

TABLE II. CARBANIONS DERIVED FROM KETONES  
A. Enolate from Acetone

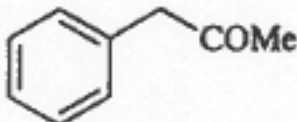
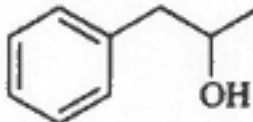
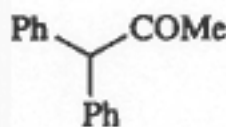
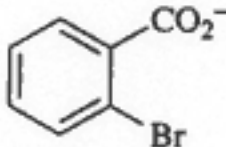
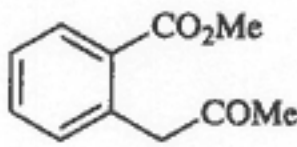
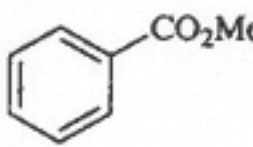
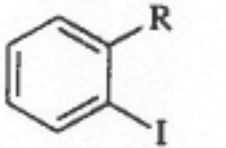
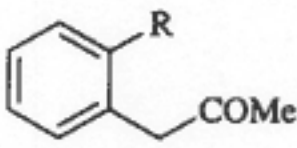
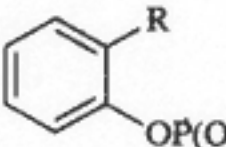
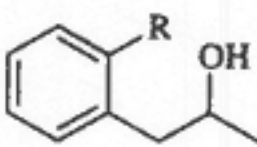
Substrate	Nucleophile	Conditions	Product(s) and Yield(s) (%)	Refs.	
PhF	$^-\text{CH}_2\text{COMe}$	K, NH <sub>3</sub> , -78°	 I (3) +  II (46)	94	
		<i>hν</i> , NH <sub>3</sub> , 3.3 h	I (60) + C <sub>6</sub> H <sub>6</sub> III (31)	136	
PhCl		1. K, NH <sub>3</sub> , -78° 2. [O]	I (68)	94	
		<i>hν</i> , NH <sub>3</sub> , 3 h	I (61) + III (31) +  IV (5)	136	
PhBr		K, NH <sub>3</sub> , -78°	I (67) + II (10) + IV (14)	94	
		<i>hν</i> , NH <sub>3</sub> , 11 min	I (85) + IV (14)	136	
		<i>hν</i> , NH <sub>3</sub> :Et <sub>2</sub> O 1:1, 0.5 h	I (41) + III (13)	108	
		<i>hν</i> , DMSO, 1 h	I (61) + IV (6)	108	
PhI		1. K, NH <sub>3</sub> , -78° 2. [O]	I (71)	94	
		<i>hν</i> , NH <sub>3</sub> , 5 min	I (67) + III (20) + IV (10)	136	
		<i>hν</i> , DMSO, 1 h	I (88) + IV (4)	138, 340, 41	
		DMSO, 1 h	I (50)	62	
		FeCl <sub>2</sub> , DMSO, 10 min	I (60)	75	
PhOPh		K, NH <sub>3</sub> , -78°	II (5)	94	
		<i>hν</i> , NH <sub>3</sub> , 4.2 h	I (14) + PhOH (20)	136	
PhOP(O)(OEt) <sub>2</sub>		K, NH <sub>3</sub>	I (5) + II (56) + III (27)	108, 94	
		<i>hν</i> , NH <sub>3</sub> , 4.2 h	I (13) + III (11) + PhOH (71)	136	
PhSPh		K, NH <sub>3</sub> , -78°	I (18) + II (71) + PhSH (84)	94	
		<i>hν</i> , NH <sub>3</sub> , 0.5 h	I (66) + III (26) + IV (5) + PhSH (97)	136	
Ph <sub>3</sub> S <sup>+</sup> Cl <sup>-</sup>		<i>hν</i> , NH <sub>3</sub> , 1.2 h	I (75) + PhSH (52) + Ph <sub>2</sub> S (22)	136	
PhSePh		<i>hν</i> , NH <sub>3</sub> , 3.3 h	I (95) + PhSeH (83)	136	
$\text{PhNMe}_3^+\text{I}^-$		K, NH <sub>3</sub>	I (46) + II (18) + IV (7)	94	
		<i>hν</i> , NH <sub>3</sub> , 1 h	I (57) + III (37)	136	
PhN <sub>2</sub> SBu- <i>t</i>		Lab. light, DMSO, 1.7 h	I (75)	58	
		1. <i>hν</i> , NH <sub>3</sub> , 1.5 h 2. CH <sub>2</sub> N <sub>2</sub>	 (85) +  (10)	168, 108	
		<i>hν</i> , NH <sub>3</sub>	 I + PhR II		
<u>R</u>			<u>I</u>	<u>II</u>	
OMe		5-15 min	(67)	(—)	291
CH <sub>2</sub> NHCO <sub>2</sub> Et		0.8 h	(58)	(11)	291, 290
CH <sub>2</sub> NHCOMe		0.7 h	(30)	(28)	291
		K, NH <sub>3</sub>	I +  II + PhR III	108	
<u>R</u>			<u>I</u>	<u>II</u>	<u>III</u>
Pr- <i>i</i>			(6)	(24)	(54)
Bu- <i>t</i>			(1)	(4)	(73)



