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Sr>Ca>Mg>Be Atomic radii increase going down a group, because successively larger valence-shell orbitals are occupied by electrons. For example, strontium has electrons in the fifth shell, which contains much larger orbitals than the fourth, third, second, or first shells.

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himself into a real jam by secretly taking Shiloh in and keeping him in a homemade pen on the ...

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A 5E lesson plan from Middle School Chemistry. Key Concepts . The attractions between the protons and electrons of atoms can cause an electron to move completely from one atom to the other.

Energy Levels, Electrons, and Ionic Bonding | Chapter 4 ...

§112.31. Implementation of Texas Essential Knowledge and Skills for Science, High School. (a) The provisions of this subchapter shall be implemented by school districts.

19 TAC Chapter 112, Subchapter C - Texas Education Agency

Engage Introduce students to the idea that electrons surround the nucleus of an atom in regions called energy levels. Review with students that in lesson two they focused on the number of protons, neutrons, and electrons in the atoms in each element.

The Periodic Table & Energy Level Models | Chapter 4: The ...

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