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7.2 Ionic Bonds and Ionic Compounds > 27 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. Would you expect to find sodium

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View Notes - 7.2 Ionice Bons and Compounds Section Review from SCIENCE Chemistry at Prescott High. Class Section Review + 5/ Objectives 0 Explain the electrical charge of an ionic compound 0 Describe

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2 5. When 4.7 x 1011 atoms of sodium are put into excess water, how many grams of base are made? RXN: 2Na + 2H 2O o 2NaOH + H 2 6. What is the difference between the limiting reagent and the excess reagent? _____ 7. What is the limiting reagent when 0.3 grams of nitrogen gas are reacted with 1.2 grams of aluminum? RXN: N 2 + 2AI o 2AIN 8.

Name Period CHEMISTRY II FINAL EXAM REVIEW

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: 4 a. S in H 2SO 3 6 b. S in MgSO 4 2 c. S in K 2S 1 d. Cu in Cu 2S 6 e. Cr in Na 2CrO 4 5 f. N in HNO 3 4 g. C in (HCO 3 ...

7 Chemical Formulas and Chemical Compounds

Chemistry is the study of the composition, properties, and reactivity of matter. This may be your first time taking chemistry, but chances are you know a lot already from observing the world around you. We will be covering the material in a first year introductory high school or college general chemistry course.

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Chemistry – Unit 7 Review Chemical Reaction Model 1. Describe key characteristics of all chemical reactions, including the role of energy. Explain how a balanced equation represents these features (include an example). In chemical reactions, atoms of the reactants recombine to form new substances in

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