

## ***Chapter 7 Review Chemical Formulas And Compounds Answer Key***

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### **Chapter 7 Review Chemical Formulas**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 4 SHORT ANSWER

Answer the following questions in the space provided. 1. Write empirical formulas to match the following molecular formulas: CH<sub>3</sub>O<sub>2</sub> a. C<sub>2</sub>H<sub>6</sub>O<sub>4</sub> N<sub>2</sub>O<sub>5</sub> b. N<sub>2</sub>O<sub>5</sub> HgCl<sub>2</sub> c. Hg<sub>2</sub>Cl<sub>2</sub> CH<sub>2</sub> d. C<sub>6</sub>H<sub>12</sub> 2. C<sub>4</sub>H<sub>8</sub> A certain hydrocarbon has an empirical formula of CH<sub>2</sub> and a molar mass of 56.12 g/mol.

### **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium phosphate SnS<sub>2</sub> d. tin(IV) sulfide HNO<sub>2</sub> e. nitrous acid 2.

### **7 Chemical Formulas and Chemical Compounds**

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER

Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on ...

### **7 Chemical Formulas and Chemical Compounds**

The (IO<sub>3</sub>)<sub>2</sub> c. 2Fe(NO<sub>2</sub>)<sub>2</sub> part of the substance represents two iodate (IO<sub>3</sub>)<sup>-2</sup> ions, with a total charge of 2-. Therefore, (IO<sub>3</sub>)<sub>2</sub>Br<sub>4</sub> the nickel (Ni) ion has a 2+ charge. The entire representation is thus four formula units of nickel(II) iodate. 3 4 Chapter 7 Chemical Formulas 59 Concept Review and Chapter Review 21.

### **Chapter 7 Review Answers - CHAPTER CHEMICAL FORMULAS IA ...**

Modern Chemistry 57 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Label each of the following statements as True or False: \_\_\_\_ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. \_\_\_\_ b.

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In Chapter 7, students will learn how to write the chemical formula of any given compound, either ionic or molecular, given that compound's scientific name (or vice versa).

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Chapter 7 Significance of a Chemical Formula •A chemical formula indicates the relative number of atoms of each kind in a chemical compound.

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### Chapter 7 Review Chemical Formulas And Compounds Section 2

Modern Chemistry 8 Chemical Bonding CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Write empirical formulas to match the following molecular formulas: \_\_\_\_ a.  $C_2H_6O_4$  \_\_\_\_ b.  $N_2O_5$  \_\_\_\_ c.  $Hg_2Cl_2$  \_\_\_\_ d.  $C_6H_{12}$

### CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds

The Chemical Formulas and Chemical Compounds chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with chemical compounds and ...

### Holt McDougal Modern Chemistry Chapter 7: Chemical ...

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: a. S in  $H_2SO_3$  b. S in  $MgSO_4$  c. S in  $K_2S$  d. Cu in  $Cu_2S$  e. Cr in  $Na_2CrO_4$  f. N in  $HNO_3$  g. C in  $(HCO_3)$  h. N in  $(NH_4)$  2. a.

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Chapter 7 Significance of a Chemical Formula •A chemical formula indicates the relative number of atoms of each kind in a chemical compound.

### Chemical Formulas and Chemical Chapter 7 Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER 1. In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) (b) (c) (d) 2. Answer the following questions in the space provided.

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chemical formulas. In this chapter, you will be introduced to some of the rules used to identify simple chemical compounds. Significance of a Chemical Formula Recall that a chemical formula indicates the relative number of atoms of each kind in a chemical compound. For a molecular compound, the

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