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Answers to Revision Questions for Higher Chemistry ...

I don't understand the following question: "Approximately how many atoms will be present in 11.5 litres of carbon monoxide? (Take molar volume of carbon monoxide as 23 litres mole⁻¹) The possible answers are 1.5×10^{23} , 3×10^{23} , 6×10^{23} and 1.2×10^{23} ." I tried solving this question by saying that 11.5 litres is half a mole, so it should be half of Avagadro's Constant which would be 3×10^{23} ...

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Answers to Revision Questions for Higher Chemistry book by ...

Higher Chemistry (For the 2015 Examinations, some schools may be pursuing the new Higher and others may be sticking with the old Higher for its last year) The content below is based on the "old" Higher which we have been studying up until now.

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Full video lessons for the SQA Higher Chemistry Course, excellent resource for candidates looking to achieve the best possible grade !

SQAHigherChemistry - YouTube

2017 Chemistry Higher ... If an answer comes directly from the text of the question, no marks should be given. For example, in response to the question, ' A student found that 0.05 mol of propane, C₃ H₈ burned to give 82.4 kJ of energy. C₃ H₈ (g) + 5 O₂ (g) = 3 CO₂ (g) + 4 H₂O (l)

2017 Chemistry Higher Finalised Marking Instructions

Record your answers on the answer grid on Page 03 of your question and answer booklet. Reference may be made to the Chemistry Higher and Advanced Higher Data Booklet. Before leaving the examination room you must give your question and answer booklet to the Invigilator; if you do not you may lose all the marks for this paper.

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Chemistry Student Resources S5/6 | Lesmahagow High School

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Best Answer: Hi Katie, To solve this problem, you need to figure out how many moles of each reagent (NaOH and HCl) you started with. Which ever molar amount is greater is your reagent that was added in excess. For NaOH, use the equation $\text{Moles of NaOH} = \frac{\text{grams of NaOH}}{\text{MW of NaOH}}$. Therefore, $1.6 \text{ g} / 40 \text{ g} \dots$

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Answers to Revision Questions for Higher Chemistry: J.M ...

Hi! I have to crash Higher Chemistry this year as I took a sudden change of career. I wish to get into medical school and go down that route, however I was initially doing standard grade in a period of one year rather than two. I missed the first month of the first unit in the higher course due to this so I am really far behind.

Higher Chemistry 2012-2013 - The Student Room

Does anyone know where I can download the Higher Chemistry Past Papers beyond year 2007? I really need them. Update: I'm sitting it with the SQA. Follow . 4 answers 4. ... I think that this answer violates the Community Guidelines. Chat or rant, adult content, spam, insulting other members,show more.

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Chemistry is one of my best subjects. I'm hoping for an A as this is what i need to do Advanced Chemistry.I Got a B in my first prelim and a C in my second each time missing a few marks for the grade higher than that. I need to study flat out because i have Chemistry on Wednesday and then Computing the day after!!

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