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Chemical Reactions Everything that happens in an organism is based on chemical reactions. A chemical reaction is a process that changes one set of chemicals into another set of chemicals. The elements or compounds that enter into the reaction are the reactants. The elements or compounds produced by the reaction are the products. Chemical reactions involve changes in the chemical bonds that join atoms in compounds.

2.4 Chemical Reactions and Enzymes - Weebly

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measure amounts of a substance because chemical reactions often involve large numbers of small particles. 10. Circle the letter of the correct answer. If one carbon atom has an atomic mass of 12.0 amu and one oxygen atom has an atomic mass of 16.0 amu, what is the molar mass of carbon dioxide? a.28.0 amub.44.0 amu c.28.0 gd.44.0 g 11.

Chapter 7Chemical Reactions Section 7.1 Describing Reactions

We will also touch on new topics such as: why chemicals react, characteristics of chemical reactions and how these reactions can be represented. We will also analyze various safety and environmental

issues related to chemical reactions and how chemical reactions can be used to address environmental challenges.

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