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A Single-Molecule Magnet with a "Twist"

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The origin of the ferromagnetic exchange in a recently reported oxo-centered triangular Mn(III) single-molecule magnet (SMM) based on oximato ligands is fascinating since the exchange in all other complexes containing the [MnIII₃O]⁷⁺ core, including the wellknown "basic carboxylates" of general formula [MnIII3O(O2- $CR)_6L_3$]⁺ (R = Me, Et; L = py, MeCN, etc.), is antiferromagnetic.¹ We speculate that a structural distortion in the core of the molecule caused primarily by the "twisting" of the oximato ligands (with regard to the Mn^{III}₃ plane) is responsible. In order to test this hypothesis we have made a family of novel ligands based on salicylaldoxime (saoH₂) in which the oximate carbon atom has been derivatized to possess the "bulky" Me (Me-saoH₂), Et (Et-saoH₂), and Ph (Ph-saoH₂) groups, and synthesized analogues of the known hexanuclear SMM [Mn^{III}₆O₂(sao)₆(O₂CPh)₂(EtOH)₄] (1).² Complex 1, obtained upon the reaction of Mn(O₂CPh)₂·2H₂O with saoH₂ in EtOH (Figure S1) contains a nonplanar [Mn^{III}₆(μ₃-O)₂(μ-OR)₂]¹²⁺ unit of two off-set, stacked $[Mn^{III}_3(\mu_3-O)]^{7+}$ triangular subunits linked by two central oximato oxygens, with the remaining four sao²⁻ ligands bridging in a near-planar $\eta^1:\eta^1:\eta^1:\mu$ -fashion along the edges of the $[Mn^{III}_3(\mu_3-O)]^{7+}$ triangles. The four "central" metals (Mn1, Mn3) are six-coordinate and in distorted octahedral geometries, while the outermost Mn ions (Mn2) have square pyramidal geometries with an additional axial contact of \sim 3.5 Å to a phenolato oxygen. The coordination geometry of the metal ions is completed by a combination of terminal alcohols and μ -carboxylate groups. 1 displays an S = 4 spin ground state as a result of ferromagnetic exchange between the two antiferromagnetically coupled Mn^{III}₃ triangles.2

The idea was to investigate whether the additional steric bulk of the derivatized oximates would enforce structural distortions similar to those seen in the ferromagnetic Mn₃ triangle¹ and the recently obtained ferromagnetic [Mn^{III}₄(Me-sao)₄(Me-saoH)₄] "cube",³ and thus switch the magnetic behavior from antiferromagnetic to ferromagnetic. Complex [Mn^{III}₆O₂(Et-sao)₆(O₂CPh)₂(EtOH)₄(H₂O)₂]• 2EtOH (2·2EtOH) which contains the ethyl-derivatized saoH₂ (Et $saoH_2$) does indeed display ferromagnetic exchange (S = 12 ground state). Here we discuss its structure and magnetic properties. Complex 2 also crystallizes in the triclinic space group P1 and its structure (Figure 1) is analogous to that of 1.4 However, the increased steric bulk of the Et-sao²⁻ ligands causes a shortening of the phenolato oxygen (O71)-square pyramidal Mn (Mn3) distance (~2.5 Å) and severe twisting of the Mn-N-O-Mn moieties within each Mn3 subunit (Figure 1). This is evidenced by the average Mn-N-O-Mn torsion angle, which in 1 is $\alpha_v = 17.5^{\circ}$ compared to $\alpha_v = 36.5^{\circ}$ for 2. This also results in a change in the

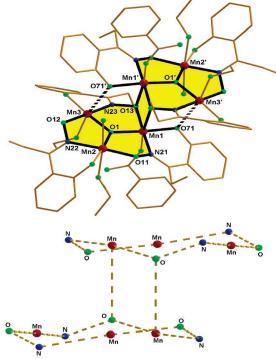


Figure 1. The molecular structure of complex 2, highlighting its core (top); the "twisted" Mn-N-O-Mn arrangement of the Etsao2- ligands in 2

coordination of the carboxylates: from μ -bridging to terminal, with the "vacant" site now occupied by an additional solvent (alcohol) molecule.

