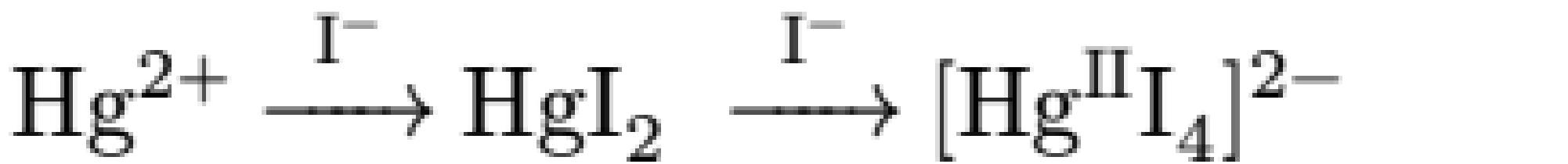


Chemical Formulae



Chemical Equations



Charges

H⁺

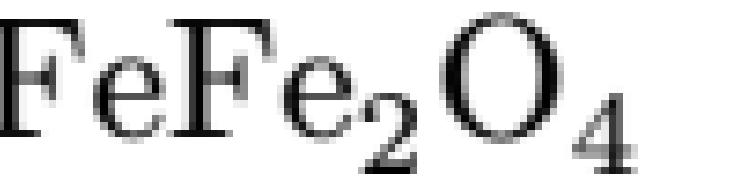
CrO₄²⁻ CrO₄²⁻

[AgCl₂]⁻

Y⁹⁹⁺ Y⁹⁹⁺

Oxidation States

II III



II III



Stoichiometric Numbers

$2\text{H}_2\text{O}$

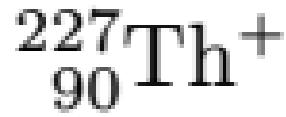
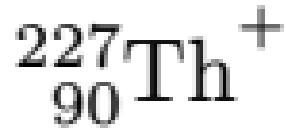
$2 \text{ H}_2\text{O}$

$2 \text{ H}_2\text{O}$

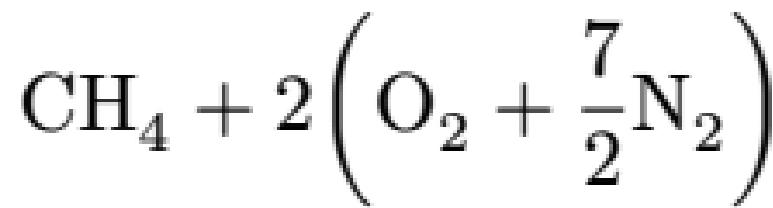
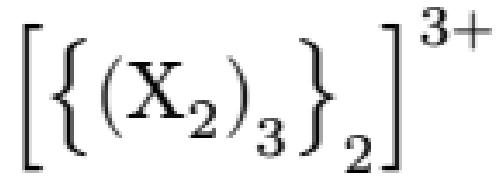
$\frac{1}{2} \text{ H}_2\text{O}$

$1/2 \text{ H}_2\text{O}$

Nuclides, Isotopes



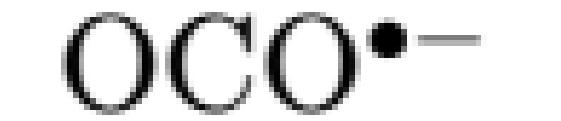
Parenthesis, Braces, Brackets



States of Aggregation



Radical Dots



Escaped Modes

NO_x is the same as NO_x

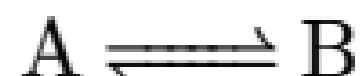
Fe^{n+} is the same as Fe^{n+}

$\text{Fe}(\text{CN})_{\frac{6}{2}}$

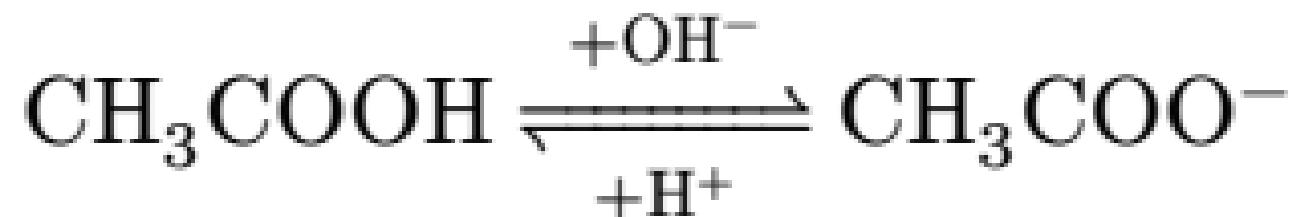
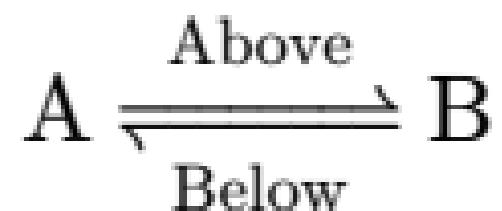
CO_2^3

$\mu\text{-Cl}$

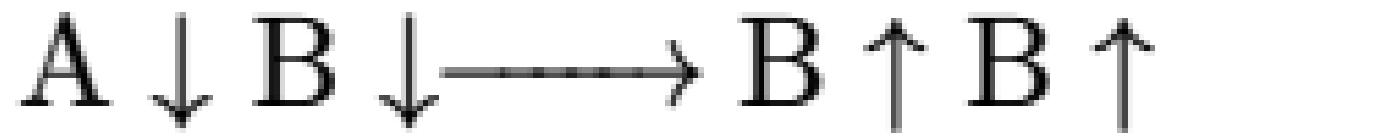
Reaction Arrows



Above/Below Arrow Text



Precipitation and Gas

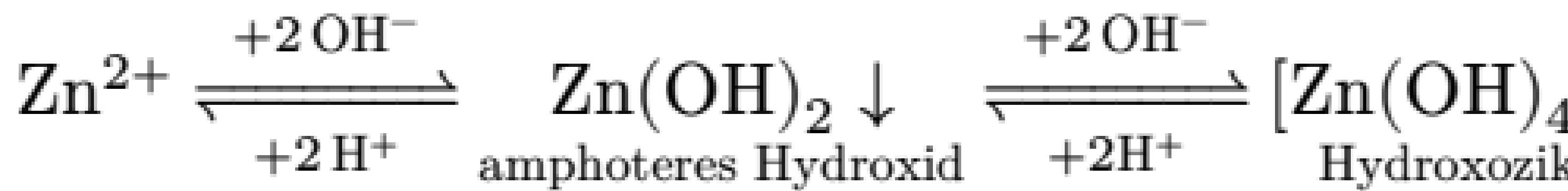


Alignments

A → B

B → C + D

More Examples



More Examples

