# Chemistry Matter And Change Chapter 4 Answers

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Chemical Thermodynamics Example 9.2 The element mercury, Hg, is a silvery liquid at room temperature. The normal freezing point of mercury is  $-38.9^{\circ}$ C, and its molar enthalpy of fusion is  $\Delta$ Hfusion = 2.29 kJ/mol. What is the entropy change of the system when 50.0 g of Hg( I) freezes at the normal freezing

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