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The maximum num	ber of electrons in the second energy level of an atom is 4. $___$	_ 4. Two
isotopes of carbon are carbon-12 and carbon-14. These isotopes differ from one another by two		
protons	5. So far, scientists have confirmed the existence of six different qua	rks
6. The atomic	number of an element is determined by its number of protons.	

CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure

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Chemistry Chapter 4 Test – Atomic Structure Democritus – (400 B.C.) Was among the first to suggest the existence of atoms and stated that matter was made up of separate discrete particles

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Chapter 4 (Atomic Structure) Test Study Guide \square An atom is the smallest particle of an element that retains the properties of that element. \square Democritus lived from 460 B.C. – 370 B.C. and was among the first to suggest the idea of atoms.

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Atomic Structure Chapter 4 Chemistry Test | Get Access To Unique Paper Atoms Smallest particles that make up matter Democritus Greek philosopher who was among the first to suggest the existence of atoms.

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TEST #4: Chapter 4.....Atomic Structure & the History of the Atom. *****Place all answers to the multiple-choice at the end of the question section. While operating cathode ray tube you make the following observations: 1. The cathode rays travel in straight lines from the negative cathode towards the positive anode.

TEST #4: Chapter 4Atomic Structure & the History of the Atom

Prentice Hall Chemistry Chapter 4 Atomic Structure. Dalton's Atomic Theory Atom The smallest particle of an element that retains its identity... Democritus (460 BC-370 BC) -Was among the first to suggest the existence of atoms (from t... How did the concept of the atom change... What instruments are used to observe in... Democritus REASONED...

chemistry chapter 4 atomic structure Flashcards and Study ...

104 Chapter 4 • The Structure of the Atom. ■ Figure 4.2 In his book A New System of Chemical Philosophy, John Dalton presented his symbols for the elements known at that time and their possible combinations.

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