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equation support your answer to question 2? 9. Use your answer to question 8 to write the base dissociation constant (K_b) expression for the reaction of benzoate ion with water. 10. Based on your estimate of the pH at the equivalence point and using the expression for K_b , determine the numerical value of K_b for the benzoate ion ($C_6H_5COO^-$).

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Chapter 11 Small-Scale Lab Section 11.3 Precipitation Reactions: Formation of Solids, page 345 Analysis 1. a. $Na_2CO_3 + 2AgNO_3 \rightarrow 2NaNO_3 + 3Ag_2CO_3(s)$ b. $2Na_3PO_4 + 3Pb(NO_3)_2 \rightarrow 6NaNO_3 + Pb_3(PO_4)_2(s)$ 2.

Chapter 11 Small-Scale Lab

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6.1 Organizing the Elements 6

Use the diagram below to answer the following questions on the lines provided. 26. Applying Concepts What two functions does the structure in Figure 3 have? ____ 27. Interpreting Graphics Would the structure in Figure 3 have come from a wilting plant? Explain. ____ 28. Interpreting Graphics What labeled structure in Figure 3

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Chapter 9 Chemical Names and Formulas 213 Section Review Objectives • Apply the rules for naming and writing formulas for binary ionic compounds • Apply the rules for naming and writing formulas for compounds with polyatomic ions Vocabulary • binary compound Part A Completion Use this completion exercise to check your understanding of the concepts and terms

05 CTR ch09 7/9/04 3:29 PM Page 211 NAMING IONS 9

Name Date Class INTERPRETING GRAPHICS Use with Section 17.2 A student performed an experiment to determine the specific heat of an unknown metal. The data she collected is organized in the table below. Use this information to answer the following questions. Quantity Trial 1 Trial 2 1. Mass of test tube + metal 118.19 g 118.21 g 2.

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9.1 Naming Ions 9 - Ms. Lara La Cueva HS Science

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