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chemical reaction release of energy as heat, release of energy as light, production of sound, change in temperature, release or absorption of electrical energy, formation of a gas, formation of a precipitate, change in color, change in odor name some signs that a chemical reaction has taken place.

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Chemical Equations and Reactions Section Quiz: Types of Chemical Reactions In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1. What type of chemical reaction results in the formation of a single product? . combustion . synthesis c. decomposition d. displacement 2.

### **brearleyhigh.kenilworthschools.com**

Convert the following word equation into a balanced chemical equation: aluminum metal copper(II) fluoride → aluminum fluoride copper metal  $2\text{Al(s)} + 3\text{CuF}_2\text{(aq)} \rightarrow 2\text{AlF}_3\text{(aq)} + 3\text{Cu(s)}$  5. One way to test the salinity of a water sample is to add a few drops of silver nitrate solution with a known concentration.

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### **Holt Physical Science Chapter 14: Chemical Reactions ...**

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Chemical Reactions - Ch. 8. I. Intro to Reactions II. Balancing Equations III. Types of Reactions IV. Reaction Energy V. Reaction Rate. Stoichiometry - Ch. 9. I. Stoichiometric Calculations II. Stoichiometry in the Real World. Gases - Ch. 10 & 11. I. Physical Properties of Gases II. The Gas Laws III.

**Mrs. J's Chemistry Page - Lecture Notes**

Modern Chemistry 2 Chemical Equations and Reactions 6. Gold has a low reactivity and therefore does not corrode over time. 7. In single-displacement reactions, if the activity of the free element is greater than that of the element in the compound, the reaction will take place. 8. Yes; because aluminum is above copper in

**CHAPTER 8 REVIEW Chemical Equations and Reactions**

Holt Chemistry 12 Chemical Equations and Reactions Section: Balancing Chemical Equations In the space provided, write the letter of the term or phrase that best answers the

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