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Word Equations Answer Key 1. Zinc and Lead (II) nitrate react to form Zinc Nitrate and Lead.  $\text{Zn} + \text{Pb}(\text{NO}_3)_2 \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Pb}$  Single Replacement 2. Aluminum Bromide and Chlorine gas react to form Aluminum Chloride and Bromine gas.  $2\text{AlBr}_3 + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3 + 3\text{Br}_2$  Single Replacement 3.

### Word Equations Answer Key Zinc and Lead (II) nitrate react ...

Balancing Chemical Equations -Answer Key Balance the equations below: 1)  $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  2)  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  3)  $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$  4)  $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$  5)  $1\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$  6)  $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$

### Balancing Chemical Equations Answer Key

Chemistry IF8766 Page 59 Instructional Fair, Inc. Title: Microsoft Word - pg 59 - Word equations.doc Author: Nathan Created Date: 3/28/2007 10:43:22 PM ...

### pg 59 - Word equations

Chemistry IF8766 9 OInstructional Fair, Inc. 59. ... SIGNIFICANT FIGURES Name ? When multiplying and dividing, limit and round to the least number of significant figures in any of the factors. 11 Example 1 :  $23.0\text{ cm} \times 432\text{ cm} \times 19\text{ cm} = 188,784\text{ cm}^3$  11 The answer is expressed as  $190,000\text{ cm}^3$  since 19 cm has only two significant fiaures.

### SIGNIFICANT FIGURES Name - Dearborn Public Schools

The Mole and Volume Directions: For gases at ... 22.4L. What volume will the following quantities of gasses occupy at STP? 1) 5.00 mole of  $\text{H}_2$  ANSWER: 112. L 2) 2.50 moles of  $\text{O}_2$  ANSWER: 56.0 L ... ANSWER: 87.5 L 8) 56.0g of  $\text{N}_2$  ANSWER: 44.8 L 9) 70. g of  $\text{CO}_2$  ANSWER: 36. L 10) 20. g of  $\text{NH}_3$  ANSWER: 26 L Chemistry IF8766 Page 51 Instructional ...

### The Mole and Volume - Ms. Agostine's Chemistry Page

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ANSWER: 0.21 mole Directions: determine the number of grams in each of the quantities below. 6) 2.8 moles of NaCl ANSWER: 160 g 7) 0.75 moles H<sub>2</sub>SO<sub>4</sub> ANSWER: 74 g 8) 2.75 moles of KMnO<sub>4</sub> ANSWER: 435. g 9) 0.55 moles of KCl ANSWER: 41. g 10) 2.3 moles of CuSO<sub>4</sub>•5H<sub>2</sub>O ANSWER: 570 g Chemistry IF8766 Page 50 Instructional Fair, Inc.

### **Moles and Mass - Ms. Agostine's Chemistry Page**

Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name \_ ... Page 39 Chemistry IF8766 Page 38 TYPES OF CHEMICAL BONDS Name-----Classify the following compound as ionic (metal + nonmetal), covalent (nonmetal + nonmetal) ... Page 40 112 ©Instructional Fair, Inc. ...

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Physical Science If8767 Answer Key rectangular-shaped object is 94 g, what is the density of iron? Answer: Physical Science If8767 13 ' @IFTSH'UCTTOPGI ... Chemistry IF8766 Page 49) \*\*\*All answers in units of g/mole. PS WORKSHEET - HHS-Physical Science - home

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### **Gram Formula Mass Instructional Fair Answer Key**

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### **Solutions to Modern Chemistry (9780030735462), Pg. 59 ...**

Chemistry IF8766 62 . STOICHIOMETRY: MASS-MASS PROBLEMS 1.2KC103 + \_302 Name How many grams of potassium chloride are produced if 25 g of potassium chlorate decompose? / 25 ken How many grams of hydrogen are necessary to react completely with 50.0 g of nitrogen in the above reaction?

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