

Chemquest 13 Answers

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40 ChemQuest 13 Name: - Yumpu

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11. Complete the following statement: Elements ending in p1 are in column number 13 or IIIA of the periodic table and elements ending in p3 are in column number 15 or VA of the periodic table. Therefore, elements ending in p2 must be in column 14 or IVA of the periodic table. 12.

ChemQuest 15 - Yav Science

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13. If protons and neutrons have the same mass, what is the approximate mass of a proton and neutron in atomic mass units (amu)? 14. The mass of is about 14 amu. Does this agree with what you determined in questions 11 and 13? 15. The charge (in the upper right hand corner of the element symbol) is -1 for a fluorine ion. Why

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ChemQuest 12 Quantum Numbers

Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% ^{127}I , 17% ^{126}I , and 3% ^{128}I . Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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ChemQuest 8: Answer Key - St. Francis Preparatory School

(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO_2 are produced, so the number of moles of CO_2 that can be produced when 4 moles of propane combusts = $4(3) = 12$.

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13. Which of the following quantities is a constant? a) $PV + T$ c) For each trial, if you multiply the pressure and volume and then divide by the temperature you get approximately 1.18. 14. Based on your answer to question 13, verify that this equation is true: This equation is called the "Combined Gas Law".

Information: Gas Pressure

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