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Chemistry Packet Answers Phase Changes - caffetorelli.com 2105232 Chemistry Packet Answers Phase Changes 2. a phase is a state of matter, such as a solid, a liquid, or a gas. which phases would you see? explain. solid ice, liquid water, and water vapor.

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Name __ ANSWER KEY __ Period ____ C:\Users\mwrooney\Desktop\Matter Homework Packet_ANSWERS.doc Identify each of the following as an example of a Physical Property or a Chemical Property. 1. Silver tarnishes when it comes in contact with hydrogen sulfide in the air.

Matter Homework Packet ANSWERS - Properties and Changes of ...

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Unit 3: Phases of Matter-key Regents hemistry 14- Z15 Mr. Murdoch Page 15 of 63 Website upload 2014 Phase Change Diagrams (Heating/Cooling Curve) Page 14: When applying CONSTANT HEAT to a solid (point A), the temperature of the solid increases (segment AB) until the melting point (point B) is reached, for example of water ice at 0° or 273 K.

Unit 3: Phases of Matter-key Regents hemistry 14 Mr ...

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Matter undergoes phase changes or phase transitions from one state of matter to another. Below is a complete list of the names of these phase changes. The most commonly known phase changes are those six between solids, liquids, and gasses. However, plasma also is a state of matter, so a complete list requires all eight total phase changes.

List of Phase Changes Between States of Matter - ThoughtCo

I can identify where a solid, liquid, gas, triple point, and critical point are on a phase change diagram; I can identify pressure as well as temperature as a determining factor for a phase of matter; I can identify the two types of energy: kinetic and potential and how they apply to particulate diagrams

Particle Movement in Matter - MRS. MACK'S CHEMISTRY

Practice Packet, Unit 3: Phase Changes & Heat 2 www.chempride.weebly.com Review (Things you need to know in order to understand the new "stuff") Particle Diagrams Draw a particle diagram of a compound of CaCl_2 , using black solid circles to represent the Ca and empty circles to represent the Cl. Draw at least five molecules of CaCl_2

Practice Packet Unit 3: Phase Changes & Heat

O o o O c: O run o o O O O o o O o -200 -100 200 300 400 500 600 700 800 900. Pressure (atmospheres) O O O

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