TABLE II. CARBANIONS DERIVED FROM KETONES (Continued)

## A. Enolate from Acetone (Continued)

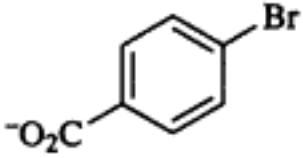
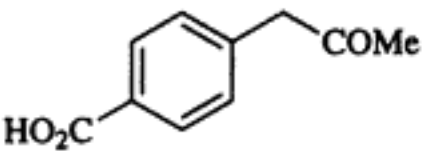
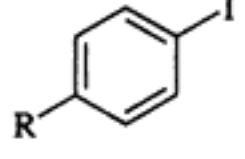
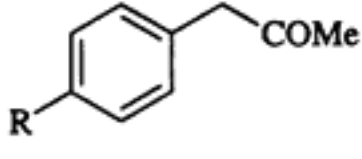
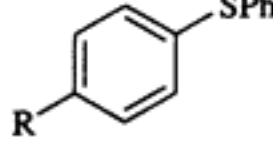
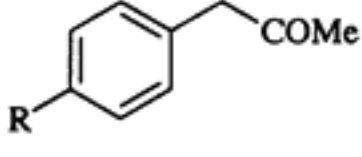
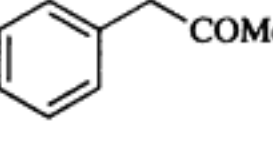
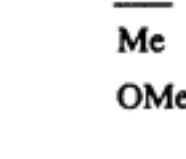
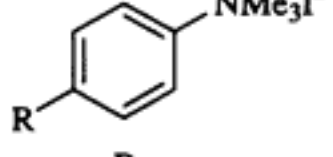
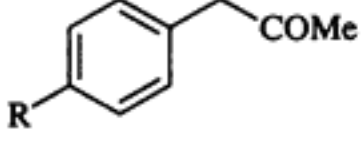
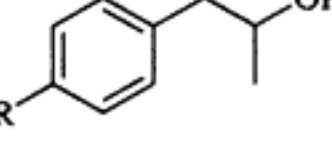
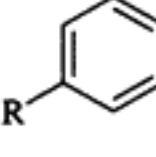
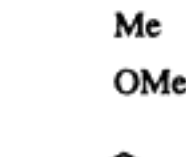
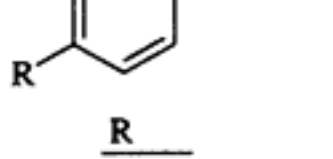
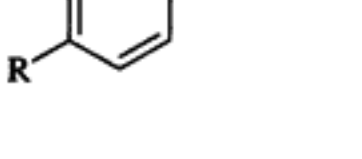
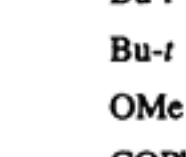
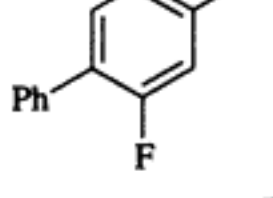
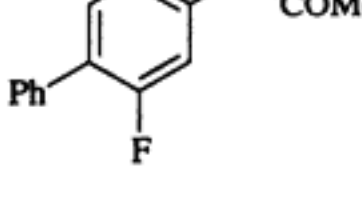
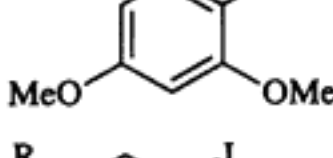
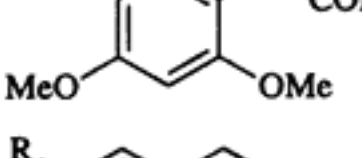
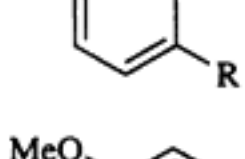
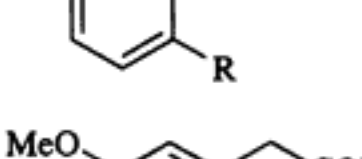
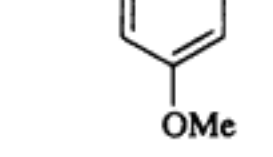
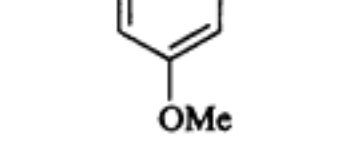
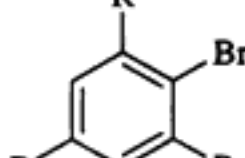
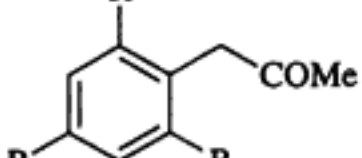
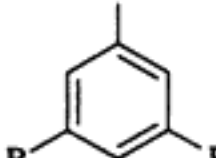
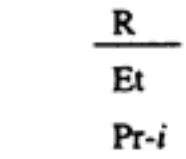
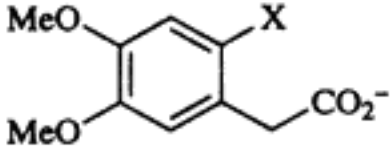
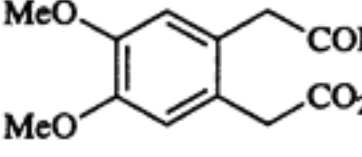
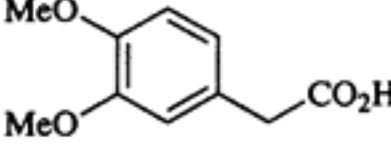
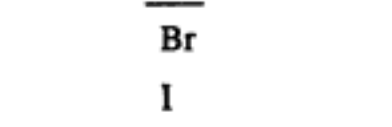
Substrate	Nucleophile	Conditions	Product(s) and Yield(s) (%)	Refs.
	$^-\text{CH}_2\text{COMe}$	1. $h\nu$ , $\text{NH}_3$ , 1.5 h 2. $\text{H}_3\text{O}^+$	 (70)	108
		$h\nu$ , $\text{NH}_3$	 $\xrightarrow{\text{R-NH}_2}$ (33) <sup>a</sup> $\xrightarrow{\text{R-NMe}_2}$ (90)	106 146
		$h\nu$ , $\text{NH}_3$	 + 	160
		1.75 h 1.5 h	(38) (50)	(28) (16)
		$\text{K}$ , $\text{NH}_3$ , $-78^\circ$	 +  + 	94
			(30) (20)	(42) (39)
		Lab. light, DMSO,		58
		0.5 h 1.5 h 1.5 h 0.75 h	(42) (86) (69) (78)	
		1. $h\nu$ , $\text{NH}_3$ , 2 h 2. $\text{MeI}$	 (62)	164
		$h\nu$ , $\text{NH}_3$ , 2 h	 (76)	108
		$h\nu$ , $\text{NH}_3$ , 2.2 h	 $\xrightarrow{\text{R-Pr-i}}$ (78) $\xrightarrow{\text{R-Bu-t}}$ (26)	108
		$h\nu$ , $\text{NH}_3$ , 1 h	 (68)	108
		$h\nu$ , $\text{NH}_3$	 + 	108
		2 h 2.5 h	(70) (2)	(22) (—)
		1. $h\nu$ , $\text{NH}_3$ 2. $\text{H}_3\text{O}^+$	 + 	149
		3 h 25 min	(60) (75-80)	(40) (20-25)

