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7. For a chemical reaction it is usually found that the reaction rate is faster at higher temperature. The rate increases because a. the concentration of reactants increase b. more reactants collide with energy equal to or greater than the activation energy c. the concentrations of products increase d.

**California Chemistry Diagnostic Test - Sierra College**

7. For a chemical reaction it is usually found that the reaction rate is faster at higher temperature. The rate increases because (a) the concentrations of reactants increase (b) more reactants collide with energy equal to or greater than the activation energy (c) the concentrations of products increase

**California Chemistry Diagnostic Test Sample Questions**

The molar mass of a substance is the mass of one mole of the substance. These test questions deal with calculating and using molar masses. An example of a molar mass might be: GMM O<sub>2</sub> = 32.0 g or KMM O<sub>2</sub> = 0.032 kg.

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Chemistry Bridging Course. 1. The symbol for the element silicon is: a) S b) Se c) Sr d) Si 2. When a metal atom forms a metallic ion, it usually: a) gains one or more protons b) loses one or more electrons c) gains one or more electrons d) loses one or more protons 3. Atom X has 12 protons in its nucleus. The formula of

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