

Comparison of Catalyst Immobilization Techniques onto Filter Media for Airborne VOCs Decomposition

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Materials Science & Technology

Research Focus

Airborne pollutants:

Volatile organic compounds

Particulate matter

Other gases

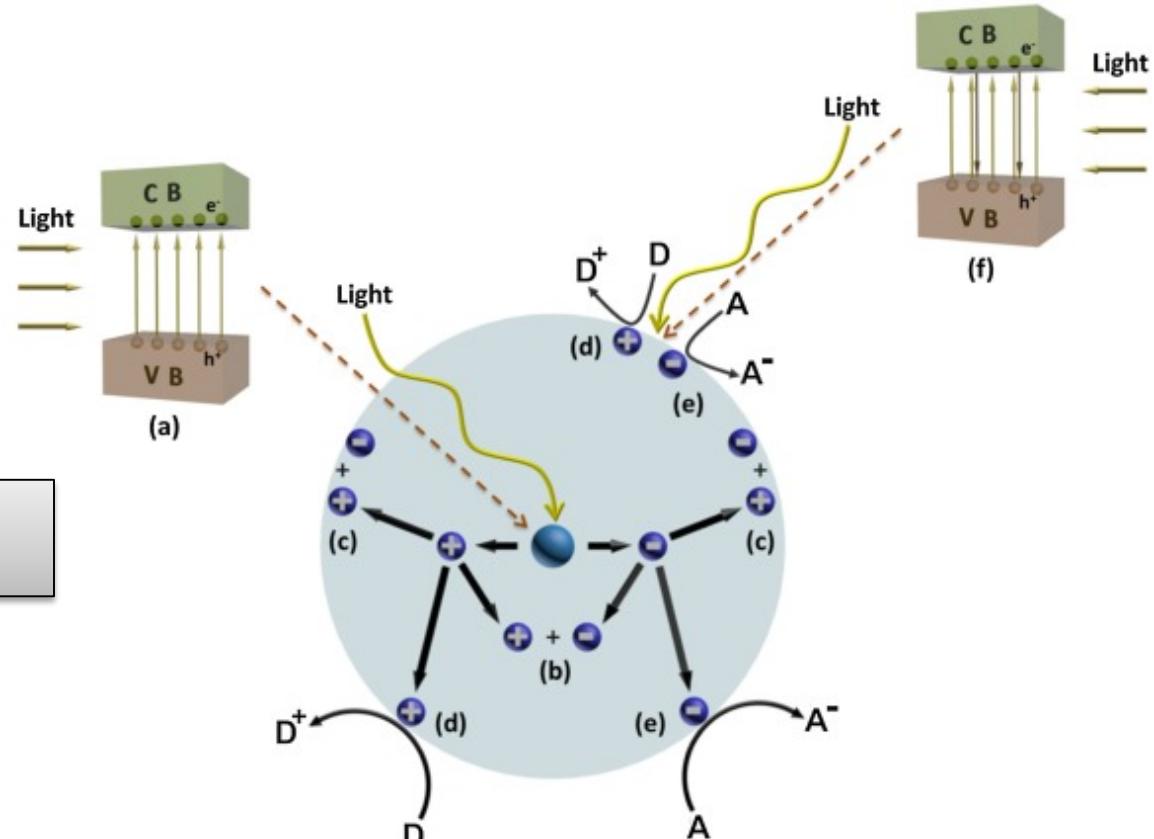
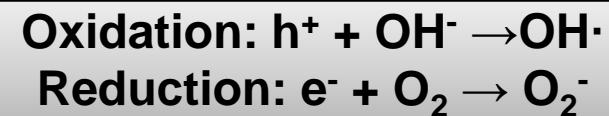
- Environmental and health concerns (allergic reaction, headache, eye, nose and throat irritation)
- Precursors for PM

Source of VOCs:

- Natural emission by plants
- Industrial processes, combustion, waste treatment, ...
- Cooking, carpets, desks, paintings, ceilings, tobacco smoke...

Photocatalysis

Titanium dioxide (TiO_2)

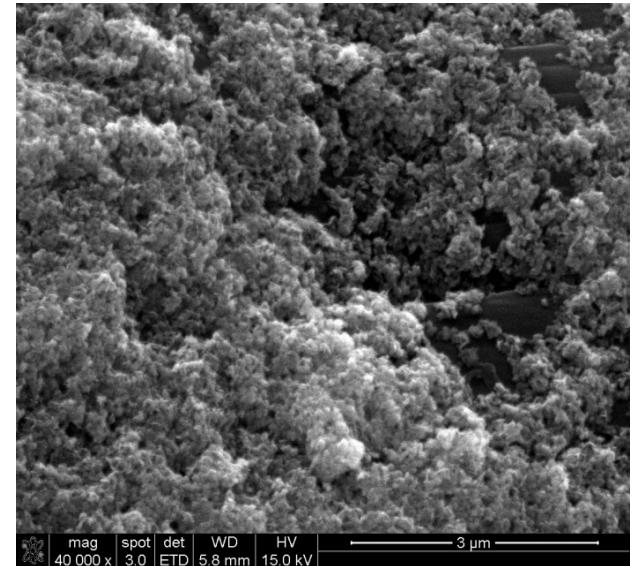
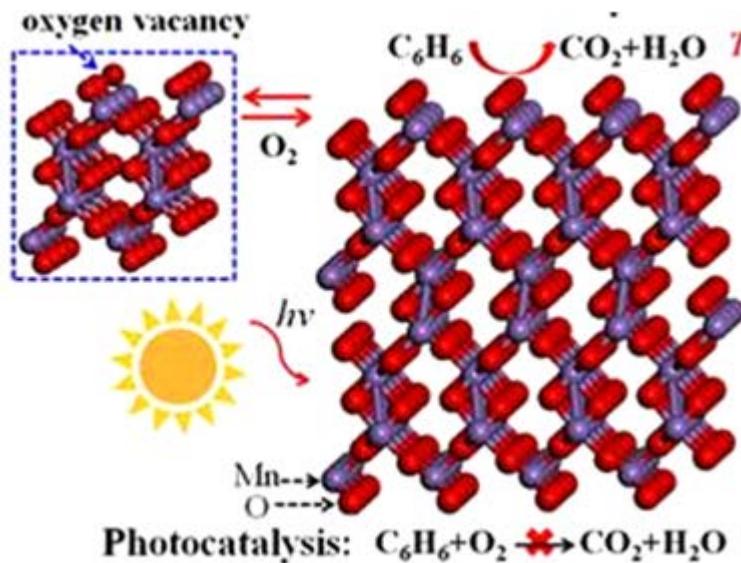


Ren, Hangjuan, et al, *J Hazard Mater* 325 (2017)
340-366.

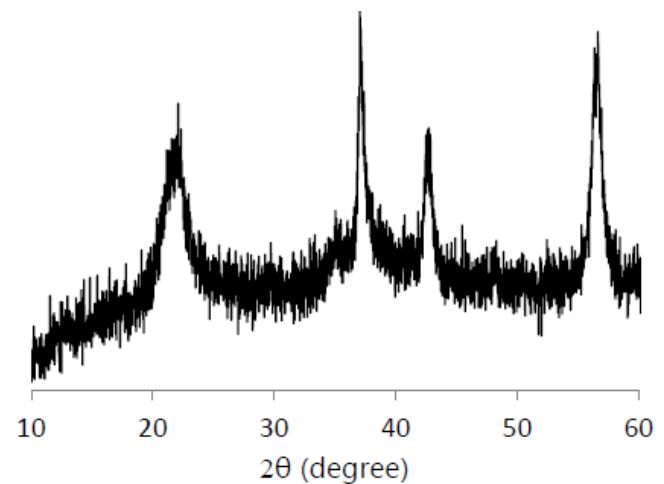
Catalysis by Visible Light

Manganese oxide (MnO_2)

A catalyst which can be activated by visible light



Ramsdellite MnO_2 – Synthesized with hydrothermal method, surface area 66.47 m^2/g (BET)



Immobilization of Catalysts

¹Properties of support material:

Physical and chemical stability

Large surface area

Filter support materials:

Glass fibre filter F9 (LydAir®MG)

Nylon mesh (SEFAR NITEX 03-30/20)

¹Immobilization techniques requirements:

