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24 ChemQuest 8 Name: _____ Date: _____ Hour: _____ Information: Structure of the Atom Note the following symbols: (they are not to scale) = proton (positive charge) = electron (negative charge) = neutron (no charge) The following three diagrams are hydrogen atoms: 1 1H 2 1H 3 1H The following three diagrams are carbon atoms: Notice the type of notation used for atoms: 12 13 14 6C 6 C 6C (6 ...

24 ChemQuest 8 Name: - Yumpu

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Chemquest #1: numbers in chemistry 1 chemquest #15: electron configurations and the periodic table 46 chemquest #26: electronegativity 76. Is it possible to answer this question: how many mg are in one km? . we saw in the last chemquest that scientific notation can be a very nice way of getting rid of mythological - solely.

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ChemQuest 8: Answer Key - St. Francis Preparatory School

Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 127.1, 17% 126.1, and 3% 128.1. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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ChemQuest 15 Name: _____ Date: _____ Hour: _____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

ChemQuest 15 - Yav Science - Home

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