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John Erickson Chemistry 12 2 © John Erickson, 2005 WS16-2BondEnergy example  $H_2(g) + F_2(g) \rightarrow 2HF(g)$  rxn = [energy used for breaking bonds] - [energy formed in making bonds] Bond Energy Name Chem Worksheet 16-2 John Mackovic (born October 1, 1943) is an American football coach and former player. He is

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Chemistry Calculation Review Name Chem Worksheet 12-1 © John Erickson, 2005 WS12-1ChemCalcReview The first step in performing a stoichiometric calculation is to write a balanced equation.

### Chemistry Calculation Review Name Chem Worksheet 12-1

What volume is needed to make 2.00 L of 1.00 M solution? 4. Concentrated HCl is 12.0 M. What volume is needed to make 2.00 L of 1.00 M solution? 5. A 0.500 M solution is to be diluted to 500.0 mL of a 0.150 M solution. How many mL of the 0.500 M solution are required? 6. A stock solution of 10.0 M NaOH is prepared.

### CHEMISTRY L E E U N I T S E V E N - CHEMSTEM - Home

$2O_2 + 2H_2O \rightarrow 4H_2O$  4. How many liters of oxygen at STP are needed to react with  $5.2 \times 10^2$  molecules of hydrogen sulfide?  $H_2S + O_2 \rightarrow SO_2 + H_2O$  5. What mass of chlorine gas is necessary to synthesize 258 L of hydrogen chloride at STP?  $H_2 + Cl_2 \rightarrow HCl$  6. If there are  $6.2 \times 10^2$  molecules of calcium carbide,  $CaC_2$ , what mass ...

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