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Holt Chemistry 6 Covalent Compounds Name Class Date Concept Review continued 10. Draw the Lewis structure for water, H 20. 11. Draw the Lewis structure for carbon monoxide, CO. 12. Draw the Lewis structure for the rocket propellant nitryl fluoride, NO 2F. 13. Draw the Lewis structures for SO 2 3 2. Answer the following item in the space ...

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Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of measure-ments of the same quantity made in the same way agree with one another. 2. Answers may vary. Look at the defense of the answer. 3. a. 7 b ...

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Concept Review: Trends in the Periodic Table 1. ionization energy 2. bond radius 3. electron affinity 4. electronegativity 5. increases, decreasing 6. d 7. a 8. c 9. The electron cloud model is based on the probability of finding an electron at a specific location. As you move far-ther out from the nucleus, the proba-bility of finding electrons ...

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Concept Review: Equilibria of Weak Acids and Bases 1. 20, water 2. OH 3. H 20, H 2PO 4, HCO 3,

HPO 4 2 4. HCOO 5. NH 4 6. NH 4 7. ClO 8. HCOOH 9. It increases. 10. It decreases. 11. stronger, stronger 12. higher, lower ... Holt Chemistry 165 Acids and Bases Answer Key TEACHER RESOURCE PAGE. Created Date:

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Answer the following questions in the space provided. 1. Why do Li, Na, K, Rb, Cs, and Fr all react with Cl in a 1:1 ratio forming sub- ... Holt Chemistry 25 The Periodic Table Section: Tour of the Periodic Table ... Concept Review continued 14. Why are the main-group elements called the representative elements? 15. Why are Group 2 elements ...

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