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2.10 What role does the molecular interaction play in a solution of alcohol and water? Sol. Alcohol and water both have strong tendency to form intermolecular hydrogen bonding. On mixing the two, a solution is formed as a result of formation of H-bonds between alcohol and H_2O molecules but these interactions are weaker and less extensive than those in pure H_2O .

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NCERT Solutions for Class 12 Chemistry Chapter 2 Solutions. NCERT IN-TEXT QUESTIONS. Question 1. Calculate the mass percent of benzene (C_6H_6) and carbon tetrachloride (CCl_4) if 22 g of benzene is dissolved in 122 g of carbon tetrachloride. Answer: Question 2.

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1. Mass percentage of a component (w/w) = $\frac{\text{Mass of component in solution}}{\text{Total mass of solution}} \times 100$
2. Volume percentage of a component (v/v) = $\frac{\text{Volume of the component}}{\text{Total volume of solution}} \times 100$
3. Mole fraction of a component (x) = $\frac{\text{Number of moles of the component}}{\text{Total number of moles}}$

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