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Holt ChemFile: Problem-Solving Workbook 57 Mole Concept Name Class Date Problem Solving continued Is the answer reasonable? Yes; 2 g of boron is about 1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain about 1/5 of an Avogadro's constant of atoms. Practice 1. Calculate the number of atoms in each of the following masses:

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Holt ChemFile: Problem-Solving Workbook 58 Mole Concept Name Class Date Problem Solving continued CONVERTING NUMBER OF ATOMS OF AN ELEMENT TO MASS Sample Problem 4 uses the progression of steps 1 → 2 → 3 to convert from the mass of an element to the number of atoms.

**Holt ChemFile Problem Solving Workbook 58 Mole Concept ...**

Holt ChemFile Problem Solving Workbook 191 Stoichiometry of Gases Name Class from HISTORY 202 at Union County College

**Holt ChemFile Problem Solving Workbook 191 Stoichiometry ...**

Holt ChemFile: Problem-Solving Workbook 122 Percentage Yield Percentage Yield Although we can write perfectly balanced equations to represent perfect reac- ... Is the answer reasonable? Yes; 83% is about 5/6, which appears to be close to the ratio 0.71/0.86. Practice 1.

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Holt ChemFile: Problem-Solving Workbook 100 Stoichiometry Name Class Date Problem Solving continued COMPUTE EVALUATE Are the units correct? Yes; the answer has the correct units of moles NH<sub>3</sub>. Is the number of significant figures correct? Yes; two significant figures is correct because data were given to two significant figures. Is the answer ...

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Holt Chemistry Chapter 9: Stoichiometry Chapter Exam Instructions. Choose your answers to the questions and click 'Next' to see the next set of questions.

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Is the answer reasonable? Yes; you would expect the pOH of an acid to be above 7, and the hydroxide ion concentration to be small. Practice 1. What is the pH of a solution with the following hydroxide ion concentrations? a.  $1 \times 10^{-5}$  M ans: 9.0 ... Holt ChemFile: Problem-Solving Workbook 271 pH

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Holt ChemFile: Problem-Solving Workbook 1 Conversions Conversions One of the aims of chemistry is to describe changes—to tell what changed, how it changed, and what it changed into. Another aim of chemistry is to look at matter ... Is the answer reasonable? Yes; the number of milligrams is 1000 times the number of grams. Practice 1.

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This reference is a must for students who need extra help, reteaching, or extra practice. The guide moves students through the same concepts as the text, but at a slower pace. More descriptive detail, along with visual algorithms, provides a more structured approach. Each chapter closes with a large bank of practice problems. Book jacket.

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