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occupies a volume of 22.4L. What volume will the following quantities of gasses occupy at STP? 1)  
5.00 mole of H<sub>2</sub> ANSWER: 112. L 2) 2.50 moles of O<sub>2</sub> ANSWER: 56.0 L 3) 0.850 moles of N<sub>2</sub>  
ANSWER: 19.0 L 4) 3.75 moles of CO<sub>2</sub> ANSWER: 84.0 L 5) 0.70 mole of NH<sub>3</sub>

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Writing Formulas From Names Directions: Write the formulas of the following compounds. 1. ammonium phosphate (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> \_\_\_\_ 2. iron (II) oxide FeO \_\_\_\_ 3. iron (III) oxide Fe<sub>2</sub>O<sub>3</sub> ...  
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### Balancing Chemical Equations Answer Key

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