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AP Chemistry - Darrell Feebeck

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The air we breathe, what we drink, as well as products in our household - these are a few examples of solutions we encounter every day. In this lesson, we will discuss what a solution is, the ...

What is a Solution in Science? - Definition & Examples ...

Moles Lab Activities - VDOE ... 1

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A battery is an electrochemical cell or series of cells that produces an electric current. In principle, any galvanic cell could be used as a battery. An ideal battery would never run down, produce an unchanging voltage, and be capable of withstanding environmental extremes of heat and humidity.

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We care about molality because freezing point depression is a colligative property, a property that depends on how many solute particles are in the solvent, not the kind of solute particles. Molality, m, is one piece of this "how many solute particles are present?" question. The Van 't Hoff factor is the second part of the "how many solute particles are present?"

Chemistry of Ice-Cream Making: Lowering the Freezing Point ...

A fictitious element that has two isotopes has an average atomic mass of 153.2. If 39.99% of the

atoms of the element have an atomic mass of 150.1, what is the atomic mass of the other isotope?

Definition of Raoult's Law | Chegg.com

How to Calculate Vapor Pressure. Have you ever left a bottle of water out in the hot sun for a few hours and heard a slight "hissing" noise when you opened it? This is caused by a principle called vapor pressure. In chemistry, vapor...

3 Easy Ways to Calculate Vapor Pressure (with Pictures)

Acids: An acid is a substance that gives hydrogen ion H + or a hydronium ion H 3 O + when dissolved in water. A substance, which has an acidic nature, contains one or more hydrogen and an anionic group in its formula.

Definition of Acids And Bases | Chegg.com

Once microorganisms have started to grow and become physiologically active they usually produce water as an end product of respiration. Thus they increase the water activity of their own immediate environment so that eventually micro-organisms requiring a high a w are able to grow and spoil a food which was initially considered to be microbiologically stable.

Growth of Micro-Organisms in Food | Microbiology

The preservation of medical laboratory blood samples, mining of sea water for magnesium, formulation of over-the-counter medicines such as Milk of Magnesia and antacids, and treating the presence of hard water in your home's water supply are just a few of the many tasks that involve controlling the equilibrium between a slightly soluble ionic solid and an aqueous solution of its ions.

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