Effective exposure to light

Stable and durable coating

Effective contact of catalyst and pollutant

Easy and energy efficient

Economical and suitable for large-scale application

¹Han, Zhenan, et al, *Aerosol Air Qual. Res* 12 (2012) 1327-1335

Materials and Methods

Catalyst immobilization

1. Concentration of catalyst suspension: 5wt%
2. Sonification for ~ 30 min
3. Stirring
4. Coating of filter media

Spray method

- Spraying distance ≈ 10 cm



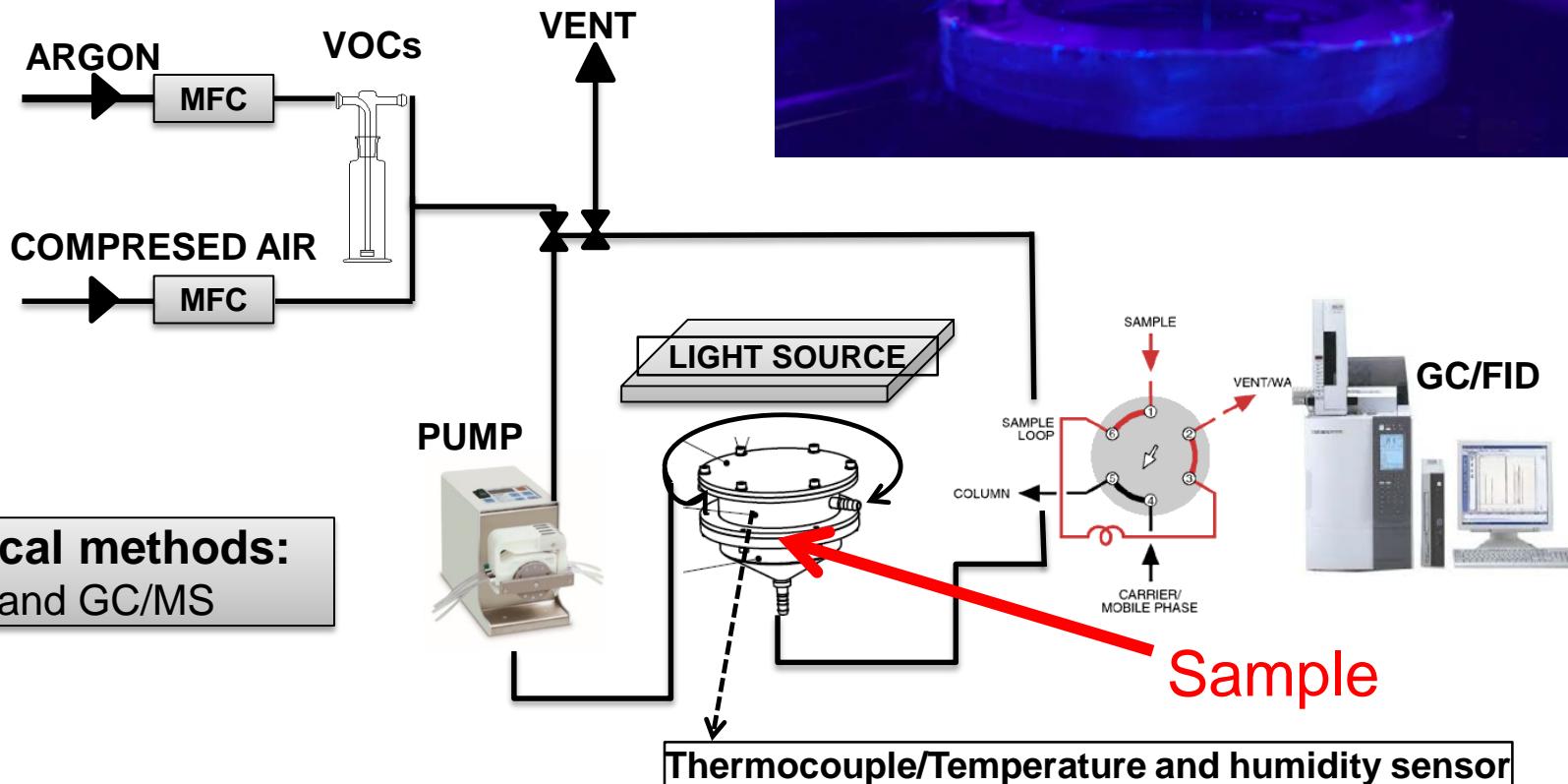
Dip-coating method



5. Drying in ambient condition (in fume hood) for 12 h

Experimental Setup

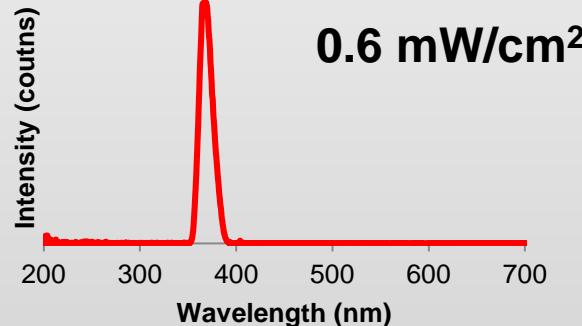
Reactor volume: **1.14 L**
Circulation flow rate: **24 mL/min**
Temperature: **22 °C**
Relative humidity: below **10 %**
VOC: **Toluene/Hexane**



