Tropical approaches to multi-scaleness of chemical reactions

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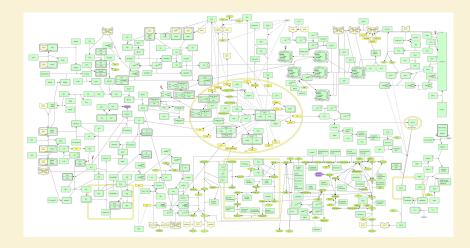
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Outline

- Linear chemical reactions networks with separated constants.
- Non-linear chemical reactions networks with multiple time scales.
- Tropical approaches.

The problem of size

Complex, large scale molecular systems.



Supagro, Oct 2013 Kitano 2004

Dynamics

- State X (numbers of molecules), x = X/V (concentrations), reactions $X \to X + v_j$, $v_j \in \mathbb{Z}^n$.
- Deterministic dynamics

$$\frac{dx}{dt} = \sum_{j=1}^{r} v_j R_j(x)$$

Dynamics

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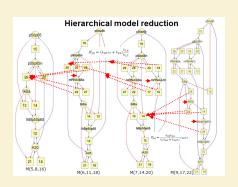
 Stochastic dynamics X(t) is a jump Markov process, of intensity

$$\lambda(x) = V \sum_{i=1}^{r} R_i(x),$$

jumps $X \rightarrow X + v_j$, jump distribution

$$p_j(x) = R_j(x) / \sum_{j=1}^r R_j(x)$$

Model reduction



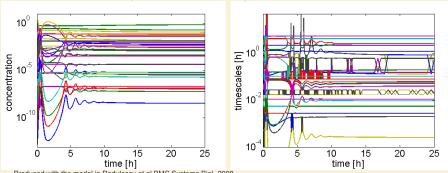
Model reduction: produce models with less variables, equations, parameters Backward pruning: define synthetic parameters that are identifiable

Multiscale networks

Our methods apply to molecular networks that have many, well separated, time and concentration scales.

Widely distributed concentrations, in log scale

Widely distributed timescales, in log scale



Produced with the model in Radulescu et al BMC Systems Biol. 2008

Aim: develop reduction methods for multi-scale models with uncertainty.

Linear networks of chemical reactions:

digraphs with linear kinetics

 A_i are reagents, c_i is concentration of A_i .

All the reactions are of the type $A_i \rightarrow A_i$ (monomolecular).

 $k_{ii} > 0$ is the reaction $A_i \rightarrow A_i$ rate constant.

The reaction rates: $w_{ji} = k_{ji}c_i$.

Kinetic equation

$$\dot{c}_i = \sum_{j,j \neq i} (k_{ij}c_j - k_{ji}c_i) \text{ or } \dot{c} = \mathbf{K}c, \tag{1}$$

Relevance for computational biology:

- Occur as subsystems of larger, nonlinear networks.
- Crude approximations obtained by linearizing networks.

Linear networks with separated constants

$$c(t) = (I^0, c(0)) + \sum_{k=1}^{n-1} r^k (I^k, c(0)) \exp(-\lambda_k t)$$

The eigenvectors of *K* specify the dynamics. Well separated constants

$$k_{l_1} \gg k_{l_2} \gg k_{l_3} \gg ...$$

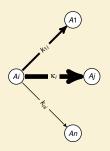
Integer labeled digraphs: each reaction arc has an integer label, specifying its position in the sequence of all reactions, ordered by speed; the lowest order is the most rapid.

Theorem: the multiscale approximation of an arbitrary linear network with separated constants is an acyclic, deterministic, integer labeled digraph.

Auxiliary discrete dynamical systems

For each A_i , $\kappa_i = \max_j \{k_{ji}\}$, $\phi(i) = \arg\max_j \{k_{ji}\}$; $\phi(i) = i$ if there is no outgoing reaction $A_i \to A_j$. ϕ determines *auxiliary dynamical system* on a set $\mathcal{A} = \{A_i\}$.

Pruning: keep only the dominating step



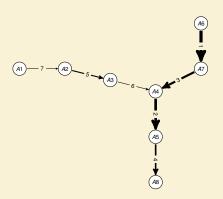


An

The auxiliary dynamical system is further decomposed into cycles C_i with basins of attraction, $Att(C_i)$: $A = \bigcup_i Att(C_i)$.

