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A fictitious element that has two isotopes has an average atomic mass of 153.2. If 39.99% of the atoms of the element have an atomic mass of 150.1, what is the atomic mass of the other isotope?

Definition of Raoult's Law | Chegg.com

Chemical Changes in the Blood During Exercise. In previous tutorials ("Hemoglobin and the Heme Group: Metal Complexes in the Blood for Oxygen Transport", "Iron Use and Storage in the Body: Ferritin and Molecular Representations", "Maintaining the Body's Chemistry: Dialysis in the Kidneys") you learned about the daily maintenance required in the blood for normal everyday activities such as ...

pH Buffers in the Blood - chemistry.wustl.edu

In chemistry, an ionic compound is a chemical compound composed of ions held together by electrostatic forces termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called cations and negatively charged ions called anions. These can be simple ions such as the sodium (Na^+) and chloride (Cl^-) in sodium chloride, or polyatomic species such as the ammonium ...

Ionic compound - Wikipedia

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Hydrogen sulfide is the chemical compound with the formula H_2S . It is a colorless chalcogen hydride gas with the characteristic foul odor of rotten eggs. It is very poisonous, corrosive, and flammable. Hydrogen sulfide is often produced from the microbial breakdown of organic matter in the absence of oxygen gas, such as in swamps and sewers; this process is commonly known as anaerobic ...

Hydrogen sulfide - Wikipedia

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