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MIXED MOLE PROBLEMS - KEY 1. a) How many grams are there in 1.5×10^{25} molecules of CO? 1.110g 1 mol 44.0 g 6.0210 molecules 1.510 molecules CO 1 mol 3 23 2 25 !=!!!! b) What volume would the CO

KEY - CP - Mixed Mole Problems

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Answer Key Chemistry If8766 Moles And Mass

MIXED PROBLEMS Name What volume of NH_3 at STP is produced if 25.0 g of N_2 is reacted with an excess of ... Chemistry IF8766 Olnstructional Fair, Inc. STOICHIOMETRY: MOLE-MOLE PROBLEMS I. $\text{N}_2 + 3\text{H}_2$ Name How many moles of hydrogen are needed to completely react with two moles

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Stoichiometry Mixed Problems Instructional Fair Answers

ANSWER: 9.3×10^{21} atoms 6) A 4.0g sample of O_2 is in a container at STP. What volume is the container? ANSWER: 2.8 L 7) How many molecules of O_2 are in the container in Problem 6? How many atoms of oxygen are in the container? ANSWER: 7.5×10^{22} molecules Chemistry IF8766 Page 53 Instructional Fair, Inc.

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MIXED MOLE PROBLEMS WORKSHEET ANSWERS CHEMISTRY IF8766

Stoichiometry: Mixed Problems (KEY) 1) $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ What volume of NH_3 at STP is produced if 25.0 of N_2 is reacted with an excess of H_2 ? 3 3 3 2 3 2 2 40.0L NH_3 1mol NH_3 22.4L NH_3 1mol N 2mol NH_3 28.0g N 25.0g N 1mol N $\times \times \times =$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ If 5.0g of KClO_3 is decomposed, what volume of O_2 is produced at STP? 2

Stoichiometry: Mixed Problems (KEY)

How many moles of oxygen are necessary to react completely with four moles of propane (C_3H_8)? 5. + 3KNO_3 + How many moles of potassium nitrate are produced when two moles of potassium phosphate react with moles of aluminum nitrate? enstructional Fair, Inc. Chemistry IF8766 62

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The Mole and Volume Directions ... What volume will the following quantities of gasses occupy at STP? 1) 5.00 mole of H_2 ANSWER: 112. L 2) 2.50 moles of O_2 ANSWER: 56.0 L 3) 0.850 moles of N ... ANSWER: 87.5 L 8) 56.0g of N 2 ANSWER: 44.8 L 9) 70. g of CO_2 ANSWER: 36. L 10) 20. g of NH_3 3 ANSWER: 26 L Chemistry IF8766 Page 51 Instructional ...

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Answer Key. Stoichiometry: Mole-Mole Problems. $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$. How many moles of hydrogen are needed to completely react with 2.0 moles of nitrogen? 6.0 moles of hydrogen . 2. $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$. How many moles of oxygen are produced by the decomposition of 6.0 moles of potassium chlorate? 9.0 moles of oxygen .

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