

1 Partial Differential Equations

The aim of this project is the computational solution of partial differential equations with focus in general mathematics or mechanics application. The user is free to choose from a variety of PDE types such as

- Generalized Second Order Partial Differential Equation that can be applied to a domain with up to 4 Dimensions (any space combination of the orthonormal coordinate system (1-2-3) plus time.)
- Generalized Transport Phenomena Equations
 - Steady State / Transient Energy Transfer
 - Steady State / Transient Momentum Transfer
 - Steady State / Transient Mass Transfer
 - Steady State / Transient Transfer of any scalar or vector ϕ via the generalized Convection-Diffusion-Reaction Equation
- Steady State / Transient Linear Elasticity
- Laplace Equation
- Poisson Equation
- Wave Equation

1.1 Generalized Second Order Partial Differential Equation

For a scalar or vector function $\phi = \phi(x_1, x_2, x_3, t)$ the general second order PDE will be of the form

$$\sum_{i,j=1}^4 A_{ij} \frac{\partial^2 \phi}{\partial x_i \partial x_j} + \sum_{i=1}^4 B_i \frac{\partial \phi}{\partial x_i} + C\phi = D \quad (1)$$

with $x_i = x_1, x_2, x_3, t$ being the 3 space dimensions plus the time direction. It should be noted that space is used in a more subtractive manner and can represent anything from physical space to frequencies or whatever the interest of the researcher is (from 1 up to 3 directions).

- $\mathbf{A} = A_{ij}$: The coefficients of the second order derivatives. In physics and mechanics applications they are associated with the molecular mechanism of transportation called diffusion.
 - A_{ij} for $i, j < 3$ are the coefficients of the second order partial derivatives at each space direction.
 - * When $A_{11} = A_{22} = A_{33}$ and $A_{ij} = A_{ji}$ the host medium is considered as isotropic meaning that it is symmetrical and homogeneous and has the same physical properties in all directions and it does not have any preferred direction of orientation. Examples of isotropic materials include most liquids and gases, as well as some solids such as glass and certain types of metals.
 - * When $A_{11} \neq A_{22} \neq A_{33}$ and $A_{ij} \neq A_{ji}$ the host medium is anisotropic meaning that it is non-symmetrical and non-homogeneous. It does not have the same physical properties in all directions and it has preferred directions of orientation. Examples of anisotropic materials some metals and ceramics
 - A_{i4}, A_{4j} are the coefficients of the second order partial derivatives with respect to time.
 - * When $A_{i4}, A_{4j} \neq 0$ the equation is transient.
 - * When $A_{i4}, A_{4j} = 0$ the equation is in steady state.

1.2 Transport / Conservation Laws

The general integral form of a Conservation Law is given by Raynold's Transport Theorem as follows:

$$\frac{d}{dt} \int_{\Omega} U d\Omega = - \oint_S \mathbf{F} \cdot \mathbf{n} dS + \int_{\Omega} Q d\Omega \quad (2)$$

where Ω is the Control Volume (CV), \mathbf{n} is the normal vector, \mathbf{F} are the forces by the flux in and out of the CV from the boundaries (surface S) and Q is the source/sink term. By implementing Gauss's Theorem (Divergence Theorem) the surface integral of the fluxes can be transformed in a volume integral and (2) can be written as follows:

$$\frac{d}{dt} \int_{\Omega} U d\Omega = - \int_{\Omega} \nabla \cdot \mathbf{F} d\Omega + \int_{\Omega} Q d\Omega \quad (3)$$

By transferring all parts of (3) in the left hand side, the integrated quantity should be equal to zero, leading to the differential form of Raynold's Transport Theorem:

$$\frac{d}{dt} U + \nabla \cdot \mathbf{F} = Q \quad (4)$$

It should be noted that the differential form is valid only under the assumption of continuous quantities. In the case of discontinuous functions the integral form should be implemented.

1.2.1 Energy Transfer Equation

Compressibility: A flow is considered as compressible if the pressure or temperature changes due to flow are large enough to cause significant density changes.

$$\text{Mach Number} = \frac{V_{flow}}{V_{sound}}$$

Generally gases become compressed after $Ma > 0.3$

1.2.2 Deravation

Total Energy in Control Volume = Internal Energy + Kinetic Energy

Internal Energy : $\rho e dx_1 dx_2 dx_3 \left[\frac{J}{Kg} \right]$

Internal Energy : $\frac{1}{2} \rho e dx_1 dx_2 dx_3 \left[\frac{J}{Kg} \right]$