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Chemistry 12 Unit 2 Notes - Equilibrium Unit 2 Notes - Equilibrium Page 3 Now what we're going to do is look at how the rate of the forward reaction changes if we put some pure N_2O_4 in a flask: If we put some pure N_2O_4 in a flask (No NO_2 yet!), there will be a high concentration of N_2O_4 . That is, there will be lots of N_2O_4 molecules to collide with each other.

Chemistry 12 Unit 2- Equilibrium Notes

acid_base_midunit_practice_test_key.pdf: File Size: 590 kb: File Type: pdf

Chemistry 12 - Mr. C's Math and Science Website

Hebden - Unit 2 (page 37-69) The strength of the chemical bonds before and after the chemical reaction. •When bonds are broken in the reactants, energy is required. Wh b d f d CHEM 0012 Lecture Notes 33 •When new bonds are formed as products, energy is released. Dynamic Equilibrium Hebden - Unit 2 (page 37-69)

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