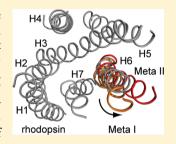


Structural Transitions of Transmembrane Helix 6 in the Formation of Metarhodopsin I

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ABSTRACT: Absorption of light by the visual pigment rhodopsin triggers a rapid cis-trans photoisomerization of its retinal chromophore and a series of conformational changes in both the retinal and protein. The largest structural change is an outward tilt of transmembrane helix H6 that increases the separation of the intracellular ends of H6 and H3 and opens up the Gprotein binding site. In the dark state of rhodopsin, Glu247 at the intracellular end of H6 forms a salt bridge with Arg135 on H3 to tether H6 in an inactive conformation. The Arg135-Glu247 interaction is broken in the active state of the receptor, and Arg135 is then stabilized by interactions with Tyr223, Met257, and Tyr306 on helices H5, H6, and H7, respectively. To address the mechanism of H6 motion, solid-state NMR measurements are undertaken of Metarhodopsin I (Meta I), the intermediate preceding the active Metarhodopsin II (Meta II)



state of the receptor. ¹³C NMR dipolar recoupling measurements reveal an interhelical contact of ¹³Cζ-Arg135 with ¹³Cε-Met257 in Meta I but not with ¹³Cζ-Tyr223 or ¹³Cζ-Tyr306. These observations suggest that helix H6 has rotated in the formation of Meta I but that structural changes involving helices H5 and H7 have not yet occurred. Together, our results provide insights into the sequence of events leading up to the outward motion of H6, a hallmark of G protein-coupled receptor activation.

■ INTRODUCTION

The visual receptors are members of the large family of G protein-coupled receptors (GPCRs). 1,2 These receptors all contain seven transmembrane helices, designated H1-H7 (Figure 1), and have the ability to activate intracellular heterotrimeric G proteins. However, the visual receptors are unique in being activated by light-induced isomerization of a covalently bound retinal chromophore rather than by binding of a diffusible ligand. Nevertheless, much of our current understanding of the structure and activation of GPCRs has emerged from structure-function studies of the visual receptors, particularly the vertebrate dim-light photoreceptor rhodopsin.

It has been known since the pioneering EPR studies of Hubbell and co-workers using site-directed spin labeling that activation of rhodopsin involves the outward rotation of transmembrane helix H6.3 However, a mechanistic understanding of how retinal isomerization is coupled to H6 motion has been lacking. The first high-resolution crystal structure of rhodopsin was obtained in 2000 of the inactive, dark state of the receptor (Figure 1).4 This structure revealed the position of highly conserved residues within the large family of Class A GPCRs but provided few clues about the mechanism of receptor activation. The crystal structure of opsin, which captures the outward rotation of H6, revealed several key elements of the activation mechanism. 5,6 In the dark state of the rhodopsin, Arg135^{3.50} forms a salt bridge with Glu247^{6.30} on H6. (The amino acid numbering used in this manuscript incorporates the residue number from the amino acid sequence of the specific receptor being discussed (e.g., Lys296) and a residue number from a generic numbering system developed by Ballesteros and Weinstein⁸⁷ (e.g., Lys296^{7,43}) that gives the position of an amino acid relative to the most conserved amino acid (designated 50) on a specific helix. In this example, Lys296^{7.43} on H7 is seven residues toward the N terminus from the most conserved residue on H7, Pro303^{7.50}. Because sequence alignments are poor for the extracellular and intracellular loops, as well as for the N and C termini, a generic superscript (e.g., EL2) is used to designate the position of a nontransmembrane residue.) In opsin, the rotation of H6 moves Glu247^{6.30} toward H5 where it forms a new electrostatic interaction with Lys231^{5.66}. H6 motion also places Met257^{6.40} in contact with Arg135^{3.50} on the transmembrane helix H3. However, this latter interaction appears to require the outward tilt of H6, which displaces the intracellular end of H6 by ~5 Å relative to H3.7 That is, the interface between H3 and H6 is tightly packed, and the outward motion of H6 allows the side chain of Arg135^{3.50} to extend toward Met257^{6.40} and make contact (see Figure 1C). In addition, the opsin crystal structure showed that the intracellular ends of both helices H5 and H7

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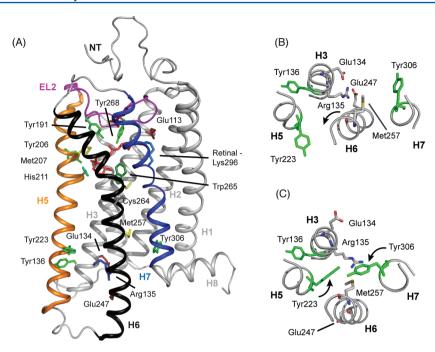


Figure 1. Crystal structures of rhodopsin (PDB access code = 1GZM) and Meta II (PDB access code = 3PXO) highlighting key residues associated with photoactivation. ⁸⁸ (A) The full receptor structure is shown. The retinal is buried between the transmembrane helices on the extracellular side of the protein with the second extracellular loop (EL2) forming a cap on the retinal binding site. The motion of helices H5, H6, and H7 during activation opens up the G-protein binding site on the intracellular side of the receptor, while helices H1—H4 form a tightly packed core. (B) Cross section of the intracellular side of rhodopsin in the region of the Arg135^{3.50}-Glu247^{6.30} salt bridge. Only selected helices are shown for clarity. (C) Structure of the intracellular side of Meta II in the region of the ionic lock. There are several differences between inactive rhodopsin (panel B) and Meta II (panel C). Rotation of H6 breaks the Arg135^{3.50}-Glu247^{6.30} contact and moves Met257^{6.40} into the H3—H6 interface. The intracellular end of H6 tilts outward, allowing the side chain of Arg135^{3.50} to extend and contact Met257^{6.40}. Rotation of H5 and H7 place Tyr223^{5.58} and Tyr306^{7.53} into contact with Arg135^{3.50}. Hydrogen bonds between these tyrosines and Arg135^{3.50} stabilize the active state of the receptor.

undergo rotation to allow two highly conserved tyrosine residues to hydrogen bond with $Arg135^{3.50}$. $Tyr223^{5.58}$ on H5 and $Tyr306^{7.53}$ on H7 rotate toward $Arg135^{3.50}$ in the active receptor and contribute to breaking of the $Arg135^{3.50}$ - $Glu247^{6.30}$ salt bridge. The combined effect of these structural changes is to open up a cavity on the intracellular side of the receptor that serves as the binding site for the C terminal tail of the $G\alpha$ subunit of transducin, the heterotrimeric G protein that associates with light-activated rhodopsin.

Several observations argue that the motions of helices H5, H6, and H7 are part of a common activation mechanism for Class A GPCRs. First, several of the residues described above are highly conserved. Arg135^{3.50} is part of the conserved ERY motif on helix H3 (i.e., residues Glu134^{3.49}-Arg135^{3.50}-Tyr136^{3.51}). Tyr223^{5.58} is the most highly conserved residue on H5, while Tyr306^{7.53} is part of a conserved NPxxY sequence on H7. Second, recent crystal structures of the β_2 -adrenergic receptor with bound agonists show an outward displacement of H6 and confirm that this signature structural change for rhodopsin activation is present in a ligand-activated GPCR.^{8,9} Nevertheless, the large changes in H6 in these crystal structures are only observed when the receptor is in complex with an antibody mimicking the G protein. The structural changes of the β_2 -adrenergic receptor¹⁰ and other ligand-activated GPCRs^{11,12} without stabilizing antibodies exhibit structural changes that are much smaller than those observed in the active state of rhodopsin. One possible explanation for the differences between rhodopsin and the ligand-activated receptors is that rhodopsin has evolved several mechanisms to reduce basal activity, including the salt bridge between Arg1353.50 and Glu247^{6,30}, which are essential for a visual receptor. In contrast,

ligand-activated receptors typically exhibit high levels of basal activity. These receptors are inherently dynamic molecules that interconvert between transiently populated conformational states. ¹³ As a result, if the visual receptors and ligand-activated receptors share a common activation mechanism, there are likely interactions that restrict H6 motion in rhodopsin that are released following photoisomerization but prior to the conformational step that generates the active receptor.

Figure 2 presents the photoreaction pathway leading from dark, inactive rhodopsin to the active Metarhodopsin II (Meta II) intermediate. In the dark state, the retinal chromophore is bound as a protonated Schiff base to Lys296^{7,43} on transmembrane helix H7. Absorption of light triggers the rapid isomerization of the retinal from the 11-cis to the all-trans configuration. The early intermediates contain a conformationally distorted all-trans retinal chromophore. 14 Calorimetric studies have shown that ~30 kcal/mol of the absorbed light energy is stored in Bathorhodopsin, the first relatively longlived intermediate. 15 The trapped energy is released as the retinal relaxes 16-19 and the surrounding amino acids reorient in the transitions to the blue-shifted intermediate, Lumirhodopsin and Metarhodopsin I (Meta I). Meta I immediately precedes the activated Meta II state. The current study focuses on the orientation and interactions involving helix H6 in Meta I to establish whether conformational changes occur in this helix leading up to the active Meta II state.

The retinal chromophore in the Meta I intermediate has an all-trans configuration and exhibits an absorption maximum ($\lambda_{\rm max}$) at 480 nm. There are no high-resolution crystal structures of Meta I. However, a 5.5 Å resolution structure of Meta I obtained by electron cryomicroscopy of 2D crystals

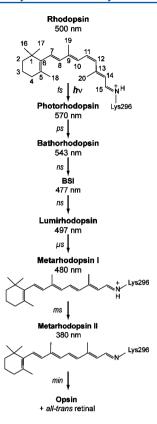


Figure 2. Photoreaction of rhodopsin. Structures of the 11-cis and all-trans retinal chromophores and the photoreaction intermediates of rhodopsin are shown. Absorption of light results in 11-cis to all-trans isomerization of the retinal. The retinal—protein complex subsequently relaxes thermally through a series of spectrally well-defined intermediates: Photorhodopsin, Bathorhodopsin, the Blue-Shifted Intermediate (BSI), Lumirhodopsin, and Meta I. Deprotonation of the retinal Schiff base nitrogen occurs in the formation of the Meta II intermediate. Meta I exists in a pH- and temperature-dependent equilibrium with the active Meta II intermediate. High pH and low temperature favor Meta I. Evidence for substantial structural changes in Meta I comes from time-resolved absorbance measurements showing that the transition to Lumirhodopsin is the last step in the photoreaction sequence that does not exhibit differences between lipid and detergent environments. In addition, Meta I is the first intermediate that cannot be fully photoconverted back to rhodopsin.

showed no significant displacements of the transmembrane helices as compared to the dark state of rhodopsin.²⁰ The largest change in Meta I relative to rhodopsin was in the region of Trp265^{6.48} on helix H6,²⁰ which suggested a local change either in the conformation of the Trp265^{6.48} side chain or in the rotational orientation of the H6 helix.

