Chemfiesta Ph Practice Worksheet Answers

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How many grams of HCl would need to be added to that swimming pool to bring the pH down from 7 to 4? (Assume the volume of the HCl is negligible) pH practice - Answers. 1) What is the pH and pOH of a 1.2×10 -3 HBr solution? pH: 2.9 pOH: 11.1. 2) What is the pH and pOH of a 2.34×10 -5 NaOH solution? pOH: 4.6 pH: 9.4×10 -1 PH and pOH of a 2.34×10 -1 PH and pOH of a 2.

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