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Selective Detection and Quantification of Modified DNA with Solid-State Nanopores

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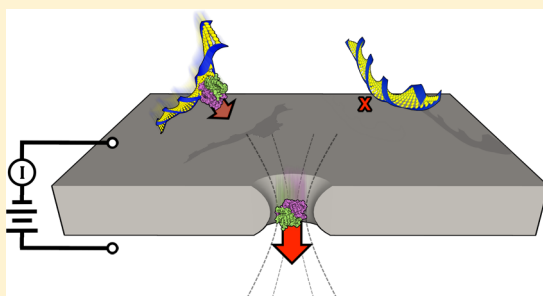
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S Supporting Information

ABSTRACT: We demonstrate a solid-state nanopore assay for the unambiguous discrimination and quantification of modified DNA. Individual streptavidin proteins are employed as high-affinity tags for DNA containing a single biotin moiety. We establish that the rate of translocation events corresponds directly to relative concentration of protein–DNA complexes and use the selectivity of our approach to quantify modified oligonucleotides from among a background of unmodified DNA in solution.



KEYWORDS: Solid-state nanopores, proteins, DNA, DNA modifications, single-molecule, quantification

Immunoprecipitation and pull-down assays are workhorses in biochemistry. With the ability to discriminate specific substrates in heterogeneous mixtures, they play important roles in a wide range of fields, including proteomics,^{1–3} epigenomics,^{4–7} and transcriptomics.^{8,9} However, despite their broad utility these well-established strategies have limitations. Besides requiring large sample sizes, they are labor-intensive and are not inherently quantitative, typically requiring subsequent polymerase chain reaction¹⁰ or enrichment¹¹ for downstream analysis. For these reasons, quantitative technologies with single-molecule sensitivity may offer important advantages.⁵

Hinging on the Coulter principle, solid-state (SS-) nanopores^{12,13} have been employed extensively for molecular resistive-pulse sensing. In these experiments, the temporary presence of a single molecule passing through a narrow pore alters the coincident flow of ions, thereby creating a unique electronic signature in the measured ionic current. Analysis of these signatures (or events) offers insight into molecular structure,^{14,15} surface charge,^{16,17} and pore-molecule interactions.^{18,19} With potential applications ranging from DNA sequencing²⁰ to biomarker identification,²¹ this powerful detection platform has been used to probe a diverse set of materials, including nucleic acids,^{22–24} proteins,^{25–27} and nonbiological nanoparticles.^{28–30}

Recently, several groups have reported progress toward SS-nanopore differentiation of structural^{31–33} or chemical^{34,35} variations in nucleic acid molecules. However, such studies often rely on subtle shifts in event characteristics that can introduce uncertainty. In this paper, we present an assay that

enables unambiguous identification of double-strand (ds) DNA modified with a single biotin moiety. Using a high-affinity protein tag (monovalent streptavidin, MS) to label modified oligonucleotides, we determine that translocation event rate correlates with protein–DNA complex formation. We speculate on the mechanism underlying this finding and then exploit its specificity to quantify modified DNA directly in solution with unmodified oligonucleotides.

SS-Nanopore Discrimination of Monobiotinylated dsDNA. Figure 1a shows a schematic of the SS-nanopore measurement approach utilized here. An electrical bias is applied across a thin-film membrane with a single nanopore immersed in electrolyte solution (see Methods). This facilitates the electrokinetic translocation of molecules (or molecular complexes) through the pore, each of which can produce an ionic current event. We first use this technique to measure MS (Figure 1b left) and monobiotinylated 90 bp dsDNA (bio90, Figure 1b center) individually at concentrations of 8 and 1 μ M, respectively. Over a range of 50–200 mV, few events can be identified for either molecule. However, when MS and bio90 are incubated together at a molar ratio of 8:1 (MS:bio90) prior to measurements, we observe a remarkable increase in the number of events per unit time (Figure 1b right). The event rate of the admixture is consistently more than an order of magnitude greater than that of either constituent molecule

