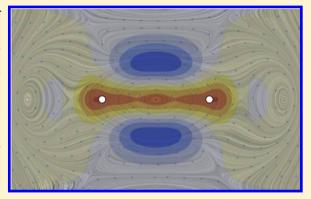


# **Current Density Functional Theory Using Meta-Generalized Gradient Exchange-Correlation Functionals**

James W. Furness,\*\*,† Joachim Verbeke,† Erik I. Tellgren,‡ Stella Stopkowicz,‡ Ulf Ekström,‡ Trygve Helgaker,‡ and Andrew M. Teale\*,†

ABSTRACT: We present the self-consistent implementation of current-dependent (hybrid) meta-generalized gradient approximation (mGGA) density functionals using London atomic orbitals. A previously proposed generalized kinetic energy density is utilized to implement mGGAs in the framework of Kohn-Sham current density functional theory (KS-CDFT). A unique feature of the nonperturbative implementation of these functionals is the ability to seamlessly explore a wide range of magnetic fields up to 1 au (~235 kT) in strength. CDFT functionals based on the TPSS and B98 forms are investigated, and their performance is assessed by comparison with accurate coupled-cluster singles, doubles, and perturbative triples (CCSD(T)) data. In the weak field regime, magnetic properties such as magnetizabilities and nuclear magnetic resonance shielding constants show modest but systematic improve-



ments over generalized gradient approximations (GGA). However, in the strong field regime, the mGGA-based forms lead to a significantly improved description of the recently proposed perpendicular paramagnetic bonding mechanism, comparing well with CCSD(T) data. In contrast to functionals based on the vorticity, these forms are found to be numerically stable, and their accuracy at high field suggests that the extension of mGGAs to CDFT via the generalized kinetic energy density should provide a useful starting point for further development of CDFT approximations.

#### 1. INTRODUCTION

The foundations of current density functional theory (CDFT) and its Kohn-Sham (KS) implementation were established in the late 1980s with the seminal works of Vignale, Rasolt, and Geldart, 1-3 where it was recognized that the exchangecorrelation functionals must depend not only on the electronic density  $\rho$  but also the paramagnetic current density  $\mathbf{j}_n$  in the presence of an electromagnetic field. Since these early works, a large number of theoretical investigations of CDFT have been presented. The foundations of CDFT have sometimes been viewed as controversial. Most recently, Pan and Sahni<sup>4-6</sup> suggested that the physical current density j, rather than j,, ought to be the fundamental variable in a CDFT and attempted to establish a Hohenberg-Kohn like theorem for this physically appealing alternative choice of variable. Unfortunately, the derivation of this theorem has been shown to be in error.<sup>7,8</sup> Furthermore, the work of Tellgren et al. showed how CDFT may be brought into Lieb's convex-conjugate formulation of density functional theory (DFT),9 further strengthening its foundations and lending key insight into the more complex relationship between the densities and potentials in the theory. In particular, it is highlighted that the lack of a Hohenberg-Kohn theorem is not an impediment to a viable CDFT. Recent theoretical works by Lieb and Schrader<sup>10</sup> and Tellgren et al.<sup>11</sup> have also addressed the issue of N-representability in CDFT.

Despite the theoretical progress in CDFT, very few practical implementations of theory have been presented. Most practical studies either have presented calculations based on fixed densities (typically computed at the Hartree-Fock or standard KS level) or have attempted to include CDFT contributions in linear-response calculations. In the context of response theory, implementations have been presented for magnetic properties by Lee et al. 12 and for excitation energies at the metageneralized gradient approximation (mGGA) level by Bates and Furche. 13 Very few fully self-consistent implementations of CDFT capable of treating systems beyond the linear-response regime have been presented. For two-dimensional systems, Vignale<sup>14</sup> has presented a self-consistent implementation of CDFT for Wigner crystals. Ferconi and Vignale<sup>15</sup> have also studied two-dimensional quantum dots in magnetic fields, using a self-consistent CDFT implementation. For three-dimensional molecular systems, we are only aware of the work by Pittalis et al. 16,17 in the context of the optimized effective potential method, the work of Zhu and Trickey<sup>18</sup> for atomic systems, and

Received: June 8, 2015 Published: August 4, 2015

<sup>&</sup>lt;sup>†</sup>School of Chemistry, University of Nottingham, University Park, Nottingham, NG7 2RD, United Kingdom

<sup>&</sup>lt;sup>‡</sup>Centre for Theoretical and Computational Chemistry, Department of Chemistry, University of Oslo, P.O. Box 1033, Blindern, N-0315, Oslo, Norway

our own implementation 19 for general atomic and molecular species.

A number of challenges arise when implementing quantum chemical methods for molecules in magnetic fields. The LONDON program<sup>20</sup> has been specifically designed to address these and is utilized throughout the present work. In particular, London atomic orbitals<sup>21–23</sup> are employed to ensure gaugeorigin invariant results. For CDFT, additional challenges arise since new forms are required for the exchange-correlation functional. Relatively few practical forms for CDFT functionals have been suggested in the literature.

In the present work, we examine the use of mGGAs and hybrid mGGAs for the exchange-correlation energy in the presence of magnetic fields. Functionals in this class depend on the orbital-dependent, canonical kinetic energy density in the absence of a magnetic field. However, as has been noted in the literature, 13,24,25 this key quantity is not gauge invariant, so some modification is required for its use in a magnetic field. One approach is to replace the canonical kinetic energy density by a generalized form including the paramagnetic current density. This quantity naturally arises in the expansion of the spherically averaged exchange hole, as derived by Dobson.<sup>26</sup> Becke has already suggested the use of this approach to produce a current-dependent generalization of the Becke-Roussel<sup>27</sup> functional. Recently, Bates and Furche<sup>13</sup> have also explored a similar generalization of the Tao-Perdew-Staroverov-Scuseria (TPSS) functional<sup>28</sup> to calculate excitation energies via response theory.

We will consider current-dependent extensions of the B98, <sup>29</sup> TPSS, <sup>28</sup> and TPSS(h) <sup>30</sup> functionals. The use of a modified current-dependent kinetic energy density is denoted by the prefix "c" throughout the remainder of this work. The nonperturbative nature of the implementation in the LONDON program allows these functionals to be tested in both weak and strong field regimes. The availability of accurate *ab initio* methodologies in the LONDON program provides a unique opportunity for the assessment and testing of CDFT functionals at field strengths up to about 1 au. In the present work, we make use of the recent implementation by Stopkowicz et al. <sup>31</sup> of coupled-cluster (CC) methods with single, double, and perturbative triple excitations [CCSD(T)] for benchmarking the CDFT approximations.

In Section 2, we review the simple generalization of mGGA functionals to the CDFT framework due to Dobson and Becke; details specific to the functionals considered in this work are collected in the Appendix. In Section 3, we outline some computational details of our calculations. Section 4 summarizes our findings, assessing the quality of the CDFT approximations by comparison with CCSD(T) data; first, in Section 4.1, we explore the performance of mGGA functionals for calculating molecular properties in the weak field regime accessible via linear response theory; next, in Section 4.2, the high-field regime is explored by considering the recently proposed perpendicular paramagnetic bonding. The interpretation of this bonding mechanism in the Kohn—Sham CDFT (KS-CDFT) framework is discussed in Section 5. Finally, concluding remarks and directions for future work are given in Section 6.

#### 2. THEORY

In this work, we consider the calculation of energies and molecular properties in the presence of a static uniform external magnetic field, **B**, which may be represented in terms of the vector potential

$$\mathbf{A}(\mathbf{r}) = \frac{1}{2}\mathbf{B} \times (\mathbf{r} - \mathbf{R}_{G}) \tag{1}$$

where  $\mathbf{R}_G$  is an arbitrary gauge origin. The London program makes use of London atomic orbitals<sup>21–23</sup> to ensure that computed energies and molecular properties are invariant with respect to choice of the gauge origin. These basis functions take the form

$$\omega_{\mu}(\mathbf{r}_{K}, \mathbf{B}, \mathbf{R}_{G}) = \exp\left[\frac{\mathrm{i}}{2}\mathbf{B} \times (\mathbf{R}_{G} - \mathbf{R}_{K}) \cdot \mathbf{r}\right] \chi_{\mu}(\mathbf{r}_{K})$$
 (2)

where  $\chi_{\mu}(\mathbf{r}_K)$  is a standard Gaussian-type orbital centered at position  $\mathbf{R}_K$ . These perturbation-dependent basis functions are used to expand the KS molecular orbitals.

