

## APPROXIMATIONS

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ABSTRACT. Only wimps specialize on the general case. Real scientists pursue examples. – Beresford Parlett [1].

### 1. PRESSURE

Consider a piston of volume  $v$  and sectional area  $a$  holding an amount  $n$  of an ideal gas and consider a force  $f$  being applied on  $a$  that compresses the fluid.

Such a force generates a pressure  $p$  on the gas. See now that  $p$  can be viewed as energy density instead of force per unit area. With this view, one finds the external energy applied to the system  $e_{\text{ext}}$  by

$$e_{\text{ext}} \sim pv.$$

This external stimulus makes the gas to perform  $pv$  work, the gas internal response  $e_{\text{int}}$ , given by

$$e_{\text{int}} \sim nk_{\text{gas}}\theta,$$

where  $k_{\text{gas}}$  represents the gas constant and  $\theta$  the gas temperature.

Thus, according to the energy conservation principle, the external stimulus must be balanced by the gas internal response:

$$e_{\text{ext}} \sim e_{\text{int}} \sim nk_{\text{gas}}\theta \implies k_{\text{gas}}\theta \sim \frac{e_{\text{ext}}}{n}.$$

That is,  $k_{\text{gas}}\theta$  is a measure of the external energy distributed per amount of gas – molar energy.

On the other hand, since by definition an ideal gas does not interact, its total internal energy equals its kinetic energy alone:

$$e_{\text{int}} \sim mu^2,$$

where  $m$  represents the gas mass and  $u$  the average velocity of the gas particles. Thus, one finds

$$mu^2 \sim nk_{\text{gas}}\theta \implies \theta \sim \frac{mu^2}{nk_{\text{gas}}} \propto u.$$

Hence, temperature can also be viewed as a measure of the mean particle velocity of the gas particles.

### 2. ENERGY

Consider a large, thin concrete slab of thickness  $l$  that is *setting*. Setting is an exothermic process that releases  $e_{\text{th}}$ , where  $\dim e_{\text{th}} = [\text{E}/\text{TL}^3]$  – thermal power density. The outside surfaces are kept at the ambient temperature, so the temperature of the walls,  $\theta_w$ , equal the ambient temperature:  $\theta_w = \theta_\infty$ . What is the maximum internal temperature?

*Guess.* Since the walls are kept at constant temperature, the process is at steady state. However, temperature ranges spatially through the slab thickness. If one measures the spatial variation by  $x$ , then the slab temperature satisfies  $\theta = \theta[x]$ .

By symmetry, the center temperature coincides with  $\theta_{\text{max}}$  at the slab center,  $x = 1/2$ , and decreases smoothly to a minimum at the walls,  $x = 0$  and  $x = l$ . This symmetry gives room to think about an inverted parabolic temperature distribution inside the slab with the parabola vertex at  $\theta_{\text{max}}$ .

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| Quantity                            | Symbol     | Dimension |
|-------------------------------------|------------|-----------|
| Slab temperature                    | $\theta$   | $\Theta$  |
| Slab thermal conduction coefficient | $k$        | $\Theta$  |
| Slab thickness                      | $l$        | $L$       |
| Wall temperature                    | $\theta_w$ | $\Theta$  |
| Setting power density               | $e_{th}$   | $E/TL^3$  |
| Position within the slab            | $x$        | $L$       |

TABLE 1. Quantities and dimensions affecting the thermal conduction of the concrete slab setting.

*Dim. Analysis.* Place a Cartesian coordinate axis running from one wall to the other covering the slab thickness. Let  $x$  measure position within  $0 \leq xl$ . Thus, since the process is at steady state,  $\theta = \theta[x]$ .

Choose the dimensional set to be  $\{E, L, T, \Theta\}$ . Hypothesize the quantities governing the phenomenon to be those listed in table 1.

As seen in table 1, according to the Pi-theorem,  $6 - 4 = 2$  dimensionless quantities can be constructed. The first one:

$$\Pi_1 = \frac{k(\theta - \theta_w)}{e_{th}l^2},$$

which measures the relationship between energy conduction and energy production. The second dimensionless quantity:

$$\Pi_2 = \frac{x}{l},$$

which is a geometric ratio.

With both dimensionless quantities, one can apply the principle of dimensional homogeneity for physical laws to find

$$\Pi_1 = \phi_\pi[\Pi_2] \implies \frac{k(\theta - \theta_w)}{e_{th}l^2} = \phi_\pi\left[\frac{x}{l}\right].$$

Scale temperatures by means of  $\Pi_1$  and lengths by  $\Pi_2$ ; *i.e.*,

$$\bar{\theta} = \Pi_1\theta \quad \text{and} \quad \bar{x} = \Pi_2x.$$

Hence, finally, the equation governing the phenomenon can be written as

$$\bar{\theta} = \phi_\pi[\bar{x}]. \quad (1)$$

where the function  $\phi_\pi$  cannot be further determined by dimensional analysis.

*Approx. Solution.* Assuming a parabolic distribution of temperatures,  $\phi_\pi$  in eq. (1) can be hypothesize to satisfy

$$\bar{\theta} = a\bar{x}^2 + b\bar{x} + c,$$

where  $\{a, b, c\}$  are dimensionless quantities to be determined.

Now, we can use a theorem in geometry that states that three points uniquely determine a parabola. Two of these points can be found from the problem statement:

$$\begin{cases} \left[ \bar{x} = 0, \bar{\theta} = 0 \right] \\ \left[ \bar{x} = 1, \bar{\theta} = 0 \right] \end{cases}.$$

Setting  $a = -1/2$  (an inverted parabola) and solving the previous systems of equations, one finds that

$$a = -\frac{1}{2}, \quad b = \frac{1}{2} \quad \text{and} \quad c = 0.$$

Replacing these values in the hypothesized  $\phi_\pi$ , one has

$$2\bar{\theta} = \bar{x}(1 - \bar{x}).$$

Now, using symmetry, when  $\bar{x} = 1/2$ , then  $\bar{\theta} = \bar{\theta}_{\max}$ :

$$\bar{\theta}_{\max} = \frac{1}{8}$$

or, returning to the dimensional quantities,  $\theta_{\max}$  can be found by

$$\frac{k(\theta_{\max} - \theta_w)}{e_{th}l^2} = \frac{1}{8}.$$

□

## REFERENCES

- [1] Michael Berry. Two-state quantum asymptotics. *Annals of the New York Academy of Sciences*, 755, April 1995.