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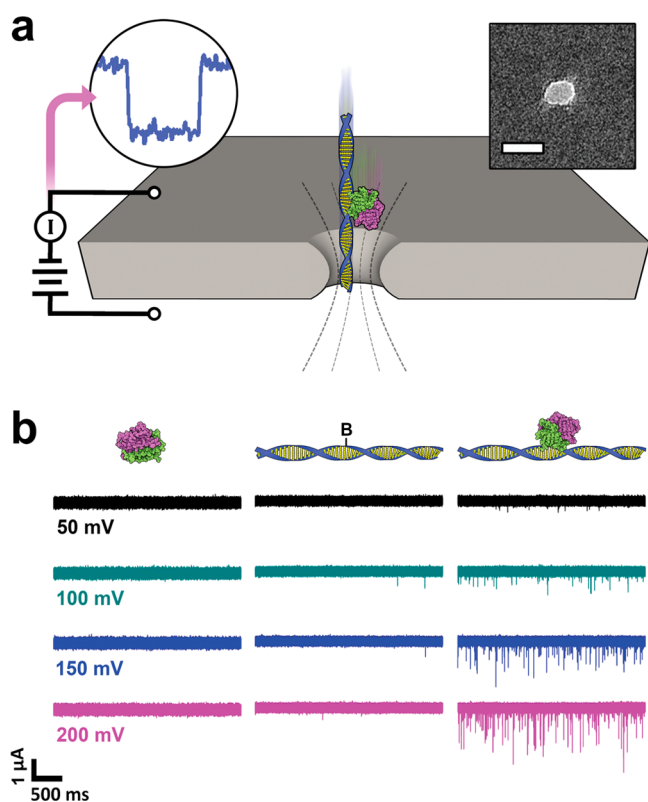


Figure 1. SS-nanopore measurement. (a) Schematic diagram of the experimental system. A bias is applied across a SS-nanopore in a SiN membrane, inducing electrokinetic translocation of protein–DNA complexes in electrolyte solution. Upper left shows the shape of a resistive pulse measured during a translocation event. Right inset: transmission electron micrograph of a typical SS-nanopore formed with conditions identical to those used here. Scale bar is 10 nm. (b) Raw current traces obtained for MS alone (left), bio90 alone (center), and MS incubated with bio90 at a molar ratio (MS:bio90) of 8:1 (right) over a range of voltages. Investigated molecules are shown schematically above each column (“B” represents the biotin moiety). Only the coincubated material produces significant events.

alone; at 200 mV applied voltage, for example, the MS/bio90 complex yields a rate of $23.3 \pm 0.9 \text{ s}^{-1}$, while the event rates of MS and bio90 individually are 0.09 ± 0.04 and $1.1 \pm 0.2 \text{ s}^{-1}$, respectively. In order to verify that the MS/bio90 events correspond to actual translocations rather than stochastic interactions between the complex and the nanopore, we reverse the polarity of the applied voltage during our measurement and observe “recapture events”^{36,37} (see Supporting Information Figure 1). Translocations are further supported by the voltage dependence³⁸ observed for MS/bio90 event durations (Supporting Information Figure 3).

What accounts for the extraordinary differences in the event rates that we observe? Although neither the MS nor bio90 alone yields a significant number of events over the investigated voltage range, we suggest differing explanations for each. For the MS, recent work²⁶ comparing experimental data to the Smoluchowski theory³⁹ suggests that the translocation of proteins through SS-nanopores typically occurs too rapidly to be resolved by conventional systems like the one employed here. Indeed, Larkin et al.²⁷ demonstrated that the enhanced time-resolution of a high-bandwidth system is capable of resolving far more protein translocation events than are detectable by standard electronics. The significant negative

charge of our MS (see Methods) will contribute to a strong electrophoretic driving force, potentially reducing translocation time further. We therefore conclude that the absence of observed MS events results from the bandwidth limitations of our apparatus, which allow a significant number of protein molecules to translocate undetected. Conversely, we attribute the small number of events measured for bio90 to infrequent translocations. Storm et al.⁴⁰ showed experimentally that dsDNA dwell times scale with molecular length as a power-law with exponent 1.27. Their results were obtained using SS-nanopores of similar diameter to those employed here and with a comparable voltage. Assuming that the scaling factor holds for small molecular lengths (Wanunu et al.¹⁹ reported similar scaling for short dsDNA), we estimate that 90 bp dsDNA should translocate in $\sim 50 \mu\text{s}$; a value that is resolvable by our electronics. Indeed, of the few events that we observe with bio90 alone, most yield a dwell time near this value (see Supporting Information Figure 4). Thus, we conclude that under the low-voltage conditions investigated here, the translocation event rate for bio90 is minimal.