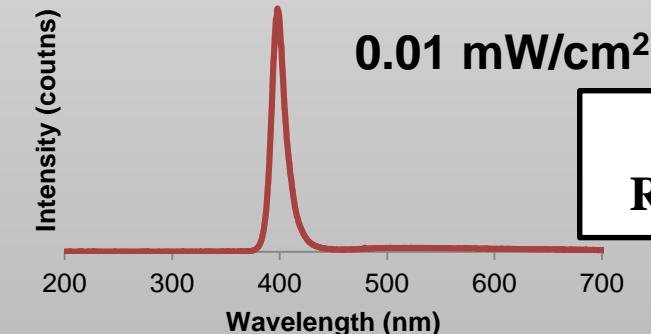
Light Intensity Measurement

UV-A light source

BLB lamp 6x 8W



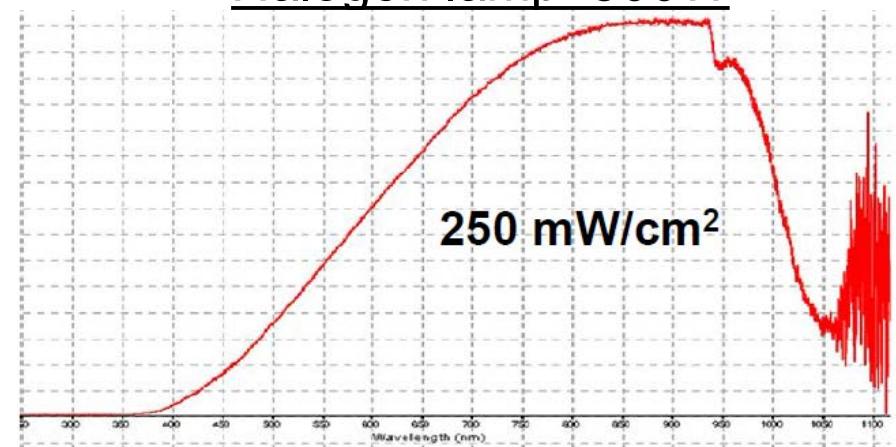
ColorBright™ UV LED strip light



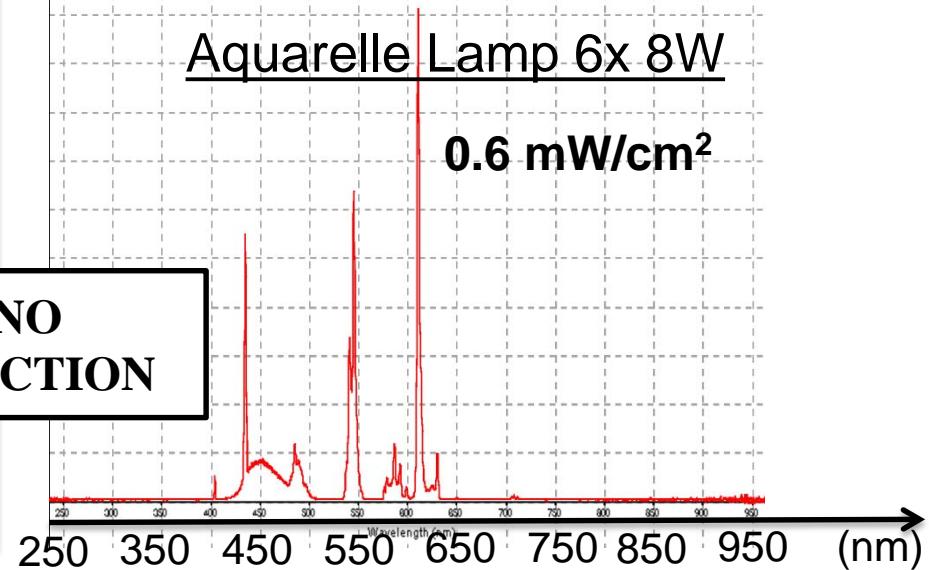
NO
REACTION

Visible light source

Halogen lamp 500W



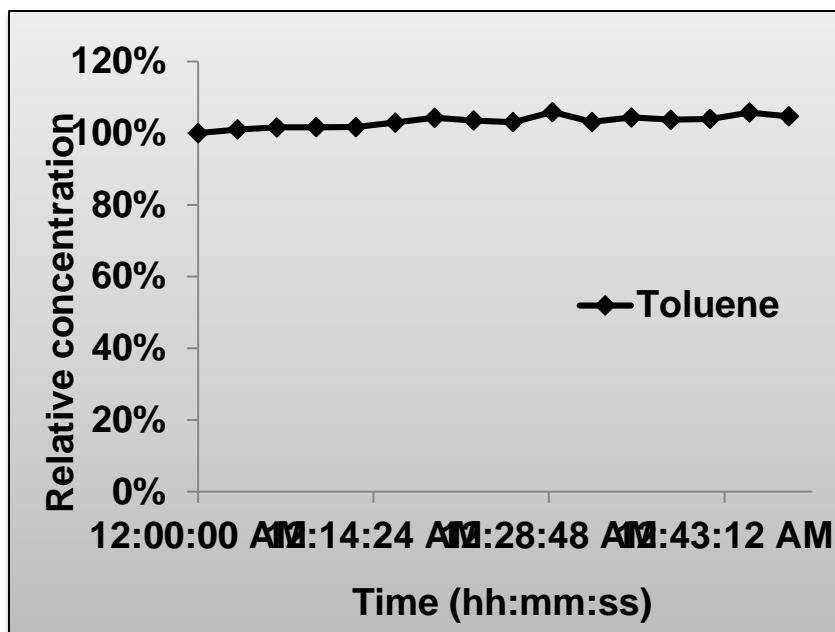
Aquarelle Lamp 6x 8W



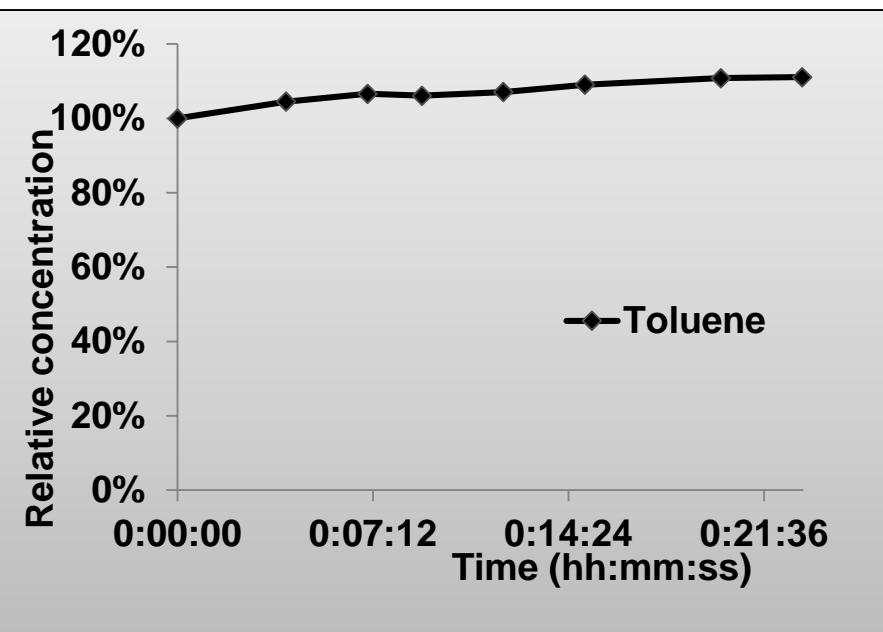
Catalysis Evaluation

Photolysis

Toluene + BLB lamp



Toluene + Halogen lamp

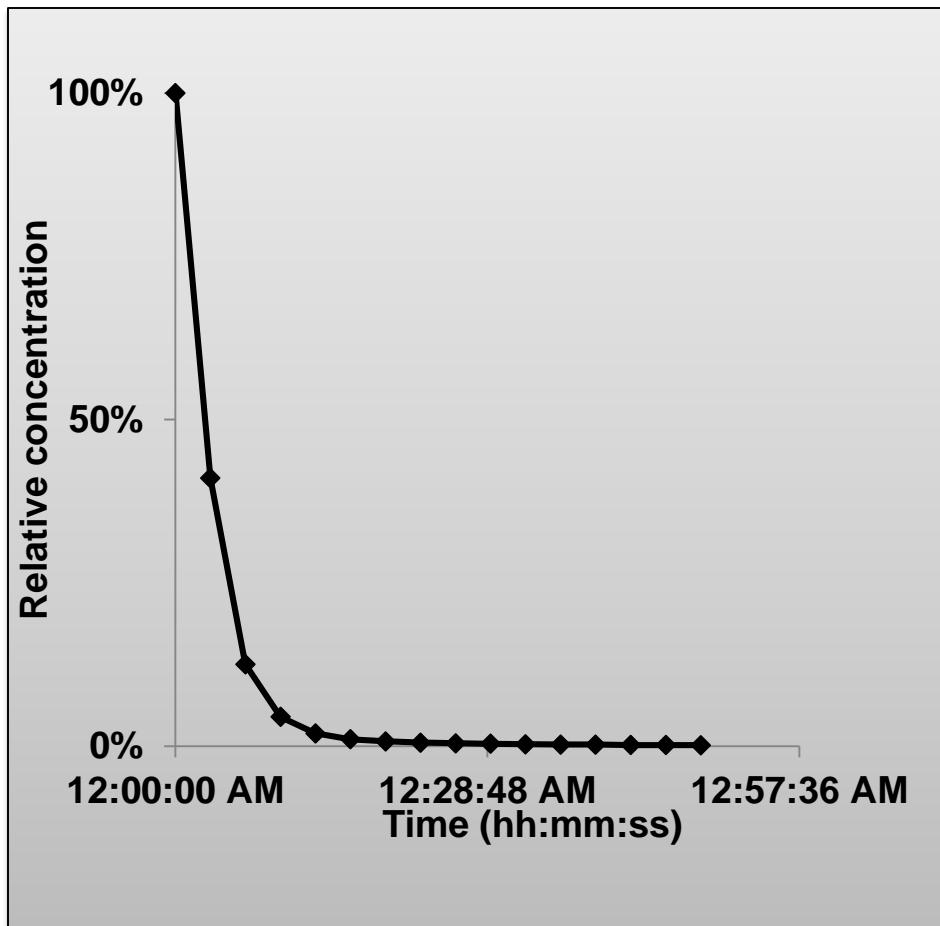


Adsorption

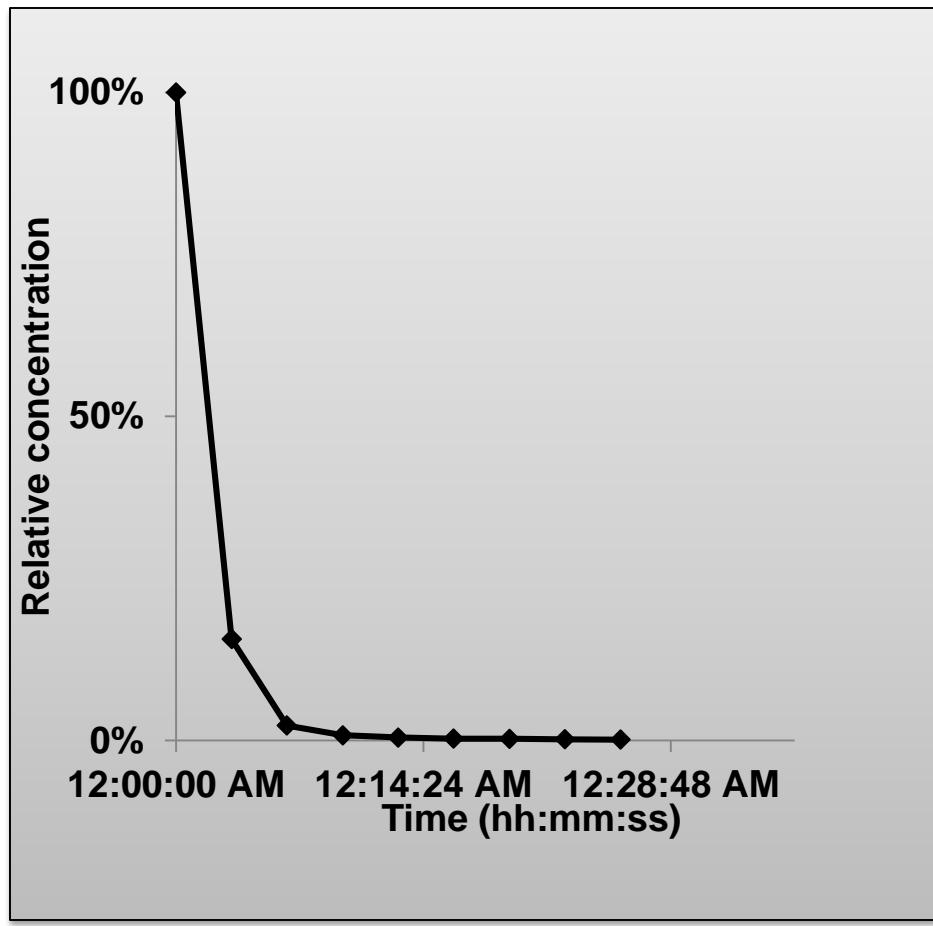
Data acquired after stable condition was reached.

