

ThermoFun: A C++/Python library for computing standard thermodynamic properties of substances and reactions across wide ranges of temperatures and pressures

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#### **Software**

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## Summary

ThermoFun is an open source library that facilitates the calculation and retrieval of standard thermodynamic properties of chemical substances, compounds, and reactions among them at a range of temperatures and pressures. The library is developed in C++ for performance, but it also has a Python API for broader and more convenient usage. It employs a variety of thermodynamic models and equations of state for solid, aqueous, surface, gaseous, and molten substances, and their reactions with input properties and parameters from various thermodynamic datasets that are collected and curated in the in ThermoHub database. The library can be used as a standalone code for searching and tabulating thermodynamic properties or linked to other modeling codes that require standard thermodynamic data as input. It offers the flexibility to use different thermodynamic datasets, including custom datasets and datasets retrieved from the online ThermoHub database, and to choose the most suitable models for different classes of substances necessary in various modeling applications. It can serve as a common source of thermodynamic models for standard properties of substances and reactions that can be easily integrated and combined, significantly improving the modeling capabilities for diverse (geo)chemical systems and over wide ranges of conditions.

### Statement of need

Thermodynamic modeling is a powerful tool that allows for the investigation of equilibrium speciation in chemical systems under conditions and time scales that cannot be reached in laboratory settings. It requires the standard thermodynamic properties of all involved substances (species) or reactions at a specific temperature (T) and pressure (P) as input. Equations of state are used to compute these properties, known as standard thermodynamic models. However, existing thermodynamic modeling codes and standalone codes may not be directly compatible, making it difficult to retrieve properties calculated with different methods and use them in different codes, or to compare and tabulate thermodynamic data from different sources. The ThermoFun library simplifies this process by facilitating the computation of standard thermodynamic properties of substances and reactions (see Figure 1), and providing fast and simple access to many thermodynamic datasets, methods, and equations of state. This allows for more efficient use of these tools in modeling codes, or for the optimization and fitting of model parameters, evaluation of the models' performance, range of applicability, and accuracy. Additionally, ThermoFun is extendable with new models and has remote access to



a curated collection of thermodynamic datasets on ThermoHub, making it a useful tool for various studies on hydrothermal processes, cementitious materials, nuclear engineering, waste incineration, radioactive and chemical management and disposal, and other (geo)chemical environments.

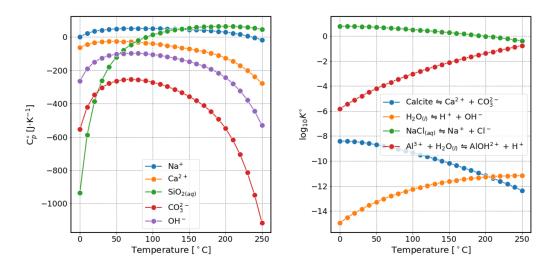


Figure 1: Example of standard thermodynamic properties of substances and reactions calculated using ThermoFun as a function of temperature along the saturated water vapor pressure curve (input parameters from aq17 thermodynamic dataset (Miron et al., 2016, 2017)).

The ThermoFun library can be linked to any C++ or Python geochemical equilibrium speciation, parameter optimization, or reactive mass-transport code as a source of standard thermodynamic properties of substances and/or reactions. As a standalone code, ThermoFun can be used simply for searching and tabulating thermodynamic properties of substances and reactions. For example, it has been used to calculate the standard properties of reactants and reactions at elevated temperatures when evaluating the performance of isocoulombic reactions, in order to extrapolate the properties of aqueous complexation reactions involving lanthanides and actinides to elevated temperatures (Miron et al., 2020). The library was also applied to evaluate the effect of using different equations of state for water-solvent on the calculated properties of dissolved aqueous species (Miron et al., 2019). ThermoFun has already been adopted as a source of thermodynamic data in the GEMS equilibrium speciation codes (Kulik et al., 2013) and the Reaktoro chemical modeling framework (Leal, 2015).

