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force that pulls adjacent parts of a liquid's surface together, decreasing surface area to the smallest possible size

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26. According to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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40. 7.05 1016 Hz 41. 1.28 s 42. 4.58 10 19 J The Periodic Law, pp. 36–45 TEST A 1. b 2. d 3. b4. 5. d 6. a 7. b8. 9. b 10. a 11. c 12. a 13. c 14. d 15. c 16. b

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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test A Chapter: The Periodic Law Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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