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Use collision theory to explain the following Liquid gasoline will burn, but gasoline vapor will explode. (Hint: this example contains two factors.) People who work in papaya-processing plants must wear gloves and masks for personal protection. Campfires are started with matches that are used to light loosely packed kindling (small pieces of very dry wood).

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A 5.000 g sample of methanol, CH_3OH , was combusted in the presence of excess oxygen in a bomb calorimeter containing 4000 g of water. The temperature of the water increased from 24.000°C to 29.765°C . The heat capacity of the calorimeter was $2657 \text{ J}/^\circ\text{C}$. The specific heat of water is $4.184 \text{ J/g}^\circ\text{C}$. Calculate the energy for the reaction in kJ/mol .

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