## **Topics in Nanoscience** Assignment-1-23 Data stroage and reading head -> GMR Hyperthermia cancer treatment—>nanoparticle, magnetic non particle, bio tag heat 1. Mention two currently available commercial products where nanoparticles are used. Write the names of the nanoparticles and their characteristics exploited for the application. (2+2+4)length/ diameter 2. Look up the diameter of a silicon atom, in pm. The latest semiconductor chips have fabricated lines as small as 6 nm. How many silicon atoms does this correspond to? (2+3) 3. What are plasmonic particles? Give two examples. PARTICLES, SPR 4. In the clinical area, PEG-intron<sup>TM</sup> is used to treat hepatitis C, multiple sclerosis, and HIV/AIDS. PEG-intron<sup>TM</sup> belongs to which special class and subclass of nanomaterials? (3) more photostable, and more absorption compared to organice dye 5. QDs are better suited as fluorescent probes than the Alexa Fluor 488 dye under prolonged illumination. Give possible reasons for intuition confinement effect high surface are (2) 6. Mention two broad features that arise from the small sizes of nanomaterials and mainly give rise to the unique properties of nanomaterials. (2) 7. When do the 'size effects' begin to appear in materials? length<a href="length">length</a><a href="characterstic">characterstic</a> length</a> (4) 8. Assuming the close-packed full-shell cluster model of atom packing, calculate the (approx.) total number of atoms that will give ~50% of surface atoms. Show the steps of 10n2 +2 calculations. 9. Define "intensive properties". Mention three intensive properties that do not obey this properties volumne, temp,colr—>melting ppint, definition in the case of nanomaterials. solubilty 10. Calculate the approximate number of atoms that will be at the surface of a spherical particle of radius, 5 nm, having around 8,000 total atoms. surface/8000=3/R 11. Mention at least two classical material properties that become quantized in some Quantum Conductance, resistance, and thermal conductance nanomaterials. 12. Surface is present in all materials, but why does one see significant effects of the surface in the nanomaterials? ratio is very much, dangling bond more reactive 13. Write down the criteria of a high-quality superhydrophobic (ultra-hydrophobic) surface. superhydrobpic, roll off<5 14. Compare the Wenzel and Cassie-Baxter models used to explain the superhydrophobicity of a surface. 15. (a) Consider a square array patterned surface that has square pillars of $s \mu m$ side length and $h \mu m$ height placed $d \mu m$ apart. Derive expressions for r, $f_1$ , and $f_2$ in terms of s, h, and d. (b) If the contact angle for the flat surface is measured to be 114°, find the apparent

surface. Given:  $s = 50 \mu \text{m}$ ,  $h = 10 \mu \text{m}$  and  $d = 150 \mu \text{m}$ .

contact angle for the patterned surface according to the Wenzel and the Cassie-Baxter equations. For the Cassie-Baxter equation, assume that the liquid covers the top surfaces of the pillars *completely*. Comment on how hydrophobicity changes with the same

(4,2)