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Holt Chemistry 6 Covalent Compounds Name Class Date Concept Review continued 10. Draw the Lewis structure for water, H₂O. 11. Draw the Lewis structure for carbon monoxide, CO. 12. Draw the Lewis structure for the rocket propellant nitryl fluoride, NO₂F. 13. Draw the Lewis structures for SO₂ and SiO₂. Answer the following item in the space ...

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and

nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

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Holt Chemistry 1 The Mole and Chemical Composition Section: Avogadro's Number and Molar Conversions Solve the following problems, and write your answer in the space provided.

Skills Worksheet Concept Review

CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

6 Chemical Bonding - Effingham County Schools / Overview

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMs r i-vi

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Holt Chemistry 5 The Mole and Chemical Composition Section: Formulas and Percentage Composition Solve the following problems, and write your answer in the space provided.

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