# Benefits of using ThermoFun

- It simplifies the calculation of standard thermodynamic properties of substances and reactions at a range of temperatures and pressures.
- It includes a variety of equations of state and other thermodynamic models that can be used to calculate standard properties under different conditions, eliminating the need for the user to implement and test these models separately in their own code or spreadsheets.
- It uses input properties and parameters from various thermodynamic datasets that are consistently kept in the online ThermoHub database, reducing the need to collect all necessary thermodynamic data from literature or to write complex scripts to import the data from different formats.
- It uses automatic differentiation to provide temperature and pressure derivatives of the calculated properties. autodiff C++ library (Leal, 2018) will be used in ThermoFun for even more capable autodiff capabilities, as it has been used in both Reaktoro (Leal, 2015) and TEQP (I. H. Bell et al., 2022).



- The thermodynamic models in ThermoFun are implemented and tested based on their original publications, and the thermodynamic data in ThermoHub is curated and documented by experts, saving time and effort and enabling more efficient use of geochemical code tools for scientific problems.
- It serves as a common source of thermodynamic models for standard properties of substances and reactions that can be easily integrated and combined, significantly improving modeling capabilities for a wide range of (geo)chemical systems and conditions.

Jupyter notebooks with examples on how to use ThermoFun can be found in the thermofun-jupyter repository. More complex Python or C++ routines that use ThermoFun can be coded. For example, ThermoFun can be coupled to a phase diagram plotter, a chemical equilibrium solver, or a parameter optimization routine for fitting the standard properties and model parameters of substances and reactions and their temperature and pressure dependence.

### Codes with similar functionalities

ThermoFun is an open source library with several benefits that serves as a common source of ever-expanding thermodynamic models for standard properties of substances and reactions. If ThermoFun does not meet your needs, there are other alternatives offering related functionality, potentially using different methods, and sometimes embedded in larger applications. Below the (not exhaustive) list of codes that share some functionality with ThermoFun:

Name	Language	Reference	Source
Cantera	C++, Python, Matlab	(Goodwin et al., 2022)	Open
CHNOSZ	R, Python	(Dick, 2019)	Open
coolprop	C++	(I. H. Bell et al., 2014)	Open
DEWPython	Python	(Chan et al., 2021)	Open
GEM-Selektor	C++	(Kulik et al., 2013)	Open
GeoTPD	R-Shiny	(Bastrakov & Dick, 2019)	Closed
PyGCC	Python	(Awolayo & Tutolo, 2022)	Open
pyromat	Python	(Martin et al., 2022)	Open
Reaktoro	C++, Python	(Leal et al., 2017)	Open
REFPPROP	FORTRAN	(Huber et al., 2022)	Closed
SteamTables	C++, Web	(megawatsoft, 2022)	Closed
SUPCRT92	Fortran	(Johnson et al., 1992)	Open
SUPCRTBL	Fortran	(Zimmer et al., 2016)	Closed
TEQP	C++, Python	(I. H. Bell et al., 2022)	Open
Thermo	Python	(C. Bell & Contributors,	Open
		2021)	
ThermoEngine	C++, Python	(Thermoengine, 2022)	Open

