

## *Chapter 17 Thermochemistry Section Review Answers*

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### **Chapter 17 Thermochemistry Section Review**

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Chapter 17 Thermochemistry183 SECTION 17.1 THE FLOW OF ENERGY ... CHAPTER 17,Thermochemistry (continued) 6. In thermochemical calculations, is the direction of heat flow given from the point of view of the system, or of the surroundings? 7. What universal law states that energy can neither be created nor destroyed

### **SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505-510)**

Chapter 10 - moles (handouts) Chapter 11 - reactions (handouts) Chapter 12 - stoichiometry (handouts) Chapter 13 - states of matter (handouts) Chapter 17 - thermochemistry (handouts) Chapter 18 - reaction rates (handouts) Chapter 19 - acids, bases, and salts (handouts) Material Science Schedule. Previous weeks schedule; Chemistry Basics ...

### **Science / Chapter 17 - thermochemistry (handouts)**

Chapter 17 Thermochemistry 429 Section Review Objectives • Explain the relationship between energy, heat, and work • Distinguish between exothermic and endothermic processes • Distinguish between heat capacity and specific heat Vocabulary Key Equations and Relationships • 1 Calorie 1 kilocalorie 1000 calories • 1 J 0.2390 cal and 4.184 J 1 cal • C m

### **05 CTR ch17 7/12/04 8:15 AM Page 429 THE FLOW OF ENERGY ...**

Ch 17 Thermochemistry Practice Test Matching Match each item with the correct statement below.  
a. calorimeter d. enthalpy b. calorie e. specific heat c. joule f. heat capacity \_\_\_\_ 1. quantity of heat needed to raise the temperature of 1 g of water by 1°C \_\_\_\_ 2. SI unit of energy \_\_\_\_ 3.

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THERMOCHEMISTRY SECTION 17.1 THE FLOW OF ENERGY-HEAT AND WORK (pages 505-510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. ~ Energy Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that ... Chapter 17 Thermochemistry ...

### **THERMOCHEMISTRY - d39smchmfovhlz.cloudfront.net**

thermochemistry chemical potential energy heat • system surroundings law of conservation of ... 530 Chapter 17 . Name Class CALCULATING HEATS OF REACTION 17.4 Section Review Objectives Apply Hess's law of heat summation to find enthalpy changes for chemical and physical processes

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17.3 15. 17.5 16. 17.5 17. Section 17.4 Hess's Law Reactants Products The change in enthalpy is the same whether the reaction takes place in one step or a series of steps. The change in enthalpy,  $\Delta H$ , is independent of pathway.

### Chapter 17 thermochemistry sections 17.3 & 17.4 - SlideShare

Chapter 17 Review "Thermochemistry" Chapter 17 Review What would likely happen (how would it feel) if you were to touch the flask in which an endothermic reaction were occurring? Standard conditions of temperature and pressure for a thermochemical equation are \_\_\_ and \_\_\_ kPa.

### Chapter 17 Review "Thermochemistry" - DBCS Mrs. Marshall

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### Chapter 17 Thermochemistry Section Review Answers

Chapter 17: Thermochemistry. Section 1: The Flow of Energy – Heat and Work. What is thermochemistry? The study of the energy changes that occur in changes of state or during chemical reactions.

## Chapter 17 Thermochemistry Section Review Answers

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