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### Chemistry Chapter 12 Stoichiometry Assessment Answers

Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a.  $\text{Na}_2\text{CO}_3 + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{CO}_3(\text{s})$  b.  $2\text{Na}_3\text{PO}_4 + 3\text{Pb}(\text{NO}_3)_2 \rightarrow 6\text{NaNO}_3 + \text{Pb}_3(\text{PO}_4)_2(\text{s})$  2. Sodium hydroxide reacts with calcium chloride to form sodium chloride and solid calcium hydroxide. 3. Mixings d, n, and o all gave no visible

### Chapter 11 Small-Scale Lab

The magic of chemistry happens in this unit! We will learn all about moles, a measure of chemical quantity. We will see how stoichiometry can be used to predict yield in a chemical reaction while doing a lot of hands on chemical reactions in the unit!

### McLaughlin, Kimberly / Stoichiometry

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