

1. **Which of the following best describes the isomer relationship between cis-1,2-dichlorocyclopropane and trans-1,2-dichlorocyclopropane?**
 - a) Enantiomers
 - b) Diastereomers
 - c) Constitutional isomers
 - d) Identical

2. **A molecule with the molecular formula C_4H_8 that has a cis-trans isomerism must be:**
 - a) An alkane
 - b) A cycloalkane or an alkene
 - c) An alkyne
 - d) An alcohol

3. **Assigning R/S configuration to a chiral center with the hydrogen atom pointing towards you requires:**
 - a) Directly assigning the priority
 - b) Reversing the final assignment (if R, call it S)
 - c) Imagining the molecule from the other side
 - d) It is impossible to assign

4. **Which compound is a meso form?**
 - a) 1,2-dichloropropane
 - b) 2,3-dichlorobutane
 - c) 1,3-dichlorobutane
 - d) 1,4-dichlorobutane

5. **The terms "erythro" and "threo" are traditionally used to describe:**
 - a) Enantiomers
 - b) Diastereomers with two chiral centers
 - c) Conformers
 - d) Meso compounds

6. **A molecule that is not superimposable on its mirror image is called:**
 - a) Achiral
 - b) Chiral
 - c) Meso
 - d) Symmetrical

7. **In a Fischer projection, horizontal lines represent bonds:**
 - a) bonds going away from the
 - b) bonds coming toward the viewer
 - c) In the plane of the paper
 - d) That are rotating freely

8. **The process by which a pure enantiomer is converted into a racemic mixture is called:**
- Resolution
 - Racemization
 - Inversion
 - Desymmetrization
9. **Which of the following statements about enantiomers is FALSE?**
- They have the same melting point.
 - They have the same boiling point.
 - They rotate plane-polarized light to the same extent but in opposite directions.
 - They interact identically with other chiral molecules.
10. **For a molecule with the formula C_6H_{12} that has no rings or double bonds, how many stereoisomers are possible if it contains one chiral center?**
- 1
 - 2
 - 3
 - 4
11. **The specific rotation of a pure (+)-enantiomer is $+40^\circ$. What is the percentage composition and observed rotation of a mixture with an observed specific rotation of $+20^\circ$?**
- 75% (+), 25% (-); $+20^\circ$
 - 50% (+), 50% (-); 0°
 - 25% (+), 75% (-); -20°
 - 100% (+); $+40^\circ$
12. **Which of the following statements about a meso compound is INCORRECT?**
- It has a plane of symmetry.
 - It is optically inactive.
 - It has no chiral centers.
 - It is achiral.
13. **The conversion of (R)-2-bromobutane to (S)-2-bromobutane involves:**
- Racemization
 - Inversion of configuration
 - Retention of configuration
 - Diastereomerization
14. **Which of the following molecules will exhibit optical activity?**
- Glyceraldehyde
 - meso-Tartaric acid
 - 1,1'-Bi-2-naphthol (BINOL)
 - cis-1,2-Dimethylcyclohexane

15. A molecule with 4 different chiral centers will have a maximum of how many stereoisomers?
- a) 4
 - b) 8
 - c) 16
 - d) 32
16. The separation of a racemic mixture into its enantiomers is known as:
- a) Resolution
 - b) Racemization
 - c) Regioselectivity
 - d) Distillation
17. Which descriptor is used for the configuration of chiral centers in carbohydrates and amino acids?
- a) R/S
 - b) D/L
 - c) E/Z
 - d) Cis/Trans
18. A molecule that lacks a plane of symmetry and a center of symmetry is:
- a) Definitely chiral
 - b) Definitely achiral
 - c) Possibly chiral
 - d) A meso compound
19. The stereochemical outcome of an SN1 reaction at a chiral center is:
- a) Complete inversion
 - b) Complete retention
 - c) Racemization
 - d) Diastereoselectivity
20. Which of the following is an example of axial chirality?
- a) Lactic acid
 - b) Alanine
 - c) Allene ($\text{H}_2\text{C}=\text{C}=\text{CH}_2$ with substituted ends)
 - d) *meso*-2,3-Butanediol
21. The Cahn-Ingold-Prelog priority for the following groups is:
- CH₂Br
 - CH₂Cl
 - CH₂F
 - CH₃
- a) -CH₂Br > -CH₂Cl > -CH₂F > -CH₃
 - b) -CH₃ > -CH₂F > -CH₂Cl > -CH₂Br

