



CRACKIT

"TARGET EXAMS!"

VERSION 2.1.0 2020



DESCRIPTION

- LAST UPDATE - 26/08/2020
- LATEST VERSION - 2.1.0
- MADE BY S.I.R.O
- COUNTRY ORIGIN > INDIA

SUBJECTS SUPPORTED

- PHYSICS
- CHEMISTRY
- MATHS
- CUSTOM

REQUIREMENTS

- WINDOWS 10
- FONTS (SETUPS ARE PROVIDED)
 - Bahnschrift Light
 - Bahnschrift
 - Bahnschrift Light SemiCondensed

SELECT SUBJECT

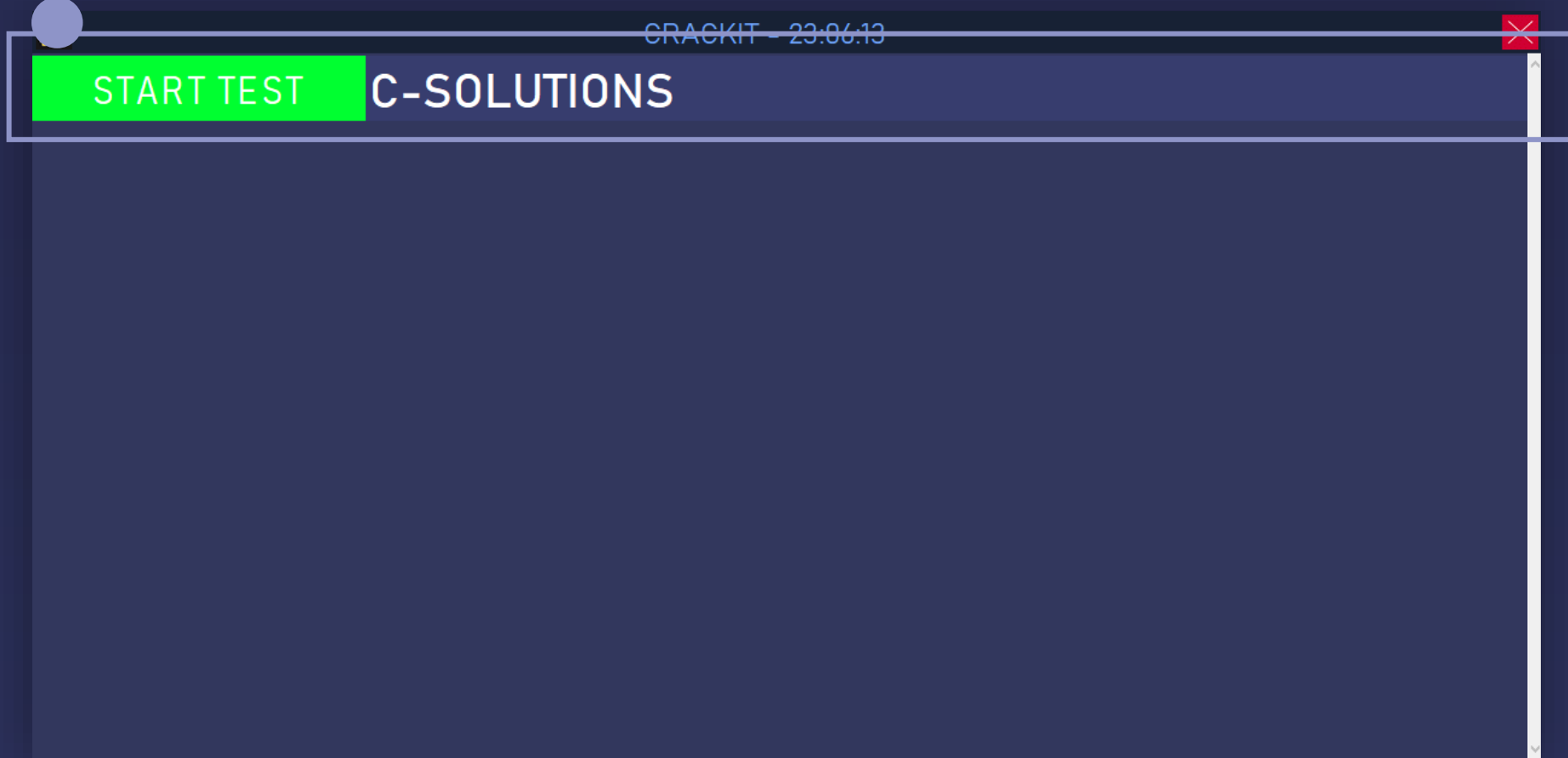


OTHER FEATURES

SELECT LEVEL

SELECT CHAPTER

