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Russian-Doll-Type Metal Carbide Endofullerene: Synthesis, Isolation, and Characterization of Sc₄C₂@C₈₀

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Endohedral metallofullerenes have broadened the range of fullerenes by virture of their novel structures and promising applications. 1,2 To date, the compositions of the endohedral clusters vary, including one metal atom, two metal atoms, and trimetallic nitride, metal carbide, and metal oxide clusters, etc. 1-6 Because endofullerenes have so many abundant endohedral clusters, they can be provided with intricate structures and properties, such as the quantum gyroscope $Sc_2C_2@C_{84}$ as well as $Sc_4(\mu-O_2)@C_{80}$ with a distorted tetrahedral scandium oxide cluster inside a carbon cage. 6,7 For a long time, inorganic nesting polyhedra and carbon onions have attracted much attention because of their distinctive endohedral architectures. However, metallofullerenes with nesting structures have not been available. 1,2 Recently, the structure of Sc₄C₂@C₈₀-I_h was predicted by density functional theory (DFT) calculations, which revealed that C2@Sc4@C80-Ih is the more realistic formula for Sc₄C₈₂. Herein, we report the synthesis, isolation, and characterization of Sc₄C₈₂ by means of mass spectrometry (MS) and UV-vis, FTIR, and ¹³C NMR spectroscopy measurements in combination with DFT calculations and show that the synthesized Sc_4C_{82} is indeed $Sc_4C_2@C_{80}$ - I_h or, more exactly, C2@Sc4@C80-Ih, i.e., a C2 unit surrounded by a Sc4 tetrahedron then encaged in an icosahedral C_{80} (C_{80} - I_h) cage.

 Sc_4C_{82} was prepared by the Krätschmer–Huffman arc discharge method and isolated by two-stage high performance liquid chromatography (HPLC) [see the Supporting Information (SI)]. The soot was promptly Soxhlet-extracted with toluene. HPLC using two columns, namely, Buckyprep and Buckyprep-M, was employed to isolate and purify the Sc_4C_{82} , respectively. The purity of the sample was confirmed by HPLC analysis and the MALDI–TOF mass spectrum (Figure 1). The MALDI–TOF mass spectrum exhibits a strong molecular ion peak at m/z 1164, accounting for the production of Sc_4C_{82} .

Figure 2 presents the 13 C NMR spectrum of Sc_4C_{82} in CS_2 at 293 K. Two signals at 137.8 and 144.7 ppm in a 1:3 intensity ratio can be clearly observed. It is unreasonable to identify the molecular structure as $Sc_4@C_{82}$, as no C_{82} cage can satisfy this 13 C NMR spectral pattern. Instead, the spectrum is fully consistent with the C_{80} - I_h cage, which has two unique carbon atoms (the triphenylenic and corannulenic sites) in a 1:3 atomic ratio. Although the Sc_4C_2 endocluster may disturb the chemical environment of the carbon atoms of the C_{80} cage, the influence can be markedly reduced via intramolecular dynamics and cage rotations. 9,11 To our knowledge, two distinct 13 C NMR signals of the C_{80} - I_h cage were also observed for $Sc_3N@C_{80}$ - I_h (137.24 and 144.57 ppm), 3a $Lu_3N@C_{80}$ - I_h (137.4

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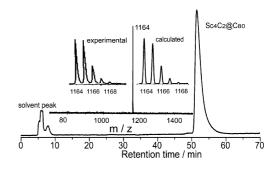


Figure 1. HPLC trace of $Sc_4C_2@C_{80}$ - I_h in a Buckyprep column. The inset shows the positive-ion MALDI-TOF mass spectrum as well as the experimental and calculated isotope distributions of $Sc_4C_2@C_{80}$ - I_h .

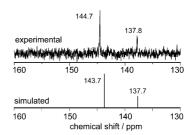


Figure 2. (top) Experimental (CS₂, 150 MHz) and (bottom) simulated 13 C NMR spectra of the C₈₀- I_h cage in Sc₄C₂@C₈₀- I_h . D₂O inside of a capillary was used as an internal lock.

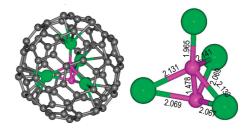


Figure 3. (left) DFT-optimized structure of $Sc_4C_2@C_{80}$ - I_h and (right) calculated Sc-C and C-C bond lengths (Å) in the encaged Sc_4C_2 moiety. Green balls represent the Sc atoms and purple balls the carbon atoms of the inner carbide moiety.

and 144.0 ppm),¹⁰ CeSc₂N@C₈₀- I_h (135.90 and 142.85 ppm),¹¹ and Sc₃C₂@C₈₀- I_h (138.9 and 145.6 ppm).^{5a} As shown in Figure 2, the DFT-predicted ¹³C NMR spectrum (with signals at 137.7 and 143.7 ppm) for the Sc₄C₂@C₈₀- I_h structural model⁹ (Figure 3) agrees well with the experimental ¹³C NMR spectrum. The ¹³C NMR chemical shifts of C₂ units within the endocluster, which were calculated to

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Table 1. Experimental and DFT-Calculated Redox Potentials (V) of Sc₄C₂@C₈₀-I_h

	$_{\rm ox}E_2$	oxE₁	$_{\rm red}E_1$	$_{\rm red}E_2$
exptl ^a DFT ^b	1.10	0.40	-1.16	-1.65
$\overline{\mathrm{DFT}}^b$	1.30	0.32	-1.33	-2.02

^a Values are given vs Fc/Fc⁺ and were obtained in o-dichlorobenzene containing 0.1 M TBAPF6 at a glassy carbon working electrode via CV at a scan rate of 100 mV s⁻¹. ^b From ref 9.

