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Description Solutions to each exercise in the Laboratory Manual, complete with diagrams, graphs and maps. Images from the Laboratory Manual are available for download in jpg format from the catalog page (below).

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CH100: Fundamentals for Chemistry Lab 2 Nomenclature File name: Ch100-Lab07-nomenclature-f07-key.doc Ionic Compounds (Metal + Non-Metal) Compound Formula Cation Formula and name Anion Formula and name Compound Name 1. RbI Rb⁺, rubidium ion I⁻, iodide ion RbI 2. Ca²⁺ Ca²⁺, calcium ion N³⁻, nitride ion Calcium nitride 3. TiCl₄ Ti⁴⁺, titanium(IV)

Laboratory #6: Naming Compounds - PCC

Doing one or more moles lab activities in each unit you teach will give students plenty of practice ... answer key. Moles Lab Activity 3: Compounds ... nonstandard lab materials and that this is not a standard practice in a chemistry lab! Answers to Selected Questions: The answers to most questions require basic conversions. The

Moles Lab Activities - VDOE

In this section students watch a 17 minute video: The Chem Games Lab Safety and answer a half-sheet paper of questions while they watch. See attached questions and answer key. This video gives an overview of some of the important things to keep in mind while working in a chemistry lab setting. It is available through discovery education and ...

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CH302 Worksheet 19 - Organic Chemistry Answer Key 1. How many structural isomers does C₄H₁₀ have? Draw them. 2 2. Name them. butane, trimethylmethane 3. How many structural isomers does C₇H₁₆ have? Draw them. 9 CH 4. Name them. In order from I-IX: Heptane, 2-Methylhexane, 3-Methylhexane, 2,3-

CH302 Worksheet 19 - Organic Chemistry Answer Key

laboratory. 2. To learn the capabilities of the different types of balances which are available in the laboratory. 3. To relate the concept of significant numbers to the accuracy of mass and volume measurements. PRINCIPLES: One of the most important operations in a chemistry laboratory is the massing of

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