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CHAPTER 11 REVIEW Molecular Composition of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The average speed of a gas molecule is most directly related to the . (a) polarity of the molecule (b) pressure of the gas (c) temperature of the gas (d) number of moles in the sample 2.

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CHAPTER 11 REVIEW Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Consider the following data table: Approximate pressure (kPa) Altitude above sea level (km) 100 0 (sea level) 50 5.5 (peak of Mt. Kilimanjaro) 25 11 (jet cruising altitude) 0.1 22 (ozone layer) a.

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CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_ The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L.

**CHAPTER 11 REVIEW Gases - Manasquan Public Schools**

Chapter 11: Gases- PAP Chemistry Review. So far what I have done... if have time will add to later. STUDY. PLAY. ... the number of collisions of molecules of the gas with the inside walls of the container. ... Chemistry Chapter 11. 48 terms. Chapter 11~Gas Laws~Chemistry. 34 terms. Chemistry Unit 9 Gas Laws.

**Chapter 11: Gases- PAP Chemistry Review Flashcards | Quizlet**

Temperature of gas depends on average kinetic energy of particles Ch 10 pg 329-330 Parameters for an ideal gas Kinetic molecular theory only applies to "ideal" gases. Most gases are only nearly ideal if pressure is not too high and temperature is not too low

**Gases - Los Angeles County High School for the Arts**

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CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:

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Review Previous Concepts Gases CHAPTER 11 1. What is a mole? What information does the molar mass of a ... 344 CHAPTER 11. Dalton's Law of Partial Pressures The pressure exerted by each gas in a mixture is called the ... of 0.28 atm when mixed with one or more gases. Dalton's law of partial pressures states that the total

**CHAPTER 11 Gases - St. Charles Parish**

Chapter 11 181 Chapter 11 - Gases Review Skills 11.1 Gases and Their Properties Ideal Gases Properties of Gases Discovering the Relationships Between Properties The Relationship Between Volume ... neon gas mixed with argon gas. (Objs 21 & 22) a. If the total pressure of the mixture of gases is 1.30 kPa and the partial pressure of neon

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