedure from intermediates 26 and 27, respectively. Methoxy analogues 53 and 54 were synthesized from their corresponding precursors by refluxing in methanol in the presence of sodium methoxide. Benzylamine 55 was synthesized from intermediate 26 by microwave-assisted heating in neat benzylamine. Compound 56 could either be prepared by hydrolysis of methoxy analogue 53 with aqueous sodium hydroxide or by substitution of chloride 26 under microwaveassisted heating with sodium hydroxide in tetrahydrofuran. A similar microwave procedure was used for the synthesis of quinoxalinone 57 from 27.

Monosubstitution of the 6 position (the 6 position mentioned in the text corresponds to the numbering in Figure 2) of the quinoxaline scaffold was also explored. For this, 2-amino-4-chloroaniline (58) was converted to 6-chloro-quinoxalin-2,3(1*H*,4*H*)-dione (59) and subsequently to its chlorinated product 2,3,6-trichloroquinoxaline (60), as has been described

<sup>&</sup>lt;sup>a</sup> Reagents and conditions: (a) brine with 9% NH<sub>4</sub>OH, Zn, DCM, reflux; (b) N-methylpiperazine, mw, 140°C.

<sup>&</sup>lt;sup>a</sup> Reagents and conditions: (a) R<sub>2</sub>COCO<sub>2</sub>H, EtOH, reflux; (b) ethylox-obutyric acid, r.t. (for compound **15**); (c) POCl<sub>3</sub>, reflux; (d) *N*-methylpiperazine, mw, 140°C.

# Scheme 3<sup>a</sup>

<sup>a</sup> Reagents and conditions: (a) N-methylpiperazine, DIPEA, THF, mw, 160°C; (b) NaH, ROH, DMF, r.t.; (c) NaH, thiophenol, DMF, r.t.; (d) NaH, phenol, DMF, r.t.; (e) NaMeO, MeOH, reflux; (f) benzylamine, mw, 160°C; (g) aq NaOH, THF, or dioxane, mw, 120°C.

in literature (Scheme 4).24 From this precursor, 2,6-dichloro-3-methoxy quinoxaline (61) was synthesized according to a modified procedure from literature.<sup>24</sup> The original procedure mentioned above failed to give 61 in sufficiently high purity. After multiple attempts, purities around 75% were found by NMR due to formation of two chloro-substituted regioisomers. However, three subsequent crystallizations of this impure material yielded desired product (61) in very high purity (>95% by NMR). Subsequent substitution of 61 with N-methylpiperazine resulted in methoxy analogue 62. Attempts to convert 62 to target compound 63 under acidic conditions failed as these conditions led to regeneration of precursor (59). However, demethylation of 62 under basic conditions, using sodium hydroxide in water, proceeded more mildly and gives 63 in good yields. Although compound **61** has been reported in literature, no evidence has been given that can confirm its proposed structure; in this particular synthesis, two regioisomers are formed of which the structures cannot be assigned unambiguously by NMR.<sup>24</sup> Absolute proof with regard to the isolated regioisomers of the target compound quinoxalinone 63, as synthesized following this scheme, could also not be obtained using standard analytical characterization. Because this information is relevant for the SAR, we decided to obtain definite proof of the structure of compound 63 by a fully regioselective synthesis (Scheme 5).

2,5-Dichloro-nitrobenzene (64) was substituted with pmethoxybenzylamine (PMB), yielding benzylamine 65, which was subsequently hydrogenated in THF with Raney nickel and hydrogen to give aniline 66. Ring closure of 66 to N-protected quinoxaline 67 was done by heating in neat diethyloxalate. Compound 67 was chlorinated with POCl<sub>3</sub> in the presence of Hünigs base and DMF to yield 68. Analogous to a procedure described in literature, removal of the PMB group from chloroquinoxaline 68 was achieved with concentrated sulfuric acid to give the desired monochloro-substituted quinoxalinone (69).<sup>25</sup> Quinoxaline 63 was finally obtained regioselectively through this route by substitution of the 2-chloro position of 69 with N-methylpiperazine. A comparison of the melting point, NMR, and LCMS data showed that the compounds obtained with both synthetic routes were indeed identical. Using the same procedure for the preparation of 63 from 69, quinoxalines 70 and 71 were obtained with N-methylhomopiperazine and bocprotected 3-aminomethylpyrrolidine, respectively. Removal of the boc group with HCl/dioxane finally gave compound 72 as the hydrochloric salt.

When **68** was directly substituted with *N*-methylpiperazine, PMB-protected compound 73 was obtained.

Finally, to enable comparison of the SAR of the methylpiperazine moiety with other series of H<sub>4</sub>R ligands, the Nethylpiperazine substituted compound 74 was prepared by the same method as had been used for 20 (Table 1) using intermediate **14** and *N*-ethylpiperazine.

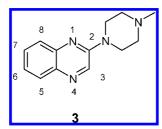
#### **Results and Discussion**

To test our model, we synthesized a series of compounds in which two-ring heterocyclic scaffolds would be directly connected to a N-methylpiperazine group. Relatively small compounds such as quinoxaline 3 would be able to fit in the H<sub>4</sub>R binding site in such a way that the heterocyclic ring could occupy the same space as the chloro-substituted phenyl ring of 1 and the indole ring of 2 while the *N*-methylpiperazine group would be able to adopt a similar conformation as in 1 and 2 (Figure 3A). Furthermore, because the tricyclic scaffold of 1 (e.g., clozapine) and its analogues are known to be very promiscuous GPCR ligands, it was also considered necessary to move away from this particular scaffold. Therefore, the hybrid scaffolds based on the structures of 1 and 2 were considered a good starting point for further exploration of the H<sub>4</sub>R binding site and evaluation of our flexible alignment model. Previous SAR studies of a number of H<sub>4</sub>R ligands showed that modification of the N-methylpiperazine moiety is very detrimental for H<sub>4</sub>R affinity. <sup>16,17</sup> We therefore initially did not alter this moiety in the newly designed compounds. In the initial series, we varied the number and the position of the heterocyclic nitrogen atoms as well as the position of the *N*-methylpiperazine moiety (Table 1). 6-Chloro-substituted compounds 5 and 6 were immediately included in our initial compounds because SAR data from the