In contrast to the low-resolution structure of Meta I, a number of biophysical studies have revealed conformational changes that stretch from the retinal binding site on the extracellular side of the receptor to the G-protein binding site on the intracellular surface in Meta I. On the extracellular side of rhodopsin, Fourier transform infrared (FTIR) spectroscopy shows that $Glu122^{3.37}$ becomes more strongly hydrogen bonded in the transition to Meta $I.^{21}$ $Glu122^{3.37}$ is located on helix H3 near the retinal β -ionone ring, and hydrogen bonds to the backbone carbonyl of His211^{5.46} on H5. These residues are part of a hydrogen-bonding network that extends to the second extracellular loop (EL2). Coupled motion of the retinal and H5

has been implicated in the transition to the active Meta II state. ²²

On the intracellular side, FTIR measurements of rhodopsin containing *p*-azido-L-phenylalanine, an engineered non-native amino acid, show that there are strong changes in polarity at the ends of helices H5 and H6 in the formation of Meta I. These changes are consistent with a rotation of H6 and movement of the intracellular end of H5 away from H3.²³ Furthermore, Meta I is the first substrate for rhodopsin kinase,²⁴ which binds to Meta I and Meta II with equal affinity,²⁵ and has been reported to be able to bind the G protein transducin.^{26,27} Together these results imply that a conformational change on the extracellular side of the receptor, induced by retinal isomerization, leads to changes on the intracellular surface of Meta I.

Solid-state NMR spectroscopy provides a complementary approach to X-ray crystallography and FTIR spectroscopy for probing the structure of Meta I and defining the structural changes that occur as light energy is channeled into rhodopsin. Previous solid-state NMR studies on Meta I were restricted to the chromophore and showed that the retinal polyene chain is in a relaxed conformation. 17,18,28,29 Here, we describe 13C and ¹⁵N solid-state magic angle spinning (MAS) NMR measurements on Meta I trapped in digitonin that target structural changes in the protein centered on helix H6. Digitonin is unusual compared to other detergents as the hydrophobic end of the digitonin is composed of a rigid spirostan steroid tail rather than flexible fatty acyl chains. It has been used extensively to stabilize the Meta I intermediate $^{30-32}$ since it was first shown to be effective in blocking the transition from Meta I to Meta II.³³ The ability of digitonin to trap Meta I is comparable to that of cholesterol previously used to trap Meta I in 2D crystals of rhodopsin.²⁰ Both digitonin and cholesterol have a rigid steroidal structure that prevents the Meta I to Meta II transition. In contrast, the transition to the Meta II intermediate is facilitated by solubilization of rhododpsin in the *n*-dodecyl- β -D-maltoside (DDM) detergent, which creates a fluid environment. Our structural measurements on Meta I can be compared with previous measurements of rhodopsin and Meta II. The comparison with rhodopsin allows us to infer what structural changes occur between the dark state and Meta I, while the comparison with Meta II allows us to infer what structural changes are associated with activation. ¹³C dipolarassisted rotational resonance (DARR) NMR measurements reveal an interhelical contact between ¹³Cζ-labeled Arg135^{3.50} and $^{13}\text{C}\varepsilon\text{-labeled Met257}^{6.40}$, indicating that the transmembrane helix H6 rotates in the formation of Meta I. However, the NMR data show that EL2 and H5 have not shifted into their active state conformations. Our results help to define the sequential structural changes occurring between rhodopsin and Meta I that prime the receptor for transition to the active Meta II state.

EXPERIMENTAL METHODS

Materials. ¹³C-labeled amino acids were purchased from Cambridge Isotope Laboratories (Andover, MA). A 5% (w/v) digitonin stock solution was prepared by dissolving 1 g of digitonin (Calbiochem, La Jolla, CA) in 20 mL of boiling distilled water. The mixture was heated at 95 °C for 5 min to ensure that all of the digitonin had dissolved. Following storage at 4 °C overnight, the solution was filtered through a 0.2 μ m sterile bottle top filter, and any insoluble material was discarded. ³⁴ The 5% (w/v) stock solution was stored at -20 °C.

Expression and Purification of ¹³C- and ¹⁵N-Labeled Rhodopsin. A stable tetracycline-inducible HEK293S cell line containing the opsin gene was used to express stable isotope labeled rhodopsin. 35 Cultured cells were grown in suspension using a bioreactor (New Brunswick Scientific). The calciumreduced³⁶ growth medium (Dulbecco's modified Eagle's medium, Sigma, St. Louis, MO) was supplemented with specific ¹³C-labeled amino acids (Cambridge Isotope Laboratories, Andover, MA), 10% dialyzed, heat-inactivated fetal bovine serum, ³⁶ Pluronic F-68 (0.1%), dextran sulfate (300 mg/L), penicillin (100 units/mL), and streptomycin (100 μ g/ mL). On day 4 after incubation, cells were fed with glucose (2.4 g/L). Opsin gene expression was induced 5 days after inoculation by addition to the growth medium of both tetracycline (2 mg/L) and sodium butyrate (5 mM). Cells were harvested on day 7, and the cell pellets were stored at -80°C.35

The cell pellet obtained from a 3 L bioreactor culture was resuspended in 150 mL of phosphate-buffered saline (PBS) containing protease inhibitors (0.4 mM PMSF and 50 $\mu g/mL$ of benzamidine). Unlabeled 11-cis retinal was added in two steps totaling 250 nmol/g of cell pellet. The rhodopsincontaining cells were then pelleted and suspended in PBS (40 mL/L of culture) containing DDM (1% w/v) for 4 h at room temperature. Subsequent purification by immunoaffinity chromatography using the rho-1D4 antibody was carried out according to existing protocols^{30–32} that were modified to lower the detergent concentration for NMR. For detergent exchange, rhodopsin was washed with 25 column volumes of PBS containing DDM (0.02% w/v), 25 column volumes of PBS containing digitonin (0.1% w/v), and 25 column volumes of PBS containing digitonin (0.02-0.05% w/v). After washing, the column was equilibrated with 10 column volumes of 2 mM phosphate buffer (pH = 7.0) containing digitonin (0.02-0.05% w/v). Rhodopsin was eluted in 2 mM phosphate buffer (pH = 7.0) containing digitonin (0.02–0.05% w/v) and 100 μ M Cterminal nonapeptide. The eluted rhodopsin fractions were pooled and concentrated to a final volume of ~400 µL using Centricon devices with a 10 kDa molecular weight cutoff (Amicon, Bedford, MA), followed by further concentration under a stream of argon gas to a volume of $\sim 100 \mu L$. All buffers were prepared fresh before purification.

Solid-State NMR Spectroscopy. NMR spectra of digitonin-solubilized rhodopsin were acquired at static magnetic field strengths of 8.5 or 14.1 T on Bruker AVANCE spectrometers using either a two-channel or a three-channel 4 mm MAS probe at spinning rates of 8-12 kHz. 1D ¹³C spectra were collected using ramped amplitude cross-polarization with a contact time of 2 ms and an acquisition time of 16 ms. Intermolecular ¹³C···¹³C correlations were obtained using the DARR pulse sequence with a mixing time of 600 ms. The ¹H radiofrequency field strength during mixing was matched to the MAS speed for each sample, satisfying the n = 1 matching condition. Two-pulse phase-modulated proton decoupling of 80-90 kHz was used during both evolution and acquisition.^{37,38} For 2D data sets, 1024 points in the f2 dimension and 64 points in the f1 dimension were acquired. The ¹³C solidstate MAS NMR spectra were externally referenced to the ¹³C resonance of neat TMS at 0.00 ppm at room temperature. Using TMS as the external reference, we calibrated the carbonyl resonance of solid glycine to 176.46 ppm. The ¹⁵N solid-state MAS NMR spectra of rhodopsin and Meta I were referenced to the ¹⁵N resonance of 5.6 M aqueous NH₄Cl at 0.0 ppm at room

temperature. 15 N-labeled glycine was used as an external reference taking the reported value of 8.1 ppm for glycine relative to 5.6 M aqueous NH₄Cl. 39 This 15 N chemical shift reference was used, instead of the IUPAC recommendation of liquid ammonia, 40 for comparison with the results of previous studies on retinal Schiff bases. 36,41

NMR measurements were first made on rhodopsin in the dark. For conversion to Meta I, the NMR rotor was ejected, the cap removed, and the sample illuminated for 1-2 min at 4 °C using a 400 W lamp with a 495 nm high pass filter. On the basis of UV/vis absorption spectra and NMR spectra of Meta I, the conversion from rhodopsin is >90%. After conversion, we estimate that there is a loss of <10% of the Meta I intermediate to Meta II and opsin before the sample is cooled to 190 K for NMR measurements. The residual rhodopsin and Meta II components in the Meta I sample do not interfere with the assignments of resonances originating from Meta I. For conversion to Meta II, samples in DDM were illuminated in the NMR rotor at room temperature. We estimate that the conversion from rhodopsin to Meta II is >90% and the loss of Meta II to opsin is <5% before the sample is cooled to 190 K.⁴² For both Meta I and Meta II, following conversion the NMR rotor is recapped, reinserted into a precooled NMR probe, and frozen within 5-10 min using cold N₂ gas. NMR spectra of dark rhodopsin, Meta I, and Meta II were obtained at a sample temperature of 190 K. The Meta I and II intermediates are completely stable at this temperature.

Restrained Molecular Dynamic Simulations of Meta I. The NMR constraints collected here were used to perform restrained molecular dynamics (MD) simulations with NAMD 2.8^{43} and employed the CHARMM27r parameters for proteins and lipids. 44-52 Retinal parameters were obtained from Saam and co-workers. 53 The simulation system started with the highresolution crystal structure of Lumirhodopsin⁵⁴ embedded in a DOPC:DOPE (1:3) CHARMM-GUI⁵⁵ bilayer by the replacement method with the cytoplasmic and extracellular (or intradiscal) regions solvated by water and ionized using VMD.⁵⁶ There were a total of 97 459 atoms: 348 protein residues, 22 crystallographic waters, 9250 TIP3P water molecules, 36 Na⁺ ions, 39 Cl⁻ ions, and 240 lipids (60 DOPC and 180 DOPE). Simulations were run at a temperature of 303 K and a pressure of 1 bar in the NPAT ensemble, in which the pressure, temperature, and area of the membrane were kept constant, using Langevin dynamics and the particle mesh Ewald method for full system periodic electrostatics. We arrive at the model of Meta I by equilibrating the membraneprotein system and then applying the restraints obtained previously by NMR^{22,57,58} and described in the Results (e.g., Met257-Arg135, 5 Å with k = 10 kcal/mol Å²). To allow the lipids to equilibrate, the simulations were run for a total of 5 ns with decreasing constraints on all heavy atoms, protein backbone, and retinal from 20 to 0.1. A time step of 1 fs was used for the first two steps of equilibration and then switched to 2 fs. To allow the protein to equilibrate, the simulations were run for an additional 6 ns before restraints were applied. Restraints were applied with the extra bonds feature in NAMD and with k values between 10 and 20 kcal/mol $Å^2$ and run for 2 ns.

