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A pH indicator is a halochromic chemical compound added in small amounts to a solution so the pH (acidity or basicity) of the solution can be determined visually. Hence, a pH indicator is a chemical detector for hydronium ions (H_3O^+) or hydrogen ions (H^+) in the Arrhenius model. Normally, the indicator causes the color of the solution to change depending on the pH.

pH indicator - Wikipedia

Properties. Nickel(II) hydroxide has two well-characterized polymorphs, α and β . The α structure consists of $\text{Ni}(\text{OH})_2$ layers with intercalated anions or water. The β form adopts a hexagonal close-packed structure of Ni^{2+} and OH^- ions. In the presence of water, the α polymorph typically recrystallizes to the β form.

Nickel(II) hydroxide - Wikipedia

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