Table 2. SAR of Quinoxalines Substituted at the R3 Position

no.	R1	R2	R3	$pK_i \pm SEM^a$
		histamine		$7.92 \pm 0.07$
		thioperamide		$7.20 \pm 0.06$
3	Н	N-methylpiperazine	Н	$6.05 \pm 0.07$
18	H	N-methylpiperazine	$CH_3$	$6.70 \pm 0.02$
19	Н	N-methylpiperazine	$C_6H_5$	$4.99 \pm 0.06$
20	H	N-methylpiperazine	$CH_2C_6H_5$	$7.40 \pm 0.04$
51	H	N-methylpiperazine	SPh	$6.44 \pm 0.02$
55	Н	N-methylpiperazine	NHCH <sub>2</sub> Ph	$5.13 \pm 0.04$
28	Н	N-methylpiperazine	OPh	$6.49 \pm 0.02$
29	Н	<i>N</i> -methylpiperazine	OCH <sub>2</sub> Ph	$6.53 \pm 0.05$
30	Н	<i>N</i> -methylpiperazine	$O(CH_2)_2Ph$	$6.32 \pm 0.03$
31	Н	<i>N</i> -methylpiperazine	$O(CH_2)_4Ph$	$5.53 \pm 0.15$
32	Н	<i>N</i> -methylpiperazine	OCH <sub>2</sub> -2-Pyridyl	$6.33 \pm 0.05$
33	Н	<i>N</i> -methylpiperazine	OCH <sub>2</sub> -3-Pyridyl	$5.53 \pm 0.03$
34	Н	<i>N</i> -methylpiperazine	OCH <sub>2</sub> -4-Pyridyl	$5.36 \pm 0.11$
35	Н	<i>N</i> -methylpiperazine	$OCH_2$ -(4- $OCH_3$ -Ph)	$6.15 \pm 0.40$
36	Н	<i>N</i> -methylpiperazine	OCH <sub>2</sub> -(4-CH <sub>3</sub> -Ph)	$5.86 \pm 0.16$
37	H	<i>N</i> -methylpiperazine	OCH <sub>2</sub> -(4-Cl-Ph)	$5.64 \pm 0.05$
38	H	<i>N</i> -methylpiperazine	OCH <sub>2</sub> -(3-Cl-Ph)	$6.57 \pm 0.05$
39	H	<i>N</i> -methylpiperazine	O(3-pyridyl)	$5.89 \pm 0.03$
40	Н	<i>N</i> -methylpiperazine	O(4-Cl-Ph)	$5.63 \pm 0.06$ $5.63 \pm 0.06$
41	Н	<i>N</i> -methylpiperazine	O(3,4-Cl-Ph)	$5.77 \pm 0.03$
42	H	<i>N</i> -methylpiperazine	O(4-F-Ph)	$5.80 \pm 0.04$
43	H	<i>N</i> -methylpiperazine  N-methylpiperazine	$O(3-CH_3-Ph)$	$6.24 \pm 0.08$
44	H	<i>N</i> -methylpiperazine	O(4-CH <sub>3</sub> -Ph)	$5.66 \pm 0.08$
45	п Н	<i>N</i> -methylpiperazine <i>N</i> -methylpiperazine	O-(4-OCH <sub>3</sub> -Ph)	$5.60 \pm 0.08$ $5.63 \pm 0.20$
46	п Н			
	н Н	N-methylpiperazine	O-(3-N,N-dimethylaniline)	$5.81 \pm 0.02$
47		N-methylpiperazine	O-cyclohexyl	$4.88 \pm 0.03$
48	H	N-methylpiperazine	OCH <sub>2</sub> CH(CH <sub>3</sub> ) <sub>2</sub>	$5.24 \pm 0.03$
49	H	<i>N</i> -methylpiperazine	OCH <sub>2</sub> CH <sub>3</sub>	$6.64 \pm 0.08$
50	H	N-methylpiperazine	$O(CH_2)_3$ - $N$ , $N$ -dimethylamine	$5.40 \pm 0.19$
52	H	N-methylpiperazine	$OCH_3$	$6.47 \pm 0.02$
56	Н	N-methylpiperazine	OH	$7.21 \pm 0.03$
62	6-Cl	<i>N</i> -methylpiperazine	OCH <sub>3</sub>	$7.58 \pm 0.04$
63	6-C1	N-methylpiperazine	OH	$7.93 \pm 0.05$
57	6,7-Cl	N-methylpiperazine	ОН	$8.25 \pm 0.07$
21	6,7-Cl	N-methylpiperazine	$(CH_2)_2Ph$	$5.40 \pm 0.04$
22	6,7-Cl	N-methylpiperazine	$CH_3$	$7.20 \pm 0.09$
52	6,7-Cl	N-methylpiperazine	OPh	$5.93 \pm 0.11$
54	6,7-Cl	N-methylpiperazine	$OCH_3$	$7.24 \pm 0.03$
23	H	N-methylpiperazine	CF <sub>3</sub>	$5.60 \pm 0.02$
26	Н	N-methylpiperazine	Cl	$6.64 \pm 0.03$
73		structure as depicted in S	Scheme 5	$6.32 \pm 0.06$
70	6-Cl	<i>N</i> -methylhomopiperazine	ОН	$6.01 \pm 0.07$
72	6-Cl	aminomethylpyrrolidine	ОН	$5.83 \pm 0.08$
74	Н	<i>N</i> -ethylpiperazine	$CH_2C_6H_5$	$5.57 \pm 0.01$

<sup>&</sup>lt;sup>a</sup> Measured by displacement of [<sup>3</sup>H]histamine binding using membranes of HEK cells transiently expressing the human  $H_4R$ .  $pK_i$ 's are calculated from at least three independent measurements as the mean  $\pm$  SEM.



**Figure 2.** 2-(4-Methylpiperazin-1-yl)quinoxaline (3) as a new lead compound for  $H_4R$  ligands.

reported series of  $H_4R$  ligands indicated that the introduction of a chlorine atom on the all-carbon six-membered ring of several heterocyclic scaffolds could be beneficial for  $H_4R$  affinity. <sup>16–18</sup>

After the synthesis and pharmacological evaluation of the compounds in Table 1, we identified 2-(4-methylpiperazin-1-yl)quinoxaline (3) as a new lead compound for the development

of H<sub>4</sub>R ligands. Fragment 3, together with its chloro analogue 6 has the highest ligand efficiency and thus can be considered the most promising for further optimization.<sup>26</sup>

The introduction of a second nitrogen at the four position of quinoline 4 enhanced the affinity 5-fold, as can be seen in quinoxaline 3. The introduction of a second nitrogen atom at the three position gave quinazoline 9, which showed slightly decreased H<sub>4</sub>R binding properties. As had been proposed before, the introduction of a chlorine atom at the 6-position was beneficial and resulted in an increase in H<sub>4</sub>R affinity of approximately 10-fold in both the quinoline (compare 4 and 5) and quinoxaline (compare 3 and 6) scaffolds. Isoquinoline 7 was inactive at the H<sub>4</sub>R because the methylpiperazine moiety is most likely positioned unfavorably relative to the heterocyclic ring.

The alignment model suggested that substitution of quinoxaline 3 with an additional aromatic ring system could occupy the aromatic pocket, which is also occupied by one of the

# Scheme 4<sup>a</sup>

$$CI \longrightarrow NH_{2}$$

$$SR \longrightarrow SP \longrightarrow GO$$

$$CI \longrightarrow NH_{2}$$

$$SR \longrightarrow SP \longrightarrow GO$$

$$CI \longrightarrow NH_{2}$$

<sup>a</sup> Reagents and conditions: (a) diethyloxalate, reflux; (b) POCl<sub>3</sub>, reflux; (c) NaOMe, MeOH, 50°C; (d) N-methylpiperazine, DIPEA, THF, mw; (e) 5% NaOH, 70°C.