Degradation of Toluene

MnO_2



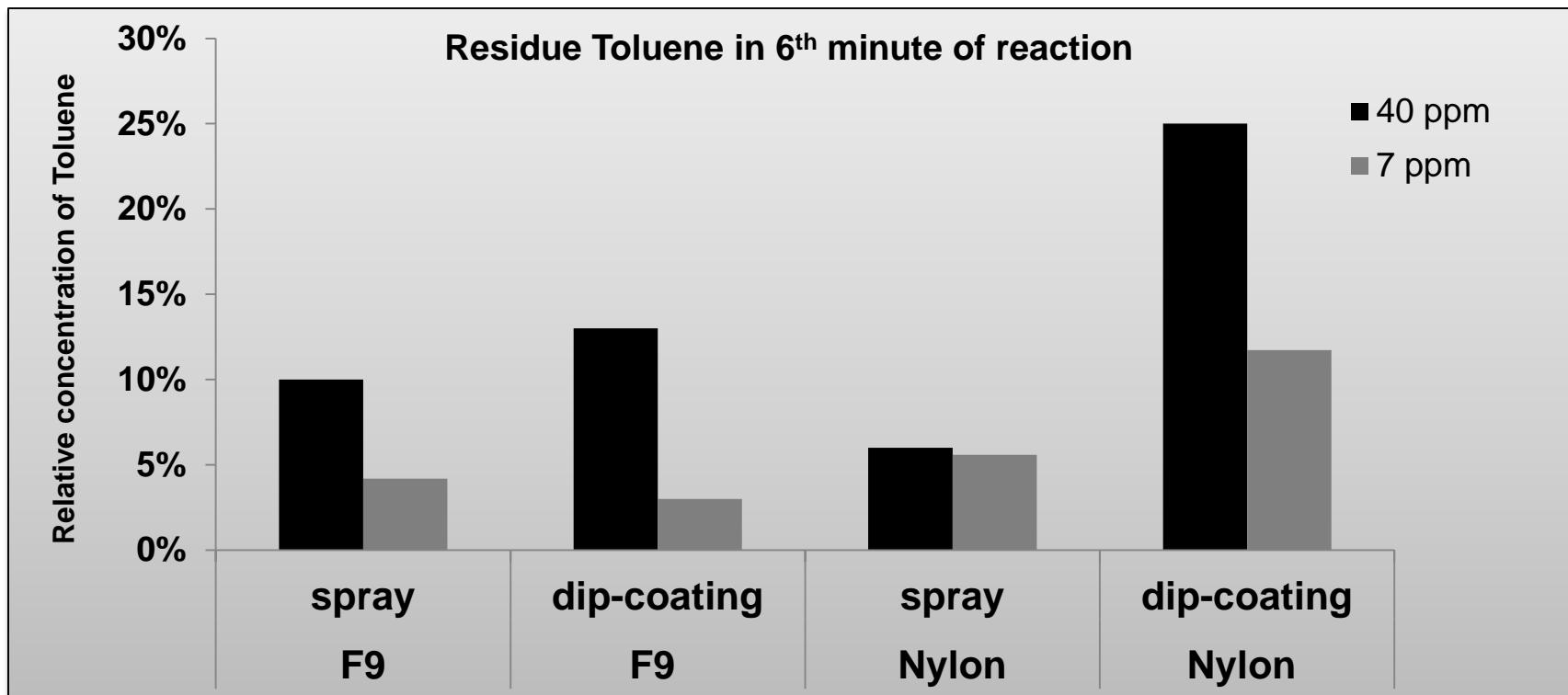
TiO_2



Photocatalysis MnO₂

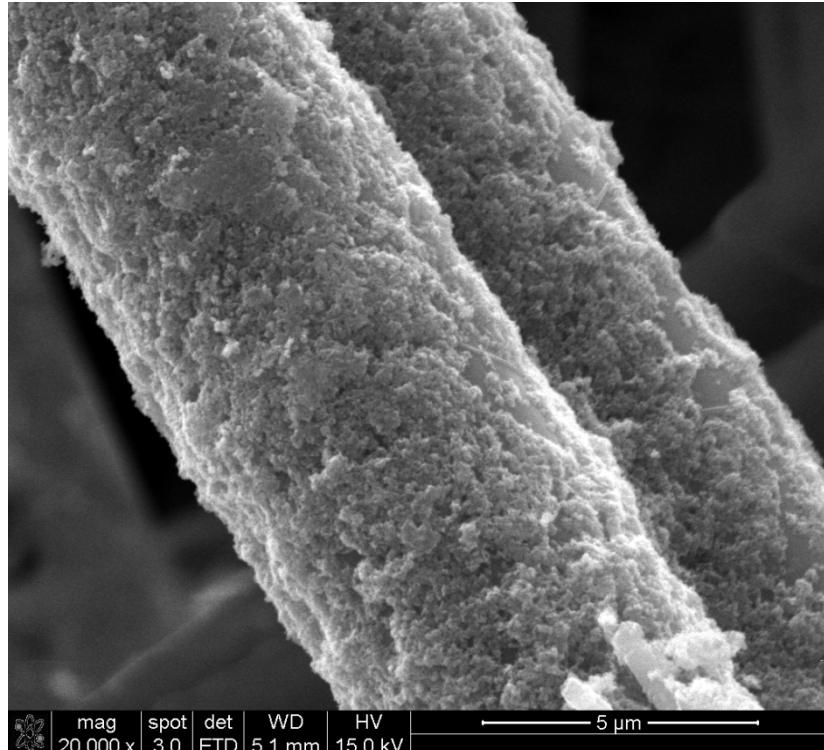
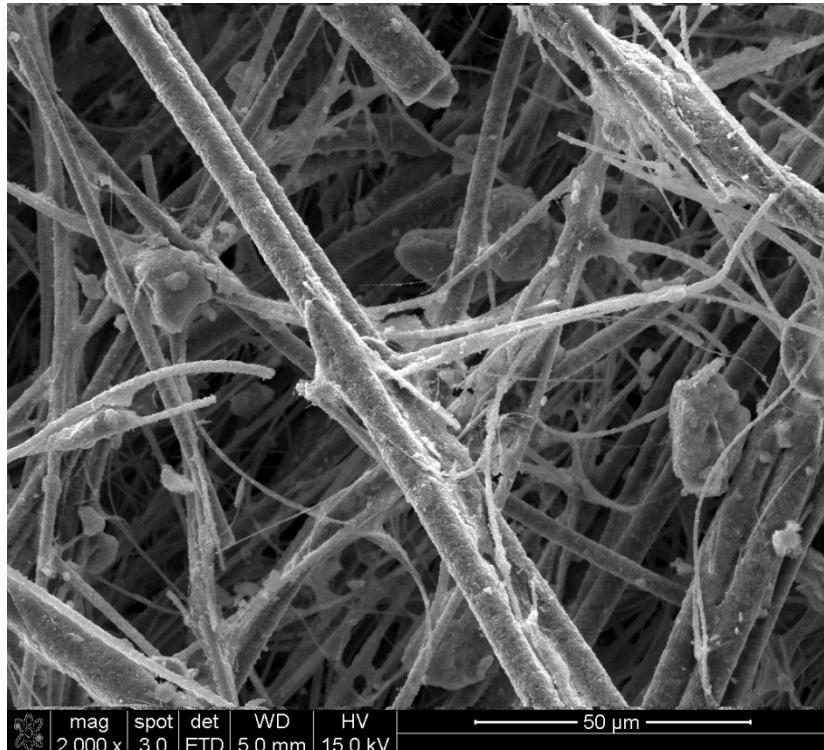
Type of filter	Type of coating	MnO ₂ load, w _{cat} (g)	Number of coating cycles
F9	spray	0.0984	6
F9	dip-coating	0.0957	2
Nylon	spray	0.1047	11
Nylon	dip-coating	0.0904	3

