## MAHARASHTRA STATE BOARD OF TECHNICAL EDUCATION

## **Summer-13 EXAMINATION**

## Model Answer

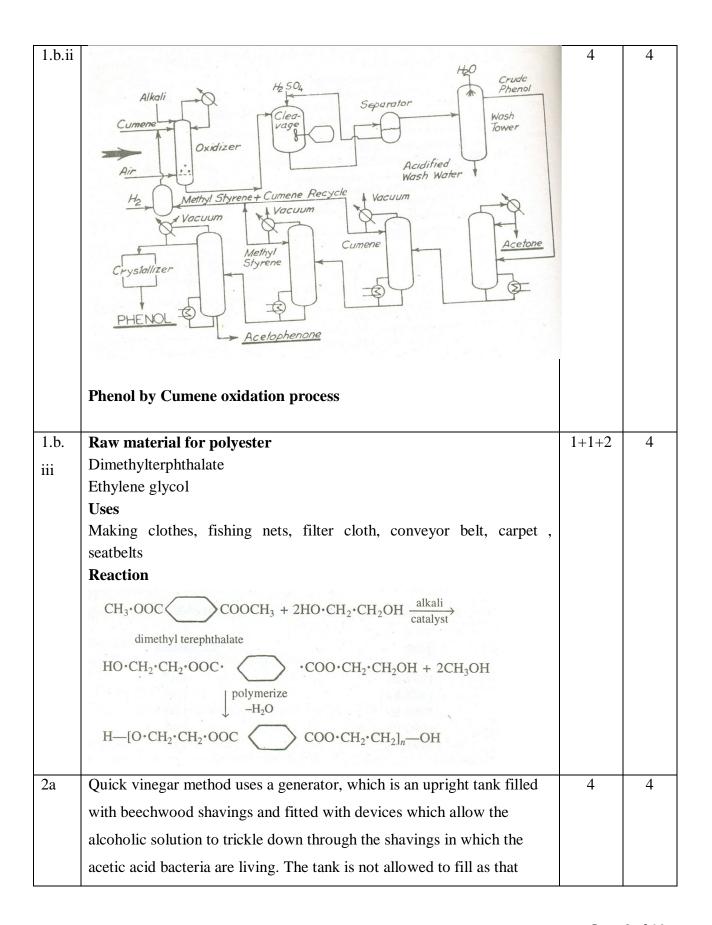
Subject & code:TOC (12127)

## Important instructions to examiners:

- 1. The answers should be examined by keywords and not as word to word as given in the model answer scheme.
- 2. The model answer and the answer written by candidate may vary, but the examiner may try to assess the understanding level of the candidate.
- 3. The language errors such as grammatical, spelling errors should not given more importance.
- 4. While assessing figures, examiner may give credit for principal components indicated in a figure. The figures drawn by candidate and model answer may vary. The examiner may give credit for any equivalent figure drawn.
- 5. Credits may be given step wise for numerical problems. In some cases, the assumed constant values may vary and there may be some difference in the candidate's answer and model answer.
- 6. In case of some questions credit may be given by judgment of relevant answer based on candidates understanding.

Q	Answe	er		mark	Total
no:					marks
1a.	S.No	Paint	Varnishes	Any	2
i	1.	Paint is the mechanical dispersion mixture of one or more pigments in a vehicle.	Varnish is a homogenous colloidal dispersion solution of resin in oils or thinner or both.	one differe nce	
	2.	A paint contains pigment.	Varnish do not Contain Pigments.		
	3.	Paint Produce an opaque film.	Varnish produce transparent film.		
	4.	In paints pigments are dispersed in drying oils.	In varnishes resins are dispersed in oils or spirits.		
ii.	1	s a lignocellulosic fibrous materia		2	2

