Chemquest 50 Answer Key

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144 ChemQuest 47 Name: ____ Date: ____ Hour: ____ Critical Thinking Questions 1. If K c for a given reaction is very large would there be a large amount of products or reactants in the mixture Large amounts of products. 2.

144 ChemQuest 47 Name: - Yumpu

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148 ChemQuest 49 Name: ____ Date: ____ Hour: ____ Information: Measuring Acidity Many people have heard of "pH", and many people know that it probably refers to acidity, but not very many understand what exactly it measures. The pH is a measure of the concentration of hydrogen ions (H +) in a solution.

148 ChemQuest 49 Name: - Yumpu

Title: ChemQuest 8: Answer Key Author: Kosiorowski Last modified by: Kosiorowski Created Date: 8/28/2008 11:49:00 PM Company: Sony Electronics, Inc.

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Again consider the combustion of propane: C3H8 \pm 5 O2 (3 CO2 \pm 4 H2O. If you had 10 moles of propane to burn, you would need 50 moles of oxygen according to the ratio in the balanced equation. If you only had 20 moles of oxygen you could not combust all 10 moles of propane. The reaction has been . limited

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a) Given your answers to questions 4-6, what happens to the rate of the reaction if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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ChemQuest 8 Information: Structure of the Atom Note the following symbols: (they are not to scale)

... 50 Number of neutrons 10 24 35 30 50 61 Hour Net charge 2+ Number of electrons 013 301 Chemical symbol 31 15 45 21 64 cu 29 131 54 237 93 16 2- 70 31 52 Cr 24 5b Mass number 231 108 ... Round all answers to two decimal places. Record your answer

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ChemQuest 15 Name: ____ Date: ____ Hour: ____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready. Remember that a row on the periodic table

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