

***Holt Chemistry Concept Review Classifying Chemical Reactions
Answers***

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Concept Review j j. Section: What is ChēiSti'Y Cornpiete each statement below by underlining the correct word or phrase in brackets. 1. A chemical is any substance that has [definit,e, indefinite} composition. Changes in chemicals, or chemical reactions, take place [only in test tubes, all. 2.

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Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of measure-ments of the same quantity made in the same way agree with one another. 2. Answers may vary. Look at the defense of the answer. 3. a. 7 b ...

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Concept Reviews SECTION: THE NATURE OF CHEMICAL REACTIONS 1. a. reactants: Fe 2O 3 and Al; products: Fe and Al 2O 3 b. reactants: AgNO 3 and H 2SO 4; products: Ag 2SO 4 and HNO 3 2. When the natural gas burns, the bonds in the molecule are broken and the

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Concept Reviews SECTION: WHAT IS MATTER? 1. a. heterogeneous b. homogeneous c. heterogeneous d. heterogeneous e. homogeneousf. 2. An atom is the smallest particle that has the properties of an element, while a molecule is the smallest unit of a substance that has the properties of that substance. Molecules consist of one atom or two or more ...

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Holt Chemistry 5 Covalent Compounds Section: Drawing and Naming Molecules Complete each statement below by choosing a term from the following list. Terms may be used more than once. ... Concept Review: Covalent Bonds 1. A covalent bond forms when two or more valence electrons are attracted

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Holt Chemistry 6 Stoichiometry Section: Limiting Reactants and Percentage Yield Complete each statement below by choosing a term from the following list. ... Concept Review: Stoichiometry and Cars 1. c 2. b 3. d 4. b 5. d 6. If there is too much oxygen and not enough gasoline, the engine will stall.

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Concept Review: Avogadro's Number and Molar Conversions 1. 2.41 10²⁴ atoms Al 2. 9.33 31023 atoms Na 3. 44.0 mol F 4. 70.7 mol H 2 5. 291 mol K 6. 877 g Sb 7. 119 g U 8. 167 g Fr

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Concept Review: Measurements and Calculations in Chemistry 1. Accuracy is the extent to which a measurement approaches the true value of a quantity; precision is the extent to which a series of measure-ments of the same quantity made in the same way agree with one another. 2. Answers may vary. Look at the defense of the answer. 3. a. 7 b ...

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

Holt Chemistry 13 Chemical Equations and Reactions Name Class Date Quiz continued ____ 6. Which statement is true for the balanced formula equation? a. The coefficients are the same for each substance. b. ... Quiz—Section: Classifying Chemical Reactions 1. c6. 2. b 7. a 3. b 8. c 4. a 9. d 5. a 10. b

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