

Half Life Problems Answer Key

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HALF-LIFE PROBLEMS Name Block 1. An isotope of cesium (cesium-137) has a half-life of 30 years. If 1,0 g of cesium-137 disintegrates over a period of 90 years, how many g of cesium-137 would remain?

HALF-LIFE PROBLEMS

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Problem #3: Os-182 has a half-life of 21.5 hours. How many grams of a 10.0 gram sample would have decayed after exactly three half-lives? Solution: $(1/2)^3 = 0.125$ (the amount remaining after 3 half-lives) $10.0 \text{ g} \times 0.125 = 1.25 \text{ g}$ remain $10.0 \text{ g} - 1.25 \text{ g} = 8.75 \text{ g}$ have decayed Note that the length of the half-life played no role in this calculation.

ChemTeam: Half-Life Problems #1 - 10

Half Life Problems Answer Key Half Life Problems Answer Key HALF-LIFE PROBLEMS Name Block 1. An isotope of cesium (cesium-137) has a half-life of 30 years. If 1,0 g of cesium-137 disintegrates over a period of 90 years, how many g of cesium-137 would remain? HALF-LIFE PROBLEMS The Results for Half Life Problems Answerkey. Structure Worksheet ...

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Carbon-14 has a half-life of 5730 years. That is, if you take one gram of C-14, half of it will decay in 5730 years. Cobalt-60 5 years Protactinium-226 2 minutes Iodine-131 8 days Americium-242 6 hours Tin-126 100,000 years This quiz covers half-life. Use the above information to answer the following questions.

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Half Life Worksheet ... (60 seconds) there are 12 half lives. Therefore the answer is: 4096 1 2 1 12 2) Iodine-131 has a half life of 8 days. What fraction of the original sample would remain at the end of 32 days? Using the same fraction, you must figure out n. If the half life is 8 days, ... the first problem each half life is 28 days ...

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