

Nuclear phenomenology

Nuclides

Nuclides are typically written as:

$${}^A_Z Y$$

where we note that:

A is the **mass/nucleon number** (# of nucleons)
 Z is the **proton/atomic number** (# of protons)
 N is the **neutron number** (# of neutrons)

with $\mathbf{A} = \mathbf{Z} + \mathbf{N}$. We also note that isotopes with: same A = isobars; same Z = isotopes; same N = isotones. Some elements have multiple isotopes, with different stability and abundance.

Nuclear shapes and sizes

Nuclei may be treated as static charge distributions with normalization:

$$\int f(\mathbf{r}) d^3\mathbf{r} = Ze \tag{1}$$

where e is the electron charge. Under the Born approximation, the cross-section $\frac{d\sigma}{d\Omega}$ is given by:

$$\left(\frac{d\sigma}{d\Omega}\right)_0 = \frac{Z^2 \alpha^2 (\hbar c)^2}{4\beta^4 E^2 \sin^4(\theta/2)} \tag{2}$$

including the electron spin, the Mott cross-section is given by:

$$\left(\frac{d\sigma}{d\Omega}\right)_{\text{Mott}} = \left(\frac{d\sigma}{d\Omega}\right)_0 [1 - \beta^2 \sin^2(\theta/2)] \tag{3}$$

In the nonrelativistic limit with no spin dependence, the Rutherford cross-section is given by:

$$\left(\frac{d\sigma}{d\Omega}\right)_{\text{Rutherford}} = \frac{(\hbar c)^2 (\alpha Z)^2}{4m^2 v^4 \sin^4(\theta/2)} \tag{4}$$

Given $\mathbf{q} = \mathbf{p} - \mathbf{p}'$ where \mathbf{p} and \mathbf{p}' are initial and final electron momenta, the form factor $F(\mathbf{q}^2)$ (Fourier transform of charge distribution) is given by:

$$F(\mathbf{q}^2) = \frac{1}{Ze} \int e^{i\mathbf{q}\cdot\mathbf{r}/\hbar} f(\mathbf{r}) d^3\mathbf{r} \quad (5)$$

$$F(\mathbf{q}^2) = \frac{4\pi\hbar}{Ze q} \int_0^\infty r \rho(r) \sin\left(\frac{qr}{\hbar}\right) dr \quad (6)$$

$$F(\mathbf{q}^2) = \frac{4\pi}{Ze} \int_0^\infty f(r) r^2 dr - \frac{4\pi\mathbf{q}^2}{6Ze\hbar^2} \int_0^\infty f(r) r^4 dr + \dots \quad (7)$$

$$(8)$$

Experimental cross-section may be approximately related to the Mott cross-section by:

$$\left(\frac{d\sigma}{d\Omega}\right)_{\text{expt}} = \left(\frac{d\sigma}{d\Omega}\right)_{\text{Mott}} |F(\mathbf{q}^2)| \quad (9)$$

Representing the nuclear charge distribution as a hard sphere where $\rho(r)$ is constant for $r \leq a$ and 0 otherwise, then the form factor simplifies to $F(\mathbf{q}^2) = 3[\sin(b) - b \cos(b)]b^{-3}$.

The charge distribution may be obtained from the form factor using:

$$f(\mathbf{r}) = \frac{Ze}{(2\pi)^3} \int F(\mathbf{q}^2) e^{-i\mathbf{q}\cdot\mathbf{r}/\hbar} d^3\mathbf{q} \quad (10)$$

which can be simplified into:

$$f(r) = \rho_{\text{ch}}(r) = \frac{\rho_{\text{ch}}^0}{1 + e^{(r-a)/b}} \quad (11)$$

where $a \approx 1.07A^{1/3}$ fm and $b \approx 0.54$ fm.

Another important quantity would be the mean square charge radius:

$$\langle r^2 \rangle = \frac{1}{Ze} \int r^2 f(\mathbf{r}) d^3\mathbf{r} \quad (12)$$

$$\langle r^2 \rangle = -6\hbar^2 \left. \frac{dF(\mathbf{q}^2)}{d\mathbf{q}^2} \right|_{\mathbf{q}^2=0} \quad (13)$$

for very small values of \mathbf{q}^2 , for medium and heavy nuclei, $\langle r^2 \rangle^{1/2} = 0.94A^{1/3}$ fm.

Semi-empirical mass formula (SEMF)

α emissions

Radioactive decay