

## *Chapter 9 Chemistry Answer Key*

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### Chapter 9 Chemistry Answer Key

Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 9 1. Draw Lewis structures for each of the following structures and assign formal charges to each atom: a) SF<sub>2</sub> 20 electrons 0 0 0 b) NH<sub>2</sub> OH (N and O are bonded to one another) 14 electrons 0 0 0 0 0 c) PO<sub>4</sub><sup>3-</sup>

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### Chapter 9 ChemiCal C and C Formulas - Mark Bishop

Chapter 9: Standard Review Worksheet 1. Answers will vary. An example is included below: 2H<sub>2</sub>O<sub>2</sub> (aq) → 2H<sub>2</sub>O(l) + O<sub>2</sub> (g) This describes the decomposition reaction of hydrogen peroxide. Microscopic: Two molecules of hydrogen peroxide (in aqueous solution) decompose to produce two molecules of liquid water and one molecule of oxygen gas.

### Chapter 9: Standard Review Worksheet

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e. molar mass of  $\text{Fe}(\text{NO}_3)_3 = 241.9 \text{ g}$   $272 \text{ g Fe}(\text{NO}_3)_3$   $1 \text{ mol } 241.9 \text{ g} = 1.12 \text{ mol Fe}(\text{NO}_3)_3$  f. molar mass of  $\text{FeSO}_4 = 151.9 \text{ g}$   $12.7 \text{ mg FeSO}_4$   $1 \text{ g } 1000 \text{ mg } 1 \text{ mol } 151.9 \text{ g} = 8.36 \times 10^{-5} \text{ mol FeSO}_4$  g. molar mass of  $\text{LiOH} = 23.95 \text{ g}$   $6.91 \times 10^3 \text{ g LiOH}$   $1 \text{ mol } 23.95 \text{ g} = 288.5 = 289 \text{ mol LiOH}$  14. a. molar mass of  $\text{CaCO}_3 = 100.1 \text{ g}$   $2.21 \times 10^{-4} \text{ mol CaCO}_3$   $100.1 \text{ g } 1 \text{ mol} = 0 \dots$

**Chapter 9 Chemical Quantities - Francis Howell High School**

Chapter 9 Practice Test - Naming and Writing Chemical Formulas Matching Match each item with the correct statement below. Match each item with the correct statement below. a. monatomic ion f. cation b. acid g. binary compound c. base h. anion d. law of definite proportions i. polyatomic ion e. law of multiple proportions \_\_\_\_ 1.

**Chapter 9 Practice Test - Naming and Writing Chemical Formulas**

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Chapter 9 – Chemical Calculations and Chemical Formulas 119 Chapter 9 Map Chapter Checklist Read the Review Skills section. If there is any skill mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter quickly before the lecture that describes it.

**Chapter 9 Chemical Calculations and Chemical Formulas**

144 Chemistry: Matter and Change • Chapter 9 Solutions Manual CHAPTER 9 SOLUTIONS MANUAL 24. Challenge  $\text{Al(s)} + \text{Pb(NO}_3)_2(\text{aq}) \rightarrow$  0 Yes. Al is above Pb in the activity series. ... Explain your answer.  $3\text{Ni} + 2\text{AuBr}_3 \rightarrow 3\text{NiBr}_2 + 2\text{Au}$  The reaction does occur because nickel is more reactive than gold. Section 9.3 Reactions in Aqueous Solutions

**Chemical ReactionsChemical Reactions**

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