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We give some schematic solutions of exercises from chapters 1 to 10 of "Introduction to Statistical Physics", by Silvio R. A. Salinas, -rst published by Springer, New York, in 2001. We also add a number of corrections and some new exercises. Additional corrections and suggestions are warmly welcomed. Silvio Salinas

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Physics 3341 Thermodynamics and Statistical Physics | Itai ...

Correspondingly, the chemical potential is an energy per mole, and the equations of state use the. 1.1. INTRODUCTION. 3 gas constant R . In statistical mechanics we use N for the number of particles in the system. The chemical potential is now an energy per particle, and the equations contain the Boltzmann constant k_B .

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