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Ionic bond because the atom with high electronegativity will attract an electron away from the atom with low electronegativity. Example: NaF (between a metal and a nonmetal) 4. What type of bond (covalent or ionic) would you expect to form between two atoms of somewhat high electronegativity Explain and give an example.

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Information: Definition of Electronegativity. Electronegativity. is a measure of how much an atom attracts an electron. The higher the electronegativity, the greater the atom's attraction for electrons. Atoms that become negative ions have a much greater electronegativity than atoms that become positive ions. ... ChemQuest 26 ...

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