# Chapter 5 Review The Periodic Law Mixed Answer Key

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1. \_\_\_\_ In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design.

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What are the answers to Chapter 5 Review Periodic Law 5-2 Worksheet? 1. a. The group 14 element in Period 4 - Ge [AR] 4s2 3d10 4p2 b. The only metal in Group 15 - Bi [Xe] 6s2 4f14 5d10 6p3 c. The ...

#### What are the answers to Chapter 5 Review Periodic Law 5-2 ...

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Review-Chapter 6 The Periodic Table 1. How did Mendeleev arrange the elements?—By atomic mass. 2. In the modern periodic table, elements are ordered—By atomic number. 3. The modern periodic law states—Elements are arranged in order of increasing elements there is periodic repletion of their physical & chemical properties. 4.

# **Review-Chapter 5 The Periodic Table**

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#### Chapter 5 Review The Periodic Law Section 3 - sbdc.calpoly.edu

CHAPTER 5 The Periodic Law. SECTION 1 History of the Periodic Table. OBJECTIVES. 1. Explain the roles of Mendeleev and Moseley in the development of the periodic table 2. Describe the modern

periodic table. 3. Explain how the periodic law can be used to predict the physical and chemical properties of elements.

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Modern Chemistry 33 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number.

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Chapter 6 Review "The Periodic Table" Chapter 6 Review Which of the following groupings contains only representative elements: a) Cu, Co, Cd, or b) Al, Mg, Li? What is typically true about the electron configurations of the representative elements? The metals in Groups 1A, 2A, and 3A electrons when forming ions.

#### Chapter 6 Review "The Periodic Table" - DBCS Mrs. Marshall

Holt Modern Chemistry Review CHAPTER 5: THE PERIODIC LAW The following pages contain the bulk (but not all) of the information for the chapter 5 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test.

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CHAPTER 5 REVIEW The Periodic Law SECTION 2 SHORT ANSWER Use this periodic table to answer the following questions in the space provided. 1. Identify the element and write the noble-gas notation for each of the following: a. the Group 14 element in Period 4CHAPTER 5 REVIEW The

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CHAPTER 5 REVIEW. The Periodic Law. SECTION 5-3. SHORT ANSWER Answer the following questions in the space provided. 1. When an electron is added to a neutral atom, energy is . (a) always absorbed (c) either absorbed or released. (b) always released (d) burned away.

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