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SOLUTIONS - CHAPTER 6 Problems (Problems 1-11 cover material for Exam 2) 1) Which of the following bonds will have the smaller value for bond length? a) C-O or C=O b) N=N or N(N c) C-O or C-S. a) C=O For bonds between the same two atoms, the higher the bond order the shorter the bond.


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Chapter 1 Case Solutions Case 1-1 The one-page opening statement to the board's audit committee should focus on the pending independent opinion of Deloitte & Touche. The explanatory paragraph of the audit report expressing doubt as to whether GM is a going concern far outweighs the auditor's unqualified opinion that the financial statements conform to GAAP in all material respects.

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84) Using table 16.1, pick an indicator for use in the titration of each of the following bases with a strong acid. a) CH₃NH₂ b) NaOH c) C₆H₅NH₂ . 86) Write balanced equations and expressions for K_{sp} for the dissolution of each ionic compound. a) CaCO₃ b) PbCl₂ c) AgI

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