The KS approach to DFT can be extended to CDFT by searching for a noninteracting system of electrons with the same charge and current densities as the physical interacting system:  $(\rho_s(\mathbf{r}), \mathbf{j}_{p,s}(\mathbf{r})) = (\rho(\mathbf{r}), \mathbf{j}_{p}(\mathbf{r}))$ . For pure (orbital independent) exchange-correlation functionals, the KS equations can be written as

$$\left[\frac{1}{2}p^2 + \frac{1}{2}\{\mathbf{p}, \mathbf{A}_{\mathrm{s}}\} + u_{\mathrm{s}} + \mathbf{s} \cdot [\nabla \times \mathbf{A}_{\mathrm{s}}]\right] \varphi_p = \varepsilon_p \varphi_p \tag{3}$$

where the KS potentials  $(u_s, A_s)$  are defined as

$$u_{\rm s} = v_{\rm ext} + \frac{1}{2}A_{\rm ext}^2 + v_{\rm J} + v_{\rm xc}, \quad \mathbf{A}_{\rm s} = \mathbf{A}_{\rm ext} + \mathbf{A}_{\rm xc}$$
 (4)

where  $(\nu_{ext}, A_{ext})$  are the physical potentials,  $(\nu_{xc}, A_{xc})$  are the exchange-correlation potentials

$$\nu_{xc}(\mathbf{r}) = \frac{\delta E_{xc}[\rho, \mathbf{j}_p]}{\delta \rho(\mathbf{r})}, \quad \mathbf{A}_{xc}(\mathbf{r}) = \frac{\delta E_{xc}[\rho, \mathbf{j}_p]}{\delta \mathbf{j}_p(\mathbf{r})}$$
(5)

and  $v_{\rm J}$  is the Coulomb potential.

A central question that immediately arises in CDFT is how the exchange-correlation functional must be modified to include current effects. While the paramagnetic current density is a valid quantity on which to base the universal density functional, it can also be shown that the exchange-correlation component must be independently gauge invariant. This places a significant constraint on the manner in which this quantity may enter any approximate CDFT functional. In contrast to standard DFT, relatively few CDFT functionals have been proposed. The majority of these are based on the vorticity

$$\nu(\mathbf{r}) = \nabla \times \left(\frac{\mathbf{j}_{p}(\mathbf{r})}{\rho(\mathbf{r})}\right)$$
(6)

witl

$$\mathbf{A}_{\mathrm{xc}} = \frac{1}{\rho} \nabla \times \frac{\delta E_{\mathrm{xc}}[\rho, \nu]}{\delta \nu} \tag{7}$$

as proposed by Vignale, Rasolt, and Geldart (VRG).<sup>3</sup> The original VRG form for the exchange-correlation energy was parametrized using Monte Carlo simulations of the high-density limit.<sup>1</sup> A number of reparameterizations for this form have been suggested based on accurate calculations in the high-density regime. <sup>12,32–34</sup> Higuchi and Higuchi<sup>35</sup> (HH) have also presented a vorticity-dependent form, derived to obey known exact relations for the CDFT exchange-correlation functional.

While the vorticity is a theoretically convenient choice to ensure the gauge invariance of the exchange-correlation energy, it has been observed that in practical self-consistent calculations it can lead to significant numerical stability issues. 18,19 How severe these issues are depends on the exact parametrization of the functional form; however, in all cases, some degree of numerical regularization is required to ensure that the selfconsistent field solution of the KS equations can be obtained. Furthermore, molecular properties computed by such calculations exhibit an unacceptably strong dependence on the regularization parameters, with no obvious convergence toward a single value. Clearly, this raises questions as to how appropriate such forms are for use in quantum chemical calculations.

Most practical mGGAs make use of the kinetic energy density

$$\tau_{\sigma} = \sum_{i}^{\text{occ}} \nabla \varphi_{i\sigma}^{*} \cdot \nabla \varphi_{i\sigma} \tag{8}$$

in their construction, where  $\varphi_i$  are the occupied KS orbitals and  $\sigma$  is the electron spin index. This term is gauge-dependent, and an unmodified mGGA-type functional form, therefore, cannot be used to describe a system with a nonzero magnetic vector potential. To resolve this issue, the gauge independence of the exchange-correlation functional must be restored. A natural modification, which can be applied to any mGGA dependent on the kinetic energy density  $\tau(\mathbf{r})$ , arises in the work of Dobson, 26,36 who generalized the expansion of the exchange hole to include the case of nonzero current densities.

The spherically averaged exchange hole at zero field can be modeled using a Taylor expansion<sup>37</sup> and is commonly considered<sup>27,29,38</sup> up to the quadratic term

$$Q_{\sigma} = \frac{1}{6} \left[ \nabla^2 \rho_{\sigma} - 2\tau_{\sigma} + \frac{(\nabla \rho_{\sigma})^2}{2\rho_{\sigma}} \right]$$
 (9)

This expansion can be generalized to nonzero field, and the curvature term becomes2

$$Q_{\sigma} = \frac{1}{6} \left[ \nabla^2 \rho_{\sigma} - 2\tau_{\sigma} + \frac{(\nabla \rho_{\sigma})^2}{2\rho_{\sigma}} + \frac{2|\mathbf{j}_{p\sigma}|^2}{\rho_{\sigma}} \right]$$
(10)

where  $\mathbf{j}_{v\sigma}$  is the paramagnetic current density

$$\mathbf{j}_{p\sigma} = -\frac{i}{2} \sum_{i}^{\text{occ}} \left[ \varphi_{i\sigma}^* \nabla \varphi_{i\sigma} - \varphi_{i\sigma} \nabla \varphi_{i\sigma}^* \right] \tag{11}$$

Comparing eqs 9 and 10, it is possible to identify a correction to the conventional  $\tau(\mathbf{r})$  that is gauge-invariant and may be utilized in mGGA functionals

$$\tau_{\sigma} \to \tilde{\tau}_{\sigma} = \tau_{\sigma} - \frac{|\mathbf{j}_{p\sigma}|^2}{\rho_{\sigma}}$$
(12)

The use of eq 12 has been put forward many times in the literature. Becke suggested its use in the Becke-Roussel model<sup>27</sup> to generate a current-dependent analogue of this functional. He also suggested that this quantity could be used to define a current-dependent inhomogeneity parameter in the more empirical B98 functional.<sup>29</sup> It has also been suggested for use to generalize the TPSS functional<sup>28</sup> by Tao.<sup>25</sup> Recently, Bates and Furche<sup>13</sup> considered the application of the resulting

cTPSS functional in the calculation of excitation energies via response theory.

In the present work, we consider the application of mGGA functionals with the modification in eq 12 to calculate magnetic properties in the weak and strong field regimes in a nonperturbative manner. For an mGGA functional that depends on the current only through the kinetic energy density, the exchange-correlation functional may be written as

$$E_{xc}[\rho, \mathbf{j}_{p}, \tau] = \int f_{xc}(\rho(\mathbf{r}), \nabla \rho(\mathbf{r}), \mathbf{j}_{p}(\mathbf{r}), \tau(\mathbf{r})) d\mathbf{r}$$
$$= \int g_{xc}(\rho(\mathbf{r}), \nabla \rho(\mathbf{r}), \tilde{\tau}(\mathbf{r})) d\mathbf{r}$$
(13)

and the KS eigenvalue problem takes the form (disregarding spin dependence)

$$\left[\frac{1}{2}\mathbf{p}\cdot(1+\eta_{xc})\mathbf{p}+\frac{1}{2}\{\mathbf{p},\mathbf{A}_{ext}+\mathbf{A}_{xc}\}+u_{s}\right]\varphi_{p}=\varepsilon_{p}\varphi_{p}$$
(14)

$$\eta_{\rm xc} = \frac{\partial f_{\rm xc}}{\partial \tau} = \frac{\partial g_{\rm xc}}{\partial \tilde{\tau}} \tag{15}$$

$$\mathbf{A}_{\mathrm{xc}} = \frac{\partial f_{\mathrm{xc}}}{\partial \mathbf{j}_{p}} = -2\eta_{\mathrm{xc}} \frac{\mathbf{j}_{p}}{\rho} \tag{16}$$

Unlike for the VRG functional, the vector potential  $\mathbf{A}_{xc}$  is now gauge-dependent. However, the gauge dependence of  $A_{xc}$ cancels against the gauge dependence of the  $\eta_{xc}$ -dependent modification to the canonical kinetic energy. The generalization to spin-resolved densities is straightforward.