	waste paper.		
iii.	Addition polymerization and Chain polymerisation	1+1	2
	Pigment, Solvent, thinner, Film former, Plasticizer, Pigment extender,	Any	2
iv	Antishrinking agent, Antifoaming agent	two	
V	High Prssure process, Low pressure process and Ziegler process	Any	2
		two	
vi	Soap is sodium or potassium salt of fatty acid. Raw material for soap is oil, caustic, perfume, fillers, brighteners	1+1	2
vii	Making clothes, fishing nets, filter cloth, conveyor belt, carpet,	Any	2
	seatbelts	two	
viii	Invertase and zymase	1+1	2
1 b i	Lacquer is a liquid covering consisting of a natural or synthetic derivative which is combined with a solvent that can be used to seal, cover or finish a surface. Once the solvent evaporates the lacquer forms a smooth clear (although coloring can be added) surface which is resistant to water and many liquids.  Constituents  Film forming agent- To form durable film	2+2	4
	Extender- to maintain viscosity		
	Solvent- to dissolve film forming material and as carrier		
	Plasticizer- To maintain plasticity of film		
	Pgment- To give colour		



would exclude oxygen which is necessary for the fermentation. Near the bottom of the generator are holes which allow air to be drawn in. the air rises through the generator and is used by the acetic acid bacteria to	
rises through the generator and is used by the acetic acid bacteria to	
oxidise the alcohol. This oxidization also releases considerable amounts	
of heat which must be controlled to avoid causing damage to the	
bacteria.	
2.b Cleansing action of soap 4	4
Soap ions consist of two parts that is the head that consists of the anion	
region, ionic and also called the hydrophilic region which dissolves in	
water. Another part is the tail that consists of hydrocarbon region and its	
molecule has covalent characteristics. It's also called the hydrophobic region which dissolves in grease or oil(dirt)	
The soap molecules will dissolves in water and reduces the surface	
tension of water. Water wets the dirty surface. The hydrophilic region	
dissolves in water whereas the hydrophobic region dissolves in dirt such	
as grease. Grease is lifted off the surface of the material and suspended	
in water. The tail region emulsifies and breaks up the grease into small	
drops.	
XXIven abeles also restor realizable will attract the second constitution	
When shaken, the water molecules will attract the soap ions and cause the grease to detach from the surface of the material. The soap bubbles	
help to float the grease emulsion in the water. When rinsed, the grease	
will be removed together with the water.	
we Water	
lonic end	
[polar and hydrophilic]	
Hydro carbon chain [non-polar and hydrophobic]	
Cleaning Action of a Soap	
2.c Benzene direct oxidation 4	4

	Benzene sulfonate caustic fusion		
	Chloro benzene caustic hydrolysis		
	Rashig		
	Toluene oxidation		
	Cumene paraoxidation		
2.d	-	1+1+2	4
2.0	<b>Mechanical:</b> Debarked wood is defibered by mechanical grinders. Pulp is of poor quality	1+1+2	4
	Semi chemical: Mild chemical treatment is given. It is easy to separate		
	cellulose from wood.		
	<b>Chemical:</b> Cellulose from wood is derived by using digestion liquor.		
	Hemicellulose and lignin is dissolved in liquor. Quality of pulp obtained		
	is high. After digestion chemicals are recovered from digested liquor to		
	avoid environment problem. Sulphate (Kraft) and sulphite are two		
	chemical processes		
2.e	Ethyl acetate	4	4
	Here Affection Affection Because Separations (column)  Separation Separations (column)  Fisher  Washe (Siepe)		
2.f	Solution polymerization is avoided due to:	4	4
	It reduces monomer concentration which results in decreasing		
	the rate of the reaction and the degree of polymerization.		
	Solvent may cause chain reactions		
	Difficulty in separation of polymer and solvent		
	Include cost of expensive separation		
3.a	Manufacturing of paper by Foundriner process	4	4