RESULTS

Trapping of the Metarhodopsin I Intermediate. The Meta I intermediate is characterized by a visible absorption maximum at 480 nm. Figure 3A presents absorption spectra

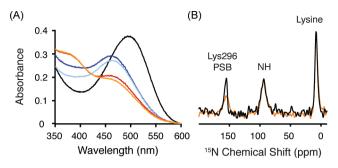


Figure 3. Trapping of the Meta I intermediate in digitonin. (A) UV–vis spectra provide a means to follow the conversion of rhodopsin ($\lambda_{\rm max}$ = 500 nm, black line) to Meta I ($\lambda_{\rm max}$ = 480 nm, blue line) after illumination by light (>495 nm) at 4 °C. Meta I remains stable in digitonin for over 30 min at 4 °C (light blue line). Conversion at 20 °C leads to a mixture of Meta I and Meta II (red line), which does not change appreciably over 30 min (orange line). (B) One-dimensional ¹⁵N spectra of rhodopsin (black) and Meta I (orange) labeled at ¹⁵Nζ-lysine are shown that were obtained using ¹H–¹⁵N cross-polarization. The ¹⁵Nζ-Lys296^{7,43} chemical shift is observed as a distinct narrow peak at 156.8 ppm in rhodopsin (black). In Meta I, the resonance shifts slightly and broadens. The ¹⁵Nζ resonances for the other ¹⁵Nζ-labeled lysines in rhodopsin are observed as a broad peak at ~8.0 ppm.

showing the conversion of rhodopsin ($\lambda_{\rm max}=500$ nm, black line) to Meta I ($\lambda_{\rm max}=480$ nm, blue line) following illumination by light with wavelengths longer than 495 nm. The stability of Meta I in digitonin is temperature dependent. At 4 °C, rhodopsin is fully converted to the Meta I intermediate and is stable for over 30 min (light blue line). When rhodopsin in digitonin is illuminated at 20 °C, a mixture of Meta I and Meta II ($\lambda_{\rm max}=380$ nm) is formed (red line). This mixture is stable for over 30 min (orange line). The temperature dependence of the transition to Meta II has previously been characterized. 30,59,60 The Meta II substrate that is obtained at 20 °C in digitonin likely corresponds to Meta IIa on the basis of EPR measurements showing that digitonin effectively blocks the large increase in the mobility of a spin label at residue

250^{6.33} near the intracellular end of H6.⁶⁰ The outward tilt of H6 is associated with the Meta IIa to Meta IIb transition.⁶⁰

A second characteristic feature of the Meta I state is that the retinal chromophore has an all-trans configuration and is attached to Lys296^{7,43} as a protonated Schiff base. To address the protonation state of the retinal Schiff base linkage, Figure 3B presents the $^{15}{\rm N}$ MAS NMR spectra of rhodopsin (black) and Meta I (orange) containing $^{15}{\rm N}\zeta$ -labeled lysine. The $^{15}{\rm N}\zeta$ -Lys296^{7,43} resonance in rhodopsin (black) is observed at 156.8 ppm. In Meta I, the $^{15}{\rm N}\zeta$ -Lys296^{7,43} resonance broadens and shifts slightly to 155.7 ppm. The similar $^{15}{\rm N}$ chemical shifts between rhodopsin and Meta I demonstrate that the Schiff base nitrogen is protonated. Deprotonation of the Schiff base has a dramatic effect on the $^{15}{\rm N}\zeta$ -Lys296^{7,43} chemical shift, which is observed at 282.2 ppm in Meta II.

We also measured the ¹³C chemical shifts of Meta I regenerated with several selectively ¹³C-labeled retinals (Figure 4). There is a good correspondence of the NMR frequencies with other studies on Meta I.^{17,18,28} For example, the ¹³C10 and ¹³C11 retinal resonances in Meta I at 130.6 and 139.2 ppm, respectively, are in agreement with previous measurements of Meta I trapped in lipids.¹⁷ Although both saturated lipids and digitonin result in observable broadening of the ¹³C resonances of the polyene chain,²⁸ we can show in Figure 4 that in digitonin there is good conversion from rhodopsin (>90%) with less than 10% decay to Meta II or opsin. The ability to photoconvert rhodopsin to Meta I may be associated with the reduction in light scattering from detergent micelles, as compared to lipid multilayers.

Structural Changes in the Region of EL2. The second extracellular loop (EL2) of rhodopsin is part of the retinal binding pocket.⁴ NMR measurements on Meta II have revealed structural changes in EL2²² and in the vicinity of the retinal chromophore.⁶¹ We compare our previous results on rhodopsin and Meta II with NMR measurements of the Meta I intermediate in the region of EL2.

Characteristic changes in ¹³C chemical shifts have been observed previously in NMR difference spectra between

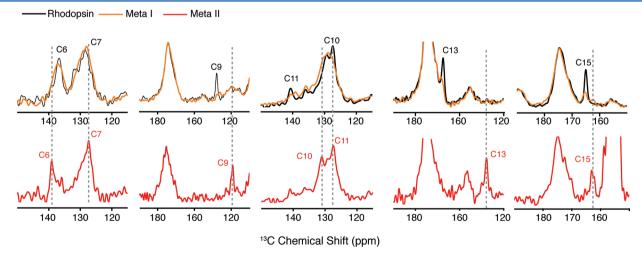


Figure 4. One-dimensional 13 C MAS NMR spectra of rhodopsin and Meta I. Rhodopsin was regenerated with 11 -cis retinal selectively 13 C labeled at different carbons along the polyene chain and the β-ionone ring. Overlap of the 13 C MAS NMR spectra of rhodopsin (black) and Meta I (orange) shows that most of the retinal resonances broaden considerably in Meta I as compared to the sharp narrow resonances observed in rhodopsin and Meta II (red). There is nearly complete conversion of rhodopsin to Meta I. For example, the 13 C9 retinal resonance in rhodopsin falls in an uncrowded region of the spectrum. The residual intensity of the 13 C9 resonance in the Meta I spectrum is <10% of its original intensity in rhodopsin. Also, there is <10% conversion of rhodopsin to Meta II. For example, the spectra of Meta I containing 13 C13-labeled retinal show a complete absence of a resonance associated with Meta II.

rhodopsin and Meta II for tyrosine, methionine, glycine, and cysteine on EL2 and EL3. ^{22,61} Figure 5A shows the difference

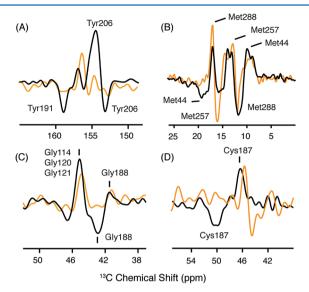


Figure 5. ¹³C MAS difference spectra of rhodopsin, Meta I, and Meta II. Difference spectra are shown for rhodopsin-Meta I (orange) and rhodopsin-Meta II (black) using rhodopsin isotopically labeled with ¹³Cζ-tyrosine, ¹³Cε-methionine, ¹³Cα-glycine, and ¹³Cβ-cysteine. Positive peaks correspond to rhodopsin, and negative peaks correspond to Meta I or Meta II. (A) The tyrosine ¹³Cζ difference spectrum between rhodopsin and Meta II exhibits two positive and two negative resonances. The two Meta II resonances at 158.9 and 153.1 ppm have been assigned to Tyr191^{EL2} and Tyr206^{5.41}, respectively.²² These resonances are not observed in the rhodopsin— Meta I difference spectrum. (B) The methionine $^{13}\text{C}\varepsilon$ difference spectrum between rhodopsin and Meta II exhibits three distinct positive and two negative resonances. The resonances associated with Met44 and Met288 have been assigned in both rhodopsin and Meta II.²² The Met257 resonance does not change between rhodopsin and Meta II. A strong negative cross peak is observed at 15.8 ppm and assigned to Met257 in Meta I. (C) The glycine 13 Clpha difference spectrum exhibits two positive and two negative resonances. The upfield resonances in rhodopsin at 41.5 ppm and in Meta II at 42.9 ppm have been assigned to Gly188^{EL2}. The appearance of a negative resonance at the position of Gly188^{EL2} is not observed in the rhodopsin–Meta I difference spectrum. (D) In the cysteine 13 C β difference spectrum between rhodopsin and Meta II, the chemical shift of Cys187^{EL2} changes from 46.8 to 50.1 ppm.²² The disulfide bridge between Cys187^{EL2} and Cys110^{3.25} tethers EL2 to the helical bundle.

spectra for $^{13}C\zeta$ -tyrosine between rhodopsin and Meta I (orange) and between rhodopsin and Meta II (black). Positive peaks correspond to rhodopsin, and negative peaks correspond to Meta II. The $^{13}C\zeta$ -tyrosine chemical shift is sensitive to hydrogen bonding of the side chain $C\zeta$ -OH group. There are two distinct resonances that appear upon the formation of Meta II. The downfield resonance at 158.9 ppm corresponds to Tyr191 on EL2 (i.e., Tyr191 $^{\text{EL2}}$), which becomes more strongly hydrogen bonded in Meta II. The upfield resonance at 153.1 ppm corresponds to Tyr206 $^{5.41}$, which becomes more weakly hydrogen bonded in Meta II. These two resonances are not observed in Meta I indicating that EL2 and H5 have not shifted into the conformation characteristic of the active state.

Figure 5B presents difference spectra for $^{13}C\varepsilon$ -methionine between rhodopsin and Meta I (orange) and between rhodopsin and Meta II (black). In the rhodopsin–Meta II

difference spectrum, we have previously assigned the positive resonance at 17.2 ppm (rhodopsin) and the negative resonance at 12.8 ppm (Meta II) to Met288^{7,35} based on its disappearance in the M288L mutant.²² Met288^{7,35} is located on H7 and is at the interface between EL2 and the extracellular end of H7. The rhodopsin–Meta I difference spectrum indicates that Met288 in Meta I has moved into an environment characteristic of the active state, suggesting local structural changes in H6, H7, or EL2.

The chemical shift for Met257^{6.40} in Meta II was assigned at 14.7 ppm based on a cross-peak with Arg135^{3.50}. A corresponding negative resonance is not observed in the difference spectrum because the resonance does not change between rhodopsin and Meta II. In contrast, a strong negative cross peak is observed at 15.8 ppm in Meta I (discussed below).

In the rhodopsin–Meta II difference spectrum, we have also assigned the positive resonance at 10.5 ppm (rhodopsin) and negative resonance at 19.2 ppm (Meta II) to Met44^{1.39}. Met44^{1.39} is located near the retinal Schiff base. The positive resonance at 10.5 ppm is observed in the rhodopsin–Meta I difference spectrum. However, the negative resonance at 19.2 ppm is not observed in Meta I. These results suggest that the environment of Met44^{1.39} near the protonated Schiff base in Meta I is different from rhodopsin but does not yet correspond to the active Meta II state.