#### Scheme 5<sup>a</sup>

<sup>a</sup> Reagents and conditions: (a) p-methoxybenzylamine, 1-propanol, reflux; (b) Raney Nickel/H<sub>2</sub>, THF, r.t.; (c) diethyloxalate, reflux; (d) POCl<sub>3</sub>, DIPEA, toluene, reflux; (e) NHR<sub>1</sub>R<sub>2</sub>, DIPEA, EtOAC, microwave, 120°C; (f) H<sub>2</sub>SO<sub>4</sub>, r.t.; (g) 2 M HCl in Et<sub>2</sub>O, r.t.

aromatic rings of **1** when it binds to the H<sub>4</sub>R (Figure 3B). Analogues of quinoxaline **3** could be readily synthesized with a variety of substituents at the 3-position. To probe the aromatic pocket, we therefore initially synthesized the methyl (**18**), phenyl (**19**), and benzyl (**20**) analogues of quinoxaline **3**. The methyl analogue showed an almost 10-fold increase in H<sub>4</sub>R affinity compared to **3**, whereas phenyl substitution in **19** is not well

tolerated and leads to a drop in  $H_4R$  affinity. Interestingly, benzyl-substituted analogue **20** bound to the  $H_4R$  with high affinity (33 nM), suggesting that an additional interaction with an aromatic pocket had indeed been found. Therefore, the investigation was extended by introducing various substituents as well as heteroatoms such as oxygen, nitrogen, and sulfur at the 3-position (Table 2).

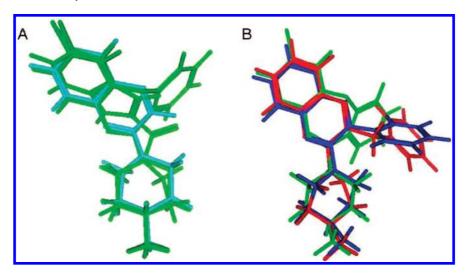


Figure 3. (A) Superposition of 1 and 2 (both in green) with quinoxaline 3 (in light blue) can be seen. Quinoxaline 3 is able to occupy the shared aromatic pockets of 1 and 2, while all three N-methylpiperazine moieties are in the same position. (B) Quinoxalines substituted with a phenyl (compound 19 in blue) and benzyl (compound 20 in red) moiety at the 3-position and their superposition with 1.

The introduction of oxygen (28) or sulfur (51) in the benzyl moiety of 20 was detrimental for H<sub>4</sub>R affinity, but both compounds bind with higher affinity than quinoxaline 3, suggesting that a large number of substituents at the 3-position indeed enhance H<sub>4</sub>R affinity.

Benzylamine analogue 55 did not improve binding, whereas the benzyl alcohol substituted analogue 29 showed about a 5-fold enhancement of affinity. The introduction of extra carbon atoms in the alkoxy spacer of 29 showed that two carbons (compare 29 and 30) were tolerated but four carbons (compare 29 and 31) were detrimental for H<sub>4</sub>R binding. Because compound 29 had quite reasonable H<sub>4</sub>R affinity, we investigated whether the properties of the aromatic ring of the benzyl alcohol moiety could be optimized for H<sub>4</sub>R interaction. 2-Pyridyl analogue 32 had similar affinity to 29, whereas a nitrogen in the 3- or 4-position of the pyridine ring (compounds 33 and 34) decreased the  $H_4R$  affinity. Analogues 35–37 with  $-OCH_3$ , -CH<sub>3</sub>, or chlorine substituents did not show improved H<sub>4</sub>R affinity.

Parallel to the synthesis of the benzyl alcohols, we synthesized a series of pyridinol and phenol-substituted compounds based on 28 in which we also introduced a number of different substituents. In these series (compounds 39-46), none of the synthesized compounds showed an affinity that was higher than **28**.

Several quinoxalines were synthesized with aliphatic ethers that enhanced the binding affinity of 3 only slightly when substituted with an ethoxy (49) or methoxy (52) group. However, when the size of these substituents was increased to isobutoxy (48) or cyclohexyloxy (47), the affinity dropped up to 10-fold for compound 47. The introduction of a flexible aliphatic side chain with a basic amine (compound 50) at the 3-position of quinoxaline 3 was not tolerated.

More interesting, quinoxalinone 56 had an affinity almost comparable to benzylquinoxaline 20. It was anticipated that the introduction of a chlorine atom on the heterocyclic core would improve the H<sub>4</sub>R affinity after having optimized the R<sub>2</sub> position as had been the case in other series. 16,17 By this rationale, compounds 62 and 63 were synthesized. The 6-chloro substituent gave an enhancement in H<sub>4</sub>R affinity of 14- and 5-fold in compounds **62** and **63**, respectively. We also synthesized 6,7dichloro-substituted quinoxaline 57 and found that it was even more potent, showing an H<sub>4</sub>R affinity of 6 nM.

To see if the introduction of two chlorine atoms on the quinoxaline heterocycle would give a linear SAR, we synthesized phenethyl-, methyl-, phenoxy- and methoxy analogues 21, 22, 52, and 54. As can be seen in Table 2, two chlorine substituents increased the affinity of the methoxy (52) and methyl (22) substituted quinoxalines but not of the phenylethyl (21) and phenoxy (22) substituted quinoxalines. In the last two compounds, this modulation appeared to even be quite detrimental for H<sub>4</sub>R affinity. During the course of these investigations, a series of substituted quinoxalinones was reported by Johnson and Johnson scientists in patent literature as new and very potent H<sub>4</sub>R antagonists. Although only a small series of compounds is reported, quinoxalines 63 and 57 are also reported as H<sub>4</sub>R antagonists with nanomolar affinities (31 and 32 nM, respectively).<sup>27</sup>

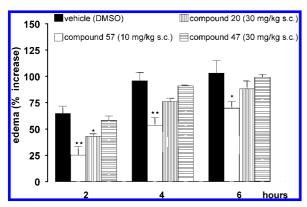
The SAR of the methylpiperazine group was studied to a limited extent because even minor changes such as the replacement of the methyl group by an ethyl group (compare 20 and 74) is very detrimental for H<sub>4</sub>R affinity. The sensitivity to chemical modulation of the N-methylpiperazine group was seen before in benzimidazole, indole, and dibenzo-oxazepine series that were reported as H<sub>4</sub>R ligands. 17,16,28 However, patent literature reports that in a series of aminopyrimidines and benzofuropyrimidines, the N-methylpiperazine moiety can be successfully replaced with 3-aminomethylpyrollidine and several other bioisosteres. 29,30 In the quinoxaline series, attempts to find a suitable bioisostere were unsuccesfull because homopiperazine 70 and aminomethylpyrollidine 72 were found to have affinities that were about 100-fold lower than N-methylpiperazine 63.

For further pharmacological characterization, quinoxalines 20 and 57 were selected from Table 2. The affinity for the other histaminergic receptors was determined in order to assess the selectivity of these ligands for the H<sub>4</sub>R over the other histamine receptor subtypes (Table 3). Although compound 57 showed limited inhibition of radioligand binding to the other histaminergic receptors, at a concentration of 10  $\mu$ M, compound 20 effectively displaces [<sup>3</sup>H]mepyramine binding to the H<sub>1</sub>R. Compound 20 was designed with a working model that made use of compound 1 (Figure 3), which had been demonstrated to have a high affinity (p $K_i = 8.11 \pm 0.10$ ) for the H<sub>1</sub>R.<sup>16</sup> Therefore, we were prompted to determine the  $pK_i$  at the  $H_1R$ , which was found to be  $6.13 \pm 0.1$  and showed that 20 also possessed some affinity for the H<sub>1</sub>R. As was already pointed

**Table 3.** Pharmacological Evaluation of Selected Quinoxalines at the Four Human Histaminergic Receptor Subtypes

compound	H <sub>1</sub> <sup>a</sup> (% inhibition)	$\begin{array}{c} H_2 \\ (\% \ inhibition) \end{array}$	H <sub>3</sub> (% inhibition)	$H_4^b$ (p $K_i \pm SEM$ )
20	86 <sup>c</sup> 7	62	26	$7.40 \pm 0.04$
57		42	64	$8.25 \pm 0.07$

 $^a$ % Inhibition of radioligand binding to the human H<sub>1</sub>-H<sub>3</sub> receptors was determined in duplo at a concentration of 10  $\mu$ M of ligand.  $^b$  Values are taken from Table 2.  $^c$  The pK<sub>i</sub> of **20** at the human H<sub>1</sub>R was determined to be 6.13  $\pm$  0.1 (n = 3).