1-st case: acyclic auxiliary dynamic systems

All cycles C_i are point attractors.



$$r_{\Phi(j)}^i = rac{\kappa_j}{\kappa_{\Phi(j)} - \kappa_i} r_j^i$$
 go along the flow $l_j^i = rac{\kappa_j}{\kappa_j - \kappa_i} l_{\Phi(j)}^i$ go opposite to the flow.

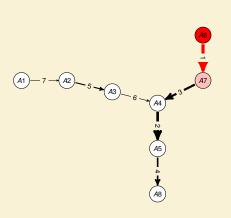
For instance:

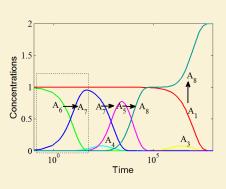
$$I^{1} \approx (1,0,0,0,0,0,0,0)$$

$$r^{1} \approx (1,0,0,0,0,0,0,-1)$$

$$I^{5} \approx (0,0,0,1,1,1,1,0)$$

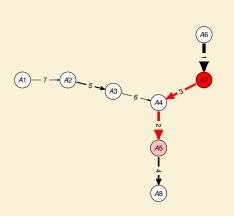
$$r^{5} \approx (0,0,0,0,1,0,0,-1)$$

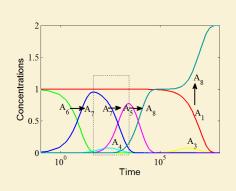




$$I^6 \approx (0,0,0,0,0,1,0,0)$$

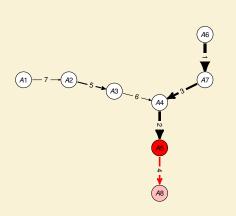
 $r^6 \approx (0,0,0,0,0,-1,1,0)$

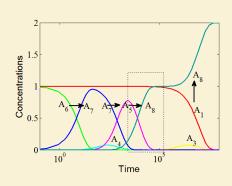




$$I^7 \approx (0,0,0,0,0,1,1,0)$$

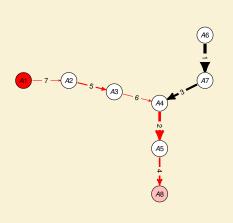
 $r^7 \approx (0,0,0,0,1,0,-1,0)$

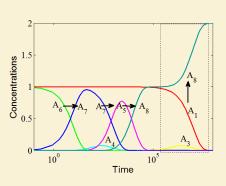




$$I^5 \approx (0,0,0,1,0,1,1,0)$$

 $r^5 \approx (0,0,0,0,-1,0,0,1)$



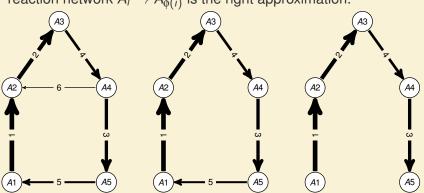


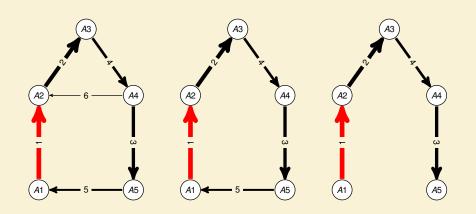
$$I^1 \approx (1,0,0,0,0,0,0,0)$$

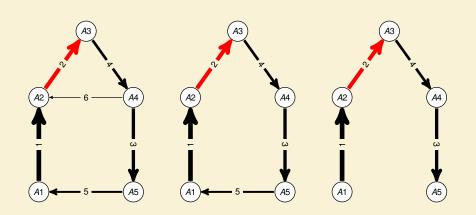
 $r^1 \approx (-1,0,0,0,0,0,0,1)$

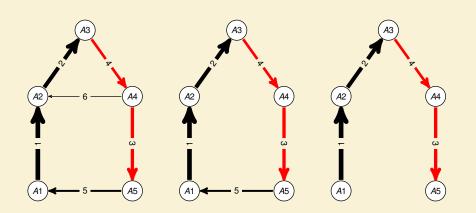
2-nd case: C_i are sinks in the initial network

Delete the limiting steps from cycles C_j . The obtained *acyclic* reaction network $A_i \to A_{\phi(i)}$ is the right approximation.

