# Chapter 9 Covalent Bonding Study Guide Answer Key

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242 Chapter 9 Covalent Bonding Figure 9-1 The overall force between two atoms is the result of electron-electron repulsion, nucleus-nucleus repulsion, and nucleus-electron attraction. The arrows in this diagram show the net force acting on two fluorine atoms as they move toward each other. The atoms are too far away from each other to have ...

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#### **Chapter 9 Covalent Bonding Study Guide Answers**

Study Guide Chapter 9 – Covalent Bonding Study Guide Chapter 9 – Covalent Bonding 1. Describe covalent bonds and be able to distinguish between single, double, triple, sigma and pi bonds.

#### Study Guide Chapter 9 - Covalent Bonding

Chapter 9 - Covalent Bonding: Orbitals . 9.1 Hybridization and the Localized Electron Model . A. Hybridization 1. The mixing of two or more atomic orbitals of similar energies on the same atom to produce new orbitals of equal energies B. Hybrid Orbitals 1. Orbitals of equal energy produced by the combination of two or more

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