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The common-ion effect says that if you add Cl-to the solution above, then less AgCl would dissolve. For example, if you add NaCl to a saturated solution of AgCl, then some AgCl will crash out of solution. Another example: more AgCl can dissolve in pure water than in water containing Cl-ions.

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A Review of General Chemistry - Homepage | Wiley

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Review of solutions to global warming, air pollution, and ...

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