

## *Chemquest 37 Answer Key*

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### Chemquest 37 Answer Key

86 ChemQuest 29 Name: \_\_\_\_\_ Date: \_\_\_\_\_ Hour: \_\_\_\_\_ Introduction Questions 1. Write the equation for the reaction of sodium and chlorine to form sodium chloride.  $\text{Na} + \text{Cl}_2 \rightarrow \text{NaCl}$  2. Write the equation for the reaction of calcium nitride and sodium chloride to produce calcium chloride and sodium nitride.

### 86 ChemQuest 29 Name: - Yumpu

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ChemQuest 37 SslgDcll?g E"a1ue Date Hour I ntroduction Questions Information: Subscripts and Coefficients A subscript is a small number that tells you how many atoms are in a compound. For example, in  $\text{CaCl}_2$  the two is the subscript and it tells us that there are two chloride ions bonded to one calcium. A coefficient tells also tells us how many atoms or compounds there are, but in a different way.

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112 Critical Thinking Questions 13. If 128g of a certain gas in a container with a volume of 21.5 L has a pressure of 132 kPa and a temperature of 45 o C, what is the molar mass of the gas  $MPV = mRT$   $M(132)(21.5) = (128)(8.31)(318)$   $M = 119.2$  g/mol 14.

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