The dichotomy between these two explanations offers a possible mechanism for the increase in event rate observed for the admixture of MS and bio90. When the molecules form a complex, the large net electrical force experienced by MS alone is countered by a significant viscous drag imparted by the bio90. We therefore propose that the combination of these forces results in translocations that are slow in comparison to MS alone and are thereby resolvable with our apparatus. Importantly, this results in selective isolation of target DNA rather than exclusion determined, for example, by SS-nanopore dimensions.²¹

In order to investigate the system further, we next perform a series of SS-nanopore measurements in which MS is titrated against a constant amount ($1 \mu\text{M}$) of bio90. Over all investigated voltages, we observe that the measured event rate rises dramatically up to a molar ratio of 1:1 (Figure 2a). However, from unity up to a molar ratio of 8:1 (MS:bio90), additional MS does not increase the event rate further. This is a result of the limited supply of dsDNA needed to form nucleoprotein complexes; the protein has an extremely low off rate⁴¹ ($\sim 10^{-5} \text{ s}^{-1}$) and each oligonucleotide contains only a single biotin moiety, so we expect that nearly all bio90 in solution will be bound at or above an equimolar concentration. Comparing our translocation results to an electromobility shift assay (EMSA) performed with MS and bio90 over the same stoichiometric range, we observe a strikingly similar trend (Figure 2b). These data support our assertion that virtually all observed translocation events for the admixture correspond to MS/bio90 complexes. Additional evidence of the high specificity of this approach is provided by control measurements in which nonbiotinylated dsDNA incubated with MS yields a negligible event rate, equivalent to bio90 alone (see Supporting Information Figure 5).

Selective Quantification of Modified Oligonucleotides. In Figure 3, we examine admixture event rates up to a molar ratio of 1:1 and find a linear dependence on applied voltage. This implies that the capture process for the MS/bio90 complex is governed by diffusion rather than by interactions with the pore, which is in agreement with previous studies.³⁶ Importantly, the observed trend offers a route toward quantification of MS/bio90 complexes in solution. It has been demonstrated elsewhere^{19,26,32} that event frequency varies with molecular concentration. Because nearly all events

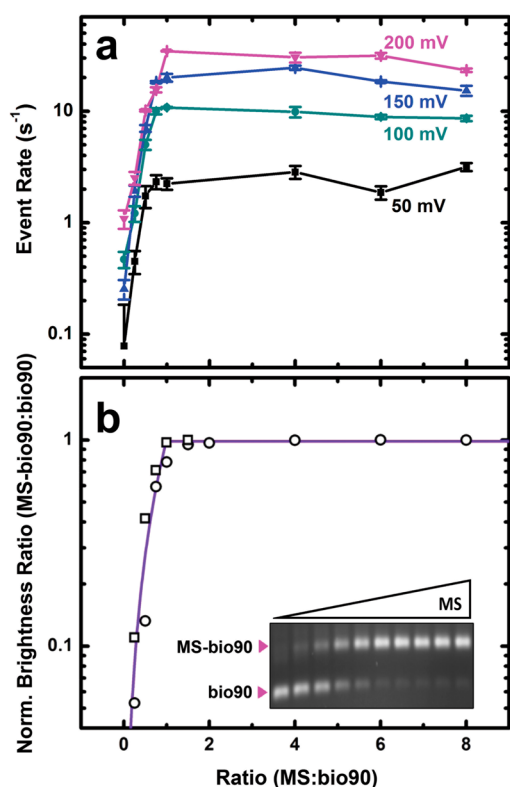


Figure 2. MS/bio90 titration experiments. (a) Semilog plot showing stoichiometric dependence of SS-nanopore translocation event rate over a range of applied voltages. (b) Semilog plot (normalized) showing stoichiometric dependence of band intensity for MS/bio90 complex relative to bio90 measured from EMSA. Circles and squares are data from two separate assays. Inset shows an example gel (circle symbols in plot).

