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A Novel and Efficient Route to Zafirlukast

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Abstract:

Zafirlukast, an important drug for allergic pulmonary disorders such as asthma, is synthesized by a five-step, high-yielding, and inexpensive process.

Introduction

Zafirlukast is a potential drug used in the treatment of pulmonary disorders such as asthma.¹ Zafirlukast acts by antagonizing the pharmacological actions of one or more of the arachidonic acid metabolites known as leukotrienes.² The reported synthetic method for this anti-asthma drug involved eight stages (Scheme 1), esterification of 1 using acetyl chloride in methanol to give 2, which upon bromination using free bromine in carbon tetrachloride yielded bromo compound 3. Condensation of 3 with 5-nitro indole (4) in the presence of silver oxide and dioxane yielded the corresponding product 5 which on methylation with methyl iodide afforded N-methylated compound 6, whose nitro group reduction gave amino compound 7. Reaction of 7 with cyclopentyl chloroformate afforded the corresponding amide ester 8. Methyl ester hydrolysis in 8 with lithium hydroxide in THF/MeOH mixture to give the corresponding acid 9, and its subsequent condensation with o-toluenesulfonamide afforded zafirlukast (10). This process suffers with the disadvantage of producing some unwanted byproducts such as polysubstituted indoles.³ Further, this process involves column chromatography purification, which is not very practical for industrial scale-up. In addition to this, usage of hazardous chemicals such as acetyl chloride, bromine, and sodium hydride make this method less attractive for industrial scale-up. A recent patent described a multistep method for

the preparation of zafirlukast, which involves an expensive reagent, 1-[3-(dimethyl-amino) propyl]-3-ethylcarbodiimide hydrochloride.⁴ Herein we report a five-step, high-yielding, and inexpensive process for Zafirlukast.

Results and Discussion

In our approach we have identified 3-methoxy-4-methylbenzoic acid (1), a readily accessible and commercially available compound, as an apt starting material. Bromination of 1 with different brominating agents such as bromine or N-bromosuccinamide with different solvents such as dichloromethane, choroform, ethylenedichloride, carbon tetrachloride and dioxane combined with dibenzylperoxide at different reaction conditions yielded low-quality and poor yields, and the reaction did not proceed to completion. When we employed 1,3-dibromo-5,5-dimethylhydantoin and 2,2'-azobis-isobutyronitrile in chloroform, the reaction afforded bromomethylbenzoic acid derivative (11) in 81% yield. This brominating agent is relatively cheaper and easier to handle. As it supplies two bromine atoms, its half equivalent is sufficient to promote the reaction. Condensation of 11 with N-methyl-5-nitro indole (12) in dioxane in the presence of Ag₂O as per reported conditions resulted in several impurities along with the indole derivative 13. However, when the reaction was conducted in the presence of zinc bromide at 60-65 °C, it furnished indole derivative (13) in 62% yield. The required N-methyl-5-nitro indole 12 is readily obtained from the methylation of commercially available 5-nitroindole.4 The presence of any alkyl group as the indole nitrogen effectively arrested the formation of N-arylkyl byproducts. Further, the selectively of the reaction at the third position of the indole moiety may also be attributed to the presence of an N-methyl group which sterically hinders the attack at the second position in the indole moiety. The inexpensive and commercially available dicyclohexylcarbodiimide facilitated a facile condensation of indole-substituted benzoic acid derivative 13 with o-toluenesulfonamide in the presence of 4-(dimethylamino)pyridine as catalyst to give required intermediate 14 in 63% yield. Catalytic hydrogenation of 14 using Raney-nickel in methanol smoothly afforded amino compound 15 which, on reaction with cyclopentylchloroformate, yielded the desired product 10 in 89% yield (Scheme 2).

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Scheme 1a

^a Reagents and conditions: a) AcCl, MeOH; b) Br₂/CCl₄; c) Ag₂O/dioxane; d) CH₃I/THF; e) Pd/C, MeOH, H₂; f) cyclopentyl chloroformate/N-methylmorpholine; g) LiOH•H₂O/THF:MeOH; h) 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride, DMAP, o-toluenesulfonamide, CH₂Cl₂.

Scheme 2ª

^a Reagents and conditions: i) DBDMH/AIBN/CHCl₃, reflux, 81%; ii) zinc bromide/dioxane/60-65 °C, 62%; iii) dicyclohexyl carbodiimide, DMAP, o-toluenesulfonamide, CH₂Cl₂, 63%; iv) Ra-Ni,THF, H₂, 88%; v) cyclopentylchloroformate/N-methylmorpholine, CH₂Cl₂.

