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Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 9 1. Draw Lewis structures for each of the following structures and assign formal charges to each atom: a) SF 2 20 electrons 0 0 0 b) NH 2 OH (N and O are bonded to one another) 14 electrons 0 0 0 0 0 0 PO 4 3-

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Chemistry 101 ANSWER KEY REVIEW QUESTIONS Chapter 10 1. Draw Lewis structures and determine the molecular geometry of each molecule or ion shown below: A)

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Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 10 1. Draw Lewis structures and determine the molecular geometry of each molecule or ion shown below: A) ClO 2 – 20 electrons Bent (from tetrahedral) B) ICl 3 28 electrons T-shaped (from trigonal bipyramidal) C) TeF 4 34 electrons See-saw (from trigonal bipyramidal) 2.

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Academic Chemistry Mr. Gensits Section Review 7.2 Answer Key 1. electrostatic forces 2. oppositely 3. ionic bonds 4. neutral 5. formula unit 6. crystals 7. high 8. large 9. stable 10. molten 11. AT 12. ST 13. AT 14.

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The goal of chemistry is to accumulate knowledge. _____13. Biochemistry is the study of the chemistry of living organisms. 10 9 8 7 6 5 4 3 2 1 • chemistry • organic chemistry • inorganic chemistry • analytical chemistry • physical chemistry • biochemistry CHEMISTRY SECTION REVIEW 1.1

1.1 Chemistry Section Review - LPS

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