Óscar's wonderful end of master's thesis

Óscar Iglesias González April 5, 2020

Óscar's Room Editions

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Colophon

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The harmony of the world is made manifest in Form and Number, and the heart and soul and all the poetry of Natural Philosophy are embodied in the concept of mathematical beauty.

– D'Arcy Wentworth Thompson

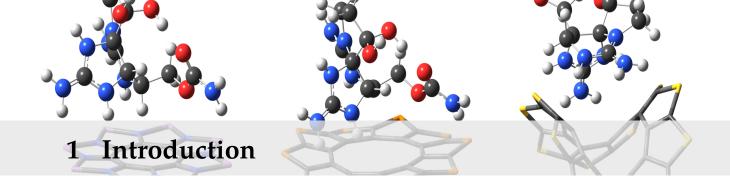
Preface

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1.1 Introduction to marine toxins and saxitoxin (STX)

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1.2 The work up until now

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1.3 Pushing onwards

2 Computational methods and specifications

2.1 General techniques

Table 2.1: Computational techniques

Calculation	Technique	Spec.	Functional	Basis set
Geometry optimizations			M06-2X	def2TZVPP
NICS calculations	DFT	GIAO	b3lyp	base

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2.2 Software

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2.3 Hardware

All of the calculations were perfomed using either the Centro de Supercomputación de Galicia's (CESGA) infraestructures, or the propietary cluster of the S3 research group.

S3's cluster is composed of nodes that include 16 Intel®Xeon®CPU E5-2630 v3 2.40GHz cores and 60 GB of RAM.

CESGA's

3.1 Introduction and origin

The search for effective and interesting SERS substrates ended up leading us to a novel class of molecules researched and presented in the year 2006 by Chernichenko and his colleagues.[1] The first representative of this family, nicknamed as "sulflower", is the ocatathio[8]circulene. This highly symmetric structure, which may be described as a form of carbon sulfide and as a belt of annulated thiophene cycles, is claimed to have great stability, high symmetry and unusual electronic properties.

From a synthetic point of view, sulflower also proved to be simple and straightforward to develop despite its complex appearance: starting from tetrathiophene, sulphurizing its free sites and acidificating ot get polythiol, and removing the excess sulfur by vacuum pyrolysis. This allowed the team to achieve yields of 56% starting from commercially available reagents.

Interestingly, the team proposes that it could be possible to prepare materials with diverse electronic properties by using different types of heteroatoms and varying on the basic structure of the molecule. Such a statement made apparent the potential of this family of molecules: highly symmetrical, stable, surface-like structures with variable electronic behavior could act as suitable SERS substrates. This chapter is entirely dedicated to that premise: the study and characterization of sulflower and sulflower-like molecules, which from now on I will collectively refer to as "sunflowers". By designing, generating and studying our own family of sunflowers, we will be contributing to characterize a novel and interesting group of molecules, and we may be able to identify an ideal SERS environment for STX.

[1]: Chernichenko et al. (2006), "'Sulflower": A New Form of Carbon Sulfide'

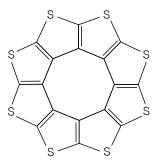


Figure 3.1: Structure of sulflower

3.2 Sunflower design

To start designing a family of molecules, first we must clearly state their defining pattern. For this purpose, we adopt Chernichenko et al.'s own proposal: "a novel class of heterolytic circulenes". We start expanding the model by answering the question: are thiophene based circulenes with other than 8 rings stable enough to be worth considering? The answer is in the original paper itself. Figure 3.2, which was recreated to use our calculation level and adapt to the style of the document, shows that 8 ring structures are the most stable alongside 9. However, considering their low relative energies and the fact that they have an even number of electrons (which would greatly simplify later calculations), 10 and 12 ring sunflowers were also chosen as part of the study.

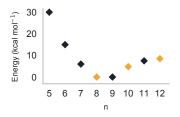
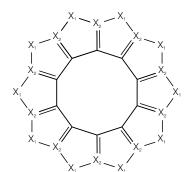


Figure 3.2: Strain of thiophenic circulenes with n rings

$$X_1$$
 X_2
 X_1
 X_2
 X_3
 X_4
 X_2
 X_4
 X_2
 X_4
 X_4
 X_2
 X_4
 X_4
 X_4
 X_4
 X_5
 X_5
 X_5
 X_5
 X_5
 X_7
 X_8
 X_8
 X_8



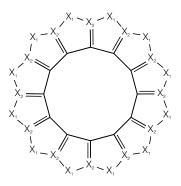


Figure 3.3: From left to right, general structures of the 8, 10 and 12 ring sunflowers

Table 3.1: Subset of the sunflower family that is going to be studied

X_1	X_2	8	n 10	12
S	С	S08	S10	S12
Se	C	Se08	Se10	Se12
As	C	As08	As10	As12
As	N	AsN08	AsN10	AsN12
P	C	P08	P10	P12
P	N	PN08	PN10	PN12

Expanding upon this idea to allow for further heteroatom substitution, we ended up with the templates in Figure 3.3. The possibilities were numerous, but we settled for S, Se, As and P substitutions on X_1 sites, and N substitutions in some cases in X_2 sites. A full table detailing all of the structures that were generated and will comprise this study can be found in Table 3.1, as well as the short names or IDs that were given to each species based on its composition and number of petals for the sake of abbreviation.