**Figure 4.** Anti-inflammatory effect of compounds **20** and **57** and inactive control **47** on paw edema induced by subplantar injection of carrageenan (1% in CMC) in rats. Data are expressed as mean  $\pm$  SEM n=6 rats per group. Comparisons between multiple groups were made by using one-way analysis of variance (ANOVA), followed by Dunnett's test. \*P < 0.05 compared with vehicle-treated animals (Student t test for grouped data).

out earlier, it can be envisioned that dual action  $H_1/H_4$  ligands might have great potential in the treatment of several histamine-mediated allergic and inflammatory disorders.<sup>31</sup>

Compound **20** and **57** were selected to study the anti-inflammatory effect on the new series of  $H_4R$  ligands in vivo. After 2 h both compounds, respectively, coadministered at 10 mg/kg and 30 mg/kg with carrageenan, caused a significant inhibition of carrageenan-induced edema when compared to the vehicle (Figure 4). The effect of **57** was still found to be significant 6 h after administration. Compound **47** (p $K_i = 4.88 \pm 0.03$ ), which was selected for this study as an inactive control, displayed no significant inhibition of the carrageenan-induced edema.

# Conclusion

Albeit limited by the available GPCR protein structural information and the complications when applying biophysical screening approaches to these targets, the growing of a micromolar affinity fragment into nanomolar ligands can be considered a fragment based design approach that has been successfully used in many GPCR projects before.<sup>2,33</sup> In this work, we have described the discovery of a series of new histamine H<sub>4</sub>R ligands starting from an in silico flexible alignment model of the known H<sub>4</sub>R agonist 1 and antagonist 2. It was used to design a series of small heterocyclic fragments as H<sub>4</sub> receptor–ligands, of which 2-(4-methylpiperazin-1-yl)quinoxaline (3) binds to the  $H_4R$  with a  $K_i$  in the submicromolar range. The growing of fragments into more druglike compounds by introducing a variety of substituents onto this new scaffold was guided by the in silico model, thereby strengthening the rational design. This approach led to a series of potent H<sub>4</sub>R ligands with affinities in the low nanomolar range. Two of these compounds, 6,7dichloro-3-(4-methylpiperazin-1-yl)quinoxalin-2(1*H*)-one (57) and 2-benzyl-3-(4-methyl-piperazin-1-yl)quinoxaline (20) were evaluated in vivo and displayed significant anti-inflammatory activity in the carrageenan induced paw edema model in rats.

## **Experimental Section**

In Vivo Pharmacology. Carrageenan-Induced Edema Model. Animals. Male Wistar rats (180–200 g; Harlan-Italy, Milan) were housed under controlled standard conditions (23 °C temperature, 12 h light/dark cycle, and 65% humidity). Food and water were provided ad libitum. The experiments received the approval of the local Animal Ethics Committee of the University of Parma, Italy.

**Induction of Acute Inflammation.** Inflammation was induced in fasted rats by subplantar injection of carrageenan (0.1 mL of 1% suspension in carboxymethycellulose) into the left hind paw. As previously described, carrageenan-induced edema was measured with a plethysmometer (Basile, Comerio, Italy) immediately prior to the injection of carrageenan and thereafter at 2, 4, and 6 h. <sup>33</sup> Edema was expressed for each animal as % increase in paw volume after carrageenan injection relative to the preinjection value, considered as 100.

**Compound Administration.** Equivalent volumes (0.1 mL per 100 g) of the test compounds or vehicle were administered subcutaneously (s.c.) in separate groups of rats immediately prior to carrageenan injection. Before use solutions of the histamine H<sub>4</sub> receptor–ligands in DMSO were freshly prepared.

**Data Analysis.** Data are expressed as mean  $\pm$  SEM. A value of P < 0.05 was considered statistically significant. The software package Prism GraphPad 3.0 (GraphPad Software Inc., San Diego, CA) was used to process data.

In Vitro Pharmacology. Radioligand Displacement Studies at the Human  $H_1$ ,  $H_2$ , and  $H_3$  Receptors. Single-point radioligand displacement studies at the human  $H_1$ ,  $H_2$ , and  $H_3$  receptors were preformed by Cerep (Le Bois l'Evêque, France) at a concentration of  $10 \, \mu M$  of ligand. These measurements were performed in duplo. The  $pK_i$  at the  $H_1R$  was determined according to a procedure described in literature.<sup>34</sup>

Radioligand Displacement Studies at the Human H<sub>4</sub> Receptor. Cell Culture and Transfection. HEK 293T cells were maintained in Dulbecco's modified Eagle medium (DMEM) supplemented with 10% fetal bovine serum (FBS), 50 IU/mL penicillin, and 50 µg/mL streptomycin in 5% CO<sub>2</sub> humidified atmosphere at 37 °C. Approximately 4 million cells were seeded in a 10 cm dish and cultured overnight before transfection. For transfection of each dish of cells, the transfection mixture was prepared in 1 mL serum-free DMEM and contained 5  $\mu$ g of human H<sub>4</sub>R receptor plasmid and 15 μL of 1 mg/mL 25 kDa linear polyethyleneimine (Polyscience, Inc., USA). The mixture was incubated for 10-15 min at room temperature before it was added into the monoloyer cell culture loaded with 5 mL of fresh cell culture medium. Two days after transfection, the cells were washed with PBS containing 1 mM EDTA, collected as pellet by centrifuging, and stored at -20 °C until use.

[³H] Histamine Binding Assay. For the radioligand binding study, pellets of transfected cells were homogenized in  $H_4R$  binding buffer (100 mM Tris-HCl, pH 7.4). The saturation binding assay was performed using different concentrations of [³H]histamine (Perkin-Elmer Life Science, Inc., USA), while nonspecific binding was determined by incubation in the presence of 3–10  $\mu$ M of compound 2 in a total assay volume of 200  $\mu$ L. For the displacement binding assay, the membranes were typically incubated with  $10^{-4}-10^{-11}$  M of ligands (stock concentration was 10 mM 1 DMSO) in the presence of [³H]histamine in a total volume of 200  $\mu$ L. The reaction mixtures were incubated for 1 h at room temperature (22 °C), and harvested on 96-well glass fiber C plates that were pretreated with 0.3% 750 kDa PEI. The binding assay data were analyzed using Prism 4.0 (Graphpad Software Inc., USA).

**Flexible Alignment.** The alignment model was created with Molecular Operating Environment 2006.08 (MOE) from Chemical Computing Group (Montreal, Canada). The compounds in Figure 3 were aligned with the flexible alignment module of MOE using

default parameters and similarity term "partial charge" added with a weight factor of 1. The alignment with highest mutual similarity score F (F = 32.1926, S = 159.991, U = 127.7983) was selected and refined further using the "refine existing alignment" option with an energy cutoff of 7.0 kcal/mol.