$$w_{\text{cat}} = 0.097 \pm 0.01 \text{ g}$$



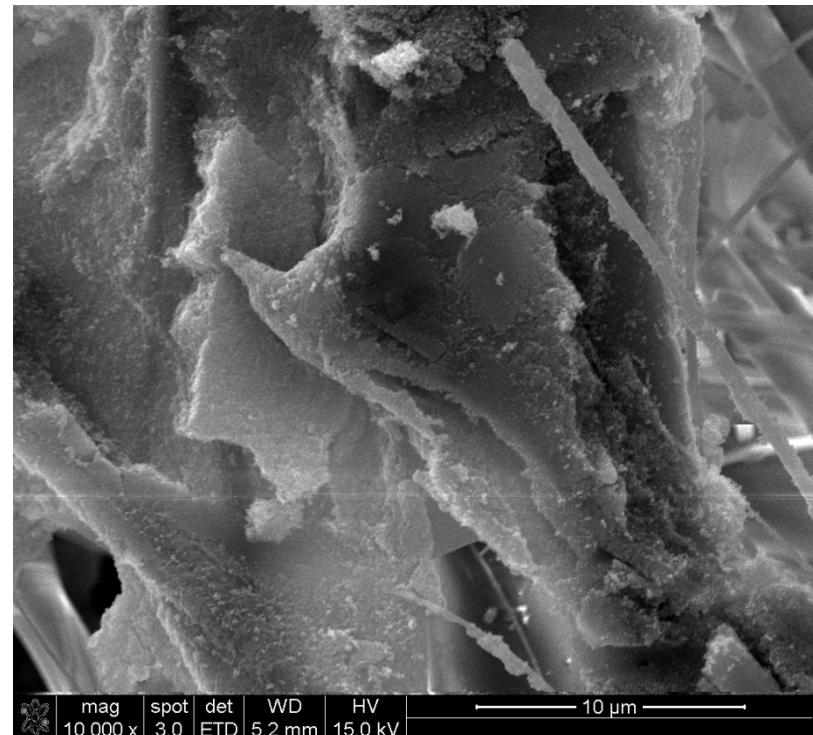
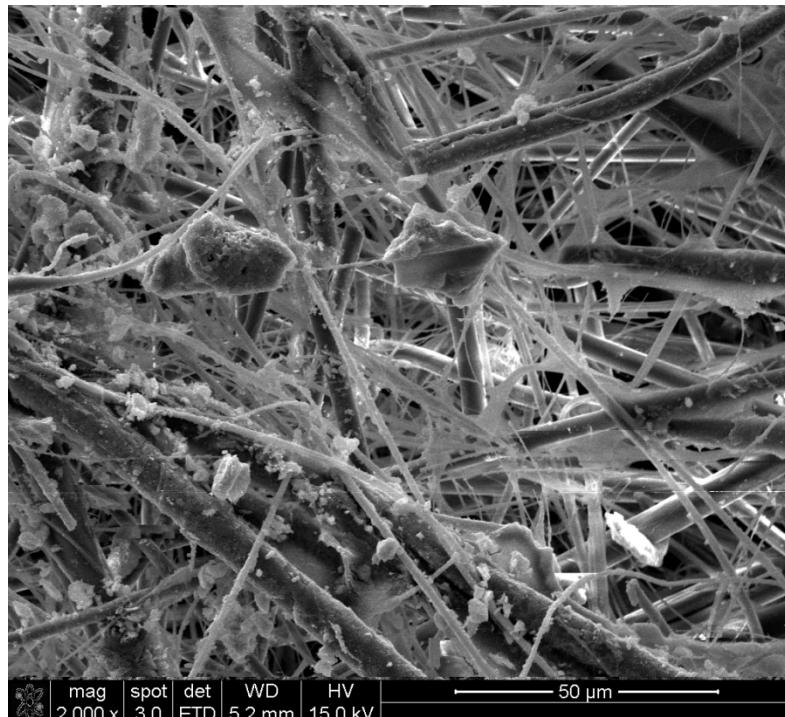
SEM

Type of filter	Type of coating	MnO ₂ load, m _{cat} (g)	Number of coating cycles	Degradation efficiency (%)
F9	spray	0.0984	6	90
F9	dip-coating	0.0957	2	87
Nylon	spray	0.1047	11	94
Nylon	dip-coating	0.0904	3	75



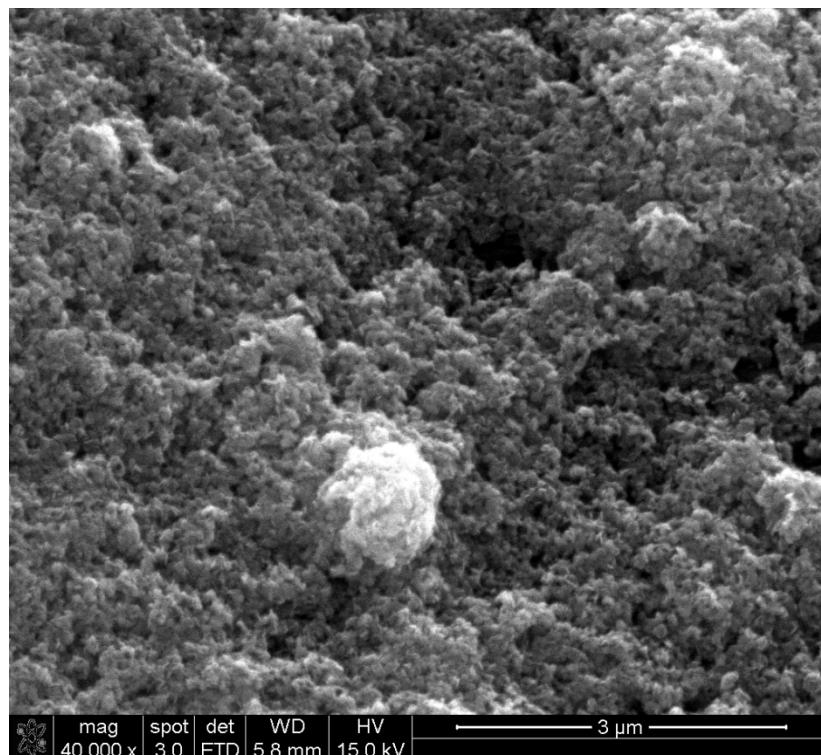
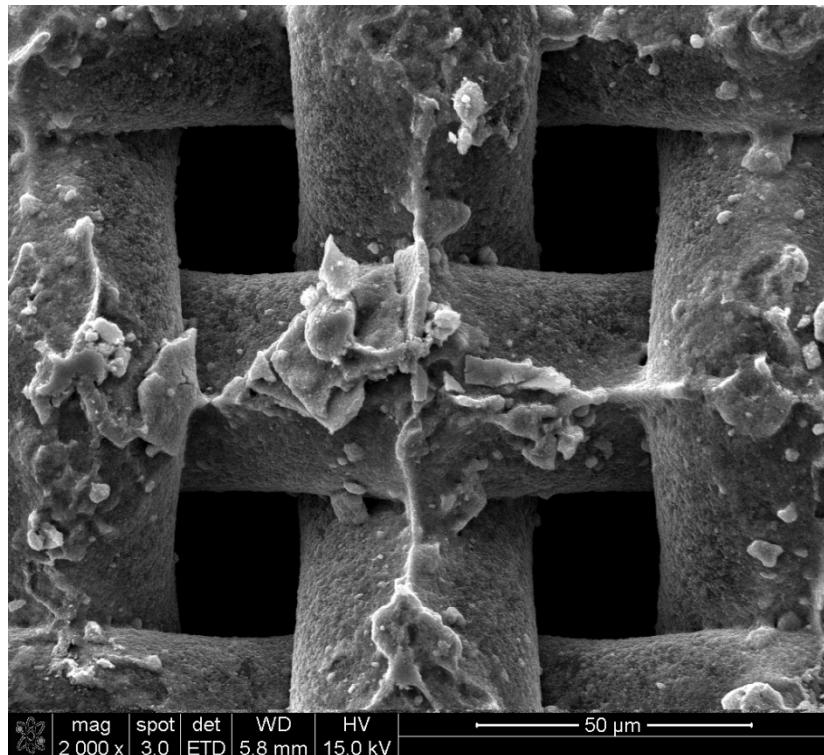
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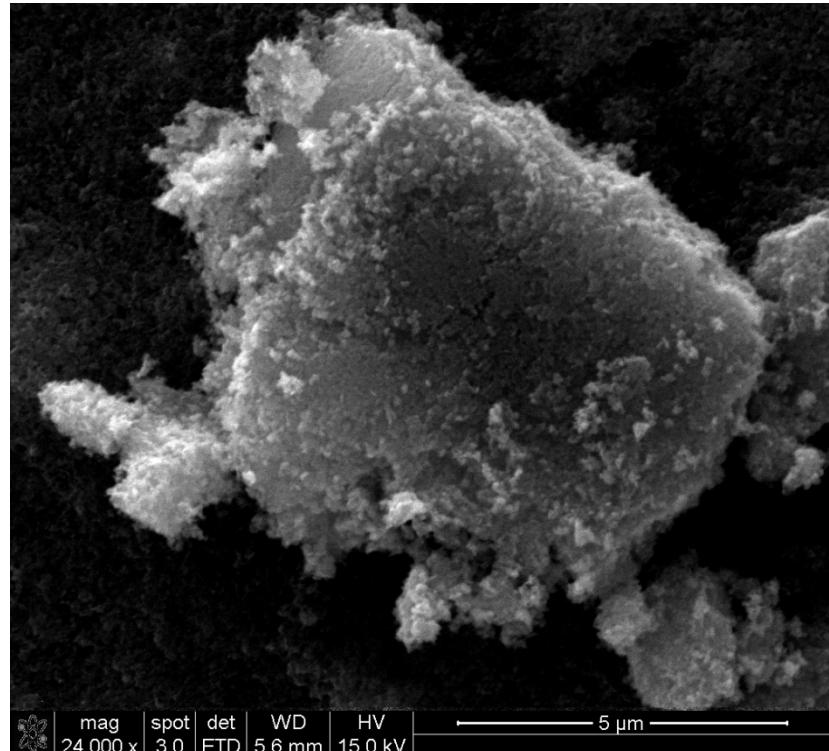
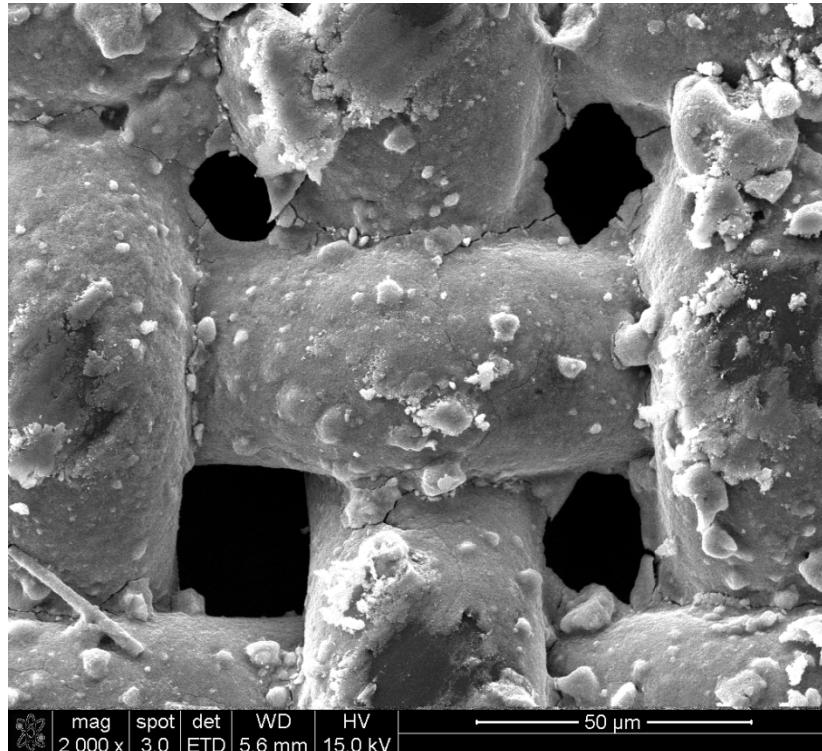
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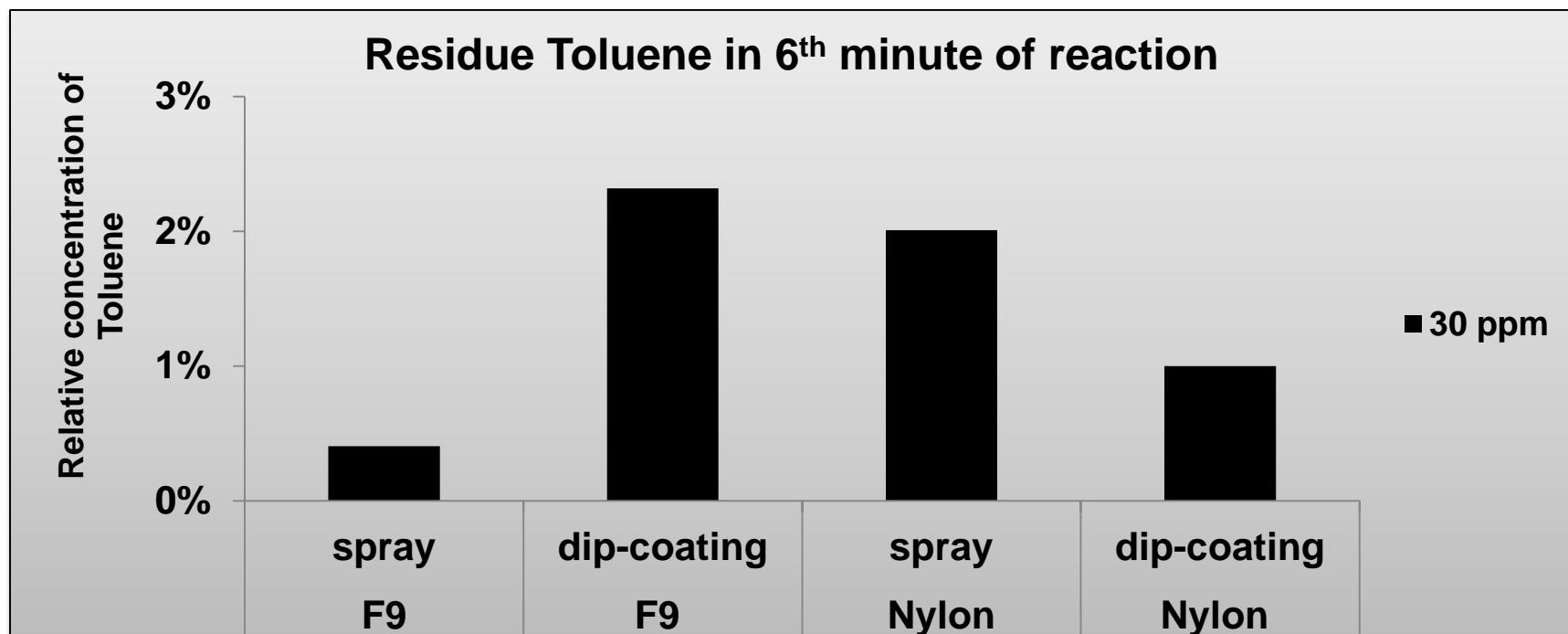
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Photocatalysis TiO₂

