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### **Chemquest 18 Answer Key**

56 ChemQuest 18 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Information: Ions Figure 1: Below are four Bohr diagrams of atoms and ions. The two diagrams on the left are atoms; the two on the right are ions. 12 protons 12 protons Atom A Ion of Atom A 17 protons 17 protons Atom B Ion of Atom B  
Critical Thinking Questions 1.

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48 18. There are three major divisions on the periodic table: the "s block", the "d block" and the "p block". Where are these blocks of elements located?

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18.9985 amu Ion 12 protons 12 neutrons 24 12 Mg 23.9968 amu 10 neutrons mass = 19 18.9980 amu Atom 12 protons 12 neutrons 12 Mg mass = 23.9978 amu Critical Thinking Questions 10. What is structurally different between an atom and an ion? Note: This is the ONLY structural difference between an atom and an ion. 11.

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a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO<sub>2</sub> are produced, so the number of moles of CO<sub>2</sub> that can be produced when 4 moles of propane combusts =  $4(3) = 12$ .

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