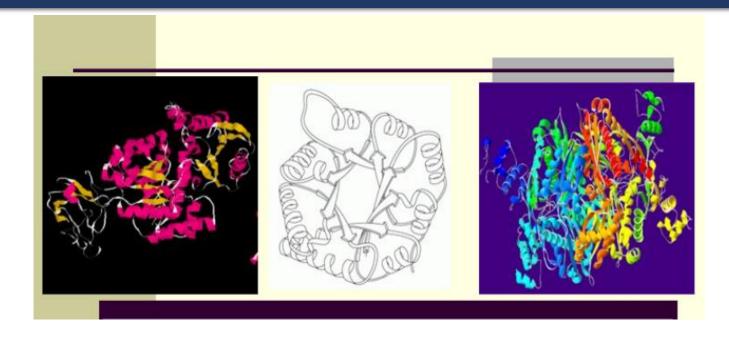
ENZYMES INHIBITION

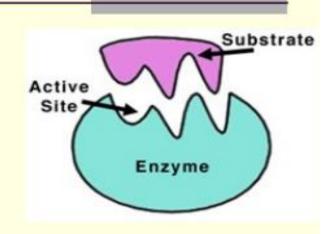


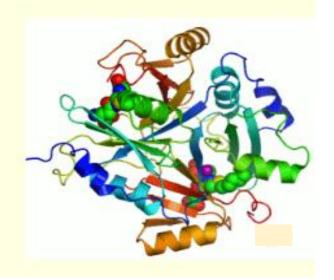
Dr. Zarish Noreen

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Enzymes

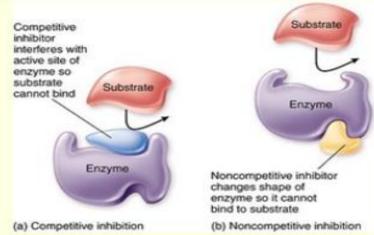
- Enzymes are biomolecules that catalyze, increase the rates of chemical reactions without being altered during the reaction.
 - Almost all enzymes are proteins;
 - Enzymes are essential to life.
- In enzymatic reactions, the molecules at the beginning of the process are called **substrates**, and the enzyme converts them into different molecules, the **products**.



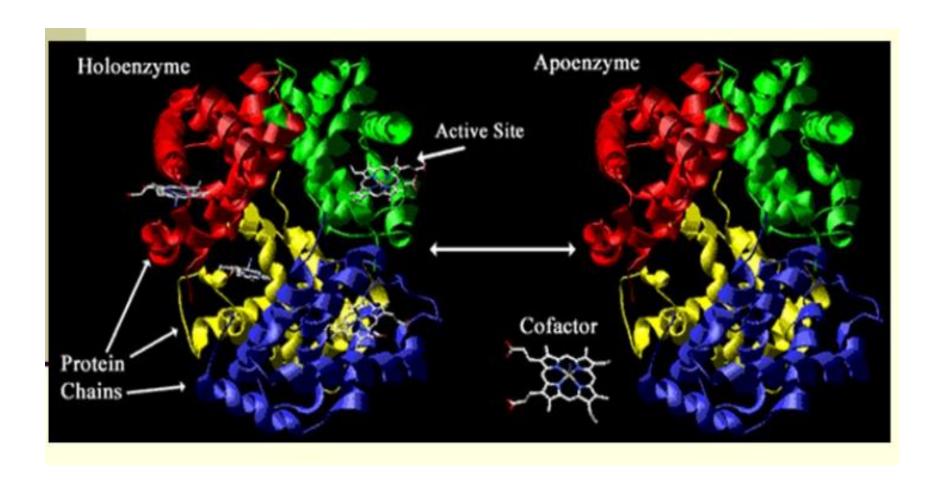


Enzymes activity

- Enzyme activity can be affected by other molecules.
 - Inhibitors are molecules that decrease enzyme activity.
 - Inducers are molecules that increase activity. Many drugs and poisons are enzyme inhibitors.
- Activity is also affected by temperature, chemical environment (e.g. pH),



Three-Dimensional Structure of Enzyme



Enzyme Structure

- Most enzymes are proteins
- Enzymes may require a *non-peptide component*as a cofactor. The peptide component is called the apoenzyme, the cofactor is called as the coenzyme and the combined functional unit is the holoenzyme
- Cofactors that are tightly bound to the polypeptide are called prosthetic groups. Such proteins are called as <u>complex or conjugated</u> <u>proteins</u>. Proteins without prosthetic groups are <u>simple proteins</u>

Enzymes Inhibitors and Inducers

Enzyme Inhibitor	Enzyme Inducer
Cimetidine	Rifampicin
Ketoconazole	Carbamazepine
Fluconazole	Phenobarbital
Miconazole	Phenytoin
Macrolides(except Azithromycin)	Griseofulvin
Fluoroquinolones(except Levofloxacin)	Smoking
	Chronic alcoholism

- Inhibitors are substances that binds to an enzyme and interfere in its activity.
- It can prevent formation of Enzyme-Substrate complex or can prevent ES breakdown to enzyme + product.
- Inhibitors are chemicals that reduce the rate of enzymic reactions.

- E + S → ES Complex → Enzyme + Product.
- E + S + I → ESI or EI → No product formation.
- There can be irreversible or reversible inhibitors.