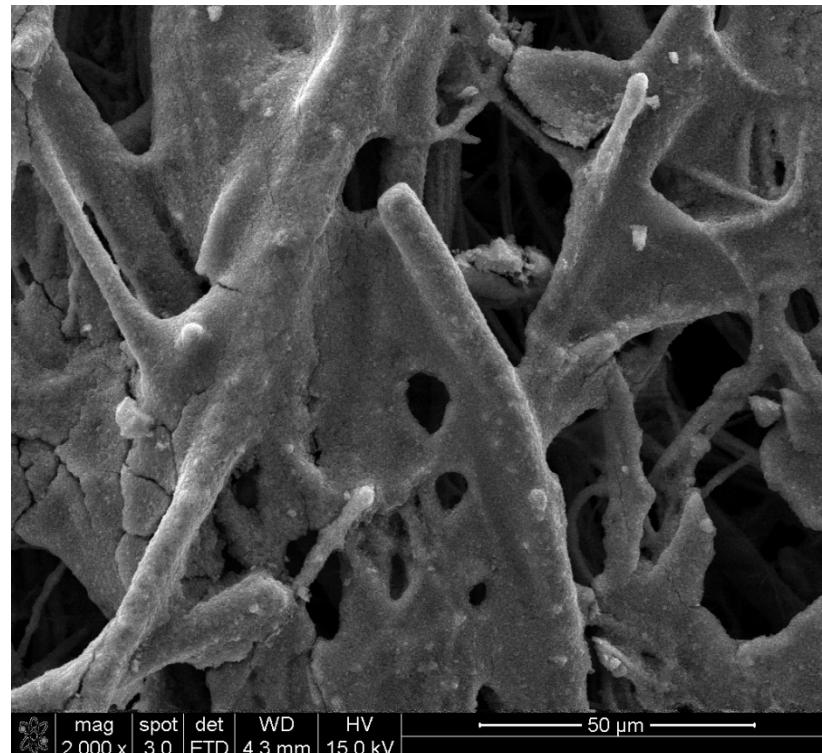
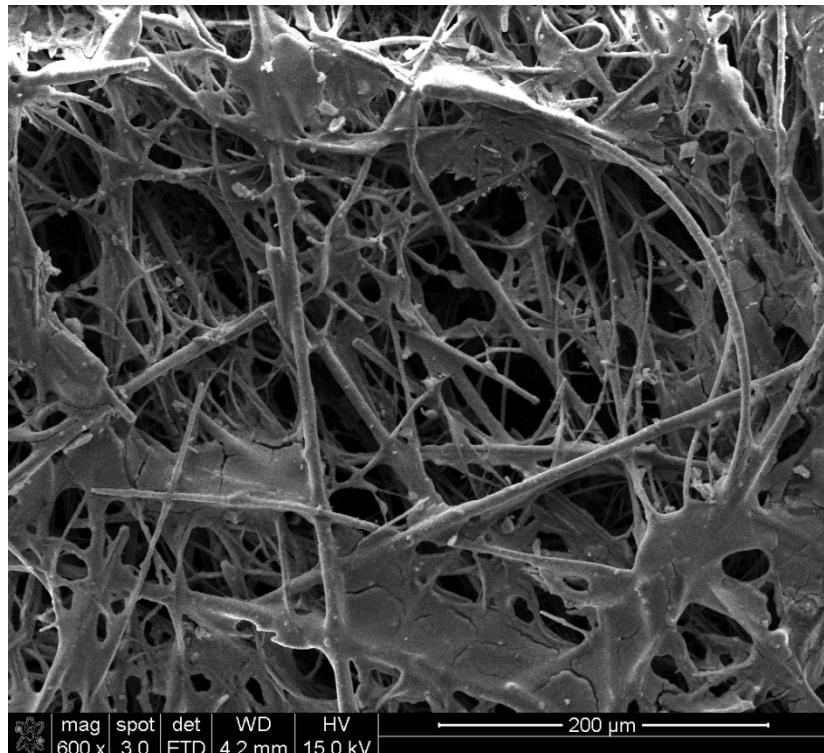
Type of filter	Type of coating	TiO ₂ load, m _{cat} (g)	Number of coating cycles
F9	spray	0.1160	9
F9	dip-coating	0.1009	3
Nylon	spray	0.1000	4
Nylon	dip-coating	0.1612	1