TABLE II. CARBANIONS DERIVED FROM KETONES (*Continued*)A. Enolate from Acetone (*Continued*)

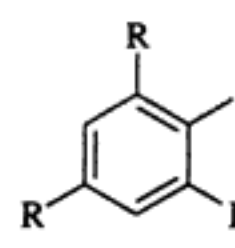
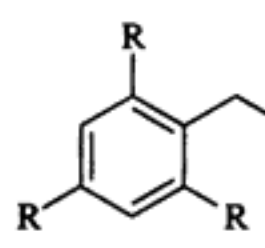
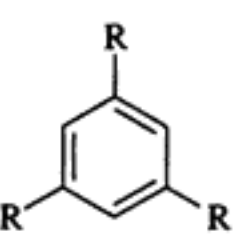
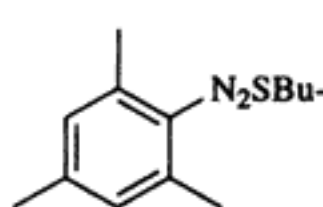
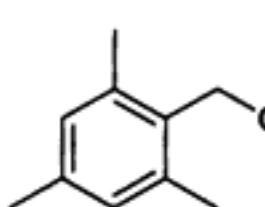
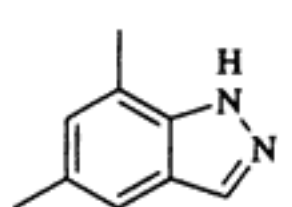
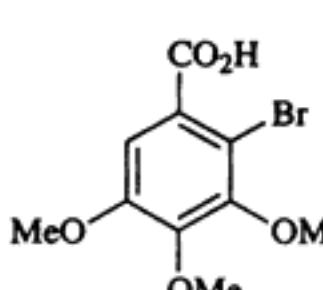
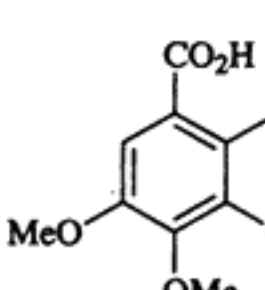
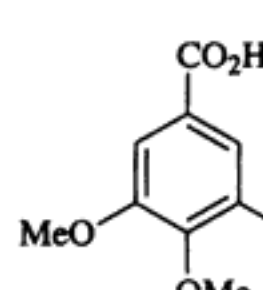
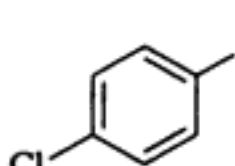
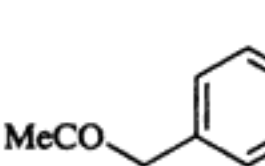
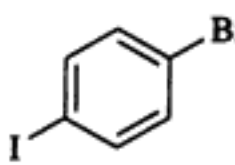
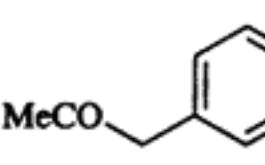
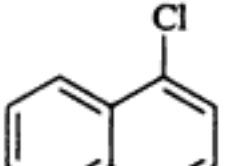
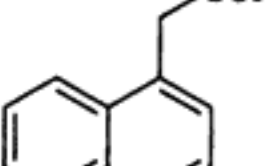
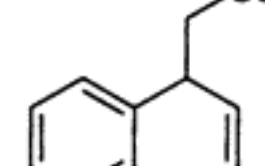
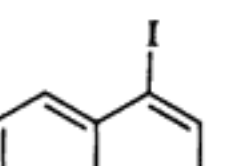
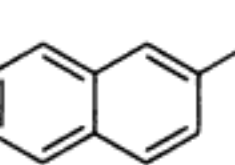
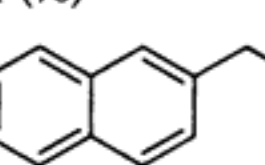
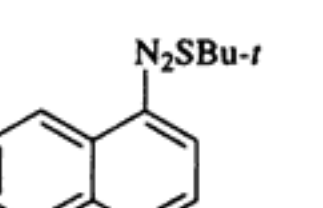
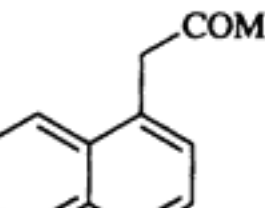
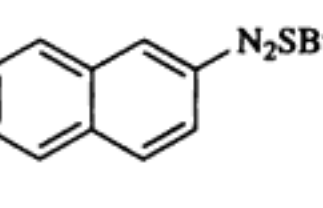
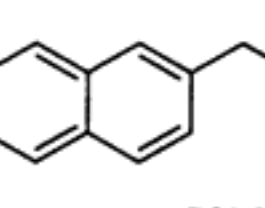
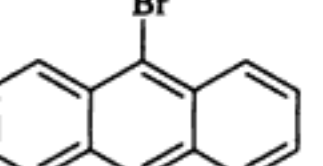
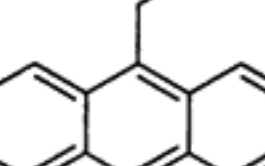
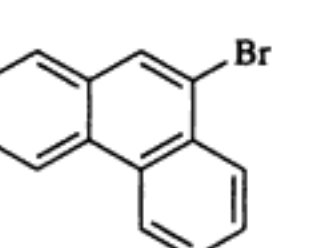
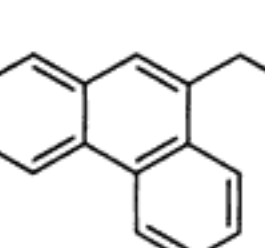
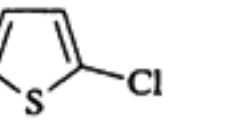
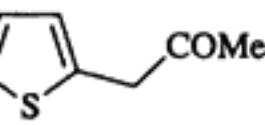
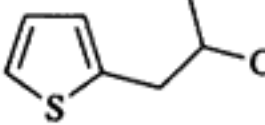

