

Name:- OMKAR GUJJA

Roll No.- 71

SUBJECT :- EIMM
MTRX

1 Explain toughening mechanics in ceramics.

→ Ceramics are now more brittle than most metals and plastics. The irreversible work associated with plastic deformation is not presented in ceramics. Hence, the methods that improve the toughness of ceramics are different from metals. These several toughening mechanisms called crack deflection, micracking toughening, transformation toughening and crack bridging.

→ Crack deflection:- In polycrystalline ceramics, the crack can propagate in an intergranular way. The associated irreversible work per unit area is $2\gamma - \gamma_b$, where γ is the surface energy of material and γ_b is the grain boundary energy. As the irreversible work is decreased because of grain boundary energy, the fracture area is increased in intergranular crack propagation, which furtherly improves the toughness of ceramics.

→ Micracking: Micracking means that the formation of microcrack before the main crack can toughen the ceramic.

Additional microcracks will cause stress to concentrate in front of the main crack. This leads to additional irreversible work required for crack propagation.

Transformation toughness: Partially stabilized zirconia is composed of tetragonal phase at high temp and monoclinic phase and cubic phase at lower temp. In equilibrium. In some components, the onset temp, in equilibrium, of tetragonal monoclinic matensite transformation at velocities hypothesized to approached at that of sound in the material.

Crack Bridging:- When a crack propagates in an irregular path; some grains of each side of main crack may protrude into other side. This leads to additional work for a complete fracture. There are some other approaches to improve the toughness of ceramics through crack bridging.

2) Compare ceramics with metals with regards to their properties and structure.

→ Comparison metals Vs Ceramics.

METALS	CERAMICS.
1) Large no. of free electron.	1) Captive electrons.
2) Metallic bond.	2) Ionic / Covalent bond is present poor conducting.
3) Uniform atoms	3) Different size atoms
4) High tensile strength	4) Poor tensile strength
5) Good ductility	5) Poor ductility
6) Impact strength	6) Impact strength is poor.
7) Relatively high weight	7) Lower weight
8) Moderate hardness	8) Extreme Hardness
9) Good electrical conductivity	9) Poor conductivity.