Figure 5C presents the rhodopsin-Meta II difference spectrum (black) for ${}^{13}\text{C}\alpha$ -glycine. We have assigned the positive resonance at 41.5 ppm and the negative resonance at 42.9 ppm to Gly188^{EL2} based on its disappearance in the G188A mutant. ⁶⁴ The appearance of the positive peak at ~42 ppm in the rhodopsin-Meta I spectrum (orange) indicates that Gly188^{EL2} located in EL2 is influenced by the conversion to Meta I. However, the absence of the negative peak at 42.9 ppm argues that the EL2 has not yet adopted the final conformation observed in the Meta II state.²² The large positive peak at 45.5 ppm in Meta II is associated with several glycines on H3 (Gly114^{3.29}, Gly120^{3.35}, and Gly121^{3.36}).^{22,61} The observation of a positive peak at this position in the rhodopsin-Meta I difference spectrum suggests a possible structural change in H3 in the transition to Meta I. Gly114^{3.29} on H3 is adjacent to Glu113^{3.28} and one helical turn from Cys110^{3.25}. Changes in the interactions of the protonated Schiff base with its counterion are likely to induce structural changes in the helix backbone in this region. For example, the largest changes in the crystal structure of Lumirhodopsin, relative to rhodopsin, were observed in two helical turns of H3 centered on Gly120^{3.35} and Gly121^{3.36}.54

Figure 5D shows the changes in the $^{13}C\beta$ resonances of the highly conserved Cys110 $^{3.25}$ -Cys187 EL2 disulfide group. Cys110 $^{3.25}$ is at the extracellular end of H3, and Cys187 EL2 is part of the β 4 strand of EL2. The Cys187 EL2 $^{13}C\beta$ resonance shifts from 46.8 to 50.1 ppm upon conversion of wild-type rhodopsin to Meta II trapped in DDM. 22 In digitonin, the dark state $^{13}C\beta$ resonance of Cys187 EL2 is at 45.5 ppm and shifts to 44.4 ppm in Meta I. The change in chemical shift of Cys187 EL2 suggests a conformation or environment distinct from either the dark state or Meta II.

Two-dimensional (2D) MAS NMR dipolar recoupling experiments provide support for the conclusion that EL2 has not yet shifted into its active state conformation. Figure 6A presents slices through the $^{13}C\zeta$ -tyrosine diagonal resonance from 2D DARR NMR spectra of rhodopsin (black) and Meta I (orange). In rhodopsin, at least four (of six possible) $^{13}C\zeta$

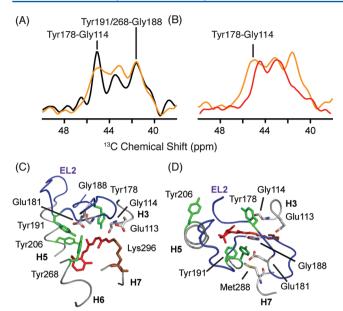


Figure 6. Two-dimensional DARR NMR spectra of ¹³Cζ-tyrosine and 13 C α -glycine-labeled rhodopsin. Slices through the diagonal resonances of $^{13}\text{C}\zeta$ -tyrosine showing through-space $^{13}\text{C}\cdots ^{13}\text{C}$ contacts with $^{13}\text{C}\alpha\text{-glycine}$ are shown for rhodopsin (black) and Meta I (orange) in panel (A) and for Meta I (orange) and Meta II (red) in panel (B). Cross-peaks between $^{13}C\zeta$ -tyrosine and $^{13}C\alpha$ -glycine provide a way to monitor changes on the extracellular side of the rhodopsin. In the rhodopsin crystal structure (1U19), there are six 13 C ζ -tyrosine- 13 C α glycine contacts located in the extracellular region of rhodopsin involving five tyrosines and five glycines: Tyr10^{NT}-Gly3^{NT}, 3.9 Å; Tyr10^{NT}-Gly280^{EL3}, 4.4 Å; Tyr29^{NT}-Gly101^{EL1}, 4.0 Å; Tyr178^{EL2}-Gly114^{3.29}, 4.5 Å; Tyr191^{EL2}-Gly188^{EL2}, 5.2 Å; Tyr268^{6.51}-Gly188 ^{EL2}, 5.3 Å. The slices shown in (A) reveal that the Tyr-Gly contacts are similar between rhodopsin and Meta I. (C, D) Structure of the extracellular side of rhodopsin (PDB access code = 1GZM) viewed from the side (C) and from the extracellular surface. The structure highlights the position of EL2 over the retinal binding site and the positions of several of the amino acids discussed.

Tyr- 13 C α Gly cross-peaks are detected. The cross-peaks at 42.0 and 45.5 ppm are assigned to contacts between Tyr268 $^{6.48}$ and Gly188 EL2 and between Tyr178 EL2 and Gly114 $^{3.29}$, respectively. In Meta I, these contacts are observed. The most substantial change between rhodopsin and Meta I is broadening of the Tyr178 EL2 -Gly114 $^{3.29}$ cross-peak. In Meta II (Figure 6B, red), the Gly188 EL2 -Tyr268 $^{6.51}$ and the Gly188 EL2 -Tyr191 EL2 cross-peaks are absent. The loss of these cross-peaks is attributed to a reorganization of the EL2 hydrogen-bonding network upon activation, respectively.

Structural Changes in Transmembrane Helices H5 and H6. The difference spectra of 13 C ζ -tyrosine and 13 C ε -methionine in Figures 5A and B also provide insights into the structural changes occurring in the transmembrane helices. Tyr206^{5,41} on H5 does not adopt its active state conformation in Meta I (Figure 5A). This tyrosine is part of a hydrogenbonding network with His211^{5,46} (H5), Glu122^{3,37} (H3), Trp126^{3,41} (H3), and Ala166^{4,55} (H4). The conversion to Meta II is associated with a rearrangement of this network such that the side chain of His211 hydrogen bonds directly with the side chain of Glu122^{3,37} ^{65,66} This rearrangement appears to be driven by direct contact of the retinal β-ionone ring with H5. 22,61 13 C dipolar couplings reveal close contacts between

Met207^{5.42} and the retinal C6 and C7 carbons upon conversion to Meta II.^{22,61}

The rhodopsin–Meta I difference spectrum of $^{13}\text{C}\varepsilon$ -methionine (Figure 5B) reveals a large negative resonance at 15.8 ppm. This frequency is close to that of Met257^{6.40} in Meta II.⁶³ The tentative assignment of the 15.8 ppm resonance in Meta I to $^{13}\text{C}\varepsilon$ -Met257^{6.40} raises the possibility that H6 has rotated in Meta I to place the Met257^{6.40} side chain in an environment similar to that in Meta II.

Figure 7A presents difference spectra between rhodopsin and Meta I (orange) and between rhodopsin and Meta II (black)

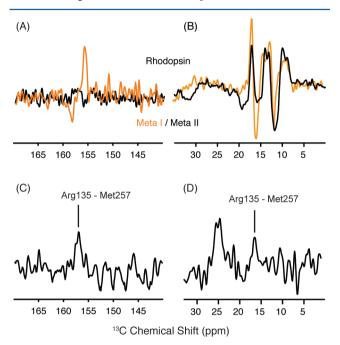


Figure 7. 13 C ζ -arginine and 13 C ε -methionine changes in the transition to Meta I. (A) Rhodopsin–Meta I (orange) and rhodopsin–Meta II (black) difference spectra are shown in the region of 13 C ζ -arginine. The 13 C ζ -arginine chemical shifts do not appear to differ between rhodopsin and Meta II, as indicated by the flat baseline in the region of the difference spectrum. In contrast, the rhodopsin–Meta I difference spectrum reveals that a single 13 C ζ -arginine has changed in the transition to Meta I. (B) Rhodopsin–Meta I (orange) and rhodopsin–Meta II (black) difference spectra are shown in the region of 13 C ε -methionine. A strong negative peak at 15.7 ppm corresponds to a new 13 C ε -methionine resonance in Meta I. (C,D) Slices extracted from a 2D DARR NMR spectrum reveal a cross peak between 13 C ζ -Arg135 $^{3.50}$ and 13 C ε -Met257 $^{6.40}$ in Meta I on both sides of the diagonal in the 2D spectrum.

for rhodopsin containing $^{13}C\zeta$ -labeled arginine. As above, the positive peaks correspond to rhodopsin, and the negative peaks correspond to Meta I or Meta II. The rhodopsin–Meta II difference spectrum in the region of the $^{13}C\zeta$ arginine chemical shift does not reveal any substantial changes indicating the chemical shift of Arg135 $^{3.50}$ is similar in rhodopsin and Meta II.

In contrast to the absence of changes in the rhodopsin–Meta II difference spectrum, the rhodopsin–Meta I difference spectrum reveals a change in chemical shift of an arginine $^{13}C\zeta$ resonance from 155.7 ppm in rhodopsin to 157.3 ppm in Meta I. The line width of this resonance is consistent with it corresponding to a single arginine residue. To confirm that this arginine is Arg135^{3.50}, we obtained 2D DARR NMR spectra of rhodopsin labeled with both $^{13}C\zeta$ -arginine and $^{13}C\varepsilon$ -methio-

nine. Figures 7C and 7D present slices from the 2D DARR NMR spectrum of Meta I in the regions of the $^{13}\mathrm{C}\zeta\text{-}\mathrm{arginine}$ and $^{13}\mathrm{C}\varepsilon\text{-}\mathrm{methionine}$ cross-peaks, respectively. Figure 7C reveals a $^{13}\mathrm{C}\zeta\text{-}\mathrm{Arg135}^{3.50}-^{13}\mathrm{C}\varepsilon\text{-}\mathrm{Met257}^{6.40}$ cross-peak at 156.9 ppm in the slice taken through the $^{13}\mathrm{C}\varepsilon\text{-}\mathrm{methionine}$ diagonal resonance, while Figure 7D reveals a $^{13}\mathrm{C}\varepsilon\text{-}\mathrm{methionine}-^{13}\mathrm{C}\zeta\text{-}\mathrm{arginine}$ cross-peak at 16.2 ppm in the slice taken through the $^{13}\mathrm{C}\zeta\text{-}\mathrm{arginine}$ diagonal resonance. The intensities of the $^{13}\mathrm{C}\varepsilon\text{-}\mathrm{methionine}-^{13}\mathrm{C}\zeta\text{-}\mathrm{arginine}$ cross-peaks in Meta I relative to the diagonal resonances are roughly the same intensity as in Meta II 63 where the internuclear distance is 4.5–5 Å. We attribute the appearance of a $^{13}\mathrm{C}\zeta\text{-}\mathrm{Arg135}^{3.50}-^{13}\mathrm{C}\varepsilon\text{-}\mathrm{Met257}^{6.40}$ cross-peak to rotation of H6 (see Discussion).

