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Most ionic compounds are crystalline solids at room temperature. In general, ionic compounds have high melting points because the ions have a strong attraction for one another. Ionic compounds conduct an electric current when melted or in an aqueous solution because the ions are then free to move. After reading Lesson 7.2, answer the following questions.

BONDING AND INTERACTIONS

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•In contrast to ionic compounds, which are formed from a metal combined with a nonmetal, most molecular compounds are composed of atoms of two or more nonmetals. Comparing Molecular and Ionic Compounds Molecular compounds tend to have relatively lower melting and boiling points than ionic compounds.

Covalent Bonding - Pittsfield High School

Exam 0610 Answers Hcs12 9s12 Solution Manual.Ionic Bonding Workbook Answers Pearson Education 7.2 Ionic Bonds and Ionic Compounds Essential Understanding Ionic compounds are the result of ionic bonds forming between oppositely charged ions.

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lonic compounds exist as collections of positively and negatively charged ions arranged in repeating three-dimensional patterns. A formula unit is the lowest whole-number ratio of ions in an ionic compound.

SECTION 7.1 IONS (pages 187-193)

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Chapter 6 Chemical Bonds Section 6.1 Ionic Bonding

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