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The first time seeing formulas and names of ionic compounds can be confusing. However, upon further investigation, there are actually simplistic formulated steps to do so. Discover a systematic method of writing formulas for ionic compounds without the need to experimentally determine each formula using cation and anion cards.

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8. Write the formula for the ionic compound, using subscripts to indicate the number of each kind of ion used (from Columns 2 and 3). Follow the rules for writing ionic formulas listed in the background material. In our example, the formula is AlCl 3. Record the chemical formula for the ionic compound in Column 6 of the data table. 9.

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2 and 3). Follow the rules for writing ionic formulas listed in the background material. In our example, the formula is AIC13. Record the chemical formula for the ionic compound in Column 6 of the data table. 7. Write the name of the ionic compound following the rules for naming ionic

compounds discussed in the background infor- mation.

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