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CHAPTER 5 Electrons in Atoms + KEY Chemistry: Matter and Change 1 Supplemental Problems 1. Orange light has a frequency of  $4.8 \times 10^{14} \text{ s}^{-1}$ . What is the energy of one quantum of orange light?

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ation shown in Figure 5-5, you should note that energy increases with increasing frequency. Thus, looking back at Figure 5-3, the violet light, with its greater frequency, has more energy than the red light. This relationship between frequency and energy will be explained in the next section. 120 Chapter 5 Electrons in Atoms Figure 5-4

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