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Balancing Chemical Equations -Answer Key Balance the equations below: 1) $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ 3) $2\text{NaCl} + 1\text{F}_2 \rightarrow 2\text{NaF} + 1\text{Cl}_2$ 4) $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 5) $1\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$ 6) $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + 1\text{Al}_2(\text{SO}_4)_3$

Balancing Chemical Equations Answer Key

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Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System. 1. CaCO₃ calcium carbonate____ 2. KCl potassium chloride____ 3. FeSO₄ iron (II) ...
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