

## *Classifying The Elements Answer Key*

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### **Classifying The Elements Answer Key**

6.2 Classifying the Elements. Prentice Hall Chemistry, 2005. ... Into what four classes can elements be sorted based on their electron configurations? representative elements, noble gases, transition metals, and inner transition metals.

### **6.2 Classifying the Elements Flashcards | Quizlet**

Summary Sheet on last page. CLASSIFICATION OF MATTER AND ELEMENTS – ANSWER KEY I. Find the odd one out and give reasons for your answer a. Gold , Silicon, Germanium Gold is the odd one out as it is a metal while silicon and germanium are metalloids b.

### **Answer - Elements - Students Practice**

section 6.1 organizing the elements (pages 155-160) This section describes the development of the periodic table and explains the periodic law. It also describes the classification of elements into metals, nonmetals, and metalloids.

### **SECTION 6.1 ORGANIZING THE ELEMENTS (pages 155-160) (page 155)**

The Group 1A elements are called alkali metals, and the Group 2A elements are called alkaline earth metals. The name alkali comes from the Arabic al aqali, meaning “the ashes.” Wood ashes are rich in compounds of the alkali metals sodium and potassium. Some groups of non- metals also have special names. halogens.

### **6.2 Classifying the Elements 6 - Henry County School District**

6.2 Classifying the Elements > 13 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. . Electrons play a key role in determining the ...

### **What can you learn about each element from the periodic table?**

There are six periods in a periodic table. Most of the elements in the periodic table are metals. The elements within a period have similar properties. :1 t t 1\* k]; “M I Pa C Matching Match each description in Column B to the correct term in Column A. Column A Column B b 14. metals lie vertical column of elements in the periodic table 15.

### **6.1 and 6.2 Section Review - groups The periodic table ...**

Section 6.2 Classification of the Elements In your textbook, read about organizing the elements by electron configuration. Use the periodic table on pages 156-157 in your textbook to match each element in Column A with the element in Column B that has the most similar chemical properties. Column A Column B 1. arsenic (As) 2. bromine (Br)

### **Section 6.2 Classification of the Elements**

Classification of Elements. Classify each element as a metal, non-metal or metalloid based on it's position on the Periodic Table. Also classify the family it is in. M = Metal NM = Non-Metal MO = Metalloid. A.

### **Classification of Elements Classify each element as a ...**

6.2 Classification of the Elements in the Periodic Table. Organizing P.T. by Electron Configuration The s, p, d, and f blocks. STUDY. PLAY. Terms in this set (...) Electron Configuration & Groups. The number of valence electrons corresponds to the Group # for Representative Groups (1,2;13-18)

### **6.2 Classification of the Elements in the Periodic Table ...**

To preview this answer key, ... Classifying Matter Answer Key. 1. ... Homogeneous; 3. Elements and compounds that have fixed compositions can be called mixtures. substances. alloys. aqueous solutions. 4. A compound differs from a mixture in that a compound always ...

### **Classifying Matter Answer Key - HelpTeaching.com**

Group 18 elements have both their s orbitals and p orbitals completely filled with electrons. This configuration is very stable, thus, the group 18 elements are very unreactive. 23. 40 24. groups 3-8

25. The seven f orbitals hold a maximum of 14 electrons. 26. 1s 2s 2p 3s 3p 4 Section 6.3  
Periodic Trends 1. c 2. c 3. d 4. b 5. a 6. a 7.

### Ch 6 Study Guide answers

Section 6.2 Classification of the Elements In your textbook, read about organizing the elements by electron configuration. Use the periodic table on pages 156–157 in your textbook to match each element in Column A with the element in Column B that has the most similar chemical properties.  
Column A Column B 1. arsenic (As) 2. bromine (Br) 3 ...

### Section 6.2 Classification of the Elements

Answer the following questions in the space provided. 19. ... • Classify elements based on electron configuration ... Key to periodic table A Key to periodic table B Thousands of periodic tables have been published since Mendeleev published his table. Each one is a little different from the rest.

### 05 CTR ch06 7/9/04 3:27 PM Page 131 ORGANIZING THE ELEMENTS 6

ber of elements? How would chemists know when they had discovered all the elements? To begin to answer these questions, chemists needed to find a logical way to organize the elements. Chemists used the properties of elements to sort them into groups. In 1829, a German chemist, J. W. Dobereiner (1780–1849), published a classification system.

### 6.1 Organizing the Elements 6

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