# Ionic Metallic Bonding Pearson Answers

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7.2 Ionic Bonds and Ionic Compounds Essential Understanding Ionic compounds are the result of ionic bonds forming between oppositely charged ions. Lesson Summary Formation of Ionic Compounds An ionic compound is made up of anions and cations and has an overall charge of 0. The electrostatic attraction between an anion and a cation is an ionic bond.

# **BONDING AND INTERACTIONS**

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7 Ionic and Metallic Bonding Vocabulary Review. STUDY. PLAY. Compounds composed of cations and anions. Ionic compounds. The attraction of free-floating valence electrons for positively charged metal ions. Metallic bonds. The lowest whole-number ratio of ions in an ionic compound. Formula unit.

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# Ionic and Metallic Bonding - Pittsfield High School

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7.2 Ionic Bonds and Ionic Compounds > 27 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. Would you expect to find sodium

#### 7.2 Ionic Bonds and Ionic Compounds > CHEMISTRY YOU

b. held together by ionic bonds. d. all of the above $\_\_\_16$	6. Which of these is nota characteristic of
most ionic compounds? a. solid at room temperature b. ha	as a low melting point c. conducts an
electric current when melted d. produced by reaction bety	ween metallic and nonmetallic elements
17. A metallic bond is a bond between:	

#### 15 Ionic Bonding and Ionic Compounds Chapter Test A - LPS

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