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Page 2 of 11 Chem 121 Final Exam (2) 1) The product of  $(0.023023 \times 4.531)$  has 2.3 subtracted from it. Correct answer (sig. figs.) in the box. (2) 2) Write down the name (not the symbol) of any transition metal element. (2) 3) The density of aluminum metal is  $2.70 \text{ g/cm}^3$ . Calculate the volume of one mole of aluminum. (3) 4) The dianion of an unknown element contains 54 electrons and 75 neutrons.

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CHEM121 Final Exam Study Guide page 1 of 11 CHEM 121: Final Exam Study Guide Chapter 1 • Know the Scientific method • Know the definitions for hypothesis, scientific law, and scientific theory. Chapter 2 • Length, mass, weight, volume - Know  $1\text{ cm}^3 \equiv 1\text{ mL}$  and  $1\text{ dm}^3 \equiv 1\text{ L}$  • Significant figures and Scientific notation

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Page 1 of 13 Chem 121 Final Exam (2) 1) A cube measures 3.21 cm on one side. Calculate its volume in liters ( $\text{cm}^3 = \text{mL}$ ) and put the answer in the box. (2) 2) The charge on a phosphorus atom is neutral when making 3 or 5 bonds. Yet is a positively charged when making 4 bonds. Account for this behaviour using Lewis structures.

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