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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain about 1/5 of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

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CHEMFILE MINI GUIDE TO PROBLEM SOLVING Is the answer ... CHEMFILE MINI-GUIDE TO PROBLEM SOLVING CHAPTER 2 Significant Figures A lever balance used to weigh a truckload of stone may be accurate to the nearest 100 kg, giving a reading of 15 200 kg, for instance. The measure-ment should be written in such a way that a person looking at it will un-

Chemfile Mini Guide To Problem Solving Answer Key

From the coefficients, I took them through calculating the simplest mass relationship to show that it was the same as all the other trials. Using the mass relationship, I asked them to answer a stoichiometry question I had asked earlier in the unit...most were able to quickly see the answer.

ChemFiles - A simple site...

Holt ChemFile: Problem-Solving Workbook 100 Stoichiometry Name Class Date Problem Solving continued COMPUTE EVALUATE Are the units correct? Yes; the answer has the correct units of moles NH 3. Is the number of significant figures correct? Yes; two significant figures is correct because data were given to two significant figures. Is the answer ...

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HOLT CHEMFILE MOLE CONCEPT ANSWER GUIDE

CHEMFILE MINI-GUIDE TO PROBLEM SOLVING •Is the answer reasonable? Yes; the calculation can be esti-mated as (3/4) (6)(40) (3/4) 240 180. 1. What mass of glucose, C 6H 12 O6, would be required to prepare 5.000 103 L of a 0.215 M solution? ans: 1.94 105 g 2. What mass of magnesium bromide would be required to prepare 720. mL of a 0.0939 M

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Holt ChemFile: Problem-Solving Workbook 60 Mole Concept Name Class Date Problem Solving continued EVALUATE Are the units correct? Yes; the answer has the correct units of moles CO 2. Is the number of significant figures correct? Yes; the number of significant figures is correct because the mass of CO 2 was given to three significant figures.

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Name Date Class 2 of 8 CHEMFILE MINI-GUIDE TO PROBLEM SOLVING SAMPLE PROBLEM 1 What is the molarity of a solution that is made by diluting 50.00 mL of a 4.74 M \dots

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