Chemquest 15 Answers

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15. Complete the following: Elements ending in d2 are in column number 4 or IVB and those ending in d4 are in column number 6 or VIB. Therefore, elements ending in d3 are in column number 5 or VB. 7 Elements ending in d must be in column number 9 or VIIIB. 16. Notice from question 6 that an element that ends in 3s is in the 3rd row. Also, from ...

ChemQuest 15 - Yav Science

46 ChemQuest 15 Name: ____ Date: ____ Hour: ____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready.

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15. The charge (in the upper right hand corner of the element symbol) is –1 for a fluorine ion. Why isn't it +1 or some other number? 16. What is the charge on every atom? Why is this the charge? 17. How do you determine the charge on an ion? 18. An oxygen ion has a –2 charge. (Use your periodic table if necessary)

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