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A fictitious element that has two isotopes has an average atomic mass of 153.2. If 39.99% of the atoms of the element have an atomic mass of 150.1, what is the atomic mass of the other isotope?

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1.	<b>1</b> ]	ባcmurry (	Organic	Chemistry	8th	edition)	2.	Kreyszig[]		

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Water is the chemical substance with chemical formula H 2 O; one molecule of water has two hydrogen atoms covalently bonded to a single oxygen atom. Water is a tasteless, odorless liquid at ambient temperature and pressure. Liquid water has weak absorption bands at wavelengths of around 750 nm which cause it to appear to have a blue colour. This can easily be observed in a water-filled bath or ...

### Properties of water - Wikipedia

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### Nonmetal - Wikipedia

1. \_\_\_ 8\_ \_\_\_(Mcmurry Organic Chemistry 8th edition) 2. Kreyszig\_ \_\_\_ \_\_ \_\_ \_\_ \_

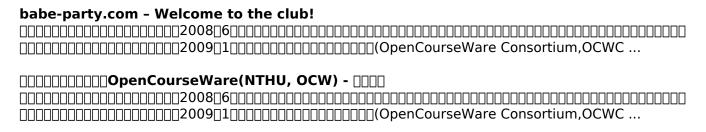
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