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Chapter 6 Assessment Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. When an atom loses an electron, it forms a(n) a. anion. c. polyatomic ion. b. cation.. d. neutral ion. 2. The charge on a chloride ion in is

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f. mass of 6 mol C = 6 (12.01 g) = 72.06 g mass of 6 mol H = 6 (1.008 g) = 6.048 g mass of 1 mol O = 16.00 g = 16.00 g molar mass of C6H6O = 94.11 g 21. a. molar mass of SO3 = 80.07 g 49.2 mg SO3 1 g 1000 mg 1 mol 80.07 g = 6.14 10-4 mol SO 3 b. molar mass of PbO2 = 239.2 g 7.44 x 104 kg PbO 2 1000 g 1 kg 1 mol 239.2 g = 3.11 105 mol PbO 2

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LawThe Periodic Table and Periodic Law CHAPTER 6 SOLUTIONS MANUAL Section 6.1 Development of the Modern Periodic Table ... Section 6.1 Assessment page 181 1. Describe the development of the modern

The Periodic Table and Periodic LawThe Periodic Table and ...

174 Chapter 6 • The Periodic Table and Periodic Law Section 66.1.1 Development of the Modern Periodic Table MAIN Idea The periodic table evolved over time as scientists discovered more useful ways to compare and organize the elements.

Chapter 6: The Periodic Table and Periodic Law

CP Chemistry I, Chapter#6 "Chemical Bonding" Test Questions Bank Practice Quiz. This is a practice quiz of all the potential multiple choice questions that a student taking Mr. New's Chemistry I course should expect on the chapter exam for chapter #6 from textbook "Modern Chemistry" (Holt, 2006).

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