Conclusions

In conclusion, we have provided a novel, cost-effective, and industrially scaleable approach to the anti-asthama drug, zafirlukast.

Experimental Section

The ^1H and $^{13}\text{CNMR}$ spectra were measured in CDCl₃ and DMSO, using 200 and 50 MHz, respectively, on a Varian Gemini 200 MHz FT NMR spectrometer; the chemical shifts are reported in δ ppm relative to TMS. The FT-IR spectra were recorded in the solid state as KBR dispersion using Perkin-Elmer 1650 FT-IR spectrophotometer. The mass

spectrum (70 eV) was recorded on HP-5989A LC-MS spectrometer. The CHN analysis was carried out on a Perkin-Elmer model 2400S analyzer. The melting points were determined by using the capillary method on POLMON (model MP-96) melting point apparatus. The solvents and reagents were used without further purification.

4-Bromomethyl-3-methoxybenzoic Acid (11). A mixture of 3-methoxy-4-methyl benzoic acid (**1**, 50.0 g, 0.30 mol), 1,3-dibromo-5,5-dimethylhydantoin (55.0 g, 0.192 mol), and 2,2'-azobisisobutyronitrile (5.0 g, 0.030 mol) in chloroform (250 mL) was heated under reflux for 4 h. The reaction mixture was cooled to 25-35 °C, the formed byproduct was

filtered, and the filtrate was washed with water (150 mL). The organic layer was separated and concentrated under vacuum. The obtained residue was triturated with petroleum ether (150 mL), and the obtained solid was filtered and dried under vacuum: yield 60.0 g, 81%; mp 184–190 °C.

 1 H NMR (CDCl₃): δ 8.9 (br s, 1H), 7.7–7.3 (m, 3H), 4.59 (s, 2H), 3.9 (s, 3H); 13 C NMR: δ 179.2, 166.9,157.1, 132.6, 131.0, 130.5, 121.7, 55.8, 28.61. MS: m/e = 245 M $^{+}$. Anal. Calcd for C₉H₉BrO₃: C, 44.11; H, 3.70. Found: C, 44.08; H, 3.67.

3-Methoxy-4-(*N***-methyl-5-nitroindole-3-yl)benzoic Acid (13).** A mixture of *N*-methyl-5-nitroindole **(12,** 50.0 g, 0.284mol), 4-bromomethyl-3-methoxybenzoic acid (125.0 g, 0.510 mol), zinc bromide (37.5 g, 0.161 mol), and dioxane (250 mL) was stirred at 60–65 °C for 14 h. Reaction mass was filtered through hyflow and washed with dioxane (100 mL). The filtrate was evaporated and dissolved in dichloromethane (150 mL), basified with sodium hydroxide (10.0 g, 0.25 mol) solution. The basic aqueous layer was acidified with hydrochloric acid (12.5 mL) and was extracted with dichloromethane (250 mL). The organic layer was washed with water (250 mL) and concentrated, and a fine solid was isolated in methanol (150 mL): yield 60.0 g, 62%; mp 234–240 °C.

¹H NMR (CDCl₃): δ 8.5 (s, 1H), 8.05 (d, 1H), 7.6–7.2 (m, 4H), 6.9 (s, 1H), 4.2 (s, 2H), 3.9 (s, 3H), 3.85 (s, 3H); ¹³C NMR: δ 171.0, 156.6, 140.2, 139.4, 134.0, 130.2, 129.7, 125.9, 121.7, 118.7, 115.3, 110.8, 55.4, 32.7, 24.4. MS: m/e = 341 M⁺. Anal. Calcd for C₁₈H₁₆N₂O₅: C, 63.52; H, 4.74; N, 8.23. Found: C, 63.40; H, 4.68; N, 8.19.

*N*1-{3-Methyl-4-[(*N*-methyl-5-nitro-1*H*-3-indolyl)methylbenzoyl}-2-methyl-1-benzenesulfonamide (14). A mixture of 3-methoxy-4-(*N*-methyl-5-nitroindole-3-yl)benzoic acid (13, 50.0 g, 0.147 mol), 4-(dimethylamino)pyridine (5.0 g, 0.040 mol), 1,3-dicyclohexylcarbodiimide (30.0 g, 0.145 mol), and o-toluenesulfonamide (25.0 g, 0.146 mol) in dichloromethane (250 mL) was stirred under reflux for 5 h. After the reaction was cooled to 25-35 °C, the solid was filtered, the filtrate was washed with a solution of hydrochloric acid (25 mL) and water (225 mL) followed by rinsing with water, and the organic layer was concentrated under vacuum. The resulting crude material was dissolved in methanol (150 mL), cooled to 0 °C for 30 min; the byproduct (dicyclohexylurea) was filtered, and the solid was washed with cool methanol (25 mL), and the filtrate was concentrated under vacuum: yield 45.5 g, 63%.

