

Chemquest 28 Answers

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Answer the questions that follow based on this chemical equation: $\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$. a) Why can't NaMg be produced? Na + cannot bond with Mg²⁺ because they are both positive. b) Why can't NaCl₂ be produced? Na + only requires one Cl⁻.

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Answer the questions that follow based on this chemical equation: $\text{Na} + \text{MgCl}_2 \rightarrow \text{NaCl} + \text{Mg}$. Why can't NaMg be produced? (Hint: Think about the charges) Why can't NaCl₂ be produced? Are NaCl and Mg the only products that can be produced? ... ChemQuest 28 ...

ChemQuest 28 - AP Chemistry

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56 ChemQuest 18 Name: _____ Date: _____ Hour: _____ Information: Ions Figure 1: Below are four Bohr diagrams of atoms and ions. The two diagrams on the left are atoms; the two on the right are ions. 12 protons 12 protons Atom A Ion of Atom A 17 protons 17 protons Atom B Ion of Atom B Critical Thinking Questions 1.

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View CHEMICAL_REACTIONS_Chemquest_Answers.ppt from ENGLISH 1201 at Northmont High School. CHEMICAL REACTIONS ChemQuest 28 $\text{LiBr} + \text{P} \rightarrow \text{Li}_3\text{P} + \text{Br}_2$ Why is bromine written as Br_2 on the right side? Bromine

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Chemquest #1: numbers in chemistry 1 chemquest #15: electron configurations and the periodic table 46 chemquest #26: electronegativity 76. Is it possible to answer this question: how many mg are in one km? . we saw in the last chemquest that scientific notation can be a very nice way of getting rid mythological - solely.

Electronegativity Chemquest 26 Answer Key, Chemquest 26 ...

Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% ^{127}I , 17% ^{126}I , and 3% ^{128}I . Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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