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I'm now considering going back to study Chemistry, and I thought it might be a good idea to see how much of it I had retained from High School. So I downloaded one of the past WACE exams and spent about 3 hours doing it, went to see how I did but realized there were no answer guides!

Is there somewhere I can get the answer guide for past ...

Justify your answer. Yes. Addition of an excess of 0.20 M AgNO_3 (aq) will precipitate all of the I^- ion present in the solution because AgI is insoluble, as evidenced by its low value of K_{sp} . 1 point is earned for the correct answer with a valid justification. (ii) The student only has access to one KI

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School Curriculum and Standards Authority | Past WACE ...

hey guys, just wondering does anyone have any of the answers to the past chemistry WACE exams. I can only find the exams on SCSA and not the answers and am really struggling with them! Any help would be greatly appreciated. Thank you

Past chemistry WACE answers - Studentbox

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