

COMPULSORY SCIENCE

PHYSICS

FORCE

1. $F = G \frac{m_1 m_2}{d^2}$

Where, F = (Gravitational force)

G = (Gravitational constant)

m_1 = (Mass of one object)

m_2 = (Mass of another object)

d = (Distance between two objects from their centers)

2. $g = \frac{GM}{R^2}$ $g = \frac{GM}{(R+h)^2}$

Where, g = (Acceleration due to gravity)

G = (Gravitational constant)

M (Mass of a planet or satellite)

R (Radius of the planet or satellite)

h = (Height from the surface of the earth/planet)

3. $I = \frac{GM}{R^2}$ $I = \frac{GM}{(R+h)^2}$

Where, I = (Gravitational field intensity)

h = (Height from the surface of the earth/planet)

4. $W = m \times g$

Where, W = (Weight of an object)

m = (Mass of an object)

g = (Acceleration)

5. $W = \text{mass} \times \text{gravitational field intensity}$

6. $g' = \left(\frac{R}{R+h}\right)^2$

Where, g' = (Value of acceleration due to gravity at a certain height (h))(from the earth's surface)

g = (Value of ' g ' on the surface of the earth/planet)

R=(Radius of the earth/planet)

h = (Height of the earth/planet)

$$7. \quad h = ut + \frac{1}{2}gt^2$$

Where, h = (Height from the earth's surface)

u= (Initial velocity)

t = (Time taken)

g = (Acceleration due to gravity)

S.I. UNITS

1. (Force) = N (Newton)

2. (Gravitational Force) = N

3. (Acceleration due to gravity) = m/s² OR m/s.s OR ms⁻²

4. (Gravitational Constant) = Nm²/kg² OR Nm²/kg⁻²

5. (Gravitational field intensity) =N/kg OR Nkg⁻¹

6. (Mass) = kg

7. (Weight)= N

PRESSURE

$$P = \frac{F}{A} \quad \text{Pressure} = \frac{\text{Thrust}}{\text{Area}}$$

$$P = h d g$$

Where, P=Pressure

h=(Height of the liquid column)

d= (Density of the liquid)

g=(Acceleration due to gravity)

$$3) \quad \frac{F_2}{F_1} = \frac{A_2}{A_1}$$

Where, F_1 = (Force applied on, small piston)

F_2 = (Force exerted on large piston)

A_1 = (Cross-sectional area of small piston)

A_2 = (Cross-sectional area of large piston)

$$4) \quad U = A d g (h_2 - h_1)$$

$$\text{Upthrust} = A d g (h_2 - h_1)$$

Where, A= (Cross-sectional area of the immersed body)

d = (Density of the liquid)

g = (Acceleration due to gravity)

h_1 = (Height of the liquid column above the upper surface of the immersed body)

h_2 = (Height of the liquid column above the lower surface of the immersed body)

5) Upthrust = Vdg

Where,

V = Volume of the immersed body or volume of the liquid displaced.

6) Upthrust = Weight of the liquid displaced (mg)

7) Density = $\frac{\text{Mass (m)}}{\text{Volume (v)}}$

8) R. D of substance = $\frac{\text{Mass of any volume of the substance}}{\text{Mass of an equal volume of water at } 4^\circ \text{C}}$

S.I Units

Pressure = N/m^2 or Nm^{-2} or Pa

Upthrust: N

Volume: m^3 (cubic metre)

Density: kg/m^3 or kgm^{-3}

Area: m^2 (square metre)

HEAT

1) $Q = m.s. dt$

Where, Q (Amount of heat)

m = (Mass of an object)

s = (Specific heat capacity)

dt = (Change in temperature)

2. $m_1 s_1 t_1 = m_2 s_2 t_2$ OR $m_1 s_1 (t_1 - t) = m_2 s_2 (t - t_2)$

Where, m_1 = (Mass of hot body)

s_1 = (Specific heat capacity of hot body)

t_1 = (Temperature of hot body)

m_2 = (Mass of cold body)

s_2 = (Specific heat capacity of cold body)

t_2 = (Temperature of cold body)

t = (Final temperature of the mixture)

S.I. UNITS

- 1) Heat : Joule(J)
- 2) Temperature: (K)
- 3) Specific heat capacity: (J/kg°C OR J (kg°C)-1)
- 4) Heat capacity: J/°C OR J°C⁻¹
- 5) Mass: (kg)