In both active opsin and Meta II, the rotation of H6 that places Met257^{6.40} in contact with Arg135^{3.50} is accompanied by the rotation of H5 and H7.⁶³ This rotation has implications for two salt bridges formed by Arg135^{3.50} in the dark state of rhodopsin. The first salt bridge between Arg135^{3.50} and Glu247^{6.30}, often referred to as the ionic lock, must be broken upon rotation of H6. The second salt bridge between Arg135^{3.50} and Glu134^{3.49} is disrupted with the rotation of Tyr223^{5.58} on H5 and Tyr306^{7.53} on H7. Tyr223^{5.58} and Tyr306^{7.53} form hydrogen bonds with the arginine guanidinium side chain to stabilize Arg135^{3.50} in a protonated state.⁶³

To test whether there are parallel changes in Meta I, 2D DARR NMR spectra were obtained for rhodopsin, Meta I, and Meta II using rhodopsin labeled with $^{13}C\varepsilon$ -methionine and $^{13}C\zeta$ -tyrosine. Figure 8 presents slices through the $^{13}C\varepsilon$ -

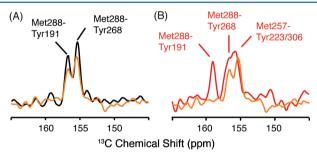


Figure 8. Two-dimensional 2D DARR NMR spectra of 13 C ξ -tyrosine and 13 C ε -methionine-labeled rhodopsin. (A) Slices through the 13 C ξ -tyrosine diagonal resonance from 2D DARR NMR spectra of rhodopsin (black) and Meta I (orange) labeled with 13 C ξ -tyrosine and 13 C ε -methionine highlight the region of Tyr–Met cross-peaks. In rhodopsin, we observe cross-peaks between Met288 $^{7.35}$ –Tyr191 $^{\text{EL}2}$ and Met288 $^{7.35}$ –Tyr268 $^{6.51}$. Upon conversion to Meta I, the intensity of the cross-peaks does not change appreciably. (B) Slices through the 13 C ξ -tyrosine diagonal resonance from 2D DARR NMR spectra of rhodopsin (black) and Meta II (red) are shown using the same labeling strategy as in (A).

methionine diagonal resonance corresponding to Met288^{7,35} in the ¹³Cζ-tyrosine region of rhodopsin (Figure 8A; black) at 17.2 ppm and of Meta I (Figure 8A,B; orange) at 12.8 ppm. In Figure 8A, we observe two cross-peaks that we assign to the two tyrosines closest to Met288^{7,35} (i.e., Tyr268^{6,48} at 3.9 Å and Tyr191^{EL2} at 5.2 Å).⁶³ Upon conversion to Meta I, these crosspeaks do not change position or intensity. However, in Meta II the Met288^{7,35}—Tyr191^{EL2} and Met288^{7,35}—Tyr268^{6,48} crosspeaks shift downfield, and two overlapping cross-peaks assigned to Met257^{6,40}—Tyr223^{5,58} and Met257^{6,40}—Tyr306^{7,53} appear at 155.7 ppm.⁶³

We interpret the results in Figures 7 and 8 in terms of rotation of H6 in Meta I without the corresponding rotation of Tyr223^{5.58} on H5 and Tyr306^{7.53} on H7. This interpretation is consistent with several observations. First, while we observe an Arg135^{3.50}-Met257^{6.40} cross-peak in both Meta I and Meta II, we only observe the Met257^{6.40}-Tyr223^{5.58} and Met257^{6.40}-Tyr306^{7.53} cross-peaks in Meta II. Second, we observe a change in a ¹³Cζ-arginine resonance in Meta I, but not in Meta II. We interpret this as due to breaking of the Arg135^{3.50}-Glu247^{6.30} interaction upon formation of Meta I and the re-establishing of strong hydrogen bonding interactions of Arg135^{3.50} with Tyr223^{5.58} and Tyr306^{7.53} in Meta II. Third, the rhodopsin-Meta I difference spectrum for ${}^{13}\text{C}\varepsilon\text{-methionine}$ of wild-type rhodopsin in Figure 7A is remarkably similar to the rhodopsin— Meta II difference spectrum for $^{13}\text{C}\varepsilon\text{-methionine}$ of the Y223F mutant. 63 In the Y223F mutant, there is a rapid decay of the Meta II state to opsin, which suggests that in wild-type Meta II the Tyr223^{5.58}-Arg135^{3.50} interaction is holding H5 in an active orientation, whereas in Meta II of the Y223F mutant, H5 has rotated back to an inactive orientation. FTIR measurements of the Y223F mutant show a shift in the pH-dependent equilibrium between Meta I and Meta II back toward Meta I.⁶³

Although we do not observe rotation of Tyr223^{5.58} on H5 in Meta I, ¹³C-difference and DARR NMR spectra of Meta I and Meta II labeled with ¹³C β -cysteine show there is a change in the H4–H5 interface. In the 1D ¹³C difference spectra of ¹³C β -cysteine-labeled rhodopsin, the C β carbons exhibit considerable changes in both Meta I andMeta II (Figure 9A). We can assign

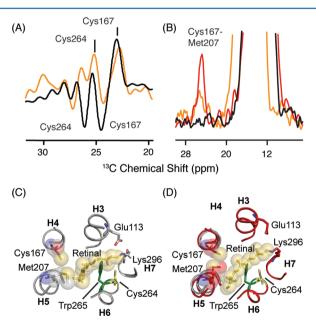


Figure 9. Chemical shift changes in 13 Cβ-cysteine in the transition from rhodopsin to Meta I. (A) Rhodopsin–Meta I (orange) and rhodopsin–Meta II (black) difference spectra in the region of reduced 13 Cβ-cysteine resonances. (B) Slices taken through 2D 13 C DARR NMR spectra of rhodopsin (black), Meta I (orange), and Meta II (red) exhibit a cross-peak between Met207 $^{5.42}$ and Cys167 $^{4.56}$. The intensity of the Cys167 $^{4.56}$ –Met207 $^{5.42}$ cross-peak in Meta I is intermediate between rhodopsin and Meta II, suggesting that there is partial rotation of H5. (C) Structure of rhodopsin (PDB access code = 1GZM) in the region of the retinal binding site. The structure highlights the positions of Cys167 $^{4.56}$ and Met207 $^{5.42}$. (D) Structure of Meta II (PDB access code = 3PXO) showing displacement of H5 and closer interaction of Cys167 $^{4.56}$ and Met207 $^{5.42}$.

the positive peaks at 23.2 and 25.8 ppm to Cys167^{4.56} and Cys264^{6.47}, respectively. These assignments were made based on the interhelical cross-peaks to His211^{5.46} and Tyr301^{7.48}, respectively.²² In Meta II, these resonances shift to 24.4 and 26.2 ppm, respectively.

Met207^{5,42} can serve as a probe for the motion of H5 relative to H4. In the formation of Meta II, a strong cross-peak is observed between 13 C ε -Met207^{5,42} on H5 and 13 C β -Cys167^{4,56} on H4 (Figure 9B). The side chain of Met207^{5,42} rotates from its original position toward H4 to give space for the β -ionone ring of the retinal to pack against H5 in Meta II. In contrast, only a weak cross-peak is observed between Met207^{5,42} and Cys167^{4,56} in Meta I (Figure 9B). These data show that although the ligand-binding pocket in the region of H5 has responded to retinal isomerization the motion is not as large as that observed in Meta II.

DISCUSSION

In this study, solid-state ¹³C and ¹⁵N NMR measurements of the Meta I intermediate are used to address the structural changes that precede receptor activation in rhodopsin. We find that the largest conformational change in Meta I is rotation of transmembrane helix H6. The chemical shift changes associated with EL2 and the rotations of Tyr223^{5.58} on H5 and Tyr306^{7.53} on H7, which are observed in Meta II, have not occurred in Meta I. We discuss these observations in connection with the sequence of events that occur in the transition from rhodopsin to Meta I and from Meta I to the active Meta II conformation.

EL2 Has Not Adopted an Active Conformation in Meta I. In rhodopsin, EL2 folds into two β -strands (β 3 and β 4) that form a plug over the retinal binding site and prevent access of hydroxylamine to the protonated Schiff base linkage. The β -strands are constrained in the dark state by a conserved disulfide bond between Cys110^{3.25} (H3) and Cys187^{EL2} (β 4) and a network of hydrogen bonding interactions. NMR chemical shift measurements of Meta II indicate a rearrangement of hydrogen bonding interactions between EL2 and amino acids on transmembrane helices H5 and H6.²² These changes in EL2, relative to rhodopsin, have suggested that EL2 adopts an active conformation that is coupled to the motion of transmembrane helices H5, H6, and H7.²²

Several observations in the current study suggest that EL2 has not adopted an active conformation in Meta I. First, the characteristic Meta II chemical shifts of Cys187^{EL2}, Gly188^{EL2}, and Tyr191^{EL2} on the β 4 strand of EL2 are not observed in Meta I. Second, a contact between Gly188^{EL2} on EL2 and Tyr268^{6.48} at the extracellular end of H6, which is present in rhodopsin, is still observed in Meta I but absent in Meta II. Third, the Met288^{7.35}–Tyr268^{6.48} and Met288^{7.35}–Tyr191^{EL2} cross-peaks have similar intensities in rhodopsin and Meta I, indicating that EL2 has not markedly changed its position relative to the extracellular ends of H6 and H7. The only indication of substantial change in the vicinity of EL2 is the chemical shift change of Met288, which may reflect changes in the position of H6 and H7, as discussed below, rather than changes in the position of EL2.

The NMR measurements indicating that EL2 has not changed conformation are consistent with recent studies on the accessibility of the retinal Schiff base to hydroxylamine.⁶⁷ In Meta II, hydroxylamine is able to hydrolyze the Schiff base linkage between the retinal chromophore and the side chain of Lys296^{7,43}. However, in both rhodopsin and Meta I, the protonated Schiff base is not accessible to hydroxylamine.⁶⁷

(This conclusion disagrees with an earlier study⁶⁸ where hydroxylamine was shown to react at 275 K with a metarhodopsin intermediate, presumably Meta I, in a pH-independent fashion. Later experiments on Meta I trapped at 240 K were not able to confirm this result,⁶⁷ which opened up the possibility that in the earlier experiments the reactive metarhodopsin intermediate was Meta III, which is known to react with hydroxylamine.⁵⁹) As a result, the increased accessibility of the Schiff base to hydroxylamine in Meta II is likely due to a change in the packing interactions of EL2 with the retinal.²² In this regard, Meta I with EL2 tightly packed against the retinal may serve as a model for Class A GPCRs where the main function of EL2 seems to be as an agonist diffusion barrier as in the angiotensin II type 1 receptor.⁶⁹

Although EL2 does not appear to have reached the structure adopted in Meta II, there are marked changes in the NMR resonances associated with the retinal chromophore. In rhodopsin and Meta II, narrow, distinct NMR resonances are observed, implying that there are well-defined conformations associated with the active and inactive states of rhodopsin. In contrast, we observe substantial broadening of the ¹⁵N and ¹³C resonances of the Schiff base and retinal in Meta I. The broadening of the ¹³C resonances was previously interpreted in terms of heterogeneity in the conformation of the polyene chain of the retinal chromophore.²⁸ Alternatively, broadening of the ¹⁵Nζ-Lys296^{7.43} and ¹³C-retinal resonances may reflect the loss of a single well-defined hydrogen bonding contact with Glu113^{3,28} and water in the binding pocket. As such, the broadening may be associated with the formation of a complex counterion involving Glu113^{3,28}, Glu181^{EL2}, and water.

The concept of a complex counterion has previously been

The concept of a complex counterion has previously been discussed. Yan et al. 32 proposed that there is a switch in the Schiff base counterion from Glu113 $^{3.28}$ to Glu181 $^{\rm EL2}$ in the formation of Meta I. In the dark state of rhodopsin, Glu113 $^{3.28}$ is part of a hydrogen-bonding network stretching to Glu181 $^{\rm EL2}$ via the backbone carbonyl of Cys187 $^{\rm EL2}$ and a structural water molecule. 70 Yan et al. 32 have observed a dramatic shift in the p $K_{\rm a}$ of the protonated Schiff base upon mutation of Glu181 $^{\rm EL2}$ in EL2 and have suggested that Glu181 $^{\rm EL2}$ is the predominant counterion in the Meta I state. Lüdeke et al. 71 have found that Glu181 $^{\rm EL2}$ is unprotonated in both rhodopsin and Meta I and have argued that together Glu113 $^{3.28}$ and Glu181 $^{\rm EL2}$ form a complex counterion.

