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dissolves. The ions, already present, separate from one another. Ionization is the process of forming ions from the solute molecules by the action of the

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water to produce a sulfuric acid solution that falls to the ground as rain or snow. $\text{SO}_3(\text{g}) + \text{H}_2\text{O}(\text{l}) \rightarrow \text{H}_2\text{SO}_4(\text{aq})$ 37. Have a sour taste; change the color of acid-base indicators; some react with

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