General Remarks. Carrageenan was purchased from Sigma-Aldrich (St. Louis, MO). Other chemicals and reagents were obtained from commercial suppliers and were used without further purification. Yields given are isolated yields unless mentioned otherwise. Flash column chromatography was typically carried out on an Argonaut Flashmaster II flash chromatography system, using prepacked Isolute Flash Si II columns with the UV detector operating at 254 nm. All melting points are uncorrected and were measured on an Optimelt automated melting point system from Stanford Research Systems. All <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were measured on a Brüker 200 or Brüker 400.

Analytical HPLC-MS analyses were conducted using a Shimadzu LC-8A preparative liquid chromatograph pump system with a Shimadzu SPD-10AV UV-vis detector with the MS detection performed with a Shimadzu LCMS-2010 liquid chromatograph—mass spectrometer. The buffer mentioned under conditions I and II is a 0.4% (w/v) NH<sub>4</sub>CO<sub>3</sub> solution in water, adjusted to pH 8.0 with NH<sub>4</sub>OH. The analyses were performed using the following two conditions:

Condition I: an Xbridge (C18)5 $\mu$ m column (100 mm  $\times$  4.6 mm) with the following two solvents: solvent A, 90% MeCN-10%; solvent B, 90% water-10% buffer; flow rate = 2.0 mL/min; start: 5% A, linear gradient to 90% A in 10 min, then 10 min at 90% A, then 10 min at 5% A. Total run time: 30 min.

Condition II: Xbridge (C18)5 $\mu$ m column (100 mm × 4.6 mm) with the following two solvents: solvent A, MeCN with 0.1% formic acid; solvent B, water with 0.1% formic acid; flow rate = 2.0 mL/min; start: 5% A, linear gradient to 90% A in 10 min, then 5 min at 90% A, then 5 min at 5% A. Total run time: 20 min. Compound purities under both conditions were calculated as the percentage peak area of the analyzed compound by UV detection at 254 nm.

Analytical HPLC-MS analyses for condition III were carried out on Agilent 1100 Series HPLC-MS system including an Agilent G1315B DAD. Condition III: XBridge (C18)3.5 $\mu$ m column (2.1 mm  $\times$  50 mm) with the following two solvents: solvent A, a 5 mM solution of NH<sub>4</sub>HCO<sub>3</sub> in water set to pH 9.0 using 19 mM NH<sub>3</sub>; solvent B, 100% MeCN. The MS detection was performed with an Agilent G1956B LC/MSD SL using a multimode ion source. Flow rate = 1.2 mL/min; start: 95% A, linear gradient to 5% A in 1.25 min, then 0.75 min at 5% A, followed by 1.0 min 95% A, 5% B. Total run time: 3 min.

**Synthetic Methods. 2-(4-Methylpiperazin-1-yl)quinoxaline** (3). 2-Chloroquinoxaline (309 mg, 1.88 mmol) and *N*-methylpiperazine (2.0 mL) were added to a microwave tube and heated at 160 °C for 5 min. The resulting mixture was diluted with water and extracted with EtOAc. Drying over Na<sub>2</sub>SO<sub>4</sub> and evaporation of the solvent yielded 402 mg (1.76 mmol, 94%) of a brown solid. Mp 108.6–110.9 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ (ppm) 8.56 (s, 1H), 7.85 (dd, J = 8.2 Hz, J = 1.3 Hz, 1H), 7.66 (dd, J = 8.4 Hz, J = 1.3 Hz, 1H), 7.59–7.50 (m, 1H), 7.41–7.24 (m, 1H), 3.79 (t, J = 5.1 Hz, 4H), 2.54 (t, J = 5.1 Hz, 4H), 2.34 (s, 3H); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ (ppm) 152.09, 141.46, 136.66, 135.55, 129.92, 128.50, 126.32, 124.60, 54.61, 46.07, 44.48.

**2-(4-Methylpiperazin-1-yl)quinazoline (9).** A solution of 2,4-dichloroquinazoline (1.2 g, 6.03 mmol) in DCM (24 mL) covered with brine (24 mL) containing 9% NH<sub>4</sub>OH was treated with powdered zinc (1.2 g), and the resulting biphasic mixture was then refluxed for 3.5 h. After cooling to r.t., it was filtered through celite and the organic layer was removed under reduced pressure. The residue was diluted with EtOAc and washed with 1 N HCl solution, after which it was dried and concentrated to dryness. Without purification, *N*-methylpiperazine (4.0 mL) was added, causing a crystalline solid to separate almost immediately. The product was diluted with ethyl acetate and washed with NaHCO<sub>3</sub> and water. Drying over Na<sub>2</sub>SO<sub>4</sub> and rotary evaporation yielded the crude title compound. Purification over SiO<sub>2</sub> yielded 180 mg (0.79 mmol,

13%) of a light-orange solid; mp 74.4–76.0 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 8.81 (s, 1H), 7.52–7.40 (m, 3H), 7.04–6.96 (m, 1H), 3.85 (t, J = 5.0 Hz, 4H), 2.34 (t, J = 5.0 Hz, 4H), 2.17 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 161.05, 158.86, 151.98, 133.68, 127.06, 125.41, 122.13, 119.32, 54.78, 46.01, 43.69; MS (ESI) m/z 229 (M + H)<sup>+</sup>.

2-Benzyl-3-(4-methylpiperazin-1-yl)quinoxaline (20). 3-Benzylquinoxalin-2(1*H*)-one (**14**) (998 mg, 4.2 mmol) and POCl<sub>3</sub> (6.0 mL) were added to a round-bottom flask and heated at reflux. After 5 h, the mixture was allowed to cool to r.t. and was poured onto crushed ice. Concentrated ammonia was then used to adjust the mixture to pH 6, and the obtained suspension was left to stir vigorously overnight at r.t. After filtration of the suspension and washing of the obtained solids with water, the product was dried in vacuo yielding 1.06 g (41.6 mmol, 99%) of 2-benzyl-3chloroquinoxaline as a white solid. This was used directly in the next step without further purification. A microwave tube charged with 2-benzyl-3-chloroquinoxaline (400 mg, 1.57 mmol) and N-methylpiperazine (3.0 mL) was heated at 120 °C for 5 min. Then water (25 mL) was added and the aqueous layer was extracted with EtOAc. The combined organic extracts were dried with brine and Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the organic solvents and subsequent purification over SiO<sub>2</sub> (EtOAc 90%, Et<sub>3</sub>N 5%, MeOH 5%) yielded 446 mg (1.40 mmol, 89%) of a yellow solid; mp 89.0–90.0 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.82 (dd, J = 1.4 Hz, J = 8.3 Hz, 1H), 7.62-7.46 (m, 2H), 7.34-7.14 (m, 5H), 4.35 (s, 2H), 3.32 (t, J =4.8 Hz, 4H), 2.57 (t, J = 4.8 Hz), 2.35 (s, 3H). <sup>13</sup>C NMR (DMSO $d_6$ )  $\delta$  (ppm): 155.89, 151.33, 140,04, 138.80, 138.24, 128.97, 128.63, 128.29, 128.16, 127.18, 126.69, 126.31, 54.75, 49.59, 45.98, 40.71; MS (ESI) m/z 319 (M + H)<sup>+</sup>.

