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A B; nucleus: a dense, positively charged mass located in the center of an atom: proton: a positively charged subatomic particle that is found in the nucleus of an atom

Quia - Chapter 4: Atomic Structure Vocabulary Review

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Chapter 4: Atomic Structure Worksheet . Answer the following questions, circle the best answer. 1) Which of the following are isotopes of each other? a) ^{14}C and ^{14}N b) ^3H and ^4He c) ^2H and ^1H d) none of these . 2) The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is .

Name Chapter 4: Atomic Structure Worksheet Answer the ...

Dalton's Atomic Theory (experiment based!) 3) Atoms of different elements combine in simple whole-number ratios to form chemical compounds 4) In chemical reactions, atoms are combined, separated, or rearranged – but never changed into atoms of another element.

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104 Chapter 4 • The Structure of the Atom Figure 4.2 In his book A New System of Chemical Philosophy, John Dalton presented his symbols for the elements known at that time and their possible combinations.

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