

Chemistry If8766 Significant Figures Answer Key

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Chemistry If8766 Significant Figures Answer

When multiplying and dividing, limit and round to the least number of significant figures in any of the factors. 11 Example 1 : $23.0 \text{ cm} \times 432 \text{ cm} \times 19 \text{ cm} = 188,784 \text{ cm}^3$ 11. The answer is expressed as $190,000 \text{ cm}^3$ since 19 cm has. only two significant figures.

SIGNIFICANT FIGURES Name - Dearborn Public Schools

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Arkansas State University Department of Chemistry and Physics. 2. Determine the number of significant figures in each of the following: a) 3427 b) 0.00456 c) 123,453 d) 172 e) 0.000984 f) 0.502 g) 3100.0×102 h) 0.0114×104 i) 107.2 j) 0.0000455 k) 2205.2 l) 30.0×10^{-2} m) 0.982×10^{-3} n) 0.0473 o) 650.502 p) 3.03×10^{-1} q) 20.4×105 r) 1.29 s)...

Worksheet 1 - Scientific notation/ significant figures

Name: _____. Calculations using Significant Figures. When multiplying and dividing,

limit and round to the least number of significant figures for any of the factors. Example 1: 23.0 cm x 432 cm x 19 cm = 188,784 cm³ The answer is expressed as 190,000 cm³ since 19 cm has only two significant figures.

Calculations using Significant Figures

Mr. Kent's Chemistry Pages. This site contains information for AP Chemistry, Regents Chemistry and Consumer Chemistry at Seafood High School. The pages include how to use significant figures in calculations among other things.

Calculations Using Significant Figures (Sig Figs)

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