2-Chloro-3-(4-methylpiperazin-1-yl)quinoxaline (26). To a solution of 2,3-dichloroquinoxaline (6.60 g, 33.2 mmol) in toluene (45 mL) was added triethylamine (4.6 mL, 33.0 mmol) and the solution was heated at reflux. N-methylpiperazine (3.7 mL, 33.4 mmol) was added and heating was continued for 2 h. The resulting mixture was diluted with saturated NaHCO<sub>3</sub> solution. The organic layer was separated, and the aqueous layer was extracted with ethyl acetate. The combined organic layers were washed with brine, dried over MgSO<sub>4</sub>, and concentrated in vacuo. Purification of the residue over SiO<sub>2</sub> (90% EtOAc, 5% Et<sub>3</sub>N, 5% MeOH) (EtOAc) yielded 7.70 g (29.3 mmol, 88%) of a yellow solid; mp 67.4–68.6 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.87–7.78 (m, 2H), 7.65–7.47 (m, 2H), 3.60 (t, J = 4.9 Hz, 4H), 2.63 (t, J = 4.9 Hz, 4H), 2.37 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 152.38, 141.46, 139.99, 138.03, 129.94, 127.54, 127.13, 126.87, 54.55, 48.77, 45.96. MS (ESI) m/z  $263 (M + H)^{+}$ 

**2,6,7-Trichloro-3-(4-methylpiperazin-1-yl)quinoxaline** (27). A microwave tube charged with 2,3,6,7-tetrachloro-quinoxaline (500 mg, 1.87 mmol), *N*-methylpiperazine (0.21 mL, 1.89 mmol), DIPEA (0.37 mL, 2.12), and THF (3.0 mL) was heated at 140 °C. After 5 min, the reaction mixture was diluted with water and extracted with EtOAc. The combined organic layers were washed with brine and dried over Na<sub>2</sub>SO<sub>4</sub>. The extracts were concentrated and filtered over a short column of SiO<sub>2</sub> (EtOAc as eluent) Evaporation of the solvent yielded 523 mg (1.58 mmol, 84%) of the title compound. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.90 (s, 1H), 7.87 (s, 1H), 3.61 (t, J = 4.8 Hz, 1H), 2.62 (t, J = 4.8 Hz, 1H), 2.36 (s, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 152.66, 142.30, 138.93, 136.46, 134.28, 130.89, 128.05, 127.42, 54.22, 48.60, 45.89.

**2-(4-Methylpiperazin-1-yl)-3-phenoxyquinoxaline (28).** To a round-bottom flask were added 2-chloro-3-(4-methylpiperazin-1-yl)quinoxaline (**26**) (1.0 g, 3.81 mmol), phenol (0.43 g, 4.57 mmol), DMF (3.0 mL), and NaOH (0.23 g, 5.75 mmol). The resulting suspension was stirred at r.t. for 2 h. After completion, the mixture was diluted with water and the aqueous layer was extracted with EtOAc. The combined organic extracts were thoroughly washed with water and brine and then dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the solvent gave an oil that was purified over SiO<sub>2</sub> (90% EtOAc, 5% Et<sub>3</sub>N, 5% MeOH) yielding 0.89 g (2.78 mmol, 73%) of a yellow solid; mp 96.8–97.6 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.73 (dd, J = 8.1 Hz, J = 1.2 Hz, 1H), 7.56–7.20 (m, 8H), 3.82 (t, J = 4.9 Hz,

4H), 2.61 (t, J = 4.9 Hz, 4H), 2.36 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 152.61, 149.95, 147.02, 138.95, 135.79, 129.37, 127.18, 126.49, 126.00, 125.62, 125.03, 121.60, 54.88, 47.67, 46.04; MS (ESI) m/z 321 (M + H)<sup>+</sup>.

2-Ethoxy-3-(4-methylpiperazin-1-yl)quinoxaline (49). To 2chloro-3-(4-methylpiperazin-1-yl)quinoxaline (26) (355 mg, 1.28 mmol) in ethanol (5.0 mL) was added NaOH (54 mg, 1.35 mmol, in 1.0 mL water) and the obtained solution was heated at reflux for 2 h. After cooling to room temperature, the solution was added to a separatory funnel containing EtOAc and water and the aqueous phase was extracted with EtOAc. The combined organic phases were washed with water and brine and dried over MgSO<sub>4</sub>. Purification over SiO<sub>2</sub> (EtOAc 90%, MeOH 5%, Et<sub>3</sub>N 5%) yielded 214 mg (0.79 mmol, 61%) of the title compound as a yellow oil. As a side-product, 30 mg (0.12 mmol, 9%) of compound 56 was also isolated. Analytical data for 49: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.70-7.62 (m, 2H), 7.44-7.30 (m, 2H), 4.54 (q, J = 7.1 Hz, 2H), 3.73 (t, J = 4.9 Hz, 4H), 2.57 (t, J = 5.0 Hz, 4H), 2.34 (s, 3H), 1.46 (t, J = 7.0 Hz, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 150.34, 146.87, 138.04, 136.36, 126.13, 126.01, 125.81, 125.38, 62.25, 54.81, 47.46, 46.01, 14.23; MS (ESI) m/z 273 (M + H)<sup>+</sup>

**2-Methoxy-3-(4-methylpiperazin-1-yl)quinoxaline (52).** 2-Chloro-3-(4-methylpiperazin-1-yl)quinoxaline **(26)** (1.0 g, 3.81 mmol) and a 5.4 M solution of NaOMe in MeOH (0.9 mL, 4.86 mmol) were dissolved in MeOH (20 mL) and heated at reflux. After 3 h, the obtained mixture was cooled to r.t. and diluted with EtOAc (60 mL). The organic layer was then washed with water and brine and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the organics yielded 932 mg (3.61 mmol, 95%) of a light-brown solid; mp 74.0–76.3 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.71–7.65 (m, 2H), 7.45–7.33 (m, 2H), 4.09 (s, 3H), 3.71 (t, J = 4.9 Hz, 4H), 2.57 (t, J = 5.0 Hz, 4H), 2.34 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 150.77, 146.95, 138.14, 136.36, 126.30, 126.07, 125.85, 125.51, 54.87, 53.52, 47.50, 46.05; MS (ESI) m/z 259 (M + H)<sup>+</sup>.

3-(4-Methylpiperazin-1-yl)quinoxalin-2(1*H*)-one (56). Method 1: Synthesis from 2-Chloro-3-(4-methylpiperazin-1-yl)quinoxaline. A microwave tube containing 2-chloro-3-(4-methylpiperazin-1-yl)quinoxaline (26) (150 mg, 0.57 mmol), NaOH (60 mg, 1.5 mmol), THF (2.0 mL), and water (0.5 mL) was heated at 120 °C for 20 min. After completion, the solution was partitioned between EtOAc and water and the aqueous layer was extracted with EtOAc. After drying over Na<sub>2</sub>SO<sub>4</sub>, the solvent was evaporated and the residue was purified over SiO<sub>2</sub> (EtOAc 90%, Et<sub>3</sub>N 5%, MeOH 5%) yielding 100 mg (0.41 mmol, 72%) of a white solid. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 11.29 (bs, 1H), 7.53–7.48 (m, 1H), 7.24–7.08 (m, 3H), 4.03 (t, J = 4.9 Hz, 4H), 2.58 (t, J = 5.0– Hz, 4H), 2.34 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 153.41, 150.74, 133.05, 128.32, 125.75, 124.91, 124.02, 114.21, 54.95, 46.50, 45.99; MS (ESI) m/z 245 (M + H)<sup>+</sup>.