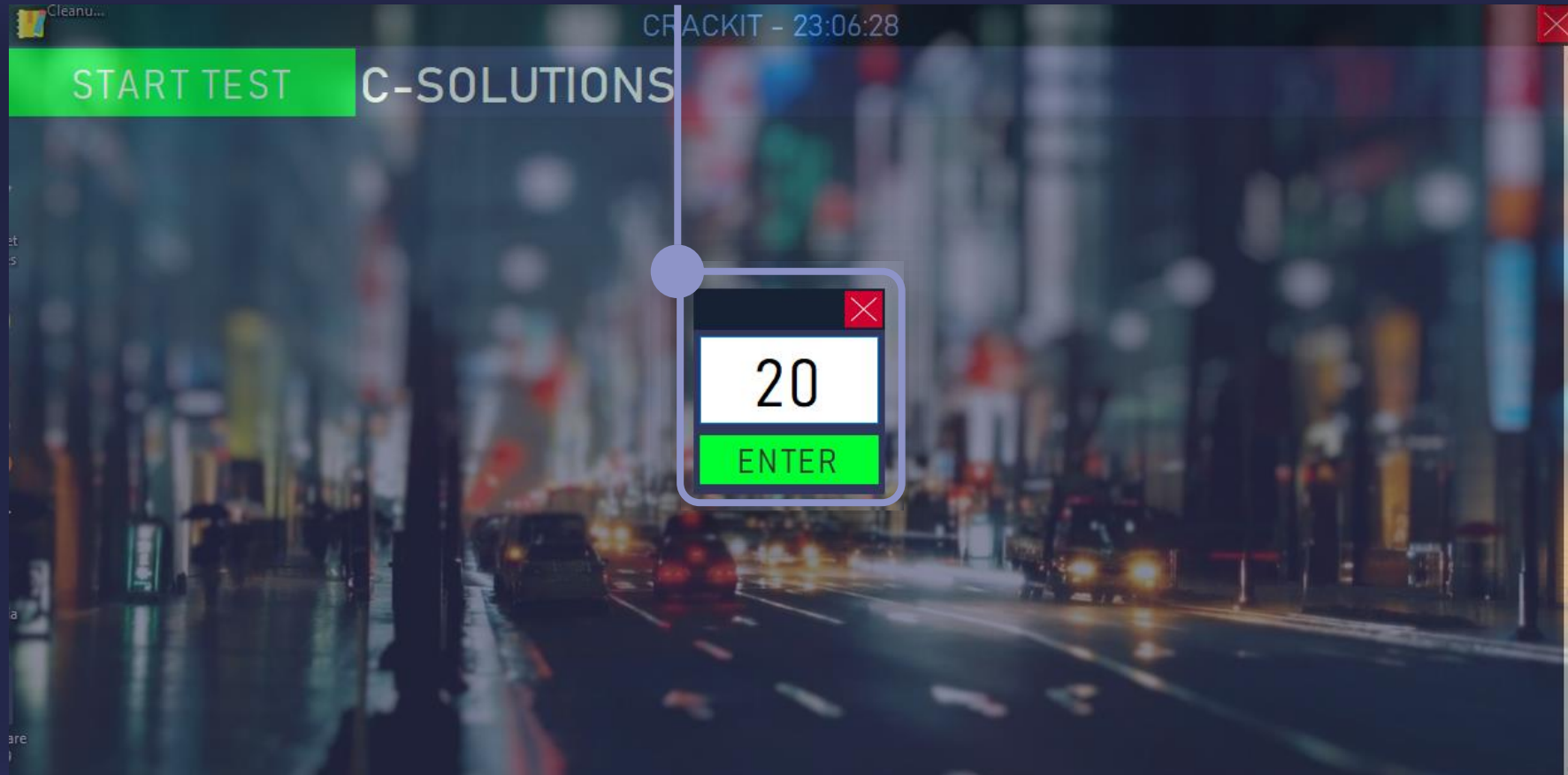
(Chapter will appear only if u have enough questions stored.)



SHORTCUT KEYS:-

1. "Esc" - TO GO BACK TO HOME PAGE

ENTER TOTAL QUESTIONS FOR TEST



QUESTION PANEL

QUESTION

OPTION BUTTONS

00:01:31

< 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 >

23) Two aqueous solutions, A and B, are separated by a semi-permeable membrane. The osmotic pressure of solution A immediately begins to decrease. Which of the following statement is true ?

