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Solutions are homogeneous (single-phase) mixtures of two or more components. For convenience, we often refer to the majority component as the solvent; minority components are solutes. But there is really no fundamental distinction between them.

Solutions and Concentrations - Chem1

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In chemistry, an ideal solution or ideal mixture is a solution with thermodynamic properties analogous to those of a mixture of ideal gases. The enthalpy of mixing is zero as is the volume change on mixing by definition; the closer to zero the enthalpy of mixing is, the more "ideal" the behaviour of the solution becomes. The vapor pressure of the solution obeys Raoult's law, and the activity ...

Ideal solution - Wikipedia

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Second Semester / advanced 20 All about chemical equilibrium This thorough treatment sets out the underlying concepts without invoking thermodynamics or complicated calculations; considerable emphasis is placed on the distinction between Q and K. The section on equilibrium calculations contains problem examples illustrating techniques such as iterative and graphical solutions of polynomials ...

Chem1 Online Textbook Main Menu

Definition. The molality (b), of a solution is defined as the amount of substance (in moles) of solute, n solute, divided by the mass (in kg) of the solvent, m solvent: $b = \frac{n}{m}$. In the cases of solutions with more than one solvent, molality can be defined for the mixed solvent considered as a pure pseudo-solvent.

Molality - Wikipedia

The rusting of barbed wire occurs over many years. [2] Collision Theory. To understand the kinetics of chemical reactions, and the factors that affect kinetics, we should first examine what happens during a reaction on the molecular level.

Factors that Affect the Rate of Reactions - Introductory ...

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Experimental Procedure. Tip: As you do your experiment, take a few pictures of yourself in action and of your experimental setup. Use the pictures to help make your science fair display board more interesting and informative. Get the salt, sugar, sand, and measuring teaspoon ready to use nearby.

What Makes Ice Melt Fastest? | Science Project

The air we breathe, what we drink, as well as products in our household - these are a few examples of solutions we encounter every day. In this lesson, we will discuss what a solution is, the ...

What is a Solution in Science? - Definition & Examples ...

Exam Description: The Chemistry CLEP covers the material commonly found in a one year college Chemistry course. You'll be expected to understand reaction types, equilibrium, kinetics, states and structure of matter, stoichiometry and equations..

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For a solution, Raoult's law relates the relative concentrations of the components in solution with their relative vapor pressures above the solution.

Definition of Raoult's Law | Chegg.com

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