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Chapter 20: Electrochemistry

14 Study Guide for An Introduction to Chemistry Exercise 2.3 - Cations and Anions: Identify each of the following as a cation or an anion, and determine the charge on each. (Obj 22) a magnesium atom with 12 protons and 10 electrons $(+12) + (-10) = +2$ This is a +2 cation. a fluorine atom with 9 protons and 10 electrons

Chapter 2 The Structure of Matter and the Chemical Elements

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