

## *Chemfax Leftover Aluminum Wire Answer*

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### **Chemfax Leftover Aluminum Wire Answer**

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### **Chemfax Leftover Aluminum Wire Answer - getpremise.com**

So we did a lab in class and I need help finding the final answers. The data table says: the initial mass of  $\text{CuCl}_2(\text{g})$  was .52g Initial mass of aluminum wire (g) was .74 The mass of "leftover" aluminum wire (g) was .68 The actual mass of aluminum reacted (g) was .6 The actual mass of recovered dry copper (g) .22 I need help finding : A) The moles of  $\text{CuCl}_2$  B) Moles of aluminum that should react ...

### **Chemistry lab help please please help D;? | Yahoo Answers**

With the Leftover Aluminum Wire Stoichiometry Laboratory Kit, students perform a reaction between aluminum wire and a copper salt. Students find the amount of aluminum that should react and compare to the amount of "leftover" aluminum.

### **Leftover Aluminum Wire—A Stoichiometry Super Value Kit**

Rinse the aluminum wire with tap water and dry the wire with a paper towel. Once dry, measure and record the actual "leftover" mass of the aluminum wire in the data table. Dispose the used aluminum wire by rolling into a ball (to eliminate sharp ends) and place in the trash. Rinse the graduated cylinder and test tube with tap water.

### **LEFTOVER ALUMINUM WIRE - A Stoichiometry Lab**

Leftover Aluminum Wire. CAUTION:  $\text{CuCl}_2$  solution is toxic by ingestion and is a body tissue irritant. Avoid contact with body tissues. Wear safety goggles, apron, and gloves for this activity.

### **Leftover Aluminum Wire | Stoichiometry (1.3K views)**

Leftover Aluminum Wire Pre-Lab. STUDY. PLAY. Terms in this set (...) What is the purpose of the lab? The purpose of the lab is to compare the actual mass of aluminum reacted and the actual mass of copper produced with that predicted from the balanced chemical equation using stoichiometric calculations.

### **Leftover Aluminum Wire Pre-Lab Flashcards | Quizlet**

Lab 14 Aluminum Leftovers – A Quantitative Analysis. Honors Chemistry. ... Using your answer from b, calculate the number of moles of water can theoretically be produced from the number of moles of  $\text{O}_2$  you have. ... Initial mass of aluminum wire (g) Mass of left over aluminum wire (g) Actual mass of aluminum reacted (g) Mass of watch glass and ...

### **Lab format - Hudson City Schools / Homepage**

Moles of Aluminum and Copper Moles of Al Moles of Cu The materials for Aluminum and Copper are available as a traditional Student Laboratory Kit from Flinn Scientific, Inc. Catalog No. Description AP4678 Leftover Aluminum Wire—Student Laboratory Kit Consult your Flinn Scientific Catalog/Reference Manual for current prices.

### **Aluminum and Copper - Flinn Sci**

- After performing this demonstration to the class, have students perform the Leftover Aluminum Wire Stoichiometry Lab, available as a student laboratory kit, Catalog No. AP4678, from Flinn Scientific. This lab uses the same reaction to teach concepts of moles, limiting reactants, and stoichiometry. Discussion

### **Foiled Again - Flinn Scientific**

Chemfax is a Calgary, Alberta based company that has grown tremendously over the past decade.

Chemfax proudly operates out of a 60,000 sq. ft. State-of-the-Art facility which includes one of the largest Class 1, Division 1, flammable liquid handling areas in Canada.

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