- c) $-\text{CH}_2\text{Br} > -\text{CH}_2\text{F} > -\text{CH}_2\text{Cl} > -\text{CH}_3$
d) $-\text{CH}_2\text{Cl} > -\text{CH}_2\text{Br} > -\text{CH}_2\text{F} > -\text{CH}_3$

22. If a molecule has an S_4 improper rotation axis, it is:

- a) Chiral
- b) Achiral
- c) Optically active
- d) A meso compound

23. The terms "syn" and "anti" are often used to describe the stereochemistry of:

- a) Aldol addition products
- b) Epoxidation products
- c) Molecules with two chiral centers
- d) All of the above

24. A sample with an enantiomeric excess (ee) of 80% of the (-)-enantiomer means the mixture contains:

- a) 80% (-) and 20% (+)
- b) 90% (-) and 10% (+)
- c) 80% racemic mixture and 20% (-)
- d) 20% (-) and 80% racemic

25. Which of the following is true for diastereomers?

- a) They have identical chemical properties in all environments.
- b) They have the same specific rotation.
- c) They have different physical properties.
- d) They are mirror images.

26. The number of stereoisomers for 1,2-dimethylcyclopropane is:

- a) 2
- b) 3
- c) 4
- d) 0

27. A chiral molecule must belong to which point group?

- a) A point group containing a plane of symmetry (σ)
- b) A point group containing an inversion center (i)
- c) A point group containing an improper axis (S_n)
- d) A point group containing only proper rotation axes (C_n)

28. A reaction that produces a single stereoisomer from an achiral starting material, without the influence of a chiral agent, is:

- a) A stereospecific reaction
- b) A stereoselective reaction

- c) An enantioselective reaction
 - d) An impossible reaction
29. **In the Fischer projection of a D-sugar, the hydroxyl group on the chiral center farthest from the carbonyl group is:**
- a) On the left
 - b) On the right
 - c) Undefined
 - d) Always equatorial
30. **The phenomenon where a molecule exists in two stereoisomeric forms that can interconvert by rotation about a single bond, but the barrier to rotation is high enough to allow isolation of the isomers, is called:**
- a) Conformational isomerism
 - b) Atropisomerism
 - c) Tautomerism
 - d) Ring-flip isomerism
31. **A molecule with the point group C_2 is:**
- a) Achiral
 - b) Chiral
 - c) A meso compound
 - d) Optically inactive
32. **. Which of the following techniques is most commonly used to determine the enantiomeric purity of a sample?**
- a) Infrared (IR) Spectroscopy
 - b) Nuclear Magnetic Resonance (NMR) with a chiral solvent/chiral shift reagent
 - c) Mass Spectrometry (MS)
 - d) Ultraviolet-Visible (UV-Vis) Spectroscopy
33. **The process where a substrate reacts with different stereoisomers of a reagent at different rates is an example of:**
- a) Steric hindrance
 - b) Stereoelectronic effects
 - c) Diastereoselectivity
 - d) Regioselectivity
34. **The descriptor for the stereochemistry of a trigonal bipyramidal molecule with five different substituents is based on:**
- a) R/S nomenclature
 - b) E/Z nomenclature
 - c) The Berry pseudorotation
 - d) The apical-equatorial priority

35. **The product of a syn addition of osmium tetroxide (OsO_4) to a *cis*-alkene is:**
- a) A meso compound
 - b) A pair of enantiomers
 - c) A pair of diastereomers
 - d) An achiral molecule
36. **A molecule that is chiral due to its overall molecular shape (helical structure), rather than a chiral center, exemplifies:**
- a) Central Chirality
 - b) Axial Chirality
 - c) Planar Chirality
 - d) Helical Chirality
37. **The stereochemical outcome of a hydroboration-oxidation reaction on an achiral alkene is:**
- a) A single enantiomer
 - b) A racemic mixture
 - c) A meso compound
 - d) A diastereomeric mixture
38. **The term "epimer" refers to:**
- a) Enantiomers that differ at every chiral center
 - b) Diastereomers that differ in configuration at exactly one chiral center
 - c) A special type of meso compound
 - d) Stereoisomers of epoxides
39. **Which of the following is the best method for determining the absolute configuration of a chiral molecule?**
- a) Boiling Point Determination
 - b) Polarimetry
 - c) X-ray Crystallography
 - d) Refractive Index Measurement
40. **. A molecule with two stereogenic centers, one of which is pseudoasymmetric, has how many stereoisomers?**
- a) 2
 - b) 3
 - c) 4
 - d) 5