

Cleavage	Reagent
$G > A^{a*}$	DMS followed by heating at pH 7/0.1 M alkali at 90 °C
$A > G^{a*}$	DMS + acid/alkali
$C + T^a$	Hydrazine at 20 °C
C^a	Hydrazine + 2 M NaCl
G^b	DMS
$G + A^b$	Acid
$C + T^b$	Hydrazine
C^b	Hydrazine + salt
$A > C^b$	Sodium hydroxide
$G > A^b$	DMS heating at pH 7
G^c	Methylene Blue
T^c	Osmium tetroxide
$T \gg G, C^{d,e,f}$	10^{-4} M $KMnO_4$ in H_2O
C^d	$N_2H_4-H_2O$ (3:1 v/v), 5 M $N_2H_4.HOAc$
$C^{d,e}$	3 M $NH_2OH-HCl$ in H_2O , pH 6.0
$T > G \gg A, C^g$	1 M Cyclohexylamine in H_2O + UV irradiation
T^h	1 M Spermine in H_2O + UV irradiation
$G > T^h$	1 M Methylamine in H_2O + UV irradiation
T^i	0.5 M $NaBH_4$ in H_2O , pH 8–10
$T \gg C^{i,j}$	2–3 M H_2O_2 in carbonate buffer, pH 9.6
C^j	2–3 M H_2O_2 in carbonate buffer, pH 8.3 or pH 7.4
$G^{c,k}$	0.1 % Methylene Blue + visible light
$G^{l,f}$	4 % DMS in formate buffer, pH 3.5
$G \gg C^m$	0.3 % Diethyl pyrocarbonate in cacodylate buffer, pH 8 at 90 °C
$A + G^m$	0.1 % Diethyl pyrocarbonate in acetate buffer, pH 5 at 90 °C
$A + G^{n,f}$	60–80 % Aqueous formic acid
$A + G^e$	Citrate buffer, pH 4 at 80 °C
$A + G^o$	2–3 % Diphenylamine in 66 % formic acid
G^p	0.5 % DMS in 50 mM cacodylate buffer, pH 8
$A + G^p$	2 % Diphenylamine in 66 % formic acid
$C + T^p$	$N_2H_4-H_2O$ (7:4 v/v)
A^q	K_2PdCl_4 at pH 2.0