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Review of Chemistry of Matter (Chapter 2,& 3 Test Review)ANSWERS Directions: Review and check all your answer with this answer sheet to get ready for the test on Chemistry of Matter (Chapter 2 & 3). Enjoy. 1. What tinny, tiny things make up all the elements on the periodic table?

### Review of Chemistry of Matter (Chapter 2,& 3 Test Review ...

AP Chemistry - Practice Test: Chapter 3 Name\_\_\_\_\_ MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1)Silver has two naturally occurring isotopes with the following isotopic masses: 107 47 Ar 107 47 Ar 106.90509 108.9047 The average atomic mass of silver is 107.8682 amu.

### AP Chemistry - Practice Test: Chapter 3 Ar

View Test Prep - SCH4U Chapter 3 Chapter Test Answer Key from CHEMISTRY SCH4U at St Francis Xavier Secondary School. BLM 3-6: Chapter 3 Test Answer Key 1. d 2. b 3. c 4. d 5. a 6. c 7. d 8. c 9. a

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AP Chemistry Chapter 3 Review Questions Multiple-choice exercise. Choose the correct answer for each question. You will need a calculator and a periodic table. Show all questions <= => Butane, CH<sub>3</sub>-CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>3</sub>, is a hydrocarbon fuel used in lighters. How many moles of molecules are there in a 350. gram sample of butane?

### AP Chemistry Chapter 3 Review Questions - ScienceGeek.net

These are the answers and explanations to the practice test on Chapters 1 - 3, which can be found here: <https://goo.gl/NgVq75>

### Chapters 1 - 3 Practice Test

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**Holt Chemistry Chapter 3: Atoms and Moles - Practice Test ...**

Chemistry 101. ANSWER KEY. 1. REVIEW QUESTIONS. Chapter 10. 1. Draw Lewis structures and determine the molecular geometry of each molecule or.

**Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 3 1 ...**

AP Chemistry-Chapter 3 MC Practice Questions Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. Combustion analysis of 0.800 grams of an unknown hydrocarbon yields 2.613 g CO<sub>2</sub> and 0.778 g H<sub>2</sub>O. What is the percent composition of the hydrocarbon?

**AP Chemistry-Chapter 3 MC Practice Questions - Quia**

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