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13. nonpolar covalent 14. polar covalent 15. dipole 16. nonpolar covalent 17. polar covalent 18. ionic 19. ionic 20. nonpolar covalent 21. nonpolar covalent 22. polar covalent 23. polar covalent Concept Review: Drawing and Naming Molecules 1. valence 2. Lewis 3. resonance 4. double 5. triple 6. single 7. unshared 8. 9. HI CIIIH H H H H H IC O H H H

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b ...

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