We consider here three mGGA functionals, cB98, cTPSS, and cTPSS(h), where the prefix "c" denotes the use of the modified  $\tilde{\tau}_{\sigma}$  in eq 12. The Appendix gives some details of the respective functional forms, showing how these modifications

#### 3. COMPUTATIONAL DETAILS

Unless otherwise indicated, all calculations in this work use the uncontracted, Cartesian, aug-cc-pCVTZ basis set. 39,40 All DFT calculations have been performed using the LONDON<sup>20</sup> program. This code utilizes the XCFun library for the evaluation of the density functionals and their derivatives. The modifications of eq 12 and the functionals cB98 and cTPSS have been added to the XCFun library. In addition, we investigate the use of a hybrid form of cTPSS, denoted cTPSS(h), based on the TPSS(h) functional of ref 42. The quality of the CDFT functionals cB98, cTPSS, and cTPSS(h) is assessed by comparison with CCSD(T) data. For comparison, Hartree-Fock (HF), local density approximation (LDA), Perdew-Burke-Ernzerhof (PBE),<sup>43</sup> and Keal-Tozer-3 (KT3)<sup>44</sup> density functional results are also presented. The latter is of particular interest since it is specifically designed for the calculation of nuclear magnetic resonance (NMR) shielding constants.

The performance of these approximations will be considered in two regimes: the weak field regime accessible by linear response calculations and the strong field regime only accessible via nonperturbative calculations. In the weak field regime, we will consider the calculation of molecular magnetizabilities and NMR shielding constants for the 26 small molecules in Table 1. Errors for these quantities are presented relative to the benchmark data of ref 45. Results are also

Table 1. Test Set of Molecules for Which Accurate Benchmark CCSD(T) Data from Ref 45 Was Available

HF	CO	$N_2$	$H_2O$	HCN	HOF	LiH
$NH_3$	$H_2CO$	$CH_4$	$C_2H_4$	AlF	$CH_3F$	$C_3H_4$
FCCH	FCN	$H_2S$	HCP	HFCO	$H_2C_2O$	LiF
	$N_2O$	OCS	$H_4C_2O$	PN	$SO_2$	

compared with those including corrections from the Tao–Perdew parametrization<sup>33</sup> of the Vignale–Rasolt–Geldart functional,<sup>2</sup> taken from ref 19. In the strong field regime, we consider the prediction of perpendicular paramagnetic bonding<sup>46</sup> in a field strength of 1 au perpendicular to the internuclear axes of H<sub>2</sub>, He<sub>2</sub>, HeNe, and Ne<sub>2</sub>. These nonperturbative calculations are assessed against CCSD(T) results computed using the implementation of ref 31 in the LONDON program.

#### 4. RESULTS

**4.1. The Weak Field Regime: Magnetic Properties.** We commence by considering the molecular magnetizabilities and NMR shielding constants of the 26 small molecules in Table 1. The magnetizability tensor elements,  $\xi_{\alpha,\beta}$ , are defined as

$$\xi_{\alpha,\beta} = \frac{\partial^2 E(\mathbf{B})}{\partial B_{\alpha} \partial B_{\beta}} \bigg|_{\mathbf{B} = \mathbf{0}} \tag{17}$$

where  $\alpha$  and  $\beta$  label Cartesian components of the tensor and magnetic field. The NMR shielding tensor for a given nucleus K is defined by

$$\sigma_{K;\alpha,\beta} = \left. \frac{\partial^2 E(\mathbf{B}, \, \mathbf{M}_K)}{\partial B_\alpha \partial M_{K,\beta}} \right|_{\mathbf{B} = \mathbf{0}, \mathbf{M}_K = \mathbf{0}}$$
(18)

where  $\mathbf{M}_K$  is the nuclear magnetic moment of nucleus K. These properties can be accessed nonperturbatively in the London program by explicit calculation of the energy in the presence of the perturbing fields. Details of this procedure are given in ref 19. Here, we compute the properties in the same manner, facilitating a comparison with previous results.

Given that for many density functional approximations these singlet second-order magnetic response properties can be accessed by standard linear response methods in a variety of programs, this approach may seem cumbersome. However, it should be noted that the implementation of the new CDFT

approaches in this framework is much more straightforward, requiring only an implementation of the functional and the derivatives required for construction of the KS matrix. More importantly, as we will see in Section 4.2, this nonperturbative approach allows us to seamlessly explore the behavior of new approximations in much stronger fields: those inaccessible via linear response theory. This means that London provides a powerful test bed for new CDFT functionals.

To quantify the accuracy of the DFT approaches for the calculation of these properties, we compare our results with the CCSD(T) benchmark values of ref 45. Specifically, we use the values at the CCSD(T)/aug-cc-pCV[TQ]Z level, which have been extrapolated to the basis set limit using the procedure of refs 45 and 47.

The errors in the calculated magnetizabilities and NMR shielding constants are summarized in Table 2 and presented graphically in Figures 1 and 2 as box—whisker plots. In these plots, individual points represent the errors for each system relative to the reference values, the upper and lower fences of the whiskers denote the maximum positive and negative errors, respectively, and the colored boxes enclose errors between the 25 and 75% quantiles. Mean and median errors are marked in the plots by horizontal black and white lines, respectively. Gray diamonds are used to represent the confidence intervals.

For the molecular magnetizabilities, it is clear from the error measures in Table 2 and their representation in Figure 1 that none of the functionals offers high accuracy. The GGA functionals PBE and KT3 in particular do not offer significant improvements over LDA. While their minimum and maximum errors are slightly improved, the mean errors actually deteriorate. Similar observations were made in ref 47 for these type of functionals. The B97-2 functional gives slightly reduced errors, which is consistent with previous conclusions<sup>4</sup> that for magnetizabilities the inclusion of HF exchange may be beneficial. At the GGA level, the underlying functionals are already gauge-invariant but do not depend explicitly on the paramagnetic current density. To introduce this dependence, the VRG functional may be added. In our earlier work, 19 we found that this correction can be numerically problematic and that the most stable parametrization of this functional to date is that put forward by Tao and Perdew,<sup>33</sup> denoted VRG(TP). For comparison, we include here the PBE + VRG(TP) and KT3 + VRG(TP) results. It is clear that the inclusion of the VRG(TP)correction actually worsens the agreement of the results with CCSD(T) reference data. At the mGGA level, the inclusion of

Table 2. Mean Error (ME), Mean Absolute Error (MAE), and Standard Deviation (SD) of Magnetic Properties Relative to the CCSD(T) Benchmark Data<sup>a</sup> of Refs 45 and 47

	mag	netizability (10 <sup>-30</sup> J 7	$\Gamma^{-2}$ )	N	NMR shielding (ppm)	
	ME	MAE	SD	ME	MAE	SD
HF	-3.06	6.35	7.29	-15.18	21.40	40.97
LDA	5.01	9.18	10.86	-24.81	24.85	30.00
PBE	6.75	8.81	9.03	-19.66	19.78	21.46
PBE + VRG(TP)	7.85	9.61	9.44	-20.33	20.46	22.43
KT3	8.18	8.95	7.83	-6.53	8.94	13.13
KT3 + VRG(TP)	9.18	9.83	8.19	-7.45	9.18	13.37
B97-2	5.46	5.84	5.96	-16.34	16.48	20.60
cB98	0.52	4.84	6.58	-12.44	12.66	17.79
cTPSS	7.13	7.51	6.76	-14.14	14.35	15.61
cTPSS(h)	6.41	6.51	6.00	-14.33	14.52	16.68

<sup>&</sup>lt;sup>a</sup>See also Figures 1 and 2.

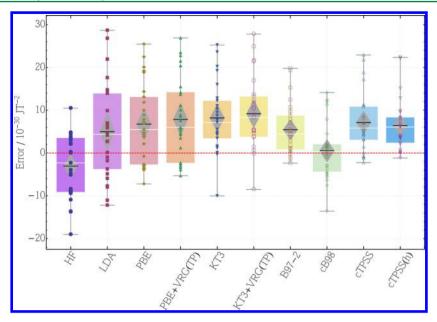


Figure 1. Box—whisker plot of errors in (C)DFT molecular magnetizabilities relative to the extrapolated CCSD(T) values of ref 47. All calculations use the uncontracted aug-cc-pCVTZ basis set. See the text for further details.

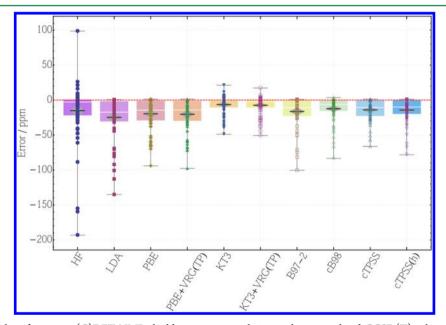


Figure 2. Box—whisker plot of errors in (C)DFT NMR shielding constants relative to the extrapolated CCSD(T) values of ref 45. All calculations use the uncontracted aug-cc-pCVTZ basis set. See the text for further details.

current is mandatory to ensure the exchange-correlation evaluation is gauge-independent.