$$w_{\text{cat}} = 0.12 \pm 0.03 \text{ g}$$



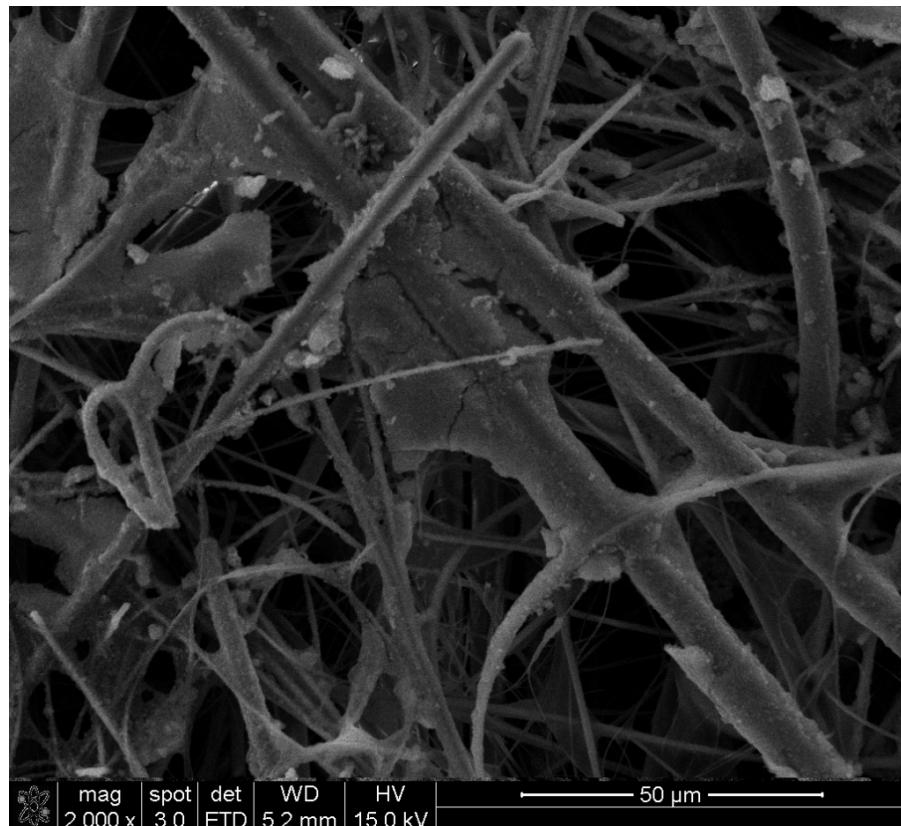
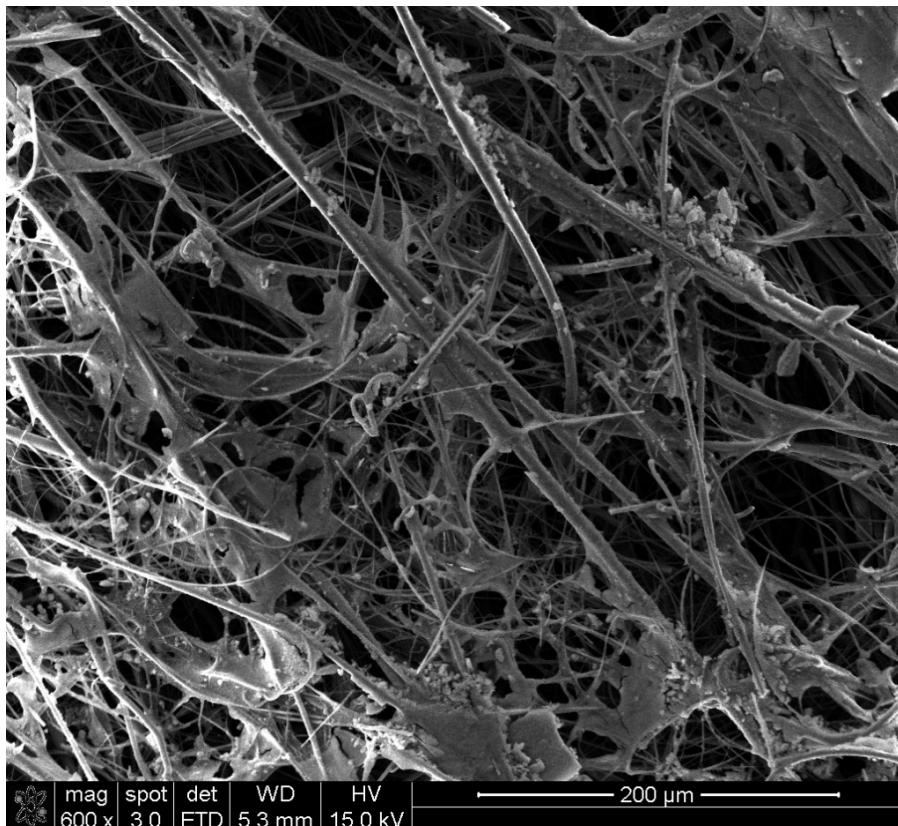
SEM

Type of filter	Type of coating	TiO ₂ load, m _{cat} (g)	Number of coating cycles	Degradation efficiency (%)
F9	spray	0.1160	6	99.5
F9	dip-coating	0.1009	2	97.5
Nylon	spray	0.1000	11	98
Nylon	dip-coating	0.1612	3	99