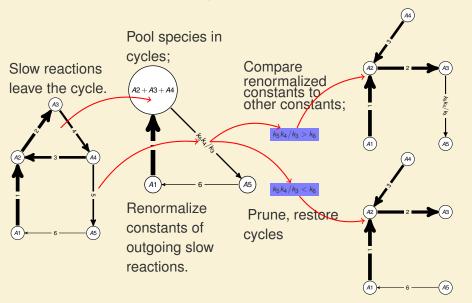








3-rd case: some of C_j are not sinks



Nonlinear chemical networks

Polynomial kinetics (e.g. mass action law)

$$\frac{dx_i}{dt} = P_i(x) = \sum_{j=1}^{M_i} k_{ij} x^{\alpha_{ij}}$$

 $\alpha_{ij} \in \mathbb{N}^n$ are multi-indices, $k_{ij} \in \mathbb{R}$ are kinetic parameters.

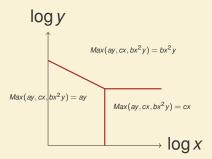
- ► Timescales are not inverses of parameters in the model. They involve concentrations and can change with time.
- Exploit some form of equilibration. Use invariant manifolds.
- Develop a theory of dominance based on tropical ideas.

Litvinov-Maslov tropicalization

Heuristics: replace multivariate polynomials by piece-wise smooth, max-plux polynomials.

$$\sum_{lpha \in \mathcal{A}} \mathsf{a}_lpha x^lpha o \mathsf{exp}[\mathit{max}_{lpha \in \mathcal{A}}\{\mathit{log}(\mathsf{a}_lpha) + < \mathit{log}(x), lpha > \}].$$

Tropical manifold: the set of points where max-plus polynomials are not smooth.



Tropical manifold of $ay + cx + bx^2y$ on "logarithmic paper".

Tropicalized Systems

► Two terms tropicalization (Savageau 2009)

$$\frac{dx_i}{dt} = DomP_i^+(x) - DomP_i^-(x)$$

► Complete, or one term tropicalization (Noel et al 2011)

coefficient of the monomial for which the maximum is attained.

$$\frac{dx_i}{dt} = DomP_i(x)$$

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where Dom\left[\sum_{\alpha\in A_i}a_{i,\alpha}x^{\alpha}\right]=sign(a_{i,\alpha_{max}})exp[max_{\alpha\in A_i}\{log(|a_{i,\alpha}|)+< u,\alpha>\}].
u=(logx_1,\ldots,logx_n), and a_{i,\alpha_{max}},\alpha_{max}\in A_i denote the
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Is the heuristics justified?

$$\frac{dx_i}{dt} = P_i(x) = \sum_{j=1}^{M_i} \bar{k}_{ij} \varepsilon^{\gamma_{ij}} x^{\alpha_{ij}}, \qquad (2)$$

where $\varepsilon > 0$ is a small number.

Definition. The system (2) is permanent, if

$$C_- < \bar{x}_i(t) < C_+$$
, for all $t > T_0(x(0))$ and for all i ,

where $x_i = \epsilon^{a_i} \bar{x}_i$ for constants $C_{\pm} > 0$ and T_0 uniform in ϵ as $\epsilon \to 0$.

Theorem If the system (2) is permanent, then the complete or two terms tropicalization are good approximations: error vanishes like ε^{γ} , and one has orbital topological equivalency between initial and tropicalized trajectories.

Tropical truncation and tropical equilibration

Polynomial kinetics (e.g. mass action law)

$$\frac{dx_i}{dt} = \sum_{j=1}^{M_i} k_{ij} x^{\alpha_{ij}}$$

 $\varepsilon > 0$ is a small number. Renormalize concentrations $x_i = \varepsilon^{a_i} \bar{x}_i$ where powers a_i are unknown.

$$\frac{d\bar{x}_i}{dt} = \sum_{j=1}^{M_i} \varepsilon^{\mu_{ij}} \bar{k}_{ij} \bar{\mathbf{x}}^{\alpha^{ij}},$$

$$\mu_{ij} = \gamma_{ij} + \sum_{l=1}^{n} \alpha_l^{ij} a_l - a_i.$$

Tropical truncation

Keep only the minimal powers of ε , ie m_i terms. For instance, if $m_i = 1$

$$\frac{d\bar{x}_i}{dt} = \varepsilon^{\mu_i} k_{ij(i)} \bar{\mathbf{x}}^{\alpha^{ij(i)}},$$

where j(i) is the index of the unique term with minimum degree in ε ,

$$\mu_i = \min_j (\gamma_{ij} + \sum_{l=1}^n \alpha_l^{ij} a_l) - a_i.$$

Tropical equilibration problem

Find a pair (j, j'), $j \neq j'$ such that

- i) $\mu_{ij} = \mu_{ij'}$,
- ii) $\mu_{ij} \leq \mu_{il}$ for all $l \neq j, j'$,
- iii) $k_{ij}k_{ij'}<0$.

