## Question Format & QP Setter Information

Name of Examination	Continuous Assessment Test - 2 (CAT-2), Fall 2021-22 Semester, (Dec 2021)					
Slot: D1+TD1	Course Mode: Online		Class Number(s): VL2021220106468			
Course Code:	BCHY101L	Course Title:	Engineering Chemistry			

General Instructions (if any):1. OPEN BOOK Examinations

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Q. No	Sub- divisi	Question Text	Mark s	modul e	level	СО	BL
	on		3				
1.		Arrange the carbocations of methyl, ethyl, isopropyl and t- butyl groups based on their relative stability. Explain the factors responsible for the same.	10	3	Е	1	
		(OR)					L3
		Write a comment on the hybridization, geometry and stability of methyl anion, benzyl anion, ylide, allylli anion and cyclopentadienyl anion.	10	>	a.		
		(OP)	1	1111	IJ.		L3
		(OR) Heterocyclic compounds varied in nature with respect	10	L Lie			
		to aromaticity. Write the structure and give your views on the aromatic nature of (i)aziridine (ii) azitidine (iii) pyrrole (iv) pyridine (v) azete		1)11			L3
		(OR)					
		<ul><li>(i)Simple alkyl radicals differ in terms of their stability.</li><li>Explain the contributing parameters.</li><li>(ii) Discuss any three reactions in which free radicals participate</li></ul>	5				
2.		Distinguish the working principles of electrochemical and electrolytic cells with suitable diagrams and	10	4	T	5	L3
		reactions. Discuss one application for each.					L4
		(OR)	~				
	I	Our mobile phones are built with rechargeable batteries. Delineate the structure and functioning of the batteries with suitable diagrams and reactions.	10	Н	Α		
		(OB)					L4
		Supercapacitors have high power density and longer life compared to batteries. Rationalize the difference with appropriate structural and functional features.	10				L4
		(OR)					
		Both photovoltaic cell and dye sensitized cell operate with semiconductors. (i) Outline the difference in their working principle with suitable diagrams. (ii) Give an	10				

		example of semiconducting material used in both the cells					L4
3.	a	An assignment of green coloration of a fabric is given. Suggest a dye material and describe the process.	5	3, 4	M	1, 5	
	b	How do Ruthenium-Polypyridine and I <sup>-</sup> /I <sub>3</sub> <sup>-</sup> redox couple improve the efficiency of TiO <sub>2</sub> based solar cell.	5				L3 and L4
		(OR)					
	a	If an option is given to use natural or synthetic dye for dyeing a fabric which one do you choose and why?	5				
	b	Why do we use Si based n- type and p-type semiconductors in photovoltaic cell instead of Si itself? Describe zone refining method.	5				L3 and L4
		(OR)					
	a	Explain why the reagents such as Na <sub>2</sub> CO <sub>3</sub> , NaNO <sub>2</sub> , HCl are used in methyl orange synthesis?	5				L3 and
		How does one drive energy from a fuel cell? What are	Name of Street				L3 and L4
	b	the points need to be considered to use it as a valuable energy source in automotive?	5		a.		L
	a	Provide the empirical formula of indigotin and its molecular weight. Explain why PCl <sub>5</sub> and Tn/CH-COOH are used in its synthesis.	5	Ш	ש		L3 and
	b	Zn/CH <sub>3</sub> COOH are used in its synthesis.  How is a fuel cell constructed? List out the solid	5		1		L3 and L4
		materials used as the essential components in SOFC.					

moderated by

Signature with date

## PAJAMA PADHAI