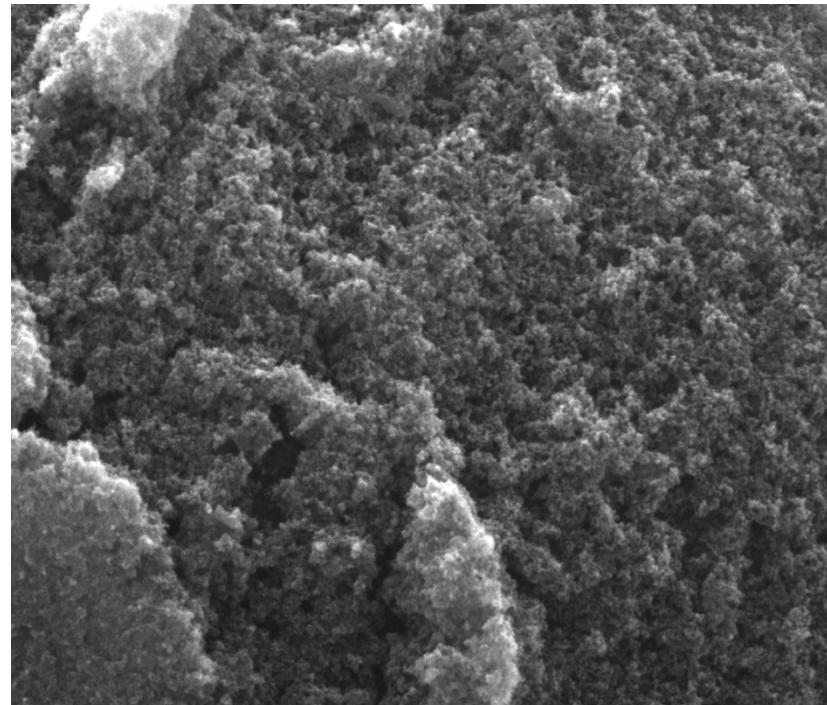
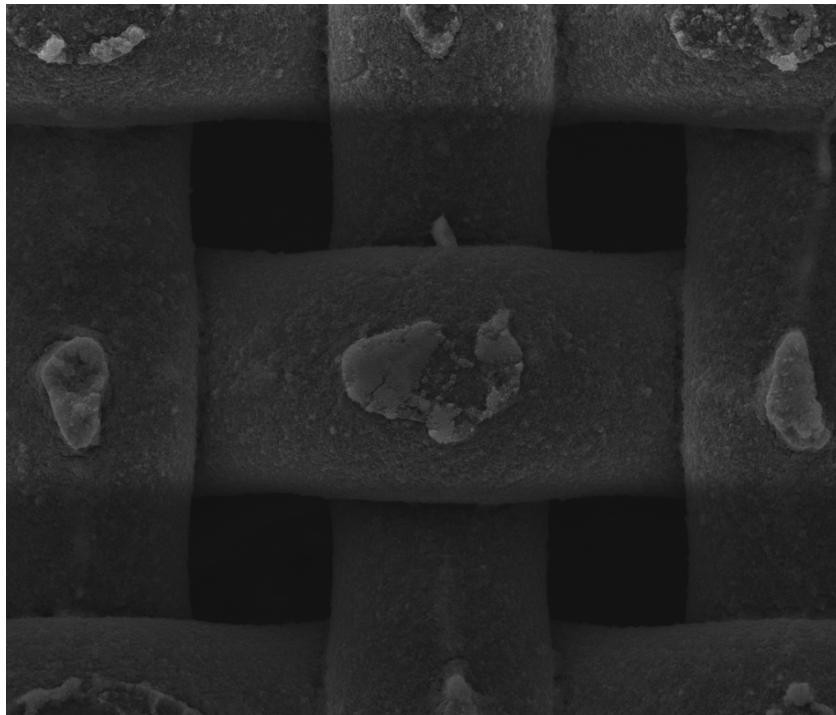
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Type of filter	Type of coating	TiO _x load, m _{cat} (g)	Number of coating cycles	Degradation efficiency (%)
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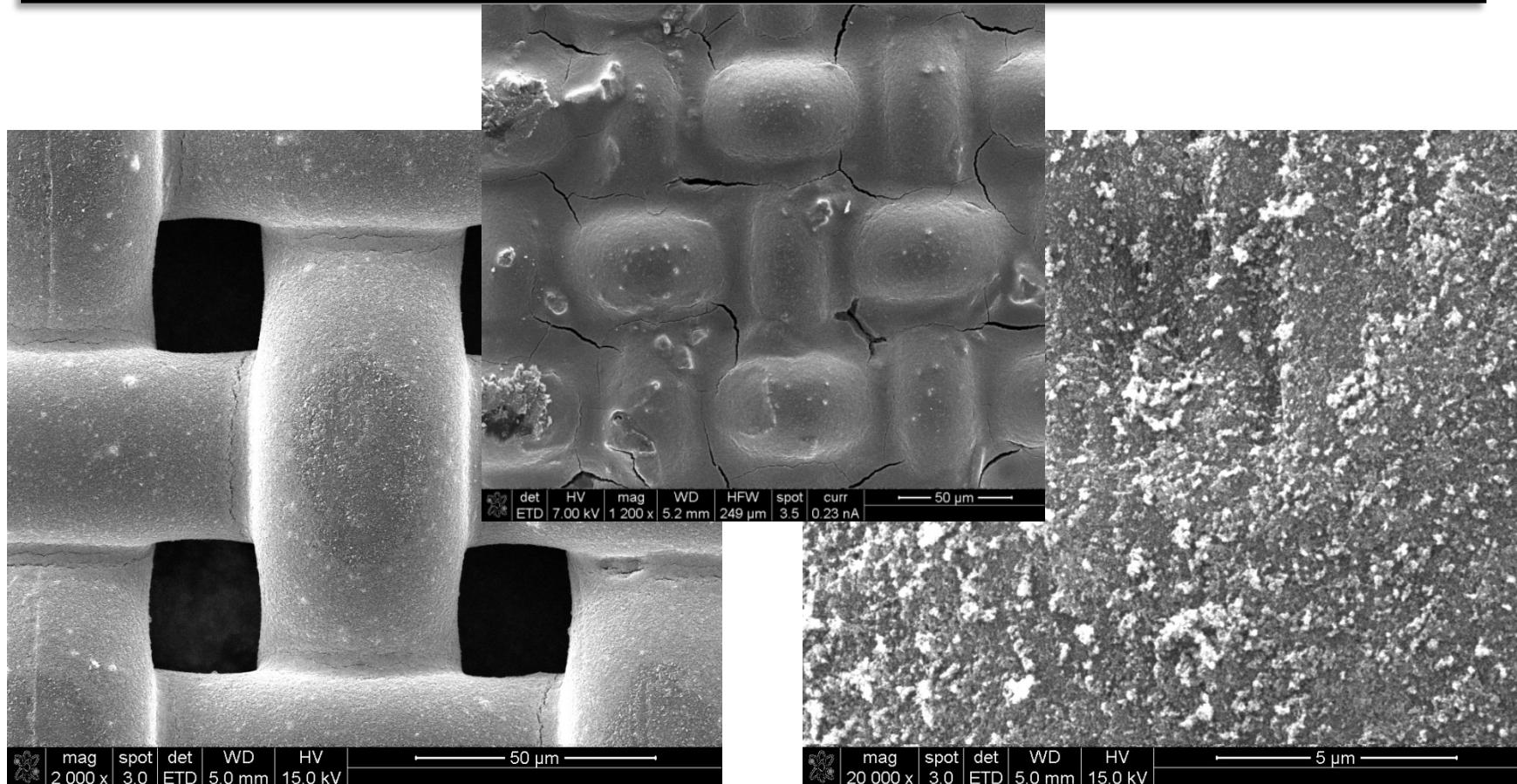
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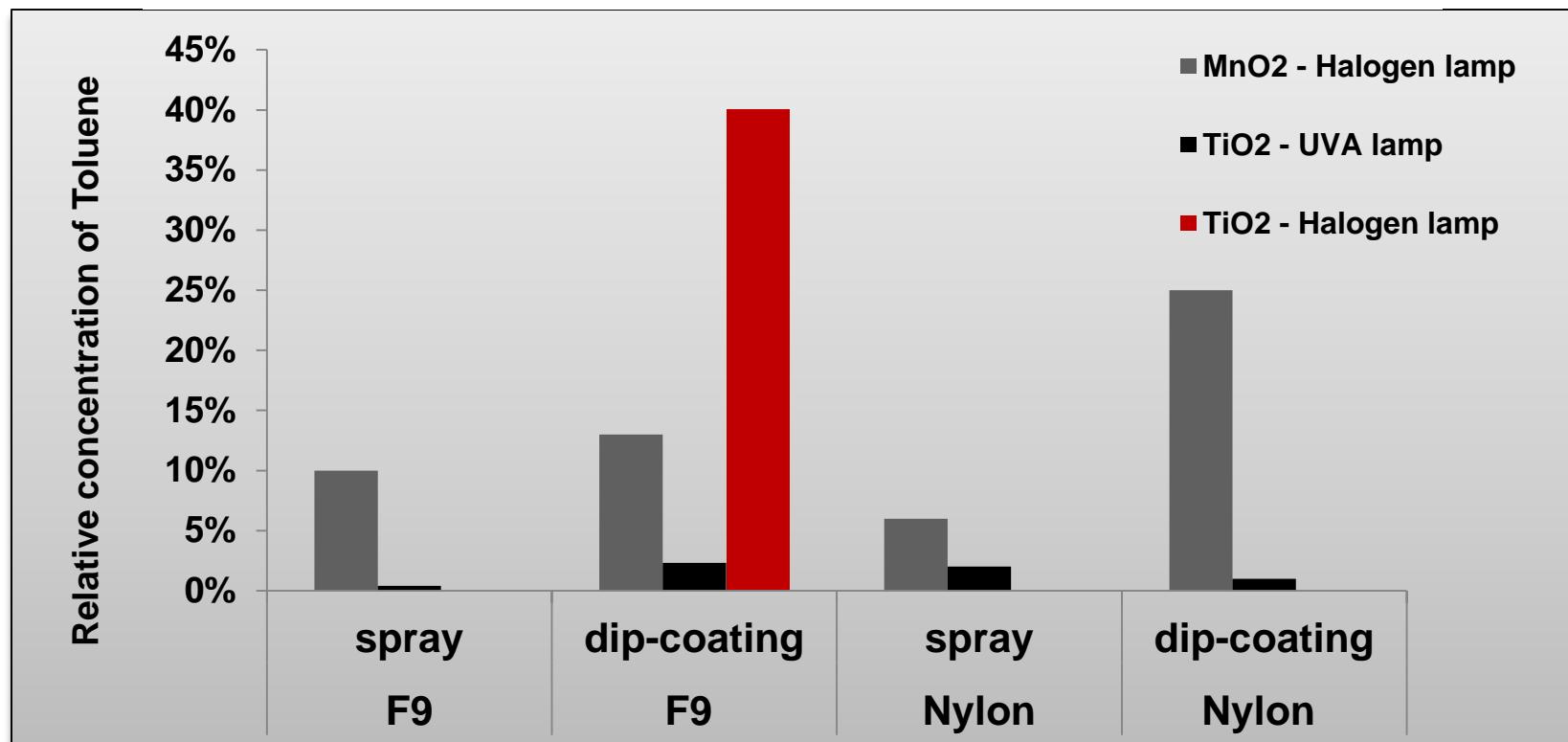
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Nylon	dip-coating	0.1612	3	99



MnO_2 vs. TiO_2

	C_{Toluene} (ppm)	W_{cat} (g)	Surface area (m^2/g)
MnO_2	40 ± 4	0.097 ± 0.01	66
TiO_2	30 ± 5	0.12 ± 0.03	35-65



Conclusion

Dip-coating method

Formation of agglomerates

Spray coating method

More uniform dispersion of catalyst

Less agglomerates

Better performance

TiO_2 more efficient under UV-light irradiance and less efficient under irradiance with halogen lamp than MnO_2