# Chapter 5 Test The Periodic Law Answers

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#### **Chapter 5 Test The Periodic**

Periodic Table. an arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group. Lanthanides. the 14 elements with atomic numbers 58 to 71. Actinides. the 14 elements with atomic numbers 90 to 103.

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5. Mendeleev's periodic table is based on \_\_\_\_\_ (a) Atomic number (b) Atomic weight (c) Equivalent weight (d) None of these. 6. The two places left empty by Mendeleev in the periodic table are (a) Aluminum and silicon (b) Gallium and germanium (c) Arsenic and antimony (d) Antimony and bismuth . 7. Gallium was named by Mendeleev as (a) Eka ...

#### Chapter - 5 Periodic Classification of Elements MCQ Test 1 ...

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#### Chapter 5: The Periodic Table - Practice Test Questions ...

Chapter 5 TEST: The Periodic Table Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the nucleus. d. atomic mass. \_\_\_\_2.

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# Chapter 5 Test The Periodic Law - sbdc.calpoly.edu

Holt McDougal Modern Chemistry Chapter 5: The Periodic Law Chapter Exam. Holt McDougal Modern Chemistry Chapter 5: The Periodic Law / Practice Exam. Exam Instructions: Choose your answers to the questions and click 'Next' to see the next set of questions.

# Holt McDougal Modern Chemistry Chapter 5: The Periodic Law ...

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

#### 5 The Periodic Law - Jefferson Township Public Schools

\_\_\_\_ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. \_\_\_\_ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

# Chemistry Chapter 5 Exam - Quia

Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. In his periodic table, Mendeleev did not list all of the elements in order

#### Assessment Chapter Test B - Wag & Paws

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# **Chapter 5 Review The Periodic Table Answers - examget.net**

Chapter 5 Periodic Classification of Elements – Test Time limit: 0 Quiz-summary 0 of 29 questions completed Questions: 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 ...

#### Chapter 5 Periodic Classification of Elements - Test ...

This is a quick review of all the sections on chapter 5 of my honors chemistry notes. There are some very important things in this video, including periodic law, periodic trends, and groups of ...

#### Chapter 5: Periodic Law (Chem in 15 minutes or less)

3. Atoms of the elements belonging to the same group of periodic table will have (a) Same number of protons (b) Same number of electrons in the valence shell (c) Same number of neutrons (d) Same number of electrons. 4. Oxides of metal are \_\_\_\_ in nature. (a) Basic (b) Acidic (c) Amphoteric (d) None of these. 5.

# Chapter - 5 Periodic Classification of Elements MCQ Test 2 ...

Chapter 5 Pre-Test Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. Mendeleev organized the chemical elements based on their

#### Chapter 5 Pre-Test - mchsapchemistry.com

5 Chapter Test The Periodic Table Multiple Choice On the line at the left, write the letter of the answer that best completes each statement. 1. The properties of elements reoccur periodically when the elements are arranged in order of increasing . a. atomic mass. b. atomic number. c. atomic radius. d. electron affinity. 2.

#### 5 Chapter Test - MRS. Nairn

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Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. Mendeleev organized the chemical elements based on their a. symbols.

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