

Chapter 22 Review Organic Chemistry Section 2 Answers

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Chapter 22-Assignment E: Summary and Review In Chapter 6, we urged you to learn the system when studying inorganic nomenclature. Memorize the minimum needed. In organic chemistry, that minimum is the list of functional groups. A functional group summary is given in Table 22.5, Section 22.14. To what are functional groups attached?

Chapter 22

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Chapter 22 Review Questions: 1. How is organic chemistry different from inorganic chemistry? What type of molecules doe organic chemists study? a. Organic chemistry studies substances with carbon inorganic chemistry studies substances that do not contain carbon. 2.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

Chapter 22: Organic Chemistry 4. Refer to Sections 22.1 and 22.2 and Example 22.2. For a hydrocarbon with n carbons, the number of hydrogens in an alkane equals $2n + 2$, in alkene it equals $2n$, in an alkyne it equals $2n - 2$. This assumes the hydrocarbons are not cyclic and contain at most, one double bond.

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Chapter 22: Organic and Biological Molecules 22.1: Alkanes: Saturated Hydrocarbons Hydrocarbons: - compounds that contains hydrogen and carbon atoms. - it may contain oxygen, nitrogen and other halogen atoms. In complex organic compound, it may even contain transition metals.

Chapter 22 Organic and Biological Molecules Notes (answers)

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