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In-chapter exercises. E9.1: The electrophilic carbon is primary, and a carbocation intermediate would not be stabilized by resonance. Therefore it is reasonable to predict an S_N2 mechanism.. Note: in later chapters, especially in chapters 10 and 12, we will see many more reactions involving ATP. As we will see then, the SAM-forming reaction above is somewhat unusual – most reactions with ...

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144 Chemistry: Matter and Change • Chapter 9 Solutions Manual CHAPTER 9 SOLUTIONS MANUAL 24. Challenge Al(s) Pb(NO₃)₂(aq) 0 Yes. Al is above Pb in the activity series. 2Al(s) 3Pb(NO₃)₂(s) 0 3Pb(s) 2Al(NO₃)₃(aq) Write the balanced chemical equations for the following double-replacement reactions. LiI(aq) AgNO₃(aq) 25.

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Q: What are the concentrations of each of the ions in a saturated solution of PbI₂, given that the K_{sp} of PbI₂ is 8.5 x10⁻⁹ at 25degC. If 5g pbl₂ are dissolved in water to make 1L of solution at 25degC, would it be sat, unsat, or supersat

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MANUAL Section 12.3 Liquids and Solids pages 415–424 Section 12.3 Assessment page 424 18. Contrast the arrangement of particles in solids and liquids. The particles are closer together in solids than in liquids because of intermolecular attractions.

Chemistry Matter And Change Chapter 9 Solutions Manual

Instructor Solutions Manual for Chemistry for Engineering Students, 3rd Edition 9-2 ANS In the first case q (heat) is -73 J and in the second case $q = -3\text{ J}$. It would appear that heat is not added to the system in either case. 9.28 When an electrical appliance whose power usage is X watts is run for Y seconds, it uses $X Y$ joules of energy. The energy unit used by electrical utilities in ...

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