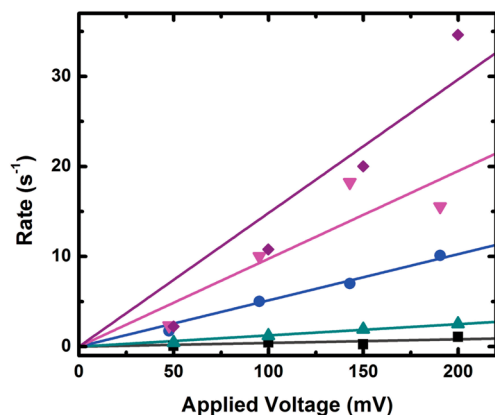


Figure 3. Voltage dependence of event rate. Event rates corresponding to specific MS/bio90 stoichiometries plotted over a range of applied voltages. From bottom to top, molar ratios (MS:bio90) are 0:1 (squares), 1:4 (upward triangles), 1:2 (circles), 3:4 (downward triangles), and 1:1 (diamonds). Solid lines are linear fits to each ratio set (color indicated).

observed in our system can be attributed exclusively to the translocation of complexes, the linear fits in Figure 3 link the concentration of MS/bio90 in solution to specific event rates produced at a given voltage. The measurements described thus far have been performed in a protein-limited regime (MS:bio90 < 1:1), and so the measured event rate has facilitated quantification of MS/bio90 complexes in a background of

unconjugated bio90. However, the same approach could in principal be used to quantify biotinylated oligonucleotides in a heterogeneous solution with nonbiotinylated DNA as well.

To investigate this possibility, we conduct a blind test on two samples prepared by a third party. Each of these samples contains a different mixture of biotinylated and nonbiotinylated 90 bp dsDNA mixed to a total concentration of 1 μ M (equivalent to that of the measurements described above). To ensure that all bio90 is conjugated, both solutions are incubated with MS at a concentration of 4 μ M. As described in the previous sections, MS alone produces a negligible number of measurable events, and so excess protein does not perturb the measurements. SS-nanopore analysis reveals a linear relationship between applied voltage and event rate for both samples, as expected (see Supporting Information Figure 6). Comparing the event rates obtained from the two blind samples to our prior measurements (Figure 4), we derive a value for the bio90

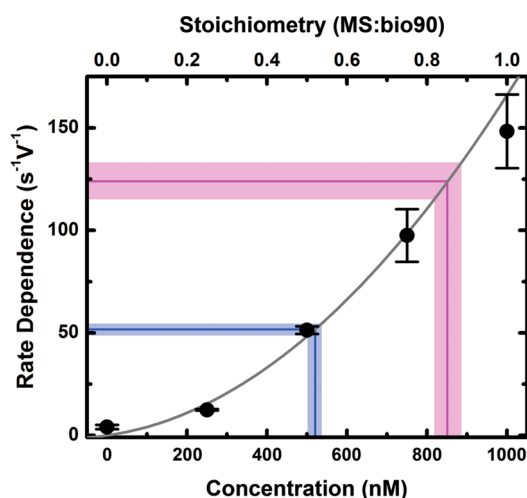


Figure 4. Selective determination of bio90 concentration within a mixture. Solid data points indicate event rate dependence (i.e., slopes from Figure 3) plotted against MS/bio90 concentration. Solid gray line is a second-order polynomial fit to the data. Experimentally derived event rate values for samples prepared in blind experiments are shown as magenta (Sample 1) and blue (Sample 2) lines extending from y-axis. Shaded regions represent error.

concentration in each: 850 ± 35 nM in Sample 1 and 520 ± 20 nM in Sample 2. Remarkably, these experimentally determined concentrations are in excellent agreement with the prepared values of 800 ± 20 and 480 ± 20 nM, respectively. These results demonstrate that our SS-nanopore approach is uniquely capable of quantifying DNA having single nucleotide biotin modifications selectively, even within a mixed sample.

Discussion. In this study, we have demonstrated highly specific detection and quantification of monobiotinylated dsDNA using SS-nanopores. We performed our detection through selective conjugation with a streptavidin protein containing a single biotin-binding domain. Under the experimental conditions investigated, we observed that the MS/bio90 complex produced a cascade of easily resolved translocation events, which is in stark contrast to either constituent molecule individually. While the DNA constructs used here featured an internal biotin near the center of the molecule (see Methods), additional measurements indicate that the effect is not dependent on the position of the biotin moiety. The variation in translocation behavior was attributed to

structure-dependent differences in net driving force, resulting in a dramatic increase in resolvable translocation events for the nucleoprotein complex as compared to DNA or protein alone. By studying stoichiometric effects, we showed that measured event rate correlated to the (partial) concentration of MS/bio90 conjugates in solution. Finally, we exploited the specificity of our measurement and the relationship between event rate and concentration to achieve selective quantification of modified oligonucleotides within a background of unmodified DNA. Upon examination of two different blind samples, we found excellent agreement with prepared values, validating our approach as a quantitative detection technique. An analogue to conventional approaches like immunoprecipitation, our technique offers unique advantages, including molecular sensitivity and intrinsic quantification. We anticipate that our methodology will have direct impact on a variety of fields including diagnostic biomarker detection and gene profiling.

