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the Properties of Metals pages 225–228 Section 7.4 Assessment pages 228 40. Contrast the structures of ionic compounds and metals. Ions in ionic compounds are arranged in a repeating pattern of alternating charges, whereas metals consist of fixed cations surrounded by a sea of mobile, or delocalized, electrons. 41.

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7.2 Ionic Bonds and Ionic Compounds MAIN Idea Oppositely charged ions attract each other, forming electrically neutral ionic compounds. 7.3 Names and Formulas for Ionic Compounds MAIN Idea In written names and formulas for ionic compounds, the cation appears first, followed by the anion. 7.4 Metallic Bonds and the Properties of Metals

Chapter 7: Ionic Compounds and Metals

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