¹H NMR (CDCl₃): δ 9.05 (br s, 1H), 8.51 (d, 1H), 8.25 (d, 1H), 8.11 (d, 1H), 7.5–7.1 (m, 7H), 6.92 (s, 1H), 4.10 (s, 2H), 3.88 (s, 3H), 3.78 (s, 3H), 2.68 (s, 3H). ¹³C NMR:

δ 164.6, 156.3, 140.2, 139.1, 136.8, 134.1, 131.7, 130.3, 129.1, 126.3, 125.4, 120.5, 115.7, 115.0, 109.0, 55.1, 32.5, 24.3, 19.5. MS: m/e = 493 M⁺. Anal. Calcd for C₂₅H₂₃-N₃O₆S: C, 60.84; H, 4.70; N, 8.51. Found: C, 60.72; H, 4.65; N, 8.46.

N1-{4-[(5-Amino-N-methyl-1H-3-indolyl)methyl]-3-methylbenzoyl}-2-methyl-1-benzenesulfonamide (15). A solution of N-{3-methyl-4-[(N-methyl-5-nitro-1H-3-indolyl)methylbenzoyl}-2-methyl-1-benzenesulfonamide (14, 40.0 g, 0.081 mol) tetrahydrofuran (120 mL), and Raney-nickel (8 mL) was placed under par hydrogenation apparatus under hydrogen (3.0 kg) pressure at 25-35 °C for 2 h. The reaction mass was filtered through a hyflow bed, and the catalyst was washed with tetrahydrofuran (40 mL). The combined filtrates were concentrated, and the obtained residue was triturated with petroleum ether (150 mL) resulting in 33.2 g, 88% yield of compound 15 as a fine solid.

¹H NMR: δ 7.93 (d, 1H), 7.57–6.67 (m, 10H), 3.88 (s, 2H), 3.85 (s, 3H), 3.69 (s, 3H), 2.49 (s, 3H); ¹³C NMR: δ 164.6, 157.2, 144.7, 138.3, 138.0, 136.3, 135.2, 134.8, 134.6, 132.1, 129.5, 125.6, 125.0, 121.3, 114.7, 110.8, 108.0, 55.6, 32.5, 22.9, 20.3. MS: m/e = 463 M⁺. Anal. Calcd for C₂₅H₂₅N₃O₄S: C, 64.78; H, 5.44; N, 9.06. Found: C, 64.70; H, 5.38; N, 8.99.

[3[[2-Methoxy-4-[[[(2-methyl phenyl)sulfonyl]amino]-carbonyl]phenyl]methyl]-*N*-methyl-1*H*-indole-5-yl]carbamicacid Cyclopentylester (10). To a solution of *N*1-{4-[(5-amino-*N*-methyl-1*H*-3-indolyl)methyl]-3-methylbenzoyl}-2-methyl-1-benzenesulfonamide (15, 25.0 g, 0.053 mol) was added dichloromethane (125 mL), *N*-methylmorpholine (25 mL, 0.227 mol), and cyclopentylchloroformate (11.25 mL, 0.087 mol) at 25-35 °C under nitrogen atmosphere, and the solution was stirred for 30 min. The reaction mixture was quenched slowly with a solution of hydrochloric acid (12.5 mL) and water (112.5 mL). The separated organic layer was washed with water (125 mL), and the solvent was completely removed under vacuum: yield as a solid 27.5 g, 89%.

¹H NMR (CDCl₃): δ 9.67 (s, 1H), 8.23 (d, 1H), 7.01–7.51 (m, 9H), 6.73 (s, 1H), 6.58 (s, 1H), 5.08–5.23 (m, 1H), 3.98 (s, 2H), 3.79 (s, 3H), 3.73 (q, 4H), 3.67 (s, 3H), 2.66 (s, 3H), 1.45–1.87 (m, 8H), 1.22 (t, 3H). ¹³C NMR: δ 164.6, 156.3, 135, 127, 109, 77, 55, 32.4, 23.3. MS: m/e = 575 M⁺. Anal. Calcd for C₃₁H₃₃N₃O₆S: C, 64.68; H, 5.78; N, 7.30. Found: C, 64.61; H, 5.68; N, 7.28.

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