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26. According to the kinetic-molecular theory, the particles in a liquid can change relative positions but still are influenced by attractive forces.

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SHORT ANSWER Answer the following questions in the space provided. 1. Use the concept of potential energy to describe how a covalent bond forms between two atoms. As the atoms involved in the formation of a covalent bond approach each other, the electron-proton attraction is stronger than the electron-electron and proton-proton repulsions.

**6 Chemical Bonding - Effingham County Schools / Overview**

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