

Chemistry If0235 Answers

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Balancing Chemical Equations -Answer Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

Balancing Chemical Equations Answer Key

Name: _____ The Mole and Volume Directions: For gases at STP (273K and 1.00 atm pressure), one mole occupies a volume of 22.4L. What volume will the following quantities of gasses occupy at STP? 1) 5.00 mole of H

The Mole and Volume - Ms. Agostine's Chemistry Page

Answers: COMBINED GAS LAW Remember to convert all temperatures to Kelvin. P 1 V 1 T 1 P 2 V 2 T 2 1 1.5 atm 3.0 L 20. C 293K 2.5 atm 1.9 L 30. C 303K 2 720 torr 256 mL 25 C 298 K 8.0x10² torr 250 mL 50. C 323 K 3 600. mmHg 2.5 L 22 C 295 K 760 mmHg 1.8 L 270 K 4 1.2 atm 750 mL 0.0 C 273.0 K 2.0 atm 500. mL 25 C 298 K 5 95 kPa 4.0 L

Answers: COMBINED GAS LAW - newburyparkhighschool.net

Place your final answer in the FORMULA MASS COLUMN. CHEMISTRY COMPUTING FORMULA MASS WORKSHEET Problem Set-up example: Find the formula mass of Ca(NO₃)₂ Ca: 1 x 40.1 = 40.1 ... CHEMISTRY MOLAR RATIOS WORKSHEET 2 1 1 2 2 2 2 3 3 2. The Mathematics of Chemical Equations

CHEMISTRY COMPUTING FORMULA MASS WORKSHEET - ISD 622

Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead. $\text{Zn} + \text{Pb(NO}_3)_2 \rightarrow \text{Zn(NO}_3)_2 + \text{Pb}$ Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas. $2\text{AlBr}_3 + 3 \text{Cl}_2 \rightarrow 2\text{AlCl}_3 + 3\text{Br}_2$ Single Replacement 3.

Word Equations Answer Key Zinc and Lead (II) nitrate react ...

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