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f. mass of 6 mol C = 6 (12.01 g) = 72.06 g mass of 6 mol H = 6 (1.008 g) = 6.048 g mass of 1 mol O = 16.00 g = 16.00 g molar mass of C₆H₆O = 94.11 g 21. a. molar mass of SO₃ = 80.07 g 49.2 mg SO₃ 1 g 1000 mg 1 mol 80.07 g = 6.14 10⁻⁴ mol SO₃ b. molar mass of PbO₂ = 239.2 g 7.44 x 10⁴ kg PbO 2 1000 g 1 kg 1 mol 239.2 g = 3.11 10⁵ mol PbO 2

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Modern Chemistry 33 Chapter Test Name Class Date Chapter Test B, continued 15. The energy state of an atom is called its ground state. 16. The number of waves that pass a point in one second is called. 17. When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

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