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Holt Chemistry 6 Covalent Compounds Name Class Date Concept Review continued 10. Draw the Lewis structure for water, H<sub>2</sub>O. 11. Draw the Lewis structure for carbon monoxide, CO. 12. Draw the Lewis structure for the rocket propellant nitryl fluoride, NO<sub>2</sub>F. 13. Draw the Lewis structures for SO<sub>2</sub> and SiO<sub>2</sub>. Answer the following item in the space ...

**Skills Worksheet Concept Review - Home - Default**

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and

nonconducting a metallic crystal (c) typically has the lowest melting point of the four

### **10 States of Matter - Ms. Agostine's Chemistry Page**

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### **Skills Worksheet Concept Review**

CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds.

### **6 Chemical Bonding - Effingham County Schools / Overview**

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### **Modern chemistry chapter 12 section 2 review answers ...**

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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Holt Chemistry 5 The Mole and Chemical Composition Section: Formulas and Percentage Composition Solve the following problems, and write your answer in the space provided.

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