Substrate	Nucleophile	Conditions	Product(s) and Yield(s) (%)	Refs.
 $\frac{\text{R}}{\text{Me}}$ OMe Pr- <i>i</i>	$^-\text{CH}_2\text{COMe}$	$h\nu$ , $\text{NH}_3$  0.5 h 1.5 h 3 h	 +  (82) (10) (92) (—) (16) (37)	108
		Lab light, DMSO, 2 h	 (30) +  (50)	58
		1. $h\nu$ , $\text{NH}_3$ 2. $\text{H}_3\text{O}^+$	 (78) +  (11)	168
		$h\nu$ , $\text{NH}_3$ , 3 h	 (42)	166
		$h\nu$ , $\text{NH}_3$ , 4 h	 (32)	166
		K, $\text{NH}_3$	 I (23) +  (69) <sup>b</sup>	45
		$h\nu$ , $\text{NH}_3$ , 1.5 h	I (88)	45
		Na(Hg), $\text{NH}_3$	I (98)	101
		K, $\text{NH}_3$	I (6) + II (84) <sup>b</sup>	45
		$h\nu$ , $\text{NH}_3$ , 1 h	I (76)	108
		$h\nu$ , $\text{NH}_3$ , 1.5 h	 (75)	108
		Lab light, DMSO, 1 h	 (75)	58
		Lab light, DMSO, 0.5 h	 (76)	58
		$h\nu$ , $\text{NH}_3$ , 1 h	 (98)	108
		$h\nu$ , $\text{NH}_3$ , 1.3 h	 (62)	108
		Na, $\text{NH}_3$ , $-78^\circ$	 I (2) +  (22) +  (46)	152
		$h\nu$ , $\text{NH}_3$ , 1 h	I (17)	152

TABLE II. CARBANIONS DERIVED FROM KETONES (Continued)

