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### Definition of Raoult's Law | Chegg.com

Chemical Changes in the Blood During Exercise. In previous tutorials ("Hemoglobin and the Heme Group: Metal Complexes in the Blood for Oxygen Transport", "Iron Use and Storage in the Body: Ferritin and Molecular Representations", "Maintaining the Body's Chemistry: Dialysis in the Kidneys") you learned about the daily maintenance required in the blood for normal everyday activities such as ...

### pH Buffers in the Blood - chemistry.wustl.edu

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Nickel(II) chloride (or just nickel chloride), is the chemical compound  $\text{NiCl}_2$ . The anhydrous salt is yellow, but the more familiar hydrate  $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$  is green. Nickel(II) chloride, in various forms, is the most important source of nickel for chemical synthesis. The nickel chlorides are deliquescent, absorbing moisture from the air to form a solution. ...

### Nickel(II) chloride - Wikipedia

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Hydrogen sulfide is the chemical compound with the formula  $\text{H}_2\text{S}$ . It is a colorless chalcogen hydride gas with the characteristic foul odor of rotten eggs. It is very poisonous, corrosive, and flammable. Hydrogen sulfide is often produced from the microbial breakdown of organic matter in

the absence of oxygen gas, such as in swamps and sewers; this process is commonly known as anaerobic ...

### Hydrogen sulfide - Wikipedia

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