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Use the periodic table below to answer the questions in this Chapter Test. In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

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Assessment Chapter Test B - Wag & Paws

Organic chemistry the study of carbon-containing compounds (typically chains of carbon atoms) and their properties. (22) Osmosis the flow of solvent into a solution through a semipermeable membrane. (11.6) Osmotic pressure (p) the pressure that must be applied to a solution to stop osmosis; $p = MRT$. (11.6) Ostwald ...

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together by chemical bonds. Carbon dioxide and water are two examples, but answers will vary. 3. Rutherford's experiment involved shooting a beam of particles at a thin sheet of metal foil. According to the then-current "plum pudding" model of the atom, most of these positively charged particles should have passed right through the foil.

Chapter 3: Standard Review Worksheet

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