## A. Enolate from Acetone (Continued)

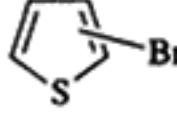
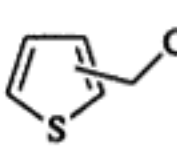
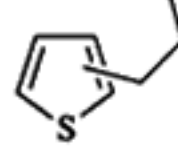
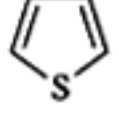
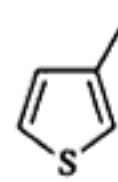
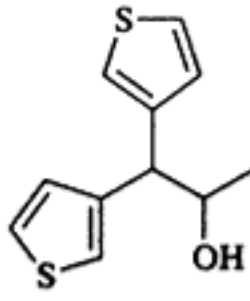
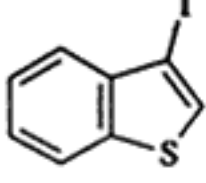
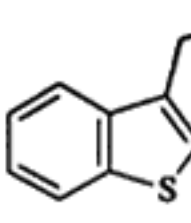
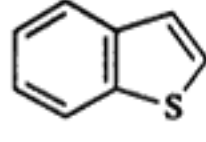
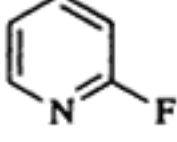
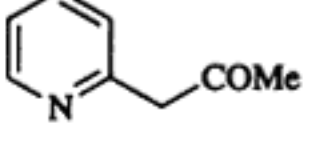
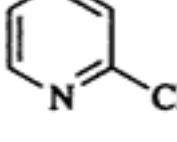
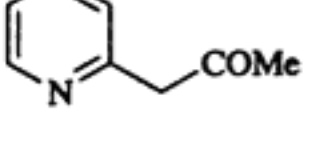
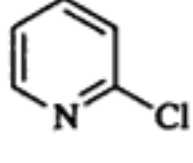
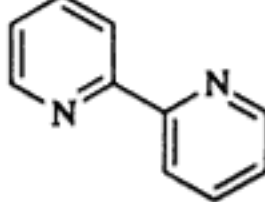
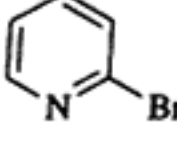
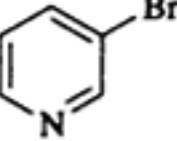
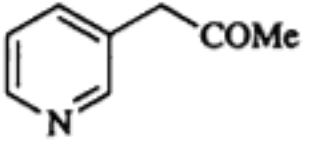
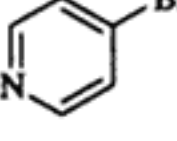
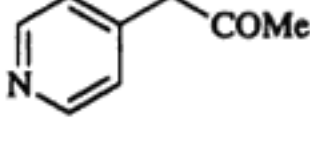
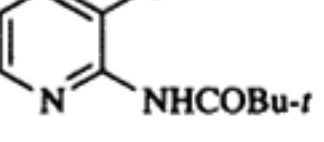
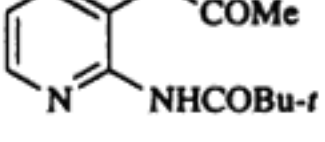
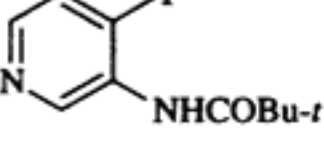
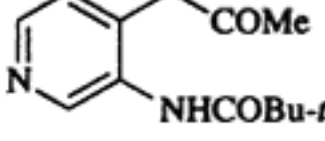
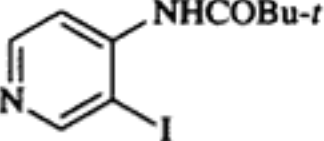
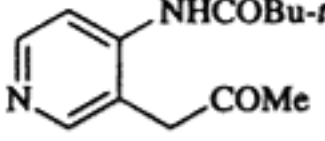
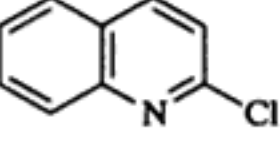
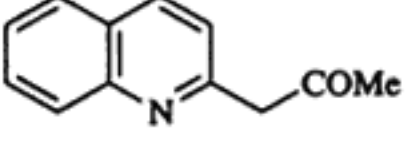
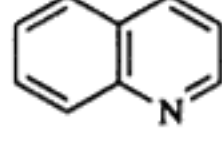
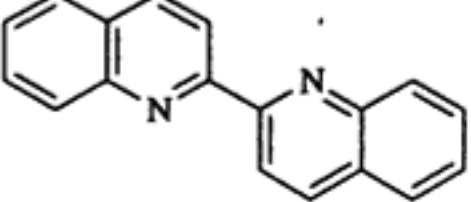
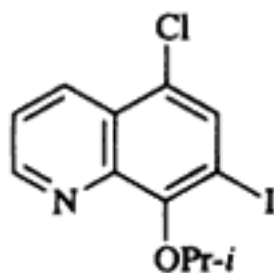
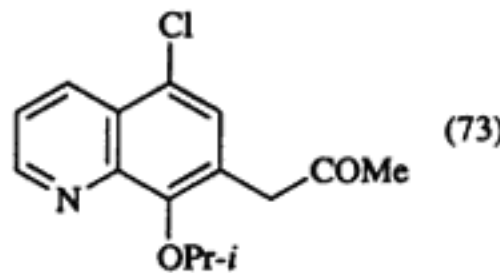
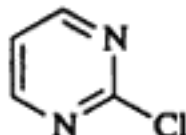
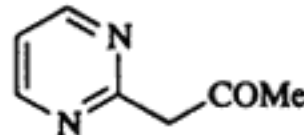
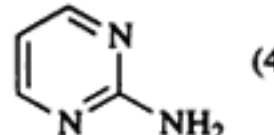
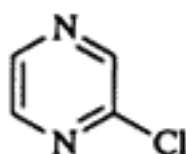
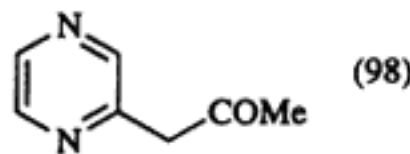