The minimum degree is attained at least twice and the minimal degree terms have opposite signs.

$$\frac{d\bar{\mathbf{x}}_i}{dt} = \varepsilon^{\mu_i} (|k_{ij}|\bar{\mathbf{x}}^{\alpha^{ij}} - |k_{ij'}|\bar{\mathbf{x}}^{\alpha^{ij'}})$$

The tropically truncated system is binomial (toric system).

Tropical equilibration and model reduction

Lemma1. We have permanency only if we have tropical equilibration of at least two terms, one positive (production), the other negative (degradation).

Lemma2. In the case of tropical equilibration of two terms (toric systems), and if $0 = \mu_1 < \mu_2 < ... < \mu_{n-1} < \mu_n$ (total separation) each variable follows

$$\frac{dx}{dt} = b_1(z)x^{\beta_1} - b_2(z)x^{\beta_2}, \quad b_1, b_2 > 0,$$

where z is a slow function of time and we have permanency iff $\beta_1 < \beta_2$.

Theorem. If the above conditions are satisfied, then the system has an attractive invariant manifold. The corresponding equilibration can be used to reduce the model. Fast variables are tropically equilibrated.

Michaelis-Menten enzymatic reaction

$$S+E \stackrel{k_1}{\underset{k_{-1}}{\rightleftharpoons}} ES \stackrel{k_2}{\rightarrow} P+E,$$

Using the conserved quantities $e_0 = [E] + [ES]$, $s_0 = [S] + [ES] + [P]$, we get the system

$$x' = -k_1x(e_0 - y) + k_{-1}y, \quad y' = k_1x(e_0 - y) - (k_{-1} + k_2)y.$$

where x = [S] and y = [SE], and some constraints

$$0 \le y \le e_0, 0 \le x + y \le s_0, 0 \le x.$$

The tropical equilibration problem

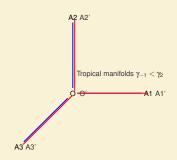
After rescalings
$$x = \bar{x}\epsilon^{a_1}$$
, $y = \bar{y}\epsilon^{a_2}$, $k_1 = \bar{k}_1\epsilon^{\gamma_1}$, $k_{-1} = \bar{k}_{-1}\epsilon^{\gamma_{-1}}$, $e_0 = \bar{e}_0\epsilon^{\gamma_e}$, $s_0 = \bar{s}_0\epsilon^{\gamma_s}$, we get

$$\begin{split} \bar{x}' &= -\bar{k}_1 \bar{e}_0 \epsilon^{\gamma_1 + \gamma_e} \bar{x} + \bar{k}_1 \epsilon^{\gamma_1 + a_2} \bar{x} \bar{y} + \bar{k}_{-1} \epsilon^{\gamma_{-1} + a_2 - a_1} \bar{y}, \\ \bar{y}' &= \bar{k}_1 \bar{e}_0 \epsilon^{\gamma_1 + \gamma_e + a_1 - a_2} \bar{x} - \bar{k}_1 \epsilon^{\gamma_1 + a_1} \bar{x} \bar{y} - (\bar{k}_{-1} \epsilon^{\gamma_{-1}} + \bar{k}_2 \epsilon^{\gamma_2}) \bar{y}. \end{split}$$

that leads to the equilibration equations

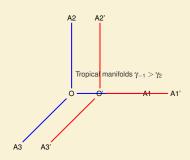
$$\begin{array}{rl} \gamma_{1} + \gamma_{e} &= \min(\gamma_{1} + a_{2}, \gamma_{-1} + a_{2} - a_{1}), \\ \gamma_{1} + \gamma_{e} + a_{1} - a_{2} &= \min(\gamma_{1} + a_{1}, \min(\gamma_{-1}, \gamma_{2})). \end{array}$$

