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temperature. The normal freezing point of mercury is  $-38.9^{\circ}\text{C}$ , and its molar enthalpy of fusion is  $\Delta H_{\text{fusion}} = 2.29 \text{ kJ/mol}$ . What is the entropy change of the system when 50.0 g of  $\text{Hg(l)}$  freezes at the normal freezing

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