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Holt ChemFile: Problem-Solving Workbook 55 Mole Concept. Name Class Date. Problem Solving continued. CONVERTING THE AMOUNT OF AN ELEMENT IN MOLES TO THE NUMBER OF ATOMS In Sample Problem 3, you were asked to determine the number of moles in 1.204 1024 atoms of lithium.

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CHAPTER 4 Problem Solving Although engineers use science to solve problems, ... The approach to solving an engineering problem should proceed in an orderly, . where Cd is the drag coefficient (dimensionless), r is the air density (kg/m3), A is the pro- . st.

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Holt ChemFile: Problem-Solving Workbook 3 Conversions Name Class Date Problem Solving continued The correct conversion factor is the one that when multiplied by the given quantity causes the units to cancel. COMPUTE 45.5 g calcium chloride 100 1 0 g mg 45 500 mg calcium chloride EVALUATE Are the units correct? Yes; milligrams are the desired units.

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Name Class Date. Problem Solving continued. MOLARITY Molarity is the most common way to express concentration in chemistry. Molarity is the number of moles of solute per liter of solution and is given as a number fol- lowed by a capital M. A 2 M solution of nitric acid contains 2 mol of HNO. 3 per liter of solution.

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