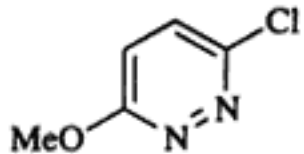
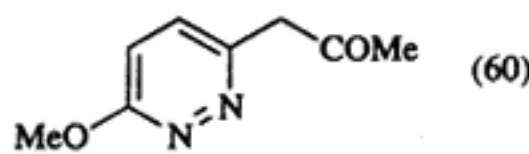
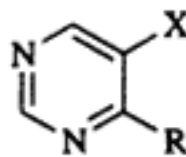
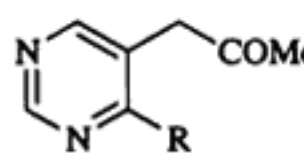
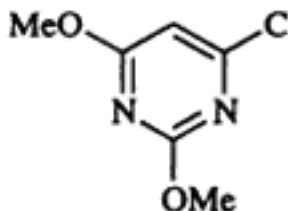
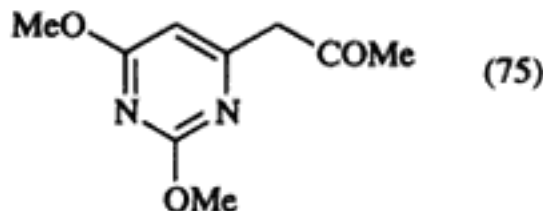
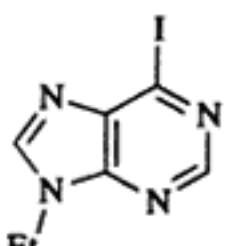
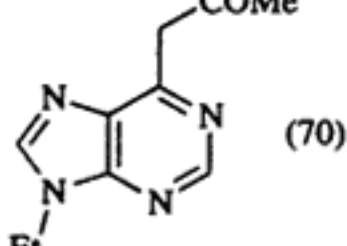
Substrate	Nucleophile	Conditions	Product(s) and Yield(s) (%)	Refs.
	$^-\text{CH}_2\text{COMe}$		 <b>I</b> +  <b>II</b> +  <b>III</b>	
		Na, NH <sub>3</sub> , -78°	2-Br, <b>I</b> (4) + <b>II</b> + (3) + <b>III</b> (70) +  (10)	152
		<i>hν</i> , NH <sub>3</sub> , 1 h	2-Br, <b>I</b> (31)	152
		K, NH <sub>3</sub> , -78°	3-Br, <b>I</b> (29) + <b>II</b> (23) + <b>III</b> (29) +  <b>IV</b> (10)	152
		<i>hν</i> , NH <sub>3</sub> , 1 h	3-Br, <b>I</b> (51) + <b>IV</b> (25)	152
		DMSO, 1 h	 (12) +  (23)	151
		<i>hν</i> , NH <sub>3</sub> , 2 h	 (40)	137
		<i>hν</i> , NH <sub>3</sub> , 1 h	 <b>I</b> (85)	137
		Na(Hg), NH <sub>3</sub>	<b>I</b> (15) +  (10) +  (55)	101
		K, NH <sub>3</sub>	<b>I</b> (4)	137
		<i>hν</i> , NH <sub>3</sub> , 15 min	<b>I</b> (100)	137
		<i>hν</i> , THF, 15 min	<b>I</b> (64)	144
		<i>hν</i> , NH <sub>3</sub> , 15 min	 (65)	137
		<i>hν</i> , NH <sub>3</sub> , 15 min	 (28)	137
		<i>hν</i> , NH <sub>3</sub>	 (90)	139
		<i>hν</i> , NH <sub>3</sub>	 (95)	139
		<i>hν</i> , NH <sub>3</sub>	 (90)	139
		K, NH <sub>3</sub>	 <b>I</b> (43) +  (29)	142
		<i>hν</i> , NH <sub>3</sub> , <sup>c</sup> 1 h	<b>I</b> (90)	142
		<i>hν</i> , NH <sub>3</sub> , <sup>d</sup> 1 h	<b>I</b> (62)	157
		Na(Hg), NH <sub>3</sub>	<b>I</b> (49) +  (50)	101