LIGHT

$$1) \text{Power of a lens (P)} = \frac{1}{\text{focal length (f) of the lens (in metre)}}$$

$$2) \frac{1}{f} = \frac{1}{u} + \frac{1}{v}$$

Where, f = Focal length of a lens)

u = (Object distance)

v = (Image distance)

$$3) \text{Magnification (m)} = \frac{\text{height of image } h_2}{\text{height of object } h_1}$$

$$4) \text{Magnification (m)} = \frac{\text{image distance}}{\text{object distance}}$$

S.I UNITS

1) Focal length: m (metre)

Power of a lens: D (diopetre)

Magnification: No unit

CURRENT ELECTRICITY AND MAGNETISM

$$\text{Current} = \frac{\text{Electric power (P)}}{\text{Potential difference (V)}}$$

$$\text{Electric power Consumption} = P \times N \times t$$

Where,

P= Electric power consumed (in kilowatt)

N= Number of appliances used

t = time in hour

$$\frac{V_2}{V_1} = \frac{n_2}{n_1}$$

Were,

V_2 = Secondary Voltage

V_1 = Primary Voltage

n_2 = Number of turns in secondary coil

n_1 = Number of turns in primary coil

S.I. UNITS

Current = ampere (A)

Potential Difference = volt (V)

Resistance = ohm (Ω)

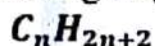
Electric power consumption = kWh (kilowatt hour)

CHEMISTRY

CARBON AND ITS COMPOUNDS

Saturated hydrocarbon:

Alkane (ane' at last and compounds having single bond)



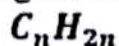
Were,

H= Hydrogen

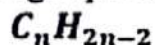
n= number of carbon atoms

Unsaturated hydrocarbon:

Alkyne ('ene' at last and compounds having double bond)



Alkyne ('yne' at last and compounds having triple bond)



Alcohol: $C_nH_{2n+1} + OH$

Alkyl Ether: $C_nH_{2n+1} + OC_nH_{2n+1}$

यौगिक (Compound)	अणुसूत्र (Molecular Formulae)	संरचनात्मक सूत्र (Structural Formulae)
मिथेन (Methane)	CH_4	$\begin{array}{c} \text{H} \\ \\ \text{H} - \text{C} - \text{H} \\ \\ \text{H} \end{array}$
इथेन (Ethane)	C_2H_6	$\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H} - \text{C} - \text{C} - \text{H} \\ \quad \\ \text{H} \quad \text{H} \end{array}$
प्रोपेन (Propane)	C_3H_8	$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \\ \quad \quad \\ \text{H} - \text{C} - \text{C} - \text{C} - \text{H} \\ \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \end{array}$
ब्यूटेन (Butane)	C_4H_{10}	$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \\ \quad \quad \quad \\ \text{H} - \text{C} - \text{C} - \text{C} - \text{C} - \text{H} \\ \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \end{array}$
पेन्टेन (Pentane)	C_5H_{12}	$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \\ \quad \quad \quad \quad \\ \text{H} - \text{C} - \text{C} - \text{C} - \text{C} - \text{C} - \text{H} \\ \quad \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \end{array}$

इथाइलिन (इथिन) (Ethylene or Ethene)	C_2H_4	$\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{C} = \text{C} \\ \quad \\ \text{H} \quad \text{H} \end{array}$
एसिटायलिन (Acetylene)	C_2H_2	$\text{H} - \text{C} \equiv \text{C} - \text{H}$
ग्लिसरोल (Glycerol)	$\text{C}_3\text{H}_8(\text{OH})_3$	$\begin{array}{c} \text{OH} \quad \text{OH} \quad \text{OH} \\ \quad \quad \\ \text{H} - \text{C} - \text{C} - \text{C} - \text{H} \\ \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \end{array}$
इथाईल अल्कोहल (Ethyl alcohol)	$\text{C}_2\text{H}_5\text{OH}$	$\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H} - \text{C} - \text{C} - \text{OH} \\ \quad \\ \text{H} \quad \text{H} \end{array}$
डाई इथाईल इथर (Diethyl Ether)	$\text{C}_2\text{H}_5 - \text{O} - \text{C}_2\text{H}_5$	$\begin{array}{c} \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \\ \quad \quad \quad \\ \text{H} - \text{C} - \text{C} - \text{O} - \text{C} - \text{C} - \text{H} \\ \quad \quad \quad \\ \text{H} \quad \text{H} \quad \text{H} \quad \text{H} \end{array}$

"The meeting of two personalities is like the contact of two chemical substances: if there is any reaction, both are transformed."

~ Carl Jung