Variable-temperature dc magnetic susceptibility data were collected on 2 in the temperature range 5-300 K in an applied field of 0.1 T (Figure S2). The room temperature $\chi_{\rm M}T$ value of 19.24 cm³ K mol⁻¹ is slightly above that expected for six noninteracting Mn^{III} ions (18 cm³ K mol⁻¹). Upon cooling the value of $\chi_{\rm M}T$ increases gradually to 23.72 cm³ K mol⁻¹ at \sim 70 K, below which it increases rapidly to a maximum value of 69.57 cm³ K mol⁻¹ at 6.5 K, before decreasing slightly to 69.36 cm³ K mol⁻¹ at 5 K. This behavior is indicative of ferromagnetic exchange between the metal centers with the low-temperature maximum suggesting an S = 12 ground state. In order to confirm the ground state of the molecule, magnetization data were collected in the ranges 0.5-7 T and 1.8-7 K, and these are plotted as reduced magnetization $(M/N\mu_B)$ versus H/T in Figure 2. The data were fit by a matrix diagonalization method to a model that assumes only the ground state is populated, includes axial zero-field splitting $(D\hat{S}_z^2)$ and the Zeeman interaction, and carries out a full powder average. The corresponding Hamiltonian is given by eq 1,

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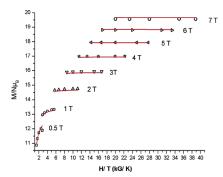


Figure 2. Plot of reduced magnetization $(M/N\mu_B)$ versus H/T for 2 in the field and temperature ranges 0.5–7 T and 1.8–7 K. The solid lines correspond to the fit of the data.

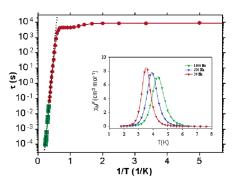


Figure 3. Out-of-phase $(\chi_M^{''})$ as susceptibility measurements in the 1.8–8 K and 50–1000 Hz ranges (insert); Arrhenius plot using as (green) and dc (red) data. The dashed line is the fit of the thermally activated region to eq 2.

$$\hat{H} = D\hat{S}_z^2 + g\mu_B \mu_0 \hat{S} \cdot H \tag{1}$$

where D is the axial anisotropy, $\mu_{\rm B}$ is the Bohr magneton, $\mu_{\rm 0}$ is the vacuum permeability, \hat{S}_z is the easy-axis spin operator, and H is the applied field. The best fit gave S=12, g=1.99 and $D=-0.43~{\rm cm}^{-1}$.

Ac susceptibility measurements were performed in the $1.8-16~\rm K$ range under a $3.5~\rm G$ ac field oscillating at $50-1000~\rm Hz$. The value of the in-phase $(\chi_{\rm M}'T)$ signal (Figure S3) increases with decreasing temperature to $\sim 6~\rm K$ where it then plateaus at $\sim 4~\rm K$ before displaying a frequency-dependent decrease below this temperature. This is indicative of the presence of excited states of smaller S values relatively close to the ground state. Extrapolation of the in-phase $(\chi_{\rm M}'T)$ signal to $0~\rm K$ from the plateau region gives a value of $\sim 74~\rm cm^3~\rm K~mol^{-1}$ indicative of an S=12 ground state, in agreement with the dc data (Figure S3). Fully visible, frequency-dependent out-of-phase $(\chi_{\rm M}'')$ ac susceptibility signals are seen below $\sim 6~\rm K$ with the peak at $1000~\rm Hz$ occurring at $\sim 4.5~\rm K$ (Figure 3).

Single-crystal hysteresis loop and relaxation measurements were performed on 2 using a micro-SQUID setup.⁵ Hysteresis loops were observed whose coercivity was strongly temperature (Figure 4) and sweep-rate (Figure S4) dependent, with a blocking temperature of ~ 3.5 K. Relaxation data determined from a combination of single-crystal dc and powder ac measurements were fitted to the Arrhenius relationship (eq 2, Figure 3),

$$\tau = \tau_0 \exp(U_{\text{eff}}/kT) \tag{2}$$

where $U_{\rm eff}$ is the effective relaxation barrier, τ is the relaxation time, and k is the Boltzmann constant, giving $U_{\rm eff} = 53.1$ K and $\tau_0 = 8 \times 10^{-10}$ s. The theoretical upper limit of $U = S^2 |D| = 88.5$ K is strongly reduced by the presence of low-lying excited states as

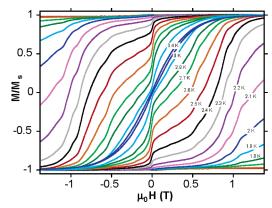


Figure 4. Magnetization versus field hysteresis loops for a single crystal of **2** at the indicated temperatures in a field sweep rate of 0.14 T s^{-1} . M is normalized to its saturation value.

evidenced by the fine structure in the hysteresis loops which are indicative of tunneling between excited-state multiplets. Detailed analyses of these will appear in a full paper.

