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17-1 CHAPTER 17 EQUILIBRIUM: THE EXTENT OF CHEMICAL REACTIONS 17.1 If the rate of the forward reaction exceeds the rate of reverse reaction, products are formed faster than they are consumed. The change in reaction conditions results in more products and less reactants. A change in reaction

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| time (hours) | 0 | 0.2 | 0.4 | 0.6 | 0.8 | 1 | 1.2 | 1.4 | 1.6 | 1.8 | 2 |
|--------------|---|-----|-----|-----|-----|---|-----|-----|-----|-----|---|
| 0 | 0 | 0.5 | 1 | 1.5 | 2 | 0 | 0.5 | 1 | 1.5 | 2 | 0 |

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