- (a) The solvent molecules are moving from the solution of higher osmotic pressure to that of lower osmotic pressure.
- (b) The initial osmotic pressure of solution B is greater than that of solution A.
- (c) Solvent molecules are moving from solution B into solution A.
- (d) Both (a) and (b) are true statements.

☐ OPTION (A) ☐ OPTION (B)

☒ OPTION (C) ☐ OPTION (D)

ANSWERED: 16 UNANSWERED: 9

ENDTEST BUTTON

QUESTION STATUS

ENDTEST BUTTON

TEST RESULT

Avast Cleanup 00:02:42

8 9 10 11 12 13 14 15 16 17 18 19 20 21

20) If two liquids A ($P_A^\circ = 100$ torr) and B ($P_B^\circ = 200$ torr) which are completely miscible with each other (each one will behave independently of the other) are present in a closed system, the total vapour pressure of the system will be:

(a) less than 100 torr (b) greater than 200 torr
(c) between 100 to 200 torr (d) 300 torr

☐ OPTION (A) ☐ OPTION (B)
☐ OPTION (C) ☐ OPTION (D)

ANSWERED: 17 UNANSWERED: 8 TIME TOOK: 0 SEC CORRECT OPTION: (D)

✖ YOUR RESULT

ANSWERED: 17

UNANSWERED: 8

CORRECT: 4

INCORRECT: 13

SPEED(SECS/Q): 0.05

MARKS: 3/100

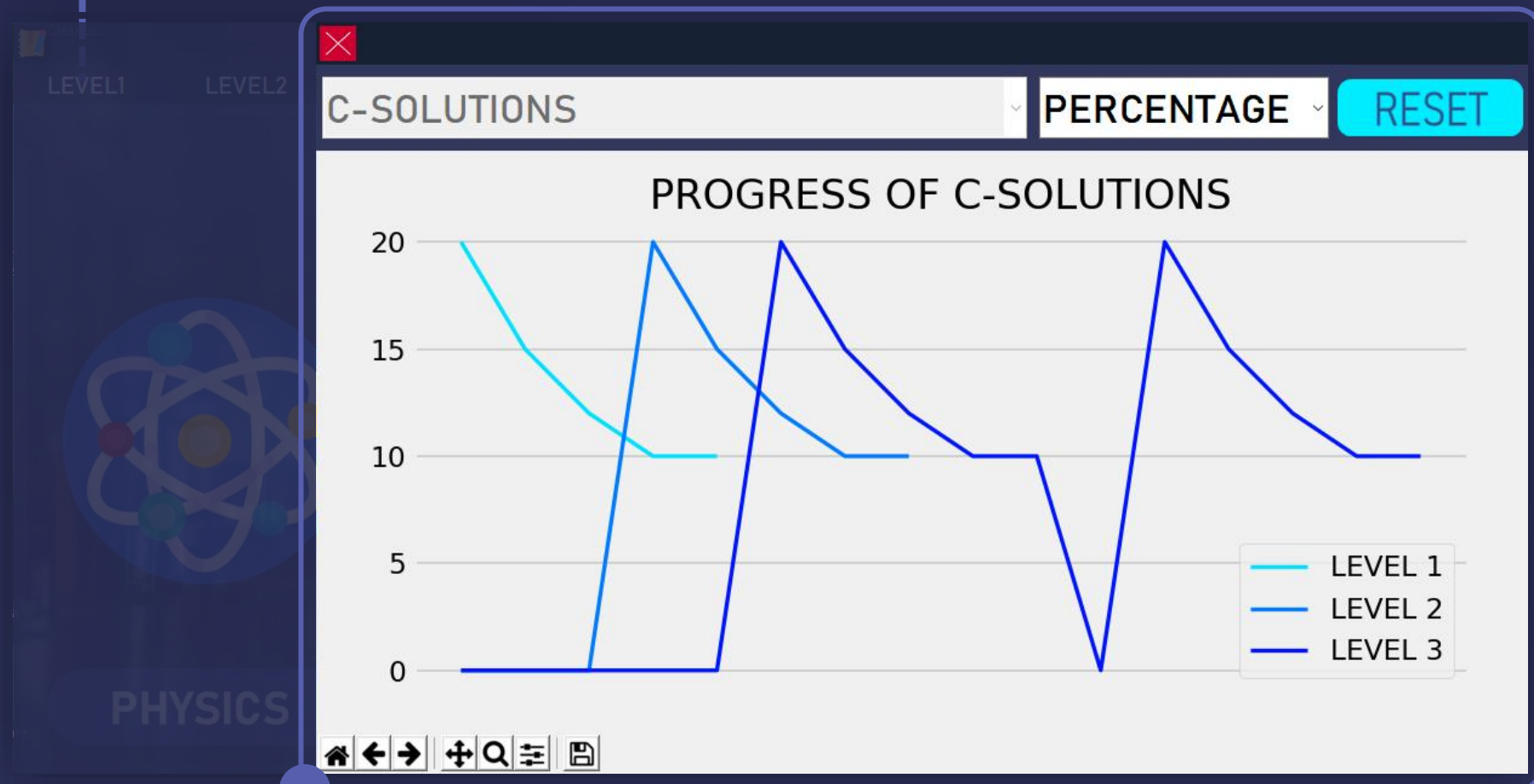
ANALYSIS

ADD CHAPTERS

ADD QUESTIONS

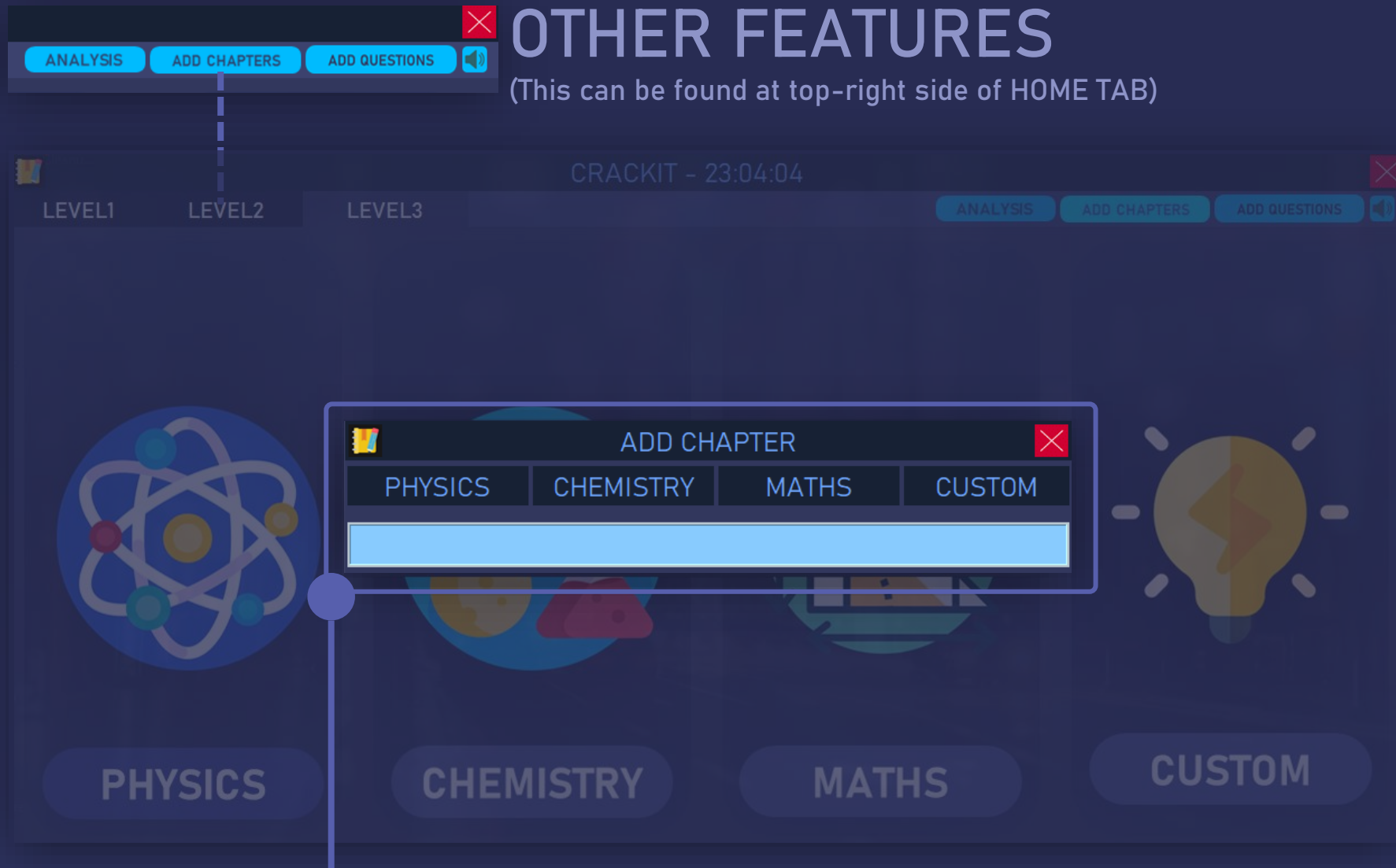
OTHER FEATURES

(This can be found at top-right side of HOME TAB)



