**Bayesian Docking-Score Predictor**

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**Goal**

According to Dr. Bill Basener, “the goal of the project is for you to apply Bayesian machine learning to a real dataset in an advanced way. This means that the project should show you can apply probabilistic reasoning to a nontrivial problem of your choosing.”

Naomi Ohashi and Tom Lever have developed Bayesian Docking-Score Predictor. Bayesian Docking-Score Predictor receives the Simplified Molecular Input Line Entry System (SMILES) of a ligand and provides a docking score of how well that ligand binds to a protein.

Our predictor receives the SMILES of a ligand. A SMILES is a string representing the chemical structure of a ligand. A ligand is a molecule that binds to another molecule that is usually larger. There is one SMILES per ligand. There is one ligand per SMILES.

A docking score is a measure of how well a ligand binds to another molecule. A docking score is a change in the molar Gibbs free energy of a compound. A docking score is measured in kilocalories per mole. According to Dr. Ryan Weil, is amazing and is okay. See below *Appendix 1: Exploring Change In Molar Gibbs Free Energy*.

**Data**

Our data set is *Data\_Frame\_Of\_SMILESs\_Docking\_Scores\_And\_Other\_Data.csv* (DFSDSAOD). Our data set is associated with one protein and with one site of that protein at which ligands bind. Each row in our data set encapsulates data relating to a ligand docking to the protein. There is one ligand per row. There is one row per ligand. Each row has an Entry ID. There is one Entry ID per ligand. There is one ligand per Entry ID.

**Implementation**

A SMILES of a ligand is converted into a numerical representation of the chemical structure of the ligand. The numerical representation is a vector of the number of occurrences of certain common substructures of molecules. One substructure is a structure of carbon atoms. Each vector of numbers of occurrences of substructures is folded into a vector of 1024 numbers of occurrences of substructures.

We construct a feature matrix where each row contains a docking score and 1024 numbers of occurrences of substructures.

We train and predict with a *Bayesian Model Using A Bayesian Additive Regression Trees Model*. This model is implemented using Python packages PyMC and pymc\_bart. In this model, a pymc\_bart.BART model with trees is trained on a vector of docking scores and a matrix of numbers of occurrences of substructures and is used to define an expected value of response values . We define a prior probability density distribution for a standard deviation as a half-normal distribution itself with a standard deviation of . We define a likelihood as a normal distribution with mean and standard deviation . We use PyMC to sample values of and from the posterior probability density distribution and to sample values of from the posterior predictive probability density distribution . The number of values of sampled is the product of a number of chains (e.g., ), a number of samples per chain (e.g., ), and a number of testing observations (e.g., ). The number of chains is equal to the number of cores on PyMC’s host machine. We find vectors of averages and standard deviations of values of sampled. Each vector has length equal to the number of testing observations.

Interestingly, in the original paper *BART: Bayesian Additive Regression Trees*, “BART’s many features are illustrated with a bake-off against competing methods of different datasets, with a simulation experiment and on a drug discovery classification problem.” According to *Statistical Learning: 8.6 Bayesian Additive Regression Trees*, “It turns out that the BART method can be viewed as a Bayesian approach to fitting an ensemble of trees: each time we randomly perturb a tree in order to fit the residuals, we are in fact drawing a new tree from posterior distribution. Furthermore, the BART algorithm can be viewed as a Markov Chain Monte Carlo procedure for fitting the BART model. We typically choose large values for [number of iterations] and [number of trees per iteration] , and a moderate value for [number of burn-in iterations] : for instance, , , and are reasonable choices. BART has been shown to have impressive out-of-box performance – that is, it performs well with minimal tuning.”

**Applications**

A trained Bayesian Docking-Score Predictor could be used for developing a training data set of SMILES’s and docking scores of ligands for another predictor.

Bayesian Docking-Score Predictor could be trained on (e.g., ) pairs of SMILES and docking score and used to predict docking scores of ligands based on many (e.g., ) other SMILES’s. Rows of data representing docking of ligands could be sorted in ascending order by predicted docking score. The first rows of data corresponding to the lowest docking scores could be added to the training data set.

Bayesian Docking-Score Predictor may be used to calculate Bayesian predictions, estimate uncertainty, or in optimization of hyperparameters of Bayesian models.

**Appendix 1: Exploring Change In Molar Gibbs Free Energy**

The enthalpy of a compound is the sum of the compound’s internal energy and the pressure energy of the compound. The pressure energy is the energy required to establish the compound’s physical dimensions. The pressure energy is the product of the pressure on the compound by its surroundings and the volume of the compound.

The molar internal energy of a compound is the internal energy of the compound per mole.

The molar volume is the volume of the compound per mole.

The molar pressure energy of a compound is the pressure energy of the compound per mole.

The molar enthalpy of a compound is the enthalpy of the compound per mole. The molar enthalpy of a compound is the sum of the compound’s molar internal energy and molar pressure energy.

The change in enthalpy of a reaction is the difference between the sum of the enthalpies of the products of the reaction and the sum of the enthalpies of the reactants.

The change in molar enthalpy of a reaction is the difference between the sum of the molar enthalpies of the products and the sum of the molar enthalpies of the reactants.

The entropy of a compound is a measure of uncertainty, disorder, or mixedupness of the compound. The entropy measures the degree to which the probability of the compound being in a particular microstate is spread out over different microstates. A microstate specifies all molecular details about the system including the position and velocity of every molecule. The more such states are available to the compound with appreciable probability, the greater the entropy.

is the probability that the compound is in the th state according to the Boltzmann distribution.

is the number of microstates whose energy equals the compound’s energy.

For an isolated system, .

The entropy of a compound is a quantity that satisfies “an infinitesimal change in entropy is equal to the ratio of an infinitesimal quantity of heat in a reversible reaction and the temperature of the compound”.

The molar entropy of a compound is the entropy of the compound per mole.

The change in entropy of a reaction is the difference between the sum of the entropies of the products of the reaction and the sum of the entropies of the reactants.

The change in molar entropy of a reaction is the difference between the sum of the molar entropies of the products of the reaction and the sum of the molar entropies of the reactants.

The Gibbs free energy of a compound at some time is the difference between the enthalpy of the compound and the product of the temperature and entropy of the compound.

The molar Gibbs free energy is the difference between the molar enthalpy of the compound and the product of the temperature and the molar entropy of the compound.

The change in Gibbs free energy of a reaction is the maximum amount of free and useful energy available to do non-volume-expansion work that can be extracted from the reactants at fixed temperature and pressure, which can be attained only in a completely reversible process. The change in Gibbs free energy of a reaction is the difference between the change in enthalpy of the reaction and the change in the product of the temperature and entropy of the compound.

The change in molar Gibbs free energy of a compound is the difference between the change in molar enthalpy of the compound and the change in the product of the temperature and molar entropy of the compound.