Quasi-equilibrium approximation



No	Condition	Truncated system	Regime
1	$a_1 < \gamma_{-1} - \gamma_1$ $a_2 = \gamma_e$	$x' = \varepsilon^{\gamma_1 + \gamma_e} \left(-\bar{k}_1 \bar{e}_0 \bar{x} + \bar{k}_1 \bar{x} \bar{y} \right)$ $y' = \varepsilon^{\gamma_1 + a_1} \left(\bar{k}_1 \bar{e}_0 \bar{x} - \bar{k}_1 \bar{x} \bar{y} \right)$	QE saturated
2	$a_1 > \gamma_{-1} - \gamma_1$ $a_2 = a_1 + \gamma_e + \gamma_1 - \gamma_{-1}$	$x' = \varepsilon^{\gamma_1 + \gamma_e} \left(-\bar{k}_1 \bar{e}_0 \bar{x} + \bar{k}_{-1} \bar{y} \right)$ $y' = -\varepsilon^{\gamma_{-1}} \left(\bar{k}_1 \bar{e}_0 \bar{x} - \bar{k}_{-1} \bar{y} \right)$	QE linear

Quasi-steady state approximation



No	Condition	Truncated system	Regime		
1	$a_1 < \gamma_2 - \gamma_1$ $a_2 = \gamma_e$	$x' = \varepsilon^{\gamma_1 + \gamma_e} \left(-\bar{k}_1 \bar{e}_0 \bar{x} + \bar{k}_1 \bar{x} \bar{y} \right)$ $y' = \varepsilon^{\gamma_1 + a_1} \left(\bar{k}_1 \bar{e}_0 \bar{x} - \bar{k}_1 \bar{x} \bar{y} \right)$	y QSS if $a_1 < \gamma_e$		
2	$\gamma_2 - \gamma_1 < a_1 < \gamma_{-1} - \gamma_1$ $a_2 = \gamma_e$	$x' = \varepsilon^{\gamma_1 + \gamma_e} \left(-\bar{k}_1 \bar{e}_0 \bar{x} + \bar{k}_1 \bar{x} \bar{y} \right)$ $y' = -\varepsilon^{\gamma_2} \bar{k}_2 \bar{y}$	x QSS if $\gamma_2 > \gamma_1 + \gamma_e$		
3	$a_1 > \gamma_{-1} - \gamma_1$ $a_2 = a_1 + \gamma_e + \gamma_1 - \gamma_{-1}$	$x' = \varepsilon^{\gamma_1 + \gamma_e} \left(-\bar{k}_1 \bar{e}_0 \bar{x} + \bar{k}_{-1} \bar{y} \right)$ $y' = -\varepsilon^{\gamma_2} \bar{k}_2 \bar{y}$	x QSS if $\gamma_2 > \gamma_1 + \gamma_e$		
4	$a_1 > \gamma_2 - \gamma_1$ $a_2 = a_1 + \gamma_e + \gamma_1 - \gamma_2$	$x' = -\varepsilon^{\gamma_1 + \gamma_e} \bar{k}_1 \bar{e}_0 \bar{x}$ $y' = \varepsilon^{\gamma_2} (\bar{k}_1 \bar{e}_0 \bar{x} - \bar{k}_2 \bar{y})$	y QSS if $\gamma_2 < \gamma_1 + \gamma_e$		

Tropical equilibrations and model reduction

TTS at quasi-equilibrium : pruning
$$\bar{x}' = \varepsilon^{\gamma_1 + \gamma_e} (-\bar{k}_1 \bar{e}_0 \bar{x} + \bar{k}_{-1} \bar{y})$$
 $\bar{y}' = \varepsilon^{\gamma_{-1}} (\bar{k}_1 \bar{e}_0 \bar{x} - \bar{k}_{-1} \bar{y})$

New slow variable
$$z = x + y$$
: pooling $\bar{z}' = -\epsilon^{\gamma_2 + a_2 - \gamma_s} \bar{k}_2 \bar{y}$.

Eliminate x, y, obtain the reduced model $z' = -k_2/(1 + k_{-1}/(k_1e_0))z$.

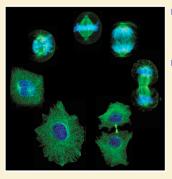
Results on Biomodels.net

Two methods: Newton polytope, reified constraints.

