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CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

6 Chemical Bonding - Effingham County Schools / Overview

40. 7.05 1016 Hz 41. 1.28 s 42. 4.58 10 19 J The Periodic Law, pp. 36-45 TEST A 1. b 2. d 3. b4. 5. d 6. a 7. b8. 9. b 10. a 11. c 12. a 13. c 14. d 15. c 16. b

Assessment Chapter Test B - clarkchargers.org

CHAPTER 10 REVIEW States of Matter SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

10 States of Matter - Ms. Agostine's Chemistry Page

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or

heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar
2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

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triple double single resonance Lewis valence unshared 1. An electron in the outermost energy level of an atom that can participate in

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