

Grams Moles Calculations Answer Key

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Grams Moles Calculations Answer Key

Grams/Moles Calculations – Answer Key Given the following, name the compound and find the number of moles: 1) 30 grams of H_3PO_4 2) 25 grams of HF (hydrofluoric acid) 1.25 moles HF 3) 110 grams of NaHCO_3 (sodium bicarbonate or baking soda) 1.31 moles 4) 1.1 grams of FeCl_3 (iron (III) chloride) 0.0068 moles

Grams/Moles Calculations Worksheet III

Grams/Moles Calculations – Answer Key Given the following, find the number of moles: 1) 30 grams of H_3PO_4 0.31 moles 2) 25 grams of HF 1.25 moles 3) 110 grams of NaHCO_3 1.31 moles 4) 1.1 grams of FeCl_3 0.0068 moles 5) 987 grams of Ba(OH)_2 3.80 moles 6) 564 grams of copper 0.11 moles 7) 12.3 grams of CO_2 0.28 moles 8) 89 grams of $\text{Pb(CH}_3\text{COO)}_4$ 0 ...

Grams/Moles Calculations – Answer Key

Mole to Grams, Grams to Moles Conversions Worksheet What are the molecular weights of the following compounds? ... Mole Calculation Worksheet – Answer Key

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Mole to Grams Grams to Moles Conversions Worksheet Answer ...

Mole Calculation Worksheet – Answer Key 1) How many moles are in 15 grams of lithium? 0.46 moles 2) How many grams are in 2.4 moles of sulfur? 77.0 grams 3) How many moles are in 22 grams of argon? 0.55 moles 4) How many grams are in 88.1 moles of magnesium? 2141 grams 5) How many moles are in 2.3 grams of phosphorus? 0.074 moles

Mole Conversion Worksheet #1 Answer Key

Mole to Grams, Grams to Moles Conversions Worksheet ... How many grams are in 4.5 moles of sodium fluoride, NaF ? 10) How many moles are in 98.3 grams of aluminum hydroxide, Al(OH)_3 ... Mole Calculation Worksheet – Answer Key What are the molecular weights of the following compounds? 1) NaOH 23 + 16 + 1 = 40.1 grams 2) H_3PO_4

Mole to Grams, Grams to Moles Conversions Worksheet

189 grams. 16) How many moles are in 98.3 grams of aluminum hydroxide, Al(OH)_3 ? 1.26 moles. 17) How many grams are in 0.02 moles of beryllium iodide, BeI_2 ? 5.2 grams. 18) How many moles are in 68 grams of copper (II) hydroxide, Cu(OH)_2 ? 0.70 moles. 19) How many grams are in 3.3 moles of potassium sulfide, K_2S ? 364.0 grams. 20) How many moles ...

Mole Calculation Worksheet – Answer Key

Molar Mass Worksheet – Answer Key Calculate the molar masses of the following chemicals: 1) Cl_2 ... Combined Mole Calculations – Answer Key 1) How many molecules are there in 24 grams of FeF_3 ? 1.28×10^{23} molecules ... “Grams/mole” is better, because any macroscopic amount of a substance is ...

Molar Mass Worksheet Answer Key - River Dell Regional ...

Mole Calculation Worksheet – Answer Key 1) How many moles are in 15 grams of lithium? 0.46 moles 2) How many grams are in 2.4 moles of sulfur? 77.0 grams 3) How many moles are in 22 grams of argon? 0.55 moles 4) How many grams are in 88.1 moles of magnesium? 2141 grams 5) How many moles are in 2.3 grams of phosphorus? 0.074 moles

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Mole to Grams, Grams to Moles Conversions Worksheet What are the molecular weights of the

following compounds? ... Mole Calculation Worksheet – Answer Key What are the molecular weights of the following compounds? 1) NaOH $22.99 + 16.00 + 1.01 = 40.00$ grams/mol 2) H_3PO_4

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Mole to Grams Grams to Moles Conversions Worksheet Answer ...

Worksheet on Moles and Stoichiometry Three conversions to remember about chemicals – usually used when we are asking about just one chemical: Problems with balanced reactions usually follow this chart – a chemical reaction is involved in these problems: Note questions 1 and 2 are on stoichiometry. There is also another worksheet on that

Worksheet on Moles and Stoichiometry

Moles, Molecules, and Grams Worksheet – Answer Key 1) How many molecules are there in 24 grams of FeF_3 ? 1.28×10^{23} molecules 2) How many molecules are there in 450 grams of Na_2SO_4 ? 1.91×10^{24} molecules 3) How many grams are there in 2.3×10^{24} atoms of silver? 421 grams 4) How many grams are there in 7.4×10^{23} molecules of AgNO_3 ? 209 grams

Moles, Molecules, and Grams Worksheet - nclark.net

Grams/Moles Calculations – Answer Key Given the following, find the number of moles: 1) 30 grams of H_3PO_4 0.31 moles 2) 25 grams of HF 1.25 moles 3) 110 grams of NaHCO_3 1.31 moles 4) 1.1 grams of FeCl_3 0.0068 moles 5) 987 grams of Ca(OH)_2 3.80 moles 6) 564 grams of copper 0.11 moles 7) 12.3 grams of CO_2 0.28 moles 8) 89 grams of $\text{Pb(CH}_3\text{COO)}_4$ 0 ...

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