Rotation of H6 Occurs in the Formation of Meta I. A hallmark of rhodopsin activation is the outward displacement of helix H6 from the helical bundle. Cross-linking of the intracellular ends of H3 and H6 blocks this motion and prevents receptor activation. Hubbell and co-workers have demonstrated that the outward motion of H6 occurs following the internal proton transfer from the retinal protonated Schiff base to $Glu113^{3.28}$.

In the cryo-EM structure of Meta I, Schertler and co-workers found that there was increased electron density on the side of H6 facing the β -ionone ring at the level of Trp265^{6,48}. This density is consistent with the displacement of Trp265^{6,48} as seen in the structure of active opsin. However, there was no change observed in the position of the intracellular end of H6, as in opsin, indicating that H6 does not tilt outward in Meta I. One way to reconcile the cryo-EM structure of Meta I showing a displacement of Trp265^{6,48} with the NMR results showing a contact between Met257^{6,40} and Arg135^{3,50} is to propose that H6 rotates in Meta I without an outward displacement.

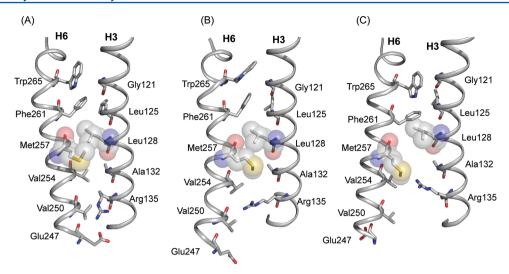


Figure 10. Structural changes in H3 and H6 upon rhodopsin activation. Packing interactions are shown between H3 and H6 in the crystal structure of rhodopsin (PDB access code = 1GZM) (A), of Meta I (B), and of Meta II (PDB access code = 3PXO) (C). The Meta I structure is based on MD simulations guided by NMR constraints. NMR measurements between Arg135^{3.50} and Met257^{6.40}, in combination with azido labeling studies²³ and EPR³ measurements of Meta I and Meta II, are consistent with rotation of H6. Trp265^{6.48} in the conserved CWxP motif on H6 is locked in place in dark rhodopsin by the 11-*cis* retinal chromophore. Retinal isomerization releases the packing constraints on the Trp265^{6.48} indole ring. Gly121^{3.36} on H3 is strictly conserved in the visual receptors. The small side chain facilitates packing of the Trp265^{6.48} side chain in dark rhodopsin but does not hinder H6 rotation. Leu128^{3.43} and Met257^{6.40} are closely packed in dark rhodopsin. Leu128^{3.43} is highly conserved (78%) in GPCRs and is part of a tightly packed transmembrane core. The rotation of H6 is facilitated by the small side chain at position 132^{3.48}. This position is highly conserved in the GPCRs as either an alanine (36%) or a serine (50%). The Arg135^{3.50} side chain interacts with Glu247^{6.30} in the dark and is prevented from moving toward the Met257^{6.40} by Val254^{6.37}. Val254^{6.37} is conserved (63%) as a β-branched amino acid across the Class A GPCRs. Rotation of H6 breaks the Arg135^{3.50}-Glu247^{6.30} interaction and removes the steric interaction with Val254^{6.37}. The side chain of Glu247^{6.30} is in an intermediate position between Arg135^{3.50} and Lys231^{5.66}; a Glu247^{6.30}-Lys231^{5.66} salt bridge forms in Meta II.

The displacement of the Trp265^{6.48} side chain may also be reflected in recent deuterium NMR studies of Brown and coworkers.⁷⁴ In rhodopsin, the Trp265^{6.48} side chain is packed against the C18 and C20 methyl groups. These methyl groups exhibit much more restricted rotational motion than the C19 methyl group in the 6-s-trans, 12-s-trans, 11-cis retinal chromophore in rhodopsin because of steric clashes within the retinal molecule (e.g., the C18 methyl group clashes with the C8H proton in a twisted 6-s-cis geometry, and the C20 methyl group clashes with the C10H proton in a twisted 12-strans geometry). However, differences in the activation energy (E_a) for rotational motion of these methyl groups between rhodopsin and Meta I and between Meta I and Meta II may have contributions from steric clashes with the protein.⁷⁴ The E_a of the C19 methyl group increases in the transition from rhodopsin to Meta I and from Meta I to Meta II and is in fact greater than the E_a for the C20 methyl group in Meta II. These observations are consistent with ${}^{13}C\cdots{}^{13}C$ distance measurements showing that the Trp265^{6.48} side chain moves from a position in contact with the C20 methyl group to a position in contact with the C19 methyl group upon receptor activation.^{58,75}

The proposed rotation of H6 in Meta I is in agreement with recent azido labeling studies on rhodopsin where *p*-azido-L-phenylalanine is incorporated into positions at the intracellular end of H6.²³ These studies show that there are large changes in the vibrations of the azido label at position 250^{6.33} between Lumirhodopsin and Meta I. However, no further changes in the azido vibrations are observed in the formation of Meta II, suggesting that structural changes in this region of the receptor are complete in Meta I. These results on azido-labeled rhodopsin were interpreted in terms of a rotation of H6 to break the Arg135^{3.50}-Glu247^{6.30} salt bridge.

Figure 10 shows the residues in the interface between H3 and H6. The distance between the labeled $^{13}C\varepsilon\text{-Met}257^{6.40}-^{13}C\zeta\text{-Arg}135^{3.50}$ carbons is $\sim\!11$ Å in dark rhodopsin (dashed line in Figure 10A). The conversion to Meta I brings the labeled ¹³C sites to within ~6 Å. Rotation of H6 alone does not reduce the distance between these sites, arguing that the side chains of Arg135^{3.50} and Met257^{6.40} reorient in Meta I. MD simulations guided by NMR restraints suggest that the side chain of Met2576.40 ratchets past Leu128^{3,43} (Figure 10B). The rotation of H6 changes the packing interactions in the H3 and H6 interface in several ways that facilitate side chain motion. The proposed rotation of H6 moves Val254^{6.37} away from Arg135^{3.50}, which otherwise blocks the motion of the arginine side chain toward Met257^{6.40}, and also moves Met257^{6.40} toward Ala132^{3.47}. The small side chain of Ala132^{3,47} provides space for bending of the Met257^{6,40} side chain toward Arg135^{3,50}. As a result, the H3–H6 ionic lock between Arg135^{3,50} and Glu247^{6,30} is broken, while the Lys231^{5.66} and Glu247^{6.30} interaction seen in the active form of opsin is not formed in Meta I despite a movement of these residues toward each other. The loss of the Glu2476.30 salt bridge in Meta I can explain the observed increased hydration specific for this intermediate.⁷⁶

H5 Has Not Adopted an Active Conformation in Meta **I.** One of the triggers for activation of rhodopsin is the steric interaction between the β -ionone ring of the retinal and transmembrane helix H5 in the region of Met207^{5.42} and Phe212^{5.47}. Contact of the β -ionone ring with H5 leads to rearrangement of hydrogen bonds between H3 and H5 on both the extracellular and intracellular sides of the receptor. On the extracellular end of H5, Tyr206^{5.41} and His211^{5.46} form hydrogen bonds with Trp126^{3.41} and Glu122^{3.37} on H3 in the dark state of rhodopsin. These hydrogen bonds are disrupted

upon activation. On the intracellular end of H5, Tyr223^{5.58} is oriented toward the surrounding lipid membrane in the dark state of rhodopsin but rotates upon activation into the helical bundle to hydrogen bond with Arg135^{3.50} of the conserved ERY sequence. The question is whether these changes occur in concert with rotation of H6 in Meta I or with the outward displacement of H6 in Meta II.

Ŷe et al.²³ observed only small changes in the azido vibrations between rhodopsin and Meta I when the azido label was incorporated at position 227^{5.62} on H5, whereas much larger changes were observed upon formation of Meta II. These results differ from those using the azido probes on H6 (discussed above) and were interpreted in terms of a small movement of H5 in response to rotation of H6. In agreement with these studies, we observe a small increase in intensity of the Cys167^{4.56}—Met207^{5.42} cross-peak in the formation of Meta I but a much larger increase in the formation of Meta II (Figure 9B). We interpret the change of the Cys167^{4.56}—Met207^{5.42} contact in terms of displacement of H5 relative to H4 and rearrangement of the position of the Met207^{5.42} side chain (Figures 9C and D) but conclude that the H5 helix has not completely shifted into its active state orientation.

Support for the conclusion that H5 has not adopted an active conformation in Meta I comes from both NMR and FTIR spectroscopy. The NMR chemical shift of Tyr206^{5.41} provides a probe of the hydrogen-bonding network centered on His211^{5.46}, where a distinctive upfield chemical shift of Tyr206^{5.41} is observed in the rhodopsin–Meta II difference spectrum. We do not observe this chemical shift change in the rhodopsin-Meta I difference spectrum (Figure 5A). In addition, we do not observe the contact between Tyr223^{5.58} and Met257^{6.40} in Meta I that is associated with the rotation of Tyr223^{5,58} toward Arg135^{3,50} observed in Meta II (see Figure 1C). In the deuterium NMR studies cited above, 74 the E_a for rotation of the C18 methyl group increases in Meta I and then decreases in Meta II. These changes were interpreted in terms of a steric clash of the β -ionone ring with H5 in Meta I and then displacement of H5 (to lessen the steric clash with the β ionone ring) in Meta II.

In FTIR studies on Meta I and Meta II, the vibrational frequency at 1734 cm⁻¹ associated with the Glu122^{3,37} carboxyl group has provided an excellent probe of the hydrogen bonding interactions between H3 and H5. Siebert and co-workers found that the 1734 cm⁻¹ vibration shifts to 1701 cm⁻¹ in the transition from rhodopsin to Meta I indicating that Glu122^{3,37} becomes more strongly hydrogen bonded.^{21,77} However, in the transition to Meta II, the Glu122^{3,37} vibration shifts to 1745 cm⁻¹, a signature of weaker hydrogen bonding. These results agree with both the NMR and azido-labeling studies described above, showing that the position or orientation of H5 changes between rhodopsin and Meta I and then undergoes further changes between Meta I and Meta II.

Sequence of Events in the Formation of the Active Meta II State. One of the challenges for understanding how light activates the visual receptor rhodopsin has been to delineate the chain of molecular events leading to the fully active Meta II conformation. The pioneering work of Hubbell and co-workers revealed that the defining motion is the outward tilting of H6.³ However, the crystal structures of dark rhodopsin and its early photointermediates have provided few clues as to the sequence of the events that drive this motion. The crystal structures of Bathorhodopsin, Lumirhodopsin, and Meta I²⁰ showed only subtle changes from the structure of

the dark state of rhodopsin. The picture that has emerged over the past decade is that the structural changes in the receptor leading up to Meta II only involve slight rearrangements of the side chains in the binding site to accommodate the all-*trans* chromophore. In contrast, the more recent structures of active $\operatorname{opsin}^{5,6}$ and Meta II 66,79 capture substantial structural changes on the intracellular side of the receptor.