436 curated models, 55 have purely polynomial kinetics.

Found	# models	Variables (avg/min/max)	Time (avg/min/max)
yes	23	17.348/3/ 86	0.486/0.004/2.803
no	32	17.812/1/194	0.099/0.000/1.934

Cell cycle model



- Cell proliferation is regulated by cyclic activity of cyclins
- A simple model, Tyson 91

$$y'_{1} = \varepsilon^{-3}k_{9}y_{2} - \varepsilon^{-6}k_{8}y_{1} + k_{6}y_{3},$$

$$y'_{2} = \varepsilon^{-6}k_{8}y_{1} - \varepsilon^{-3}k_{9}y_{2} - \varepsilon^{-2}k_{3}y_{2}y_{5},$$

$$y'_{3} = \varepsilon^{2}k'_{4}y_{4} + \varepsilon^{-2}k_{4}y_{4}y_{3}^{2} - k_{6}y_{3},$$

$$y'_{4} = -\varepsilon^{2}k'_{4}y_{4} - \varepsilon^{-2}k_{4}y_{4}y_{3}^{2} + \varepsilon^{-2}k_{3}y_{2}y_{5},$$

$$y'_{5} = \varepsilon^{2}k_{1} - \varepsilon^{-2}k_{3}y_{2}y_{5}.$$
Historical complex (active MPF) v_{4}

 y_1 cdc2, y_2 p-cdc2, y_3 cyclin-p:cdc2 complex (active MPF), y_4 cyclin-p:cdc2-p complex (inactive MPF), y_5 cyclin.

Tropical equilibrations

Rescale $y_i = \varepsilon^{a_i} \bar{y}_i$ and get

$$\begin{split} \bar{y}_1' &= \epsilon^{-3+a_2-a_1} k_9 \bar{y}_2 - \epsilon^{-6} k_8 y_1 + k_6 \epsilon^{a_3-a_1} \bar{y}_3, \\ \bar{y}_2' &= \epsilon^{-6+a_1-a_2} k_8 \bar{y}_1 - \epsilon^{-3} k_9 \bar{y}_2 - \epsilon^{-2+a_5} k_3 \bar{y}_2 \bar{y}_5, \\ \bar{y}_3' &= \epsilon^{2+a_4-a_3} k_4' \bar{y}_4 + \epsilon^{-2+a_3+a_4} k_4 \bar{y}_4 \bar{y}_3^2 - k_6 \bar{y}_3, \\ \bar{y}_4' &= -\epsilon^2 k_4' \bar{y}_4 - \epsilon^{-2+2a_3} k_4 \bar{y}_4 \bar{y}_3^2 + \epsilon^{-2+a_2+a_5-a_4} k_3 \bar{y}_2 \bar{y}_5, \\ \bar{y}_5' &= \epsilon^{2-a_5} k_1 - \epsilon^{-2+a_2} k_3 \bar{y}_2 \bar{y}_5. \end{split}$$

Full tropical equilibration

$$a_1 = 3$$
, $a_2 = 0$, $a_3 = 2$, $a_4 = 0$, $a_5 = 4$.

Tropical equilibrations

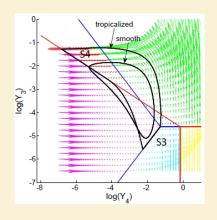
Rescaled equations

$$\begin{split} \bar{y}_1' &= \epsilon^{-6} (k_9 \bar{y}_2 - k_8 \bar{y}_1) + k_6 \epsilon^{-1} \bar{y}_3, \, \bar{y}_2' = \epsilon^{-3} (k_8 \bar{y}_1 - k_9 \bar{y}_2) - \epsilon^2 k_3 \bar{y}_2 \bar{y}_5, \\ \bar{y}_3' &= k_4' \bar{y}_4 + k_4 \bar{y}_4 \bar{y}_3^2 - k_6 \bar{y}_3, \, \bar{y}_4' = \epsilon^2 (-k_4' \bar{y}_4 - k_4 \bar{y}_4 \bar{y}_3^2 + k_3 \bar{y}_2 \bar{y}_5), \\ \bar{y}_5' &= \epsilon^{-2} (k_1 - k_3 \bar{y}_2 \bar{y}_5). \end{split}$$

