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Kinetic Molecular Theory(POGIL) - PO G IL K inetiC M... This preview shows pages 1-2. Sign up to view the full content. Modified from Foundations of Chemistry by David Hanson Page 1 of 3 POGIL: Kinetic Molecular Theory Learning Objectives □ Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases.

### **Kinetic Molecular Theory(POGIL) - PO G IL K inetiC M ...**

Modified from Foundations of Chemistry by David Hanson Page 1 of 3. POGIL: Kinetic Molecular Theory. Learning Objectives. □ Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases. □ Develop an understanding of the postulates of the kinetic molecular theory.

### **POGIL: Kinetic Molecular Theory**

Review Questions Answer KEY 1. Kinetic molecular theory is the theory that explains the motion of solids, liquids, and gases. 2. KMT explains the differences between properties of solids, liquids, and gases by examining how the particles are moving under similar conditions. 3. a. Solid – a phase of matter with a definite shape and volume.

### **Review Questions Answer KEY**

TOPIC: Kinetic-Molecular Theory (KMT) Do Now: packet page \_\_ Relationship between Temp and Average Kinetic Energy Models Scientists use models because simpler than real life Kinetic-molecular theory can be extended to all phases of matter Kinetic-Molecular Theory (KMT) as it pertains to ideal gases There are 4 assumptions to the KMT Real vs. ideal gas Ideal gas obeys assumptions of KMT all the ...

### **Kinetic-Molecular Theory - Wappingers Central School District**

Structure and Dynamics) as needed, and ChemActivity Exercises & Problems Pages 640-644 Chp15 ChemActivity on Nuclear Chemistry Kinetics ... Summary of General Chemistry Redesign Project

### **Pogil Answer Key Chemistry Kinetics - pdfsdocuments2.com**

Kinetic Molecular Theory Answer Key. There are constant forces of attraction between the molecules. There are constant forces of repulsion between the molecules. There are constant forces of attraction and repulsion between the molecules. There are no attractive or repulsive forces between the molecules.

### **Kinetic Molecular Theory Answer Key - HelpTeaching.com**

Kinetic Molecular Theory Worksheet 1. The word kinetic comes from a Greek word that means “to move.” The kinetic molecular theory is based upon the assumption that particles of matter (atoms or molecules) are in constant random motion. 2. Of the three states of matter, which one has the most kinetic energy? Gas 3.

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