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Chemistry: The Molecular Nature of Matter and Change

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In classical physics and general chemistry, matter is any substance that has mass and takes up space by having volume. All everyday objects that can be touched are ultimately composed of atoms, which are made up of interacting subatomic particles, and in everyday as well as scientific usage, "matter" generally includes atoms and anything made up of them, and any particles (or combination of ...

Matter - Wikipedia

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Chemistry 11 Answer Key - Vancouver School Board

Change gases to STP. Calculate partial pressure of a gas using Dalton's Law. Calculate stoichiometric problems using density of a gas at STP. Find the molar volume of a gas (laboratory) Know the concept of Graham's Law (no calculation) Chemistry Gas Laws Review Name ____ Work the following problems.

Chemistry Gas Laws Review

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