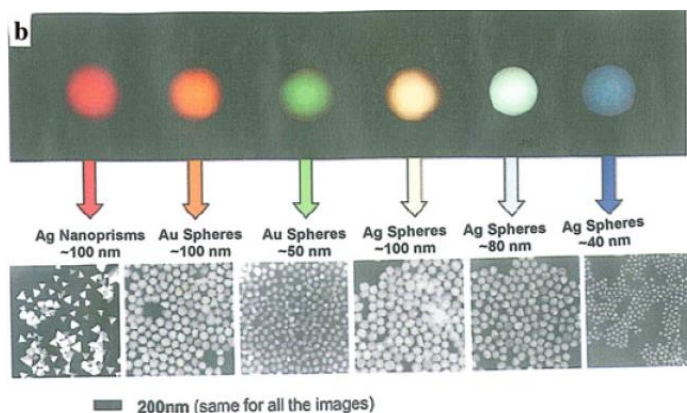


## Topics in Nanosciences\_Assignment\_24-3

### Problems

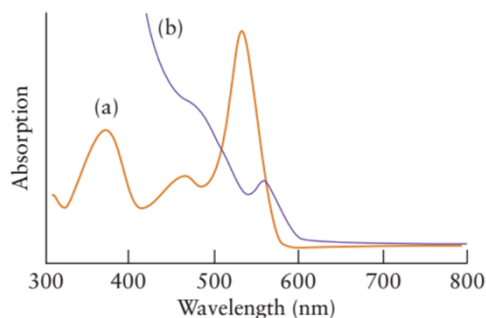
P1. Distinguish between the mechanisms responsible for the appearance of colors in the suspensions of silver or gold metal nanoparticles and semiconductor quantum dots. [6]

P2. See the adjoining Figure. From the Figure, conclude about the factors of the nanocrystals that can be controlled and manipulated to produce nanoparticles with distinct light-scattering profiles. [3]



P3. Why do very small nanoparticles (<1–2 nm) not show the LSPR? [2]

P4. Below are two absorption plots. One was obtained from a solution of an organic dye and the other from a quantum dot suspension. Which plot was obtained from which solution? Explain your reasoning. [2,2]



P5. Discuss about the conditions when maximum enhancement of the local field can be observed on a spherical plasmonic nanoparticle. [4]

P6. Discuss the roles of the plasmonic nanoparticles in the Surface-Enhanced Raman Scattering (SERS) spectroscopy. [8]

P7. Prove that the entire series of spherical fullerenes containing  $60 + (k \times 6)$  atoms, where  $k = 0, 2, 3, 4, \dots$ , etc., satisfies Euler's Theorem. [8]

P8. Given that a carbon atom can be assumed to have a diameter of 0.22 nm, estimate the diameter of a  $C_{60}$  molecule. [3]

P9. Draw the unit cell (smallest graphene sheet structure) of a (6, 6) tube and a (9, 0) tube. What kind of tubes are these? [4,4]

P10. In the following, the chiral vectors of some CNTs are given:

(i)  $c = 9 a_1$  (ii)  $c = 9 a_2$  (iii)  $c = 9a_1 + 9a_2$  (iv) (10, 10) (v)  $c = 10a_1 + 9a_2$  and (vi) (9, 7).

Arrange these tubes in the order of increasing diameter.

[12]