

## *Chemactivity 6 Atomic Size Answers*

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Answers to ChemActivity 1 – The Nuclear Atom- Chapter 5 1. 6 2. 6,7,7 3. 6,6,7 4. If more -1 it is a negative ion, if more +1 it is a positive ion. Z = the number of protons 7. The same number of protons 8. The neutrons differ in number which means the mass number is different. Mass number = protons + neutrons 10. Within the nucleus.

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ChemActivity 7 Atomic Number and Atomic Mass Are all atoms of an element the same?) 6. Model 1: Natural abundances of isotopes Each element found in nature occurs as a vary appreciably from place to place in the in the Sun are much different than on the Earth.

### **Solved: ChemActivity 7 Atomic Number And Atomic Mass Are A ...**

7 (regardless of your answer for Gd!). ChemActivity 12 1.<sup>13</sup>C has a small nucleus consisting of 6 protons (positively charged) and 7 neutrons (no charge). <sup>13</sup>C has six electrons in shells around the nucleus.

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Chemactivity 5 Shell Model 2 Answers Chemactivity 5 Shell Model 2 Chemactivity 5: The Shell Model (II) Core Charge, Valence Electrons, Valence shell, etc. STUDY. PLAY. What ... Chemactivity 6: Atomic Size. Chemactivity 5: The Shell Model (II) Flashcards | Quizlet ChemActivity 5 The Shell Model (II) 25 Information As described above, the outer ...

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