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Acid and Base Worksheet - Answers. 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$  -  $\text{HNO}_3$  and  $\text{NO}_3^-$  make one pair.  $\text{OH}^-$  and  $\text{H}_2\text{O}$  make the other. b)  $\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{NH}_3^+ + \text{OH}^-$  -  $\text{CH}_3\text{NH}_2$  and  $\text{CH}_3\text{NH}_3^+$  make ...

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