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$P_{ideal} = 1.0004$, $P_1 = nRT/(V - nb)$ $P_1 = 1.0029$, $P_2 = (a \cdot n^2)/V^2$ $P_2 = 0.0129$, $P_{waals} = P_1 + P_2$ $P_{waals} = 1.0158$ The ideal gas law predicts a pressure of 1.0004 atmospheres, while the van der Waals model predicts 1.0158 atmospheres.

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Introduction to MATLAB for Engineers, by William J. Palm III, is aimed at beginning engineering students and practicing engineers' who want to learn MATLAB. The text assumes no prior programming knowledge. Based on Professor Palm's experience in teaching MATLAB with freshmen, the text contains many examples of engineering applications of MATLAB.

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