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2) $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$ 3) $2 \text{NaCl} + 1 \text{F}_2 \rightarrow 2 \text{NaF} + 1 \text{Cl}_2$ 4) $2 \text{H}_2 + 1 \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$ 5) $1 \text{Pb(OH)}_2 + 2 \text{HCl} \rightarrow 2 \text{H}_2\text{O} + 1 \text{PbCl}_2$ 6) $2 \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + 1 \text{Al}_2(\text{SO}_4)_3$

Balancing Chemical Equations Answer Key

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Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name _ ... Page 39 Chemistry IF8766 Page 38 TYPES OF CHEMICAL BONDS Name-----Classify the following compounds as ionic (metal + metal), covalent (nonmetal + ... Page 40 112 ©Instructional Fair, Inc. ...

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