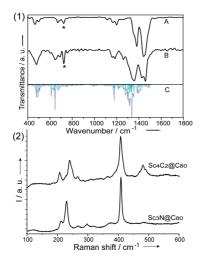


Figure 4. (1) FTIR spectra of (A) $Sc_3C_2@C_{80}$ - I_h and (B) $Sc_4C_2@C_{80}$ - I_h along with (C) the simulated IR spectrum of Sc₄C₂@C₈₀-I_h. The asterisks show the solvent peaks. (2) Low-energy Raman spectra of Sc₄C₂@C₈₀-I_h and Sc₃N@C₈₀-I_h (laser wavelength 633 nm).

appear at 226.1 and 326.7 ppm, were not detected because of the spin-rotation interaction and low signal-to-noise ratio. 5e,9

Previous DFT computations have shown that Sc₄C₂@C₈₀-I_h has a valence state of $[C_2]^{6-}@[Sc^{3+}]_4@[C_{80}\text{-}I_h]^{6-}$ with a wide HOMO-LUMO gap.9 Table 1 lists the redox potentials of $Sc_4C_2@C_{80}$ - I_h obtained by cyclic voltammetry (CV) and DFT computations. The measured electrochemical band gap of 1.56 eV, along with the DFT prediction (1.65 eV), indicates that Sc₄C₂@C₈₀- I_h is a very stable metallofullerene.

The FTIR spectra of $Sc_4C_2@C_{80}$ - I_h and $Sc_3C_2@C_{80}$ - I_h and the DFT-simulated IR spectrum of Sc₄C₂@C₈₀ are shown in Figure 4. First of all, the experimental FTIR spectrum of Sc₄C₂@C₈₀ agrees well with the previously DFT-computed IR spectrum⁹ of Sc₄C₂@C₈₀- I_h (Figure 4). The peaks ranging from 1600 to 1000 cm⁻¹ can be considered as a group of tangential cage modes of the outer carbon cage. 12 The characteristic vibrations of the tangential cage modes are approximately at 1184, 1359, 1459, and 1513 cm⁻¹ for $Sc_4C_2@C_{80}$ - I_h . These tangential cage modes of $Sc_4C_2@C_{80}$ - I_h exhibit quite a degree of resemblance to those of the well-known species $M_3N@C_{80}$ - I_h (M = Sc, Dy, Tm, Gd). ¹² The vibrational frequency at 480 cm⁻¹ can be assigned to the radial cage mode. 12 Signals at 600, 645, and 688 cm⁻¹ represent asymmetric Sc-C_{carbide} stretching vibrations, indicating that there are three kinds of Sc-Ccarbide stretching modes in Sc₄C₂@C₈₀-I_h. This assignment is in line with the calculated structure, which presents three groups of Sc-C_{carbide} bonds according to the various bond lengths (Figure 3).

The low-energy Raman spectrum of $Sc_4C_2@C_{80}$ - I_h (Figure 4) shows a resemblance to those of $Sc_3N@C_{80}$ - I_h , $Dy_3N@C_{80}$ - I_h , and $Sc_3CH@C_{80}-I_h$. The line groups at ~481 and 240 cm⁻¹ are characteristic of the A_g(1)- and H_g(1)-derived cage modes of C₈₀- I_h , respectively.¹³ It should be noted that the cage modes show obvious splittings, such as those of the A_g(1)-derived modes at 470, 481, and 490 cm $^{-1}$ and the $H_g(1)$ -derived cage modes at 230, 240, and 249 cm⁻¹. Such big splittings suggest that distortions in the icosahedral C₈₀ cage result from presence of the big Sc₄C₂ cluster. ^{13c} Two groups of lines at \sim 208 and 407 cm⁻¹ also give splittings caused by the complex structure of the Sc₄C₂ cluster (Figure 3). The medium line at 208 cm⁻¹ and weak line at 187 cm⁻¹ can be assigned to the Sc₄C₂ translation modes, and the group of lines around 407 cm $^{-1}$ represents the $\nu_s(Sc-C)$ modes of the Sc_4C_2 moiety.¹³ Therefore, the Raman spectral analysis confirms the $Sc_4C_2@C_{80}$ - I_h structure.

In conclusion, for the first time we have produced stable $Sc_4C_2@C_{80}$ - I_h and characterized it as a metal carbide endofullerene by FTIR and Raman spectra in combination with DFT calculations. Furthermore, DFT calculations have demonstrated that this molecule has a Russian-doll-type structure, C₂@Sc₄@C₈₀. To the best of our knowledge, $Sc_4C_2@C_{80}$ - I_h is the first metallofullerene that exhibits a Russian-doll nesting structure.

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Supporting Information Available: Experimental details, UV-vis-NIR spectrum, and CV and HPLC data. This material is available free of charge via the Internet at http://pubs.acs.org.

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