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### **Chemactivity 32 Molarity Answers With Work**

ChemActivity 32 Molarity 185 Figure 1: Cations and anions in a solid crystal and in water. ions in a crystal Hno ions in aqueous solution Critical Thinking Questions 1. ionic compound, NaCl or Na<sub>2</sub>SO<sub>4</sub>, does Figure 1 represent? Explain your reasoning clearly. 2. a) What is the identity of the cation in Figure 1?

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ChemActivity 32 c) there are  $14.1 \times 10^{-2}$  moles of sodium in the water. a) What is the molarity of the sodium sulfate solution in Model 2? Molarity 181 b) What is the molarity of the sulfate ions in the solution in Model 2? c) What is the molarity of the sodium ions in the solution in Model 2?

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ChemActivity 32 Molarity Which is more concentrated with respect to sodium ions, 50.0 g of NaCl in 500.0 mL of solution or 59.0 g of Na<sub>2</sub>SO<sub>4</sub> in 500.0 mL of solution? Which is more concentrated with respect to sodium ions, 0.50 M NaCl or 0.30 M Na<sub>2</sub>SO<sub>4</sub>? Which CTQ was easier to answer, 8 or 9? Why?

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