Method 2: Synthesis from 2-Methoxy-3-(4-methylpiperazin-1-yl)quinoxaline. 2-Methoxy-3-(4-methylpiperazin-1-yl)-quinoxaline (52) (100 mg, 0.39 mmol) and 5% NaOH in water (5.0 mL) were heated at reflux for 20 h. After cooling to room temperature, the pH was adjusted to 7.0 with diluted HCl. The aqueous solution was extracted with DCM, and the combined extracts were washed with brine. Drying over Na<sub>2</sub>SO<sub>4</sub> and evaporation of the solvent yielded 84 mg (0.34 mmol, 88%) of a white solid. An analytical sample was recrystallized from EtOAc to yield white needles; mp 206.4–207.8 °C.

**6,7-Dichloro-3-(4-methylpiperazin-1-yl)quinoxalin-2(1H)-one (57).** To a microwave tube were added 2,6,7-trichloro-3-(4-methylpiperazin-1-yl)-quinoxaline (**26**) (212 mg, 0.64 mmol), dioxane (3.0 mL), NaOH (52 mg, 1.30 mmol), and water (1.0 mL). The mixture was heated in the microwave for 5 min at 120 °C and diluted with water. The obtained suspension was thoroughly extracted with DCM, and the organic layers were dried with Na<sub>2</sub>SO<sub>4</sub>. The residue that was obtained after evaporation of the solvent was triturated with EtOAc to yield 114 mg (0.36 mmol, 57%) of an off-white solid; mp 262.4–263.6 °C. <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  (ppm): 7,52 (s, 1H), 7.26 (s, 1H), 3.92 (m, 4H), 2.41 (t, J = 4.9 Hz, 4H), 2.19 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 151.51,

150.96, 132.27, 128.87, 125.20, 125.20, 124.43, 114.87, 54.32, 45.70, 45.42; MS (ESI) 313 m/z (M + H) $^+$ .

**2,6-Dichloro-3-methoxyquinoxaline (61).** A 5.4 M solution of NaOMe in methanol (2.9 mL, 15.7 mmol) was diluted with methanol (36 mL) and added over a 6 h period to a slurry of 2,3,6-trichloroquinoxaline (3.0 g, 12.8 mmol) in methanol (36 mL) at 50 °C. After addition, the obtained solution was refluxed for 16 h and the solvent was evaporated. The residue was taken up in CHCl<sub>3</sub> (100 mL) and washed with water and brine. After drying over Na<sub>2</sub>SO<sub>4</sub>, the organic layer was removed and the residue was purified over SiO<sub>2</sub> (toluene) to yield a white solid. After three crystallizations, 369 mg (1.61 mmol, 13%) of the title compound was obtained as white needles (purity > 95% by NMR); mp 110.9–111.4 °C. ¹H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.86–7.82 (m, 2H), 7.51 (dd, J = 8.8 Hz, J = 2.2 Hz, 1H), 4.14 (s, 3H).

**6-Chloro-3-methoxy-2-(4-methylpiperazin-1-yl)quinoxaline** (**62**). To a microwave tube were added 2,6-dichloro-3-methoxy-quinoxaline (**61**) (200 mg, 0.87 mmol), *N*-methylpiperazine (0.2 mL, 1.80 mmol), and THF (3.0 mL). After heating the solution for 5 min at 140 °C, the obtained suspension was diluted with EtOAc (30 mL) and washed with saturated NaHCO<sub>3</sub>, water, and brine. Drying with Na<sub>2</sub>SO<sub>4</sub> and evaporation of the organic layer yielded an off-white solid. Purification over SiO<sub>2</sub> (EtOAc 90%, MeOH 5%, Et<sub>3</sub>N 5%) yielded 168 mg (0.57 mmol, 66%) of a white-yellow solid; mp 99.9–101.6 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ (ppm): 7.66 (d, *J* = 2.3 Hz, 1H), 7.59 (d, *J* = 8.7 Hz, 1H), 7.34 (dd, *J* = 8.8 Hz, *J* = 2.4 Hz, 1H), 4.01 (s, 3H), 3.71 (t, *J* = 5.0 Hz, 4H), 2.56 (t, *J* = 5.0 Hz, 4H), 2.34 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ (ppm): 151.24, 146.83, 136.80, 136.73, 130.49, 127.02, 126.85, 125.09, 54.85, 53.72, 47.42, 46.03; MS (ESI) *mlz* 293 (M + H)<sup>+</sup>.

6-Chloro-3-(4-methylpiperazin-1-yl)quinoxalin-2(1*H*)-one (63). Method 1: Synthesis from 6-Chloro-3-methoxy-3-(4-methylpiperazin-1-yl)quinoxaline (62). 6-Chloro-3-methoxy-3-(4-methylpiperazin-1-yl)quinoxaline (62) (135 mg, 0.46 mmol) and 5% aqueous NaOH (10 mL) were heated at 70 °C. After 4 h, the clear solution was neutralized (pH 7) with diluted HCl and extracted with DCM. The combined extracts were then washed with brine. After drying over Na<sub>2</sub>SO<sub>4</sub> and evaporation of the organics, the residue was purified over SiO<sub>2</sub> (EtOAc 90%, MeOH 5%, Et<sub>3</sub>N 5%) to yield 35 mg (0.13 mmol, 28%) of the title compound as a white solid; mp 228.5–230.0 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 11.23 (bs, 1H), 7.40 (d, J = 8.5 Hz, 1H), 7.14 (dd, J = 8.6 Hz, J = 2.2 Hz, 1H), 7.03(d, J = 2.2 Hz, 1H), 4.03 (t, J = 4.8 Hz, 4H), 2.58 (t, J = 5.0 Hz,4H), 2.35 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 153.32, 150.36, 131.74, 129.94, 128.91, 126.74, 124.40, 113.91, 54.89, 46.47, 45.93; MS (ESI) m/z 279 (M + H)<sup>+</sup>.

Method 2: Synthesis from 3,7-Dichloroquinoxalin-2(1*H*)-one (69). *N*-Methylpiperazine (1.0 mL), 3,7-dichloro-quinoxalin-2(1*H*)-one (69) (300 mg, 1.40 mmol), and EtOAc (2.0 mL) were added to a microwave tube and were heated at 120 °C for 10 min. After cooling to r.t., the mixture was diluted with EtOAc and washed with water and brine. Drying of the organic phase over Na<sub>2</sub>SO<sub>4</sub> and evaporation of the solvent gave the crude product that was recrystallized from EtOAc to yield 227 mg (0.81 mmol, 58%) of off-white crystals; mp 230.0–232.0 °C. ¹H NMR (CDCl<sub>3</sub>) δ (ppm): 11.01 (bs, 1H), 7.36 (d, J = 8.6 Hz, 1H), 7.10 (dd, J = 8.6 Hz, J = 2.2 Hz, 1H), 7.03 (d, J = 2.1 Hz, 1H), 3.98 (t, J = 4.9 Hz, 4H), 2.52 (t, J = 5.0 Hz, 4H), 2.30 (s, 3H). ¹³C NMR (CDCl<sub>3</sub>) δ (ppm): 153.26, 150.38, 131.75, 129.91, 128.91, 126.73, 124.38, 113.88, 54.90, 46.49, 45.94; MS (ESI) m/z 279 (M + H)+.

