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Stoichiometry Page 1 of 12 The Advanced Placement Examination in Chemistry Part I. What is a possible molecular formula of the hydrocarbon? (A) CH<sub>4</sub>. Chapter 3 Stoichiometry. Write the molecular formula and calculate the formula weight for. empirical formula? For 100.0 g of benzoic acid: Balancing Equations Worksheet and Key 1.

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Stoichiometry Practice Worksheet Solve the following stoichiometry grams-grams problems: 1) Using the following equation:  $2 \text{ NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{ H}_2\text{O} + \text{Na}_2\text{SO}_4$  How many grams of sodium sulfate will be formed if you start with 200 grams of sodium hydroxide and you have an excess of sulfuric acid?

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A chemistry tutorial designed to help learn the basic principles of balancing chemical equations, along with examples and methods of balancing different chemical equations.

### **Balancing Chemical Equations - Chemistry Tutorial**

Balancing. In chemistry, chemical reactions are frequently written as an equation, using chemical symbols. The reactants are displayed on the left side of the equation and the products are shown on the right, with the separation of either a single or double arrow that signifies the direction of the reaction.

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