Here, we investigate the cB98, cTPSS, and cTPSS(h) functionals; the TPSS and TPSS(h) forms are similar in performance to PBE, offering marginal improvements on some error measures, with TPSS(h) performing slightly better than TPSS. The cB98 form gives the best performance of all the functionals considered. It is noteworthy that in the mGGA class the mean errors reduce as more HF exchange is included in the functional, with cTPSS, cTPSS(h), and cB98 containing 0, 10, and 19.85% HF exchange, respectively. This suggests that the treatment of exchange may be the dominant factor in the errors. Since the treatment of long-range exchange is not rectified in the transition from GGA to mGGA type functionals (and only partially corrected by a global admixture), then it

may be that this factor far outweighs any improvements due to the inclusion of current effects.

It is worth emphasizing that in the course of our investigation we found that the implementation of the mGGA functionals including a generalized kinetic energy density was straightforward. In particular, we found that no special care was required with respect to numerics compared with standard functionals and that in practical use the functionals are robust and that self-consistent calculations using these functionals converge without significant difficulty. This sharply contrasts the behavior for the VRG functionals as investigated in refs 18 and 19.

The results for NMR shielding constants are presented in Figure 2. Here, we see LDA is poor, as expected. In addition, KT3, which was designed for these properties, is the best performing functional, significantly improving over the standard

Table 3. Dissociation Energies and Equilibrium Bond Lengths for  $H_2$  and Rare Gas Dimers in a 1 au Magnetic Field Perpendicular to the Internuclear Axis

		$R_{ m e}/a_0$			$D_{ m e}/{ m mE_h}$			
	H <sub>2</sub>	He <sub>2</sub>	NeHe	Ne <sub>2</sub>	$H_2$	He <sub>2</sub>	NeHe	Ne <sub>2</sub>
HF	2.709	3.296	3.543	3.773	2.333	0.218	0.482	0.978
LDA	2.377	2.550	2.846	3.062	14.807	8.231	11.817	19.730
PBE	2.515	2.810	3.080	3.294	5.976	2.463	3.938	7.128
KT3	2.512	2.852	3.121	3.342	4.628	2.661	3.937	6.527
cB98	2.636	3.203	3.472	3.676	1.324	1.011	1.514	2.050
cTPSS	2.567	2.864	3.124	3.346	5.250	1.307	2.344	4.577
cTPSSh	2.562	2.879	3.154	3.379	5.258	1.245	2.130	3.990
CCSD(T)	2.584	2.977	3.248	3.487	4.551	1.259	2.217	4.016
FCI <sup>a</sup>	2.584	2.975			4.551	1.271		

<sup>a</sup>The He<sub>2</sub> FCI calculations use the aug-cc-pVTZ basis set.

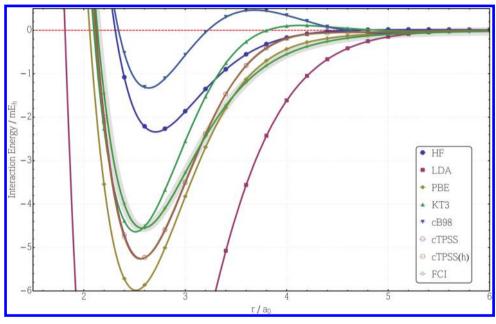


Figure 3. Potential energy curve for the  ${}^3\Sigma_u^+(1\sigma_g 1\sigma_u^*)$  state of the  $H_2$  molecule in a magnetic field of 1 au perpendicular to the bonding axis for a variety of methods with the uncontracted aug-cc-pCVTZ basis set.

PBE GGA functional and the B97-2 hybrid functional. In this case, we see the addition of the VRG(TP) correction to PBE and KT3 has little effect, very slightly deteriorating the results. The mGGA results for cB98, cTPSS, and cTPSS(h) are intermediate between PBE and KT3, offering small systematic improvements over PBE. Again, B98 produces the best results of the mGGA functionals.

On the whole, the quality of the mGGA results at modest field strengths may be regarded as disappointing. The overall errors suggest that mGGAs may offer modest improvements over conventional GGA functionals such as PBE, although they cannot compete with GGAs tailored to specific properties. This is broadly consistent with findings by Bates and Furche for the calculation of excitation energies using cTPSS. <sup>13</sup> Since current corrections are known to be relatively small, it is important that the underlying functional should be relatively accurate. For the mGGAs considered here, there are known weaknesses (for example, in the treatment of long-range exchange) that may obscure the effect of the current dependence. For the case of NMR shielding constants, a more detailed analysis of the significance of current effects and how these interplay with errors in a range of density functionals is presented in ref 48.

We will now examine how these functionals perform when the magnetic field becomes much higher and has a stronger effect on the electronic structure.

4.2. The Strong Field Regime: Paramagnetic Bonding. One approach to explore whether the inclusion of current effects via the modified kinetic energy density of eq 12 is physically reasonable is to increase the strength of the magnetic field. Lange et al. 46 have recently performed full configuration interaction (FCI) calculations at high field that have uncovered a new mechanism for chemical bonding in the presence of a strong magnetic field. This new bonding has been termed perpendicular paramagnetic bonding and occurs at field strengths similar to those found on some white dwarf stars. Since this work, Murdin et al.<sup>49</sup> have shown that phosphorus and selenium doped silicon semiconductors can produce a viable laboratory analogue of free hydrogen<sup>49</sup> and helium<sup>50</sup> in strong magnetic fields. The description of these types of systems via quantum chemistry will require less computationally demanding approaches, and CDFT is one strong candidate for the simulation of these systems.

To investigate the performance of cB98, cTPSS, and cTPSS(h) in strong magnetic fields, potential energy profiles

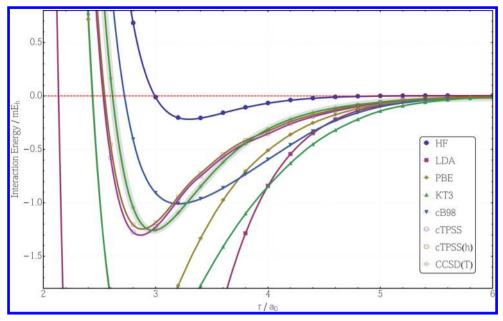


Figure 4. Potential energy curve for the lowest  ${}^{1}\Sigma_{g}^{+}$  state of He<sub>2</sub> in a magnetic field of 1 au perpendicular to the bonding axis for a variety of methods with the uncontracted aug-cc-pCVTZ basis set.

were calculated for H2, He2, NeHe, and Ne2. In particular, we consider the  ${}^{3}\Sigma_{u}^{+}(1\sigma_{\sigma}1\sigma_{u}^{*})$  state of H<sub>2</sub> and the lowest  ${}^{1}\Sigma_{\sigma}^{+}$  states of He2, HeNe, and Ne2. Each of these states is repulsive or weakly dispersion bound in the absence of a magnetic field but becomes more strongly bound when a field is applied. We note that only the  $^3\Sigma_u^+(1\sigma_{_{\!\it u}}1\sigma_{_{\!\it u}}^*)$  state of  ${\rm H_2}$  is an overall ground state in the presence of the field. These states were compared against results from accurate CCSD(T) potential energy curves calculated using a recent nonperturbative implementation by Stopkowicz et al.<sup>31</sup> in the London program. For comparison, we have also generated similar profiles with standard LDA and GGA density functionals as well as with HF theory. The calculated potential energy curves for H<sub>2</sub>, He<sub>2</sub>, NeHe, and Ne<sub>2</sub> are shown in Figures 3, 4, 5, and 6, respectively. Equilibrium bond lengths and dissociation energies were determined numerically and are presented in Table 3.

The  ${}^{3}\Sigma_{u}^{+}(1\sigma_{a}1\sigma_{u}^{*})$  state of the H<sub>2</sub> molecule in a perpendicular field was examined at the FCI level by Lange et al. 46 The potential energy curves for this state are shown in Figure 3. HF strongly underbinds this state in comparison with the FCI data. In contrast, LDA strongly overbinds, a tendency that is largely corrected by the PBE functional and further improved by the cTPSS and cTPSS(h) models. These trends are reflected in the equilibrium bond lengths and dissociation energies in Table 3. Although they are not highly accurate, the cTPSS and cTPSS(h) models give a reasonable qualitative description of the potential energy curve. The empirically parametrized KT3 functional is interesting because it gives simultaneously a reasonable estimate of both the equilibrium bond length and dissociation energy. However, at intermediate separation, an unphysical barrier is observed. For B98, an even more pronounced barrier is present, and the potential energy curve is generally even less accurate than Hartree-Fock theory. This may suggest that heavily parametrized functional forms, determined to perform well at zero field, may not be the best candidates for use in strong-field CDFT studies.

