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718 Chapter 18 Nuclear Chemistry + + + + Energy p + p + n + n 24 He2+ For many of the lighter elements, the possession of an equal number of protons and neutrons leads to stable atoms. For example, carbon-12 atoms, 6C 12, with six protons and six neutrons, and oxygen-16 atoms, 8O 16, with eight protons and eight neutrons, are both very stable.

#### Chapter 18 NuClear Chemistry - An Introduction to Chemistry

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#### Chemistry 151. General Chemistry I - chymist.com

At 25°, the free energy of formation of gaseous water is -229 kJ/mol. Calculate  $\Delta G$  for the following reaction if the hydrogen is supplied at 4.00 atm and the oxygen is supplied at 3.00 atm, while the water produced is at 1.00 atm pressure.

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that are introduced in this section. Each blank can be completed with a term, short phrase, or number. Chemistry is the study of the of matter and the 1. that matter undergoes. Chemistry has traditionally been 2. divided into areas of study. Organic chemistry is the study 3. of substances that contain, while inorganic chemistry is 4.

#### 1.1 Chemistry Section Review - LPS

chemistry. From aluminum to xenon, we explain the properties and composition of the substances that make up all matter. Our study guides are available online and in book form at barnesandnoble.com.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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