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Holt McDougal Modern Chemistry 1 Chapter Test Assessment Chapter Test B Chapter: The Periodic Law PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. In his periodic table, Mendeleev did not list all of the elements in order

Assessment Chapter Test B - Wag & Paws

Taken from the book Modern Chemistry by Holt, Rinehart, and Winston on Chapters 4 and 5, which deals with electrons and the periodic table. Includes the chapter vocabulary and a few other useful things.

Modern Chemistry (Holt, Rinehart, and Winston): Chapter 4 ...

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Modern Chemistry 1 The Periodic Law CHAPTER 5 The Periodic Law SECTION 1 History of the Periodic Table OBJECTIVES 1. Explain the roles of Mendeleev and Moseley in the development of the periodic table 2. Describe the modern periodic table. 3.

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40. 7.05 1016 Hz 41. 1.28 s 42. 4.58 10 19 J The Periodic Law, pp. 36-45 TEST A 1. b 2. d 3. b4. 5. d 6. a 7. b8. 9. b 10. a 11. c 12. a 13. c 14. d 15. c 16. b

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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