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Stoichiometry 379 CHAPTER 12 Assessment 36 Chapter 12 stoichiometry test b answer key. a. Two formula units KClO3 decom- pose to form two formula units KCl and three molecules O2. b Chapter 12 stoichiometry test b answer key. Four molecules NH3 react with six molecules NO to form five mol-

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1. d It might be easiest to balance the equation with mostly whole numbers: 2 NH 3 +  $\frac{7}{2}$ O 2  $\rightarrow$  2NO 2 + 3H 2O.The question asks about the amount of oxygen reacting with ONE mole of ammonia, thus cut the  $\frac{7}{2}$  (3.5) of oxygen in half to 1.75

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