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Problem Example 1. The Normal Saline solution used in medicine for nasal irrigation, wound cleaning and intravenous drips is a 0.91% (w/v) solution of sodium chloride in water. How would you prepare 1.5 L of this solution? Solution: The solution will contain 0.91 g of NaCl in 100 mL of water, or 9.1 g in 1 L. Thus you will add $(1.5 \times 9.1\text{g}) = 13.6\text{ g}$ of NaCl to 1.5 L of water.

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Once microorganisms have started to grow and become physiologically active they usually produce water as an end product of respiration. Thus they increase the water activity of their own immediate environment so that eventually micro-organisms requiring a high a_w are able to grow and spoil a food which was initially considered to be microbiologically stable.

Growth of Micro-Organisms in Food | Microbiology

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