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1. all matter is composed of extremely small particles called atoms 2. atoms of a given element are identical in size, mass, and other properties; atoms of diff. elements differ in size, mass, and other prop. 3. atoms cannot be subdivided, created, or destroyed 4. atoms of diff. elements combine in simple whole number ratios to form chemical ...

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learn the answer to this question, you need to understand some of the basic ideas of chemistry and know how chemists—and most scientists, for that matter—solve problems. 6 Chapter 1 Introduction to Chemistry CFC-11 (ppt) 280 240 200 160 120 Concentration of CFCs in the Atmosphere 1977 1979 1981 1983 1985 1987 1989 1991 1993 1995 Year 450 ...

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a. heating the test tube gently to prevent the solution from boiling over b. pointing the test tube

away from others so that no one is injured c. placing a stopper in a test tube to prevent gas from escaping d. holding the test tube with test tube clamps to avoid touching hot objects 16.

Name Honors Chemistry // Practice Test - Chapter 1

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Chemistry Chapter 1 Test Review Multiple Choice Identify the choice that best completes the statement or answers the question. Put the LETTER of the correct answer in the blank. ____ 1. Inorganic chemistry is the study of a. non-carbon related compounds. b. the chemistry of living things. c. mathematical modeling. d.

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Chemistry 101 ANSWER KEY 1 REVIEW QUESTIONS Chapter 3 1 ...

4 Chapter 1 An Introduction to Chemistry. 1.2 Suggestions for Studying Chemistry 5 SPECIAL TOPIC 1.1 Green Chemistry ... should work to test yourself on these tasks. The answers to these exercises are found at the end of the book. If you have trouble with a practice exercise, look more closely ...

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