Examining the potential energy curves for  $He_2$  in Figure 4, we see that HF tends to underbind, with a bond length of  $3.30a_0$ , compared with the CCSD(T) value of  $2.98a_0$ . Similarly,

the HF dissociation energy of 0.218 mE<sub>h</sub> is much smaller than the corresponding CCSD(T) value of 1.259 mE<sub>h</sub>. For this small system, we were able to compare the CCSD(T) results with FCI values (calculated in the slightly smaller aug-cc-pVTZ basis); as expected, the agreement is excellent: the corresponding potential energy curves are essentially indistinguishable on the scale of Figure 4. For LDA, we see a strong tendency to overbind, giving much too short  $R_{\rm e}$  values and much too large estimates of  $D_{\rm e}$ . The GGA functionals PBE and KT3 show considerable improvement over LDA; however, they still strongly overbind. The improvement for the mGGA functionals is striking; in particular, TPSS and TPSS(h) give a good qualitative description of the potential energy curve. The corresponding  $R_{\rm e}$  and  $D_{\rm e}$  values indicate that there still remains a tendency toward overbinding, but this is greatly reduced.

The cB98 functional tends to show more significant underbinding. Here, we note that the arguments used in the construction of cB98 and cTPSS are rather different. In particular, cB98 is an empirically parametrized functional (see the Appendix), whereas cTPSS is constructed based on the satisfaction of known exact conditions. In this work, we have used the parameters determined in ref 29 to define the cB98 form. These parameters were determined at zero field and from post-LDA calculations; as a result, they may not be optimal for fully self-consistent calculations in the presence of a magnetic field. On the other hand, the cTPSS functional is designed to satisfy selected constraints at zero field, and it could be argued that in the presence of a field both the B98- and TPSS-based functionals are open to further optimization, although this is beyond the scope of the present work.

The stability of the mGGA functionals is particularly evident in the strong field regime when one compares the present results for He<sub>2</sub> with those for the VRG-based estimates in Figure 7 of ref 19. The VRG approaches led to very difficult SCF convergence and complex potential energy curves with a strong unphysical overbinding. The mGGAs considered here are unproblematic in practical application and yield results surprisingly close to the CCSD(T) estimates.

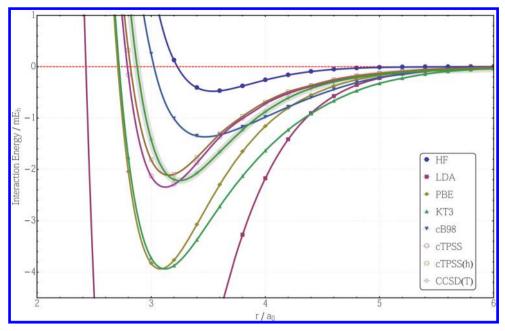


Figure 5. Potential energy curve for the lowest  ${}^{1}\Sigma_{g}^{+}$  state of the NeHe dimer in a magnetic field of 1 au perpendicular to the bonding axis for a variety of methods with the uncontracted aug-cc-pCVTZ basis set.

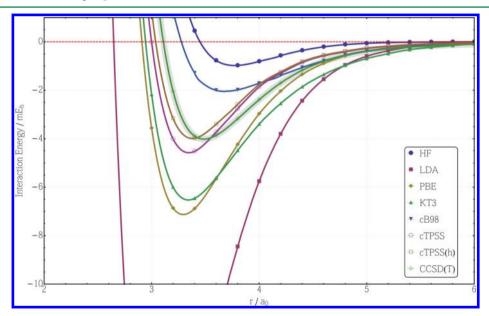


Figure 6. Potential energy curve for the lowest  ${}^{1}\Sigma_{g}^{+}$  state of Ne<sub>2</sub> in a magnetic field of 1 au perpendicular to the bonding axis for a variety of methods with the uncontracted aug-cc-pCVTZ basis set.

Similar qualitative trends are observed for the NeHe and NeNe dimers in Figures 5 and 6. Again, LDA and GGA functionals are not sufficiently accurate for practical use, and the mGGA functionals provide a large improvement. The cTPSS and cTPSS(h) results remain impressive, with cTPSS(h) being consistently slightly more accurate than cTPSS. This trend is reflected in both the potential energy curves and the  $R_{\rm e}$  and  $D_{\rm e}$  values in Table 3.

# 5. INTERPRETATION OF PARAMAGNETIC BONDING IN THE KS-CDFT FRAMEWORK

The mGGA CDFT functionals offer a computationally cheap correlated method for the examination of the exotic bonding mechanisms observed in a strong magnetic field. In many areas of chemistry, the nature of bonding, chemical reactions, spectra, and properties of molecular species are interpreted qualitatively in terms of orbital interactions. We now consider the extent to which information from KS-CDFT calculations can aid in simple interpretation of the perpendicular paramagnetic bonding interactions.

We begin by considering a molecular orbital analysis of the perpendicular paramagnetic bonding. KS-CDFT calculations provide a simple set of canonical molecular orbitals, which can be used to construct the electronic density via

$$\rho(\mathbf{r}) = \sum_{i} \varphi_{i}(\mathbf{r}) \ \varphi_{i}^{*}(\mathbf{r}) = \sum_{\gamma \zeta} \omega_{\gamma}(\mathbf{r}) \ D_{\gamma \zeta} \omega_{\zeta}^{*}(\mathbf{r})$$
(19)

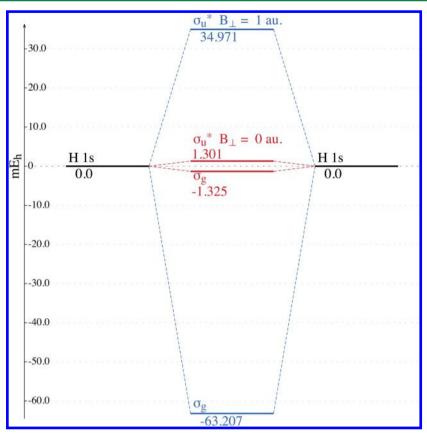


Figure 7. Relative orbital energies for the occupied  $\sigma_g$  and  $\sigma_u^*$  CDFT molecular orbitals for  $H_2$  relative to the atomic 1s orbitals, with (blue) and without (red) a field of 1.0 au perpendicular to the interatomic axis.

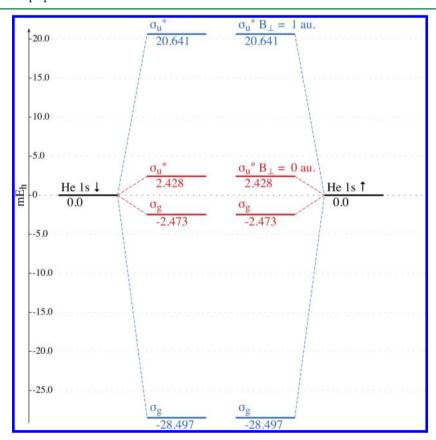


Figure 8. Relative orbital energies for the occupied  $\sigma_g$  and  $\sigma_u^*$  CDFT molecular orbitals for He<sub>2</sub> relative to the atomic 1s orbitals, with (blue) and without (red) a field of 1.0 au perpendicular to the interatomic axis.

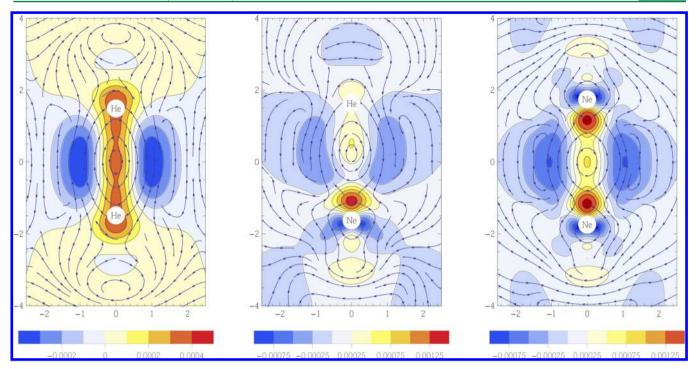


Figure 9.  $\Delta \rho_{B_1}(\mathbf{r})$  (colored contours) and  $\Delta \mathbf{j}_{B_1}(\mathbf{r})$  (streamlines) for the rare gas dimers He<sub>2</sub> (left), HeNe (middle), and Ne<sub>2</sub> (right) in a  $B_{\perp}$  = 1.0 au magnetic field. The internuclear axis is aligned with the *z*-axis, and the plots are show in the *yz* plane intersecting the atomic positions.