	Pulp With 99.5% H <sub>2</sub> O Water Pressure roll Suction roll  Steam heated drying roll Finished Paper range % H <sub>2</sub> O 90 80-82 60-65 5-6 5-6 5-6		
3.b	Catalyst is used in Ziegler process is di-ethyl aluminium chloride and titanium tetrachloride.	2	4
	The product from reactor is taken in flash drum, here water is added to destroy residual catalyst. The bottom product from flash drum isaq.  Slurry of polythene which is separated from aq. filterate. The overhead product of flash drum containing hydrocarbon solvent is trated in fractioner drier.	2	
3.c	Acid Value of oil: - It is the number of milligrams of KOH required to neutralize the free acid present in one gram of oil.	1	4
	Importance of acid value:i)Acid value is the measure of free fatty acid presentin an oil and a fat. Low value indicates freshness of the oil or fat.  ii) Acidity in animal or vegetable or vegetable oils increases by oxidation or by hydrolysis of the oil. It imparts bad odur to the oil.	1	
	<b>Iodine Value</b> : - It is a number of milligrams of iodine absorbed by 100 grams of oil for complete saturation.	1	
	Importance of Iodine value-:	1	
	i) Iodine value is the measure of unsaturation of oil or fat.		
	ii) It helps in classification of oils Thus,		
	1) An oil containing one double bond has iodine value < 90 ➤ Non – drying oil		

	2) An oil containing two double bonds has iodine value <140→		
	Semi-drying oil		
	3) An oil co containing three double bonds has iodine value >140 →		
	drying oil		
3.d	<b>Pigments: -</b> It finely divided solids generally made up metal oxides .It is used to give colour to paint. <b>Drying oil: -</b> These are unsaturated oils. It is used to form protective	1	4
	film and give gloss.	1	
	<b>Thinners or solvent</b> : - It is alcohols or turpentine. is used to dissolve polymers in paint and to disperse pigments (emulsion formation). It adjust viscosity, form thin film	1	
	<b>Plasticizer: -</b> These are polymers. Used to impart elasticity to paint.	1	
	oil		
3.e	Steam  Barometric Leg  Barometric Leg  Catalyst  Oil Slurry  Make-up Nickel Catalyst  Recycle Catalyst  Decolorization  Filter Aid Fullers Earth Carbon  Vanaspati or Partially Hydrogenated Oil	4	4

4.a	Chemical reaction: $[C_6H_7O_2(OH)_3]_n+nNaOH \longrightarrow [C_6H_7O_2(OH)_3NaOH]_n$ Cellulose alkali Alkali cellulose	4	4
	$[C_6H_7O_2(OH)_3NaOH]_n + nCS_2 \longrightarrow \begin{pmatrix} OC_6H_9O_4 \\ C=S \end{pmatrix} + nH_2O$		
	SNa n $C=S+ nH_2O + nH_2SO_4 \rightarrow [C_6H_{10}O_5]_n + nCS_2 + nNaHSO_4$ Viscose fibre		
	SNa n		
4.b	Manufacturing of oil -:	2	4
	i) Solvent Extraction		
	Cakes obtained by pressing operations contain 5-10% oils can be		
	recovered by heating		
	the cake with volatile hydrocarbon like benzene, petroleum ether. The		
	common solvent for edible oil is Hexane.		
	The use of chlorinated solvents mainly to decrease the explosion and		
	fire hazard did not prove much satisfactory. The solvent used should not		
	make yhe oil toxic for the application.		
	ii) <b>Refining</b> -:The colour and flavor to fats of edible non edible oils is	2	
	mainly due to presence of non-glycerides compounds. Free fatty		
	acids, waxes, coloured bodies, gossypol compounds are responsible for		
	the undesired properties of fat or oils, used for edible purposes & industrial applications. Most of those compounds are removed by		
	treatment with aq. Solution of caustic soda at 40-85 deg.C		
	treatment with aq. Solution of caustic soda at 40-65 deg.c		

