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= 8.0 g H 2 c. How many moles of nitrogen are required to produce 11.7 moles of NH 3? ? mol N 2 = 11.7 mol NH 3! 1 mol N 2 2 mol NH 3 = <math>5.85 mol N 2 d. How many moles of nitrogen are required to produce 11.7 grams of NH 3? ? mol N 2 = 11.7 g NH 3! 1 mol NH 3 17.0 g NH 3! 1 mol N 2 2 mol NH 3 = <math>0.344 mol N 2 e. How many grams of hydrogen are ...

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