

Half Life Practice Answer Key

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Half Life of Radioactive Isotopes.docx Sh«jir Open New! Edit your docs with Microsoft Word Online. Half Life of Radioactive Isotopes Name: Date: 1. Carbon -14 is a radioactive isotope found in small amounts in all living things. When the living thing dies, the carbon -14 begins to decay at a steady rate with a half -life of 5,730 years ...

HALF-LIFE PROBLEMS

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Problem #3: Os-182 has a half-life of 21.5 hours. How many grams of a 10.0 gram sample would have decayed after exactly three half-lives? Solution: $(1/2)^3 = 0.125$ (the amount remaining after 3 half-lives) $10.0 \text{ g} \times 0.125 = 1.25 \text{ g}$ remain $10.0 \text{ g} - 1.25 \text{ g} = 8.75 \text{ g}$ have decayed Note that the length of the half-life played no role in this calculation.

ChemTeam: Half-Life Problems #1 - 10

Half Life Worksheet - Extra Practice: Solutions 1) Fluorine-21 has a half life of approximately 5 seconds. What fraction of the original nuclei would remain after 1 minute? The answer is solved by creating the fraction 2^n 1. Where n = the number of half lives. If each half life is 5 seconds, then in one minute (60 seconds) there are 12 half ...

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Monohybrid Cross Practice Problems Worksheet Answers - Repaying the debt and becoming financially independent has many important components. The absolute most essential of the would be creating a budget. A funding offers you a plan of exactly where your money is moving and exactly where it needs to go.

Monohybrid Cross Practice Problems Worksheet Answers ...

Carbon-14 has a half-life of 5730 years. That is, if you take one gram of C-14, half of it will decay in 5730 years. Cobalt-60 5 years Protactinium-226 2 minutes Iodine-131 8 days Americium-242 6 hours Tin-126 100,000 years This quiz covers half-life. Use the above information to answer the

following questions.

Nuclear Chemistry : Half-Life Quiz - Softschools.com

HALF-LIFE WORKSHEET 1. What is radioactivity? 2. What is half-life? ... Use the following graph to answer questions 7-10. 7. How long is a half-life for carbon-14? ____ 8. If only 25% of the carbon-14 remains, how old is the material containing the carbon-14? ____ 9. If a sample originally had 120 atoms of carbon-14, how many atoms will remain ...

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