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pH and pOH The pH of a solution indicates how acidic or basic that solution is. pH range of 0-7 acidic 7 neutral 7-14 basic Since ... $[H^+]$ pH $[OH^-]$ pOH Acidic or Basic 1. $10^{-5} M$ 5 $10^{-9} M$ 9 Acidic 2. $10^{-7} M$ 7 $10^{-7} M$ 7 Neutral 3. $10^{-10} M$ 10 $10^{-4} M$ 4 Basic 4. $10^{-2} M$ 2 $10^{-12} M$ 12 Acidic 5.

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