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Chemistry 12 Tutorial 10—Solutions Chemistry 12—Tutorial 10 Solutions Page 3 of 5 pages 3 Ksp  $4.9 \times 10^{-17}$   $4.3 \times 10^{-9}$  3. a) Calculate the molar solubility (solubility in moles/Litre) of  $\text{Fe}(\text{OH})_2$  in water.  $s + s + 2s$  (where "s" is the molar solubility)

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