Here, we note that the occupied KS orbitals  $\varphi_i(\mathbf{r})$  can be complex in the presence of a magnetic field. In the second equality, the density is expressed in terms of the one-particle density matrix  $D_{\gamma\zeta}$  and the basis functions  $\omega_{\gamma}(\mathbf{r})$ . We will commence by considering how the molecular orbital energies associated with  $H_2$  and the rare gas dimers change upon application of a magnetic field as the perpendicular paramagnetic bonding in Section 5.1 evolves. Since the orbitals themselves can be complex in the presence of a field, we then proceed in Section 5.2 to analyze the bonding in terms of the changes in electronic density of eq 19 as a function of field, which is naturally a real observable quantity.

5.1. KS-CDFT Molecular Orbital Analysis. KS molecular orbitals have been widely used as an interpretive aid in chemical applications throughout the literature. The KS orbitals are defined to minimize the noninteracting kinetic energy and yield the physical electronic density via eq 19. They also have appealing properties; for example, the highest occupied MO energy is minus the first ionization potential (IP,)51,52 and the remaining orbital energies can be interpreted as Koopman's type approximations to higher IPs. 53 The extent to which these properties hold for general practical approximations has been a subject of debate in the literature, <sup>54,55</sup> as has the interpretation of KS virtual orbitals<sup>56</sup> owing to the role of the integer discontinuity,<sup>51</sup> which is missing from common approximations. However, from a practical standpoint, it is widely accepted that interpretations based on occupied KS orbitals (to which we limit the following discussion) are theoretically justified, and their utility has been borne out in many practical applications.

We now consider how the KS orbital energies change upon application of a perpendicular magnetic field of 1 au. Given that the cTPSS-based models seem to be the most reliable of those studied in the present work, we consider how the orbital energies from this functional change in Figures 7 and 8 for H<sub>2</sub>.

and He<sub>2</sub>, respectively. For the H<sub>2</sub> molecule in the  $^3\Sigma^+_u(1\sigma_g1\sigma^*_u)$  state, we consider the energy of the occupied  $\sigma_g$  and  $\sigma^*_u$  orbitals upon bonding. In particular, we plot orbital energies in the absence of a field and in a perpendicular field of 1 au relative to those of the atomic orbitals in the same field. We have plotted the orbital energies at an internuclear separation  $R_{\rm e}=2.564$  au, consistent with the cTPSS equilibrium bond length in the presence of a field.

We see that in the absence of a field the singly occupied  $\sigma_g$  orbital is stabilized by 1.33 mE<sub>h</sub>, whereas the singly occupied  $\sigma_u^*$  is destabilized by 1.30 mE<sub>h</sub> relative to the 1s hydrogen orbitals. This is consistent with a net bond order of zero and a repulsive profile for the corresponding potential energy curve. In the perpendicular field of 1 au, we see that, relative to the hydrogen 1s orbital in the same field, the  $\sigma_g$  orbital is stabilized by 63.21 mE<sub>h</sub>, whereas the  $\sigma_u^*$  orbital is destabilized by 34.97 mE<sub>h</sub>. This greater stabilization of the  $\sigma_g$  orbital leads to a net bonding interaction, consistent with the analysis of ref 46. This illustrates that the KS orbitals can be a useful tool in rationalizing this exotic bonding phenomenon.

A similar analysis can be carried out for the lowest  $^1\Sigma_g^+$  state of He<sub>2</sub> and is presented in Figure 8. In this plot, we have separated the spin-down and spin-up orbitals and defined their energies relative to atomic orbitals of the same spin in the same field. Defined in this way, the offset between the orbitals of different spin due to the Zeeman interaction is removed from the plot. Again, the energies correspond to the cTPSS He<sub>2</sub> equilibrium internuclear separation  $R_e = 2.864$  au in the presence of a perpendicular field. In the absence of a field, we see that, again, the relative stabilization of  $\sigma_g$  and destabilization of  $\sigma_u^*$  are approximately compensatory, whereas in the presence of a field, the  $\sigma_g$  orbital is more stabilized than the  $\sigma_u^*$  orbital is destabilized. It is also clear that the extra stabilization of the  $\sigma_g$  orbital of  $\sim$ 8 mE<sub>h</sub> is considerably less than the  $\sim$ 28 mE<sub>h</sub> observed for H<sub>2</sub>. This is consistent with the strength of binding

exhibited for these species in Figures 3 and 4 as well as in Table 3.

Similar orbital energy diagrams may be constructed for the HeNe and NeNe systems; however, they become significantly more complex due to large differences between the orbital energies in HeNe and the splitting of the p-orbitals in NeNe. We therefore consider an alternative visualization of the bonding effects in these systems based on the charge and (physical) current density differences.

**5.2. Electron Density Analysis.** For each of the species  $\text{He}_2$  HeNe, and  $\text{Ne}_2$  we have performed calculations of the electronic density and current density in the presence of varying perpendicular magnetic fields  $(B_\perp)$  using the cTPSS functional at the corresponding equilibrium geometries. Here, we consider the density change  $\Delta \rho_{B_\perp}(\mathbf{r})$  for each system relative to the isolated atoms in the same field

$$\Delta \rho_{B_{\perp}}(\mathbf{r}) = \rho_{B_{\perp}}^{\text{dimer}}(\mathbf{r}) - \sum_{i=1}^{N_{\text{atoms}}} \rho_{B_{\perp}}^{i}(\mathbf{r})$$
(20)

This difference density allows for a visualization of the nature of the paramagnetic bonding in these systems. In a similar manner, one can consider the (gauge invariant) physical current density difference  $\Delta j$ 

$$\Delta \mathbf{j}_{B_{\perp}}(\mathbf{r}) = \mathbf{j}_{B_{\perp}}^{\text{dimer}}(\mathbf{r}) - \sum_{i=1}^{N_{\text{storms}}} \mathbf{j}_{B_{\perp}}^{i}(\mathbf{r})$$
(21)

In Figure 9, we present plots of the density differences in eqs 20 and 21 for each of the species with  $B_{\perp}=1.0$  au. The shading of the contours represents the buildup (red) or depletion of the charge density (blue), relative to two noninteracting atoms in the same field. In all three cases, there is a clear buildup of density between the atoms consistent with bonding. The charge density difference is elongated along the field, above and below the plane of the plots. The streamlines show the vector field associated with the current density difference. Paratropic circulations are clearly visible over the center of the bonds where charge density accumulates, and diatropic circulations are visible in regions where the charge density is depleted. We have confirmed that for higher fields the alignment between para-/diatropic circulations and charge accumulation/depletion becomes more pronounced.

This picture of the perpendicular paramagnetic bonding suggests that as the charge density is elongated along the field the constituent atoms may approach one another more closely. As they do so, they experience a greater nuclear—nuclear repulsion, which may be screened by a rearrangement of the charge density toward the bond center. This rearrangement is accompanied by consistent para- and diatropic current circulations. The cTPSS functional used in this work provides a simple, computationally cheap route to perform analysis of the bonding encountered in the strong field regime.

#### 6. CONCLUSIONS

In this work, we have implemented a previously detailed <sup>25,26,36</sup> modification to the kinetic energy density term in mGGA functionals to perform nonperturbative cDFT calculations in a stable, gauge-invariant manner using the London program. <sup>20</sup> The modified mGGA functionals cB98, cTPSS, and cTPSS(h) show a level of accuracy in predicting weak field magnetic properties that is competitive with existing GGA functionals without any additional fitting. The functionals cTPSS and

cTPSS(h) show excellent prediction of perpendicular paramagnetic bonding behavior, suggesting that the modification of the kinetic energy density is a viable route for the incorporation of current effects in standard mGGA density functionals. In contrast to vorticity-dependent forms previously studied, the functionals exhibited excellent numerical stability in the finite field setting without the need for delicate numerical regularizations.

While the mGGA results show considerable promise in the high-field regime, their performance in the weak field regime is perhaps disappointing, leading to only modest improvements of conventional GGA forms. To some extent, this may be due to the fact that mGGAs contain many of the shortcomings associated with GGA forms. For example, the functionals still have potentials with incorrect asymptotic behavior, particularly for the exchange contribution. Since the current effects at weak field are not dominant but rather add small corrections to the predominantly Coulombic exchange and correlation interactions, then it may be necessary to further improve the underlying functional forms before the true impact of the current terms can be assessed in this regime.

