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Use the following three acids to answer questions 33 and 34: iodic acid hypoiodous acid periodic acid 33. Give the formulas for these three acids. 34. List the acids in order of increasing acid strength.

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Amino acid side chains can affect the shape of a protein by ______ 8. When an enzyme interacts with a substrate, the enzyme _____ 9. When a protein is denatured by heat, it _____ 10. Genetic information is stored in nucleic acids by the a. arrangement of the bases in a sequence. b. hydrogen bonding between bases. c. arrangement of sugars in a sequence. d.

One product of the reaction of amino acids to form a ...

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reactants is the Lewis acid? Is this reactant also a Brønsted-Lowry acid? Explain your answers. BCl 3 is the Lewis acid because it accepts an electron pair in forming a covalent bond. It is not a Brønsted-Lowry acid, because it is not donating a proton. e. Which of the reactants is the Lewis base? Explain your answer. NF

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Modern Chemistry 121 Acids and Bases CHAPTER 14 REVIEW Acids and Bases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Answer the following questions according to the Brønsted-Lowry definitions of acids and bases: _____ a. What is the conjugate base of H 2SO 3? _____ b.

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