ANALYSIS

(Available only if you have given test on that particular chapter)



OTHER FEATURES

(This can be found at top-right side of HOME TAB)

ADD NEW CHAPTER

(Select the 'SUBJECT', Enter the 'CHAPTER NAME', Hit 'ENTER BUTTON' on your keyboard)

ANALYSIS

ADD CHAPTERS

ADD QUESTIONS

OTHER FEATURES

(This can be found at top-right side of HOME TAB)

QUESTION PATH: documents/5656.jpg

SUBJECT:

PHYSICS

CHEMISTRY

MATHS

CUSTOM

CHAPTER:

M-ELLIPSE

LEVEL:

LEVEL1

LEVEL2

LEVEL3

CORRECT:

OPTION (A)

OPTION (B)

OPTION (C)

OPTION (D)

SAVE QUESTION

RECENT ACTIVITY

EMPTY!

SELECT SUBJECT

RECENT ACTIVITY

SELECT LEVEL

SELECT
CORRECT OPTION

SHORTCUT KEYS:-

1. "o"/"O" - TO OPEN QUESTION PATH DIRECTLY
2. "ENTER" - TO SAVE THE QUESTION

CRACKIT - 22:48:37

CRACKIT - 23:06:13

CRACKIT - 23:06:28

00:01:31

00:02:09

00:02:42

20)

If two liquids A ($P_A^\circ = 100$ torr) and B ($P_B^\circ = 200$ torr) which are completely immiscible each other (each one will behave independently of the other) are present in a closed container, the total vapour pressure of the system will be:

(a) less than 100 torr (b) greater than 200 torr

(c) between 100 to 200 torr (d) 300 torr

OPTION (A)

OPTION (B)

OPTION (C)

OPTION (D)

YOUR RESULT

ANSWERED: 17

UNANSWERED: 8

CORRECT: 4

INCORRECT: 13

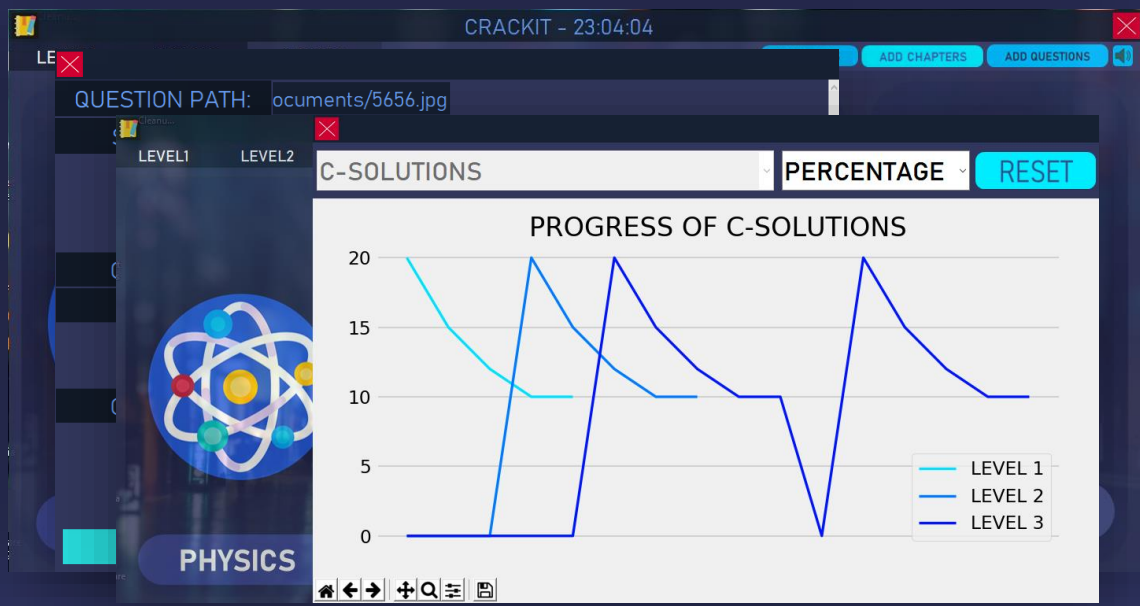
SPEED(SECS/Q): 0.05

MARKS: 3/100

ANSWERED: 17 UNANSWERED: 8 TIME TOOK: 0 SEC CORRECT OPTION: (D)

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V2.1 2020





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