

## *Chemfiesta Ph Answers*

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### Chemfiesta Ph Answers

pH practice - Answers. 1) What is the pH and pOH of a  $1.2 \times 10^{-3}$  HBr solution? pH: 2.9 pOH: 11.1. 2) What is the pH and pOH of a  $2.34 \times 10^{-5}$  NaOH solution? pOH: 4.6 pH: 9.4. 3) What is the pH and pOH of a solution made by adding water to 15 grams of hydroiodic acid until the volume of the solution is 2500 mL? pH: 1.6 pOH: 12.4

### pH practice - Awesome Science Teacher Resources

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