

21415

17340

3 Hours/100 Marks

Instructions: (1) All questions are compulsory.

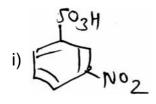
- (2) Illustrate your answers with **neat** sketches **wherever** necessary.
- (3) Figures to the **right** indicate **full** marks.

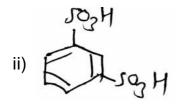
MARKS

1. Answer any ten:

20

- a) Define Aryl Halides. Give example.
- b) Write functional group in Aryl diazonium ion. Also write structure of benzene diazonium chloride.
- c) Explain Aliphatic compounds and Aromatic compounds.
- d) Write sulphonation reaction in case of Aniline.
- e) Name the compounds:





- f) Explain how m-dinitrobenzene is prepared from nitrobenzene.
- g) Write resonating structures of Naphthalene.
- h) Explain Anthraquinone. How is it prepared?
- i) Write structure of:
 - i) Phenolphthalein
 - ii) Picric acid.



MARKS

- j) Convert Aniline into m-Nitro aniline.
- k) Write physical properties of Toluene.
- I) Write uses of Benzoic acid.

2. Answer any four:

16

- a) Explain how is chlorobenzene prepared by direct Halogenation. Also write uses of chlorobenzene.
- b) Explain following reactions in case of Toluene:
 - i) Catalytic Hydrogenation
 - ii) Friedel craft alkylation.
- c) Explain preparation of Nitrobenzene from Benzene with mechanism.
- d) Write chemical reactions of Benzene sulphonic acid with:
 - i) Br₂/FeBr₃
 - ii) $HNO_3/H_2SO_4/\Delta$
 - iii) Fuming H₂SO₄/200°C
 - iv) NaOH.
- e) Write physical and chemical properties of Anthracene.
- f) Write physical properties and uses of Benzene sulphonic acid.

3. Answer any four:

16

- a) Explain how is Benzene prepared from:
 - i) Acetylene
 - ii) Phenol.
- b) Explain how is Benzoic acid prepared from:
 - i) Toluene
 - ii) Phenyl Magnesium Bromide.

MARKS

c) Name the compounds:

- d) Write chemical properties of Aniline.
- e) Write physical properties of Naphthalene. Also write its uses.
- f) Write physical properties and uses of phenol.

4. Answer any four:

16

- a) Convert:
 - i) Nitrobenzene to Aniline
 - ii) Chlorobenzene to Aniline.
- b) Write chemical reactions of phenol with:
 - i) Zn dust
 - ii) Br₂/H₂O
 - iii) Br_2/CS_2
 - iv) Conc. HNO₃.
- c) Explain how Alizarin prepared from Anthracene.
- d) Explain distillation of coal tar.



MARKS

- e) Discuss what happens when Benzene is treated with:
 - i) Acetyl chloride/AlCl₃
 - ii) CH₃Cl/AlCl₃.
- f) Explain Diazotization reaction.

5. Answer any four:

16

- a) Write physical properties and uses of Benzene.
- b) Explain how does benzoic acid react with:
 - i) HNO₃/H₂SO₄
 - ii) LiAlH₄
 - iii) PCI₅
 - iv) C₂H₅OH/H⁺.
- c) Explain preparation of Benzene diazonium chloride. Also write its uses.
- d) Explain Acidity of phenol.
- e) Write physical and chemical properties of chlorobenzene.
- f) Explain Basicity of Aniline. Also write its resonance hybrid.

6. Answer any four:

16

- a) Explain Reduction reaction of Nitrobenzene in acidic, alkaline and neutral medium.
- b) Convert Benzene into Benzene sulphonic acid. Also write mechanism of the reaction.
- c) Write physical and chemical properties of benzene diazonium chloride.
- d) Explain following reactions in case of Naphthalene:
 - i) Nitration
 - ii) Sulphonation.
- e) Write structure of:
 - i) α -Bromo Naphthalene
 - ii) Anthraquinone.
- f) Explain how will you show Anthracene has three Benzene rings fused in orthoposition.