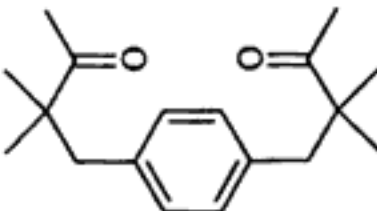
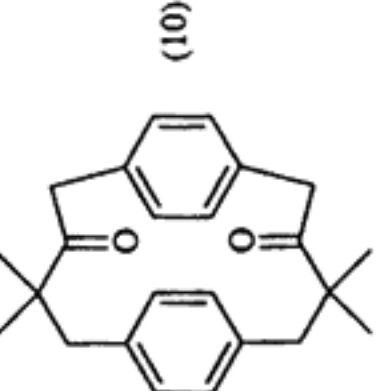
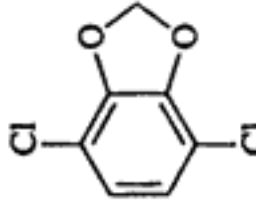
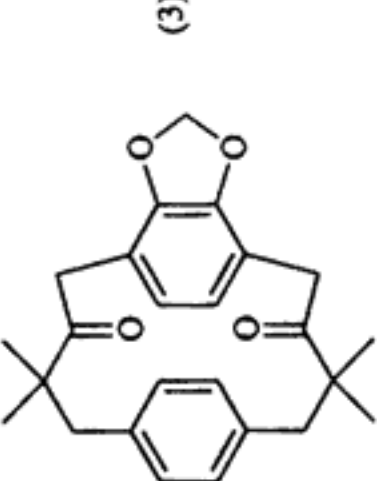
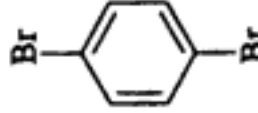
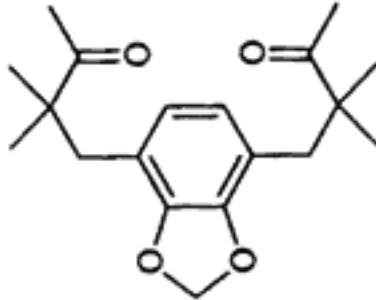
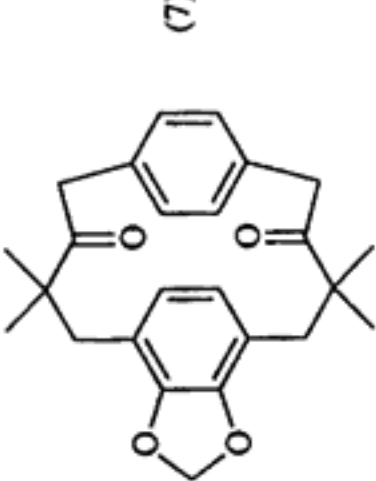
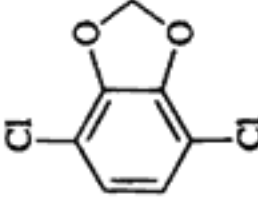
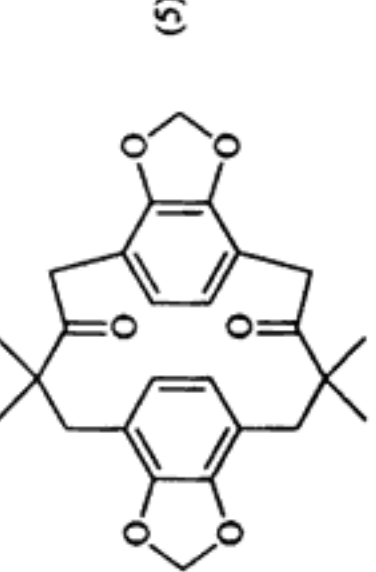
TABLE II. CARBANIONS DERIVED FROM KETONES (Continued)