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### References

- Awolayo, A. N., & Tutolo, B. M. (2022). PyGeochemCalc: A python package for geochemical thermodynamic calculations from ambient to deep earth conditions. *Chemical Geology*, 606, 120984. https://doi.org/10.1016/j.chemgeo.2022.120984
- Bastrakov, E., & Dick, J. (2019, August). GeoTPD an interactive online tool for geochemical modelling for the broad geological community. Goldschmidt abstracts, 2019.
- Bell, C., & Contributors. (2021). Thermo: Chemical properties component of chemical engineering design library (ChEDL). https://github.com/CalebBell/thermo
- Bell, I. H., Deiters, U. K., & Leal, A. M. M. (2022). Implementing an Equation of State without Derivatives: teqp. *Industrial and Engineering Chemistry Research*, 61(17), 6010–6027. https://doi.org/10.1021/acs.iecr.2c00237
- Bell, I. H., Wronski, J., Quoilin, S., & Lemort, V. (2014). Pure and pseudo-pure fluid thermophysical property evaluation and the open-source thermophysical property library CoolProp. *Industrial & Engineering Chemistry Research*, *53*(6), 2498–2508. https://doi.org/10.1021/ie4033999
- Chan, A., Daswani, M. M., & Vance, S. (2021). *DEWPython: A python implementation of the deep earth water model and application to ocean worlds.* arXiv. https://doi.org/10.48550/ARXIV.2105.14096
- Dick, J. M. (2019). CHNOSZ: Thermodynamic calculations and diagrams for geochemistry. Frontiers in Earth Science, 7. https://doi.org/10.3389/feart.2019.00180
- Goodwin, D. G., Moffat, H. K., Schoegl, I., Speth, R. L., & Weber, B. W. (2022). *Cantera: An object-oriented software toolkit for chemical kinetics, thermodynamics, and transport processes.* https://www.cantera.org. https://doi.org/10.5281/zenodo.6387882
- Huber, M. L., Lemmon, E. W., Bell, I. H., & McLinden, M. O. (2022). The NIST REFPROP database for highly accurate properties of industrially important fluids. *Industrial & Engineering Chemistry Research*, 61(42), 15449–15472. https://doi.org/10.1021/acs.iecr. 2c01427
- Johnson, J. W., Oelkers, E. H., & Helgeson, H. C. (1992). SUPCRT92: A software package for calculating the standard molal thermodynamic properties of minerals, gases, aqueous species, and reactions from 1 to 5000 bar and 0 to 1000°c. *Computers & Geosciences*, 18(7), 899–947. https://doi.org/10.1016/0098-3004(92)90029-Q
- Kulik, D. A., Wagner, T., Dmytrieva, S. V., Kosakowski, G., Hingerl, F. F., Chudnenko, K. V., & Berner, U. R. (2013). GEM-selektor geochemical modeling package: Revised algorithm and GEMS3K numerical kernel for coupled simulation codes. *Computational Geosciences*, 17, 1–24. https://doi.org/10.1007/s10596-012-9310-6
- Leal, A. M. M. (2015). Reaktoro: An open-source unified framework for modeling chemically reactive systems. https://reaktoro.org. https://reaktoro.org
- Leal, A. M. M. (2018). autodiff, a modern, fast and expressive C++ library for automatic differentiation. https://autodiff.github.io. https://autodiff.github.io
- Leal, A. M. M., Kulik, D. A., Smith, W. R., & Saar, M. O. (2017). An overview of computational methods for chemical equilibrium and kinetic calculations for geochemical and reactive transport modeling. *Pure and Applied Chemistry*, 89, 597–643. https://doi.org/10.1515/pac-2016-1107
- Martin, C., Ranalli, J., & Moore, J. (2022). PYroMat: A python package for thermodynamic properties. *Journal of Open Source Software*, 7(79), 4757. https://doi.org/10.21105/joss.04757



- megawatsoft. (2022). Thermodynamic properties of fluids. In *Thermodynamic Properties of Fluids*. https://www.megawatsoft.com/
- Miron, G. D., Kulik, D. A., & Thoenen, T. (2020). Generating isocoulombic reactions as a tool for systematic evaluation of temperature trends of thermodynamic properties: Application to aquocomplexes of lanthanides and actinides. *Geochimica Et Cosmochimica Acta*, 286, 119–142. https://doi.org/10.1016/j.gca.2020.07.020
- Miron, G. D., Leal, A. M. M., & Yapparova, A. (2019). Thermodynamic properties of aqueous species calculated using the HKF model: How do different thermodynamic and electrostatic models for solvent water affect calculated aqueous properties? *Geofluids*, 2019, 1–24. https://doi.org/10.1155/2019/5750390
- Miron, G. D., Wagner, T., Kulik, D. A., & Heinrich, C. A. (2016). Internally consistent thermodynamic data for aqueous species in the system na-k-al-si-o-h-cl. *Geochimica Et Cosmochimica Acta*, 187. https://doi.org/10.1016/j.gca.2016.04.026
- Miron, G. D., Wagner, T., Kulik, D. A., & Lothenbach, B. (2017). An internally consistent thermodynamic dataset for aqueous species in the system ca-mg-na-k-al-si-o-h-c-cl to 800 °c and 5 kbar. *American Journal of Science*, 317, 755–806. https://doi.org/10.2475/07.2017.01
- Thermoengine, C. C. T. (2022). ThermoEngine: Software for Model Building and Computational Thermodynamics Supporting Applications in the Earth Sciences (Version 1.0.0). Zenodo. https://doi.org/10.5281/zenodo.6527840
- Zimmer, K., Zhang, Y., Lu, P., Chen, Y., Zhang, G., Dalkilic, M., & Zhu, C. (2016). SUPCRTBL: A revised and extended thermodynamic dataset and software package of SUPCRT92. *Computers & Geosciences*, 90, 97–111. https://doi.org/10.1016/j.cageo. 2016.02.013