**5-Chloro-***N***-(4-methoxybenzyl)-2-nitroaniline (65).** A solution of 3,4-dichloro-nitrobenzene (37.0 g, 0.19 mol) and 4-methoxybenzylamine (50 mL, 0.38 mol) in *n*-propanol (200 mL) was heated at reflux for 16 h, after which the solvent was removed under reduced pressure. The residue was taken up in EtOAc and washed with water and brine. Drying of the organic phase over Na<sub>2</sub>SO<sub>4</sub> and removal of the solvent yielded a yellow solid that was recrystallized from abs EtOH to yield 42.6 g (0.15 mol, 76%) of yellow crystals; mp 70.0–72.0 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 8.33 (br s, 1H), 8.11 (d, J = 9.1 Hz, 1H), 7.27–7.23 (m, 2H), 6.89 (d,

J = 8.7 Hz, 2H), 6.82 (d, J = 2.1 Hz, 1H), 6.61 (dd, J = 9.1 Hz, J = 2.1 Hz, 1H), 4.41 (d, J = 5.4 Hz, 2H), 3.80 (s, 3H).

**5-Chloro-***N***'-(4-methoxybenzyl)benzene-1,2-diamine (65).** To a solution of **64** (8.7 g, 29.7 mmol) in THF (150 mL) in a round-bottom flask was added Raney nickel (2.0 g of a 50% suspension in water) and the resulting mixture was stirred vigorously. Hydrogen gas was added with a balloon until TLC indicated complete conversion of the starting material (16 h). The catalyst was removed by filtration, and the reaction mixture was evaporated to dryness, yielding 7.74 g (29.5 mmol, 98%) of a dark-brown solid that was used in the next step without further purification. <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  (ppm): 7.27 (d, J = 8.7 Hz, 2H), 6.89 (d, J = 8.7 Hz, 2H), 6.50 (d, J = 8.2 Hz, 1H), 6.37 (dd, J = 8.1 Hz, J = 2.2 Hz, 1H), 6.26 (d, J = 2.3 Hz, 1H), 5.33 (m, 1H), 4.71 (s, 2H), 4.21 (d, J = 6.1 Hz, 2H), 3.73 (s, 3H).

**7-Chloro-1-(4-methoxybenzyl)quinoxaline-2,3(1***H***,4***H***)-dione (67). Crude 66 (7.7 g, 0.029 mol) and diethyloxalate (16.0 mL, 154 mmol) were heated at 140 °C for 16 h. After completion, the reaction mixture was carefully cooled to r.t. upon which the desired product crystallized from the reaction flask. The product was filtered off and washed with cold EtOH to yield 8.35 g (26.4 mmol, 91%) of the title compound as white crystals; mp 236.0–238.0 °C. ¹H NMR (DMSO-d\_6): \delta (ppm): 12.18 (br s, 1H), 7.28–7.13 (m, 5H), 6.88 (d, J = 8.7 Hz, 2H), 5.30 (s, 2H), 3.71 (s, 3H). ¹³C NMR (DMSO-d\_6) \delta (ppm): 158.25, 155.39, 153.17, 127.87, 127.27, 126.96, 126.58, 124.84, 123.06, 116.77, 114.89, 113.84, 54.80, 44.74** 

**3,7-Dichloro-1-***N***-(4-methoxybenzyl)quinoxalin-2-one (68).** Compound **67** (6.46 g, 20.4 mmol) and DIPEA (6.8 mL, 39.0 mmol) were dissolved in DMF (5.0 mL), after which POCl<sub>3</sub> (6.0 mL, 64.4 mmol) was carefully added. The resulting mixture was heated at 95 °C for 60 min and carefully poured onto crushed ice. After vigorous stirring, the obtained suspension was extracted with DCM, and the combined organic extracts were washed with water and brine and dried over Na<sub>2</sub>SO<sub>4</sub>. The solvent was reduced to about 20 mL and the solution was filtered over a pad of silica using DCM with 5% of methanol to elute the compound. Evaporation of the collected filtrate yielded 5.15 g (15.4 mmol, 75%) of the title compound as a white solid; mp 182.0–183.0 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 7.80 (d, J = 8.6 Hz, 1H), 7.59 (d, J = 2.0 Hz, 1H), 7.43 (dd, J = 8.6 Hz, 2H, 5.45 (s, 2H), 3.71 (s, 3H).

**3,6-Dichloroquinoxalin-2(1H)-one (69).** Compound **68** (1.0 g, 2.98 mmol) was dissolved in concentrated  $H_2SO_4$  (6.0 mL) and stirred at room temperature. After 20 min, the deep-red reaction mixture was carefully poured onto a small amount of crushed ice and neutralized with 15 N sodium hydroxide solution while cooling at 0 °C. A cloudy suspension was obtained that was extracted thoroughly with *n*-butanol. The combined organic layers were washed with a small amount of brine and dried over  $Na_2SO_4$ . Removal of the solvent yielded 640 mg (100%) of the crude title compound as a light-yellow solid. <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  (ppm): 7.73 (d, J = 8.6 Hz, 1H), 7.39–7.30 (m, 2H); <sup>13</sup>C NMR (DMSO- $d_6$ )  $\delta$  (ppm): 150.80, 149.27, 134.57, 133.15, 129.40, 129.29, 123.58, 114.67.

7-Chloro-3-(3-(methylamino)pyrrolidin-1-yl)quinoxalin-2(1H)one Hydrochloride (72). To compound 69 (100 mg, 0.47 mmol) in a microwave tube were added Boc-aminomethylpyrrolidine (103 mg, 0.51 mmol), DIPEA (0.9 mL, 0.51 mmol), and EtOAc (2.0 mL), and the resulting suspension was heated at 120 °C for 10 min. After cooling to room temperature, the mixture was diluted with EtOAc and washed with water and brine. Drying of the organic phase over Na<sub>2</sub>SO<sub>4</sub> and rotary evaporation gave the crude product that was purified over SiO<sub>2</sub> (EtOAC:hexanes, 1:4) to yield 124 mg (64%) of the Boc-protected intermediate (72) as a light-yellow solid. This compound was then stirred in 2 M HCl ether (2.0 mL) and dioxane (2.0 mL) until removal of the Boc group was completed (LCMS). Evaporation of the solvents yielded the hydrochloric salt as a yellow-white solid.  $^{1}$ H NMR (400 MHz, 341 K, D<sub>2</sub>O)  $\delta$  (ppm): 7.85 (d, J = 8.8 Hz, 1H), 7.62 (dd, J = 8.8 Hz, J = 2.2 Hz, 1H), 7.58 (d, J = 2.1 Hz, 1H), 4.92 (dd, J = 14.1 Hz, J = 6.8 Hz, 1H), 4.79 (dd, J=14.2 Hz, J=5.0 Hz, 1H), 4.59–4.54 (m, 1H), 4.50–4.41 (m, 2H), 3.18 (s, 3H), 3.00–2.92 (m, 1H), 2.75–2.66 (m, 1H).  $^{13}$ C NMR (D<sub>2</sub>O)  $\delta$  (ppm): 153.49, 147.74, 132.71, 128.45, 126.48, 125.11, 121.49, 116.65, 59.01, 54.58, 51.10, 33.14, 28.02; MS (ESI) m/z 279 (M + H) $^+$ .

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Supporting Information Available: LCMS purity data for compounds 5, 7, 9, 18–23, 26, 28–57, 62, 63, 70, and 72–74 and synthetic procedures for compounds 4–7, 18, 19, 21–32, 29–48, 50–52, 54, 55, 70, 73, and 74. This material is available free of charge via the Internet at http://pubs.acs.org.

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