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Chapter 21

Nuclear Chemistry (AP) FAQ. Q: How do I identify the type of a nuclear reaction? A: First of all, you need to identify the particle (alpha, beta, gamma, positron, proton, neutron, electron). Capture is when the particle is added to the main atom in order to make it split (a reactant). Emission is when the split produces the particle (a product).

AP Chemistry Study Guide: Chapter 21, Nuclear Chemistry

Chemistry 11 Final Exam Study Guide Chemistry 11 - Final Exam Study Guide Page 15 When Electronegativities of bonding atoms are the same (as they are in diatomic molecules) or close to the same, they SHARE electrons. Bonds formed when atoms share electrons are called Covalent Bonds .

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bond is indicated by the lowest-numbered carbon atom involved in the bond cis isomerism: identical substituents are on the same side of a double bond trans isomerism: identical substituents are on ...

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Chapters 21-25 Resources

122 Chemistry: Matter and Change • Chapter 21 Study Guide for Content Mastery In your textbook, read about the chemistry of voltaic cells. For each item in Column A, write the letter of the matching item in Column B. Column A Column B 10. One of the two parts of an electrochemical cell, where either oxidation or reduction takes place 11.

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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