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Holt Chemistry 1 Matter and Energy ... 14. Substance A has a specific heat of 0.650 J/g K, and substance B has a specific ... substance, which substance will have a higher temperature? Why? Concept Review: Energy 1. energy 2. physical 3. chemical 4. endothermic 5. exothermic 6. kinetic 7. transferred 8. In any chemical or physical change,

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Holt Chemistry 6 Covalent Compounds Name Class Date Concept Review continued 10. Draw the Lewis structure for water, H₂O. 11. Draw the Lewis structure for carbon monoxide, CO. 12. Draw the Lewis structure for the rocket propellant nitryl fluoride, NO₂F. 13. Draw the Lewis structures for SO₂ 3 2. Answer the following item in the space ...

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Holt Chemistry 4 Matter and Energy Name Class Date ... 14. Substance B will have the higher temperature. Each gram of substance B ... Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the

Skills Worksheet Concept Review - Woodland Hills School ...

Holt Chemistry 26 The Periodic Table Name Class Date Concept Review continued 14. Why are the main-group elements called the representative elements? 15. Why are Group 2 elements less reactive than Group 1 elements? 16. Explain why a helium atmosphere is used in welding instead of an oxygen-

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Holt Chemistry 6 Stoichiometry Section: Limiting Reactants and Percentage Yield ... 14. 1.34 g Ag (1 mol Ag/107.87 g Ag) (3 mol AgNO₃/3 mol Ag) (169.88 g AgNO₃/1 mol AgNO₃) 2.11 g AgNO₃ percentage yield (actual yield/theoretical yield) x 100 (2.10 g AgNO₃ ... Concept Review: Stoichiometry and Cars 1. c 2. b 3. d 4. b 5. d 6. If there is too ...

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