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Holt Chemistry Chapter 11: States of Matter and ...

the force that acts on the surface of a liquid and that tends.... the change of a substance from a liquid to a gas. the temperature and pressure at which a liquid and a gas are i.... the change of state from a gas to a liquid.

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a force that acts on the surface f a liquid and that tends to minimize the area of the surface. evaporation. the process by which water changes from liquid form to an atmospheric gas. boiling point. the temperature at which a liquid boils at sea level. condensation. the process of changing from a gaseous to a liquid or solid state.

Holt Chemistry Ch. 11 States of Matter and Intermolecular ...

These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

Section 2.6.2 - Intermolecular Forces Example 2 Holt Chemistry Ch. 11 States of Matter and Intermolecular Forces. chemistry. STUDY. PLAY. surface tension. a force that acts on the surface f a

liquid and that tends to minimize the area of the surface.... (chemistry) a change directly from the solid to the gaseous state without becoming liquid ...

Chemistry Concept Review Section Intermolecular Forces Answers

Water has a greater adhesion to glass than to car wax. The forces of adhesion and cohesion will pull water up a narrow glass tube, called a capillary tube, shown in Figure 3. The adhesion of the water molecules to the molecules that make up the glass tube pulls water molecules up the sides of the tube.

CHAPTER

The units cover: the science of chemistry, matter and energy, atoms and moles, the Periodic Table, ions and ionic compounds, covalent compounds, the mole and chemical composition, chemical equations and reactions, stoichiometry, causes of change, states of matter and intermolecular forces, gases, solutions, chemical equilibrium, acids and bases ...

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Intermolecular Forces • List the substances BaCl₂, H₂, CO, HF, and Ne in order of increasing boiling points. • The attractive forces are stronger for ionic substances than for molecular ones • The intermolecular forces of the remaining substances depend on molecular weight, polarity, and hydrogen bonding. The

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