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2) 2.50 moles of O₂ ANSWER: 56.0 L 3) 0.850 moles of N₂ ANSWER: 19.0 L 4) 3.75 moles of CO₂ ANSWER: 84.0 L 5) 0.70 mole of NH₃ ANSWER: 16 L 6) 12.0 g of H₂ ANSWER: 133 L 7) 125. g of O₂ ANSWER: 87.5 L 8) 56.0g of N₂ ANSWER: 44.8 L 9) 70. g of CO₂ ANSWER: 36. L 10) 20. g of NH₃ ANSWER: 26 L Chemistry IF8766 Page 51 Instructional Fair, Inc.

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Chemistry IF8766 OInstructional Fair, Inc. STOICHIOMETRY: MOLE-MOLE PROBLEMS I. N₂ + 3H₂ Name How many moles of hydrogen are needed to completely react with two moles of nitrogen? 2.0 + 302 How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?

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ANSWER: 0.99 mole 5) 53 g of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ANSWER: 0.21 mole Directions: determine the number of grams in each of the quantities below. 6) 2.8 moles of NaCl ANSWER: 160 g 7) 0.75 moles H_2SO_4 ANSWER: 74 g 8) 2.75 moles of KMnO_4 ANSWER: 435. g 9) 0.55 moles of KCl ANSWER: 41. g 10) 2.3 moles of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ ANSWER: 570 g Chemistry IF8766 ...

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MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in 1.5×10^{25} molecules of CO_2 ? 1.110g 1 mol 44.0 g 6.0210molecules 1.510molecules CO_2 mol 3 23 2 25!=!!!! b) What volume would the CO_2

KEY - CP - Mixed Mole Problems

TELIER'S PRINCIPLE Name telier's Principle states that when a system at equilibrium is subjected to a stress, the m will shift its equilibrium point in order to relieve the stress.

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