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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about  $\frac{1}{5}$  of the molar mass of boron. Therefore, 2.00 g boron will contain about  $\frac{1}{5}$  of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

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Holt ChemFile: Problem-Solving Workbook 122 Percentage Yield ... It can be produced by passing chlorine gas over heated mercury(II) oxide according to the following equation: ... Is the answer reasonable? Yes; 83% is about 5/6, which appears to be close to the ratio 0.71/0.86.

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