

## *Chapter 20 Worksheet Redox Answers*

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## Chapter 20 Worksheet Redox Answers

Chapter 20 Worksheet: Redox ANSWERS I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also.

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Chapter 20 Worksheet: Redox. I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also. 1.  $2\text{Sr} + \text{O}_2$  2.  $2\text{Li} + \text{S}$  3.  $2\text{SrO} + \text{Li}_2\text{S}$  4.  $2\text{Cs} + \text{Br}_2$

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Chapter 20 Worksheet: Redox ANSWERS I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also.  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$  Sr0 to  $\text{Sr}^{2+}$ ; oxidized/reducing agent  $\text{O}_2$  to  $\text{O}^{2-}$ ; reduced/ox. ag. 1.  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$  2.  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$  Li0 to  $\text{Li}^{+}$ ; oxidized/red. ag.

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636 Chapter 20 Redox Reactions Figure 20-1 The reaction of magnesium and oxygen involves a transfer of electrons from magnesium to oxygen. Therefore, this reaction is an oxidation-reduction reaction. Using the classifications given in Chapter 10, this redox reaction also is classified as a combustion reaction.  $\text{X X X X X X X X } 2\text{Mg} + 2\text{MgO}$  Each ...

## Chapter 20: Redox Reactions - Neshaminy School District

Chapter 20 Worksheet: Redox I. Determine what is oxidized and what is reduced in each reaction. Identify the oxidizing agent and the reducing agent, also. 1.  $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$  2.  $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$  3.  $2\text{Cs} + \text{Br}_2 \rightarrow 2\text{CsBr}$  4.  $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$  5.  $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$  6.

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