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Radioactivity •Radioactivity is the process by which nuclei emit particles and rays as they break down. •The name of the penetrating rays emitted by a radioactive source is called radiation. •A radioactive isotope is an unstable atom which breaks down on its own, releasing energy and/or

Chapter 25 - Nuclear Chemistry

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806 Chapter 25 Nuclear Chemistry Figure 25-2 Both Pierre and Marie Curie played important roles in founding the field of nuclear chemistry. Marie Curie went on to show that unlike chemical reactions, radioactivity is not affected by changes in physical conditions such as temperature and pressure. She is the only person in history to receive

Chapter 25: Nuclear Chemistry - Jayne Heier

Answer Key for Nuclear Chemistry Worksheet #1: Nuclear Decay Processes Chem 160 – K. Marr Key Questions 1. What does the symbol “ e” tell the reader? (i.e., What does the superscript 0 mean? What can a subscript -1 possibly mean? Why is the beta particle symbolized with an e?)

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How does nuclear radiation affect atoms? Nuclear radiation can ionize atoms.15. Is the following sentence true or false? One potential danger of radon gas is that prolonged exposure to it can lead to lung cancer. trueDetecting Nuclear Radiation (page 297)16. Name two devices that are used to detect nuclear radiation.a. Geiger counters b.

Chapter 10 Nuclear Chemistry Section 10.1 Radioactivity ...

9.5 Lesson9.5TransitionElements 57 9.6 Lesson9.6LanthanideandActinideSeries 57 10 Trends on the Periodic Table Worksheets 59

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