

Introduction To Matlab For Engineers Chapter 3 Solutions

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State the problem concisely. Specify the data to be used by the program. Specify the information to be generated by the program. Work through the solution steps by hand or with a calculator; use a simpler set of data if. Write and run the program. Check the output of the program with your hand ...

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$P_{ideal} = 1.0004$, $P_1 = nRT/(V - nb)$ $P_1 = 1.0029$, $P_2 = (a*n^2)/V^2$ $P_2 = 0.0129$, $P_{waals} = P_1 + P_2$ $P_{waals} = 1.0158$ The ideal gas law predicts a pressure of 1.0004 atmospheres, while the van der Waals model predicts 1.0158 atmospheres.

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