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Hint: add your answer to question 1 and four times your answer to question 2. 4 + 4(7) = 4 + 28 = 32 4. Verify that there are 32 electrons pictured in the drawing of CCl4. Count each dot in the picture above: there are 24. There are also 8 more electrons represented by the 4 lines (each line represents two electrons). 5.

# **ChemQuest 23: Drawing Covalent Compounds**

Hint: add your answer to question 1 and four times your answer to question 2. Verify that there are 32 electrons pictured in the drawing of CCl4. Find the sum of all the valence electrons for NF3. (Add how many valence electrons one nitrogen atom has with the valence electrons for three fluorine atoms.) ... ChemQuest 23 ...

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) ... If we react 23.0g of sodium metal with plenty of chlorine gas, calculations tell us that we should get about 58.5g of sodium chloride. ... ChemQuest 33 ...

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