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Chemistry 101 Final Exam Study Guide 2017 of the other reactant. $2\text{Ca(s)} + \text{O}_2\text{(g)} \rightarrow 2\text{CaO(s)}$
Answer: ____ 157. For the reaction, calculate how many moles of the product form when 0.042 mol of O_2 completely reacts.

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Unit 1...Measurement & Classification of Matter

the periodic table to the electron configuration of atoms. You will identify the s-, p-, d-, and f-blocks of the periodic table. Why It's Important The periodic table is the single most powerful chemistry reference tool available to you. Understanding its organization and interpreting its data will greatly aid you in your study of chemistry.

Chapter 6: The Periodic Table and Periodic Law - irion-isd.org

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6) Solutions which have water as the solvent are called ____ solutions. a) aqueous b) wet c)

hydrated d) numerous 14) A solution that has not reached the limit of solute that will dissolve in it is said to be _____. a) saturated b) supersaturated c) concentrated d) unsaturated 30) Vinegar is a solution of acetic acid in water and typically contains 5% acetic acid by mass.

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