We expect that this approach to include current dependence should play a central role in the future development of new CDFT functionals, and many avenues are open for development. Obvious possibilities include the generalization of rangeseparated mGGA functionals<sup>57,58</sup> to obtain a better balance of errors among exchange, correlation, and current contributions and the reparameterization of functionals either empirically or via the consideration of alternative exact conditions in their construction. In the latter category, we note that the presence of a magnetic field causes compression of the electronic density in the two dimensions perpendicular to the field and elongation along it. As a result, nonuniform coordinate scaling relations may be one example of conditions that may provide a powerful tool in further development. The LONDON program<sup>20</sup> provides a powerful platform for this work since CDFT approaches can be calibrated against accurate ab initio data for a range of field strengths, where experimental data may be scarce. In the future, we hope that these relatively inexpensive CDFT approaches may then be applied to the study of larger systems, such as those in refs 49 and 50.

### APPENDIX

#### A. Meta-GGAs in CDFT

We present the key working equations defining the B98 and TPSS functionals, indicating how the modified  $\tilde{\tau}_{\sigma}$  of eq 12 enters each functional.

*A.1. cB98.* The B98 functional<sup>29</sup> has a general construction that is similar to the popular B97 functional form;<sup>59</sup> however, instead of using reduced spin-density gradients, it makes use of an inhomogeneity parameter  $q_{\sigma}$ 

$$q_{\sigma} = \frac{(Q_{\sigma} - Q_{\sigma}^{\text{UEG}})}{|Q_{\sigma}^{\text{UEG}}|} \tag{22}$$

where the exchange hole curvature for the uniform electron gas takes the simple form

$$Q_{\sigma}^{\text{UEG}} = -\frac{1}{5} (6\pi^2)^{2/3} \rho_{\sigma}^{5/3} \tag{23}$$

and  $Q_{\sigma}$  is defined in eq 9. This inhomogeneity factor,  $q_{\sigma}$  controls the enhancement or attenuation of the exchange and correlation energy over the uniform gas values.

The exchange component of the functional takes the form

$$E_{\mathbf{x},\sigma} = \int \! e_{\mathbf{x},\sigma}^{\mathrm{UEG}}(\rho_{\!\sigma}) \; \mathbf{g}_{\mathbf{x},\sigma}(q_{\!\sigma}) \; \mathrm{d}\mathbf{r} \tag{24} \label{eq:24}$$

and the opposite- and same-spin components of the correlation energy take the forms

$$E_{c,\alpha\beta} = \int e_{c,\alpha\beta}^{\text{UEG}}(\rho_{\alpha}, \rho_{\beta}) \, g_{c,\alpha\beta}(q_{\text{avg}}) \, d\mathbf{r}$$
(25)

$$E_{c,\sigma\sigma} = \int e_{c,\sigma\sigma}^{\text{UEG}}(\rho_{\sigma}) f_{\sigma}^{\text{SCC}} g_{c,\sigma\sigma}(q_{\sigma}) \, d\mathbf{r}$$
(26)

The functions  $g_{x,\sigma}(q_{\sigma})$ ,  $g_{c,\alpha\beta}(q_{avg})$ , and  $g_{c,\sigma\sigma}(q_{\sigma})$  are dimensionless inhomogeneity correction factors depending on  $q_{\sigma}$  and  $q_{avg} = (1/2)(q_{\alpha} + q_{\beta})$ . In addition, the same-spin correlation energy contains a self-correlation correction (SCC) factor

$$f_{\sigma}^{\text{SCC}} = \left[ \tau_{\sigma} - \frac{1}{4} \frac{(\nabla \rho_{\sigma})^2}{\rho_{\sigma}} \right] / \tau_{\sigma}$$
(27)

This factor varies between 0 and 1 and vanishes in one-orbital regions, ensuring the functional is self-correlation-free. For convenience in deriving fitted forms for the inhomogeneity correction factors *g*, Becke proposed the following transformation to a finite interval

$$w = \frac{\gamma q}{\sqrt{1 + \gamma^2 q^2}} \tag{28}$$

where  $\gamma$  is a parameter to be determined, and separate transformations are carried out for the exchange, opposite-spin correlation, and like-spin correlation, respectively, using the appropriate definitions of q as in eqs 24 and 25. On the basis of calculations of atomic exchange-correlation energies, Becke proposed the values  $\gamma_{x,\sigma} = 0.11$ ,  $\gamma_{c,\alpha\beta} = 0.14$ , and  $\gamma_{c,\sigma\sigma} = 0.16$ . The inhomogeneity corrections g were then determined by fitting a power series expansion of the form

$$g = \sum_{i=0}^{m} c_i w^i \tag{29}$$

with m = 2 chosen to prevent unphysical overfitting. This fitting was carried out using the G2 thermochemical data set and basis set-free, post-LSDA calculations. The optimal parameters can be found in ref 29.

Here, we use this parametrization directly but note that in future these parameters could be reoptimized based on self-consistent data. In addition, an amount of Hartree–Fock exchange is included with weight  $c_{\rm x}=0.1985$ . The resulting B98 functional, therefore, possesses a high degree of nonlocality and may be classified as a hybrid mGGA functional with dependence not only on  $\tau$  but also on the laplacian of the density  $\nabla^2 \rho$ .

The original definition of B98 utilized the zero-field exchange hole curvature of eq 9 in its definition. However, it was noted that this form can be readily extended to include current effects via eq 10, and it is this avenue that we explore in the present work. Unless otherwise stated, we employ this modified form throughout and denote it as cB98.

A.2. cTPSS. One of the most widely used meta-GGA functionals is due to Tao, Perdew, Staroverov, and Scuseria (TPSS).<sup>28</sup> This functional is designed to satisfy exact constraints without empirical parameters and as such is an interesting candidate to study in the context of generalization to finite magnetic field strengths where much less is known about

the performance of approximate functionals. In this functional, the ratio

$$z = 2\tau^{W}/\tau, \quad \tau = \sum_{\sigma} \tau_{\sigma}, \quad \tau^{W} = \frac{1}{8} \frac{|\nabla \rho|^{2}}{\rho}$$
 (30)

plays a key role as a dimensionless inhomogeneity parameter, along with

$$p = \frac{|\nabla \rho|^2}{4(3\pi^2)^{2/3}\rho^{8/3}} = s^2 \tag{31}$$

Note that throughout this work we use the definition of  $\tau$  in eq 8, which does not include the factor of  $^{1}/_{2}$  commonly employed. The exchange functional then takes the form

$$E_{\rm x}[\rho] = \int \! \rho \epsilon_{\rm x}^{\rm UEG}(\rho) \, F_{\rm x}(p,z) \tag{32}$$

where the precise details of the form chosen for the enhancement factor  $F_x(p, z)$  can be found in ref 28. The correlation energy takes the form

$$\begin{split} E_{\rm c}[\rho_{\alpha},\,\rho_{\beta},\,\nabla\rho_{\alpha},\,\nabla\rho_{\beta},\,\tau] &= \int\!\!\rho\varepsilon_{\rm c}^{\rm revPKZB}(\rho_{\alpha},\,\rho_{\beta},\,\nabla\rho_{\alpha},\,\nabla\rho_{\beta},\,\tau) \\ &\times \left[1 + d\varepsilon_{\rm c}^{\rm revPKZB}(\rho_{\alpha},\,\rho_{\beta},\,\nabla\rho_{\alpha},\,\nabla\rho_{\beta},\,\tau)(\tau^{\rm W}/\tau)^{3}\right]{\rm d}\mathbf{r} \end{split} \tag{33}$$

where  $d = 2.8 \text{ hartree}^{-1}$ .

Using the replacement in eq 12 leads to modifications in the exchange contribution via z in eq 30 and in the correlation energy as shown in eq 33. This modified form is noted cTPSS and is consistent with that used in the response implementation of ref 13.

## AUTHOR INFORMATION

#### **Corresponding Authors**

\*(J.W.F.) E-mail: pcxjf1@nottingham.ac.uk.

\*(A.M.T.) E-mail: andrew.teale@nottingham.ac.uk.

#### **Funding**

This work was supported by the Norwegian Research Council through the CoE Centre for Theoretical and Computational Chemistry (CTCC) grant nos. 179568/V30 and 171185/V30 and through the European Research Council under the European Union Seventh Framework Program through the Advanced Grant ABACUS, ERC grant agreement no. 267683.

#### Notes

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

A.M.T. is grateful for support from the Royal Society University Research Fellowship scheme. We are grateful for access to the University of Nottingham High Performance Computing Facility.