4 c	H <sub>2</sub> O	4	4
4.c	CO <sub>2</sub> scrubber  Fermenter  H <sub>2</sub> O  Mixer  G0°  Flash  Condenser  Alcohol  Fractionator  H <sub>2</sub> O  Continuous  Cooker  Water	4	4
4d	Paints are usually applied one coat on top of another and each coat has a specific function purpose.  The primer is applied directly onto the cleaned steel surface. Its purpose is to wet the surface and to provide good adhesion for subsequently applied coats.  In the case of primers for steel surfaces, these are also usually required to provide corrosion inhibition.  The intermediate coats (or undercoats) are applied to 'build' the total film thickness of the system. Generally, the thicker the coating the longer the life. This may involve the application of several coats. The finishing coats provide the first line defence against the environment and also determine the final appearance in	4	4
4.e	Classification of plastics:-	2	4
	i)Thermoplastics-:The thermoplastics are those which when heated		
	begin to soften at atemp. Of 60 deg.C then can be moulded without any		

	change in chemical structure.		
	e.g. Acrylics, cellulosics, fluorocarbons, polyethylene, nylon,		
	polyvinyls		
	ii) Thermosetting plastics-: The thermosetting	2	
	. e.g. Alkydes , Epoxides Furah, Phenolics, Polysters		
4.f	Detergent builders-: Detergent builders are complex phosphates which	2	4
	increase detergent power. These are water softners which keep away Ca		
	& Mg ions. They prevent redosition of soil from the wash water from		
	the fabric.		
	<b>Detergent brighteners</b> -: Brighteners are used to increase effectiveness	1	
	of soaps & detergents. Fabric brighteners are fluroscent dyes which		
	make the fabric look brighter because they have the ability to convert U.		
	V. light. to visible light.		
	<b>Detergent additives</b> -: Synthic detergents in granular form contains 2-9	1	
	% additives corrosion or tarnish inhibitors etc.corrosion inhibitors such		
	sodium silicate protect metal &washer parts dishes from the action of		
	detergent & water.		
5.a	Toluene Oxidation process for production of phenol	1	8
	Raw material:		
	(i) Toluene (ii) Air (iii) cobalt naphthenate and cupric benzoate catalyst		
	Reaction:	2	
	(a) Oxidation to benzoic acid		
	CH <sub>a</sub> + 1-1/2O <sub>2</sub> → COOH + H <sub>2</sub> O		
	naphthenate benzoic acid.		
	(b) Oxidation of benzoic acid to phenol		
	COOH + 1/2 O <sub>2</sub> $\xrightarrow{220^{\circ}\text{C}}$ C <sub>6</sub> H <sub>5</sub> OH + CO <sub>8</sub>	5	
	3.2.3. Quantitative requirements		
	Process:- Two stage oxidation process is used in the first fresh and		

		,	
	recycle toluene are mixed with small amount of cobalt naphthenate		
	catalyst and charged to reactor which liquid filled tower through air is		
	spared. Cooling tubes are provided to remove the exothermic heat of		
	reaction.		
	The reactor is run at 150 0C and 3 atm. Excess air is used but toluene		
	conversion is 40% this give by product. The yield of benzoic acid is		
	about 90%.		
	Off gases are vented and it is passes through condenser for collection of		
	water and toluene. Liquid from reactor passes to a distillation column		
	which strips the toluene and other by product at the bottom of column .		
	Purified benzoic acid is separated by extracting the bottoms with hot		
	water then crystallizing and filtering the crude benzoic acid.		
	To make phenol the crude acid is melted mixed with cupric benzoate		
	catalyst then charged to an air-sparged tower containing cooling tubes		
	and mechanical agitation. Reactor condition are 220 0 C and 1.3-1.7		
	atms.		
	Phenol product is obtained by continuously distilling the reactor liquor		
	into a fractionating column where unreacted benzoic acid is returned to		
	the reactor. Phenol is withdrawn as a bottoms product .		
5.b	Production of Alcohol from Molasses:-	2	8
	Raw material		
	Molasses (Black strap) It is a mother liquor left after the		
	removal of sugar crystals. By product of sugar industry. Contain		
	about 55% sugar.		
	Yeast commonly used for fermentation which produce large		
	amount of alcohol.		
	Following steps are carried out during process	6	
	(i) Preparation of inoculum From selected strains o yeast the		
	inoculum is prepared, in which starter containing yeast in its		
	long phase. The yeast developed in the seed tank should be pure		

