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Chapter 6 The Periodic Table

left to right (row) -- 7 periods in the table. Patterns within a period (6.1) -properties of elements change within a period as you move left to right. -properties within a period repeat as you move from one period to the next.

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152 Chapter 6 The Periodic Table and Periodic Law Figure 6-1. A resident of London, England invented the word smog to describe the city's filthy air, a combination of smoke and natural fog. The quality of London's air became so poor that in 1952 about 4000 Londoners died during a four-day period.

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Chapter 6 the periodic table. 12. Areas of the periodic table Three classes of elements are: 1) metals, 2) nonmetals, and 3) metalloids 1) Metals: electrical conductors, have luster, ductile, malleable 2) Nonmetals: generally brittle and non-lustrous, poor conductors of heat and electricity 13.

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The Periodic Table and Periodic Law. 150 Chapter 6. What You'll Learn. You will explain why elements in a group have similar properties. You will relate the group and period trends seen in the periodic table to the electron configuration of atoms. You will identify the s-, p-, d-, and f-blocks of the periodic table.

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Clark, Smith (CC-BY-SA 4.0) GCC CHM 130 Chapter 6: Periodic Table page 1 Chapter 6 – Periodic Table 6.1 Periodic Table Shape The current periodic table is arranged in order of increasing atomic number. This represents the increasing number of protons in the nucleus.

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174 Chapter 6 • The Periodic Table and Periodic Law Section 66.1.1 Development of the Modern Periodic Table MAIN Idea The periodic table evolved over time as scientists discovered more useful ways to compare and organize the elements.

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Study Guide - Chapter 6 – The Periodic Table and Periodic Law Section 6.1 Development of the Modern Periodic Table 1. octaves 2. eight 3. nine 4. accepted 5. Dmitri Mendeleev 6. atomic mass 7.

elements 8. atomic number 9. Henry Moseley 10. protons 11. periodic law 12. properties 13. 14.007 u

Ch 6 Study Guide answers

Chapter 6 - The Periodic Table 1. Chapter 6: The Periodic Table By Kendon Smith Columbia Central HS Brooklyn, MI 2. A. Development of the Periodic Table 1. The first periodic table: (p. 156) a. Created by Dimitri Mendeleev in the 1800's. b. Elements were listed in order of increasing atomic mass. c.

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