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3.3 Study of geometry

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Table 3.2: Bond length statistics for each studied species

		C-	·C	C-	X_2	X ₁ -2	X_2
		mean	std	mean	std	mean	std
	ring	0.000	0.000	0.000	0.000	0.000	0.000
S	08	0.000	0.000	0.000	0.000	0.000	0.000
	10	0.000	0.000	0.000	0.000	0.000	0.000
	12	0.000	0.000	0.000	0.000	0.000	0.000
	ring	0.000	0.000	0.000	0.000	0.000	0.000
Se	08	0.000	0.000	0.000	0.000	0.000	0.000
	10	0.000	0.000	0.000	0.000	0.000	0.000
	12	0.000	0.000	0.000	0.000	0.000	0.000
	ring	0.000	0.000	0.000	0.000	0.000	0.000
As	08	0.000	0.000	0.000	0.000	0.000	0.000
	10	0.000	0.000	0.000	0.000	0.000	0.000
	12	0.000	0.000	0.000	0.000	0.000	0.000
	ring	0.000	0.000	0.000	0.000	0.000	0.000
AsN	08	0.000	0.000	0.000	0.000	0.000	0.000
	10	0.000	0.000	0.000	0.000	0.000	0.000
	12	0.000	0.000	0.000	0.000	0.000	0.000
	ring	0.000	0.000	0.000	0.000	0.000	0.000
P	08	0.000	0.000	0.000	0.000	0.000	0.000
	10	0.000	0.000	0.000	0.000	0.000	0.000
	12	0.000	0.000	0.000	0.000	0.000	0.000
	ring	0.000	0.000	0.000	0.000	0.000	0.000
PN	08	0.000	0.000	0.000	0.000	0.000	0.000
	10	0.000	0.000	0.000	0.000	0.000	0.000
	12	0.000	0.000	0.000	0.000	0.000	0.000

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3.4 Study of aromaticity

Aromaticity is a property of molecules that have a ring or a chain of resonance bonds that increases their stability. Initially associated with benzene, it's typically found in flat ring structures (although there are other varieties such as homoaromatic and three-dimensional aromatic systems).

As π electrons in aromatic systems have free circulation, an external magnetic field perpendicular to the main plane of the system is able to induce a ring current. The NICS techinque as a probe for aromaticity is based on the idea that NMR chemical shifts are influenced by said ring currents, as a consecuence of Ampère's law. Ring currents generate their own magnetic field, which will weaken or strengthen the effect of the

external field resulting in decreased or increased shifts. Aromatic systems will experience shielding on the inside of the ring and deshielding on the outside. Antiaromatic systems will experience the opposite.

Nucleus-Independent Chemical Shift (NICS)

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The basics of NICS

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A custom solution for non-planar molecules

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AICD

The basics of AICD

the original language. There is no need for special content, but the length of words should match the language.

Application and comparison of results

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3.5 Spectroscopic characterization

Vibrational spectroscopy

Raman spectra

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Decomposition of key modes

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VEDA

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Electronic spectroscopy

UV spectra

Having obtained a general characterization of the sunflower-type molecules, it's time to get back to the problem at hand and start looking into how they can be applied.

4.1 Study of adsorption

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Sampling and optimization

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Maxwell-Boltzmann statistics

Basis Set Superposition Error correction

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4.2 Study of non-covalent interactions

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4.3 Study of UV behavior

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General UV spectroscopy

Charge transfer analysis

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4.4 Resonance Raman

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Generation and comparison of spectra

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Combined resonance graphs

Final selection



A Some more blindtext

Bibliography

Here are the references in citation order.

[1] Konstantin Yu. Chernichenko et al. "Sulflower": A New Form of Carbon Sulfide'. In: *Angew. Chem. Int. Ed.* 45 (2006), pp. 7367–7370 (cited on page 5).

Notation

The next list describes several symbols that will be later used within the body of the document.

- c Speed of light in a vacuum inertial frame
- *h* Planck constant

Greek Letters with Pronounciation

Character	Name	Character	Name
α	alpha <i>AL-fuh</i>	ν	nu NEW
β	beta BAY-tuh	ξ , Ξ	xi KSIGH
γ, Γ	gamma GAM-muh	O	omicron OM-uh-CRON
δ , Δ	delta DEL-tuh	π , Π	pi <i>PIE</i>
ϵ	epsilon EP-suh-lon	ρ	rho ROW
ζ	zeta ZAY-tuh	σ, Σ	sigma SIG-muh
η	eta AY-tuh	τ	tau TOW (as in cow)
θ, Θ	theta THAY-tuh	v, Υ	upsilon OOP-suh-LON
ι	iota eye-OH-tuh	ϕ , Φ	phi FEE, or FI (as in hi)
к	kappa KAP-uh	X	chi KI (as in hi)
λ , Λ	lambda <i>LAM-duh</i>	ψ , Ψ	psi SIGH, or PSIGH
μ	mu MEW	ω, Ω	omega oh-MAY-guh

Capitals shown are the ones that differ from Roman capitals.