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Worksheet- Nuclear Decay Instructions: Fill in the table below and then use it to figure out what is happening during each type of decay-  $\alpha$  (alpha),  $\beta$  (beta), and  $\gamma$  (gamma) Parent Isotope Particle emitted New, Daughter isotope Alpha, Beta, or gamma Decay? # of protons lost or gained by "parent" Change in mass number a. Ra He 222Rn 86 ...

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Radioactive Decay Answer Key. During beta decay, the mass number and atomic number of the antineutrino are zero. As a proton undergoes radioactive decay to form a positron and a neutron, the positron represents an electron with a positive charge. Write the nuclear equation for the alpha decay of Uranium-238.

### **Radioactive Decay Answer Key - HelpTeaching.com**

AP Chemistry : Radioactive Decay and Nuclear Chemistry Study concepts, example questions & explanations for AP Chemistry. CREATE AN ACCOUNT Create Tests & Flashcards. Home Embed ... In order to answer this question, we'll need to employ the equation for first-order decay. The reason we know that this is a first order decay process is because we ...

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