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At the completion of this episode's lesson(s), you should be able to: • Describe and explain the process of forming a solution using the terms solute and solvent. • Compare the properties of suspensions, colloids, and solutions. • List the factors that affect the rate at which a solid solute dissolves in a liquid solvent.

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Department of Chemistry

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Chemistry - Wikipedia

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Calculating the concentration of solutions in moles per litre (molarity, mol/L, M) tutorial suitable for chemistry students

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You can tell suspensions from colloids and solutions because the components of suspensions will eventually separate. Colloids can be distinguished from solutions using the Tyndall effect. A beam of light passing through a true solution, such as air, is not visible.

Solutions, Suspensions, Colloids, and Dispersions

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