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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

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CHAPTER 8 REVIEW Chemical Equations and Reactions Teacher Notes and Answers Chapter 8 SECTION 1 SHORT ANSWER 1. a. d b. a c. b d. f e. e f. c 2. 8,4,9 3. a. 12 atoms ... Modern Chemistry 2 Chemical Equations and Reactions 6. Gold has a low reactivity and therefore does

CHAPTER 8 REVIEW Chemical Equations and Reactions

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

8. a 9. d 10. c Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of measure-ments of the same quantity made in the same way agree with one another. 2. Answers may vary. Look at the defense of the answer ...

Skills Worksheet Concept Review - Woodland Hills School ...

Modern Chemistry 32 Chapter Test Name Class Date Chapter Test B, continued _____ 6. When electrons change energy states, the amount of energy given off or absorbed is equal to a. hc. b. hv. c. n. d. cv. _____ 7. Max Planck proposed that "hot" objects radiate energy in small, specific amounts called a. quanta. b. waves. c. hertz. d ...

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