	Environment issue:-	4	
	(iv)For decorating stalls in fairs, showcases of shops etc.		
	exhibitions		
	(iii) In school and colleges for mounting drawing and photographs in		
	(ii)For decorations		
	(i)For inner packaging like television, computer, washing machine		
5.c	Uses	4	8
	a dry ice.		
	obtained during distillation.CO2 is obtained and is converted as		
	alcohol. To prepared absolute ethanol the 5% water is removed.  (vi) By products:- n-amyl alcohol and isoamyl alcohol is		
	ethyl alcohol. It contain 60% alcohol after distillation get 95%		
	(v) Recovery:- The fermented mesh is distilled to obtained pure		
	Glucose Ethanol		
	C6H12O6 2C2H5OH+2CO2		
	sucrose Glucose Fructose		
	(iv) C12H22O16 C6H22O6 +C6H12O6		
	and medium.		
	for 50 hr at 30 to 40 0C in fermenter after mixing yeast starter		
	exothermic no temp. contral is required. Reaction is carried out		
	pushes air out and anaerobic fermentation starts The reaction is		
	fermentation.so it is carried out in the absence of oxygen.CO2		
	(iii) Fermentation: - Alcoholic fermentation is the anaerobic		
	add to improve quality of fermentation. PH value adjusted to 4 to 5 by adding H2SO4.		
	Nutrients such as ammonium sulphate or ammonium phosphate		
	10 to 18%. Can be used directly as fermentation medium.		
	(ii)Preparation of medium: The molasses is diluted with water to		
	and free from contamination and mutation.		

	Thermocol is a bio-nondegradable product and is one more enemy of		
	environment.But after its use its thrown away in dust bin and then travel		
	to dumping ground to remain buried in the heap of trash. As it is bio-		
	nondegradable product it remain intact.		
i.a	Rusching process	3	8
	This process has two vapour phase catalytic stage. Purified benzene is		
	fed to a heater reactor containing ferric chloride catalyst. chloranation		
	with HCL-O2 at 2200C occurs with residence time to produce 10-20%		
	conversion of benzene. Fractionation separates unreacted benzene from		
	chlorobenzene and polychlorobenzene		
	The crude chlorobenzene is scrubbed with phenol, water washed, and		
	sent to second catalytic stage. Here it is hydrolysed. Phenol from the		
	hydrolyzer it is washed with water then extracted by benzene,and		
	finally purified by two stage distillation.		
	Neat diagram	3	
	3.3.1. Reference flow spect: Figure 174-5		
	Benzene  Benzene  Benzene  Air Hydro-Chlorination Recactor  Makeup HCI Recycle HCI Waste (Ø-Ch)  Furfied Ø-O  Purified Ø-O  Puri		
	Uses:- (i) Plastic industries (ii) Chemical intermediates for herbicides,		
	insecticides, pharmaceuticals and dyestuffs.(ii) Petroleum refining.	2	
	msecuciocs, pharmaceuticais and dyesturis.(ii) retroieum terming.		

6.b	The solvent is separated from the product stream by heating,	2	8
	evaporation or fractionation, and residual traces amounts are		
	subsequently removed from the raffinate by stream stripping		
	Description	6	
6.c		1.5	8
	(i) PVC:- Used for manufacturing raincoats, hand bags, plastic	mark	
	dolls, curtain cloths and vinyl flooring. Good electric insulator.	each	
	(ii) Polystyrene:-It is used for manful. of polystyrene in films and		
	fomas.		
	(iii) Polyethylene:-High pressure polyethylene is used for producing		
	household utensils, bowls and bckets and also packing film. Low		
	pressure polyethylene is used for beer crates.		
	(iv) Polesters:- In textile manfu Can be bland with cotton and wool		
	called as teryool for fishing nets, filter cloths, conveyor belts etc.		
	(v) Thermocol:-(i)For inner packaging like television,computer,		
	washing machine(ii)For decorations		