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enstructional Fair, Inc. 5. 35 g of CuSOa,5H20 Determine the number of grams in each of the quantities below. 1, 2.5 moles of NaCl 2. 0.50 moles of 3, 1.70 moles of KMn04 4, 0.25 moles of KCl e, ST 5. 3.2 moles of CuSO405H20 Chemistry IF87óó

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Unit 11 - Organic Chemistry

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r i c Page 4:2 ANSWER KEY WRITING FORMULAS (CRISS-CROSS METHOD) Name IONIC BONDING Name Ionicbondingoccurswhenametcl hcnstersoneormoreelectrons10anonmelolln on ...

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ELECTRON CONFIGURATION Name (LEVEL ONE)

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Balancing Chemical Equations -Answer Key Balance the equations below: 1) 1 N 2 + 3 H 2 \rightarrow 2 NH 3 2) 2 KClO 3 \rightarrow 2 KCl + 3 O 2 3) 2 NaCl + 1 F 2 \rightarrow 2 NaF + 1 Cl 2 4) 2 H 2 + 1 O 2 \rightarrow 2 H 2 O 5) 1 Pb(OH) 2 + 2 HCl \rightarrow 2 H 2 O + 1 PbCl 2 6) 2 AlBr 3 + 3 K 2 SO 4 \rightarrow 6 KBr + 1 Al 2 (SO 4) 3

Balancing Chemical Equations Answer Key

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Ionic Bonding Instructional Fair Inc Answers - WordPress.com

SUBSTANCES VS. MIXTURES A substance is matter for which a chemical formula can be written. Elements and compounds are substances. Mixtures can be in any proportion, and the parts are not chemically bonded. Classify the following as to whether it is a substance or a mixture by writing S or M in the space provided. 1. sodium 2. water 3. soil 4 ...

SUBSTANCES VS. MIXTURES - Mendham Borough Schools

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