

7019

BOARD DIPLOMA EXAMINATION, (C-20) SEPTEMBER/OCTOBER-2021

DCE - FIRST YEAR EXAMINATION

ENGINEERING CHEMISTRY AND ENVIRONMENTAL STUDIES

Time: 3 hours]

PART—A

3×10=30

- Instructions: (1) Answer all questions.
 - (2) Each question carries three marks.
 - (3) Answers should be brief and straight to the point and shall not exceed five simple sentences.
 - What are fundamental particles? Name them. 1.
 - 2. Define solute, solvent and solution.
 - 3. Define Arrhenius acid and base. Give one example for each.
 - 4. What are non-electrolytes? Give examples.
 - 5. Write any three disadvantages of using hard water in industries.
 - 6. Write the names of monomers in polythene and teflon.
 - 7. Write the composition and uses of water gas.
 - 8. Write the composition and applications of vinegar.
 - 9. Define COD and BOD.
 - 10. Define producers and consumers. Give examples.

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PART—B 8×5=40

Instructions: (1) Answer all questions.

- (2) Each question carries eight marks.
- (3) Answers should be comprehensive and criterion for valuation is the content but not the length of the answer.
- 11. (a) State and explain Aufbau's principle and Hund's rule.

OR

- (b) Define ionic bond. Explain the formation of NaCl.
- 12. (a) Define normality. If 98 grams of H₂SO₄ is present in 5 litres of solution. Find the normality of solution.

OR

- (b) Explain Lewis acid-base theory.
- 13. (a) Define alloy. Write the composition and uses of (i) German silver and (ii) Nichrome.

OR

- (b) Define galvanic cell. Explain its structure and working.
- 14. (a) Define rusting of iron. Explain the mechanism of rusting of iron.

OR

- (b) Explain permutit process of removal of hardness of water.
- 15. (a) Define elastomer. Write preparation and uses of Buna-S.

OR

(b) Explain (i) greenhouse effect and (ii) acid rain.

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PART—C 10×1=10

Instruction: (1) Answer the following question that carries ten marks.

Explain addition polymerisation and condensation polymerisation.
Explain with one example each.