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What causes a gas to exert pressure? a. collisions b. density c. temperature d. elevation ____ 2. The SI unit for pressure is a. newton. b. mm Hg. c. pascal. d. liter. ____ 3. The pressure exerted by a gas does not depend on a. temperature. b. volume. c. number of moles present. d. the identity of the gas. ____ 4. At sea level, the average ...

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Mrs. J's Chemistry Page - Lecture Notes

Concept Review j j. Gas particles may move apart to fill any container they occupy. This behavior occurs because gas particles are [close together, far apart] and are [attracted, not strongly attracted] to one another. 6. [Physical, Chemical] changes are changes in which the identity of a

substance does not change.

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Properties of a Gas. Pressure is a measure of the amount of force per unit area. The pressure of a gas is the amount of force the gas exerts on a surface within its volume. Gases with high pressure exert more force than gas with low pressure. The SI unit of pressure is the pascal (Symbol Pa).

Chemistry Study Guide for Gases - ThoughtCo

CHAPTER 10 REVIEW Physical Characteristics of Gases Modern Chemistry 109 Chapter Test
Chapter: Solutions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

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