The two hallmarks of Meta II formation are the proton transfer from the protonated Schiff base to Glu113^{3,28} and the outward motion of H6. Both steps are set up in Meta I. The observed Schiff base ¹⁵N chemical shift in Meta I, along with vibrational spectroscopy studies, ^{19,32,71} favors a slightly stronger protonated Schiff base—counterion interaction. The interaction may involve both Glu113^{3,28} and Glu181^{EL2} as part of a complex counterion. As a consequence of a rearrangement in the electrostatic interactions near the protonated Schiff base, Glu113^{3,28} may move into a more hydrophobic environment in Meta I and become the driving force for Schiff base deprotonation. Deprotonation of the Schiff base is the first of two protonation switches that must be triggered for rhodopsin activation. ⁸⁰

The NMR evidence for rotation of H6 in the Meta I intermediate supports the conclusions drawn from previous azido labeling measurements. Both studies suggest that H6 rotation disrupts the intracellular salt bridge between Arg135^{3.50} and Glu247^{6.30}. In contrast, the electrostatic interaction between Arg135^{3.50} and Glu134^{3.49}, both highly conserved residues, remains intact in Meta I and is arguably more important for stabilizing the inactive state of the receptor. Protonation of Glu134^{3.49} in Meta II has been described as the second switch required for rhodopsin activation. The NMR studies presented above indicate that the inward rotation of Tyr223^{5.58} on H5 and Tyr306^{7.53} on H7 has not occurred in Meta I and consequently must be associated with the outward motion of H6 upon activation.

Finally, the observed structural transitions in the formation of Meta I provide insights into the conservation of residues in the ligand-activated GPCRs and common elements of receptor activation. The ionic lock between Arg135^{3.50} and Glu247^{6.30} is not highly conserved. Even in those receptor subfamilies where these complementary charged residues are conserved, crystal structures often do not reveal a direct interaction. 82-85 For example, Schertler and co-workers have recently reported crystal structures of two inactive forms of the β_1 -adreneric receptor, one with an ionic lock between Arg^{3.50}-Glu^{6.30} and the other without.⁸⁶ The two forms were found with several inverse agonists bound. The lack of a stabilizing electrostatic interaction between H3 and H6 in the diffusible ligandactivated receptors may be associated with their generally high levels of basal activity relative to the dark state of rhodopsin. In contrast, Arg135^{3.50} and Glu134^{3.49} are part of the highly conserved D/ERY sequence on H3, and their interaction appears to have a much more dramatic effect in modulating receptor activity. The observation of a direct interaction in Meta I between Arg135^{3.50} and Met257^{6.40}, a nonconserved but highly important residue in the visual receptors, suggests that in the ligand-activated receptors there are subfamily-specific interactions that regulate receptor activation.

CONCLUSIONS

The work presented here describes solid-state NMR studies on the Meta I intermediate in the photoreaction of the visual receptor rhodopsin. Meta I is the intermediate immediately preceding Meta II, the active state of the receptor. By comparing our current results with previous NMR measurements on Meta II, we are able to define the structural changes that result in receptor activation. The major conclusions of our studies are that transmembrane helix H6 has rotated in the formation of Meta I to break the intracellular ionic lock between Arg135^{3.50} and Glu247^{6.30} but that structural changes involving helices H5 and H7 have not yet occurred. The motion of H6, along with several additional smaller changes in structure, appears to prime the receptor for the transition to an active conformation.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) Smith, S. O. Annu. Rev. Biophys. 2010, 39, 309-328.
- (2) Palczewski, K. Annu. Rev. Biochem. 2006, 75, 743-767.
- (3) Farrens, D. L.; Altenbach, C.; Yang, K.; Hubbell, W. L.; Khorana, H. G. Science 1996, 274, 768–770.
- (4) Palczewski, K.; Kumasaka, T.; Hori, T.; Behnke, C. A.; Motoshima, H.; Fox, B. A.; Le Trong, I.; Teller, D. C.; Okada, T.; Stenkamp, R. E.; Yamamoto, M.; Miyano, M. Science 2000, 289, 739–745.
- (5) Scheerer, P.; Park, J. H.; Hildebrand, P. W.; Kim, Y. J.; Krauss, N.; Choe, H. W.; Hofmann, K. P.; Ernst, O. P. *Nature* **2008**, *455*, 497–502.
- (6) Park, J. H.; Scheerer, P.; Hofmann, K. P.; Choe, H. W.; Ernst, O. P. Nature **2008**, 454, 183–187.
- (7) Altenbach, C.; Kusnetzow, A. K.; Ernst, O. P.; Hofmann, K. P.; Hubbell, W. L. *Proc. Natl. Acad. Sci. U.S.A.* **2008**, *105*, 7439–7444.
- (8) Rasmussen, S. G. F.; Choi, H. J.; Fung, J. J.; Pardon, E.; Casarosa, P.; Chae, P. S.; DeVree, B. T.; Rosenbaum, D. M.; Thian, F. S.; Kobilka, T. S.; Schnapp, A.; Konetzki, I.; Sunahara, R. K.; Gellman, S. H.; Pautsch, A.; Steyaert, J.; Weis, W. I.; Kobilka, B. K. *Nature* **2011**, 469, 175–180.
- (9) Rasmussen, S. G. F.; DeVree, B. T.; Zou, Y.; Kruse, A. C.; Chung, K. Y.; Kobilka, T. S.; Thian, F. S.; Chae, P. S.; Pardon, E.; Calinski, D.; Mathiesen, J. M.; Shah, S. T. A.; Lyons, J. A.; Caffrey, M.; Gellman, S. H.; Steyaert, J.; Skiniotis, G.; Weis, W. I.; Sunahara, R. K.; Kobilka, B. K. Nature 2011, 477, 549–555.
- (10) Rosenbaum, D. M.; Zhang, C.; Lyons, J. A.; Holl, R.; Aragao, D.; Arlow, D. H.; Rasmussen, S. G. F.; Choi, H. J.; DeVree, B. T.; Sunahara, R. K.; Chae, P. S.; Gellman, S. H.; Dror, R. O.; Shaw, D. E.; Weis, W. I.; Caffrey, M.; Gmeiner, P.; Kobilka, B. K. *Nature* **2011**, *469*, 236–240.
- (11) Xu, F.; Wu, H. X.; Katritch, V.; Han, G. W.; Jacobson, K. A.; Gao, Z. G.; Cherezov, V.; Stevens, R. C. Science 2011, 332, 322–327.

- (12) Warne, T.; Moukhametzianov, R.; Baker, J. G.; Nehmé, R.; Edwards, P. C.; Leslie, A. G. W.; Schertler, G. F. X.; Tate, C. G. *Nature* **2011**, 469, 241–244.
- (13) Deupi, X.; Kobilka, B. K. Physiology 2010, 25, 293-303.
- (14) Eyring, G.; Curry, B.; Broek, A.; Lugtenburg, J.; Mathies, R. Biochemistry 1982, 21, 384-393.
- (15) Cooper, A. Nature 1979, 282, 531-533.
- (16) Doukas, A. G.; Aton, B.; Callender, R. H.; Ebrey, T. G. Biochemistry 1978, 17, 2430-2435.
- (17) Verdegem, P. J. E.; Bovee-Geurts, P. H. M.; De Grip, W. J.; Lugtenburg, J.; de Groot, H. J. M. *Biochemistry* **1999**, 38, 11316—11324.
- (18) Feng, X.; Verdegem, P. J. E.; Eden, M.; Sandström, D.; Lee, Y. K.; Bovee-Geurts, P. H. M.; De Grip, W. J.; Lugtenburg, J.; de Groot, H. J. M.; Levitt, M. H. *J. Biomol. NMR* **2000**, *16*, 1–8.
- (19) Pan, D. H.; Mathies, R. A. Biochemistry 2001, 40, 7929-7936.
- (20) Ruprecht, J. J.; Mielke, T.; Vogel, R.; Villa, C.; Schertler, G. F. X. *EMBO J.* **2004**, *23*, 3609–3620.
- (21) Beck, M.; Sakmar, T. P.; Siebert, F. Biochemistry 1998, 37, 7630-7639.
- (22) Ahuja, S.; Hornak, V.; Yan, E. C. Y.; Syrett, N.; Goncalves, J. A.; Hirshfeld, A.; Ziliox, M.; Sakmar, T. P.; Sheves, M.; Reeves, P. J.; Smith, S. O.; Eilers, M. *Nat. Struct. Mol. Biol.* **2009**, *16*, 168–175.
- (23) Ye, S. X.; Zaitseva, E.; Caltabiano, G.; Schertler, G. F. X.; Sakmar, T. P.; Deupi, X.; Vogel, R. *Nature* **2010**, 464, 1386–1389.
- (24) Paulsen, R.; Bentrop, J. Nature 1983, 302, 417-419.
- (25) Pulvermüller, A.; Palczewski, K.; Hofmann, K. P. *Biochemistry* **1993**, 32, 14082–14088.
- (26) Tachibanaki, S.; Imai, H.; Mizukami, T.; Okada, T.; Imamoto, Y.; Matsuda, T.; Fukada, Y.; Terakita, A.; Shichida, Y. *Biochemistry* **1997**, *36*, 14173–14180.
- (27) Tachibanaki, S.; Imai, H.; Terakita, A.; Shichida, Y. FEBS Lett. **1998**, 425, 126–130.
- (28) Spooner, P. J. R.; Sharples, J. M.; Goodall, S. C.; Seedorf, H.; Verhoeven, M. A.; Lugtenburg, J.; Bovee-Geurts, P. H. M.; DeGrip, W. J.; Watts, A. *Biochemistry* **2003**, *42*, 13371–13378.
- (29) Salgado, G. F. J.; Struts, A. V.; Tanaka, K.; Krane, S.; Nakanishi, K.; Brown, M. F. *J. Am. Chem. Soc.* **2006**, *128*, 11067–11071.
- (30) Franke, R. R.; Sakmar, T. P.; Graham, R. M.; Khorana, H. G. J. Biol. Chem. 1992, 267, 14767–14774.
- (31) Resek, J. F.; Farahbakhsh, Z. T.; Hubbell, W. L.; Khorana, H. G. *Biochemistry* **1993**, 32, 12025–12032.
- (32) Yan, E. C. Y.; Kazmi, M. A.; Ganim, Z.; Hou, J. M.; Pan, D. H.; Chang, B. S. W.; Sakmar, T. P.; Mathies, R. A. *Proc. Natl. Acad. Sci. U.S.A.* **2003**, *100*, 9262–9267.
- (33) Matthews, R. G.; Hubbard, R.; Brown, P. K.; Wald, G. J. Gen. Physiol. 1963, 47, 215-240.
- (34) Kiefer, H.; Krieger, J.; Olszewski, J. D.; Von Heijne, G.; Prestwich, G. D.; Breer, H. *Biochemistry* **1996**, 35, 16077–16084.
- (35) Reeves, P. J.; Kim, J. M.; Khorana, H. G. Proc. Natl. Acad. Sci. U.S.A. 2002, 99, 13413–13418.
- (36) Eilers, M.; Reeves, P. J.; Ying, W. W.; Khorana, H. G.; Smith, S. O. *Proc. Natl. Acad. Sci. U.S.A.* **1999**, *96*, 487–492.
- (37) Bennett, A. E.; Rienstra, C. M.; Auger, M.; Lakshmi, K. V.; Griffin, R. G. *I. Chem. Phys.* **1995**, *103*, 6951–6958.
- (38) Fung, B. M.; Khitrin, A. K.; Ermolaev, K. J. Magn. Reson. 2000, 142, 97-101.
- (39) Wang, J. X.; Balazs, Y. S.; Thompson, L. K. Biochemistry 1997, 36, 1699-1703.
- (40) Markley, J. L.; Bax, A.; Arata, Y.; Hilbers, C. W.; Kaptein, R.; Sykes, B. D.; Wright, P. E.; Wüthrich, K. J. Biomol. NMR 1998, 12, 1–23
- (41) Harbison, G. S.; Herzfeld, J.; Griffin, R. G. Biochemistry 1983, 22, 1-4.
- (42) Getmanova, E.; Patel, A. B.; Klein-Seetharaman, J.; Loewen, M. C.; Reeves, P. J.; Friedman, N.; Sheves, M.; Smith, S. O.; Khorana, H. G. *Biochemistry* **2004**, 43, 1126–1133.