In conclusion, the deliberate structural distortion of a Mn_6 compound via the use of a bulky salicylaldoxime derivative switches the intra-triangular magnetic exchange from antiferromagnetic to ferromagnetic resulting in an S=12 ground state and an anisotropy barrier ($U_{\rm eff}$) approaching that of the Mn_{12} family.^{7,8} The combined results thus suggest that the intelligent use of ligands that can cause targeted distortions to the core structures of Mn_6 (and other) clusters may prove to be a viable means to raise the blocking temperatures of SMMs to record values.

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Supporting Information Available: Crystallographic details in CIF format; synthetic procedures: and magnetism data. This material is available free of charge via the Internet at http://pubs.acs.org.

References

- Stamatatos, T. C.; Foguet-Albiol, D.; Stoumpos, C. C.; Raptopoulou, C. P.; Terzis, A.; Wernsdorfer, W.; Perlepes, S. P.; Christou, G. J. Am. Chem. Soc. 2005, 127, 15380.
- (2) Milios, C. J.; Raptopoulou, C. P.; Terzis, A.; Lloret, F.; Vicente, R.; Perlepes, S. P.; Escuer, A. Angew. Chem., Int. Ed. 2003, 43, 210.
- (3) Milios, C. J.; Prescimone, A.; Mishra, A.; Parsons, S.; Wernsdorfer, W.; Christou, G.; Perlepes, S. P.; Brechin, E. K. Chem. Commun. 2006. http://dx.doi.org/10.1039/b611174b.
- (4) Anal. Calcd (found) for dried **2** (solvent free): C 50.62 (50.78), H 5.14 (5.01), N 4.66 (4.51). Crystal data for **1**·2CH₃CH₂OH: C_{80} H₁₀₄Mn₆N₆O₂₆, FW = 1895.30 g mol⁻¹, T = 150 K, Triclinic P1, a = 12.4947(9) Å, b = 13.2846(9) Å, c = 14.5047(11) Å, α = 71.488(4)°, β = 82.305(4)°, γ = 68.687(4)°, V = 2126.4(3) ų, d_{calcd} = 1.480 g cm⁻³, independent refelections 11741 [R(int) = 0.034], data 8011, parameters 547, final R1 = 0.0339 for 8011 reflections with F > $4\sigma(F)$. Conventional R [F > $4\sigma(F)$], R1 = 0.0339 [8011 data].
- (5) Wernsdorfer, W. Adv. Chem. Phys. 2001, 118, 99.
- (6) For reviews see: (a) Christou, G.; Gatteschi, D.; Hendrickson, D. N.; Sessoli, R. MRS Bull. 2000, 25, 66. (b) Gatteschi, D.; Sessoli, R. Angew. Chem., Int. Ed. 2003, 42, 268. (c) Aromí, G.; Brechin, E. K. Struct. Bonding 2006, 122, 1. (d) Bircher, R.; Chabousant, G.; Dobe, C.; Güdel, H. U.; Ochsenbein, S. T.; Sieber, A.; Waldmann, O. Adv. Funct. Mater. 2006, 16, 209.
- (7) (a) Caneschi, A.; Gatteschi, D.; Sessoli, R.; Barra, A. L.; Brunnel, L. C.; Guillot, M. J. Am. Chem. Soc. 1991, 113, 5873. (b) Sessoli, R.; Tsai, H. L.; Schake, A. R.; Wang, S. Y.; Vincent, J. B.; Folting, K.; Gatteschi, D.; Christou, G.; Hendrickson, D. N. J. Am. Chem. Soc. 1993, 115, 1804. (c) Sessoli, R.; Gatteschi, D.; Caneschi, A.; Novak, M. A. Nature 1993, 365, 141.
- (8) Chakov, N. E.; Lee, S.-C.; Harter, A. G.; Kuhns, P. L.; Reyes, A. P.; Hill, S. O.; Dalal, N. S.; Wernsdorfer, W.; Abboud, K.; Christou, G. J. Am. Chem. Soc. 2006, 128, 6975.

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