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Class 12 Chemistry. Solid State NCERT InText Solution. Question: 1.1 - Why are solids rigid? Answer: The particles of solids are close packed and can only oscillate about their fixed positions. These properties make solids rigid. Question: 1.2 - Why do solids have a definite volume?

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Class 12 Chemistry. Solution In Text Questions 2. Question - 2.5 - Calculate (a) molality, (b) molarity and (c) mole fraction of KI if the density of 20% (mass/mass) aqueous KI is 1.202 g mL⁻¹. Answer: Given, 20% (mass/mass) aqueous solution of KI. This means 20 gm of KI is dissolved in 80 gm of water.

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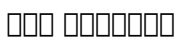


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Question 3.11: Suggest a list of metals that are extracted electrolytically. Question 3.12: Consider the reaction: $\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+ + 6\text{e}^- \rightarrow 2\text{Cr}^{3+} + 8\text{H}_2\text{O}$ What is the quantity of electricity in coulombs needed to reduce 1 mol of $\text{Cr}_2\text{O}_7^{2-}$? Question 3.14: Suggest two materials other than hydrogen that can be used as fuels in fuel cells.

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