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4. The bond energy (in MJ/mole) is the energy required to break one mole of the specified bonds. Triple bonds share six electrons between two atoms and are stronger (have greater bond energy) than double bonds, which share four electrons between two atoms.

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Mr. Apthorp's Liberty High School chemistry page.

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158 ChemActivity 27 Intermolecular Forces Exercises Based on the data in Table 1, predict the

boiling points of a) heptane, ethanol, $\text{CH}_3\text{CH}_2\text{OH}$ c) 2-octanone, $\text{CH}_3\text{CCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ Both cis-1,2,-dichloroethylene and trans-1,2,-dichloroethylene have the same molecular formula: $\text{C}_2\text{H}_2\text{Cl}_2$. However, the cis compound has a dipole moment,

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