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Isotopes Ions And Atoms Worksheet 2 Answer Key

Model 1 that supports your answer. NO, CARBON-12 AND CARBON-13 ARE ISOTOPES OF CARBON BUT HAVE DIFFERENT MASS NUMBERS. 12. Considering your answers to Question 11, work with your group to write a definition of isotope using a sentence. ISOTOPES ARE ATOMS OF THE SAME ELEMENT WITH DIFFERENT MASS NUMBERS. 13.

Isotopes - KEY - vonsteuben.org

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13, and 14 refer to the Mass Number d. How many protons and neutrons are in the first isotope? 6 protons & 6 neutrons e.

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Isotope Practice Worksheet Name: 1. 2. D. 4. 13 12 Here are three isotopes of an element: a. The element is: b. The number 6 refers to the c. The numbers 12, 13, and 14 refer to the d. How many protons and neutrons are in the first isotope? (O e. How many protons and neutrons are in the second isotope? f.

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Fill in the isotope names and any missing information, including isotope numbers from the chart. Use your periodic table and the information provided. # of protons 25 # of neutrons 17 15 # of electrons # of protons 32 # of neutrons ... Isotopes Worksheet ...

Isotopes Worksheet - St. Francis Preparatory School

Isotopes and Atomic Mass - PhET Interactive Simulations

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How can you tell one isotope from another? Use the sim to learn about isotopes and how abundance relates to the average atomic mass of an element. Sample Learning Goals Define "isotope" using mass number, atomic number, number of protons, neutrons and electrons. Given information about an element, find the mass and name of an isotope.

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