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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain about 1/5 of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

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this information to perform calculations and to answer the questions at the end of the experiment. SAFETY ... Holt ChemFile B viii Safety in the Chemistry Laboratory mc06SE\_cFMme\_i-xviii.qxd 1/31/05 11:59 AM Page viii. 19. If you get a chemical in your eyes, walk immediately to the eyewash

## mc06SE cFMme i-xviii

Holt ChemFile: Problem-Solving Workbook 122 Percentage Yield Percentage Yield Although we can write perfectly balanced equations to represent perfect reac- ... Is the answer reasonable? Yes; 83% is about 5/6, which appears to be close to the ratio 0.71/0.86. Practice 1.

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Holt ChemFile: Problem-Solving Workbook 193 Stoichiometry of Gases Name Class Date Problem Solving continued Additional Problems 1. The industrial production of ammonia proceeds according to the following equation. N 2 (g) 3H 2 (g)  $\rightarrow$  2NH 3 (g) a. What volume of nitrogen at STP is needed to react with 57.0 mL of hydro-gen measured at STP? b.

## Holt ChemFile Problem Solving Workbook 191 Stoichiometry ...

Holt ChemFile: Problem-Solving Workbook 56 Mole Concept Name Class Date Problem Solving continued Sample Problem 4 How many boron atoms are there in 2.00 g of boron? Solution ANALYZE What is given in the problem? mass of boron in grams What are you asked to find? number of boron atoms PLAN What steps are needed to convert from grams of B to ...

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