Methods. SS-Nanopore Device Fabrication and Electrical Measurement. Nanopores were fabricated using a technique described elsewhere.⁴² Briefly, the beam of a scanning helium ion microscope (Carl Zeiss Orion Plus) was focused on a suspended silicon nitride thin film membrane (thickness 30 nm) in a silicon support chip. Calibrated exposure times were used to mill nanopores with diameters ranging from 7.3 to 7.7 nm. The support chip containing an individual pore was then positioned in a custom flow cell with fluid access to both sides of the membrane. Measurement solution (900 mM NaCl and 6 mM PBS buffer) was introduced on either side of the flow cell, and Ag/AgCl electrodes were immersed in the solution. Electrical measurements (Axopatch 200B) were used to verify that the device exhibited low RMS noise (typically <20 pA) and linear current–voltage characteristics that matched the calibrated nanopore diameter.³² Translocation measurements were performed by replacing the solution on one side of the device with measurement solution containing biomolecules. Current was recorded at a bandwidth of 200 kHz and filtered at 100 kHz with a four-pole Bessel filter. Analysis was performed with custom software with which we applied an additional low-pass filter of 25 kHz to all measurements. The event threshold for analysis was set at 4 standard deviations and events with durations from 12 to 2000 μ s were considered.

Biomolecules. Bio90 oligonucleotides were purchased (Integrated DNA Technologies, Coralville, IA) with the sequence: TGT ATA CCA TGG CCA GGA TCC TGG GCC ATC TGG TAT^B GTA ATT CAT AAA GAA TTC TCA TTC TGC AGG TGC ACA TGT TAA CAC TAG TCG TGA. The T^B represents a single internal biotinylated dT. The opposing strand (forming the dsDNA) contained no modified nucleotides. The nonbiotinylated oligonucleotide used in the mixture (blind measurements) had the same sequence but with no biotin moiety. The streptavidin variant employed (SAe1D3) contained one active biotin-binding site⁴¹ and was supplied by the Howarth lab (Oxford University). This mutant protein (54.5 kDa) retains binding affinity and stability similar to wild-type streptavidin and contains a hexaglutamate tag⁴³ used for isolation that imparts a negative charge of $-17.1e$ under comparable pH conditions.

Electrophoretic Mobility Shift Assay. MS was incubated in 1 \times PBS buffer with bio90 for 20 min at room temperature at molar ratios ranging from 0:1 to 8:1 (MS:bio90). The mixtures were then loaded onto a 1.5% agarose gel with ethidium bromide for visualization. The buffer reservoir of the electro-

phoresis unit was submerged in an ice bath to minimize dissociation of the protein–DNA complex.

■ ASSOCIATED CONTENT

● Supporting Information

Additional figures depicting recapture of translocated MS-DNA constructs, scatter plots as a function of applied voltage and stoichiometry, dwell time analysis, control measurements with nonbiotinylated DNA, and analysis of blind sample measurements. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

A.T.C. and O.K.Z. contributed equally.

Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) Peng, J.; Schwartz, D.; Elias, J. E.; Thoreen, C. C.; Cheng, D.; Marsischky, G.; Roelofs, J.; Finley, D.; Gygi, S. P. *Nat. Biotechnol.* **2003**, *21* (8), 921–926.
- (2) Rual, J.-F.; Venkatesan, K.; Hao, T.; Hirozane-Kishikawa, T.; Dricot, A.; Li, N.; Berriz, G. F.; Gibbons, F. D.; Dreze, M.; Ayivi-Guedehoussou, N.; Klitgord, N.; Simon, C.; Boxem, M.; Milstein, S.; Rosenberg, J.; Goldberg, D. S.; Zhang, L. V.; Wong, S. L.; Franklin, G.; Li, S.; Albala, J. S.; Lim, J.; Fraughton, C.; Llamas, E.; Cevik, S.; Bex, C.; Lamesch, P.; Sikorski, R. S.; Vandenhaute, J.; Zoghbi, H. Y.; Smolyar, A.; Bosak, S.; Sequerra, R.; Doucette-Stamm, L.; Cusick, M. E.; Hill, D. E.; Roth, F. P.; Vidal, M. *Nature* **2005**, *437* (7062), 1173–1178.
- (3) Zhang, B.; Park, B.-H.; Karpins, T.; Samatova, N. F. *Bioinformatics* **2008**, *24* (7), 979–986.
- (4) Jacinto, F. V.; Ballestar, E.; Esteller, M. *BioTechniques* **2008**, *44* (1), 35–43.
- (5) Laird, P. W. *Nat. Rev. Genet.* **2010**, *11* (3), 191–203.
- (6) Song, C.-X.; Yi, C.; He, C. *Nat. Biotechnol.* **2012**, *30* (11), 1107–1116.
- (7) Williams, K.; Christensen, J.; Pedersen, M. T.; Johansen, J. V.; Cloos, P. A. C.; Rappilber, J.; Helin, K. *Nature* **2011**, *473* (7347), 343–348.
- (8) Hirai, M. Y.; Klein, M.; Fujikawa, Y.; Yano, M.; Goodenowe, D. B.; Yamazaki, Y.; Kanaya, S.; Nakamura, Y.; Kitayama, M.; Suzuki, H.; Sakurai, N.; Shibata, D.; Tokuhisa, J.; Reichelt, M.; Gershenzon, J.; Papenbrock, J.; Saito, K. *J. Biol. Chem.* **2005**, *280* (27), 25590–25595.
- (9) Zhao, J.; Ohsumi, T. K.; Kung, J. T.; Ogawa, Y.; Grau, D. J.; Sarma, K.; Song, J. J.; Kingston, R. E.; Borowsky, M.; Lee, J. T. *Mol. Cell* **2010**, *40* (6), 939–953.
- (10) Haring, M.; Offermann, S.; Danker, T.; Horst, I.; Peterhansel, C.; Stam, M. *Plant Methods* **2007**, *3*, 11.

