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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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The law of conservation of matter states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations  $Gizmo^{TM}$ , type the chemical formulas into the text boxes of the

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Question: How are chemical equations balanced? 1. Balance: Turn on Show histograms. The equation is balanced when there are equal numbers of each type of atom represented on each side of the equation. In the Gizmo, use the up and down arrows to adjust the numbers of hydrogen, oxygen, and water molecules until the equation is balanced.

# Student Exploration: Balancing Chemical Equations

Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass. Launch Gizmo

## Chemical Equations Gizmo: Lesson Info: ExploreLearning

Balancing Chemical Equations –Answer Key Balance the equations below: 1) 1 N 2 + 3 H 2  $\rightarrow$  2 NH 3 2) 2 KClO 3  $\rightarrow$  2 KCl + 3 O 2 3) 2 NaCl + 1 F 2  $\rightarrow$  2 NaF + 1 Cl 2 4) 2 H 2 + 1 O 2  $\rightarrow$  2 H 2 O 5) 1

Pb(OH) 2 + 2 HCl  $\rightarrow$  2 H 2 O + 1 PbCl 2 6) 2 AlBr 3 + 3 K 2 SO 4  $\rightarrow$  6 KBr + 1 Al 2 (SO 4) 3

## **Balancing Chemical Equations Answer Key**

This is the fourth lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions.

# **Tenth grade Lesson Gizmo: Modeling Chemical Reactions**

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Student Exploration Balancing Chemical Equations At the end of this balancing chemical equations lesson plan, students will be able to recognize whether a chemical equation containing coefficients is balanced or not, and how that relates to the law of conservation of mass. BALANCING CHEMICAL EQUATIONS LESSON PLAN – A COMPLETE ...

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