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1—6 Review and Working With Numbers Complete the following statements by inserting either "are" or "are not" in the blanks provided. 1. Zeros between two significant digits significant. 2. Zeros to the right of a decimal point that precede all nonzero digits 3. All nonzero digits significant. 4. One or more final zeros used after the decimal ...

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Name Date Class 8—1 Practice Problems 1. 2. 3. 4. What is the molecular shape of carbon tetrachloride (CC14)? What are the bond angles in this molecule?

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2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the dry cell battery. The chemical energy in the ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas 7. kinetic . 8. f . 9. c 10. b 11. a . 12. e .

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3—3 Review and Reinforcement (continued) Use the periodic table to determine how many protons, neutrons, and electrons are present in each of the following atoms. Write your answers in the spaces provided. Atom 12. iodine-125 13. potassium-39 14. iron-56 Protons Neutrons Electrons Write the chemical symbol for each of the ions described below.

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