- (43) Phillips, J. C.; Braun, R.; Wang, W.; Gumbart, J.; Tajkhorshid, E.; Villa, E.; Chipot, C.; Skeel, R. D.; Kale, L.; Schulten, K. J. Comput. Chem. 2005, 26, 1781–1802.
- (44) MacKerell, A. D.; Feig, M.; Brooks, C. L. J. Comput. Chem. 2004, 25, 1400–1415.
- (45) MacKerell, A. D.; Bashford, D.; Bellott, M.; Dunbrack, R. L.; Evanseck, J. D.; Field, M. J.; Fischer, S.; Gao, J.; Guo, H.; Ha, S.; Joseph-McCarthy, D.; Kuchnir, L.; Kuczera, K.; Lau, F. T. K.; Mattos, C.; Michnick, S.; Ngo, T.; Nguyen, D. T.; Prodhom, B.; Reiher, W. E.; Roux, B.; Schlenkrich, M.; Smith, J. C.; Stote, R.; Straub, J.; Watanabe, M.; Wiorkiewicz-Kuczera, J.; Yin, D.; Karplus, M. J. Phys. Chem. B 1998, 102, 3586–3616.
- (46) Yin, D. X.; Mackerell, A. D. J. Comput. Chem. 1998, 19, 334-348.
- (47) Feller, S. E.; Gawrisch, K.; MacKerell, A. D. J. Am. Chem. Soc. **2002**, 124, 318–326.
- (48) Feller, S. E.; MacKerell, A. D. J. Phys. Chem. B 2000, 104, 7510-7515.
- (49) Feller, S. E.; Yin, D. X.; Pastor, R. W.; MacKerell, A. D. *Biophys. J.* **1997**, *73*, 2269–2279.
- (50) MacKerell, A. D. J. Chim. Phys. Phys.-Chim. Biol. 1997, 94, 1436–1447.
- (51) Beglov, D.; Roux, B. J. Chem. Phys. 1994, 100, 9050-9063.
- (52) Klauda, J. B.; Brooks, B. R.; MacKerell, A. D.; Venable, R. M.; Pastor, R. W. J. Phys. Chem. B 2005, 109, 5300-5311.
- (53) Saam, J.; Tajkhorshid, E.; Hayashi, S.; Schulten, K. *Biophys. J.* **2002**, 83, 3097–3112.
- (54) Nakamichi, H.; Okada, T. Proc. Natl. Acad. Sci. U.S.A. 2006, 103, 12729–12734.
- (55) Jo, S.; Lim, J. B.; Klauda, J. B.; Im, W. Biophys. J. **2009**, 97, 50–58.
- (56) Humphrey, W.; Dalke, A.; Schulten, K. J. Mol. Graphics 1996, 14, 33-38.
- (57) Ahuja, S.; Eilers, M.; Hirshfeld, A.; Yan, E. C. Y.; Ziliox, M.; Sakmar, T. P.; Sheves, M.; Smith, S. O. *J. Am. Chem. Soc.* **2009**, *131*, 15160–15169.
- (58) Hornak, V.; Ahuja, S.; Eilers, M.; Goncalves, J. A.; Sheves, M.; Reeves, P. J.; Smith, S. O. *J. Mol. Biol.* **2010**, 396, 510–527.
- (59) Vogel, R.; Siebert, F.; Mathias, G.; Tavan, P.; Fan, G. B.; Sheves, M. Biochemistry 2003, 42, 9863–9874.
- (60) Kusnetzow, A. K.; Altenbach, C.; Hubbell, W. L. *Biochemistry* **2006**, 45, 5538–5550.
- (61) Ahuja, S.; Crocker, E.; Eilers, M.; Hornak, V.; Hirshfeld, A.; Ziliox, M.; Syrett, N.; Reeves, P. J.; Khorana, H. G.; Sheves, M.; Smith, S. O. *J. Biol. Chem.* **2009**, 284, 10190–10201.
- (62) Herzfeld, J.; Das Gupta, S. K.; Farrar, M. R.; Harbison, G. S.; McDermott, A. E.; Pelletier, S. L.; Raleigh, D. P.; Smith, S. O.; Winkel, C.; Lugtenburg, J.; Griffin, R. G. *Biochemistry* **1990**, 29, 5567–5574.
- (63) Goncalves, J. A.; South, K.; Ahuja, S.; Zaitseva, E.; Opefi, C. A.; Eilers, M.; Vogel, R.; Reeves, P. J.; Smith, S. O. *Proc. Natl. Acad. Sci. U.S.A.* **2010**, *107*, 19861–19866.
- (64) Goncalves, J. A.; Ahuja, S.; Erfani, S.; Eilers, M.; Smith, S. O. Prog. Nucl. Magn. Reson. Spectrosc. 2010, 57, 159–180.
- (65) Patel, A. B.; Crocker, E.; Reeves, P. J.; Getmanova, E. V.; Eilers, M.; Khorana, H. G.; Smith, S. O. *J. Mol. Biol.* **2005**, 347, 803–812.
- (66) Choe, H. W.; Kim, Y. J.; Park, J. H.; Morizumi, T.; Pai, E. F.; Krauss, N.; Hofmann, K. P.; Scheerer, P.; Ernst, O. P. *Nature* **2011**, 471, 651–655.
- (67) Katayama, K.; Furutani, Y.; Kandori, H. J. Phys. Chem. B 2010, 114, 9039–9046.
- (68) Ratner, V. L.; Bagirov, I. G.; Fesenko, E. E. Vision Res. 1981, 21, 251–253.
- (69) Unal, H.; Jagannathan, R.; Bhat, M. B.; Karnik, S. S. J. Biol. Chem. **2010**, 285, 16341–16350.
- (70) Okada, T.; Fujiyoshi, Y.; Silow, M.; Navarro, J.; Landau, E. M.; Shichida, Y. *Proc. Natl. Acad. Sci. U.S.A.* **2002**, *99*, 5982–5987.
- (71) Lüdeke, S.; Beck, R.; Yan, E. C. Y.; Sakmar, T. P.; Siebert, F.; Vogel, R. J. Mol. Biol. **2005**, 353, 345–356.

- (72) Knierim, B.; Hofmann, K. P.; Ernst, O. P.; Hubbell, W. L. *Proc. Natl. Acad. Sci. U.S.A.* **2007**, *104*, 20290–20295.
- (73) Sheikh, S. P.; Zvyaga, T. A.; Lichtarge, O.; Sakmar, T. P.; Bourne, H. R. *Nature* **1996**, 383, 347–350.
- (74) Struts, A. V.; Salgado, G. F. J.; Martínez-Mayorga, K.; Brown, M. F. Nat. Struct. Mol. Biol. 2011, 18, 392–394.
- (75) Crocker, E.; Eilers, M.; Ahuja, S.; Hornak, V.; Hirshfeld, A.; Sheves, M.; Smith, S. O. *J. Mol. Biol.* **2006**, 357, 163–172.
- (76) Grossfield, A.; Pitman, M. C.; Feller, S. E.; Soubias, O.; Gawrisch, K. J. Mol. Biol. 2008, 381, 478–486.
- (77) Fahmy, K.; Jäger, F.; Beck, M.; Zvyaga, T. A.; Sakmar, T. P.; Siebert, F. *Proc. Natl. Acad. Sci. U.S.A.* **1993**, *90*, 10206–10210.
- (78) Nakamichi, H.; Okada, T. Angew. Chem., Int. Ed. 2006, 45, 4270-4273.
- (79) Standfuss, J.; Edwards, P. C.; D'Antona, A.; Fransen, M.; Xie, G.; Oprian, D. D.; Schertler, G. F. *Nature* **2011**, *471*, *656*–*660*.
- (80) Mahalingam, M.; Martinez-Mayorga, K.; Brown, M. F.; Vogel, R. *Proc. Natl. Acad. Sci. U.S.A.* **2008**, *105*, 17795–17800.
- (81) Ye, S. X.; Huber, T.; Vogel, R.; Sakmar, T. P. Nat. Chem. Biol. 2009, 5, 397-399.
- (82) Cherezov, V.; Rosenbaum, D. M.; Hanson, M. A.; Rasmussen, S. G. F.; Thian, F. S.; Kobilka, T. S.; Choi, H. J.; Kuhn, P.; Weis, W. I.; Kobilka, B. K.; Stevens, R. C. *Science* **2007**, *318*, 1258–1265.
- (83) Warne, T.; Serrano-Vega, M. J.; Baker, J. G.; Moukhametzianov, R.; Edwards, P. C.; Henderson, R.; Leslie, A. G.; Tate, C. G.; Schertler, G. F. X. *Nature* **2008**, *454*, 486–491.
- (84) Jaakola, V. P.; Griffith, M. T.; Hanson, M. A.; Cherezov, V.; Chien, E. Y. T.; Lane, J. R.; Ijzerman, A. P.; Stevens, R. C. *Science* **2008**, 322, 1211–1217.
- (85) Chien, E. Y. T.; Liu, W.; Zhao, Q. A.; Katritch, V.; Han, G. W.; Hanson, M. A.; Shi, L.; Newman, A. H.; Javitch, J. A.; Cherezov, V.; Stevens, R. C. Science **2010**, 330, 1091–1095.
- (86) Moukhametzianov, R.; Warne, T.; Edwards, P. C.; Serrano-Vega, M. J.; Leslie, A. G.; Tate, C. G.; Schertler, G. F. X. Proc. Natl. Acad. Sci. U.S.A. 2011, 108, 8228–8232.
- (87) Ballesteros, J. A.; Weinstein, H. Methods Neurosci. 1995, 25, 366-428.
- (88) Ahuja, S.; Smith, S. O. Trends Pharmacol. Sci. 2009, 30, 494–502.
- (89) Parkes, J. H.; Liebman, P. A. Biochemistry 1984, 23, 5054-5061.
 (90) Epps, J.; Lewis, J. W.; Szundi, I.; Kliger, D. S. Photochem.

Photobiol. 2006, 82, 1436-1441.

(91) Furutani, Y.; Kandori, H.; Shichida, Y. *Biochemistry* **2003**, *42*, 8494–8500.