

Chemquest Answer Key

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Chemquest Answer Key

What is the difference between qualitative and quantitative observations (Your answers should reveal an understanding of the definitions for qualitative and quantitative.) Qualitative refers to the general quality of a characteristic (tall, short, long, wide, etc.) whereas quantitative includes the quantity of the characteristic (how tall, how ...

ChemQuestsanswers - Yumpu

Title: ChemQuest 8: Answer Key Author: Kosiorowski Last modified by: Kosiorowski Created Date: 8/28/2008 11:49:00 PM Company: Sony Electronics, Inc.

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ChemQuest 15 Name: _____ Date: _____ Hour: _____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

ChemQuest 15 - Yav Science - Home

(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = 4(3) = 12.

ChemQuest 33 - baileykaze.webs.com

130 ChemQuest 43 Name: _____ Date: _____ Hour: _____ Information: Reaction Order Below is a table of data corresponding to the following balanced equation: $2\text{ClO}_2(\text{aq}) + 2\text{OH}^-(\text{aq}) \rightarrow \text{ClO}_3^-(\text{aq}) + \text{ClO}_2^-(\text{aq}) + \text{H}_2\text{O}(\text{l})$... What happens to the rate of a reaction as the concentrations of the reactants increases Justify your answer with data ...

130 ChemQuest 43 Name: - Yumpu

Anatomy of a ChemQuest . How ChemQuests are Designed to Enhance Student Learning. To give you an idea of how a ChemQuest is designed, examine the following excerpt from Chemistry Inquiry #8: Structure of Atoms. (Click on the previous link to download a copy.) At this point, the students have not been taught about the structure of an atom.

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Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 127.1 and 17% 126.9, and 3% 128.1. Calculate the average atomic mass of iodine. 2. Calculate the average atomic mass of lithium, which occurs as two isotopes that have the following

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