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The AP Chemistry Lab Manual: AP Chemistry Guided Inquiry Experiments: Applying the Science Practices features 16 labs where students explore chemical concepts, questions of interest, correct lab techniques and safety procedures. Teachers may choose any of the guided inquiry labs from this manual to satisfy the course requirement of students performing six guided inquiry labs.

### **AP Chemistry Lab Manual | AP Central — The College Board**

AP Chemistry Labs Up until May 2006, laboratory situations were specifically tested in question #5 on the AP exam, could also come up in parts of other free-response questions, and appeared in a few multiple-choice items. In the new exam format for 2007 you should expect a laboratory based situation as

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LHS AP Chemistry. Search this site. Navigation. AP Chemistry Homepage. ... AP Chemistry Homepage > ... The purpose of this lab activity is to investigate how buffers are made, the pH range in which they are effective, and their buffer capacity. Background Information.

### **Lab #16 - Properties of Buffer Solutions - LHS AP Chemistry**

AP Chemistry Laboratory #21 Introduction Oxidation-reduction reactions form a major class of chemical reactions. From the reactions of oxygen with sugars, fats, and proteins that provide energy for life to the corrosion of metals, many important reactions involve the processes of oxidation and reduction.

### **AP Chemistry Laboratory #21 - Bergen**

AP Chemistry Lab #2 - Inquiry Acid/Base Titrations This lab will not be written into your lab notebook, but will be placed into your chemistry binder as a guided inquiry lab. A general procedure for this titration is described below and will be outlined during our class discussion.

### **AP Chemistry Lab #2 - Inquiry Acid/Base Titrations**

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AP Chemistry Lab #10 Page 2 of 7. colorless condition persists until all of the thiosulfate ions have reacted. Once all of the thiosulfate ions have reacted, more iodine is produced; and the starch indicator reacts and turns dark blue. Carrying out reaction (1) in the presence of thiosulfate ion and starch produces a chemical "clock".

### **doctortang.com**

AP Chemistry Lab #14 Page 2 of 8. Stirring Solutions Each time a reagent is added to a test tube, the solution needs to be stirred. It is important to mix the solutions at the top and the bottom of the test tube. A stirring rod that is flattened at the bottom can be used as a plunger to effectively mix solutions in narrow test tubes.

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