#### **■** REFERENCES

- (1) Vignale, G.; Rasolt, M. Phys. Rev. Lett. 1987, 59, 2360-2363.
- (2) Vignale, G.; Rasolt, M. Phys. Rev. B: Condens. Matter Mater. Phys. 1988, 37, 10685-10696.
- (3) Vignale, G.; Rasolt, M.; Geldart, D. J. W. Phys. Rev. B: Condens. Matter Mater. Phys. 1988, 37, 2502.
- (4) Pan, X.; Sahni, V. Int. J. Quantum Chem. 2010, 110, 2833-2843.
- (5) Pan, X.-Y.; Sahni, V. J. Phys. Chem. Solids 2012, 73, 630-634.
- (6) Pan, X.-Y.; Sahni, V. Int. J. Quantum Chem. 2013, 113, 1424–1425.

- (7) Tellgren, E. I.; Kvaal, S.; Sagvolden, E.; Ekström, U.; Teale, A. M.; Helgaker, T. Phys. Rev. A: At., Mol., Opt. Phys. 2012, 86, 062506.
- (8) Vignale, G.; Ullrich, C. a.; Capelle, K. Int. J. Quantum Chem. **2013**, 113, 1422–1423.
- (9) Lieb, E. H. Int. J. Quantum Chem. 1983, 24, 243-277.
- (10) Lieb, E. H.; Schrader, R. Phys. Rev. A: At., Mol., Opt. Phys. 2013, 88, 032516.
- (11) Tellgren, E. I.; Kvaal, S.; Helgaker, T. Phys. Rev. A: At., Mol., Opt. Phys. 2014, 89, 012515.
- (12) Lee, A. M.; Handy, N. C.; Colwell, S. M. J. Chem. Phys. 1995, 103, 10095-10109.
- (13) Bates, J. E.; Furche, F. J. Chem. Phys. 2012, 137, 164105—164115.
- (14) Vignale, G. Phys. Rev. B: Condens. Matter Mater. Phys. 1993, 47, 10105-10111.
- (15) Ferconi, M.; Vignale, G. Phys. Rev. B: Condens. Matter Mater. Phys. 1994, 50, 14722-14725.
- (16) Pittalis, S.; Kurth, S.; Helbig, N.; Gross, E. K. U. *Phys. Rev. A: At., Mol., Opt. Phys.* **2006**, 74, 062511.
- (17) Pittalis, S.; Kurth, S.; Sharma, S.; Gross, E. K. U. J. Chem. Phys. **2007**, 127, 124103–124107.
- (18) Zhu, W.; Zhang, L.; Trickey, S. B. Phys. Rev. A: At., Mol., Opt. Phys. 2014, 90, 022504.
- (19) Tellgren, E. I.; Teale, A. M.; Furness, J. W.; Lange, K. K.; Ekström, U.; Helgaker, T. J. Chem. Phys. **2014**, 140, 034101–034112.
- (20) Tellgren, E. I.; Soncini, A.; Helgaker, T. J. Chem. Phys. 2008, 129, 154114–154124.
- (21) London, F. J. Phys. Radium 1937, 8, 397-409.
- (22) Ditchfield, R. Mol. Phys. 1974, 27, 789-807.
- (23) Ditchfield, R. J. Chem. Phys. 1976, 65, 3123-3133.
- (24) Maximoff, S. N.; Scuseria, G. E. Chem. Phys. Lett. 2004, 390, 408–412.
- (25) Tao, J. Phys. Rev. B: Condens. Matter Mater. Phys. 2005, 71, 205107.
- (26) Dobson, J. F. J. Chem. Phys. 1993, 98, 8870-8872.
- (27) Becke, A. D. Can. J. Chem. 1996, 74, 995-997.
- (28) Tao, J.; Perdew, J.; Staroverov, V.; Scuseria, G. E. Phys. Rev. Lett. **2003**, *91*, 146401.
- (29) Becke, A. D. J. Chem. Phys. 1998, 109, 2092-2098.
- (30) Staroverov, V. N.; Scuseria, G. E.; Tao, J.; Perdew, J. P. J. Chem. Phys. **2003**, 119, 12129–12137.
- (31) Stopkowicz, S.; Gauss, J.; Lange, K. K.; Tellgren, E. I.; Helgaker, T. J. Chem. Phys. **2015**, DOI: 10.1063/1.4928056.
- (32) Orestes, E.; Marcasso, T.; Capelle, K. Phys. Rev. A: At., Mol., Opt. Phys. 2003, 68, 022105.
- (33) Tao, J.; Perdew, J. Phys. Rev. Lett. 2005, 95, 196403.
- (34) Tao, J.; Vignale, G. Phys. Rev. B: Condens. Matter Mater. Phys. 2006, 74, 193108.
- (35) Higuchi, K.; Miyasita, M.; Kodera, M.; Higuchi, M. J. Magn. Magn. Mater. **2007**, 310, 1065–1066.
- (36) Dobson, J. F. J. Phys.: Condens. Matter 1992, 4, 7877-7890.
- (37) Becke, A. D. Int. J. Quantum Chem. 1983, 23, 1915-1922.
- (38) Becke, A. D.; Roussel, M. R. Phys. Rev. A: At., Mol., Opt. Phys. **1989**, 39, 3761–3767.
- (39) Dunning, T. H. J. Chem. Phys. 1989, 90, 1007-1023.
- (40) Kendall, R. A.; Dunning, T. H., Jr.; Harrison, R. J. J. Chem. Phys. **1992**, *96*, *6796*–*6806*.
- (41) Ekström, U.; Visscher, L.; Bast, R.; Thorvaldsen, A. J.; Ruud, K. J. Chem. Theory Comput. 2010, 6, 1971–1980.
- (42) Staroverov, V. N.; Scuseria, G. E.; Tao, J.; Perdew, J. P. J. Chem. Phys. **2003**, 119, 12129–12137.
- (43) Perdew, J.; Burke, K.; Ernzerhof, M. Phys. Rev. Lett. 1996, 77, 3865-3868.
- (44) Keal, T. W.; Tozer, D. J. J. Chem. Phys. 2004, 121, 5654-5660.
- (45) Teale, A. M.; Lutnæ s, O. B.; Helgaker, T.; Tozer, D. J.; Gauss, J. J. Chem. Phys. **2013**, 138, 024111–024132.
- (46) Lange, K. K.; Tellgren, E. I.; Hoffmann, M. R.; Helgaker, T. Science (Washington, DC, U. S.) 2012, 337, 327-331.

- (47) Lutnaes, O. B.; Teale, A. M.; Helgaker, T.; Tozer, D. J.; Ruud, K.; Gauss, J. J. Chem. Phys. **2009**, 131, 144104—144120.
- (48) Reimann, S.; Ekström, U.; Stopkowicz, S.; Teale, A. M.; Borgoo, A.; Helgaker, T. *Phys. Chem. Chem. Phys.* **2015**, *17*, 18834–18842.
- (49) Murdin, B. N.; Li, J.; Pang, M. L. Y.; Bowyer, E. T.; Litvinenko, K. L.; Clowes, S. K.; Engelkamp, H.; Pidgeon, C. R.; Galbraith, I.; Abrosimov, N. V.; Riemann, H.; Pavlov, S. G.; Hübers, H.-W.; Murdin, P. G. Nat. Commun. 2013, 4, 1469–1477.
- (50) Litvinenko, K. L.; Pang, M.; Li, J.; Bowyer, E.; Engelkamp, H.; Shuman, V. B.; Portsel, L. M.; Lodygin, a. N.; Astrov, Y. a.; Pavlov, S. G.; Hübers, H.-W.; Pidgeon, C. R.; Murdin, B. N. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2014**, *90*, 115204.
- (51) Perdew, J. P.; Parr, R. G.; Levy, M.; Balduz, J. L. Phys. Rev. Lett. 1982, 49, 1691–1694.
- (52) Almbladh, C. O.; von Barth, U. Phys. Rev. B: Condens. Matter Mater. Phys. 1985, 31, 3231-3244.
- (53) Bartlett, R. J.; Grabowski, I.; Hirata, S.; Ivanov, S. J. Chem. Phys. **2005**, 122, 034104–034116.
- (54) Baerends, E. J.; Gritsenko, O. V. J. Phys. Chem. A 1997, 101, 5383-5403.
- (55) Stowasser, R.; Hoffmann, R. J. Am. Chem. Soc. 1999, 121, 3414-3420
- (56) van Meer, R.; Gritsenko, O. V.; Baerends, E. J. J. Chem. Theory Comput. **2014**, 10, 4432–4441.
- (57) Goll, E.; Ernst, M.; Moegle-Hofacker, F.; Stoll, H. *J. Chem. Phys.* **2009**, *130*, 234112–234119.
- (58) Peverati, R.; Truhlar, D. G. J. Phys. Chem. Lett. 2011, 2, 2810–2817.
- (59) Becke, A. D. J. Chem. Phys. 1997, 107, 8554-8560.