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### **Chapter 15 Chemical Equilibrium**

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water molecules: if the attraction of the water molecules for the positive ion is stronger than the attraction of the negative ions near it in the crystal, the ion leaves the crystal and enters solution. After entering solution, the dissolved ion is surrounded completely by water molecules, which tends to prevent the ion from reentering the ...

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168 Guided Reading and Study Workbook CHAPTER 16, Solutions (continued) 7. ... water increases to 39.2 g per 100 g of water at 100°C from 36.2 g per 100 g of ... Chapter 16 Solutions 169 15. Describe the two diagrams of a bottled carbonated beverage below as greater

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