- (11) Selbach, M.; Mann, M. *Nat. Methods* **2006**, 3 (12), 981–983.
- (12) Dekker, C. *Nat. Nanotechnol.* **2007**, 2 (4), 209–215.
- (13) Wanunu, M. *Phys. Life Rev.* **2012**, 9 (2), 125–158.
- (14) Freedman, K. J.; Haq, S. R.; Edel, J. B.; Jemth, P.; Kim, M. J. *Sci. Rep.* **2013**, 3, 1638.
- (15) Fologea, D.; Brandin, E.; Uplinger, J.; Branton, D.; Li, J. *Electrophoresis* **2007**, 28 (18), 3186–3192.
- (16) Hoogerheide, D. P.; Garaj, S.; Golovchenko, J. A. *Phys. Rev. Lett.* **2009**, 102 (25), 256804.
- (17) Smeets, R. M. M.; Keyser, U. F.; Krapf, D.; Wu, M. Y.; Dekker, N. H.; Dekker, C. *Nano Lett.* **2006**, 6 (1), 89–95.
- (18) Larkin, J.; Henley, R.; Bell, D. C.; Cohen-Karni, T.; Rosenstein, J. K.; Wanunu, M. *ACS Nano* **2013**, 7 (11), 10121–10128.
- (19) Wanunu, M.; Sutin, J.; McNally, B.; Chow, A.; Meller, A. *Biophys. J.* **2008**, 95 (10), 4716–4725.
- (20) Branton, D.; Deamer, D. W.; Marziali, A.; Bayley, H.; Benner, S. A.; Butler, T.; Di Ventra, M.; Garaj, S.; Hibbs, A.; Huang, X. H.; Jovanovich, S. B.; Krstic, P. S.; Lindsay, S.; Ling, X. S. S.; Mastrangelo, C. H.; Meller, A.; Oliver, J. S.; Pershin, Y. V.; Ramsey, J. M.; Riehn, R.; Soni, G. V.; Tabard-Cossa, V.; Wanunu, M.; Wiggin, M.; Schloss, J. A. *Nat. Biotechnol.* **2008**, 26 (10), 1146–1153.
- (21) Niedzwiecki, D. J.; Iyer, R.; Borer, P. N.; Movileanu, L. *ACS Nano* **2013**, 7 (4), 3341–3350.
- (22) Li, J.; Gershow, M.; Stein, D.; Brandin, E.; Golovchenko, J. A. *Nat. Mater.* **2003**, 2 (9), 611–615.
- (23) Skinner, G. M.; van den Hout, M.; Broekmans, O.; Dekker, C.; Dekker, N. H. *Nano Lett.* **2009**, 9 (8), 2953–2960.
- (24) Storm, A. J.; Chen, J. H.; Zandbergen, H. W.; Dekker, C. *Phys. Rev. E: Stat., Nonlinear, Soft Matter Phys.* **2005**, 71 (5 Pt 1), 051903.
- (25) Fologea, D.; Ledden, B.; McNabb, D. S.; Li, J. *Appl. Phys. Lett.* **2007**, 91 (5), 053901.
- (26) Plesa, C.; Kowalczyk, S. W.; Zinsmeister, R.; Grosberg, A. Y.; Rabin, Y.; Dekker, C. *Nano Lett.* **2013**, 13 (2), 658–663.
- (27) Larkin, J.; Henley, R. Y.; Muthukumar, M.; Rosenstein, J. K.; Wanunu, M. *Biophys. J.* **2014**, 106 (3), 696–704.
- (28) Bacri, L.; Oukhaled, A. G.; Schiedt, B.; Patriarche, G.; Bourhis, E.; Gierak, J.; Pelta, J.; Auvray, L. *J. Phys. Chem. B* **2011**, 115 (12), 2890–2898.
- (29) Hall, A. R.; Keegstra, J. M.; Duch, M. C.; Hersam, M. C.; Dekker, C. *Nano Lett.* **2011**, 11 (6), 2446–2450.
- (30) Prabhu, A. S.; Jubery, T. Z. N.; Freedman, K. J.; Mulero, R.; Dutta, P.; Kim, M. J. *J. Phys.: Condens. Matter* **2010**, 22 (45), 454107.
- (31) Venta, K.; Shemer, G.; Puster, M.; Rodríguez-Manzo, J. A.; Balan, A.; Rosenstein, J. K.; Shepard, K.; Drndić, M. *ACS Nano* **2013**, 7 (5), 4629–4636.
- (32) Wanunu, M.; Dadosh, T.; Ray, V.; Jin, J.; McReynolds, L.; Drndić, M. *Nat. Nanotechnol.* **2010**, 5 (11), 807–814.
- (33) Zhao, Q.; Sigalov, G.; Dimitrov, V.; Dorvel, B.; Mirsaidov, U.; Sligar, S.; Aksimentiev, A.; Timp, G. *Nano Lett.* **2007**, 7 (6), 1680–1685.
- (34) Shim, J.; Humphreys, G. I.; Venkatesan, B. M.; Munz, J. M.; Zou, X.; Sathe, C.; Schulten, K.; Kosari, F.; Nardulli, A. M.; Vasmatazis, G.; Bashir, R. *Sci. Rep.* **2013**, 3, 1389.
- (35) Wanunu, M.; Cohen-Karni, D.; Johnson, R. R.; Fields, L.; Benner, J.; Peterman, N.; Zheng, Y.; Klein, M. L.; Drndić, M. *J. Am. Chem. Soc.* **2011**, 133 (3), 486–492.
- (36) Gershow, M.; Golovchenko, J. A. *Nat. Nanotechnol.* **2007**, 2 (12), 775–779.
- (37) Plesa, C.; Cornelissen, L.; Tuijtel, M. W.; Dekker, C. *Nanotechnology* **2013**, 24 (47), 475101.
- (38) Carlsen, A. T.; Zahid, O. K.; Ruzicka, J.; Taylor, E. W.; Hall, A. R. *ACS Nano* **2014**, DOI: 10.1021/nn501694n.
- (39) Smoluchowski, M. v. Z. *Phys. Chem.* **1917**, 92, 129–168.
- (40) Storm, A. J.; Storm, C.; Chen, J.; Zandbergen, H.; Joanny, J.-F.; Dekker, C. *Nano Lett.* **2005**, 5 (7), 1193–1197.
- (41) Howarth, M.; Chinnapen, D. J. F.; Gerrow, K.; Dorrestein, P. C.; Grandy, M. R.; Kelleher, N. L.; El-Husseini, A.; Ting, A. Y. *Nat. Methods* **2006**, 3 (4), 267–273.
- (42) Yang, J.; Ferranti, D. C.; Stern, L. A.; Sanford, C. A.; Huang, J.; Ren, Z.; Qin, L.-C.; Hall, A. R. *Nanotechnology* **2011**, 22 (28), 285310.
- (43) Fairhead, M.; Krndija, D.; Lowe, E. D.; Howarth, M. J. *Mol. Biol.* **2014**, 426 (1), 199–214.