## A. Enolate from Acetone (Continued)

Substrate	Nucleophile	Conditions	Product(s) and Yield(s) (%)	Refs.																								
	$^-\text{CH}_2\text{COMe}$	$h\nu$ , 1 h		144																								
		THF	I (82)																									
		DMF	I (74)																									
		DME	I (28)																									
		$\text{Et}_2\text{O}$	I (9)																									
		$\text{C}_6\text{H}_6$	I (4)																									
		DMSO, 5 min	I (37)																									
		$h\nu$ , $\text{NH}_3$ , 1 h	 (73)	155																								
		$h\nu$ , $\text{NH}_3$ , 15 min	 I (15) +  (4)	55																								
		$h\nu$ , THF, 15 min, $0^\circ$	I (61)	55																								
		$\text{NH}_3$ , 15 min	 (98)	55																								
		$\text{NH}_3$ , 15 min	 (60)	55																								
				54																								
<table><tr><th>X</th><th>R</th></tr><tr><td>Cl</td><td>Bu-<i>t</i></td></tr><tr><td>Cl</td><td>Bu-<i>t</i></td></tr><tr><td>Cl</td><td>Ph</td></tr><tr><td>Cl</td><td>Ph</td></tr><tr><td>Cl</td><td>Ph</td></tr><tr><td>Br</td><td>Bu-<i>t</i></td></tr><tr><td>Br</td><td>Bu-<i>t</i></td></tr><tr><td>Br</td><td>Bu-<i>t</i></td></tr><tr><td>Br</td><td>Ph</td></tr><tr><td>Br</td><td>Ph</td></tr><tr><td>Br</td><td>Ph</td></tr></table>					X	R	Cl	Bu- <i>t</i>	Cl	Bu- <i>t</i>	Cl	Ph	Cl	Ph	Cl	Ph	Br	Bu- <i>t</i>	Br	Bu- <i>t</i>	Br	Bu- <i>t</i>	Br	Ph	Br	Ph	Br	Ph
X	R																											
Cl	Bu- <i>t</i>																											
Cl	Bu- <i>t</i>																											
Cl	Ph																											
Cl	Ph																											
Cl	Ph																											
Br	Bu- <i>t</i>																											
Br	Bu- <i>t</i>																											
Br	Bu- <i>t</i>																											
Br	Ph																											
Br	Ph																											
Br	Ph																											
		K, $\text{NH}_3$	(42)																									
		$h\nu$ , $\text{NH}_3$ , 1.25 h	(60-65)																									
		$\text{NH}_3$ , 16 h	(39)																									
		K, $\text{NH}_3$	(47)																									
		$h\nu$ , $\text{NH}_3$ , 1.25 h	(20-25)																									
		$\text{NH}_3$ , 16 h	(20-25)																									
		K, $\text{NH}_3$	(30)																									
		$h\nu$ , $\text{NH}_3$ , 1.25 h	(70-75)																									
		$\text{NH}_3$ , 16 h	(30)																									
		K, $\text{NH}_3$	(42)																									
		$h\nu$ , $\text{NH}_3$ , 1.25 h	(25-30)																									
		$h\nu$ , $\text{NH}_3$ , 15 min	 (75)	140																								
		$h\nu$ , $\text{NH}_3$ , 0.5 h	 (70)	153, 154																								

<sup>a</sup> This compound decomposes in contact with air.<sup>b</sup> This product is a mixture of dihydro and tetrahydro derivatives.<sup>c</sup> The potassium salt of the nucleophile was used.<sup>d</sup> A  $\text{LiNH}_2$ :nucleophile ratio of 3 was used.

TABLE IX. RING CLOSURE REACTIONS (Continued)  
C. Miscellaneous (Continued)

Substrate	Nucleophile	Conditions	Product(s) and Yield(s) (%)	Refs.
		$h\nu$ , <i>t</i> -BuOK, NH <sub>3</sub> , 15 min	 (10)	307
		$h\nu$ , <i>t</i> -BuOK, NH <sub>3</sub> , 15 min	 (3)	307
		$h\nu$ , <i>t</i> -BuOK, NH <sub>3</sub> , 15 min	 (7)	307
		$h\nu$ , <i>t</i> -BuOK, NH <sub>3</sub> , 15 min	 (5)	307

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