

## *Chemquest 20 Answer Key*

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Is your answer to question 3 consistent with the data in the table Hopefully you see that triple bonds require more energy to break—they are harder to break than other bonds. 5. What is the relationship between bond order and single, double, and triple bonds A single bond has a bond order of 1, a double bond has bond order of 2, and a triple ...

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20 60 126 Chemical Atomic number Net charge 2+ 3+ symbol 32 19 83 10. Which two fundamental particles (e , , no) have the same mass? 11. Which particle accounts for virtually all of any atoms volume? 12. Exactly how does an atom of P differ from ho more e 13. Given a neutral iron atom (Fe), what must be done in order to make it into a Fe<sup>2+</sup> cation?

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