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Use the following passage and table to answer questions 11-14. A student studying electron  
configurations in atoms created Table 1 in order to summarize the various facts she learned about  
principal energy levels, energy sublevels, and orbitals. 11. Based on Table 1 what is the relationship  
between the principal energy level number (n)

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Use the following passage and table to answer questions 11-13. A student researched the  
properties of both ionic and covalent compounds. To summarize his findings, he created a table  
comparing the characteristics of the two compounds. Table 2 is shown below. Table 2 Comparison  
of Ionic and Covalent Compounds

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