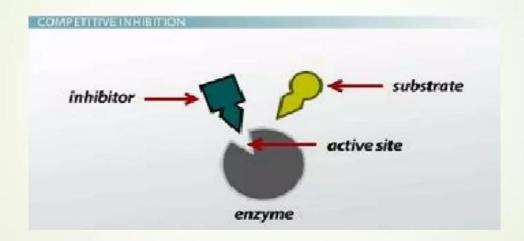
- Irreversible bind to enzyme through covalent bond.
- Reversible bind to enzyme through non-covalent interactions.

- Reversible enzyme inhibitors
- Inhibition of enzyme activity in which the inhibiting molecular entity can associate and dissociate from the protein's binding site.
- There are 3 types of reversible inhibitors
- (1) Competitive inhibition
- (2) Uncompetitive inhibition
- (3) Non-competitive inhibition

(1) Competitive inhibition

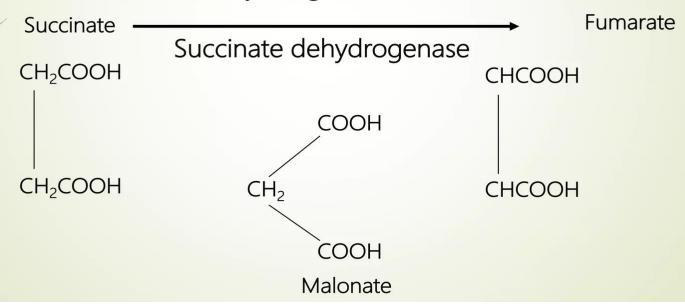
- These compete with the substrate molecules for the active site.
- The inhibitors action is proportional to its concentration.
- It can reverse inhibition by adding more substrate.

(1) Competitive inhibition



(1) Competitive inhibition

■ E.g. - Malonate is a competitive inhibitor for succinate dehydrogenase.



(2) Uncompetitive inhibition

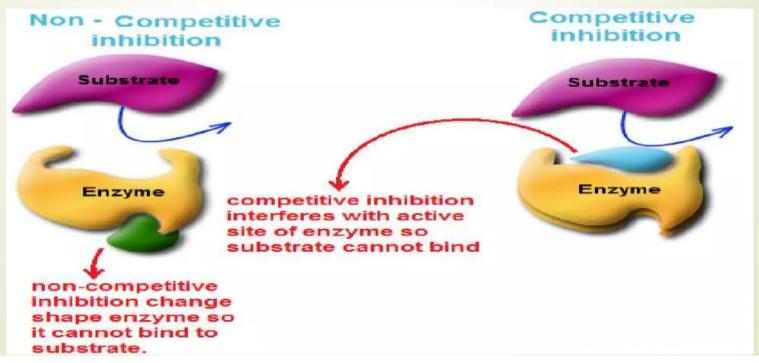
- Inhibitor binds only to Enzyme-Substrate complex.
- Binding site for inhibitor is created only upon substrate binding.
- There's no EI complex, only E, ES, and ESI, but ESI can't make product.
- E.g. Tetramethylene sulfoxide & 3-butylthiolene-1-oxide are uncompetitive inhibitors of liver alcohaldehydrogenase.

(3) Non-competitive inhibition

- These can bind to enzyme or enzyme-substrate complex.
- Enzyme can bind both substrate and inhibitor, but ESI complex can't make product.
- Inhibits by binding irreversibly to the enzyme but not at the active site.

(3) Non-competitive inhibition

E.g.- Cyanide combines with the iron in the enzymes cytochrome oxidase.



Enzyme Inhibition

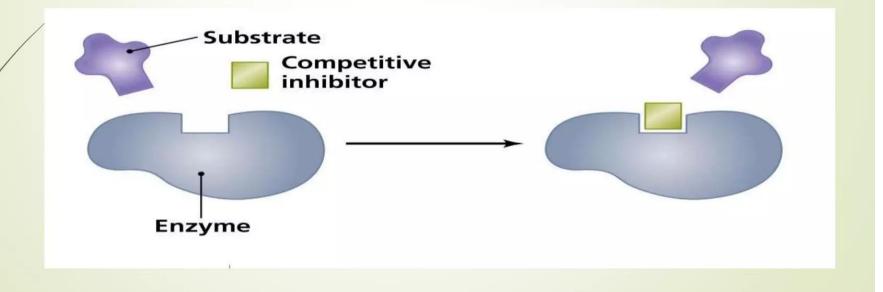
- Irreversible enzyme inhibitors
- The binding of an inhibitor can stop a substrate from entering the enzyme's active site from catalyzing its reaction.
- Irreversible inhibitors usually react with the enzyme and change it chemically.

Irreversible enzyme inhibitors

- They bind covalently and destroy the functional group on the enzyme that is essential for its activity.
- They could be
- (1) Competitive
- ► (2) Allosteric

(1) Competitive

Permanent binding to enzyme active site.



(2) Allosteric

- Permanent binding to enzyme allosteric site.
- They changes the shape of the enzyme.



Example of irreversible inhibitors

Oxalic and citric acid inhibit blood clotting by forming complexes with calcium ions necessary for the enzyme metal ion activator.

