

## *Chemistry Matter And Change Section Answers*

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Solutions Manual Chemistry: Matter and Change • Chapter 7 103 Section 7.1 Ion Formation pages 206–209 Section 7.1 Assessment page 209 1. Compare the stability of a lithium atom with that of its ion, Li . The Li<sup>+</sup> ion is more stable because it has a complete octet. 2. Describe two different causes of the force of attraction in a chemical bond.

### **Ionic Compounds and MetalsIonic Compounds and Metals**

Chemistry: Matter and Change Teacher Guide and Answers 8 molar mass KCO<sub>3</sub> 99.11 g/mol KCO<sub>3</sub> n molar mass of molecular formula/molar mass of empirical formula 198.22 g/mol/99.11 g/mol 2(KCO<sub>3</sub>)<sub>n</sub> The molecular formula of the compound is K<sub>2</sub>C<sub>2</sub>O<sub>6</sub>. Section 10.5 The Formula for a Hydrate 1. hydrate 2. hydration 3. water molecules

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Section 6.1 continued In your textbook, read about the modern periodic table. Use the information in the box on the left taken from the periodic table to complete the table on the right. Nitrogen ... Chemistry: Matter and Change Chapter 6 . Subject: Image Created Date:

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