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Modern Chemistry 112 Chapter Test Name Class Date Chapter Test B, continued 27. Compare the properties of solutions, suspensions, and colloids. 28. Ethanol dissolves in water, but carbon tetrachloride does not. What can you conclude about ethanol and carbon tetrachloride? PART IV Write the answers to the following questions on the line to the ...

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SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton repulsions.

6 Chemical Bonding - Effingham County Schools / Overview

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Assessment Chapter Test A - Kettering City School District

Modern Chemistry 35 Chapter Test Name Class Date Chapter Test B, continued Write the orbital notation for the following elements in the space provided. 37. lithium, atomic number 3 38. carbon, atomic number 6 39. neon, atomic number 10 PART VI Write the answers to the questions on the line to the left, and show your work in the space provided.

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