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a. Find the volume of 1.0M tartaric acid, H2C4H4O6, solution that when diluted to a total volume of 250.0ml produces a solution that is 0.050M in tartaric acid.b. Find the mass of potassium...

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Name ____ Lab Section ____ CHEM 1515 1 Spring 2001 Experiment #3: Shifting Reactions (Adapted from Exp. I-4 from Inquiries in Chemistry, 3rd edition) Problem Statement: How can we shift reactions forward and backward? I. Data Collections and Analysis A. Put about 50 mL of KSCN (potassium thiocyanate solution labeled 0.0005 ...

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