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DRAFT $5/7/2014\ 10:27:00\ PM\ 4$ Question 4 (a) P V = n R T (1.04)(1.00) = n (0.08206)(1100) n = 0.0115 moles (1) (b) Disagree. The reaction is at equilibrium so there must be some CaCO 3(s) left in the container. (It is entirely possible to answer this via stoichiometry, proving that if there was total decomposition, that the moles and

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January 2014 Chemistry Regents #1-10. Highlight to reveal answers and explanations . Questions 1-10 Questions 11-20 Questions 21-30 Questions 31-40 Questions 41-50. Questions 51-54 Questions 55-57 Questions 58-61 Questions 62-63 Questions 64-65 Questions 66-70.

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answer with justification. (b) As the mass of the Sn electrode decreases, where does the mass go? The atoms on the Sn electrode are going into the solution as Sn2+ ions. 1 point is earned for the correct answer. (c) In the expanded view of the center portion of the salt bridge shown in the diagram below, draw and label

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