Reduced model

$$\begin{split} \bar{y}_3' &= k_4' \bar{y}_4 + k_4 \bar{y}_4 \bar{y}_3^2 - k_6 \bar{y}_3, \, \bar{y}_4' = \epsilon^2 (-k_4' \bar{y}_4 - k_4 \bar{y}_4 \bar{y}_3^2 + k_1), \\ k_1 &= k_3 \bar{y}_2 \bar{y}_5, \, k_9 \bar{y}_2 = k_8 \bar{y}_1, \, \epsilon^3 \bar{y}_1(t) + \bar{y}_2(t) + \epsilon^2 \bar{y}_3(t) + \bar{y}_4(t) = 1. \end{split}$$

Tropicalization of the 2D reduced cell cycle model



- the smooth and tropicalized trajectories are qualitatively similar.
- ▶ y₃, y₄ are excitable, not equilibrated in the same time.

Partial tropical equilibrations

Rescale $y_i = \varepsilon^{a_i} \bar{y}_i$ and get

$$\begin{split} \bar{y}_1' &= \epsilon^{-3+a_2-a_1} k_9 \bar{y}_2 - \epsilon^{-6} k_8 y_1 + k_6 \epsilon^{a_3-a_1} \bar{y}_3, \\ \bar{y}_2' &= \epsilon^{-6+a_1-a_2} k_8 \bar{y}_1 - \epsilon^{-3} k_9 \bar{y}_2 - \epsilon^{-2+a_5} k_3 \bar{y}_2 \bar{y}_5, \\ \bar{y}_3' &= \epsilon^{2+a_4-a_3} k_4' \bar{y}_4 + \epsilon^{-2+a_3+a_4} k_4 \bar{y}_4 \bar{y}_3^2 - k_6 \bar{y}_3, \\ \bar{y}_4' &= -\epsilon^2 k_4' \bar{y}_4 - \epsilon^{-2+2a_3} k_4 \bar{y}_4 \bar{y}_3^2 + \epsilon^{-2+a_2+a_5-a_4} k_3 \bar{y}_2 \bar{y}_5, \\ \bar{y}_5' &= \epsilon^{2-a_5} k_1 - \epsilon^{-2+a_2} k_3 \bar{y}_2 \bar{y}_5. \end{split}$$

From the equilibration of y_1 , y_2 , y_5 we find $a_1 = 3$, $a_2 = 0$, $a_5 = 4$.

Partial tropical equilibrations

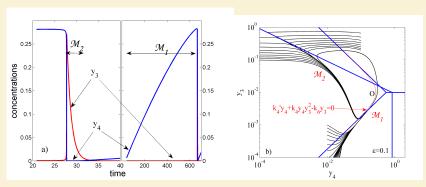
if y_3 is equilibrated, then $a_3 = 2$, $a_4 = 0$ and

$$\bar{y}_3' = k_4' \bar{y}_4 + k_4 \bar{y}_4 \bar{y}_3^2 - k_6 \bar{y}_3, \, \bar{y}_4' = \epsilon^2 (-k_4' \bar{y}_4 - k_4 \bar{y}_4 \bar{y}_3^2 + k_1).$$

if y_4 is equilibrated, then $a_3 = 0$, $a_4 = 4$ and

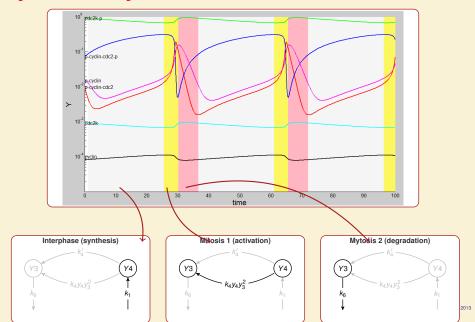
$$\tilde{y}_3' = \epsilon^6 k_4' \tilde{y}_4 + \epsilon^2 k_4 \tilde{y}_4 \tilde{y}_3^2 - k_6 \tilde{y}_3, \ \tilde{y}_4' = -\epsilon^2 k_4' \tilde{y}_4 + \epsilon^{-2} (-k_4 \tilde{y}_4 \tilde{y}_3^2 + k_1).$$

Tropical equilibrations and invariant manifolds



Tropical equilibrations correspond to locally attractive invariant manifolds and slower parts of the cycle.

Hybrid cell cycle model



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