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Chemical thermodynamics - Wikipedia

In chemistry, an ideal solution or ideal mixture is a solution with thermodynamic properties analogous to those of a mixture of ideal gases. The enthalpy of mixing is zero as is the volume change on mixing by definition; the closer to zero the enthalpy of mixing is, the more "ideal" the behaviour of the solution becomes. The vapor pressure of the solution obeys Raoult's law, and the activity ...

Ideal solution - Wikipedia

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Water structure science: References 3201 - 3300

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Structure, properties, spectra, suppliers and links for: m-Xylene, 108-38-3, 1330-20-7.

m-Xylene | C8H10 | ChemSpider

Colorless liquid with an aromatic odor. NIOSH ZE2450000: colourless liquid Oxford University Chemical Safety Data (No longer updated) More details

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Who Gets to Graduate? - The New York Times

The aggregation of proteins into amyloid fibrils and their deposition into plaques and intracellular inclusions is the hallmark of amyloid disease. Recent advances in structural biology techniques ...

A new era for understanding amyloid structures and disease ...

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Chlorure d'hydrogène - Wikipédia

The circular RNA ciRS-7 has been reported to be involved in the pathogenesis of various tumors, including gastric and colorectal cancer. However, the role of ciRS-7 in esophageal squamous cell ...

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Scope of review. This comprehensive review summarizes fundamental principles of glutathione catalysis and compares the structures and mechanisms of glutathione-dependent enzymes, including glutathione reductase, glutaredoxins, glutathione peroxidases, peroxiredoxins, glyoxalases 1 and 2, glutathione transferases and MAPEG.

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